



## *In vivo* hydrogen peroxide diffusivity in brain tissue supports volume signaling activity

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### ABSTRACT

Hydrogen peroxide is a major redox signaling molecule underlying a novel paradigm of cell function and communication. A role for H<sub>2</sub>O<sub>2</sub> as an intercellular signaling molecule and neuromodulator in the brain has become increasingly apparent, with evidence showing this biological oxidant to regulate neuronal polarity, connectivity, synaptic transmission and tuning of neuronal networks. This notion is supported by its ability to diffuse in the extracellular space, from source of production to target. It is, thus, crucial to understand extracellular H<sub>2</sub>O<sub>2</sub> concentration dynamics in the living brain and the factors which shape its diffusion pattern and half-life. To address this issue, we have used a novel microsensor to measure H<sub>2</sub>O<sub>2</sub> concentration dynamics in the brain extracellular matrix both in an *ex vivo* model using rodent brain slices and *in vivo*. We found that exogenously applied H<sub>2</sub>O<sub>2</sub> is removed from the extracellular space with an average half-life of  $t_{1/2} = 2.2$  s *in vivo*. We determined the *in vivo* effective diffusion coefficient of H<sub>2</sub>O<sub>2</sub> to be  $D^* = 2.5 \times 10^{-5}$  cm<sup>2</sup> s<sup>-1</sup>. This allows it to diffuse over 100 μm in the extracellular space within its half-life. Considering this, we can tentatively place H<sub>2</sub>O<sub>2</sub> within the class of volume neurotransmitters, connecting all cell types within the complex network of brain tissue, regardless of whether they are physically connected. These quantitative details of H<sub>2</sub>O<sub>2</sub> diffusion and half-life in the brain allow us to interpret the physiology of the redox signal and lay the pavement to then address dysregulation in redox homeostasis associated with disease processes.

### 1. Introduction

Hydrogen peroxide – H<sub>2</sub>O<sub>2</sub> – is a major biological oxidant which plays a pleiotropic role in redox regulation of cell functions. Its intracellular concentration is maintained in the steady-state low nanomolar range through regulation of production and efficient redundant removal systems [1]. H<sub>2</sub>O<sub>2</sub> is a strong 2-electron oxidant, but its high activation energy renders it relatively stable and limits its reaction with most biological molecules. In the brain, this oxidant has been shown to regulate neuronal polarity, connectivity, synaptic transmission in addition to tuning of neuronal networks (reviewed in Refs. [2,3]), and redox imbalance or oxidative stress are strongly associated with neurodegeneration and can undermine deleterious neuroinflammatory responses [4]. In the striatum H<sub>2</sub>O<sub>2</sub> is proposed to be released from medium spiny neurons in response to glutamate depolarization, resulting in the down-regulation of dopamine release from midbrain afferents

(reviewed in Ref. [5]).

The intracellular concentration of H<sub>2</sub>O<sub>2</sub> is maintained in low nM levels due to the activity of molecular sinks [6,7], while extracellular concentrations may be higher. Some reports point towards a blood plasma level as high as 1–5 μM [8], however careful experiments and critical analysis suggest that blood plasma levels of H<sub>2</sub>O<sub>2</sub> may be in the low nM range, with the exception of situations such as inflammation or infection [9–11]. Also, the concentration of H<sub>2</sub>O<sub>2</sub> varies in mammalian cells within different compartments: lowest [H<sub>2</sub>O<sub>2</sub>] is found in the cytosol (80 pM–2.2 nM), while inside the endoplasmic reticulum, 700 nM concentration has been reported [6,12,13]. Steep concentration gradients are expected to be found in the cytosol centered at H<sub>2</sub>O<sub>2</sub> entry or production sites [13,14]. Although H<sub>2</sub>O<sub>2</sub> can cross biological membranes through passive diffusion [15,16], sharp gradients across biomembranes were identified in the seminal work by Antunes and Cadenas [17]. Higher transport rates are afforded by several aquaporins, coined

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“peroxiporins”, shown to facilitate H<sub>2</sub>O<sub>2</sub> diffusion across biomembranes [18]. The occurrence of H<sub>2</sub>O<sub>2</sub> transport systems across the cell membranes advances the notion that the extracellular concentration of H<sub>2</sub>O<sub>2</sub> impacts dynamically on intracellular signaling pathways in neighboring cells.

Enzymes responsible for removal of H<sub>2</sub>O<sub>2</sub> include peroxiredoxins (Prx) and glutathione peroxidases (GPx), both of which display high second-order rate constants in the range of 10<sup>5</sup>–10<sup>8</sup> M<sup>-1</sup> s<sup>-1</sup> relative to H<sub>2</sub>O<sub>2</sub> [19]. Peroxiredoxins can transmit the oxidizing equivalents to other target proteins, acting as redox relays in thiol-dependent H<sub>2</sub>O<sub>2</sub> signaling [20]. Catalase (CAT) is a heme protein that, in its main catalytic cycle, catalyzes the 2-step dismutation of two H<sub>2</sub>O<sub>2</sub> molecules to water and O<sub>2</sub> with no net consumption of cellular reducing equivalents [21,22]. Finally, peroxidase enzymes use H<sub>2</sub>O<sub>2</sub> as a substrate to generate oxidants, as is the case of myeloperoxidase found in neutrophils that produces hypochlorous acid as part of the pathogen defense mechanism [23].

Inherent to the notion of H<sub>2</sub>O<sub>2</sub> as an intercellular signaling molecule is its diffusion in the extracellular space. To understand the distribution and range of influence of a particular diffusible compound in brain tissue one must know its effective diffusion coefficient (*D*<sup>\*</sup>) and the relative importance of diffusion vs clearance processes that participate in the removal and/or degradation/decomposition from the extracellular space (ECS). Diffusion in the ECS can be modified by removal across the blood brain barrier, uptake into cells or binding to receptors (concepts reviewed in Ref. [24]). Here we have investigated H<sub>2</sub>O<sub>2</sub> concentration dynamics in the brain ECS in the striatum both in an acute slice preparation and *in vivo*. To this purpose we have used novel ruthenium purple modified carbon fiber microelectrodes (RP-CFM) which we have previously designed and characterized for their electrocatalytic properties towards the selective detection of H<sub>2</sub>O<sub>2</sub> concentration with high spatial and temporal resolution [25]. We determined the half-life of H<sub>2</sub>O<sub>2</sub> in brain tissue and used mathematical modelling to determine *D*<sup>\*</sup> *in vivo*, thus inferring its diffusional spread in the brain.

## 2. Materials and methods

**Reagents and Solutions:** Reagents were analytical grade and obtained from Sigma-Aldrich. Argon was provided by Air Liquide, Portugal and Carbox (95% O<sub>2</sub>/5%CO<sub>2</sub>) was obtained from Linde Sogas, Portugal. All solutions were prepared in ultra-pure deionized water (≥18 MΩ cm) from a Milli-Q system (Millipore Company, Bedford, MA, USA). Microelectrode *in vitro* evaluations were performed in phosphate buffered saline (PBS) with the following composition (in mM): 140 NaCl, 2.7 KCl, 8.1 Na<sub>2</sub>HPO<sub>4</sub>, 1.8 KH<sub>2</sub>PO<sub>4</sub>, pH 7.4. Media for hippocampal slice experiments was artificial cerebrospinal fluid (aCSF) composed of (in mM): 124 NaCl, 2 KCl, 25 NaHCO<sub>3</sub>, 1.25 KH<sub>2</sub>PO<sub>4</sub>, 1.5 CaCl<sub>2</sub> and 10 D-glucose. For dissection and recovery, a modified aCSF was used to increase viability. Composition of this aCSF was (in mM): 124 NaCl, 2 KCl, 25 NaHCO<sub>3</sub>, 1.25 KH<sub>2</sub>PO<sub>4</sub>, 0.5 CaCl<sub>2</sub>, 10 MgSO<sub>4</sub>, 0.2 AA, 1 GSH and 10 D-glucose. In all cases, aCSF was continuously bubbled with humidified Carbox for pH buffering (pH 7.4) and oxygenation.

**Animals:** All the procedures in this study were performed in accordance with the European Union Council Directive for the Care and Use of Laboratory animals, 2010/63/EU, and were approved by the local ethics committee (ORBEA) and the Portuguese Directorate-General for Food and Veterinary. A total of 21 male Wistar rats aged 8–12 weeks and weighing 300–390 g (Charles-River Laboratories, Barcelona, Spain) were used in these experiments: 6 for slices and 15 for *in vivo*. While in the animal facility, animal husbandry conditions were as follows: housed in pairs in filter-topped type III Makrolon cages in the local vivarium with controlled environmental conditions, namely a temperature of 22–24 °C, relative humidity of 45–65%, air exchange rate of 15 times per hour, 12 h light/dark cycle, and with standard rat chow diet (4RF21-GLP Mucedola, SRL, Settimo Milanese, Italy) and chlorinated water available *ad libitum*.

**Carbon Fiber Microelectrode Fabrication, Modification and Evaluation:** Carbon fiber microelectrodes (CFM) were fabricated as previously described [26]. Briefly, a single carbon fiber (30 μm o.d.; Textron Lowell, MA, USA) was inserted into a borosilicate glass capillary (1.16 mm i.d. and 2.0 mm o.d.; Harvard Apparatus, Holliston, MA, USA) and cleaned with acetone. Each capillary was pulled using a vertical puller (Harvard Apparatus, UK) and the protruding carbon fiber was cut to a tip length of approx. 150 μm. The electrical contact between the carbon fiber and the copper wire was provided by conductive silver paint (RS, Northants, UK). The microelectrodes were tested for general recording properties in 0.05 M PBS Lite (in mM: 10 Na<sub>2</sub>HPO<sub>4</sub>, 40 NaH<sub>2</sub>PO<sub>4</sub>, and 100 NaCl, pH 7.4) by fast cyclic voltammetry at a scan rate of 200 V s<sup>-1</sup>, between -0.4 and +1.6 V vs Ag/AgCl for 30 cycles.

Ruthenium purple (RP) modification of CFM was achieved as described in Ref. [25]. Briefly, a solution of 1 mM K<sub>4</sub> [Ru<sup>II</sup>(CN)<sub>6</sub>] prepared in 35 mM KCl (N<sub>2</sub>-purged) was mixed, under vigorous stirring, with a solution of 1 mM FeCl<sub>3</sub> in 35 mM KCl (N<sub>2</sub>-purged). The pH of the resulting colloidal suspension was adjusted to 2.0 using 1 M HCl. This solution was placed in an ultrasound bath and RuCl<sub>3</sub> was added from a 1 mM solution to give a final concentration of 20 μM. This solution was used within 4 h of preparation. Electrodeposition was carried out by potential cycling between -0.2 and +1.0 V vs Ag/AgCl at a scan rate of 50 mV s<sup>-1</sup>. The cycles were repeated until the redox peaks no longer increased in height. The modified CFM (CFM-RP) was rinsed with distilled water and placed in a solution of 1 mM RuCl<sub>3</sub> in 35 mM KCl. Between 2 and 4 further cycles were performed under the same conditions to ensure the stability of the deposited film. For coating with Nafion®, the tip of the CFM-RP was dipped into a 5% solution of Nafion® in aliphatic alcohols for 5 s and then dried at 100 °C for 15 min. These sensors were designated CFM-RP-Nafion®.

**Preparation of Rat Striatum Slices:** Following decapitation under deep anesthesia (isoflurane), the brain of the animal was rapidly removed and placed in ice-cold, Carbox bubbled modified aCSF. The cerebellum was removed, the two hemispheres were separated and mounted in the pre-chilled stage of a vibratome (Vibroslice, World Precision Instruments) and submerged in the chamber filled with ice cold modified aCSF continuously bubbled with Carbox. Brain slices containing the striatum were obtained with a thickness of 400 μm and transferred to a pre-incubation chamber (BSC-PC; Harvard Apparatus) filled with modified aCSF. Slices were allowed to recover under these conditions for at least 1 h prior to recording.

**Recording H<sub>2</sub>O<sub>2</sub> in Rat Striatum Slices:** An individual slice was placed in a recording chamber (BSC-BU with BSC-ZT top, Harvard Apparatus) and perfused with aCSF continuously bubbled with humidified Carbox and at a flow rate of 1.5–2 mL min<sup>-1</sup>. The temperature of the chamber was maintained at 32 °C (temperature controller model TC-202A, Harvard Apparatus). A recording array comprised of a CFM-RP-Nafion® and a CFM mounted at a tip-to-tip distance of 100 μm was lowered into the tissue with the aid of a micromanipulator so as to guarantee that the totality of the active surface was in the tissue core. Recording of the reduction current was initiated and once a stable baseline was obtained, an H<sub>2</sub>O<sub>2</sub> solution (1 mM in aCSF) was pressure ejected (10 Psi, 1s) via a Picospritzer II (General Valve, Fairfield, NJ) through a pulled micropipette, placed between the two working electrodes.

### 2.1. *In vivo* recording of hydrogen peroxide in the brain of anesthetized rats

The experimental setup used for amperometric monitoring of H<sub>2</sub>O<sub>2</sub> *in vivo* in anesthetized rats was similar to that used in previous studies [27]. Briefly, an animal was anesthetized with urethane 1.25 g kg<sup>-1</sup> (i.p.) and placed in a stereotaxic apparatus. Body temperature was maintained at 37 °C with a heated pad coupled to a Gaymar Heating Pump (Braintree Scientific, Inc., USA). The skull was exposed by a midline scalp incision and retraction of the skin and temporal muscle. Bleeding was

controlled using a Bovie® cautery. A craniotomy was made over the cerebral cortex with an area of roughly  $7 \text{ mm}^2$  (AP: +3.0 to +0.5 mm; ML:  $\pm 1$  to  $\pm 4$  mm; relative to bregma [28]) with removal of the overlying meninges. An additional small burr-hole was drilled in a site remote from the recording area for the insertion of a miniature Ag/AgCl reference electrode in the subdural space. The cortical surface was maintained wet with saline soaked cotton balls. An array comprising a CFM-RP-Nafion® and a pulled micropipette (average distance of  $75 \pm 10 \mu\text{m}$ ) was lowered into the dorsal medial striatum (AP: +1.5–2.0 mm; ML:  $\pm 1.5 - \pm 2.5$  mm and DV:  $-4.0$  to  $-6.0$  mm relative to bregma). After insertion of the array into the brain, the baseline was allowed to stabilize for at least 20 min. The mean display frequency was set at 4–20 Hz. An  $\text{H}_2\text{O}_2$  solution (1 mM in aCSF) was pressure ejected via a Picospritzer II (General Valve, Fairfield, NJ) through a pulled micropipette, with a volume dispensed of 3–125 nL. In experiments in which cardiac arrest was performed during recording session, the rat received a lethal dose of euthanasia solution (0.5 mL of sodium pentobarbital  $200 \text{ g mL}^{-1}$ , i.p.).

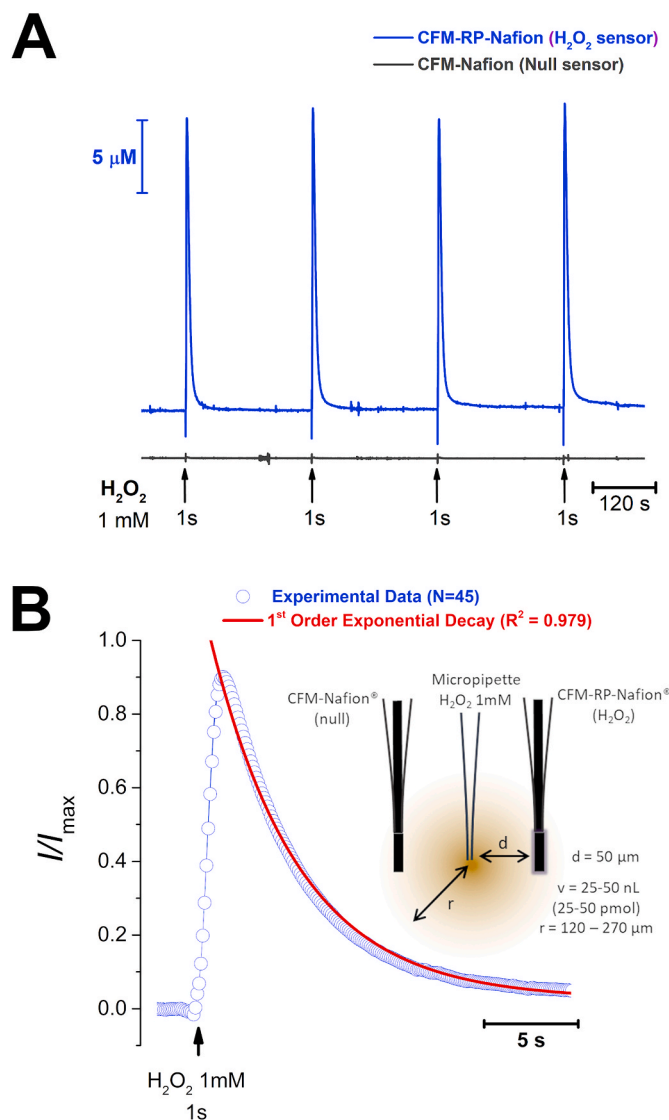
**Electrochemical Instrumentation:** Electrochemical procedures including RP electrodeposition and CFM evaluation by FCV were performed using a Multi PalmSens4 Potentiostat (PalmSens, The Netherlands) controlled by MultiTrace v.4.2 software (PalmSens, The Netherlands). A 3-electrode electrochemical cell was used, comprised of the working electrode, an Ag/AgCl in 3 M NaCl reference electrode (RE-5B, BAS Inc., IN, USA), and a Pt wire as auxiliary electrode. Amperometric recording in slices and calibration in the slice recording chamber were performed using a FAST16mkII potentiostat (Quanteon, KY, USA) in a 2-electrode electrochemical cell configuration comprised of the working electrode and an Ag/AgCl pellet reference electrode. Amperometric recording *in vivo* in the anesthetized rodent brain was performed using a FAST16mkIII potentiostat (Quanteon, KY, USA) in a 2-electrode electrochemical cell comprised of a working electrode and an Ag/AgCl miniature pseudo reference electrode. The latter was produced by anodization of the exposed tip of the Teflon-coated Ag wire in 1 M HCl saturated with NaCl, which, when in contact with cerebrospinal fluid in the brain containing chloride ions, develops an Ag/AgCl half-cell. Both FAST16 potentiostats were controlled by FAST2014 software (Quanteon, KY, USA). Working potential for monitoring  $\text{H}_2\text{O}_2$  was set at  $-0.2 \text{ V}$  vs Ag/AgCl [25].

**Data Analysis.** Data analysis was performed using MultiTrace v.4.2, FAST Analysis version 6.0, OriginPro 2016, and GraphPad 5.0. Values are given as the mean  $\pm$  SEM unless otherwise stated. Normality of data was confirmed using the D'Agostino and Pearson omnibus normality test ( $\alpha = 0.05$ ). Calculated parameters were statistically evaluated by using One-way ANOVA followed by Dunnett post-hoc test. The number of repetitions is indicated in each individual determination. The sensitivity of CFM-RP-Nafion® toward  $\text{H}_2\text{O}_2$  reduction was determined by linear regression analysis in the range of 0–50  $\mu\text{M}$ . The diffusion of  $\text{H}_2\text{O}_2$  from a spherical source was modeled using OriginPro 2016, based on equation (1), which describes diffusion from a spherical source [30].

### 3. Results

#### 3.1. Uptake and removal of extracellular Hydrogen Peroxide in rat brain striatum slices

We evaluated  $\text{H}_2\text{O}_2$  concentration dynamics in rat striatum slices using previously described CFM-RP-Nafion® sensors [25] in order to investigate how brain tissue handles an increase in extracellular  $\text{H}_2\text{O}_2$ . To this purpose, we constructed an array composed of a  $\text{H}_2\text{O}_2$  sensor and a null sensor with a tip-to-tip distance of ca.  $100 \mu\text{m}$ . This array was lowered into the tissue core, in the dorsal-medial region of the striatal slice. A micropipette with a tip diameter of 10–20  $\mu\text{m}$  and filled with a 1 mM  $\text{H}_2\text{O}_2$  solution was placed between the 2 sensors and a volume of solution (25–50 nL) was pressure ejected for a period of 1 s, as depicted in the inset of Fig. 1B. As can be observed in Fig. 1A, repeated  $\text{H}_2\text{O}_2$



**Fig. 1.** - A - Recording of  $\text{H}_2\text{O}_2$  concentration dynamics in striatal slices evoked by exogenous local application of a 1 mM  $\text{H}_2\text{O}_2$  solution. Blue trace represents the amperometric current measured at the CFM-RP-Nafion®  $\text{H}_2\text{O}_2$  sensor while the grey trace represents the current recorded at a null sensor. Repeated application of  $\text{H}_2\text{O}_2$  through a micropipette placed between the 2 sensors produced transient and reproducible increase in  $\text{H}_2\text{O}_2$  in the brain tissue. B - Average of 44 individual normalized signals (blue) and first order exponential decay function (red). Inset is a schematic representation of the experimental design used, including the microelectrodes and ejection pipette between the two, where  $d$  is the distance between the micropipette tip and each of the sensors,  $v$  is the volume of the  $\text{H}_2\text{O}_2$  solution ejected from the micropipette and  $r$  is the calculated radius of the bolus of  $\text{H}_2\text{O}_2$  ejected. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

ejections (3 min inter-pulse interval) resulted in transient increases in  $\text{H}_2\text{O}_2$  in the tissue recorded at the CFM-RP-Nafion® sensor, with no change in the current recorded at the null sensor.

The decay phase of the transient  $\text{H}_2\text{O}_2$  signal was suitably fitted to a 1<sup>st</sup> order exponential decay function, shown as the superimposed red line over the experimental data in blue circles in Fig. 1B. From this fitting we determined the time constant of the signal decay,  $k = 0.32 \pm 0.02 \text{ s}^{-1}$ , as well as the decay half-time,  $t_{1/2} = 2.5 \pm 0.1 \text{ s}$  ( $N = 44$  signals, obtained from 6 slices).

In order to discriminate the role of brain tissue tortuosity vs intracellular processes governing  $\text{H}_2\text{O}_2$  uptake and breakdown, we repeated

the above experiment in a matrix mimicking tissue tortuosity but devoid of cellular components. We obtained 400  $\mu\text{m}$  slices from a 0.2% agarose block [29,30], which were placed in the slice recording chamber and evaluated  $\text{H}_2\text{O}_2$  signal obtained as with striatum slices. As summarized in Table 1, the recorded  $\text{H}_2\text{O}_2$  signals showed slower decay kinetics and the time constant of  $\text{H}_2\text{O}_2$  decay was significantly decreased as compared to that found in striatal slices ( $k = 0.11 \pm 0.01 \text{ s}^{-1}$ ) with increase in the signal half-time ( $t_{1/2} = 6.5 \pm 0.3 \text{ s}$ ,  $N = 5$ ,  $p < 0.0001$ ). Additionally, we determined the  $\text{H}_2\text{O}_2$  signal profile in metabolically compromised striatum slices perfused in 0 mM Glucose, 10 mM Saccharose and 10 mM  $\text{CN}^-$ . Under these conditions, the  $\text{H}_2\text{O}_2$  signal was similar to that observed in agarose slices and the decay time constant ( $k = 0.12 \pm 0.01 \text{ s}^{-1}$ ) and signal half-time ( $t_{1/2} = 5.8 \pm 0.5 \text{ s}$ ) were significantly changed as compared to control tissue ( $p < 0.0001$ ). These results support a role for cellular components in shaping extracellular  $\text{H}_2\text{O}_2$  concentration dynamics.

### 3.2. Hydrogen Peroxide removal In Vivo in the rat striatum

Slice preparations preserve tissue cytoarchitecture and allow us to understand the role of neural cells (neurons and astrocytes) in modulating the concentration dynamics of diffusible species such as  $\text{H}_2\text{O}_2$ . However, they are devoid of functional vasculature. Hence, we investigated  $\text{H}_2\text{O}_2$  signal profiles *in vivo* in the striatum of anesthetized rats. Similar to what was previously observed in striatal slices, local pressure ejection of a 1 mM solution resulted in a transient increase in  $\text{H}_2\text{O}_2$  recorded at the CFM-RP-Nafion®  $\text{H}_2\text{O}_2$  sensor inserted in the dorsal medial striatum *in vivo* in the anesthetized rat brain and repetitive ejections produced reproducible signals *in vivo* (Fig. S1).

The decay phase of the signal was well adjusted to a 1<sup>st</sup> order exponential decay function (Fig. 2A) and average time constant was determined to be  $k = 0.34 \pm 0.01 \text{ s}^{-1}$ , corresponding to an average  $t_{1/2} = 2.2 \pm 0.1 \text{ s}$  ( $N = 121$  signals, obtained from 12 recording sessions, Table 1), similar to that observed in slices. Furthermore, the average peak concentration of  $\text{H}_2\text{O}_2$  measured at the detector was  $30.5 \pm 2.5 \mu\text{M}$ . We repeated this paradigm using a higher concentration of  $\text{H}_2\text{O}_2$  (10 mM) and observed that the average time constant significantly decreased to  $k = 0.20 \pm 0.02 \text{ s}^{-1}$  ( $N = 43$  signals, obtained from 3 recording sessions, Table 1) and the average  $\text{H}_2\text{O}_2$  peak concentration was increased to  $105.9 \pm 30.4 \mu\text{M}$ . This result suggests that for 10 mM  $\text{H}_2\text{O}_2$ , the Prx pool becomes partially hyperoxidized [31], which might hinder its participation in  $\text{H}_2\text{O}_2$  removal, resulting in slower  $\text{H}_2\text{O}_2$  clearance by the tissue. For the remainder of the experiments we used a 1 mM  $\text{H}_2\text{O}_2$  solution.

Induction of cardiac arrest resulted in rapid loss of spontaneous brain wave activity accompanied by a gradual decrease of the  $k$  values

**Table 1**

Comparison of decay time constant ( $k$ ) and half-life ( $t_{1/2}$ ) of  $\text{H}_2\text{O}_2$  signals obtained in response to puff ejections of a 1 mM  $\text{H}_2\text{O}_2$  solution, unless otherwise indicated. Data were obtained in *ex vivo* acute striatal slices and 0.2% agarose slices, and *in vivo* in the dorsal medial striatum in anesthetized rat and in a 0.2% agarose block at 37 °C. Values represent mean  $\pm$  SEM. \* $p < 0.0001$  as compared to control slices and # $p < 0.001$  as compared to *in vivo* control (1 mM  $\text{H}_2\text{O}_2$ ).

Experimental Model (N)		$k / \text{s}^{-1}$	$t_{1/2} / \text{s}$
<i>Ex vivo</i> - Striatum Slices	Control (44)	$0.32 \pm 0.02$	$2.5 \pm 0.1$
	MC (5)	$0.12 \pm 0.01^*$	$5.8 \pm 0.5^*$
	Agarose 0.2% (5)	$0.11 \pm 0.01^*$	$6.5 \pm 0.3^*$
<i>In vivo</i> - Dorsal Medial Striatum	Control (121)	$0.34 \pm 0.01$	$2.2 \pm 0.1$
	10 mM $\text{H}_2\text{O}_2$ (42)	$0.23 \pm 0.02^\#$	$3.5 \pm 0.2^\#$
	A.D. (19)	$0.21 \pm 0.03^\#$	$4.8 \pm 0.7^\#$
	Agarose 0.2%	$0.04 \pm$	$18.6 \pm$
	(26)	$0.004^\#$	$1.2^\#$

MC – metabolically compromised slices, perfused with 0 mM glucose, 10 mM saccharose and 10 mM  $\text{CN}^-$ .

A.D. – after death evoked by lethal dose of pentobarbital.

(Fig. 2B). As summarized in Table 1, the average time constant value stabilized at  $k = 0.21 \pm 0.03$  ( $N = 19$ ), a value significantly lower when compared to the control situation, but similar to that observed in slices following metabolic poisoning ( $p = 0.87$ ) and in a 0.2% agarose block ( $p = 0.83$ ) used as a mimetic of brain tissue tortuosity. The onset of decrease in  $k$  values coincided with the decrease in power observed in the spectrogram, known to reflect brain wave activity [32–34].

### 3.3. Estimation of Hydrogen Peroxide diffusion coefficient in vivo in the striatum

In order to determine *de in vivo* effective diffusion coefficient ( $D^*$ ) of exogenously applied  $\text{H}_2\text{O}_2$  in striatum tissue, the experimental data were input to Eq. (1) where  $C$  and  $C_0$  represent the instantaneous and initial concentration,  $r$  is the radius of the sphere of  $\text{H}_2\text{O}_2$  ejected,  $d$  is the distance between source (pipette) and recorder ( $\text{H}_2\text{O}_2$  sensor),  $t$  is time,  $D^*$  the effective diffusion coefficient and  $\lambda$  represents the inactivation constant.

$$C = C_0 e^{-\lambda t} \left[ \frac{1}{2} \left( \text{erf} \left( \frac{r+d}{2\sqrt{D^*t}} \right) - \text{erf} \left( \frac{r-d}{2\sqrt{D^*t}} \right) \right) - \frac{1}{d} \sqrt{\frac{D^*t}{\pi}} \left( \frac{e^{-\frac{(r-d)^2}{4D^*t}}}{e^{4D^*t}} - e^{-\frac{(r+d)^2}{4D^*t}} \right) \right] \quad (1)$$

The experimental design was adjusted to guarantee  $r < d$ , as illustrated in Fig. 3B. Recordings were obtained both *in vivo*, in the striatum of anesthetized rats and from a 0.2% agarose gel block at 37 °C. As shown in Fig. 3A, in both cases we observed a good adjustment between the experimental data (open circles) and the fit to Eq. (1) (line).

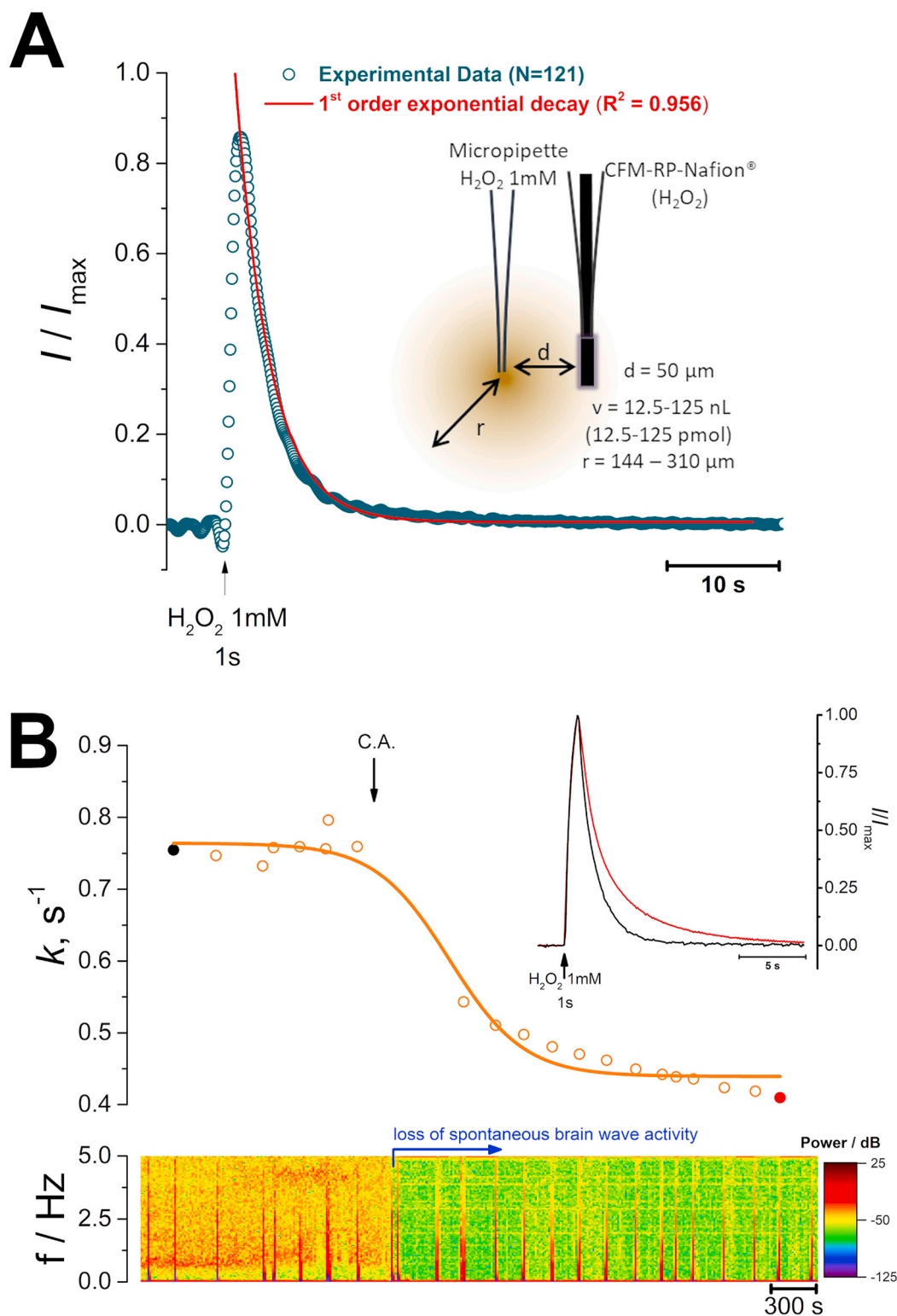
We determined the  $D^*$  *in vivo* of  $\text{H}_2\text{O}_2$  to be  $(2.5 \pm 0.1) \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$  with an average inactivation constant of  $\lambda = 0.19 \pm 0.01 \text{ s}^{-1}$  ( $N = 57$  obtained from 5 recording sessions). Interestingly, this value of  $D^*$  was found to be significantly higher than that calculated from data obtained in agarose gel, where we observe  $D^*_{\text{agarose}} = (1.7 \pm 0.1 \times 10^{-5}) \text{ cm}^2 \text{ s}^{-1}$  and an inactivation constant of  $\lambda = 0.005 \pm 0.002 \text{ s}^{-1}$  ( $N = 38$ ).

From the Einstein-Smoluchowski Eq. (2) one can define the average square of the distance travelled ( $\overline{x^2}$ ) by an individual molecule as a function  $D$  and  $t$  [35]. As can be appreciated in Fig. 4,  $\text{H}_2\text{O}_2$  can be classified in the same range as small volume neurotransmitters such as nitric oxide ( $\text{*NO}$ ), displaying a relatively large displacement within its  $t_{1/2}$  (112  $\mu\text{m}$ ) when compared to other neurotransmitters such as dopamine or glutamate (properties summarized in Table 2).

$$\overline{x^2} = 2Dt \quad (2)$$

## 4. Discussion

Interest in monitoring  $\text{H}_2\text{O}_2$  in brain tissue extracellular space has grown over the past years as a result of increasing recognition of its involvement not only in the etiology of neurodegenerative disorders such as Parkinson's and Alzheimer's diseases [39], but also due to the appreciation of its putative role as a diffusible neuromodulator, analogous to  $\text{*NO}$  [5,40]. Recent advances in the identification of membrane  $\text{H}_2\text{O}_2$  transporters, beyond the passive diffusion through membranes, support that the extracellular  $\text{H}_2\text{O}_2$  dynamics might modulate intracellular redox signaling pathways in cells localized within the volume of influence of  $\text{H}_2\text{O}_2$ . Here, we have investigated the extracellular concentration dynamics of  $\text{H}_2\text{O}_2$  in the striatum, a brain region where  $\text{H}_2\text{O}_2$  has been proposed to act as an intercellular messenger molecule, modulating dopamine release from midbrain afferent fibers [41,42], focusing on determining  $\text{H}_2\text{O}_2$  half-life time in brain tissue in an *ex vivo* model (striatal slices) and *in vivo* in the striatum of anesthetized rats. In addition, we have determined the effective diffusion coefficient of  $\text{H}_2\text{O}_2$  in brain tissue *in vivo*. We used an artificial peroxidase type electrochemical microsensor [25] to monitor real-time  $\text{H}_2\text{O}_2$  concentration dynamics in the brain extracellular space with high sensitivity and selectivity towards  $\text{H}_2\text{O}_2$  and high spatial and temporal resolution

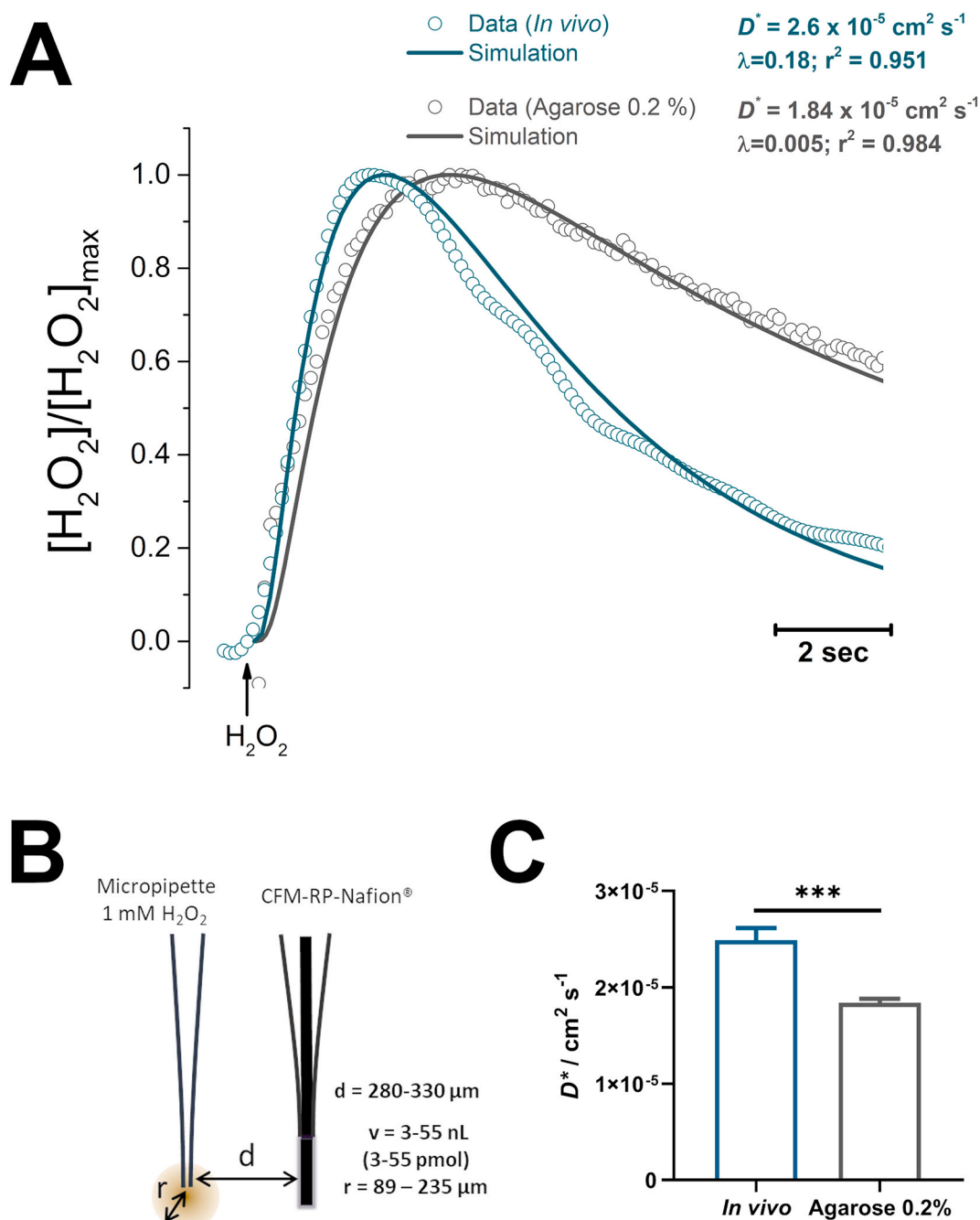


**Fig. 2.** – **A** – Recording of  $\text{H}_2\text{O}_2$  concentration dynamics *in vivo* in the dorsal medial striatum (AP: +1.8, ML: –2.0; DV: –4.5 mm) evoked by exogenous local application of a  $\text{H}_2\text{O}_2$  1 mM solution. Fitting of experimental data to a 1<sup>st</sup> order decay function (red line). Inset is a schematic representation of the experimental design used, including the microelectrodes and ejection pipette between the two, where  $d$  is the distance between the micropipette tip and the CFM-RP-Nafion<sup>®</sup> sensor,  $v$  is the volume of the  $\text{H}_2\text{O}_2$  solution ejected from the micropipette and  $r$  is the calculated radius of the bolus of  $\text{H}_2\text{O}_2$  ejected. **B** – Representative trace of the observed change in time constant value before and after induction of cardiac arrest as a function of time (top). Inset are superimposed the first (black) and last (red) signals. On the bottom is the power spectrum of the 10 Hz amperometric recording, showing the marked loss in power following cardiac arrest, indicative of loss of spontaneous brain wave activity. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

resulting from the reduced sensor size (o.d.  $33\ \mu\text{m}$ ) and the fast-sampling ability of amperometry. In both experimental models, we coupled a micropipette to the microsensors to apply a controlled pressure ejection of a nL volume of a  $\text{H}_2\text{O}_2$  solution at a known distance from the detector. Through mathematical fitting of the data, to the best of our knowledge, we provide the first *in vivo* data regarding the effective or apparent diffusion coefficient of  $\text{H}_2\text{O}_2$  in the living brain.

#### Hydrogen Peroxide is Rapidly Removed from the Extracellular

**Space by Metabolically Active Cells.** In the first set of experiments, we investigated the extracellular concentration dynamics of  $\text{H}_2\text{O}_2$  in an *ex vivo* model of acute striatum slices. This experimental model has the advantage of preserving the brain tissue cytoarchitectural integrity, namely connectivity and intercellular communication. Because cerebrovascular blood flow is absent in this model, it allows us to better understand the role of neural cells (astrocytes and neurons) without the contribution of blood flow. Our data revealed that a puff application of a

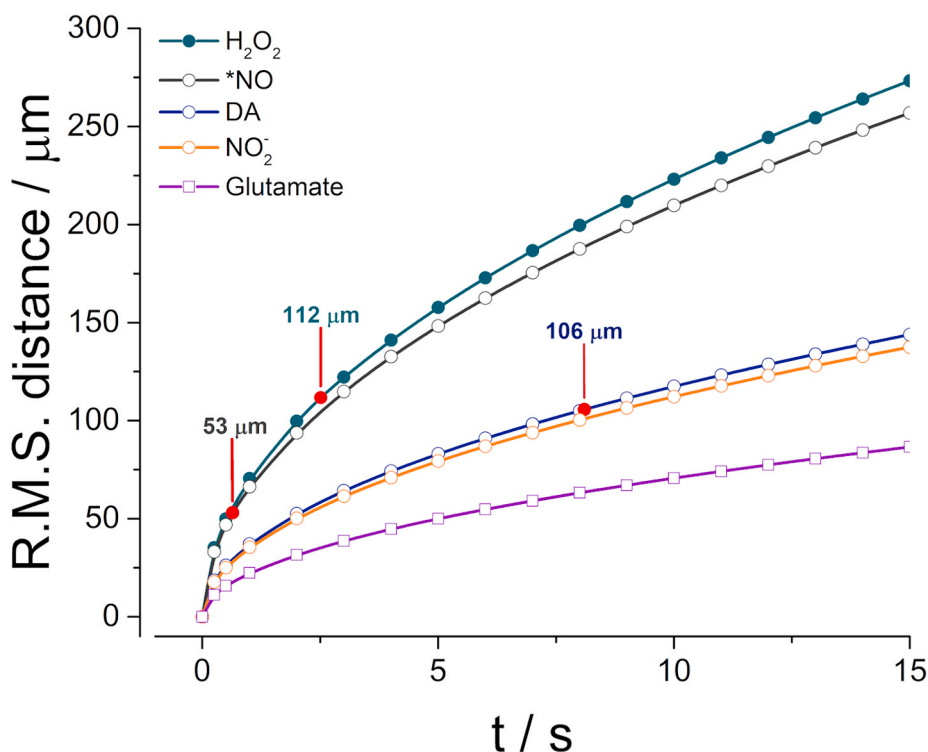


**Fig. 3.** – **A** - Representative recordings of  $\text{H}_2\text{O}_2$  concentration dynamics obtained *in vivo* in the striatum (blue-green) and in a 0.2% agarose block at 37 °C (grey). Circles represent experimental data while lines correspond to the theoretical trace obtained by modelling expression (1). **B** - Experimental design used, comprising a micropipette filled with a 1 mM  $\text{H}_2\text{O}_2$  solution coupled to a CFM-RP-Nafion® sensor for  $\text{H}_2\text{O}_2$  detection. Interval of distance ( $d$ ) between source and detector as well as for the volume and radius of applied  $\text{H}_2\text{O}_2$  bolus are discriminated. As drawn,  $d = 300 \mu\text{m}$  and  $r = 100 \mu\text{m}$ . **C** - Effective diffusion coefficient values calculated by modelling experimental data with Eq. (1). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

$\text{H}_2\text{O}_2$  solution in the proximity of the detector resulted in a transient increase in  $\text{H}_2\text{O}_2$  concentration, revealing that extracellular  $\text{H}_2\text{O}_2$  diffuses away from the point of origin and is rapidly removed. Removal of exogenously applied  $\text{H}_2\text{O}_2$  was found to be the result of intracellular factors, as the decay was significantly slower in a 0.2% agarose slice. Agarose has been widely used to mimic brain tissue tortuosity and to evaluate its effect on the diffusion of chemical species [29,30]. In line with this observation, in metabolically compromised striatal slices perfused with  $\text{CN}^-$  and 0 mM glucose as a strategy to hinder energy metabolism,  $\text{H}_2\text{O}_2$  decay was found to be similar to that observed in agarose. Taken together, these results support the notion that

extracellular  $\text{H}_2\text{O}_2$  enters cells, namely neurons and/or astrocytes, and is removed in metabolically competent cells, presumably through the activity of enzymatic systems such as GPx, Prx and CAT.

Similarly, *in vivo*, we also observed that puff application of an  $\text{H}_2\text{O}_2$  solution resulted in a transient increase in extracellular  $\text{H}_2\text{O}_2$ , with a decay rate superior to that found in a 0.2% agarose block. Following cardiac arrest, the decay rate of the signal decreased progressively, but remained above that found in agarose. We speculate that this is the result of  $\text{H}_2\text{O}_2$  removal by erythrocytes, namely due to the activity of catalase which does not consume redox equivalents when degrading  $\text{H}_2\text{O}_2$  [43]. While in neurons and astrocytes, catalase is mainly found in



**Fig. 4.** Calculated root-mean-square (R.M.S.) distance of random movement for  $\text{H}_2\text{O}_2$  using Eq. (2) and considering  $D^*$  values calculated in present study. For comparative purposes we have also plotted data for  $\text{*NO}$  (grey), dopamine (DA, blue),  $\text{NO}_2^-$  (orange) and glutamate (purple). Red circles represent distance travelled by each analyte for their respective extracellular half-life time ( $t_{1/2}$ ). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

**Table 2**

Comparative analysis of half-life and apparent diffusion coefficient for  $\text{H}_2\text{O}_2$  and other species acting as volume transmission molecules ( $\text{*NO}$  and dopamine) as well as the fast-excitatory neurotransmitter glutamate. Nitrite has been included to show the effect of charge on a small molecule.

Analyte	Experimental Model	$t_{1/2}$ (s)	$D^*$ ( $\times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$ )	Ref
$\text{H}_2\text{O}_2$	<i>In vivo</i> in the rat striatum	2.2	2.5	Present Study
$\text{*NO}$	<i>In vivo</i> in the rat cortex	0.6	3.3	[36]
Dopamine	Midbrain slices from Guinea pig	8.1	0.69	[37]
Nitrite	<i>In vivo</i> in the rat cortex	n.d.	0.63	[36]
Glutamate	Mathematical Modelling	n.d.	0.25	[38]

peroxisomes and thus less accessible to exogenously derived  $\text{H}_2\text{O}_2$ , erythrocytes do not have membrane organelles and CAT is found in the cytosol. Even after cardiac arrest, erythrocytes present in blood vessels may, temporarily, contribute to  $\text{H}_2\text{O}_2$  removal in brain tissue.

#### Half-Life of Hydrogen Peroxide in the Brain Extracellular Space:

The average half-life of extracellular  $\text{H}_2\text{O}_2$  in slices and *in vivo* were 2.5 and 2.2 s respectively, values that are not statistically different from each other. Neuromodulators such as  $\text{*NO}$  have been shown to have a half-life time of 0.42 and 0.75 s in cortex and hippocampus, respectively [44] while dopamine is reported to have  $t_{1/2}$  values ranging from 7.4 to 8.1 s [37]. The removal of each one of the messengers is governed by different biochemical and cellular pathways.  $\text{*NO}$  inactivation is tightly coupled to its facile reaction with hemoglobin in circulating erythrocytes [44,45].  $\text{*NO}$  produced in response to glutamatergic neurotransmission evokes vasodilation, increasing local blood flow and the concentration of hemoglobin [46,47]. Additionally,  $\text{*NO}$  does not require transporters to cross biomembranes: due to its hydrophobic nature, the free radical readily crosses phospholipid bilayers or uses them as fast diffusion highways [36,48]. Dopamine clearance from the extracellular space depends on the activity of the neuronal dopamine transporter (DAT), a member of the neurotransmitter sodium symporter

protein superfamily (reviewed in Ref. [49]). While  $\text{H}_2\text{O}_2$  can, to a limited extent, cross biomembranes by simple diffusion, it is currently accepted that it can gain rapid entry into the cells through water channels – aquaporins (AQP). These are membrane channels that facilitate the passage of water and other non-charged solutes across biological membranes down their concentration gradients [50]. Within the group of channels, those that display permeability for  $\text{H}_2\text{O}_2$  (AQP3, AQP8 and AQP9) have been coined peroxiporins [51–53]. Considering the large concentration gradient existing between extra- and intracellular compartments, one can expect any increase in extracellular  $\text{H}_2\text{O}_2$  to rapidly enter the cell, where it is rapidly decomposed by GPx, Prx or CAT. The existence of peroxiporins helps to understand the smaller  $t_{1/2}$  of  $\text{H}_2\text{O}_2$  as compared to dopamine, for example.

**Effective Diffusion Coefficient and Inactivation Constant of Hydrogen Peroxide in Brain Tissue:** To operate as an intercellular signaling molecule,  $\text{H}_2\text{O}_2$  must diffuse in the extracellular space. We have previously investigated the diffusion of  $\text{*NO}$  in brain tissue, using both experimental-based and mathematical modelling strategies to better understand its concentration dynamics in the brain [36,44,54]. We adjusted data obtained in a 0.2% agarose block and *in vivo* using a previously described mathematical model for diffusion of a chemical species from a spherical source (equation (1) [55]) to determine and compare the effective diffusion coefficient of  $\text{H}_2\text{O}_2$  ( $D^*$ ) and its inactivation constant ( $\lambda$ ). Data obtained in agarose revealed an average diffusion coefficient of  $1.7 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$ , in line with that reported for aqueous buffer of  $1.8 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$  [56]. Additionally, the fact that the inactivation constant in agarose is close to zero ( $\lambda = 0.005$ ) corroborated that inactivation mechanisms did not contribute to shaping of  $\text{H}_2\text{O}_2$  diffusion in agarose, as expected. Modeling data obtained *in vivo* revealed a  $\lambda$  value of  $\lambda = 0.19 \text{ s}^{-1}$ , which is a range similar to the decay constant calculated from the exponential fit of data from slices and *in vivo* (ca.  $0.3 \text{ s}^{-1}$ ), implying that inactivation plays an important role in shaping extracellular concentration dynamics.

We found the *in vivo* effective diffusion coefficient of  $\text{H}_2\text{O}_2$  to be slightly higher than that observed in agarose, assuming an average value of  $D^* = 2.5 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$ . This is a curious observation, as one would expect to observe hindrance of diffusion in the extracellular space

as compared to free diffusion in buffer (reported to be  $D = 1.8 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$  at  $37^\circ \text{C}$  [56]).

**Hydrogen Peroxide is a Volume Transmission Signaling Molecule:** From Eq. (2) we can observe that the mean distance a single molecule of  $\text{H}_2\text{O}_2$  can travel *in vivo* in the brain is similar to that we have previously described for the highly diffusible intercellular messenger  $\bullet\text{NO}$  [36] and much higher than what is expected for neuromodulators such as dopamine or a fast synaptic neurotransmitter such as glutamate. Interestingly, because the inactivation constant for  $\text{H}_2\text{O}_2$  is lower than that for  $\bullet\text{NO}$ , which is rapidly removed by rapid reaction with oxyhemoglobin ( $k = 3.4 \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$  [57]),  $\text{H}_2\text{O}_2$  may in fact diffuse larger distances in the brain tissue.

Considering this, we can tentatively place  $\text{H}_2\text{O}_2$  within the class of volume transmitters in the brain, of which  $\bullet\text{NO}$  is a paradigmatic example. Volume transmission is a form of intercellular communication in which the signaling molecule diffuses in the extracellular space, connecting all cell types in the brain, in what Agnati and Fuxe have coined “complex cellular networks” [58,59]. The diffusion or migration of the signal is terminated when the signaling molecule is removed by enzymes, cleared by brain capillaries or taken up by transporters. In support of this notion, several cell studies support the role of  $\text{H}_2\text{O}_2$  as a neuromodulator in the brain. Micromolar levels of  $\text{H}_2\text{O}_2$  (1–50  $\mu\text{M}$ ) have been shown to modulate synaptic plasticity by inducing the release of  $\text{Ca}^{2+}$  from intracellular stores which can then activate a myriad of  $\text{Ca}^{2+}$ -dependent proteins and alter neuronal  $\text{Ca}^{2+}$  permeability [3]. Additionally,  $\text{H}_2\text{O}_2$  has been shown to be important in modulating dopamine release in the striatum in a signaling pathway that includes glutamate-evoked  $\text{H}_2\text{O}_2$  production (presumably in mitochondria) and diffusion from medium spiny neurons to modulate dopamine release from midbrain axons via opening of  $\text{K}_{\text{ATP}}$  channels [41,42,60]. The present work consolidates these findings, providing quantitative and direct bases on how extracellular  $\text{H}_2\text{O}_2$  diffuses in the extracellular space and highlighting the major contributors towards shaping its concentration dynamics *in vivo*.

## 5. Conclusions

Hydrogen peroxide has been established as a ubiquitous redox signaling molecule in a new paradigm of cell communication. However, a fine line separates oxidative eustress (physiological signaling) from and oxidative distress (disease processes). In the present work, using a novel sensing strategy based on an artificial-peroxidase amperometric microsensor, we have investigated the quantitative determinants of the extracellular concentration dynamics of  $\text{H}_2\text{O}_2$  in the brain tissue. Our data reveal the *in vivo* effective diffusion coefficient and half-life of  $\text{H}_2\text{O}_2$ , supporting a role as a volume signaling molecule in the striatum. The concept of the emerging field of redox medicine, which aims to apply redox modulation as a therapeutic strategy requires that we first comprehend signaling pathways by combining experimental data with mathematical modeling to deconstruct signals associated with specific redox molecules and determine their spatial and temporal pattern of change in tissue.

## Author contributions

**AL** – Corresponding author; Experiment design and conception; Experiment performance; Data Analysis; Writing of manuscript.

**EF** – Microelectrode construction and evaluation; data analysis.

**AS** – Writing and critical revision of manuscript.

**JL** – Experimental design; critical revision of manuscript.

**RB** – Experiment design and conception; data analysis; critical revision of manuscript; funding.

## Declaration of competing interest

The authors declare no conflict of interests.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.redox.2022.102250>.

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