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## Structure Reports

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2-Amino-*N*,3-dimethylbenzamide

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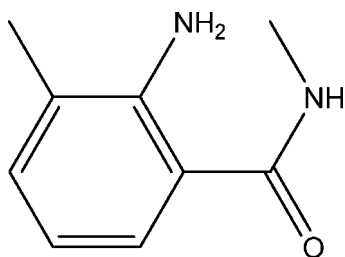
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Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.042;  $wR$  factor = 0.111; data-to-parameter ratio = 17.3.

In the title compound,  $\text{C}_9\text{H}_{12}\text{N}_2\text{O}$ , the mean plane through the amide group and the benzene ring form a dihedral angle of  $33.93(7)^\circ$ . An intramolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond is present. In the crystal, molecules are linked by  $\text{N}-\text{H}\cdots\text{N}$  and  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds, forming double-stranded chains parallel to the  $b$  axis.

## Related literature

For background to substituted anthranilamides, see: Gnamm *et al.* (2012); Lahm *et al.* (2005); Norman *et al.* (1996); Roe *et al.* (1999). For the synthesis, see: Staiger & Wagner (1953); Coppola (1980); Witt & Bergman (2000).



## Experimental

## Crystal data

$\text{C}_9\text{H}_{12}\text{N}_2\text{O}$   
 $M_r = 164.21$   
 Monoclinic,  $P2_1$   
 $a = 9.833(6)$  Å  
 $b = 5.011(3)$  Å  
 $c = 9.841(6)$  Å  
 $\beta = 118.27(1)^\circ$

$V = 427.1(4)$  Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.36 \times 0.13 \times 0.10$  mm

## Data collection

Rigaku MM007-HF CCD (Saturn 724+) diffractometer  
 Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2007)  
 $T_{\min} = 0.970$ ,  $T_{\max} = 0.992$

3828 measured reflections  
 1939 independent reflections  
 1884 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.036$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.111$   
 $S = 1.05$   
 1939 reflections  
 112 parameters

1 restraint  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.30$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.22$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1A}\cdots\text{O1}$	0.88	2.21	2.785 (2)	123
$\text{N1}-\text{H1B}\cdots\text{N1}^{\text{i}}$	0.88	2.44	3.240 (2)	151
$\text{N2}-\text{H2}\cdots\text{O1}^{\text{ii}}$	0.88	2.18	2.858 (2)	133

Symmetry codes: (i)  $-x, y - \frac{1}{2}, -z + 1$ ; (ii)  $x, y + 1, z$ .

Data collection: *CrystalClear* (Rigaku, 2007); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ5025).

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## supplementary materials

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## 2-Amino-*N*,3-dimethylbenzamide

Xiang-dong Mei, Yan-hui Liang and Ke-Bin Li

### Comment

Anthranilamide-based derivatives exhibit interesting biological activities such as antibacterial, antifungal, antiviral and insecticidal effects (Gnamm *et al.*, 2012; Lahm *et al.*, 2005; Norman *et al.*, 1996; Roe *et al.*, 1999). We report here the crystal structure of the title compound 2-amino-*N*,3-dimethylbenzamide, an important organic intermediate in the synthesis of medicines, agricultural chemicals, and animal drugs.

In the title compound (Fig. 1), the least-square mean plane through the amide group (C8/C9/O1/N2) form a dihedral angle of 33.93 (7)° with the benzene ring. The molecular conformation is stabilized by an intramolecular N—H···O hydrogen bond (Table 1). In the crystal structure, intermolecular N—H···N and N—H···O hydrogen interactions link molecules into double chains running parallel to the *b* axis.

### Experimental

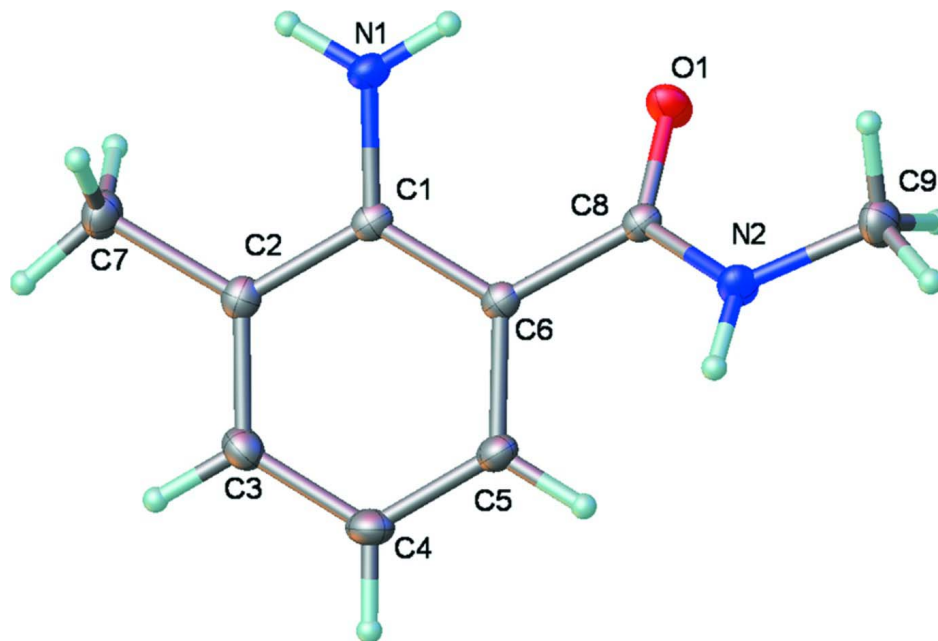
The title compound was prepared according to the literature method (Witt & Bergman, 2000) by stirring isatoic anhydride with aqueous methylamine. Isatoic anhydride was prepared by reaction of anthranilic acid with triphosgene in good yield (Coppola, 1980). The title compound (0.2 g) was dissolved in ethanol (50 ml) at room temperature. Colourless blocks were obtained through slow evaporation after two weeks.

### Refinement

The H atoms were placed at calculated positions, with C—H = 0.93–0.98 Å, and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2–1.5U_{\text{eq}}(\text{C})$ . 841 Friedel pairs were not merged.

### Computing details

Data collection: *CrystalClear* (Rigaku, 2007); cell refinement: *CrystalClear* (Rigaku, 2007); data reduction: *CrystalClear* (Rigaku, 2007); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008).

**Figure 1**

The molecular structure of the title compound showing 50% probability displacement ellipsoids and atom-numbering scheme.

## 2-Amino-*N*,3-dimethylbenzamide

### Crystal data

$C_9H_{12}N_2O$

$M_r = 164.21$

Monoclinic,  $P2_1$

Hall symbol:  $P\ 2_1yb$

$a = 9.833\ (6)\ \text{\AA}$

$b = 5.011\ (3)\ \text{\AA}$

$c = 9.841\ (6)\ \text{\AA}$

$\beta = 118.27\ (1)^\circ$

$V = 427.1\ (4)\ \text{\AA}^3$

$Z = 2$

$F(000) = 176$

$D_x = 1.277\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2240 reflections

$\theta = 2.4\text{--}32.7^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 173\ \text{K}$

Rod, colourless

$0.36 \times 0.13 \times 0.10\ \text{mm}$

### Data collection

Rigaku MM007-HF CCD (Saturn 724+) diffractometer

Radiation source: rotating anode

Confocal monochromator

$\omega$  scans at fixed  $\chi = 45^\circ$

Absorption correction: multi-scan

(*CrystalClear*; Rigaku, 2007)

$T_{\min} = 0.970$ ,  $T_{\max} = 0.992$

3828 measured reflections

1939 independent reflections

1884 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.036$

$\theta_{\max} = 27.5^\circ$ ,  $\theta_{\min} = 2.4^\circ$

$h = -12 \rightarrow 12$

$k = -6 \rightarrow 6$

$l = -12 \rightarrow 12$

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.042$	H-atom parameters constrained
$wR(F^2) = 0.111$	$w = 1/[\sigma^2(F_o^2) + (0.0574P)^2 + 0.1149P]$
$S = 1.05$	where $P = (F_o^2 + 2F_c^2)/3$
1939 reflections	$(\Delta/\sigma)_{\max} < 0.001$
112 parameters	$\Delta\rho_{\max} = 0.30 \text{ e } \text{\AA}^{-3}$
1 restraint	$\Delta\rho_{\min} = -0.22 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.18145 (16)	0.6816 (2)	0.25039 (14)	0.0310 (3)
N1	0.08945 (16)	0.5876 (3)	0.47375 (16)	0.0267 (3)
H1A	0.0480	0.5886	0.3725	0.032*
H1B	0.0504	0.4853	0.5193	0.032*
N2	0.22924 (17)	1.1192 (3)	0.24907 (16)	0.0276 (3)
H2	0.2598	1.2682	0.3025	0.033*
C1	0.21634 (17)	0.7472 (3)	0.56154 (18)	0.0215 (3)
C2	0.27443 (18)	0.7589 (3)	0.72312 (19)	0.0245 (3)
C3	0.3954 (2)	0.9304 (4)	0.80932 (19)	0.0300 (4)
H3	0.4353	0.9365	0.9181	0.036*
C4	0.4600 (2)	1.0941 (4)	0.7407 (2)	0.0325 (4)
H4	0.5431	1.2102	0.8019	0.039*
C5	0.40169 (18)	1.0856 (4)	0.58261 (19)	0.0266 (3)
H5	0.4443	1.1990	0.5351	0.032*
C6	0.28106 (17)	0.9128 (3)	0.49133 (17)	0.0211 (3)
C7	0.2030 (2)	0.5880 (4)	0.7989 (2)	0.0330 (4)
H7B	0.0947	0.6394	0.7603	0.050*
H7C	0.2593	0.6139	0.9109	0.050*
H7A	0.2085	0.3999	0.7748	0.050*
C8	0.22609 (17)	0.8953 (3)	0.32129 (18)	0.0214 (3)
C9	0.1841 (2)	1.1254 (4)	0.08586 (19)	0.0343 (4)
H9A	0.0791	1.0551	0.0268	0.051*
H9B	0.2555	1.0156	0.0663	0.051*
H9C	0.1873	1.3098	0.0542	0.051*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0437 (7)	0.0195 (6)	0.0277 (6)	-0.0021 (5)	0.0152 (5)	-0.0023 (4)
N1	0.0268 (7)	0.0240 (6)	0.0291 (7)	-0.0067 (6)	0.0130 (5)	-0.0007 (6)
N2	0.0377 (8)	0.0188 (7)	0.0282 (7)	-0.0033 (6)	0.0173 (6)	-0.0010 (6)
C1	0.0194 (7)	0.0172 (7)	0.0290 (8)	0.0029 (6)	0.0124 (6)	0.0013 (6)
C2	0.0247 (8)	0.0230 (8)	0.0278 (8)	0.0043 (7)	0.0140 (7)	0.0030 (6)
C3	0.0287 (8)	0.0356 (9)	0.0235 (7)	-0.0004 (7)	0.0106 (7)	-0.0026 (7)
C4	0.0271 (8)	0.0344 (9)	0.0328 (8)	-0.0085 (8)	0.0115 (7)	-0.0077 (8)
C5	0.0260 (7)	0.0256 (7)	0.0314 (8)	-0.0056 (7)	0.0162 (7)	-0.0032 (7)
C6	0.0208 (7)	0.0185 (7)	0.0255 (7)	0.0018 (6)	0.0123 (6)	-0.0004 (6)
C7	0.0387 (9)	0.0323 (9)	0.0313 (8)	-0.0008 (8)	0.0192 (7)	0.0052 (8)
C8	0.0207 (7)	0.0181 (7)	0.0272 (7)	0.0016 (6)	0.0127 (6)	0.0007 (6)
C9	0.0431 (10)	0.0314 (10)	0.0286 (8)	-0.0010 (8)	0.0171 (8)	0.0043 (7)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

O1—C8	1.239 (2)	C3—H3	0.9500
N1—C1	1.386 (2)	C4—C5	1.381 (2)
N1—H1A	0.8800	C4—H4	0.9500
N1—H1B	0.8800	C5—C6	1.398 (2)
N2—C8	1.337 (2)	C5—H5	0.9500
N2—C9	1.450 (2)	C6—C8	1.498 (2)
N2—H2	0.8800	C7—H7B	0.9800
C1—C6	1.410 (2)	C7—H7C	0.9800
C1—C2	1.413 (2)	C7—H7A	0.9800
C2—C3	1.384 (2)	C9—H9A	0.9800
C2—C7	1.510 (2)	C9—H9B	0.9800
C3—C4	1.392 (3)	C9—H9C	0.9800
C1—N1—H1A	120.0	C6—C5—H5	119.5
C1—N1—H1B	120.0	C5—C6—C1	119.49 (14)
H1A—N1—H1B	120.0	C5—C6—C8	120.10 (14)
C8—N2—C9	122.35 (15)	C1—C6—C8	120.36 (14)
C8—N2—H2	118.8	C2—C7—H7B	109.5
C9—N2—H2	118.8	C2—C7—H7C	109.5
N1—C1—C6	121.07 (15)	H7B—C7—H7C	109.5
N1—C1—C2	119.33 (14)	C2—C7—H7A	109.5
C6—C1—C2	119.43 (14)	H7B—C7—H7A	109.5
C3—C2—C1	119.16 (14)	H7C—C7—H7A	109.5
C3—C2—C7	121.00 (16)	O1—C8—N2	121.14 (14)
C1—C2—C7	119.84 (15)	O1—C8—C6	121.59 (14)
C2—C3—C4	121.71 (15)	N2—C8—C6	117.26 (14)
C2—C3—H3	119.1	N2—C9—H9A	109.5
C4—C3—H3	119.1	N2—C9—H9B	109.5
C5—C4—C3	119.12 (16)	H9A—C9—H9B	109.5
C5—C4—H4	120.4	N2—C9—H9C	109.5
C3—C4—H4	120.4	H9A—C9—H9C	109.5
C4—C5—C6	121.09 (15)	H9B—C9—H9C	109.5

C4—C5—H5	119.5		
N1—C1—C2—C3	-176.23 (16)	N1—C1—C6—C5	175.30 (15)
C6—C1—C2—C3	-0.9 (2)	C2—C1—C6—C5	0.1 (2)
N1—C1—C2—C7	3.0 (2)	N1—C1—C6—C8	-7.4 (2)
C6—C1—C2—C7	178.30 (16)	C2—C1—C6—C8	177.33 (13)
C1—C2—C3—C4	0.8 (3)	C9—N2—C8—O1	-1.2 (3)
C7—C2—C3—C4	-178.38 (18)	C9—N2—C8—C6	177.79 (15)
C2—C3—C4—C5	0.1 (3)	C5—C6—C8—O1	144.60 (17)
C3—C4—C5—C6	-1.0 (3)	C1—C6—C8—O1	-32.7 (2)
C4—C5—C6—C1	0.9 (3)	C5—C6—C8—N2	-34.4 (2)
C4—C5—C6—C8	-176.35 (16)	C1—C6—C8—N2	148.33 (15)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1 <i>A</i> ...O1	0.88	2.21	2.785 (2)	123
N1—H1 <i>B</i> ...N1 <sup>i</sup>	0.88	2.44	3.240 (2)	151
N2—H2...O1 <sup>ii</sup>	0.88	2.18	2.858 (2)	133

Symmetry codes: (i)  $-x, y-1/2, -z+1$ ; (ii)  $x, y+1, z$ .