Hyperkalemia by β -Fluoroethylacetate: A Report of 2 Cases

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When acidemia occurs associated with mineral acids, there is a predictable increase in the serum K^+ concentration. In uncomplicated cases of acidosis associated with nonmineral organic acids, the serum K^+ concentration usually remains within the normal range. The mechanism of this different effect of inneral and organic acidemias on the transmembrane movement of K^+ remains undefined.

We observed increases of serum K^+ concentration in 2 patients who ingested β -fluoroethylacetate solution for the purpose of committing suicide.

Key Words: β-fluoroethylacetate. Metabolic acidosis. Hyperkalemia

INTRODUCTION

In a case of acidosis induced by the administration of a mineral acid, such as Hol or NH $_4$ cl, acidemia itself increases serum K $^+$ concentration. $^{1+5)}$ It is well known that hyper-kalemia is caused by the release of intracellular K $^+$ into extracellular fluid. In contrast, in animal experiments involving the administration of nonmineral acids, such as lactic acid $^{6)}$ and methylmalonic acid $^{7)}$ there was no increase in serum K $^+$ concetration by a nonmineral acid clinically except in diabetic ketoacidosis $^{9)}$.

We report 2 cases of hyperkalemia which developed after β -fluoroethylacetate was ingested in which serum K⁺ was increased to 5.57 meq I and 6.7 meq/I, respectively, with severe metabolic acidosis

CASE 1

A 59-year-old woman was admitted to the hospital via the emergency room in a semicoma with vomiting and intermittent general convul-

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sions, following the ingestion of 6.0 gm of β -fluoroethylacetate (Kilatol F^R) in a suicide attempt.

She had used no medication previously. There was no history of convulsions or renal disease.

On physical examination, she was obviously semicornatose without cyanosis. The oral mucosa appeared burned. On chest examination she was found to have fine, moist rales and arrhythmia without a murmur. Neurological examination revealed decreased deep tendon reflex.

The temperature was 37°C, the pulse 92, and the respirations 25/min. The blood pressure was 110/70 mmHg.

The laboratory findings on admission were as follows: hemoglobin 13.6 gm%, hematocrit 39.2%, white cell count 10,000, platelet count 40,200, total protein 6.2 gm, and albumin 3.9 gm%. Blood gas and electrolytes findings are shown in Table 1. Urinalysis disclosed (+) protein and (+) sugar.

On admission, a Levin tube was inserted, and a gastric lavage was done. She was in metabolic acidosis (pH 7.26, HCO₃ 17 meq/1, pCO₂ 38 mmHg). Serum K⁺ concentration was 5.57 meq/l. Therefore, 24 meq/l of HCO₃ was given every 2 hours for 24 hours along with 10% glucose solution and 5% dextrose in saline.

Forty milligrams of Lasix was also prescribed

every 8 hours for 24 hours. On the second hospital day the pH was 7.34, HCO₃ 19 meq/l, pCO₂36 mmHg and serum K⁺ 4.9 meq/l. Bicarbonate and hypertonic glucose solution were given. On the 3rd hospital day she was in a clear mental state. Sixty milligrams of Lidocaine was given to control severe intermittent ventricular premature contraction. Prednisolone was given to relieve esophageal edema, and penicillin G to prevent infection. On the 5th hospital day there was a marked improvement in the ECG, and Lidocaine was withdrawn. On the 6th hospital day the results of the laboratory examinations were pH 7.36, HCO₃ 24 meq/l, pCO₂ 43 mmHg and serum K⁺ 3.6 meq/l. She began to eat a normal diet on the 7th day.

An esophagogram showed a mild stricture of the entire esophagus.

The patient had recovered completely after 52 days in the hospital and was discharged.

CASE 2

A 38-year-old woman brought to the emergency room with convulsions, following the ingestion of 20 ml of β -fluoroethylacetate solution 12 hours before in a suicide attempt.

She had used no medication previously. There was no history of convulsions or renal disease.

She was agitated without cyanosis. Chest examination revealed fine moist rales and arrhythmia without a murmur.

The temperature was 36.5°C, the pulse 86, and the respirations 24/min. The laboratory findings on admission were as follows: hemoglobin 13.2

gm%, hematocrit 37.2%, white blood cell count 6, 900, platelet count 26,200, total protein 6.5 gm%, and albumin 3.5 gm%.

The BUN was 32 mg% and creatinine was 2.0 mg%. Sodium was 158 meq/I and K was 6.7 meq/I. The findings on an arterial blood gas analysis were pH 6.83, pCO $_2$ 90 mmHg, HCO $_3$ 15 mmol/I and PO $_2$ 55 mmHg (Table 1).

Urinalysis revealed (+) proteinuria. Electrocardiogram showed atrial fibrillation.

Two hours after arrival, she expired.

DISCUSSION

In acidosis associated with mineral acids. acidemia, in which there is a predictable increase in serum K+ concentration occurs while in uncomplicated cases of acidosis associated with organic acids, the serum K+ concentration usually remains within the normal range. This difference is thought to be caused by the difference of membrane permeabilities between mineral acid and organic acid. Normally, there exist some mechanisms to maintain the H+balances between intracellular and extracellular fluid and some mechanisms to maintain the balanced K+ ratio. The change in the H+ ratio may cause the change in serum K+ concentration. Recently, Alder et al. 10) reported that there may be an undetermined relationship between intra-and extracellular H+/K+ ratio, but the definite relationship between serum pH and K+ concentration remains unclarified. In 1958 Tobin⁵⁾ infused HCI, NH₁CI and acetic acid intravenously into nephrectomized cats.

Table 1. Clinical Information and Biochemical Parameters in the Plasma on Admission

	Case 1	Case 2
Age, Sex	59. F	38. F
Amount of β -FEA	60ml	20ml
Onset to ER	2 hours	12hours
Chief complaints	semicoma and convulsion	convulsion
ECG findings	PVC	atrial fibrillation
pH in arterial	7.26	6.83
pCO ₂ mmHg	38	90
HCO ₃ mmol/I	17	15
Na meq/I	130	158
K meq	5.57	6.7
BUN mg%	24	32
Creatinine mg%	1.2	2.0
Outcome	Recovered	Expired after 2 hours

He reported that there was a significant increase in serum K+concentration in a HCl, NH₄Cl group after 2 hours and no change in that of the lactic acid and acetic acid groups. In 1980 Oster et al.11,12) infused HCI, NH4CI, and acetic acid into dogs. They reported an increase in serum K+ concentration in the HCl and NH₄Cl group but a decrease in it in the initial period in the lactic acid group. Cohen et al.13) mentioned in his review of a physiological phenomenon of lactate transport through cellular membrane, that lactic acid enters into cells in 2 forms, as ionized lactate and as unionized lactic acid. In 1983 Oh et al.14) reported the development of hyperkalemia in dogs in a severe acidosis state (pH 6.9, HCO₃ 6.0 meq/l) which was induced by giving lactic acid infusions. In 1984 Kang¹⁵⁾ induced acidosis by giving factic acid infusions to dogs for 4 hours. He reported that the serum K+ concentration decreased below the baseline for the first 2 hours, before it started increasing. He suggested that hyperkalemia might develop if severe acidosis induced by nonmineral acid were to have been prolonged further. There have been a few reports to the effect that hyperkalemia has developed in humans with organic acidosis, especially diabetic ketoacidosis or lactic acidosis.

The fluoroacetate¹⁶¹ which the patients had taken is a highly toxic, nonmineral acid which comes from the South African Plant, Dichapetalum Cymosum. It is intended to be used as a rodent-cide. The human lethal dose is 5 mg/kg. When administered to living subjects it interrupts the cellular metabolic process by blocking the citric acid cycle. Fluoroacetate itself is nontoxic, but once transformed into fluorocitrate acid it is highly toxic, causing generalized convulsions and severe ventricular arrhythmia.

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