# Synthesis, Nicotinic Acetylcholine Receptor Binding, and Antinociceptive Properties of $\mathbf{2}^{\prime}$-Fluoro-3'-(substituted pyridinyl)-7deschloroepibatidine Analogues 

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S Supporting Information


#### Abstract

Fluoro-3-(substituted pyridine)epibatidine analogues 7a-e and 8a-e were synthesized, and their in vitro and in vivo $n A C h R$ properties were determined. $2^{\prime}$ -Fluoro-3'-(4"-pyridinyl)deschloroepibatidine (7a) and $2^{\prime}$-fluoro- $3^{\prime}$-( $3^{\prime \prime}$-pyridinyl)deschloroepibatidine (8a) were synthesized as bioisosteres of the $4^{\prime}$-nitrophenyl lead compounds $5 \mathbf{5 a}$ and $5 \mathbf{g}$. Comparison of the in vitro nAChR properties of 7 a and $\mathbf{8 a}$ to those of 5 a and 5 g showed that 7 a and 8 a had in vitro nAChR properties similar to those of 5 a and 5 g but both were more selective for the $\alpha 4 \beta 2$-nAChR relative to the $\alpha 3 \beta 4$ - and $\alpha 7-\mathrm{nAChRs}$ than $\mathbf{5 a}$ and $\mathbf{5 g}$. The in vivo nAChR properties in mice of $\mathbf{7 a}$ were similar to those of $\mathbf{5 a}$. In contrast, 8a was an agonist in all four mouse acute tests, whereas 5 g was active only in a spontaneous activity test. In addition, $\mathbf{5 g}$ was a nicotine antagonist in both the tail-flick and hot-plate tests, whereas $8 \mathbf{a}$ was an antagonist only in the tailflick test.


## INTRODUCTION

Tobacco use continues to be the leading cause of preventable deaths in the United States as well as globally. Current statistics reveal that smoking-related diseases are responsible for nearly 6 million premature deaths globally annually. ${ }^{1}$ In the United States, tobacco use is responsible for nearly 1 in 5 deaths that is approximately 443000 premature deaths. ${ }^{2}$ In 2010, an estimated $23 \%$ ( 58.3 million) of U.S. adults were current cigarette smokers. ${ }^{3}$ Tobacco use increases the risk of multiple cancers such as cancers of the lung, mouth, nasal cavities, larynx, pharynx, esophagus, stomach, colorectum, liver, pancreas, kidney, bladder, uterine, cervix, ovary, and myeloid cells. Therefore, smoking accounts for at least $30 \%$ of all cancer deaths and $87 \%$ of lung cancer deaths. ${ }^{4}$

Due to the well-documented negative heath consequences, approximately $70 \%$ of smokers want to quit and about $40 \%$ try to quit every year. Of those who try to quit, only about $7 \%$ stay off nicotine for more than a year. The vast majority do not make it even a week without cigarettes. The major factor that is attributed to the initiation and sustaining of smoking is the presence of nicotine ( $\mathbf{1}$ ), the addictive substance in tobacco. Nicotine can produce a myriad of behavioral effects and is unquestionably one of the most popular and powerful reinforcing agents. Both the psychological and physiological effects of tobacco smoke are a result of nicotine's activation of various nicotinic acetylcholine receptor (nAChR) subtypes. For
example, nicotine interacts with $\alpha 4 \beta 2-, \alpha 4 \beta 2 \alpha 6^{*}$-, $\alpha 4 \beta 2 \alpha 5^{*}$, and $\alpha 7-\mathrm{nAChR}$ in the dopaminergic mesolimbic pathway, a brain system thought to mediate the pleasurable and rewarding effects of most substances of abuse, including nicotine. ${ }^{5}$ In addition, currently, the few treatments for nicotine dependence include nicotine replacement therapies (NRT); the antidepressant buproprion ${ }^{6,7}$ (2), which acts as a dopamine uptake inhibitor in addition to its properties as a nicotinic antagonist of $\alpha 3 \beta 4$ - and $\alpha 4 \beta 2$-nAChRs; ${ }^{8}$ and the FDA-approved varenicline ${ }^{9,10}$ (3), which acts as a partial nicotine agonist at the $\alpha 4 \beta 2$ and a full agonist at the $\alpha 3 \beta 4$ - and $\alpha 7$-nAChRs. ${ }^{11}$ In addition, varenicline has affinity for $\alpha 6 \beta 2^{*}$-nAChR equal to that at $\alpha 4 \beta 2$ nAChR , but functionally varenicline was more potent in stimulating $\alpha 6 \beta 2^{*}$ versus $\alpha 4 \beta 2^{*}$ mediated $\left[{ }^{3} \mathrm{H}\right]$ dopamine release from rat striatal synaptosomes. ${ }^{12}$ However, side effects such as gastrointestinal disturbances (nausea and vomiting) and neuropsychiatric effects (trouble sleeping, unusual dreams, violent or suicidal ideation) were frequently reported with the use of varenicline. In addition, recent evidence suggests that varenicline produces increased risk of heart attack, stroke, and/ or other cardiovascular problems. ${ }^{13}$ Therefore, there is need for development of new and improved pharmacotherapies for smoking cessation.

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Chart 1. Structures of Compounds 1-4, 5a-g, 6, 7a-e, and 8a-e


1


2

3

N $X$

6

7a, $X=H$
7a, $X=H$
7b, $X=F$
7c, $X=\mathrm{Cl}$
7d, $X=\mathrm{NH}_{2}$
7e, $X=\mathrm{CH}_{3} \mathrm{O}$

5a, $X=\mathrm{NO}_{2}, Y=\mathrm{H}$
b, $X=C F_{3}, Y=H$
c, $X=C N, Y=H$
d, $X=\mathrm{SO}_{2} \mathrm{CH}_{3}, Y=\mathrm{H}$
e, $X=\mathrm{SO}_{2} \mathrm{NH}_{2}, Y=\mathrm{H}$
f, $X=\mathrm{SO}_{2} \mathrm{CF}_{3}, Y=\mathrm{H}$
g, $X=H, Y=\mathrm{NO}_{2}$

Scheme $1^{a}$

${ }^{a}$ Reagents and conditions: (a) $\mathrm{Pd}(\mathrm{OAc})_{2}, \mathrm{P}(o \text {-tolyl })_{3}$, pyridinyl boronic acid, $\mathrm{Na}_{2} \mathrm{CO}_{3}, \mathrm{DME}, \mathrm{H}_{2} \mathrm{O}, 8{ }^{\circ} \mathrm{C}, 5 \mathrm{~h}$ for the preparation of $\mathbf{1 1 b}, \mathbf{1 1 c}, \mathbf{1 2 b}$, and 12c; (b) pyridinyl boronic acid, $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}, \mathrm{~K}_{2} \mathrm{CO}_{3}$, toluene, $\mathrm{EtOH}, \mathrm{H}_{2} \mathrm{O}$, reflux, 24 h for the preparation of 11a, 11d and 12a; (c) 70\% HFpyridine, $\mathrm{NaNO}_{2}$; (d) 2-aminopyridine-5-pinacol boronic ester, $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}, \mathrm{~K}_{2} \mathrm{CO}_{3}, 1,4$-dioxane, $\mathrm{H}_{2} \mathrm{O}, 110{ }^{\circ} \mathrm{C}, 18 \mathrm{~h}$.

The natural alkaloid epibatidine ( 4 , exo-2-( $2^{\prime}$-chloro-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane) is an important lead structure in the development of pharmacotherapies for treating nicotine addiction as well as other central nervous system (CNS) disorders including Alzheimers and Parkinson's diseases, pain, schizophrenia, anxiety, depression, and Tourette's syndrome among others. ${ }^{14}$ Since its isolation and structural determination in 1992, ${ }^{15}$ epibatidine has drawn a
lot of interest because of its very high affinity for the $\alpha 4 \beta 2^{*}$ nAChRs. ${ }^{16,17}$ In previous studies, we reported the synthesis, nAChR binding affinity, and pharmacological properties of a number of epibatidine analogues. ${ }^{18,19}$ Interestingly, some analogues retained high affinity for nAChR but unlike epibatidine showed no agonist activity in the acute mouse antinociception test and were antagonists of nicotine-induced antinociception in these assays. ${ }^{18-20}$ For example, we identified

Scheme $2^{a}$

${ }^{a}$ Reagents and conditions: (a) bis(pinacolato)diboron, $\mathrm{Pd}_{2} \mathrm{dba}_{3}$ ( $3 \mathrm{~mol} \%$ ), XPhos ( $16 \mathrm{~mol} \%$ ), 1,4-dioxane, $110{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}$; (b) $\mathrm{Pd}_{2} \mathrm{dba} 3(3 \mathrm{~mol} \%)$, $\mathrm{K}_{3} \mathrm{PO}_{4}$ (2.3 equiv), 1,4-dioxane, $110{ }^{\circ} \mathrm{C}, 18 \mathrm{~h}$.
$2^{\prime}$-fluoro- $3^{\prime}$-(4-nitrophenyl)deschloroepibatidine (5a), also referred to as RTI-7527-102 and 4-nitro-PFEB, as an nAChR ligand with a $K_{\mathrm{i}}$ value of 0.009 nM for inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding. This compound also showed potent antagonism of nicotine-induced antinociception in the tail-flick and hot-plate tests in mice. ${ }^{21}$ In a separate study, we showed that 5 a was a competitive antagonist of human $\alpha 4 \beta 2$-nAChRs with a potency 17 -fold higher than that of dihydro- $\beta$ erythroidine (6) with very low efficacy at $\alpha 3 \beta 4$ - and $\alpha 7$ nAChRs. ${ }^{22}$ In a more recent study, the $\alpha 4 \beta 2-\mathrm{nAChR}$ antagonist 5a attenuated the discriminative stimulus effects of nicotine, reduced nicotine's ability to facilitate intracranial selfstimulation (ICSS), blocked conditioned place preference (CPP) produced by nicotine in mice, and dose-dependently blocked nicotine self-administration in rats. ${ }^{23}$ Thus, 5a has both in vitro and in vivo properties thought to be favorable for a potential pharmacotherapy to treat smokers. However, the presence of a nitro-substituted phenyl group, a system that is associated with toxicity via partial reduction in vivo to the hydroxylamine, which can undergo metabolic activation to an electrophilic nitroso species of $\mathbf{5 a}$, raises concern about its future development. In a recent study, we reported that replacement of the 4-nitro group in $\mathbf{5 a}$ by other strong electron-withdrawing groups led to compounds $\mathbf{5 b} \mathbf{- g}$ that retained high affinity for $\alpha 4 \beta 2$-nAChRs and potent antagonist activity in the tail-flick test. ${ }^{24}$

In this study, we report the synthesis, nAChR binding, and pharmacological properties of compounds $7 \mathbf{a}-\mathbf{e}$ and $8 \mathbf{8}-\mathbf{e}$. Compound 7a is a bioisosteric analogue of 5 a where the nitrophenyl group has been replaced by a pyridine nitrogen. Compound 8a is a similar bioisosteric analogue of 5 g , a compound that has a $K_{\mathrm{i}}$ value of 0.053 nM of affinity for inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding and $\mathrm{AD}_{50}$ values of 0.5 and $130 \mu \mathrm{~g} / \mathrm{kg}$ in the tail-flick and hot-plate tests. ${ }^{24}$ The syntheses and evaluation of analogues $7 \mathbf{b}-\mathbf{e}$ and $\mathbf{8 b}-\mathbf{e}$ allowed a determination of the effects of electron-withdrawing and -donating groups on the pyridine ring. See Chart 1 for the structures of the compounds described in the above paragraphs.

Chemistry. The synthetic route to the $7 \mathbf{a}-\mathbf{c}$ and $\mathbf{8 a}-\mathbf{e}$ analogues commenced with the intermediate 7 -tert-butoxycar-bonyl-2-exo-(2'-amino-3'-bromo-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (9) prepared in several steps from $N$-Boc pyrrole as reported in earlier work. ${ }^{25,26}$ As outlined in Scheme 1, the Suzuki cross-couplings of the haloboronic acids, that is, 2-
fluoropyridine-5-boronic acid, 2-fluoropyridine-4-boronic acid, 2-chloropyridine-5-boronic acid, and 2-chloropyridine-4-boronic acid with the $2^{\prime}$-amino- $3^{\prime}$-bromo compound 9 , carried out in the presence of palladium diacetate, tri-( $(0$-tolyl $)$ phosphine, and sodium carbonate, heated at $80^{\circ} \mathrm{C}$ in 1,2 -dimethoxyethane and water for 5 h furnished the bipyridine intermediates 11 b , 11c, 12b, and 12c. ${ }^{27}$ Suzuki cross-coupling of pyridine-4boronic acid, pyridine-3-boronic acid, and 2-methoxypyridine5 -boronic acids with 9 in the presence of tetrakis(triphenylphosphine) palladium( 0 ) as the catalyst, potassium carbonate as the base, and toluene $(15 \mathrm{~mL})$, ethanol $(1.5 \mathrm{~mL})$, and water $(1.5 \mathrm{~mL})$ as solvents and heating at reflux for 24 h in a sealed tube provided the cross-coupled products 11a, 11d, and 12a in good yields. Conversion of the amino group to the fluoro group along with a concomitant removal of the tertbutyloxycarbonyl protecting group in the intermediates 11a-d and 12a-c performed through the diazotization reaction with sodium nitrite in the presence of hydrogen fluoride in pyridine ( $70 \%$ ) furnished the products $8 \mathbf{a}-\mathbf{e}$ and $\mathbf{7 a}-\mathbf{c}$. Compound $\mathbf{8 d}$ was synthesized by subjecting 2 -exo-( $2^{\prime}$-fluoro- $3^{\prime}$-bromo-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane $10^{21}$ to a SuzukiMiyaura cross-coupling with 2 -aminopyridine-5-pinacol boronic ester in the presence of tetrakis(triphenylphosphine) palladium(0), potassium carbonate, 1,4-dioxane, and water, heated at $110^{\circ} \mathrm{C}$ in a sealed tube overnight. The reaction furnished the diamine 8 d in a $67 \%$ yield (Scheme 1).

The synthesis of the 2 -exo-[2'-fluoro-3'-(2-aminopyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane 7d, was accomplished in a "one-pot" reaction that combined the borylation and the Suzuki-Miyaura steps (Scheme 2). The borylation reaction was accomplished using Buchwald's dialkylphosphinobiphenyl ligand, 2-dicyclohexylphosphino- $2^{\prime}, 4^{\prime}, 6^{\prime}$-triisopropylbiphenyl (XPhos), and tris(dibenzylideneacetone)dipalladium $(0)$ as the catalytic system. ${ }^{28,29}$ Cross-coupling of 2-amino-4bromopryidine (13) and bis(pinacolato)diborane in the presence of XPhos, tris(dibenzylideneacetone)dipalladium (0), and potassium acetate heated at $110^{\circ} \mathrm{C}$ in 1,4-dioxane converted 13 to the boronic ester, which was carried on to the next step directly by addition of $2^{\prime}$-fluoro- $3^{\prime}$-bromo intermediate 10, tribasic potassium phosphate as base, and an additional $3 \mathrm{~mol} \%$ of tris(dibenzylideneacetone)dipalladium (0) and heating at $110^{\circ} \mathrm{C}$ for 18 h to provide 7 d . Compound 7 e was synthesized as shown in Scheme 3. Borylation of $\mathbf{1 4}$ was achieved by cross-coupling with bis(pinacolato)diborane heated

## Scheme $3^{a}$


${ }^{a}$ Reagents and conditions: (a) bis(pinacolato)diboron, KOAc, $\mathrm{PdCl}_{2}(\mathrm{dppf})$, DMF, $80{ }^{\circ} \mathrm{C}$; (b) $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}, \mathrm{~K}_{2} \mathrm{CO}_{3}, 1,4$-dioxane, $\mathrm{H}_{2} \mathrm{O}, 110^{\circ} \mathrm{C}, 18 \mathrm{~h}$.
at $80{ }^{\circ} \mathrm{C}$ in the presence of $1,1^{\prime}$-bis(diphenylphosphino)-ferrocene-palladium(II)dichloride, potassium acetate, and dimethylformamide as the solvent to provide the pinacol boronic ester 15 (Scheme 3). Compound $\mathbf{1 5}$ was cross-coupled with $2^{\prime}$-fluoro- $3^{\prime}$-bromo intermediate 10 to furnish 2 -exo- $\left[2^{\prime}\right.$ -
fluoro-3'-(2-methoxypyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (7e).

Biology. The inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding at $\alpha 4 \beta 22^{*}$ nAChRs were conducted as previously reported. ${ }^{21}$ The binding assays were performed using tissue homogenates prepared from freshly collected cerebral cortices from adult male Sprague-Dawley rats. These homogenates were frozen at $-80^{\circ} \mathrm{C}$ until use. It should be noted that, although this brain region contains a variety of nAChRs, $\alpha 4 \beta 2$ is the predominant ( $>90 \%$ ) subtype. The epibatidine analogues were tested for agonist and antagonist activity at rat $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$ nAChRs in an in vitro electrophysiology assay as previously described. ${ }^{24}$ The epibatidine analogues were tested in mice for their effects on body temperature and two pain models (tailflick and hot-plate assays) after acute administration as previously described. ${ }^{21}$ For the antagonist experiments, male ICR adult mice were pretreated subcutaneously (sc) with either saline or epibatidine analogues 10 min before nicotine. Nicotine was administered at a dose of $2.5 \mathrm{mg} / \mathrm{kg}$, sc (an $\mathrm{ED}_{84}$ dose), and mice were tested 5 min later. $\mathrm{ED}_{50}$ and $\mathrm{AD}_{50}$ values with $95 \%$ confidence limits were determined.

## RESULTS AND DISCUSSION

The nAChR binding affinities and the functional nicotinic pharmacological properties of $2^{\prime}$-fluoro- $3^{\prime}$-(substituted pyridine)deschloroepibatidine analogues $7 \mathbf{a}-\mathbf{e}$ and $8 \mathbf{a}-\mathbf{e}$ were determined. The $K_{\mathrm{i}}$ values for the inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine

Table 1. Radioligand Binding and Efficacy Profile Data for $\mathbf{2}^{\prime}$-Fluoro-3'-(substituted pyridine)deschloroepibatidine Analogues


5


7


8

|  |  |  |  | agonist activity at $100 \mu \mathrm{M}$ (\% of max ACh activity) ${ }^{d}$ |  |  | antagonist activity at $100 \mu \mathrm{M}$ (\% ACh response remaining $)^{e}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compd ${ }^{\text {a }}$ | X | Y | $\alpha 4 \beta 2 *\left[{ }^{3} \mathrm{H}\right] \mathrm{epibatidine}^{b}\left(K_{\mathrm{i}}, \mathrm{nM}\right) \text { (Hill }$ | $\alpha 4 \beta 2$ | $\alpha 3 \beta 4$ | $\alpha 7$ | $\alpha 4 \beta 2$ | $\alpha 3 \beta 4$ | $\alpha 7$ |
| nicotine ${ }^{c}$ |  |  | $1.50 \pm 0.30$ | $40 \pm 1$ | $37 \pm 3$ | $43 \pm 5$ | nd | nd | nd |
| nat-epibatidine |  |  | $0.026 \pm 0.002$ | $131 \pm 13$ | $97 \pm 4$ | $150 \pm 8$ | nd | nd | nd |
| varenicline |  |  | $0.12 \pm 0.02$ | $13 \pm 0.4$ | $66 \pm 4$ | $74 \pm 5$ | $38 \pm 2$ | nd | nd |
| 5a ${ }^{f}$ | $\mathrm{NO}_{2}$ | H | $0.009 \pm 0.001$ | 0 | $4 \pm 1$ | $6 \pm 1$ | $6 \pm 1$ | $9 \pm 2$ | $55 \pm 6$ |
| $5 \mathrm{~g}^{f}$ | H | $\mathrm{NO}_{2}$ | $0.053 \pm 0.004$ | $1.0 \pm 0.1$ | $1.3 \pm 0.2$ | $6 \pm 1$ | $5 \pm 1$ | $2.1 \pm 0.4$ | $17 \pm 3$ |
| 7a | H |  | $0.12 \pm 0.03$ | $1.5 \pm 0.3$ | $7 \pm 1$ | $8 \pm 2$ | $6 \pm 1$ | $34 \pm 4$ | $45 \pm 4$ |
| 7 b | F |  | $0.067 \pm 0.01$ | $3 \pm 0.3$ | $5 \pm 0.4$ | $2 \pm 0.5$ | $10 \pm 1$ | $27 \pm 2$ | $44 \pm 4$ |
| 7 c | Cl |  | $1.18 \pm 0.14$ | $1.2 \pm 0.2$ | $4 \pm 0.4$ | $3 \pm 0.3$ | $7 \pm 1$ | $20 \pm 1$ | $31 \pm 3$ |
| 7 d | $\mathrm{NH}_{2}$ |  | $0.13 \pm 0.005$ | $3.9 \pm 0.5$ | 0 | $10 \pm 4$ | $8 \pm 1$ | $9 \pm 1$ | $32 \pm 10$ |
| 7 e | $\mathrm{CH}_{3}$ |  | $0.04 \pm 0.012$ | 0 | $1.5 \pm 0.2$ | $2.1 \pm 0.8$ | $4 \pm 1$ | $6 \pm 1$ | $12 \pm 4$ |
| 8a | H |  | $0.35 \pm 0.038$ | $2 \pm 0.3$ | $3 \pm 0.5$ | $7 \pm 2$ | $7 \pm 1$ | $15 \pm 2$ | $54 \pm 9$ |
| 8b | F |  | $0.049 \pm 0.02$ | $9 \pm 1$ | $9 \pm 1$ | $8 \pm 0.7$ | $24 \pm 4$ | $73 \pm 8$ | $75 \pm 18$ |
| 8c | Cl |  | $0.063 \pm 0.08$ | $1 \pm 0.2$ | $14 \pm 1$ | 0 | $8 \pm 1$ | $23 \pm 2$ | $32 \pm 4$ |
| 8d | $\mathrm{NH}_{2}$ |  | $0.25 \pm 0.033$ | $2.7 \pm 0.3$ | $1.3 \pm 0.3$ | $5 \pm 2$ | $7 \pm 1$ | $9 \pm 1$ | $50 \pm 6$ |
| 8 e | $\mathrm{CH}_{3} \mathrm{O}$ |  | $0.13 \pm 0.027$ | $5 \pm 0.1$ | $9 \pm 2$ | $22 \pm 4$ | $14 \pm 1$ | $23 \pm 6$ | $41 \pm 2$ |

${ }^{a}$ All compounds were tested as their $( \pm)$-isomers. ${ }^{b}$ The $K_{\mathrm{d}}$ for $( \pm)-\left[{ }^{3} \mathrm{H}\right]$ epibatidine is $0.02 \mathrm{nM} .{ }^{c}$ Data taken from ref 21 . ${ }^{d}$ Assessed by comparing the current response to $100 \mu \mathrm{M}$ of each compound to the mean current response of three preceding applications of ACh , applied at an $\mathrm{EC}_{20}$ concentration ( $20 \mu \mathrm{M}$ for $\alpha 4 \beta 2,110 \mu \mathrm{M}$ for $\alpha 3 \beta 4$ ) or an $\mathrm{EC}_{50}$ concentration ( $300 \mu \mathrm{M}$ for $\alpha 7$ ) and expressed as a percentage of the maximal response to ACh . ${ }^{e}$ Assessed by comparing the current response to an $\mathrm{EC}_{50}$ concentration of $\mathrm{ACh}(70 \mu \mathrm{M}$ for $\alpha 4 \beta 2,200 \mu \mathrm{M}$ for $\alpha 3 \beta 4,300 \mu \mathrm{M}$ for $\alpha 7)$ in the presence of $100 \mu \mathrm{M}$ of each compound to the mean current response of three preceding applications of ACh alone. ${ }^{f}$ Data taken from ref 24.
binding at the $\alpha 4 \beta 2^{*}$-nAChRs for compounds $7 \mathbf{a}-\mathbf{e}$ and $8 \mathbf{a}-\mathbf{e}$ along with reference compounds nat-epibatidine (4), varenicline (3), and the lead compounds 5 a and 5 g are listed in Table 1. The reference standards nat-epibatidine and varenicline and lead compounds $5 \mathbf{a}$ and 5 g have $K_{\mathrm{i}}$ values of $0.026,0.12,0.009$, and 0.053 nM for the $\alpha 4 \beta 22^{*}$-nAChR, respectively. ${ }^{24}$ The pyridine bioisosteres $\mathbf{7 a}$ and $\mathbf{8 a}$ of $\mathbf{5 a}$ and $\mathbf{5 b}$, respectively, had $K_{\mathrm{i}}$ values of 0.12 and 0.35 nM , respectively, for $\alpha 4 \beta 2^{*}$-nAChR. Even though 7a and 8a have $K_{i}$ values slightly less than $5 a$ and $\mathbf{5 g}$, respectively, their $K_{\mathrm{i}}$ values are still subnanomolar and thus have potent affinity for nAChRs labeled by $\left[{ }^{3} \mathrm{H}\right]$ epibatidine.

Substitution of the $4^{\prime}$-pyridyl and $3^{\prime}$ 'pyridyl rings of analogues $7 \mathbf{a}$ and $8 \mathbf{a}$, respectively, with $3^{\prime}$ - and $4^{\prime}$-substituents, respectively, had only small effects on $\alpha 4 \beta 2^{*}$-nAChR binding affinity. The $K_{\mathrm{i}}$ values varied from 0.04 to 1.18 nM . With the exception of the $3^{\prime}$-chloro analogue 7 c , all the $3^{\prime}$-substituted $4^{\prime}$ pyridyl analogues $7 \mathbf{b}-\mathbf{e}$ had very high affinity for $\alpha 4 \beta 2^{*}$ nAChRs . The $K_{\mathrm{i}}$ values ranged from 0.04 for 7 e to 0.13 nM for 7d. Even the $3^{\prime}$-chloro substituted analogue had a $K_{i}$ value of 1.18 nM . All the $4^{\prime}$-substituted $3^{\prime}$-pyridyl analogues had $K_{i}$ values of $0.04-0.35 \mathrm{nM}$ and thus very high affinity for $\alpha 4 \beta 2^{*}$ nAChR. The presence of $3^{\prime}$-substituents on the $4^{\prime}$-pyridyl analogues $\mathbf{7 b}$-e or $4^{\prime}$-substituents on the $3^{\prime}$-pyridyl analogues $\mathbf{8 b} \mathbf{- e}$ did not show any clear structure affinity patterns. In the case of the $7 \mathbf{a}-\mathbf{e}$ series, the two highest affinity compounds were the electron-withdrawing $3^{\prime}$-fluoro analogue $7 \mathbf{b}$ ( $K_{i}=$ $0.067)$ and the electron-donating methoxy analogue $7 \mathrm{e}\left(K_{\mathrm{i}}=\right.$ 0.04 ). For the $4^{\prime}$-substituted $3^{\prime}$-pyridyl analogues, the $4^{\prime}$-fluoro and $4^{\prime}$-chloro electron-withdrawing analogues $\mathbf{8 b}$ and $8 \mathbf{c}$ ( $K_{\mathrm{i}}=$ 0.049 and 0.063 nM ) had higher $\alpha 4 \beta 2^{*}$-nAChR affinity than the $3^{\prime}$-amino and $3^{\prime}$-methoxy electron-donating analogues $\mathbf{8 d}$ and $8 \mathbf{e}\left(K_{\mathrm{i}}=0.25\right.$ and 0.13 nM$)$. However, all four compounds have subnanomolar $K_{\mathrm{i}}$ values.

The receptor subtype selectivity of $7 \mathbf{a}-\mathbf{e}$ and $\mathbf{8 a - e}$ was assessed in an electrophysiological assay using rat $\alpha 4 \beta 2-, \alpha 3 \beta 4-$, and $\alpha 7$-nAChRs expressed in Xenopus oocytes and assayed by a two-electrode voltage clamp. Compounds were compared to previously determined values for nicotine, nat-epibatidine, varenicline, and compounds 5a and 5 g (Table 1). Current responses to a high concentration ( $100 \mu \mathrm{M}$ ) of each compound were compared to the maximum response that can be achieved with acetylcholine. All compounds differed dramatically from nat-epibatidine (a full agonist at $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$ nAChRs ). Compounds $7 \mathrm{a}, 7 \mathrm{c}, 7 \mathrm{e}, 8 \mathrm{a}$, and 8 c displayed little or no agonist activity at $\alpha 4 \beta 2$ in this initial screen, while $7 \mathbf{b}, 7 \mathrm{~d}$, $\mathbf{8 b}$, and $\mathbf{8 d} \mathbf{- e}$ had a low level of agonist activity at this subtype. Compounds $7 \mathbf{d}$ and $8 \mathbf{d}$ had little or no agonist activity at $\alpha 3 \beta 4$, while $7 \mathbf{a}-\mathbf{c}, \mathbf{8 a - c}$, and $\mathbf{8 e}$ had a low level of agonist activity at this subtype. At $\alpha 7$-nAChRs, compounds $7 \mathbf{b}, 7 \mathbf{e}$, and $8 \mathbf{c}$ had little or no agonist activity, compounds $7 \mathbf{a}, 7 \mathbf{c}-\mathbf{d}, 8 \mathbf{a}-\mathbf{b}$, and $\mathbf{8 d}$ displayed low levels of agonist activity, and compound $8 \mathbf{e}$ showed a moderate level of agonist activity ( $22 \pm 4 \%$ of the maximal acetylcholine response). Compounds $7 \mathbf{a}-\mathbf{e}$ and $\mathbf{8 a}-\mathbf{e}$ all showed lower agonist activity at $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$ nAChR than did varenicline (a partial agonist at $\alpha 4 \beta 2$ - and a full agonist at $\alpha 3 \beta 4-$ and $\alpha 7$-nAChRs).

As an initial screen of antagonist properties, we measured the current response to an $\mathrm{EC}_{50}$ concentration of acetylcholine in the presence of $100 \mu \mathrm{M}$ of each compound and compared this to a preceding current response to acetylcholine alone (Table 1). Compounds $7 \mathbf{a}-\mathbf{e}$ and $\mathbf{8 a - e}$ antagonized, to varying extents, the $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$-nAChR subtypes in this preliminary screen. This contrasts with varenicline, which can
antagonize $\alpha 4 \beta 2$ receptors but is a full agonist at $\alpha 3 \beta 4$ - and $\alpha 7$ nAChRs. The ability of 8 e to both activate and antagonize the $\alpha 7-\mathrm{nAChR}$ subtype indicates that 8 e is a partial agonist at this receptor. The results from this initial screen suggested that some compounds in this series may be selective $\alpha 4 \beta 2$ antagonists, and 7a, 7c, and 8a were selected for more detailed studies.

We examined the subtype selectivity of antagonist activity of compounds $7 \mathrm{a}, 7 \mathrm{c}$, and 8 a in more detail by generating concentration-inhibition curves (Table 2) and compared the

Table 2. Comparison of Antagonist Potency ( $\mathrm{IC}_{50}$ Values) for Several Epibatidine Analogues at $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$ nAChRs

|  | antagonist activity ${ }^{a} \mathrm{IC}_{50}(\mu \mathrm{M})$ |  |  |
| :--- | :--- | :--- | :--- |
| compd | $\alpha 4 \beta 2$ | $\alpha 3 \beta 4$ | $\alpha 7$ |
| varenicline | $0.20 \pm 0.03^{b}$ | $d$ | $d$ |
| $\mathbf{5 a}$ | $3.2 \pm 0.2^{c}$ | $7.9 \pm 0.5^{c}$ | $32 \pm 12^{c}$ |
| 5 g | $4.3 \pm 0.6$ | $3.9 \pm 0.3$ | $23 \pm 5$ |
| 7 a | $1.4 \pm 0.1$ | $8 \pm 1$ | $75 \pm 16$ |
| 7 c | $2.0 \pm 0.4$ | $8.8 \pm 0.9$ | $56 \pm 10$ |
| $\mathbf{8 a}$ | $1.7 \pm 0.2$ | $18 \pm 3$ | $99 \pm 24$ |

${ }^{a}$ Antagonist activity of $7 \mathrm{a}, 7 \mathrm{c}$, and 8 a was assessed in the in vitro electrophysiology assay at a range of concentrations to generate concentration-inhibition curves. Data were fit to the following equation: $I=I_{\max } /\left[1+\left(\mathrm{IC}_{50} / X\right)^{n}\right]$, where $I$ is the current response at a compound concentration $(X), I_{\max }$ is the maximum current, $\mathrm{IC}_{50}$ is the compound concentration producing half-maximal inhibition of the current response, and $n$ is the Hill coefficient. ${ }^{b}$ Data taken from ref 11. ${ }^{c}$ Data taken from ref 24. ${ }^{d}$ Varenicline is an agonist at $\alpha 3 \beta 4$ - and $\alpha 7$ nAChRs, with an $\mathrm{EC}_{50}$ of $55 \pm 8$ and $18 \pm 6 \mu \mathrm{M}$, respectively (ref 11 ).
$\mathrm{IC}_{50}$ values to the lead nitro compounds $\mathbf{5 a}$ and $\mathbf{5 g}$. Compound 7 a , the bioisosteric analogue of 5 a , where the nitro group has been replaced by a pyridine nitrogen, displayed an improved $\alpha 4 \beta 2$ selectivity. While 5a was 2.5 -fold selective for $\alpha 4 \beta 2$ over $\alpha 3 \beta 4$ and 10 -fold selective for $\alpha 4 \beta 2$ over $\alpha 7$, compound 7 a was 5.7-fold selective for $\alpha 4 \beta 2$ over $\alpha 3 \beta 4$ and 54-fold selective for $\alpha 4 \beta 2$ over $\alpha 7$. Similarly, compound 8a, the bioisosteric analogue of 5 g , where the nitro group has been replaced by a pyridine nitrogen, displayed an improved $\alpha 4 \beta 2$ selectivity. While 5 g was nonselective between $\alpha 4 \beta 2$ and $\alpha 3 \beta 4$ and 5 -fold selective for $\alpha 4 \beta 2$ over $\alpha 7$, compound 8 a was 11 -fold selective for $\alpha 4 \beta 2$ over $\alpha 3 \beta 4$ and 58 -fold selective for $\alpha 4 \beta 2$ over $\alpha 7$. We also examined 7 c but found it to be less selective for $\alpha 4 \beta 2$ than was 7a. None of these compounds was as potent an antagonist as varenicline at $\alpha 4 \beta 2$-nAChRs. The ability of these compounds to antagonize $\alpha 3 \beta 4$ - and $\alpha 7$-nAChRs differs markedly from varenicline, which is a full agonist at these subtypes.

The $3^{\prime}$-substituted $4^{\prime}$-pyridyl analogues $7 \mathbf{a}-\mathbf{e}$ and the $4^{\prime}$ substituted 3-pyridyl analogues ( $\mathbf{8 a - e}$ ) were evaluated for their in vivo nAChR properties in mice, and the results were compared to the properties of lead compounds 5 a and 5 g and varenicline (Table 3). Similar to 5a, the pyridine bioisostere 7a does not have agonist activity in the tail-flick and hot-plate tests but like 5a and varenicline did show activity in the hypothermia and spontaneous-activity tests. The $\mathrm{ED}_{50}$ values for 7 a in the hypothermia and spontaneous-activity tests were 1.69 and 0.38 $\mathrm{mg} / \mathrm{kg}$ compared to 0.21 and $0.22 \mathrm{mg} / \mathrm{kg}$ for 5 a . Similar to 5 a , 7a antagonized nicotine-induced antinociception in the tail-flick and hot-plate tests with $\mathrm{AD}_{50}$ values of 12 and $290 \mu \mathrm{~g} / \mathrm{kg}$, respectively, compared to 3 and $120 \mu \mathrm{~g} / \mathrm{kg}$ for $\mathbf{5 a}$. Thus, 7 a is a
Table 3. Antinociception, Hypothermia, and Spontaneous Activity Profile Data for $\mathbf{2}^{\prime}$-Fluoro-3'-(substituted pyridine)deschloroepibatidine Analogues


| /kg) |  | $\mathrm{AD}_{50}(\mu \mathrm{~g} / \mathrm{kg})$ |  |
| :---: | :---: | :---: | :---: |
| hypothermia | spontaneous activity | tail-flick | hot-plate |
| 1.0 (0.6-2.1) | 0.5 (0.15-0.78) |  |  |
| 0.004 (0.002-0.008) | 0.001 (0.0005-0.005) |  |  |
| 2.8 | 2.1 | 0.2 | 470 |
| 0.21 (0.04-1.9) | $0.22(0.04 \pm 1.2)$ | 3 (0.8-45) | 120 (10-900) |
| 0\% @ 10 | $6.5(5.3 \pm 8.3)$ | 0.5 (0.3-5) | 130 (50-290) |
| 1.69 (1.1-2.6) | 0.38 (0.2-2.7) | 12 (10-172) | 290 (19-991) |
| 1.58 (0.97-2.1) | 0.17 (0.08-1.5) | $4(0.1-70)$ | 117 (110-1100) |
| 2.74 (1.89-3.5) | 1.01 (0.27-3.7) | 320 (45-3262) | 1370 (180-1430) |
| 1.87 (0.1-35) | 0.61 (0.04-9.1) | 9 (0.4-19) | 10\% @ 10000 |
| 8.5 (1.9-38.6) | 1.82 (0.4-8.4) | 0.3 (0.02-5.7) | 40\% @ 10000 |
| 3.7 (2.9-4.5) | 0.69 (0.4-12.8) | 3 (0.5-24) | 10\%@1000 |
| 0.68 (0.52-1.1) | 0.38 (0.13-1.1) | 1\% @ 100 | 1\% @ 100 |
| 3.11 (1.5-5.1) | 1.58 (0.5-4.4) | 9 (2-38) | 2001 (297-3610) |
| 2.8 (2-3.8) | 184 (0.5-6.3) | 30 (3-35) | 50\%@10 |
| 0.77 (0.51-1.2) | 0.53 (0.19-1.1) | $21(3-125)$ | 0\%@100 |

very good bioisostere analogue of $\mathbf{5 a}$. In contrast, the $3^{\prime}$-pyridyl compound 8 a , which is the pyridine bioisostere of 5 g , was an agonist in all four mice acute tests, whereas 5 g was active only in the spontaneous-activity test. In addition, whereas $\mathbf{5 g}$ was an antagonist of nicotine-induced antinociception in the tail-flick and hot-plate tests with $\mathrm{AD}_{50}$ values of 0.5 and $130 \mu \mathrm{~g} / \mathrm{kg}$, respectively, $\mathbf{8 a}$ was an antagonist in only the tail-flick test with an $\mathrm{AD}_{50}$ value of $3 \mu \mathrm{~g} / \mathrm{kg}$. Somewhat surprisingly, the in vivo properties of $\mathbf{8 a}$ are quite different from those of $\mathbf{5 g}$, and thus, even though it is an interesting partial agonist, it is not a good bioisosteric analogue of 5 g in the mice in vivo test. These discrepancies between the in vivo and in vitro effects of 5 g and 8a could also be related to many factors such as differences in the metabolic profile and brain penetrability of these two analogues as well as the fact that we used expressed receptors systems and not native receptors preparations. Furthermore, since the agonistic response of nicotine in these two tests is largely mediated by $\alpha 4 \beta 2^{*}$ nAChR subtypes, it is possible that 5 g and 8 a differ in their affinity/activity at the various $\alpha 4 \beta 2^{*}$ nAChR subtypes mediating their pharmacological responses. In addition, in vivo regulation of nicotinic receptors such as $\alpha 4 \beta 2^{*}$ nAChR subtypes by these two analogues may differ also. For example, it is possible that, since $\mathbf{5 g}$ was more potent than $\mathbf{8 a}$ as a functional blocker in the tail-flick and hot-plate tests, in vivo desensitization/blockade of $\alpha 4 \beta 2^{*}$ receptors by 5 g is more pronounced. We recently reported that varenicline (3) and sazetidine, two $\alpha 4 \beta 2^{*}$ nicotinic partial agonists that differ in their desensitization properties, differ in their potency to act as functional antagonists of nicotine in these tests. ${ }^{30,31}$

Similar to the unsubstituted analogue $7 \mathbf{a}$, none of the $3^{\prime}$ substituted 4-pyridyl analogues $7 \mathbf{b}-\mathbf{e}$ had any agonist activity in the tail-flick and hot-plate tests (Table 3). In the case of the $4^{\prime}$ substituted $3^{\prime}$-pyridyl analogues, the $4^{\prime}$-fluoro and $4^{\prime}$-methoxy analogues $\mathbf{8 b}$ and $8 \mathbf{e}$ had agonist activity in the tail-flick and hot-plate tests. Similar to $5 \mathbf{5 a}, 5 \mathrm{~g}$, and varenicline, $\mathbf{7 b}-\mathbf{e}$ and $\mathbf{8 b}-\mathbf{e}$ had activity in the hypothermia and spontaneous-activity tests.

Analogues $7 \mathbf{b}-\mathbf{e}$ and $8 \mathbf{c}-\mathbf{e}$ antagonized nicotine-induced antinociception in the tail-flick test (Table 3). Analogues 7b-c and 8 c also antagonized nicotine-induced antinociception in the hot-plate test with $\mathrm{AD}_{50}$ values ranging from 117 to 1370 $\mu \mathrm{g} / \mathrm{kg}$. Compounds $7 \mathbf{a}$ and $7 \mathbf{b}$, which have $\mathrm{AD}_{50}$ values of 290 and $117 \mu \mathrm{~g} / \mathrm{kg}$, respectively, in the hot-plate test, compared to $470 \mu \mathrm{~g} / \mathrm{kg}$ for varenicline strongly suggest that these compounds have good brain penetration. Unlike any of the other pyridine substituted analogues, the $3^{\prime}$-fluoro- $4^{\prime}$-pyridyl analogue $\mathbf{8 b}$ was an agonist in all four acute mouse tests and had no antagonist properties.

Calculated physicochemical properties such as lipophilicity ( $\operatorname{clog} \mathrm{P}$ ), topological polar surface area (TPSA), and derived values such as $\log B B$ can be used as an indication of the potential of a compound for development as a CNS drug. These molecular descriptors were calculated for lead compounds 5 a and 5 g , bioisosteric analogues 7 a and 8 a , respectively, as well as compounds $\mathbf{7 b}-\mathbf{e}, \mathbf{8 b}-\mathbf{e}$ and reference compounds nicotine, epibatidine, and varenicline (Table 4). In general, CNS drugs have a clogP in the range $2-4,{ }^{32}$ TPSA less than $76 \AA \AA^{33}$ and $\operatorname{logBB}$ greater than $-1 .{ }^{34}$ All of the compounds have clog P values within or close to the desirable range and $\operatorname{logBB}$ values between -0.54 and +0.08 . In addition, all of the compounds have TPSA values of less than $76 \AA$ A comparison of the $\log \mathrm{P}$, TPSA, and $\log \mathrm{BB}$ values of $1.99,37.81$, and -0.12 , respectively, for bioisosteric analogues 7 a and 8 a to

Table 4. Calculated Physiochemical Properties of 5a, 5g, 7ae, 8a-e, Nicotine, Nat Epibatidine, and Varenicline

| compd | $\log \mathrm{P}^{a}$ | TPSA ${ }^{\text {a }}$ | $\operatorname{logBB}{ }^{b}$ |
| :---: | :---: | :---: | :---: |
| nicotine | 1.16 | 16.13 | 0.08 |
| epibatidine | 1.84 | 24.92 | 0.05 |
| varenicline | 1.01 | 37.81 | -0.27 |
| 5a | 3.14 | 70.74 | -0.43 |
| 5 g | 3.14 | 70.74 | -0.43 |
| 7 a | 1.99 | 37.81 | -0.12 |
| 7 b | 2.52 | 37.81 | -0.04 |
| 7 d | 1.75 | 63.83 | -0.54 |
| 7 c | 2.81 | 37.81 | 0.01 |
| 7 e | 2.12 | 37.81 | -0.10 |
| 8a | 1.99 | 37.81 | -0.12 |
| 8 b | 2.52 | 37.81 | -0.04 |
| 8 c | 2.81 | 37.81 | 0.01 |
| 8d | 1.75 | 63.83 | -0.54 |
| 8 e | 2.42 | 47.04 | -0.19 |
| ${ }^{a}$ ChemAxon Calculator Plugins, Marvin 6.1.0, 2013. ${ }^{b} \operatorname{logBB}=$ $-0.0148 \times$ TPSA $+0.152 \times \operatorname{clog} P+0.139$ (from ref 34 ). |  |  |  |

the corresponding values of $3.14,70.74$, and -0.43 for lead compounds $\mathbf{5 a}$ and $\mathbf{5 g}$ show that these two bioisosteric analogues ( $7 \mathbf{a}$ and 8a) have at least as favorable if not better calculated physicochemical properties than lead compounds 5 a and 5 g . In addition, both 7 a and 8 a have calculated $\log \mathrm{BB}$ values somewhat better than that of varenicline.

In summary, $2^{\prime}$-fluoro-3-(substituted pyridine)epibatidine analogues $7 \mathbf{a}-\mathbf{e}$ and $\mathbf{8 a}-\mathbf{e}$ were synthesized and evaluated for the ability to inhibit $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding to nAChR, tested for agonist and antagonist activity at $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$-, and $\alpha 7$ nAChR in an electrophysiology assay, and evaluated for agonist effects in the tail-flick, hot-plate, spontaneous-activity, and hypothermia tests in the mouse and as antagonists of nicotineinduced antinociception in the tail-flick and hot-plate tests in the mouse. A comparison of the $n A C h R$ binding and electrophysiology of bioisosteres $7 \mathbf{a}$ and $8 \mathbf{a}$ to those of the nitrophenyl lead compounds $\mathbf{5 a}$ and $\mathbf{5 g}$, respectively, showed that 7 a and 8 a had in vitro nAChR properties similar to those of 5 a and 5 g but were more selective for the $\alpha 4 \beta 2$-nAChR relative to the $\alpha 3 \beta 4$ - and $\alpha 7$-nAChRs than $\mathbf{5 a}$ and $\mathbf{5 g}$. Similar to 5a, 7a did not have agonist activity in the tail-flick and hot-plate tests and like 5a was a potent antagonist of nicotine-induced antinociception in these two tests. Thus, 7a is a very good bioisosteric analogue of 5 a . In contrast, 8 a unlike 5 g was an agonist in both the tail-flick and hot-plate tests and was an antagonist of nicotine-induced antinociception only in the tailflick test, whereas 5 g was an antagonist in both tests. Even though 8a is not a good bioisosteric analogue of $\mathbf{5 g}$ in the mice test, it is an interesting partial agonist. A comparison of the $\mathrm{AD}_{50}$ value of 7 a to that of varenicline in the hot-plate test, strongly suggest that this compound penetrates the brain in mice. Calculated $\log B B$ values for $7 a$ and $8 a$ also suggest that this compound will have good blood-brain barrier penetration. Since both nAChR antagonists and partial agonists are of interest as possible pharmacotherapies for treating smokers, both 7 a and 8a are candidates for development as pharmacotherapies to treat nicotine addiction.

## EXPERIMENTAL SECTION

Melting points were determined on a Mel-temp (Laboratory Devices, Inc.) capillary tube apparatus. NMR spectra were recorded on a Bruker

Avance 300 or AMX 500 spectrometer using tetramethylsilane as internal standard. Mass spectra were determined on a Perkin-Elmer Sciex API 150EX mass spectrometer outfitted with APCI and ESI sources. Melting point was determined on a Laboratory Devices MELTEMP II. Elemental analyses were carried out by Atlantic Microlab, Inc., Norcross GA. The purity of the compounds ( $>95 \%$ ) was established by elemental analysis. Analytical thin-layer chromatography (TLC) was carried out on plates precoated with silica gel ( $60 \mathrm{~F}_{254}$ ). TLC visualization was accomplished with a UV lamp or in an iodine chamber. Purifications by flash chromatography were performed on a Combiflash Teledyne ISCO instrument.

Suzuki Cross-Coupling Reaction: General Procedure (Method A). To a resealable reaction vessel under nitrogen was added 1.0 equiv of 7-tert-butoxycarbonyl-2-exo-( $2^{\prime}$-amino- $3^{\prime}$-bromo- 5 '-pyridin-yl)-7-azabicyclo[2.2.1] heptane $(9), \mathrm{Pd}(\mathrm{OAc})_{2}(0.1$ equiv $), \mathrm{P}(o \text {-tolyl })_{3}$ ( 0.2 equiv), sodium carbonate ( 2.0 equiv) and the respective pyridinyl boronic acid ( 1.6 equiv), DME ( 6 mL ), and water ( 0.7 mL ). The mixture was degassed through bubbling nitrogen, sealed, and heated on a sand bath at $80^{\circ} \mathrm{C}$ for 5 h . The mixture was cooled, poured into 20 mL of a saturated aqueous solution of $\mathrm{NaHCO}_{3}$, and extracted with $\mathrm{EtOAc}(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$ and filtered through Celite, and the solvent was removed under reduced pressure. The resultant residue was purified on silica gel by flash chromatography eluted with $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ ( $50: 1$ to $10: 1$ ).

Suzuki Cross-Coupling Reaction: General Procedure (Method B). To a resealable reaction vessel under nitrogen was added 1.0 equiv of the $3^{\prime}$-bromo compound $9, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(10 \mathrm{~mol} \%), \mathrm{K}_{2} \mathrm{CO}_{3}$ (2.0 equiv) and the respective pyridinyl boronic acid ( 1.3 equiv), toluene ( 12 mL ), ethanol ( 1.5 mL ), and water $(1.5 \mathrm{~mL})$. The mixture was degassed through bubbling nitrogen, sealed, and heated on a sand bath at $110^{\circ} \mathrm{C}$. After 24 h , the mixture was cooled, poured into 30 mL of $\mathrm{H}_{2} \mathrm{O}$, and extracted with EtOAc $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$ and filtered through Celite, and the solvent was removed in vacuo. The resultant residue was purified by flash chromatography on a silica gel column using hexanesisopropanol ( $80: 20$ to $25: 75$ ) or $\mathrm{CHCl}_{3}-\mathrm{MeOH}(30: 1$ to $10: 1)$ as the eluent.

General Procedure C: Removal of the Boc-Protecting Group. A solution of the Boc-protected compound in methylene chloride (5 mL ) was treated with TFA ( 1.5 mL ) and stirred at room temperature overnight. In some cases the solution was heated at $40^{\circ} \mathrm{C}$ for 2 h and then stirred at room temperature overnight. The solvent was then removed in vacuo, and the residue was treated with a solution of $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$ and extracted with $\mathrm{CHCl}_{3}-\mathrm{MeOH}(10 \%)(3 \times 30$ $\mathrm{mL})$. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo, and the residue was purified by flash chromatography using a silica gel column eluted with $\mathrm{CHCl}_{3}-\mathrm{MeOH}(10 \%)$ to provide the respective amine.

2-exo-[2'-Fluoro-3'-(pyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (7a) Fumarate. A solution of 12a (378 mg, 1.03 mmol, 1.0 equiv) in $70 \% \mathrm{HF}$ in pyridine $(1.5 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $806 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$ and extracted with EtOAc $(3 \times 100 \mathrm{~mL})$. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on a silica gel column using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $192 \mathrm{mg}(69 \%)$ of 7 a as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.50-1.78(\mathrm{~m}, 6 \mathrm{H})$, $2.01-2.08$ (dd, $J=9.0,11.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.02-3.07 (dd, $J=8.7,5.2 \mathrm{~Hz}$, $1 \mathrm{H}), 3.66(\mathrm{~s}, 1 \mathrm{H}), 3.77(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.70-7.73(\mathrm{~m}, 2 \mathrm{H}), 8.13(\mathrm{dd}, J=$ $2.4,9.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.18(\mathrm{~s}, 1 \mathrm{H}), 8.64(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=1.5$ $\mathrm{Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 30.0,31.8,41.1,45.7,57.9,63.7$, 121.2, 125.1, 141.3, 142.1, 144.2, 147.8, 148.0, 150.6, 158.6, 161.7; MS (ESI) $m / z 270.2(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $7 \mathbf{a}(302 \mathrm{mg}, 1.12 \mathrm{mmol})$ in chloroform ( 2 mL ) was placed in vial and treated with 1.1 equiv of fumaric acid $(0.65 \mathrm{M}$ in $\mathrm{MeOH})$. After 24 h , the white solid obtained was recrystallized from a $\mathrm{MeOH}-\mathrm{Et}_{2} \mathrm{O}$ mixture to provide the salt $7 \mathbf{a} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white solid:
mp 192-195 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.86-2.22(\mathrm{~m}$, $6 \mathrm{H}), 2.44-2.51(\mathrm{dd}, J=9.0,11.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.50-3.55(\mathrm{~m}, 1 \mathrm{H}), 4.35$ (br s, 1H), $4.56(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 7.72-7.75(\mathrm{~m}, 2 \mathrm{H})$, $8.20(\mathrm{dd}, J=2.4,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.27(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.67(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 27.0,29.0,37.7,43.4,60.2,64.1,121.6,125.1$, 136.2, 137.6, 141.3, 143.9, 147.8, 148.0, 150.7, 159.0, 162.2, 171.4; MS (ESI) $m / z 270.1$ [( $\mathrm{M}-$ fumaric $)^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{FN}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ ]. Anal. $\left(\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{FN}_{3} \mathrm{O}_{4} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(2-fluoropyridin-4-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (7b) Fumarate. A solution of 12b (230 $\mathrm{mg}, 0.60 \mathrm{mmol}, 1.0$ equiv) in $70 \% \mathrm{HF}$ in pyridine ( 1.5 mL ) was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $413 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$ and extracted with EtOAc ( $3 \times$ 100 mL ). The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on a silica gel column using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $121 \mathrm{mg}(70 \%)$ of $7 \mathbf{b}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.56-1.68(\mathrm{~m}$, $6 \mathrm{H}), 1.92-1.98(\mathrm{dd}, J=9.1,11.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.81-2.86(\mathrm{~m}, 1 \mathrm{H}), 3.60$ $(\mathrm{s}, 1 \mathrm{H}), 3.83(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.17(\mathrm{~d}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (ddd, $J=1.6$, $4.9,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.15-8.19(\mathrm{~m}, 2 \mathrm{H}), 8.23(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.4,31.5,40.7,44.2,56.4,62.9,109.0,119.4,121.1$, 139.6, 141.5, 147.5, 157.1, 160.3, 162.6, 162.7; MS (ESI) $m / z 288.3$ $(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $7 \mathbf{b}(141 \mathrm{mg}, 0.49 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in a vial was treated with 1.2 equiv of fumaric acid $(0.65 \mathrm{M})$ in MeOH , and the vial was allowed to stand in a refrigerator overnight. The excess solvent was then removed in vacuo from the salt and then redissolved in a minimal amount of MeOH , and the fumarate salt was recrystallized from MeOH using diethyl ether to provide $110 \mathrm{mg}(55 \%)$ of the salt $7 \mathbf{b} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white crystalline solid: mp 203-205 ${ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.87-2.20(\mathrm{~m}, 5 \mathrm{H}), 2.45-2.50(\mathrm{dd}, J=9.3$, $13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.50-3.53(\mathrm{~m}, 1 \mathrm{H}), 4.34-4.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.56(\mathrm{~d}, J=$ $3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.64(\mathrm{~s}, 2 \mathrm{H}), 7.41(\mathrm{~s}, 1 \mathrm{H}), 7.61-7.63(\mathrm{~m}, 1 \mathrm{H}), 8.21(\mathrm{dd}$, $J=2.4,9.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~d}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.32(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 25.8,27.8,36.5,42.2,59.0,62.8,109.4,121.6$, 135.0, 136.5, 140.1, 147.2, 147.8, 158.3, 160.2, 163.4, 165.3, 170.2; MS (ESI) $m / z 288.3$ [( $\mathrm{M}-$ fumaric $)^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~F}_{2} \mathrm{~N}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ ]. Anal. $\left(\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~F}_{2} \mathrm{~N}_{3} \mathrm{O}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(2-chloropyridin-4-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (7c) Fumarate. A solution of 12c (130 $\mathrm{mg}, 0.32 \mathrm{mmol}, 1.0$ equiv) in $70 \% \mathrm{HF}$ in pyridine ( 1.5 mL ) was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $224 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times$ 100 mL ). The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on a silica gel column using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $86 \mathrm{mg}(87 \%)$ of 7 c as a colorless oil. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.54-1.67(\mathrm{~m}$, $6 \mathrm{H}), 1.92-1.98(\mathrm{dd}, J=9.1,11.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.81-2.86(\mathrm{~m}, 1 \mathrm{H}), 3.60$ $(\mathrm{s}, 1 \mathrm{H}), 3.83(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.46(\mathrm{dd}, J=1.2,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~s}, 1 \mathrm{H})$, $8.12-8.15(\mathrm{~m}, 2 \mathrm{H}), 8.47(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ 30.4, 31.6, 40.7, 44.3, 56.4, 62.9, 119.2, 122.1, 139.6, 141.5, 145.1, 147.2, 149.9, 152.1, 157.1, 160.3; MS (ESI) $m / z 304.3(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $7 \mathrm{c}(106 \mathrm{mg}, 0.35 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in a vial was treated with 1.2 equiv of fumaric acid $(0.65 \mathrm{M})$ in MeOH , and the vial was allowed to stand in a refrigerator overnight. The excess solvent was then removed in vacuo from the salt and then redissolved in a minimal amount of MeOH , and the fumarate salt was recrystallized from MeOH using diethyl ether to provide $62 \mathrm{mg}(42 \%)$ of the salt 7 c $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white crystalline solid: mp $193-194{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $(500$ $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.87-2.21(\mathrm{~m}, 5 \mathrm{H}), 2.45-2.50(\mathrm{dd}, J=9.2,13.2 \mathrm{~Hz}$, $1 \mathrm{H}), 3.50-3.53(\mathrm{~m}, 1 \mathrm{H}), 4.34-4.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.56(\mathrm{~d}, J=3.9 \mathrm{~Hz}$, $1 \mathrm{H}), 6.63(\mathrm{~s}, 2 \mathrm{H}), 7.67(\mathrm{dd}, J=1.4,9.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{~s}, 1 \mathrm{H}), 8.21$ (dd, $J=2.4,9.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.48(\mathrm{~d}, J=4.9 \mathrm{~Hz}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 25.7,27.8,36.5,42.2,59.0,62.9,119.4$,
122.7, 135.0, 136.6, 140.1, 145.4, 147.3, 149.9, 151.8, 158.3, 160.3, 170.1; MS (ESI) $m / z 304.0$ [(M - fumaric $)^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClFN}_{3}$. $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$. Anal. $\left(\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{ClFN}_{3} \mathrm{O}_{4} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-Fluoro-3-(2'-amino-4'-pyridinyl)deschloroepibatidine (7d) Hydrochloride. A solution of 2-amino-4-bromopyridine ( 200 mg , $1.16 \mathrm{mmol}, 1.0$ equiv), bispinacolato diborane $(307 \mathrm{mg}, 1.21 \mathrm{mmol}$, 1.05 equiv), $\mathrm{Pd}_{2} \mathrm{dba}_{3}(36 \mathrm{mg}, 0.035 \mathrm{mmol}, 3 \mathrm{~mol} \%)$, Xphos ( 88 mg , $0.185 \mathrm{mmol}, 16 \mathrm{~mol} \%$ ), and KOAc ( $272 \mathrm{mg}, 2.77 \mathrm{mmol}, 2.4 \mathrm{mmol}$ ) in dioxane placed in a resealable pressure vessel was degassed through bubbling nitrogen for 40 min then heated at $110^{\circ} \mathrm{C}$ for 4 h . A TLC check revealed that all the bromopyridine had been converted to the boronic ester. The reaction was allowed to cool to room temperature, and $\mathrm{K}_{3} \mathrm{PO}_{4}$ ( $613 \mathrm{mg}, 2.89 \mathrm{mmol}, 2.5$ equiv), a solution of $10(270 \mathrm{mg}$, $1.0 \mathrm{mmol}, 0.87$ equiv) in dioxanes, an additional $3 \mathrm{~mol} \%$ of $\mathrm{Pd}_{2} \mathrm{dba}_{3}$, and $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})$ were added to the reaction. The mixture was degassed for 30 min and heated for 18 h at $110^{\circ} \mathrm{C}$. The reaction was cooled to room temperature and extracted with EtOAC $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$ and filtered through Celite, and the solvent was removed in vacuo. Two purifications of the residue by flash chromatography through an ISCO column using $\mathrm{CHCl}_{3}-\mathrm{MeOH}(10: 1)$ as the eluent provided 60 $\mathrm{mg}(21 \%)$ of $7 \mathbf{d}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $1.51-1.71(\mathrm{~m}, 5 \mathrm{H}), 1.90-1.97(\mathrm{~m}, 1 \mathrm{H}), 2.36(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 2.80-2.85$ (dd, $J=3.8,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.61(\mathrm{~s}, 1 \mathrm{H}), 3.81(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.66$ (br s, 2H), $6.72(\mathrm{~s}, 1 \mathrm{H}), 6.84(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{dd}, J=2.3,9.5$ $\mathrm{Hz}, 1 \mathrm{H}), 8.11(\mathrm{~s}, 1 \mathrm{H}), 8.13(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ 30.2, 31.4, 40.5, 44.3, 56.5, 62.9, 108.1, 113.9, 139.4, 140.6, 143.7, 145.9, 148.5, 157.4, 158.8, 160.6; MS (ESI) $\mathrm{m} / z 285.5(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $7 \mathbf{d}(122 \mathrm{mg}, 0.43 \mathrm{mmol})$ in chloroform in a vial was treated with a 2.0 equiv solution of HCl in diethyl ether and allowed to stand at room temperature. The excess solvent was filtered off, and the obtained salt was washed with ether and then dried to provide 94 mg of the salt $7 \mathrm{~d} \cdot \mathrm{HCl}$ as a white solid: mp $205-208{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.83-2.28(\mathrm{~m}, 5 \mathrm{H}), 2.46-2.53(\mathrm{dd}, J=3.8,9.6 \mathrm{~Hz}$, $1 \mathrm{H}), 3.52-3.57(\mathrm{dd}, J=3.1,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.37(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.59(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.02-7.05(\mathrm{dd}, J=1.6,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~s}$, 1H) $7.98(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}) 8.16(\mathrm{dd}, J=2.3,9.2 \mathrm{~Hz}, 1 \mathrm{H}) 8.28(\mathrm{~s}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{CNMR}\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 26.8,28.9,37.5,43.3,60.5,64.2,112.0$, 113.8, 137.4, 141.3, 143.2, 148.0, 148.2, 158.8, 158.9, 162.1; MS (ESI) $m / z 285.7\left[(\mathrm{M}-\mathrm{HCl})^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{FN}_{4} \cdot 2 \mathrm{HCl}\right]$. Anal. $\left(\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{Cl}_{2} \mathrm{FN}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(2-methoxypyridin-4-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (7e) Fumarate. To a resealable reaction pressure vessel under nitrogen was added compound $10(180 \mathrm{mg}, 0.66$ $\mathrm{mmol}, 1.0$ equiv), compound $16(188 \mathrm{mg}, 0.80 \mathrm{mmol}, 1.2$ equiv), $\left(\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(38 \mathrm{mg}, 0.03 \mathrm{mmol}, 5 \mathrm{~mol} \%), \mathrm{K}_{2} \mathrm{CO}_{3}(184 \mathrm{mg}, 1.33\right.$ mmol, 2.0 equiv), 1,4-dioxane ( 10 mL ), and water ( 0.80 mL ). The reaction mixture was degassed through bubbling nitrogen for 40 min , sealed, and heated over a sand bath at $110^{\circ} \mathrm{C}$ for 18 h . After cooling, the solvent was removed under reduced pressure, and to the residue was added 20 mL of $\mathrm{H}_{2} \mathrm{O}$. The organic product was extracted using $\mathrm{EtOAc}(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$ and filtered through Celite, and the solvent was removed in vacuo. Purification by flash chromatography on silica gel using $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ as the eluent provided $100 \mathrm{mg}(50 \%)$ of 7 e as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.51-1.68(\mathrm{~m}, 5 \mathrm{H})$, 1.89-1.96 (dd, 3.8, 9.6 Hz, 1H), 1.98 (broad signal 1H), 2.79-2.84 (dd, $J=3.4,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{~s}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 6.96$ $(\mathrm{s}, 1 \mathrm{H}), 7.07-7.10(\mathrm{dt}, J=5.3,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.06(\mathrm{dd}, J=2.4,9.6 \mathrm{~Hz}$, $1 \mathrm{H}), 8.11(\mathrm{~s}, 1 \mathrm{H}), 8.21(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 30.2$. 31.4, 40.5, 44.3, 53.5, 56.4, 62.9, 110.5, 116.6, 139.6, 140.8, 144.6, 146.2, 146.4, 147.2, 160.5, 164.6; MS (ESI) $m / z(300.4)(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $7 \mathbf{e}(156 \mathrm{mg}, 0.52 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in a vial was treated with 1.2 equiv of fumaric acid $(0.65 \mathrm{M})$ in MeOH , and the vial was allowed to stand in a refrigerator overnight. The excess solvent was then removed in vacuo from the salt that was then redissolved in a minimal amount of MeOH , and the fumarate salt was recrystallized from MeOH using diethyl ether to provide 164 mg (74\%) of the salt $7 \mathrm{e} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white solid: mp $160-164{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.85-2.19(\mathrm{~m}, 5 \mathrm{H}), 2.43-2.50(\mathrm{dd}, J=9.3,13.2 \mathrm{~Hz}, 1 \mathrm{H})$,
$3.48-3.53(\mathrm{~m}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 4.34(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.55(\mathrm{~s}, 1 \mathrm{H}), 6.65$ $(\mathrm{s}, 2 \mathrm{H}), 7.07(\mathrm{~s}, 1 \mathrm{H}), 7.22(\mathrm{dd}, J=1.2,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 8.12(\mathrm{~d}, J=9.2$ $\mathrm{Hz}, 1 \mathrm{H}), 8.22-8.23(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 26.9,29.0,37.7$, 43.4, 54.2, 60.2, 64.1, 111.6, 117.9, 136.1, 137.5, 141.2, 145.9, 147.5, 147.6, 148.3, 162.2, 166.2, 171.1; MS (ESI) $m / z 300.3$ [(M fumaric $)^{+}, \mathrm{M}=\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{FN}_{3} \mathrm{O} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ ]. Anal. $\left(\mathrm{C}_{21} \mathrm{H}_{22} \mathrm{FN}_{3} \mathrm{O}_{5}\right) \mathrm{C}, \mathrm{H}$, N.

2-exo-[2'-Fluoro-3'-(pyridin-3-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (8a) Hemifumarate. A solution of 11a ( 394 mg , $1.08 \mathrm{mmol}, 1.0$ equiv) in $70 \% \mathrm{HF}$ in pyridine $(1.5 \mathrm{~mL})$ was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $742 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times 40 \mathrm{~mL})$. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide 203 mg (70\%) of $\mathbf{8 a}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.49-1.79(\mathrm{~m}, 6 \mathrm{H}), 2.01-2.08(\mathrm{dd}, J=9.1$, $11.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.02-3.07(\mathrm{dd}, J=3.3,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.67(\mathrm{~s}, 1 \mathrm{H}), 3.77$ (br s, 1 H ), $7.54(\mathrm{dd}, J=2.6,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.08-8.15(\mathrm{~m}, 3 \mathrm{H}), 8.58-$ $8.60(\mathrm{~d}, 2 \mathrm{H}), 8.58(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.80(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 29.9,31.8,40.6,41.1,45.7,57.8,63.9,121.1,125.2,138.4$, 141.4, 142.0, 147.0, 150.1, 158.7, 161.8; MS (ESI) $m / z 270.3$ (M + H) ${ }^{+}$.

A solution of $8 \mathbf{a}(246 \mathrm{mg}, 0.91 \mathrm{mmol})$ in chloroform $(2 \mathrm{~mL})$ was placed in vial and treated with 1.1 equiv of fumaric acid $(0.65 \mathrm{M}$ in MeOH ). After 24 h , the white solid obtained was recrystallized from MeOH using $\mathrm{Et}_{2} \mathrm{O}$ to provide the salt $\mathbf{8 a} \cdot 0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white solid: mp 155-159 ${ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 1.86-2.22(\mathrm{~m}$, $6 \mathrm{H}), 2.44-2.51(\mathrm{dd}, J=9.0,11.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.49-3.54(\mathrm{dd}, J=3.0,5.1$ $\mathrm{Hz}, 1 \mathrm{H}), 4.35(\mathrm{br}$ s, 1 H$), 4.56(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 7.56-$ $7.60(\mathrm{dd}, J=2.3,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.12-8.16(\mathrm{~m}, 2 \mathrm{H}), 8.23(\mathrm{~s}, 1 \mathrm{H}), 8.61$ (dd, $J=1.4,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.81(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 27.4$, 29.4, 38.2, 43.8, 59.8, 64.0, 121.0, 125.4, 136.8, 138.2, 138.5, 141.4, 147.0, 147.2, 150.0, 159.1, 162.3, 171.5; MS (ESI) $m / z 270.2$ [(M fumaric $\left.)^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{FN}_{3} \cdot 0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}\right]$. Anal. $\left(\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{FN}_{3} \mathrm{O}_{2} \cdot 0.5\right.$ $\left.\mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(6-fluoropyridin-3-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (8b) Hemifumarate. Compound 11b ( $250 \mathrm{mg}, 0.65 \mathrm{mmol}, 1.0$ equiv) was placed in a plastic vessel and was treated dropwise with 1.5 mL of $70 \% \mathrm{HF}$ in pyridine, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $449 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times 40$ $\mathrm{mL})$. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $170 \mathrm{mg}(91 \%)$ of $\mathbf{8 b}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.54-1.70(\mathrm{~m}, 6 \mathrm{H})$, 1.92-1.99 (dd, $J=9.0,11.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.82-2.87$ (m, 1H), 3.61 ( s , $1 \mathrm{H}), 3.83(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.04(\mathrm{dd}, J=3.0,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.99-8.09(\mathrm{~m}$, $2 \mathrm{H}), 8.14(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.42(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ $30.3,31.5,40.6,44.3,56.4,62.9,109.3,118.5,139.5,141.3,145.8$, 147.5, 157.3, 160.4, 161.7, 164.9; MS (ESI) $\mathrm{m} / \mathrm{z} 288.3(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $\mathbf{8 b}(198 \mathrm{mg}, 0.69 \mathrm{mmol})$ in chloroform $(2 \mathrm{~mL})$ was placed in a vial and treated with 1.1 equiv of fumaric acid $(0.65 \mathrm{M}$ in $\mathrm{MeOH})$. After 24 h , the white solid obtained was recrystallized from MeOH using $\mathrm{Et}_{2} \mathrm{O}$ to provide $200 \mathrm{mg}(84 \%)$ of the salt $\mathbf{8 b} \cdot 0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white crystalline solid: mp 197-199 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.81-2.15(\mathrm{~m}, 5 \mathrm{H}), 2.38-2.43(\mathrm{dd}, J=9.3,13.2 \mathrm{~Hz}, 1 \mathrm{H})$, $3.42-3.46(\mathrm{~m}, 1 \mathrm{H}), 4.43(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.57(\mathrm{~s}, 1 \mathrm{H}), 7.21(\mathrm{dd}, J=2.4,8.3$ $\mathrm{Hz}, 1 \mathrm{H}), 8.14(\mathrm{dd}, J=2.4,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.21-8.25(\mathrm{~m}, 2 \mathrm{H}), 8.48(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 27.5,29.5,38.3,43.8,59.9,64.1,111.0$, 120.5, 137.0, 141.4, 143.8, 147.2, 148.8, 159.7, 161.6, 164.1, 166.0, 174.0; MS (ESI) $m / z 288.3$ [(M - fumaric $)^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~F}_{2} \mathrm{~N}_{3}$. $0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ ]. Anal. $\left(\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~F}_{2} \mathrm{~N}_{3} \mathrm{O}_{2}\right)$ C, $\mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(6-chloropyridin-3-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (8c) Fumarate. Compound 11c (300
$\mathrm{mg}, 0.75 \mathrm{mmol}, 1.0$ equiv) was placed in a plastic vessel and was treated dropwise with 3 mL of $70 \% \mathrm{HF}$ in pyridine, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $559 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and extracted with EtOAc $(3 \times 40$ $\mathrm{mL})$. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $142 \mathrm{mg}(62 \%)$ of 8 c as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.54-1.71(\mathrm{~m}, 6 \mathrm{H})$, $1.92-1.98(\mathrm{dd}, J=9.1,11.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.81-2.86(\mathrm{~m}, 1 \mathrm{H}), 3.61(\mathrm{~s}$, $1 \mathrm{H}), 3.81(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.42(\mathrm{dd}, J=0.6,8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{ddd}, J=0.8$, $4.1,8.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 8.06 (dd, $J=2.4,9.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.15$ (br s, 1H), 8.58 (br s, 1H); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.4,31.5,40.6,44.3,56.4,62.9$, 118.5, 124.1, 129.2, 139.5, 141.3, 146.1, 149.2, 151.2, 157.3, 160.5; MS (ESI) $m / z 304.3(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $8 \mathrm{c}(138 \mathrm{mg}, 0.46 \mathrm{mmol})$ in chloroform $(2 \mathrm{~mL})$ was placed in a vial and treated with 1.1 equiv of fumaric acid $(0.65 \mathrm{M}$ in MeOH ). After 24 h , the white solid obtained was recrystallized from MeOH using $\mathrm{Et}_{2} \mathrm{O}$ to provide 105 mg (55\%) of the salt of 8 c $0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ as a white crystalline solid: $\mathrm{mp} 194-195{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.89-2.20(\mathrm{~m}, 5 \mathrm{H}), 2.45-2.49(\mathrm{dd}, J=9.2$, $13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.49-3.52(\mathrm{dd}, J=3.5,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.34(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $4.56(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~s}, 2 \mathrm{H}), 7.60(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.09-$ $8.15(\mathrm{~m}, 2 \mathrm{H}), 8.23(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.64(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 27.1,29.1,37.8,43.5,60.3,64.2,120.4,125.8,130.6$, $136.3,137.8,141.3,147.4,150.6,152.6,159.8,161.7,171.5$; MS (ESI) $m / z 304.5\left[(\mathrm{M}-\text { fumaric })^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClFN}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}\right]$. Anal. $\left(\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{ClFN}_{3} \mathrm{O}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2'-Fluoro-3'-(2"-amino-5"-pyridinyl)deschloroepibatidine (8d) Hydrochloride. To a resealable reaction pressure vessel under nitrogen was added 2-exo-( 2 '-fluoro- $3^{\prime}$-bromo)-7-azabicyclo[2.2.1]heptane ( 10 ) $\left(125 \mathrm{mg}, 0.46 \mathrm{mmol}, 1.0\right.$ equiv), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(27 \mathrm{mg}, 5$ $\mathrm{mol} \%), \mathrm{K}_{2} \mathrm{CO}_{3}$ ( $128 \mathrm{mg}, 0.92 \mathrm{mmol}, 2.0$ equiv), 1,4-dioxane ( 10 mL ), water ( 0.80 mL ) , and 2-aminopyridine-5-pinacolate boronic ester (122 $\mathrm{mg}, 0.55 \mathrm{mmol}, 1.2$ equiv). The mixture was degassed through bubbling nitrogen for 40 min and heated at $110^{\circ} \mathrm{C}$ for 18 h . After cooling, the solvent was removed under reduced pressure, and to the residue was added 20 mL of $\mathrm{H}_{2} \mathrm{O}$. The organic product was extracted using EtOAc $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered through Celite, and the solvent was removed in vacuo. Purification by flash chromatography on silica gel using $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ as the eluent provided $88 \mathrm{mg}(67 \%)$ of the desired product 8 d as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.47-$ $1.67(\mathrm{~m}, 5 \mathrm{H}), 1.85-1.92(\mathrm{~m}, 2 \mathrm{H}), 2.76-2.80(\mathrm{dd}, J=3.8,5.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.56(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~s}, 2 \mathrm{H}), 6.53(\mathrm{~d}, J=$ $8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{dt}, J=1.9,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{dd}, J=2.3,9.5 \mathrm{~Hz}$, $1 \mathrm{H}), 7.98(\mathrm{~s}, 1 \mathrm{H}), 8.23(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{CNMR}\left(\mathrm{CDCl}_{3}\right) \delta 30.2,31.4,40.5$, 44.5, 56.4, 62.8, 108.2, 120.2, 138.0, 138.7, 140.7, 144.2, 147.9, 157.5, 158.3, 160.6; MS (ESI) $m / z 285.7(\mathrm{M}+\mathrm{H})^{+}$.

A solution of the diamine $8 \mathbf{d}(217 \mathrm{mg}, 0.76 \mathrm{mmol})$ in chloroform in a vial was treated with a 2.0 equiv solution of HCl in diethyl ether and allowed to stand at room temperature. The excess solvent was filtered off, and the obtained salt washed with ether and then dried to provide $246 \mathrm{mg}(90 \%)$ of $\mathbf{8 d} \cdot \mathrm{HCl}$ as a white solid: $\mathrm{mp} 202-205{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.88-2.24(\mathrm{~m}, 5 \mathrm{H}), 2.44-2.52(\mathrm{dd}, J=3.8,9.6$ $\mathrm{Hz}, 1 \mathrm{H}), 3.51-3.56(\mathrm{dd}, J=3.1,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.37(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H})$, 4.58 (d, $J=2.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.11 (dd, $J=1.9,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.18-8.28$ (m, $4 \mathrm{H}) ;{ }^{13} \mathrm{CNMR}\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 26.8,28.9,37.6,43.3,60.5,64.4,114.7$, 119.3, 120.4, 137.6, 140.6, 145.1, 147.2, 155.8, 158.9, 162.1; MS (ESI) $m / z 285.6\left[(\mathrm{M}-\mathrm{HCl})^{+}, \mathrm{M}=\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{FN}_{4} \cdot 2 \mathrm{HCl}\right]$. Anal. $\left(\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{Cl}_{2} \mathrm{FN}_{4} \cdot 1.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-exo-[2'-Fluoro-3'-(6-methoxypyridin-3-yl)-5'-pyridinyl]-7azabicyclo[2.2.1]heptane (8e) Hemifumarate. Compound 11d ( $480 \mathrm{mg}, 1.21 \mathrm{mmol}, 1.0$ equiv) was placed in a plastic vessel and was treated dropwise with 3 mL of $70 \% \mathrm{HF}$ in pyridine, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min . Sodium nitrite ( $835 \mathrm{mg}, 10$ equiv) was added in small portions, and the reaction mixture was stirred at room temperature. After 1 h , the mixture was poured into a $1: 1$ aqueous
solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$ and extracted with EtOAc $(3 \times$ 100 mL ). The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluent to provide $227 \mathrm{mg}(94 \%)$ of $\mathbf{8 e}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.48-1.76(\mathrm{~m}, 6 \mathrm{H})$, $1.94-2.05(\mathrm{~m}, 2 \mathrm{H}), 2.96-3.01(\mathrm{dd}, J=3.4,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H})$, $3.77(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.83(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{tt}, J=0.7,1.7,8.7 \mathrm{~Hz}$, $1 \mathrm{H}), 7.99(\mathrm{dd}, J=2.4,9.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.34(\mathrm{~d}, J$ $=1.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta 29.9,31.7,40.9,45.7,54.3,57.7$, 63.7, 111.7, 121.3, 124.5, 140.8, 141.6, 145.8, 147.9, 158.6, 161.8, 165.5; MS (ESI) $m / z 300.3(\mathrm{M}+\mathrm{H})^{+}$.

A solution of $8 \mathrm{e}(169 \mathrm{mg}, 0.53 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in a vial was treated with a 1.2 equiv of fumaric acid $(0.65 \mathrm{M})$ in MeOH , and the vial was allowed to stand in a refrigerator overnight. The excess solvent was removed in vacuo from the salt that was then redissolved in a minimal amount of MeOH , and the fumarate salt was recrystallized from MeOH using diethyl ether to provide 159 mg of the salt $8 \mathbf{8}$ $0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}: \mathrm{mp} 193-195{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 300 MHz , methanol- $d_{4}$ ) $\delta$ $1.80-2.15(\mathrm{~m}, 6 \mathrm{H}), 2.36-2.43(\mathrm{dd}, J=9.3,13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.40-3.45$ $(\mathrm{m}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 4.27(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.42(\mathrm{~s}, 1 \mathrm{H}), 6.61(\mathrm{~s}, 1 \mathrm{H}), 6.91$ (dd, $J=0.7,7.6 \mathrm{~Hz} \mathrm{1H}$ ), $7.95(\mathrm{dt}, J=0.8,2.4,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.06$ (dd, $J$ $=1.9,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.41(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (methanol- $d_{4}$ ) $\delta 26.9,29.0,37.7,43.4,54.3,60.2,64.1,111.7,124.2$, 136.2, 137.4, 140.6, 140.8, 145.8, 148.0, 159.1, 162.3, 165.8, 171.3; MS (ESI) $m / z 300.5\left[(\mathrm{M}-\text { fumaric })^{+}, \mathrm{M}=\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{FN}_{3} \mathrm{O} \cdot 0.5 \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}\right]$. Anal. $\left(\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{FN}_{3} \mathrm{O}_{3} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(pyridin-3-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (11a). A solution of compound 9 ( $322 \mathrm{mg}, 0.87 \mathrm{mmol}, 1.0$ equiv), pyridine-3-boronic acid ( $140 \mathrm{mg}, 1.14 \mathrm{mmol}, 1.3$ equiv), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(50 \mathrm{mg}, 0.044 \mathrm{mmol}$, $5 \mathrm{~mol} \%)$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(242 \mathrm{mg}, 1.75 \mathrm{mmol}, 2.0$ equiv) in toluene ( 10 $\mathrm{mL}), \mathrm{EtOH}(2 \mathrm{~mL})$, and water $(2 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The mixture was sealed and heated over a sand bath at $110^{\circ} \mathrm{C}$ for 22 h . After cooling to room temperature, 20 mL of $\mathrm{H}_{2} \mathrm{O}$ was added, and the organic product was extracted with EtOAc (3 $\times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $i$ PrOH -hexanes as the eluent to provide $263 \mathrm{mg}(82 \%)$ of $11 \mathrm{a} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.41$ (br s, 9 H$), 1.48-1.61(\mathrm{~m}, 2 \mathrm{H})$, $1.75-1.86(\mathrm{~m}, 3 \mathrm{H}), 1.96-2.04(\mathrm{~m}, 1 \mathrm{H}), 2.79-2.83(\mathrm{dd}, J=3.8,5.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 4.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.66(\mathrm{~s}, 2 \mathrm{NH}), 7.34(\mathrm{~d}, J=2.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.38(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{dt}, J=7.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.96$ $(\mathrm{d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.59(\mathrm{dd}, J=4.9,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.69(\mathrm{~d}, J=1.8 \mathrm{~Hz}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.3$ (3 C), 28.8, 29.7, 40.3, 44.9, 55.9, 62.2, 79.5, 118.0, 123.6, 132.1, 134.1, 136.2, 136.9, 146.6, 148.9, 149.7, 154.5, 154.9; MS (ESI) $m / z 367.6(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(6-fluoropyridin-3-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (11b). A solution of compound 9 ( $241 \mathrm{mg}, 0.65 \mathrm{mmol}, 1.0$ equiv), 5 -fluoropyridine-4boronic acid ( $148 \mathrm{mg}, 1.05 \mathrm{mmol}, 1.6$ equiv), $\mathrm{Pd}(\mathrm{OAc})_{2}(15 \mathrm{mg}, 0.065$ mmol, $10 \mathrm{~mol} \%), \mathrm{P}(o \text {-tolyl })_{3}(40 \mathrm{mg}, 0.131 \mathrm{mmol}, 20 \mathrm{~mol} \%)$, and $\mathrm{Na}_{2} \mathrm{CO}_{3}(139 \mathrm{mg}, 1.31 \mathrm{mmol}, 2.0$ equiv) in DME $(8 \mathrm{~mL})$ and water $(0.9 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The mixture was sealed and heated over a sand bath at $80^{\circ} \mathrm{C}$ for 5 h . After cooling to room temperature, the mixture was poured into a saturated aqueous solution of $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times$ 30 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using EtOAc-hexanes as the eluent to provide $250 \mathrm{mg}(99 \%)$ of $\mathbf{1 1 b}$. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 1.39(\mathrm{br} \mathrm{s}, 9 \mathrm{H}), 1.51-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.81-1.85(\mathrm{~m}, 3 \mathrm{H})$, 1.94-2.00 (m, 1H), 2.79-2.84 (m, 1H), $4.16(\mathrm{~s}, 1 \mathrm{H}), 4.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $4.70(\mathrm{~s}, 2 \mathrm{NH}), 7.02(\mathrm{dd}, J=2.9,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{~d}, J=2.25 \mathrm{~Hz}$, $1 \mathrm{H}), 7.91$ (ddd, $J=2.5,8.4,16 \mathrm{~Hz}, 1 \mathrm{H}), 7.96(\mathrm{~d}, J=2.25 \mathrm{~Hz}, 1 \mathrm{H})$, $8.28(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.2(3 \mathrm{C}), 28.8,29.7$, 40.3, 44.8, 55.9, 62.1, 79.5, 109.5, 116.8, 132.0, 136.9, 141.5, 146.8, 147.5, 154.6, 154.9, 161.3, 164.5; MS (ESI) $m / z 385.5(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(6-chloropyridin-3-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (11c). A solution of compound 9 ( $304 \mathrm{mg}, 0.83 \mathrm{mmol}, 1.0$ equiv), 5 -chloropyridine-4boronic acid ( $208 \mathrm{mg}, 1.32 \mathrm{mmol}, 1.6$ equiv $), \mathrm{Pd}(\mathrm{OAc})_{2}(19 \mathrm{mg}, 0.083$ $\mathrm{mmol}, 10 \mathrm{~mol} \%), \mathrm{P}(o \text {-tolyl })_{3}(51 \mathrm{mg}, 0.166 \mathrm{mmol}, 20 \mathrm{~mol} \%)$, and $\mathrm{Na}_{2} \mathrm{CO}_{3}(176 \mathrm{mg}, 1.66 \mathrm{mmol}, 2.0$ equiv) in DME $(6 \mathrm{~mL})$ and water $(0.7 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The mixture was sealed and heated over a sand bath at $80^{\circ} \mathrm{C}$ for 5 h . After cooling to room temperature, the mixture was poured into a saturated aqueous solution of $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times$ 30 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using EtOAc-hexanes as the eluent to provide 305 mg (99\%) of 11 c as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.31(\mathrm{br} \mathrm{s}, 9 \mathrm{H}), 1.43-1.50(\mathrm{~m}, 2 \mathrm{H}), 1.72-1.76$ $(\mathrm{m}, 3 \mathrm{H}), 1.85-1.92(\mathrm{~m}, 1 \mathrm{H}), 2.70-2.74(\mathrm{~m}, 1 \mathrm{H}), 4.06(\mathrm{~s}, 1 \mathrm{H}), 4.26$ (br s, 1H), $4.60(\mathrm{~s}, 2 \mathrm{NH}), 7.25(\mathrm{~d}, J=2.25 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.71(\mathrm{dd}, J=2.5,8.21 \mathrm{H}), 7.88(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.38(\mathrm{~d}, J$ $=2.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.3$ (3 C), 28.8, 29.7, 40.3, 44.8, 55.9, 62.1, 79.5, 116.7, 124.2, 132.2, 133.1, 136.9, 139.0, 147.0, 149.5, 150.6, 154.4, 155.0; MS (ESI) $m / z 401.5(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(6-methoxypyri-din-3-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (11d). A solution of 9 ( $337 \mathrm{mg}, 0.92 \mathrm{mmol}, 1.0$ equiv), 2-methoxypyridine-5boronic acid ( $182 \mathrm{mg}, 1.2 \mathrm{mmol}, 1.3$ equiv), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(53 \mathrm{mg}, 0.046$ $\mathrm{mmol}, 5 \mathrm{~mol} \%)$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(253 \mathrm{mg}, 1.83 \mathrm{mmol}, 2.0$ equiv) in toluene $(12 \mathrm{~mL}), \mathrm{EtOH}(2 \mathrm{~mL})$, and $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$ was placed in a resealable pressure vessel and degassed through bubbling nitrogen for 20 min . The vessel was sealed and placed on a sand bath that was heated at $110{ }^{\circ} \mathrm{C}$ overnight. After cooling to room temperature, $\mathrm{H}_{2} \mathrm{O}$ $(20 \mathrm{~mL})$ was added, and the organic product was extracted with EtOAc $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The residue was purified by flash chromatography on silica gel using EtOAc-hexanes as the eluent to furnish compound 11 d ( 310 mg , $92 \%)$ as a colorless oil. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.39(\mathrm{br} \mathrm{s}, 9 \mathrm{H})$, $1.50-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.76-1.88(\mathrm{~m}, 3 \mathrm{H}), 1.91-1.98(\mathrm{~m}, 1 \mathrm{H}), 2.77-$ $2.81(\mathrm{dd}, J=3.7,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 4.34(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}), 4.78$ (s, 2 NH ), 6.79 (d, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.65(\mathrm{dd}, J=8.4,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.93(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}) 8.22(\mathrm{~d}, J=2.3$ $\mathrm{Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.1$ (3 C), 28.4, 28.6, 40.1, 44.8, 53.3, 55.3, 62.1, 79.3, 110.8, 118.1, 126.9, 128.4, 131.8, 136.7, 138.9, 145.8, 146.6, 154.8, 163.5; MS (ESI) $m / z 397.5(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(pyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (12a). A solution of compound 9 ( $354 \mathrm{mg}, 00.96 \mathrm{mmol}, 1.0$ equiv), pyridine-4-boronic acid ( 154 mg , $1.25 \mathrm{mmol}, 1.3$ equiv), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(56 \mathrm{mg}, 0.048 \mathrm{mmol}$, $5 \mathrm{~mol} \%)$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(266 \mathrm{mg}, 1.92 \mathrm{mmol}, 2.0$ equiv) in toluene ( 10 $\mathrm{mL}), \mathrm{EtOH}(2 \mathrm{~mL})$, and water $(2 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The mixture was sealed and heated over a sand bath at $110^{\circ} \mathrm{C}$ for 22 h . After cooling to room temperature, 20 mL of $\mathrm{H}_{2} \mathrm{O}$ was added, and the organic product was extracted with EtOAc (3 $\times 30 \mathrm{~mL}$ ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using $i$ -$\mathrm{PrOH}-$ hexanes as the eluent to provide 340 mg ( $97 \%$ ) of $\mathbf{1 2 a} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.39$ (br s, 9 H ), $1.44-1.59(\mathrm{~m}, 2 \mathrm{H})$, $1.81-1.84(\mathrm{~m}, 3 \mathrm{H}), 1.93-2.00(\mathrm{~m}, 1 \mathrm{H}), 2.79-2.84(\mathrm{dd}, J=3.8,5.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 4.36(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.67(\mathrm{~s}, 2 \mathrm{NH}), 7.39-7.43(\mathrm{~m}$, $3 \mathrm{H}), 7.99(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.66(\mathrm{dd}, J=6.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.3$ (3 C), 28.8, 29.7, 40.4, 44.8, 55.8, 62.1, 79.5, 118.7, 123.4 (2 C), 132.2, 136.5, 146.4, 147.2, 150.5 (2 C), 153.9, 154.9; MS (ESI) $m / z 367.6(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(2-fluoropyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (12b). A solution of compound 9 ( $319 \mathrm{mg}, 0.87 \mathrm{mmol}, 1.0$ equiv), 2-fluoropyridine-4boronic acid ( $196 \mathrm{mg}, 1.39 \mathrm{mmol}, 1.6$ equiv $), \mathrm{Pd}(\mathrm{OAc})_{2}(20 \mathrm{mg}, 0.087$ $\mathrm{mmol}, 10 \mathrm{~mol} \%), \mathrm{P}(o \text {-tolyl })_{3}(53 \mathrm{mg}, 0.173 \mathrm{mmol}, 20 \mathrm{~mol} \%)$, and $\mathrm{Na}_{2} \mathrm{CO}_{3}(184 \mathrm{mg}, 1.73 \mathrm{mmol}$, 2.0 equiv) in DME $(6 \mathrm{~mL})$ and water $(0.7 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The
mixture was sealed and heated over a sand bath at $80^{\circ} \mathrm{C}$ for 5 h . After cooling to room temperature, the mixture was poured into a saturated aqueous solution of $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times$ 30 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using EtOAc-hexanes as the eluent to provide 300 mg ( $92 \%$ ) of $\mathbf{1 2 b}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.4(\mathrm{br} \mathrm{s}, 9 \mathrm{H}), 1.52-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.82-1.84$ $(\mathrm{m}, 3 \mathrm{H}), 1.94-1.98(\mathrm{~m}, 1 \mathrm{H}), 2.79-2.84(\mathrm{~m}, 1 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 4.36$ (br s, 1H), $4.77(\mathrm{~s}, 2 \mathrm{NH}), 7.06(\mathrm{~s}, 1 \mathrm{H}), 7.34$ (ddd, $J=1.6,5.13,8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 7.41(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.0(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=$ $5.16 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.3$ (3 C), 28.8, 29.7, 40.5, 44.8, 55.9, 62.1, 79.7, 108.8, 121.1, 132.5, 136.5, 147.8, 148.3, 152.0, 153.7, 155.0, 162.8, 166.0. MS (ESI) $m / z 385.3(\mathrm{M}+\mathrm{H})^{+}$.

7-tert-Butoxycarbonyl-2-exo-[2'-amino-3'-(2-chloropyridin-4-yl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (12c). A solution of compound 9 ( $192 \mathrm{mg}, 0.52 \mathrm{mmol}, 1.0$ equiv), 2-chloropyridine-4boronic acid ( $131 \mathrm{mg}, 0.83 \mathrm{mmol}, 1.6$ equiv), $\mathrm{Pd}(\mathrm{OAc})_{2}(12 \mathrm{mg}, 0.052$ mmol, $10 \mathrm{~mol} \%), \mathrm{P}(o \text {-tolyl })_{3}(32 \mathrm{mg}, 0.104 \mathrm{mmol}, 20 \mathrm{~mol} \%)$, and $\mathrm{Na}_{2} \mathrm{CO}_{3}(111 \mathrm{mg}, 1.04 \mathrm{mmol}, 2.0$ equiv) in DME ( 6 mL ) and water $(0.7 \mathrm{~mL})$ was degassed through bubbling nitrogen for 20 min . The mixture was sealed and heated over a sand bath at $80^{\circ} \mathrm{C}$ for 5 h . After cooling to room temperature, the mixture was poured into a saturated aqueous solution of $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and extracted with $\mathrm{EtOAc}(3 \times$ 30 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered through Celite, and concentrated in vacuo. The resultant residue was purified by flash chromatography on silica gel using EtOAc-hexanes as the eluent to provide $112 \mathrm{mg}(54 \%)$ of $\mathbf{1 2 c}$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.41(\mathrm{br} \mathrm{s}, 9 \mathrm{H}), 1.49-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.77-1.83$ $(\mathrm{m}, 3 \mathrm{H}), 1.94-2.00(\mathrm{~m}, 1 \mathrm{H}), 2.78-2.83(\mathrm{~m}, 1 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 4.36$ (br s, 1 H$), 4.54(\mathrm{~s}, 2 \mathrm{NH}), 7.37(\mathrm{dd}, J=1.4,5.13 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=$ $2.22 \mathrm{~Hz}, 1 \mathrm{H}), 7.45(\mathrm{~s}, 1 \mathrm{H}), 8.0(\mathrm{~d}, J=2.22 \mathrm{~Hz}, 1 \mathrm{H}), 8.44(\mathrm{~d}, J=5.10$ $\mathrm{Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.3$ (3 C), 28.8, 29.7, 40.5, 44.8, 55.9, 62.1, 79.6, 117.4, 122.0, 123.8, 132.4, 136.5, 147.8, 149.6, 150.2, 152.4, 153.8, 154.9; MS (ESI) $m / z 401.3(\mathrm{M}+\mathrm{H})^{+}$.

Preparation of 2-Methoxypyidine-4-boronic Acid Pinacol Ester (15). A solution of 4-bromo-2-methoxypyridine (14) (462 mg, $2.46 \mathrm{mmol}, 1.0$ equiv), bis(pinacolato) diboron ( $749 \mathrm{mg}, 2.95 \mathrm{mmol}$, 1.2 equiv), $\mathrm{PdCl}_{2}$ (dppf) $(54 \mathrm{mg}, 0.074 \mathrm{mmol}, 3 \mathrm{~mol} \%)$, and KOAc ( $724 \mathrm{mg}, 7.37 \mathrm{mmol}, 3.0$ equiv) in DMF ( 6 mL ) in a resealable pressure vessel was degassed through bubbling nitrogen for 20 min . The reaction mixture was sealed and heated over a sand bath at $85^{\circ} \mathrm{C}$ overnight. After cooling to room temperature, the mixture was diluted with EtOAc and filtered through a plug of Celite and anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was removed in vacuo, and the residue was purified by flash chromatography on silica gel using $\mathrm{EtOAc}-\mathrm{MeOH}$ as the eluent to provide $427.4 \mathrm{mg}(74 \%)$ of 15 as a brownish oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.34(\mathrm{~s}, 12 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H})$, $7.18(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.18(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H})$.
$\left[{ }^{3} \mathrm{H}\right]$ Epibatidine Binding Assay. The inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding at rat brain $\alpha 4 \beta 2^{*}$-nAChRs was conducted as previously reported. ${ }^{21}$

Electrophysiology. The electrophysiology assays with rat $\alpha 4 \beta 2$-, $\alpha 3 \beta 4$, and $\alpha 7$-nAChRs were conducted as previously described. ${ }^{24}$

In Vivo Test. The antinociception (tail-flick and hot-plate), locomotor, and body temperature tests were all conducted in mice as previously described. ${ }^{21}$

## ASSOCIATED CONTENT

## Supporting Information

Elemental analysis data. This material is available free of charge via the Internet at http://pubs.acs.org.

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## Notes

The authors declare no competing financial interest.

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## ABBREVIATIONS USED

NRT, nicotine replacement therapy; CDC, Centers for Disease Control; nAChR, nicotinic acetylcholine receptor; ICSS, intracranial self-stimulation; CPP, conditioned place preference; DD, drug discrimination; SA, self-administration; $\mathrm{PdCl}_{2}$ (dppf), $1,1^{\prime}$-bis(diphenylphosphino)ferrocene-palladium(II) dichloride; MCPBA, meta-chloroperoxybenzoic acid; MPE, maximum potential effect; $\mathrm{DH} \beta \mathrm{E}$, dihydro- $\beta$-erythroidine; $\mathrm{AD}_{50}$, the antagonist dose that blocks $50 \%$ of the nicotine response

## REFERENCES

(1) World Health Organization. WHO Report on the Global Tobacco Epidemic, 2011; Warning about the Dangers of Tobacco; 2011.
(2) Center for Disease Control and Prevention.. Vital Signs: Current cigarette smoking among adults aged $\geq 18$ years: United States, 2009. Morbidity and Mortality Weekly Report 2010, 59, 1135-1140.
(3) SAMHSA. (Substance Abuse and Mental Health Services Administration) Results from the 2010 National Survey on Drug Use and Health: Summary of National Findings; U. S. Department of Health and Human Services: Washington, DC, 2011.
(4) American Cancer Society. Cancer Facts \& Figures 2011; American Cancer Society: Atlanta, Georgia, 2011.
(5) De Biasi, M.; Dani, J. A. Reward, addiction, withdrawal to nicotine. Annu. Rev. Neurosci. 2011, 34, 105-130.
(6) Hurt, R. D.; Sachs, D. P.; Glover, E. D.; Offord, K. P.; Johnston, J. A.; Dale, L. C.; Khayrallah, M. A.; Schroeder, D. R.; Glover, P. N.; Sullivan, C. R.; Croghan, I. T.; Sullivan, P. M. A comparison of sustained-release bupropion and placebo for smoking cessation. $N$. Engl. J. Med. 1997, 337, 1195-1202.
(7) Hayford, K. E.; Patten, C. A.; Rummans, T. A.; Schroeder, D. R.; Offord, K. P.; Croghan, I. T.; Glover, E. D.; Sachs, D. P.; Hurt, R. D. Efficacy of bupropion for smoking cessation in smokers with a former history of major depression or alcoholism. Br. J. Psychiatry 1999, 174, 173-178.
(8) Slemmer, J. E.; Martin, B. R.; Damaj, M. I. Bupropion is a nicotinic antagonist. J. Pharmacol. Exp. Ther. 2000, 295, 321-327.
(9) Coe, J. W.; Brooks, P. R.; Vetelino, M. G.; Wirtz, M. C.; Arnold, E. P.; Huang, J.; Sands, S. B.; Davis, T. I.; Lebel, L. A.; Fox, C. B.; Shrikhande, A.; Heym, J. H.; Schaeffer, E.; Rollema, H.; Lu, Y.; Mansbach, R. S.; Chambers, L. K.; Rovetti, C. C.; Schulz, D. W.; Tingley, F. D., III; O'Neill, B. T. Varenicline: An $\alpha 4 \beta 2$ nicotinic receptor partial agonist for smoking cessation. J. Med. Chem. 2005, 48, 3474-3477.
(10) Coe, J. W.; Brooks, P. R.; Wirtz, M. C.; Bashore, C. G.; Bianco, K. E.; Vetelino, M. G.; Arnold, E. P.; Lebel, L. A.; Fox, C. B.; Tingley, F. D., III; Schulz, D. W.; Davis, T. I.; Sands, S. B.; Mansbach, R. S.; Rollema, H.; O'Neill, B. T. 3,5-Bicyclic aryl piperidines: A novel class of $\alpha 4 \beta 2$ neuronal nicotinic receptor partial agonists for smoking cessation. Bioorg. Med. Chem. Lett. 2005, 15, 4889-4897.
(11) Mihalak, K. B.; Carroll, F. I.; Luetje, C. W. Varenicline is a partial agonist at $\alpha 4 \beta 2$ and a full agonist at $\alpha 7$ neuronal nicotinic receptors. Mol. Pharmacol. 2006, 70, 801-805.
(12) Bordia, T.; Hrachova, M.; Chin, M.; McIntosh, J. M.; Quik, M. Varenicline is a potent partial agonist at $\alpha 6 \beta 2^{*}$ nicotinic acetylcholine
receptors in rat and monkey striatum. J. Pharmacol. Exp. Ther. 2012, 342, 327-334.
(13) U.S. Food and Drug Administration. FDA Drug Safety Communication: Chantix (varenicline) may increase the risk of certain cardiovascular adverse events in patients with cardiovascular disease (Safety Announcement June 16, 2011). http://www.fda.gov/ Drugs/Drugsafety/ucm259161.htm (accessed September 2013).
(14) Lloyd, G. K.; Williams, M. Neuronal nicotinic acetylcholine receptors as novel drug targets. J. Pharmacol. Exp. Ther. 2000, 292, 461-467.
(15) Spade, T. E.; Garraffo, H. M.; Edwards, M. W.; Yeh, H. J. C.; Pannell, L.; Daly, J. W. Epibatidine: A novel (chloropyridyl)azabicycloheptane with potent analgesic activity from an Ecuadoran poison frog. J. Am. Chem. Soc. 1992, 114, 3475-3478.
(16) Badio, B.; Daly, J. W. Epibatidine, a potent analgetic and nicotinic agonist. Mol. Pharmacol. 1994, 45, 563-569.
(17) Badio, B.; Shi, D.; Garraffo, M.; Daly, J. W. Antinociceptive effects of the alkaloid epibatidine: Further studies on involvement of nicotinic receptors. Drug Dev. Res. 1995, 36, 46-59.
(18) Carroll, F. I. Epibatidine structure-activity relationships. Bioorg. Med. Chem. Lett. 2004, 14, 1889-1896.
(19) Carroll, F. I. Epibatidine analogs synthesized for characterization of nicotinic pharmacophores-A review. Heterocycles 2009, 79, 99120.
(20) Carroll, F. I.; Lee, J. R.; Navarro, H. A.; Brieaddy, L. E.; Abraham, P.; Damaj, M. I.; Martin, B. R. Synthesis, nicotinic acetylcholine receptor binding, and antinociceptive properties of 2 -exo-2-( $2^{\prime}$-substituted-3'-phenyl-5'-pyridinyl)-7-azabicyclo[2.2.1]heptanes. Novel nicotinic antagonist. J. Med. Chem. 2001, 44, 40394041.
(21) Carroll, F. I.; Ware, R.; Brieaddy, L. E.; Navarro, H. A.; Damaj, M. I.; Martin, B. R. Synthesis, nicotinic acetylcholine receptor binding, and antinociceptive properties of $2^{\prime}$-fluoro- $3^{\prime}$-(substituted phenyl)deschloroepibatidine analogs. Novel nicotinic antagonist. J. Med. Chem. 2004, 47, 4588-4594.
(22) Abdrakhmanova, G. R.; Damaj, M. I.; Carroll, F. I.; Martin, B. R. 2-Fluoro-3-(4-nitro-phenyl)deschloroepibatidine is a novel potent competitive antagonist of human neuronal $\alpha 4 \beta 2$ nAChRs. Mol. Pharmacol. 2006, 69, 1945-1952.
(23) Tobey, K. M.; Walentiny, D. M.; Wiley, J. L.; Carroll, F. I.; Damaj, M. I.; Azar, M. R.; Koob, G. F.; George, O.; Harris, L. S.; Vann, R. E. Effects of the specific $\alpha 4 \beta 2 \mathrm{nAChR}$ antagonist, 2-fluoro-3-(4nitrophenyl)deschloroepibatidine, on nicotine reward-related behaviors. Psychopharmacology (Berlin, Ger.) 2012, 223, 159-168.
(24) Ondachi, P.; Castro, A.; Luetje, C. W.; Damaj, M. I.; Mascarella, S. W.; Navarro, H. A.; Carroll, F. I. Synthesis and nicotinic acetylcholine receptor in vitro and in vivo pharmacological properties of 2'-fluoro-3'-(substituted phenyl)deschloroepibatidine analogues of 2'-fluoro-3'-(4-nitrophenyl)deschloroepibatidine. J. Med. Chem. 2012, 55, 6512-6522.
(25) Brieaddy, L. E.; Liang, F.; Abraham, P.; Lee, J. R.; Carroll, F. I. New synthesis of 7-(tert-butoxycarbonyl)-7-azabicyclo[2.2.1]hept-2ene. A key intermediate in the synthesis of epibatidine and analogs. Tetrahedron Lett. 1998, 39, 5321-5322.
(26) Carroll, F. I.; Liang, F.; Navarro, H. A.; Brieaddy, L. E.; Abraham, P.; Damaj, M. I.; Martin, B. R. Synthesis, nicotinic acetylcholine receptor binding, and antinociceptive properties of 2 -exo-2-(2'-substituted $5^{\prime}$-pyridinyl)-7-azabicyclo[2.2.1]heptanes. Epibatidine analogues. J. Med. Chem. 2001, 44, 2229-2237.
(27) Gao, Y.; Horti, A. G.; Kuwabara, H.; Ravert, H. T.; Hilton, J.; Holt, D. P.; Kumar, A.; Alexander, M.; Endres, C. J.; Wong, D. F.; Dannals, R. F. Derivatives of (-)-7-methyl-2-(5-(pyridinyl)pyridin-3-yl)-7-azabicyclo[2.2.1]heptane are potential ligands for positron emission tomography imaging of extrathalamic nicotinic acetylcholine receptors. J. Med. Chem. 2007, 50, 3814-3824.
(28) Billingsley, K. L.; Barder, T. E.; Buchwald, S. L. Palladiumcatalyzed borylation of aryl chlorides: Scope, applications, and computational studies. Angew. Chem., Int. Ed. 2007, 46, 5359-5363.
(29) Martin, T.; Laguerre, C.; Hoarau, C.; Marsais, F. Highly efficient borylation Suzuki coupling process for 4-bromo-2-ketothiazoles: straightforward access to micrococcinate and saramycetate esters. Org. Lett. 2009, 11, 3690-3693.
(30) AlSharari, S. D.; Carroll, F. I.; McIntosh, J. M.; Damaj, M. I. The antinociceptive effects of nicotinic partial agonists varenicline and sazetidine-A in murine acute and tonic pain models. J. Pharmacol. Exp. Ther. 2012, 342, 742-749.
(31) Ortiz, N. C.; O'Neill, H. C.; Marks, M. J.; Grady, S. R. Varenicline blocks $\beta 2^{*}$-nAChR-mediated response and activates $\beta 4^{*}$ -nAChR-mediated responses in mice in vivo. Nicotine Tob. Res. 2012, 14, 711-719.
(32) Summerfield, S. G.; Read, K.; Begley, D. J.; Obradovic, T.; Hidalgo, I. J.; Coggon, S.; Lewis, A. V.; Porter, R. A.; Jeffrey, P. Central nervous system drug disposition: The relationship between in situ brain permeability and brain free fraction. J. Pharmacol. Exp. Ther. 2007, 322, 205-213.
(33) Ghose, A. K.; Herbertz, T.; Hudkins, R. L.; Dorsey, B. D.; Mallamo, J. P. Knowledge-based, central nervous system (CNS) lead selection and lead optimization for CNS drug discovery. ACS Chem. Neurosci. 2012, 3, 50-68.
(34) Clark, D. E. Rapid calculation of polar molecular surface area and its application to the prediction of transport phenomena. 2. Prediction of blood-brain barrier penetration. J. Pharm. Sci. 1999, 88, 815-821.

