







## Article

# Thermoelectric Properties of Cu<sub>2</sub>Te Nanoparticle Incorporated N-Type Bi<sub>2</sub>Te<sub>2.7</sub>Se<sub>0.3</sub>

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**Abstract:** To develop highly efficient thermoelectric materials, the generation of homogeneous heterostructures in a matrix is considered to mitigate the interdependency of the thermoelectric compartments. In this study, Cu<sub>2</sub>Te nanoparticles were introduced onto Bi<sub>2</sub>Te<sub>2.7</sub>Se<sub>0.3</sub> n-type materials and their thermoelectric properties were investigated in terms of the amount of Cu<sub>2</sub>Te nanoparticles. A homogeneous dispersion of Cu<sub>2</sub>Te nanoparticles was obtained up to 0.4 wt.% Cu<sub>2</sub>Te, whereas the Cu<sub>2</sub>Te nanoparticles tended to agglomerate with each other at greater than 0.6 wt.% Cu<sub>2</sub>Te. The highest power factor was obtained under the optimal dispersion conditions (0.4 wt.% Cu<sub>2</sub>Te incorporation), which was considered to originate from the potential barrier on the interface between Cu<sub>2</sub>Te and Bi<sub>2</sub>Te<sub>2.7</sub>Se<sub>0.3</sub>. The Cu<sub>2</sub>Te incorporation also reduced the lattice thermal conductivity, and the dimensionless figure of merit *ZT* was increased to 0.75 at 374 K for 0.4 wt.% Cu<sub>2</sub>Te incorporation compared with that of 0.65 at 425 K for pristine Bi<sub>2</sub>Te<sub>2.7</sub>Se<sub>0.3</sub>. This approach could also be an effective means of controlling the temperature dependence of *ZT*, which could be modulated against target applications.

**Keywords:** thermoelectric; n-type; Cu<sub>2</sub>Te; composite

## 1. Introduction

Thermoelectric (TE) technology enables direct solid-state conversion without any moving parts or harmful emissions between heat and electrical energy and shows great potential for applications in waste heat recovery. The conversion efficiency of TE devices depends on the performance of TE materials as represented by the dimensionless figure of merit  $ZT = S^2\sigma T / \kappa_{tot}$ , where  $\sigma$ ,  $S$ ,  $T$ , and  $\kappa_{tot}$  are the electrical conductivity, Seebeck coefficient, temperature, and total thermal conductivity, respectively. To realize a high *ZT*, it is desirable to have a high power factor ( $S^2\sigma$ ) and low thermal conductivity ( $\kappa_{tot}$ ) [1,2]. However, each parameter has a trade-off relation, which makes it difficult to achieve high TE performance for practical applications.

$\text{Bi}_2\text{Te}_3$ -based TE materials are thought to be the only materials that can be considered for cooling or low-temperature energy-harvesting applications. In particular, low-temperature heat below 400 °C accounts for more than 70% of industrial waste heat, which is the case for  $\text{Bi}_2\text{Te}_3$ -related material systems [3]. Extensive theoretical and experimental studies such as electronic band structure engineering [4,5], doping [6,7], nanostructuring [8,9], and nanocomposite fabrication [10,11] have been suggested to optimize TE performance. Kim et al. reported a significant reduction in lattice thermal conductivity without any deterioration of the electrical properties by introducing dense dislocation arrays into the simple composition of bulk p-type  $\text{Bi}_{0.5}\text{Sb}_{1.5}\text{Te}_3$  and obtained the highest  $ZT$  value of  $\sim 1.9$  near room temperature [12]. Although significant progress in thermal conductivity reduction has been attained [12–14], it remains necessary to boost the electrical properties of TE materials to achieve efficient power generation and cooling devices. The carrier filtering mechanism alters the Seebeck coefficient by introducing interfaces between the TE matrix and secondary nanophases, and the band bending on the interfaces induces the filtering of low-energy carriers [15–20]. The secondary nanophases also strengthen the phonon scattering to reduce thermal conductivity, which could significantly enhance  $ZT$ . The effect of the nano-phase in TE materials has been reported on PbTe-based TE materials, where the SrTe nano-precipitate enhances  $ZT$  up to  $\sim 2.5$  at high temperatures [18]. For  $\text{Bi}_2\text{Te}_3$ -based materials, the addition of TE to p-type  $\text{Bi}_{0.5}\text{Sb}_{1.5}\text{Te}_3$  thin films prepared by a pulsed laser deposition technique significantly enhances the Seebeck coefficient [19]. However, the nano-precipitate and vacuum deposition approaches require delicate processes that cannot be applied practically. In this study, we introduced  $\text{Cu}_2\text{Te}$  nanoparticles (NPs) into an n-type  $\text{Bi}_2\text{Te}_{2.7}\text{Se}_{0.3}$  (BTS) matrix to enhance TE performance. Several papers [21–23] have reported the existence of  $\text{Cu}_2\text{Te}$  phase when excess Cu is incorporated on Bi-Te-based TE materials. However, they do not discuss the direct effect of  $\text{Cu}_2\text{Te}$  incorporation.  $\text{Cu}_2\text{Te}$  NPs, which were synthesized by the organic-free chemical method, could be considered as efficient additives that would provide benefits such as modulating the carrier concentration and enhancing phonon scattering. In addition, the incorporation of  $\text{Cu}_2\text{Te}$  NPs only requires a simple process that can easily be employed to alter TE properties according to the target applications.

## 2. Materials and Methods

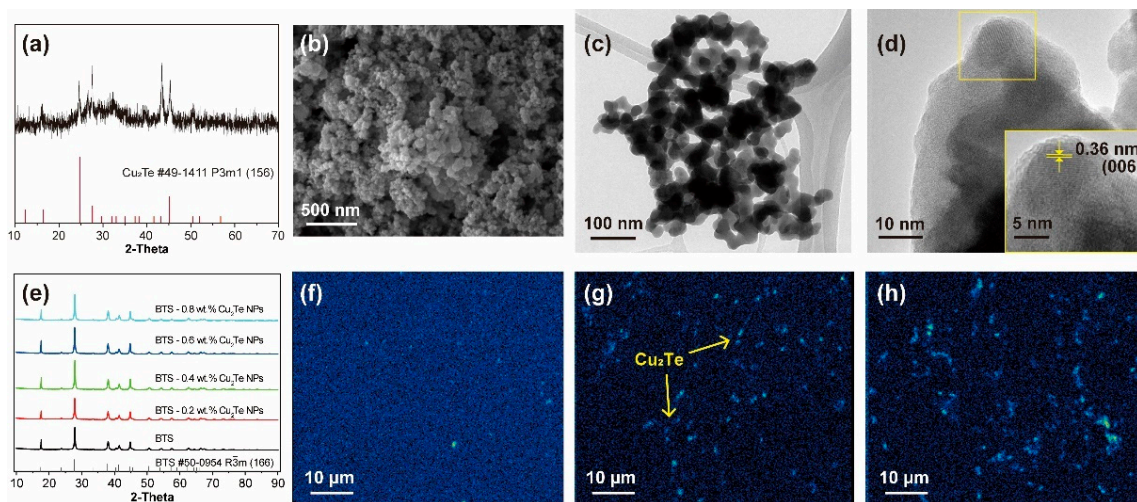
The BTS matrix was synthesized using a conventional melting-quenching process. Raw elements with stoichiometric ratios (Bi, 99.999%, 5 N Plus; Te, 99.999%, 5 N Plus; Se, 99.999%, 5 N Plus) were heated at 1000 °C for 6 h under a vacuum in fused silica tubes, and the melts were quenched in a water vessel and finally ground into fine powders using ball milling.

The  $\text{Cu}_2\text{Te}$  NPs were synthesized using the chemical reduction method. Under an Ar atmosphere, 1 g Te was dissolved in 0.3 M  $\text{NaBH}_4$  aqueous solution. Further, 0.2 M  $\text{CuCl}_2 \cdot \text{H}_2\text{O}$  aqueous solution was slowly poured into the above Te solution and stirred for 30 min. After remaining for an additional 30 min, the precipitates were centrifuged and washed with ethanol several times. For the  $\text{Cu}_2\text{Te}$  NP dispersion on the BTS matrix, the obtained  $\text{Cu}_2\text{Te}$  NPs were dispersed in ethanol and mixed with BTS by wet grinding. The mixture was dried in a vacuum oven, and fine powders were collected. The obtained powders were sintered by spark plasma sintering at 773 K for 3 min under a pressure of 60 MPa.

The consolidated samples were analysed by X-ray diffraction (XRD,  $\text{Cu K}\alpha$ , 1.5406 Å, New D8 Advance, Bruker, Billerica, MA, USA), scanning electron microscopy (SEM, JSM-7600F, JEOL, Tokyo, Japan), electron probe microanalysis (EPMA, 30 kV, JXA-8530F PLUS, JEOL, Tokyo, Japan), and transmission electron microscopy (TEM, 200 kV, Tecnai G2-20, FEI, Hillsboro, Oregon, USA). The TE properties were measured using a ZEM-3 (ULVAC-RIKO, Methuen, MA, USA) for the electrical parts and the laser-flash method (LFA, DLF 1300, TA, New Castle, DE, USA) for the thermal parts. Hall measurements were conducted to obtain the carrier concentration (HT-Hall, ResiTest 8300, Toyo Corporation, Tokyo, Japan).

### 3. Results and Discussions

Figure 1a shows the XRD pattern of as-prepared  $\text{Cu}_2\text{Te}$  NPs obtained by the chemical reduction method. Our  $\text{Cu}_2\text{Te}$  NPs has a hexagonal structure with the space group of  $P3m1$  (JCPDS # 49-1441). The XRD peaks are not strong due to the small size of  $\text{Cu}_2\text{Te}$  NPs. The size of as-prepared  $\text{Cu}_2\text{Te}$  is 20–50 nm from the SEM and TEM images (Figure 1b,c), and the high resolution TEM image shows the lattice distance of 0.36 nm, which corresponds to (006) plane of  $\text{Cu}_2\text{Te}$  phase. Figure 1e presents the powder XRD patterns of the BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NP ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ) composite materials. All samples show typical BTS patterns (rhombohedral structure, space group of  $R\bar{3}m$ , JCPDS # 50-0594) without any additional peaks. The lattice parameters of the BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NP ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ) are summarized in Table 1, and there are no significant changes in lattice parameters by  $\text{Cu}_2\text{Te}$  NPs incorporation. The small size and low content of  $\text{Cu}_2\text{Te}$  NPs in the composite powder were not detectable by XRD (Figure 1e). We analysed the  $\text{Cu}_2\text{Te}$  nanophases on BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NP ( $x = 0.2, 0.4,$  and  $0.6$ ) bulk samples by EPMA (Figure 1f–h), where the light parts of the Cu mapping were matched with the  $\text{Cu}_2\text{Te}$  nanophases.  $\text{Cu}_2\text{Te}$  NPs can be easily observed on the parent bulk materials, and as the amount of  $\text{Cu}_2\text{Te}$  NPs increases, a homogeneous distribution of  $\text{Cu}_2\text{Te}$  NPs can be found. However, for the BTS-0.6 wt.%  $\text{Cu}_2\text{Te}$  NPs sample, some  $\text{Cu}_2\text{Te}$  NPs exhibit agglomeration, which means that a large amount (more than 0.6 wt.%) of  $\text{Cu}_2\text{Te}$  NPs on BTS cannot be dispersed uniformly.



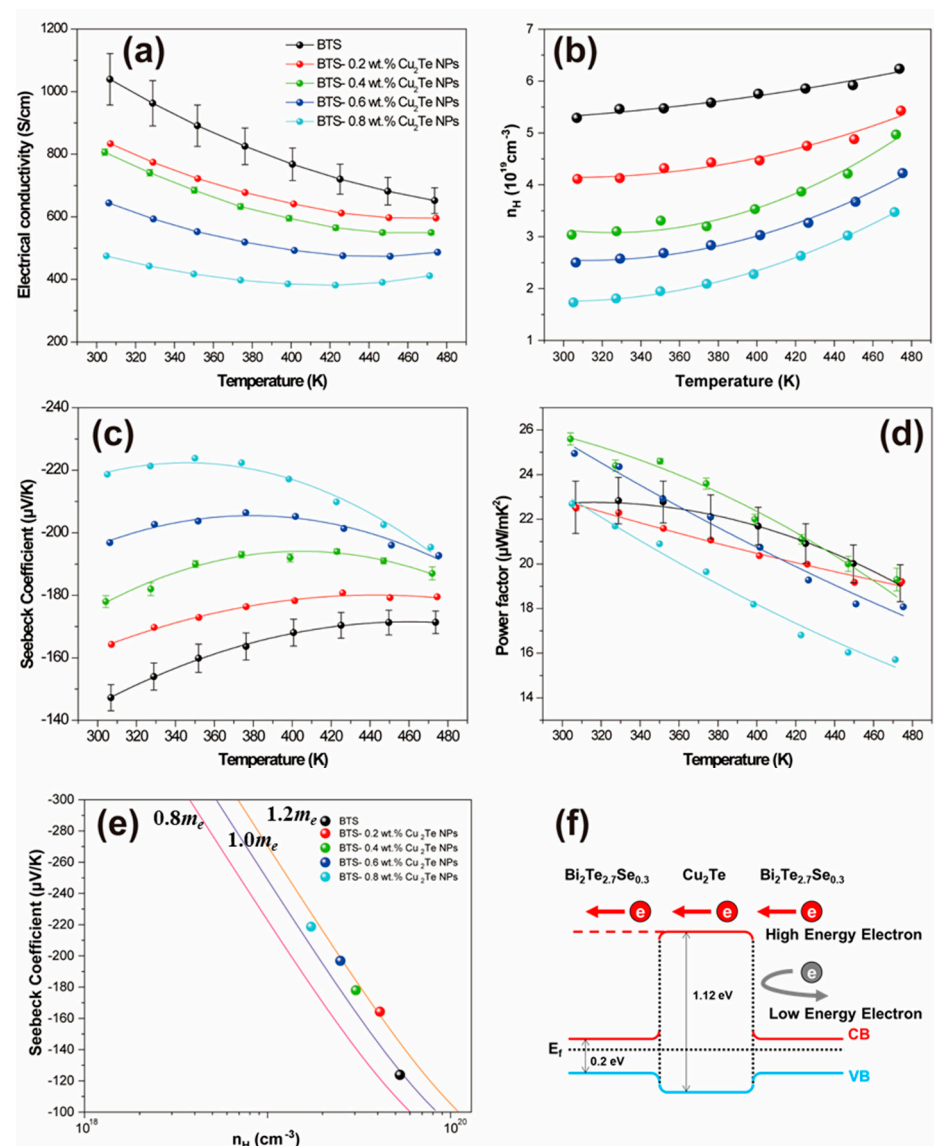
**Figure 1.** (a) XRD pattern, (b) SEM image, (c) TEM image, and (d) high resolution TEM image of  $\text{Cu}_2\text{Te}$  NPs, (e) XRD patterns of BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ), Cu K mapping images of EPMA for (f) BTS-0.2 wt.%  $\text{Cu}_2\text{Te}$  NPs, (g) BTS-0.4 wt.%  $\text{Cu}_2\text{Te}$  NPs, and (h) BTS-0.6 wt.%  $\text{Cu}_2\text{Te}$  NPs, where the light dots correspond to elemental Cu.

**Table 1.** The lattice parameters of BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ).

	Lattice Constant for a (Å)	Lattice Constant for c (Å)
BTS	4.3619	30.3632
BTS-0.2 wt.% $\text{Cu}_2\text{Te}$ NPs	4.3614	30.3604
BTS-0.4 wt.% $\text{Cu}_2\text{Te}$ NPs	4.3615	30.3641
BTS-0.6 wt.% $\text{Cu}_2\text{Te}$ NPs	4.3610	30.3606
BTS-0.8 wt.% $\text{Cu}_2\text{Te}$ NPs	4.3614	30.3650

Figure 2a presents the temperature dependence of the electrical conductivity ( $\sigma$ ) for BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ). The electrical conductivity gradually decreases over the entire temperature range studied here as the  $\text{Cu}_2\text{Te}$  NP incorporation increases. The electrical conductivities of BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4,$  and  $0.6$ ) decrease with increasing temperature, indicating semi-metallic or metallic conduction

behaviour. However, the sample with a high amount of  $\text{Cu}_2\text{Te}$  NPs (0.8 wt.%) incorporated shows semiconducting behaviour at a high temperature due to the carrier concentration decrease shown in Figure 2b. The carrier concentration obtained from the Hall measurements decreases as the amount of  $\text{Cu}_2\text{Te}$  NPs increases. The Cu intercalation in the van der Waals gap between Te–Te shows donor-like behaviour, while our  $\text{Cu}_2\text{Te}$  incorporation shows acceptor-like behaviour. The temperature dependence of the Seebeck coefficient ( $S$ ) for the BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NP ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ) samples is displayed in Figure 2c. All samples exhibit negative  $S$  values, indicating that electrons constitute the majority of the charge carriers, which is consistent with the signs of the respective Hall measurements. The temperature at which the  $S$  peaks ( $T_{max}$ ) are shifted to lower temperatures with increasing  $\text{Cu}_2\text{Te}$  incorporation, and the absolute value of  $S$  maximum ( $S_{max}$ ) increases as the amount of incorporated  $\text{Cu}_2\text{Te}$  increases. Typically, BTS show anisotropic nature of TE properties due to Te vacancies and antisite defects [24]. It is noted that the reproducibility (collecting the results for more than 5 different batches) of electrical properties is greatly enhanced through  $\text{Cu}_2\text{Te}$  introduction.



**Figure 2.** Electrical properties of BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ). (a) Electrical conductivities, (b) electron carrier concentrations obtained by Hall measurement, (c) Seebeck coefficients, (d) power factors, (e) Pisarenko's plot, and (f) band diagram of BTS and  $\text{Cu}_2\text{Te}$  NPs interfaces.



Usually, the  $T_{max}$  shift to lower temperatures originates from increased bipolar conduction, but the corresponding  $S_{max}$  should also be decreased. Improvement of the bipolar conduction cannot explain the observed increase in  $S_{max}$  with increasing  $\text{Cu}_2\text{Te}$ . From Pisarenko's relation [1,25], one can estimate the density of states (DOS) effective mass with the assumption of a single parabolic band model using the following Equation (1):

$$S = \frac{8\pi^2 k_B^2}{3eh^2} \left(\frac{\pi}{3n}\right)^{2/3} m_d^* T \quad (1)$$

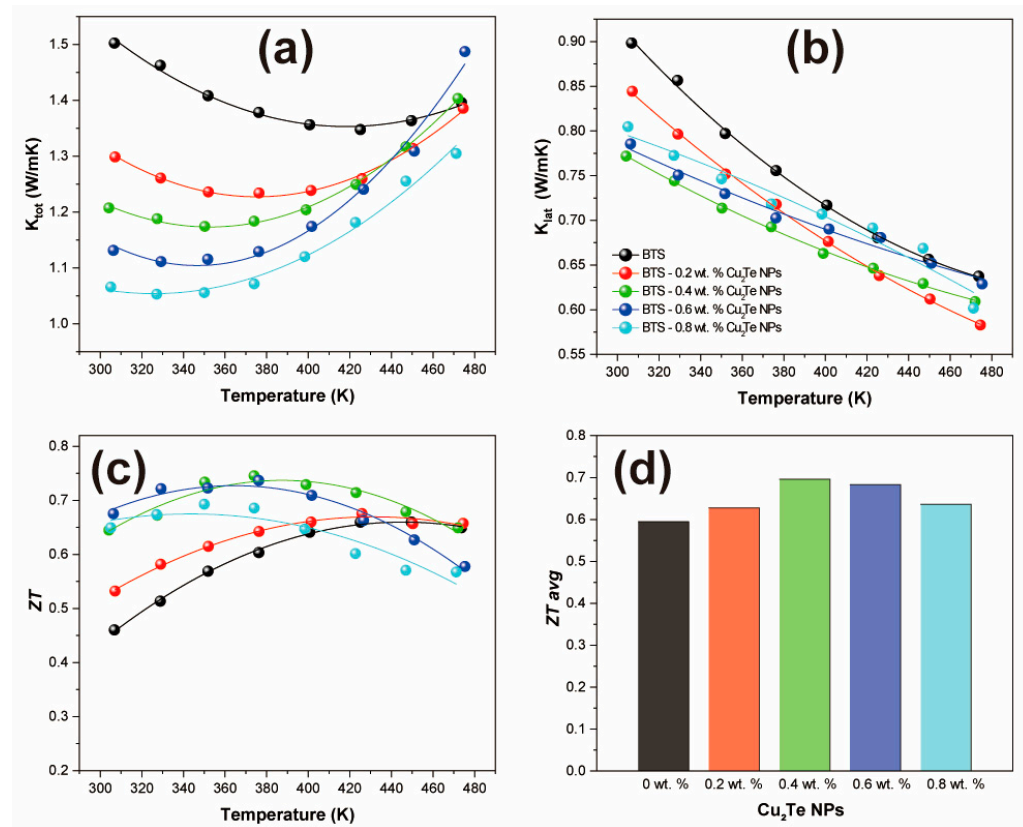
where  $h$  is the Planck constant,  $k_B$  is the Boltzmann constant,  $m_d^*$  is the DOS effective mass,  $e$  is the electronic charge, and  $n$  is the carrier concentration. As shown in Figure 2e, the pristine BTS has an  $m_d^*$  value of  $0.97 m_e$  ( $m_e$  is the electron rest mass), whereas  $\text{Cu}_2\text{Te}$  NP incorporation increases the  $m_d^*$  value of  $1.16 m_e$  in the BTS-0.4 wt.%  $\text{Cu}_2\text{Te}$  NP sample at room temperature. Therefore, the increase in  $m_d^*$  with  $\text{Cu}_2\text{Te}$  incorporation can enhance  $S_{max}$ , even when the bipolar conduction becomes strong. The  $n$  reduction observed with the incorporation of  $\text{Cu}_2\text{Te}$  is also responsible for the increase in  $S_{max}$ . The  $m_d^*$  change could be attributed to the engineered band structure interface between the BTS matrix and  $\text{Cu}_2\text{Te}$  NPs. In this context, we can suggest a band diagram that describes the interfacial band bending between the BTS and  $\text{Cu}_2\text{Te}$  NPs (Figure 2f). The electron energy barrier, i.e., the hetero-interface of the conduction bands between the BTS and  $\text{Cu}_2\text{Te}$  NPs, filters the low-energy carriers. The electron affinity and band gap of BTS and  $\text{Cu}_2\text{Te}$  were obtained from the literature [26–29]. A previous theoretical study of PbTe TE materials showed that ~1.5 nm nanoinclusion significantly enhanced the  $S$  value [15]. The electrostatic potential only affected the interface between the nanoinclusion and the matrix, and larger nanoinclusions were less effective than smaller nanoinclusions. Our synthesized  $\text{Cu}_2\text{Te}$  NPs had sizes of ~50 nm, so we expected that the smaller  $\text{Cu}_2\text{Te}$  NPs could have a greater effect. The BTS-0.4 wt.%  $\text{Cu}_2\text{Te}$  NP sample has the highest calculated power factor ( $S^2\sigma$ ), which is ~15% higher at room temperature than that of the pristine BTS sample.

The temperature dependence of the total thermal conductivity ( $\kappa_{tot}$ ) of BTS- $x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ) is shown in Figure 3a. All samples display very low  $\kappa_{tot}$  values over the entire temperature range studied here and also exhibit upturns. The occurrence temperatures of the upturns in the thermal conductivity data move to lower temperatures with increasing  $\text{Cu}_2\text{Te}$  incorporation amounts, which is consistent with the behaviour of  $S$  and  $\sigma$ . This increase in thermal conductivity is due to the bipolar diffusion of the carriers and is still present in the BTS sample at elevated temperatures, as shown in Figure 3a. This increase in  $\kappa_{tot}$  with temperature is attributable to the thermal energy transported by electron–hole pairs, which is equal to the energy of the gap between the hot and cold sides of the sample. This bipolar diffusion phenomenon results in an increase in the heat transfer beyond what is expected from the normal carrier contribution ( $\kappa_{ele}$ ) to  $\kappa$  defined by the Wiedemann-Franz law [30],  $\kappa_{ele} = L\sigma T$ , where  $L$  is the Lorenz number. The Lorenz number is estimated by assuming a single parabolic band and acoustic phonon scattering using the following Equations (2)–(4):

$$S = \pm \frac{k_B}{e} \left[ \frac{(r + 5/2)F_{r+3/2}(\xi)}{(r + 3/2)F_{r+1/2}(\xi)} - \xi \right] \quad (2)$$

$$F_n(\xi) = \int_0^\infty \frac{x^n}{1 + e^{(x-\xi)}} dx \quad (3)$$

$$L = \left(\frac{k_B}{e}\right)^2 \left[ \frac{(r + 7/2)F_{r+5/2}(\xi)}{(r + 3/2)F_{r+1/2}(\xi)} - \left(\frac{(r + 5/2)F_{r+3/2}(\xi)}{(r + 3/2)F_{r+1/2}(\xi)}\right)^2 \right] \quad (4)$$



**Figure 3.** (a) Total thermal conductivities, (b) lattice thermal conductivities, (c) ZT values, and (d) average ZT values between room temperature and 470 K of BTS- $x$  wt.% Cu<sub>2</sub>Te NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ).

Here,  $\xi$ ,  $F_n(\xi)$ , and  $r$  are the reduced Fermi energy ( $(E_v - E_F)/k_B T$ ), Fermi integral of order  $n$ , and scattering parameter, respectively. We set  $r = 0-1/2$  for acoustic phonon scattering. The temperature dependence of the lattice thermal conductivity ( $\kappa_{lat}$ ) can be estimated by deducting the bipolar conduction portion ( $\kappa_{bp}$ ) and electronic portion ( $\kappa_{ele}$ ) of thermal transport from the total thermal conductivity;  $\kappa_{lat} = \kappa_{tot} - \kappa_{bp} - \kappa_{ele}$ .  $\kappa_{bp}$  was calculated using the below Equation (5).

$$\kappa_{bp} = \left( \sum_i S_i^2 \sigma_i - S^2 \sigma \right) T. \quad (5)$$

$S_i$ ,  $\sigma_i$ ,  $S$ , and  $\sigma$  are the Seebeck coefficient and electrical conductivity of an individual band, total Seebeck coefficient, and electrical conductivity from both the conduction and valence bands, respectively.  $S$  and  $\sigma$  are in turn defined as follows Equations (6) and (7):

$$\sigma = \sum_i \sigma_i \quad (6)$$

$$S = \frac{\sum_i S_i \sigma_i}{\sum_i \sigma_i} \quad (7)$$

Using the two-band (TB) model (an extension of the single parabolic band model), which includes one valence band and one conduction band, band parameters such as the density-of-states effective mass ( $m_d^*$ ) and non-degenerate mobility ( $\mu_0$ ) of each band were obtained. Specifically, the  $m_d^*$  and  $\mu_0$  of individual band were estimated by fitting the TB model to Hall carrier concentration ( $n_H$ )-dependent  $S$  and  $n_H$ -dependent  $\sigma$  measurements, respectively [31]. The band gap between the valence and conduction bands required in the TB model was adopted from the literature [32]. Once  $m_{d,i}^*$  and  $\mu_{0,i}$  ( $i =$  valence and conduction bands) were estimated, corresponding  $S_i$  and  $\sigma_i$  ( $i =$  valence and conduction

bands) were calculated. Furthermore, theoretical total  $S$  and  $\sigma$  (which agree well with experimental  $S$  and  $\sigma$ ) were computed with  $S_i$  and  $\sigma_i$  according to Equations (6) and (7). Finally, the calculated  $S_i$ ,  $\sigma_i$ ,  $S$ , and  $\sigma$  were substituted back into Equation (5) to estimate  $\kappa_{bp}$ .

The incorporation of  $\text{Cu}_2\text{Te}$  NPs encourages bipolar conduction and, consequently,  $\kappa_{lat}$  (shown in Figure 3b), after  $\text{Cu}_2\text{Te}$  NP incorporation is reduced over the entire temperature range due to enhanced phonon scattering. The reduction in  $\kappa_{lat}$  occurs up to 0.4 wt.%  $\text{Cu}_2\text{Te}$  NPs incorporation, whereas more  $\text{Cu}_2\text{Te}$  NPs produce a negative phonon scattering effect by agglomeration of  $\text{Cu}_2\text{Te}$  NPs. Generally, it is acceptable that the introduction of secondary phases decreases  $\kappa_{lat}$  by phonon scattering. However, the size of the secondary phase is an important factor for enhancing phonon scattering. Particles with sizes larger than hundreds of nanometres have a limited effect on  $\kappa_{lat}$  [33]. Our  $\kappa_{lat}$  results after  $\text{Cu}_2\text{Te}$  agglomeration (more than 0.6 wt.%  $\text{Cu}_2\text{Te}$  NPs) could be described similarly. The size of the  $\text{Cu}_2\text{Te}$  NPs was  $\sim 50$  nm, and if several  $\text{Cu}_2\text{Te}$  NPs were agglomerated, the size was easily greater than a hundred nanometres. In addition, the previous report also shows that reduced particle size with the same volume fraction could decrease thermal conductivity more [34]. Therefore, a homogeneous dispersion of small  $\text{Cu}_2\text{Te}$  NPs is necessary for encouraging phonon scattering and enhancing the electrical properties.

Collecting the above effects,  $ZT$  is shown as a function of temperature for  $\text{BTS}-x$  wt.%  $\text{Cu}_2\text{Te}$  NPs ( $x = 0, 0.2, 0.4, 0.6,$  and  $0.8$ ) in Figure 3c. The highest  $ZT$  of 0.75 is observed at 374 K for the  $\text{BTS}-0.4$  wt.%  $\text{Cu}_2\text{Te}$  NPs sample, which is 15% higher than that of the pristine sample. It is noteworthy that the maximum  $ZT$  temperature for each sample is modulated by the introduction of  $\text{Cu}_2\text{Te}$  NPs, decreasing as the amount of  $\text{Cu}_2\text{Te}$  NPs increases. The simple incorporation of  $\text{Cu}_2\text{Te}$  can easily modulate the temperature dependence of n-type TE materials, and we suggest that the TE properties can be precisely controlled according to the target application, such as cooling or low-temperature energy harvesting. The average  $ZT$  value ( $ZT_{avg}$ ) in the temperature range studied here is shown in Figure 3d.  $ZT_{avg}$  reaches 0.70 for the  $\text{BTS}-0.4$  wt.%  $\text{Cu}_2\text{Te}$  NPs sample, representing 19% enhancement compared to the pristine BTS matrix.

#### 4. Conclusions

In conclusion, we investigated the TE properties of n-type BTS by incorporating  $\text{Cu}_2\text{Te}$  NPs. The incorporation of  $\text{Cu}_2\text{Te}$  NPs encourages the Seebeck coefficient and power factor, and we thought that the band bending between the interface of the BTS and  $\text{Cu}_2\text{Te}$  NPs could filter the low-energy electron carrier. The  $\text{Cu}_2\text{Te}$  NPs on the BTS matrix also affect  $\kappa_{lat}$  by encouraging phonon scattering. Together with modulating the electrical and thermal properties, the maximum  $ZT$  value reaches 0.75 at 374 K for the  $\text{BTS}-0.4$  wt.%  $\text{Cu}_2\text{Te}$  NPs sample, 15% higher than that of pristine BTS (0.65 at 425 K). In addition, the temperature dependence of  $ZT$  can be controlled by the  $\text{Cu}_2\text{Te}$  NP dispersion just before the sintering process, which is a great advantage for target applications. Further advances could be expected by developing the synthesis of  $\text{Cu}_2\text{Te}$  NPs with sizes of several nanometres.

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