



Research article

Effect of ultrasound and steam treatments on bioaccessibility of β -carotene and physicochemical parameters in orange-fleshed sweet potato juiceEvelyn Alicia Rios-Romero^a, Luz Araceli Ochoa-Martínez^{a,*}, Luis Arturo Bello-Pérez^b, Juliana Morales-Castro^a, Armando Quintero-Ramos^c, José Alberto Gallegos-Infante^a^a Tecnológico Nacional de México/I T de Durango, División de Estudios de Posgrado de Investigación, Blvd. Felipe Pescador 1830 Ote, Colonia Nueva Vizcaya, C.P. 34080, Durango, Dgo. Mexico^b Instituto Politécnico Nacional, Centro de Desarrollo de Productos Bióticos, Km. 8.5 Carretera Yautepec-Jojutla, C.P. 62731, Yautepec, Morelos, Mexico^c Universidad Autónoma de Chihuahua, Facultad de Ciencias Químicas, Campus II, Circuito Universitario s/n, C.P. 31125, Chihuahua, Chih, Mexico

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ABSTRACT

Several health benefits have been associated to orange-fleshed sweet potato owing to the existence of various bioactive compounds, including β -carotene. The purpose of this study was to establish the effect of ultrasound and steam treatment on the bioaccessibility of β -carotene, total polyphenols, antioxidant activity, polyphenol oxidase, peroxidase activity, and color in the orange-fleshed sweet potato juice. Sweet potato juice was processed using ultrasound (0.66 W cm⁻² and 8 min), steam (2 min), and a combination of steam and ultrasound. The bioaccessibility of β -carotene was increased in processed sweet potato juice, with samples treated by ultrasound showing the highest bioaccessibility (76.6%). Processing had no effect on antioxidant or enzyme activity, but resulted in significant changes in the color of the juice. As a processing technology, ultrasound enables preservation or improvement of the quality of sweet potato juice, and when combined with other treatments, facilitates the development of new products.

1. Introduction

Sweet potato (*Ipomoea batatas* L.) is known as one of the most important crops around the world, together with rice, wheat, potato, maize, and cassava. It has been cultivated due to its nutritious and health-benefits for humans (Wang et al., 2016). Sweet potato is commonly consumed in México, where it is cooked with molasses or consumed as crystallized candies. To take advantage of its nutritional value, there is a need to identify alternative uses of this product. *Ipomoea batatas* L. has high concentration of starch, and also contains dietary fiber, vitamins, minerals, and antioxidants, such as carotenoids, anthocyanins, phenolic acids, tocopherol (Teow et al., 2007). In orange-fleshed sweet potato, β -carotene is the main pigment, as well as a major precursor of vitamin A (Kidmose et al., 2007). The favorable effects of bioactive compounds are dependent on their bioavailability; therefore, it is necessary to evaluate the percentage of their liberation from the food source and their capacity to be absorbed through the intestinal wall (Hervert-Hernández et al., 2010). It is possible to examine the bioaccessibility of carotenoids by computing the quantity shifted to the micelle fraction following a simulated *in vitro* digestion methodology (Pugliese et al., 2013). Previous

studies have reported the retention and bioaccessibility of β -carotene in sweet potato and products based on this, subjected to different treatments (Bechoff et al., 2011; Failla et al., 2009; Trancoso-Reyes et al., 2016; Zhang et al., 2018). However, there have been no reports on sweet potato juice.

Juices from fruit and vegetables are commonly used as agents to carry high concentrations of bioactive compounds, since juices are an appropriate form for their intake (Anaya-Esparza et al., 2017). The selection of the processing method is crucial since it can be used to reduce the negative physical and chemical effects induced by excessive thermal treatments. Blanching is an essential phase in the processing of vegetables and vegetable commodities, including juices, to inactivate enzymes and microorganisms, to remove entrapped air, and preserve color (Bhat and Sharma, 2016). Enzyme inactivation, mainly of polyphenol oxidase, represents a quality parameter in the processing of fruits and vegetables. Previous reports have indicated that the inactivation of enzymes is difficult to achieve only through the use of ultrasound (Bi et al., 2015; Cheng et al., 2007; Ríos-Romero et al., 2018). Nevertheless, blanching can also negatively affect the heat sensitive nutrients, texture, water soluble components, and the value and biological activity of the final

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products. This depends on the thermal supply to the product. Alternative technologies are gaining popularity as food processing techniques, such as ultrasonic waves on juices that can not only maintain but also enhance the nutritional value (Jabbar et al., 2014). Ultrasound is a form of vibrational energy in the frequency range of 20–100 kHz with a sound intensity of 10–1000 W cm⁻² (O'Donnell et al., 2010). The benefits of ultrasound are attributed to acoustic cavitation, which consists of the formation, growth, and collapse of microbubbles due to pressure changes (Chemat et al., 2017). Ultrasound treatment can enhance improvements in food products, either by increasing or maintaining the nutritional quality, and increasing the bioaccessibility of bioactive compounds. In this context, the objective of this study was to determine the effect of ultrasound and steam treatment on the bioaccessibility of β-carotene, antioxidant activity, polyphenol oxidase and peroxidase activity, and color in sweet potato juice.

2. Materials and methods

2.1. Materials

HPLC type solvents (methanol, acetonitrile, and 2-propanol) were acquired from J.T. Baker (Baker-Mallinckrodt, Mexico). The β-carotene standard, hydrogen peroxide (H₂O₂), chlorogenic acid (5-cafeoylquinic acid), ferrous chloride (II) (FeCl₂), linoleic acid, 2,6-Di-tert methyl phenol (BHT), ammonium thiocyanate (NH₄SCN), pyrogallol, 1-2-dihydroxybenzene-butyl-4-(catechol) acid, 2-2'azobis (2-amino-propane) dihydrochloride (AAPH), 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox), fluorescein, porcine pepsin, α-amylase, and pancreatin were obtained from Sigma-Aldrich (St. Louis, MO, USA). The rest of substances used were of analytical brand. The orange-fleshed sweet potatoes variety was from the northeast region of Mexico and was bought in a local supermarket in Durango, Mexico. The same lot was used for all experiments. The roots were stored in a cold room (6 ± 2 °C) until processing.

2.2. Obtention of sweet potato juice

Sweet potatoes were washed, peeled, and cut in 4-cm-wide sections. In order to get the juice, the sweet potatoes were placed in an extractor (TURMIX® D.F. México). The resulting juice immediately was processed and subjected to three types of treatment: (i) ultrasound waves, applied with an ultrasonic device (UP200Ht; Hielscher Ultrasound Technology, Germany) attached to 40 mm diameter probe (treatment time of 8 min, ultrasonic intensity of 0.66 W cm⁻², and a constant frequency of 26 kHz) (Ríos-Romero et al., 2018); (ii) steam treatment, performed using a domestic steamer for 2 min; (iii) a combination of steam and ultrasound treatment. A sample of juice without treatment was physicochemically characterized and used as a control. Treated juices treatment were kept in the dark at -20 °C up to analysis. For the analysis of *in vitro* bioaccessibility, the processed sweet potato juices were lyophilized.

2.3. Physicochemical characterization (pH, soluble solids, titratable acidity)

The pH was measured using a potentiometer (Microprocessor pH Meter 212; HANNA Instruments, Romania). Measurements of soluble solids were determined as °Brix using a manual refractometer (ATAGO PAL-1; Tokyo Tech. Award, Japan) at 25 ± 1 °C. Titratable acidity (TA) was measured with sodium hydroxide (0.1 N) up to a pH of 8.2 ± 0.1. It was expressed as grams of citric acid per 100 mL of sample.

2.4. Extraction and analysis of β-carotene

The β-carotene extraction from the treated samples was performed as reported by Rios-Romero et al. (2018). Briefly, sweet potato juice (3 mL), 0.2 g of sodium carbonate and 20 mL of methanol were mixed and

filtered. Then, the solids were washed twice using 20 mL of methanol. After, the residue was washed three times with 15 mL of acetone/hexane (1:1, v/v) and BHT (0.1%). The supernatant from the series of washings was placed in a separation funnel containing 20 mL of sodium sulfate solution (10% w/v). Subsequently, 100 mL of distilled water was added and carefully mixed until 400 mL was completed. Then, 10 mL of petroleum ether was poured. Finally, the organic phase was separated and evaporated at 40 °C in a rotary evaporator. Subsequently, the samples were mixed with acetonitrile and methanol (85:15, v/v). The β-carotene content was measured by ultra-high-performance liquid chromatography (UPLC). The liquid chromatography system involved a sample administrator (5 °C) and a binary solvent administrator connected with a tandem Xevo TQ-S triple quadrupole mass spectrometer (Waters Corp., Milford, MA, USA). The column used was an Acquity UPLC HSS T3 (2.1 mm × 100 mm × 1.8 μm) (Waters Corp., Milford, MA, USA) at 35 °C. The gradient system with the mobile phase consisted of solvent A (acetonitrile:methanol) (85/15, v/v) and solvent B (2-propanol, 100%). The ramp program consisted of 95:5 (A:B) at initial conditions, 95:5 (A:B) from 0 to 2.0 min, 50:50 (A:B) from 2 to 3.5 min, and 95:5 (A:B) from 3.5 to 4.5 min, the flow rate was 0.5 mL min⁻¹. The atmospheric pressure chemical ionization conditions were as follows: positive polarity; Corona, 3.5 kV; cone, 30 V; source temperature, 150 °C; probe temperature, 550 °C; desolvation and cone gas, 550 L h⁻¹ and 150 L h⁻¹, correspondingly, and collision gas 0.15 mL min⁻¹. For the determination of β-carotene, the calibration curve was constructed. The method was linear in the range from 4 to 20 μg mL⁻¹ with a correlation coefficient of 0.9979. The UPLC and tandem Xevo TQ-S triple quadrupole mass spectrometer control and data processing were carried out using MassLinx (Water Corp., Milford, MA, USA) software.

2.5. *In vitro* digestion of β-carotene

The percentage of β-carotene bioaccessibility in the sweet potato juice was evaluated based on the standardized static *in vitro* digestion method suggested in a consensus report by Minekus et al. (2014) with some modifications. Samples of sweet potato lyophilized juice (1 g) were exposed to imitated oral, gastric, and small intestinal stages of digestion. Briefly, Simulated Salivary Fluid stock solution (3.5 mL) at pH 7, α-amylase solution of 1500 U mL⁻¹ (0.5 mL), CaCl₂ (25 μL, 0.3 M), and water (975 μL) were added to the sample. Afterwards the mixture was gently shaken at 37 °C for 2 min. Next, oral bolus was combined with 7.5 mL of simulated gastric fluid stock solution at pH 3, 1.6 mL of pepsin solution (25000 U mL⁻¹), and CaCl₂ (5 μL, 0.3 M) adjusted to pH 3 with hydrochloric acid 1 M, followed by the addition of water to obtain a final amount of 20 mL. The mixture was then shaken at 37 °C for 2 h. For the small intestinal stage, 10 mL of simulated intestinal fluid stock solution at 37 °C and pH 7, pancreatin solution 5 mL (800 U mL⁻¹ trypsin activity), bile extract (2.5 mL), and CaCl₂ (40 μL, 0.3 M) were mixed. The pH was settled to pH 7 with NaOH, 1 M and some water was supplemented to obtain a final amount of 40 mL. Finally, the sample was agitated at 37 °C for 2 h. The extraction of carotenoids was performed on the entire micellar fraction left, with the addition of diethyl ether (20mL) and NaCl (10mL, 10%, w/v). The mixture was agitated in a vortex for 1 min and then centrifuged at 4410 × g at 4 °C for 30 min. Then, the micellar fraction was moved to a beaker containing anhydrous sodium sulfate and concentrated by rotary evaporation. Quantification of β-carotene was performed as described in the previous section. The bioaccessibility percentage was calculated using Eq. (1):

$$\% \text{Bioaccessibility} = (\beta - \text{carotene digested}) / (\beta - \text{carotene in juice}) * 100 \quad (1)$$

2.6. Analysis of total polyphenols

Analysis of total polyphenols were performed by using the Folin-Ciocalteu method (Singleton and Rossi, 1965). Twenty five microliters

Table 1. Physicochemical characteristics of sweet potato juice under different treatments.

Process	pH	Titrateable acidity (%)	Soluble solids (° Bx)
Control	6.00 ± 0.01 ^a	0.093 ± 0.006 ^a	15.8 ± 0.1 ^a
Ultrasound	5.96 ± 0.03 ^a	0.093 ± 0.006 ^a	15.8 ± 0.0 ^a
Steam	5.99 ± 0.03 ^a	0.096 ± 0.007 ^a	15.8 ± 0.1 ^a
Steam + Ultrasound	5.95 ± 0.03 ^a	0.098 ± 0.006 ^a	15.7 ± 0.1 ^a

Mean ± standard deviation followed by different superscript letter in the same column indicate that values are significantly different ($p < 0.05$.) by Tukey's test.

of the juice were added in a 96-well microplate, then 80 μ L of distilled water and 5 μ L of Folin-Ciocalteu reagent were poured. The above mixture was left to rest at room temperature for 5 min. After, 80 μ L of sodium carbonate (7.5% w/v) was supplemented and allowed to stand for 30 min. The absorbance was immediately determined at 750 nm using a microplate reader (Daigger Automated Microplate Reader). The resulted values were reported as mg equivalents of chlorogenic acid per gram of dry matter.

2.7. Analysis of antioxidant activity (FTC and ORAC)

The antioxidant activity measured by ferric thiocyanate (FTC), which determine the oxidation of a lipid system, was carried out according to Rumbaoa et al. (2009). For the reaction, the sample (1 mL) was mixed with linoleic acid solution (1 mL, 2.51% v/v) in ethanol (99.5%), 2 mL of 0.05 M phosphate buffer (pH 7.0), and distilled water (1 mL). The mixture was kept in the dark in a water bath at 40 °C. The control was prepared with 1 mL of methanol instead of sample. Afterwards, 0.1 mL from the mixture was taken, and ethanol (9.7 mL, 75% v/v) was added. Then, 0.1 mL of NH₄SCN (30% v/v), and 0.1 mL of FeCl₂ (20 mM) in HCl (3.5%, v/v) were added. Finally, after 3 min, the absorbance was recorded at 500 nm using a spectrophotometer (DR5000; HACH, Loveland, CO). This process was replicated every 24 h until the solution attained the maximum level of absorbance. The percentage of inhibition was calculated using Eq. (2):

$$\text{Activity (\%)} = (1 - (\Delta A_{\text{sample}} / \Delta A_{\text{control}})) * 100 \quad (2)$$

The oxygen radical absorbance capacity (ORAC) method was performed as stated by Garrett et al. (2010). The needed concentration of the AAPH solution was prepared at 79.65 mmol L⁻¹. For fluorescein solution, 22.5 mg were dissolved in 50 mL of phosphate buffer (75 mM, pH 7.4). A second stock of fluorescein solution was prepared with 50 μ L of the first solution and 10 mL of phosphate buffer. Then, 320 μ L of the second solution and 20 mL of the phosphate buffer were mixed. Finally, each hole of the 96-well black microplate was filled with 20 μ L of sample and 200 μ L of fluorescein solution. The sample and the fluorescein were stored at 37 °C for 15 min, the reading of fluorescein solution was taken as the zero time. The reaction initiated when 75 μ L of AAPH was poured to each well. The measurements were taken using a microplate reader (SYNERGY HT, Bio-Tek Instruments, Inc., Winooski, VT) with excitation wavelength of 485 nm and with emission wavelength of 528 nm every 3.5 min until 35 cycles were completed. The ORAC results were reported as μ mol equivalents of Trolox g⁻¹ of dry matter.

2.8. Analysis of enzymatic residual activity (polyphenol oxidase and peroxidase)

The polyphenol oxidase activity (PPO) was analysed in the sonicated samples according to the method reported by Abid et al. (2014a, b). The samples were centrifuged at 1000 × g for 10 min. The mixture for the reaction consisted of 1.5 mL of sample, 0.5 catechol (3 mL, 0.05 M), and potassium phosphate buffer (3 mL, 0.2 M, pH 6.8). The increment in the absorbance at 410 nm was determined at time zero and every 30 s for 10 min in a spectrophotometer (DR5000; HACH, Loveland, CO).

The peroxidase activity (POD) was measured using the methodology presented by Kwak et al. (1995) with pyrogallol as a substrate. Briefly, the samples of sweet potato juice were centrifuged at 1000 × g for 10 min. The mixture for the reaction consisted of 0.10 mL of the centrifuged sample, water (2.1 mL), potassium phosphate buffer (0.32 mL, 100 mM, pH 6.0), pyrogallol (0.32 mL, 5% v/v) and 0.16 mL of H₂O₂ (0.50% v/v). The absorbance was measured at 420 using a spectrophotometer (DR5000; HACH, Loveland, CO).

2.9. Analysis of color

The color determination of the juices was measured using a colorimeter (Hunter Lab Color Flex EZ 45/0; Hunter Associates Laboratory Inc., Virginia, USA) which is based on three color axes, called L*, a*, and b*, at room temperature. The device was adjusted with white and black tiles as reference. The values of color were expressed as L* (brightness/darkness), a* (red/green), and b* (yellow/blue). In addition, the total color difference (TCD), chrome, and hue angle were calculated using Eqs. (3), (4), and (5), respectively, where L₀, a₀, and b₀ are the values for the juice used as control.

$$TCD = ((L^* - L_0)^2 + (a^* - a_0)^2 + (b^* - b_0)^2)^{1/2} \quad (3)$$

$$Chrome = (a^{*2} + b^{*2})^{1/2} \quad (4)$$

$$Hue\ angle = \tan^{-1}(a^* / b^*) \quad (5)$$

2.10. Statistical analysis

The experiments were established under a complete random unifactorial design. All the treatments were performed in duplicate, and the variable compounds were analyzed three times. The obtained values were expressed as the mean ± standard deviation. The results were analyzed by one-tail analysis of variance (ANOVA) and the difference between means was analyzed using Tukey's test with a significance of $p < 0.05$.

3. Results and discussion

3.1. Physicochemical characterization of sweet potato juice

According to the physicochemical analysis of the sweet potato juice, the pH of juice was 6.0 ± 0.01. The total soluble solids were 15.8 ± 0.1 °Brix, while the percentage of titrateable acidity of the sweet potato juices was 0.093 ± 0.006. Table 1 presents the physicochemical characterization of the sweet potato juice after processing. The juice samples showed no changes in the pH, acidity, and soluble solid parameters.

3.2. Effect of treatment on β -carotene and bioaccessibility

Table 2 shows the β -carotene content in sweet potato juice subjected to various processing methods. Ultrasound process improved the extraction of β -carotene by 21.3% with respect to the control sample. The juice treated with the combination of steam and ultrasound presented an

Table 2. Effect of processing on the content and bioaccessibility of β -carotene in sweet potato juice.

Process	β -carotene ($\mu\text{g g}^{-1}$)	Bioaccessibility (%)
Control	246.3 \pm 16.7 ^{ab}	10.3 \pm 0.7 ^a
Ultrasound	298.9 \pm 28.2 ^c	76.6 \pm 5.2 ^c
Steam	239.2 \pm 22.9 ^a	15.9 \pm 3.2 ^b
Steam + Ultrasound	286.8 \pm 13.5 ^{bc}	34.1 \pm 3.4 ^d

Mean \pm standard deviation followed by different superscript letter in the same column indicate that values are significantly different ($p < 0.05$.) by Tukey's test.

increase of 16.4% in the β -carotene content in respect to the control sample. Several authors (Abid et al., 2014a, b; Jabbar et al., 2014; Santhirasegaram et al., 2013; Rios-Romero et al., 2018) have pointed out similar increases in the content of carotenoids in sonicated juices attributed to the blow up of the cell tissues. Treatment with steam resulted in a non-significant change of 2.8%, which may be due to the light processing conditions (only 2 min).

The bioaccessibility of carotenoids from vegetable commodities is the main consequence from the disruption of cells during preparation, such as processing and chewing (Zaccari et al., 2015). The effect of treatment in the percentage of bioaccessibility is presented in Table 2. A notable increase in the bioaccessibility of β -carotene was detected in the processed samples. Samples processed with ultrasound showed the highest increase in the percentage of bioaccessibility, reaching a value of 76.64%. The improvement in the bioaccessibility of β -carotene could be because the acoustic waves, which, under the given conditions, did not damage the structure of the β -carotene. However, when the steam and ultrasound treatments were combined, β -carotene could undergo thermal and chemical oxidation, decreasing its bioaccessibility. Food processing modifies the structure of the food matrix, which may cause changes in the bioaccessibility of their bioactive compounds. Specifically, ultrasound treatment may improve the bioaccessibility of carotenoids, as reported by Campoli et al. (2018) in sonicated guava juice, they reported an important increase in lycopene bioaccessibility. Additionally, Gille et al. (2019) reported that the ultrasound extraction of carotenoids from *P. tricornutum* biomass caused a significant increase of up to three-fold in the bioaccessibility of β -carotene. Similarly, Mercado-Mercado et al. (2018) observed that the ultrasound treatment of mango by-products improved the bioaccessibility of β -carotene, with an increase of 32.6% in mango peel and 44.0% in mango paste compared with the control sample. The mechanisms proposed for the liberation of carotenoids are as follows: (1) the modification of the cell by the cavitation phenomenon; (2) breakdown of the glycosidic linkages in dietary fiber; (3) the exposure of ester bonds, lipids, and peptides to the outside; (4) an increment in the enzyme activity of pepsin and pancreatin; (5) hydrolysis of α ,1-4 glycosidic bonds present in the amylose and amylopectin chains; (6) decrease in the viscosity of digestible fiber in the intestinal fragment; (7) liberation of carotenoids present in soluble digestible fiber (Mercado-Mercado et al., 2018).

3.3. Processing effect on total polyphenols, antioxidant activity, and enzymatic activity

The effect of ultrasound and steam processing on the total polyphenols in the sweet potato juices is indicated in Table 3. During the

cavitation phenomenon, the formation of free radicals results in a reduction in the amount of bioactive compounds. This was detected in the samples of sweet potato juice treated with ultrasound (a reduction of 11.9% in the total polyphenols content). In addition, the degradation of these compounds may occur as a result of oxidation caused by the enzymes released in the process but which were not inactivated. In previous studies, Dias et al. (2015) and Fonteles et al. (2012) also communicated a reduction in the content of these compounds. The results concerning the effect of steam and the combination of steam and ultrasound treatment on sweet potato juice showed that phenolic compounds were not significantly different. Ma et al. (2013) reported that blanching in a water bath at 86 °C for 10 min increased the total polyphenol content, where blanching assisted in the dissolution of polyphenols into the juice. Bhat and Sharma (2016) evaluated the effect of combining blanching and ultrasound treatment on total polyphenols in gourd juice and found that the extraction and preservation of total polyphenols was appreciably higher in the processed samples than in the control sample.

The antioxidant activity of sweet potato juices is shown in Table 3. Due to its lipophilic characteristics and its capacity to scavenge peroxy radicals, the ferric thiocyanate method, which has the ability to capture peroxy radicals and to react with polyunsaturated fatty acids, was used to measure the antioxidant activities of carotenoids in sweet potato juice. The inhibition of linoleic acid oxidation in sweet potato juices processed with ultrasound and steam presented insignificant changes compared to the control sample. This insinuate that the existence of non-phenolic components aided to the antioxidant activity of the juice. The ORAC method was used to evaluate the antioxidant activity attributed to the polyphenolic compounds. This assay tests the ability of antioxidants present in juice to preserve fluorescein from oxidative injury. According to the results obtained using the ORAC method, the antioxidant activity of the treated samples showed insignificant changes in comparison to the control sample. The antioxidant activity was mainly attributed to non-phenolic compounds, such as β -carotene, but also to the phenolic compounds present in the juice. In a previous study, Rios-Romero et al. (2018) reported the same tendency in sweet potato juice sonicated under different conditions.

With regards to enzyme inactivation (Table 4), the processing conditions did not inactivate polyphenol oxidase or peroxidase, most likely due to the changes in protein conformation, which promoted the combination of substrate and enzyme. Earlier studies (Yu et al., 2013; Wang et al., 2018) have indicated that ultrasound under mild conditions does not inactivate enzymes. Even in the current study, ultrasound treatment combined with steam was not enough to enhance the inactivation of enzymes. Failure to inactivate the enzymes can have detrimental effects

Table 3. Effect of processing on the total polyphenols content and antioxidant activity in sweet potato juice.

Process	Total polyphenols (mg Chlorogenic ac. g^{-1} DM)	ORAC ($\mu\text{mol TROLOX g}^{-1}$ DM)	Lipid peroxidation Inhibition (%)
Control	7.5 \pm 0.3 ^a	1.34 \pm 0.04 ^a	97.7 \pm 4.8 ^a
Ultrasound	6.6 \pm 0.3 ^b	1.33 \pm 0.04 ^a	99.0 \pm 1.1 ^a
Steam	7.4 \pm 0.2 ^a	1.33 \pm 0.03 ^a	97.0 \pm 2.8 ^a
Steam + Ultrasound	7.8 \pm 0.5 ^a	1.32 \pm 0.04 ^a	98.8 \pm 2.6 ^a

Mean \pm standard deviation followed by different superscript letter in the same column indicate that values are significantly different ($p < 0.05$.) by Tukey's test.

Table 4. Effect of processing on the residual activity of polyphenol oxidase and peroxidase in sweet potato juice.

Process	Polyphenol oxidase residual activity (%)	Peroxidase residual activity (%)
Control	100.0 ± 00 ^a	100.0 ± 00 ^a
Ultrasound	98.7 ± 3.2 ^a	97.8 ± 3.0 ^a
Steam	108.4 ± 6.4 ^b	108.1 ± 7.8 ^b
Steam + Ultrasound	92.0 ± 5.1 ^a	104.5 ± 7.9 ^{ab}

Mean ± standard deviation followed by different superscript letter in the same column indicate that values are significantly different ($p < 0.05$.) by Tukey's test.

Table 5. Effect of processing on the color characteristics in sweet potato juice.

Process	Color characteristics					
	L*	a*	b*	TCD	Chrome	Hue angle
Control	47.7 ± 0.5 ^a	21.6 ± 0.8 ^a	35.3 ± 0.7 ^a	—	41.0 ± 1.5 ^b	58.1 ± 0.4 ^a
Ultrasound	46.2 ± 0.6 ^{ab}	20.9 ± 0.8 ^{ab}	33.6 ± 0.2 ^b	2.5 ± 0.2 ^a	38.4 ± 2.0 ^a	58.0 ± 0.6 ^a
Steam	45.5 ± 0.8 ^{bc}	19.3 ± 0.7 ^{bc}	31.6 ± 0.5 ^c	4.9 ± 0.9 ^b	37.3 ± 1.7 ^a	58.5 ± 0.4 ^a
Steam + Ultrasound	44.9 ± 0.6 ^c	19.7 ± 0.4 ^c	32.3 ± 0.7 ^c	4.5 ± 0.8 ^b	37.8 ± 0.9 ^a	58.5 ± 0.1 ^a

Mean ± standard deviation followed by different superscript letter in the same column indicate that values are significantly different ($p < 0.05$.) by Tukey's test.

* TCD: Total Color Difference.

on juice quality, including the production of an undesirable color, off-flavor, and off-odor.

3.4. Effect of processing on color

Color is a visible signal of the grade of fruit juices and represents an important characteristic in consumer pleasure. Thus, color can point up the acceptance degree of fruit juice (Dias et al., 2015). The effect of the processing treatments on the color of sweet potato juice are summarized in Table 5. The processed samples showed lower color results for brightness (L*), redness (a*), and yellowness (b*) with respect to the control. A similar trend of reducing color values was formerly found by Aadil et al. (2013) and Song et al. (2018) in grapefruit and sweet corn juice, respectively. Differences in visible color can be categorized based on the total color difference (TCD). Choi et al. (2002) pointed out that TCD values > 2 corresponded to perceptible differences in product impression. In this study, the processes induced “visible” color changes in sweet potato juice, with noticeable differences between the applied processes. Processing can cause physical, chemical, and biological reactions that affect the color of sweet potato juice. In this investigation study, the changes observed in the color of the sweet potato juice may be caused by cavitation or thermal degradation, which result in the destruction of pigments. Chrome specify the grade of saturation, purity, or visible strength of color, and is described as the level of alteration from gray to a pure chromatic color (Cruz-Cansino et al., 2015). Therefore, the control presented the highest value, representing a high color saturation. Similarly, Mohideen et al. (2015) observed lower Chrome values in blueberry juice processed using ultrasound. With regard to Hue angle, which represents tonality, none of the juices presented any significant changes, and tonality was maintained for all treatments. Ornelas-Paz et al. (2007) reported elevated a* results and small Hue angle values in mango, which were corresponded with a high β -carotene results in mango flesh. However, in the present study, because of the slight changes in the color parameters, it was not possible to investigate this behavior.

4. Conclusions

In the present investigation, the effects of steam and ultrasonic treatments on the bioaccessibility of β -carotene, antioxidant activity, enzyme activity, and color in orange-fleshed sweet potato juice were investigated. As a result, the bioaccessibility of β -carotene was found to increase after processing, with sonication being the most effective. All of

the treatments applied successfully preserved the antioxidant properties of the juices. However, not all of the conditions use for the treatments were effective in inactivating polyphenol oxidase and peroxidase. After processing, the samples of sweet potato juice preserved their characteristic color, and our results indicate that the best processing method was ultrasound. The results presented in this study provide an insight into the advantages of consuming sweet potato juice combined with other juices, enhancing the taste and functional characteristics of this juice due to the high carotenoid content.

Declarations

Author contribution statement

Evelyn Alicia Rios-Romero: Performed the experiments; Wrote the paper.

Luz Araceli Ochoa-Martínez: Conceived and designed the experiments; Contributed reagents, materials, analysis tools or data; Wrote the paper.

Luis Arturo Bello-Pérez: Conceived and designed the experiments.

Juliana Morales-Castro: Contributed reagents, materials, analysis tools or data.

Armando Quintero-Ramos: Performed the experiments.

José Alberto Gallegos-Infante: Analyzed and interpreted the data.

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Data will be made available on request.

Declaration of interests statement

The authors declare no conflict of interest.

Additional information

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