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New Micropeptins with Anti-Neuroinflammatory Activity Isolated from a Cyanobacterial Bloom

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INTRODUCTION

Neuropathic pain, or nerve pain, resulting from damage to or dysfunction of the somatosensory nervous system, is notoriously difficult to treat. Neuroinflammation is key in the initiation and persistence of neuropathic pain.¹ Recent research has shown that the prevalence of neuropathic pain in the United States is approximately 10% of adults, and out of those reporting neuropathic pain, over 40% were treated with either weak or strong opioid drugs.² This prevalence of neuropathic pain increases in specific cohorts with 33% of those with chronic back pain displaying characteristic signs of neuropathic pain.³ Additionally, chronic, uncontrolled neuroinflammation is an indicative sign of neurodegenerative diseases such as Alzheimer's disease and Parkinson's disease.^{4,5} Our group is interested in the discovery of new specialized metabolites from marine and aquatic microbiomes that are effective in reducing neuroinflammation. While we have had some success in this approach by mining marine cyanobacterial blooms for antineuroinflammatory agents,⁶ freshwater cyanobacterial blooms offer additional chemical space for discovery, especially in terms of cyclic peptides.⁷ These peptides belong to several classes including the microcystins, aeruginosins, microginins, anabaenopeptins, micropeptins, microviridins, and cyclamides. More than 600 peptides and peptidic compounds have been described from various cyanobacterial taxa.7 Many of these peptides are hepatotoxic, such as the microcystins, which can cause non-alcoholic liver disease and acute liver failure.^{8,9} Consumption of contaminated drinking water is considered the major route of human exposure to microcystins, and the EPA has set concentration limits on these toxins (1.6 μ g/L =

natural product libraries for therapeutic hit and lead identification.

Do Not Drink; 20 μ g/L = Do Not Use). While neurotoxins and hepatotoxins are a significant public health concern, these blooms are comprised of a complex milieu of taxonomically diverse microbes, each possessing various biosynthetic capabilities. Large-scale biomass sampling from groups like Biosortia has revealed that freshwater cyanobacterial blooms are more chemically complex than previously assumed, and novel molecules with potent biological activities can be isolated and characterized.^{10,11} Work on this "metabolite mining" from complex aquatic microbiomes envisions the bloom environment as an ecosystem level complex co-culture experiment in which the aquatic microbiome produces an equally complex suite of secondary metabolites.

The use of prefractionated natural product libraries has shown promise in increasing the potential of identifying lead structures following a bioassay-guided discovery approach.¹² Biosortia Pharmaceuticals has created a prefractionated library from cyanobacterial biomass dominated by *Microcystis aeruginosa* originally isolated from a wastewater management facility in Muskegon, Michigan. Previous mining of this material resulted in the isolation and structure elucidation of new, highly cytotoxic microcystin toxins.¹⁰ A similar workflow

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Figure 1. Structures of micropeptins 982 (1) and 957 (2).

was implemented for biomass collected from Grand Lake St. Mary's in Ohio, and this effort resulted in the identification and characterization of the exquisitely potent cyanobufalin toxins.¹¹ Each chromatography fraction in the library contained a rarefied mixture of chemicals allowing hits to be identified from less complex starting materials compared to crude extracts or more complex fractions. Pairing the prefractionated library with metabolite profiling tools such as liquid chromatography tandem mass spectrometry (LC-MS/ MS)-based molecular networking afforded the opportunity for rapid annotation and dereplication.¹³ Implementing these approaches allowed our group to isolate and characterize two new micropeptins (1 and 2) and the previously discovered micropeptin 996 (Figures 1 and S1). All three metabolites showed potent inhibition of the inflammatory response in BV-2 murine microglial cells.

RESULTS AND DISCUSSION

Structure Elucidation of Micropeptins 996, 1, and 2. Biosortia's chromatography fraction library consisted of over 150 individual samples, each with a rarefied collection of microbial metabolites. Following a bioassay-guided isolation approach, we prioritized fractions with low cytotoxicity and strong anti-inflammatory effects in the BV-2 murine microglial cell line. Fraction D29 showed no significant cytotoxicity and inhibited the production of nitric oxide species (NOS) in the cells by greater than 50% at a 10 μ g/mL dose (Figures S2 and S3).

While we were using a prefractionated library for screening. the samples were still relatively complex, likely due to the complex consortia of bacteria and other microbes in the original biomass. To rapidly annotate and visualize the chemical space in the fraction of interest (D29), we subjected the fraction to LC-MS/MS analysis followed by molecular networking. The resultant network showed five separate clusters with clusters 1 and 2 showing 15 and 9 individual nodes, respectively, and parent masses between m/z 900–1100 (Figure S4). Comparing the precursor m/z values and previous molecular networks available in the literature, we putatively identified the metabolites in clusters 1 and 2 as micropeptins or cyanopeptolins.^{14,15} Node 1019 and 1005 showed high ion counts in mass spectrometry analysis and were abundant in the high-performance liquid chromatography (HPLC) profile, and these metabolites were targeted for isolation (cf. Figures 2, S4, and S5).

Following the examination of the molecular networks, additional chromatographic procedures and testing led to the isolation of three metabolites (Figures S5 and S6). ¹H NMR and ¹³C NMR analyses indicated that these molecules were peptidic in nature and that they were closely related to each



Figure 2. Molecular networking cluster showing micropeptin 982 (1) and micropeptin 996 as nodes 1005 $(M + Na^+)$ and 1019 $(M + Na^+)$, respectively. These nodes are enlarged and in yellow.

other in chemical structure (cf. Figures S7, S8, S11, S12, and S20).

The first metabolite of interest (Fraction D29, peak 7) gave an m/z of 1019.4854 $[M + Na]^+$ following high-resolution electrospray ionization mass spectrometry (HRESIMS) analysis, suggesting a molecular formula of $C_{52}H_{68}N_8O_{12}$ (Figure S9). Dereplication efforts based on mass spectrometry data (Figures S9 and S10) identified micropeptin 996 as the likely metabolite and further examination of ¹H NMR and ¹³C NMR spectra confirmed the identification as these data matched previously published data exactly.¹⁶

The second metabolite (1) (Fraction D29, peak 6.1) gave an m/z of 1005.4698 [M + Na]⁺ following HRESIMS analysis, suggesting a molecular formula of $C_{51}H_{66}N_8O_{12}$ and 23° of unsaturation (Figure S18). Dereplication efforts did not readily identify a known micropeptin with this m/z value. The molecule was 14 Da less than micropeptin 996 and the ¹H NMR and ¹³C NMR spectra of 1 and micropeptin 996 were nearly identical, and the presumption was a difference of a CH₂ group between the two compounds. Examination of twodimensional (2D) NMR data including HSQC, HMBC, COSY, TOCSY, and NOESY spectra (Table 1 and Figures S1 and S13-S17) established spin systems and connections for the following sequence of amino acids: valine, N-methyl phenylalanine, phenylalanine, amino hydroxy piperidone (Ahp), tyrosine, threonine, glutamine, and butyric acid, the difference being the substitution of a tyrosine in 1 for the homotyrosine in micropeptin 996. The planar structure of 1 was also supported by MS/MS fragmentation data, which identified a key fragment (m/z 404.1832) as Ahp-Phe-N-MePhe-H₂O (Figure S19).¹⁶ The relative configuration of the Ahp residue was assigned based on NOE correlations between

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Table 1. NMR Data for Micropeptin 982 (1) (500 MHz for ¹H NMR, 125 MHz for ¹³C NMR, DMSO-d₆)

position	$\delta_{\rm C}$, mult	$\delta_{\rm H^{\prime}}$ mult, J (Hz)	HMBC	COSY	position	$\delta_{\rm C}$ mult	$\delta_{\rm H^{\prime}}$ mult, J (Hz)	HMBC	COSY
Val-1	171.9, C				4b		1.50, m		3a, 4a, 5
2	55.8, CH	4.65, dd (9.6, 4.5)	1	3	5	73.7, CH	5.02, ovlp		4a, 4b, Ahp- OH
3	30.8, CH	2.00, m		4, 5	NH		7.04, d (9.0)	Tyr-1	2
4	19.1, CH ₃	0.84, d (6.8)		3, 5	OH Tvr-1	169.4 C	6.03, br		5
5	17.2, CH ₃	0.70, d (6.8)		3, 4	2	53.8, CH	4.30, m		3a, 3b, Tyr- NH
NH N-MePhe-1	168.8. C	7.42 (ovlp)		2	3a	34.9, CH2	3.10, dd (14.3, 4.2)	2, 4, 5, 9	2, 3b
2	60.5. CH	5.01. ovlp		3a, 3b	3b	0112	2.53. m	2, 4, 5, 9	2. 3a
- 3a	33.8,	3.20, m		2, 3b	4	128.1, C		_, ,, e, ,	_) 0.0
3h	CH ₂	2.84 ovlp		2.32	5/9	129.7, CH	6.90, d (8.4)	6, 7, 8	6, 8
4	137.7. C	2.01, 011p		2, 54	6/8	114.9.	6.56. d (8.4)	4, 5, 7, 9	5, 9
5/9	129.7,	7.23, d (7.3)	6, 7, 8	6, 8	-, -	CH		., _, ., .	-, -
	CH	, , , ,	, ,	,	7	155.5, C			
6/8	129.3,	7.39, m	4, 5, 9	5, 9	NH		8.52, d (8.9)	Thr-1	1
_	СН		6.0	6.0	Thr-1	168.7, C			
7	126.3, CH	7.30, t (7.4)	6, 8	6, 8	2	54.0, CH	4.53, d (9.6)	1	Thr-NH
N-Me	30.3.	2.78. s	2. Phe-1		3	72.0, CH	5.37, m	4, Val-1	4
Dho 1	CH ₃	2.70, 8	2, 1110 1		4	17.5, CH ₃	1.10, d (6.5)	2, 3	3
Phe-1	170.2, C	4.72 11 (11 (2 21	NH		7.46, d (9.5)	Gln-1	2
L	50.0, CH	4.72, dd (11.0, 4.4)	1	3a, 3D	Gln-1	171.7, C			
3a	35.1, CH ₂	2.83, ovlp	2	2, 3b	2	52.6, CH	4.28, ovlp		3a, 3b, Gln- NH
3b		1.66, ovlp		2, 3a	3a	27.6, CH ₂	1.86, m	2, 5	2, 3b, 4
4	136.S, C		_	6.0	3b		1.67, m	2, 5	3a, 4
5/9	129.2, CH	6.//, m		6, 8	4	31.5,	2.10, ovlp	2, 3, 5	3a, 3b
6/8	127.7, CH	7.17, d (7.4)	4, 5, 9	5, 9	5	СН ₂ 173.7, С			
7	126.2, CH	7.13, d (7.1)	5, 9		NH		8.13, d (7.7)	BTA-1, Gln-2	
Ahp-1	168.9, C				NH ₂		7.27, s		2
2	48.8, CH	3.59, m	1	3a, 3b, Ahp-	BTA-1	172.4, C	0.11	1.2.4	2
3a	21.4,	2.40, m		2, 3b, 4a, 4b	2	37.2, CH ₂	2.11, ovip	1, 3, 4	3
a1	CH_2				3	18.7, CH	1.55, m	1, 2, 4	2
3b		1.61, m		2, 3a	4	13.7	$0.90 \pm (7.4)$	1 2 3	3
4a	29.1, CH ₂	1.68, m		3a, 4b, 5	т	CH ₃	0.20, t (7.4)	1, 2, 3	5

H-2_{Ahp}, H-4a_{Ahp}, and H-5_{Ahp}, showing that these hydrogens were coplanar. Additionally, the NH proton of the Ahp residue showed an NOE correlation to H-3a_{Ahp}, which showed an NOE correlation to the OH group attached to C-5_{Ahp} (Figure S1). This was consistent with previously published research assigning the relative configuration of the Ahp residue.^{16,17} Compound 1 was given the name micropeptin 982, which is a general convention to designate these micropeptins by molecular weight.

The third metabolite (2) (Fraction D29, peak 6.2) gave an m/z of 980.4858 [M + Na]⁺, suggesting a molecular formula of $C_{49}H_{67}N_9O_{11}$ and 21° of unsaturation (Figure S24). Examination of the ¹H NMR spectrum showed additional resonances in 2 in the region from 7.0 to 7.5 ppm and two shielded doublets (δ_H 0.88 and 0.75) that were not present in 1 (Figure S20). Examination of the ¹H NMR spectrum and ¹H-¹H 2D NMR spectra (COSY, TOCSY, NOESY, Figures S21–S23) established spin systems consistent with the following amino acids: valine, *N*-methyltryptophan, phenyl-

alanine, amino hydroxy piperidone (Ahp), valine, threonine, glutamine, and butyric acid. The sequence of amino acids was established via NOE correlations, MS/MS data, and analogy to previously characterized micropeptins. An NOE correlation between Ahp-OH ($\delta_{\rm H}$ 5.85) and H-9 of Phe ($\delta_{\rm H}$ 6.78) supported the fact that these residues were connected. The MS/MS data showed a fragment of m/z 443.1931, which supported the fragmentation of Ahp-Phe-N-MeTrp-H₂O (Figure S25). This left two positions in the peptide ring to be occupied by valines along with ester formation by threonine and the Gln-BTA side chain. The relative configuration of the Ahp residue was identical to that of 1 based on NOE correlations and 2 was given the name micropeptin 957.

The absolute configuration of the amino acids in 1 and 2 was determined by the acid hydrolysis of both compounds. Next, we treated the hydrolysates with Marfey's reagent (L-FDAA) to generate derivatives, and finally, we completed a chromatographic comparison of the derivatized hydrolysates to authentic L- and D-amino acid standards (Figures S26 and



Figure 3. Anti-inflammatory activity of micropeptin 982 (MP 982) **1**, micropeptin 957 (MP 957) **2**, and micropeptin 996 (MP 996) tested against BV-2 murine microglial cells. All data are expressed as mean \pm standard error (n = 3), and the significance was reported by analysis of variance (ANOVA) followed with Dunnett's multiple comparison procedure. Significance ($\alpha = 0.05$) as compared with LPS, $p \le 0.0002$ (***) and $p \le 0.0001$ (****).

S27). All amino acids in both 1 and 2 matched L standards, and the configuration of the Ahp residues in both compounds is proposed based on the relative configuration and comparison to previously characterized micropeptins.

Biological Activity. Each micropeptin tested (1, 2, and micropeptin 996) showed the ability to reduce NOS in BV-2 microglial cells following stimulation with bacterial LPS (Figure 3). Approximately, a 50% reduction in NOS was observed for doses of each metabolite at 10 μ M, and a significant reduction in NOS was shown at concentrations as low as 0.1 μ M. We tested each metabolite for the ability to inhibit the protease plasmin with the rationale that perhaps the anti-inflammatory activity was the result of the inhibition of this protease in the cell. The activation of plasminogen to the protease plasmin occurs in cellular inflammatory responses, and plasminogen activation has shown the ability to activate microglial cells.^{18,19} Additionally, other micropeptins have shown the ability to inhibit plasmin.²⁰ However, in vitro enzyme inhibition assays did not show inhibition when concentrations of the micropeptins tested were up to 10 μ M (Figure S28).

In this report, we have isolated and characterized two new micropeptins (1 and 2), and we have evaluated the ability of three micropeptins (1, 2, micropeptin 996) to reduce inflammation in BV-2 microglial cells. Micropeptins are a large class of cyanobacterial peptides with over 82 variants⁷ and a complicated nomenclature that includes alternative names to micropeptin such as cyanopeptolin, aeruginopeptin, anabaenopeptilide, oscillapeptilide, and oscillapeptin, among several others.⁷ However, despite the many names, the chemistry is more consistent in this group as the class is characterized by the 3-amino-6-hydroxy-2-piperidone (Ahp) residue and an ester linkage formed by the β -hydroxy group of threonine and the carboxy group of the terminal amino acid. There are four other ring amino acids (six total), which can be variable and the penultimate residue (from the point of view of biosynthesis) is N-methylated. There are two additional residues not in the ring: one an amino acid and the other a fatty acid. These characteristics are also observed in 1 and 2. Intriguingly, while these metabolites are generally thought to be sequestered to cyanobacterial metabolism, a recent

cyanopeptolin isolated from a Streptomyces olivochromogenes strain provokes thoughts as to the evolutionary history of this class of molecules.²¹ A consistent biological activity observed in this class is protease inhibition in contrast to the phosphatase inhibition observed from microcystins.²² For instance, micropeptin 996 was previously shown as a potent inhibitor of chymotrypsin with an IC₅₀ of 0.64 μ M.¹⁶ Micropeptin T2 showed potent inhibition of plasmin (IC_{50} = 0.1 μ g/mL), while micropeptin T1 showed no ability to inhibit this protease $(IC_{50} > 100 \ \mu g/mL)$.²⁰ The difference between the two metabolites was a substitution of lysine in T2 instead of the tyrosine in T1 (in the same amino acid position as tyrosine in 1). We observed no plasmin inhibition from 1, 2, or micropeptin 996, and this may be due to specific amino acid residues conferring this activity to specific micropeptins. Additional structure-activity relationship studies should prove illuminating with respect to protease inhibition. Other micropeptins such as kyanamide, symplostatin 5, and tutuilamides A-C all showed potent inhibition of elastase.17,23,24

Symplostatin 5 additionally alleviated inflammation in an epithelial lung airway model system.²² We observed a reduction in inflammation in microglial cells after treatment with the micropeptins reported in this study, and further exploration of their mechanism of action in microglial cells is warranted. For realistic therapeutic development of CNS drugs, transport through the blood-brain barrier is essential. While microcystins have been shown to rely on transportation through organic anionic transporters (OATs) to move into cells,²⁵ less is known about the same transport of micropeptins. However, it will be essential to determine the ability of micropeptins to cross the BBB to be realistic agents for reducing neuroinflammation. In addition to the fraction (D29) that was further mined in this report, there are several other fractions of interest that had the ability to reduce NOS levels in LPS-stimulated microglial cells (Figure S3). Fraction D29 was chosen for further purification because it showed the strongest reduction in NOS when tested at 10 μ g/mL and also showed no significant cytotoxic effects at the same concentration.

CONCLUSIONS

The metabolite profiling approach (LC-MS/MS molecular networking) was implemented in our workflow and we quickly identified the micropeptin class in our active fraction. This mass spectrometry-guided approach can be paired with bioassay data to identify likely bioactive molecules in samples with many analytes.²⁶ Molecular networking has been used to describe the chemical space in cyanobacterial harmful algal blooms, 27,28 and innovative approaches using this technology will continue to increase the efficiency of therapeutic discovery efforts. There are many other likely micropeptins in this active sample (Figure S4), and further isolations will allow for structure-activity relationship studies to be carried out. There were clearly potent cytotoxins as well in the prefractionated library, which may be of interest as potential chemotherapeutics or may be explored for their environmental toxicity (Figure S2). High content and information-rich mass spectrometry workflows paired with additional bioassays will be necessary to fully exploit the potential of the wealth of metabolites in the prefractionated library.

MATERIALS AND METHODS

General Experimental Procedures. Optical rotations were acquired using a Jasco P-2000 polarimeter. UV spectra were measured using a Beckman Coulter DU-800 spectrophotometer. NMR spectra were recorded using a Varian 500 MHz instrument. The chemical shifts reported for 1, 2, and micropeptin 996 were referenced to the residual solvent peaks of DMSO- d_6 ($\delta_{\rm H}$ 2.50 and $\delta_{\rm C}$ 39.5). HRESIMS analysis was performed using an AB SCIEX TripleTOF 4600 mass spectrometer with Analyst software. MS/MS data were recorded on this same instrument using the product ion function. LC-MS/MS was performed using a ThermoFisher LTQ XL mass spectrometer with an electrospray ionization (ESI) source. Semi-preparative and analytical HPLC was carried out using a Dionex UltiMate 3000 HPLC system equipped with a micro vacuum degasser, an autosampler, and a diode-array detector.

Collection and Processing of Cyanobacterial Biomass into the Prefractionated Library. Biomass was collected from a holding lagoon at the Muskegon County Wastewater Management facility on Maple Island Road, Muskegon, Michigan, and concentration of biomass was completed as previously described.¹⁰ The biomass was lyophilized and sent to AnalytiCon Discovery (Potsdam, Germany) for library fractionation. Lyophilized biomass powder (10 kg) was repeatedly extracted with methanol, and separation over HP-20 eluting in 10% MeOH steps resulted in a medium polarity fraction. Next, preparative HPLC generated the prefractionated library, which was partitioned into deep well plates for initial biological testing.

LC-MS/MS-Based Molecular Networking. Fraction D29 was subjected to LC-MS/MS analysis with a specific scan event recording MS/MS spectra in data-dependent acquisition mode on a ThermoFisher LTQ XL mass spectrometer with an electrospray ionization (ESI) source coupled to a Dionex UltiMate 3000 HPLC system equipped with a micro vacuum degasser, an autosampler, and a diode-array detector. A Kinetex 5 μ m C18 column (150 × 4.6 mm²) was used for separation of analytes. The LC method consisted of a linear gradient from 15 to 100% CH₃CN in water + 0.1% formic acid over 20 min, followed by an isocratic period at 100% CH₃CN

of 5 min. The flow rate was held constant at 0.4 mL/min. The MS spray voltage was 3.5 kV with a capillary temperature of 325 °C. For the MS/MS component, the CID isolation width was 1.0 and the collision energy was 35.0 eV. The raw data files were converted to mzXML format using MSConvert from the ProteoWizard suite (http://proteowizard.sourceforge.net/ tools.shtml). The molecular network was generated using the online platform at the Global Natural Products Social Molecular Networking website (gnps.ucsd.edu).²⁷ The data were processed by removing all MS/MS fragment ions within ± 17 Da of the precursor m/z. MS/MS spectra were filtered by choosing only the top six fragment ions in the ± 50 Da window throughout the spectrum. The precursor ion mass tolerance was set to 2.0 Da with an MS/MS fragment ion tolerance of 0.5 Da. A network was then created in which edges were filtered to have a cosine score above 0.6 and more than two matched peaks. The spectra in the network were then searched against GNPS' spectral libraries. The library spectra were filtered in the same manner as the input data. All matches kept between network spectra and library spectra were required to have a score above 0.6 and at least two matched peaks. The network was visualized using the Browser Network Visualizer tool available on the gnps website²⁹ and then imported into the program Cytoscape for additional analysis.

Isolation of 1 and 2. Fraction D29 from the prefractionated library was subjected to reversed phase semipreparative HPLC using a Kinetex 5 μ m C18 column (250 \times 10 mm²) and an isocratic method. Mobile phase: 65% CH₃CN/35% H₂O with 0.05% formic acid added to each solvent and flow rate of 3 mL/min. Seven HPLC peaks (1-7)were collected and subjected to the anti-inflammatory assay and an evaluation of purity using ¹H NMR and LC-MS. This first purification resulted in the isolation of 14 mg of micropeptin 996 (t_{R} , 26.5 min, peak 7). Peak 6 (t_{R} , 25 min) was further purified using a YMC 5 μ m Chiral ART Cellulose-SB column (250 \times 4.6 mm²) using an isocratic method. Mobile phase: 65% CH₃CN/35% H₂O with 0.05% formic acid added to each solvent and flow rate of 1 mL/min. Final purification resulted in the isolation of 12 mg of 1 (t_{R} , 5.25 min, peak 6.1) and 2 mg of 2 ($t_{\rm R}$, 10.75 min, peak 6.2). *Micropeptin 982* (1). Colorless oil; $[\alpha]_{\rm D}^{23}$ -65 (c 0.1,

Micropeptin 982 (1). Colorless oil; $[α]_D^{23}$ -65 (*c* 0.1, MeOH); UV (MeOH) $λ_{max}$ (log ε) 228 (4.8), 278 (4.1) nm; ¹H NMR (500 MHz, DMSO-*d*₆), and ¹³C NMR (125 MHz, DMSO-*d*₆), see Table 1; HRESIMS *m*/*z* 1005.4698 [M + Na]⁺ (calculated for C₅₁H₆₆N₈O₁₂Na, 1005.4698).

Micropeptin 957 (2). Colorless oil; $[\alpha]_D^{23}$ –23 (*c* 0.1, MeOH); UV (MeOH) λ_{max} (log ε) 224 (4.8), 274 (3.9) nm; ¹H NMR (500 MHz, DMSO-*d*₆), see Table S1; HRESIMS *m*/*z* 980.4858 [M + Na]⁺ (calculated for C₄₉H₆₇N₉O₁₁Na, 980.4858).

Marfey's Analysis. To determine the absolute configuration of the α -amino acids in 1 and 2, 0.5 mg of each compound was reconstituted in 0.5 mL of 6 N HCl and heated at 110 °C for 15 h. The hydrolysate was dried, reconstituted in 300 μ L of 1 M NaHCO₃ solution, and treated with 50 μ L of 1 mg/mL solution of N- α -(2,4-dinitro-5-fluorophenyl)-L-alanine amide (L-FDAA) in acetone, followed by heating at 40 °C for 1 h. Next, the mixture was cooled to room temperature and quenched with 100 μ L of 1 N HCl. The hydrolysate was dried and reconstituted in 200 μ L of CH₃CN and filtered through a 0.2 μ m filter. The hydrolysate and the L-FDAA derivatized α -amino acid standards were subjected to HPLC analysis (20% H₂O/CH₃CN + 0.1% FA to 50% H₂O/CH₃CN + 0.1% FA

over 30 min and then back to initial conditions from 31 to 36 min; Kinetex C18 column, 150 × 4.6 mm², flow rate 0.6 mL/min). The retention time (min) of the hydrolysate of 1 matched L-Val (20.4; D-Val, 23.1), L-N-MePhe (23.6; D-N-MePhe, 23.9), L-Phe (23.3; D-Phe, 25.1), L-Tyr (18.4, D-Tyr, 19.6), L-Thr (12.5; D-Thr, 15.3; L-allo-Thr, 12.7; D-allo-Thr, 13.8), and L-Gln (11.6; D-Gln, 12.0). The retention times (min) of the hydrolysate of 2 matched L-Val (20.4; D-Val, 23.1), L-N-MeTrp (21.6; D-N-MeTrp, 22.4), L-Phe (23.3; D-Phe, 25.1), L-Thr (12.5; D-Tyr, 15.3; L-allo-Thr, 12.7; D-allo-Thr, 13.8), and L-Gln (11.6; D-Gln, 12.0).

Biological Assays. BV-2 murine microglial cells were cultured as previously described.⁶ The cytotoxicity of the library fractions and pure compounds to the cells was measured using the CellTiter-Glo Luminescent Cell Viability Assay (CTG, Promega, Fitchburg, WI). BV-2 cells were treated with trypsin and seeded in a white-walled 96-well plate (Corning, Corning, NY) at 100 000 cells/mL in DMEM/F-12 media. Cells were allowed to adhere for 24 h in the incubator, followed by treatment with the fractions at 10 μ g/mL concentrations and eventually pure compounds at final concentrations of 10, 1, and 0.1 μ M. After dosing, the cells were incubated with fractions or pure compounds for 24 h. Following incubation, the cell viability was measured following the manufacturer's protocols. The CTG buffer and substrate were combined, and 100 μ L of the mixed reagent was added to each well, followed by 2 min of slow shaking on an orbital shaker and then 10 min of nonshaking equilibration time. The luminescence was measured using a SpectraMax M2 plate reader. Cell viability was then reported as a percentage of vehicle control (dimethyl sulfoxide (DMSO)). To test for antiinflammatory activity, the Griess assay was used. The Griess assay reagent kit was purchased from Promega Corporation (Fitchburg, WI). BV-2 cells were plated in a 24-well plate at 100 000 cells/mL in complete media. Cells were treated with fractions or pure compounds for 1 h at the same concentrations used for the cell viability assays. After incubation, BV-2 microglia were exposed to LPS at 1 μ g/mL for 23 h. After incubation, 50 μ L aliquots were taken from each well and transferred to a clear 96-well plate where Griess reagents were added for total nitric oxide concentration determination compared to the control wells of DMSO. Statistical significance was determined using one-way ANOVA followed by a Dunnett's post hoc test, and the analysis was performed using the program Prism (San Diego, CA).

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ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.1c02025.

NMR data for micropeptin 957 (2) (Table S1); structures of micropeptins 982 (1), 957 (2), and 996 (Figure S1); cell viability of Biosortia chromatography fractions tested against BV-2 murine microglial cells (Figure S2); ability of Biosortia chromatography fractions to reduce nitric oxide species (NOS) in BV-2 murine microglial cells (Figure S3); full LC-MS/MS molecular network (Figure S4); HPLC isolation of compounds in chromatography fraction D29 (Figure S5); final purification of compounds 1 (6.1) and 2 (6.2) from fraction D29 peak 6 (Figure S6); ¹H NMR, ¹³C NMR, HRESIMS, and MS/MS data for micropeptin 996 (Figures S7-S10); ¹H NMR, ¹³C NMR, HSQC, HMBC, COSY, TOCSY, NOESY, HRESIMS, and MS/MS data for micropeptin 982 (1) (Figures S11-S19); ¹H NMR, COSY, TOCSY, NOESY, HRESIMS, and MS/MS of micropeptin 957 (2) (Figures S20-S25); Marfey's analysis of 1 and 2 (Figures S26 and S27); and dose-response of micropeptins 996, 1, and 2 tested for plasmin inhibition (Figure S28) (PDF)

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Notes

The authors declare no competing financial interest.

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