

Synthesis of Carbazoles by a Diverted Bischler–Napieralski Cascade Reaction

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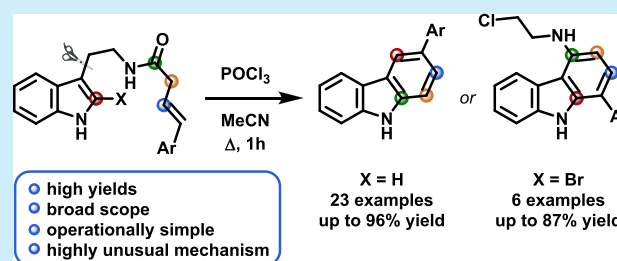
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ABSTRACT: An unforeseen twist in a seemingly trivial Bischler–Napieralski reaction led to the selective formation of an unexpected carbazole product. The reaction proved to be general, providing access to a range of diversely substituted carbazoles from readily available substrates. Judicious variation of substituents revealed a complex cascade mechanism comprising no less than 10 elementary steps, that could be diverted in multiple ways toward various other carbazole derivatives.



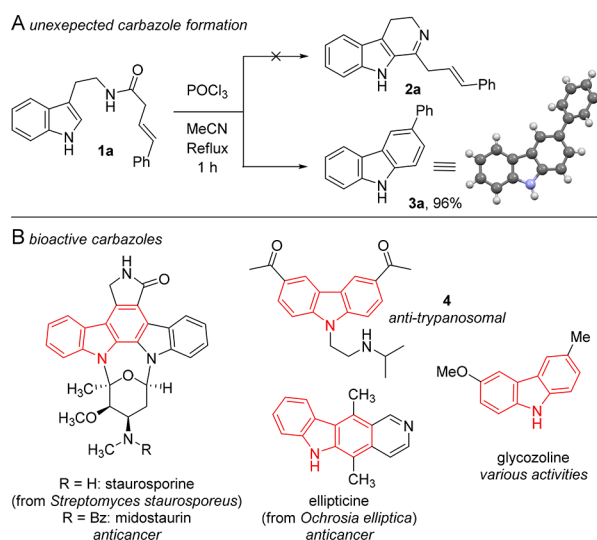
Since its first report in 1893, the Bischler–Napieralski reaction has been widely employed for the synthesis of dihydro- β -carbolines and -isoquinolines owing to its robustness and broad functional group tolerance.¹ Even currently, the Bischler–Napieralski reaction and its contemporary variations are still the object of intensive study in many areas, including natural product synthesis.² In light of our interest in bioactive indole alkaloids and related compounds,³ we employed the Bischler–Napieralski reaction to access a series of dihydro- β -carbolines. However, when we subjected styrylacetamide **1a** to typical Bischler–Napieralski conditions (POCl_3 , MeCN, reflux, 1 h) we serendipitously found near-quantitative formation of 3-phenylcarbazole (**3a**) instead of the expected dihydro- β -carboline **2a** (Scheme 1A). The structure of **3a** was confirmed by ^1H and ^{13}C NMR, HRMS, and X-ray crystallography.

Although carbazoles are less common than the related indoles among natural products and medicinal compounds, various carbazoles displaying interesting properties have been reported (Scheme 1B).⁴ Notable examples include the anticancer natural products staurosporine⁵ (and its clinically used semisynthetic derivative, midostaurin⁶) and ellipticine.⁷

Recently, carbazole derivative **4** was identified as a lead for new antitrypanosomiasis drugs,⁸ while glycozoline is known for its antibacterial, antifungal, antifeedant, and anti-inflammatory properties.⁹ Typical methods for the synthesis of carbazoles involve high temperature, long reaction times, and often metal catalysis (sometimes replaced by iodine or Lewis acids).^{10,11} Intrigued by our preliminary result, we decided to further explore the synthetic potential of this novel, mild, and metal-free route to carbazoles in more detail.

Puzzled by the surprising, but highly efficient formation of **3a**, we set out to investigate the generality of the process. A series of diversely substituted tryptamides **1a–t** was subjected to the reaction conditions (POCl_3 , MeCN, reflux, 1 h).

Scheme 1. (A) Unexpected Carbazole Formation; (B) Bioactive Carbazoles

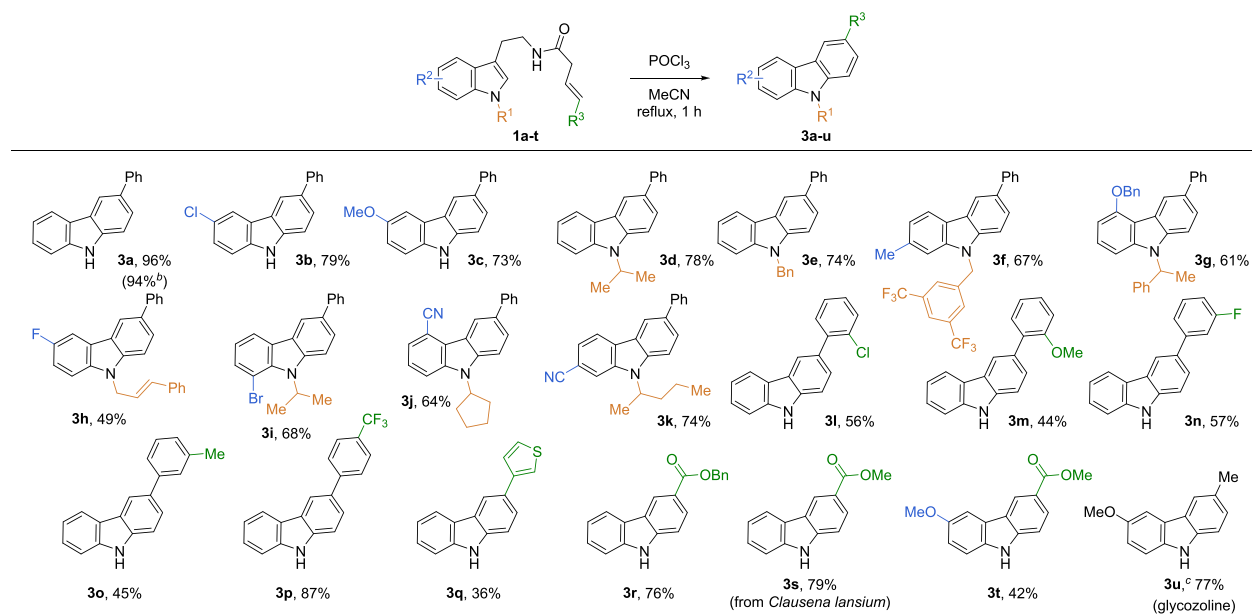


Pleasingly, we observed that all substrates underwent full conversion within 1 h (Scheme 2). Both electron-withdrawing and electron-donating substituents on the indole (R^2) are tolerated without significant influence on the yield, affording the corresponding products **3b–k** in mostly good yield, with 5-

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Scheme 2. Scope of the Carbazole Formation^a

^aAll reactions were performed with 0.2 mmol of **1a–t**, 0.3 mmol of POCl_3 , refluxing in MeCN for 1 h. ^bPerformed on a 2 mmol scale. ^cObtained by treatment of **2t** with LiAlH_4 .

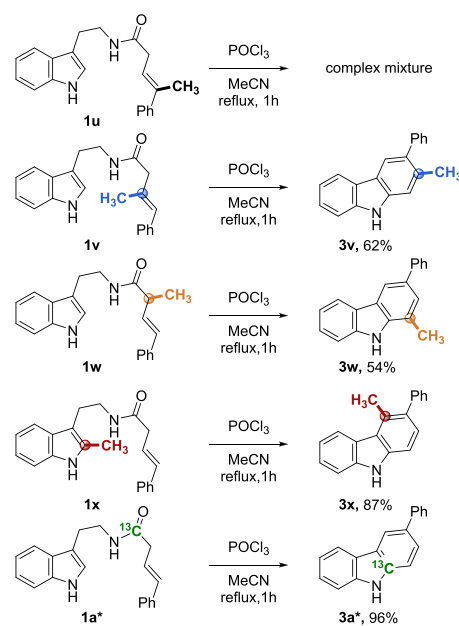
fluoro substitution giving the lowest yield (**3h**, 49%). Similarly, *N*-alkyl substituents had very little effect on the reaction outcome (**3d–k**). The effect of varying R^3 substitution is more significant. Electron-deficient arenes as R^3 substituents perform best (**3l**, **3n**, and especially **3p**). In contrast, products bearing an electron-rich aryl (**3m**, **3o**) or 3-thienyl R^3 substituent (**3q**) were obtained in lower yields. Interestingly, esters as the R^3 substituent were also able to promote the transformation, affording the corresponding carbazoles in high yield when the indole core is unsubstituted (**3r,s**, $\text{R}^1 = \text{R}^2 = \text{H}$) and in moderate yield when a 5-methoxy group is present (**3t**). Treatment of **3t** with LiAlH_4 afforded the natural product glycozoline (**3u**) which, together with **3s**, has been isolated from *Clausena lansium*.¹²

Once we established the generality of the reaction, we began our mechanistic investigation by the systematic variation of the substitution of the styrylacetic acid moiety in **1a** (Scheme 3). Reaction of the γ -methyl-substituted styrylacetic acid moiety in **1u** led to a complex reaction mixture containing traces of the corresponding regular Bischler–Napieralski product, but no carbazole derivatives. Reaction of the β -methyl-substituted substrate **1v** gave 2-methylcarbazole **3v**, while α -methyl-substituted styrylacetic acid moiety **1w** afforded 1-methylcarbazole **3w**.

These results may be rationalized by either transfer of the cinnamyl moiety to the indole C2 position or a complete rearrangement of the starting material involving ring opening of the indole moiety. The reaction of **1x**, bearing a methyl substituent at the indole C2 position, surprisingly afforded 4-methyl-3-phenylcarbazole (**3x**). The formation of **3x** can only be rationalized by a methyl migration or ring opening of the indole. Finally, we employed ^{13}C -labeled substrate **1a***¹³ and observed the incorporation of the ^{13}C label at the 9a position of carbazole **3a***.

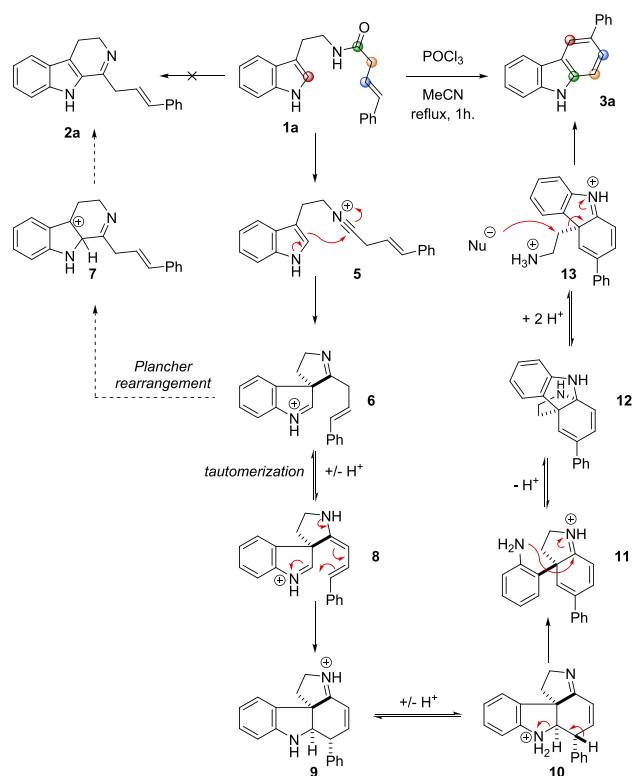
Based on the results summarized in Scheme 3 and relevant prior literature,¹⁴ we could postulate a mechanism to rationalize the formation of **3a** from **1a** (Scheme 4). Plausibly, the reaction is initiated by the formation of nitrilium ion **5**,

Scheme 3. Systematic Methyl Substitution



which undergoes attack by the indole C3 position to give spiroindolenine derivative **6**. In the Bischler–Napieralski reaction, **6** undergoes a rapid Plancher rearrangement, leading to dihydro- β -carboline **2a** after deprotonation of **7**. In this case, however, the presence of the styryl moiety makes tautomerization to **8** more favorable. The resulting vinylogous enamine attacks the protonated indolenine, leading to formation of the tetracyclic scaffold **9**. Then, β -elimination of the (protonated) aromatic amine takes place, opening up the indoline ring in **10**. The resulting aniline **11** subsequently undergoes imine transfer (via the bridged aminal **12**) to form the carbazole framework. The resulting dihydrocarbazole **13** finally undergoes attack by

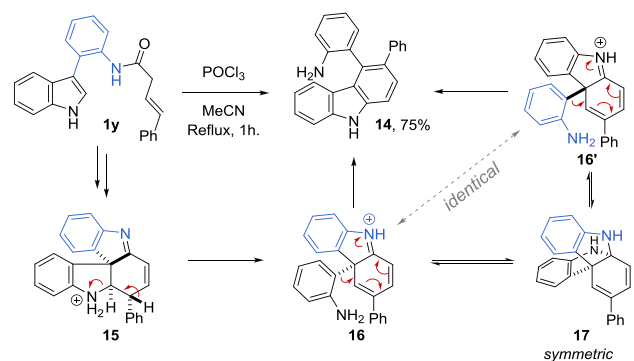
Scheme 4. Postulated Mechanism



an unidentified nucleophile (most likely chloride) to give **3a** with aromatization as a strong thermodynamic driving force.

Once we established a plausible mechanism, we realized that this complex, multistep transformation offers numerous opportunities for interruption or diversion of the reaction by judicious selection of substituents. First, we explored the possibility of diverting the cascade process by considering the equilibrium between **11**, **12**, and **13** that ultimately leads to the formation of **3a**. We reasoned that the nucleophilic attack that takes place on the sp^3 carbon of **13** could be avoided if the aliphatic linker is replaced by an aromatic one. Indeed, subjecting the phenylene-linked amide **1y** to the cyclization conditions afforded carbazole **14** in 75% yield (Scheme 5). Based on the above-mentioned considerations, we expected that the cascade would proceed analogously to the formation of **3a** until intermediate **17** and be interrupted at that stage. However, aromatization proved too great a driving force also in this case. As S_N2 substitution is not possible in this case (cf. **13**

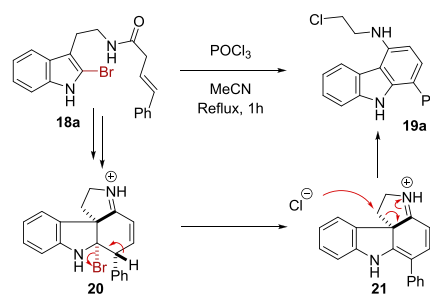
Scheme 5. Aromatic Linker Diversion



to **3a**, Scheme 4), the 1,2-aryl migration of the aniline fragment in **16** would re-establish the aromaticity of the system in the final stage. It is interesting to note that amination intermediate **17** has an internal mirror plane and the two iminium species **16** and **16'** are identical, thus leading to the formation of a single carbazole product (**14**).

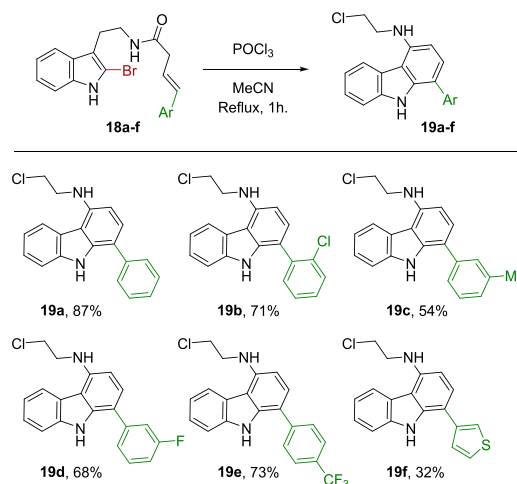
Next, we focused our attention on intermediate **10** (Scheme 4), the tetracyclic core of which is present in a variety of natural products.¹⁵ To target this scaffold, ring opening of the indole (leading to **11**, Scheme 4) needs to be prevented. Thus, we synthesized C2 Br-substituted styrylacamide **18a** to offer an alternative elimination pathway, interrupting the cascade at this stage. Indeed, the reaction of **18a** does undergo a diverted pathway; however, the product was again a carbazole (**19a**, Scheme 6), albeit with yet another surprising substitution

Scheme 6. C2 Bromide Diversion



pattern. The formation of **19a** could be rationalized by an alternative evolution of intermediate **20**. At this point, elimination of HBr is favored over indoline ring opening, leading to **21**. Similarly to the conversion of **13** to **3a** (Scheme 4), attack of a chloride anion would terminate the cascade to give carbazole **19a**.

We then proceeded to demonstrate the generality of this alternative transformation (Scheme 7). All desired products **19a–f** were obtained in moderate to very good yield, although we observed higher yields for products bearing electron-withdrawing substituents such as halogens and CF_3 (**19b**, **19d**, **19e**), yet the highest yield was observed for the unsubstituted

Scheme 7. Scope of C2 Bromide Diversion^a

^aAll reactions were performed with 0.2 mmol of **18a–f**, 0.3 mmol of $POCl_3$, refluxing in MeCN for 1 h.

product **19a**. In contrast, the presence of a methyl substituent led to a lower yield (**19c**), whereas replacing the phenyl ring with a thienyl moiety reduced the yield significantly (**19f**).

In conclusion, we report the serendipitous discovery of a diverted Bischler–Napieralski cascade reaction yielding carbazoles. The method features metal-free conditions, good yields, and high functional group tolerance. Systematic experimentation allowed us to confidently establish a complex multistep reaction mechanism, which allowed for straightforward further diversion or interruption of the reaction pathway to give different carbazole regioisomers. Efforts to further exploit the tetracyclic intermediates in the reaction in the total synthesis of indole alkaloids are currently ongoing in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c00785>.

Experimental details and characterization data, ¹H and ¹³C NMR spectra, 2D NMR spectra (PDF)

FAIR data, including the primary NMR FID files, for compounds **1a–y**, **3a–x**, **14**, **18a–f** and **19a–f** (ZIP)

Accession Codes

CCDC 2062984 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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