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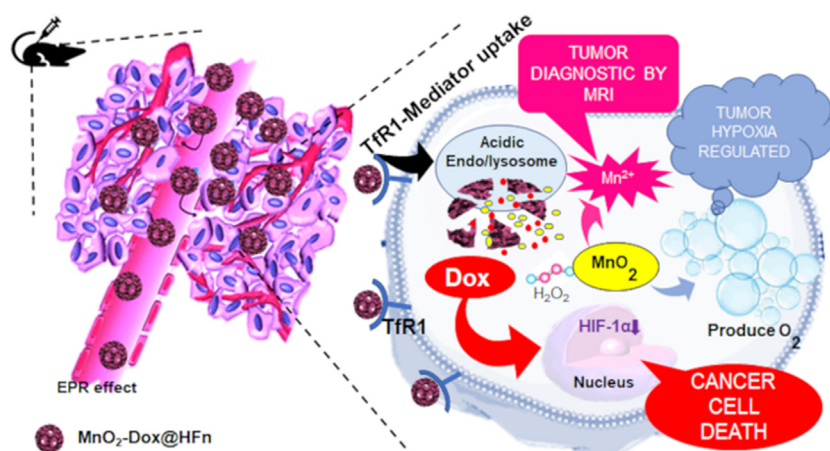
Innate tumor-targeted nanozyme overcoming tumor hypoxia for cancer theranostic use

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GRAPHICAL ABSTRACT



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ABSTRACT

Introduction: Hypoxic tumor microenvironment (TME) is the major contributor to cancer metastasis, resistance to chemotherapy, and recurrence of tumors. So far, no approved treatment has been available to overcome tumor hypoxia.

Objectives: The present study aimed to relieve tumor hypoxia via a nanozyme theranostic nanomaterial as well as providing magnetic resonance imaging (MRI)-guided therapy.

Methods: Manganese dioxide (MnO_2) was used for its intrinsic enzymatic activity co-loaded with the anti-cancer drug Doxorubicin (Dox) within the recombinant heavy-chain apoferritin cavity to form MnO_2 -Dox@HFnanoparticles. Following the synthesis of the nanomaterial, different characterizations were performed as well as its nanozyme-like ability. This nanoplatform recognizes tumor cells through the transferrin receptors 1 (TFR1) which are highly expressed on the surface of most cancer cells. The cellular uptake was confirmed by flow cytometry and fluorescence spectroscopy. *In vitro* and *in vivo* studies have been investigated to evaluate the hypoxia regulation, MRI ability and anti-tumor activity of MnO_2 -Dox@HFnanoparticles.

Results: Being a TME-responsive nanomaterial, MnO_2 -Dox@HFnanoparticles exerted both peroxidase and catalase activity that mainly produce massive oxygen and Mn^{2+} ions. Respectively, these products relieve the unfavorable tumor hypoxia and also exhibit T1-weighted MRI with a high longitudinal relaxivity of

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33.40 mM. s⁻¹. The utility of MnO₂-Dox@HF_n was broadened with their efficient anti-cancer activity proved both *in vitro* and *in vivo*.

Conclusions: MnO₂-Dox@HF_n successfully overcome tumor hypoxia with double potentials enzymatic ability and diagnostic capacity. This investigation could ignite the future application for cancer theranostic nanozyme therapy.

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Introduction

Tumor microenvironment (TME) is recognized jointly as the hallmark of solid tumors that is regularly featured by low levels of oxygen (hypoxia), acidosis (lysosome pH 4–5) coupled with a high concentration of reactive oxygen species (e.g., H₂O₂) [1–3]. Essentially, tumor hypoxia severely weakens cancer treatment by promoting chemotherapy resistance, tumor metastasis and tumor recurrence [4,5]. Since hypoxic cancer cells adapt their metabolism through chronic overproduction of hypoxia-inducible-factor-1 (HIF-1 α) which involves multiple cellular regulations. For instance, HIF-1 α regulates the up-expression of vascular endothelial protein (VEGF) which stimulates angiogenesis, resulting in proliferation and invasiveness of malignant cells [6,7]. Consequently, oxygen-carrying and oxygen-generating nanoparticles are developed to enhance tumor oxygenation since this factor seriously exacerbates cancer diagnosis and therapy [8–10]. However, most of these synthesized nanomaterials are limiting by the lower and short-time capacity to produce oxygen (O₂), thus any effect impacts the tumor invasion. Using TME-abundant metabolite (H₂O₂ and H⁺) as a substrate is a valuable strategy to provide massive and continuous O₂ production as well as regulating this unfavorable TME [11–13]. Therefore, exploiting the intrinsic catalytic ability of manganese dioxide (MnO₂) via the conversion of endogenous H₂O₂ to generate O₂ has been conceived as one of the most effective O₂-generating methods [14–16]. Particularly under acidic conditions, MnO₂ not only acting as a catalyst but also as a reactant to produce Mn²⁺ ions. The resulting Mn²⁺ ions can improve the proton relaxation rate to achieve T1-weighted magnetic resonance imaging (MRI) for specific tumor diagnosis [17,18]. MRI is amongst the useful clinical diagnostic methods for the imaging of soft tissue, especially at early-stage tumors, as it can penetrate deeply and identify tumor lesions with high resolution [19]. Furthermore, the utilization of Mn²⁺ as an MRI contrast agent is unsharable because Mn is one of the trace minerals required by the organism. Also, the innocuous Mn²⁺ ions are water-soluble and simply excreted by the kidneys [11] this means no fear of long-term toxicity risk when used for *in vivo* applications.

Owning all of those abilities, MnO₂ has been recently brought into tumor cells with various matrices, either directly as MnO₂ nanostructures or loaded into nanocarriers [13,15,16]. A wide range of nanocarriers have been used to deliver MnO₂ to the tumor site, such as PAMAM dendrimer [20] hyaluronic acid [21] cancer cell membrane [22] and mesoporous silica nanoparticle [23]. However, inorganic nanocarriers are often struggled to meet clinical expectations because they may cause an inflammatory response or neurotoxic reactions [24]. Additionally, most carriers need additional ligand functionalization to target tumor cells, they face a number of issues such as reproducibility, surface characterization and immune reactions [25]. Therefore, the use of endogenous vehicles already presents in the body like apoferritin, an empty shell of ferritin protein, which possesses a great innate active targeting ability toward tumors would be a shorter alternative route to deliver MnO₂ safely.

The inner surface of the recombinant heavy-chain apoferritin (HF_n) has a high negative charge density which could serve as a

suitable vehicle for the positively charged MnO₂. As being ubiquitous found in all living organisms, the potentiality of HF_n to be used as a tumor-targeted nanocarrier has been widely explored. First, the inner cavity of HF_n is large enough to accumulate MnO₂ with an outer diameter of ~12 nm, making it ideal for passive tumor targeting through enhanced permeability and retention (EPR) effect [26]. The remarkable flexibility of the HF_n protein shell also allows its use as an excellent drug delivery carrier which may encapsulate larger compounds than its size [27]. Second, the protein structure is a hollow nanocage composed of 24 self-assembling subunits that decay under an acidic pH and recombine in a neutral environment, promising TME selective drug delivery [28]. Naturally, HF_n binds specifically to transferrin receptors 1 (TfR1) via clathrin-coated pits to be internalized within the cancer cells [29]. It is well-known that TfR1 are highly expressed by > 100-fold on the surface of most common cancer cells due to their increased iron demand compared to normal cells [30,31]. TfR1 has therefore been advised as a privileged targeting marker for the tumor-selective delivery to increase the accumulation of nanomaterial in the tumor site and minimize the side effects on normal tissues [32]. Recently, nanozyme has spawned larger advances in nanotechnology research. Outline that nanozyme are small nanomaterials between 1 and 100 nm size owning enzymatic-like properties [33]. Taken together, nanozyme-based theranostic material could be a better alternative to treat cancer especially by modulating the unfavorable TME.

In the present work, a versatile theranostic nanozyme material in which MnO₂ and Doxorubicin (Dox) were co-loaded within the HF_n nanocage to form MnO₂-Dox@HF_n was designed for the first time.

Experimental

Synthesis of MnO₂-Dox@HF_n

MnO₂ was formed by a simple redox reaction between potassium permanganate (KMnO₄) and poly-allylamine hydrochloride (PAH). Shortly, 80 μ L of KMnO₄ (10 mg mL⁻¹) was added in 20 mL deionized water and 80 μ L PAH (23.6 mg mL⁻¹) under moderate stirring at 37 °C to allow MnO₂ growth. After 30 min reaction, the MnO₂ solution was centrifuged and the precipitate was collected in 2 mL deionized water, MnO₂ was obtained. To disassemble the HF_n subunits, the pH of the solution was lowered to 3.5 by adding 0.1 M HCl. Then, MnO₂ (6.64 mM) and Dox (1.5 mM) were added to the protein solution (10 mg mL⁻¹), the molar ratio of MnO₂: Dox: HF_n was set as 1:2:5. The mixture was stirred for 15 min and the pH was finally adjusted to 7.4 using 0.1 M NaCl to allow the self-assembly of the HF_n nanocage. The resulting solution was stirred for an additional 2 h at room temperature. To remove the unloaded molecules, the mixture was dialyzed against a storage buffer for 24 h. The supernatant was analyzed on a Sephadex 100 column for size-exclusion chromatography (SEC). The fraction volume was set at 1 mL, and the absorbance of each eluted fraction was recorded at 280, 370 and 480 nm by UV spectra.

Enzymatic activity

The production of Mn^{2+} in the tubes containing MnO_2 -Dox@HF_n under different pH (7.4, 6.5 and 5.0) with or without the presence of 100 μM H_2O_2 was investigated. To assay the O_2 generation from varied concentrations of MnO_2 -Dox@HF_n, $[Ru(dpp)3]Cl_2$ was used as an O_2 sensing probe. Furthermore, MnO_2 -Dox@HF_n was incubated in PBS solution pH 5.0 adding 100 μM H_2O_2 to mimic the TME. After 30 min, the O_2 bubble generation was examined macroscopically. While, the peroxidase-like activity of MnO_2 -Dox@HF_n was measured through the catalytic oxidization of 3,3',5,5'-tetramethylbenzidine (TMB) with H_2O_2 . The experiment was performed using TMB (5 μM) and MnO_2 -Dox@HF_n ($MnO_2 = 45 \mu M$) in a reaction volume of 3 mL buffer solution (100 mM acetate buffer, pH 5.0) with 100 μM H_2O_2 . The reaction solution was detected using the UV-vis spectra at $\lambda = 650$ nm. Additionally, the kinetic data were obtained by varying the concentrations of H_2O_2 (5, 10, 25, 50 mM) and then calculated based on the Michaelis-Menten and the Lineweaver-Burk plot method.

MRI in vitro

To determine the proton relaxation rates (r_1) of MnO_2 -Dox@HF_n, the longitudinal relaxation time T_1 (s) has been investigated under different concentrations at pH 6.0 and pH 7.4. The r_1 was determined as the slope of the concentration-dependent longitudinal relaxation time T_1 curve.

In vitro cytotoxicity

The human cervical cancer cells (HeLa), human ovarian cancer cells (SKOV3) and 4 T1 murine breast cancer cells (4 T1) were originally obtained from the American Type Culture Collection (ATCC) and cultured with the recommended medium at 37 °C within 5% CO_2 atmosphere.

In vitro cell viability was determined by standard 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide (MTT) assay after 48 h incubation under normal and hypoxic environment. A fresh medium containing $CoCl_2$ (100 μM) was used to mimic hypoxia. The half-maximal inhibitory concentration (IC_{50}) values were calculated on HeLa, 4 T1 and SKOV3 cancer cells.

Cellular uptake

To visualize the cellular uptake of the nanozyme, the surface of the protein carrier was labeled with fluorescein-5-maleimide (FAM) as previously conducted [34].

Flow cytometry was performed to analyze quantitatively the cellular uptake of MnO_2 -Dox@HF_n. HeLa was used as TfR1-positive cells which overexpressed TfR1, while SKOV-3 as TfR1-negative cells which have lower TfR1 expression levels. Cells were co-incubated with FAM- MnO_2 -Dox@HF_n for different times (1, 2, and 4 h). After washes with cold PBS, cells were collected by trypsinization, and then analyzed immediately. To study the interaction and endocytosis mechanism between cancer cells and MnO_2 -Dox@HF_n nanomaterials, HeLa cells were seeded into six-well plates (5×10^4 cells per well). Firstly, for the analysis of energy-dependent mechanism: the cells were divided into three groups; the control group was co-cultivated directly with FAM- MnO_2 -Dox@HF_n (1.5 mg mL^{-1}) at 37 °C for 2 h; the second group of cells was co-cultured with the above nanomaterial at 4 °C for 2 h (low temperature inhibits cell physiology and metabolic activity) and the third group of cells was pretreated with sodium azide (1 mg mL^{-1}) for 1 h (sodium azide inhibits intracellular ATP synthesis), followed by incubation of FAM- MnO_2 -Dox@HF_n for 2 h. Secondly, for the endocytosis mechanism analysis: cells were pre-

treated with endocytic inhibitors, including chlorpromazine (8.5 μg mL^{-1}), genistein (56.75 μg mL^{-1}) and amiloride (133 μg mL^{-1}) for 1 h prior to FAM- MnO_2 -Dox@HF_n (1.5 mg mL^{-1}) administration and further incubated for 2 h [35]. Quantitative analysis by flow cytometry was carried out using the above-mentioned method. Compared with the control cells without inhibitors, the decrease in the uptake rate in the presence of inhibitors demonstrated the corresponding endocytosis pathways. Besides, the qualitative intracellular uptake of FAM- MnO_2 -Dox@HF_n was monitored by confocal laser scanning microscopy (CLSM).

Cellular hypoxia amelioration

The intracellular O_2 generation of MnO_2 -Dox@HF_n was investigated by using $[Ru(dpp)3]Cl_2$ fluorescence. HeLa cells were ventilated with nitrogen for 3 h first to induce severe hypoxia. Then incubated with MnO_2 -Dox@HF_n (10 μg mL^{-1}) for 3, 6, 12 and 24 h and further incubated with 1 μM $[Ru(dpp)3]Cl_2$ for 4 h to assay the O_2 content. Finally, cells were rinsed and treated with 4% paraformaldehyde for CLSM imaging.

Western blot analysis

HeLa cells were incubated with $CoCl_2$ (100 μM) to induce severe hypoxia since $CoCl_2$ inhibits PHD enzyme which degrades HIF-1 α [36]. After 48 h incubation with different formulations, HIF-1 α expression was assessed by western blot (WB) as previously conducted [5].

Ethics statement

All experiments involving animals were conducted according to the ethical policies and regulations provided by the Guide for Care and Use of Laboratory Animals, approved by China Pharmaceutical University (Approval no. 2019-1205).

In vivo study

Female Balb/c mice weighing 18–20 g were used and the tumor-bearing Balb/c mice model was established by subcutaneously injecting 4 T1 tumor cells (1×10^6 cells suspended in 100 μL of saline) into the right flank breast of each mouse. The experiment was performed when the tumor volume reached ≈ 200 mm^3 .

MRI in vivo imaging

A clinical MRI scanner was used to observe MR imaging. Intra-tumor, intramuscular, and intravenous (via tail vein) injection of MnO_2 -Dox@HF_n (0.2 mM MnO_2) was embarked. T1-weighted imaging was done with the following parameters: TR = 760 ms; TE = 16.9 ms; field of view = 100 cm^2 ; matrix: 256 mm \times 256 mm; slice thickness = 3 mm and space = 0.5 mm.

In vivo Anti-tumor activity

The mice were randomly divided into five groups ($n = 3$) and treated intravenously with different formulations. The doses of MnO_2 and Dox were kept at 2.5 mg kg^{-1} and 2.5 mg kg^{-1} , respectively, in those groups and the treatment was studied for 10 days.

To evaluate the apoptotic response in tumor tissues, terminal deoxynucleotidyl transferase dUTP-biotin nick end labeling (TUNEL) staining and hematoxylin and eosin (H&E) staining was conducted. The hypoxia in tumor tissues was analyzed through the expression of HIF-1 α protein by immunohistochemistry. More-

over, staining for the Ki-67 antigen was used to evaluate the tumor cell proliferation.

Statistical analysis

Data are presented as mean \pm standard deviation (SD) of the mean for results obtained from three independent trials unless otherwise indicated. Student's *t*-test (two-tailed) or analysis of variance (ANOVA) was used to determine statistical significance between two or more groups, respectively. The results of *p*-values < 0.05 were considered statistically significant.

Results and discussion

Synthesis and characterization of MnO₂-Dox@HF_n

The production of recombinant human HF_n from *E. coli* ended up with 30 mg L⁻¹ protein. The purification of HF_n was revealed via SDS-PAGE (S1, Supplementary material). In the effluent eluted by 300 mM Imidazole, high purity protein subunits were observed at the band 29 kDa, in line with the molecular weight of the recombinant HF_n monomer [37]. Protein determination at 280 nm by Nanodrop displayed only one peak of protein (S2), confirming the successful purification of HF_n. Spherical cage-like was observed through TEM analysis of HF_n in S3, with a mean size diameter of 12.05 \pm 1.3 nm.

MnO₂ and Dox were loaded into the HF_n protein cavity by the easiest pH change method. The SEC profile showed the co-elution of the protein nanocage, MnO₂ and Dox in fraction 4 (Fig. 1A), suggesting the combination of MnO₂-Dox@HF_n nanomaterial. The SDS-PAGE of various formulations revealed a similar band at MW \sim 29 kDa, as did the blank HF_n (Fig. 1B), showing that encapsulation of the molecules had no impact on the HF_n subunit structure. Correlate with this result, the native PAGE of MnO₂-Dox@HF_n also displayed an identical band to that of HF_n alone, indicating that the HF_n nanocage retained its structural integrity and was unaltered during the core loading (S4). Additionally, HF_n and MnO₂-Dox@HF_n exhibited a similar CD spectrum in Fig. 1C, indicating that the secondary structure of the protein did not change after mineralization. According to the UV-vis spectra of the MnO₂-Dox@HF_n synthesis, the typical peak of KMnO₄ (\sim 525 nm) has disappeared and a new visible absorption at \sim 370 nm was observed, attributed to the formation of MnO₂ (Fig. 1D). The emerging peaks at 280 nm, 370 nm, and 480 nm highlighted the presence of HF_n protein, MnO₂, and Dox, respectively. Furthermore, the full XPS of MnO₂-Dox@HF_n indicated the presence of spectrum C, O, N, and Mn elements in Fig. 1E. Notably, the Mn 2p spectrum defined two peaks centered at 653.8 eV and 642.0 eV (Fig. 1F), corresponding respectively to Mn 2p_{1/2} and Mn 2p_{3/2}, characteristic of MnO₂. The spin energy separation of 11.8 eV indicated that the oxidation state of Mn was mainly + 4 [16]. No signal of Mn₂O₃ (641 eV) nor KMnO₄ (647 eV) were present observed in the XPS spectra, further confirming that the present MnO₂ is in the 4+ oxidation state. Mn concentration in MnO₂-Dox@HF_n was quantified by ICP-AES and found to be 0.4 mM. A similar Dox loaded rate of 48.4% was achieved as evaluated by the calibration curve (S5). Indeed, the encapsulation of Dox into HF_n nanocages depends on their electrostatic interactions. The conjugation of the hydrophobic Dox to the N-terminus of HF_n protein is expected to result in a self-assembly behavior. Because the pKa of Dox is 8.3, it can easily be attached to the negatively charged internal surface of HF_n in the mildly acidic loading buffer. Besides, the protein surface charge of blank HF_n shifted from -14.99 ± 2.65 mV to -21.02 ± 1.37 mV after MnO₂ loading (Fig. 1G), indicating the electrostatic interaction between HF_n and MnO₂ conjugation. The Mn atom basic unit of MnO₂ could

be linked to the O atoms of the amino acid of HF_n protein via an intermolecular hydrogen bonding [38]. Furthermore, the positive charge of MnO₂ (31.19 ± 2.05 mV) was hidden when it's encapsulated within the MnO₂-Dox@HF_n nanozyme (-24.68 ± 1.58 mV). According to TEM, MnO₂-Dox@HF_n was found to be uniform with an external diameter of 10 \sim 12 nm (Fig. 1H), a perfect size to exhibit strong catalytic activity [33]. Fig. 1I shows that MnO₂-Dox@HF_n dispersed well without any obvious aggregation in PBS, physiological saline, FBS and DMEM, with polydispersity lower than 0.3. The results suggest that MnO₂-Dox@HF_n was stable in physiological circumstances and suitable for intravenous administration. Both blank HF_n and MnO₂-Dox@HF_n in FBS and DMEM medium possess a mean hydrodynamic diameter around 13 nm with a negative zeta potential charge (S6), which further proved the stability of the nanoparticles. Eventually, negligible particle size fluctuations were observed between the MnO₂-Dox@HF_n stored for 4 weeks and the freshly prepared material (S7), confirming their long-term stability.

TME-response payload release

The drug release behavior of Dox from MnO₂-Dox@HF_n was quantified by UV-vis spectrophotometer and the particle size were measured to monitor the disintegration of the nanozyme. As seen in Fig. 2A, differences in solution pH can lead to variance in the drug release behavior. At pH 7.4, no significant release of Dox was observed within 48 h and the particle size

was fairly stable (S8). Contrarily under acidic pH, a burst release occurred in the first six hours due to the dissociation of the MnO₂-Dox@HF_n system, which was confirmed by the increase in particle size to 154.62 \pm 1.57 nm at pH 5.0. This result is consistent with the previous report due to the HF_n nanoshell disintegration under an acidic environment [34]. To evaluate the Dox release mechanism of MnO₂-Dox@HF_n, the cumulative release profile was fitted to the five kinetic equations [39]. The Korsmeyer-Peppas model was found to represent the kinetics of drug release from MnO₂-Dox@HF_n (Table S1). This model provides further information on the value of release exponent 'n', which was < 0.4 , indicating that the release mechanism could be described as a Fickian diffusion mechanism [39]. Simultaneously, the degradation behavior of MnO₂ in MnO₂-Dox@HF_n was investigated by quantifying the release of Mn using AAS under various solutions to mimic TME. It can be seen in Fig. 2B that MnO₂ was degraded rapidly at pH 5.0 and pH 6.5, particularly in the presence of H₂O₂. These results confirmed the ultrasensitive response of MnO₂-Dox@HF_n to pH and H₂O₂, characteristic of TME.

Enzymatic activity

It is well known that MnO₂ possesses intrinsic catalytic activity and produce colorless Mn²⁺ ions in the mildly acidic environment. Considering this fact, MnO₂-Dox@HF_n was placed under different pH solutions with or without the presence of H₂O₂. After 3 h, it was found that at pH 6.5 and pH 5.0 in presence of H₂O₂, the solution was becoming colorless (S9). This result demonstrates the reactivity of MnO₂-Dox@HF_n nanozyme in an acidic H₂O₂ condition that induces the production of uncolored Mn²⁺ ions. Besides, the catalytic O₂ generation of MnO₂-Dox@HF_n was monitored using [Ru(dpp)3]Cl₂, an O₂ detector sensor that reduced its luminescence by molecular O₂ [40]. As shown in Fig. 2C, [Ru(dpp)3]Cl₂ was gradually quenched as the concentration of MnO₂-Dox@HF_n increases due to the generation of O₂, in accordance with the previous report [41]. Notably, once located in the cancer cell, intact MnO₂-Dox@HF_n can exert catalase activity since H₂O₂ could diffuse into the HF_n cavity through the hydrophilic channels and interacts with the contained MnO₂ [32]. Nevertheless, oxygen bub-

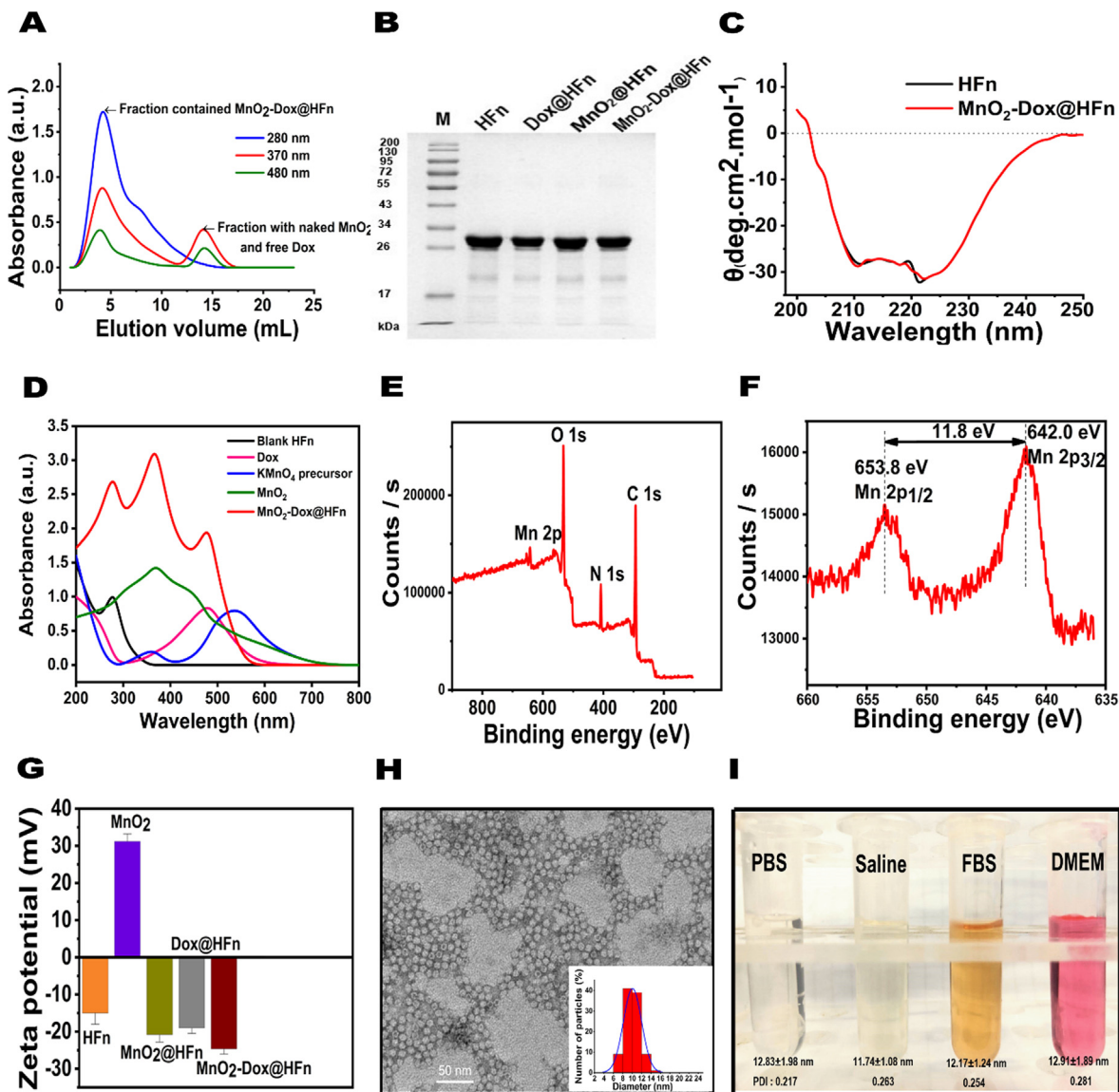


Fig. 1. Characterization of MnO₂-Dox@HFfn. (A) Size-exclusion chromatography of MnO₂-Dox@HFfn. The elution profile was obtained by the absorption recorded at 280, 370, and 480 nm, indicating the presence of HFfn protein, MnO₂, and Dox, respectively. (B) SDS-PAGE protein analysis of various formulations. (C) Circular dichroism spectra of HFfn and MnO₂-Dox@HFfn with the two characteristics minima of α -helical protein around 210 nm and 221 nm. (D) UV spectra for MnO₂-Dox@HFfn synthesis. (E) XPS analysis of MnO₂-Dox@HFfn full scan and (F) Mn 2p core-level spectra. (G) Zeta potential of various formulations. (H) TEM image of MnO₂-Dox@HFfn negatively stained with uranyl acetate. Inset represents the TEM size distribution. (I) Optical image of MnO₂-Dox@HFfn dispersed in PBS, saline, FBS, and DMEM solutions with hydrodynamic particle size and polydispersity values.

bles were macroscopically observed in the acidic H₂O₂ solution after incubation with MnO₂-Dox@HFfn as seen in Fig. 2D, vividly demonstrating the strong ability of MnO₂-Dox@HFfn to generate O₂. This powerful O₂ generation could be explained by the equation in S10-A, B where both the catalytic activity and the redox reaction of MnO₂-Dox@HFfn are conspicuous.

On the other hand, the second enzymatic activity of MnO₂-Dox@HFfn was evaluated by the catalytic oxidation of TMB substrate in a tumor-mimicking microenvironment. As illustrated in Fig. 2E, MnO₂-Dox@HFfn functioned as an excellent peroxidase mimetic material by developing a blue color product with maximum absorption at 652 nm, typical of the oxidized TMB. This result was validated by the equation in S10-C. Moreover, the real-time change in the substrate TMB absorption was monitored under different concentrations of H₂O₂ (S11), from which the time-course absorbance of oxidized TMB was collected (Fig. 2F). The peroxidase

kinetic parameters were plotted according to the Michaelis-Menten curve (Fig. 2G) following the equation

$$V_0 = V_{max} \times \frac{[S]}{(K_m + [S])}$$

where V₀ is the initial velocity, V_{max} is the maximal reaction velocity, [S] is the concentration of substrate and K_m is the Michaelis constant. From the Lineweaver-Burk plot (Fig. 2H), K_m and V_{max} of MnO₂-Dox@HFfn were calculated to be 28.87 × 10⁻⁶ M s⁻¹ and 5.23 mM, respectively. As illustrated in the equation S10-C, the peroxidase ability of MnO₂-Dox@HFfn catalyze the endogenous H₂O₂ to produce hydroxyl radical, a kind of reactive oxygen species, thereby leading to intracellular oxidative damage [12] which could increase the anti-tumor activity of MnO₂-Dox@HFfn.

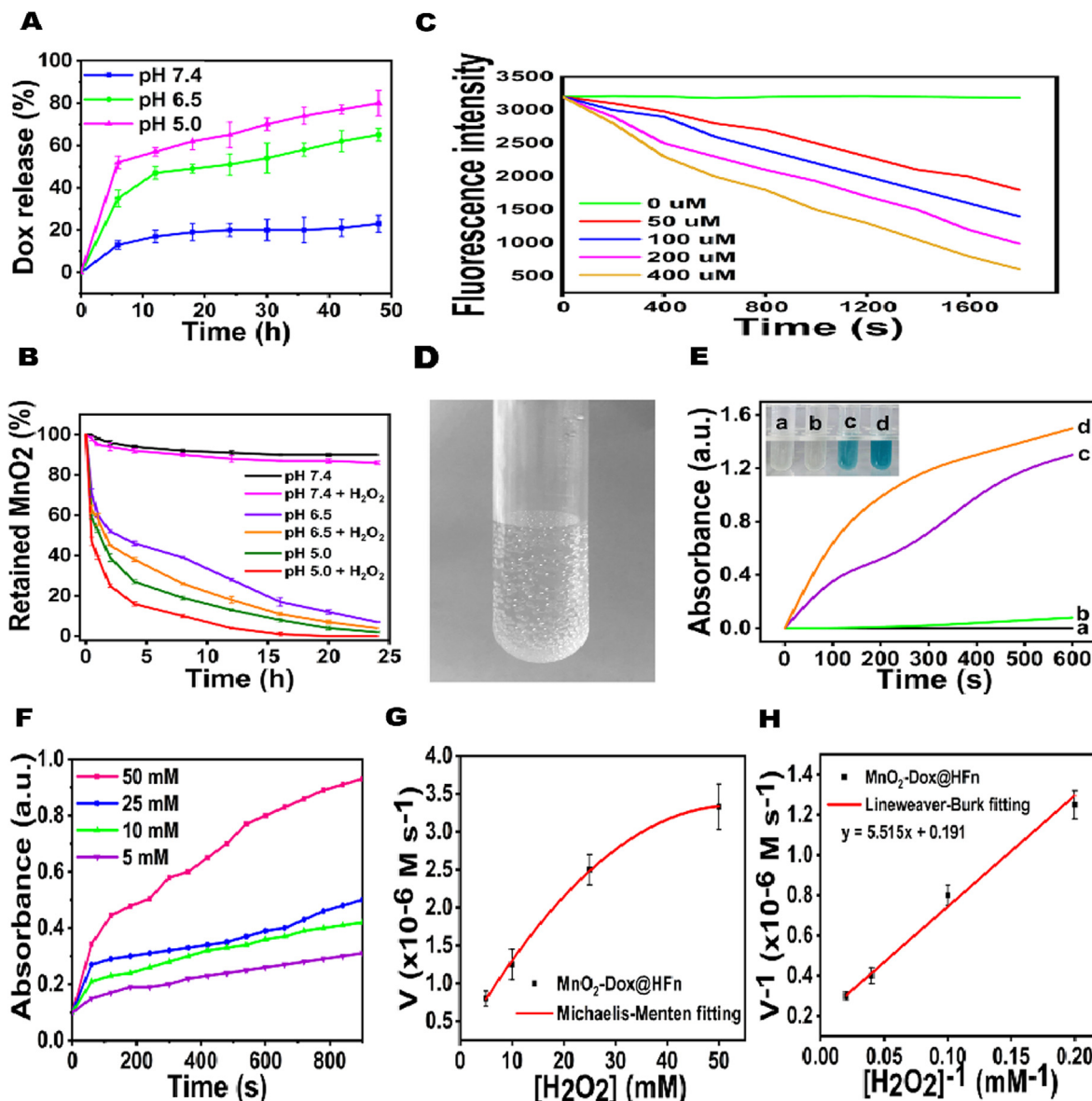


Fig. 2. Disassociation behavior of $\text{MnO}_2\text{-Dox@HFfn}$. (A) The release profile of Dox from $\text{MnO}_2\text{-Dox@HFfn}$ at different pH. (B) MnO_2 decomposition quantified by Mn concentration using AAS. (C) O_2 generation from varied concentrations of $\text{MnO}_2\text{-Dox@HFfn}$. $[\text{Ru}(\text{dpp})_3]\text{Cl}_2$ fluorescence was quenched by the O_2 generated from $\text{MnO}_2\text{-Dox@HFfn}$. (D) O_2 bubbles were observed after incubation of $\text{MnO}_2\text{-Dox@HFfn}$ in a tumor mimic microenvironment condition. (E) Peroxidase-like activity of $\text{MnO}_2\text{-Dox@HFfn}$ measuring by the absorbance of oxidized TMB at 652 nm in the following solutions: a: TMB + H_2O_2 ; b: TMB + $\text{MnO}_2\text{-Dox@HFfn}$; c: TMB + H_2O_2 + $\text{MnO}_2\text{-Dox@HFfn}$; d: TMB + H_2O_2 + $\text{MnO}_2\text{-Dox@HFfn}$. The inset is a photograph of colored products. (F) TMB derived oxidation products overtime. (G) Michaelis-Menten and (H) Lineweaver-Burk fitting. The error bars represent the SD for three measurements.

In vitro MRI and cell viability

Taking advantage of the produced Mn^{2+} ions, which is an excellent T1-shortening agent, MRI *in vitro* study was embarked. The obvious concentration-based brightening effect in T1-weighted MR images of $\text{MnO}_2\text{-Dox@HFfn}$ at pH 6.0 was observed with a high relaxivity r_1 of $33.40 \text{ mM}^{-1} \text{ s}^{-1}$ (Fig. 3A-B). In contrast, no signal was found under pH 7.4, which may be attributed to the non-dissociation of $\text{MnO}_2\text{-Dox@HFfn}$ in physiological conditions.

Next, the non-toxicity of the blank nanocarrier was studied via MTT assay after 48 h incubation. As shown in Fig. 3C, cell viability remained above 90% even after incubation with concentrated HFfn protein (2.5 mg mL^{-1}). The nanoformulation cell death effect in a normal environment was evaluated towards HeLa, 4 T1 and SKOV3 cells as shown in S12. $\text{MnO}_2\text{-Dox@HFfn}$ nanomaterial exhibited unsatisfactory effects of cytotoxicity under normoxia. Comparatively,

the cell viability effect of $\text{MnO}_2\text{-Dox@HFfn}$ in the cancer cells under hypoxic conditions showed apparent dose-dependent anti-tumor activity (Fig. 3D-F). Intriguingly, $\text{MnO}_2\text{-Dox@HFfn}$ nanzyme demonstrated higher cytotoxicity on HeLa and 4 T1 cancer cells compared to the naked Dox. This enhancement could be attributed to a great cellular uptake of the encapsulated drug, improving its accumulation and anti-cancer efficiency. Furthermore, IC_{50} of HeLa cells treated with $\text{MnO}_2\text{-Dox@HFfn}$ was 5-fold lower than the SKOV3 cells (Fig. 3G) due to the difference of Tfr1 expression on their surface [42,43] allowing the cellular internalization of $\text{MnO}_2\text{-Dox@HFfn}$. Indeed, both Dox@HFfn and $\text{MnO}_2\text{-Dox@HFfn}$ possessed Tfr1-targeting ligand to enter the cells. However, it is well-known that cancer hypoxia induces drug resistance, therefore the difference in cytotoxicity result between Dox@HFfn and $\text{MnO}_2\text{-Dox@HFfn}$ could be explained by the increased efficiency of Dox since the unfavorable hypoxia condition was relieved and also

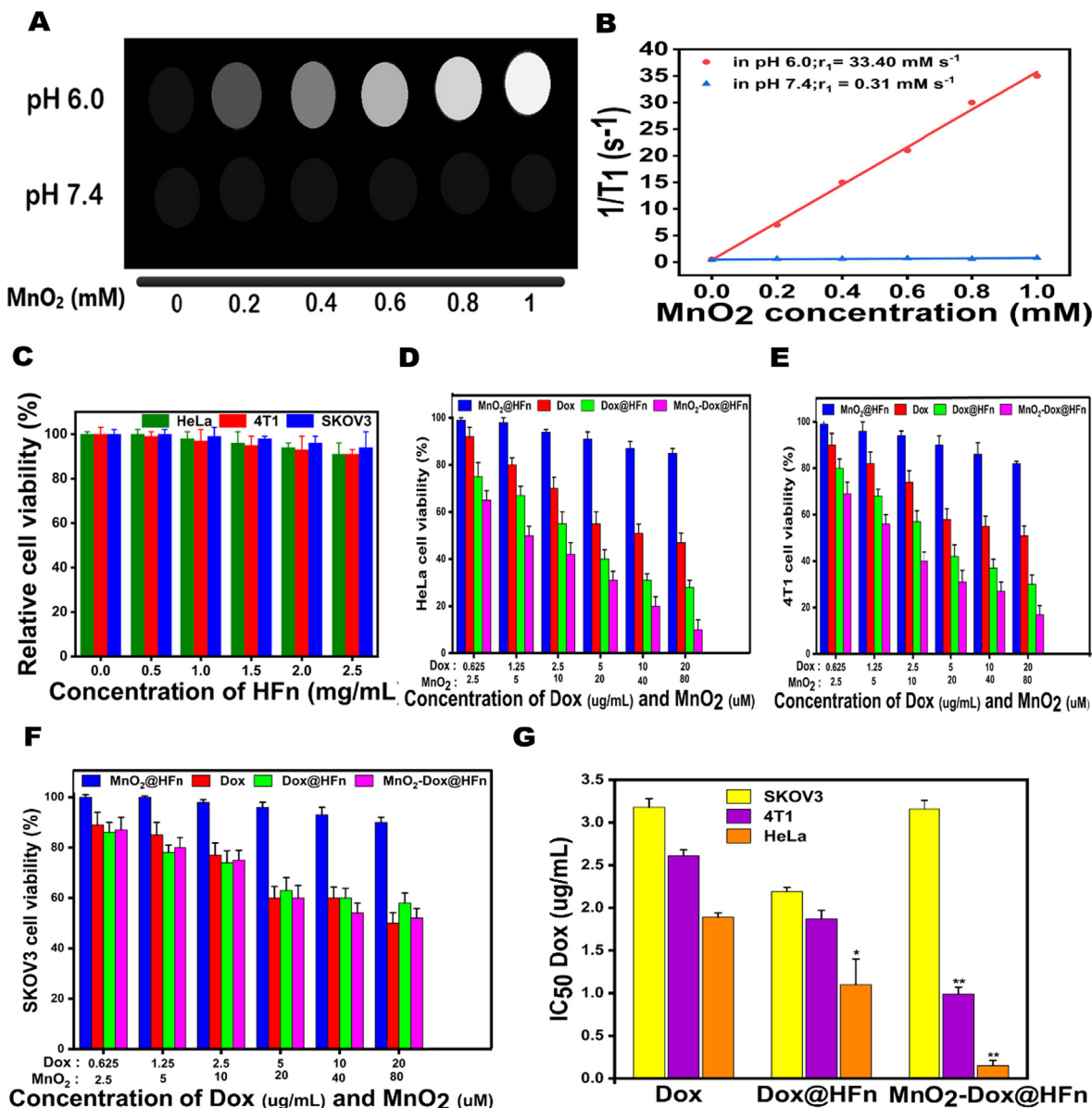


Fig. 3. *In vitro* study. (A) T1-weighted MRI *in vitro* with various MnO₂ concentrations under pH 6.0 and pH 7.4. (B) Concentration-dependent longitudinal relaxation time T1, based on readings from A. (C) Cytotoxicity of HFn carrier toward HeLa, 4 T1, and SKOV3 cancer cells. (D-F) Cell viability of three cell lines under hypoxic conditions determined by MTT assay exposed to various concentrations of nanomaterials for 48 h. The Dox concentration axis represents the quantity of Dox in each group for Dox contained formulations as well as the MnO₂ concentration axis for MnO₂ contained nanomaterials. (G) IC₅₀ values of various cancer cells based on Dox concentration (*p < 0.05, **p < 0.01 vs SKOV3). Data are presented as means ± SD (n = 3).

due to the cellular damage caused by free radical produced by MnO₂.

Cellular binding of MnO₂-Dox@HFn

The quantitative cellular uptake of MnO₂-Dox@HFn was performed by flow cytometry after successfully labeled the nanozyme with FAM fluorescent probe (S13). To confirm whether Tfr1 receptors mediate the binding of MnO₂-Dox@HFn toward cancer cells, cellular uptake was quantified in high Tfr1-expressing HeLa cells and low Tfr1-expressing SKOV3 cells [42,43]. As the incubation time increased, the cellular internalization was greatly increased (Fig. 4A-B), meanwhile, MnO₂-Dox@HFn carried a time-dependent cellular uptake. Significantly, MnO₂-Dox@HFn exhibited high cellular uptake potentials in HeLa cells compared to SKOV3 cells (Fig. 4C), showing that the internalization process

was mediated by Tfr1 affinity. Additionally, the effect of temperature on cellular uptake at 37 °C and 4 °C has been studied, as low temperatures may inhibit the activity of cell transport proteins. As illustrated in Fig. 4D and E, the mean fluorescence intensity (MFI) at 37 °C was significantly higher than that at 4 °C. This data suggested that the uptake of MnO₂-Dox@HFn possibly involving transport proteins, which may be related to energy-dependent processes [44]. To elucidate the absorption mechanism of MnO₂-Dox@HFn, cells were pretreated with sodium azide, chlorpromazine, genistein, and amiloride to inhibit energy-dependent endocytosis, clathrin-mediated endocytosis, caveolae-mediated endocytosis, and macropinocytosis, respectively [35]. As seen in Fig. 4D and E, the presence of sodium azide remarkably reduced the cellular uptake of MnO₂-Dox@HFn, indicating an energy-dependent mechanism. Chlorpromazine pretreatment also resulted in a significant MFI decrease of MnO₂-Dox@HFn uptake,

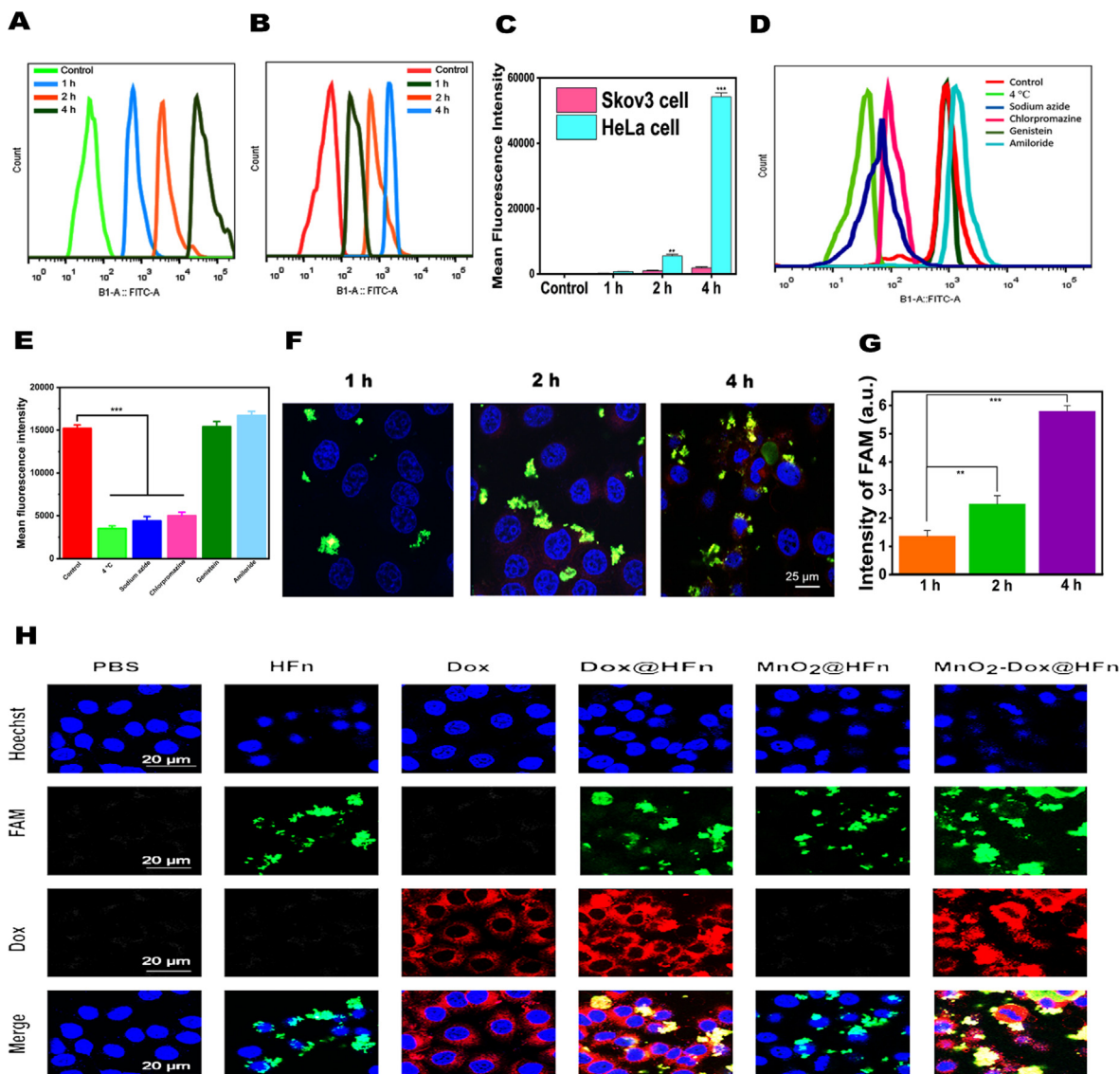


Fig. 4. Flow cytometry analysis of the cellular uptake of FAM-MnO₂-Dox@HFn in (A) HeLa and (B) SKOV3 cancer cells. (C) Quantitative analysis of FAM-MnO₂-Dox@HFn binding according to A and B. (D) Cellular uptake of FAM-MnO₂-Dox@HFn by flow cytometry in HeLa cells incubated with various endocytosis inhibitors. (E) The relative fluorescent intensity of FAM-MnO₂-Dox@HFn with the selective inhibitors according to D (**p < 0.01, ***p < 0.001 vs control) (n = 3). (F) Internalization of FAM-MnO₂-Dox@HFn in HeLa cells after incubation for 1, 2 and 4 h via CLSM. The nucleus was stained with Hoechst 33,342 (blue). The uptake of nanozyme was visualized by HFn protein carrier labeled with FAM (green) and the red fluorescence of Dox. (G) Quantitative analysis of FAM intensity according to D using ImageJ software (**p < 0.01, ***p < 0.001 vs 1 h). (H) HeLa cells uptake of various nanomaterials after 4 h incubation visualized by CLSM.

suggesting that the cell entry pathway was mainly involved in receptor- and clathrin-mediated endocytosis. Besides, pretreatment of the cells with genistein and amiloride did not affect the uptake of MnO₂-Dox@HFn nanomaterials. Based on these findings, we suggest that MnO₂-Dox@HFn are mainly bound with TfR1 receptors and entering cells by TfR1 clathrin-mediated endocytosis. Subsequently, the co-localization study of MnO₂-Dox@HFn using CLSM displayed an obvious cellular internalization trend over time (Fig. 4F-G), which demonstrated the nuclear translocation of the nanozyme. The accumulation of Dox fluorescence released from MnO₂-Dox@HFn inside the cell nuclei was observed (Fig. 4H), which was stronger than the naked Dox. It demonstrates that the specific binding of HFn nanocage through the TfR1-mediated endocytosis enhanced cellular drug accumulation compared to the passive diffusion performed by free Dox. Indeed, these results can explain the enhanced anti-tumor efficiency of MnO₂-Dox@HFn nanozyme.

Cellular oxygen generation

As stated previously, hypoxia has serious consequences in tumor progression, hence the intracellular oxygenation of MnO₂-Dox@HFn was examined. The cellular O₂ level was assessed by [Ru(dpp)3]Cl₂ probe, which decreases its luminescence intensity when the O₂ content increase [45]. As obviously illustrated in Fig. 5A-B, MnO₂-Dox@HFn triggered the generation of in situ O₂ and the hypoxic environment was completely removed after 24 h incubation. To further confirm such findings, the expression of O₂-regulated HIF-1α protein was quantified after inducing severe hypoxia with CoCl₂. Interestingly, the HIF-1α expression level of cells treated with MnO₂-Dox@HFn was significantly down-regulated (Fig. 5C-D), while for the control and Dox@HFn treated, cells were still under hypoxic challenge. Cells contain very little amount of HIF-1α when O₂ levels are high, hence this result con-

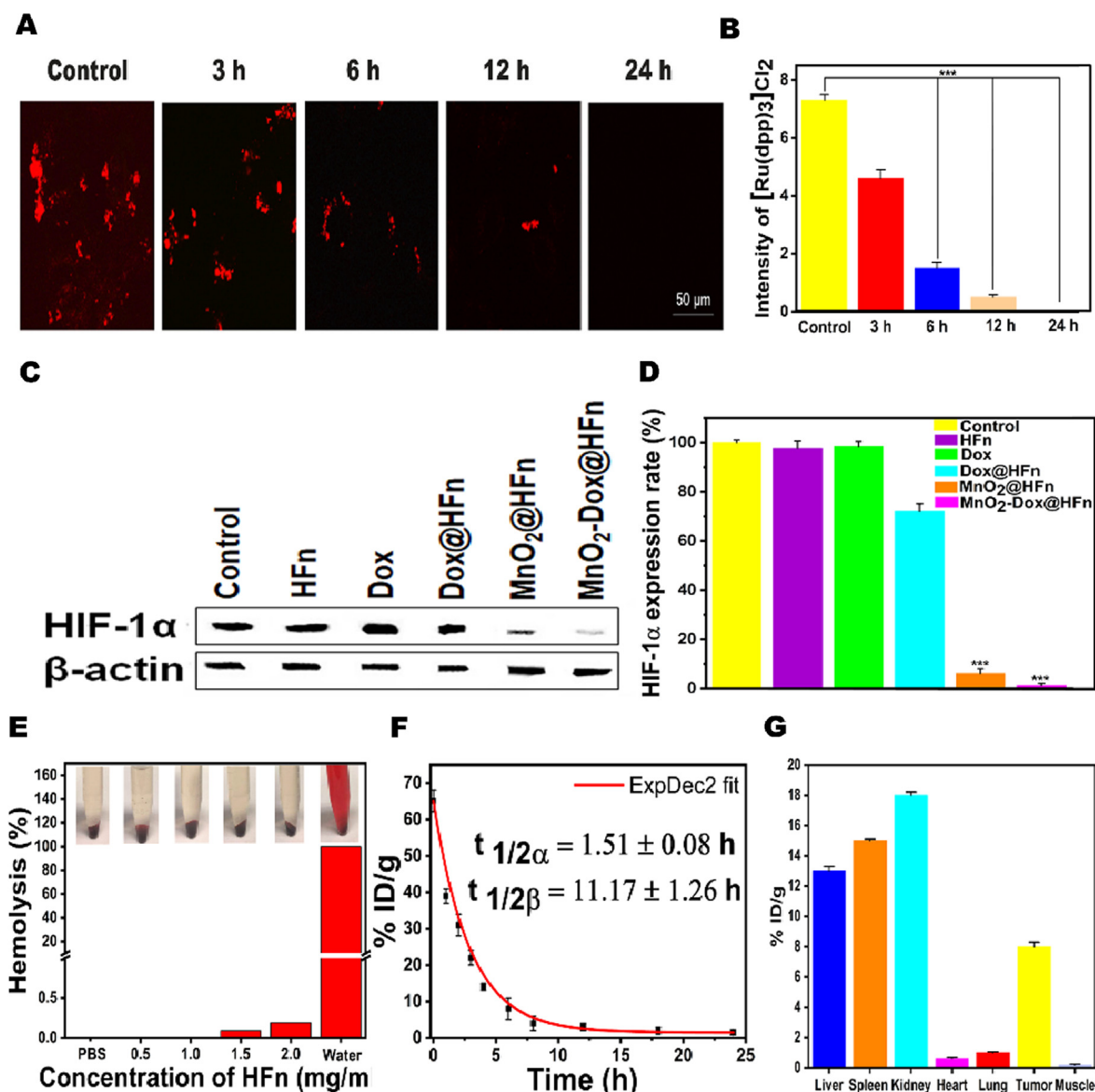


Fig. 5. (A) Intracellular hypoxia assessed by [Ru(dpp)₃]Cl₂ fluorescence probe at different incubation times of MnO₂-Dox@HFn. (B) Quantitative evaluation of fluorescence intensity according to A (***) $p < 0.001$ vs control). (C) Western blot analysis of HIF-1 α expression in HeLa cells under different treatments. β -actin was used as a loading control. (D) Quantitative detection of HIF-1 α based on C (***) $p < 0.001$ vs control). (E) Hemolytic activity of MnO₂-Dox@HFn using PBS as reference and water as a 100% hemolysis value. (F) Blood circulation curve of MnO₂-Dox@HFn based on Mn quantification by AAS. (G) Tissue biodistribution of MnO₂-Dox@HFn in mice after intravenous injection.

firmly that MnO₂-Dox@HFn nanozyme successfully alleviates the hypoxia status of cancer cells.

Hemocompatibility, pharmacokinetic and biodistribution evaluation *in vivo*

To determine the potential toxicity of MnO₂-Dox@HFn during systemic circulation for *in vivo* study, hemolysis behavior was first investigated. No obvious hemolytic effect was observed even at the highest concentration of MnO₂-Dox@HFn (2 mg mL⁻¹) as does PBS (Fig. 5E), confirming the good hemocompatibility of MnO₂-Dox@HFn. Then, the blood circulation of MnO₂-Dox@HFn was analyzed in healthy mice by measuring the concentration of Mn in the body via AAS. The concentration of MnO₂-Dox@HFn decreased over time following the two-compartment model with circulating half-lives of $t_{1/2\alpha} = 1.51 \pm 0.08$ h and $t_{1/2\beta} = 11.17 \pm 1.26$ h (Fig. 5F). This long circulation time of MnO₂-Dox@HFn in the blood allows them

to accumulate in the tumor site via the EPR effect. To confirm this hypothesis, the tissue biodistribution of MnO₂-Dox@HFn was examined. As can be seen in Fig. 5G, a high Mn level in the tumor tissue and kidneys was found, indicating that MnO₂-Dox@HFn was accumulated in the tumor region and excreted with renal clearance.

In vivo tumor imaging by MRI

To demonstrate the tumor-specific selectivity of MnO₂-Dox@HFn as a potent MRI T1-shortening agent, the nanomaterial was injected intratumorally and also into the muscle tissue on the opposite side. A remarkable brighter appearance was observed in the tumor region after only 20 min post-injection (Fig. 6A), the time for the nanozyme shell to dissociate, permitting real-time evaluation of the tumor. However, the non-tumor area with the same dose of MnO₂-Dox@HFn injected did not show any obvious

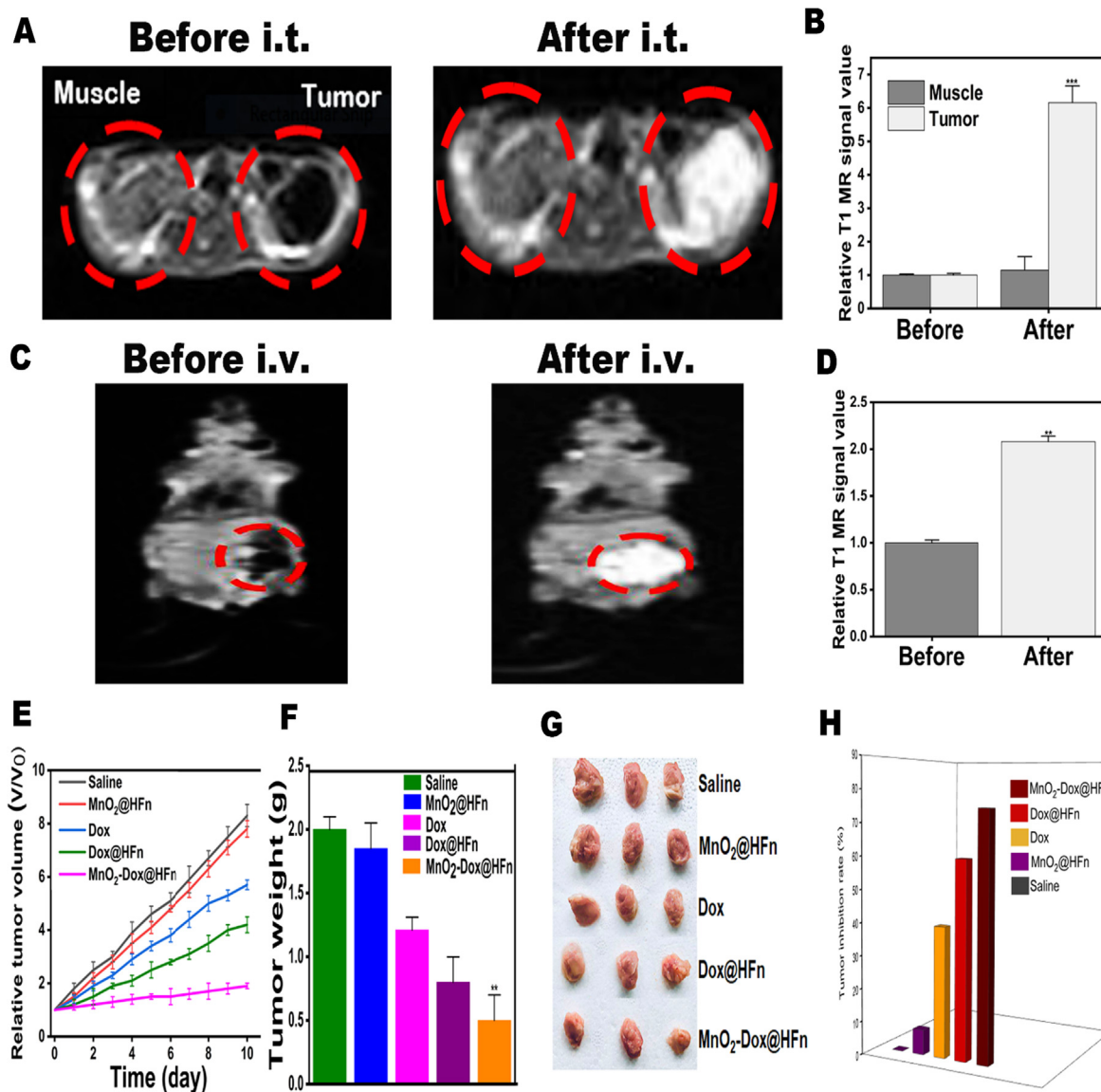


Fig. 6. T1-MR images of 4 T1 tumor-bearing mice through (A) local and (C) intravenous injection of MnO₂-Dox@HFn. (B) and (D) Quantification of T1-MRI signals via intratumorally and intravenously injection according to A and C, respectively (n = 3; **p < 0.01, ***p < 0.001 vs before). (E) Tumor growth in various treated groups (n = 3). (F) Tumor weight at the endpoint of the study (**p < 0.01). (G) Representative tumor images were harvested from mice after the treatment. (H) Tumor growth inhibition rate.

T1 signal (Fig. 6B). This result demonstrated the selective delivery of MnO₂-Dox@HFn within cancer tissues only, leaving normal tissue unharmed. Consecutively, after 6 h post-intravenous injection of MnO₂-Dox@HFn, the tumor site was strongly illuminated (Fig. 6C-D), further indicating its precise tumor distinction and cancer accumulation.

In vivo study for Anti- tumor activity

Having proved the specific tumor-targeting ability of MnO₂-Dox@HFn, their potential *in vivo* anti-tumor activity was studied. The variation trend of tumor volume during the treatment is shown in Fig. 6E. According to the tumor weight and photographs of tumor tissues excised after the treatment (Fig. 6F-G), MnO₂-Dox@HFn treated group showed the greatest therapeutic activity with the highest tumor inhibition rate of 78.5% (Fig. 6H). The enhancement anti-tumor ability of MnO₂-Dox@HFn compared to Dox@HFn could be explained by the effect of MnO₂ to relieve tumor hypoxia. Since the chemotherapeutic activity of Dox has been demonstrated to be less effective under hypoxic rather than

in normoxic environment [46,47]. Additionally, the antitumor efficacy was detected by TUNEL assay and H&E staining. The results are consistent with the *in vitro* study, which shows the unsatisfied tumor cell killing capacity of free Dox, and improved when encapsulated within the HFfn nanocage (Dox@HFfn) for the tumor-targeting delivery (Fig. 7A). Besides, the weak anti-tumor ability of MnO₂@HFfn could be explained by the free radicals produced from MnO₂, leading to cancer cell damage. Contrarily, an obvious cell apoptosis signal was noticed in the tumor tissue treated with MnO₂-Dox@HFfn nanozyme. Below, the histological analysis after H&E staining from MnO₂-Dox@HFfn-treated mice showed their potentiality to suppress tumor growth. The ability of MnO₂-Dox@HFfn to regulate hypoxic conditions in tumor tissue was also evaluated through immunohistochemistry of HIF-1α and Ki-67. MnO₂-Dox@HFfn-treated mice decreased until 91.06% the expression of HIF-1α as compared with the saline group (Fig. 7B), due to their *in vivo* tumor oxygenation ability. Benefiting from this down-regulation of HIF-1α protein, cell migration was inhibited as proved by the suppression of Ki-67 proliferative cells in the MnO₂-Dox@HFfn group, showing remarkable interest to prevent

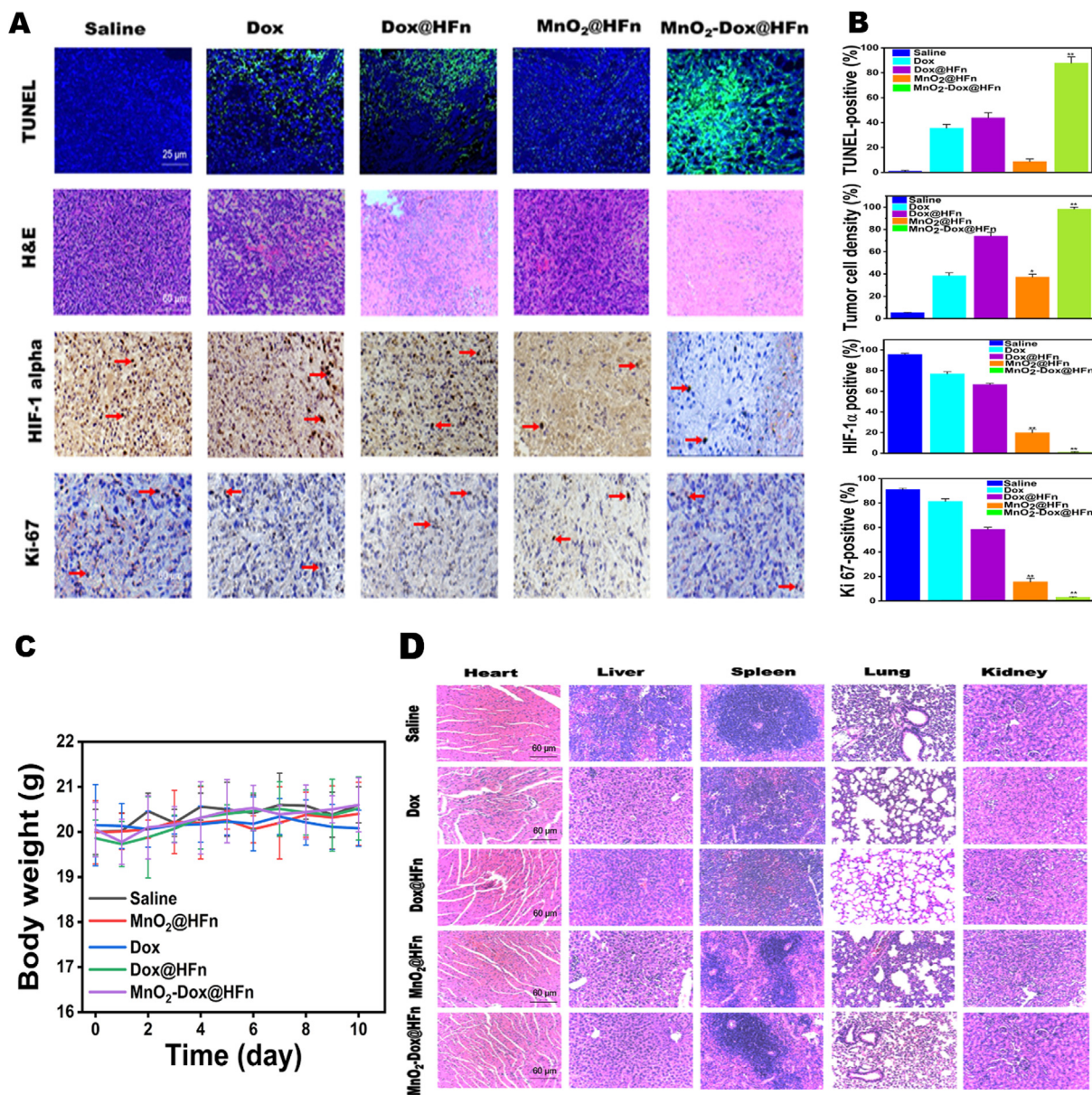


Fig. 7. Histological study and biosafety. (A) Immunofluorescence analysis of the resected tumor tissues. TUNEL assays evaluating apoptotic cells where nuclei were stained blue (DAPI staining), and apoptotic cells were stained green (TUNEL staining). H&E staining assay. Immunohistochemical staining of HIF-1 α and Ki-67 (brown-stained cells). (B) Quantification of the positive cell rate in the TUNEL apoptosis; H&E; HIF-1 α and Ki-67 staining (n = 3; **p < 0.01 or *p < 0.05 vs free Dox). (C) Body weight during the whole treatment. (D) H&E staining of major organs tissue sections.

tumor metastasis. Taken together, the expected duration of a single dose therapeutic effect of MnO₂-Dox@HFfn might be around 48 h as deduced from the *in vitro* cytotoxicity study. The cellular oxygen generation also demonstrated that after 24 h incubation with MnO₂-Dox@HFfn, the hypoxia was completely relieved; as well as confirmed the western blot analysis during which the HIF-1 α expression level of cells treated with MnO₂-Dox@HFfn for 48 h was significantly down-regulated. Further proved with the *in vivo* study, while the mouse was treated with MnO₂-Dox@HFfn every 2 days during the treatment period and significant therapeutic activity was observed.

Biosafety

Finally, the biosafety of MnO₂-Dox@HFfn was evaluated through the body weight and the H&E staining of major organs. As shown in Fig. 7C, there were no significant changes in the body weight in all

tested groups. Likewise, based on the histological H&E analysis, no noticeable toxicity nor abnormality was observed in all of the groups after the whole therapeutic period (Fig. 7D). These results engaged the excellent biocompatibility and good safety profile of MnO₂-Dox@HFfn.

Conclusions

In summary, the MnO₂-Dox@HFfn nanozyme was formulated in a facile manner through pH change. Compared to other approaches, this study highlights several important concepts that may promote the exploration of theranostic nanozyme for cancer therapy. First, MnO₂-Dox@HFfn showed peroxidase and catalase-like activity to generate superior O₂ indispensable to overcome tumor hypoxia. Secondly, the nanozyme provides tumor-specific MR imaging-guided cancer treatment. Thirdly, MnO₂-Dox@HFfn exhibited potential anti-cancer ability, which could be attributed

to the intracellular oxidase damage from peroxidase-activity and the chemotherapeutic effect of Dox. Lastly, MnO₂-Dox@HF_n reduces malignant cell proliferation *in vivo*, which is beneficial for clinical tumor therapy.

Compliance with Ethics Requirements

Animal study Ethics statement: All experiments involving animals were conducted according to the ethical policies and regulations provided by the Guide for Care and Use of Laboratory Animals, approved by China Pharmaceutical University (Approval no. 2019-1205).

CRediT authorship contribution statement

Hanitrarimalala Veroniaina: Conceptualization, Methodology, Software, Data curation, Writing - original draft. **Zhenghong Wu:** Supervision, Visualization, Investigation. **Xiaole Qi:** Supervision, Validation, Writing - review & editing.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jare.2021.02.004>.

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