



Asymmetric synthesis of CF₂-functionalized aziridines by combined strong Brønsted acid catalysis

Xing-Fa Tan^{1,2}, Fa-Guang Zhang^{*1,2} and Jun-An Ma^{*1,2}

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Address:

¹Department of Chemistry, Tianjin Key Laboratory of Molecular Optoelectronic Sciences, and Tianjin Collaborative Innovation Centre of Chemical Science & Engineering, Tianjin University, Tianjin 300072, China and ²Joint School of NUS & TJU, International Campus of Tianjin University, Fuzhou 350207, China

Email:

Fa-Guang Zhang^{*} - zhangfg1987@tju.edu.cn; Jun-An Ma^{*} - majun_an68@tju.edu.cn

^{*} Corresponding author

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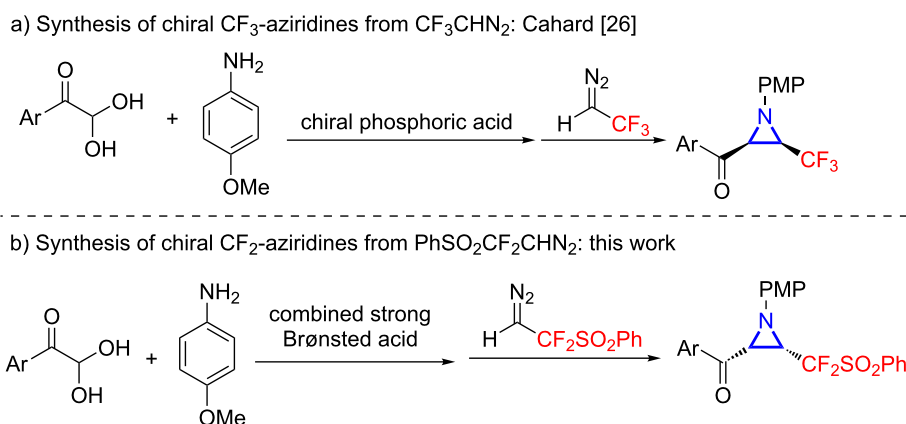
Abstract

A diastereo- and enantioselective approach to access chiral CF₂-functionalized aziridines from difluorodiaoethyl phenyl sulfone (PhSO₂CF₂CHN₂) and in situ-formed aldimines is described. This multicomponent reaction is enabled by a combined strong Brønsted acid catalytic platform consisting of a chiral disulfonimide and 2-carboxyphenylboronic acid. The optical purity of the obtained CF₂-substituted aziridines could be further improved by a practical dissolution–filtration procedure.

Introduction

Chiral aziridines are prevalently found in natural products and artificially made bioactive molecules, thus receiving significant attention in the past decades [1-6]. Among them, the introduction of fluorine or fluoroalkyl groups into three-membered N-heterocycles has emerged as an attractive direction due to the unique fluorine effect in pharmaceuticals and biology [7-11]. In this context, it is not surprising that the syntheses of trifluoromethylaziridines have been pursued from versatile precursors [12-25]. However, catalytic asymmetric approaches to chiral CF₃-functionalized aziridines have only been reported by Cahard in 2012, who utilized trifluorodiaoethane (CF₃CHN₂)

as the nucleophile to react with aldimines catalyzed by chiral phosphoric acid (Scheme 1a) [26]. In comparison, there is a significant dearth of available synthetic approaches to CF₂-functionalized aziridines, particularly in a stereocontrolled manner. Indeed, a handful of reported methods document the employment of difluoromethylamines, difluoromethyl phenyl sulfone, and difluoromethyl vinyl sulfonium salts as the fluorinating partner en route to various CF₂-substituted aziridines [27-31], and a general protocol to chiral CF₂-aziridines remains an unsolved challenge. Thus, herein we report a diastereo- and enantioselective aza-Darzens reaction between in situ-generated



Scheme 1: Preparation of chiral aziridines from fluorinated diazo reagents.

aldimines and our recently developed difluorodiazo reagent PhSO₂CF₂CHN₂ acting as the difluorinated nucleophile [32–35], providing access to a variety of chiral CF₂-functionalized aziridines under mild conditions (Scheme 1b). The key to this multicomponent transformation hinges upon the discovery of a combined strong Brønsted acid system comprised of a chiral disulfonimide and 2-carboxyphenylboronic acid.

Results and Discussion

We commenced the desired one-pot transformation by conducting the model reaction between phenylglyoxal monohydrate (**1a**), 4-methoxyaniline (**2a**), and PhSO₂CF₂CHN₂ (**3**, Ps-DFA). Initial screenings were focused on the evaluation of various chiral phosphoric acids that have proven effective in similar aza-Darzens reactions of diazo esters and trifluorodiazoethane [36–39]. Unfortunately, these endeavors resulted in either no

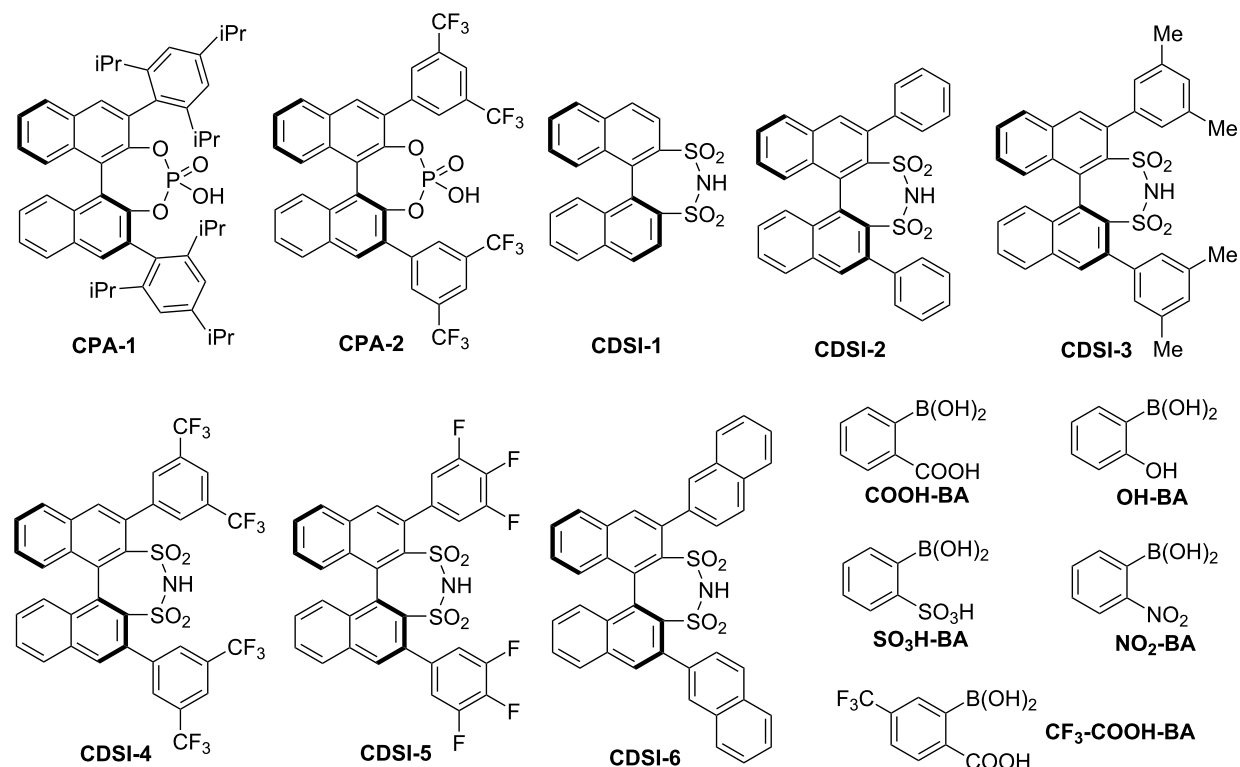
conversion or no enantioselectivity at all. As arylboronic acids have been harnessed to enhance the Brønsted acidity in asymmetric organocatalysis in combination with chiral diols or chiral aminoalcohols [40–44], we envisioned that the simultaneous use of arylboronic acids and chiral Brønsted acids may bring about a complementary catalytic platform. Encouragingly, the targeted CF₂-functionalized aziridine **4a** was obtained in up to 51% ee and high diastereoselectivity, albeit in a low yield (Table 1, entries 1 and 2). The difficulty in further improving the conversions might be ascribed to the limited Brønsted acidity of chiral phosphoric acids. Bearing this in mind, we then turned our attention to chiral disulfonimides developed by List, which have been established as a unique type of stronger Brønsted acids [45]. Putting it into practice, a range of BINOL-derived disulfonimides was used as the chiral additive in combination with 2-carboxyphenylboronic acid (**COOH-BA**) in the

Table 1: Representative screening results of the asymmetric aziridination reaction of PhSO₂CF₂CHN₂.^a

entry	arylboronic acid (mol %)	chiral Brønsted acid (mol %)	yield of 4a (%) ^b	ee (%) of 4a and dr of crude mixture ^c
1	COOH-BA (8)	CPA-1 (5)	24	51, 13:1
2	COOH-BA (8)	CPA-2 (5)	28	25, 11:1
3	COOH-BA (8)	CDSI-1 (5)	21	41, 19:1
4	COOH-BA (8)	CDSI-2 (5)	50	41, 9:1
5 ^d	COOH-BA (8)	CDSI-3 (5)	16	60, 5:1
6 ^d	COOH-BA (8)	CDSI-4 (5)	64	73, 13:1
7	COOH-BA (8)	CDSI-5 (5)	34	33, 10:1

Table 1: Representative screening results of the asymmetric aziridination reaction of PhSO₂CF₂CHN₂.^a (continued)

8	COOH-BA (8)	CDSI-6 (5)	47	52, 9:1
9	OH-BA (8)	CDSI-4 (5)	63	68, 28:1
10 ^d	SO₃H-BA (8)	CDSI-4 (5)	62	66, 16:1
11 ^d	NO₂-BA (8)	CDSI-4 (5)	45	62, 16:1
12	CF₃-COOH-BA (8)	CDSI-4 (5)	81	67, 8:1
13 ^e	COOH-BA (8)	CDSI-4 (5)	60	47, 5:1
14 ^f	COOH-BA (8)	CDSI-4 (5)	trace	n.d.
15 ^d	COOH-BA (8)	CDSI-4 (10)	65	70, 12:1
16 ^d	–	CDSI-4 (5)	10	60, >20:1

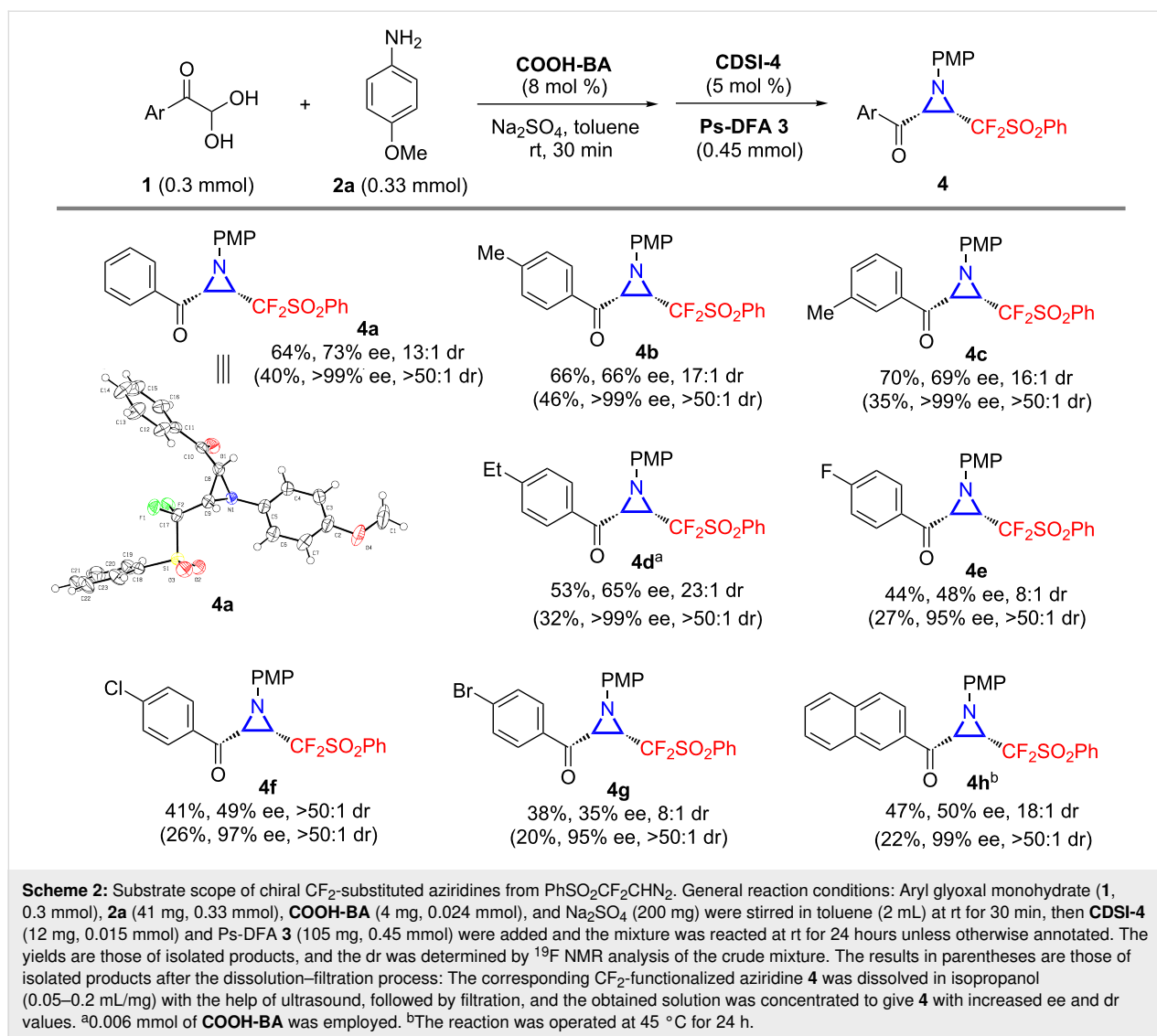


^aGeneral reaction conditions: **1a** (8 mg, 0.05 mmol, 1.0 equiv), **2a** (7 mg, 0.055 mmol), arylboronic acid (0.004 mmol), and Na₂SO₄ (40 mg) was stirred in toluene (1 mL) at rt for 30 min, then the chiral Brønsted acid (0.0025 mmol) and **3** (18 mg, 0.075 mmol) were added and the mixture was reacted at rt for 12 hours unless otherwise noted; ^byield of isolated product **4a** was given for entries labelled with d; hexafluorobenzene was used as an internal standard to determine the yield in other cases; ^cee of **4a** was determined by chiral HPLC analysis, and the dr of the crude reaction mixture was probed by ¹⁹F NMR analysis; ^d0.3 mmol scale of reaction was conducted: **1a** (46 mg, 0.3 mmol, 1.0 equiv), **2a** (41 mg, 0.33 mmol), arylboronic acid (0.024 mmol), and Na₂SO₄ (200 mg) was stirred in toluene (2 mL) at rt for 30 min, then the chiral Brønsted acid (0.015 mmol) and **3** (105 mg, 0.45 mmol) were added and the mixture was reacted at rt for 12–24 hours; ^eCH₂Cl₂ was used as the solvent; ^freaction was operated at 0 °C.

model reaction (Table 1, entries 3–8). We were pleased to find that **CDSI-4** gave the most promising result in terms of both yield and enantioselectivity (64% isolated yield with 73% ee, Table 1, entry 6). An examination on various arylboronic acids, solvent, temperature, and catalyst loadings resulted in no obvious improvement (Table 1, entries 9–15). Among them, the highest yield of **4a** was observed (81%, Table 1, entry 12), albeit with slightly reduced ee value. This enhancement in catalytic activity could be attributed to the increased Brønsted acidity when the strong electron-withdrawing trifluoromethyl group was placed on the benzene ring of the arylboronic acid.

Removing the boronic acid from the reaction system leads to a dramatic decrease in both yield and enantiocontrol (Table 1, entry 16).

The challenge to further improve the enantioselectivity promoted us to search for other practical solutions. Considering the poor solubility of **4a** in organic solvents, a dissolution–filtration process with isopropanol was found to be workable for increasing the final ee value. This simple procedure could afford **4a** with excellent enantiopurity as a single diastereoisomer (>99% ee, >50:1 dr, Scheme 2). By the aid of the



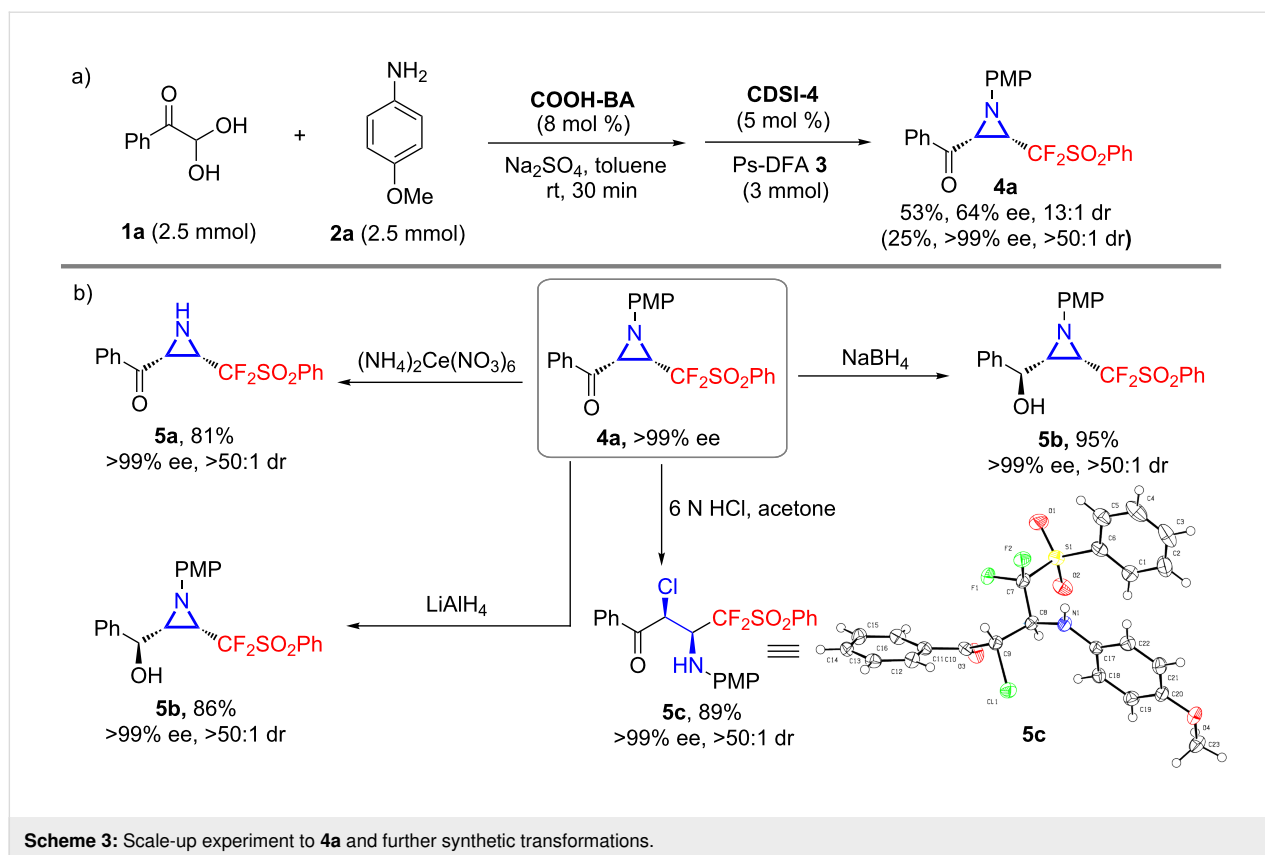
developed one-pot aza-Darzens reaction and dissolution–filtration operation, a series of optically-pure CF₂-aziridines **4b–h** were furnished in moderate overall yields with uniformly excellent ee and dr values, including alkyl or halogen-substituted phenyl and 2-naphthyl ketones (Scheme 2). Unfortunately, phenylglyoxal monohydrates bearing strong electron-withdrawing groups were not compatible with the current conditions. X-ray analysis of aziridine **4a** confirmed the absolute configuration of the chiral centers, pointing at a *cis*-aziridination process [46].

Scaled-up experiments with model substrate **1a** also proved to be feasible, delivering the chiral CF₂-aziridine **4a** with comparable results (Scheme 3a). The 4-methoxyphenyl group of **4a** was cleaved smoothly with ceric ammonium nitrate, giving the free aziridine **5a** in 81% yield while maintaining the ee value. The reduction of the carbonyl moiety with either NaBH₄ or

LiAlH₄ produced hydroxy-substituted CF₂-functionalized aziridine **5b** in excellent yield with exclusive diastereoselectivity [47]. Furthermore, the ring-opening of **4a** under acidic conditions underwent well and gave rise to CF₂-functionalized α -chloro- β -amino ketone **5c** in 89% yield with >99% ee and >50:1 dr (confirmed by X-ray spectroscopy) [46].

Conclusion

In summary, an array of chiral CF₂-functionalized aziridines was constructed from in situ-formed aldimines and difluorodiazoethyl phenyl sulfone under mild conditions by a combined strong Brønsted acid system consisting of chiral disulfonimide and 2-carboxyphenylboronic acid. The optical purity of the obtained CF₂-substituted aziridines could be further improved by a practical dissolution–filtration procedure. Substrate expansion and mechanistic investigation are underway and will be reported in due course.



Experimental

General procedure for the preparation of chiral CF₂-functionalized aziridines **4:** To a 25 mL Schlenk tube equipped with a stirring bar were added 2,2-dihydroxy-1-arylethan-1-one (**1**, 0.3 mmol, 1 equiv), 4-methoxyaniline (**2a**, 40.6 mg, 0.33 mmol), 2-boronobenzoic acid (**COOH-BA**, 3.98 mg, 0.024 mmol), anhydrous Na₂SO₄ (200 mg) and toluene (1 mL) at room temperature under an argon atmosphere. After reacting for 30 minutes at room temperature, ((2-diazo-1,1-difluoroethyl)sulfonyl)benzene (**Ps-DFA 3**, 104.5 mg, 77.4 μL, 0.45 mmol) was added with a micro syringe and **CDSI-4** (12.3 mg, 0.015 mmol) in toluene (1 mL) was added dropwise. The reaction was allowed to stir for 24 hours at room temperature under an argon atmosphere until the consumption of substrates was completed (as monitored by TLC). The reaction mixture was quenched with saturated aq NaHCO₃ and extracted with ethyl acetate three times. The combined organic layer was washed with water and brine, and then dried over anhydrous Na₂SO₄, filtered and evaporated under vacuum. The residue was purified by neutral alumina column chromatography (eluting with dichloromethane/petroleum ether) to give CF₂-substituted aziridine **4**. The enantiomeric excess was determined by chiral HPLC analysis. See Supporting Information File 1 for the dissolution–filtration procedure for each compound.

Supporting Information

Supporting Information File 1

Experimental procedures, compound characterization, NMR spectra of all new compounds, and HPLC traces.
[\[https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S1.pdf\]](https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S1.pdf)

Supporting Information File 2

X-ray data for compound **4a**.
[\[https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S2.cif\]](https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S2.cif)

Supporting Information File 3

X-ray data for compound **5c**.
[\[https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S3.cif\]](https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-60-S3.cif)

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ORCID® iDs

Fa-Guang Zhang - <https://orcid.org/0000-0002-0251-0456>

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- CCDC 1983642 (4a) and 1983643 (5c) contain the supplementary crystallographic data. The crystallographic data can be obtained free of charge via http://www.ccdc.cam.ac.uk/data_request/cif. It should be mentioned that the stereochemistry of 3-CF₂-aziridines in this reaction (2*R*,3*S*) are opposite to that of 3-CF₃-aziridines (2*S*,3*R*) in Cahard's report ([26]). We hypothesized that the enantiodetermining step in our reaction might be controlled through an in situ formed chiral boronate complex from chiral disulfonimide and 2-carboxyphenylboronic acid. Further efforts to improve the level of stereoselectivity and detailed mechanistic elucidation are still undergoing in our lab.

47. The stereochemistry of compound **5b** was determined based on the ^1H NMR, ^{19}F NMR, and 2D NOE analysis. In the 2D NOE spectrum, the correlation between HO (4.83 ppm) and HC–N (2.7 ppm) was observed, whereas no correlation between HC–O (3.7 ppm) and HC–CF₂ (3.02 ppm) was found.

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