

Article

# Structures and Biogenesis of Fallaxosides D<sub>4</sub>, D<sub>5</sub>, D<sub>6</sub> and D<sub>7</sub>, Trisulfated Non-Holostane Triterpene Glycosides from the Sea Cucumber *Cucumaria fallax*

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**Abstract:** Four new trisulfated triterpene glycosides, fallaxosides D<sub>4</sub> (1), D<sub>5</sub> (2), D<sub>6</sub> (3) and D<sub>7</sub> (4) have been isolated from the sea cucumber *Cucumaria fallax* (Cucumariidae, Dendrochirotida). The structures of the glycosides have been elucidated by 2D NMR spectroscopy and HRESIMS. All the glycosides have the lanostane aglycones of a rare non-holostane type with 7(8)-, 8(9)- or 9(11)-double bonds, one or two hydroxyl groups occupying unusual positions in the polycyclic nucleus and shortened or normal side chains. The pentasaccharide carbohydrate moieties of 1–4 have three sulfate groups. The cytotoxic activity of glycosides 1–4 against the ascite form of mouse Ehrlich carcinoma cells and mouse spleen lymphocytes and hemolytic activity against mouse erythrocytes have been studied.

**Keywords:** sea cucumbers; *Cucumaria fallax*; triterpene glycosides; fallaxosides; non-holostane glycosides

## 1. Introduction

Triterpene glycosides are characteristic metabolites of sea cucumbers (class Holothurioidea, phylum Echinodermata). Majority of them contain holostane type aglycones (lanostane derivatives with 18(20)-lactone) [1], but rare non-holostane aglycones, i.e., lanostane derivatives have not 18(20)-lactone [2–7].

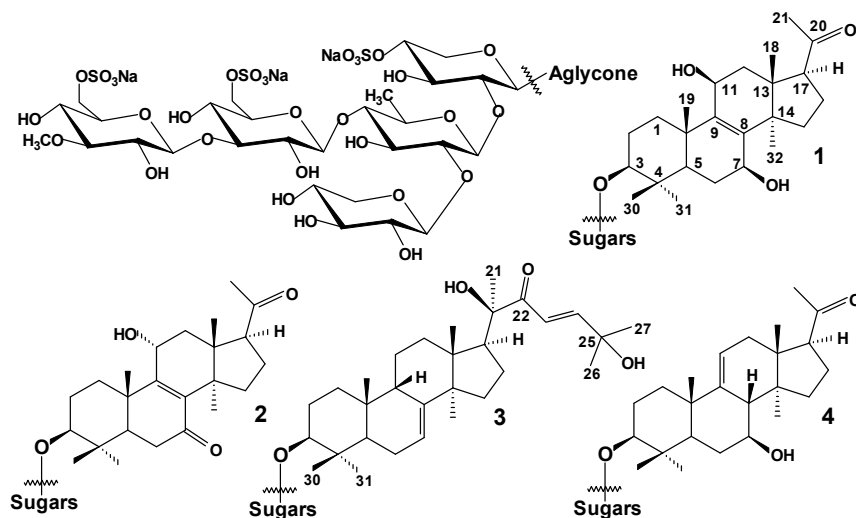
Recently we have started studies on the Far-Eastern sea cucumber *Cucumaria fallax* [6] which contains exclusively non-holostane oligoglycosides having unusual double bond positions and uncommon sites of oxidation in their aglycone moieties. Herein we report the isolation of four new trisulfated glycosides, fallaxosides D<sub>4</sub>–D<sub>7</sub> (compounds 1–4) with earlier unknown aglycones and their structures, established by analysis of <sup>1</sup>H-, <sup>13</sup>C-NMR and 2D NMR (<sup>1</sup>H–<sup>1</sup>H COSY, 1D TOCSY, HMBC, HSQC, ROESY) spectra and confirmed by HRESI mass spectrometry. The biogenesis of these unusual metabolites is also discussed.

## 2. Results and Discussion

The sea cucumbers were extracted with 70% ethanol under reflux during 5 h. The concentrated extract was sequentially submitted to the column chromatography on Polychrom-1 (powdered Teflon) in H<sub>2</sub>O→50% ethanol in order to eliminate salts and polar impurities and on Si gel using

CHCl<sub>3</sub>/EtOH/H<sub>2</sub>O (100:125:25 and 100:150:50) as mobile phases to obtain the fraction containing polar trisulfated pentaosides (glycosides belonging to the group A<sub>7</sub>). Further separation of the fraction by HPLC on a semi-preparative reversed phase column using MeOH/H<sub>2</sub>O/NH<sub>4</sub>OAc (1 M water solution) as mobile phase in ratio 60/39/1 gave the subfractions A<sub>7</sub>I–A<sub>7</sub>V. Each of the subfractions was rechromatographed using HPLC. The HPLC of subfraction A<sub>7</sub>I with the same solvent system in ratio of 35/64/1 gave fallaxoside D<sub>4</sub> (1) and fallaxoside D<sub>5</sub> (2). The HPLC of subfraction A<sub>7</sub>II using the solvent system in ratio of 50/49/1 followed by 45/54/1 and 47/51/2 gave fallaxoside D<sub>7</sub> (4). The HPLC of subfraction A<sub>7</sub>V with the same solvents in ratio of 58/41/1 gave fallaxoside D<sub>6</sub> (3).

The presence five characteristic doublet signals at  $\delta_H$  4.76–5.22 (1H, d,  $J = 6.9$ –8.4 Hz), correlated by HSQC spectra with the signals of anomeric carbons at  $\delta_C$  102.0–105.1 in the <sup>1</sup>H-NMR spectra of the carbohydrate chains of fallaxosides D<sub>4</sub> (1), D<sub>5</sub> (2), D<sub>6</sub> (3) and D<sub>7</sub> (4) (Scheme 1) and known fallaxosides D<sub>1</sub> and D<sub>2</sub> [7] were indicative of a pentasaccharide chain and  $\beta$ -configurations of glycosidic bonds. The positions of all the interglycosidic linkages and the place of linkage of the carbohydrate chain to an aglycone were deduced by the analysis of the ROESY and HMBC spectra of the carbohydrate parts of the glycosides (Table 1). Indeed, the cross-peaks between H-1 of the first monosaccharide residue (xylose) and H-3 (C-3) of an aglycone; H-1 of the second monosaccharide residue (quinovose) and H-2 (C-2) of the first monosaccharide residue (xylose); H-1 of the third monosaccharide residue (glucose) and H-4 (C-4) of the second monosaccharide residue (quinovose); H-1 of the fourth monosaccharide residue (3-O-methylglucose) and H-3 (C-3) of the third monosaccharide unit (glucose); H-1 of the fifth monosaccharide residue (xylose) and H-2 (C-2) of the second monosaccharide residue (quinovose) were observed. The  $\delta_C$  values characteristic of  $\alpha$ - and  $\beta$ -shifting effects of sulfate groups were observed for C-4 ( $\delta_C$  76.1) and C-5 ( $\delta_C$  64.0) of the first xylose residue, for C-6 ( $\delta_C$  67.3) and C-5 ( $\delta_C$  74.8) of the glucose residue and C-6 ( $\delta_C$  67.0) and C-5 ( $\delta_C$  75.5) of terminal 3-O-methylglucose residue. These data indicated the presence of a pentasaccharide carbohydrate chain with three sulfate groups.



**Scheme 1.** Structures of the glycosides: 1—fallaxoside D<sub>4</sub>; 2—fallaxoside D<sub>5</sub>; 3—fallaxoside D<sub>6</sub>; 4—fallaxoside D<sub>7</sub>.

The <sup>13</sup>C-NMR spectra of the carbohydrate parts of compounds 1–4 (Table 1) were identical to each other and coincided with those of cucumariosides of the group A<sub>7</sub> isolated first from *Cucumaria japonica* [8] and with the corresponding spectra of fallaxosides D<sub>1</sub> and D<sub>2</sub> having the same carbohydrate chain [6]. The structure of such a carbohydrate chain was previously elucidated by desulfation whereby a known desulfated derivative was obtained. This derivative was obtained from a monosulfated glycoside where very detailed chemical evidence of the sugar sequence was obtained by a variety of methods including specific enzymatic hydrolysis, periodate oxidation and Smith degradation, etc. All the obtained progenins were characterized by <sup>13</sup>C-NMR [8]

**Table 1.** NMR Spectroscopic data (700 MHz, C<sub>5</sub>D<sub>5</sub>N/D<sub>2</sub>O (4/1 *v/v*)) of the carbohydrate moieties of **1–4**.

Atom	$\delta_C$ mult. <sup>a,b,c</sup>	$\delta_H$ mult. (J in Hz) <sup>d</sup>	HMBC	ROESY
Xyl1 (1→C-3)				
1	104.7 CH	4.76 d (6.9)	C: 3	H-3, H-3, 5 Xyl1
2	<b>81.3</b> CH	3.97 t (8.6)	C: 1 Qui2; C: 1, 3 Xyl1	H-1 Qui2; H-4 Xyl1
3	75.2 CH	4.31 t (8.6)	C: 2, 4 Xyl1	H-1, 5 Xyl1
4	76.1 CH	4.98 dd (8.6, 13.8)	C: 3 Xyl1	H-2 Xyl1
5	64.0 CH <sub>2</sub>	4.76 d (11.2)	C: 1, 3 Xyl1	
		3.87 dd (8.6, 11.2)		H-1, 3 Xyl1
Qui2 (1→2Xyl1)				
1	102.0 CH	5.20 d (7.8)	C: 2 Xyl1	H-2 Xyl1; H-3, 5 Qui2
2	82.4 CH	3.92 t (8.6)	C: 1 Xyl5; C: 1, 3 Qui2	H-1 Xyl5; H-4 Qui2
3	75.2 CH	3.98 t (8.6)	C: 2, 4 Qui2	H-5 Qui2
4	<b>86.3</b> CH	3.43 t (8.6)	C: 1 Glc3; C: 3, 5, 6 Qui2	H-1 Glc3; H-2, 6 Qui2
5	70.8 CH	3.56 dd (6.0, 9.5)		H-1, 3, 6 Qui2
6	17.8 CH <sub>3</sub>	1.55 d (6.0)	C: 4, 5 Qui2	H-4, 5 Qui2
Glc3 (1→4Qui2)				
1	103.9 CH	4.78 d (7.8)	C: 4 Qui2	H-4 Qui2; H-5 Glc3
2	73.4 CH	3.81 m	C: 1, 3 Glc3	
3	<b>86.5</b> CH	4.13 t (8.6)	C: 1 MeGlc4; C: 2, 4 Glc3	H-1 MeGlc4; H-1 Glc3
4	69.1 CH	3.80 t (8.6)	C: 5, 6 Glc3	
5	74.8 CH	4.08 m		H-1 Glc3
6	67.3 CH <sub>2</sub>	4.94 d (11.2)		
		4.59 dd (6.9, 11.2)	C: 5 Glc3	H-4 Glc3
MeGlc4 (1→3Glc3)				
1	104.7 CH	5.15 d (8.4)	C: 3 Glc3	H-3 Glc3; H-3, 5 MeGlc4
2	74.3 CH	3.78 t (8.4)	C: 1 MeGlc4	H-4 MeGlc4
3	86.3 CH	3.64 t (9.3)	C: 2, 4 MeGlc4, OMe	H-1, 5 MeGlc4
4	69.8 CH	4.01 m	C: 3, 5 MeGlc4	H-2, 6 MeGlc4
5	75.5 CH	4.01 m	C: 4, 6 MeGlc4	H-1, 3 MeGlc4
6	67.0 CH <sub>2</sub>	4.92 brd (11.0)	C: 4, 5 MeGlc4	
		4.75 brd (8.4)	C: 5 MeGlc4	
OMe	60.4 CH <sub>3</sub>	3.80 s	C: 3 MeGlc4	
Xyl5 (1→2Qui2)				
1	105.1 CH	5.22 d (7.6)	C: 2 Qui2	H-2 Qui2; H-3, 5 Xyl5
2	74.8 CH	3.91 t (7.6)	C: 1, 3 Xyl5	
3	76.3 CH	4.07 t (8.4)	C: 2, 4 Xyl5	H-1, 5 Xyl5
4	70.1 CH	4.05 m	C: 3 Xyl5	H-2 Xyl5
5	66.4 CH <sub>2</sub>	4.28 dd (5.1, 11.8)	C: 1, 4 Xyl5	
		3.66 t (9.3)	C: 1, 3, 4 Xyl5	H-1, 3 Xyl5

<sup>a</sup> Recorded at 176.04 MHz; <sup>b</sup> Bold = interglycosidic positions; <sup>c</sup> Italic = sulfate position; <sup>d</sup> Multiplicity by 1D TOCSY.

The NMR spectra of the aglycone part of fallaxoside D<sub>4</sub> (**1**) revealed the presence of 24 carbon atoms (Table 2), including in six methylenes, five methines and six methyl groups as well as seven quaternary carbon signals, that corresponded to the 22,23,24,25,26,27-hexanorlanostane type of aglycones previously found in glycosides of *C. fallax* [6]. The resonances of two olefinic carbons with  $\delta_C$  139.9 (C-8) and 142.9 (C-9) were assigned to the tetrasubstituted double bond. The HMBC correlations between H-32 ( $\delta_H$  1.00, 3H, s) and C-8 and between H-11 ( $\delta_H$  4.81, 1H, brd,  $J = 6.8$  Hz) and C-9 as well as between H-19 ( $\delta_H$  1.57, 3H, s) and C-9 confirmed the 8(9)-double bond position. The signal at  $\delta_C$  211.8 indicated the presence of a keto group. The signals at  $\delta_H$  2.83 (1H, t,  $J = 8.8$  Hz, H-17) and  $\delta_C$  58.9 (C-17) were assigned to a methine group adjacent to a ketone group. The HMBC correlations from H-17 and H-21 ( $\delta_H$  2.14, 3H, s) to the carbon at  $\delta_C$  211.8 (C-20) allowed the positioning of the group at C-20. There were two downshifted resonances in the <sup>13</sup>C-NMR spectrum of **1** at  $\delta_C$  67.7 (C-7) and  $\delta_C$  65.2 (C-11) corresponding to the oxygen bearing allylic type methines suggesting the presence of two hydroxyls. Their positions at C-7 and C-11 were corroborated by the correlations from H-6b ( $\delta_H$  1.99, 1H, m) to C-7 and from H-12b ( $\delta_H$  2.27, 1H, d,  $J = 14.2$  Hz) to C-11 in the HMBC

spectrum as well as ROESY correlations between H-1(2.47 m) and H-11 and between H-5 (1.07 dd, 2.7, 15.6 Hz) and H-7. Seeing that absolute configuration at C-5 with  $\alpha$ -orientation of hydrogen, as well as configurations of C-10, C-13, C-14, and C-17 stereocenters in the sea cucumber triterpene glycosides were established earlier [9], the H-5–H-7 H-15 $\alpha$ –H7 and H-32–H-7 NOESY correlations are indicative of  $\beta$ -orientation of hydroxyl group at C-7. The configuration of the C-11 stereocenter in **1** was proposed by analysis of the ROESY spectrum and MM2 optimized models of **1**. The  $\beta$ -orientation of hydroxyl group at C-11 was proposed, based on the comparison of MM2 optimized models of aglycones with  $\alpha$ - and  $\beta$ -oriented hydroxyls. The observed in the ROESY spectrum of **1** correlations from H-11 to H-1 and *vice versa* would be realized for both 11- $\alpha$  and 11 $\beta$ -hydroxyls. However, the correlations from H-11 with methyl groups H-19 and H-18 should be the realized in the ROESY spectrum at 11 $\alpha$ -hydroxy orientation. Nevertheless, these correlations were absent in the ROESY spectrum of **1**. Hence the  $\beta$ -orientation of the hydroxyl group at C-11 is the most probable.

**Table 2.** NMR Spectroscopic data (700 MHz, C<sub>5</sub>D<sub>5</sub>N/D<sub>2</sub>O (4/1 *v/v*)) of the aglycone moiety of fallaxoside D<sub>4</sub> (**1**).

Position	$\delta_C$ Mult. <sup>a</sup>	$\delta_H$ Mult. (J in Hz)	HMBC	ROESY
1	34.9 CH <sub>2</sub>	2.47 m 1.32 m	C: 18	H-11, H-19 H-3, H-5, H-11
2	26.6 CH <sub>2</sub>	1.90 m 1.79 m		H-19, H-30
3	88.6 CH	3.13 dd (4.7, 12.2)	C: 30, 31, C: 1 Xyl1	H-1, H-5, H-31, H-1 Xyl1
4	39.5 C			
5	51.4 CH	1.07 dd (2.7, 15.6)	C: 4, 19, 30	H-1, H-3, H-7, H-31
6	30.8 CH <sub>2</sub>	2.35 m 1.99 m	C: 5, 7, 10	H-31 H-19, H-30
7	67.7 CH	4.55 t (8.1)	C: 8, 9	H-5, H-15, H-32
8	139.9 C			
9	142.9 C			
10	38.3 C			
11	65.2 CH	4.81 brd (6.8)	C: 8, 9, 13	H-1
12	42.8 CH <sub>2</sub>	2.53 dd (7.5, 14.2) 2.27 d (14.2)	C: 13, 18 C: 9, 11, 13, 14, 18	H-17, H-32 H-18, H-21
13	45.2 C			
14	51.5 C			
15	32.0 CH <sub>2</sub>	2.71 brd (10.8) $\beta$ 1.60 brt (10.8) $\alpha$	C: 14, 32	H-18 H-7, H-32
16	22.6 CH <sub>2</sub>	2.41 m 1.68 m		H-18 H-32
17	58.9 CH	2.83 t (8.8)	C: 12, 13, 16, 18, 20, 21	H-12, H-21, H-32
18	19.7 CH <sub>3</sub>	1.23 s	C: 12, 13, 14, 17	H-12, H-15, H-16, H-19, H-21
19	22.2 CH <sub>3</sub>	1.57 s	C: 1, 5, 9, 10	H-1, H-2, H-5, H-6, H-18, H-30
20	211.8 C			
21	31.3 CH <sub>3</sub>	2.14 s	C: 17, 20	
30	16.4 CH <sub>3</sub>	0.99 s	C: 3, 4, 5, 31	H-2, H-6, H-19, H-31
31	27.9 CH <sub>3</sub>	1.15 s	C: 3, 4, 5, 30	H-3, H-5, H-6, H-30
32	26.1 CH <sub>3</sub>	1.00 s	C: 8, 13, 14, 15	H-7, H-12, H-15, H-16, H-17

<sup>a</sup> Recorded at 176.04 MHz.

The HR ESI MS (negative and positive ion modes) of fallaxoside D<sub>4</sub> (**1**) exhibited ion peaks [M<sub>3</sub>Na – H<sub>2</sub>O – Na]<sup>–</sup> at *m/z* 1403.3543 (calc. 1403.3572), [M<sub>3</sub>Na – H<sub>2</sub>O – 2Na]<sup>2–</sup> at *m/z* 690.1841 (calc. 690.1840) and [M<sub>3</sub>Na – H<sub>2</sub>O – 3Na]<sup>3–</sup> at *m/z* 452.4596 (calc. 452.4596). The molecular formula of **1** was deduced as C<sub>53</sub>H<sub>83</sub>O<sub>35</sub>S<sub>3</sub>Na<sub>3</sub> based on the HR ESI MS and <sup>13</sup>C-NMR spectroscopic data. However, the ions observed in the mass spectra corresponded to the loss of water upon ionization of fallaxoside D<sub>4</sub> (**1**), presumably due to the lability of the dihydroxy-ene-fragment in the rings B and C of its aglycone. The structure of the carbohydrate chain of **1** was also confirmed by fragment ion peaks at *m/z* 929, corresponding to a monodesulfated carbohydrate chain [MeGlcOSO<sub>3</sub>Na + GlcOSO<sub>3</sub>Na +

Qui + Xyl + XylOSO<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub>]<sup>–</sup> or [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>)]<sup>–</sup>. Other fragment ions at *m/z* 827.2 corresponded to a didesulfated carbohydrate chain [MeGlcOSO<sub>3</sub>Na + GlcOSO<sub>3</sub>Na + Qui + Xyl + XylOSO<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> + H]<sup>–</sup> or [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) + H]<sup>–</sup>; at *m/z* 797.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – Xyl + H]<sup>–</sup>; at *m/z* 695.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – Xyl + 2H]<sup>–</sup>; at *m/z* 563.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl + 3H]<sup>–</sup>; at *m/z* 519 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl – Qui + H]<sup>–</sup>; at *m/z* 417.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl – Qui + 2H]<sup>–</sup>, observed in the ESI MS/MS spectra of **1**.

Taken together all these data indicate that fallaxoside D<sub>4</sub> (**1**) is 22,23,24,25,26,27-hexanor-3β-O-{6-O-sodium sulfate-3-O-methyl-β-D-glucopyranosyl-(1→3)-6-O-sodium sulfate-β-D-glucopyranosyl-(1→4)-[β-D-xylopyranosyl-(1→2)]-β-D-quinovopyranosyl-(1→2)-4-O-sodium sulfate-β-D-xylopyranosyl}-7β,11β-dihydroxy-20-ketolanost-8-ene.

The structure of the aglycone moiety of fallaxoside D<sub>5</sub> (**2**) was proved to be similar and NMR spectra also confirmed the 22,23,24,25,26,27-hexanorlanostane skeleton system of its aglycone. The difference in the spectra (Table 3) was in the presence of an additional signal of a ketone group at δ<sub>C</sub> 201.4 instead of one of hydroxyl group signals in the <sup>13</sup>C-NMR spectrum.

**Table 3.** NMR Spectroscopic data (700 MHz, C<sub>5</sub>D<sub>5</sub>N/D<sub>2</sub>O (4/1 *v/v*)) of the aglycone moiety of fallaxoside D<sub>5</sub> (**2**).

Position	δ <sub>C</sub> Mult. <sup>a</sup>	δ <sub>H</sub> Mult. (J in Hz)	HMBC	ROESY
1	34.1 CH <sub>2</sub>	2.29 m 2.01 m	C: 10 C: 3	H-3, H-5 H-11
2	26.5 CH <sub>2</sub>	2.00 m 1.77 m	C: 30, C: 1 Xyl1	H-19 H-19, H-30
3	87.8 CH	3.13 dd (4.0; 11.6)	C: 4, 30, C: 1 Xyl1	H-1, H-5, H-31, H-1 Xyl1
4	39.5 C			
5	50.5 CH	1.83 dd (3.9; 13.5)	C: 4, 6, 10, 19, 30, 31	H-1, H-3, H-31
6	36.9 CH <sub>2</sub>	2.56 m 2.50 m	C: 5, 7, 10 C: 5, 7, 10	H-19, H-30
7	201.4 C			
8	140.0 C			
9	163.1 C			
10	40.6 C			
11	64.1 CH	4.70 dd (5.4; 8.9)	C: 8, 9, 10, 13	H-1, H-18, H-19
12	43.5 CH <sub>2</sub>	2.58 dd (8.8; 13.1) 2.40 dd (5.1; 13.4)	C: 9, 11, 13, 18 C: 11, 13, 17, 18	H-18, H-21 H-17, H-32
13	48.5 C			
14	48.7 C			
15	33.4 CH <sub>2</sub>	2.30 m 1.86 m	C: 14, 17, 32	H-32 H-18
16	22.1 CH <sub>2</sub>	2.27 m 1.68 m		H-18 H-32
17	58.6 CH	2.91 t (9.0)	C: 12, 13, 16, 18, 20	H-12, H-21, H-32
18	18.6 CH <sub>3</sub>	0.58 s	C: 12, 13, 14, 17	H-12, H-15, H-16, H-19, H-21
19	19.8 CH <sub>3</sub>	1.10 s	C: 1, 5, 9, 10	H-2, H-6, H-18
20	211.7 C			
21	31.2 CH <sub>3</sub>	2.15 s	C: 17, 20	H-17, H-18
30	16.0 CH <sub>3</sub>	1.01 s	C: 3, 4, 5, 31	H-2, H-6
31	27.0 CH <sub>3</sub>	1.05 s	C: 3, 4, 5, 30	H-3, H-5, H-6
32	25.0 CH <sub>3</sub>	1.40 s	C: 8, 13, 14, 15	H-12, H-15, H-17

<sup>a</sup> Recorded at 176.04 MHz.

The signals of the quaternary olefinic carbons indicated the presence of 8(9)-double bond and were downshifted (δ<sub>C</sub> 140.0 and 163.1) when compared with those in the <sup>13</sup>C-NMR spectrum of **1**, suggesting that the ketone group adjoins the double bond. Additionally, the signals at δ<sub>H</sub> 2.56 (1H, m, H-6a) and δ<sub>H</sub> 2.50 (1H, m, H-6b) demonstrated the formation of a spin coupled system in the COSY



spectrum with the signal of H-5 (1H, dd,  $J = 3.9, 13.5$  Hz) only and were also noticeably downshifted. These data indicated the closeness of these atoms to the ketone group that was positioned at C-7. The HMBC correlation from both H<sub>2</sub>-6 to C-7 ( $\delta_C$  201.4) and C-10 ( $\delta_C$  40.6) confirmed the ketone position. Hence, the hydroxyl group at C-7 of the aglycone of **1** was substituted with a ketone group in the aglycone of **2**. The signal at  $\delta_C$  64.1 (C-11) in the <sup>13</sup>C-NMR spectrum of **2**, correlated by the HSQC with the resonance at  $\delta_H$  4.70 (1H, dd,  $J = 5.4, 8.9$  Hz, H-11) indicated the presence of oxygen bearing methine group. The HMBC correlations from both H-12 ( $\delta_H$  2.58, 1H, dd  $J = 8.8, 13.1$  Hz, H-12a) and ( $\delta_H$  2.40, 1H, dd  $J = 5.1, 13.4$  Hz, H-12b) to C-11 corroborated the attachment of hydroxyl group to C-11.

The orientation of hydroxyl group was proposed as  $\alpha$  based on the ROESY spectrum. Clear NOE correlations from H-11 to both  $\beta$ -oriented methyl groups Me-18 and Me-19 were observed. Hence the hydroxyl groups at C-11 are opposite oriented in fallaxosides D<sub>4</sub> (**1**) and D<sub>5</sub> (**2**) when compared each other. The HRESIMS (negative ion mode) of fallaxoside D<sub>5</sub> (**2**) exhibited quasi-molecular ion peaks at  $m/z$  1419.3502 (calc. 1419.3521)  $[M_{3Na} - Na]^-$ , at  $m/z$  698.1806 (calc. 698.1815)  $[M_{3Na} - 2Na]^{2-}$  and at  $m/z$  457.7911 (calc. 457.7912)  $[M_{3Na} - 3Na]^{3-}$ . This and <sup>13</sup>C-NMR spectroscopic data allowed the determination of the molecular formula of **2** as C<sub>53</sub>H<sub>81</sub>O<sub>35</sub>S<sub>3</sub>Na<sub>3</sub>. The presence of sulfate groups and sequence of monosaccharide units in the carbohydrate chain of **2** was confirmed by fragmentary ions at  $m/z$  1299.4  $[M_{3Na} - Na - HSO_4Na]^-$ , 1179.4  $[M_{3Na} - Na - 2HSO_4Na]^-$ , 911.1  $[M_{3Na} - Na_2SO_4 - agl + H]^-$ ; 877.3  $[M_{3Na} - Na - MeGlcOSO_3Na - GlcOSO_3Na + H]^-$ ; 797.1  $[M_{3Na} - Na_2SO_4 - agl - Xyl + H]^-$ ; 745.3  $[M_{3Na} - Na - MeGlcOSO_3Na - GlcOSO_3Na - Xyl + 2H]^-$ ; 665.1  $[M_{3Na} - Na_2SO_4 - agl - 2Xyl + 2H]^-$ ; 599.3  $[M_{3Na} - Na - MeGlcOSO_3Na - GlcOSO_3Na - Qui - Xyl + H]^-$ ; 519  $[M_{3Na} - Na_2SO_4 - agl - 2Xyl - Qui + H]^-$  detected in the ESI MS/MS (negative ion mode) spectrum. All these data indicate that fallaxoside D<sub>5</sub> (**2**) is 22,23,24,25,26, 27-hexanor-3 $\beta$ -O-[6-O-sodium sulfate-3-O-methyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-6-O-sodium sulfate- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-quinovopyranosyl-(1 $\rightarrow$ 2)-4-O-sodium sulfate- $\beta$ -D-xylopyranosyl-11 $\alpha$ -hydroxy-7,20-diketolanost-8-ene.

The <sup>13</sup>C-NMR spectrum of the aglycone part of fallaxoside D<sub>6</sub> (**3**) revealed the presence of 30 carbons indicating the presence of an aglycone with normal non-shortened side chain (Table 4). The absence of a  $\gamma$ -lactone was deduced from the absence of the signal at  $\approx \delta_C$  176 and contemporary presence of the resonances of Me-18 ( $\delta_C$  24.8, C-18,  $\delta_H$  1.28, 3H, s, H-18) in the <sup>13</sup>C- and <sup>1</sup>H-NMR spectra, respectively, indicating the aglycone of fallaxoside D<sub>6</sub> (**3**) to be of the lanostane type without a lactone, in contrast with aglycones of the majority of sea cucumber glycosides. The resonances of an olefinic methine group at  $\delta_C$  122.2 (C-7) and  $\delta_H$  5.59 (1H, m, H-7) in NMR spectra as well as of an olefinic quaternary carbon at  $\delta_C$  149.7 (C-8) in the <sup>13</sup>C-NMR spectrum were assigned to the 7(8)-double bond typical of many sea cucumber glycosides [1,9]. The presence of an isolated H-5/H<sub>2</sub>-6/H-7/H-9/H-11/H-12 spin system in the COSY spectrum of **3** as well as the HMBC correlation between H-32 ( $\delta_H$  1.03, 3H, s) and C-8 confirmed the double bond position. The signal at  $\delta_C$  205.3 and resonances of two olefinic tertiary carbons at  $\delta_C$  119.9 and 156.5 in the <sup>13</sup>C-NMR spectrum indicated the presence of a keto group and an additional double bond, correspondingly. Extensive analysis of the COSY, HSQC and HMBC spectra of **3** revealed these functionalities could be positioned only in the side chain of the aglycone. Actually the multiplicity and the coupling constants of olefinic protons at  $\delta_H$  7.41 (1H, d,  $J = 15.5$  Hz, H-23) and  $\delta_H$  7.40 (1H, d,  $J = 15.5$  Hz, H-24) were indicative of a *Z*-configured 23(24)-double bond vicinal on the one hand to the keto group (C-22) and from the other hand to the quaternary oxygen bearing carbon at  $\delta_C$  70.5 (C-25). The presence of the resonances of two heminal methyl groups at  $\delta_H$  1.47 (3H, s, H-26) and  $\delta_H$  1.48 (3H, s, H-27) in the <sup>1</sup>H-NMR spectrum correlated by HSQC spectrum with corresponding carbons resonances at  $\delta_C$  29.3 (C-26) and  $\delta_C$  29.2 (C-27) as well as the HMBC correlations from H-26 and H-27 to C-24 and C-25 and from H-23 and H-24 to C-22 and C-25 confirmed the structure of side chain having 22-keto- and 25-hydroxyl groups and 23*Z*,24-double bond. The correlations observed in the ROESY spectrum of **3** (Table 4) corroborated the configurations of stereocenters C-3, C-5, C-10, C-13, C-14, C-17 as well as (20*R*) configuration (NOE between H-16 and H-23 and vice versa) established earlier for similar

aglycone of frondoside C [10]. A ROESY correlations of a signal at 2.28 brd,  $J = 12.8$  Hz, H-9 with signals of methyl group at C-10 and C-14 established the rare  $9\beta$ -H configuration in **3** characteristic of 7(8)-unsaturated sea cucumber glycosides [1,9,11]. Thus, these data allowed us to determine all the structural and stereochemical features of this unusual aglycone.

**Table 4.** NMR spectroscopic data (700 MHz,  $C_5D_5N/D_2O$  (4/1  $v/v$ )) of the aglycone moiety of fallaxoside  $D_6$  (**3**).

Position	$\delta_C$ Mult. <sup>a</sup>	$\delta_H$ Mult. (J in Hz)	HMBC	ROESY
1	35.5 CH <sub>2</sub>	1.32 m 1.27 m		H-11
2	27.0 CH <sub>2</sub>	1.95 m 1.78 m		H-19, H-30
3	88.9 CH	3.14 dd (4.1, 11.6)	C: 30, C: 1 Xyl1	H-5, H-31, H-1 Xyl1
4	39.5 C			
5	49.7 CH	0.84 brd (12.7)	C: 1, 4, 30	H-1, H-3, H-31
6	23.1 CH <sub>2</sub>	1.92 m 1.83 m		H-31 H-19, H-30
7	122.2 CH	5.59 m	C: 9, 14	H-15, H-32
8	149.7 C			
9	48.1 CH	2.28 brd (12.8)		H-18, H-19
10	35.6 C			
11	22.8 CH <sub>2</sub>	1.63 m 1.38 m		H-32
12	34.8 CH <sub>2</sub>	1.96 m 1.74 m		H-17, H-21 H-18
13	45.3 C			
14	52.9 C			
15	33.4 CH <sub>2</sub>	1.59 m 1.50 m		H-18 H-7, H-32
16	22.2 CH <sub>2</sub>	1.93 m 1.54 m		H-23 H-17, H-32
17	52.8 CH	2.39 t (8.8)	C: 14, 16, 18	H-21, H-32
18	24.8 CH <sub>3</sub>	1.28 s	C: 12, 14, 17	H-9, H-12, H-16, H-21
19	24.5 CH <sub>3</sub>	0.93 s	C: 1, 5, 9, 10	H-2, H-6, H-9
20	80.5 C			
21	24.5 CH <sub>3</sub>	1.62 s	C: 17, 20, 22	H-12, H-17, H-18
22	205.3 C			
23	119.9 C	7.41 d (15.5)	C: 22, 24, 25	
24	156.5 C	7.40 d (15.5)	C: 22, 23, 25	H-17, H-21, H-26, H-27, H-16
25	70.5 C			
26	29.3 CH <sub>3</sub>	1.47 s	C: 24, 25, 27	
27	29.2 CH <sub>3</sub>	1.48 s	C: 24, 25, 26	
30	17.4 CH <sub>3</sub>	1.02 s	C: 3, 4, 5, 31	H-2, H-6
31	28.7 CH <sub>3</sub>	1.17 s	C: 3, 4, 5, 30	H-3, H-5, H-6
32	30.7 CH <sub>3</sub>	1.03 s	C: 8, 13, 14, 15	H-7, H-11, H-12, H-15, H-17

<sup>a</sup> Recorded at 176.04 MHz.

The HRESIMS (negative ion mode) of fallaxoside  $D_6$  (**3**) exhibited pseudomolecular ion peaks  $[M_{3Na} - 2Na]^{2-}$  at  $m/z$  740.2279 (calc. 740.2284) and  $[M_{3Na} - 3Na]^{3-}$  at  $m/z$  485.8224 (calc. 485.8225). This and  $^{13}C$ -NMR spectroscopic data allowed the determination of the molecular formula of **3** as  $C_{59}H_{93}O_{35}S_3Na_3$ . The presence of sulfate groups and sequence of monosaccharide units in the carbohydrate chain of **3** was confirmed by the fragmentary ion peaks at  $m/z$  1383.4  $[M_{3Na} - NaHSO_4]^-$ ; 827.2  $[M_{3Na} - Na_2SO_4 - NaSO_3 - agl + H]^-$ ; 797.1  $[M_{3Na} - Na_2SO_4 - agl - Xyl + H]^-$ ; 695.2  $[M_{3Na} - Na_2SO_4 - NaSO_3 - agl - Xyl + 2H]^-$ ; 563.1  $[M_{3Na} - Na_2SO_4 - NaSO_3 - agl - 2Xyl + 3H]^-$ ; 519  $[M_{3Na} - Na_2SO_4 - agl - 2Xyl - Qui + H]^-$  detected in the ESI MS/MS (negative ion mode) spectrum.

All these data indicate that fallaxoside D<sub>6</sub> (**3**) is 3 $\beta$ -O-{6-O-sodium sulfate-3-O-methyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-6-O-sodium sulfate- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 4)-[ $\beta$ -D-xylo-pyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-quinovopyranosyl-(1 $\rightarrow$ 2)-4-O-sodium sulfate- $\beta$ -D-xylopyranosyl]-22-keto-25-hydroxylanosta-7,23Z-diene.

The structure of the aglycone moiety of fallaxoside D<sub>7</sub> (**4**) was found by extensive NMR spectroscopy (Table 5) to be similar to that of **1–2**, indicating the presence of a 22,23,24,25,26,27-hexanorlanostane aglycone with the keto group at C-20 ( $\delta_C$  212.5). The double bond was positioned as 9(11) according to the signal of the quaternary olefinic carbon at  $\delta_C$  147.6 (C-9) and olefinic methine resonances at  $\delta_C$  116.0 (C-11) in the <sup>13</sup>C-NMR spectrum and  $\delta_H$  5.27 (1H, m, H-11) in the <sup>1</sup>H-NMR spectrum. The long range COSY correlation H-8/H-11 as well as the characteristic correlation from Me-19 ( $\delta_H$  1.00, 3H, s, H-19) to C-9 confirmed the double bond position. The signal at  $\delta_C$  71.6 (C-7) in the <sup>13</sup>C-NMR spectrum of **4**, correlated by the HSQC with the resonance at  $\delta_H$  3.83 (1H, m, H-7) indicated the attachment of hydroxyl group to this carbon. Its position as C-7 was confirmed by the COSY spectrum, where the signals of spin coupled system H-5/H<sub>2</sub>-6/H-7/H-8 were observed.

**Table 5.** NMR spectroscopic data (700 MHz, C<sub>5</sub>D<sub>5</sub>N/D<sub>2</sub>O (4/1 *v/v*)) of Aglycone Moiety of fallaxoside D<sub>7</sub> (**4**).

Position	$\delta_C$ Mult. <sup>a</sup>	$\delta_H$ Mult. (J in Hz)	HMBC	ROESY
1	36.4 CH <sub>2</sub>	1.60 m 1.24 m		H-11, H-19 H-3, H-5, H-11
2	26.7 CH <sub>2</sub>	2.00 m 1.77 m		H-19, H-30
3	88.5 CH	3.10 dd (4.1; 11.8)		H-1, H-5, H-31, H-1 Xyl1
4	39.5 C			
5	49.3 CH	0.86 brd (13.7)		H-1, H-3, H-7, H-31
6	31.8 CH <sub>2</sub>	2.15 m 1.79 m		H-31
7	71.6 CH	3.83 m		H-5, H-32
8	49.8 CH	2.38 m		H-15, H-18, H-19
9	147.6 C			
10	38.9 C			
11	116.0 CH	5.27 m	C: 10, 13	H-1
12	35.8 CH <sub>2</sub>	2.34 m 1.94 m		H-32 H-18
13	46.8 C			
14	47.3 C			
15	36.6 CH <sub>2</sub>	2.05 m 1.94 m		H-32
16	22.5 CH <sub>2</sub>	2.31 m 1.62 m		H-18 H-32
17	59.3 CH	2.92 t (9.0)	C: 13, 18	H-12, H-32
18	16.4 CH <sub>3</sub>	0.59 s	C: 12, 13, 17	H-8, H-12, H-19
19	21.8 CH <sub>3</sub>	1.00 s	C: 5, 9, 10	H-1, H-2, H-8, H-18
20	212.5 C			
21	31.2 CH <sub>3</sub>	2.18 s	C: 17, 20	
30	16.6 CH <sub>3</sub>	0.94 s	C: 3, 4, 5, 31	H-2, H-31
31	27.9 CH <sub>3</sub>	1.14 s	C: 3, 4, 5, 30	H-3, H-5, H-6, H-30
32	18.3 CH <sub>3</sub>	0.99 s	C: 8, 13, 15	H-7, H-15, H-16, H-17

<sup>a</sup> Recorded at 176.04 MHz

The orientation of hydroxyl group was proposed as  $\beta$  based on the ROESY spectrum. Clear NOE correlations from H-7 to both  $\alpha$ -oriented H-5 ( $\delta_H$  0.86, 1H, brd, *J* = 13.7 Hz) and Me-32 ( $\delta_H$  0.99, 3H, s) and vice versa were observed. (8*S*)-Configuration was proposed based on the correlations between H-8 and the both Me-18 and Me-19, observed in the ROESY spectrum of **4**.



The HR ESI MS (negative ion mode) of fallaxoside D<sub>7</sub> (**4**) exhibited ion peaks  $[M_{3Na} - 2Na]^{2-}$  at  $m/z$  691.1917 (calc. 691.1918) and  $[M_{3Na} - 3Na]^{3-}$  at  $m/z$  453.1320 (calc. 453.1315). This and <sup>13</sup>C-NMR spectroscopic data allowed the determination of the molecular formula of **4** as C<sub>53</sub>H<sub>83</sub>O<sub>34</sub>S<sub>3</sub>Na<sub>3</sub>. The identity of the carbohydrate chain structure of fallaxoside D<sub>7</sub> (**4**) to those of **1–3** confirmed by the presence of the fragmentary ion peaks with the same  $m/z$ : 797.1, 695.2, 665.1, 563.1, 519.0 and 417.1 in the ESI MS/MS (negative ion mode) spectrum of **4**.

On the base of all the above discussed data, the structure of fallaxoside D<sub>7</sub> (**4**) was established to be 22,23,24,25,26,27-hexanor-3 $\beta$ -O-{6-O-sodium sulfate-3-O-methyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-6-O-sodium sulfate- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 4)-[ $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-quinovopyranosyl-(1 $\rightarrow$ 2)-4-O-sodium sulfate- $\beta$ -D-xylopyranosyl}-7 $\beta$ -hydroxy-20-ketolanost-9(11)-ene.

The cytotoxic activities of fallaxosides D<sub>4</sub> (**1**), D<sub>5</sub> (**2**), D<sub>6</sub> (**3**) and D<sub>7</sub> (**4**) against mouse spleen lymphocytes and ascite form of mouse Ehrlich carcinoma cells along with hemolytic activity against mouse erythrocytes were studied. None of these glycosides were active in these tests at the dosage studied (IC<sub>50</sub> > 100  $\mu$ M/mL). This could be explained by the absence of the 18(20)-lactone moiety that is essential for the membranolytic action of sea cucumber glycosides and by the presence of three sulfate groups that decreased the membranolytic activity of the glycosides [12].

It is most probable that holostane type glycosides play a role in chemical defense against predators because of their strong membranolytic activities and not only as regulators of oocyte maturation. Non-holostane glycosides may play only a regulatory role and are evolutionary precursors of holostane glycosides [12].

In summary, four novel triterpene glycosides **1–4** have been isolated from the sea cucumber *Cucumaria fallax* along with the series of other uncommon glycosides [6]. All of these compounds contain new non-holostane aglycones with some unprecedented structural features. It is obvious that the isolated glycosides, including the earlier studied fallaxoside C<sub>2</sub> and D<sub>2</sub> (having a 7-keto-8(9)-en fragment), C<sub>1</sub> and D<sub>1</sub> (having a 7,11-diketo-8(9)-en fragment) [6], D<sub>4</sub> (**1**) (with a 7,11-dihydroxy-8(9)-en fragment), D<sub>5</sub> (**2**) (with a 7-keto-11-hydroxy-8(9)-en fragment) and D<sub>7</sub> (**4**) (with a 7-hydroxy-9(11)-en fragment), are biogenetically related with each other. They have oxygen-containing functional groups (ketones or hydroxyls) at the same C-7 and/or C-7 and C-11 positions of aglycones. Another unusual structural feature of many isolated glycosides includes the presence of a 8(9)-double bond in the majority of glycosides isolated from this species. This suggests that in *C. fallax*, the protosterol cation, formed from squalene, is cyclized not only into 9(11)-unsaturated parkeol or into 9 $\beta$ -H-lanosta-7,24-dienol, as shown earlier in other sea cucumbers [13], but also into lanosta-8(9),24-diene-3 $\beta$ -ol (lanosterol).

### 3. Experimental Section

#### 3.1. General Experimental Procedures

Melting points were determined with a Kofler-Thermogenerate apparatus (Leica VMTG, Wien, Austria). Specific rotation was measured on a 343 Polarimeter (Perkin-Elmer Corporation, Waltham, MA, USA). NMR spectra were recorded on an Advance III-700 spectrometer (Bruker Bio Spin, Rheinstetten, Germany) at 700.13 MHz/176.04 MHz (<sup>1</sup>H/<sup>13</sup>C) in C<sub>5</sub>D<sub>5</sub>N/D<sub>2</sub>O (4/1) with TMS as an internal reference ( $\delta = 0$ ). The ESIMS (negative ion mode) were recorded using an Agilent 6510 Q-TOF apparatus (Agilent Corporation, Palo Alto, CA, USA), 50% MeOH was used as the solvent, sample concentration 0.01 mg/mL. HPLC was performed using an Agilent 1100 chromatograph equipped with a differential refractometer on a Ascentis RP Amide (10  $\times$  250 mm, 5  $\mu$ m) column (Supelco, Bellefonte, PA, USA).

#### 3.2. Animal Material

The samples of *Cucumaria fallax* (Cucumariidae, Dendrochirotida) were collected during the 41-st scientific cruise of the research vessel *Akademik Oparin* in the Pacific Ocean near Black Brothers Islands,

Kurile Islands (46°23'9'' N, 150°46'25'' E) on 20 July, 2011 using Sigsbee trawl from the depth of 150 m (collector Kirill Minin, P.P. Shyrshov Institute of Oceanology of the Russian Academy of Sciences, Moscow, Russia). The sea cucumber taxonomic identification was carried out by Dr. Vadim G. Stepanov (Kamchatka Department of the Institute of Geography, Far East Division of the Russian Academy of Sciences, Petropavlovsk-Kamchatsky, Russia), the voucher specimen is deposited in collection of Kamchatka Department of the Institute of Geography.

### 3.3. Extraction and Isolation

The sea cucumbers were minced and extracted with 70% ethanol under reflux during 5 h. The concentrated in vacuo ethanolic extract of *C. fallax* (28.2 g of dry wt after ethanol extraction) was sequentially submitted to column chromatography on Polychrom-1 (powdered Teflon) in H<sub>2</sub>O→50% ethanol and on Si gel using CHCl<sub>3</sub>/EtOH/H<sub>2</sub>O (100:125:25 and 100:150:50) as mobile phases to obtain the fraction A<sub>7</sub> (165 mg). Further separation of the fraction by HPLC on a semi-preparative Supelco Ascentis RP-Amide (10 × 250 mm) reverse phase column using MeOH/H<sub>2</sub>O/NH<sub>4</sub>OAc (1 M water solution) as mobile phase in ratio 60/39/1 gave the subfractions A<sub>7</sub>I–A<sub>7</sub>V, each of them was subsequently rechromatographed. The HPLC of subfraction A<sub>7</sub>I with the same solvent system in ratio of 35/64/1 gave 3.5 mg of fallaxoside D<sub>4</sub> (1) and 3.5 mg of fallaxoside D<sub>5</sub> (2). The HPLC of subfraction A<sub>7</sub>II using the solvent system in ratio of 50/49/1 followed by 45/54/1 and 47/51/2 gave 1.8 mg of fallaxoside D<sub>7</sub> (4). The HPLC of subfraction A<sub>7</sub>V with the same solvents in ratio of 58/41/1 gave 7.2 mg of fallaxoside D<sub>6</sub> (3).

*Fallaxoside D<sub>4</sub>* (1): Colorless powder; MP: 212–214 °C;  $[\alpha]_D^{20}$  –2 (c 0.1, 50% MeOH); NMR: See Tables 1 and 2; HRESIMS (–) *m/z* 1403.3543 (calc. for C<sub>53</sub>H<sub>83</sub>O<sub>35</sub>S<sub>3</sub>Na<sub>3</sub>, 1403.3572) [M<sub>3</sub>Na – H<sub>2</sub>O – Na]<sup>–</sup>, *m/z* 690.1841 (calc. 690.1840) [M<sub>3</sub>Na – H<sub>2</sub>O – 2Na]<sup>2–</sup>, 452.4596 (calc. 452.4596) [M<sub>3</sub>Na – H<sub>2</sub>O – 3Na]<sup>3–</sup>; ESI MS/MS (–) *m/z* 929 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>)]<sup>–</sup>; 827.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) + H]<sup>–</sup>; 797.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – Xyl + H]<sup>–</sup>; 695.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – Xyl + 2H]<sup>–</sup>; 563.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl + 3H]<sup>–</sup>; 519 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl – Qui + H]<sup>–</sup>; 417.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl (C<sub>24</sub>H<sub>37</sub>O<sub>3</sub>) – 2Xyl – Qui + 2H]<sup>–</sup>.

*Fallaxoside D<sub>5</sub>* (2): Colorless powder; MP: 206–208 °C;  $[\alpha]_D^{20}$  –3 (c 0.1, 50% MeOH); NMR: see Tables 1 and 3; HRESIMS (–) *m/z* 1419.3502 (calc. for C<sub>53</sub>H<sub>81</sub>O<sub>35</sub>S<sub>3</sub>Na<sub>3</sub>, 1419.3521) [M<sub>3</sub>Na – Na]<sup>–</sup>, 698.1806 (calc. 698.1815) [M<sub>3</sub>Na – 2Na]<sup>2–</sup>, 457.7911 (calc. 457.7912) [M<sub>3</sub>Na – 3Na]<sup>3–</sup>; ESI MS/MS (–) *m/z* 1299.4 [M<sub>3</sub>Na – Na – HSO<sub>4</sub>Na]<sup>–</sup>, 1179.4 [M<sub>3</sub>Na – Na – 2HSO<sub>4</sub>Na]<sup>–</sup>, 911.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl + H]<sup>–</sup>; 877.3 [M<sub>3</sub>Na – Na – MeGlcOSO<sub>3</sub>Na – GlcOSO<sub>3</sub>Na + H]<sup>–</sup>; 797.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – Xyl + H]<sup>–</sup>; 745.3 [M<sub>3</sub>Na – Na – MeGlcOSO<sub>3</sub>Na – GlcOSO<sub>3</sub>Na – Xyl + 2H]<sup>–</sup>; 665.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – 2Xyl + 2H]<sup>–</sup>; 599.3 [M<sub>3</sub>Na – Na – MeGlcOSO<sub>3</sub>Na – GlcOSO<sub>3</sub>Na – Qui – Xyl + H]<sup>–</sup>; 519 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – 2Xyl – Qui + H]<sup>–</sup>.

*Fallaxoside D<sub>6</sub>* (3): Colorless powder; MP: 225–227 °C;  $[\alpha]_D^{20}$  –8 (c 0.1, 50% MeOH); NMR: see Tables 1 and 4; HRESIMS (–) *m/z* 740.2279 (calc. for C<sub>59</sub>H<sub>93</sub>O<sub>35</sub>S<sub>3</sub>Na<sub>3</sub>, 740.2284) [M<sub>3</sub>Na – 2Na]<sup>2–</sup>, 485.8224 (calc. 485.8225) [M<sub>3</sub>Na – 3Na]<sup>3–</sup>; ESI MS/MS (–) *m/z* 1383.4 [M<sub>3</sub>Na – NaHSO<sub>4</sub>]<sup>–</sup>; 827.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl + H]<sup>–</sup>; 797.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – Xyl + H]<sup>–</sup>; 695.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl – Xyl + 2H]<sup>–</sup>; 563.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl – 2Xyl + 3H]<sup>–</sup>; 519 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – 2Xyl – Qui + H]<sup>–</sup>.

*Fallaxoside D<sub>7</sub>* (4): Colorless powder; MP: 210–212 °C;  $[\alpha]_D^{20}$  –5 (c 0.1, 50% MeOH); NMR: See Tables 1 and 5; HRESIMS (–) *m/z* 691.1917 (calc. for C<sub>53</sub>H<sub>83</sub>O<sub>34</sub>S<sub>3</sub>Na<sub>3</sub>, 691.1918) [M<sub>3</sub>Na – 2Na]<sup>2–</sup>, 453.1320 (calc. 453.1315) [M<sub>3</sub>Na – 3Na]<sup>3–</sup>; ESI MS/MS (–) *m/z* 797.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – Xyl + H]<sup>–</sup>; 695.2 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – NaSO<sub>3</sub> – agl – Xyl + 2H]<sup>–</sup>; 665.1 [M<sub>3</sub>Na – Na<sub>2</sub>SO<sub>4</sub> – agl – 2Xyl + 2H]<sup>–</sup>; 563.1

$[M_{3Na} - Na_2SO_4 - NaSO_3 - agl - 2Xyl + 3H]^-$ , 519.0  $[M_{3Na} - Na_2SO_4 - agl - 2Xyl - Qui + H]^-$ ; 417.1  $[M_{3Na} - Na_2SO_4 - NaSO_3 - agl - 2Xyl - Qui + 2H]^-$ .

### 3.4. Cell Culture

Splenocytes from CD-1 line mice were used. The spleen were isolated from mice and centrifuged ( $450 \times g$ ) for 5 min. The splenocytes were washed three times and resuspended in phosphate-buffered saline, pH 7.2–7.4 at final concentration  $2\text{--}5 \times 10^6$  cells/mL.

### 3.5. Cytotoxicity Activity

Water solutions (10  $\mu$ L), containing different concentrations (0.12–100.00  $\mu$ M) of the tested substances and 90  $\mu$ L of the cell suspension were added to the wells of a 96-well plate and incubated for 1 h at 37  $^\circ$ C. After the incubation, 10  $\mu$ L of the cell suspension were added to 10  $\mu$ L of trypan blue solution (0.4% in PBS) and placed on a slide. After 1–5 min incubation the amount of alive and dead cells were calculated with an Imager A1 optical microscope (Carl Zeiss, Oberkochen, Germany) using the AxioVision (Carl Zeiss) 4.7.1 software. The cytotoxic activity of the substances was calculated as the ratio of the dead cells to general cells amount. ED<sub>50</sub> was calculated with SigmaPlot 10.0 software.

### 3.6. Hemolytic Activity

Blood was taken from a CD-1 mouse. The erythrocytes were washed three times with 0.9% NaCl, centrifuged ( $450 \times g$ ) on a LABOFUGE 400R centrifuge (Heraeus, Hanau, Germany) for 5 min [14] followed by resuspension in phosphate-buffered saline (PBS), pH 7.2–7.4. Erythrocytes were used at a concentration providing an optical density of 1.5 at 700 nm for a non-hemolyzed sample. Twenty  $\mu$ L of a water solution of test substance with a fixed concentration (0.12–100.00  $\mu$ M) were added to a well of a 96-well plate containing 180  $\mu$ L of the erythrocyte suspension and incubated for 1 h at 37  $^\circ$ C. The plates were centrifuged ( $900 \times g$ ) on a LMC-3000 laboratory centrifuge (Biosan, Riga, Latvia) for 10 min as proposed in [15]. Ten  $\mu$ L of the supernatant were placed on special microplate with a plate bottom for determination of the optical density on a Multiskan FC spectrophotometer (Termo Fisher Scientific, Waltham, MA, USA) at  $\lambda = 570$  nm. ED<sub>50</sub> values were calculated using the SigmaPlot 3.0 software. Triton X-100 (Biolot, Saint Petersburg, Russian Federation) at a concentration of 1%, causing 100% cell hemolysis was used as positive control. The erythrocyte suspension in phosphate-buffered saline, pH 7.2–7.4 (PBS) with 20  $\mu$ L of the solvent without a tested compound was used as negative control.

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**Conflicts of Interest:** The authors declare no conflict of interest.

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**Sample Availability:** Samples of the compounds are available from the authors.



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