

Erratum

Erratum: Sanz del Olmo, N.; et al. Antioxidant and Antibacterial Properties of Carbosilane Dendrimers Functionalized with Polyphenolic Moieties. *Pharmaceutics* 2020, 12, 698

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Due to an error during production, in Section 1.1, letters that were in italics refer to the H or C atom that corresponds to the chemical displacement in NMR have not been fully displayed. The corrected part should be:

1. Materials and Methods

1.1. Synthesis and Characterization of Polyphenolic Dendrimers (1–6)

All reactions were carried out under an inert atmosphere and solvents of reactions were bought in dry conditions. NMR experiments were carried out on a Varian 500 Hz instrument. Chemical shifts (δ) are given in ppm. As a reference, deuterated methanol solvent (CD_3OD) was used. Assignment of resonances was done from HSQC, HMBC, and COSY NMR experiments. Elemental analyses were performed on a LECO CHNS-932 instrument. Mass spectra were obtained using the ESI-TOF technique from an Agilent 6210 TOF LC/MS instrument in MeOH, and ACN/ H_2O 0.1% formic acid as the mobile phase.

Spectra obtained through the different techniques for the new compounds described in this work are collected in the Supplementary Materials (Figures S1–S42).

1.1.1. Synthesis of $\text{G}_1\text{-[Si(CH}_2\text{)}_3\text{NH(CO)CH=CHCH}_2\text{Ph(OH)(OCH}_3\text{))}_4$ (1)

To achieve the synthesis of compound **1**, the activation of ferulic acid (179.0 mg, 0.922 mmol) with EDCI-HCl (176.7 mg, 0.922 mmol) and HOBt (124.6 mg, 0.922 mmol), using dry DMF as solvent for the reaction, was carried out in the first place. The mixture was stirred at room temperature for one hour. Afterwards, a DMF solution mixture of dendrimer functionalized with amine groups $\text{G}_1\text{-[Si(CH}_2\text{)}_3\text{NH}_2\text{]}_4$ (127.0 mg, 0.192 mmol) and NEt_3 (0.768 mmol) was added dropwise under stirring and at 0 °C. After five minutes in these conditions, the mixture was kept at 60 °C, overnight. The compound was purified by size exclusion chromatography in DMF, obtaining compound **1** as a brown solid (178.4 mg, 68%). $^1\text{H-NMR}$ (CD_3OD): δ (ppm) 0.00 (s, 24H, $\text{NHCH}_2\text{CH}_2\text{CH}_2\text{Si(CH}_3\text{)}_2$), 0.51–0.67 (m, overlapping of signals, 24H, $\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $\text{NHCH}_2\text{CH}_2\text{CH}_2\text{Si(CH}_3\text{)}_2$),

1.39 (m, 8H, SiCH₂CH₂CH₂Si), 1.57 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)₂), 3.29 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)₂), 3.88 (s, 12H, OCH₃), 6.48 (d, 4H, ³J_(H-H) = 15.7 Hz, PhCH=CH(CO)NH), 6.81 (dd, 4H, ³J_(H-H) = 8.2 Hz, ⁵J_(H-H) = 1.8 Hz, 1H_{Ar}, *ortho*-OH), 7.04 (d, 4H, ³J_(H-H) = 8.2 Hz, 1H_{Ar}, *para*-OCH₃), 7.12 (d, 4H, ⁵J_(H-H) = 1.8 Hz, 1H_{Ar}, *ortho*-OCH₃), 7.47 (d, 4H, ³J_(H-H) = 15.7 Hz, PhCH=CH(CO)NH). ¹³C-NMR (CD₃OD): δ (ppm) −3.1 ((CH₃)₂SiCH₂CH₂CH₂NH), 13.6 ((CH₃)₂SiCH₂CH₂CH₂NH), 18.6, 19.9, 21.1 (SiCH₂CH₂CH₂Si), 25.2 ((CH₃)₂SiCH₂CH₂CH₂NH), 43.9 ((CH₃)₂SiCH₂CH₂CH₂NH), 56.4 (OCH₃), 111.6 (C_{Ar}, *ortho*-OCH₃), 116.5 (C_{Ar}, *ortho*-OH), 118.9 (C_{Ar}, PhCH=CH(CO)NH), 123.2, (C_{Ar}, *para*-OCH₃), 128.3 (C_{ipso}, *para*-OH), 142.0 (PhCH=CH(CO)NH), 149.3 (C_{ipso}), 149.8 (C_{ipso}), 169.1 (NHC=O). {¹H-¹⁵N}-HMBC-NMR (CD₃OD): δ (ppm) −259.4 (NHC=O). MS: [M + H]⁺ = 1366.7266 u (Calc. 1366.7264 u), [M + Na]⁺ = 1388.7061 u (Calc. 1388.7063 u). Elemental Analysis (%): Calc for C₇₂H₁₁₂N₄O₁₂Si₅ (1366.13 g/mol). C, 63.30; H, 8.26; N, 4.10. Exp.: C, 62.9; H, 8.26; N, 5.05.

1.1.2. Synthesis of G₁-[Si(CH₂)₃NH(CO)CH=CHCH₂Ph(OH)₂]₄ (2)

Dendrimer **2** was prepared through the same method as described for **1** by using the following reagents: caffeic acid (255.1 mg, 1.416 mmol), EDCI·HCl (271.4 mg, 1.416 mmol), HOBT (191.3 mg, 1.416 mmol), G₁-[Si(CH₂)₃NH₂]₂ (195.3 mg, 0.295 mmol) and NEt₃ (1.180 mmol). Compound **2** was obtained as a brown solid (231.9 mg, 61%). ¹H-NMR (CD₃OD): δ (ppm) −0.04 (s, 24H, NHCH₂CH₂CH₂Si(CH₃)₂), 0.48–0.67 (m, overlapping of signals, 24H, SiCH₂CH₂CH₂Si and NHCH₂CH₂CH₂Si(CH₃)₂), 1.36 (m, 8H, SiCH₂CH₂CH₂Si), 1.53 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)), 3.24 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)₂), 6.37 (d, 4H, ³J_(H-H) = 15.7 Hz, PhCH=CH(CO)NH), 6.75 (d, 4H, ³J_(H-H) = 8.2 Hz, 1H_{Ar}, *meta*-CH=CH), 6.89 (dd, 4H, ³J_(H-H) = 8.2 Hz, ⁵J_(H-H) = 2.0 Hz, 1H_{Ar}, *ortho*-CH=CH), 7.00 (d, 4H, ⁵J_(H-H) = 2.0 Hz, 1H_{Ar}, *ortho*-CH=CH, *ortho*-OH), 7.38 (d, 4H, ³J_(H-H) = 15.7 Hz, PhCH=CH(CO)NH). ¹³C-NMR (CD₃OD): δ (ppm) −3.1 ((CH₃)₂SiCH₂CH₂CH₂NH), 13.6 ((CH₃)₂SiCH₂CH₂CH₂NH), 18.5, 19.9, 21.0 (SiCH₂CH₂CH₂Si), 25.2 ((CH₃)₂SiCH₂CH₂CH₂NH), 43.9 ((CH₃)₂SiCH₂CH₂CH₂NH), 115.1 (C_{Ar}, *ortho*-OH, *ortho*-CH=CH), 116.5 (C_{Ar}, *meta*-CH=CH), 118.5 (PhCH=CH(CO)NH), 122.1, (C_{Ar}, *ortho*-CH=CH), 128.3 (C_{ipso}, *meta*-OH, *para*-OH), 142.1 (PhCH=CH(CO)NH), 146.7 (C_{ipso}), 148.7 (C_{ipso}), 169.2 (NHC=O). {¹H-¹⁵N}-HMBC-NMR (CD₃OD): δ (ppm) −259.5 (NHC=O). MS: [M + H]⁺ = 1310.6618 u (Calc. 1310.6670 u), [M + Na]⁺ = 1332.6431 u (Calc. 1332.6433 u). Elemental Analysis (%): Calc for C₆₈H₁₀₄N₄O₁₂Si₅ (1310.02 g/mol). C, 62.35; H, 8.00; N, 4.28. Exp.: C, 60.23; H, 7.52; N, 4.50.

1.1.3. Synthesis of G₁-[Si(CH₂)₃NH(CO)Ph(OH)₃]₄ (3)

Dendrimer **3** was prepared through the same method as described for **1** by using the following reagents: gallic acid (252.9 mg, 1.344 mmol), EDCI·HCl (257.6 mg, 1.344 mmol), HOBT (181.6 mg, 1.344 mmol), G₁-[Si(CH₂)₃NH₂]₄ (185.0 mg, 0.280 mmol) and NEt₃ (1.120 mmol). Compound **3** was obtained as a brown solid (248.9 mg, 70%). ¹H-NMR (CD₃OD): δ (ppm) 0.00 (s, 24H, NHCH₂CH₂CH₂Si(CH₃)₂), 0.46–0.71 (m, overlapping of signals, 24H, SiCH₂CH₂CH₂Si and NHCH₂CH₂CH₂Si(CH₃)₂), 1.40 (m, 8H, SiCH₂CH₂CH₂Si), 1.59 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)), 3.31 (m, 8H, NHCH₂CH₂CH₂Si(CH₃)₂), 6.88 (s, 8H, H_{Ar}), 7.07–7.32 (m, overlapping of signals, 16H, NH and Ph(OH)). ¹³C-NMR (CD₃OD): δ (ppm) −3.1 ((CH₃)₂SiCH₂CH₂CH₂NH), 13.6 ((CH₃)₂SiCH₂CH₂CH₂NH), 18.6, 19.9, 21.1 (SiCH₂CH₂CH₂Si), 25.2 ((CH₃)₂SiCH₂CH₂CH₂NH), 44.3 ((CH₃)₂SiCH₂CH₂CH₂NH), 107.8 (C_{Ar}, *ortho*-OH), 126.4 (C_{ipso}, *para*-OH), 146.6 (C_{ipso}), 170.5 (NHC=O). {¹H-¹⁵N}-HMBC-NMR (CD₃OD): δ (ppm) −267.0 (NHC=O). MS: [M + H]⁺ = 1269.5762 u (Calc. 1269.5740 u). Elemental Analysis (%): Calc for C₆₀H₉₆N₄O₁₆Si₅ (1269.87 g/mol). C, 56.75; H, 7.62; N, 4.41. Exp.: C, 56.60; H, 7.38; N, 4.95.

1.1.4. Synthesis of G₂-[Si(CH₂)₃NH(CO)CH=CHCH₂Ph(OH)(OCH₃)]₈ (4)

Dendrimer **4** was prepared through the same method as described for **1** by using the following reagents: ferulic acid (259.5 mg, 1.336 mmol), EDCI·HCl (255.6 mg, 1.336 mmol), HOBT (180.6 mg, 1.366 mmol), G₂-[Si(CH₂)₃NH₂]₈ (166.0 mg, 0.139 mmol) and NEt₃

(1.112 mmol). Compound **4** was obtained as a brown solid (296.3 mg, 70%). $^1\text{H-NMR}$ (CD_3OD): δ (ppm) -0.03 (s, 12H, $\text{CH}_3(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 0.00 (s, 48H, $-(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 0.48 – 0.70 (m, overlapping of signals, 64H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$, $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$ and $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 1.30 – 1.47 , (m, 24H, overlapping of signals, $\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 1.52 – 1.62 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 3.29 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 3.86 (s, 24H, OCH_3), 6.48 (d, 8H, $^3J_{(\text{H-H})} = 16.2$ Hz, $\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 6.80 (d, 8H, $^3J_{(\text{H-H})} = 7.9$ Hz, 1H_{Ar} , *ortho*-OH), 7.02 (d, 8H, $^3J_{(\text{H-H})} = 7.9$ Hz, 1H_{Ar} , *para*- OCH_3), 7.10 (s, 8H, 1H_{Ar} , *ortho*- OCH_3), 7.47 (d, 8H, $^3J_{(\text{H-H})} = 16.2$ Hz, $\text{PhCH}=\text{CH}(\text{CO})\text{NH}$). $^{13}\text{C-NMR}$ (CD_3OD): δ (ppm) -4.1 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), -2.8 ($\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 13.7 ($-(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 19.8 , 20.0 , 20.1 , 21.1 ($\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $-\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 25.3 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 44.0 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 56.4 ($-\text{OCH}_3$), 111.6 (C_{Ar} , *ortho*- OCH_3), 116.5 (C_{Ar} , *ortho*-OH), 119.0 (C_{Ar} , $\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 123.2 , (C_{Ar} , *para*- OCH_3), 128.3 (C_{ipso} , *para*-OH), 142.0 ($\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 149.2 (C_{ipso}), 149.8 (C_{ipso}), 169.0 ($\text{NHC}=\text{O}$). $\{^1\text{H-}^{15}\text{N}\}$ -HMBC-NMR (CD_3OD): δ (ppm) -259.6 ($\text{NHC}=\text{O}$). Elemental Analysis (%): Calc for $\text{C}_{160}\text{H}_{260}\text{N}_8\text{O}_{24}\text{Si}_{13}$ (3044.98 g/mol). C, 63.11; H, 8.61; N, 3.68. Exp.: C, 64.56; H, 9.01; N, 3.96.

1.1.5. Synthesis of G_2 -[Si(CH₂)₃NH(CO)CH=CHCH₂Ph(OH)₂]₈ (**5**)

Dendrimer **5** was prepared through the same method as described for **1** by using the following reagents: caffeic acid (300.1 mg, 1.666 mmol), EDCI·HCl (318.7 mg, 1.666 mmol), HOBt (225.1 mg, 1.666 mmol), G_2 -[Si(CH₂)₃NH₂]₈ (207.0 mg, 0.174 mmol) and NEt_3 (1.392 mmol). Compound **5** was obtained as a brown solid (218.6 mg, 65%). $^1\text{H-NMR}$ (CD_3OD): δ (ppm) -0.04 (s, 12H, $\text{CH}_3(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 0.00 (s, 48H, $-(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 0.47 – 0.69 (m, overlapping of signals, 64H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$, $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$ and $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 1.34 – 1.46 , (m, 24H, overlapping of signals, $\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 1.49 – 1.62 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 3.27 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 6.40 (d, 8H, $^3J_{(\text{H-H})} = 15.7$ Hz, $\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 6.77 (d, 8H, $^3J_{(\text{H-H})} = 8.1$ Hz, 1H_{Ar} , *meta*-CH=CH), 6.91 (d, 8H, $^3J_{(\text{H-H})} = 8.1$ Hz, 1H_{Ar} , *ortho*-CH=CH), 7.03 (s, 8H, 1H_{Ar} , *ortho*-CH=CH, *ortho*-OH), 7.41 (d, 8H, $^3J_{(\text{H-H})} = 15.7$ Hz, $\text{PhCH}=\text{CH}(\text{CO})\text{NH}$). $^{13}\text{C-NMR}$ (CD_3OD): δ (ppm) -4.2 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), -2.9 ($\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 13.7 ($-(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 19.8 , 20.0 , 20.1 , 21.1 ($\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $-\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 25.2 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 44.0 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 115.2 (C_{Ar} , *ortho*-OH, *ortho*-CH=CH), 116.5 (C_{Ar} , *meta*-CH=CH), 118.6 ($\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 122.1 , (C_{Ar} , *ortho*-CH=CH), 128.4 (C_{ipso} , *meta*-OH, *para*-OH), 142.1 ($\text{PhCH}=\text{CH}(\text{CO})\text{NH}$), 146.7 (C_{ipso}), 148.7 (C_{ipso}), 169.2 ($\text{NHC}=\text{O}$). $\{^1\text{H-}^{15}\text{N}\}$ -HMBC-NMR (CD_3OD): δ (ppm) -259.6 ($\text{NHC}=\text{O}$). Elemental Analysis (%): Calc for $\text{C}_{152}\text{H}_{244}\text{N}_8\text{O}_{24}\text{Si}_{13}$ (2932.76 g/mol). C, 62.25; H, 8.39; N, 3.82. Exp.: C, 60.60; H, 8.49; N, 4.89.

1.1.6. Synthesis of G_2 -[Si(CH₂)₃NH(CO)Ph(OH)₃]₈ (**6**)

Dendrimer **6** was prepared through the same method as described for **1** by using the following reagents: gallic acid (385.4 mg, 2.049 mmol), EDCI·HCl (391.9 mg, 2.049 mmol), HOBt (272.7 mg, 2.049 mmol), G_2 -[Si(CH₂)₃NH₂]₈ (352.0 mg, 0.213 mmol) and NEt_3 (1.704 mmol). Compound **6** was obtained as a brown solid (455.7 mg, 75%). $^1\text{H-NMR}$ (CD_3OD): δ (ppm) -0.07 (s, 12H, $\text{CH}_3(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), -0.03 (s, 48H, $-(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 0.45 – 0.68 (m, overlapping of signals, 64H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$, $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$ and $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 1.31 – 1.44 , (m, 24H, overlapping of signals, $\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 1.51 – 1.65 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 3.27 (m, 16H, $(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 6.85 (s, 16H), 7.03 (broad s, overlapping of signals, 32H, NH and Ph(OH)). $^{13}\text{C-NMR}$ (CD_3OD): δ (ppm). -4.3 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), -2.9 ($\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 13.7 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 19.8 , 19.9 , 20.1 , 20.2 ($\text{SiCH}_2\text{CH}_2\text{CH}_2\text{Si}$ and $-\text{CH}_3\text{Si}(\text{CH}_2\text{CH}_2\text{CH}_2\text{Si})_2$), 25.3 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 44.3 ($(\text{CH}_3)_2\text{SiCH}_2\text{CH}_2\text{CH}_2\text{NH}$), 107.8 (C_{Ar} , *ortho*-OH), 126.4 (C_{ipso} , *para*-OH), 146.6 (C_{ipso} , *ortho*-OH), 170.4 ($\text{NHC}=\text{O}$). $\{^1\text{H-}^{15}\text{N}\}$ -HMBC-NMR (CD_3OD): δ (ppm) -266.6 ($\text{NHC}=\text{O}$). Elemental Analysis (%):

Calc for $C_{136}H_{228}N_8O_{32}Si_{13}$ (2852.45 g/mol). C, 57.27; H, 8.06; N, 3.93. Exp.: C, 56.77; H, 8.98; N, 4.79.

The article will be updated and the original [1] will remain on the webpage.

Supplementary Materials: The following are available online at <https://www.mdpi.com/1999-4923/13/1/121/s1>, Figure S1. Mass Spectrometry (ESI-TOF) of dendritic polyphenol (1). Figure S2. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S3. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S4. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S5. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S6. $\{^1H-^1H\}$ -COSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S7. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S8. $\{^1H-^{13}C\}$ -HMBC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (1). Figure S9. Mass Spectrometry (ESI-TOF) of dendritic polyphenol (2). Figure S10. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S11. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S12. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S13. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S14. $\{^1H-^1H\}$ -COSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S15. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (2). Figure S16. Mass Spectrometry (ESI-TOF) of dendritic polyphenol (3). Figure S17. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S18. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S19. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S20. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S21. $\{^1H-^1H\}$ -COSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S22. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (3). Figure S23. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S24. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S25. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S26. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S27. $\{^1H-^1H\}$ -COSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S28. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S29. $\{^1H-^{13}C\}$ -HMBC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (4). Figure S30. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S31. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S32. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S33. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S34. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S35. $\{^1H-^{13}C\}$ -HMBC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (5). Figure S36. 1H -NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S37. ^{13}C -NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S38. $\{^1H-^{15}N\}$ -HMBC-NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S39. 1H -DOSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S40. $\{^1H-^1H\}$ -COSY-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S41. $\{^1H-^{13}C\}$ -HSQC-2D-NMR (500 MHz, CD₃OD) of dendritic polyphenol (6). Figure S42. (A) A representative calibration curve of inhibition of DPPH by Trolox standards. Representative results of at least three independent experiments are shown. (B) Graphics with equations line for compound G1-[Si(CH₂)₃NH(CO)Ph(OH)₃]₄ (3).

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