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# Experimental Investigation of the Emission and Performance Characteristics of a DI Diesel Engine Fueled with the Vachellia nilotica Seed Oil Methyl Ester and Diesel Blends

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**ABSTRACT:** The rapid growth in industrialization steadily increased the energy demand. The world's population ultimately depends on petroleum as a major share of fuel for transportation and industrialization. Even though it is widely used in various sectors, its emission into the atmosphere creates serious problems in the form of acid rain, smog, etc. This present experimental investigation highlights the utilization of *Vachellia nilotica* seed oil methyl ester (VNSOME) synthesized from *Vachellia nilotica* seed oil (VNSO) fueled in a diesel engine to assess the emission and performance characteristics. VNSOME is produced using the alkaline catalyst (NaOH) transesterification technique. Four different fuel blends of biodiesel, namely, VNSOME5, VNSOME10, VNSOME15, and VNSOME20, were prepared and fueled in an unmodified engine. The engine brake thermal efficiency is lower, the brake-specific fuel consumption (BSFC) using VNSOME20 is higher, and the temperature



of exhaust gas emitted after combustion is increased. The thermal efficiency is reduced by 7.34% with increased BSFC and exhaust gas temperature (EGT) of 9.3 and 14.28%, respectively, as compared to diesel fuel. Similarly, using an optimized biodiesel blend (VNSOME20), the emission emitted such as HC and CO is reduced by 19.14 and 22.2%, respectively. However, the engine fueled with the VNSOME20 biodiesel blend increased the level of  $CO_2$  and  $NO_x$  emitted into the atmosphere when compared to diesel fuel.

## **1. INTRODUCTION**

Fossil fuels are largely used in industries, transportation, power plants, and automobiles. Industries and power plants use fossil fuels for power generation. Agricultural equipment such as pesticide sprayers and water pumps use compact engines, which are normally run by fossil fuels. Due to the rapid growth in the agricultural, industrial, and automobile sectors, the consumption of these fuels is increased. The demand for biofuels is rapidly increasing in view of depleting natural resources. These fossil fuels, especially diesel and petroleum, are extensively used for energy production. However, the emissions from the combustion of fuels are the principal causes of global warming and many environmental consequences. Various bio-oils are produced from biological resources, crops, byproducts from the forest, and feedstocks. The transesterification process or alcoholysis was the available best method to treat the oil to reduce the viscosity and remove the free fatty methyl ester acid and glycerol contents.<sup>1–</sup>

Biodiesel is produced from various feedstocks of plants, including karanja, jatropha, soya bean, lemon seed, pumpkin seed, neem, pongamia, rape seed oils, etc. Oil obtained from conventional and nonconventional sources is also used to produce biodiesel.<sup>9–25</sup> Ong et al.<sup>26</sup> optimized the production of *Calophyllum inophyllum* biodiesel and studied the feasibility of its utilization in an internal combustion engine. The biodiesel production was optimized using RSM, and the operating parameters that include catalyst concentration, molar ratio, reaction time, and reaction temperature on yield were analyzed. The composition of biodiesel in diesel fuel varied from 10 to 50%. The results showed that, using 10% *Calophyllum inophyllum* in diesel fuel, BSFC and EGT were lower than those in diesel fuel, whereas NO<sub>x</sub> was higher than that in diesel fuel. The characterization and optimization of *Calophyllum inophyllum–Ceiba pentandra* oil were investigated

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© 2021 The Authors. Published by American Chemical Society by Ong et al.<sup>27</sup> ANN and ACO techniques were used to optimize the process variables such as the methanol–oil ratio, catalyst concentration, and reaction time on biodiesel yield. The recent challenges and opportunities in the production of biodiesel derived from agricultural products and microalgae using ionic liquids were reviewed by Ong et al.<sup>28</sup> Transesterification using microwave irradiation on the synthesis of biodiesel produced from *Ceiba pentandra* oil was optimized by Silitonga et al.<sup>29</sup> The performance and emission characteristics of biodiesel–bioethanol and diesel blends in diesel engines using the K-Extreme learning method were studied by Silitonga et al.<sup>30</sup>

The influential effect of different Mimusops elangi methyl ester (MEME) fuel blends on the performance, emission, and combustion parameters of a DI diesel engine was studied by Krupakaran et al.<sup>31</sup> In their study, the MEME was synthesized by the process of transesterification, and the test fuel blends were prepared with diesel in the volume of 10% (90% diesel-10% MEME), 20% (80% diesel-20% MEME), 30% (70% diesel-30% MEME), 40% (60% diesel-40% MEME), and 100% MEME. The fuel properties were evaluated and validated with the limits of ASTM standards. Engine test results revealed that, on operating the engine in the full load with 20MEME, the engine exhibited 5.12 and 4.18% higher BSEC and BTE, respectively, when correlated with base fuel (diesel). Results also indicated that the inside cylinder pressure developed and rate of heat release were superior while using 20MEME biodiesel blend at the full load condition. A significant reduction of HC (5.26%), CO (16.6%), and smoke (6.2%) resulted in juxtaposition with base fuel despite marginally higher CO<sub>2</sub> (5.26%) and NO<sub>x</sub> (4.8%) at the maximum load of the engine operation.

The impact of oil derived from *Cymbopogon martinii* [Palmarosa methyl ester (PMO)] as an alternative fuel for a diesel engine was experimentally carried out by Sathiyamoor-thy et al.<sup>32</sup> The performance analysis showed that, compared to diesel, an extensive augmentation in BSFC was exhibited for PMO25 (7.41%), PMO50 (19.35%), and PMO100 (26.47%). The results of combustion revealed that the longer duration of combustion and shorter ignition delay were noticed with the use of PMO fuel blends, which are responsible for the considerable increment in the heat release and pressure of combustion. The test engine produces lower CO, HC, and smoke (19.02, 31.25, and 29.75%) for PMO100. Furthermore, a minimal increment in NO<sub>x</sub> (15.21%) was noticed for PMO blends contrasted with the base diesel fuel.

The impact of using *Pithecellobium dulce* biodiesel (PDBD) in lower concentrations blended with diesel fuel on the engine performance and emission characteristics was experimentally investigated by Sekhar et al.<sup>33</sup> On a volume basis, three fuel blends, namely, PDBD5, PDBD10, and PDBD20, were prepared and fueled in an unmodified diesel engine. On comparing the test results of PDBD blends over diesel fuel, it was revealed that the PDBD20 fuel blend produced lower HC (17.64%), CO (19.64%), and NO<sub>x</sub> (6.73%). Finally, it was contemplated that the PDBD20 diesel blend could be a successful fuel for use in an unmodified engine.

The extraction of biodiesel from *Oenothera lamarckiana* oil (OLO) in a diesel engine with an engine power of 9 kW was experimentally carried out by Hoseini et al.<sup>34</sup> to assess its performance and emissions. Two blends, namely, B10 and B20, along with diesel fuel, were tested with various engine loads. Results showed that, using the B20 blend, the SFC was

lower than that in diesel fuel by about 6.8%. Similarly, by operating the engine at the full load condition, the HC and CO are lower compared to those of diesel by about 37.28 and 10.13%, respectively. However, with the engine operating at the peak load, the other emissions emitted, such as  $CO_2$  and  $NO_{xy}$  were improved by about 7.9 and 4.6%, respectively.

Performance analysis on a diesel engine fueled with sandapple-based biodiesel blends was experimentally carried out by Ogunkunle and Ahmed.<sup>35</sup> Sand apple oil was blended in the volume concentration of 5, 10, 15, and 20 along with diesel fuel. The engine tests were implemented by changing the load of the engine from 0 to 100%. Results on performance revealed that increased BSFC and lower BTE were observed using the biodiesel blend.

The biodiesel synthesized from *juliflora* seed oil was exploited as an additional source of fuel in a DI diesel engine by Asokan et al.<sup>36</sup> In their study, the experiments were conducted with four different blends (B20, B30, B40, and B100) along with pure diesel. The test engine outcomes detected that the B20 fuel exhibits marginally higher BSFC (3.7%) and lower BTE (6.7%) at 100% load than those of diesel. In regard to emission, the B20 fuel generates minimum HC, CO, and smoke and slightly higher NO<sub>x</sub> than diesel fuel at 100% loading condition. From the conclusion, they recommended that the B20 fuel be utilized as a substitute for traditional diesel fuel.

Lalambari (LA) oil-based biodiesel and their blends (LA20, LA40, LA60, LA80, and LA100) were tested on a CI engine to study the effect on performance, combustion, and exhaust emissions. Test outcomes revealed that there are a devaluation in BTE (by 4.4%) and an increase in BSFC (by 2.5%) for LA20 than diesel fuel while the engine is operated at the peak load (3.7 kW). Also, pressure and heat release during the combustion were lesser for all the LA blends at 100% load. Furthermore, a reduction in smoke emission of about 16.6% and marginal increase in CO<sub>2</sub> of 3.5% were recorded for LA20 at 3.7 kW compared to diesel fuel. Finally, the test results proved that LA100 and their blends could be incorporated as an additional fuel source for CI engines.<sup>37</sup>

Vachellia nilotica belongs to a member of the Leguminosae family, which grows to around 15–18 and 2–3 m in height and diameter, respectively. Its color commonly determines the age of the bark. The slaty green color of the bark indicates that the tree is immature, while the black color indicates that the tree is mature. In a matured tree, longitudinal gaps uncovering the inward gray-pinkish slash, radiating a ruddy low-quality gum, were observed. The tree leaves are bi-pinnate, 3-10 sets and 1.3-3.8 cm long, leaflets of 10-20 sets and 2-5 mm in length. Flowers of the tree are in globulous heads with sparkling golden yellow shading and have a diameter of 1.2-1.5 cm. When the tree is immature, the pods are green and tomatoes, while the mature tree pods are dark greenish. Pods are indehiscent, profoundly choked between the seed and offering a necklace view and a length of 7-15 cm. The pod consists of 8-12 seeds, packed, elliptical, and misty brown colored with strong testa. The Vachellia nilotica can bloom and fruit in a few years (2-3 years) after the period of germination. After the high-precipitation years, the rapid growth of fruits occurs. Normally, blooming starts during the months of March and June, while the shaping of the pods happens between July and December. During the periods of June and November, the leaf falls after it is completely dry. The seed pods can drop from the tree between October and January.<sup>38,39</sup>

From the review of various literature, researchers are majorly focusing on various new-generation feedstocks to produce biodiesel. Researchers are also focusing on the various techniques to produce biodiesel along with the traditional method. The main objective of the study is to determine the physiochemical properties of Vachellia nilotica biodiesel and the feasibility of using the prepared fuel on performance and emission characteristics of diesel engine. This study mainly focuses on the extraction of biodiesel produced from Vachellia nilotica seed oil and the suitability of Vachellia nilotica seed oil methyl ester (VNSOME) in different concentrations blended into diesel fuel to assess the engine performance and emission produced during the combustion. A competitive study is carried out in the engine fueled with diesel, VNSOME5, VNSOME10, VNSOME15, and VNSOME20. As per the ASTM standards, the properties of biodiesel were estimated. It is also noted that the produced biodiesel is suggested as a novel feedstock as an alternative source for a diesel engine without any engine modification.

#### 2. MATERIALS AND METHODS

**2.1. Extraction of Vachellia nilotica Seed Oil (VNSO).** The Vachellia nilotica seeds utilized in this study are collected from the agriculture fields of rural villages nearby Tirunelveli District, Tamil Nadu, India. Initially, screening is done to remove the dust particles and undersized particles, and the seeds are thoroughly washed with water. The washed seeds are heated at an air temperature of 50 °C to remove the moisture content using an air blower. Then, the seeds are fed into a mechanical expeller to extract the Vachellia nilotica seed oil (VNSO). Further, the VNSO is refined by evacuating sticky substances and then neutralized for the betterment of its purity.

2.2. Preparation of VNSOME through a Biodiesel Production Plant. The VNSOME is produced through a small-scale biodiesel plant at CSIR-CLRI, Chennai, India. The detailed procedure is discussed below:

- The VNSOME is produced from the VNSO by a singlestage transesterification process due to the lower free fatty acid (2%) content.
- Sodium hydroxide (NaOH) is used as a catalyst and methanol is used as a solvent for the synthesis of VNSO into VNSOME.
- In the initial stages of the experimentation, the methoxide (CH<sub>3</sub>OH and NaOH) solution is prepared based on the free fatty acid in the VNSO.
- Then, 5 L of VNSO is fed into the reaction heater 1 (RH1), as shown in Figure 1. Then, it is heated up to 60 °C. After reaching 60 °C, the methoxide solution is fed into the RH1 (Figure 1).
- The mixture (VNSO and methoxide solution) is then maintained in the temperature range of 60–65 °C.
- The process is carried out until the formation of glycerol. The glycerol formation can be identified by the appearance of brownish color during circulation.
- After glycerol evolution, the admixture is allowed to settle for 3-5 h.
- The VNSOME produced is at the top of the RH1, and the glycerol is settled at the bottom, and it is drained out.
- Then, the VNSOME is fed into the water heater tank 1 (WH1), in which water is available at a temperature of



Figure 1. Biodiesel plant.

50 °C. Water washing of the product is performed in WH1, and then the VNSOME is fed into the drying heater tank 1 (DH1). In DH1, the VNSOME is heated up to 110 °C to remove the moisture content. Finally, the VNSOME is collected from the plant and stored in containers.

**2.3. Characterization of VNSOME.** The chemical characterization of VNSOME has been performed to quantify the methyl esters present in the product. The characterization of VNSOME is analytically studied using Fourier transform infrared (FT-IR) spectrometry (test method: ASTM E1252) and gas chromatography-mass spectrometry (GC-MS) (test method: ASTM E2997-16).

2.3.1. GC-MS of VNSOM. Gas chromatography analysis of VNSOME is carried out to evaluate the composition of methyl esters in VNSOME. The working conditions of GC-MS are shown in Table 1.

Table 2 shows the saturated and unsaturated methyl esters present in the VNSOME. From Table 2, it is clear that the VNSOME contains higher unsaturated fatty acid methyl esters

# Table 1. GC-MS Operating Conditions<sup>33</sup>

property	specification
injection	split ratio of 1:10 at 280 °C
column	capillary column Elite-5
column dimension	30 m $\times$ 0.25 mm i.d. $\times$ 250 $\mu m$ film thickness
carrier gas	helium
column flow rate	1 mL/min
detector	electron ionization
electron energy	70 eV
mass range	40-450 amu
oven temperature	initial temperature of 60 °C increased to 150 °C (hold for 2 min). Further, temperature is raised to 4 °C/min up to 280 °C and kept constant for 5 min with a total run time of 54.5 min.

Table 2. Fatty Acid Methyl Ester Composition of VNSOME

fatty acid methyl ester compound	molecular formula	composition (%)				
linoleic acid methyl ester	$C_{19}H_{34}O_2$	67.35				
elaidic acid methyl ester	$C_{19}H_{36}O_2$	10.22				
total unsaturated fatty acid meth	total unsaturated fatty acid methyl esters					
palmitic acid methyl ester	$C_{17}H_{34}O_2$	15.10				
isostearic acid methyl ester	$C_{19}H_{38}O_2$	5.18				
arachidic acid methyl ester	$C_{21}H_{42}O_2$	0.95				
behenic acid methyl ester	$C_{23}H_{46}O_2$	0.66				
lignoceric acid methyl ester	$C_{25}H_{50}O_2$	0.54				
total saturated fatty acid methyl	22.43					

(around 77.57%). Hence, the VNSOME is suitable for cold weather conditions.

2.3.2. FT-IR Spectrometry of VNSOME. The FT-IR spectrometry results of Vachellia nilotica seed oil and the functional group with peak are tabulated in Table 3, and Figure 2 shows the FT-IR spectrum of VNSOME. The Perkin Elmer RXIFT-IR spectrometer was used to develop the VNSOME spectrum.<sup>33</sup>

Table 3. Functional Group and Peak Identification

wave number	functional group
850-650	phenols of aromatic group
1300-1400	C–O alcohols
1500-1400	C-C
1750-1600	С=0
3000-2800	CH stretching of CH <sub>3</sub> and CH <sub>2</sub>
3400	OH, NH, and water impurities

It can be observed that the peak detected at 3400 cm<sup>-1</sup> clearly states the presence of -OH, -NH, and water impurities. Similarly, a stretch in -CH,  $CH_3$ , and  $CH_2$  is found between the wavelength of 3000 and 2800 cm<sup>-1</sup>. There is a peak at 2926 and 2854 cm<sup>-1</sup> suggesting the existence of  $CH_3$  and  $CH_2$ . The presence of >C=O is found in the wavelength between 1750 and 1600 cm<sup>-1</sup> as a peak is observed. The peak ranging from 1500 to 1400 cm<sup>-1</sup> classifies the C–C stretching vibrations. Also, the C–O vibrations can be noticed between the wavelengths of 1300 and 1400 cm<sup>-1</sup>. The peaks of phenols of aromatic group vibration can be noticed with wavelength ranging from 850 to 650 cm<sup>-1</sup>.

**2.4. Properties of VNSO and VNSOME.** The VNSOME is mixed with diesel fuel on volume basis of 5, 10, 15, and 20% by using a lab stirrer (Remi RQ-121/D) at 750 rpm for 2 h. The physiochemical properties of the diesel fuel, VNSO, and VNSOME are analyzed based upon the ASTM D6751-02 standards, and they are represented in Table 4.

The calorific value (lower heating value) of VNSO and VNSOME is lower than that of the diesel fuel due to the existence of oxygen particles in the structure. The flash point of VNSO and VNSOME is detected to be 237 and 185  $^{\circ}$ C, respectively, making the products secure for transportation. Copper strip corrosion shows the value of class 1a for VNSO and VNSOME and is responsible for lower corrosiveness on the engine parts. Finally, it was found that the reaming fuel properties of VNSO and VNSOME are within the standard limits of ASTM D6751-02.

**2.5. Test Engine and Experimentation.** An experimental investigation is conducted on a stationary direct injection (DI) multifuel engine [single cylinder, four strokes, and compres-



Figure 2. FT-IR spectra for VNSOME.

#### Table 4. Fuel Properties of VNSO and VNSOME

		Vachell	ia nilotica	Ceiba pe	entandra <sup>a</sup>	Papaver	somniferum L <sup>b</sup>		
property	diesel	VNSO	VNSOME	ССРО	CPME	RPSO	PSOME	limits	test method
kinematic viscosity at 40 $^\circ C (mm^2/s)$	2.40	40.52	4.505	18.74	4.69	5.03	4.13	1.90-6.0	ASTM D445
fire point (°C)	57	285	188	NA	NA	NA	NA		ASTM D93
flash point (°C)	46	237	185	186.5	158.5	234	172	130 min	ASTM D93
pour point (°C)	-12	7	-6	NA	-2	-14	-13	-15 to 10	ASTM D97
cloud point (°C)	0	12	5	NA	-3	-12	-15	-3 to 12	ASTM D2500
copper strip corrosion	1a	1a	1a	NA	1a	NA	NA	class 3	ASTM D130
specific gravity at 15 °C	0.825	0.885	0.860	0.906	0.883	0.919	0.874	0.86-0.90	ASTM D1298
cetane index	53	57	62			53	58	47 (min)	ASTM D976
calorific value (MJ/kg)	41.30	36.88	38.5	38.672	40.276	39.6	40.365	39-43	ASTM D240
ash content (% by mass)	0.012	0.0178	0.0165	NA	NA	NA	NA	0.02max	ASTM D129
carbon residue (wt %)	0.014	0.0260	0.0235	NA	NA	NA	NA	0.05max	ASTM D524
<sup><i>a</i></sup> Ref 29. <sup><i>b</i></sup> Ref 40.									

sion ratio (CR) of 17.5:1]. It is loaded with an electrical dynamometer (eddy current type), and the engine is operated at 1500 rpm (constant speed). The details of the experimental test rig used in the present study are tabulated in Table 5, and

#### **Table 5. Test Engine Specifications**

engine make	Kirloskar
capacity	661 cc
engine type	single-cylinder, four-stroke, and multifuel engine
rated power	3.5 kW (5 BHP)
compression ratio range	12 to 18
bore/stroke	87.5/110 mm
type of cooling	water
engine rated speed	1500 rpm
injection variation	0–25° bTDC
injection pressure	220 bar

Figure 3 shows the schematic diagram. On various engine loads, a series of experiments are carried out on the diesel engine using diesel fuel and biodiesel blends (VNSOME5, VNSOME10, VNSOME15, and VNSOME20). Exhaust gas temperature is measured using a K-type thermocouple attached to the tailpipe of the engine. The exhaust gas analyzer (AVL Digas 444) is connected to the tailpipe to determine the



Figure 3. Schematic diagram of the test engine setup.

emissions (CO<sub>2</sub>, NO<sub>x</sub>, HC, and CO) produced during combustion of the test engine. The smoke emissions are assessed by the smoke meter (AVL 437). The range and accuracy of the instruments to measure the emissions produced (exhaust gas analyzer and smoke meter) are tabulated in Table 6.

Table 6.	Range	and	Accuracy	of	Gas	Analyzer	and	Smoke
Meter	-							

parameter to be measured	range	resolution/accuracy
CO <sub>2</sub>	0-20 vol %	0.01 vol %
NO <sub>x</sub>	0-5000 ppm	1 ppm
HC	0–20,000 ppm	2 ppm
CO	0-10 vol %	0.01 vol %
smoke meter		
smoke density	0-100%	0.01%

#### 3. RESULTS AND DISCUSSIONS

**3.1. Performance Analysis of VNSOME Blends.** The performance analysis on a diesel engine operated at various loads using VNSOME blends and diesel fuel has been conducted, and the test results are discussed below:

The brake-specific fuel consumption (BSFC) of the engine using VNSOME fuel blends and diesel for different loads of engine operation is shown in Figure 4. For all the test fuels, the BSFC decreases as the loading is increased. On operating the engine at the maximum load, the BSFC of different concentration of biodiesel shows an increase of about 2.5, 4.8, 7.2, and 9.3% usingVNSOME5, VNSOME10, VNSOME15, and VNSOME20, respectively. Viscosity, specific gravity, and calorific value are the specific parameters of fuel properties that affect the BSFC of any type of diesel engine. From the results of BSFC, it is observed that BSFC from the engine fueled with VNSOME and its blends is higher than that with diesel fuel as the specific gravity and viscosity are higher.<sup>41,42</sup> Also, it is noted from the properties of VNSOME such as higher viscosity, density of the prepared fuel, and its blends that have a lower calorific value.

The BTE of the engine completely depends on the physical parameters such as evaporation, atomization, and combustion inside the cylinder. Figure 5 shows the disparity of brake thermal efficiency (BTE) related to different BP of the test engine at 1500 rpm. BTE is a function of BP and thermal input



Figure 4. BSFC variation of the test fuels.



Figure 5. BTE variation of the test fuels.

(product of volume flow rate and calorific value of the fuel). At a particular load of operation, the BP remains constant; hence, the BTE of the test fuels strongly depends on the calorific value.<sup>43–46</sup> It is found that the increase in the blend concentration of VNSOME with diesel fuel decreases the calorific value and it is less compared to that of diesel fuel. Therefore, the BTE is lower than the diesel for all the VNSOME blends. At the BP of 3.5 kW, BTEs of VNSOME5, VNSOME10, VNSOME15, and VNSOME20 are 1.97, 5.49, 6.37, and 7.34%, respectively, lower than the diesel.

The exhaust gas temperature (EGT) of using VNSOME fuel blends and diesel for different loads of engine operation is shown in Figure 6. With increased loading condition, the EGT of test fuels increases. Similarly, with the increase in the blend of VNSOME with diesel, the EGT significantly increases at all load conditions. The enriched oxygen content available in the VNSOME blends improves the process of combustion, which resulted in greater EGT.<sup>47</sup> At the maximum loading condition (3.5 kW), the EGTs of VNSOME5, VNSOME10, VNSOME15, and VNSOME20 are 3.22, 6.25, 9.63, and 14.28% higher than the EGT of the diesel fuel, respectively.

**3.2. Exhaust Emission Analysis of VNSOME Blends.** The engine exhausts emitted from the diesel engine such as



Figure 6. EGT variation of the test fuels.

hydrocarbons (HC), carbon monoxide (CO), oxides of nitrogen (NO<sub>x</sub>), smoke, and carbon dioxide (CO<sub>2</sub>) are analyzed for the VNSOME5, VNSOME10, VNSOME15, and VNSOME20 fuel blends in comparison with neat diesel, and the obtained results are discussed below:

The variation in carbon monoxide (CO) emission in regard to BP is outlined in Figure 7. CO emission is higher at lower



Figure 7. CO emission variation of the test fuels.

loads and is increased gradually at higher loads for all the test fuels. The possible increase in the CO emission at high loads is due to the rich fuel mixture present in the cylinder during combustion. The test outcomes revealed that the CO emission formed during the combustion with VNSOME fuel blends is lower than that with diesel fuel. At 3.5 kW, the VNSOME5, VNSOME10, VNSOME15, and VNSOME20 generates 5.55, 11.11, 16.66, and 22.22% lower CO emission when correlated to diesel fuel operation. The inbuilt oxygen composition in the VNSOME fuel blends improves the poststoichiometric combustion and thus lessens the CO emission.<sup>48,49</sup>

The hydrocarbon (HC) deviation using VNSOME fuel blends and diesel for different loads of engine operation is displayed in Figure 8. From the pictorial view, it is very clear that the VNSOME fuel blends produce lower HC emission



Figure 8. HC emission variation of the test fuels.

than diesel fuel. At the maximum load (3.5 kW), a reduction of 4.25, 8.51, 14.90, and 19.14% HC emission was depicted, respectively, for VNSOME5, VNSOME10, VNSOME15, and VNSOME20 in contrast with diesel. The oxygen composition in the VNSOME blends leads to enhance the combustion process, and it is responsible for the lower HC emission.<sup>50-52</sup>

The formation of oxides of nitrogen  $(NO_x)$  emitted after combustion using VNSOME fuel blends and diesel for different loads of engine operation is plotted in Figure 9.



**Figure 9.**  $NO_x$  variation of the test fuels.

From Figure 9, it can be detected that the formation of oxides of nitrogen (NO<sub>x</sub>) is higher with VNSOME biodiesel blends with increased engine load. Also, it is found that the formation of NO<sub>x</sub> using diesel fuel is lower than that using VNSOME fuel blends. The oxygen content already available in the VNSOME fuel blend is the main cause of the increased formation of NO<sub>x</sub>. The availability of oxygen content present in the biodiesel blend leads to complete and enhanced combustion that simultaneously increased the combustion temperature.<sup>53–55</sup> At 3.5 kW load, when compared to diesel fuel, the formation of NO<sub>x</sub> emission is found to be higher by about 1.94, 3.80, 5.60, and 7.34% for VNSOME5, VNSOME10, VNSOME15, and VNSOME20, respectively. The fluctuation in smoke emission for the VNSOME fuel blends and diesel for different loads of engine operation is displayed in Figure 10. With the possible increase in the engine



Figure 10. Smoke variation of the test fuels.

load, the smoke is also increased for all the test fuels since the smoke opacity strongly depends on engine load. In general, the smoke and  $NO_x$  are trade-offs. The smoke emission was reduced by 4.42, 7.37, 9.30, and 11.20% for VNSOME5, VNSOME10, VNSOME15, and VNSOME20, respectively, in contrast with the pure diesel fuel at the maximum load (3.5 kW). The reduction in C–H ratio, superfluous oxygen composition, and absence of sulfur content in the VNSOME blends are the important causes of the effective reduction in smoke emissions.<sup>49,56,57</sup>

The disparity of carbon dioxide  $(CO_2)$  emission of the test fuels (VNSOME5, VNSOME10, VNSOME15, VNSOME20, and diesel fuel) at different BPs is exhibited in Figure 11. When the load increases in the test, the engine emits higher  $CO_2$ emissions for all the test fuels. At the full load (3.5 kW) condition, the VNSOME5, VNSOME10, VNSOME15, and VNSOME20 blends produce 2.34, 4.58, 6.01, and 7.40% higher  $CO_2$  emission compared to diesel fuel. This may be due



Figure 11. CO<sub>2</sub> variation of the test fuels.

to the improved combustion of VNSOME blends than diesel fuel.  $^{49}$ 

## 4. CONCLUSIONS

In this study, an attempt was made to prepare the new biodiesel (VNSOME) fueled in a DI diesel engine for assessing the emission and performance characteristics. Based on the experimental results from the engine analysis, the conclusions are as follows:

- The blends of VNSOME in diesel fuel exhibited lower BTEs using VNSOME in different volume concentrations on diesel compared to neat diesel fuel. For VNSOME20, 7.34% marginally lower BTE is detected at 3.5 kW. Similarly, higher BSFC and EGT were found for all the VNSOME blends.
- The CO emissions produced from the engine tested with VNSOME in different volume concentrations are lower than that with neat diesel. At 3.5 kW, VNSOME20 emits 22.22% lower CO than that of diesel fuel.
- The formation of HC from the engine with different blends of VNSOME fuel is significantly lower than that with neat diesel. The VNSOME20 fuel blend generates 19.14% lower HC compared to diesel fuel at the maximum load of the engine.
- $NO_x$  emissions are marginally increased for VNSOME fuel blends. When compared to neat diesel, the  $NO_x$  formed is higher by about 7.34% than the VNSOME20 blend with the engine operated at the peak load.
- Using VNSOME fuel blends in the engine, the smoke emission is considerably reduced. There is an 11.20% reduction of smoke identified for VNSOME20 compared to diesel fuel at the full load of the engine.
- CO<sub>2</sub> emissions are slightly higher for all VNSOME fuel blends. At the full load (3.5 kW), VNSOME20 develops 7.40% higher CO<sub>2</sub> than diesel fuel.

**4.1. Future recommendations.** It can be concluded that the biodiesel fuel blend (VNSOME20) was found to be an alternative source for diesel fuel and the diesel engine can be fueled with this blend without making any modification.

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## **Author Contributions**

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

### Notes

The authors declare no competing financial interest.

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