



# Deciphering the conformational landscape of few selected aromatic noncoded amino acids (NCAAs) for applications in rational design of peptide therapeutics

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## Abstract

Amino acids are the essential building blocks of both synthetic and natural peptides, which are crucial for biological functions and also important as biological probes for mapping the complex protein–protein interactions (PPIs) in both prokaryotic and eukaryotic systems. Mapping the PPIs through the chemical biology approach provides pharmacologically relevant peptides, which can have agonistic or antagonistic effects on the targeted biological systems. It is evidenced that  $\geq 60$  peptide-based drugs have been approved by the US-FDA so far, and the number will improve further in the foreseeable future, as  $\geq 140$  peptides are currently in clinical trials. However, natural peptides often require fine-tuning of their pharmacological properties by strategically replacing the  $\alpha_L$ -amino acids of the peptides with non-coded amino acids (NCAA), for which codons are absent in the genetic code for biosynthesis of proteins, prior to their applications as therapeutics. Considering the diverse repertoire of the NCAs, the conformational space of many NCAs is yet to be explored systematically in the context of the rational design of therapeutic peptides. The current study deciphers the conformational landscape of a few such C $\alpha$ -substituted aromatic NCAs (Ing: 2-indanyl-L-Glycine; Bpa: 4-benzoyl-L-phenylalanine; Aic: 2-aminoindane-2-carboxylic acid) both in the context of tripeptides and model synthetic peptide sequences, using alanine (Ala) and proline (Pro) as the reference. The combined data obtained from the computational and biophysical studies indicate the general success of this approach, which can be exploited further to rationally design optimized peptide sequences of unusual architecture with potent antimicrobial, antiviral, gluco-regulatory, immunomodulatory, and anti-inflammatory activities.

**Keywords** Non-coded amino acids · Peptide design · Peptide therapeutics · Circular dichroism · Fluorescence

## Introduction

Amino acids are the primary building blocks of all proteins involved in various biological processes in living organisms. Twenty-three  $\alpha$ -amino acids are commonly found in proteins, from which twenty  $\alpha$ -amino acids are called proteinogenic, for which codons are available in the genetic code for biosynthesis of proteins. The remaining three are post-translationally modified  $\alpha$ -amino acids, sometimes found only in matured proteins. Interestingly,  $\geq 140$  amino acids

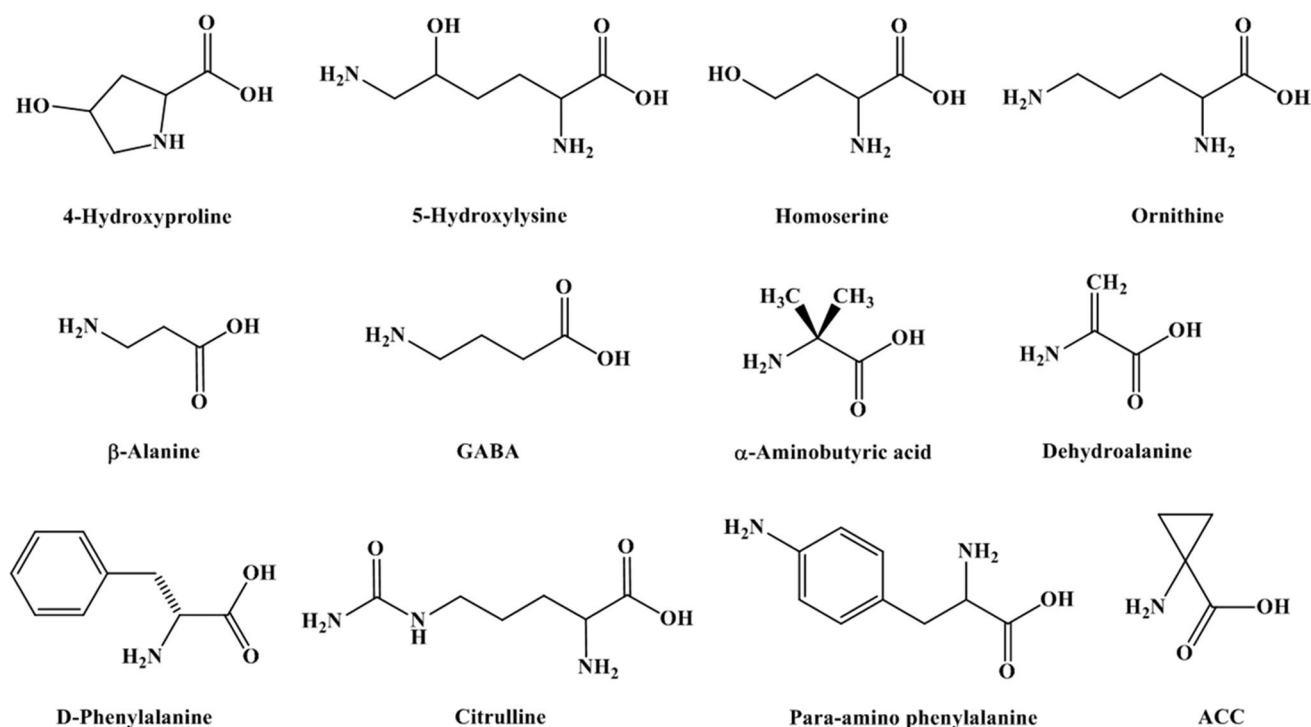
are found in natural proteins due to the post-translational modifications of the naturally occurring twenty amino acids (Ambrogelly et al. 2007). For unclear reasons, nature has always chosen  $\alpha_L$ -amino acids, glycine being the achiral exception, in the biosynthesis of proteins. However, many  $\alpha_L$  and non- $\alpha_L$ -amino acids are also found in the living organisms for which codons are absent in the genetic code for protein synthesis. Such amino acids are collectively classified as non-proteinogenic/unnatural/non-coded amino acids (NCAs) (Bell et al. 2008; Blaskovich 2016). Examples of a few such NCAs have been listed below in Fig. 1.

Notably,  $\geq 1000$  NCAs are naturally found in various plants and microorganisms (Bell 2003; Lu and Freeland 2006; Pollegioni and Servi 2012). The existence of NCAs in living organisms has immensely inspired chemical biologists over the years for peptidomimetic manipulation, which has expanded the chemical and conformational landscapes of the native  $\alpha$ -amino acids for potential usage in biological

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**Fig. 1** Non-coded amino acids (NCAAs) commonly found in living organisms

applications. Interestingly, several NCAAs have been put to the test for enhancing the target-specific biological activity compared to their natural counterparts (Blaskovich 2016). Further, NCAAs can have numerous structural diversity and functional versatility and, thus, are often used as building blocks and molecular scaffolds in constructing combinatorial libraries (Goodman et al. 2007; Johnson et al. 2010). In addition, NCAAs are often incorporated into the peptides to improve the specific backbone conformational propensity and enhance the overall structural stability of the peptides (Balaram 1992; Rana et al. 2005; Vasudev et al. 2011). Indeed, the usage of NCAAs appears to be advantageous in various applications ranging from foldamer design and advanced material to peptide-based drugs (Clerici et al. 2016; Gellman 1998; Li and Yang 2006). NCAAs are widely used in peptidomimetics, and pharmacologically active analogs of the native peptides, as they confer better bioavailability and thermostability. No wonder that many of these NCAAs are also critical components in pharmaceuticals and developmental drugs (Blaskovich 2016) such as antitumor, antimicrobial, osteoporosis, diabetes, and hypertension (Avan et al. 2014; Vlieghe et al. 2010).

In the world of pharmacopeia, peptides as therapeutics bridge the gap between the traditional small molecules and the biologics, as it has a distinct biochemical and therapeutic profile. With  $\geq 60$  FDA-approved drugs in the market, peptides as therapeutics had a glorious past, and current trends

of  $\geq 140$  peptides in clinical trials suggest that peptides also have a fabulous future (Fosgerau and Hoffmann 2015). In general, approval of peptides as targeted therapeutics has a better success rate compared to the traditional small molecules, as peptides are generally known to be physiologically active and efficacious signaling molecules that can modulate cellular signaling by selectively binding to a plethora of transmembrane proteins like G-protein coupled receptors (Davenport et al. 2020) (GPCR) and ion channels, including some enzymes (Fosgerau and Hoffmann 2015; Lau and Dunn 2018). In addition, by the nature of their chemical constituent, peptides are chemically diverse and can be manufactured with ease due to established chemistry. Peptides are accepted to be safer as therapeutics as they have a relatively low immunogenicity profile compared to small molecules (Purcell et al. 2007). More importantly, peptides can selectively target larger surface areas typically involved in protein–protein interactions due to their intermediate size. From the therapeutic standpoint, natural peptides provide high levels of biodiversity, biological activity, specificity, low immunogenicity, and toxicity in therapeutic usages. However, peptides made of natural amino acids tend to have low metabolic stability due to the susceptibility toward proteolysis, leading to a short half-life, poor epithelial absorption, and rapid elimination from the body (Lien and Lowman 2003; Antosova et al. 2009). Altogether, it all adds up to the current bottlenecks of the peptide drug market, which is

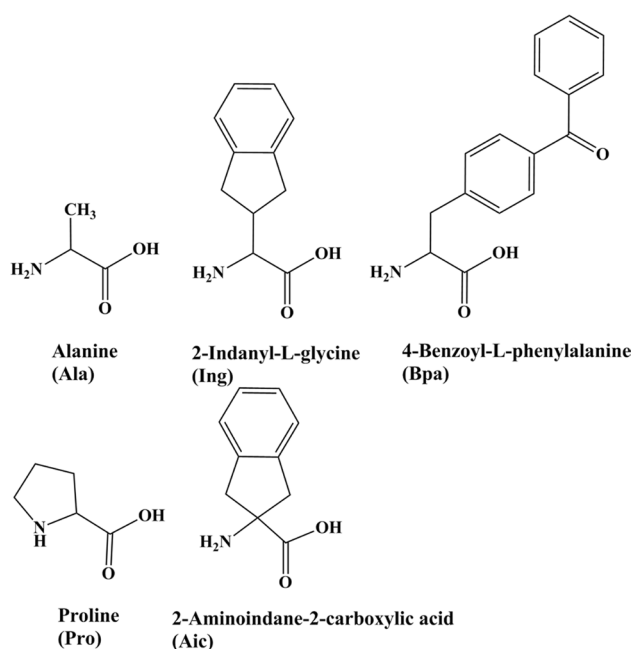
mainly limited to injectables. Though many of the shortcomings associated with natural peptides are being addressed by the designer synthetic peptides, the oral bioavailability of the peptides remains a challenge due to the complex chemical-biological landscape of the gastrointestinal tract. In principle, the utilization of NCAAs in the rational design of therapeutic peptides (Fosgerau and Hoffmann 2015) can substantially overcome some of the limitations encountered by the natural peptides, enhancing their potential for therapeutic usage (Stevenazzi et al. 2014; Cavaco et al. 2018; Stone and Deber 2017). In addition, with the help of evolving formulations and drug delivery platforms, the scope of the peptide drug administration can expand beyond the traditional injectables to oral, intranasal, and transdermal routes.

The issue of bioavailability is currently being addressed by several strategies, out of which rational introduction of NCAAs to the natural peptide backbone appears to be quite promising to improve the oral bioavailability of the peptides (Ding et al. 2020). The presence of NCAAs can influence the membrane permeability and improve the binding affinity of the peptides toward the target proteins by inducing desired conformational changes. In addition, strategic positioning of the NCAAs in the peptide sequences through rational design can also introduce resistance to degradation in the gastrointestinal tract and improve plasma half-life, absorption, and immunogenicity profile of the peptide therapeutics (Di 2015; Knudsen et al. 2000). For example, stereochemical mutation (Durani 2008) of L to D-amino acids has been shown to substantially improve the pharmacokinetic property of octreotide compared to somatostatin, a drug prescribed for treating gastrointestinal tumors (Katz and Erstad 1989; Mahalakshmi and Balaram 2006). Further, simple chemical modification of the terminal amino acids (Rader et al. 2018) like *N*-acetylation and *N*-amidation, including *N*-methylation of amides, has been shown to improve resistance to proteolysis and delayed renal clearance in the case of cyclosporin, a drug prescribed to tame the immune system during organ transplant. Similarly, incorporating D-amino acids and NCAA like 2-naphthyl alanine through rational design has produced LY2510924 peptide, a potent CXCR4 receptor antagonist for treating tumors (Peng et al. 2015). In addition, the introduction of hydrocinnamic acid to PMX53, a potent cyclic peptide antagonist of the C5a receptor (C5aR1) (Liu et al. 2018; Rana et al. 2016), has been shown to improve its oral bioavailability substantially in animal studies (Kumar et al. 2020). Further, the presence of NCAAs like Aib has been shown to significantly improve the therapeutic index of glucagon-like peptide-1 (GLP-1) mimetics in the treatment of type-II diabetes (T2DM), including several other peptide-based drugs (Brunel et al. 2019; Sebokova et al. 2010). Similarly, the activity of the compstatin, a potent cyclic peptide inhibitor

of the complement protein C3, has been shown to improve remarkably by introducing 1-methyl tryptophan at position 4 as a replacement for the valine (Katragadda et al. 2006). In addition, the usage of  $\alpha$ ,  $\beta$ -dehydrophenylalanine in the rational design of peptides has also been shown to improve the antimicrobial activity of the short peptides (Mathur et al. 2007; Sharma et al. 2012). Further, incorporation of the NCAAs like L-ornithine and L-kynurenine into the antibiotics like daptomycin, often prescribed to treat life-threatening bacterial infections, indicates the untapped potential of the NCAAs (Miller et al. 2016; Gray and Wenzel 2020).

Nevertheless, several synthetically modified NCAAs are still there in the literature (Blaskovich 2016), whose basic functionality as a building block of designer peptides or foldamers has not been systematically explored to evaluate their scope in the future biotechnological applications. Interestingly, several NCAAs in free form and embedded in peptides are often found as the central component of many marketed drugs (Gupta and Chauhan 2011). It is also noteworthy that the NCAAs with aromatic structures are often incorporated into proteins through biosynthetic pathways. In general, NCAAs with mono and disubstituted bulky aromatic side chains at the C $\alpha$ -position tend to be conformationally constrained, which is considered disadvantageous for racemization and sterically challenging for proteolysis. Therefore, the inclusion of such NCAAs to the peptide backbone improves the bioavailability of the natural peptides (Ramesh et al. 2011).

In this context, the current study has attempted to systematically explore the backbone conformational propensities of a few aromatic NCAAs (Fig. 2), such as 2-indanyl-L-Glycine (Ing), 4-benzoyl-L-phenyl alanine (Bpa), and 2-amino indane-2-carboxylic acid (Aic) for potential application in de novo design of bioactive peptides (Saghyan and Langer 2016), by recruiting the canonical amino acids like alanine (Ala) and proline (Pro) as the reference. Among the selected NCAAs, Bpa is a photo labeling aromatic amino acid, which has been utilized to study PPIs (Chin et al. 2002; Grunbeck et al. 2011; Joiner et al. 2019; Kauer et al. 1986). Further, both Ing and Aic are aromatic amino acids. Between the two, Aic appears to be a highly conformationally constrained sterically hindered amino acid, which is found in several drug-like molecules (Schiller et al. 1991; Kowalczyk et al. 2007). The computational and experimental data presented in the current study for the selected NCAAs in the context of model synthetic peptides indicate that the said amino acids (Ing, Bpa, and Aic) can be systematically utilized further in rational engineering of conformationally ordered biologically active small designer peptides with higher thermostability and bioavailability in a sequence-dependent manner.



**Fig. 2** Structural comparison of the naturally coded amino acids with the non-coded amino acids explored in the current study

## Results

### Conformational propensities of the selected NCAAs

Conformational propensities of the individual amino acids are usually dependent on the peptide sequence and the overall molecular environment imparted by the solvent conditions. Interestingly, the literature favors the quantum mechanical (QM) method (Revilla-Lopez et al. 2010; Bisetty et al. 2005; Tran et al. 2006) over the classical approach for reliably predicting the conformational propensities of both coded and NCAAs. Though the QM-based methods are relatively more accurate, the accuracy level is nearly similar to the molecular mechanics (MM) based methods for larger systems. In addition, the QM approach is computationally resource-intensive, even for small peptide systems in the presence of explicit water molecules. On the other hand, the computational cost required for classical MM-based molecular dynamics studies in the presence of explicit water molecules is comparatively minimal and has been successful in providing quality data that are reliable to the experimental observations (Croitoru et al. 2021; Beck et al. 2008). Thus, in the current study classical approach has been put to use for garnering the information about the accessible conformational space available to the selected NCAAs (Fig. 2), using the canonical coded amino acids as a reference in a model Ala-X-Ala (X = amino acid of interest) tripeptide system. Initially, Ala-Gly-Ala tripeptide, both acylated and amidated respectively, at the N- and the C-terminus was modeled in an extended conformation

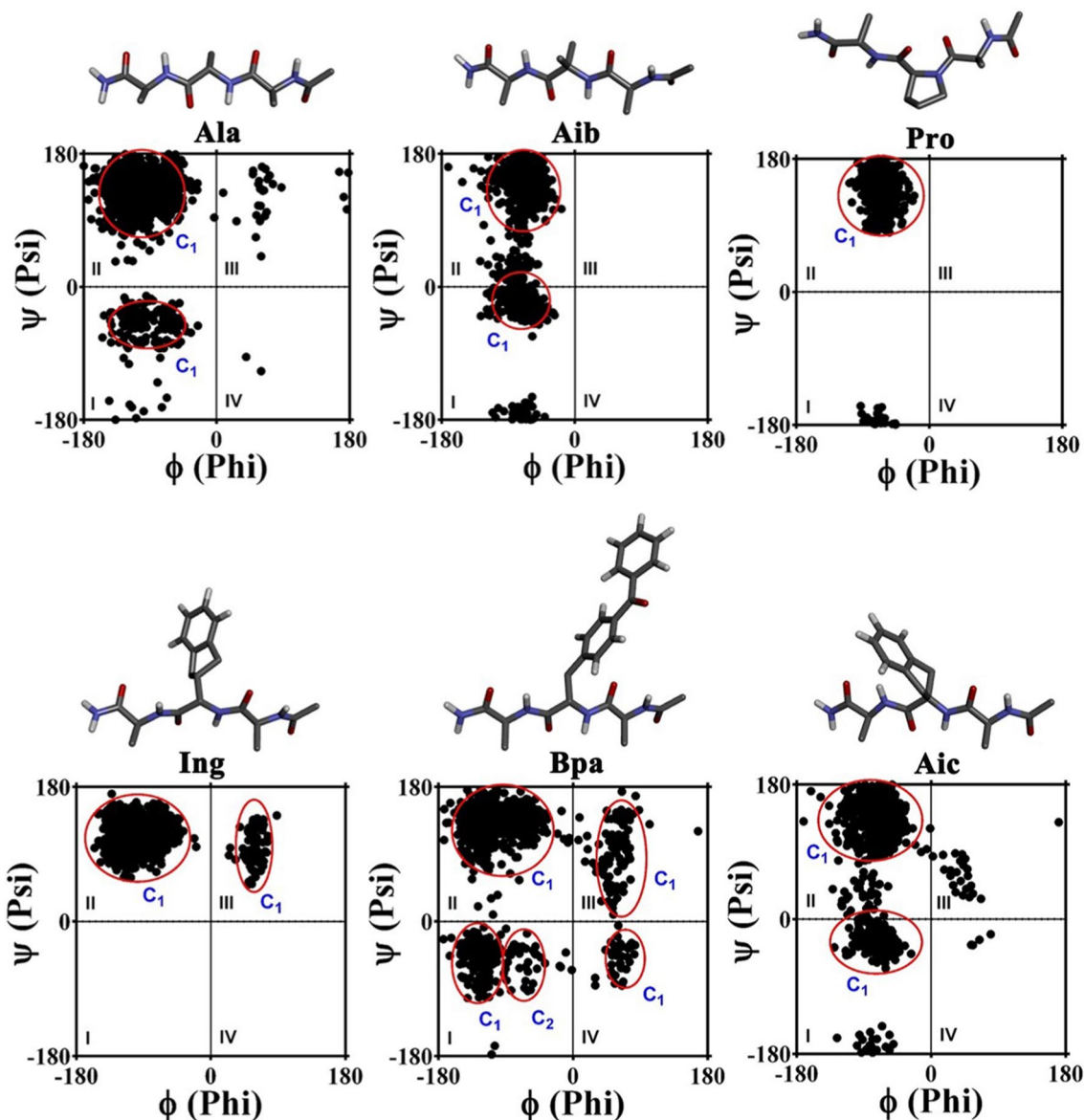
( $\phi = -120^\circ$ ,  $\psi = 120^\circ$ ) and was subsequently subjected to molecular dynamics (MD) studies over 10 ns at 300 K in explicit SPC water for probing whether the 10 ns time would be enough for the model tripeptides to reasonably access the available conformational space in the four quadrants of the Ramachandran plot (Fig. S1).

Further, the Ac-Ala-**Ala**-Ala-NH<sub>2</sub>, Ac-Ala-**Aib**-Ala-NH<sub>2</sub>, and Ac-Ala-**Pro**-Ala-NH<sub>2</sub> tripeptides were also subjected to 10 ns of MD, and the backbone dihedral angles of the highlighted amino acids were further plotted at an interval of 10 ps. The conformational propensity data presented for the coded amino acids in Fig. 3 (Table 1) strongly correlate with the observations made in the structural studies of the proteins and peptides. Thus, the Ac-Ala-**Ing**-Ala-NH<sub>2</sub>, Ac-Ala-**Bpa**-Ala-NH<sub>2</sub>, and Ac-Ala-**Aic**-Ala-NH<sub>2</sub> tripeptides, containing the non-coded amino acids like 2-indanyl-L-Glycine, 4-benzoyl-L-phenyl alanine, and 2-aminoindane-2-carboxylic acid were also subjected to 10 ns of MD at 300 K in the presence of explicit water.

In addition, the data presented for these amino acids in Fig. 3 and Table 1 indicate that both Ing and Bpa have conformational propensities broadly similar to Ala, whereas the Aic has a conformational propensity similar to Aib. It is noteworthy that while Aib has a hydrophobic side chain, Aic has a side chain that is both hydrophobic and aromatic in nature. Further, it is hypothesized that free Ing can exhibit both “endo” and “exo” conformations due to the possible pucker in the adjacent five-membered ring. However, quantum mechanical studies suggest that the “endo” conformer has lower energy than the “exo” conformer (Renfrew et al. 2012). Thus, Ing was modeled in the “endo” conformer. Interestingly, Ing in “endo” or “exo” conformer did not sample the right-handed helical conformations in the model tripeptide context. However, backbone conformations broadly similar to the left-handed helical conformations were noted in our studies (Fig. 3). In addition, despite having an elongated double aromatic side-chain structure, Bpa could access conformational space in all four quadrants of the Ramachandran plot (Table 1). The energetic comparison of the conformations (allowed/forbidden) accessed by the NCAAs was not attempted, considering the qualitative nature and context of the study. However, each trajectory was subjected to backbone conformational clustering with a cut-off of 1.2 Å. The data presented in Figs. 4, 5, and 6 indicate that in addition to the extended sheet conformations, the given NCAAs can also induce helical or polyproline-II type conformations in the tripeptide backbone (Table 2).

### Modeling of Helical peptides with the selected NCAAs

To further probe the conformational propensities of the given NCAAs in the context of the polypeptide structure,



**Fig. 3** Ramachandran plot illustrating the broad conformational landscapes of the NCAAs compared to the coded amino acids in the context of Ac-Ala-X-Ala-NH<sub>2</sub> tripeptide over 10 ns of MD at 300 K. The

red circles highlight the most populated clusters in each quadrant of the Ramachandran plot

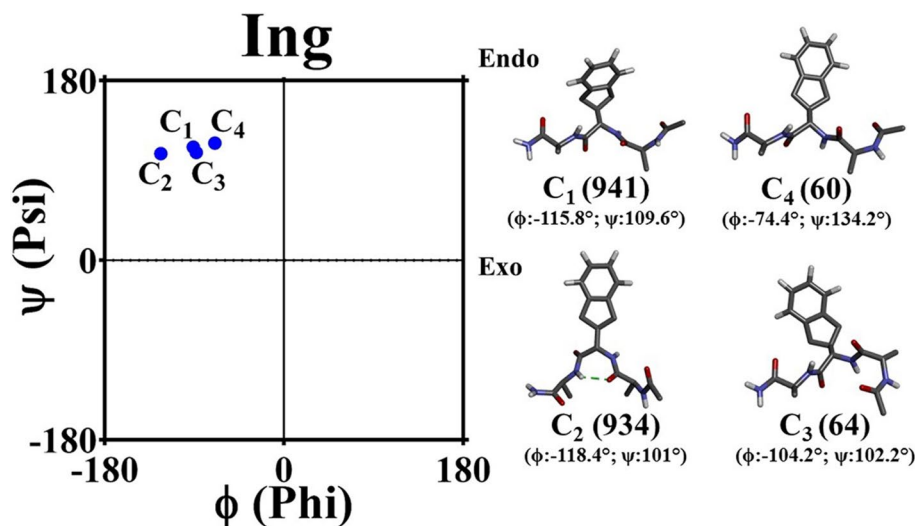
five 18-mer peptides (Hed1, Hed2, Hed3, Hed4, and Hed5) with almost identical sequences were modeled in the right-handed helical conformations. Among the five peptides, Hed1 [Ac-YGKAAAAKAAAANKAAAK-NH<sub>2</sub>] was designed to serve as the positive control. The Hed2 peptide was designed to serve as a negative control by introducing the **Ala9/Pro9** mutation in the Hed1 peptide, as proline is known to be a helix breaker under general solvent conditions. On the other hand, Hed3, Hed4, and Hed5 peptides harbored the **Ala9/Ing9**, **Ala9/Bpa9**, and **Ala9/Aic9** mutation, respectively, so the effect of these NCAAs on the helical conformation of the peptide can

be effectively probed. Subsequently, the peptides were subjected to 100 ns MD studies, each in the presence of explicit water with 0.15 M NaCl at 300 K. As presented in Fig. 7, end fraying is observed across all the peptides, except the Hed2 peptide, which demonstrates a poorly folded random coil-like structure under the given conditions. Further, in addition to the Hed1, all other peptides could also sustain helical conformation to a different extent despite having the NCAAs. Overall, the data suggest that the selected NCAAs can potentially support the helical conformation to a great extent in the context of sequence-optimized polypeptides.

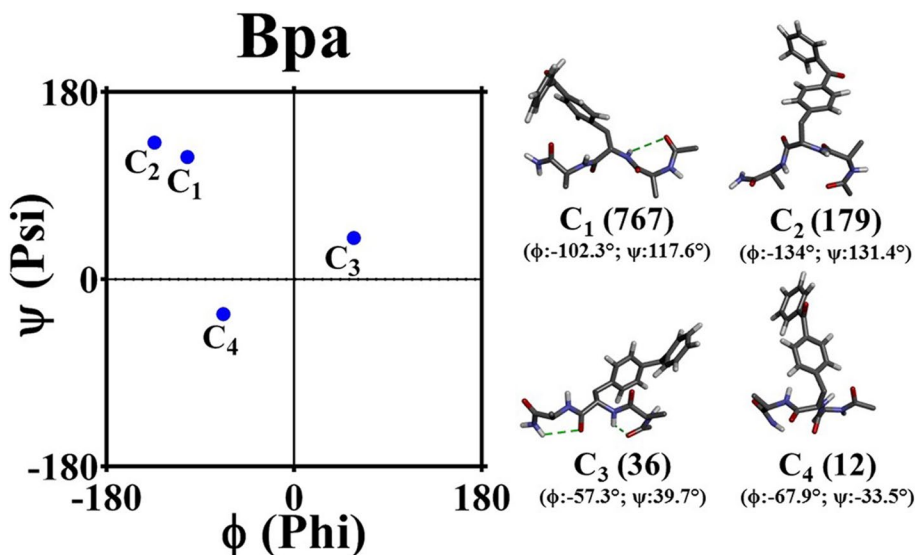
**Table 1** The  $\phi$ ,  $\psi$  angles (mean  $\pm$  SD) observed for the amino acids in the most populated clusters

Quadrant	I	II	III	IV
Ala	C1: $\phi = -94^\circ \pm 26^\circ$ $\psi = -45^\circ \pm 22^\circ$	C1: $\phi = -100^\circ \pm 26^\circ$ $\psi = 131^\circ \pm 23^\circ$		
Aib	C1: $\phi = -66^\circ \pm 10^\circ$ $\psi = -26^\circ \pm 10^\circ$	C1: $\phi = -65^\circ \pm 12^\circ$ $\psi = 144^\circ \pm 18^\circ$		
Pro		C1: $\phi = -63^\circ \pm 10^\circ$ $\psi = 139^\circ \pm 19^\circ$		
Ing		C1: $\phi = -98^\circ \pm 23^\circ$ $\psi = 118^\circ \pm 17^\circ$	C1: $\phi = 59^\circ \pm 7^\circ$ $\psi = 98^\circ \pm 23^\circ$	
Bpa	C1: $\phi = -123^\circ \pm 13^\circ$ $\psi = -48^\circ \pm 20^\circ$ C2: $\phi = -67^\circ \pm 14^\circ$ $\psi = -52^\circ \pm 20^\circ$	C1: $\phi = -103^\circ \pm 29^\circ$ $\psi = 132^\circ \pm 19^\circ$	C1: $\phi = 57^\circ \pm 11^\circ$ $\psi = 94^\circ \pm 35^\circ$	C1: $\phi = 63^\circ \pm 10^\circ$ $\psi = -46^\circ \pm 17^\circ$
Aic	C1: $\phi = -69^\circ \pm 13^\circ$ $\psi = -29^\circ \pm 10^\circ$	C1: $\phi = -70^\circ \pm 17^\circ$ $\psi = 135^\circ \pm 22^\circ$		

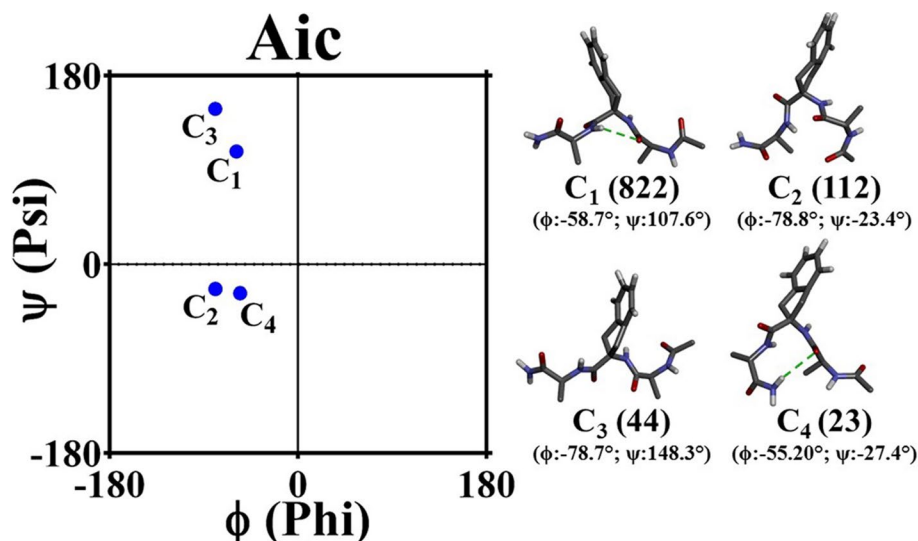
**Fig. 4** Schematic illustration of the central conformers of the most populated clusters evolved for the Ac-Ala-Ing-Ala-NH<sub>2</sub> tripeptide over the duration of MD. The data are representative of both the “endo” and “exo” trajectory. The number within the parenthesis indicates the total number of conformers for the given major cluster. The dotted green lines highlight the backbone hydrogen bonds



**Fig. 5** Schematic illustration of the central conformers of the most populated clusters evolved for the Ac-Ala-Bpa-Ala-NH<sub>2</sub> tripeptide over the duration of MD. The number within the parenthesis indicates the total number of conformers for the given major cluster. The dotted green lines highlight the backbone hydrogen bonds



**Fig. 6** Schematic illustration of the central conformers of the most populated clusters evolved for the Ac-Ala-Aic-Ala-NH<sub>2</sub> tripeptide over the duration of MD. The number within the parenthesis indicates the total number of conformers for the given major cluster. The dotted green lines highlight the backbone hydrogen bonds



**Table 2** The probable backbone conformational propensities of the selected NCAAs ascertained from the cluster analysis

Clusters	Ing	Bpa	Aic
C <sub>1</sub>	Parallel $\beta$ -sheet	Parallel $\beta$ -sheet	$\beta$ -sheet
C <sub>2</sub>	Parallel $\beta$ -sheet	Anti-parallel $\beta$ -sheet	$\alpha_R$ -helix
C <sub>3</sub>	Parallel $\beta$ -sheet	$\alpha_L$ -helix	Poly-Pro helix-II
C <sub>4</sub>	Poly-Pro helix-II	$\alpha_R$ -helix	$3_{10}$ -helix

### Preparation of the synthetic peptides

The MD data (Fig. 7) suggested that the model peptides could maintain helical conformations to a certain extent in the presence of explicit water molecules. Thus, the model peptides were synthetically prepared for experimental validation by recruiting the well-established solid-phase peptide synthesis procedures and further purified to homogeneity. Except for Hed5, all other peptides were end-capped (acylated and amidated), respectively, at the N- and C-terminus (Fig. 8). The sequence integrity of the peptides was verified by mass spectrometry (Fig. S2 to S6). The solubility check experiment indicated that the synthetic peptides are soluble in water.

It is noteworthy that acylation and amidation of the N-terminus and C-terminus of the peptides increase two amide bonds, which can not only improve the conformational stability of the short peptides but also can enhance proteolytic stability and antimicrobial potency of the synthetic peptides. In fact, NT-acylation is an established *in vivo* mechanism for moderating the structure, folding, and function of the proteins (Ree et al. 2018). However, several physiologically active peptides do not undergo NT-acylation, as it affects the binding of the peptides to their cognate receptors and subsequent signaling. Thus, as a control, the Hed5 peptide was

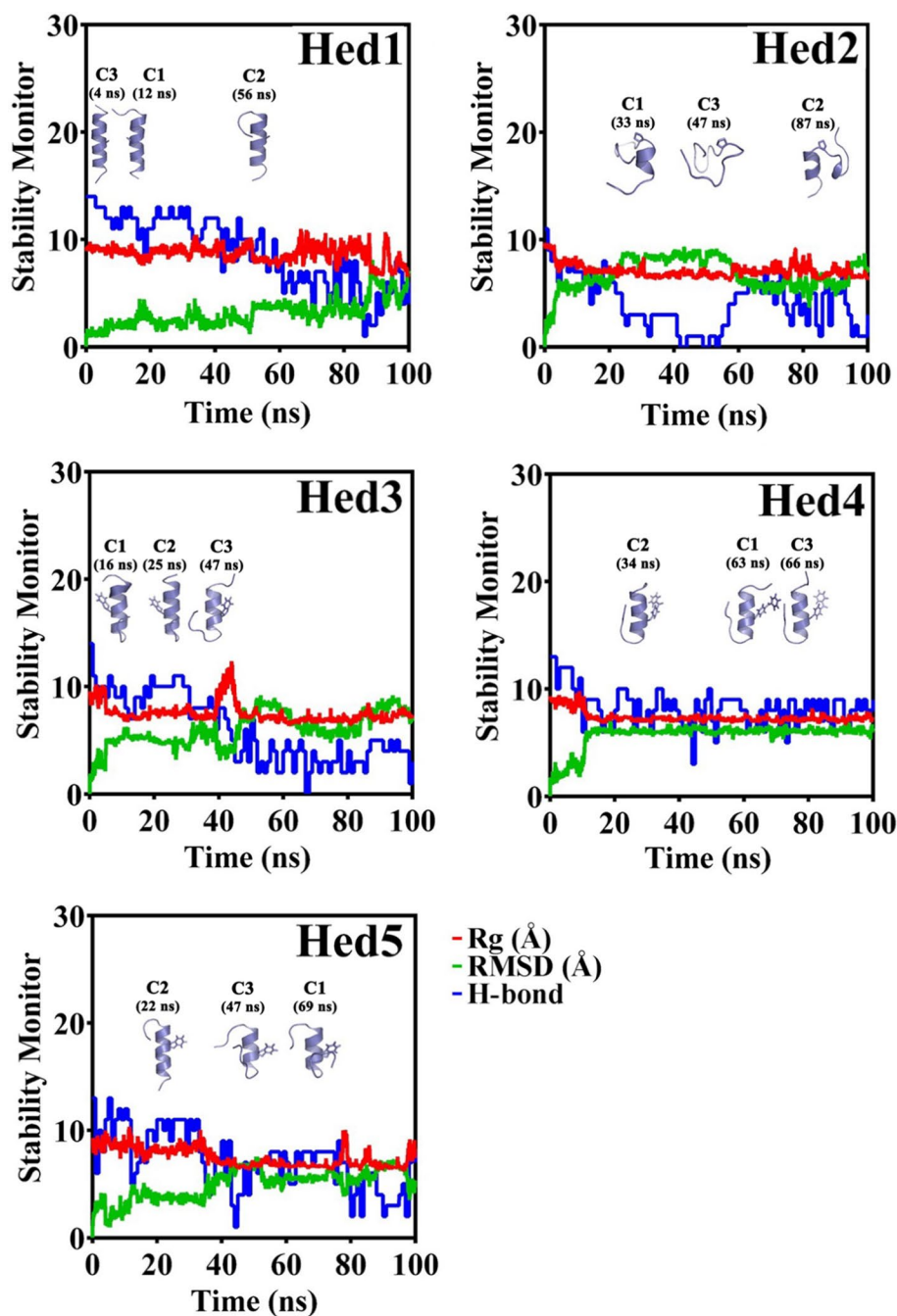
not subjected to NT-acylation or CT-amidation for a better comparative understanding of the conformational effect in different solvent conditions.

### Solution conformation of the synthetic peptides

The influence of the NCAAs on the overall conformation of the synthetic peptides was evaluated further by far-UV circular dichroism (CD) spectroscopy at 25 °C in the presence of various solvents (water, PBS buffer, methanol), cosolvents (TFE: trifluoroethanol, HFIP: hexafluoroisopropanol) and aqueous micellar conditions (SDS: sodium dodecyl sulfate). As presented in Fig. 9, characteristic negative CD bands, respectively at 208 and 222 nm, indicate helical conformation in the Hed1 peptide. On the other hand, the typical negative CD band around 190 nm confirms a random coil-like conformation for the Hed2 peptide in pure water (pH ~ 5.8) and 1×PBS, confirming the design principles.

It is noteworthy that in comparison to Hed1, Hed2 peptide harbors Ala9/Pro mutation, which is responsible for its random coil-like structure in water, as observed in the MD studies (Fig. 7). Interestingly, helical conformation is also observed for the Hed3, Hed4, and Hed5 peptides, mutated with Ala9/Ing, Ala9/Bpa, and Ala9/Aic NCAAs in pure water. However, the observed CD intensities for these peptides were marginally lower than the Hed1 peptide. However, all the peptides, including the Hed2 peptide, demonstrated perfect helical conformations in the presence of methanol, 20–30% TFE, 20–30% HFIP, including SDS micelles, indicating better conformational ordering, possibly due to the minimal end fraying in the model peptides under the influence of the cosolvents. The estimated percentage helicity of the peptides (Table S1) indicates 28–67% helicity for the Hed1, 0–47% helicity for the Hed2, 20–53% helicity for the Hed3, 9–14% helicity for the Hed4, and 14–43% helicity for

**Fig. 7** Summary of the MD analysis data for the model peptides. The decrease in the total number of helical hydrogen bonds indicates the observed end fraying of the peptides. The central conformers of the top three most populated clusters for each model peptide are presented in cartoon representation



the Hed5 peptide across all the solvent conditions at 25 °C. Though the 222/208 ratio has been calculated for the peptides under all the solvent conditions, attributing the data to delineate helical character ( $\alpha/3_{10}$ ) has not been attempted due to the simple nature of the peptide sequences. Overall the data suggest that the tested NCAs can support ordered helical conformation in a sequence-optimized short peptides, as substitution of NCAs did not alter the conformational state of the synthetic peptides.

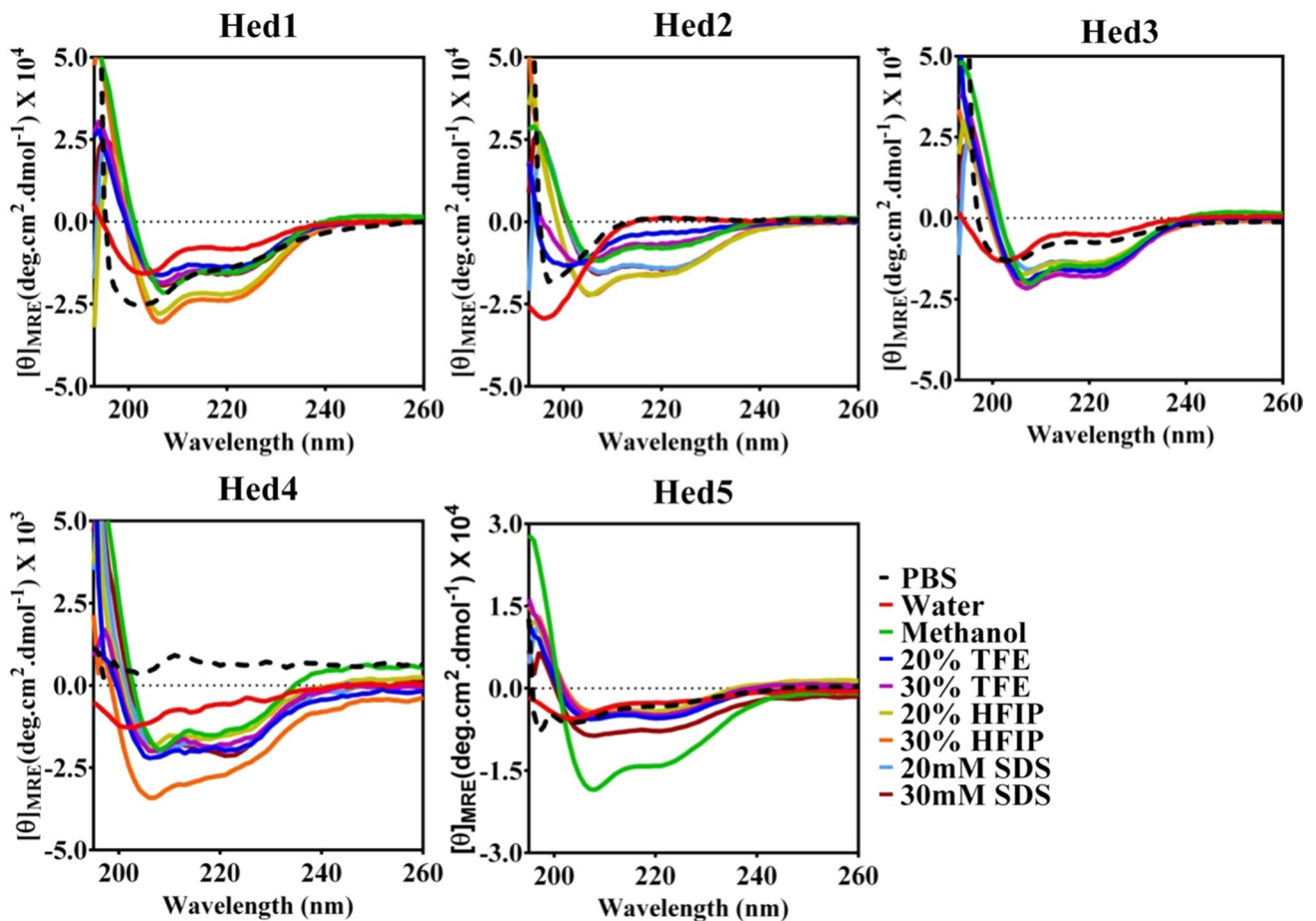
Further, it was also essential to assess whether the tested NCAs could contribute toward the thermostability of the

peptides. It is well established that in addition to the amino acid sequence, secondary structure content and folding of the peptides are both solvent and temperature-dependent. However, a highly ordered folded peptide does not consistently demonstrate higher thermostability. A well-folded protein or peptide can possess a highly asymmetric secondary structural elements such as  $\alpha$ -helix. Still, when it unfolds, it reveals loss in secondary structure, which is observable in the change in CD band intensities. Thus, recording CD spectra as a function of temperature are often helpful for evaluating the effects of mutations on the overall stability of



Peptide	Sequence	T <sub>mol. wt.</sub>	O <sub>mol. wt.</sub>	Purity
Hed1	Ac-YGKAAAAK- <b>Ala</b> -AAAKAAAAK-NH <sub>2</sub>	1644.92	1645.2	95.1%
Hed2	Ac-YGKAAAAK- <b>Pro</b> -AAAKAAAAK-NH <sub>2</sub>	1670.96	1671.2	99.0%
Hed3	Ac-YGKAAAAK- <b>Ing</b> -AAAKAAAAK-NH <sub>2</sub>	1747.10	1747.2	97.8%
Hed4	Ac-YGKAAAAK- <b>Bpa</b> -AAAKAAAAK-NH <sub>2</sub>	1825.13	1825.2	97.0%
Hed5	<sup>+</sup> <sub>3</sub> HN-YGKAAAAK- <b>Aic</b> -AAAKAAAAK-COO <sup>-</sup>	1691.98	1691.6	95.3%

**Fig. 8** Comparison of the amino acid sequences of the model synthetic peptides. The amino acids highlighted in color indicate the position-specific difference between the peptide sequences



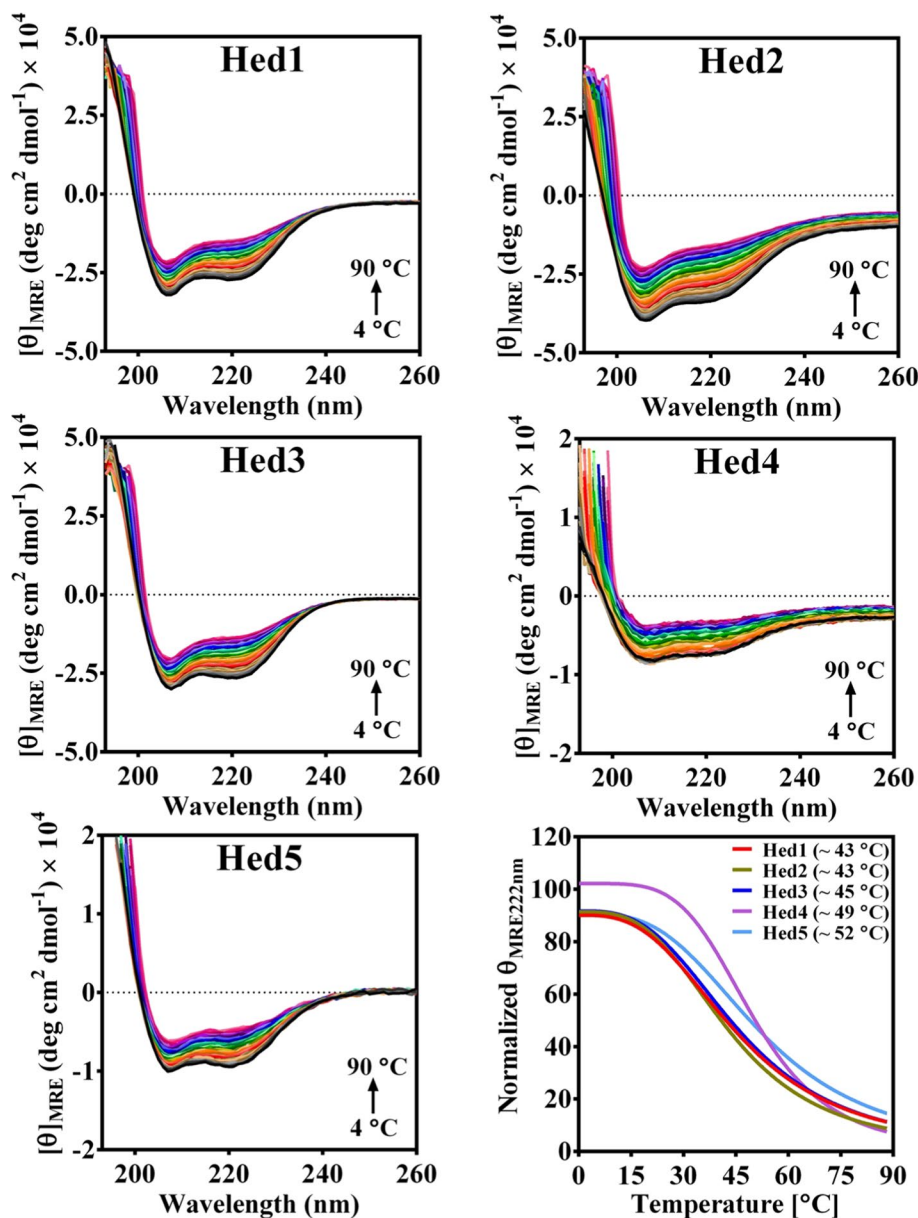
**Fig. 9** Far-UV CD signature of the model synthetic peptides (50 μM) as observed under the influence of various solvents and cosolvents at 25 °C

the peptide or protein. For a fair comparison of thermostability, all the model peptides were subjected to a temperature ramping between 4–90 °C in the presence of 20% HFIP. The data presented in Fig. 10 clearly indicate that the model peptides have a well-ordered helical conformation at the low temperature, which subsequently changes with the increase in the solvent temperature. Interestingly, no other unfolding

intermediates are observed for the peptides throughout the thermal denaturation studies.

The melting temperature of the model synthetic peptides estimated from the data and the percentage of helicity estimated at the initial and the final solvent temperature are presented in Table 3. It is noteworthy that both Hed1 and Hed2 peptides having only canonical amino acids in

**Fig. 10** Thermal denaturation (4–90 °C) profiles of the synthetic model peptides (50  $\mu$ M) recorded in 20% HFIP highlight the melting temperatures ( $T_m$ ) of the respective peptides



their sequences demonstrated an almost similar denaturation profile with an estimated  $T_m$  of 43 °C. In addition, the Hed3 peptide, which harbors Ala9/Ing mutation, also demonstrated a similar denaturation profile but with a slightly higher  $T_m$  of 45 °C. In contrast, Hed4 (Ala9/Bpa) and Hed5 (Ala9/Aic) peptides demonstrated a similar denaturation profile, which is different from the Hed1, Hed2, and Hed3 peptides. More importantly, both Hed4 and Hed5 peptides displayed  $T_m$  of 49 °C and 52 °C, which is significantly higher than the other peptides. Overall, the data suggest that careful usage of such NCAs in the rational

**Table 3** Melting temperature ( $T_m$ ) and the percentage helicity estimated from the thermal denaturation profile of the model peptides

Peptide	$T_m$ (°C)	Helicity (%) at (4 °C)	Helicity (%) at (90 °C)
Hed1	43	85	45
Hed2	43	91	47
<b>Hed3</b>	<b>45</b>	75	42
<b>Hed4</b>	<b>49</b>	26	16
<b>Hed5</b>	<b>52</b>	32	19

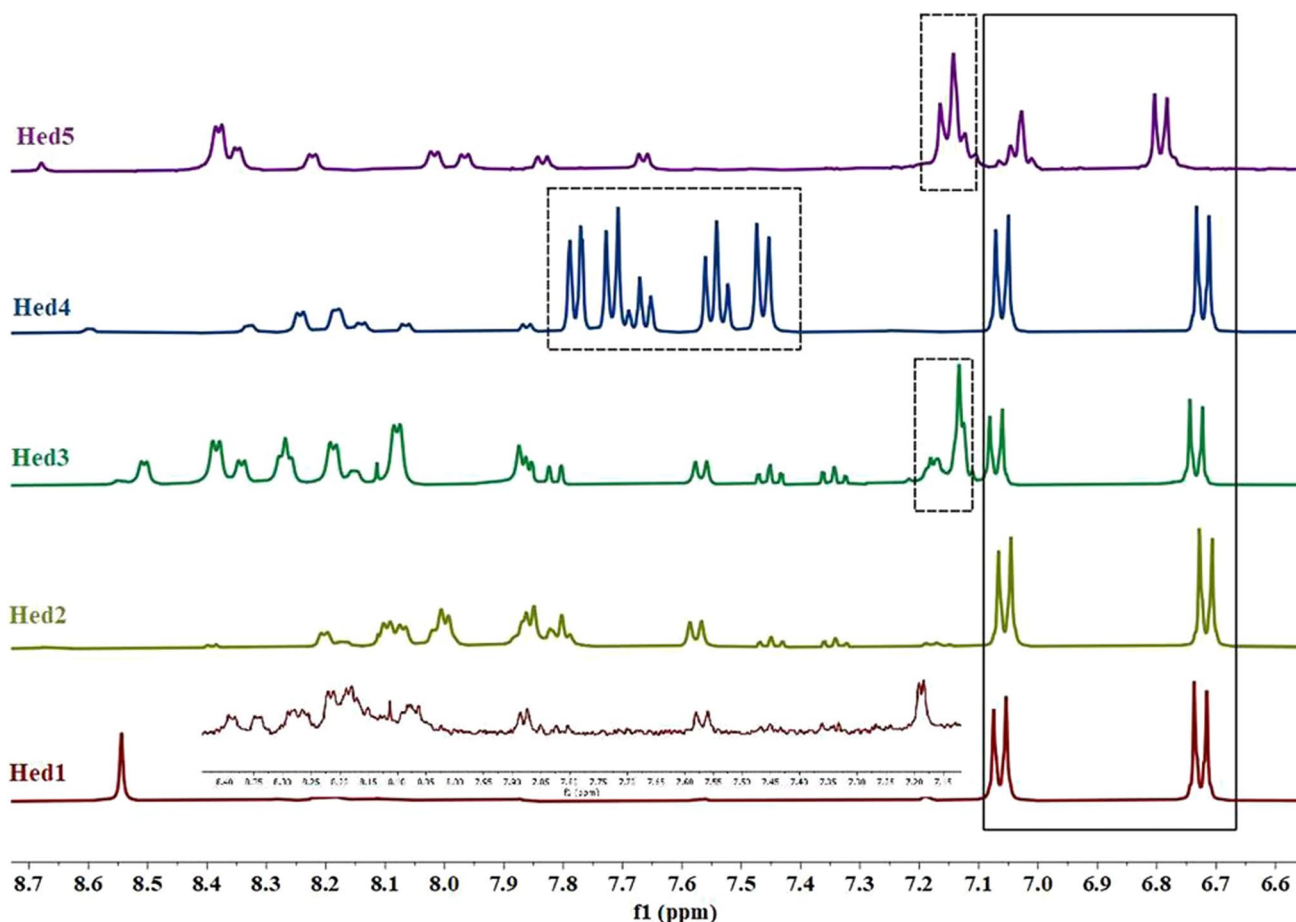
design of peptide sequences can substantially improve the physical property of the peptides.

### The $^1\text{H-NMR}$ of the model synthetic peptides

The CD data (Fig. 9) suggest that the synthetic model peptides acquire better conformational ordering in the presence of methanol. Thus, 1D  $^1\text{H-NMR}$  of the peptides was recorded in  $\text{CD}_3\text{OD}$  at 25 °C. The chemical shift dispersion observed for the peptides in 1D  $^1\text{H-NMR}$  data (Fig. 11) agrees with the CD data and strongly suggests the ordered nature of the peptides in the solution. The  $^1\text{H-NMR}$  spectra are concentration-independent in both chemical shifts and line widths in the 0.2–2 mM concentration range in  $\text{CD}_3\text{OD}$ , which suggests that the peptides are free from the aggregation in the working concentration range (Fig. S7). Given the nature of the model peptide sequences, determination of the solution structure of the peptides through 2D  $^1\text{H-NMR}$  was not attempted. However, an attempt was made to garner broad evidence from the observed (Fig. 8)

deviation of  $\text{C}^\alpha\text{H}$  chemical shift ( $\Delta\delta_{\text{H}\alpha}$ ), which is often referred to as the chemical shift index (CSI) and used as an indicator to judge the secondary structures in the peptides and proteins (Wishart et al. 1992). The hypothetical CSI plot presented for the peptides (Fig. 12) indicates that most of the amino acids are in helical conformation, and the CD signature observed for the peptides is likely due to the continuous helical conformation extended throughout the peptide backbone and is not localized to a patch of amino acids.

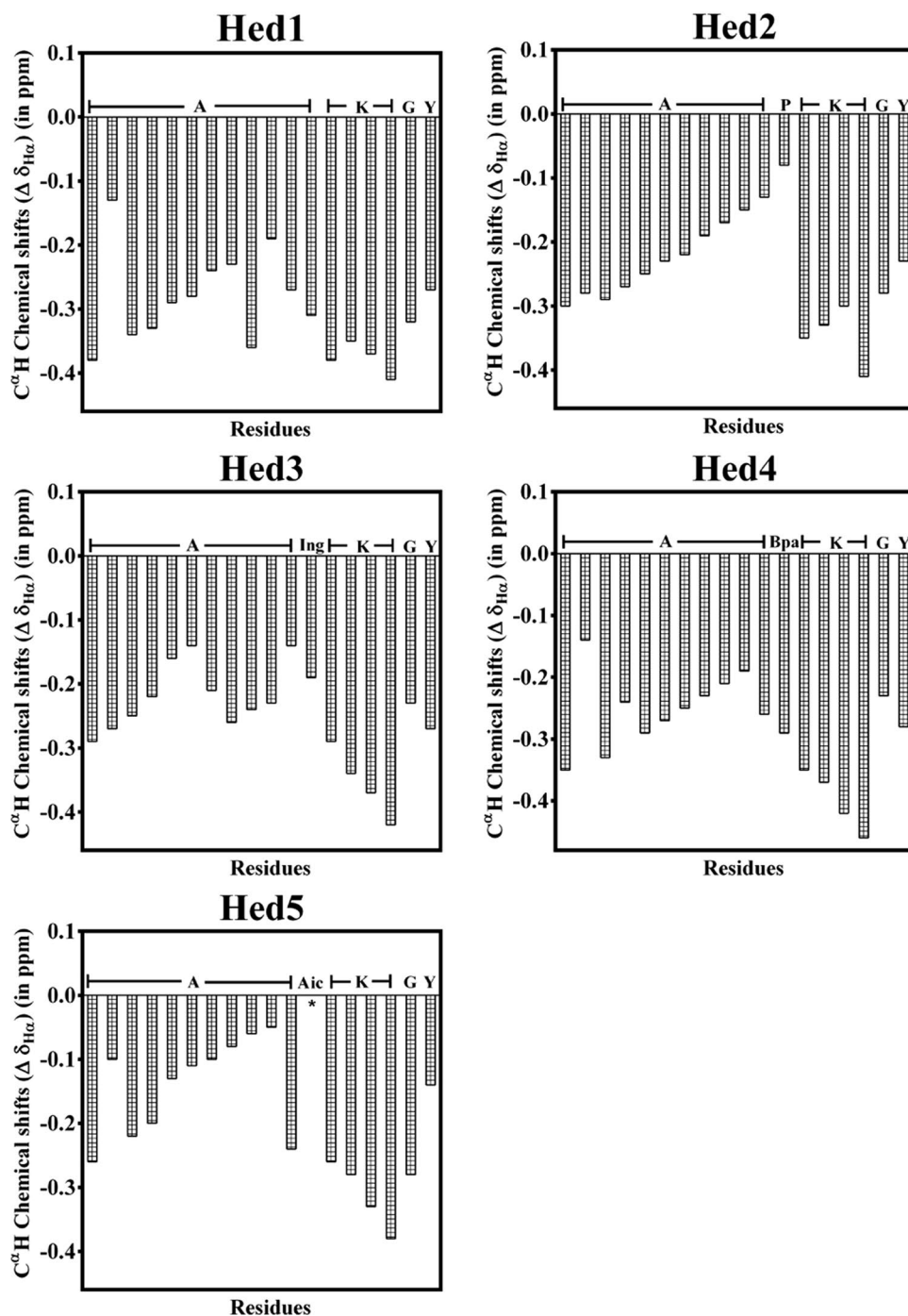
In addition, as illustrated in Fig. 7, all the peptides maintain helical conformation in the water over the duration of the MD, except Hed2, which demonstrates short helical/random conformation due to the presence of a proline at the central part of the sequence. It is well known that proline acts as a helix breaker under general solvent conditions but can support helical conformation both in the presence of micelles and organic solvents (Li et al. 1996). Thus, the CSI plot presented in Fig. 12 for the peptides in  $\text{CD}_3\text{OD}$  depicts a realistic conformational state of



**Fig. 11** The  $^1\text{H-NMR}$  of the model synthetic peptides (400 MHz, ~2 mM) in  $\text{CD}_3\text{OD}$  at 298 K highlights the chemical shift dispersion in the NH and aromatic regions. The tyrosine ar-

omatic protons for the peptides are shown in the solid box, whereas the aromatic protons for Ing (Hed3), Bpa (Hed4), and Aic (Hed5) are, respectively, shown within dotted boxes

**Fig. 12** Schematic illustration of the chemical shift index ( $\Delta\delta_{\text{H}\alpha}$ ) of the amino acids of the model synthetic peptides in reference to the average random coil values observed in  $\text{CD}_3\text{OD}$ , suggesting the helical conformation in the peptides



the individual amino acids contributing toward the overall helical conformation observed for the peptides in the CD studies (Fig. 9).

### Fluorescence spectra of the model synthetic peptides

The model synthetic peptides have a tyrosine amino acid at the *N*-terminus. Thus, the tyrosine fluorescence was recorded

for all the peptides at 50  $\mu\text{M}$  concentration in the presence of various solvents and cosolvents at 25  $^{\circ}\text{C}$  to understand the role of the tested NCCAs on the fluorescence signal of the peptides.

As noted in Fig. 13, all the peptides demonstrate an emission maximum at around 310 nm in agreement with the tyrosine fluorescence, which does not shift significantly under the influence of the different solvents and cosolvents. In agreement with the CD data, the highest fluorescence

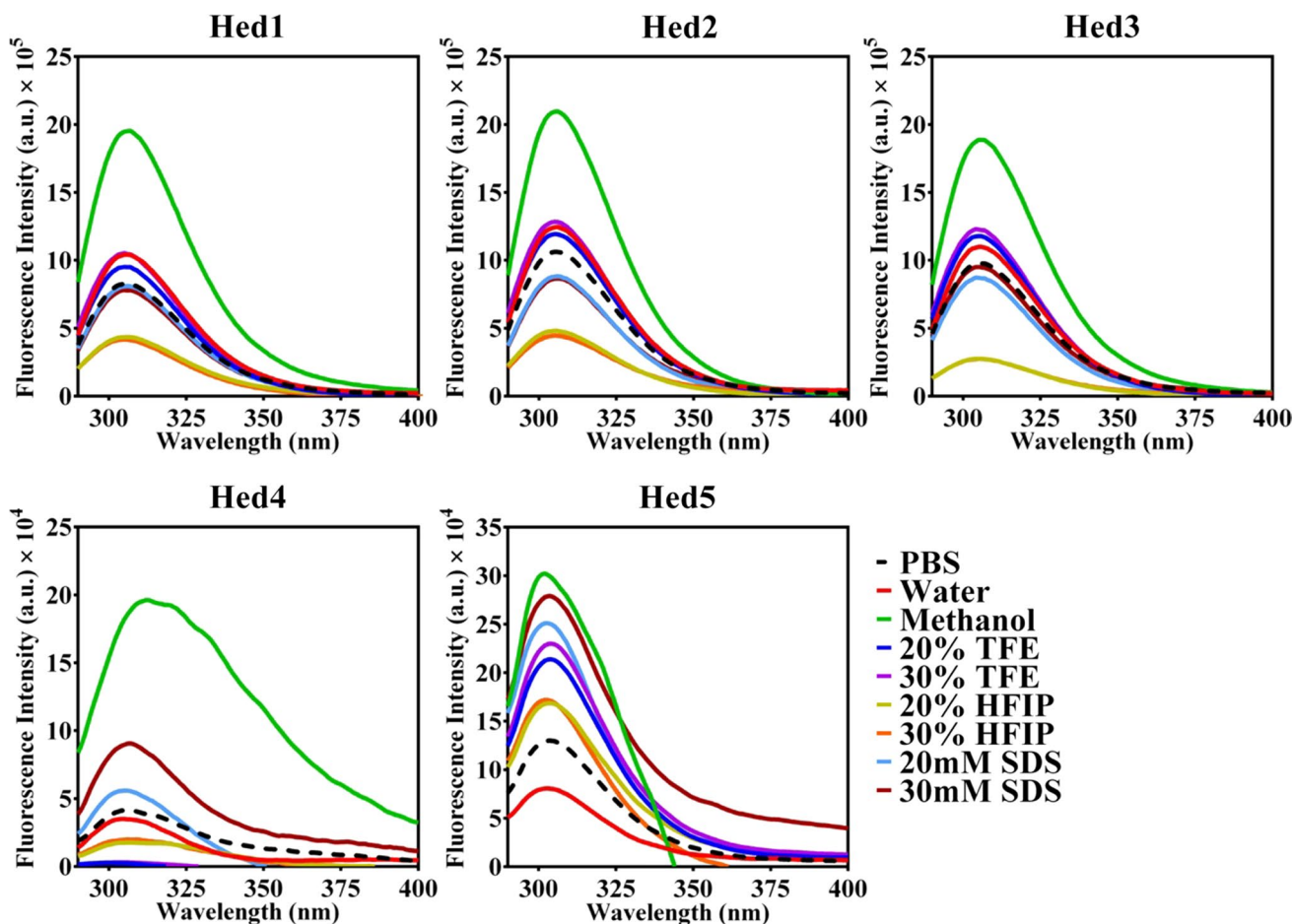


Fig. 13 Tyrosine fluorescence emission spectra of 50  $\mu\text{M}$  model synthetic peptides in the presence of various solvents and cosolvents at 25  $^{\circ}\text{C}$

intensities are also noted for the model peptides in the methanol. Despite the better conformational ordering, the fluorescence intensity of the peptides is consistently lower in the presence of TFE and HFIP compared to methanol, which indicates the solvent-induced fluorescence quenching. However, it is observed that in contrast to Hed1, Hed2, and Hed3 peptides, the fluorescence intensity of the Hed4 and Hed5 peptides in methanol is considerably lower, which could be most likely due to strong intramolecular  $\pi$ - $\pi$  interaction (Vaiana et al. 2003), involving the aromatic side chains of Bpa and Aic respectively, with the tyrosine side chain, as observed in the MD studies (Fig. 14).

## Discussion

The incorporation of the NCAAs in the rational design of peptides has the power to untap the limitless potential of the NCAAs in the discovery and engineering of peptide-based future therapeutics. Thus, it is highly pertinent to systematically explore and understand the conformational

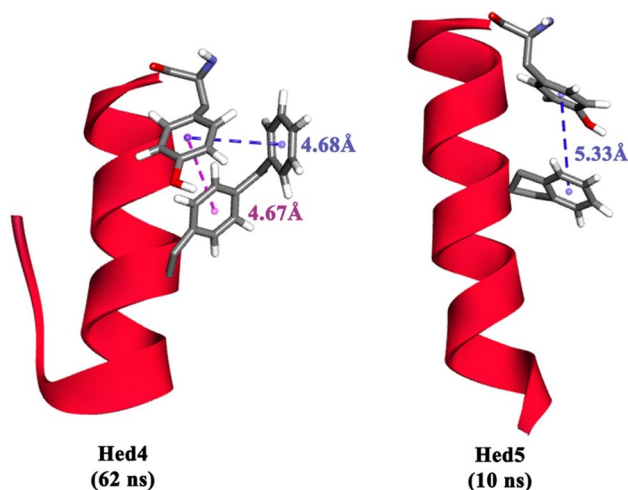
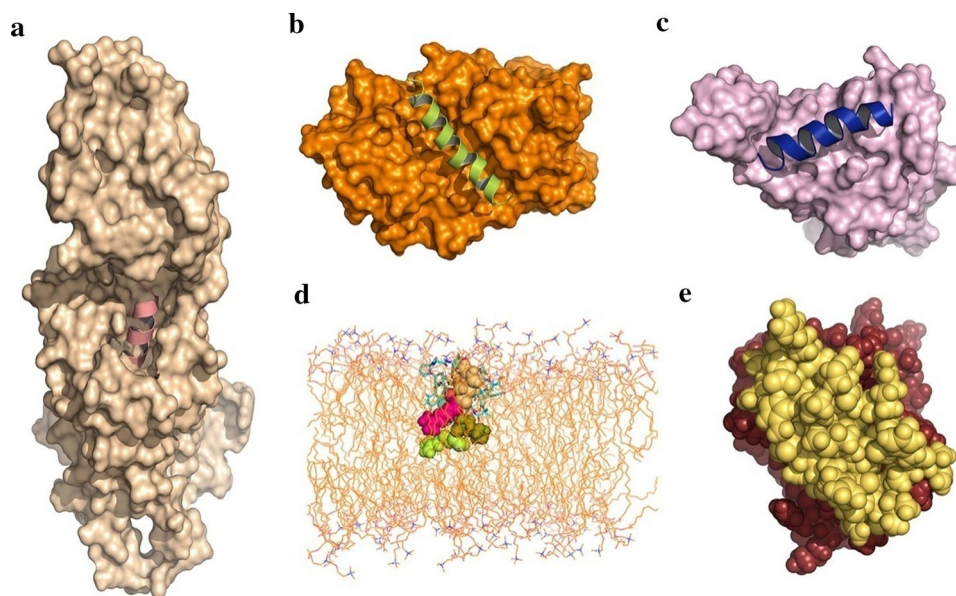


Fig. 14 Structural snapshots extracted from the MD trajectories highlight the potential intramolecular  $\pi$ - $\pi$  interaction between the tyrosine and the aromatic side chains of Bpa and Ing, respectively, in Hed4 and Hed5 peptides

propensities of the various NCAs in realistic model peptide sequences under a variety of solvent conditions. The current study has attempted to provide a glimpse of the possible approach in a few selected aromatic NCAs, which can be further expanded to other types of NCAs. The data presented for the model 18-mer peptides indicate that the selected aromatic NCAs are well tolerated in the peptide backbone and are able to support the helical conformation of the peptides across the solvent conditions. Thus, the NCAs can be further utilized in the rational engineering of short helices in sequence-optimized peptides. Nevertheless, the potential of the selected NCAs to influence  $\beta$ -sheet conformation needs to be further probed in a sequence and structure-dependent manner. In addition to the conformational ordering (Figs. 7, 9, and 12), it is also observed that the selected NCAs can enhance the thermal stability of the model peptides (Fig. 10) by several degrees. Peptides with NCAs are generally proteolytically resistant, and enhancement of thermostability of such peptides is a desired pharmacokinetic gain required for the development of peptide-based therapeutics. Further, it is noteworthy that the selected aromatic NCAs can forge both hydrophobic as well as other types of inter and intramolecular interactions, such as “ $\pi$ - $\pi$ ”, “cation- $\pi$ ”, “sulfur- $\pi$ ”, “alkyl- $\pi$ ”, including unconventional weak hydrogen bonding with the neighboring amino acids,

which can enhance the overall molecular recognition ability of the designer peptides for targeting physiologically relevant proteins, enzymes and receptors. Moreover, combining the power of computational biology, the model 18-mer peptides described in the current study can be further optimized in terms of sequence, structure, and overall length (Joshi et al. 2006) by utilizing a proper mix of both coded and all kinds of NCAs. Recruitment of the NCAs can not only improve the binding affinity and stability but also can significantly enhance the druggability and formulation capabilities of the synthetic peptides as future drugs, which can be exploited in a variety of therapeutic applications as illustrated schematically in Fig. 15.

For example, the glucagon-like peptide-binding receptor (GLP-1R) is an established drug target for the management of type-II diabetes mellitus (T2DM). GLP-1R binds to GLP-1 peptide, an incretin hormone made of 30 or 31-amino acids, and triggers the secretion of insulin from the pancreatic  $\beta$ -cells in a glucose-dependent manner. However, GLP-1 is quickly metabolized by the enzyme dipeptidyl peptidase-4 (DPP4), resulting in the inactivation of the GLP-1. The usage of NCA like Aib has been useful in the successful design of 30-mer peptides, like semaglutide and taspoglutide, which are being marketed as FDA-approved peptide drugs for the management of T2DM. The recent discovery and translation of small molecule drugs targeting GLP-1R to



**Fig. 15** Potential therapeutic applications of the sequence-optimized designer peptides  $\leq 30$  amino acids composed of both coded and NCAs. **a** Targeting the GLP-1R (wheat, PDB: 6XOX) through short designer peptide agonists (GLP-1 mimetic, salmon pink) for the management of T2DM, **b** Targeting the PPIs involving the G-protein subunits ( $G_{\alpha}$  subunit, orange, PDB: 2ODE)] and the RGS protein through a designer peptide (pea green) inhibitor, **c** Targeting the PPIs involving the ACE2 receptor and the receptor-binding domain of the

SARS-CoV2 spike protein (light pink, PDB: 6M0J) through designer small peptide inhibitors (blue), **d** Illustration of the membrane disruption potential of a designer small antimicrobial peptide (CPK) in the presence of a model POPC bilayer with a focus on the antimicrobial resistance (AMR), and **e** Design of antibody-like neutralizing peptide (yellow) targeting the complement protein C5a (brick red, PDB: 1KJS) for the treatment of chronic inflammation-induced diseases

clinical trials is very encouraging (Cong et al. 2022; Kawai et al. 2020), which suggests that conformationally ordered shorter peptide agonists with better pharmacokinetic properties can be designed for targeting the GLP-1R (Fig. 15a).

Similarly, designer peptides, both optimized in sequence and size, can be utilized as competitive inhibitors targeting pharmacologically important PPIs. For example, downstream signaling of GPCRs can be modulated alternatively by targeting the regulator of G-protein signaling (RGS) proteins (Soundararajan et al. 2008), which upon binding, enhances the intrinsic GTPase activity of the  $G\alpha$ -subunit of heterotrimeric G-protein and thus, control the duration of the agonist-induced signaling cycle of the GPCRs. Targeting the PPIs involving the G-protein and RGS protein through designer peptides (Fig. 15b) deliverable to the cytosolic environment of the cells has the potential to reduce the therapeutic dose size and the off-target actions of certain GPCR agonists. In addition, such peptides can also enhance the natural signaling output of the GPCRs challenged by the poor level of natural agonists under certain disease conditions like neuronal disorder and depression (Squires et al. 2018). A similar approach can also be followed for blocking the PPIs (Fig. 15c) of the SARS-CoV2 spike protein (Lan et al. 2020) with the host cell angiotensin-converting enzyme-2 (ACE2) receptor for minimizing the onset of severe COVID19 in desired populations. It is noteworthy that most of the peptide mimetics reported for blocking the interaction of ACE2 (Pomplun 2020) with the spike protein of the SARS-CoV2 are in the range of 25–65 amino acids, and many of them do not harbor NCAAs. Thus, there is a huge potential for exploring sequence-optimized conformationally ordered shorter peptides involving NCAAs for therapeutic applications.

Besides planning to control the current pandemic, it is also essential to stay prepared for the future health problems that are going to surface in the post-COVID19 world. Antimicrobial resistance (AMR) may aggravate further both during the pandemic and also beyond the pandemic, a major concern that requires urgent attention (Dadgostar 2019; Upert et al. 2021), as there is a dearth of new-age antimicrobials (AMP) in the developmental pipeline. Further, AMPs are preferred (Magana et al. 2020) over conventional antibiotics, as they can modulate the immune response appropriately, and they have better antibiofilm activity, including slower emergence of resistance. The perceived resistance to proteolysis, thermostability, and the ability to induce the desired conformation in short peptides suggest that careful inclusion of the NCAAs into the designer peptides (Fig. 15d) optimized in sequence and size can provide the required alternatives to the naturally secreted antimicrobial peptides (Huan et al. 2020) with low cytotoxicity and hemolytic activity both in vitro and in vivo settings. In fact, AMPs are an essential part of the immune system,

which are involved in controlling the signaling activities of proinflammatory cytokines in the eukaryotes. It is noteworthy that in response to stress, injury, and exposure to the contagions like the common influenza viruses (H5N1, H7N9), and the severe acute respiratory syndrome (SARS) coronavirus, an overwhelmed immune system can aberrantly upregulate the concentration of the complement fragment 5a (C5a), (Zhang et al. 1997) an extremely potent proinflammatory glycoprotein of the complement system, augmenting the lethal “cytokine storm” that exacerbate several chronic inflammations induced pathological conditions, such as sepsis, allergic asthma, acute lung injury, including the moderate to severe COVID19 in humans. Indeed, C5a has been found to be significantly elevated to several folds in the plasma of moderate to severe COVID19 patients (Carvelli et al. 2020) compared to the healthy subjects. Thus, C5a is considered to be a major pharmacological target, and no wonder that ~20 drug candidates targeting the various stages of the complement signaling are currently under development by several pharma majors, and most of the drugs in the process of development are antibodies. In fact, Eculizumab (Soliris; Alexion Pharmaceuticals) and IFX-1 (InflaRx) are the two FDA-approved antibodies currently available in the market that respectively targets C5, the precursor protein of C5a, or the C5a directly. C5a being an important protein of the immune system, completely shutting it down through antibodies is physiologically undesired, and thus, we have been actively pursuing the avenue of repurposing the FDA-approved drugs as potential “neutraligands” of C5a (Das and Rana 2021; Mishra and Rana 2019; Mishra et al. 2022) for therapeutically modulating the pathological stimulation of C5a-C5aR1 signaling axes. Nevertheless, antisense peptides (Fujita et al. 2004) targeting the C5a have been explored earlier, and thus, in the context of the PPIs involving C5a and C5aR1 (Das et al. 2021; Sahoo et al. 2018; Rana and Sahoo 2015), the excessive C5a produced in the body can also be neutralized with designer peptides (Fig. 15e), mimicking the function of the antibodies for maintaining the desired balance between the pharmacological and physiological concentration of C5a in the plasma for the therapeutic benefits. The examples illustrated in Fig. 15 are the snapshots of the currently ongoing studies involving small designer peptides of unusual architecture involving NCAAs, which are likely to be elaborated further in subsequent future studies through experimental validation in appropriate biological systems.

## Conclusions

Peptides, the molecular fragments of the proteins, are considered the ideal biological probes for understanding the structure–function relationships behind the complex protein–protein interactions of several biological processes.

The chemical versatility available to peptides makes them one of the go-to scaffolds for the rational design of targeted therapeutics involving both extracellular, transmembrane, and intracellular proteins. From the humble beginning of the synthesis of glycylglycine dipeptide to the marketing of the synthetic peptide lypressin as antidiuretics, peptide chemistry has rapidly advanced so far over the last 50 years (Sachdeva 2016). Though the synthesis of peptides has come a long way, peptides as therapeutics have yet to capture a sizeable market share compared to other therapeutics. Focus on the native or the analogs of the native peptide only could be one of the possible reasons behind this. Systematic expansion of the chemical space in peptides within the boundary of conformational space is possible with de novo design of small heterologous peptides, involving a strategic mix of both coded and NCAs, which can be exploited further for therapeutic applications. The above approach is exemplified in the current study with a set of selected C<sub>α</sub>-substituted aromatic NCAs in a model helical system, which can be easily expanded to any other systems by systematically screening the repertoire of synthetically available NCAs for subsequent therapeutic explorations.

## Materials and methods

### Peptide modeling and Molecular dynamics (MD) simulation

The Ala-Gly-Ala tripeptide end-capped both at the *N*-terminus (-COCH<sub>3</sub>) and the *C*-terminus (-NH<sub>2</sub>) were initially modeled in an extended conformation ( $\phi = -120^\circ$ ,  $\psi = 120^\circ$ ) using the Discovery studio (Biovia). Further, the structure of Bpa, Ing, and Aic was geometrically optimized in the Discovery studio and subsequently incorporated into the Ac-Ala-Gly-Ala-NH<sub>2</sub> tripeptide template to respectively prepare the Ac-Ala-Bpa-Ala-NH<sub>2</sub>, Ac-Ala-Ing-Ala-NH<sub>2</sub>, and Ac-Ala-Aic-Ala-NH<sub>2</sub> tripeptides. The Ac-Ala-Ala-Ala-NH<sub>2</sub>, Ac-Ala-Aib-Ala-NH<sub>2</sub>, and Ac-Ala-Pro-Ala-NH<sub>2</sub> tripeptides were also designed similarly to serve as the reference. The tripeptides were energy minimized in a cubic box under periodic boundary conditions both in the presence of vacuum and explicit simple point charge (SPC) water with density set to the value corresponding to 1 atm. Subsequently, the system was subjected to equilibration for 100 ps each at 300 K under NVT and NPT conditions prior to initiating the production molecular dynamics (MD) run over 10 ns by recruiting the GROMACS software package (Hess et al. 2008). The NCAs were incorporated into the GROMACS force field by involving a fragment built-up approach to make the system work. Subsequently, a template 18-mer synthetic peptide code named as “Hed1” was modeled as an end-capped (*N*-terminus: -COCH<sub>3</sub>; *C*-terminus: -NH<sub>2</sub>)

$\alpha$ -helix ( $\phi = -47^\circ$ ,  $\psi = -57^\circ$ ) using the Discovery studio, by relying on the reported CD spectroscopy data. The “Hed1” peptide was further used as a template for generating the other (Hed2, Hed3, Hed4, and Hed5) model helices harboring one of the NCAs at a strategic location. As described above, all the model helices were subjected to comparative MD studies over 100 ns each at 300 K. The MD trajectories were thoroughly analyzed as desired using the modules built into GROMACS. Data were plotted in GraphPad Prism (version 9 for Windows, GraphPad Software, La Jolla California USA, [www.graphpad.com](http://www.graphpad.com)). Both PyMOL (The PyMOL Molecular Graphics System, Version 2.5.2, Schrödinger, LLC) and Discovery studio (BIOVIA) software were used to process, visualize, analyze, and present the peptide structures.

### Peptide synthesis

Using the standard Fmoc chemistry over the solid phase, the designed 18-mer helical peptides (Hed1, Hed2, Hed3, Hed4, and Hed5) were synthetically prepared by recruiting the services of GenScript (NJ, USA). Except for Hed5, all peptides were acylated at the *N*-terminus and amidated at the *C*-terminus. Post cleavage, the peptides were purified, and the absolute purity ( $\geq 95\%$ ) of the peptides was judged from the analytic HPLC profile of the peptides recorded by recruiting the C18 (4.6 × 250 mm) column at 220 nm, using acetonitrile–water gradient in the presence of 0.05–0.065% trifluoroacetic acid (TFA). The ESI–MS further confirmed the identity of the peptides. All the peptides were soluble in ultrapure water and 1 × PBS (pH 7.4).

### Circular dichroism (CD) studies

The far-UV CD spectra of the peptides were recorded by the Chirascan CD spectrometer system (Applied photophysics) at 25 °C. Each sample was subjected to a minimum of three scans with a time constant of 1 s and a step size of 1 nm. The PBS buffer and Milli Q water were filtered through a 0.2 μm filter and thoroughly degassed with continuous stirring while bubbling with nitrogen gas prior to the experiment. The stock concentration of the peptides was determined using a UV spectrophotometer. The CD spectra of the peptides were recorded at a concentration of 50 μM in water (pH ~ 5.8) and methanol, including a varying percentage of cosolvents, such as trifluoroethanol (TFE), hexafluoroisopropanol (HFIP), as well as in 20–30 mM SDS micelles suspended in PBS. The CD spectra were background corrected by subtracting the blank spectra of the corresponding solvents used in the study. The thermostability of the model peptides was judged by recording the CD spectra between 4 and 90 °C in the presence of 20% HFIP with a stepping gradient of 2 °C. The sample was equilibrated for 1 min at



each temperature point before recording the ellipticity. The CD signal in millidegrees was converted to mean residue ellipticity [ $[\theta]_{\text{MRE}} \text{deg.cm}^2.\text{dmol}^{-1}$ ] using the following equation (millidegrees  $\times$  mean residual weight) / (pathlength in millimeter  $\times$  concentration in  $\text{mg ml}^{-1}$ ). The mean residual weight of a peptide is the total molecular weight divided by the total number of amides in the backbone (Greenfield 2006). All data points were plotted in GraphPad Prism (version 9), and the spectra were smoothed up to 2–3 units wherever required. The helicity percentage was calculated from  $[\theta]_{\text{MRE}}$  at 222 nm (Morrisett et al. 1973).

## Fluorescence studies

The fluorescence emission spectra of the peptides were recorded at 50  $\mu\text{M}$  concentration in the presence of various solvents and cosolvents at 25 °C using the Horiba Fluoromax spectrophotometer. The excitation and emission slit widths were set to 5 nm, with excitation wavelength set at 280 nm and emission range set between 290 and 500 nm. The fluorescence spectra presented represent an average of three scans that have been corrected for the background spectra of the respective solvents.

## Nuclear magnetic resonance (NMR) studies

The  $^1\text{H-NMR}$  of the peptides maintained at 2 mM were recorded using a Bruker AVANCE 400 MHz NMR spectrometer equipped with cryoprobe at 298 K in 99.8%  $\text{CD}_3\text{OD}$  having 0.03% v/v tetramethylsilane (TMS) as the internal reference. Each peptide was subjected to a minimum of 32 scans. In addition, the  $^1\text{H-NMR}$  of the peptides was also recorded at tenfold dilution to judge the molecular aggregation of the peptides. The raw FIDs were processed with the TOPSPIN provided by Bruker, followed by MestreNova software for presentation. Hypothetical  $\text{C}^\alpha\text{H}$  chemical shift index ( $\Delta\delta_{\text{H}\alpha}$ ) for the peptides were generated using the relation  $\Delta\delta = \delta_{\text{obs}} - \delta_{\text{coil}}$  using mean  $\delta_{\text{coil}}$  values for the given residues reported in GGXGG peptide under identical solvent conditions. (Merutka et al. 1995).

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**Author contributions** LMB conducted the biophysical experiments, and MG performed the NMR experiments. SR conducted the computational modeling and MD studies. SR wrote the manuscript and conceived the idea for the project.

## Declarations

**Conflict of interest** The authors declare no competing interests. The research reported in the manuscript does not involve human participants and/or animals and does not describe results obtained from case studies involving human participants. Thus, informed consent does not apply.

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