

OPEN

Migration and transformation of cadmium in rice - soil under different nitrogen sources in polymetallic sulfide mining areas

Xiaoxia Zhang^{1,2*}, Xuexia Zhang¹, Shuji Lv³, Lei Shi⁴ & Rongping Wang^{1,2*}

We conducted pot experiments to assess the bioavailability of cadmium (Cd) in contaminated rhizosphere soil and accumulation in rice organs in response to nitrogen (N) supply ((NH₄)₂SO₄, NH₄NO₃, NH₄Cl). The results showed that the concentration of bioavailable Cd in rice rhizosphere soil was (NH₄)₂SO₄ treatment > NH₄Cl treatment > NH₄NO₃ treatment at the same level of N application and growth period; the Cd concentration in rice roots was (NH₄)₂SO₄ treatment > NH₄NO₃ treatment > NH₄Cl treatment; and the Cd concentration in rice straw was NH₄NO₃ treatment > NH₄Cl. The Cd concentration in rice roots, straws, and seeds at the maturity stage was (NH₄)₂SO₄ treatment > NH₄Cl treatment. With the same N fertilizer, excessive N promoted Cd accumulation in rice at later growth stages. This suggested that sulfate (SO₄²⁻) influenced Cd concentration in rice. NH₄Cl application maintained a low Cd level in different rice organs with the same N level. This confirmed that NH₄Cl is a safe N source for rice planting in polymetallic sulfide mining areas. The study concludes that appropriate NH₄Cl levels for Cd-contaminated paddy soil with high-S-content could obtain rice grains with Cd concentrations below the food safety standards (0.2 or 0.4 mg·kg⁻¹).

Mining has a tremendous impact on the geochemical environment of the mine and its surrounding ecosystems¹. During the mining of polymetallic sulfide mines, metal sulfides release a large number of heavy-metal ions and acid wastewater², which flow into natural waters, such as rivers, and result in the acidification and accumulation of heavy metals in paddy soils downstream of mining areas^{3,4} as well as in heavy metals in crops exceeding the standard in agricultural products, such as rice and maize⁵⁻⁷. Excessive SO₄²⁻ retained in the soil leads to soil acidification and the formation of acid sulfate soil⁸⁻¹⁰. Plant nutrient regulation is essential to mitigate abiotic stress in sustainable agriculture. Mineral nutrients can significantly mitigate the accumulation of heavy metals in plants. In isolated studies, two essential mineral nutrients, nitrogen (N) and sulfur (S), have been reported to reduce the impact of Cd in plants and to improve overall plant growth, metabolism, and productivity under Cd exposure¹¹⁻¹³. Some studies, however, have shown a positive correlation between S sources and Cd accumulation in plants^{14,15}. Assimilation of SO₄²⁻ and nitrate (NO₃⁻) is involved in plant stress defense¹¹. The S requirement and S metabolism of plants is closely related to N metabolism. The state of one element has a significant effect on the other^{16,17}. Anjum *et al.*¹⁸ reported that assimilatory of SO₄²⁻ and NO₃⁻ reduction coordinated well in the mitigation of metal(loid) toxicity. Because of the optimization of a large N and S requirement, plants maintained healthy growth and, ultimately, a high yield under Cd stress^{19,20}. Few reports, however, are available on the interactive effects of S and N sources in the mitigation of Cd toxicity¹¹. Therefore, effort is needed to unravel the physiological mechanisms involved in the balance between Cd, S, N, and plant growth. We investigated the influence of different forms of N source on rice growth and Cd concentration in rice soil in polymetallic sulfide mining areas. We conducted pot experiments to assess the bioavailability of Cd in rhizosphere soil and examined its accumulation in different organs of *Choukoku* (*Oryza sativa* L.cv.)²¹ in response to various forms of N supply.

¹Guangdong Key Laboratory of Integrated Agro-environmental Pollution Control and Management, Guangdong Institute of Eco-environmental Science & Technology, Guangzhou, 510650, China. ²National-Regional Joint Engineering Research Center for Soil Pollution Control and Remediation in South China, Guangzhou, 510650, China. ³College of Agronomy, Hunan Agricultural University, Changsha, 410128, China. ⁴School of Metallurgy and Environment, Central South University, Changsha, 410083, China. *email: zhangxiaoxia.20088@163.com; rpwang@soil.gd.cn

	Risk screening value ^a	Risk intervention value ^a	Test value
pH	≤5.5	≤5.5	4.21
Cd/mg·kg ⁻¹	0.3	1.5	0.54
As/mg·kg ⁻¹	30	200	70.28
Cu/mg·kg ⁻¹	50		298.25
Zn/mg·kg ⁻¹	200		302.66
Cr/mg·kg ⁻¹	250	800	46.28
Pb/mg·kg ⁻¹	80	400	201.00
S/mg·kg ⁻¹			443.66
Bioavailable S/mg·kg ⁻¹			133.38

Table 1. Basic physical and chemical properties of the tested soil. ^aRisk screening values and risk intervention values for soil contamination of agricultural land, soil environmental quality national standard of the People's Republic of China, and risk control standard for soil contamination of agricultural land (GB 15618-2018).

Fertilizer	N (g N·kg ⁻¹ soil)		
	0.1 g·kg ⁻¹	0.2 g·kg ⁻¹	0.4 g·kg ⁻¹
NH ₄ Cl	7.6 g-A	15.2 g-B	30.4 g-C
(NH ₄) ₂ SO ₄	9.4 g-D	18.8 g-E	37.6 g-F
NH ₄ NO ₃	5.8 g-G	11.6 g-H	23.2 g-I
KH ₂ PO ₄	17.6 g		

Table 2. The test settings of N-level management.

Materials and Methods

Soil. We collected the soil in this study from paddy fields polluted by heavy metals located downstream of the Dabaoshan polymetallic sulfide mining area in the northern Guangdong Province. Since the 1970s, this farmland has been using acid mine wastewater from Dabaoshan for sewage irrigation. This land has had a history of sewage irrigation for more than 40 years. The level of many heavy metals, including Cd, lead (Pb), and arsenic (As), in the local soils is between the risk screening values and the risk intervention values for soil contamination of agricultural land (GB 15618-2018). In principle, agronomic control and other safe utilization measures should be taken before planting edible agricultural products that do not meet the quality and safety standards for this kind of soil pollution. The tested soil was from 0- to 20-cm-deep topsoil of a tillage layer. After air-drying, we screened the soil through a 2-mm-pore nylon sieve, and then fully mixed the soil for the pot experiment. The basic physical and chemical properties of the tested soils are shown in Table 1.

Potted experimental design. *Choukoku* (*Oryza sativa* L. cv.) is a rice variety with a high level of Cd accumulation. We treated the seeds of *Choukoku* using the following steps: (1) disinfected with 0.5% sodium hypochlorite for 20 minutes, (2) washed several times with deionized water, and (3) soaked for 24 hours. The treated seeds were cultured in damp gauze at room temperature. When they germinated to about 1 cm, we transferred the seeds to a pot containing quartz sand and cultured them in deionized water. When the seedlings grew three to four leaves, we selected seedlings with uniform growth for the soil culture experiment. We planted two plants in each pot.

We set up the N fertilizer varieties and their application levels in rice pots in nine treatments, as shown in Table 2. Each treatment had eight replicates.

We prepared 72 pot devices two days before transplanting the rice seedlings. We filled 3.8 kg of soil in a plastic pot with a diameter of 20 cm and a height of 20 cm. At the same time, we weighed N and phosphorus fertilizers in proportion to the values shown in Table 2. We dissolved the fertilizer in 1000 mL of water and added it to the soil. To ensure the growth of seedlings, we added water to the water surface after transplanting, which was 2 cm higher than the soil interface. We kept the samples in a greenhouse and watered each pot every morning and evening.

Collection and treatment of rice plants. We transplanted the rice seedlings on May 21, 2012. We collected rice plant samples at the tillering stage for analysis on July 2, 2012. When the rice plants ripened, they were harvested after three days of roasting. The ripening time of rice in each treatment was different, according to the ripening conditions, and the specific sampling dates of the ripening period were as follows: on August 23, 2012, the rice plant samples treated with A, B, D, E, G, H, and I in Table 2 were collected; on August 28, 2012, the rice plant samples treated with F in Table 2 were collected; on September 3, 2012, the rice plant samples treated with C in Table 2 were collected. Photos of the maturation stage were taken on August 23, 2012. They were provided as supplementary figures.

After collecting rice samples, we cut down the samples along the rice roots. We took out the rice roots along with the surrounding soil and collected the roots and the soil around the rhizospheric rice. We rinsed the rice roots, stems and leaves, and the grains with tap water and then rinsed them with deionized water. They were green-killed at 105 °C for 30 minutes, and then dried at 60 °C until the weight remained unchanged. Finally, the

dried roots, stems and leaves, and grains were crushed and screened through a 60-mesh nylon sieve and then stored separately in polyethylene plastic bags. The rhizosphere soil samples were air-dried at room temperature and grinded with an agate mortar after removing animal and plant residues and other impurities. After screening through 10-mesh and 100-mesh nylon sieves, we stored the rhizosphere soil samples in separate polyethylene plastic bags.

Sample testing and data analysis. We measured the soil pH value in deionized water at a soil-to-solution ratio of 1:2.5 (w: v). We determined the heavy-metal and S concentrations in the soil using an inductively coupled plasma optical emission spectrometer (ICP-OES, ICP-5000, Focused Photonics, Inc., Zhejiang, China), following $\text{HNO}_3\text{-HClO}_4\text{-HF}$ digestion. The detection limits of the ICP-OES measurements were $0.5\ \mu\text{g}\cdot\text{L}^{-1}$ for Cd; $1\ \mu\text{g}\cdot\text{L}^{-1}$ for Cr; $0.5\ \mu\text{g}\cdot\text{L}^{-1}$ for Cu; $0.5\ \mu\text{g}\cdot\text{L}^{-1}$ for As; $0.5\ \mu\text{g}\cdot\text{L}^{-1}$ for Pb; $0.2\ \mu\text{g}\cdot\text{L}^{-1}$ for Zn; and $0.2\ \text{mg}\cdot\text{L}^{-1}$ for S. We determined the bioavailable concentrations of Cd in the soil using ICP-OES, following 1:2 suspensions of soil and $0.005\ \text{mol}\cdot\text{L}^{-1}$ DTPA – $0.01\ \text{mol}\cdot\text{L}^{-1}$ CaCl_2 – $0.1\ \text{mol}\cdot\text{L}^{-1}$ TEA mixed solution leaching at 180 rotations per minute (rpm) for 2 hours. We also determined the bioavailable concentrations of S in the soil using ICP-OES, following 1:5 suspensions of soil and $0.008\ \text{mol}\cdot\text{L}^{-1}$ $\text{Ca}(\text{H}_2\text{PO}_4)_2$ – $2\ \text{mol}\cdot\text{L}^{-1}$ CH_3COOH mixed solution leaching at 180 rpm for 1 hour. The organs samples of rice were digested by a $\text{HNO}_3\text{-HClO}_4$ (3:1, v: v) mixed acid.

In the sample pretreatment process, the acid reagents were of high purity (G.R.). In the process of sample detection, we added a standard solution to every 20 samples with an internal standard method quality control. The recoveries of Cd in soil and plant samples were higher than 95%, and the relative standard deviation was more than 10%.

All of the data were statistically analyzed using Microsoft Excel 2007 and DPS 7.05 software and were expressed as means \pm standard deviation. Treatment means were compared using least significant difference at $p < 0.05$, and the graphics were drawn using Origin 8.0.

Results and Discussion

Bioavailable Cd and pH of rice rhizosphere soil under different N fertilizer treatment conditions. A large number of studies have found that different forms of N fertilizer application can cause different changes in soil pH, thus affecting the bioavailability of heavy metals in rhizosphere soil^{22–24}.

The effects of different N fertilizer treatments on bioavailable Cd content in rice rhizosphere soil at tillering and maturing stages are shown in Fig. 1. The bioavailable Cd content in rice rhizosphere soil at the tillering and maturing stages under the three N fertilizer treatment conditions were as follows: $(\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{Cl} > \text{NH}_4\text{NO}_3$. This content indicated that the supply of S significantly increased the bioavailability of Cd in the soil. This also indirectly indicated that the effect of $\text{NH}_4^+\text{-N}$ on the bioavailability of Cd in the soil was more significant than that of $\text{NO}_3^-\text{-N}$. This result is consistent with Eriksson's²⁴ finding that $\text{NH}_4^+\text{-N}$ significantly increased Cd activity in soil. With an increase in N fertilizer application, the change trend of bioavailable Cd was different. At the tillering stage, when NH_4Cl and $(\text{NH}_4)_2\text{SO}_4$ were applied, the change trend of available Cd content in rice rhizosphere soil was the same as that of the N level, which increased with an increase in the N level; however, the increase of the $(\text{NH}_4)_2\text{SO}_4$ treatment was more obvious. The content of bioavailable Cd under $(\text{NH}_4)_2\text{SO}_4$ treatments increased the most rapidly. With an increase of N application, the content of bioavailable Cd increased from $0.379\ \text{mg}\cdot\text{kg}^{-1}$ to $0.460\ \text{mg}\cdot\text{kg}^{-1}$, increasing by 21.37%. The increase in the application NH_4Cl fertilizer did not significantly increase Cd bioavailability, increasing only slightly from $0.376\ \text{mg}\cdot\text{kg}^{-1}$ to $0.383\ \text{mg}\cdot\text{kg}^{-1}$. The NH_4NO_3 application had the smallest effect on the increase of bioavailable Cd content in rice rhizosphere soil. The bioavailable Cd content in rice rhizosphere soil under $0.1\ \text{g}\cdot\text{kg}^{-1}$, $0.2\ \text{g}\cdot\text{kg}^{-1}$, and $0.4\ \text{g}\cdot\text{kg}^{-1}$ of NH_4NO_3 treatment conditions was only 82.41%, 62.35%, and 71.15% under $(\text{NH}_4)_2\text{SO}_4$ application, respectively. The content of bioavailable Cd first decreased and then increased with an increase in the N level. At the maturity stage, the bioavailable Cd content in the rhizosphere soil of rice decreased with an increase in the N application, from $0.411\ \text{mg}\cdot\text{kg}^{-1}$ and $0.182\ \text{mg}\cdot\text{kg}^{-1}$ to $0.274\ \text{mg}\cdot\text{kg}^{-1}$ and $0.137\ \text{mg}\cdot\text{kg}^{-1}$, and decreased by 33.33% and 24.72%, respectively. Under NH_4Cl fertilizer treatments, the bioavailable Cd content in rhizosphere soil first decreased and then increased, and it reached the lowest level at the treatment level of $0.2\ \text{g}\cdot\text{kg}^{-1}$. Compared with bioavailable Cd content at the tillering stage and maturity stage, the effect of $\text{NH}_4^+\text{-N}$ application to improve soil Cd bioavailability was more significant than that of $\text{NO}_3^-\text{-N}$, and the supply of S significantly increased the soil Cd bioavailability.

The effect of different N fertilizer treatments on the pH of rice rhizosphere soil is shown in Fig. 2. The pH of rice rhizosphere soil was different under conditions of the same growth period, the same N application rate, and different N fertilizer varieties. The results showed that the pH of rice rhizosphere soil under $(\text{NH}_4)_2\text{SO}_4$ and NH_4Cl treatments was lower than under NH_4NO_3 treatments. When plants absorb $\text{NH}_4^+\text{-N}$ and $\text{NO}_3^-\text{-N}$, their roots secrete different ions. When they absorb $\text{NH}_4^+\text{-N}$, it causes H^+ secretion, which results in acidification of the rice rhizosphere soil. Conversely, when they absorb $\text{NO}_3^-\text{-N}$, they secrete OH^- , which results in alkalization of the rhizosphere soil. In this experiment, these three N fertilizers introduced the same cations into the soil, and with the same N level, NH_4^+ contained in $(\text{NH}_4)_2\text{SO}_4$ and NH_4Cl was greater than NH_4NO_3 . Therefore, the acidification of the first two N fertilizers was greater than that of the latter, and the bioavailable Cd content in the corresponding soil was higher (as shown in Fig. 1).

In addition, previous studies have shown that the anions introduced by the three N fertilizers also have some influence on the adsorption of heavy metals in soil, because Cl^- can form relatively stable complexes with Cd in solution (such as CdCl^+ , CdCl_2^0 , CdCl_3^- , CdCl_4^{2-} , and so on), which causes Cd to migrate from a solid to the soil solution and to improve the solubility of Cd²⁵.

Consensus has not been reached on the effect of heavy-metal uptake by plants under S^{11–15}. Our previous studies have shown, however, that when rice is planted in polluted soils of polymetallic sulfide mining areas, both total S and bioavailable S were enriched in rice rhizosphere soil, and the S content in the rhizosphere soil was

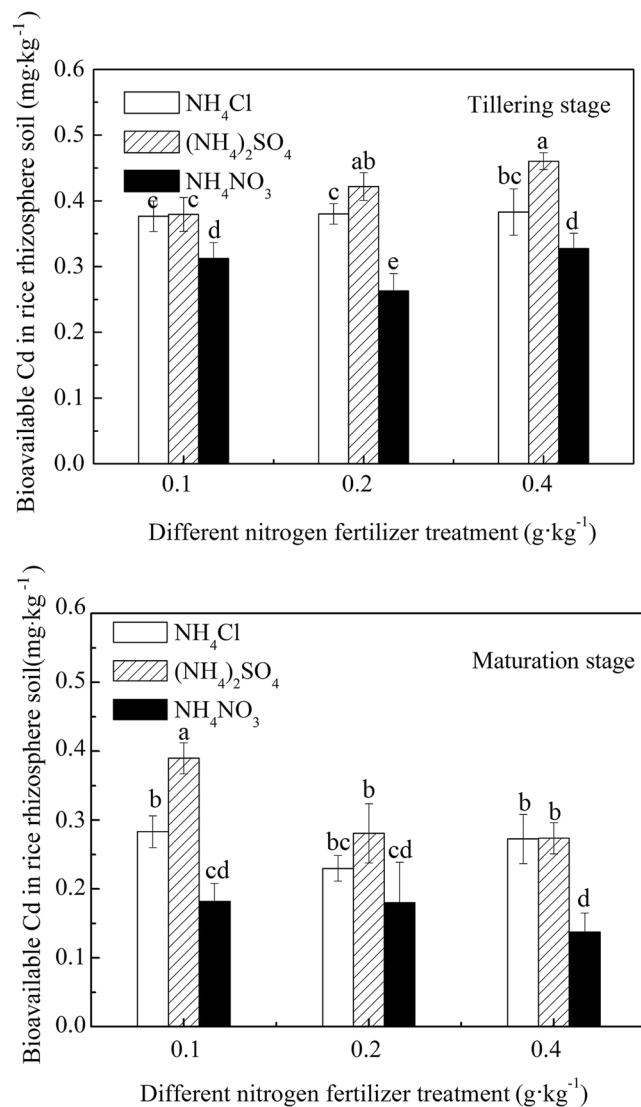


Figure 1. Concentration of bioavailable Cd in the rhizosphere soil under different N fertilizer treatment conditions.

positively correlated with Cd accumulation in rice²⁶. This result indicated that an increase in the SO_4^{2-} content also may increase Cd bioavailability. In addition, SO_4^{2-} also competed for the adsorption sites of Cd ions on the soil surface, which increased the release of Cd in soil solution²⁷. The effects of different N fertilizers on soil pH and Cd bioavailability under flooding conditions have been studied by Jiaka *et al.*²⁸. The results showed that the ability order of N fertilizer to reduce soil pH was $\text{NH}_4\text{Cl} > (\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{NO}_3$, and the ability order to increase soil Cd bioavailability under N fertilizer was $\text{NH}_4\text{Cl} > (\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{NO}_3$ ²⁸. The results of this experiment showed that the ability to increase soil Cd bioavailability was $(\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{Cl}$. These differences may have resulted from the different physicochemical properties of the paddy soil planted in the two experiments. The former soil used was from a rice paddy soil developed from gray alluvial deposits in Sichuan, and its pH was 6.57, which was a neutral paddy soil. The soil used in this experiment was paddy soil that had been irrigated by sewage from the polymetallic sulfide mining area for many years. The acid mine drainage not only caused the decreasing soil pH (pH = 4.21), but also caused the increasing S content in the soil. The total S content in the soil was 443.66 $\text{mg}\cdot\text{kg}^{-1}$, which was 1.58 times higher than the average S content in Guangdong Province. The bioavailable S content was 133.38 $\text{mg}\cdot\text{kg}^{-1}$, which was 6.2 times higher than the average bioavailable S content in Guangdong Province, with 30% of the total S content, which was higher than 10% of the average S content in natural soil. This finding indicated that the soil contained a large number of highly mobile sulfate ions. Our previous studies showed that both total and bioavailable S tended to be enriched in rice rhizosphere soil, which also resulted in the existence of Cd in rhizosphere soil in the form of higher bioavailability²⁶.

Cd concentration in rice roots under different N fertilizer treatment conditions. Figure 3 shows that the Cd concentration in rice roots was different during the same growth periods under the same amount of different N sources. The Cd concentration in rice roots was greatest under $(\text{NH}_4)_2\text{SO}_4$ application, followed

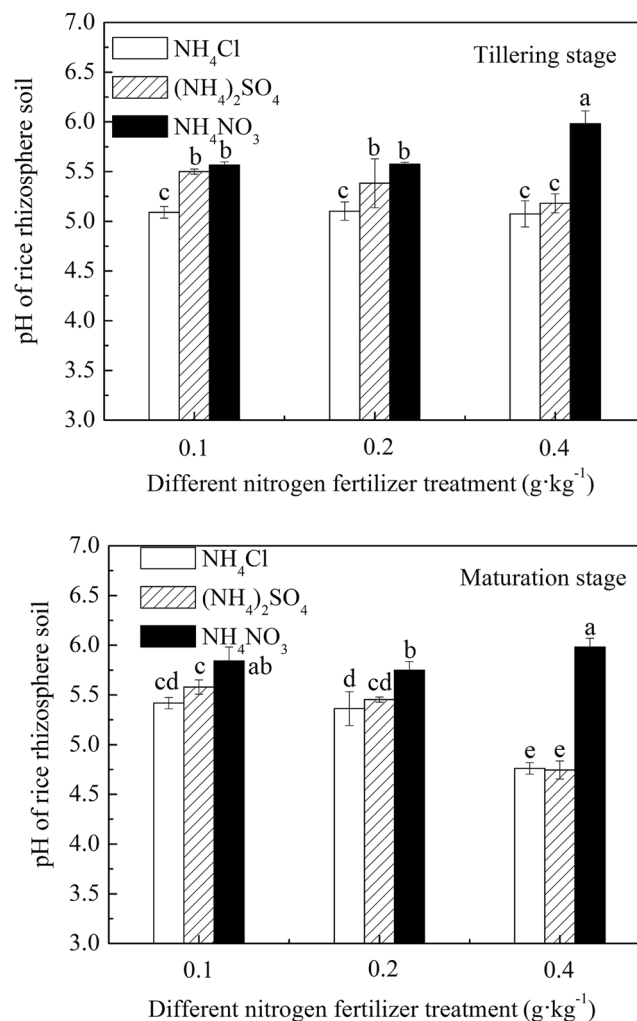


Figure 2. pH of rice rhizosphere soil under different N fertilizer treatment conditions.

by NH_4NO_3 , and NH_4Cl was the lowest. The effects of the three N fertilizer treatments on Cd concentration in rice roots were as follows: $(\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{NO}_3 > \text{NH}_4\text{Cl}$. The effects of Cd concentration in roots at different growth stages of rice under different N fertilizer treatments were different. At the tillering stage, the Cd concentration in the rice roots increased along with the increasing level of NH_4Cl , which increased from $1.19 \text{ mg}\cdot\text{kg}^{-1}$ to $1.51 \text{ mg}\cdot\text{kg}^{-1}$. The Cd content in the rice roots did not differ much under the same N level of $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 , but it decreased with an increase in the N level. This result indicated that increasing the amount of these two N fertilizers would inhibit the absorption of Cd in rice roots during the early growth period. Except for the treatment of the $0.1 \text{ g}\cdot\text{kg}^{-1}$ N application level under $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 treatments, the Cd content in rice roots increased more significantly at the ripening stage than at the tillering stage. This result indicated that the Cd concentration in the rice roots increased with rice growing, but the effects of different N fertilizer application on Cd concentration in the rice roots were different. With an increase in the NH_4Cl application treatments, the Cd content in the rice roots remained basically unchanged. With an increase in the $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 treatment conditions, the Cd content in the rice roots increased significantly. These results showed that Cd absorption in the rice roots increased under the $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 N fertilizer treatment at the later growing stage.

A comparison of Fig. 1 with Fig. 3 shows that the bioavailable Cd content in the paddy soils treated with the three N fertilizers was $(\text{NH}_4)_2\text{SO}_4$ treatment $>$ NH_4Cl treatment $>$ NH_4NO_3 treatment, whereas the Cd content in the rice roots was $(\text{NH}_4)_2\text{SO}_4 > \text{NH}_4\text{NO}_3 > \text{NH}_4\text{Cl}$. This comparison also shows that different forms of N fertilizer have different effects on the Cd content in paddy soils and rice roots. Xie *et al.*²² studied the effects of different N forms on the uptake and accumulation of Cd and zinc by hyperaccumulating plant *Cymbidium nigrum*. According to their results, NH_4^+ -N treatment decreased the pH of the rhizosphere soil more than NO_3^- -N treatment, whereas NH_4^+ -N treatment promoted the accumulation of Cd in *Cymbidium nigrum* more than NO_3^- -N treatment. In addition, Hassan *et al.*²⁹ studied the effects of N forms on Cd toxicity in rice. Their results showed that the Cd content in rice roots and stalks was $\text{Ca}(\text{NO}_3)_2$ treatment $>$ NH_4NO_3 treatment $>$ $(\text{NH}_4)_2\text{SO}_4$ treatment, which aligned with the conclusion that the Cd content in plant roots and stems was NO_3^- -N treatment $>$ NH_4^+ -N treatment. The comparative results of NH_4Cl and NH_4NO_3 treatments in this experiment were

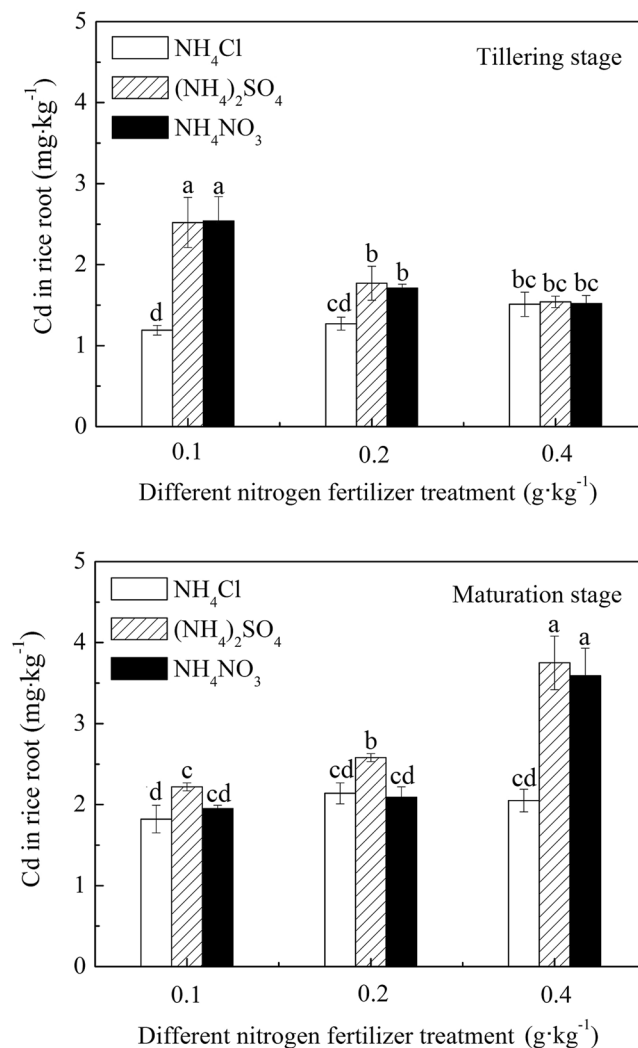


Figure 3. Cd concentration in rice roots under different N fertilizer treatment conditions.

the same as those in the noted studies. NH_4^+ and H^+ competed with Cd ions for the root adsorption sites. In contrast, NH_4^+ -N treatment could make the root secrete more malic acid chelating with Cd than with NO_3^- -N treatment^{29,30}.

In this study, the bioavailable Cd content in the rice rhizosphere soil and in rice roots was the highest in the treatment group of $(\text{NH}_4)_2\text{SO}_4$. The main reason for this difference was that the experimental soil was paddy soil that had been polluted by acid mine wastewater and contained excessive SO_4^{2-} , which promoted the absorption of Cd by the rice roots^{26,31}.

Cd concentration in rice stems and leaves under different N fertilizer treatment conditions.

Figure 4 shows the effects of different levels of N fertilizer treatments on Cd concentration in rice stems and leaves at the tillering and maturity stages. The Cd concentration in rice stems and leaves under the same growth period and N application rate was as follows: $\text{NH}_4\text{NO}_3 > \text{NH}_4\text{Cl}$. The results were the same as that for the rice roots. When the N level gradually increased, the application of NH_4NO_3 fertilizer was more conducive to the concentration and accumulation of Cd by rice stems and leaves. This result was the same as that found by Jönsson *et al.*³². The application of NH_4NO_3 was more beneficial to Cd accumulation in plant stems than that of NH_4Cl . The benefit may be related to the transport of NO_3^- in rice stems and leaves. Other studies have shown that the transport of NO_3^- to plant stems requires the participation of cations to achieve a charge balance, which will promote the transfer of Cd to the stem³². Plants usually absorb NH_4^+ -N and cause H^+ secretion, which results in soil acidification around the plant rhizosphere. When plants absorb NO_3^- -N and cause OH^- secretion, it results in rhizosphere soil alkalization. Therefore, it is generally believed that NO_3^- -N is a more suitable application than NH_4^+ -N in acid soil. This study, however, showed that in an acid-heavy metal-contaminated soil remediation project, fertilization could not be selected only by adjusting the soil pH value. The selection of N fertilizer should be considered on the basis of the effect of different forms of N fertilizer on the migration of heavy metals to plants.

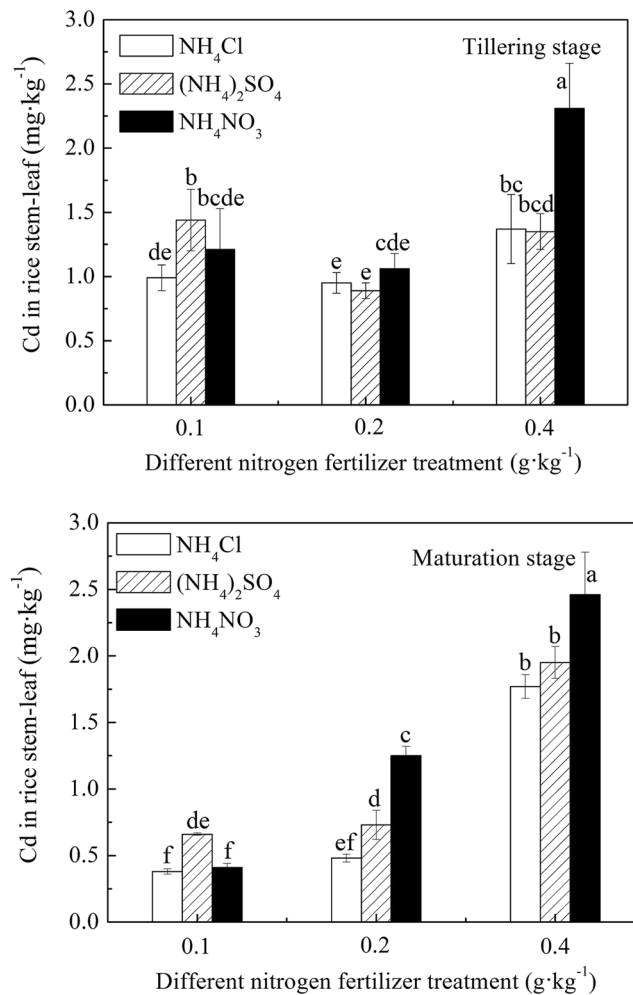


Figure 4. Cd concentration in rice stems and leaves under different N fertilizer treatment conditions.

Comparing Figs. 3 and 4, the Cd concentration in rice stems and leaves was different from that in the rice root treated with (NH₄)₂SO₄. With the 0.1 g·kg⁻¹ N treatment level, the Cd concentration in the rice stems and leaves remained as follows: (NH₄)₂SO₄ treatment > NH₄NO₃ treatment > NH₄Cl treatment, which followed the same trend as that in the rice root. The Cd content in the rice stems and leaves following the other two treatments was lower than that with the NH₄NO₃ treatments. This result suggested that application of (NH₄)₂SO₄ reduced Cd transport from the rice root to the stems and leaves. Gao studied the effect of SO₄²⁻ on Cd accumulation in rice seedlings³¹. She noted that SO₄²⁻ enhanced the retention ability of Cd in the rice root tissues and reduced the transfer of Cd to the shoots. When the content of SO₄²⁻ increased to a certain extent and exceeded the retention ability of Cd in roots, the Cd began to transfer to the shoots. This indicated that a certain mutual restriction existed between SO₄²⁻ and N applications on Cd absorption. The soil in this experiment was polluted by acid mine wastewater, and its S content, especially bioavailable S, was high, which may have hindered the transfer of Cd to shoots.

Figure 4 shows that the Cd concentration in rice stems and leaves increased with an increase in the N level at the maturity stage under the same N fertilizer variety treatment. This finding indicated that the application of N fertilizer could promote the Cd concentration in rice stems and leaves at the later stage of rice growing. Under the same growth period and N treatment level (except 0.1 g·kg⁻¹), the Cd concentration in rice stems and leaves was the highest with NH₄NO₃ treatments. Figure 4 also shows that the Cd concentration in rice stems and leaves at the tillering stage was more serious than that at the ripening stage. The content of bioavailable Cd in the soil during this period may have been higher; conversely, rice grows vigorously during the vegetative growth stage and passively absorbs more Cd.

Cd concentration in brown rice under different N fertilizer treatment conditions. The Cd concentration of brown rice under different N fertilizer treatment conditions is shown in Fig. 5. Figure 5 shows that under the N application conditions selected in this experiment, the Cd concentration in brown rice exceeded the 0.2 mg·kg⁻¹ standard (GB 2762-2017)³³, but the exceeding standard was different for the different N fertilizer varieties and N levels. Other than the 0.4 g·kg⁻¹ N application level of the NH₄NO₃ treatment, the Cd content in brown rice increased gradually with an increase in the N application level. The Cd concentration in brown rice treated with NH₄Cl increased from 0.202 mg·kg⁻¹ to 0.390 mg·kg⁻¹, but the highest Cd content in brown rice treated with NH₄NO₃ reached 0.431 mg·kg⁻¹ at the 0.2 g·kg⁻¹ N application level, and then it decreased to 0.390 mg·kg⁻¹. Therefore, the effects of NH₄NO₃ and NH₄Cl treatments on the Cd concentration in rice seed did

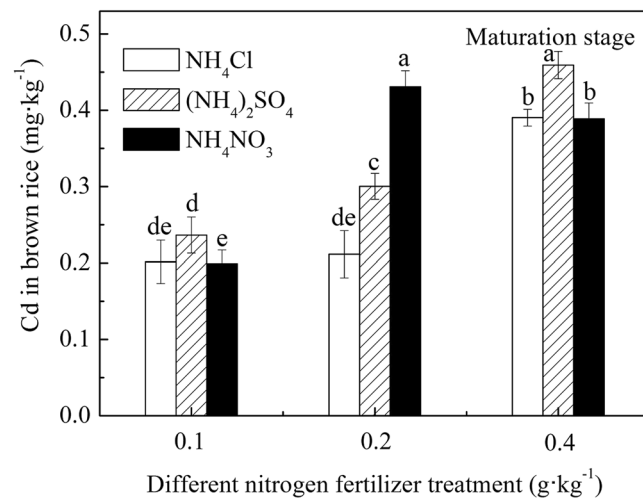


Figure 5. Cd concentration in brown rice under different N fertilizer treatment conditions.

Fertilizer	N (g N·kg ⁻¹ soil)		
	0.1 g·kg ⁻¹	0.2 g·kg ⁻¹	0.4 g·kg ⁻¹
NH ₄ Cl	109 d-A	109 d-B	(120–131) d-C
(NH ₄) ₂ SO ₄	109 d-D	109 d-E	(114–123) d-F
NH ₄ NO ₃	109 d-G	109 d-H	109 d-I

Table 3. Days of rice growth period.

not show the same regularity as that of rice straw. At the 0.4 g·kg⁻¹ N application level, the amount of N fertilizer was high, which resulted in the late maturity of rice treated with NH₄Cl and (NH₄)₂SO₄ (more than five days later than the 0.1 g·kg⁻¹ N application level and the 0.2 g·kg⁻¹ N application level). In general, the Cd concentration in brown rice was the lowest when treated with NH₄Cl.

The Cd content in brown rice treated with NH₄Cl and (NH₄)₂SO₄ increased gradually with an increase in the N application level. The Cd concentration of brown rice treated with NH₄Cl increased from 0.202 mg·kg⁻¹ to 0.390 mg·kg⁻¹, and the Cd concentration in brown rice treated with (NH₄)₂SO₄ increased from 0.236 mg·kg⁻¹ to 0.459 mg·kg⁻¹. The Cd concentration of brown rice was (NH₄)₂SO₄ treatments > NH₄Cl treatments, which was the same as that of rice roots and straws at the ripening stage. Compared with NH₄NO₃ treatments, the Cd concentration change trend in brown rice at the 0.1 g·kg⁻¹ N application level and the 0.2 g·kg⁻¹ N application level was the same as that in rice straw. At the 0.4 g·kg⁻¹ N application level, the Cd content in brown rice was the highest when treated with (NH₄)₂SO₄. This result may be due to the excessive SO₄²⁻ content, which exceeded the ability of the root to retain Cd and promoted the transfer of Cd to the shoot³¹.

No consensus has been reached on the effect of N fertilizer applications on Cd accumulation in rice. Eriksson²⁴ showed that the order of Cd uptake by plants was (NH₄)₂SO₄ > NH₄NO₃, Zeng *et al.*³⁴ showed that Cd uptake by rice seedlings treated with the three N fertilizers was NH₄NO₃ > NH₄Cl > (NH₄)₂SO₄. Jiaka *et al.*²⁸ showed that different forms of N fertilizer promoted Cd uptake and that the order of Cd content in rice straw was (NH₄)₂SO₄ > NH₄Cl > NH₄NO₃, whereas Cd content in rice grain was NH₄Cl > NH₄NO₃ > (NH₄)₂SO₄. Notably, S, as a necessary nutrient for plants as well as an important component in soil, has some restrictive effects with NH₄⁺-N and NO₃⁻-N. In paddy soils with high S content, the application of N fertilizer should consider the effect of SO₄²⁻, and NH₄Cl fertilizer seems more suitable for high-S-content soil.

As for the effect of N sources on rice growth, the photos taken of the first mature samples (collected on August 23) show that the biomass of rice treated with NH₄Cl and (NH₄)₂SO₄ was higher than that treated with NH₄NO₃. The results of Jiaka *et al.*²⁸ and Jalloh *et al.*³⁵ showed that NH₄⁺-N treatments had significantly higher grain yields, which was consistent with our results. The picture taken on August 23 and the information in Table 3 show that under the 0.4 g·kg⁻¹ N application level, the excessive N source enabled the rice treated with NH₄Cl and (NH₄)₂SO₄ to have extended growth and delayed maturity (about 5–21 days later than the 0.1 g·kg⁻¹ and 0.2 g·kg⁻¹ N application levels). Because N is the main component of chlorophyll, the greener the plant, the higher the N concentration it has³⁶. The green color of the rice was NH₄Cl treatment > (NH₄)₂SO₄ treatment > NH₄NO₃ treatment. Therefore, the N concentration in rice was NH₄Cl treatment > (NH₄)₂SO₄ treatment > NH₄NO₃ treatment, but the Cd concentration in rice was (NH₄)₂SO₄ treatment > NH₄Cl treatment ≈ NH₄NO₃ treatment. Jalloh *et al.*³⁵ also showed that NH₄⁺-N treatments had significant higher N accumulation in plant tissues than NO₃⁻-N treatments. The research results from Yang³⁷, however, showed that rice plantlets treated with NH₄⁺ had a higher growth ability with higher tillers, dry weight, photosynthetic rate, transpiration rate, and total N content by 49.1%, 9.6%, 106.5%, 9.6%, and 48.3%, respectively, when compared with NO₃⁻-treated plantlets under Cd stress. The NO₃⁻-treated rice plantlet had the strongest Cd absorption, which was 2.79 times higher than NH₄⁺ treated

in rice roots. Figures 4 and 5 show that rice treated with $0.2 \text{ g}\cdot\text{kg}^{-1}$ and $0.4 \text{ g}\cdot\text{kg}^{-1}$ NO_3^- -N had a higher Cd concentration but was less green in color, and thus its N concentration was lower. Therefore, the N concentration in aboveground rice was high, and the Cd concentration was correspondingly high under the same form of N source. For rice treated with different forms of the same N level, this phenomenon was not obvious. Figures 4 and 5 show that the Cd concentration increased significantly from $0.2 \text{ gN}\cdot\text{kg}^{-1}$ to $0.4 \text{ gN}\cdot\text{kg}^{-1}$. The probable reason for this increase was that excessive NH_4^+ enhanced Cd translocation by inhibiting root heavy-metal adenosine-triphosphatase gene expression in rice and that excessive NO_3^- enhanced Cd uptake by up-regulating the expression of iron-regulated metal transporter in rice^{37,38}.

In addition, generally the bioavailable Cd content in rice rhizosphere soil is high, which indicates that the Cd content can be absorbed by rice also is high. This study, however, showed that although the bioavailable Cd content in soil was positively correlated with the Cd absorption by rice roots, it was not strongly correlated with the Cd content in rice straw and rice seed. Therefore, it remains questionable which chemical extraction method should be used to characterize the bioavailability of Cd in soil.

Rice Cd standards are different in different countries. The current World Health Organization Cd standard is $0.4 \text{ mg}\cdot\text{kg}^{-1}$, and this standard currently is used in Japan and Taiwan. The European Union and China have a stricter standard of $0.2 \text{ mg}\cdot\text{kg}^{-1}$. Figure 5 shows that all of the Cd concentrations in the brown rice treated with NH_4Cl did not exceed $0.4 \text{ mg}\cdot\text{kg}^{-1}$. Therefore, the Cd concentration in rice can be kept at a low level by applying NH_4Cl source in high-S-content paddy soil. For field production where the experimental soil samples were collected, the Cd concentration of brown rice was between 2.35 and $3.50 \text{ mg}\cdot\text{kg}^{-1}$ under farmers' habitual way of applying fertilizer. Considering that the results of pot incubation and field production are different, the Cd concentration for aboveground pot incubation usually is lower than the concentration for field production^{26,39}. Therefore, future work is required to optimize the NH_4Cl supply concentration under field conditions to obtain rice grains with Cd concentration below the food safety standards of 0.2 or $0.4 \text{ mg}\cdot\text{kg}^{-1}$. If necessary, agricultural measures, such as low-Cd-uptake rice varieties and water management²⁶, can be used.

Conclusions

We conducted pot experiments to study the effects of different N fertilizer varieties and levels on bioavailable Cd in rice rhizosphere soil and the Cd concentration in different organs of rice at the tillering and maturing stages. The effect of NH_4^+ -N on improving the bioavailability of Cd in soil was more significant than that of NO_3^- -N; and the supply of S significantly increased the bioavailability of Cd in soil. The bioavailable Cd content in rice rhizosphere soil and in rice roots was the highest when treated with $(\text{NH}_4)_2\text{SO}_4$. However, S and N applications had some mutual restrictions on the Cd concentration. At a certain level of N, application of $(\text{NH}_4)_2\text{SO}_4$ reduced Cd transport from rice roots to shoots. When it exceeded that level, application of $(\text{NH}_4)_2\text{SO}_4$ promoted the transfer of Cd from rice roots to shoots. The Cd concentration with NH_4Cl treatment was relatively lower at different growth stages than with $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 treatment. Thus, this study suggests that NH_4Cl would be a good N source to reduce Cd phytoextraction for paddy soil with a high level of S content and likely would obtain rice grains with Cd concentration below the food safety standards of 0.2 or $0.4 \text{ mg}\cdot\text{kg}^{-1}$.

Received: 2 July 2019; Accepted: 29 January 2020;

Published online: 12 February 2020

References

- Allan, R. J. Impact of mining activities on the terrestrial and aquatic environment with emphasis on mitigation and remedial measures in *Heavy Metals* (ed. Förstner, U. *et al.*) 119–140 (Springer-Verlag Berlin Heidelberg, 1995).
- Naidu, G. *et al.* A critical review on remediation, reuse, and resource recovery from acid mine drainage. *Environ. Pollut.* **247**, 1110–1124 (2019).
- Zhou, J. M., Dang, Z., Cai, M. F. & Liu, C. Q. Soil heavy metal pollution around the Dabaoshan mine, Guangdong province, China. *Pedosphere* **17**, 588–594 (2007).
- Williams, P. N. *et al.* Occurrence and partitioning of Cd, arsenic and lead in mine impacted paddy rice: Hunan, China. *Environ. Sci. Technol.* **43**, 637–642 (2009).
- Lei, M. *et al.* Heavy metal pollution and potential health risk assessment of white rice around mine areas in Hunan Province, China. *Food Secur.* **7**, 45–54 (2015).
- Wang, Z., Qin, H. & Liu, X. Health risk assessment of heavy metals in the soil-water-rice system around the Xiaozhuang uranium mine, China. *Environ. Sci. Pollut. R.* **26**, 5904–5912 (2019).
- Han, Z. *et al.* Pollution Assessment of Heavy Metals in Soils and Plants around a Molybdenum Mine in Central China. *Pol. J. Environ. Stud.* **28**, 123–133 (2019).
- Van Breemen, N. Dissolved Aluminum in Acid Sulfate Soils and in Acid Mine Waters 1. *Soil Sci. Soc. Am. J.* **37**, 694–697 (1973).
- Sokolova, T. A. & Alekseeva, S. A. Adsorption of sulfate ions by soils (a review). *Eurasian Soil Sci.* **41**, 140–148 (2008).
- Reuss, J. O. & Johnson, D. W. Acid deposition and the acidification of soils and waters in *Ecological Studies* (Vol. 59) (ed. Billings, W. D. *et al.*) 1–8 (Springer Science & Business Media, 2012).
- Khan, M. I. R. *et al.* Modulation and significance of nitrogen and sulfur metabolism in Cd challenged plants. *Plant Growth Regul.* **78**, 1–11 (2016).
- Chen, S., Sun, L., Sun, T., Chao, L. & Guo, G. Interaction between Cd, lead and potassium fertilizer (K_2SO_4) in a soil-plant system. *Environ. Geochem. Hlth.* **29**, 435–446 (2007).
- Fan, J. L., Hu, Z. Y., Ziadi, N. & Xia, X. Excessive sulfur supply reduces Cd accumulation in brown rice (*Oryza sativa* L.). *Environ. Pollut.* **158**, 409–415 (2010).
- Nocito, F. F. *et al.* Heavy metal stress and sulfate uptake in maize roots. *Plant Physiol.* **141**, 1138–1148 (2006).
- Feng, Q., Tai, P., Li, P., Guo, Y. & Fu, S. Role of sulfur in Cd accumulation of *Tagetes erecta* L. *J. Plant Nutr.* **32**, 919–928 (2009).
- Fazili, I. S. *et al.* Interactive effect of sulfur and nitrogen on nitrogen accumulation and harvest in oilseed crops differing in nitrogen assimilation potential. *J. Plant Nutr.* **31**, 1203–1220 (2008).
- Khan, M. I. R., Nazir, F., Asgher, M., Per, T. S. & Khan, N. A. Selenium and sulfur influence ethylene formation and alleviate Cd-induced oxidative stress by improving proline and glutathione production in wheat. *J. Plant Physiol.* **173**, 9–18 (2015).
- Anjum, N. A. *et al.* Metal/metalloid stress tolerance in plants: role of ascorbate, its redox couple, and associated enzymes. *Protoplasma* **251**, 1265–1283 (2014).

19. Anjum, N. A. *et al.* Glutathione and proline can coordinately make plants withstand the joint attack of metal (loid) and salinity stresses. *Front. Plant Sci.* **5**, 662, <https://doi.org/10.3389/fpls.2014.00662> (2014).
20. Gill, S. S., Khan, N. A. & Tuteja, N. Cd at high dose perturbs growth, photosynthesis and nitrogen metabolism while at low dose it up regulates sulfur assimilation and antioxidant machinery in garden cress (*Lepidium sativum* L.). *Plant Sci.* **182**, 112–120 (2012).
21. Itou, M. The find of high Cd accumulator rice “Choukokuoku” (*Oryza Sativa* L.) and the influence of phytoremediation using *Choukokuoku* on soil contamination countermeasure (JSSPN Technological Advancement Progress Awards). *J. Sci. Soil Manure Jpn.* **84**, 359–360 (2013).
22. Xie, H. L., Jiang, R. F., Zhang, F. S., McGrath, S. P. & Zhao, F. J. Effect of nitrogen form on the rhizosphere dynamics and uptake of Cd and zinc by the hyperaccumulator *Thlaspi caerulescens*. *Plant Soil* **318**, 205–215 (2009).
23. Willaert, G. Effects of various nitrogen fertilizers on the chemical and biological activity of major and trace elements in a Cd contaminated soil. *Pedologie* **43**, 83–91 (1992).
24. Eriksson, J. E. Effects of nitrogen-containing fertilizers on solubility and plant uptake of Cd. *Water Air Soil Poll.* **49**, 355–368 (1990).
25. Norvell, W. A., Wu, J., Hopkins, D. G. & Welch, R. M. Association of Cd in durum wheat grain with soil chloride and chelate-extractable soil Cd. *Soil Sci. Soc. Am. J.* **64**, 2162–2168 (2000).
26. Zhang, X. X. *et al.* Accumulation of S, Fe and Cd in rhizosphere of rice and their uptake in rice with different water managements. *Environ. Sci.* **34**, 2837–2846 (2013).
27. Tsadilas, C. D., Karaivazoglou, N. A., Tsotsolis, N. C., Stamatiadis, S. & Samaras, V. Cd uptake by tobacco as affected by liming, N form, and year of cultivation. *Environ. Pollut.* **134**, 239–246 (2005).
28. Jiaka, L. *et al.* Effect of nitrogen fertilizer type and application rate on Cd uptake and grain yield of paddy rice. *Chin. J. Eco-Agr.* **18**, 281–285 (2010).
29. Hassan, M. J., Shafi, M., Zhang, G., Zhu, Z. & Qaisar, M. The growth and some physiological responses of rice to Cd toxicity as affected by nitrogen form. *Plant Growth Regul.* **54**, 125–132 (2008).
30. Wu, L. H., Luo, Y. M., Christie, P. & Wong, M. H. Effects of EDTA and low molecular weight organic acids on soil solution properties of a heavy metal polluted soil. *Chemosphere* **50**, 819–822 (2003).
31. Gao, M. X. Effects of sulfur supply on Cd uptake and accumulation in rice seedlings. PhD thesis, Northwest A & F University, Yangling, China (2009).
32. Jönsson, E. L. & Asp, H. Effects of pH and nitrogen on Cd uptake in potato. *Biol. Plantarum* **57**, 788–792 (2013).
33. National Health and Family Planning Commission of the People’s Republic of China & China Food and Drug Administration. National food safety standard. Maximum levels of contaminants in foods GB2762–2017 (2017).
34. Zeng, Q. R., Zhou, X. H. & Mao, X. Y. Effects of Different Nitrogen Fertilizers on the Dissolution of Heavy Metals and the Absorption of Rice Seedlings in the Contaminated Soil of Lead-zinc Mine Tailings. *Soil Fertilizer Sci. China* **3**, 7–11 (1997).
35. Jalloh, M. A., Chen, J., Zhen, F. & Zhang, G. Effect of different N fertilizer forms on antioxidant capacity and grain yield of rice growing under Cd stress. *J. Hazard. Mater.* **162**, 1081–1085 (2009).
36. Myers, J. H. Effect of physiological condition of the host plant on the ovipositional choice of the cabbage white butterfly, *Pieris rapae*. *J. Anim. Ecol.* **54**, 193–204 (1985).
37. Yang, Y. J. Effects of Nitrogen Fertilizer Form and Dosage on Cadmium Accumulation and Toxicity and Its Regulatory Mechanism in Rice. PhD thesis, Chinese Academy of Agricultural Sciences, Beijing, China (2016).
38. Yang, Y. *et al.* Excessive nitrate enhances cadmium (Cd) uptake by up-regulating the expression of OsIRT1 in rice (*Oryza sativa*). *Environ. Exp. Bot.* **122**, 141–149 (2016).
39. Zhu, S. S. Effect of rice rhizosphere soil heavy metal speciation and activity of the enzyme. Master thesis, Central South University of Forestry Science & Technology, Changsha, China (2013).

Acknowledgements

This project was supported by the Young Scientists Fund of the National Natural Science Foundation of China (41101212) and GDAS’ Project of Science and Technology Development (2019GDASYL-0105039 & 2019GDASYL-0401003).

Author contributions

Xiaoxia Zhang designed the study, prepared the material, conducted the experiments, analyzed the results, and wrote the paper; Xuexia Zhang designed the study, and revised the paper; Shuji Lv analyzed the results and revised the paper; Lei Shi provided the idea for rice planting and collection; Rongping Wang revised the paper.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information is available for this paper at <https://doi.org/10.1038/s41598-020-59409-1>.

Correspondence and requests for materials should be addressed to X.Z. or R.W.

Reprints and permissions information is available at www.nature.com/reprints.

Publisher’s note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article’s Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2020