

Communication

Methane-Mediated Vapor Transport Growth of Monolayer WSe₂ Crystals

Hyeon-Sik Jang ^{1,†}, Jae-Young Lim ^{1,†}, Seog-Gyun Kang ¹, Sang-Hwa Hyun ², Sana Sandhu ¹, Seok-Kyun Son ³, Jae-Hyun Lee ^{2,*} and Dongmok Whang ^{1,*}

¹ School of Advanced Materials Science and Engineering and SKKU Advanced Institute of Nanotechnology (SAINT), Sungkyunkwan University (SKKU), 2066, Seobu-Ro, Jangan-Gu, Suwon-Si, Gyeonggi-Do 16419, Korea; dagu1821@skku.edu (H.-S.J.); limjyyy@skku.edu (J.-Y.L.); suggyoons@skku.edu (S.-G.K.); sanasandhu_7@yahoo.co.in (S.S.)

² Department of Energy Systems Research and Department of Materials Science and Engineering, Ajou University, Suwon, Gyeonggi-Do 16499, Korea; ruche@ajou.ac.kr

³ Department of Physics, Mokpo National University, Muan-gun, Jeollanam-Do 58554, Korea; skson@mokpo.ac.kr

* Correspondence: jaehyunlee@ajou.ac.kr (J.-H.L.); dwhang@skku.edu (D.W.); Tel.: +82-31-210-2465 (J.-H.L.); +82-31-290-7399 (D.W.)

† These authors contributed equally to this work.

Received: 28 October 2019; Accepted: 13 November 2019; Published: 19 November 2019



Abstract: The electrical and optical properties of semiconducting transition metal dichalcogenides (TMDs) can be tuned by controlling their composition and the number of layers they have. Among various TMDs, the monolayer WSe₂ has a direct bandgap of 1.65 eV and exhibits p-type or bipolar behavior, depending on the type of contact metal. Despite these promising properties, a lack of efficient large-area production methods for high-quality, uniform WSe₂ hinders its practical device applications. Various methods have been investigated for the synthesis of large-area monolayer WSe₂, but the difficulty of precisely controlling solid-state TMD precursors (WO₃, MoO₃, Se, and S powders) is a major obstacle to the synthesis of uniform TMD layers. In this work, we outline our success in growing large-area, high-quality, monolayered WSe₂ by utilizing methane (CH₄) gas with precisely controlled pressure as a promoter. When compared to the catalytic growth of monolayered WSe₂ without a gas-phase promoter, the catalytic growth of the monolayered WSe₂ with a CH₄ promoter reduced the nucleation density to 1/1000 and increased the grain size of monolayer WSe₂ up to 100 μm. The significant improvement in the optical properties of the resulting WSe₂ indicates that CH₄ is a suitable candidate as a promoter for the synthesis of TMD materials, because it allows accurate gas control.

Keywords: TMD; 2D material; WSe₂; monolayer; methane promoter; single-crystal

1. Introduction

The discovery of graphene and its unique properties has triggered the development of various types of layered materials [1]. In particular, transition metal dichalcogenides (TMDs), atomically thin semiconductors of the type MX₂ (M = Mo, W; X = S, Se), have attracted considerable attention as their physical and electrical properties are tunable. Depending on their composition and thickness, two-dimensional (2D) TMDs have a variety of electrical properties ranging from metal, to insulator, to semiconductor, which could lead to a new dimension of atomic thickness for future device applications [2,3]. TMD materials have useful device characteristics, such as a high on/off ratio, a wide range of photoluminescence, and a low subthreshold voltage, making them suitable for spintronics and

optoelectronics [4]. Among the numerous TMD materials, WSe₂ has been extensively studied because its electrical transport properties can be easily adjusted from p-type to bipolar behavior depending on the type of contact metal [5–7]. Bulk WSe₂ crystallizes in the “2H” or trigonal prismatic structure (space group *P*6₃/*mmc*; *a* = 0.330 nm, *c* = 1.298 nm), in which each W atom is surrounded by six Se atoms, defining two triangular prisms. It was also reported that the energy band structure of WSe₂ can be altered according to its layer number. WSe₂ shows a direct bandgap of 1.65 eV in the monolayer, compared to an indirect bandgap of 1.2 eV in the multilayered bulk [8,9]. Similar to another 2D layered material, TMD is typically prepared using a mechanical exfoliation method. However, this top-down approach is not suitable for practical high-performance device applications, so bottom-up approaches for large-scale and mass-production have been extensively studied. The chemical vapor deposition (CVD) method is one of the bottom-up approaches that allows the synthesis of large-area TMDs. The CVD growth of TMDs has largely been studied using two different approaches. The first approach is to pre-deposit transition metal sources such as MoO₃, WO₃, etc., on the growth substrate and convert them to TMD by sulfidation (or selenization) [10–18]. The second is a noncatalytic growth method, in which a transition metal source and sulfur (or selenium) are heat-treated in a growth tube and flowed in a gaseous state to synthesize the TMD layer on a target substrate [19,20]. However, these CVD approaches have not been successful in uniform, high-quality TMD synthesis because it is difficult to control the thickness and nucleation density of TMDs [21]. Recently, to overcome such problems, many researchers have studied various types of promoters and methods applied for CVD-based TMD synthesis to control gas-phase transport of precursors and the reaction of TMD on the growth substrate [22–26]. Ling et al. reported the synthesis of highly-crystalline MoS₂ at a relatively low growth temperature (650 °C) using various aromatic molecules as seeding promoters [13]. In particular, domain size of MoS₂ increased up to 60 μm through vaporized aromatic-molecule catalysts such as perylene-3,4,9,10-tetracarboxylic acid tetrapotassium salt (PTAS) and F₁₆CuPc. They also reported that uniform monolayer MoS₂ can be synthesized on the entire area of the SiO₂/Si substrate; however, the use of such an organic catalyst leaves a residue on the growth substrate that acts as a defect of the synthesized TMD. Another limitation of this method is that it is not applicable to the growth of WS₂ and WSe₂, which require high growth temperatures. In addition, inorganic materials were also reported in assisted WSe₂ growth methods [15,27]. Liu et al. demonstrated a Cu-assisted self-limited growth (CASLG) method that allowed the synthesis of a high-quality, uniform WSe₂ monolayer while maintaining a balance between high crystallinity and fast growth rates. They explained that Cu atoms, which occupy the hexagonal sites positioned at the center of the six-membered rings of the WSe₂ surface, induce self-limited growth of WSe₂ and prevent unwanted reactions [15]. However, this approach also had disadvantages, for example, the synthesized WSe₂ had small grain sizes with multilayered regions and the vapor pressure of the solid catalyst could not be precisely controlled.

Herein, we report a catalytic growth of the large-area monolayer WSe₂ by utilizing CH₄ (methane) with precisely controlled pressure as promoter. Through a systematic investigation, it is confirmed that grain size and the nucleation density of WSe₂ can be controlled according to the ratio of carrier gases (Ar/CH₄). The gas promoter leads to synthesis of about 100 μm size domains of WSe₂ and significantly reduces nucleation density from 1.6 × 10⁵ to 1.5 × 10² mm⁻². Various analytical tools such as Raman, photoluminescence (PL), X-ray photoelectron spectroscopy (XPS), and atomic force microscopy (AFM) analysis are used to demonstrate the properties of synthesized monolayer WSe₂.

2. Materials and Methods

2.1. Preparation

The WSe₂ precursor powders (Alfa Aesar, Ward Hill, MA, U.S., 99.8%; metal basis, 10 microns) were placed on the cleaned alumina boat. Prior to the growth, the SiO₂/Si wafer (thermal oxide wafer: 300 nm SiO₂ layer on Si (100), MTI Inc., Richmond, CA, U.S.) substrate was washed by acetone, ethyl alcohol, and deionized (DI) water, for 5 min, to remove the organic residue and was then treated with

oxygen plasma (100 sccm, 100 W). High-purity Ar gas (99.999%, JC gas Inc.) and methane gas diluted in Ar (1% CH₄, 99% Ar, JC gas Inc., Suwon-si, Gyeonggi-do, Korea) were used as carrier gases.

2.2. Synthesis of WSe₂

The homemade CVD system was designed to flow gas in both directions with a three-zone furnace and a double-quartz tube (outer: 34 mm diameter, inner: 15 mm diameter tube). The WSe₂ powders were placed in an alumina boat located at the center furnace of the homemade CVD. The SiO₂/Si substrate was cut to 1 cm × 5 cm size and then placed in the left furnace, about 10 cm from the alumina boat. The CVD system was pumped to the base pressure (2×10^{-3} torr) by a rotary pump for 10 min and then filled with Ar gas to 760 torr. In the process of increasing the temperature to the WSe₂ growth temperature, the flow direction of the carrier gas (Ar 200 sccm) was reversed to prevent unwanted deposition. After the temperature reached 1050 °C, the flow direction of the carrier gas was reversed again to allow the evaporated precursor to reach the growth substrate. In the synthesis process, the experiment was carried out by flowing a different ratio of Ar and CH₄ (1% diluted at Ar) for 60 min at atmospheric pressure. After the reaction, the furnace was quenched down to room temperature while maintaining the gas flow, and the samples were collected for characterization.

2.3. Characterization of Synthesized WSe₂

The morphology and size of synthesized WSe₂ samples were characterized using optical microscopy (OM, Olympus DX51, Tokyo, Japan) and a SEM (JEOL JSM-7401F, JEOL, LTD, Tokyo, Japan) operating at 5 kV and 10 μA. The nucleation density and grain size of WSe₂ were analyzed using the Image J program tool. The thickness and surface potential of the WSe₂ monolayer were confirmed by atomic force microscopy and Kelvin probe force microscopy (KPFM) using Park NX10 (Park system, Suwon-si, Gyeonggi-do, Korea) with a Si cantilever Pt-coated tip. X-ray photoelectron spectroscopy analysis was carried out by ESCA2000 spectrometry (Termo Fisher Scientific, Waltham, Massachusetts, U.S.) using monochromatic Al-Kα radiation (1468.6 eV). Raman and photoluminescence spectra were collected with micro-Raman spectroscopy (WITEC Raman system, Ulm, Germany) using a 532 nm laser.

3. Results and Discussion

As shown in Figure 1a, monolayer WSe₂ was synthesized on the SiO₂/Si substrate by a homemade three-zone furnace CVD using WSe₂ powder as a precursor. Briefly, the CVD system can control the temperature at each zone and adjust the direction of the carrier gas as required. During the ramping process for increasing the temperature of the furnace, the carrier gas flowed from the right to left direction to prevent the evaporated precursor from reaching the growth substrate, and the flow direction of the carrier gas was reversed during the growth process to synthesize the WSe₂ monolayer. A 1 × 5 cm² SiO₂/Si growth substrate was placed 10 cm away from the alumina boat containing the precursor. The growth behavior of WSe₂ was investigated by observing the product at the same location as the growth substrate, because the morphology and density of the WSe₂ crystals depended upon the distance between the precursor and the growth substrate [12,16]. Figure 1b illustrates the catalytic growth of WSe₂ crystals via vapor-solid transport mechanism, when CH₄ gas diluted in Ar (1% CH₄, 99% Ar) was used as both a carrier gas and a promoter. Like other catalysts for the growth of 2D materials, such as the Cu substrate commonly used for graphene growth, CH₄ induces the lateral epitaxy growth of WSe₂, increasing its grain size while suppressing its vertical growth or deposition. During the synthesis of WSe₂, methyl radicals and hydrogen are produced by thermal decomposition of CH₄ at the precursor hot zone (1050 °C) [28]. Methyl radicals can react with oxygen atoms on the SiO₂ surface to form O-CH₃, reducing the nucleation site of WSe₂. In addition, carbon-related radicals can react with the unstable W vapor to form metastable metallo-organic compounds, which may induce growth of low-defect WSe₂ crystals. Hydrogen is also known to induce the growth of low-defect WSe₂ crystals while suppressing vertical growth into bilayers and multilayers by etching defective WSe₂ [29–31]. Figure 1c,d show that while randomly distributed triangular WSe₂ crystals were grown,

the size, density, and thickness uniformity of the grown crystal domains varied significantly with or without CH_4 promoters. When WSe_2 was grown without CH_4 gas, grain size of the obtained domains was less than $1\ \mu\text{m}$ and there were many multilayer regions (Figure 1c). However, when CH_4 gas was used as a promoter, WSe_2 existed mostly as a monolayer with a grain size of more than $10\ \mu\text{m}$ (Figure 1d). These results clearly show that CH_4 gas acts as a promoter for the growth of WSe_2 crystals.

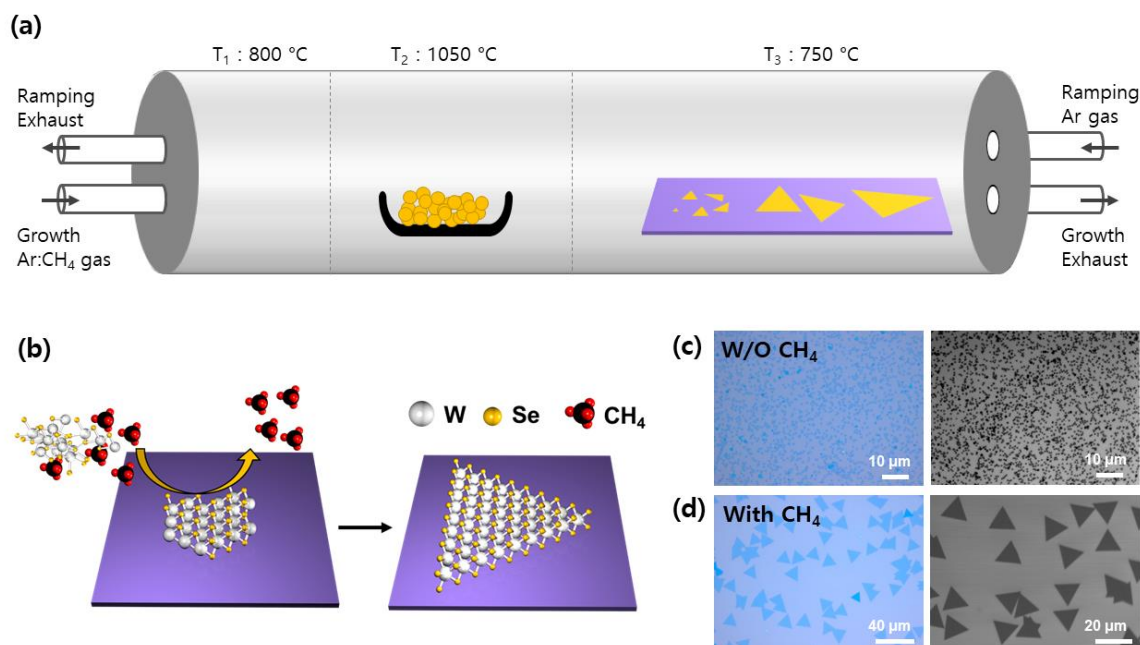


Figure 1. The schematic illustration of methane-mediated WSe_2 synthesis. (a) Sketch of homemade tube-type chemical vapor deposition (CVD) setup. (b) Schematic image of WSe_2 crystal growth by vapor-solid transport mechanism and its growth morphology difference between (c) without and (d) with methane (CH_4) gas.

As various parameters affect the CVD growth of TMDs, substrate size, carrier gas velocity, weights of precursor powders, growth time, and characterization regions were set as constant [8,10,13,18,19,22,26]. Based on this, Figure 2a–d show SEM images of WSe_2 according to the CH_4 gas ratio. Figure 2a and Figure S1 show that the WSe_2 grain size is less than $1\ \mu\text{m}$ when only Ar gas is used as the carrier gas. Figure 2a and Figure S1 also show some parts of the multilayer WSe_2 regions (dark-colored) with a nucleation density of $1.6 \times 10^5\ \text{mm}^{-2}$. By increasing the CH_4 gas to 50 sccm, the average grain size of WSe_2 was increased to $\sim 6\ \mu\text{m}$ with a triangular shape and a nucleation density of $5.5 \times 10^3\ \text{mm}^{-2}$ (Figure 2b and Figure S2). As the flow of CH_4 gas was increased to 100 sccm, the synthesized monolayer WSe_2 showed an average grain size of $9\ \mu\text{m}$ with a nucleation density of $6.8 \times 10^2\ \text{mm}^{-2}$ (Figure 2c and Figure S3). Figure 2d and Figure S3 show that the domain size of a single crystal monolayer of WSe_2 increased up to $80\ \mu\text{m}$ when flowing 150 sccm of diluted CH_4 gas. In this case, the average grain size was $52\ \mu\text{m}$ with a wide distribution due to a lower nucleation density of $156\ \text{mm}^{-2}$. From a statistical analysis of domain images in Figures S1–S4, grain size and nucleation density of WSe_2 were obtained as a function of the CH_4 gas ratio (Figure 2e and Figure S5). Generally, increasing the CH_4 gas ratio yielded a lower nucleation density of monolayer WSe_2 with a larger grain size. The catalytic effect of CH_4 on the synthesis of large-grain monolayer WSe_2 was similar to the catalytic growth of other 2D materials (graphene, h-BN, MoS_2 , WSe_2 , etc.) [11,15,32–36].

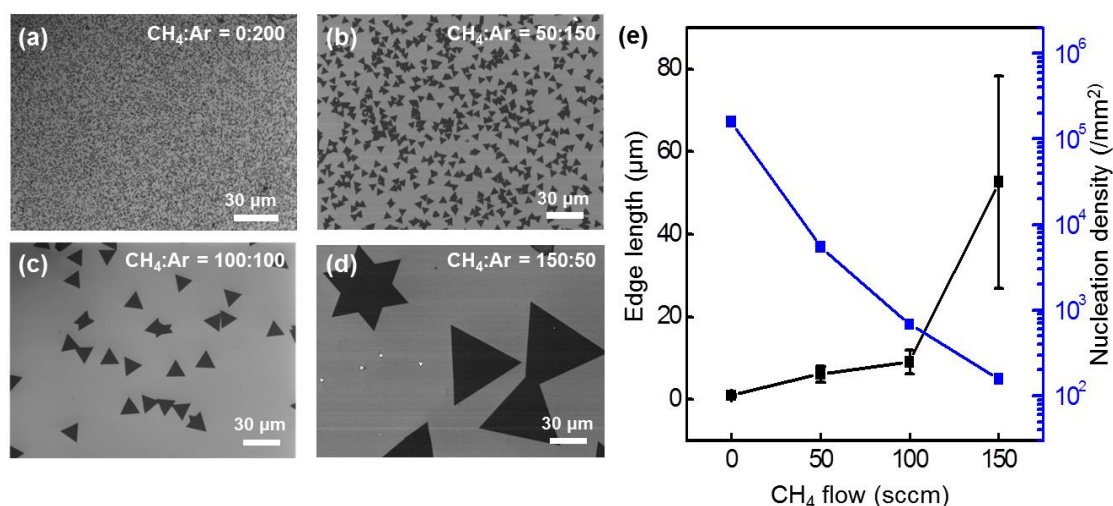


Figure 2. Size control of the WSe₂ domain by tuning the methane carrier gas ratio. Typical SEM images of WSe₂ grains synthesized on a SiO₂/Si substrate with a flow of (a) CH₄:Ar = 0:200, (b) CH₄:Ar = 50:150, (c) CH₄:Ar = 100:100, and (d) CH₄:Ar = 150:50 sccm. (e) Edge length (black) and nucleation density (blue) of WSe₂ domains as a function of the CH₄ gas ratio. The error bars represent the edge length variations of WSe₂ domains obtained at the same CH₄ gas flow.

We also investigated the effects of the CH₄ promoter on the morphological and optical properties of synthesized WSe₂ via the nondestructive analysis tools of Raman spectroscopy and PL. Figure 3a,b show the typical Raman mapping (at center wavelength: ~252 cm⁻¹) obtained with and without the CH₄ promoter, respectively. When CH₄ was used as a carrier gas, the grain size was about 80 μm with a uniform and strong intensity of E¹_{2g} peak over the synthesized WSe₂ crystals (Figure 3a). This result is consistent with the SEM results in Figure 2d. On the other hand, when only Ar was used as the carrier gas, the intensities of the measured E¹_{2g} peaks were much lower and nonuniform (Figure 3b). Figure 3c shows the differences in the typical Raman spectra of WSe₂ crystals grown with and without a CH₄ promoter. In the case of CH₄-assisted growth, Raman peaks corresponding to E¹_{2g} and A_{1g} modes of single-layered WSe₂ were observed (Figure S6). When only Ar gas was used, a relatively low E¹_{2g} peak and an additional small peak at 307 cm⁻¹ (corresponding to B¹_{2g} resonance mode of WSe₂) were observed. In general, the B¹_{2g} peak is only active on the bilayer or multilayer of WSe₂ [5,37]. We also noted that carbon-related Raman signals such as D peak (~1350 cm⁻¹), G peak (~1600 cm⁻¹), or 2D peak (~2700 cm⁻¹) were not observed. These results indicate that CH₄ acted only as a promoter during WSe₂ synthesis and did not leave other carbon-related residues. We noted that the WSe₂ growth temperature (700~750 °C) was too low to form a carbon layer by the reaction of methane on the surface of the SiO₂/Si substrate [38]. The optical properties of the synthesized WSe₂ and the effect of the CH₄ promoter were further investigated using micro-PL with a 532 nm laser.

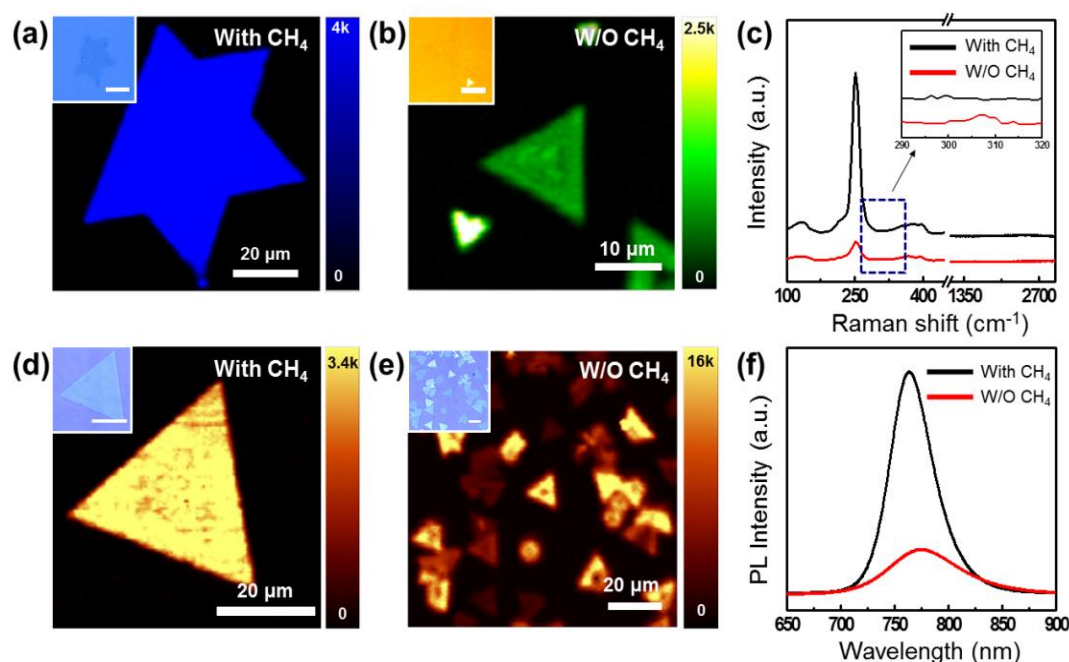


Figure 3. Raman and photoluminescence (PL) investigation of the synthesized WSe₂ according to the catalytic effects of methane. Raman mapping results of (a) CH₄:Ar = 150:50 sccm, (b) Ar gas only as carrier gas, and (c) representative Raman spectrum of each mapping result. PL mapping results of (d) CH₄:Ar = 150:50 sccm, (e) Ar gas only as carrier gas, and (f) representative PL spectrum of each mapping result. Raman and PL results were obtained from a micro-Raman instrument with a wavelength of 532 nm laser. Inset is an OM image corresponding to each mapping region.

Figure 3d shows the PL mapping of WSe₂ synthesized using a CH₄ promoter (CH₄:Ar = 150:50). The synthesized WSe₂ grain exhibited a uniform PL intensity at the 760 nm wavelength, which is equivalent to the PL value measured with exfoliated and synthesized single-crystal monolayer WSe₂ [5,37,39]. On the other hand, when only Ar (200 sccm) was used as a carrier gas, the PL of synthesized WSe₂ had a low intensity and showed a wide distribution due to the formation of bilayers and multilayers of WSe₂, as shown in Figure 3e. The synthesis effects of CH₄ gas were demonstrated from the representative PL spectrum of each PL mapping shown in Figure 3f. Based on these optical property data, it was confirmed that when using CH₄ as a promoter in the WSe₂ growth process, large WSe₂ grains with uniform monolayers can be synthesized.

As shown in the topology images obtained through tapping mode AFM, the thickness of the synthesized WSe₂ is uniform to ~0.7 nm, corresponding to the thickness of the monolayer (Figure 4a) [40,41]. A KPFM image of the monolayer WSe₂ showed a reduction in surface potential of ~300 meV in WSe₂ due to the electrostatic screening effect and charge distribution of WSe₂ (Figure 4b) [42]. The work function of the Pt-coated tip was ~4.3 eV, which was obtained by measuring the surface potential of highly oriented pyrolytic graphite (HOPG) (Figure S7). Since the work function of the SiO₂/Si substrate was 4.6 eV, the work function of the synthesized WSe₂ was estimated to be ~4.3 eV. This value is equivalent to the work function value of the exfoliated monolayer WSe₂ [43]. Figure 4c,d show the XPS results of the synthesized monolayer WSe₂ with four W-4f peaks (W⁴⁺4f_{7/2}: 32.8 eV, W⁴⁺4f_{5/2}: 34.8 eV, W⁶⁺4f_{7/2}: 36 eV, and W⁶⁺4f_{5/2}: 38.2 eV) and two Se-3D peaks (Se 3d_{5/2}: 55.1 eV and 3d_{3/2}: 55.9 eV). The two W⁴⁺4f peaks correspond to the binding energy of W bonded to Se atoms, while the two Se-3d peaks point to the binding energy of Se bonded to W atoms. The two W⁶⁺4f peaks correspond to the binding energy of the W atoms bonded to the O atoms, resulting from the exposure of the synthesized WSe₂ to air during the XPS analysis. Additionally, there was no W-4f peak at 32 eV and 34 eV, which represent the 1T phase; therefore, it can be confirmed at the WSe₂ of the 2H phase. These results are consistent with previous reports on WSe₂ [12,44].

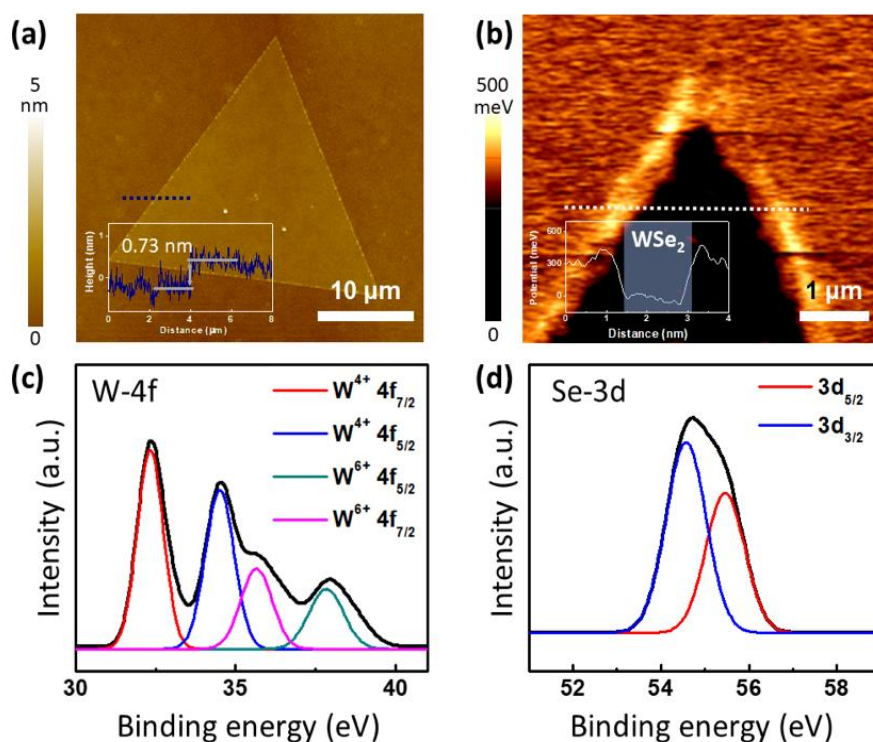


Figure 4. (a) Atomic force microscopy (AFM) image of the synthesized monolayer WSe₂. The inset graph is the height profile corresponding to the blue dot line. (b) Surface potential mapping image of WSe₂ by Kelvin probe force microscopy (KPFM). The inset graph is the surface potential profile corresponding to the white dot line. X-ray photoelectron spectroscopy (XPS) analysis of (c) W-4f and (d) Se-3d of synthesized WSe₂.

4. Conclusions

In summary, we developed a CH₄-assisted vapor transport growth method to obtain high-quality monolayer WSe₂ crystals with large domain sizes. Unlike other promoters or growth promoters previously reported (polymer, halide, and metal), CH₄ only acts as a promoter for WSe₂ growth without producing any residue. Moreover, the nucleation density of WSe₂ was tuned (from 1.6×10^5 to $1.5 \times 10^2 \text{ mm}^{-2}$) by using the gas-phase CH₄ promoter with precise flow control. The characterization of the synthesized monolayer WSe₂ by Raman, PL, and KPFM confirmed that CH₄ is a suitable candidate as a promoter for the growth of high-quality monolayer WSe₂. Finally, our CH₄-assisted growth approach may be applicable for the controlled growth of high-quality single crystals of other TMDs.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2079-4991/9/11/1642/s1>: Figures S1–S4—The SEM results at each synthesis condition for statistical analysis, Figure S5—Distribution of grain sizes in WSe₂ according to carrier gas ratios, Figure S6—Raman spectrum of synthesized monolayer WSe₂, Figure S7—The work function of Pt-coated AFM tip and reference HOPG, and Figure S8—XPS wide-range spectra of synthesized WSe₂.

Author Contributions: H.-S.J. and J.-Y.L. contributed equally to this work. H.-S.J. and S.-G.K. synthesized the samples. J.-Y.L. and S.-K.S. analyzed and performed the scanning electron microscopy, aAFM, KPFM, and XPS. H.-S.J. and S.-H.H. measured the optical properties (Raman and PL). J.-H.L. and D.W. guided the project and analyzed the results. H.-S.J., J.-Y.L., S.S., J.-H.L., and D.W. wrote the paper. All authors participated in reviewing the manuscript. All authors approved the final version of the manuscript.

Funding: This work was supported by the Korea Basic Science Institute (KBSI) National Research Facilities and Equipment Center (NFEC) grant funded by the Korean government (Ministry of Education) (No. 2019R1A6C1010031) and the National Research Foundation of Korea (NRF) grant funded by the Korean government (MSIP) (NRF-2019R1F1A1062330) and the Ajou University research fund.

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Geim, A.K. Graphene: Status and prospects. *Science* **2009**, *324*, 1530–1534. [[CrossRef](#)] [[PubMed](#)]
2. Ross, J.S.; Klement, P.; Jones, A.M.; Ghimire, N.J.; Yan, J.; Mandrus, D.G.; Taniguchi, T.; Watanabe, K.; Kitamura, K.; Yao, W.; et al. Electrically tunable excitonic light-emitting diodes based on monolayer WSe₂ p-n junctions. *Nat. Nanotechnol.* **2014**, *9*, 268–272. [[CrossRef](#)] [[PubMed](#)]
3. Baugher, B.W.; Churchill, H.O.; Yang, Y.; Jarillo-Herrero, P. Optoelectronic devices based on electrically tunable p-n diodes in a monolayer dichalcogenide. *Nat. Nanotechnol.* **2014**, *9*, 262–267. [[CrossRef](#)] [[PubMed](#)]
4. Zhang, H. Ultrathin Two-Dimensional Nanomaterials. *ACS Nano* **2015**, *9*, 9451–9469. [[CrossRef](#)]
5. Li, H.; Lu, G.; Wang, Y.; Yin, Z.; Cong, C.; He, Q.; Wang, L.; Ding, F.; Yu, T.; Zhang, H. Mechanical exfoliation and characterization of single- and few-layer nanosheets of WSe₂, TaS₂, and TaSe₂. *Small* **2013**, *9*, 1974–1981. [[CrossRef](#)]
6. Li, H.; Wu, J.; Yin, Z.; Zhang, H. Preparation and applications of mechanically exfoliated single-layer and multilayer MoS₂ and WSe₂ nanosheets. *Acc. Chem. Res.* **2014**, *47*, 1067–1075. [[CrossRef](#)]
7. Zhao, W.; Ghorannevis, Z.; Chu, L.; Toh, M.; Kloc, C.; Tan, P.H.; Eda, G. Evolution of electronic structure in atomically thin sheets of WS₂ and WSe₂. *ACS Nano* **2013**, *7*, 791–797. [[CrossRef](#)]
8. Terrones, H.; Lopez-Urias, F.; Terrones, M. Novel hetero-layered materials with tunable direct band gaps by sandwiching different metal disulfides and diselenides. *Sci. Rep.* **2013**, *3*, 1549. [[CrossRef](#)] [[PubMed](#)]
9. Fang, H.; Chuang, S.; Chang, T.C.; Takei, K.; Takahashi, T.; Javey, A. High-performance single layered WSe₂ p-FETs with chemically doped contacts. *Nano Lett.* **2012**, *12*, 3788–3792. [[CrossRef](#)]
10. Zhou, H.; Wang, C.; Shaw, J.C.; Cheng, R.; Chen, Y.; Huang, X.; Liu, Y.; Weiss, N.O.; Lin, Z.; Huang, Y.; et al. Large area growth and electrical properties of p-type WSe₂ atomic layers. *Nano Lett.* **2015**, *15*, 709–713. [[CrossRef](#)]
11. Gao, Y.; Hong, Y.L.; Yin, L.C.; Wu, Z.; Yang, Z.; Chen, M.L.; Liu, Z.; Ma, T.; Sun, D.M.; Ni, Z.; et al. Ultrafast Growth of High-Quality Monolayer WSe₂ on Au. *Adv. Mater.* **2017**, *29*, 1700990. [[CrossRef](#)]
12. Huang, J.; Yang, L.; Liu, D.; Chen, J.; Fu, Q.; Xiong, Y.; Lin, F.; Xiang, B.J.N. Large-area synthesis of monolayer WSe₂ on a SiO₂/Si substrate and its device applications. *Nanoscale* **2015**, *7*, 4193–4198. [[CrossRef](#)]
13. Ling, X.; Lee, Y.H.; Lin, Y.; Fang, W.; Yu, L.; Dresselhaus, M.S.; Kong, J. Role of the seeding promoter in MoS₂ growth by chemical vapor deposition. *Nano Lett.* **2014**, *14*, 464–472. [[CrossRef](#)]
14. Liu, B.; Fathi, M.; Chen, L.; Abbas, A.; Ma, Y.; Zhou, C. Chemical Vapor Deposition Growth of Monolayer WSe₂ with Tunable Device Characteristics and Growth Mechanism Study. *ACS Nano* **2015**, *9*, 6119–6127. [[CrossRef](#)]
15. Liu, J.; Zeng, M.; Wang, L.; Chen, Y.; Xing, Z.; Zhang, T.; Liu, Z.; Zuo, J.; Nan, F.; Mendes, R.G.; et al. Ultrafast Self-Limited Growth of Strictly Monolayer WSe₂ Crystals. *Small* **2016**, *12*, 5741–5749. [[CrossRef](#)]
16. Clark, G.; Wu, S.F.; Rivera, P.; Finney, J.; Nguyen, P.; Cobden, D.H.; Xu, X.D. Vapor-transport growth of high optical quality WSe₂ monolayers. *APL Mater.* **2014**, *2*, 101101. [[CrossRef](#)]
17. Elias, A.L.; Perea-Lopez, N.; Castro-Beltran, A.; Berkdemir, A.; Lv, R.; Feng, S.; Long, A.D.; Hayashi, T.; Kim, Y.A.; Endo, M.; et al. Controlled synthesis and transfer of large-area WS₂ sheets: From single layer to few layers. *ACS Nano* **2013**, *7*, 5235–5242. [[CrossRef](#)]
18. Li, M.-Y.; Shi, Y.; Cheng, C.-C.; Lu, L.-S.; Lin, Y.-C.; Tang, H.-L.; Tsai, M.-L.; Chu, C.-W.; Wei, K.-H.; He, J.-H. Epitaxial growth of a monolayer WSe₂-MoS₂ lateral pn junction with an atomically sharp interface. *Science* **2015**, *349*, 524–528. [[CrossRef](#)]
19. Park, J.; Choudhary, N.; Smith, J.; Lee, G.; Kim, M.; Choi, W. Thickness modulated MoS₂ grown by chemical vapor deposition for transparent and flexible electronic devices. *Appl. Phys. Lett.* **2015**, *106*, 012104. [[CrossRef](#)]
20. Zhan, Y.; Liu, Z.; Najmaei, S.; Ajayan, P.M.; Lou, J. Large-area vapor-phase growth and characterization of MoS₂ atomic layers on a SiO₂ substrate. *Small* **2012**, *8*, 966–971. [[CrossRef](#)]
21. Choi, W.; Choudhary, N.; Han, G.H.; Park, J.; Akinwande, D.; Lee, Y.H. Recent development of two-dimensional transition metal dichalcogenides and their applications. *Mater Today* **2017**, *20*, 116–130. [[CrossRef](#)]
22. Lee, Y.H.; Zhang, X.Q.; Zhang, W.; Chang, M.T.; Lin, C.T.; Chang, K.D.; Yu, Y.C.; Wang, J.T.; Chang, C.S.; Li, L.J.; et al. Synthesis of large-area MoS₂ atomic layers with chemical vapor deposition. *Adv. Mater.* **2012**, *24*, 2320–2325. [[CrossRef](#)]

23. Liu, K.K.; Zhang, W.; Lee, Y.H.; Lin, Y.C.; Chang, M.T.; Su, C.Y.; Chang, C.S.; Li, H.; Shi, Y.; Zhang, H.; et al. Growth of large-area and highly crystalline MoS₂ thin layers on insulating substrates. *Nano Lett.* **2012**, *12*, 1538–1544. [[CrossRef](#)]
24. Shi, Y.; Zhou, W.; Lu, A.Y.; Fang, W.; Lee, Y.H.; Hsu, A.L.; Kim, S.M.; Kim, K.K.; Yang, H.Y.; Li, L.J.; et al. van der Waals epitaxy of MoS₂ layers using graphene as growth templates. *Nano Lett.* **2012**, *12*, 2784–2791. [[CrossRef](#)]
25. Lee, Y.H.; Yu, L.; Wang, H.; Fang, W.; Ling, X.; Shi, Y.; Lin, C.T.; Huang, J.K.; Chang, M.T.; Chang, C.S.; et al. Synthesis and transfer of single-layer transition metal disulfides on diverse surfaces. *Nano Lett.* **2013**, *13*, 1852–1857. [[CrossRef](#)]
26. Wu, S.; Huang, C.; Aivazian, G.; Ross, J.S.; Cobden, D.H.; Xu, X. Vapor-solid growth of high optical quality MoS₂ monolayers with near-unity valley polarization. *ACS Nano* **2013**, *7*, 2768–2772. [[CrossRef](#)]
27. Li, S.S.; Wang, S.F.; Tang, D.M.; Zhao, W.J.; Xu, H.L.; Chu, L.Q.; Bando, Y.; Golberg, D.; Eda, G. Halide-assisted atmospheric pressure growth of large WSe₂ and WS₂ monolayer crystals. *Appl. Mater. Today* **2015**, *1*, 60–66. [[CrossRef](#)]
28. Cantelo, R.C. The Thermal Decomposition of Methane. *J. Phys. Chem. A* **1924**, *28*, 1036–1048. [[CrossRef](#)]
29. Permenov, D.G.; Radzig, V.A. Mechanisms of Heterogeneous Processes in the System SiO₂ + CH₄: II. Methylation of >Si=O Groups. *Kinet. Catal.* **2004**, *45*, 265–272. [[CrossRef](#)]
30. Zhang, F.; Su, D.; He, J.; Sang, Z.; Liu, Y.; Ma, Y.; Liu, R.; Yan, X. Methyl modified SiO₂ aerogel with tailored dual modal pore structure for adsorption of organic solvents. *Mater. Lett.* **2019**, *238*, 202–205. [[CrossRef](#)]
31. Li, X.; Li, X.; Zang, X.; Zhu, M.; He, Y.; Wang, K.; Xie, D.; Zhu, H. Role of hydrogen in the chemical vapor deposition growth of MoS₂ atomic layers. *Nanoscale* **2015**, *7*, 8398–8404. [[CrossRef](#)] [[PubMed](#)]
32. Lee, J.H.; Lee, E.K.; Joo, W.J.; Jang, Y.; Kim, B.S.; Lim, J.Y.; Choi, S.H.; Ahn, S.J.; Ahn, J.R.; Park, M.H.; et al. Wafer-scale growth of single-crystal monolayer graphene on reusable hydrogen-terminated germanium. *Science* **2014**, *344*, 286–289. [[CrossRef](#)] [[PubMed](#)]
33. Geng, D.; Wu, B.; Guo, Y.; Huang, L.; Xue, Y.; Chen, J.; Yu, G.; Jiang, L.; Hu, W.; Liu, Y. Uniform hexagonal graphene flakes and films grown on liquid copper surface. *Proc. Nat. Acad. Sci. USA* **2012**, *109*, 7992–7996. [[CrossRef](#)] [[PubMed](#)]
34. Wang, L.; Xu, X.; Zhang, L.; Qiao, R.; Wu, M.; Wang, Z.; Zhang, S.; Liang, J.; Zhang, Z.; Zhang, Z.; et al. Epitaxial growth of a 100-square-centimetre single-crystal hexagonal boron nitride monolayer on copper. *Nature* **2019**, *570*, 91–95. [[CrossRef](#)]
35. Chen, S.; Ji, H.; Chou, H.; Li, Q.; Li, H.; Suk, J.W.; Piner, R.; Liao, L.; Cai, W.; Ruoff, R.S. Millimeter-size single-crystal graphene by suppressing evaporative loss of Cu during low pressure chemical vapor deposition. *Adv. Mater.* **2013**, *25*, 2062–2065. [[CrossRef](#)]
36. Lee, J.S.; Choi, S.H.; Yun, S.J.; Kim, Y.I.; Boandoh, S.; Park, J.-H.; Shin, B.G.; Ko, H.; Lee, S.H.; Kim, Y.-M.; et al. Wafer-scale single-crystal hexagonal boron nitride film via self-collimated grain formation. *Science* **2018**, *362*, 817–821. [[CrossRef](#)]
37. Tonndorf, P.; Schmidt, R.; Bottger, P.; Zhang, X.; Borner, J.; Liebig, A.; Albrecht, M.; Kloc, C.; Gordan, O.; Zahn, D.R.; et al. Photoluminescence emission and Raman response of monolayer MoS₂, MoSe₂, and WSe₂. *Opt. Mater. Express* **2013**, *21*, 4908–4916. [[CrossRef](#)]
38. Pham, V.P.; Jang, H.S.; Whang, D.; Choi, J.Y. Direct growth of graphene on rigid and flexible substrates: Progress, applications, and challenges. *Chem. Soc. Rev.* **2017**, *46*, 6276–6300. [[CrossRef](#)]
39. Zeng, H.; Liu, G.B.; Dai, J.; Yan, Y.; Zhu, B.; He, R.; Xie, L.; Xu, S.; Chen, X.; Yao, W.; et al. Optical signature of symmetry variations and spin-valley coupling in atomically thin tungsten dichalcogenides. *Sci. Rep.* **2013**, *3*, 1608. [[CrossRef](#)]
40. Wang, K.; Huang, B.; Tian, M.; Ceballos, F.; Lin, M.W.; Mahjouri-Samani, M.; Boulesbaa, A.; Poretzky, A.A.; Rouleau, C.M.; Yoon, M.; et al. Interlayer Coupling in Twisted WSe₂/WS₂ Bilayer Heterostructures Revealed by Optical Spectroscopy. *ACS Nano* **2016**, *10*, 6612–6622. [[CrossRef](#)]
41. Almadori, Y.; Bendiab, N.; Grevin, B. Multimodal Kelvin Probe Force Microscopy Investigations of a Photovoltaic WSe₂/MoS₂ Type-II Interface. *ACS Appl. Mater. Interfaces* **2018**, *10*, 1363–1373. [[CrossRef](#)] [[PubMed](#)]
42. Li, F.; Qi, J.; Xu, M.; Xiao, J.; Xu, Y.; Zhang, X.; Liu, S.; Zhang, Y. Layer Dependence and Light Tuning Surface Potential of 2D MoS₂ on Various Substrates. *Small* **2017**, *13*, 1603103. [[CrossRef](#)] [[PubMed](#)]

43. Britnell, L.; Ribeiro, R.M.; Eckmann, A.; Jalil, R.; Belle, B.D.; Mishchenko, A.; Kim, Y.J.; Gorbachev, R.V.; Georgiou, T.; Morozov, S.V.; et al. Strong light-matter interactions in heterostructures of atomically thin films. *Science* **2013**, *340*, 1311–1314. [[CrossRef](#)] [[PubMed](#)]
44. Tao, L.; Meng, F.; Zhao, S.; Song, Y.; Yu, J.; Wang, X.; Liu, Z.; Wang, Y.; Li, B.; Wang, Y.J.N. Experimental and theoretical evidence for the ferromagnetic edge in WSe₂ nanosheets. *Nanoscale* **2017**, *9*, 4898–4906. [[CrossRef](#)] [[PubMed](#)]



© 2019 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).