

Carboxylate-functionalized foldamer inhibitors of HIV-1 integrase and Topoisomerase 1: artificial analogues of DNA mimic proteins

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ABSTRACT

Inspired by DNA mimic proteins, we have introduced aromatic foldamers bearing phosphonate groups as synthetic mimics of the charge surface of B-DNA and competitive inhibitors of some therapeutically relevant DNA-binding enzymes: the human DNA Topoisomerase 1 (Top1) and the human HIV-1 integrase (HIV-1 IN). We now report on variants of these anionic foldamers bearing carboxylates instead of phosphonates. Several new monomers have been synthesized with protecting groups suitable for solid phase synthesis (SPS). Six hexadecaamides have been prepared using SPS. Proof of their resemblance to B-DNA was brought by the first crystal structure of one of these DNA-mimic foldamers in its polyanionic form. While some of the foldamers were found to be as active as, or even more active than, the original phosphonate oligomers, others had no activity at all or could even stimulate enzyme activity *in vitro*. Some foldamers were found to have differential inhibitory effects on the two enzymes. These results demonstrate a strong dependence of inhibitory activity on foldamer structure and charge distribution. They open broad avenues for the development of new classes of derivatives that could inhibit the interaction of specific proteins with their DNA target thereby influencing the cellular pathways in which they are involved.

INTRODUCTION

During the last 25 years, chemists have investigated synthetic DNA analogues with Watson–Crick base-pairing abilities. These DNA mimics are generally composed of the four natural nucleobases and of a backbone different from natural deoxyribose-phosphate (1–3). They thus resemble DNA but also feature essential differences that can result in improved behaviors, which represent a fundamental principle in molecular mimicry. For instance, peptide nucleic acids (4–7) (PNAs) and locked nucleic acids (8–10) (LNAs) can bind to complementary DNA sequences better than DNA itself and have therefore been developed to interfere with nucleic acid base-pairing in biological contexts (5,11–13). This vast field focuses on one part only of the nucleic acid interactome: their interactions with other nucleic acids.

Another part of the nucleic acid interactome lies with the recognition of their surface by proteins. Protein–nucleic acid interactions are at the core of multiple cellular processes and provide numerous opportunities for therapeutic intervention. Inhibitors of protein–nucleic acid interactions include selective DNA ligands (14–16), ‘interfacial’ inhibitors that bind the protein–nucleic acid interface (17–20), and nucleic acids themselves (21–23). However, until recently, synthetic DNA analogues that would reproduce the surface features of double stranded DNA (instead of its base-pairing ability) in order to bind to DNA-binding proteins and eventually achieve competitive inhibition of protein–DNA interactions have rarely been considered (24). Yet some naturally occurring proteins called DNA mimic proteins (25–32) do exactly that by displaying helical arrays of anionic amino

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acid residues (Figure 1G, H), thus hinting at the development of artificial systems with similar properties.

Along this line, we reported foldamer-based DNA analogues that mimic the negative charge surface of double-stranded B-DNA (33). Oligoamides composed of an alternation of 8-amino-2-quinolinecarboxylic acid (Q^{Pho} , Q^{5Pho} , Figure 1A) and 8-aminomethyl-2-quinolinecarboxylic acid (${}^mQ^{Pho}$) adopt stable single helical conformations at the surface of which phosphonate side chains form a double helical array that matches the positions of phosphates in B-DNA. This resemblance opened the prospect to bind some non-sequence selective DNA-binding proteins, *i.e.* proteins that mainly recognize the shape and charge distribution of nucleic acids. The anionic foldamers eventually proved to perform much better than anticipated in that they actually *inhibited* several DNA-binding enzymes: the foldamers bind to these enzymes better than the DNA substrate itself. In particular, we observed unprecedented inhibitory activity for two therapeutically relevant enzymes: the human Topoisomerase 1 (Top1) (34) which mediates the relaxation of supercoiled DNA, and the human immunodeficiency virus 1 integrase (HIV-1 IN) (35) which catalyzes the insertion of the HIV-1 DNA formed after reverse transcription into the human genome. Inhibition by 32 unit-long foldamers – the equivalent of sixteen DNA base pairs – occurred at sub-micromolar concentrations, matching or even exceeding the performance of the best inhibitors of these enzymes, camptothecin (36) for Top1 and raltegravir (37) for HIV-1 IN, and offering access to a novel mechanism of inhibition. Indeed, camptothecin and raltegravir typically block the DNA-protein complex (19,20), whereas the foldamers compete with the DNA substrate. In contrast, other DNA-binding enzymes were inhibited to a lower extent (Topoisomerase 2, Flap endonuclease 1) or not inhibited at all (deoxyribonuclease 1, S1 nuclease, benzonase[®]) (33).

These discoveries called for further developments in particular regarding the structural basis of foldamer-mediated enzyme inhibition. It should be noted that our initial efforts aimed at making the foldamers resemble B-DNA as closely as possible, but that their inhibitory activity emerged precisely because of differences from the DNA structure: a ‘perfect’ mimic would not be more active as DNA itself, which is a rather poor inhibitor of Top1 and HIV-1 IN. Thus, phosphonates were selected as anion bearing functionalities and Q^{5Pho} was introduced in $({}^mQQ^{5Pho})_n$ sequences because it confers a smaller minor groove and a larger major groove to the mimics that match the grooves of DNA better than Q^{Pho} in $({}^mQQ^{Pho})_n$ sequences (Figure 1B–E) (33). However, the need for all these features to achieve inhibition was not evidenced. For instance, DNA mimic proteins have a less obvious resemblance to B-DNA (Figure 1G, H) (24,30,31), and possess no phosphate or phosphonate anions since negative charges are borne by aspartate and glutamate carboxylate functions. Furthermore, an intriguing and essential aspect concerns selective enzyme inhibition: can two enzymes be differentially inhibited by a given DNA mimic even though they act on the same kind of double-stranded DNA substrate?

The initial objectives of the current study were thus multiple: (i) to develop a solid phase synthesis of foldamer-based DNA mimics as a more suitable approach to the prepara-

tion of multiple variants than the initial segment doubling solution phase strategy (33); (ii) to assess the ability of carboxylic acid functionalized foldamers inspired from DNA mimic proteins to inhibit Top1 and HIV-1 IN; and (iii) to assess the dependence of enzyme inhibition on the length and position of the side chains. In the following, we report the synthesis and characterization of several new monomers and hexadecameric oligomers, including, in one case, crystallographic structural evidence of the anionic form. We discovered that carboxylate side chains can impart enhanced inhibitory activity in some cases, or no inhibition in others, or even cause enzyme activation. Importantly, we found a sequence that has differential inhibitory effects on the two enzymes. Altogether, these results further validate the novel concept of foldamer-mediated DNA surface mimicry. The variety of DNA-binding enzymes and the feasibility of foldamer modifications augur well for the development of this approach and open new avenues in molecular mimicry for protein surface recognition.

MATERIALS AND METHODS

Chemicals and reagents were used as commercially supplied without any further purification unless otherwise stated. Low loading Wang resin (0.38 mmol/g) was purchased from Novabiochem. Ghosez reagent was purchased from Sigma Aldrich. *N,N*-diisopropylethylamine (DIPEA) was distilled over calcium hydride. Anhydrous THF and CH_2Cl_2 for solid phase synthesis were dispensed from an MBRAUN SPS-800 solvent purification system. Reactions requiring anhydrous conditions were performed under nitrogen atmosphere. Melting points were determined using a Buchi B-540 melting point apparatus.

Synthesis

The detailed synthesis and characterization of monomers and oligomers are described in the Supplementary Information.

Nuclear magnetic resonance

NMR spectra were recorded on an Avance II NMR spectrometer (Bruker Biospin) with a vertical 7.05T narrow-bore/ultrashield magnet operating at 300 MHz for 1H observation, 75 MHz for ${}^{13}C$ observation and 121.4 MHz for ${}^{31}P$ observation by means of a 5-mm direct BBO H/X probe with Z gradient capabilities. Chemical shifts are reported in ppm and are referenced against residual solvent signals of $CDCl_3$ (δ_H : 7.26, δ_C : 77.0), $DMSO-d_6$ (δ_H : 2.50, δ_C : 39.4), HDO (δ_H : 4.8). ${}^{31}P$ NMR signals are referenced to PPh_3O at 27 ppm. Peak multiplicities are noted as follows: singlet (s), broad singlet (bs), doublet (d), triplet (t), quartet (q), doublet of doublets (dd), doublet of quartets (dq) and multiplet (m). Data processing was performed with Bruker TOPSPIN 2.1 software.

RP-HPLC analysis and purification

RP-HPLC quality acetonitrile and MilliQ water were used for RP-HPLC analyses and purification.

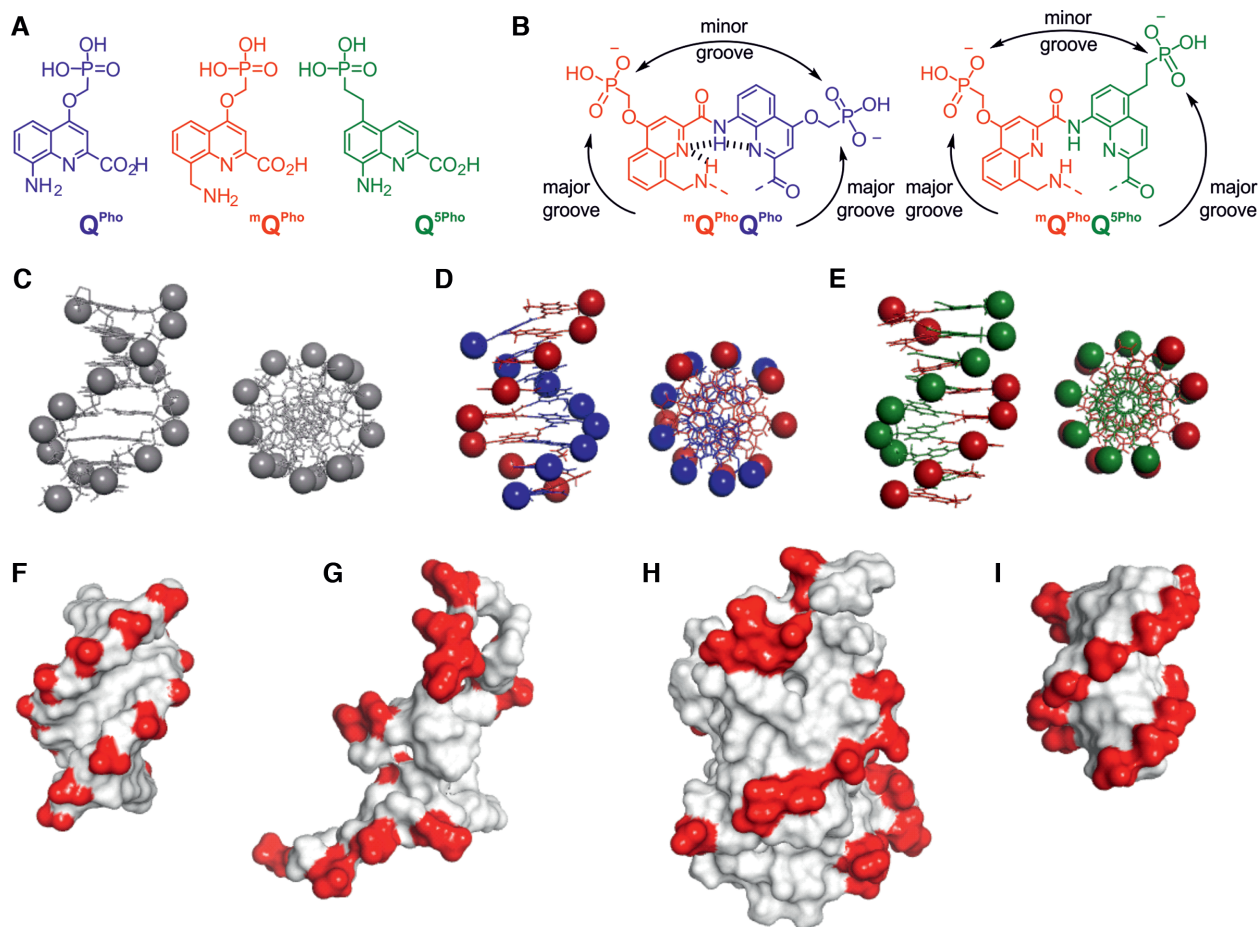


Figure 1. DNA mimics design. (A) Formulae of amino acid monomers Q^{Pho} , mQ^{Pho} and Q^{5Pho} . (B) Formulae of $mQ^{Pho}Q^{Pho}$ and $mQ^{Pho}Q^{5Pho}$ repeat units. (C–E) Top views and side views of molecular models of the structures of: (C) an eight-base-pair B-DNA; (D) $(mQ^{Pho}Q^{Pho})_8$ and (E) $(mQ^{Pho}Q^{5Pho})_8$. Phosphorus atoms are shown as large spheres. The same color code is used in (D) and (E) as in the formulae shown in (A). (F–I) Surface rendering showing negative charges (in red) of the structures of B-DNA (F), DNA mimic proteins TAFII230 (G) and HI1450 (H) and $(mQ^{Pho}Q^{5Pho})_8$ (I). All structures are shown at the same scale. The structures of B-DNA, HI1450 (PDB# 1NNV) (28) and TAFII230 (PDB# 1TBA) (29) are derived from crystal data. The structure of $(mQ^{Pho}Q^{5Pho})_8$ is an energy minimized model.

HPLC analyses of the oligomers were performed on a JASCO AS-2055 chromatography system equipped with a reverse-phase Varian Pursuit C18 column (250 × 4.6 mm, 5 μm) and a multichannel UV/VIS detector JASCO UV-2077. Oligomers were analysed using a linear gradient of 15 min starting from 0% of B and 100% of A to 20% of B and 80% of A with a flow of 1 ml/min. Solvents A and B were prepared as follows: a stock 50 mM aqueous triethylammonium acetate buffer solution at pH 8.7 was prepared by dissolving 3 ml of glacial acetic acid in 950 ml water and, while mixing, adding freshly distilled triethylamine (6.5 ml). After adjusting the pH to 8.7 with triethylamine, the volume was finally adjusted to 1 l with water. Solvent A was prepared by dilution of the stock buffer solution with water 1:3 (vol/vol) to a final concentration of 12.5 mM triethylammonium acetate, pH 8.7, in water. Solvent B was prepared by dilution of the stock buffer solution with acetonitrile 1:3 (vol/vol) to a final composition of 12.5 mM triethylammonium acetate, pH 8.7. Semi-preparative purification of oligomers was performed on a semi-preparative HPLC using a Varian Pursuit C18 column (250 × 15 mm, 5 μm)

with a flow of 4 ml/min. Oligomers were purified using linear gradients starting from 95% of A and 5% of B to different ratios of A/B which are stated for each oligomer in the Supplementary Information. Monitoring by UV detection was carried out at 254 or 300 nm using a diode array detector. Monomers were analysed with a Nucleodur C8 column (120 × 4.6 mm, 5 μm) by using a linear gradient of 20 min from 30% of D and 70% of C to 100% of D (where solvent C was H₂O + 0.1% TFA and solvent D was ACN + 0.1% TFA) with a flow of 1 ml/min.

High resolution mass spectrometry analysis and exact mass measurements

Detection of compounds was performed using a thermo Orbitrap Exactive mass spectrometer (Thermo Fisher Scientific, San Jose, USA). Electrospray ionization source (HESI-II) was used for ionization of the target compounds in positive and negative ion modes. Instrument calibration in positive mode and negative mode were done prior to sample injection (ion calibration solution 88323 and 88324, for positive and negative ions, respectively, Thermo Scientific

Pierce). The tuning parameters were set to avoid fragmentation and keep ions intact as follows: ESI voltage, +3.3 kV/−3.0 kV; sheath gas pressure, 30.0 arbitrary units and capillary temperature, 373 K. The RF of the ion guides were optimized to obtain the maximum transmission efficiencies. For the compounds of interest, a scan range of m/z 500–4000 was chosen. The automatic gain control (AGC) target for a maximum capacity in C-trap was set at 2×10^6 ions for a maximum injection time of 250 ms.

Crystallization

For crystallization experiments, a lyophilized powder of **4** was dissolved in ultra-pure water and ammonium bicarbonate. Since oligomer **4** does not possess stereogenic centres, it was expected to fold as a racemic mixture of right-handed (*P*) and left-handed (*M*) helices. Final concentration of the racemic solution to start crystallization screening was 8 mg/ml. Crystallization trials were performed using standard aqueous hanging drop vapor diffusion methods (38) in 24-well Linbro-style plates, at 293 K. Screening of crystallization conditions was carried out using commercial sparse matrix screen JBScreen Basic 1 to 4, from Jena Bioscience (39,40). For each screening condition, drops were prepared using 0.75 μ l of solution of racemic **4** and an equal volume of the crystallization reagent on a silanized glass slide, which was then suspended over a reservoir solution. Clusters of small, light yellow crystals were observed after 5 days in one of the 96 screening conditions. X-ray quality crystallogeneses was optimized by increasing the drop size by equilibration of a mixture containing 1.0 μ l of 8 mg/ml solution of **4** and 1.5 μ l of a reservoir solution of 18% w/vol polyethylene glycol 8000, 100 mM MES buffer, pH 6.5, and 200 mM calcium acetate; against 500 μ l of the reservoir solution. Crystals (Supplementary Figure S1) used for the X-ray diffraction measurements grew to a size of $0.3 \times 0.05 \times 0.05$ mm in 10 days.

Data collection, structure solution and refinement

For low temperature X-ray diffraction measurements, a single crystal was mounted using a cryo-loop after quick soaking on Paratone-N oil and flash-frozen. Diffraction data were collected at the IECB X-ray facility (UMS 3033) on a micro-focus, rotating anode Rigaku FRX diffractometer, with Cu K α radiation ($\lambda = 1.5417$ Å) and a hybrid pixel detector (PILATUS 200K), at 100 K. The crystal diffracted to a maximum resolution of 0.95 Å. Diffraction data were processed and scaled using the *CrystalClear-SM 1.36* package (41). The symmetry of the crystal was triclinic with space group $P\bar{1}$, $Z' = 1$ and unit cell parameters: a , 20.85 Å; b , 21.73 Å; c , 31.33 Å; α , 83.61°; β , 75.07°; and γ , 77.15°. The structure was solved by direct method using the program *SHELXT* (42). The phase set calculated allowed identifying most of **4**, including all side chains. The structure was refined by full-matrix least-squares method on F^2 with *SHELXL-2014* (43) within the *Olex2* suite (44). For all non-hydrogen atoms attempts to introduce anisotropic displacement parameters were made. AFIX 116 constraints were used for the $^m\text{Q}^{\text{Ac}}$ and Q^{5Ac} atoms. In case of side chains

and solvent molecules, the non-hydrogen atoms were refined with anisotropic or isotropic displacement parameters. Hydrogen atoms were included for **4** in idealized positions using HFIX and refined with a riding model. For solvent molecules, positions of hydrogen atoms were not determined. After several attempts to model the disordered water molecules, the *PLATON/SQUEEZE* procedure (45) was implemented to treat the regions with highly disordered solvent molecules. The total potential solvent accessible void volume was 4772.1 Å³ and the number of electron count per cell was 1361. DFIX, DELU, SIMU and ISOR instructions were used to model displacement parameters and the geometry of **4**. The FVAR function was used during refinement of occupancy factors of disordered parts. The final cif file was checked using IUCR's *checkcif* algorithm. The coordinates and structure factors have been deposited in the Cambridge Crystallographic Data Centre (CCDC). The accession numbers and refinement statistics are provided in Supplementary Tables S1 and S2.

Top1-catalyzed relaxation assays

Top1 catalytic activity was assessed in plasmid DNA relaxation assays according to the specifications of the supplier (Topogen, Columbus, OH, USA) with minor modifications. For each reaction, 100 ng of (−)-pcDNA3.1 supercoiled plasmid was incubated with ~3 ng of purified human recombinant Top1 in 1 \times reaction buffer (10 mM Tris-HCl, pH 7.9, 1 mM EDTA, 150 mM NaCl, 0.1% bovine serum albumin, 0.1 mM spermidine, 5% glycerol). Reactions were performed in 10 μ l final volume for 15 min at room temperature and stopped by the addition of 1 μ l of 5% SDS and 2 μ l of BlueJuice™ Gel Loading Buffer (Life Technologies, Saint-Aubin, France). Reaction mixtures were then electrophoresed in 1% agarose gels for 2 h at 25 V in 0.5 \times TAE buffer. Gels were stained by incubation in 0.5 \times TAE solution containing 2 μ g/ml ethidium bromide for 5 min and destained by incubation in fresh 0.5 \times TAE buffer containing 1 mM MgSO₄ for 30 min. Reaction products were then visualized by UV-transillumination and quantified using the ImageJ software. Percentages of relaxed DNA (R) were normalized to reactions conducted with Top1 alone (set at 100%) and plotted as a function of foldamer concentrations. Results are the mean \pm standard deviation of at least three independent experiments.

HIV-1 IN-catalyzed integration assays

HIV-1 IN was purified from bacteria using the same procedure as previously described (46). Typical concerted integration assays were carried as previously described (47) using both pBSK-Zeo target and HIV-1 U5 containing donor DNA (48). For this purpose, we used 5'-end-labeled donor DNA (10 ng), circular target DNA plasmids (50 ng) and purified IN previously diluted to 2 μ M in 1 M NaCl, 20 mM HEPES pH 7, 10 mM DTT. Then 200 nM IN was incubated for 30 min on ice with 10 ng of donor DNA and 50 ng of acceptor plasmid in 5 μ l final volume in the presence or lack of foldamers. Reaction was then started by adding 5 μ l of the reaction buffer (final concentrations 100 nM IN, 15% DMSO, 8% PEG, 10 mM MgCl₂, 20 μ M ZnCl₂, 100

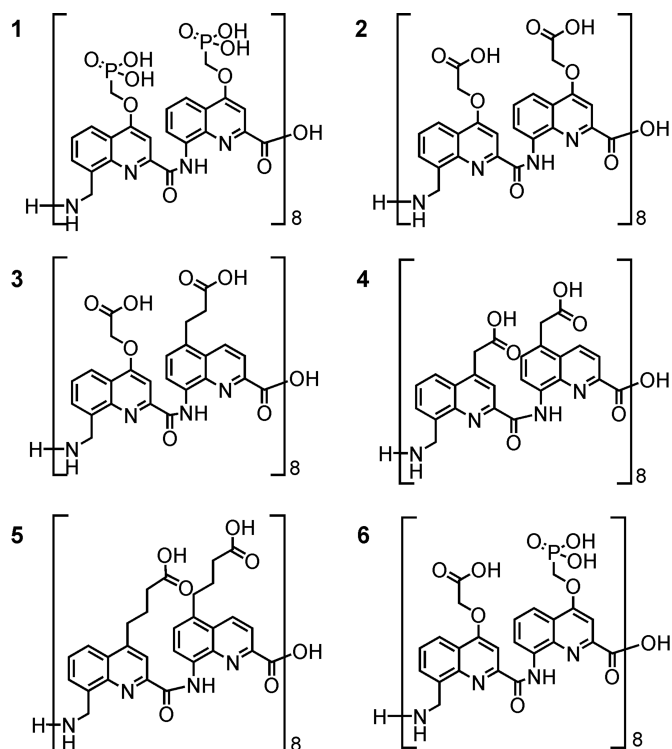


Figure 2. Formulae of DNA mimic hexadecaamides 1–6.

mM NaCl, 10 mM DTT). After the reaction, integration products were loaded onto 1.4% agarose gel. The gel was then dried and autoradiography was performed. Quantification of the integration activity was performed using the ImageJ software with the following procedure: the bands corresponding to the free substrate (S), the donor/donor (d/d), linear FSI (FSI) and circular HSI + FSI (HSI + FSI) were quantified. Results are the mean \pm standard deviation of at least three independent experiments.

RESULTS AND DISCUSSION

Design and synthesis

Previously described sequence **1** (Figure 2) bearing phosphonate side chains in position 4 of both Q and mQ units serves as the reference compound for this study. It was shown to have excellent inhibitory properties of both HIV-1 IN and Top1. Several notable differences exist between this compound and B-DNA. Phosphonates can be either mono- or di-anionic depending on the local environment (33). In addition, single point attachments of the side chains to the foldamer backbone allow for more flexible positioning of the anions than for doubly connected DNA phosphates. Also, contrary to B-DNA and to $({}^mQQ^{5Pho})_n$ sequences (Figure 1B, E, I), the two grooves of $({}^mQQ^4)_n$ differ marginally in width (Figure 1B, D): the Q^4 - mQ connectivity differs from the mQ - Q^4 connectivity by a single methylene unit. Finally, the single helical foldamer main chain may in principle be kinked upon rotation about a single bond and may thus possess larger flexibility than double stranded helical B-DNA.

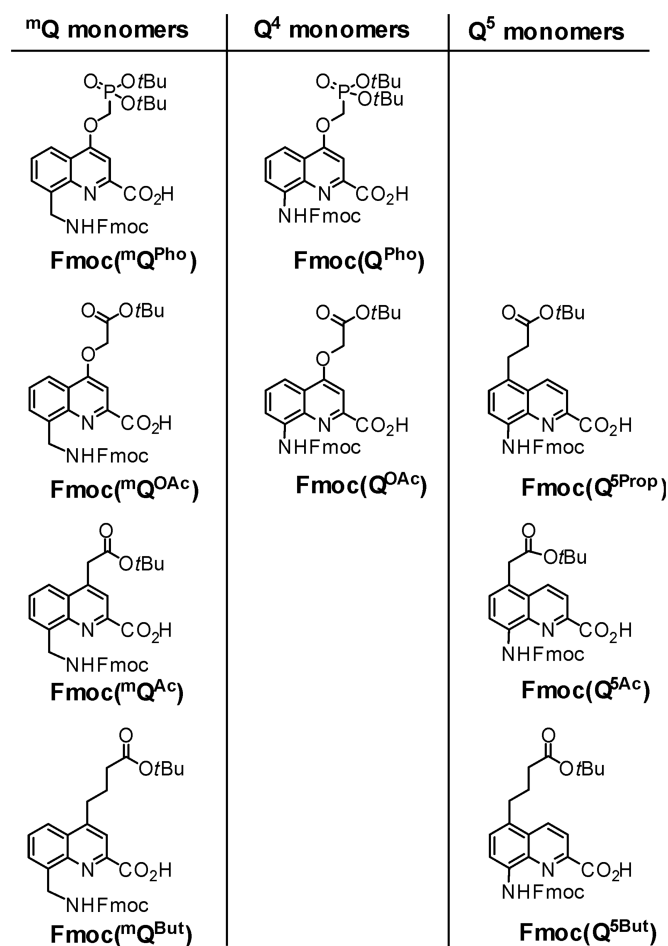


Figure 3. Formulae of Fmoc-main chain amine and *tert*-butyl-side chain acid protected monomers for SPS.

Sequences 2–6 (Figure 2) were designed by analogy with carboxylate functionalized DNA-mimic proteins. They represent variations of side chain nature and charge state (2, 6), side chain position (3, 4, 5), and side chain length (4, 5) with the hypothesis that longer side chains may adjust their conformation to form salt bridges or hydrogen bonds with the targeted enzyme. Although no accurate prediction could be made, we expected that at least some of these variations would modulate foldamer inhibitory activity, possibly in a way that can be correlated to structure.

For the preparation of these sequences, we anticipated that SPS would be less labor intensive than the previously reported solution phase synthesis, whose main advantage is to be easily amenable to scale up. SPS of Q_n sequences bearing different proteinogenic side chains has been reported before (49–51) and we thought its adaptation to $({}^mQQ)_n$ sequences would not require much development. The SPS of 1–6, entailed the preparation of the nine monomers shown in Figure 3. These bear Fmoc amine main chain protection and acid labile *tert*-butyl ester side chain protection instead of the *tert*-butyl-carbamate amine protection and diethylphosphonate ester protection used in solution (33). All of them are new except Q^{OAc} and Q^{5Ac} (49,50). Of note are the different ether or carbon-carbon bond main chain-side

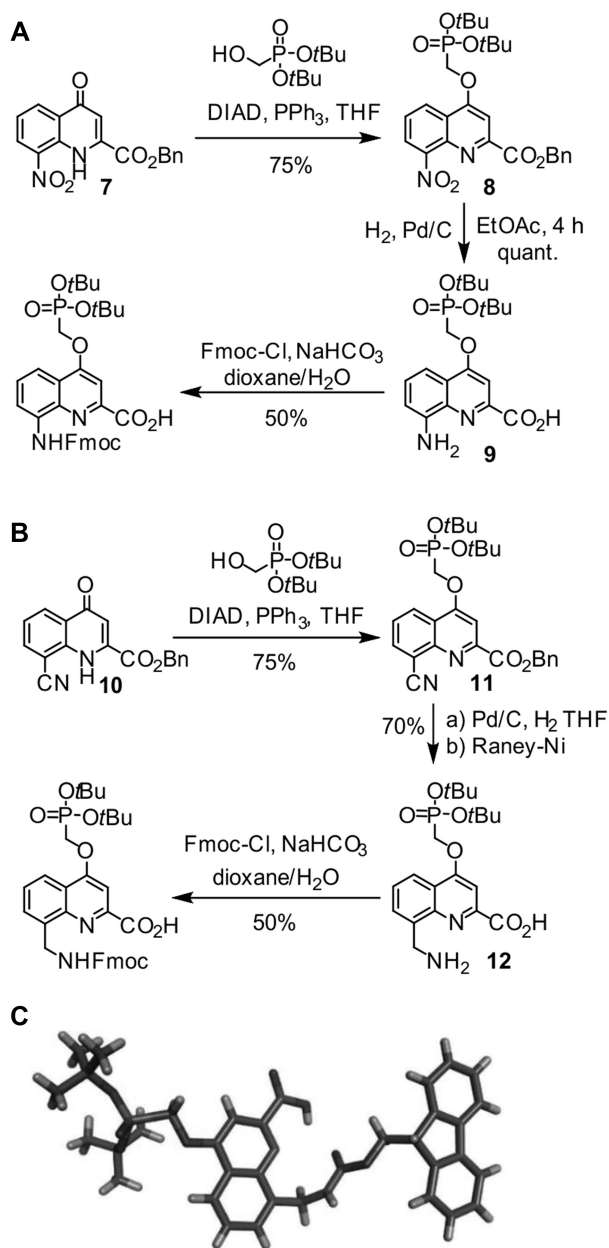


Figure 4. Synthetic pathway to (A) **Fmoc(Q^{Pho})** and (B) **Fmoc(mQ^{Pho})** monomers. Crystal structure of **Fmoc(mQ^{Pho})** monomer (C) crystallized from dichloromethane/ether. Included solvent molecules have been removed for clarity. DIAD = diisopropyl azodicarboxylate; Fmoc = fluorenylmethoxycarbonyl.

chain linkages. These do not reflect a specific design but are simply a consequence of the synthetic route followed (49,50).

The syntheses of protected **Fmoc(Q^{Pho})** and **Fmoc(mQ^{Pho})** are shown in Figure 4 as representative examples (see also Supplementary Information). Precursors **7** and **10** were prepared on a multi gram scale following previously reported procedures (52). Side chain introduction under Mitsunobu conditions, hydrogenolysis of the benzyl ester and hydrogenation of the 8-nitro or 8-cyano group followed by installation of Fmoc and purification through silica-gel chro-

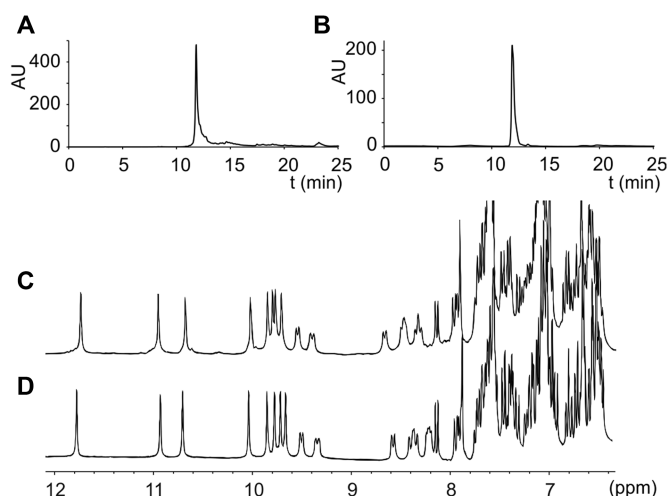


Figure 5. (A, B) RP-HPLC profiles of (A) crude and (B) pure oligomer **4**. (C, D) Carboxamide and aromatic regions of the ¹H NMR spectra of (C) crude and (D) pure oligomer **4** in H₂O/D₂O 9:1 (vol/vol), 50 mM NH₄HCO₃ at 298 K. RP-HPLC = reversed phase high-performance liquid chromatography.

matography afforded to the final compounds ready to use in SPS in *ca.* 50% overall yield and >98% purity (RP-HPLC, Supplementary Figures S48 and S49). A crystal structure of protected monomer **Fmoc(mQ^{Pho})** is shown in Figure 4C.

Drawing on our experience in microwave assisted oligoamide foldamer SPS (49–51), we prepared oligomers **1–6** on a low loading Wang resin *via* acid chloride activation of the monomers under neutral conditions with the Ghosez reagent (1-chloro-*N,N*,2-trimethylpropenylamine) (53). Removal from the resin with concomitant side chain *tert*-butyl deprotection was performed by treatment with 95:2.5:2.5 TFA/H₂O/*i*Pr₃SiH (vol/vol/vol) (TFA = trifluoroacetic acid; *i*Pr₃SiH = triisopropylsilane). The crude materials were analyzed and purified by RP-HPLC providing foldamers **1–6** in good purity (RP-HPLC purity > 95%, Supplementary Figures S43–S47). All foldamers were characterized by ¹H NMR in H₂O/D₂O, HPLC, and mass spectrometry (see Supplementary Information). Semi-preparative RP-HPLC proved to be an efficient analytical method to purify anionic foldamers providing the sequences in high purity. As a representative example, Figure 5 shows analytical data for **4** before and after purification. As a proof of the helical folding in water, NMR spectra displayed sharp amide and aromatic protons which are distributed over a wide range of chemical shift (from 12 to 9 ppm for amide protons and from 9 to 8 ppm for aromatic protons, Figure 5C and D), in agreement to our previous reported studies (49,54).

Structural characterization

Structural evidence of anionic oligomer helical folding was previously established in aqueous solution by NMR investigations of an octameric sequence (33). Folding typically results in the spreading of NMR resonances over a broad range of chemical shift values despite the repetitive nature of the sequence, as can be seen in Figure 5D and in Supplemen-

tary Figures S9–S13. The sharp lines are a good indication of the absence of foldamer aggregation even at mM concentrations, as could be expected for a polyelectrolyte. Solid state evidence of the helical conformations had also been obtained by X-ray crystallographic analyses. These structures were obtained from crystals grown in organic solvents using side chain protected oligomers. However, earlier attempts to grow crystals from unprotected anionic oligomers such as **1** have been unsuccessful. We assigned this difficulty to the relative length of the phosphonate side chain, which appears to be detrimental to crystal growth from water (50). This hypothesis was verified when attempting to grow crystals of the new carboxylate-functionalized oligomers: crystal growth from water was successful, but only in the case of hexadecaamide (${}^m\text{Q}^{\text{Ac}}\text{Q}^{\text{5Ac}}$)₈ which bears the shortest side chains. Crystals that diffracted at atomic resolution were obtained and the structure could be resolved in centrosymmetrical space group $P\bar{1}$ (Figure 6C and Supplementary Figure S1). The asymmetric unit contained a molecule **4** folded as a single helix with no noticeable difference from the structures obtained from organic solvents. As with other helical aromatic amides (50,54), and unlike with α -helical peptides (55,56), electrostatic repulsion between the multiple charges is not conducive of a denaturation, or even of a structural change. This is despite the fact that charges are close enough to influence their apparent pK_a values, as was measured for phosphonate side chains (33).

The structure of **4** also contained six Ca^{2+} ions used as an additive in crystallization drop to screen anion repulsions between adjacent molecules (Supplementary Figures S2, S4 and S5). Divalent metal ions often serve that purpose in the crystal structures of nucleic acids (57–59) and other polyanionic species (60).

The helically folded structure of (${}^m\text{Q}^{\text{Ac}}\text{Q}^{\text{5Ac}}$)₈ matched the predicted model of (${}^m\text{Q}^{\text{Pho}}\text{Q}^{\text{5Pho}}$)₈ (Figure 6D, E) especially in the central part of the sequence, and the crystal structure of protected (${}^m\text{Q}^{\text{Pho}}\text{Q}^{\text{Pho}}$)₈ obtained from organic solvents (Supplementary Figure S6). It confirmed the resemblance between these oligomers and B-DNA (Figure 6A–C). In particular, there is a good match of the minor and major groove widths when side chains are in position 5 of Q units (Supplementary Figure S3). Even solid state packing was found to be analogous to the packing of related DNA structures (Supplementary Figure S5). Nevertheless, the shorter side chains of ${}^m\text{Q}^{\text{Ac}}$ and Q^{5Ac} are conducive of an overall diameter significantly smaller than that of B-DNA (Figure 6A–C).

Inhibition of DNA-processive enzymes

We then assessed the effect of the foldamers on the catalytic transesterification of scissile phosphodiester bonds of recombinant Top1 and HIV-1 IN using dedicated *in vitro* assays, as previously described (33) (Figure 7, Table 1). These enzymes share a common strand-transferase activity. Despite scant overall sequence similarity, Top1 also shares a conserved catalytic pentad in its tertiary structure with a number of other integrases (61). We compared the effect of the new variants to that of reference oligomer **1** (${}^m\text{Q}^{\text{Pho}}\text{Q}^{\text{Pho}}$)₈ which completely inhibits Top1-mediated DNA relaxation at 1 μM concentration (Figure 7C) and

Table 1. Inhibition of Top1 and HIV-1 IN by anionic foldamers **1–6**. IC_{50} values (concentrations of each DNA mimic that induce 50% inhibition of the catalytic activity) and standard deviations were extracted from Figure 7A and B. nd: not determined due to the lack of inhibitory activity in the range of concentration investigated

	IC_{50} (μM)	
	Top1	HIV-1 IN
1 (${}^m\text{Q}^{\text{Pho}}\text{Q}^{\text{Pho}}$) ₈	0.225 \pm 0.04	2.0 \pm 0.4
2 (${}^m\text{Q}^{\text{OAc}}\text{Q}^{\text{OAc}}$) ₈	nd.	nd. (>10)
3 (${}^m\text{Q}^{\text{OAc}}\text{Q}^{\text{5Prop}}$) ₈	nd.	2.2 \pm 0.2
4 (${}^m\text{Q}^{\text{Ac}}\text{Q}^{\text{5Ac}}$) ₈	nd. (activation)	nd. (activation)
5 (${}^m\text{Q}^{\text{But}}\text{Q}^{\text{5But}}$) ₈	0.83 \pm 0.06	0.80 \pm 0.02
6 (${}^m\text{Q}^{\text{OAc}}\text{Q}^{\text{Pho}}$) ₈	0.94 \pm 0.12	0.08 \pm 0.03

HIV-1 IN integration at 10 μM concentration (Figure 7D). The results showed that the position and nature of the anions strongly affected the inhibitory activity of the DNA mimics and that, in many cases, they did so differently for the two enzymes.

A first striking result was the complete inactivity of oligomer **2** for both enzymes. Considering that **2** differs from **1** only by the carboxylate vs. phosphonate nature of the anions, this result points to an essential role of this parameter. When only half of the phosphonates are replaced by carboxylates, *i.e.* in **6** (${}^m\text{Q}^{\text{OAc}}\text{Q}^{\text{Pho}}$)₈, activity was only moderately restored for Top1 (IC_{50} of 0.2 μM for **1** and 0.94 μM for **6**). In contrast, activity was fully restored and even enhanced for HIV-1 IN (IC_{50} of 0.08 μM). This latter result suggests that optimal affinity arises at a given charge balance – neither not enough nor too many negative charges are desirable – and that this parameter is enzyme-dependent. Indeed, the IC_{50} values of **6** for the two enzymes differ by more than one order of magnitude, inhibition of HIV-1 IN being the strongest, whereas **1** inhibits Top 1 at a lower concentration than HIV-1 IN. Nevertheless, one should keep in mind that these comparisons are valid for what concerns the active concentration of **1** and **6**, but are not strictly quantitative in that the enzyme concentration and the substrate nature and concentration differ in the two assays.

It is worth stressing here how remarkable is the very occurrence of strong competitive inhibition of a DNA-binding enzyme by a DNA-mimic. The assays used to assess enzyme activity are carried out in the presence of B-DNA substrates in considerably larger amounts, if one considers both their length and concentration, than the enzyme and the foldamer used for inhibition. Even though processive enzymes may not possess very high binding to DNA to allow for their scrolling along DNA sequences (to be the best of our knowledge, the exact dissociation constant for Top1 is not known and the non-Michaelis-Menten nature of HIV-1 IN, due to its multiple oligomeric active states has not allowed for the precise determination of its affinity for DNA to date) competitive inhibition can only be achieved if the foldamer binds much stronger.

A second striking and differential effect came from changing side chain position. Shifting some carboxylate side chains of inactive **2** from position 4 to position 5 of the quinoline rings, as in **3** (${}^m\text{Q}^{\text{OAc}}\text{Q}^{\text{5Prop}}$)₈, did not restore

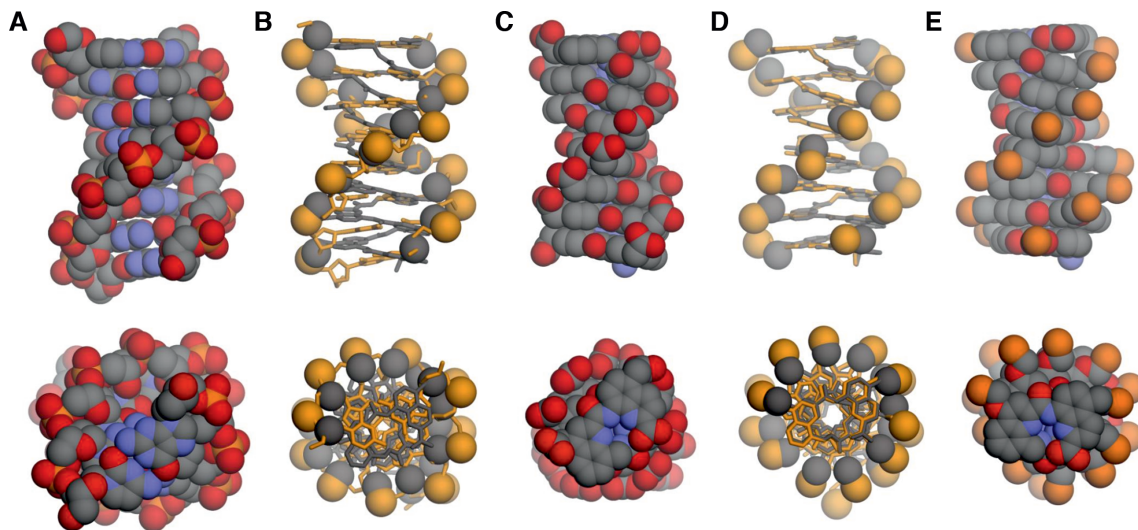


Figure 6. Side views and top views shown at the same scale of: (A) crystal structure of 8-bp B-DNA duplex d(CGCTAGCG)₂ (PDB ID: 250D) (70); (B) molecular overlay of B-DNA (ochre) and DNA mimic 4 (gray); (C) crystal structure of DNA mimic 4; (D) molecular overlay of DNA mimic 4 (gray) and a molecular model of (mQ^{Pho}Q^{5Pho})₈ (ochre) and (E) a molecular model of (mQ^{Pho}Q^{5Pho})₈. In (B) and (D), foldamers or DNA are shown in stick representations, except the side chain carboxylate carbon atoms, and the phosphonate and phosphate phosphorus atoms that are shown as spheres. Two molecules of (mQ^{Pho}Q^{5Pho})₄ as observed in the crystal (33) have been stacked to produce the model of (mQ^{Pho}Q^{5Pho})₈. Hydrogen atoms have been removed for clarity. In (E), ethyl ester functions at the side chains, *tert*-butyl-carbamate and benzyl ester functions at the N terminus and C terminus, respectively, have been removed for clarity.

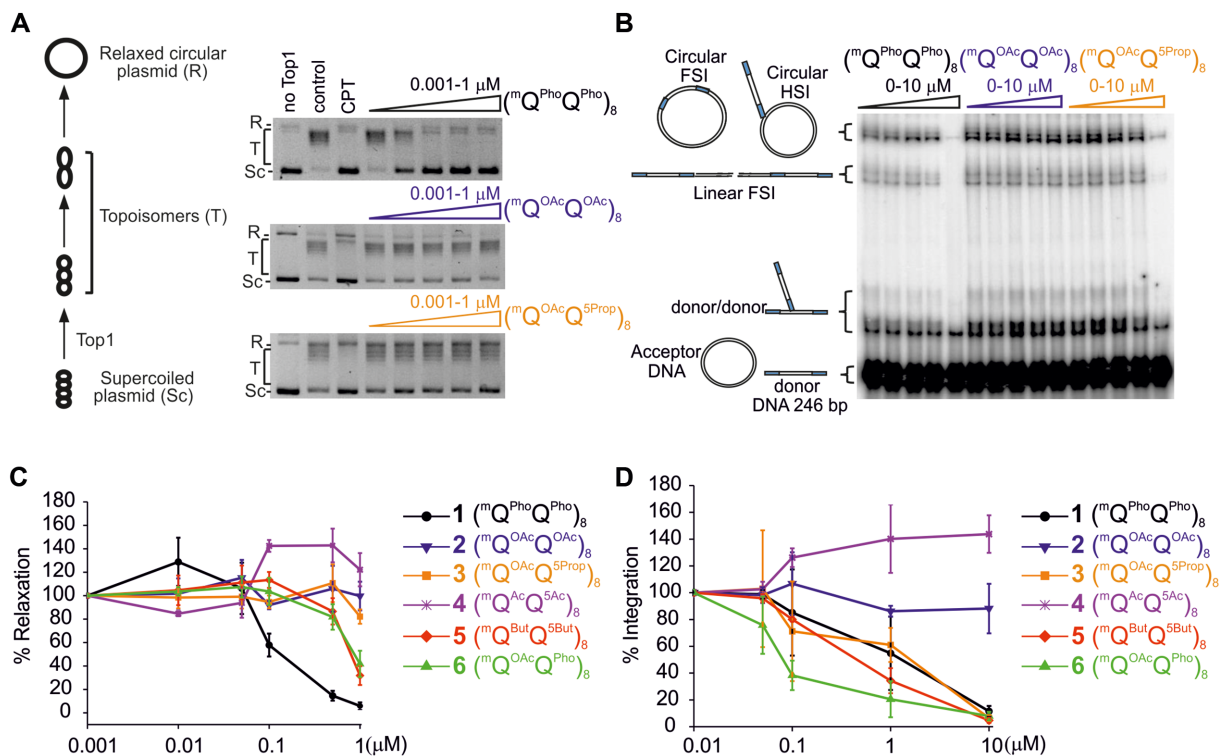


Figure 7. Enzyme inhibition. (A, B) Representative gel electrophoresis of (A) the inhibition of Top1-mediated relaxation of supercoiled circular DNA and (B) *in vitro* HIV-1 IN integration (200 nM of IN) of a radio-labelled 246-base-pair (bp) viral DNA donor (10 nM) into a 2,700 bp pZeo plasmid acceptor (2.8 nM). In (A), lane 1 is DNA alone, control lane 2 is DNA + Top1 and lane 3 is the same as lane 2 in the presence of 50 μM camptothecin (CPT). The control lane 2 shows full Top1 activity. Camptothecin (CPT) is a classical Top1 inhibitor. In (B), for each foldamer, the left lane is the control lane (*i.e.* all things equal but without any foldamer). The position and structures of the donor substrate and different products obtained after half-site (HSI), full-site (FSI) and donor/donor integration are shown. (C, D) Quantitation of (C) Top1 inhibition expressed as a percentage of relaxation, compared to the control, and (D) *in vitro* HIV-1 IN integration. In (D), the circular FSI + HSI and linear FSI products were quantified on gels using the ImageJ software and the circular and linear HSI + FSI is reported as the percentage of heterointegration quantified in the absence of foldamers. Results are the means of at least triplicate experiments ± standard deviation.

Top1 inhibition but it did restore HIV-1 IN inhibition. Thus carboxylate oligomer **3** is as active against HIV-1 IN as phosphonate oligomer **1**. The effect is also observed when the carboxylate side chains are made longer. Oligomer **5** (${}^m\text{Q}^{\text{But}}\text{Q}^{\text{5But}}$)₈ inhibits HIV-1 IN slightly better than **3**. It is also the only all-carboxylate oligomer that shows some inhibition of Top1 (IC₅₀ of 0.83 μM). In addition to **6**, oligomers **3** and **5** can thus also be considered to be more selective for HIV-1 IN.

Finally, a peculiar effect was observed when side chains are made shorter, as in **4**. This compound did not show any inhibition. On the contrary, it activated Top1 and HIV-1 IN above 0.1 μM. The fact that it occurred with both proteins in completely different assays suggests it is not an artifact. The reduced diameter of this oligomer (Figure 6C) hints at a non-optimal fit with the proteins' DNA binding site. Weaker binding would then not be conducive of competitive inhibition but it might nevertheless promote active conformations of the proteins for binding to DNA, resulting in an activation. Such an activation was indeed reported using short DNA duplexes (47), and might be associated with the presence of other DNA binding sites. Indeed, several DNA binding sites have been characterized in HIV-1 IN. The viral DNA binds the catalytic core domain while the target DNA binds both the catalytic core and the carboxyterminal domain which possesses a non-sequence specific DNA binding property and which is thus probably involved in foldamer binding.

The wide range of activities of the new oligomers demonstrate the feasibility of selective enzyme inhibition for two enzymes that have DNA as a common substrate. HIV-1 IN inhibition shows remarkable dependence on all the parameters varied (side chain position, length, and nature of the anion) whereas Top1 inhibition appears to depend mostly on the presence of the phosphonate ions.

Unfortunately, the lack of quantitative binding assays does not allow one to directly relate the observed inhibitions to protein-foldamer affinity. The results presented here also call for further screening of foldamer variations. For example, aromatic tyrosine-like or tryptophan-like foldamer side chains might also interact favorably with positively charged residues at the protein surface and further enhance selectivity. Higher throughput foldamer syntheses (e.g. automation of SPS), would facilitate such plans and is currently being developed.

CONCLUSION

Nucleic acid surface mimicry appears to have high potential for modulating protein binding to DNA. Yet it has been vastly overlooked until now, perhaps out of lack of suitable molecular scaffolds, and also because the focus of DNA mimicry has been centered on Watson-Crick base pairing ability (1–13). This situation contrasts with the significant advances in protein surface mimicry for the purpose of inhibiting protein-protein interactions (62–64). For example, α-helix mimicry and stabilization constitutes a field of its own in which foldamer-based approaches prove to be successful (65–69). The successes met in protein epitope mimicry may thus be a source of inspiration for nucleic acid surface mimicry. Along this line, the results presented

above consolidate our claim that helically folded aromatic oligoamides are valid and original scaffolds to reproduce some of the charge surface features of B-DNA. The identification of foldamer side chain patterns that elicit selective inhibition properties, that is, HIV-1 IN versus Top1, is an important milestone. This is especially true considering two proteins that act on a similar B-DNA substrate, *i.e.* at the exclusion of a well-defined DNA sequence binding site. More generally, the strong dependence of inhibition on the spatial distribution and nature of anionic side chains point to the existence of protein-specific foldamer-protein interactions, and thus to the possibility of further tailoring activity and selectivity. Next steps will be to determine the structural requirements for these foldamers to achieve stronger and more selective inhibition, a process that would require the use of crystal structure data (34,35) as well as molecular modelling. Targeting Top1 or HIV-1 IN by a new mechanism of action, namely competitive inhibition, using such foldamers would represent an alternative to existing drugs that are used in the clinic to poison these two recombinases in the treatment of cancer or AIDS, respectively. Such an alternative might help counteracting drug resistance that develops in both pathologies. In the long term, our results also point out towards the use of foldamers to inhibit the activity of other key cellular DNA-interacting enzymes that are specifically deregulated in human diseases or that are difficult to target because of their ubiquitous nature (*i.e.* transcription factors) and/or the absence of pharmacological inhibitors. Identification of those potential targets is currently under way.

DATA AVAILABILITY

Atomic coordinates and structure factors for the reported crystal structures have been deposited with the Cambridge Crystallographic Data Centre (CCDC) under accession numbers 1527431 and 1474392 for **4** and **Fmoc**(${}^m\text{Q}^{\text{Pho}}$), respectively.

SUPPLEMENTARY DATA

[Supplementary Data](#) are available at NAR Online.

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