



Advances in mercury(II)-salt-mediated cyclization reactions of unsaturated bonds

Sumana Mandal, Raju D. Chaudhari and Goutam Biswas*

Review

Open Access

Address:

Department of Chemistry, Cooch Behar Panchanan Barma University, Panchanan Nagar, Vivekananda Street, Cooch Behar – 736101, West Bengal, India

Email:

Goutam Biswas* - goutam@cbpbu.ac.in

* Corresponding author

Keywords:

carbocyclic; catalyzed; cyclization; heterocyclic; mercury(II) salts

Beilstein J. Org. Chem. **2021**, *17*, 2348–2376.

<https://doi.org/10.3762/bjoc.17.153>

Received: 24 May 2021

Accepted: 27 August 2021

Published: 09 September 2021

Associate Editor: L. Ackermann

© 2021 Mandal et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

The synthesis of complex cyclic compounds is extremely challenging for organic chemists. Many transition-metal-salt-mediated cyclizations are reported in literature. Hg(II) salts have been successfully employed in cyclizations to form complex heterocyclic and carbocyclic structures that are impossible to synthesize with other transition metal salts. In this review, we have summarized cyclization reactions that are performed with Hg(II) salts. These salts are also successfully applied in stoichiometric or catalytic amounts to form complex cyclic structures and natural products.

Introduction

The use of transition metal reagents has found considerable applications in organic synthesis [1–4] and has radically changed the realm of chemical science. It also provides a powerful tool for the construction of complex molecular frameworks [5–7]. A plethora of reviews involving transition metals such as Pd(II) [8–10], Ru(II) [11–13], Rh(III) [14–16], Mn(II) [17–19], Au(II/I) [20–22], Ag(I) [23–25] etc. in both cascade and sequential reactions have been published. Electrophilic Hg(II) salts are important reagents in organic synthesis and there is published literature establishing this fact [26–29]. However, the main drawback of Hg(II) salts, as compared to other transition metal salts, is their increased toxicity [30,31]. Hg(II) salts on the other

hand, are very cheap in comparison to other transition metal salts (Table 1) and one of the soft Lewis acids of the periodic table [32]. Hg(II) salts have already manifested some unique reactivity and therefore attracted increasing interest from chemists [33]. Many examples involving Hg(II) salts with unsaturated bonds in presence of various nucleophiles giving rise to various types of products are abound in the literature. Utilization of Hg(II) salts in the intramolecular cationic cyclization of olefinic, acetylenic, and allenic substrates having aromatic rings, nucleophiles, and heteroatoms in the neighborhood were well documented. Hg(II) reagents were also often employed in the important cyclization step during total syntheses of many

natural products [34]. Despite its wide application in ring formation reactions, only a few review articles on Hg(II)-salt-mediated cyclization reactions are available in the literature [35]. This review describes the intramolecular cyclization of unsaturated compounds in the presence of stoichiometric/catalytic amounts of Hg(II) salts. The classification of this review is based on the following topics.

- Cyclization reactions involving the stoichiometric amount of Hg(II) salts.
- Cyclization reactions involving the catalytic amount of Hg(II) salts.
- Total synthesis involving Hg(II)-salt-mediated cyclization reactions.

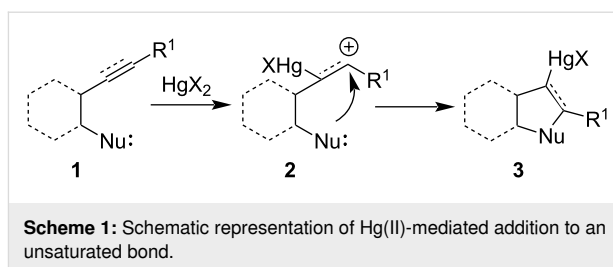
Table 1: Comparison of price of reagent grade Hg(II) salts with other transition metal salts.

Name of salts	CAS No.	Price \$/g ^a
mercury(II) chloride HgCl ₂	7487-94-7	0.67 ^b
mercury(II) acetate Hg(OAc) ₂	1600-27-7	1.35 ^b
mercuric triflate Hg(OTf) ₂	49540-00-3	12.7 ^c
mercury(II) perchlorate Hg(ClO ₄) ₂	304656-34-6	2.83 ^c
auric chloride AuCl ₃	13453-07-1	112 ^d
silver perchlorate AgClO ₄	7783-93-9	8.24 ^c
platinum chloride PtCl ₂	10025-65-7	115 ^d

^aPrices are from <http://www.sigmaaldrich.com> (as on 05-08-2021).

^b100 g, ^c25 g, ^d5 g pack size.

The generalized mechanism for cyclization reactions are, alkenes/alkynes **1** initially react with Hg(II) salts (HgX₂) leading to the formation of a mercurial carbonium ion **2** followed by the attack of an intramolecular nucleophile giving rise to a cyclized mercurio-halide complex compound **3** (Scheme 1).



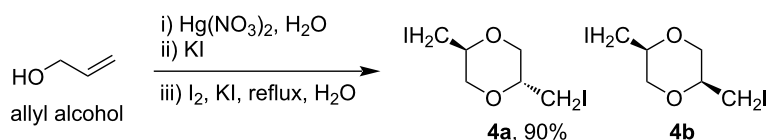
Review

Cyclization reactions involving stoichiometric amounts of Hg(II) salts

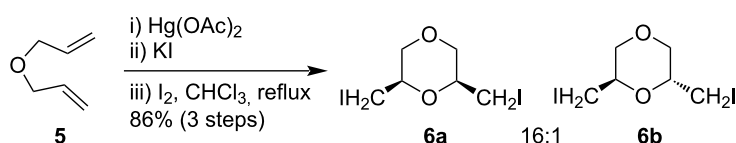
Cyclization involving alkenes (>C=C<)

In 1900 and 1901, the Sand and Biilmann group had first reported the Hg(II)-mediated cyclization of allyl alcohol using Hg(II) nitrate (Hg(NO₃)₂) in two separate publications [36,37]. The cyclized product thus formed was further treated with iodine to get *trans*-2,5-bis(iodomethyl)-*p*-dioxane (**4a**). The formation of *trans*-isomer **4a** as the major product and *cis*-isomer **4b** as a minor product was later confirmed by Summerbell et al. by repeating the same experiments (Scheme 2) [38].

Cyclization of diallyl ether **5** in presence of mercuric acetate (Hg(OAc)₂) in an aqueous medium was reported to synthesize a diastereomeric mixture of 2,6-disubstituted-*p*-dioxane **6**. The outcome of the reaction was much generalized with no detailed discussion about the ratio of diastereomeric products [39]. Later, Summerbell and co-workers modified the reaction conditions to synthesize 2,6-disubstituted dioxane derivatives **6** (*cis/trans* 16:1). Unlike 2,5-disubstituted dioxane derivatives, here the *cis*-isomer was the major product. A higher ratio of the *cis*-dioxane **6** was achieved by increasing the reaction time and acidity of the reaction medium, while elevated temperature showed no effect (Scheme 3) [40].



Scheme 2: First report of Hg(II)-mediated synthesis of 2,5-dioxane derivatives from allyl alcohol.



Scheme 3: Stepwise synthesis of 2,6-disubstituted dioxane derivatives.

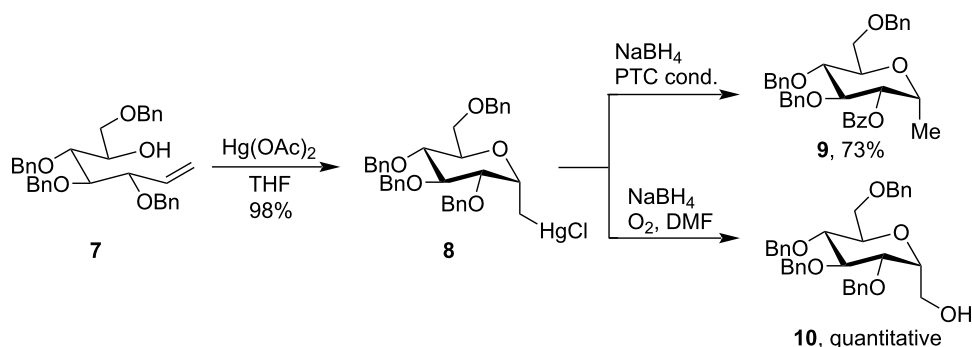
The mercuricyclization was also employed in the field of carbohydrate chemistry for the synthesis of α -D-C-glycopyranosyl derivatives. The reaction between carbohydrate alkene precursor **7** and $\text{Hg}(\text{OAc})_2$ proceeds with high stereoselectivity to give the α -D-C-glycopyranosyl derivative (1,5-*trans*-isomer) **8** as a single isomer [41]. Treatment of compound **8** with sodium borohydride (NaBH_4) under phase transfer conditions (PTC) yields compound **9** as the only product. The selectivity of α -stereochemistry was primarily due to the strong directing effect of the neighboring benzyl ether group with the $\text{Hg}(\text{OAc})_2$. When cyclic mercuric halide **8** was treated with NaBH_4 and oxygen (O_2) in DMF oxidative demercuration takes place to give alcohol **10** in quantitative yield (Scheme 4).

The methodology utilized in Scheme 5 had been successfully employed for the preparation of C-glycopyranosyl derivatives of common and uncommon sugars like α -gluco, α -manno, β -altro, β -ido, α - and β -gulo, β -talo, α -galacto, and α -allo C-glycopyranosides [42]. Vinylated derivatives of aldopentoses **11** were treated with $\text{Hg}(\text{OAc})_2$ affording the corresponding

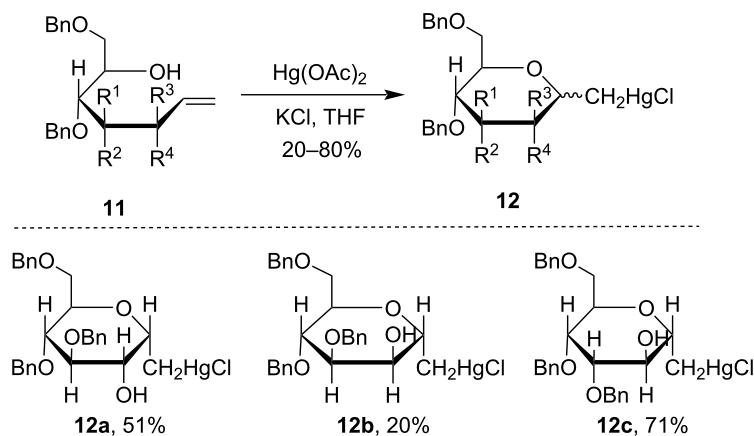
cyclized pyranosylmercuric chloride derivatives **12** (Scheme 5). This methodology can be used to synthesize rare C-glycosyl carbohydrates from easily available sugars.

In a similar manner, stereoselective cyclization of C-glycosyl amino acid derivative **13** using mercuric trifluoroacetate $\text{Hg}(\text{TFA})_2$ at room temperature was performed. The reaction proceeds primarily through stereoselective cyclization to give α -D-C-glycopyranosyl amino acid derivative **14** as the major product [43]. Nevertheless, C-mannopyranosyl derivatives cannot be achieved in a similar manner as reductive elimination forms during the mercury removal process (Scheme 6).

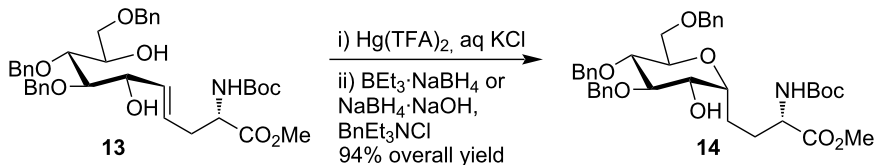
Mercury(II) salts had been effectively used to synthesize five-membered furanose derivatives with high stereoselectivity. Nicotra et al. developed $\text{Hg}(\text{OAc})_2$ -mediated cyclization of hydroxy-alkene derivative **15** to form α -D-ribose derivative **16** at room temperature (Scheme 7). They had confirmed the formation of the α -isomer of D-C-ribofuranosyl **16** predominantly over the β -isomer (α/β 95:5) [44].



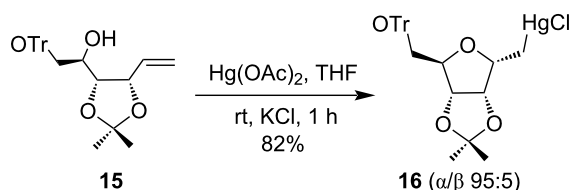
Scheme 4: Cyclization of carbohydrate alkene precursor.



Scheme 5: Hg(II)-mediated synthesis of C-glycopyranosyl derivatives.

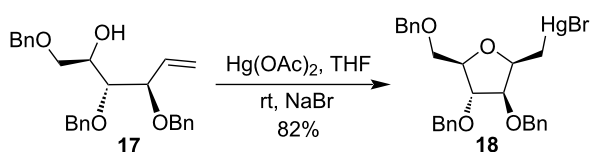


Scheme 6: Synthesis of C-glycosyl amino acid derivative using $\text{Hg}(\text{TFA})_2$.



Scheme 7: $\text{Hg}(\text{OAc})_2$ -mediated synthesis of α -D-ribose derivative.

In contrast, when Reitz and co-workers carried out a ring formation of benzylated C-arabinofuranoside derivative **17** in presence of 1.4 equiv of $\text{Hg}(\text{OAc})_2$ and sodium bromide (NaBr) at room temperature, then β -D-arabinose derivative **18** was formed as the major product (Scheme 8) [45]. The high stereoselectivity of β -derivative **18** at the anomeric position was predominantly due to the presence of the benzyl groups at the C-2 and C-3 positions of the starting material.



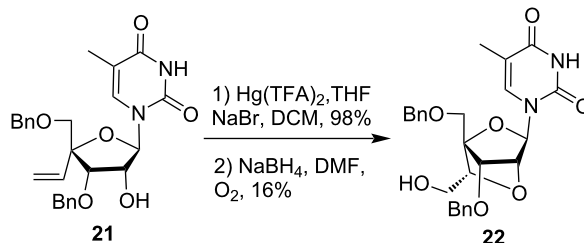
Scheme 8: Synthesis of β -D-arabinose derivative **18**.

It had been cited in many publications that the stereoselectivity of products formed due to $\text{Hg}(\text{II})$ -salt-mediated cyclization reactions of alkene-alcohol derivatives depends on several factors: the nature of the $\text{Hg}(\text{II})$ salts [46], the starting materials

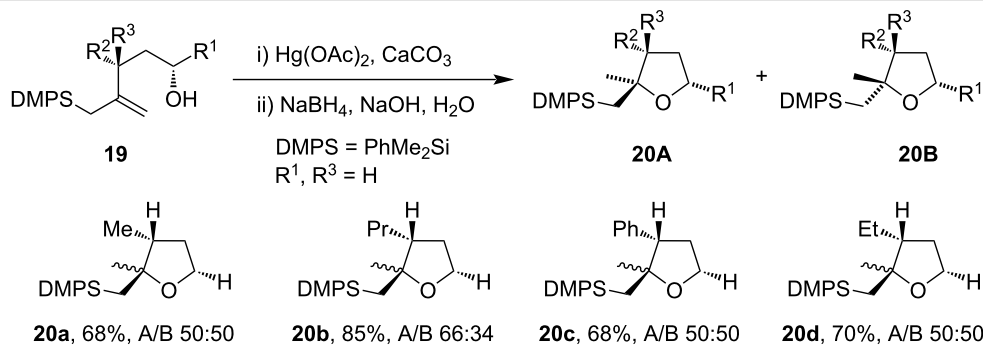
[47], and the effect of H_2O (protic solvent) in the reaction media [48].

Pulido et al. had developed a conversion of allylsilyl alcohols **19** to diastereomeric mixtures of tetrahydrofuran derivatives **20A** and **20B** (Scheme 9) [47]. It was reported that more substituted alkyl groups present in allylsilyl alcohol **20b** direct the selective formation of diastereomeric products. They also observed that changes in $\text{Hg}(\text{II})$ salts result in different ratios of the *cis*- and *trans*-isomer in the cyclized products [47]. The differences in *cis* and *trans* ratios were primarily due to the basicity of anions associated with $\text{Hg}(\text{II})$ salts [46].

A similar type of reaction methodology was employed for the formation of a bicyclic nucleoside analog. 4'-C-vinylribofuranoside derivative **21** on treatment with $\text{Hg}(\text{TFA})_2$ followed by reduction with NaBH_4 leads to the formation of bicyclic nucleoside derivative **22** (Scheme 10) [49].



Scheme 10: Synthesis of $\text{Hg}(\text{TFA})_2$ -mediated bicyclic nucleoside derivative.



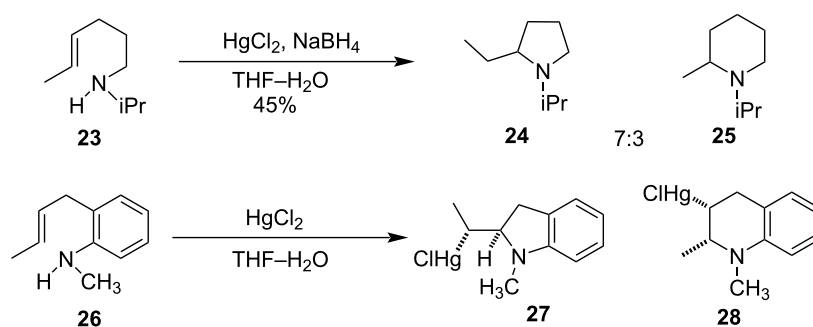
Scheme 9: $\text{Hg}(\text{OAc})_2$ -mediated synthesis of tetrahydrofuran derivatives.

Pyrrolidine and piperidine derivatives were also synthesized by Hg(II)-salt-induced cyclization. *N*-Isopropyl-1-aminohex-4-ene (**23**) on treatment with 1 equiv HgCl₂ followed by reduction with NaBH₄ yielded pyrrolidine **24** and piperidine derivative **25** in the ratio of 7:3 [50]. *N*-Methylaniline derivative **26** undergoes cyclization with HgCl₂ and gave *cis*-mercuro chlorides **27** and **28** as isolated products (Scheme 11) [51].

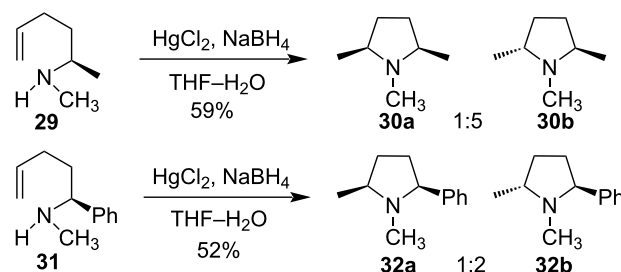
For the synthesis of diastereomeric pyrrolidine derivatives, a Hg(II) salt had been used. When HgCl₂ was added to secondary methylamine derivatives **29** and **31** followed by reduction with NaBH₄ a mixture of diastereomers **30a,b** and **32a,b** was

obtained, respectively. *trans*-Isomers were formed as major products over *cis*-isomers (Scheme 12) [52].

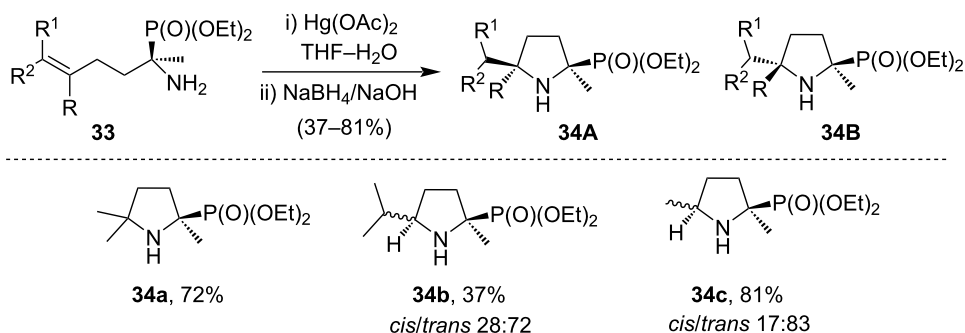
Five- and six-membered *N*-containing heterocyclic phosphonates were synthesized by intramolecular cyclization of alkenyl α -aminophosphonates in a similar way with the treatment of Hg(OAc)₂ [53]. The cyclized products **34A,B** formed from starting material **33** were regioselective and followed Markovnikoff's type addition in the reaction [54-57]. It was also reported that the formation of α -phosphorylated pyrrolidines mostly takes place in regio- and stereoselective ways depending on the reaction conditions (Scheme 13) [53,58].



Scheme 11: Synthesis of pyrrolidine and piperidine derivatives.

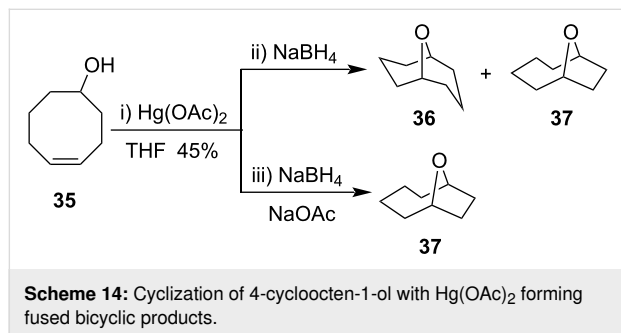


Scheme 12: HgCl₂-mediated synthesis of diastereomeric pyrrolidine derivatives.

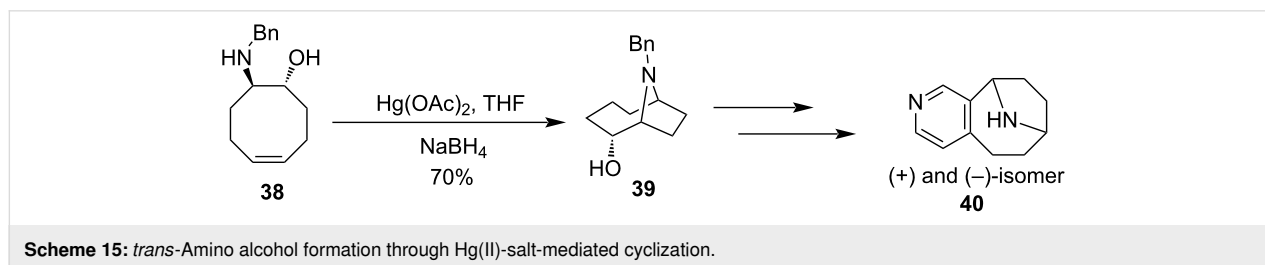


Scheme 13: HgCl₂-mediated cyclization of alkenyl α -aminophosphonates.

Cyclization of 4-cycloocten-1-ol (**35**) with $\text{Hg}(\text{OAc})_2$ resulted in formation of two types of fused bicyclic products, 9-oxabicyclo[3.3.1]nonane (**36**) and 9-oxabicyclo[4.2.1]nonane (**37**) (Scheme 14) [59,60]. It was observed that the regioselective synthesis of 9-oxabicyclo[4.2.1]nonane (**37**) was favored when NaBH_4 and sodium acetate NaOAc were present, while in absence of NaOAc , 9-oxabicyclo[3.3.1]nonane (**36**) was formed as an exclusive product.



In a similar manner, an aminomercuration reaction had been successfully employed in the cyclization of *trans*-amino alcohol **38** leading to the formation of (1*R*,2*R*,6*R*)-9-benzyl-9-azabicyclo[4.2.1]nonan-2-ol (**39**). The bicyclic derivative **39** through the number of consequent reactions formed a highly potent (+)/(-)-pyrido[3,4-*b*]homotropine (**40**), a bridged nicotinoid (Scheme 15) [61].

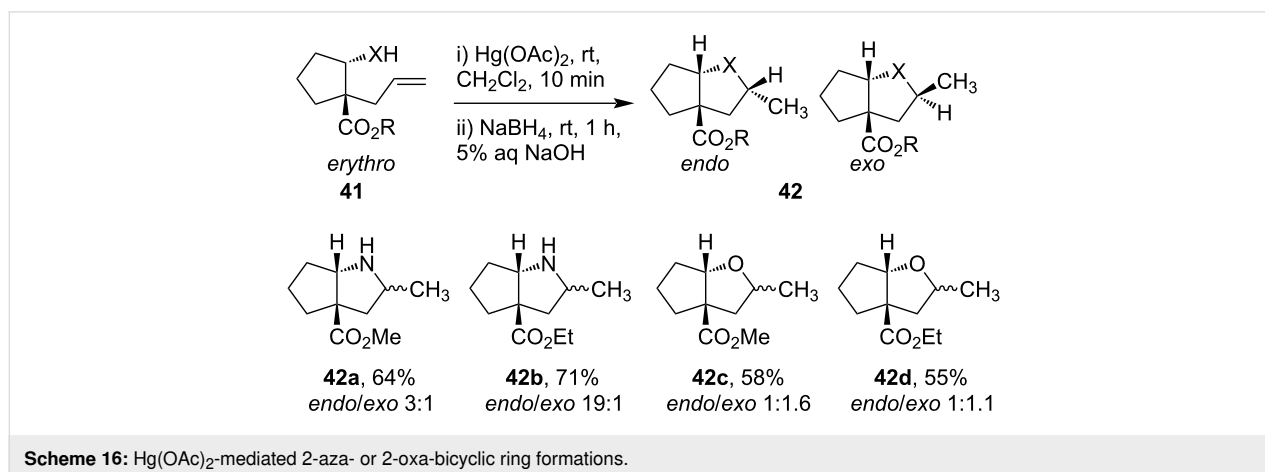


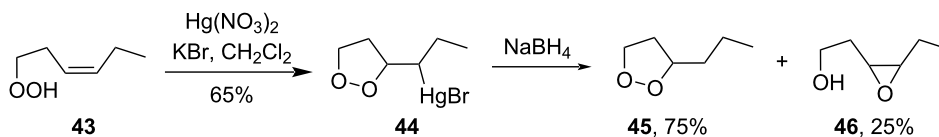
Similarly, precursor **41** at room temperature in the presence of $\text{Hg}(\text{OAc})_2$ (1 equiv) in CH_2Cl_2 cyclized to form 2-aza- or 2-oxabicyclic mercuri derivatives. Further, treatment of intermediates with NaBH_4 (4 equiv) in presence of 5% aq NaOH solution for 1 hour at room temperature produced bicyclic derivatives **42** (Scheme 16) [62]. The diastereoselectivity of the products formed depends on the heteroatoms involved; azamercuration yields more selectivity than oxymercuration.

Nixon et al. reported a $\text{Hg}(\text{II})$ -salt-induced alkenyl hydroperoxide **43** cyclization to synthesize cyclic peroxides **44** as the major product, which on further treatment with NaBH_4 gives compound **45** along with the unexpected product **46** (Scheme 17) [63].

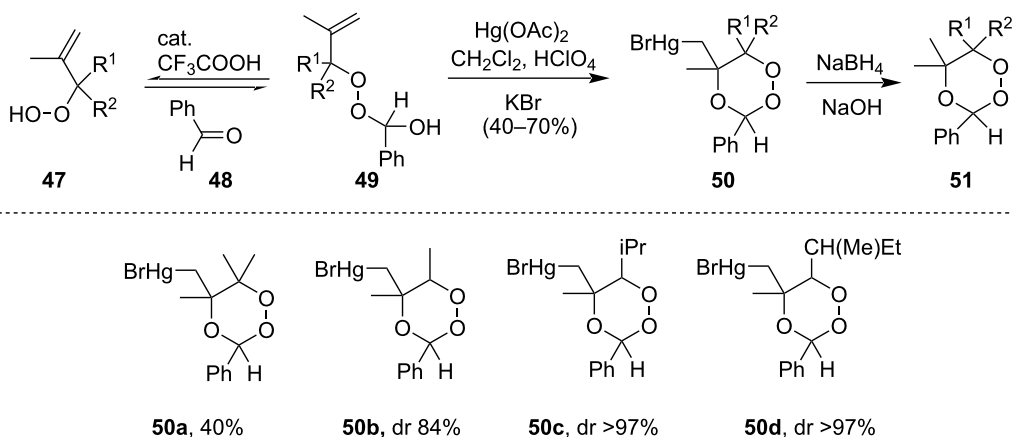
Later they had reported the reaction between alkenyl hydroperoxide **47** and aldehydes **48** to form hemiperoxyacetal **49** which subsequently reacted with $\text{Hg}(\text{OAc})_2$ and a catalytic amount of perchloric acid (HClO_4) forming cyclized product 1,2,4-trioxanemercuri bromide **50**. The product thus formed was reduced with NaBH_4 to form compound **51** (Scheme 18) [64].

Kurbanov et al. first reported the cyclization of different isoprenoid derivatives to hexahydrochromene derivatives using $\text{Hg}(\text{II})$ salts [65]. They had shown that *trans*-geranylacetone (**52**) produces *trans*-2,5,5,9-tetramethylhexahydrochromene (**53**) while the *cis*-cyclized product was formed from *cis*-



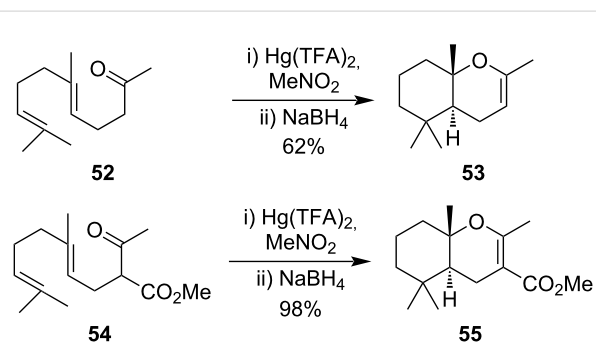


Scheme 17: Hg(II)-salt-induced cyclic peroxide formation.

Scheme 18: $\text{Hg}(\text{OAc})_2$ -mediated formation of 1,2,4-trioxanes.

isoprenoid derivatives. Later, Hoyer et al. used similar reaction conditions to cyclize dienes **54** with 1.1 equiv of $\text{Hg}(\text{TFA})_2$ to yield endocyclic enol ether **55** in almost quantitative yield [66]. They had also performed experiments with different isoprenoid derivatives (carboxylic acid, ketones alcohols, and keto esters) to form different bicyclic products (Scheme 19).

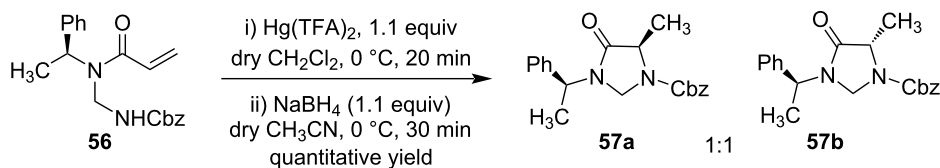
A Hg(II)-salt-promoted cyclization of amidal **56** was performed to synthesize optically active cyclic alanine derivatives **57** as 1:1 diastereomeric mixture [67]. Diastereomers were further separated by column chromatography and afforded enantiomerically pure isomers of *N*-substituted imidazolidine-4-ones **57** (Scheme 20).



Scheme 19: Endocyclic enol ether derivative formation through Hg(II) salts.

In a similar manner, *N*-Cbz-protected amine **58** undergoes cyclization using $\text{Hg}(\text{TFA})_2$ (1 equiv) and yielded a dinitrogen-containing mixture of diastereomeric alicyclic derivatives **59**.

The tetrahydropyrimidin-4(1*H*)-one-mercuri trifluoroacetate derivative **59** on successive treatment with NaBr and LiBH_4 gives a mixture of tetrahydropyrimidin-4(1*H*)-one derivatives **60** in



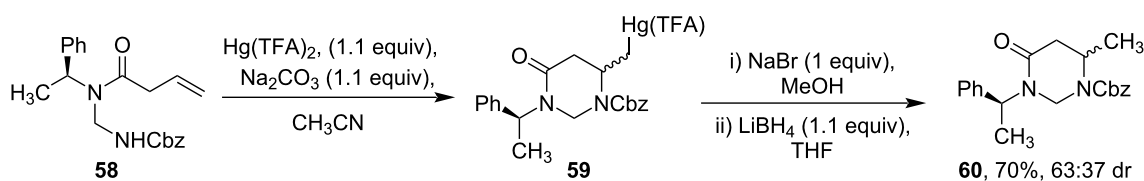
Scheme 20: Synthesis of optically active cyclic alanine derivatives.

the diastereomeric ratio of 63:37 after separation by column chromatography [68] (Scheme 21).

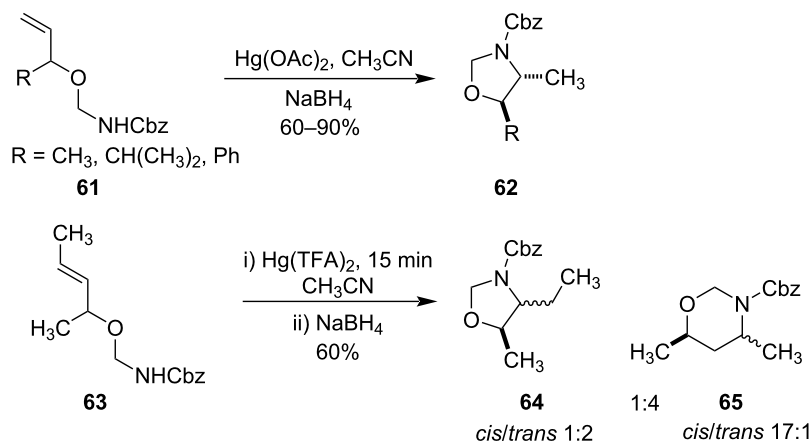
Mercury(II) salts had also been successfully utilized in the cyclization of ether derivatives **61** to form stereoselectively *trans*-4,5-disubstituted oxazolidine derivatives **62** (Scheme 22) [69]. Later it was reported, when homologous allyloxycarbamate derivative **63** was cyclized with Hg(TFA)₂ then five- and six-membered rings **64** and **65** were formed with a 1:4 ratio (Scheme 22) [70]. The greater yield of the six-membered prod-

uct was primarily due to the electron-withdrawing effect of etheral oxygen which in turn destabilizes the carbocation at the β-carbon and hence the nucleophilic attack at the γ-carbon took place.

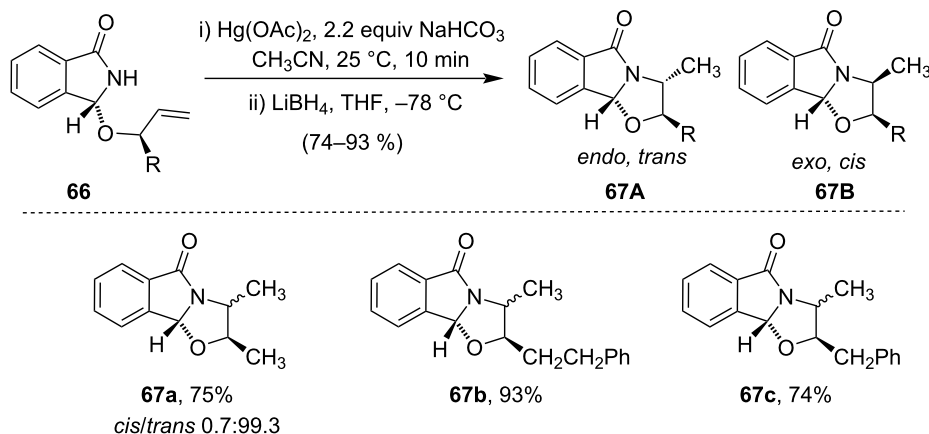
Cyclization of amide derivative **66** induced by Hg(OAc)₂ followed by reduction with LiBH₄ afforded a mixture of compounds **67A/67B** [71]. The formation of *endo,trans*-product as a major product over the *exo,cis*-isomer was primarily due to a stereoselection effect (Scheme 23). The small amount of



Scheme 21: Hg(II)-salt-mediated formation of tetrahydropyrimidin-4(1H)-one derivatives.



Scheme 22: Cyclization of ether derivatives to form stereoselective oxazolidine derivatives.



Scheme 23: Cyclization of amide derivatives induced by Hg(OAc)₂.

exo,cis-isomers **67Ba/67Bc** was only detected by HPLC analysis of the crude reaction mixture.

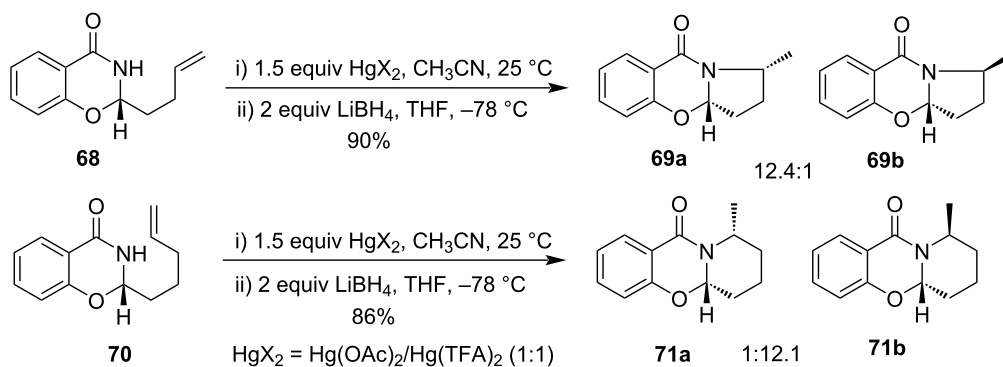
Takacs et al. observed the rapid cyclization of salicylamide-derived amidal auxiliary derivatives **68** and **70** in presence of 1.5 equiv of a 1:1 mixture of $\text{Hg}(\text{OAc})_2/\text{Hg}(\text{TFA})_2$ resulting in a diastereomeric pair of cyclized products **69** and **71**, respectively. It was observed that the *cis*-isomer was predominant in the case of the five-membered ring while the *trans*-isomer was predominant in the case of the six-membered ring formation (Scheme 24) [72].

Cyclization involving alkynes (-C≡C-)

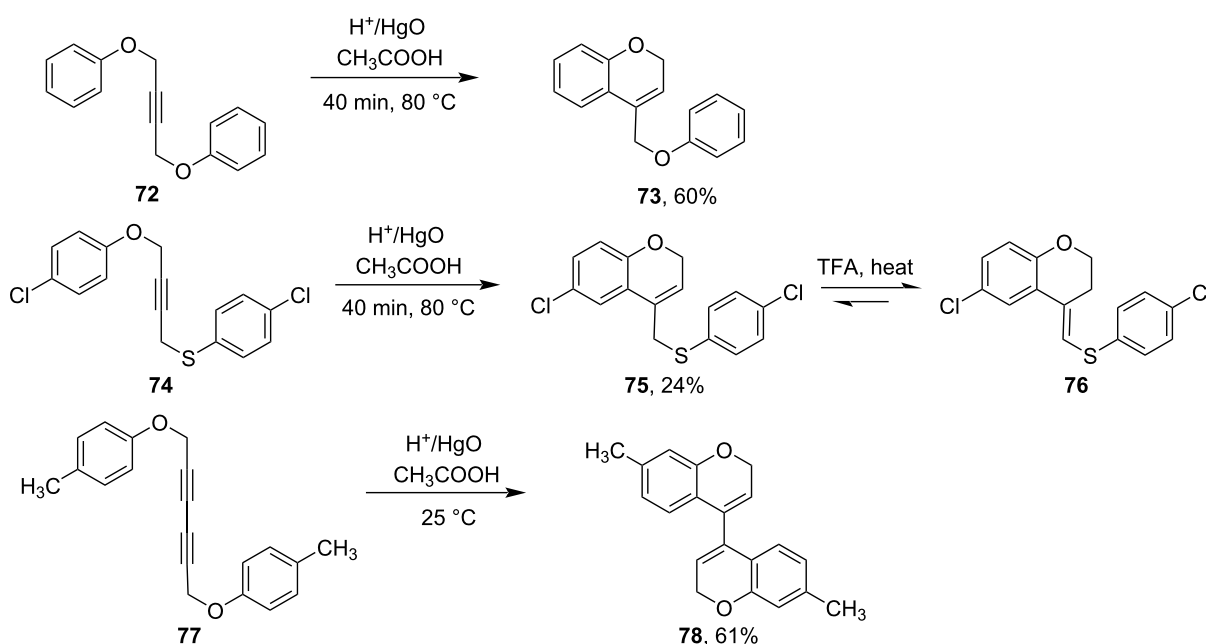
$\text{Hg}(\text{II})$ -salt-mediated cyclizations were also observed in the case of alkynes. Arylalkynes **72** were cyclized via $\text{Hg}(\text{II})$ -salt-in-

duced reactions to form benzopyran derivatives **73** [73–75]. Interestingly in the case of 1-aryloxy-4-arylthio-2-butyne derivatives **74**, selective ring closure on the oxygen side afforded 6-membered chromene derivatives **75** but no thiochromene derivatives were observed [76]. Compound **75** on refluxing with trifluoroacetic acid isomerized to form exocyclic *trans*-compound **76**. It was reported that 1,4-di(arylthio)-2-butyne derivatives only afforded usual $\text{Hg}(\text{II})$ -salt-mediated hydrated products instead of cyclized thiochromene derivatives. Balasubramanian et al. had performed a mercuric-oxide-mediated cyclization of 1,6-di(aryloxy)-2,4-hexadiyne derivatives **77** to get bichromene derivatives **78** (Scheme 25) [77].

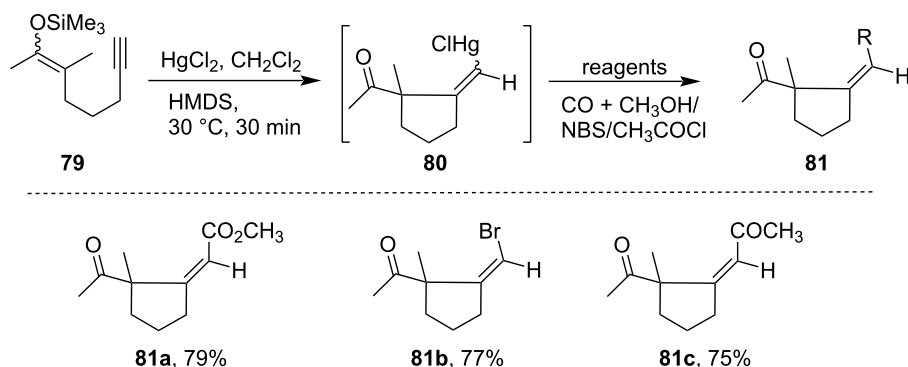
A HgCl_2 -induced cyclization also takes place for acetylenic silyl enol ether derivative **79** forming carbocyclic compounds



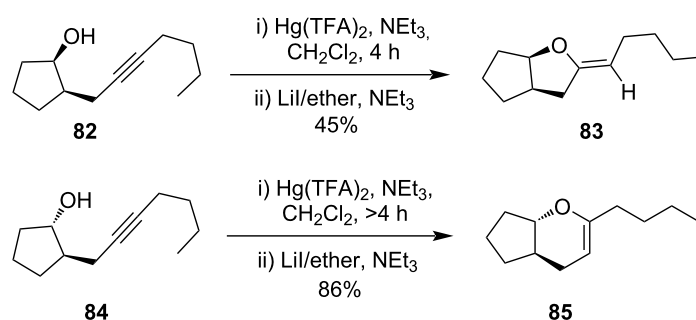
Scheme 24: $\text{Hg}(\text{OAc})_2/\text{Hg}(\text{TFA})_2$ -promoted cyclization of salicylamide-derived amidal auxiliary derivatives.



Scheme 25: $\text{Hg}(\text{II})$ -salt-mediated cyclization to form dihydrobenzopyrans.



Scheme 26: HgCl₂-induced cyclization of acetylenic silyl enol ether derivatives.



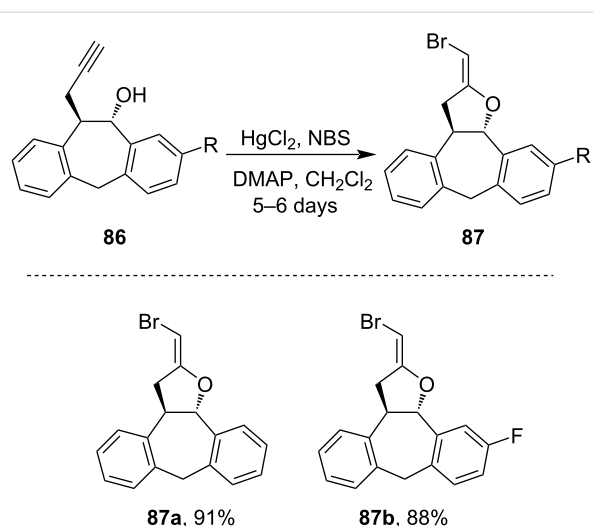
Scheme 27: Synthesis of exocyclic and endocyclic enol ether derivatives.

81 with good yields via intermediate **80** [78]. The cyclization of compounds **79** undergo regioselective addition with the triple bond in exocyclic alkene position leading to the formation of α -mercury ketone **80**, which were later functionalized by electrophilic addition (Scheme 26).

Schwartz et al. developed a new route for the synthesis of exocyclic enol ether **83** and endocyclic enol ether **85** involving a Hg(II)-induced cyclization of acetylenic alcohols **82** and **84**, respectively [79]. This work revealed that the regioselectivity of the Hg(II) addition to the alkyne may be influenced by diastereomers; in the case of *cis*-isomer **82** exocyclic enol ether **83** was formed while for *trans*-isomer **84** the reaction takes place with a much slower rate yielding endocyclic enol ether **85** as the only product (Scheme 27) [79].

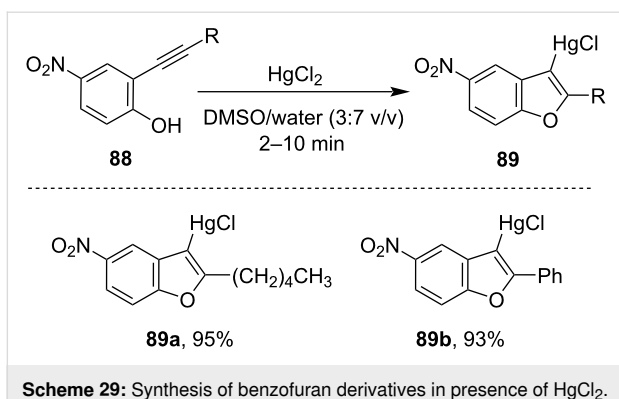
trans-Acetylenic alcohol **86** on treatment with HgCl₂ (0.5 equiv) in presence of *N*-bromosuccinimide (NBS) undergoes cyclization yielding stable bromo alkenes **87** (Scheme 28) [80,81].

Atta et al. reported the specific cyclization of ethynyl phenols **88** in presence of HgCl₂ at ambient temperature yielding benzo-



Scheme 28: Cyclization of *trans*-acetylenic alcohol by treatment with HgCl₂.

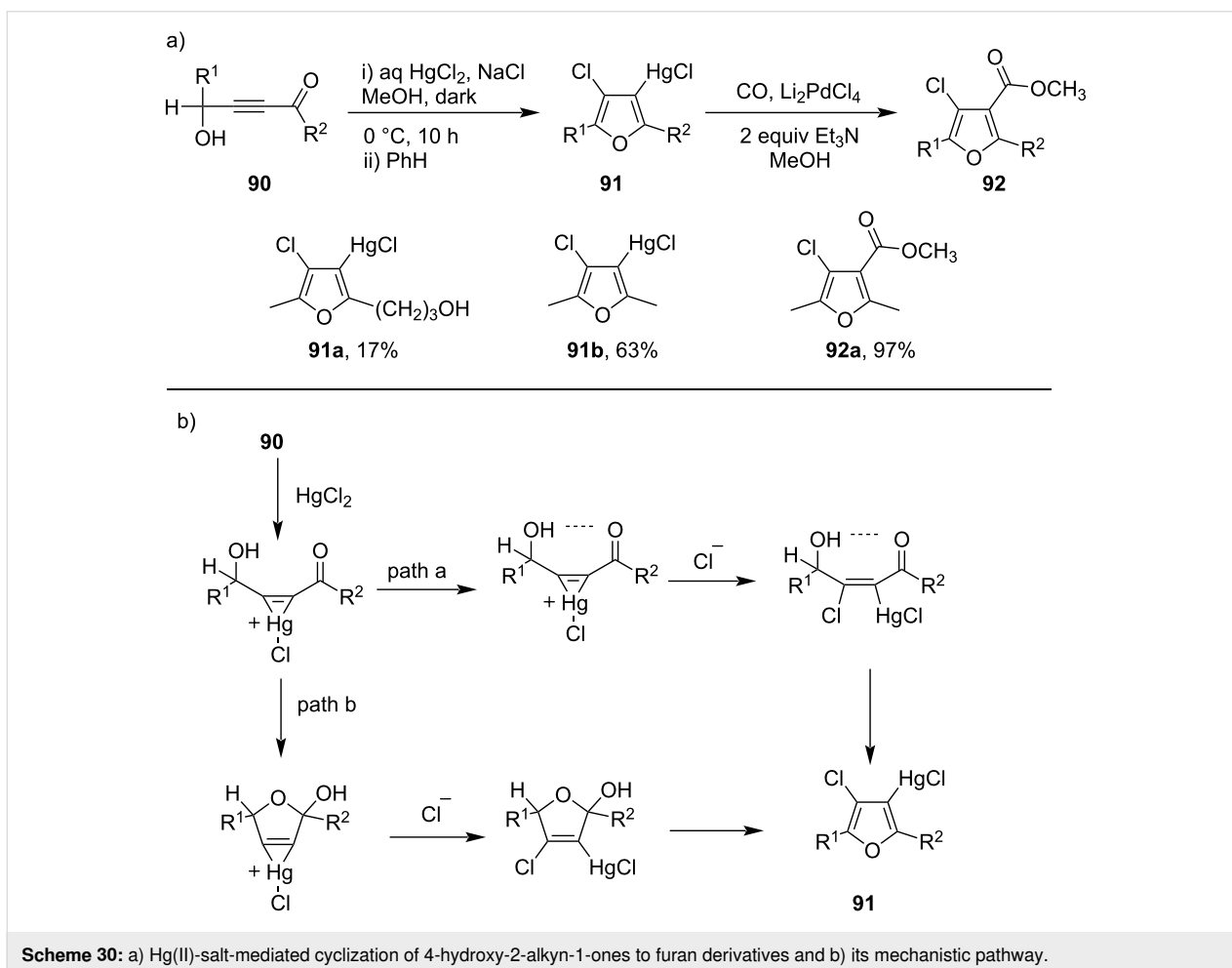
furan derivatives **89**. They had reported that the electron-withdrawing electronic effect at the aromatic ring promotes the cyclization reaction, whereas there is no cyclization in the absence of the withdrawing group (Scheme 29) [82].

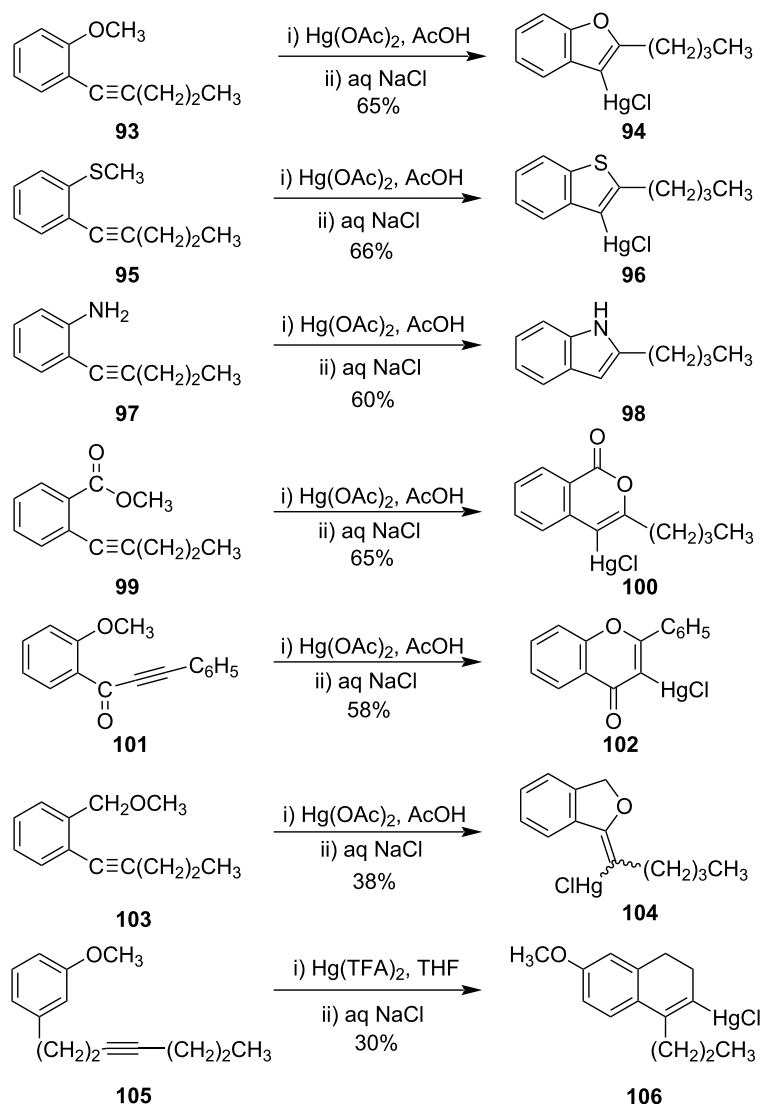


Larock et al. had performed the mercuriation of 4-hydroxy-2-alkyn-1-ones **90** with HgCl₂ to get furylmercurials **91** via *syn*-addition of acetylene, which on carbonylation yielded the furan-containing carbonyl compound **92** (Scheme 30) [83]. It was proposed that initially mercuriation of acetylene bonds via mercurinium like ions or π -complex takes place. Then the structure was stabilized through hydrogen bonding between alcohol and carbonyl groups (path a) or by hemiketal formation (path b).

Thus, the chloride ion attacks from the frontside, producing *syn*-addition molecules that, upon dehydration, formed furan rings.

Later they had reported similar mercuriation of arylacetylenes to synthesize a broad spectrum of heterocycles, namely benzofurans, benzothiophenes, isocoumarins, chromones, benzopyrans, 1,2-dihydronaphthalenes, coumarins, and coumestan including some physiologically active heterocyclic natural products like flavones [84]. In the presence of Hg(OAc)₂ in acetic acid, simple cyclization of *ortho*-substituted arylacetylenes **93**, **95**, and **97** yielded benzofurans **94**, benzothiophenes **96**, and indoles **98**, respectively. When the carbonyl group was introduced between an aryl and a methoxy group (**99**) then six-membered isocoumarin ring **100** was formed, and when a carbonyl group was introduced in between an aryl and an alkyne group (**101**), chromone derivatives **102** were formed. On the introduction of the methylene moiety in between the aryl and methoxy group, **103** yielded isofuran derivatives **104** due to the change in regioselectivity during the cyclization. Dihydronaphthalenes **106** were synthesized by cyclization of the corresponding methoxy-arylalkyne derivative **105** (Scheme 31).

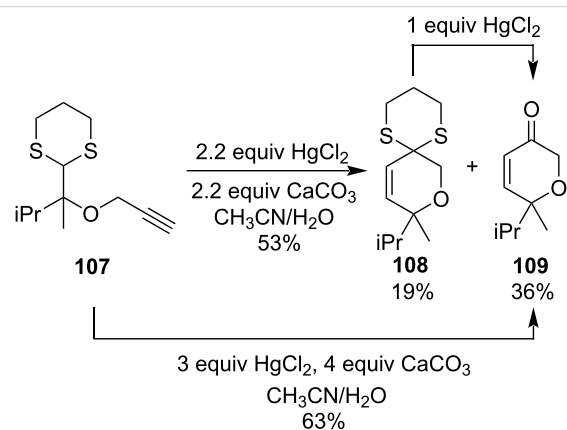




Scheme 31: Cyclization of arylacetylenes to synthesize carbocyclic and heterocyclic derivatives.

In 2005, Ghorai et al. reported the synthesis of a mixture of 6-membered as well as 5-membered heterocyclic compounds through a Hg(II)-salt-promoted unique cyclization–rearrangement reaction [85]. They had performed the HgCl₂-promoted cyclization of *O*-propargyl glycolaldehyde dithioacetals **107** via their dithioketals and dithioacetals to synthesize six-membered heterocycles **108** and **109** (Scheme 32). Five-membered dihydrofuryl aldehydes were also isolated as minor products in some examples.

Following a similar protocol, Biswas et al. later published the HgCl₂-mediated cyclization reaction of tethered alkynedithioacetals **110** to provide six- and five-membered carbocyclic and heterocyclic derivatives **111** and **112**, respectively. They had observed that the formation of five-membered rings (**112a–e**)



Scheme 32: Hg(II)-salt-promoted cyclization–rearrangement to form heterocyclic compounds.

was preferred when substituents were present at the alkyne terminus, whereas six-membered rings (**111a,b**) formed predominantly in case of unsubstituted alkyne dithioacetals (Scheme 33) [86]. They had reported the plausible mechanism for the formation of a six-membered pyranose ring follows 'path a,' while for the formation of five-membered pyrrolidine derivatives 'path b' was followed.

Cyclization involving allenes (>C=C=C<)

Apart from alkenes/alkynes, there are also examples where cyclization takes place involving allene functionalities. Some of the examples are discussed below.

Balasubramanian et al. reported the cyclization of aryl allenic ethers **113** on treatment with Hg(OTf)₂. Compound **113** reacted with 1.1 equiv of Hg(OTf)₂ at room temperature followed by reduction with alkaline NaBH₄ leading to the formation of

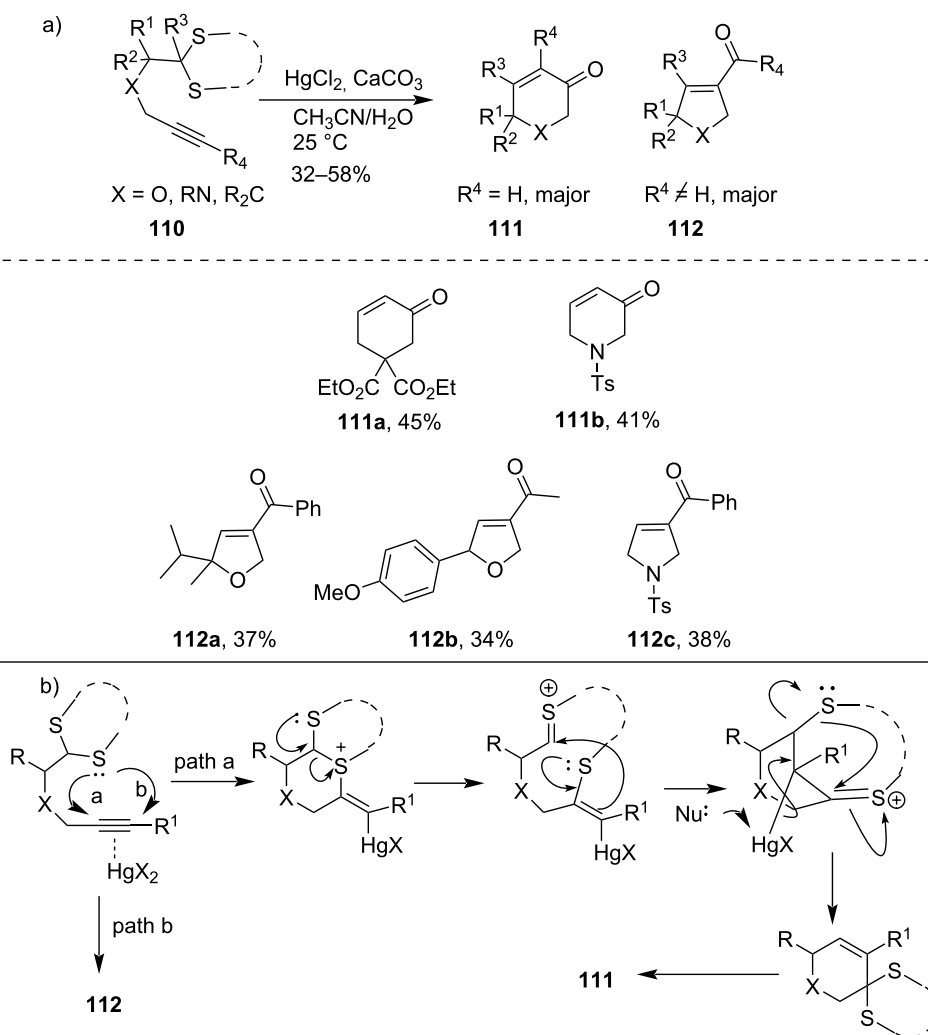
benzopyran derivatives **114** and **115**. The ratio of the formation of products depends on the temperature and time of the reaction (Scheme 34) [87].

Devan et al. had developed similar types of Hg(TFA)₂-mediated cyclizations of allene **116** at low temperature followed by reduction with alkaline NaBH₄ to form cyclized product **117** in moderate yield [88]. The reaction was believed to proceed through Hg(II) ion-promoted electrophilic cyclization (Scheme 35).

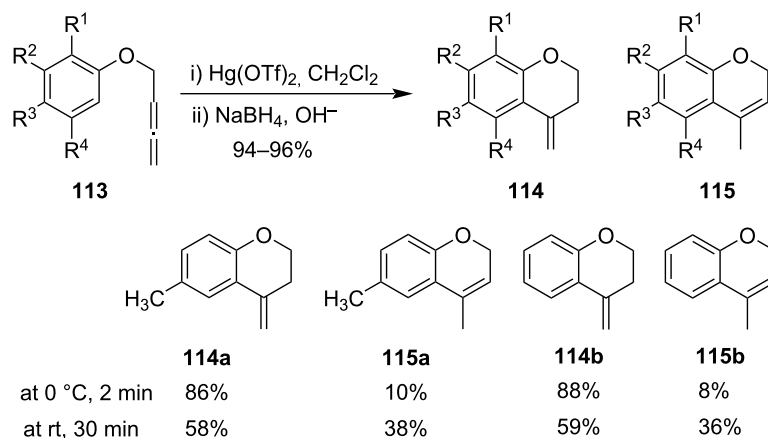
Cyclization involving catalytic Hg(II) salts

Cyclization involving alkenes (>C=C<)

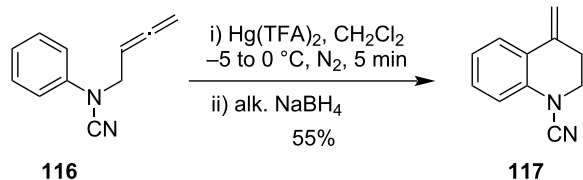
Apart from the stoichiometric amount used in cyclization, there is abundant literature where a catalytic amount of Hg(II) salt was employed for cyclization reactions between nucleophiles and unsaturated bonds.



Scheme 33: a) HgCl₂-mediated cyclization reaction of tethered alkyne dithioacetals; and b) proposed mechanism.



Scheme 34: Cyclization of aryl allenic ethers on treatment with $\text{Hg}(\text{OTf})_2$.



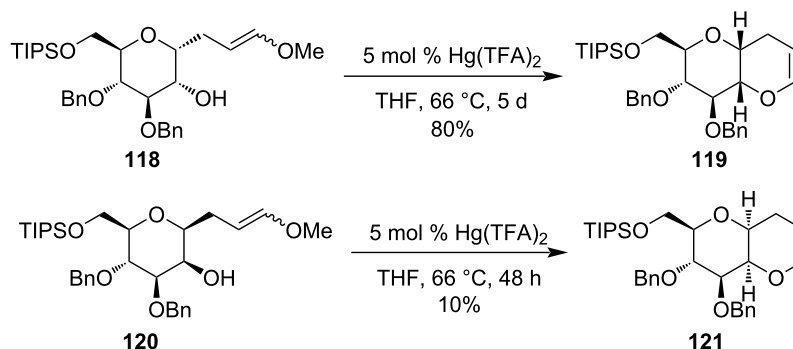
Scheme 35: $\text{Hg}(\text{TFA})_2$ -mediated cyclization of allene.

The synthesis of fused polycyclic ethers by the treatment of a catalytic amount of $\text{Hg}(\text{TFA})_2$ with suitable starting material was demonstrated by Tan et al. [89]. They had reported a $\text{Hg}(\text{II})$ -catalyzed intramolecular *trans*-etherification reaction of 2-hydroxy-1-(γ -methoxyallyl)tetrahydropyrans **118** and **120** to the corresponding bicyclic dihydropyrans **119** and **121**, respectively (Scheme 36).

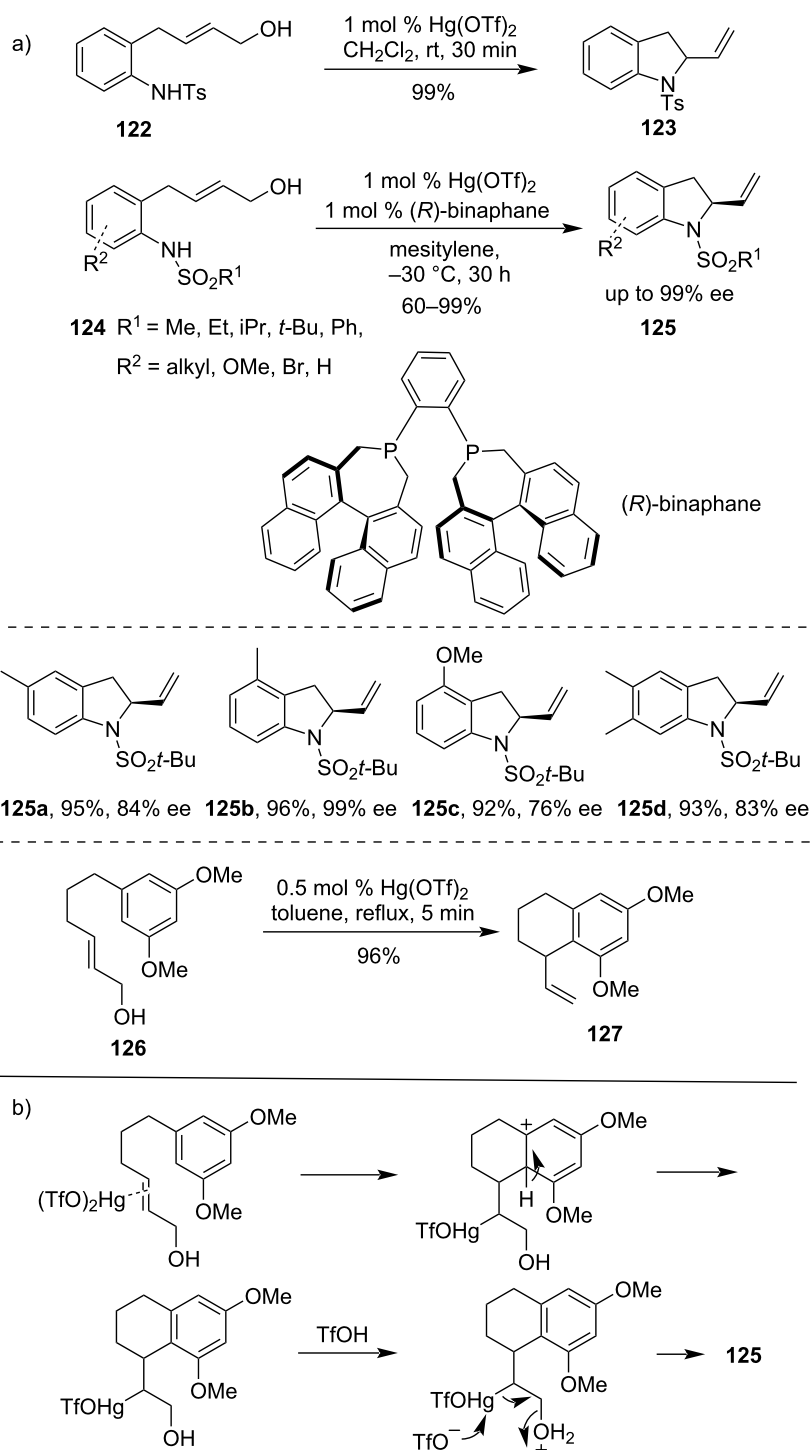
Later, Namba et al. reported the synthesis of racemic vinylindoline derivatives **123** from *N*-tosylanilinoallylic alcohol derivative **122** by using 1–2 mol % of $\text{Hg}(\text{OTf})_2$ in CH_2Cl_2 at room

temperature [90]. An asymmetric synthesis of vinylindoline derivatives **125** was achieved by utilizing chiral ligands like chiral binaphane (Scheme 37) [91]. They had observed that the formation of six- and seven-membered rings required elevated temperatures. Subsequently, the same group studied the cyclization of arylene **126** to furnish naphthalene derivative **127**. The plausible mechanism for the formation of compound **127** proceeded consecutively with π -complex formation, Friedel–Crafts type addition, deprotonation, and finally protonation of alcohol for the elimination of water to get the final product [92].

A $\text{Hg}(\text{OTf})_2$ -mediated cyclization was utilized for the synthesis of 1,4-dihydroquinoline **129** possessing a quaternary carbon center from **128** [93]. The reaction was reported via a seven-membered bicyclic hemiaminal as mentioned in the mechanism. This catalytic rearrangement protocol was successfully applied for the construction of complex carbon frameworks from various tosylanilinoallyl acetals. 4*H*-Chromene derivatives were also synthesized starting from phenol derivatives (Scheme 38).



Scheme 36: $\text{Hg}(\text{II})$ -catalyzed intramolecular *trans*-etherification reaction of 2-hydroxy-1-(γ -methoxyallyl)tetrahydropyran derivatives.

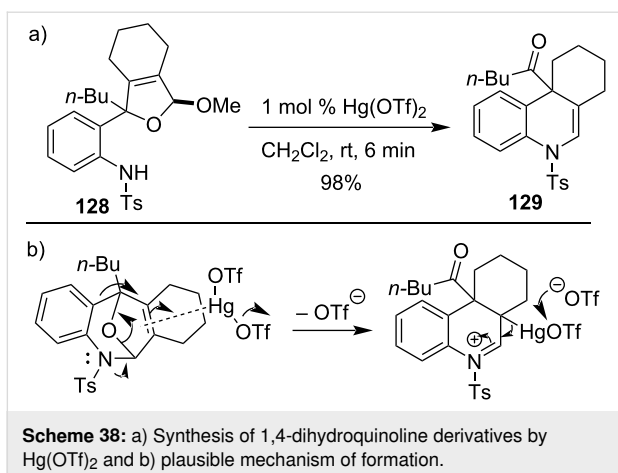


Scheme 37: a) Cyclization of alkene derivatives by catalytic Hg(OTf)_2 salts and b) mechanism of cyclization.

Cyclization involving alkynes ($-\text{C}\equiv\text{C}-$)

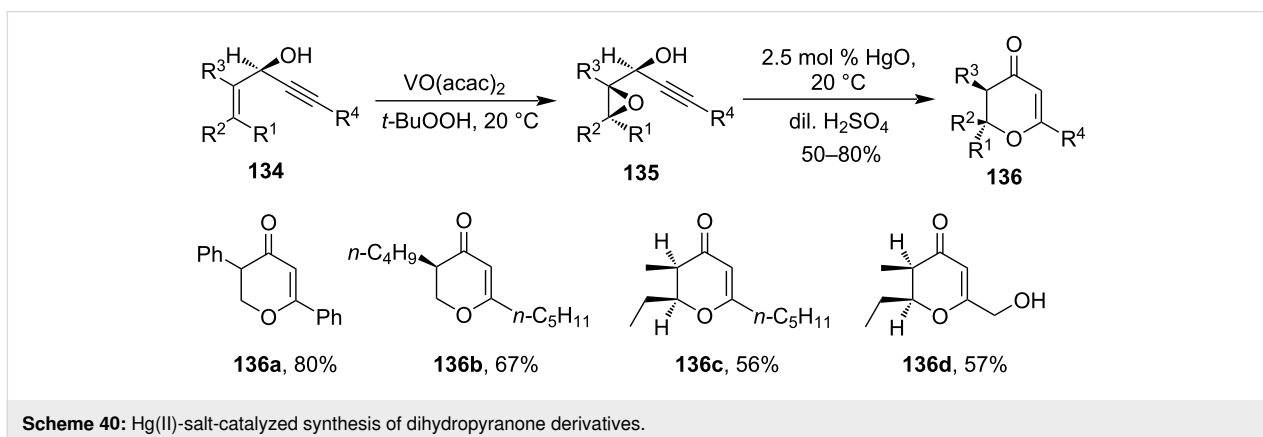
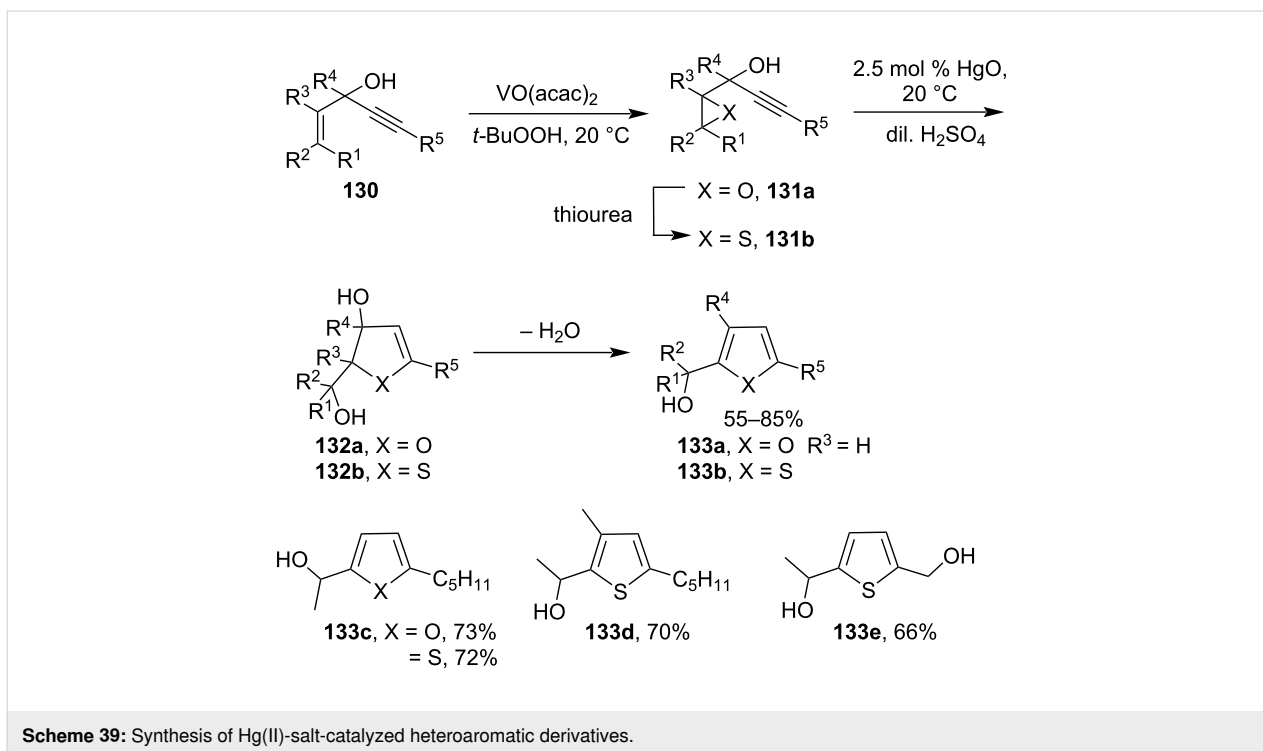
Marson et al. had developed the synthesis of substituted furans **133a–c** promoted by catalytic use of Hg(II) salt through cyclization of secondary and tertiary 1-alkynyl-2,3-epoxyalcohols **131a** [94]. This is an example of a Hg(II) -salt-catalyzed rear-

rangement of 1-alkynyl-2,3-epoxy alcohols to substituted furans. The furan **133a** was formed by dehydration of intermediate **132a** through the corresponding oxonium cation. When R^3 is an alkyl group, direct dehydration was not possible because of blocking. Later they reported a similar transformation of



thiiranes to provide a wide variety of substituted thiophenes. They had synthesized substituted thiophenes **133b** starting from thiiranes **131b** via cyclization utilizing the catalytic amount of HgO and dil. H₂SO₄ at room temperature [95]. Initially, the Hg(II)-salt-catalyzed the formation of intermediate **132b** which further proceeded by dehydration to yield thiophenes **133b** as the final product (Scheme 39).

Starting with 1-alkynyl-2,3-epoxy alcohols **135**, Marson et al. had reported a Hg(II)-salt-catalyzed rearrangement to produce 2,3-disubstituted-2,3-dihydropyranone derivatives **136**. The stereochemistry of substituents at 2,3-positions of 2,3-dihydropyranone **136** was controlled by *cis*- and *trans*-configuration of the epoxide of starting materials (Scheme 40) [96].



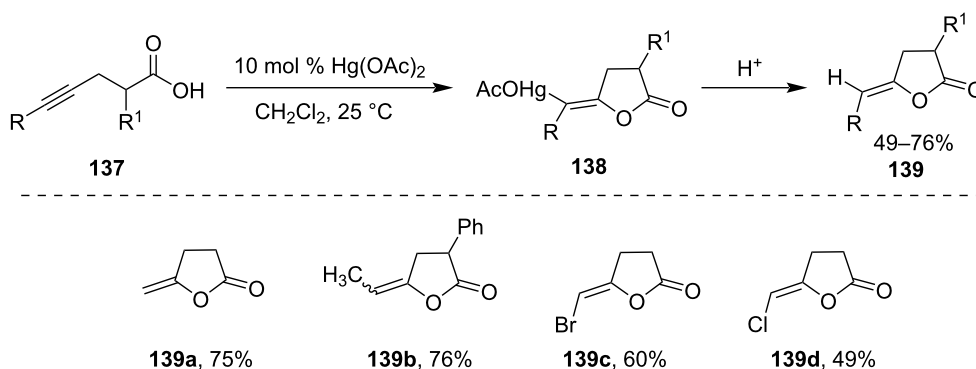
Several unsaturated lactones had been synthesized from alkynoic acids via Hg(II)-salt-catalyzed cyclization reaction. For example, simple 4-pentynoic acid derivatives **137** afforded γ -methylene butyrolactones **139** in good yields via the formation of organomercural compounds **138** using catalytic mercuric oxide, mercuric acetate, or mercuric trifluoroacetate as shown in Scheme 41 [97].

Alkyne carboxylic acids also undergo oxymercuration reactions to form furan- and pyran-like derivatives. When γ -alkyne carboxylic derivative **140** was refluxed with HgO the cyclization took place to give product **141** in almost quantitative yields [98,99]. Compound **142** under refluxing conditions gave spirocyclic compound **143** as the exclusive product (Scheme 42) [99]. Propargylic triols **144** undergo Hg(OTf)₂-catalyzed cyclization reaction, to produce monounsaturated spiroketal **145**.

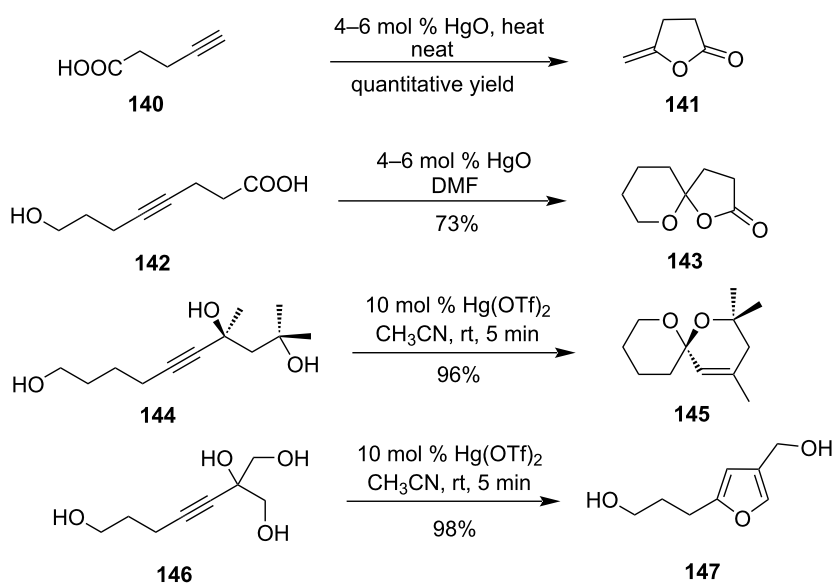
Finally in the case of 1,2-dihydroxy-3-alkynes **146** comparable reaction conditions leads to the formation of 2,5-disubstituted-furan **147** rather than mono-unsaturated spiroketals [100].

Interestingly when 1,4-dihydroxy-5-alkyne derivatives were subjected to a Hg(OTf)₂-catalyzed cyclization then oxacyclization takes place to form tetrahydropyran derivatives (Scheme 43) [101]. Later it was shown that alkynyl diol **150** when treated with 20 mol % Hg(OTf)₂ followed by Et₃SiH afforded bispyranoyl ketone **151**, but when Hg(OTf)₂ was increased (1 equiv) then fused pyran-oxocane derivative **152** was isolated [102].

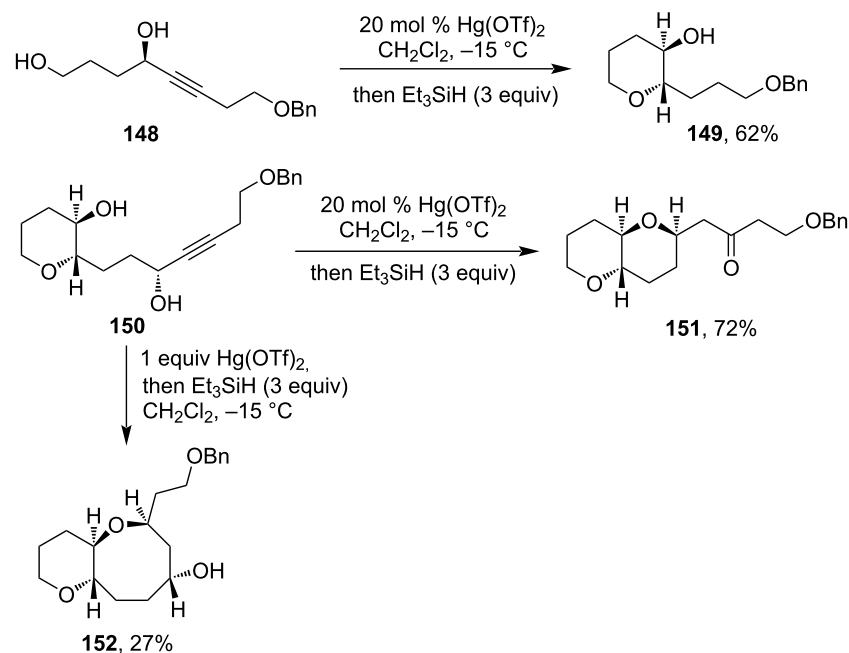
Six-membered morpholine derivatives were also synthesized by catalytic Hg(II)-salt-induced cyclization. Yamamoto and co-workers published the intermolecular cyclization of alkynyl-



Scheme 41: Hg(II)-salt-catalyzed cyclization of alkynoic acids.



Scheme 42: Hg(II)-salt-mediated cyclization of alkyne carboxylic acids and alcohol to furan, pyran, and spirocyclic derivatives.



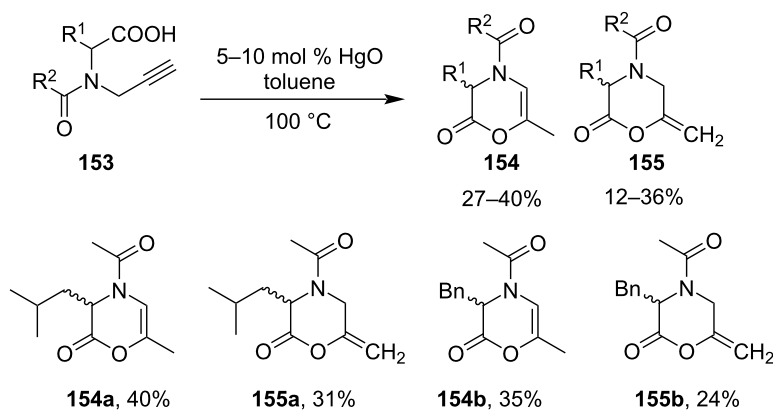
Scheme 43: Hg(II)-salt-mediated cyclization of 1,4-dihydroxy-5-alkyne derivatives.

carboxylic acid **153** to produce 6-membered morpholine type ring compound **154** and compound **155** [103]. The stereochemistry of the chiral amino acid was not conserved in the cyclized product hence it leads to the formation of racemic products with moderate yields (Scheme 44).

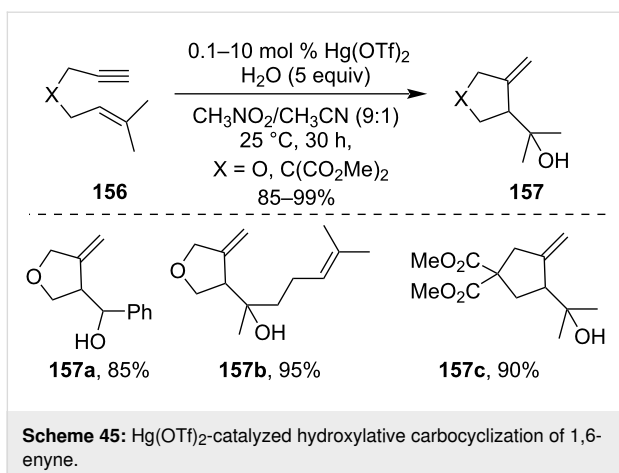
1,6-Enynes **156** underwent smooth hydroxylative carbocyclization in the presence of catalytic Hg(OTf)_2 . As an example, 3-methylene five-membered cyclic derivatives **157** were synthesized by Nishizawa et al. via the Hg(OTf)_2 -catalyzed hydroxylative carbocyclization of 1,6-enyne **156** (Scheme 45) [104]. It was observed that from 1,7-enynes, six-membered rings were

formed in low to moderate yield while from 1,8-enynes, uncyclized hydrated products were isolated, as major products along with different byproducts.

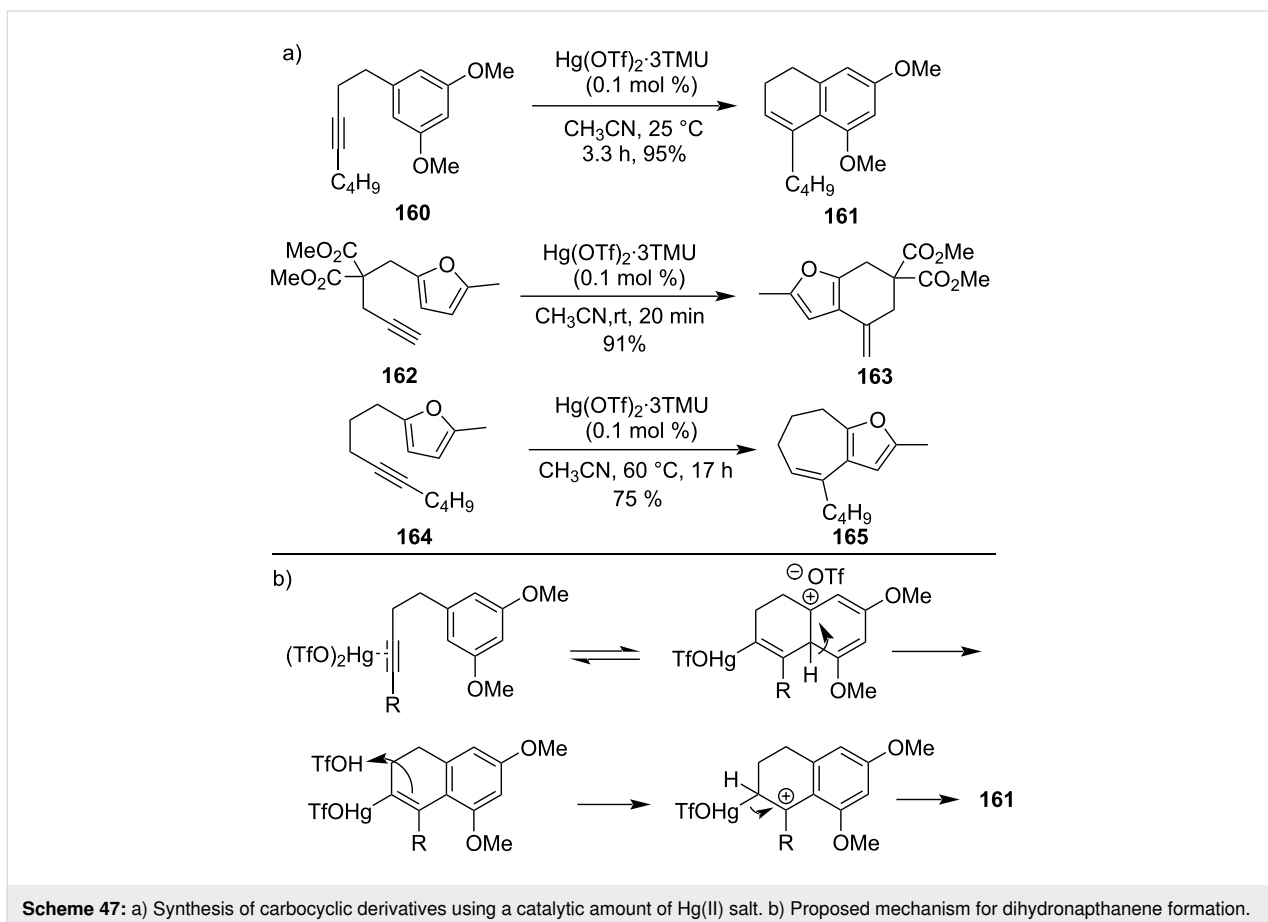
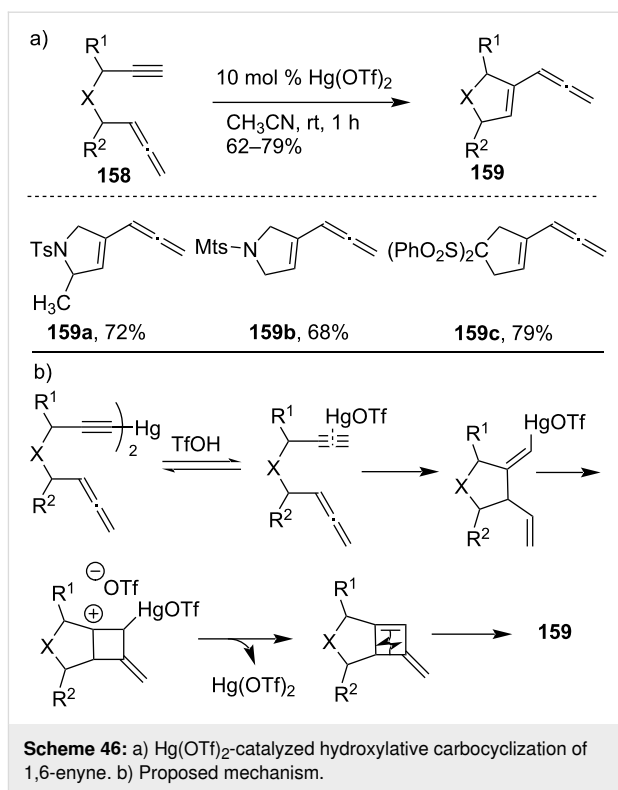
In a Hg(OTf)_2 -catalyzed process, 1,6-allenynes **158** were cycloisomerized to generate allenenes **159** in moderate to good yield (Scheme 46) [105]. However, depending on the substituents, allenene and/or unexpected triene were produced as a main product for disubstituted 1,6-allenynes. It was hypothesized based on experimental evidence that alkynes would first form a π -complex with Hg(OTf)_2 , followed by vinylmercuration, demercuration, and eventually isomerizes to allenene.



Scheme 44: Six-membered morpholine derivative formation by catalytic Hg(II)-salt-induced cyclization.

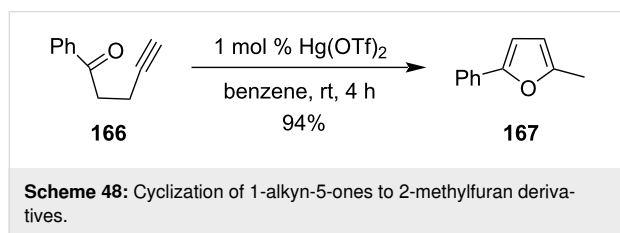


In 2003, a carbocyclization with a catalytic quantity of mercury salt was used to efficiently synthesize dihydronaphthalene derivative **161** from exemplifying benzyl derivative **160** [106]. The reported methodology was an example of the Hg(OTf)₂·(TMU)₃ complex promoting a moderate and efficient procedure for arylalkyne cyclization to directly afford dihydronaphthalene derivatives (Scheme 47). Later, Friedel–Crafts type reaction of alkynylfuran **162** and **164** were

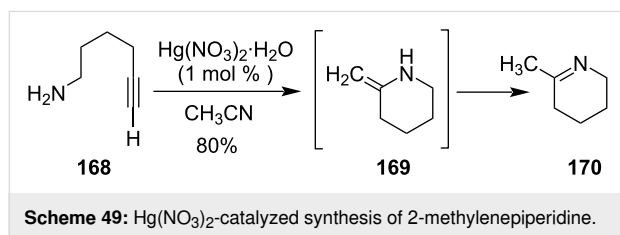


reported in presence of the $\text{Hg}(\text{OTf})_2 \cdot 0.1 \text{ Sc}(\text{OTf})_3$ complex (5 mol %) to form six- (**163**) and seven-membered rings (**165**) in good yield. For the cyclization at 2-position of the furan the $\text{Hg}(\text{OTf})_2 \cdot (\text{TMU})_3$ complex was used as catalyst [107].

Later it was reported that 1-alkyn-5-ones such as **166** also undergo an effective cyclization reaction to synthesize 2-methylfuran derivatives **167** with high yield in the presence of $\text{Hg}(\text{OTf})_2$ as the catalyst under very mild reaction conditions (Scheme 48) [108].



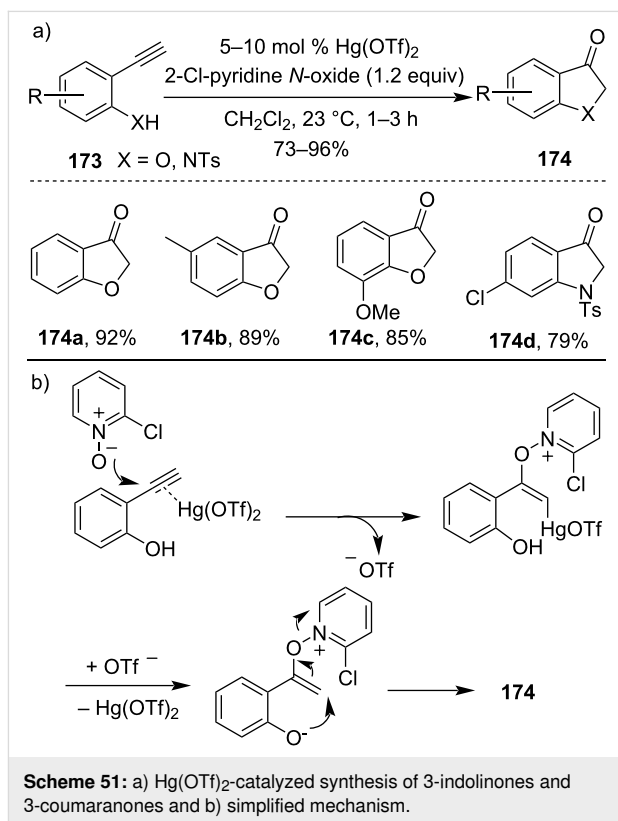
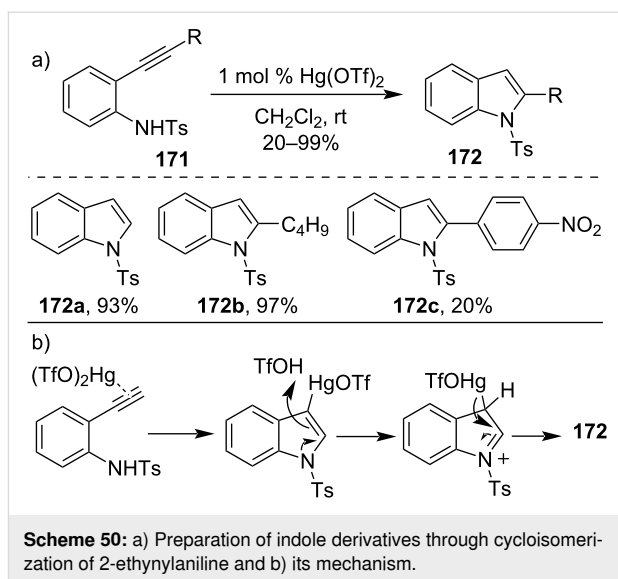
Cyclization of 6-aminohex-1-yne **168** was performed by catalytic amounts of $\text{Hg}(\text{NO}_3)_2$ and generated 2-methylenepiperidine **169** initially, which further isomerizes to form 2-methyl-1,2-dehydropiperidine (**170**, Scheme 49) [109].



Similarly, in 2007 Kurisaki et al. had also prepared indole derivatives **172** with excellent yields from 2-ethynylaniline derivatives **171** upon the treatment of catalytic amounts of $\text{Hg}(\text{OTf})_2$ at room temperature. It was an example of cycloisomerization of 2-ethynylaniline derivatives utilizing mild reaction conditions (Scheme 50) [110].

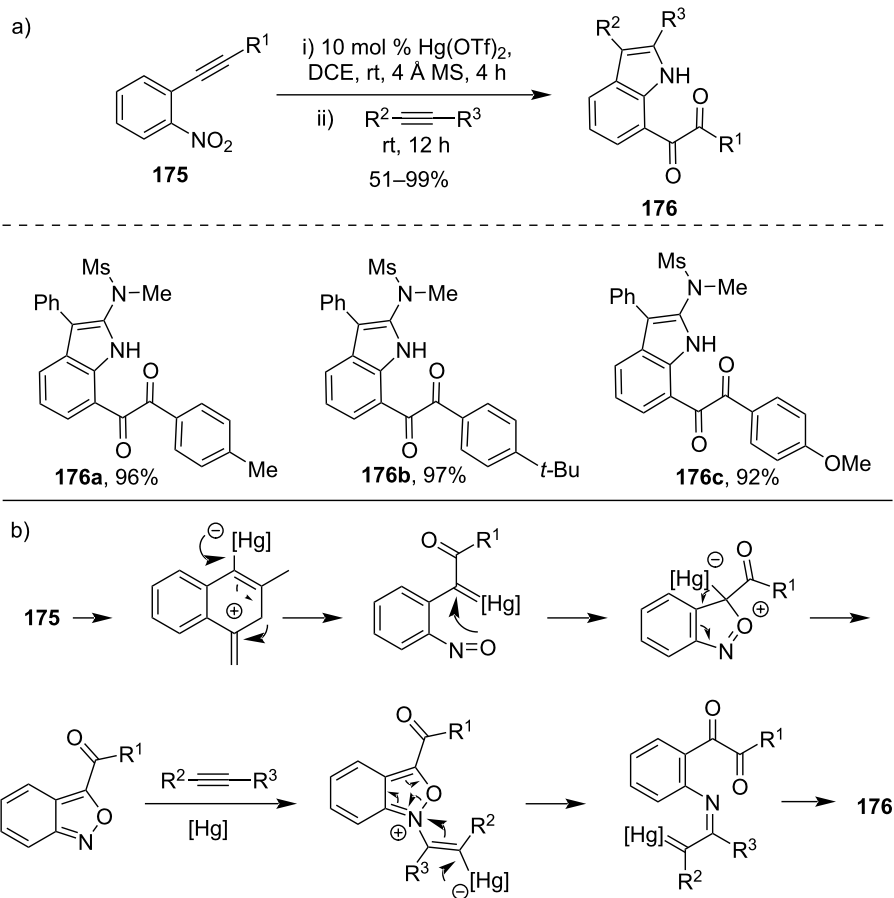
Rong et al. had demonstrated the $\text{Hg}(\text{II})$ -salt-catalyzed enolate umpolung reaction for the efficient synthesis of various 3-indolinones and 3-coumaranones **174**. They had proved that the reaction mechanism proceeds via activation of the alkynyl group with $\text{Hg}(\text{OTf})_2$ salt and addition of 2-chloropyridine *N*-oxide. The resulting activated alkynyl complex was demercurated, followed by the $\text{S}_{\text{N}}2'$ reaction thus formed undergoes demercuration to yield 3-coumaranone (Scheme 51) [111].

Recently, a $\text{Hg}(\text{OTf})_2$ -catalyzed one-pot cyclization of nitroalkyne **175** and alkyne had been reported to synthesize indole derivatives **176**. Based on this strategy, the one-pot



method to synthesize indole derivatives had been developed [112]. Similarly, benzo[*c*]isoxazole was also formed in excellent yields with high selectivity using this strategy. In these transformations, two Hg -carbene intermediates were proposed to be involved (Scheme 52).

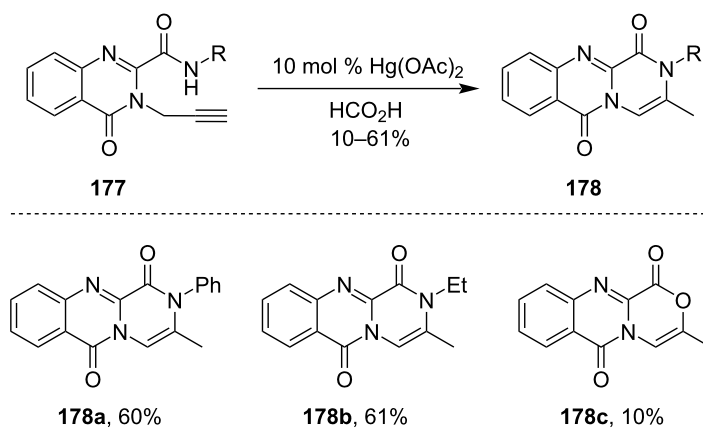
Mercury-catalyzed reactions were also well known for the formation of various complex scaffolds like tricyclic pyrazinones



Scheme 52: a) Hg(OTf)₂-catalyzed one pot cyclization of nitroalkyne and b) its plausible mechanism.

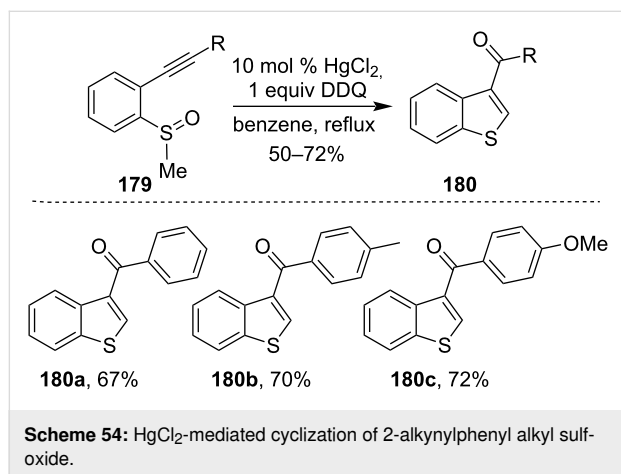
from the corresponding starting materials. For example, Zhang et al. showed that refluxing anilide **177** in presence of a catalytic amount of Hg(OAc)₂ and 90% formic acid gave the tricyclic

heterocyclic scaffold **178** [113]. It involved a two-step process with the rearrangement of the primary cyclization products (Scheme 53).



Scheme 53: Synthesis of tricyclic heterocyclic scaffolds.

In 2013, Lin et al. reported a Hg(II) chloride-mediated cyclization reaction of 2-alkynylphenyl alkyl sulfoxide **179** to synthesize benzothiophene derivatives **180** with good yields [114]. In this case, the reaction was believed to proceed via the initial formation of metal carbenoids followed by a sequential C–H insertion and then oxidation (Scheme 54). This methodology was later successfully utilized for the total synthesis of raloxifene and benzo[*b*]thiophene derivatives [115].



Cyclization involving allenes (>C=C=C<)

Hg(II) triflate salts had also been successfully employed for the aryllallene **181** cyclization by Yamamoto et al [116]. The cata-

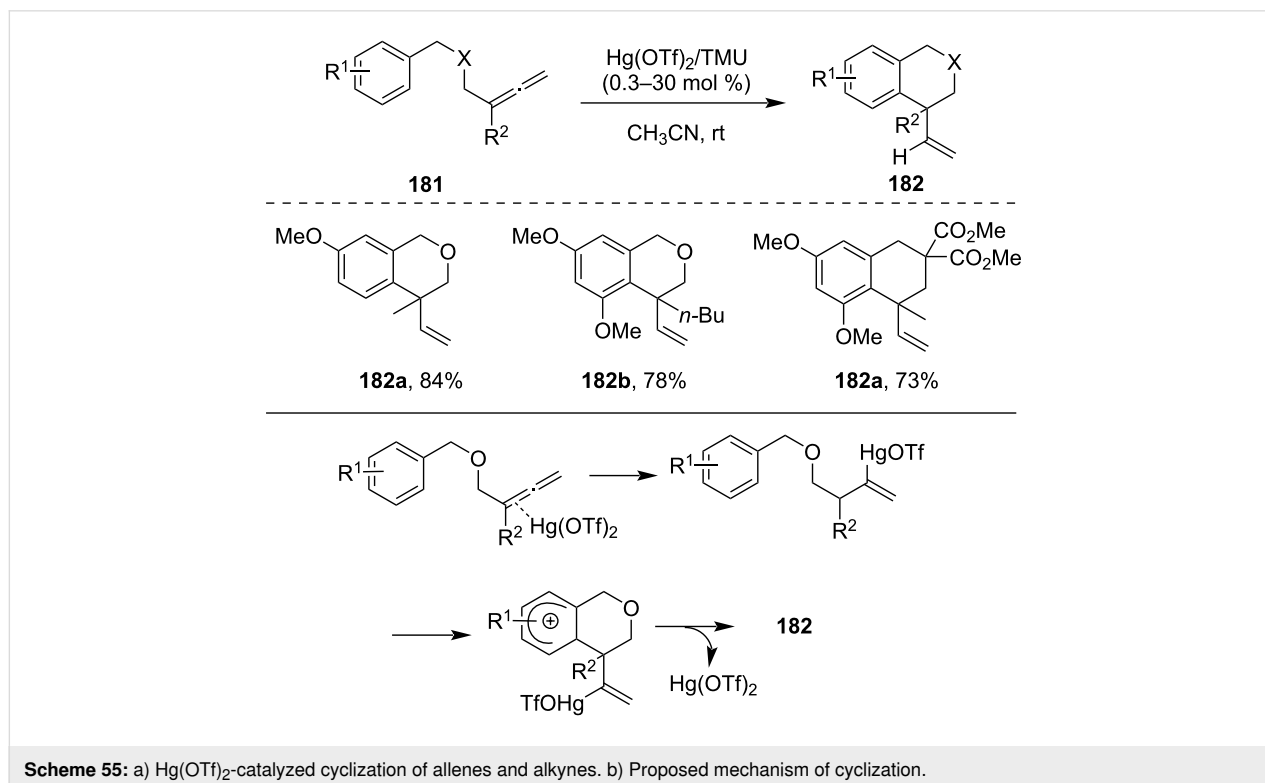
lytic pathway was proved to involve the direct H-transfer to the vinylmercury complex from the aromatic ring. It involved Hg(OTf)₂-catalyzed cyclization of aryl 1,1-disubstituted allenes with the formation of a quaternary carbon center followed by the formation of a cationic vinylmercury intermediate (Scheme 55).

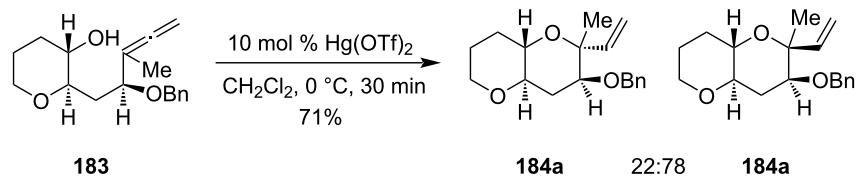
For the synthesis of stereoselective tetrahydropyran derivatives **184**, Hg(II)-catalyzed cyclization proved to be more effective than silver(I)-salt-mediated cyclization. It showed that methyl-substituted allenes undergo efficient cyclization to form polycyclic ethers under Hg(OTf)₂-catalyzed conditions at lower temperature (Scheme 56) [117].

Following a similar strategy, mercury chlorate (Hg(ClO₄)₂) had been employed successfully as a cheap alternative to precious metals salts in the cyclization of α -allenol derivatives **185** to differently substituted 2,5-dihydrofurans **186** in an efficient and selective manner. It was also shown that from enantiopure allenyl derivatives, the desired pure cyclized product was generated by utilizing the above reaction conditions (Scheme 57) [118].

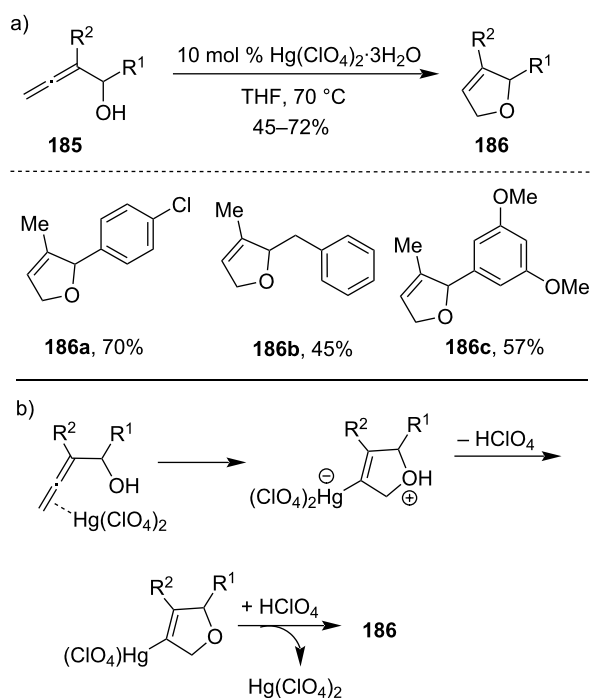
Mercury(II)-salt-mediated cyclization in total synthesis

Apart from previously described cyclization reactions, there were examples where a Hg(II)-salt-mediated cyclization had





Scheme 56: Stereoselective synthesis of tetrahydropyran derivatives.

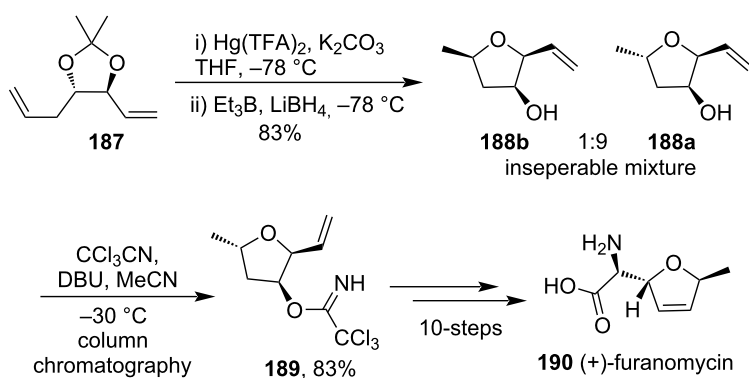
Scheme 57: a) $\text{Hg}(\text{ClO}_4)_2$ -catalyzed cyclization of α -allenol derivatives. b) Simplified mechanism.

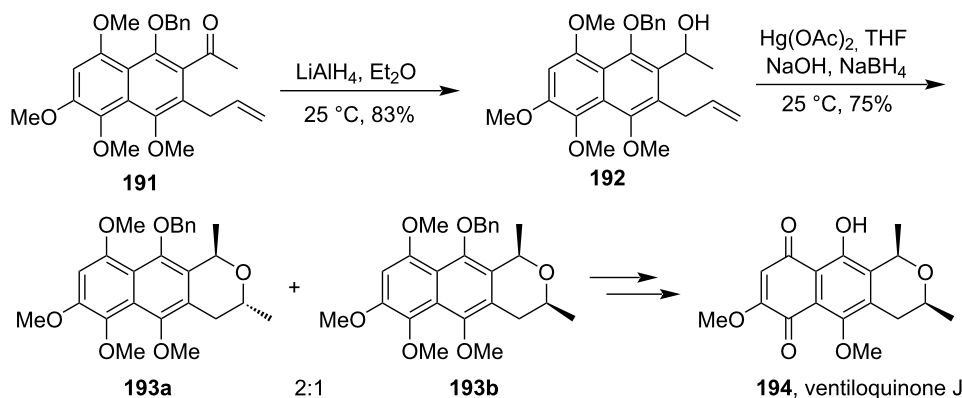
been successfully employed as one of the important steps during the total synthesis of natural products.

In 1998, a highly enantioselective total synthesis of (+)-furanomycin (**190**) was achieved through $\text{Hg}(\text{TFA})_2$ -promoted cyclization of γ -hydroxyalkene derivative **187** as an intermediate stage to give a mixture of diastereomeric tetrahydrofurans **188a** and **188b** (Scheme 58) [119].

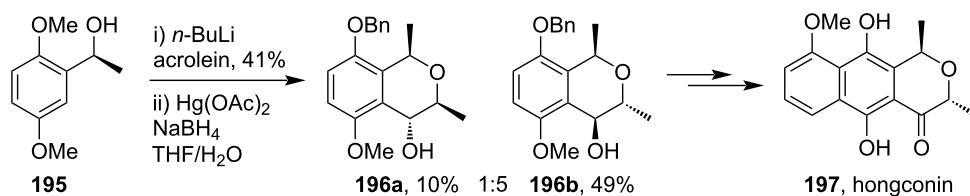
Later for the total synthesis of ventiloquinone **J** (**194**), a $\text{Hg}(\text{II})$ -salt-catalyzed intramolecular cyclization reaction of the *ortho*-allyl alcohol **192** was involved. The reaction went through the formation of a mixture of diastereoisomers **193a** and **193b**. After which the inseparable mixture of products underwent further oxidative demethylation and yielded the final products (Scheme 59) [120].

Kraus and co-workers had reported the synthesis of the racemic naphthohydroquinone hongconin (**197**) starting from the *ortho*-allyl alcohol derivative **195**. The starting material was cyclized using a $\text{Hg}(\text{II})$ salts to get an inseparable mixture of products **196a** and **196b** in the ratio of 1:5 [121]. The synthetic route proceeded by benzylic alcohol *ortho*-metalation followed by a regioselective mercuri-cyclization reaction (Scheme 60).

Scheme 58: $\text{Hg}(\text{TFA})_2$ -promoted cyclization of a γ -hydroxy alkene derivative.



Scheme 59: Synthesis Hg(II)-salt-mediated cyclization of allyl alcohol for the construction of ventiloquinone J.

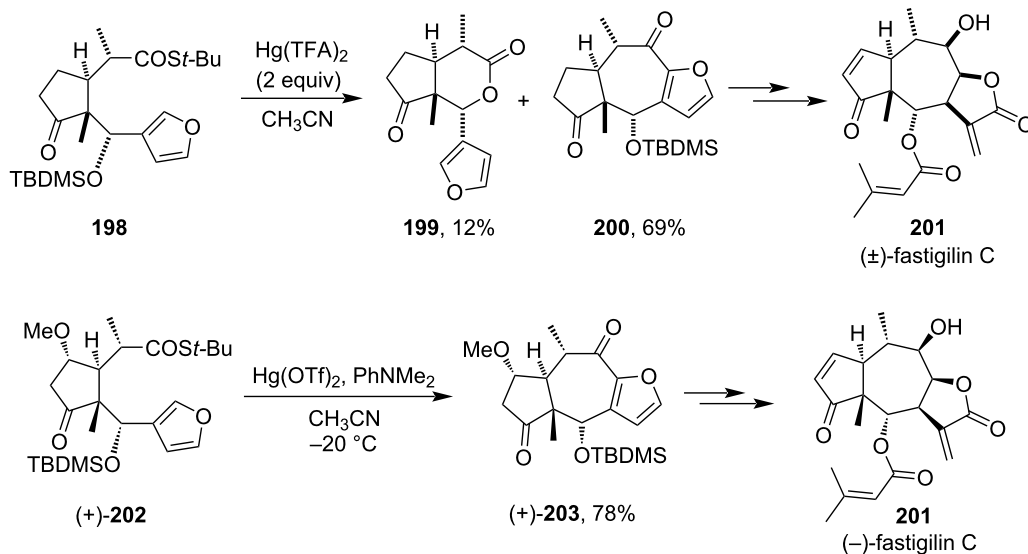


Scheme 60: Hg(OAc)₂-mediated cyclization as a key step for the synthesis of hongconin.

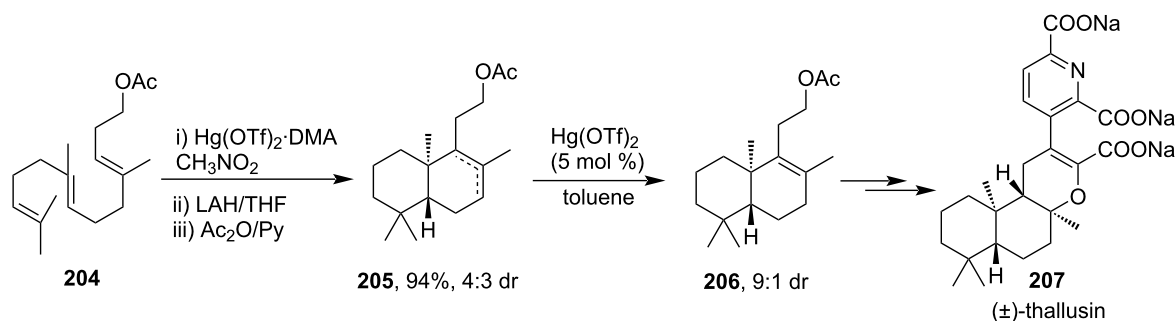
During the total syntheses of (±)-fastigilin C and (–)-fastigilin C (**201**), 2 equiv of Hg(TFA)₂ were used to synthesize key intermediate tricyclic furan compounds **199**, **200**, and **203**. Hg(TFA)₂ helped in the desired ring-formation reaction of compound **198** to afford two cyclic compounds **199** and **200** in the ratio of 6:1 with an overall 81% yield [122]. But in the case of

methoxy-substituted derivative **202** only one furan derivative **203** was formed (Scheme 61).

In 2007, Nishizawa et al. successfully utilized a Hg(OTf)₂-induced cyclization in the key step for the synthesis of (±)-thalusulin (**207**, Scheme 62). A complex mixture of Hg(OTf)₂ and



Scheme 61: Examples of Hg(II)-salt-mediated cyclized ring formation in the syntheses of (±)-fastigilin C and (–)-fastigilin C.



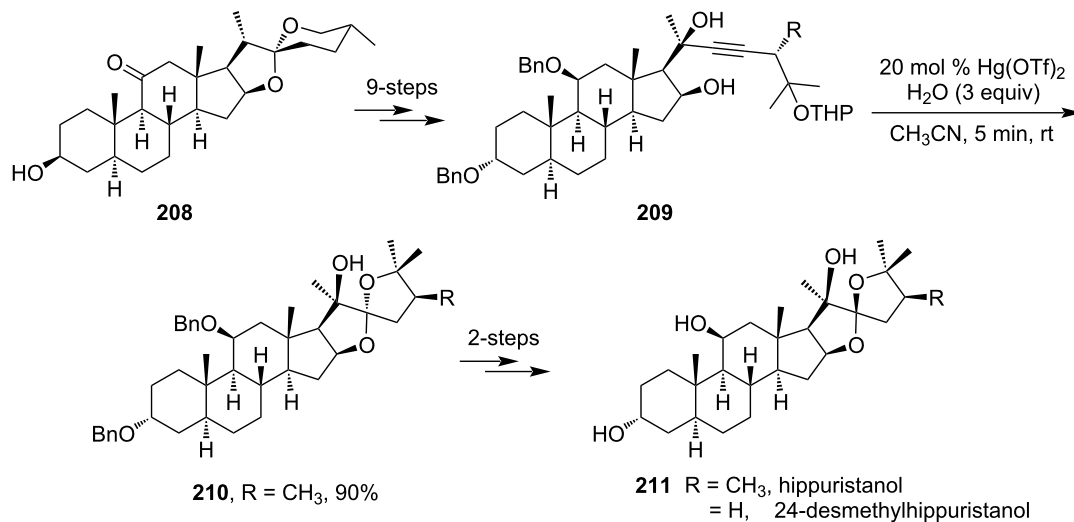
Scheme 62: Formal synthesis of (±)-thallusin.

N,N-dimethylaniline (DMA) (1.2 equiv) was initially used for olefin cyclization to produce a regio- and a stereoisomeric mixture of acetate **205** after reduction and acetylation of the crude product. The Hg(OTf)_2 -catalyzed isomerization of the double bond in compound **205** yielded thermodynamically favorable isomer **206** as a major product [123].

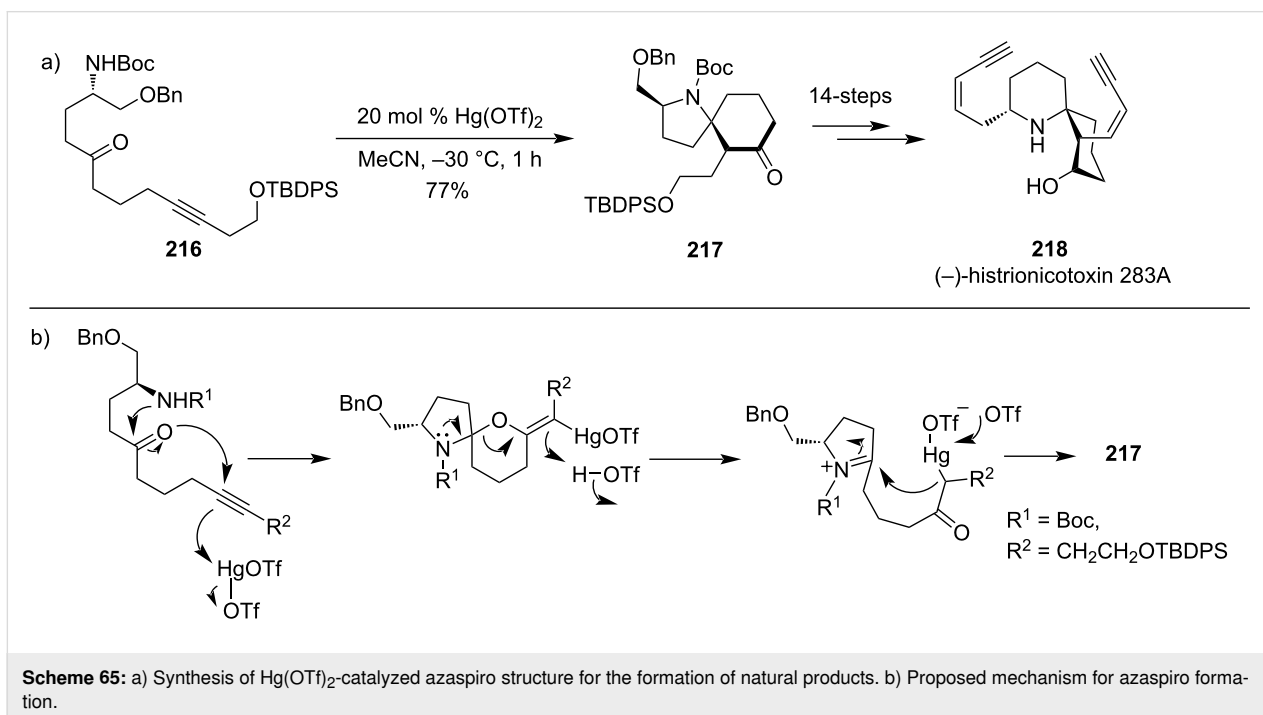
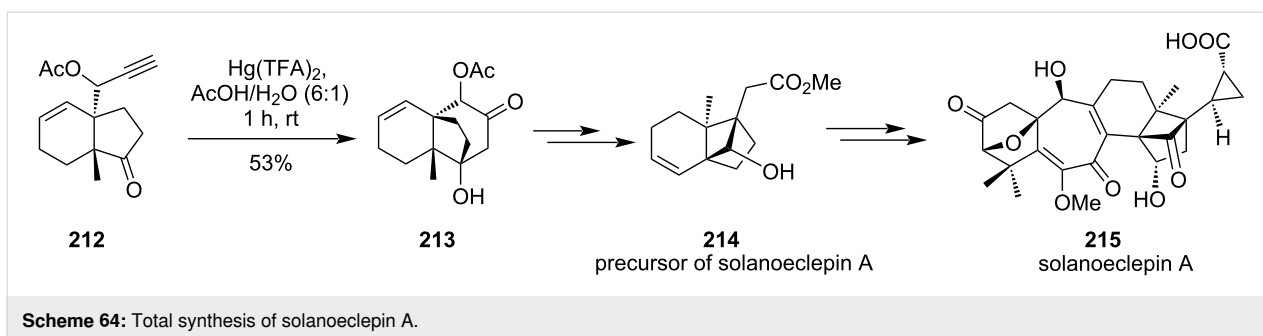
In 2010, Ravindar et al. developed the total synthesis of the steroidal natural product hippuristanol (**211**) starting from 11-ketotigogenin **208** (Scheme 63). They had utilized a Hg(OTf)_2 -catalyzed spiroketalization reaction in their key step to form the desired ketal intermediate **210** [124]. Subsequently, in 2011, the same group reported another synthesis of hippuristanol (**211**) and its analog from easily available starting materials [125]. In this work, hippuristanol and some analogs were successfully synthesized utilizing a Hg(OTf)_2 -catalyzed cascade spiroketalization step of the 3-alkyne-1,7-diol motif. The Hg(OTf)_2 -catalyzed cascade spiroketalization step was proved to be more convenient than Suárez cyclization.

A Hg(TFA)_2 -mediated cyclization was efficiently utilized for the synthesis of highly strained tricyclo[5.2.1.0^{1,6}]decene intermediate **214** containing a cyclobutane ring (Scheme 64). Compound **213** is an important precursor for the asymmetric total synthesis of solanoelepin A. The formation of β -hydroxyketone **213** was achieved by Hg(TFA)_2 -mediated cyclization of compound **212** using TFA/H₂O (1.7: 1) as the solvent. In presence of other mercury compounds like HgO and Hg(OAc)_2 no product or starting material was recovered [126].

Spiro-skeleton structures are found in many natural products and synthesizing stereoselective spiro-skeletons has always been difficult for organic chemists. Morimoto and co-workers were the first to disclose the Hg(OTf)_2 -catalyzed cycloisomerization of amino ynone to produce the azaspiro skeleton. Later, this methodology was successfully used for the synthesis of several spiro-skeleton structures. Natural products such as histri-*onicot*oxin alkaloids **218** (Scheme 65) [127,128] and lepadiformine [129,130] were being successfully synthesized using



Scheme 63: Total synthesis of hippuristanol and its analog.



this methodology for spirocyclic ring structure synthesis. The proposed mechanism proceeded initially with aminoketal formation by 6-*exo*-dig intramolecular oxymercuration, followed by Petasis–Ferrier-type cyclization, and finally nucleophilic addition of mercuric enolate to iminium results in the formation of azaspiro structure.

Conclusion

In conclusion, this review summarizes $\text{Hg}(\text{II})$ -salt-mediated cyclization reactions either for direct synthesis of cyclized products or as a part of the total synthesis of important natural products. Different $\text{Hg}(\text{II})$ salts were used stoichiometrically or catalytically depending upon the nature of functional groups and reactants. However, the reactivity of different unsaturated bonds involved in the cyclization primarily depends on the nucleophile as well as nature of functional groups attached to unsaturated bonds. When alkenes are linked to activating groups like methoxy or hydroxy, a catalytic quantity of $\text{Hg}(\text{II})$ ions is re-

quired for cyclization. Nothing can be predicted in case of alkynes; however, the presence of a strong nucleophile promotes the $\text{Hg}(\text{II})$ -salt-catalyzed cyclization of allenes in most circumstances. In cyclization reactions, $\text{Hg}(\text{OTf})_2$ showed to be the most effective and versatile of all $\text{Hg}(\text{II})$ salts. Mercury(II) salts can also be used to cyclize unsaturated bonds in a regio- and diastereoselective manner. Apart from toxicity concerns, $\text{Hg}(\text{II})$ salts are cheap, stable, and versatile in terms of reactivity, making them a viable option to similar transition metal catalysts.

Acknowledgements

The authors would like to express their gratitude to anonymous reviewers for their valuable comments about the manuscript.

Funding

This work was supported by Department of Chemistry, Cooch Behar Panchanan Barma University.

ORCID® iDs

Sumana Mandal - <https://orcid.org/0000-0001-9385-3288>Goutam Biswas - <https://orcid.org/0000-0002-1488-2003>

References

- Lautens, M.; Klute, W.; Tam, W. *Chem. Rev.* **1996**, *96*, 49–92. doi:10.1021/cr950016l
- Zhou, F.; Li, C.-J. *Chem. Sci.* **2019**, *10*, 34–46. doi:10.1039/c8sc04271c
- Lou, J.; Wang, Q.; Wu, P.; Wang, H.; Zhou, Y.-G.; Yu, Z. *Chem. Soc. Rev.* **2020**, *49*, 4307–4359. doi:10.1039/c9cs00837c
- Beller, M.; Bolm, C., Eds. *Transition Metals for Organic Synthesis: Building Blocks and Fine Chemicals*, 1st ed.; Wiley-VCH: Weinheim, Germany, 2004. doi:10.1002/9783527619405
- Sharma, R.; Kour, P.; Kumar, A. *J. Chem. Sci.* **2018**, *130*, 73. doi:10.1007/s12039-018-1466-8
- Tanaka, K., Ed. *Transition-Metal-Mediated Aromatic Ring Construction*; John Wiley & Sons: Hoboken, NJ, USA, 2013. doi:10.1002/9781118629871
- Solé, D.; Fernández, I., Eds. *Advances in Transition-Metal Mediated Heterocyclic Synthesis*; Academic Press: London, UK, 2018.
- Biffis, A.; Centomo, P.; Del Zotto, A.; Zecca, M. *Chem. Rev.* **2018**, *118*, 2249–2295. doi:10.1021/acs.chemrev.7b00443
- Saranya, S.; Rohit, K. R.; Radhika, S.; Anilkumar, G. *Org. Biomol. Chem.* **2019**, *17*, 8048–8061. doi:10.1039/c9ob01538h
- Ruiz-Castillo, P.; Buchwald, S. L. *Chem. Rev.* **2016**, *116*, 12564–12649. doi:10.1021/acs.chemrev.6b00512
- Duarah, G.; Kaishap, P. P.; Begum, T.; Gogoi, S. *Adv. Synth. Catal.* **2019**, *361*, 654–672. doi:10.1002/adsc.201800755
- Seretis, A.; Diamantopoulou, P.; Thanou, I.; Tzevelekidis, P.; Fakas, C.; Lilas, P.; Papadogianakis, G. *Front. Chem. (Lausanne, Switz.)* **2020**, *8*, 221. doi:10.3389/fchem.2020.00221
- Johansson, J. R.; Beke-Somfai, T.; Said Stålsmeden, A.; Kann, N. *Chem. Rev.* **2016**, *116*, 14726–14768. doi:10.1021/acs.chemrev.6b00466
- Fan, X.; Liu, C.-H.; Yu, Z.-X. Rhodium(I)-Catalyzed Cycloadditions Involving Vinylcyclopropanes and Their Derivatives. In *Rhodium Catalysis in Organic Synthesis*; Tanaka, K., Ed.; Wiley-VCH: Weinheim, Germany, 2019; pp 229–276. doi:10.1002/9783527811908.ch10
- Wee, A. G. H. *Curr. Org. Synth.* **2006**, *3*, 499–555. doi:10.2174/157017906778699512
- Evans, P. A., Ed. *Modern Rhodium-Catalyzed Organic Reactions*, 1st ed.; Wiley-VCH: Weinheim, Germany, 2005. doi:10.1002/3527604693
- Mondal, M.; Bora, U. *RSC Adv.* **2013**, *3*, 18716. doi:10.1039/c3ra42480d
- Carney, J. R.; Dillon, B. R.; Thomas, S. P. *Eur. J. Org. Chem.* **2016**, 3912–3929. doi:10.1002/ejoc.201600018
- Concellón, J. M.; Rodríguez-Solla, H.; del Amo, V. *Chem. – Eur. J.* **2008**, *14*, 10184–10191. doi:10.1002/chem.200800796
- Hashmi, A. S. K. *Chem. Rev.* **2007**, *107*, 3180–3211. doi:10.1021/cr000436x
- Campeau, D.; León Rayo, D. F.; Mansour, A.; Muratov, K.; Gagosz, F. *Chem. Rev.* **2021**, *121*, 8756–8867. doi:10.1021/acs.chemrev.0c00788
- Boorman, T. C.; Larrosa, I. *Chem. Soc. Rev.* **2011**, *40*, 1910–1925. doi:10.1039/c0cs00098a
- Abbiati, G.; Rossi, E. *Beilstein J. Org. Chem.* **2014**, *10*, 481–513. doi:10.3762/bjoc.10.46
- Wu, Y.-C.; Xiao, Y.-T.; Yang, Y.-Z.; Song, R.-J.; Li, J.-H. *ChemCatChem* **2020**, *12*, 5312–5329. doi:10.1002/cctc.202000900
- Sekine, K.; Yamada, T. *Chem. Soc. Rev.* **2016**, *45*, 4524–4532. doi:10.1039/c5cs00895f
- Larock, R. C. Mercury. *Comprehensive Organometallic Chemistry II*; Elsevier: Amsterdam, Netherlands, 1995; pp 389–435. doi:10.1016/b978-008046519-7.00100-3
- Vogt, R. R.; Nieuwland, J. A. *J. Am. Chem. Soc.* **1921**, *43*, 2071–2081. doi:10.1021/ja01442a010
- Ramalingan, C.; Park, Y.-T. *J. Org. Chem.* **2007**, *72*, 4536–4538. doi:10.1021/jo070297k
- Gong, Y.; Cao, Z.-Y.; Shi, Y.-B.; Zhou, F.; Zhou, Y.; Zhou, J. *Org. Chem. Front.* **2019**, *6*, 3989–3995. doi:10.1039/c9qo01049a
- Masuhara, T. *Nihon Shika Ishikai Zasshi* **1975**, *28*, 128.
- Clarkson, T. W.; Magos, L. *Crit. Rev. Toxicol.* **2006**, *36*, 609–662. doi:10.1080/10408440600845619
- Leyva-Pérez, A.; Corma, A. *Angew. Chem., Int. Ed.* **2012**, *51*, 614–635. doi:10.1002/anie.201101726
- Larock, R. C. *Angew. Chem., Int. Ed. Engl.* **1978**, *17*, 27–37. doi:10.1002/anie.197800271
- Lattes, A. *Chem. Heterocycl. Compd.* **1975**, *11*, 4–17. doi:10.1007/bf00945261
- Kaur, N. *Synth. Commun.* **2018**, *48*, 2715–2749. doi:10.1080/00397911.2018.1497657
- Sand, J. *Ber. Dtsch. Chem. Ges.* **1901**, *34*, 1385–1394. doi:10.1002/cber.19010340207
- Biilmann, E. *Ber. Dtsch. Chem. Ges.* **1900**, *33*, 1641–1655. doi:10.1002/cber.19000330235
- Summerbell, R. K.; Stephens, J. R. *J. Am. Chem. Soc.* **1954**, *76*, 6401–6407. doi:10.1021/ja01653a046
- Summerbell, R. K.; Stephens, J. R. *J. Am. Chem. Soc.* **1954**, *76*, 731–734. doi:10.1021/ja01632a028
- Summerbell, R. K.; Lestina, G.; Waite, H. J. *J. Am. Chem. Soc.* **1957**, *79*, 234–237. doi:10.1021/ja01558a061
- Pougny, J.-R.; Nassr, M. A. M.; Sinaÿ, P. *J. Chem. Soc., Chem. Commun.* **1981**, 375–376. doi:10.1039/c39810000375
- Boschetti, A.; Nicotra, F.; Panza, L.; Russo, G. *J. Org. Chem.* **1988**, *53*, 4181–4185. doi:10.1021/jo00253a005
- Nolen, E. G.; Kurish, A. J.; Potter, J. M.; Donahue, L. A.; Orlando, M. D. *Org. Lett.* **2005**, *7*, 3383–3386. doi:10.1021/ol051341q
- Nicotra, F.; Panza, L.; Ronchetti, F.; Toma, L. *Tetrahedron Lett.* **1984**, *25*, 5937–5938. doi:10.1016/s0040-4039(01)81725-1
- Reitz, A. B.; Nortey, S. O.; Maryanoff, B. E. *Tetrahedron Lett.* **1985**, *26*, 3915–3918. doi:10.1016/s0040-4039(00)98686-6
- Andrey, O.; Glanzmann, C.; Landais, Y.; Parra-Rapado, L. *Tetrahedron* **1997**, *53*, 2835–2854. doi:10.1016/s0040-4020(97)00003-3
- Pulido, F. J.; Barbero, A.; Val, P.; Diez, A.; González-Ortega, A. *Eur. J. Org. Chem.* **2012**, 5350–5356. doi:10.1002/ejoc.201200666
- Mohapatra, D. K.; Mohapatra, S.; Gurjar, M. K. *Tetrahedron Lett.* **2006**, *47*, 5943–5947. doi:10.1016/j.tetlet.2006.06.049
- Enderlin, G.; Nielsen, P. *J. Org. Chem.* **2008**, *73*, 6891–6894. doi:10.1021/jo801081t
- Perie, J. J.; Laval, J. P.; Roussel, J.; Lattes, A. *Tetrahedron* **1972**, *28*, 675–699. doi:10.1016/0040-4020(72)84031-6
- Roussel, J.; Perie, J. J.; Laval, J. P.; Lattes, A. *Tetrahedron* **1972**, *28*, 701–716. doi:10.1016/0040-4020(72)84032-8

52. Tokuda, M.; Yamada, Y.; Takagi, T.; Sugimoto, H.; Furusaki, A. *Tetrahedron* **1987**, *43*, 281–296. doi:10.1016/s0040-4020(01)89956-7
53. Roubaud, V.; Moigne, F. L.; Mercier, A.; Tordo, P. *Phosphorus, Sulfur Silicon Relat. Elem.* **1994**, *86*, 39–54. doi:10.1080/10426509408018386
54. Le Moigne, F.; Mercier, A.; Tordo, P. *Tetrahedron Lett.* **1991**, *32*, 3841–3844. doi:10.1016/s0040-4039(00)79391-9
55. Stipa, P.; Finet, J. P.; Le Moigne, F.; Tordo, P. *J. Org. Chem.* **1993**, *58*, 4465–4468. doi:10.1021/jo00068a049
56. Dembkowski, L.; Finet, J. P.; Fréjaville, C.; Le Moigne, F.; Maurin, R.; Mercier, A.; Pages, P.; Stipa, P.; Tordo, P. *Free Radical Res. Commun.* **1993**, *19* (Suppl. 1), s23–s32. doi:10.3109/10715769309056s23
57. Le Moigne, F.; Tordo, P. *J. Org. Chem.* **1994**, *59*, 3365–3367. doi:10.1021/jo00091a024
58. Roubaud, V.; Moigne, F. L.; Mercier, A.; Tordo, P. *Synth. Commun.* **1996**, *26*, 1507–1516. doi:10.1080/00397919608003517
59. Barluenga, J.; Pérez-Prieto, J.; Bayón, A. M.; Asensio, G. *Tetrahedron* **1984**, *40*, 1199–1204. doi:10.1016/s0040-4020(01)99327-5
60. Barluenga, J.; Pérez-Prieto, J.; Asensio, G.; García-Granda, S.; Salvado, M. A. *Tetrahedron* **1992**, *48*, 3813–3826. doi:10.1016/s0040-4020(01)92272-0
61. Carroll, F. I.; Hu, X.; Navarro, H. A.; Deschamps, J.; Abdrakhmanova, G. R.; Damaj, M. I.; Martin, B. R. *J. Med. Chem.* **2006**, *49*, 3244–3250. doi:10.1021/jm060122n
62. Peçanha, E. P.; Verli, H.; Rodrigues, C. R.; Barreiro, E. J.; Fraga, C. A. M. *Tetrahedron Lett.* **2002**, *43*, 1607–1611. doi:10.1016/s0040-4039(02)00074-6
63. Nixon, J. R.; Cudd, M. A.; Porter, N. A. *J. Org. Chem.* **1978**, *43*, 4048–4052. doi:10.1021/jo00415a014
64. Bloodworth, A. J.; Tallant, N. A. *J. Chem. Soc., Chem. Commun.* **1992**, 428–429. doi:10.1039/c39920000428
65. Kurbanov, M.; Semenovskiy, A. V.; Smit, W. A.; Shmelev, L. V.; Kucherov, V. F. *Tetrahedron Lett.* **1972**, *13*, 2175–2178. doi:10.1016/s0040-4039(01)84798-5
66. Hoyer, T. R.; Caruso, A. J.; Kurth, M. J. *J. Org. Chem.* **1981**, *46*, 3550–3552. doi:10.1021/jo00330a037
67. Amoroso, R.; Cardillo, G.; Tomasini, C. *Tetrahedron Lett.* **1990**, *31*, 6413–6416. doi:10.1016/s0040-4039(00)97079-5
68. Cardillo, G.; Amoroso, R.; Cardillo, G.; Tomasini, C. *Heterocycles* **1992**, *34*, 349. doi:10.3987/com-91-5930
69. Harding, K. E.; Stephens, R.; Hollingsworth, D. R. *Tetrahedron Lett.* **1984**, *25*, 4631–4632. doi:10.1016/s0040-4039(01)91218-3
70. Harding, K. E.; Hollingsworth, D. R. *Tetrahedron Lett.* **1988**, *29*, 3789–3792. doi:10.1016/s0040-4039(00)82115-2
71. Takacs, J. M.; Helle, M. A.; Yang, L. *Tetrahedron Lett.* **1989**, *30*, 1777–1780. doi:10.1016/s0040-4039(00)99577-7
72. Takacs, J. M.; Helle, M. A.; Takusagawa, F. *Tetrahedron Lett.* **1989**, *30*, 7321–7324. doi:10.1016/s0040-4039(00)70687-3
73. Majumdar, K. C.; Thyagarajan, B. S. *J. Heterocycl. Chem.* **1972**, *9*, 489–494. doi:10.1002/jhet.5570090305
74. Bates, D. K.; Jones, M. C. *J. Org. Chem.* **1978**, *43*, 3856–3861. doi:10.1021/jo00414a015
75. Bates, D. K.; Jones, M. C. *J. Org. Chem.* **1978**, *43*, 3775–3776. doi:10.1021/jo00413a032
76. Thyagarajan, B. S.; Majumdar, K. C.; Bates, D. K. *J. Heterocycl. Chem.* **1975**, *12*, 59–66. doi:10.1002/jhet.5570120110
77. Balasubramanian, K. K.; Virupaksha Reddy, K.; Nagarajan, R. *Tetrahedron Lett.* **1973**, *14*, 5003–5004. doi:10.1016/s0040-4039(01)87633-4
78. Drouin, J.; Boaventura, M. A.; Conia, J. M. *J. Am. Chem. Soc.* **1985**, *107*, 1726–1729. doi:10.1021/ja00292a045
79. Riediker, M.; Schwartz, J. J. *Am. Chem. Soc.* **1982**, *104*, 5842–5844. doi:10.1021/ja00385a074
80. Compennolle, F.; Mao, H.; Tahri, A.; Kozlecki, T.; Van der Eycken, E.; Medaer, B.; Hoornaert, G. J. *Tetrahedron Lett.* **2002**, *43*, 3011–3015. doi:10.1016/s0040-4039(02)00404-5
81. Mao, H.; Koukni, M.; Kozlecki, T.; Compennolle, F.; Hoornaert, G. J. *Tetrahedron Lett.* **2002**, *43*, 8697–8700. doi:10.1016/s0040-4039(02)02111-1
82. Atta, A. K.; Kim, S.-B.; Heo, J.; Cho, D.-G. *Org. Lett.* **2013**, *15*, 1072–1075. doi:10.1021/ol4000873
83. Larock, R. C.; Liu, C. L. *J. Org. Chem.* **1983**, *48*, 2151–2158. doi:10.1021/jo00161a007
84. Larock, R. C.; Harrison, L. W. *J. Am. Chem. Soc.* **1984**, *106*, 4218–4227. doi:10.1021/ja00327a026
85. Ghorai, S.; Bhattacharjya, A. *Org. Lett.* **2005**, *7*, 207–210. doi:10.1021/ol047893a
86. Biswas, G.; Ghorai, S.; Bhattacharjya, A. *Org. Lett.* **2006**, *8*, 313–316. doi:10.1021/ol0527274
87. Balasubramanian, T.; Balasubramanian, K. K. *Tetrahedron Lett.* **1991**, *32*, 6641–6644. doi:10.1016/0040-4039(91)80243-y
88. Devan, B.; Rajagopalan, K. *Synth. Commun.* **1994**, *24*, 1691–1700. doi:10.1080/00397919408010171
89. Tan, D. S.; Schreiber, S. L. *Tetrahedron Lett.* **2000**, *41*, 9509–9513. doi:10.1016/s0040-4039(00)01658-0
90. Namba, K.; Nakagawa, Y.; Yamamoto, H.; Imagawa, H.; Nishizawa, M. *Synlett* **2008**, 1719–1723. doi:10.1055/s-2008-1077881
91. Yamamoto, H.; Ho, E.; Namba, K.; Imagawa, H.; Nishizawa, M. *Chem. – Eur. J.* **2010**, *16*, 11271–11274. doi:10.1002/chem.201001656
92. Namba, K.; Yamamoto, H.; Sasaki, I.; Mori, K.; Imagawa, H.; Nishizawa, M. *Org. Lett.* **2008**, *10*, 1767–1770. doi:10.1021/ol800450x
93. Namba, K.; Kanaki, M.; Suto, H.; Nishizawa, M.; Tanino, K. *Org. Lett.* **2012**, *14*, 1222–1225. doi:10.1021/ol2034492
94. Marson, C. M.; Harper, S.; Wrigglesworth, R. *J. Chem. Soc., Chem. Commun.* **1994**, 1879. doi:10.1039/c39940001879
95. Marson, C. M.; Campbell, J. *Tetrahedron Lett.* **1997**, *38*, 7785–7788. doi:10.1016/s0040-4039(97)01819-4
96. Marson, C. M.; Harper, S.; Oare, C. A.; Walsgrove, T. J. *J. Org. Chem.* **1998**, *63*, 3798–3799. doi:10.1021/jo9715837
97. Krafft, G. A.; Katzenellenbogen, J. A. *J. Am. Chem. Soc.* **1981**, *103*, 5459–5466. doi:10.1021/ja00408a030
98. Yamamoto, M. *J. Chem. Soc., Chem. Commun.* **1978**, 649–650. doi:10.1039/c39780000649
99. Yamamoto, M.; Yoshitake, M.; Yamada, K. *J. Chem. Soc., Chem. Commun.* **1983**, 991–992. doi:10.1039/c39830000991
100. Ravindar, K.; Sridhar Reddy, M.; Deslongchamps, P. *Org. Lett.* **2011**, *13*, 3178–3181. doi:10.1021/ol201102x
101. McDonald, F. E.; Ishida, K.; Hurtak, J. A. *Tetrahedron* **2013**, *69*, 7746–7758. doi:10.1016/j.tet.2013.04.138
102. Hurtak, J.; McDonald, F. *Synlett* **2017**, *28*, 2951–2955. doi:10.1055/s-0036-1588562
103. Yamamoto, M.; Tanaka, S.; Naruchi, K.; Yamada, K. *Synthesis* **1982**, 850–852. doi:10.1055/s-1982-29970
104. Nishizawa, M.; Yadav, V. K.; Skwarczynski, M.; Takao, H.; Imagawa, H.; Sugihara, T. *Org. Lett.* **2003**, *5*, 1609–1611. doi:10.1021/ol034201u

105. Sim, S. H.; Lee, S. I.; Seo, J.; Chung, Y. K. *J. Org. Chem.* **2007**, *72*, 9818–9821. doi:10.1021/jo701591e
106. Nishizawa, M.; Takao, H.; Yadav, V. K.; Imagawa, H.; Sugihara, T. *Org. Lett.* **2003**, *5*, 4563–4565. doi:10.1021/ol035622e
107. Yamamoto, H.; Sasaki, I.; Imagawa, H.; Nishizawa, M. *Org. Lett.* **2007**, *9*, 1399–1402. doi:10.1021/ol070335m
108. Imagawa, H.; Kurisaki, T.; Nishizawa, M. *Org. Lett.* **2004**, *6*, 3679–3681. doi:10.1021/ol048730p
109. Müller, T. E.; Pleier, A.-K. *J. Chem. Soc., Dalton Trans.* **1999**, 583. doi:10.1039/a808938h
110. Kurisaki, T.; Naniwa, T.; Yamamoto, H.; Imagawa, H.; Nishizawa, M. *Tetrahedron Lett.* **2007**, *48*, 1871–1874. doi:10.1016/j.tetlet.2006.12.120
111. Rong, Z.; Hu, W.; Dai, N.; Qian, G. *Org. Lett.* **2020**, *22*, 3286–3290. doi:10.1021/acs.orglett.0c01096
112. Zheng, M.; Chen, K.; Zhu, S. *Synthesis* **2017**, *49*, 4173–4182. doi:10.1055/s-0036-1588416
113. Zhang, J.; Haider, N. *ARKIVOC* **2016**, No. iii, 125–133. doi:10.3998/ark.5550190.p009.495
114. Lin, C.-H.; Chen, C.-C.; Wu, M.-J. *Chem. – Eur. J.* **2013**, *19*, 2578–2581. doi:10.1002/chem.201202704
115. Wen, S.-M.; Lin, C.-H.; Chen, C.-C.; Wu, M.-J. *Tetrahedron* **2018**, *74*, 2493–2499. doi:10.1016/j.tet.2018.03.067
116. Yamamoto, H.; Ueda, M.; Yamasaki, N.; Fujii, A.; Sasaki, I.; Igawa, K.; Kasai, Y.; Imagawa, H.; Nishizawa, M. *Org. Lett.* **2016**, *18*, 2864–2867. doi:10.1021/acs.orglett.6b01144
117. Suzuki, Y.; Kuwabara, A.; Koizumi, Y.; Mori, Y. *Tetrahedron* **2013**, *69*, 9086–9095. doi:10.1016/j.tet.2013.08.026
118. Alcaide, B.; Almendros, P.; Luna, A.; Soriano, E. *J. Org. Chem.* **2015**, *80*, 7050–7057. doi:10.1021/acs.joc.5b00887
119. Kang, S. H.; Lee, S. B. *Chem. Commun.* **1998**, 761–762. doi:10.1039/a800727f
120. Giles, R. G. F.; Green, I. R.; van Eeden, N. *Synthesis* **2004**, 1601–1608. doi:10.1055/s-2004-822370
121. Kraus, G. A.; Li, J.; Gordon, M.; Jensen, J. H. *J. Org. Chem.* **1994**, *59*, 2219–2222. doi:10.1021/jo00087a044
122. Tanis, S. P.; Robinson, E. D.; McMills, M. C.; Watt, W. *J. Am. Chem. Soc.* **1992**, *114*, 8349–8362. doi:10.1021/ja00048a003
123. Nishizawa, M.; Iyenaga, T.; Kurisaki, T.; Yamamoto, H.; Sharfuddin, M.; Namba, K.; Imagawa, H.; Shizuri, Y.; Matsuo, Y. *Tetrahedron Lett.* **2007**, *48*, 4229–4233. doi:10.1016/j.tetlet.2007.04.075
124. Ravindar, K.; Reddy, M. S.; Lindqvist, L.; Pelletier, J.; Deslongchamps, P. *Org. Lett.* **2010**, *12*, 4420–4423. doi:10.1021/ol1019663
125. Ravindar, K.; Reddy, M. S.; Lindqvist, L.; Pelletier, J.; Deslongchamps, P. *J. Org. Chem.* **2011**, *76*, 1269–1284. doi:10.1021/jo102054r
126. Chuang, H.-Y.; Isobe, M. *Org. Lett.* **2014**, *16*, 4166–4169. doi:10.1021/ol501858w
127. Matsumura, K.; Nishikawa, K.; Yoshida, H.; Doe, M.; Morimoto, Y. *RSC Adv.* **2018**, *8*, 11296–11303. doi:10.1039/c8ra02011f
128. Matsumura, K.; Nishikawa, K.; Yoshida, H.; Niwa, T.; Fushii, Y.; Doe, M.; Morimoto, Y. *Chem. – Asian J.* **2021**, *16*, 1882–1886. doi:10.1002/asia.202100383
129. Nishikawa, K.; Yamauchi, K.; Kikuchi, S.; Ezaki, S.; Koyama, T.; Nokubo, H.; Matsumura, K.; Kodama, T.; Kumagai, M.; Morimoto, Y. *Chem. – Eur. J.* **2017**, *23*, 9535–9545. doi:10.1002/chem.201701475
130. Nishikawa, K.; Kikuchi, S.; Ezaki, S.; Koyama, T.; Nokubo, H.; Kodama, T.; Tachi, Y.; Morimoto, Y. *Org. Lett.* **2015**, *17*, 5772–5775. doi:10.1021/acs.orglett.5b02867

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<https://creativecommons.org/licenses/by/4.0>). Please note that the reuse, redistribution and reproduction in particular requires that the author(s) and source are credited and that individual graphics may be subject to special legal provisions.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (<https://www.beilstein-journals.org/bjoc/terms>)

The definitive version of this article is the electronic one which can be found at: <https://doi.org/10.3762/bjoc.17.153>