

On-Pathway Oligomer of Human Islet Amyloid Polypeptide Induced and Stabilized by Mechanical Rotation During MAS NMR

Samuel D. McCalpin¹, Malitha C. Dickwella Widanage¹, Riqiang Fu² and Ayyalusamy Ramamoorthy^{1*}

¹*Biophysics Program, Department of Chemistry, Biomedical Engineering, Macromolecular Science and Engineering, Michigan Neuroscience Institute, University of Michigan, Arbor, MI 48109, USA*

²*National High Magnetic Field Laboratory, Florida State University, Tallahassee, Florida 32310, USA.*

KEYWORDS. Amyloid, Islet Amyloid Polypeptide, Oligomer, Magic Angle Spinning, NMR.

ABSTRACT: Intermediates along the fibrillation pathway are generally considered to be the toxic species responsible for the pathologies of amyloid diseases. However, structural studies of these species have been hampered by heterogeneity and poor stability in standard aqueous conditions. Here, we report a novel methodology for producing stable, on-pathway oligomers of the human Type-2 Diabetes-associated islet amyloid polypeptide (*hIAPP*, or amylin) using the mechanical forces associated with magic angle spinning (MAS). The species were a heterogeneous mixture of globular and short rod-like species with significant β -sheet content and the capability of seeding *hIAPP* fibrillation. We used MAS NMR to demonstrate that the nature of the species was sensitive to sample conditions including peptide concentration, ionic strength, and buffer. The methodology should be suitable for studies of other aggregating systems.

Several degenerative diseases, including Alzheimer's Disease (AD), Parkinson's Disease (PD), and Type-2 Diabetes (T2D), are linked to the aggregation of proteins and peptides into fibrillar structures called amyloid.¹⁻⁴ While the relationship between amyloid aggregation and these diseases was established several decades ago, few successful treatments have been developed that target the fibril formation process.⁵ Structural biologists seeking to guide drug design struggle to characterize amyloid fibrils because their large size and non-crystalline nature render traditional solution NMR and crystallography unfeasible. Still, solid-state NMR and Cryo-EM have proven capable of revealing specific, atom-resolution cross- β -sheet structures in fibril cores. However, recent research indicates that intermediate aggregates along the amyloid formation pathway, called "oligomers" or "protofibrils" based on size and morphology, are

the toxic species most responsible for the pathologies of AD, PD, T2D, and other amyloidoses.⁶⁻⁹ Structural characterizations of oligomeric amyloid aggregates would greatly benefit efforts to develop treatments against amyloid diseases, but several factors have made them scarce if not nonexistent. The most significant of these factors are high heterogeneity among oligomeric species and their short lifetimes in aqueous conditions.

Magic angle spinning (MAS) solid-state NMR spectroscopy is uniquely positioned to characterize heterogeneous systems of intermediate molecular weight, slow-tumbling species like amyloid oligomers. Its sensitivity to individual nuclei in distinct chemical environments allows discrimination between differently structured conformers. Additionally, a combination of a dipolar recoupling experiment and

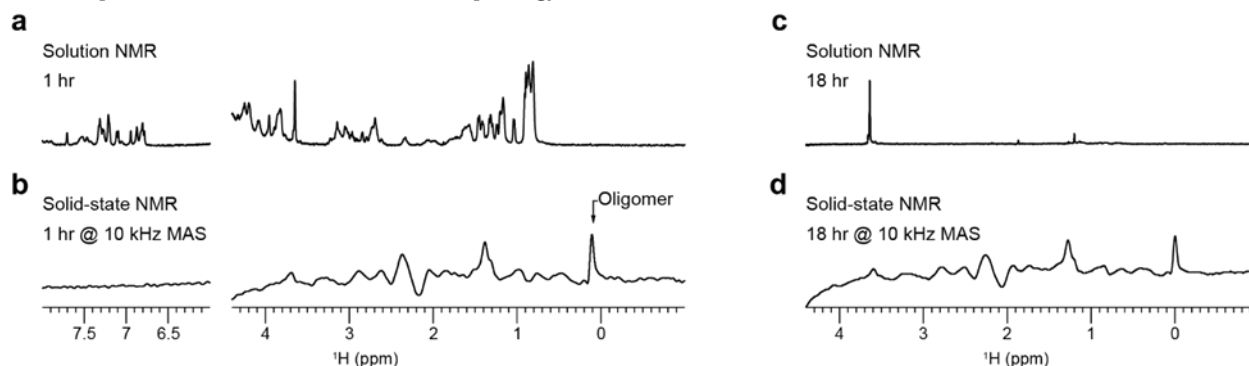


Figure 1. Visualizing the oligomeric intermediate's peaks under MAS. *hIAPP* monomers were detected from solution NMR (a, c) and MAS NMR (b, d) using ¹H spectra obtained at 1- and 18-hour timepoints on a 500 MHz spectrometer at 298 K. Samples contained 50 mM *hIAPP* in 10 mM d₁₁-tris, 100 mM NaCl, pH 7.4 buffer.

intermediate frequency MAS has been demonstrated to facilitate the separation of the oligomer signal from that of monomers and fibrils in a heterogeneous preparation of amyloid-beta.¹⁰ Typically, MAS experiments require biological samples to be lyophilized, frozen, or crystallized, but comprehensive multiphase (CMP) NMR spectroscopy enables the direct observation of mixed-phase systems with liquids.¹¹ The technique has been used to study microorganisms and environmental samples but also presents an approach to characterize heterogeneous amyloid systems consisting of soluble fast-tumbling monomers and low-order oligomeric aggregates and insoluble slow-tumbling high-order oligomers and fibrillar aggregates. The significant mechanical forces involved in MAS additionally offer a novel modulator of the amyloid formation pathway. Previous work has demonstrated the ability of small molecules, peptides, lipid membranes, metals, synthetic polymers, pH, and electric fields to alter amyloid aggregation pathways and in some cases to stabilize oligomeric intermediates for further structural studies.^{12–24} Here, we provide the first report, to our knowledge, of the ability of the mechanical forces associated with MAS to stabilize intermediate, on-pathway aggregates of human islet amyloid polypeptide (*hIAPP*) in an aqueous environment.

To compare the aggregation behavior of *hIAPP* with and without MAS, we prepared two samples and collected ¹H NMR spectra over the course of 24 hours using a standard static solution probe and a CMP probe with 10 kHz MAS (**Figure 1**). The sample conditions were identical except for 100% D₂O used in the MAS sample compared to 10% D₂O used in the solution NMR experiment. Under the conditions of the solution NMR experiment, the ¹H NMR spectrum of *hIAPP* appeared with the lineshape in **Figure 1a**, and the signal quickly decayed as the peptide aggregated within 7–8 hours. In contrast, the MAS sample immediately afforded a much broader lineshape with a significant peak near 0 ppm, typical of aggregated *hIAPP* (**Figure 1b**). This signal was also stable for at least 24 hours (**Figure 1c**). To rule out D₂O as the cause for the differences between the samples, we performed a thioflavin T fluorescence assay with buffers of varying amounts of D₂O and observed no change in the kinetics based on the percentage of D₂O (**Figure S1**). Thus, our data suggests that MAS induced the formation of an *hIAPP* aggregate.

However, due to the smaller volume of the MAS rotor, significantly less material was present in the MAS experiment than in the solution experiment, causing a noisier signal. There was also notable baseline rolling in the MAS spectrum which we hypothesized arose from conductivity through the coil due to salt in the buffer. To address these issues, we performed the MAS experiment again using a sample with a higher concentration of *hIAPP* in pure D₂O (**Figure 2a**). Interestingly, the aggregation behavior of this sample, as assessed by a series of ¹H NMR spectra collected under 10 kHz MAS, differed greatly from that of the previous sample. In this case, the ¹H NMR lineshape was initially much sharper and decayed over time to a broader, less intense lineshape which was stable after 18 hours. Subtracting the spectrum at 18 hours from the initial one revealed a lineshape which was remarkably similar to that of the solution spectrum, suggesting that the spectral changes over time resulted

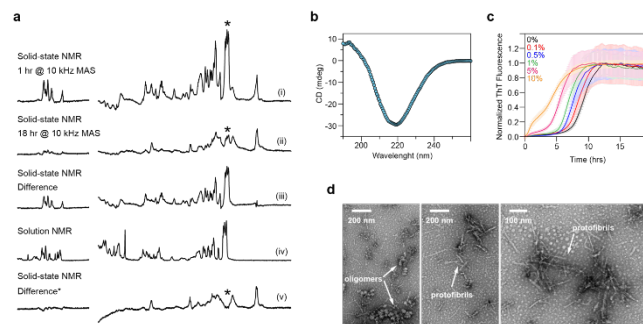


Figure 2. Heterogeneous, b-sheet structured, on-pathway *hIAPP* oligomers induced by MAS. (a) 1D ¹H NMR spectra with spectral editing showing the MAS-*hIAPP* oligomeric intermediates: (i) after 1 hour, (ii) after 18 hours, (iii) the difference spectrum of i and ii (normalized by the number of scans), (iv) solution NMR spectrum of *hIAPP*, and (v) the difference spectrum of ii and i (normalized by the peak with asterisk). (b) b-sheet structure identified using CD spectroscopy. (c) ThT fluorescence assay of *hIAPP* monomers with noted concentrations of MAS-*hIAPP* seeds. (d) TEM revealed heterogeneous MAS-*hIAPP* species; oligomers and protofibrils are labeled. For MAS NMR, 80 nmol *hIAPP* was hydrated to 2 mM in D₂O. NMR data were collected over 48 hours with 10 kHz MAS, and then the sample was removed from the rotor for further characterization. The MAS-*hIAPP* was diluted to 80 mM in D₂O for CD spectroscopy, 100 mM for TEM, and the noted molar ratios relative to 5 mM monomeric *hIAPP* in sodium phosphate buffer (10 mM NaPO₄, 100 mM NaCl, 10 mM ThT, pH 7.4) for ThT experiments.

from the depletion of *hIAPP* monomers as they associated into larger species. Additionally, we subtracted the solution spectrum from the long-time MAS spectrum, normalized to the intensity of the peak at 0.95 ppm, to obtain the lineshape of the *hIAPP* aggregates. The presence of a peak near 0 ppm, which did not vary with time, suggested that these species (denoted as MAS-*hIAPP*) were formed immediately under MAS and were stable for at least 24 hours.

We then characterized the MAS-*hIAPP* by a combination of fluorescence, CD spectroscopy, and TEM. CD exhibited a strong negative peak at 218 nm, indicative of significant b-sheet content (**Figure 2b**). The ThT fluorescence assay showed sigmoidal aggregation kinetics for *hIAPP* monomers in the presence of seeds taken from the MAS sample. In contrast, the fluorescence intensity of ThT in the presence of the MAS aggregate alone was constant over time and noticeably less than that of the monomers which formed fibrils (**Figure 2c**). Lag times of monomer aggregation decreased with increasing concentrations of MAS-*hIAPP* seeds, demonstrating that the MAS-*hIAPP* seeded aggregation of *hIAPP* monomers. Taken together, these results indicate that MAS induced the formation of on-pathway *hIAPP* oligomers with significant b-sheet content. However, TEM revealed a mixture of morphologies, including amorphous globular aggregates and rod-like protofibrillar species which were much shorter than the fibers formed by *hIAPP* under static conditions (**Figure 2d**). Based on the data presented here, we cannot determine whether one or both observed species are on-pathway. But it seems most likely that the smaller globular aggregates were the dominant contributors to the NMR and CD spectra, assuming their faster tumbling and greater aqueous solubility. Regardless, the data

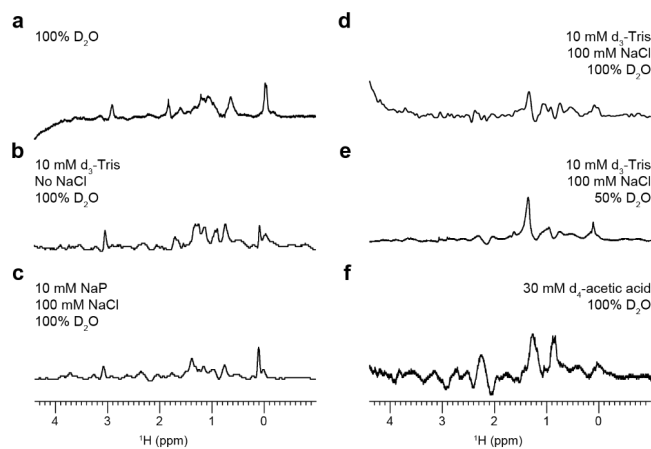


Figure 3. MAS-induced aggregation pathway altered by buffer. Solution conditions modified *hIAPP* aggregation under 10 kHz MAS. *hIAPP* concentrations are (a) 2 mM, (b) 2 mM (7.4 pH), (c) 2 mM (7.4 pH), (d) 2 mM (7.4 pH), (e) 50 μ M (7.4 pH), and (f) 50 μ M (5.5 pH). A 0-ppm peak indicated the presence of oligomeric intermediates.

confirms that MAS induced the formation of non-fibrillar oligomeric intermediates of *hIAPP*.

There were issues reproducing the MAS-*hIAPP* formation in D_2O (Figure S2), perhaps due to the use of an unbuffered solvent. Solution pH has been extensively reported to affect the kinetics of *hIAPP* aggregation and the end species formed, so it is favorable to control regardless of any issues with reproducibility.²⁵⁻²⁸ To this end, we prepared *hIAPP* samples in several buffer conditions and collected 1H MAS spectra as before. The observed 1H lineshapes for each sample condition were compiled in Figure 3. In all cases, there was a peak near 0 ppm at the initial timepoint which did not significantly change for at least 18 hours, consistent with the immediate formation of a stable oligomer regardless of buffer. The 0 ppm peak was weakest for the low *hIAPP* concentration, acetate buffer condition, which is unsurprising given that both low peptide concentration and low pH are known to reduce *hIAPP* aggregation (Figure 3f).^{26,29} Remarkably, though there was significant spectral variation across the samples, the lineshapes reported for the oligomers formed in 2 mM *hIAPP* in pure D_2O , Tris buffer (no salt), and sodium phosphate buffer (with salt) were very similar (Figure 3a-c). This suggests that a similar species, or range of species, formed under these sample conditions. Based on the differences between the spectral patterns of Figures 3d and 3e and between Figures 3a, 3b, and 3d, the peptide concentration and buffer ionic strength seemed to be the most significant factors which influenced the nature of the oligomers formed. Tris also appeared to interact with *hIAPP* and bias the aggregation pathway, but only in the presence of salt.

The amount of structural information we could obtain from 1D 1H NMR lineshapes was limited, so we collected 2D 1H - 1H RFDR MAS spectra of a MAS-*hIAPP* sample in pure D_2O (Figure 4).³⁰ These RFDR spectra (Figure 4a) exhibited several well-resolved cross-peaks which mostly saturated with 25 ms mixing (Figures 4c, S3). A comparable spectrum of MAS-*hIAPP* in buffer (Figure S4) displayed significantly fewer (but still well-resolved) cross-peaks. Considering that the primary difference between these two

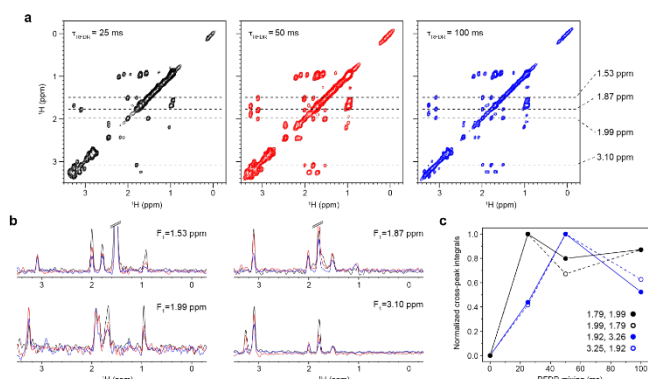


Figure 4. Dipolar recoupling from MAS-*hIAPP*. 2D 1H - 1H RFDR spectra of 2 mM *hIAPP* in 100% D_2O obtained with an 800 MHz NMR spectrometer under 15 kHz MAS at 298 K. (a) Spectra were obtained at three mixing times. (b) 1D slices and (c) build-up curves are plotted for the cross-peaks at the noted chemical shifts.

samples was a significant monomer population present in pure D_2O and not in buffer, the extra cross-peaks in the spectrum of the D_2O sample likely arose from this population. It seems unlikely that dipolar recoupling would occur so efficiently in small, fast-tumbling monomers, so it is also possible that the observed cross-peaks represented low-order oligomers or monomers which were motionally restricted by transient interactions with larger, invisible oligomers. Regardless, the RFDR spectra demonstrated that MAS-*hIAPP* samples should be amenable to structural studies. Challenges related to peak assignments could be overcome in the future with 3D heteronuclear experiments on ^{13}C -labeled peptide, and higher MAS frequency could allow full observation of the larger oligomers.

Drug design efforts against amyloid diseases would greatly benefit from structural studies of intermediate amyloid aggregates. Here, we presented a novel method to stabilize nonfibrillar species of *hIAPP* using the mechanical forces associated with MAS. We showed that the heterogeneous mixture of species included rod-like protofibrils and globular aggregates and on-pathway, beta-sheet-rich species which formed within several hours in pure D_2O , or immediately in common buffer conditions. The sample conditions, namely the peptide concentration and choice of buffer, were also found to affect the oligomers formed. 2D RFDR experiments showed that the MAS-*hIAPP* was amenable to further structural characterization. Such studies would deepen our understanding of the molecular basis of T2D. However, their feasibility in our hands was limited by the availability of isotope-labeled peptide and MAS limits on liquid samples. Significantly, the methodology reported here should be broadly applicable to studying aggregates of other amyloidogenic peptides and IDPs and is more tunable than chemical chaperones of oligomeric aggregates.

ASSOCIATED CONTENT

Supporting Information. Methods and materials, additional ThT fluorescence and MAS NMR data, discussion of MAS NMR data reproducibility. This material is available free of charge via the Internet at <http://pubs.acs.org>.

AUTHOR INFORMATION

Corresponding Author

* Correspondence and requests for materials should be addressed to A.R. (email: ramamoorthy@umich.edu)

Author Contributions

The manuscript was written through contributions of all authors.

ACKNOWLEDGMENT

This work was supported by the NIH grant 5R01DK132214-02. A portion of this work was performed at the National High Magnetic Field Laboratory, which is supported by National Science Foundation Cooperative Agreement No. DMR-2128556 and the State of Florida.

ABBREVIATIONS

hIAPP, human islet amyloid polypeptide; NMR, nuclear magnetic resonance; MAS, magic angle spinning.

REFERENCES

- Knowles, T. P. J.; Vendruscolo, M.; Dobson, C. M. The Amyloid State and Its Association with Protein Misfolding Diseases. *Nat. Rev. Mol. Cell Biol.* **2014**, *15* (6), 384–396. <https://doi.org/10.1038/nrm3810>.
- Hempel, H.; Hardy, J.; Blennow, K.; Chen, C.; Perry, G.; Kim, S. H.; Villemagne, V. L.; Aisen, P.; Vendruscolo, M.; Iwatsubo, T.; Masters, C. L.; Cho, M.; Lannfelt, L.; Cummings, J. L.; Vergallo, A. The Amyloid- β Pathway in Alzheimer's Disease. *Mol. Psychiatry* **2021**, *26* (10), 5481–5503. <https://doi.org/10.1038/s41380-021-01249-0>.
- Milardi, D.; Gazit, E.; Radford, S. E.; Xu, Y.; Gallardo, R. U.; Caffisch, A.; Westermark, G. T.; Westermark, P.; Rosa, C. L.; Ramamoorthy, A. Proteostasis of Islet Amyloid Polypeptide: A Molecular Perspective of Risk Factors and Protective Strategies for Type II Diabetes. *Chem. Rev.* **2021**, *121* (3), 1845–1893. <https://doi.org/10.1021/acs.chemrev.0c00981>.
- Mehra, S.; Sahay, S.; Maji, S. K. α -Synuclein Misfolding and Aggregation: Implications in Parkinson's Disease Pathogenesis. *Biochim. Biophys. Acta BBA - Proteins Proteomics* **2019**, *1867* (10), 890–908. <https://doi.org/10.1016/j.bbapap.2019.03.001>.
- Doig, A. J.; del Castillo-Frias, M. P.; Berthoumieu, O.; Tarus, B.; Nasica-Labouze, J.; Sterpone, F.; Nguyen, P. H.; Hooper, N. M.; Faller, P.; Derreumaux, P. Why Is Research on Amyloid- β Failing to Give New Drugs for Alzheimer's Disease? *ACS Chem. Neurosci.* **2017**, *8* (7), 1435–1437. <https://doi.org/10.1021/acschemneuro.7b00188>.
- Haataja, L.; Gurlo, T.; Huang, C. J.; Butler, P. C. Islet Amyloid in Type 2 Diabetes, and the Toxic Oligomer Hypothesis. *Endocr. Rev.* **2008**, *29* (3), 303–316. <https://doi.org/10.1210/er.2007-0037>.
- Cline, E. N.; Bicca, M. A.; Viola, K. L.; Klein, W. L. The Amyloid- β Oligomer Hypothesis: Beginning of the Third Decade. *J. Alzheimer's Dis.* **2018**, *64* (s1), S567–S610. <https://doi.org/10.3233/JAD-179941>.
- Ono, K. The Oligomer Hypothesis in α -Synucleinopathy. *Neurochem. Res.* **2017**, *42* (12), 3362–3371. <https://doi.org/10.1007/s11064-017-2382-x>.
- Biol, M.; Kumar, S.; Rhoades, E.; Miranker, A. D. Conformational Switching within Dynamic Oligomers Underpins Toxic Gain-of-Function by Diabetes-Associated Amyloid. *Nat. Commun.* **2018**, *9* (1), 1312. <https://doi.org/10.1038/s41467-018-03651-9>.
- Kotler, S. A.; Brender, J. R.; Vivekanandan, S.; Suzuki, Y.; Yamamoto, K.; Monette, M.; Krishnamoorthy, J.; Walsh, P.; Cauble, M.; Holl, M. M. B.; Marsh, E. N. G.; Ramamoorthy, A. High-Resolution NMR Characterization of Low Abundance Oligomers of Amyloid- β without Purification. *Sci. Rep.* **2015**, *5* (1), 11811. <https://doi.org/10.1038/srep11811>.
- Courtier-Murias, D.; Farooq, H.; Masoom, H.; Botana, A.; Soong, R.; Longstaffe, J. G.; Simpson, M. J.; Maas, W. E.; Fey, M.; Andrew, B.; Struppe, J.; Hutchins, H.; Krishnamurthy, S.; Kumar, R.; Monette, M.; Stronks, H. J.; Hume, A.; Simpson, A. J. Comprehensive Multiphase NMR Spectroscopy: Basic Experimental Approaches to Differentiate Phases in Heterogeneous Samples. *J. Magn. Reson.* **2012**, *217*, 61–76. <https://doi.org/10.1016/j.jmr.2012.02.009>.
- Ahmad, B.; Winkelmann, J.; Tiribilli, B.; Chiti, F. Searching for Conditions to Form Stable Protein Oligomers with Amyloid-like Characteristics: The Unexplored Basic PH. *Biochim. Biophys. Acta BBA - Proteins Proteomics* **2010**, *1804* (1), 223–234. <https://doi.org/10.1016/j.bbapap.2009.10.005>.
- Lopez del Amo, J. M.; Fink, U.; Dasari, M.; Grelle, G.; Wanker, E. E.; Bieschke, J.; Reif, B. Structural Properties of EGCG-Induced, Nontoxic Alzheimer's Disease A β Oligomers. *J. Mol. Biol.* **2012**, *421* (4), 517–524. <https://doi.org/10.1016/j.jmb.2012.01.013>.
- Chang, Y.-J.; Chen, Y.-R. The Coexistence of an Equal Amount of Alzheimer's Amyloid- β 40 and 42 Forms Structurally Stable and Toxic Oligomers through a Distinct Pathway. *FEBS J.* **2014**, *281* (11), 2674–2687. <https://doi.org/10.1111/febs.12813>.
- Nedumpully-Govindan, P.; Kakinen, A.; Pilkington, E. H.; Davis, T. P.; Chun Ke, P.; Ding, F. Stabilizing Off-Pathway Oligomers by Polyphenol Nanoassemblies for IAPP Aggregation Inhibition. *Sci. Rep.* **2016**, *6* (1), 19463. <https://doi.org/10.1038/srep19463>.
- Prade, E.; Barucker, C.; Sarkar, R.; Althoff-Ospelt, G.; Lopez del Amo, J. M.; Hossain, S.; Zhong, Y.; Multhaup, G.; Reif, B. Sulindac Sulfide Induces the Formation of Large Oligomeric Aggregates of the Alzheimer's Disease Amyloid- β Peptide Which Exhibit Reduced Neurotoxicity. *Biochemistry* **2016**, *55* (12), 1839–1849. <https://doi.org/10.1021/acs.biochem.5b01272>.
- Jayaraman, S.; Gantz, D. L.; Haupt, C.; Gursky, O. Serum Amyloid A Forms Stable Oligomers That Disrupt Vesicles at Lysosomal PH and Contribute to the Pathogenesis of Reactive Amyloidosis. *Proc. Natl. Acad. Sci.* **2017**, *114* (32), E6507–E6515. <https://doi.org/10.1073/pnas.1707120114>.
- Rodriguez Camargo, D. C.; Korshavn, K. J.; Jussupow, A.; Raltchev, K.; Goricanec, D.; Fleisch, M.; Sarkar, R.; Xue, K.; Aichler, M.; Mettenleiter, G.; Walch, A. K.; Camilloni, C.; Hagn, F.; Reif, B.; Ramamoorthy, A. Stabilization and Structural Analysis of a Membrane-Associated HIAPP Aggregation Intermediate. *eLife* **2017**, *6*. <https://doi.org/10.7554/eLife.31226>.
- Saikia, J.; Pandey, G.; Sasidharan, S.; Antony, F.; Nemade, H. B.; Kumar, S.; Chaudhary, N.; Ramakrishnan, V. Electric Field Disruption of Amyloid Aggregation: Potential Noninvasive Therapy for Alzheimer's Disease. *ACS Chem. Neurosci.* **2019**, *10* (5), 2250–2262. <https://doi.org/10.1021/acschemneuro.8b00490>.
- Ciudad, S.; Puig, E.; Botzanowski, T.; Meigooni, M.; Arango, A. S.; Do, J.; Mayzel, M.; Bayoumi, M.; Chaignepain, S.; Maglia, G.; Cianferani, S.; Orekhov, V.; Tajkhorshid, E.; Bardiaux, B.; Carulla, N. A β (1-42) Tetramer and Octamer Structures Reveal Edge Conductivity Pores as a Mechanism for Membrane Damage. *Nat. Commun.* **2020**, *11* (1), 3014. <https://doi.org/10.1038/s41467-020-16566-1>.
- Gao, Y.; Guo, C.; Watzlawik, J. O.; Randolph, P. S.; Lee, E. J.; Huang, D.; Stagg, S. M.; Zhou, H.-X.; Rosenberry, T. L.; Paravastu, A. K. Out-of-Register Parallel β -Sheets and Antiparallel β -Sheets Coexist in 150-KDa Oligomers Formed by Amyloid- β (1-42). *J. Mol. Biol.* **2020**, *432* (16), 4388–4407. <https://doi.org/10.1016/j.jmb.2020.05.018>.
- Cox, S. J.; Rodriguez Camargo, D. C.; Lee, Y.-H.; Dubini, R. C. A.; Rovó, P.; Ivanova, M. I.; Padmini, V.; Reif, B.; Ramamoorthy, A.

Small Molecule Induced Toxic Human-IAPP Species Characterized by NMR. *Chem. Commun.* **2020**, 56 (86), 13129–13132. <https://doi.org/10.1039/D0CC04803H>.

(23) Sahoo, B. R.; Souders, C. L.; Nakayama, T. W.; Deng, Z.; Linton, H.; Suladze, S.; Ivanova, M. I.; Reif, B.; Ando, T.; Martyniuk, C. J.; Ramamoorthy, A. *Conformational Tuning of Amylin by Charged Styrene-Maleic-Acid Copolymers*; 2021; p 2020.04.23.057547. <https://doi.org/10.1101/2020.04.23.057547>.

(24) Roy, D.; Maity, N. C.; Kumar, S.; Maity, A.; Ratha, B. N.; Biswas, R.; Maiti, N. C.; Mandal, A. K.; Bhunia, A. Modulatory Role of Copper on HIAPP Aggregation and Toxicity in Presence of Insulin. *Int. J. Biol. Macromol.* **2023**, 124470. <https://doi.org/10.1016/j.ijbiomac.2023.124470>.

(25) Abedini, A.; Raleigh, D. P. The Role of His-18 in Amyloid Formation by Human Islet Amyloid Polypeptide. *Biochemistry* **2005**, 44 (49), 16284–16291. <https://doi.org/10.1021/bi051432v>.

(26) Brender, J. R.; Hartman, K.; Nanga, R. P. R.; Popovych, N.; de la Salud Bea, R.; Vivekanandan, S.; Marsh, E. N. G.; Ramamoorthy, A. Role of Zinc in Human Islet Amyloid Polypeptide Aggregation. *J. Am. Chem. Soc.* **2010**, 132 (26), 8973–8983. <https://doi.org/10.1021/ja1007867>.

(27) Khemtémourian, L.; Doménech, E.; Doux, J. P. F.; Koorengevel, M. C.; Killian, J. A. Low PH Acts as Inhibitor of Membrane Damage Induced by Human Islet Amyloid Polypeptide. *J. Am. Chem. Soc.* **2011**, 133 (39), 15598–15604. <https://doi.org/10.1021/ja205007j>.

(28) Santos, J.; Iglesias, V.; Santos-Suárez, J.; Mangiagalli, M.; Brocca, S.; Pallarès, I.; Ventura, S. PH-Dependent Aggregation in Intrinsically Disordered Proteins Is Determined by Charge and Lipophilicity. *Cells* **2020**, 9 (1), 145. <https://doi.org/10.3390/cells9010145>.

(29) Wei, L.; Jiang, P.; Xu, W.; Li, H.; Zhang, H.; Yan, L.; Chan-Park, M. B.; Liu, X.-W.; Tang, K.; Mu, Y.; Pervushin, K. The Molecular Basis of Distinct Aggregation Pathways of Islet Amyloid Polypeptide *. *J. Biol. Chem.* **2011**, 286 (8), 6291–6300. <https://doi.org/10.1074/jbc.M110.166678>.

(30) Ramamoorthy, A.; Xu, J. 2D ¹H/¹H RFDR and NOESY NMR Experiments on a Membrane-Bound Antimicrobial Peptide Under Magic Angle Spinning. *J. Phys. Chem. B* **2013**, 117 (22), 6693–6700. <https://doi.org/10.1021/jp4034003>.

SYNOPSIS TOC

