

SUPPLEMENTARY INFORMATION

Bottom-up Evolution of Perovskite Clusters into High-activity Rhodium Nanoparticles toward Alkaline Hydrogen Evolution

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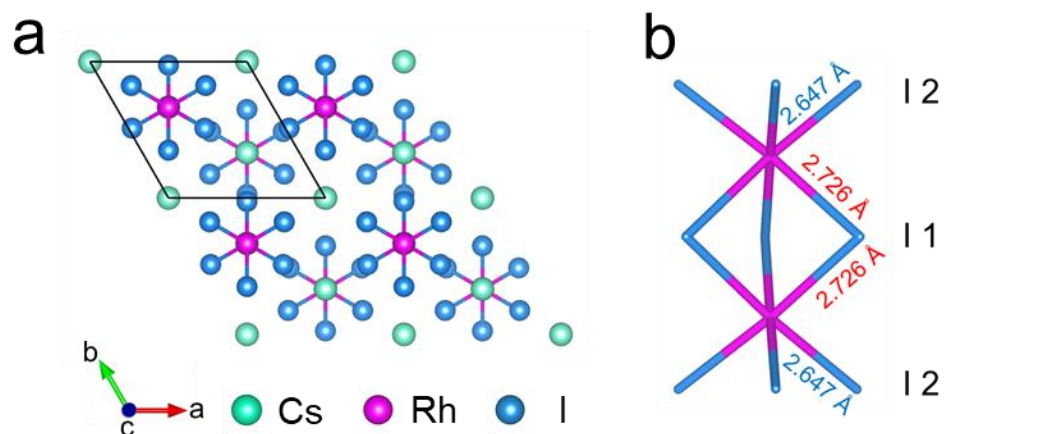
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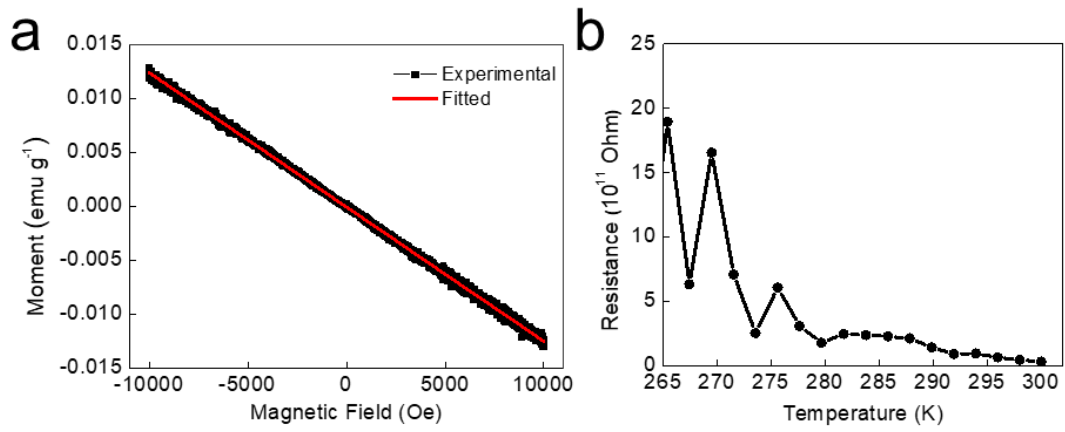
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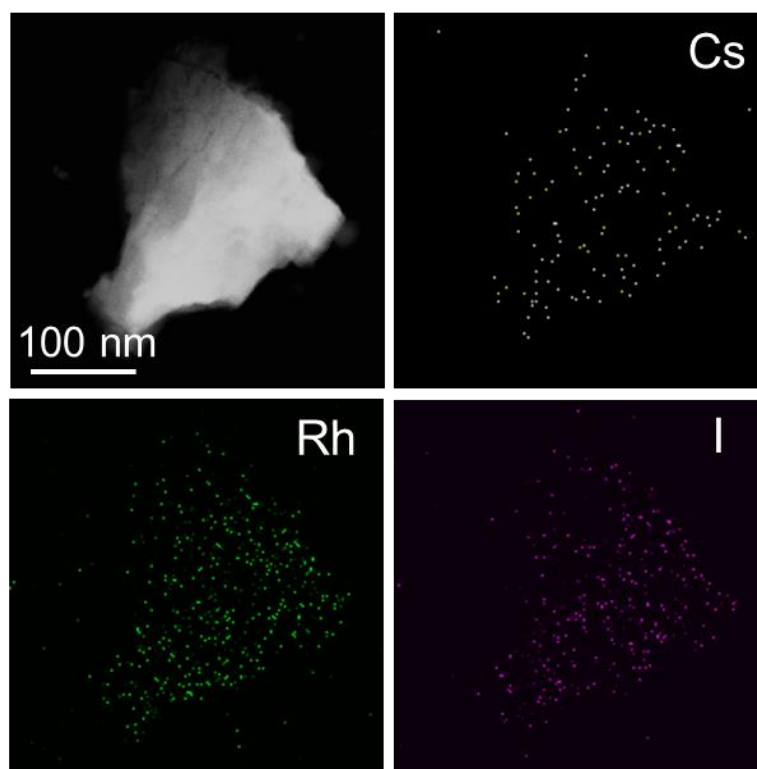
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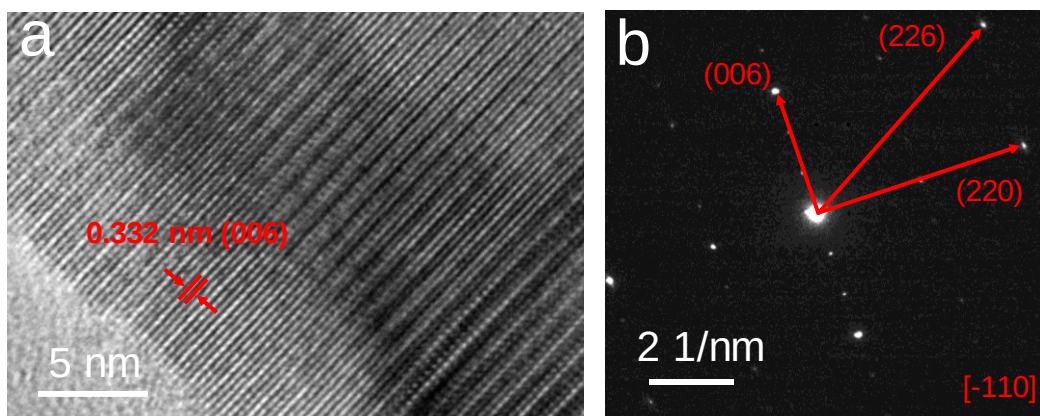
Supplementary Fig. 1 **a** Atomic structure of $\text{Cs}_3\text{Rh}_2\text{I}_9$. **b** The detail of $[\text{Rh}_2\text{I}_9]^{3-}$ bi-octahedra.



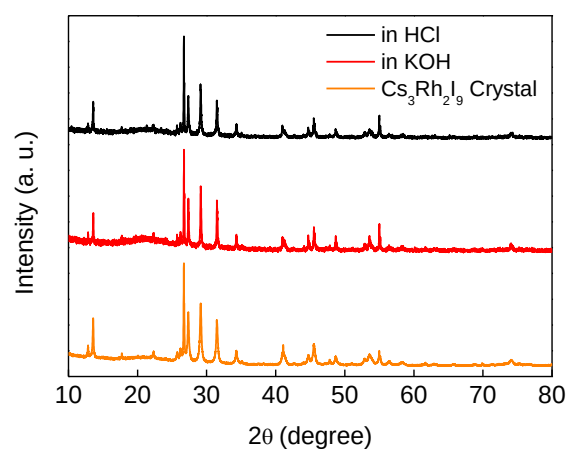
Supplementary Fig. 2 The curves of **a** moment-magnetic field at 3 K and **b** resistance-temperature of $\text{Cs}_3\text{Rh}_2\text{I}_9$ single crystal.



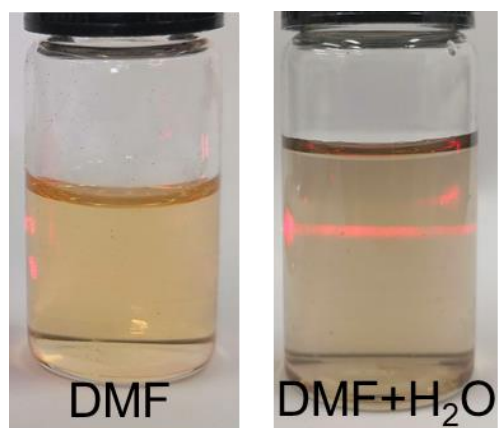
Supplementary Fig. 3 TEM-EDX mapping images of $\text{Cs}_3\text{Rh}_2\text{I}_9$.



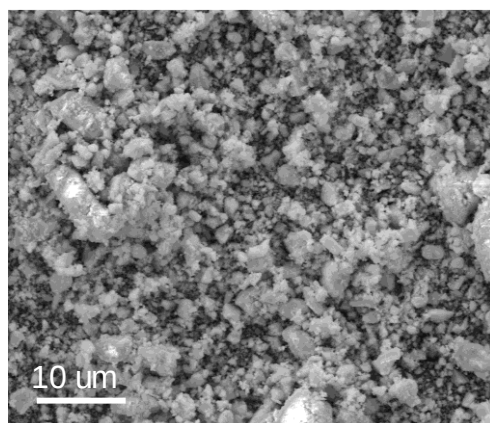
Supplementary Fig. 4 **a** HRTEM of $\text{Cs}_3\text{Rh}_2\text{I}_9$. **b** The corresponding selected area electron diffraction image.



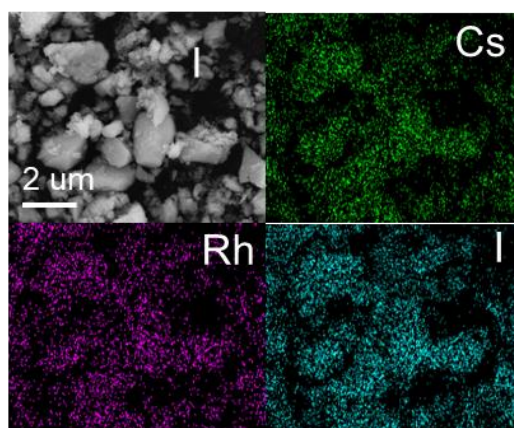
Supplementary Fig. 5 The XRD patterns of Cs₃Rh₂I₉ after immersion in 1.0 M HCl or 1.0 M KOH for 7 days.



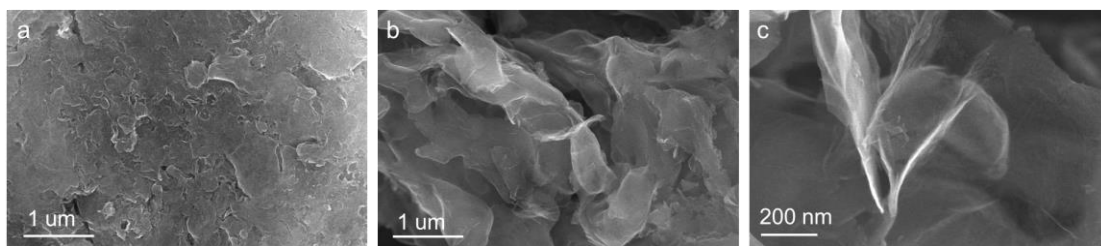
Supplementary Fig. 6 The photographs of $\text{Cs}_3\text{Rh}_2\text{I}_9$ in DMF without Tyndall effect (left) and $\text{Cs}_3\text{Rh}_2\text{I}_9$ in $\text{H}_2\text{O}/\text{DMF}$ with Tyndall effect (right).



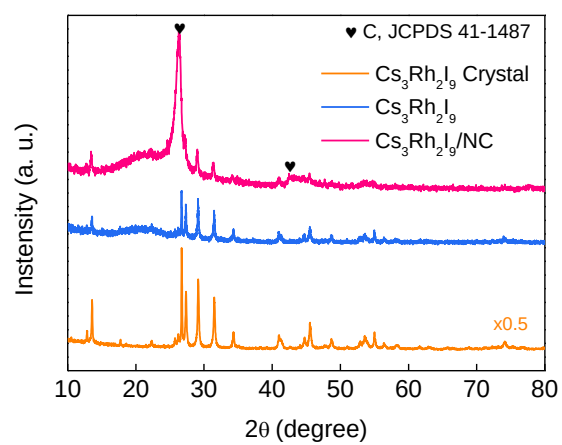
Supplementary Fig. 7 SEM image of precipitated Cs₃Rh₂I₉.



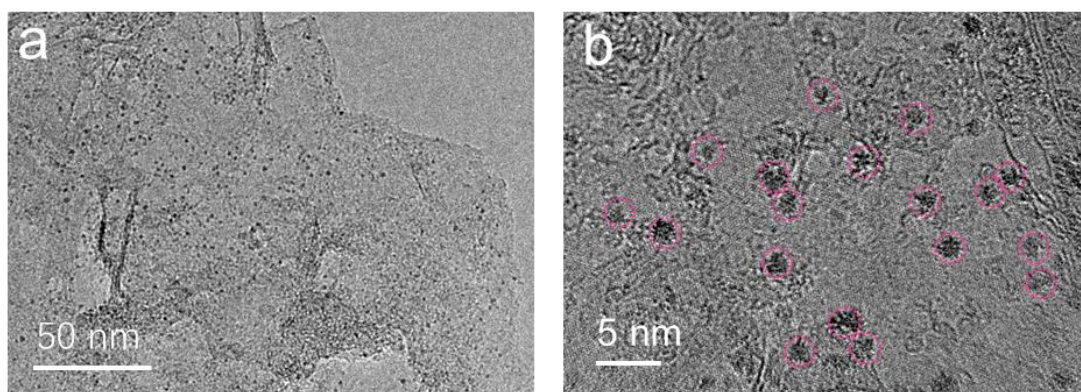
Supplementary Fig. 8 SEM-EDX mapping images of precipitated $\text{Cs}_3\text{Rh}_2\text{I}_9$.



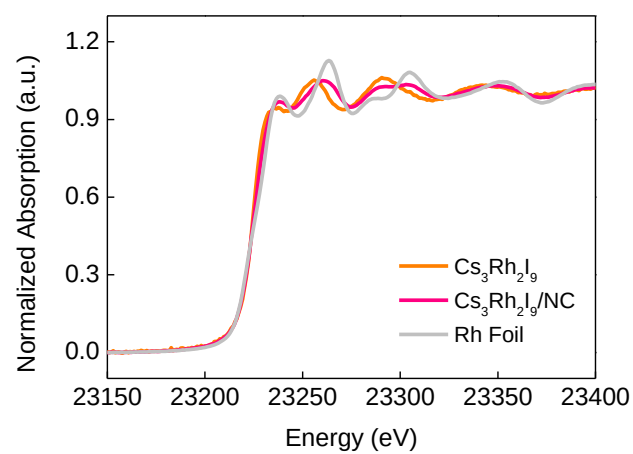
Supplementary Fig. 9 SEM images of Cs₃Rh₂I₉/NC.



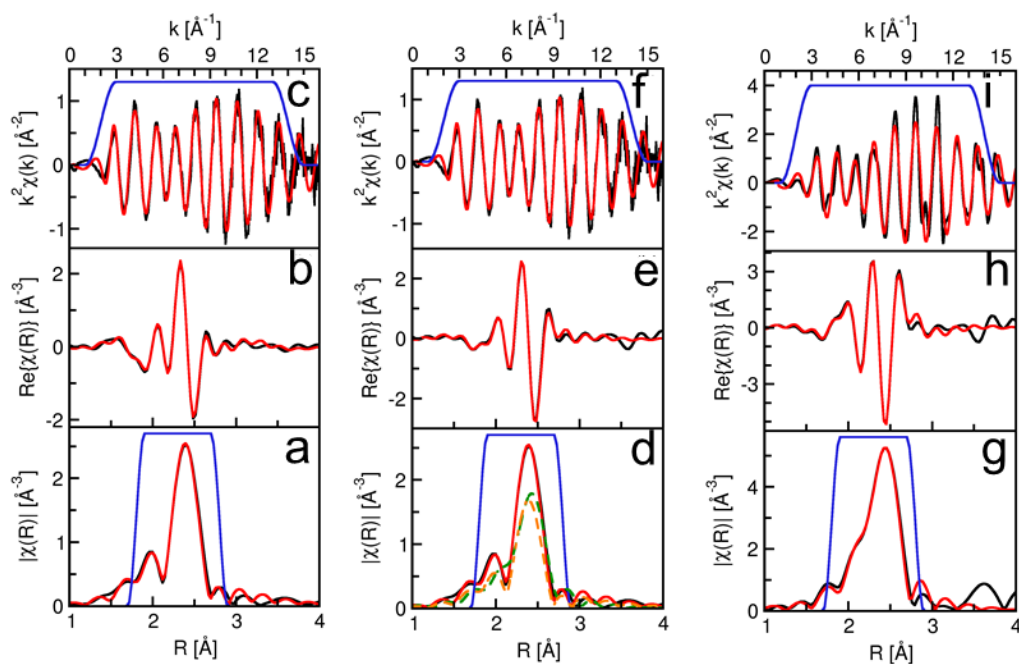
Supplementary Fig. 10 XRD patterns of $\text{Cs}_3\text{Rh}_2\text{I}_9$ crystal, $\text{Cs}_3\text{Rh}_2\text{I}_9$ and $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$.



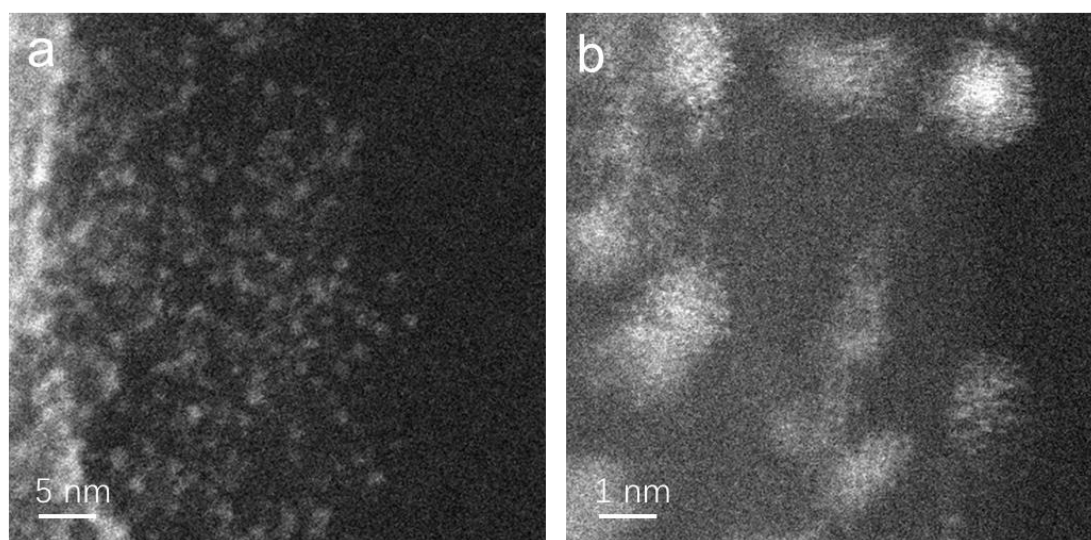
Supplementary Fig. 11 **a** TEM image of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$. **b** HRTEM image with the red circles showing the $\text{Cs}_3\text{Rh}_2\text{I}_9$ clusters.



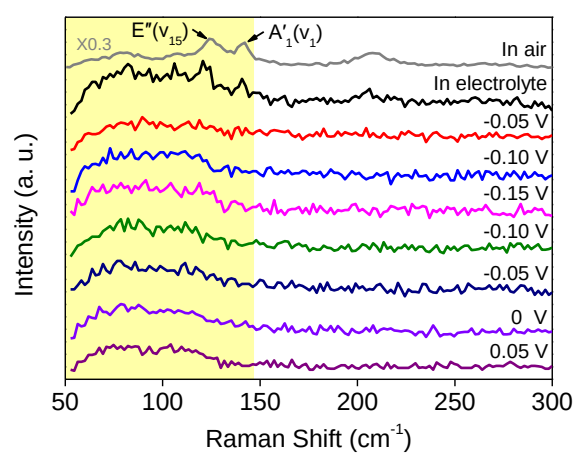
Supplementary Fig. 12 The X-ray absorption near edge structure spectra of Rh K-edge for $\text{Cs}_3\text{Rh}_2\text{I}_9$, $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$, and Rh foil.



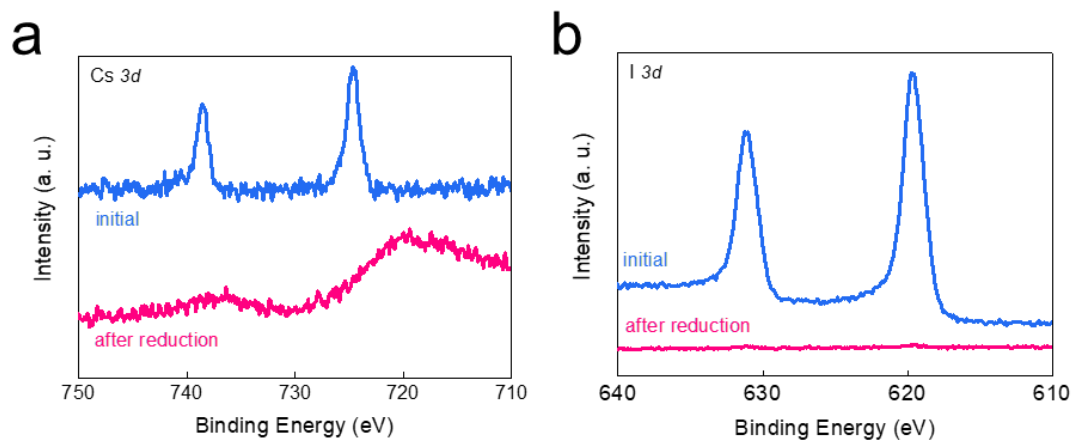
Supplementary Fig. 13 EXAFS data and fitting results. **a-c** Rh edge of $\text{Cs}_3\text{Rh}_2\text{I}_9$ sample showing magnitude of FT, real part of FT and weighted $\chi(k)$. **d-f** Rh edge of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ sample showing magnitude of FT, real part of FT and weighted $\chi(k)$. **g-i** Rh edge of Rh foil sample showing magnitude of FT, real part of FT and weighted $\chi(k)$. In all figures, the data is shown in black, the fit in red and the FT and fitting windows in blue.



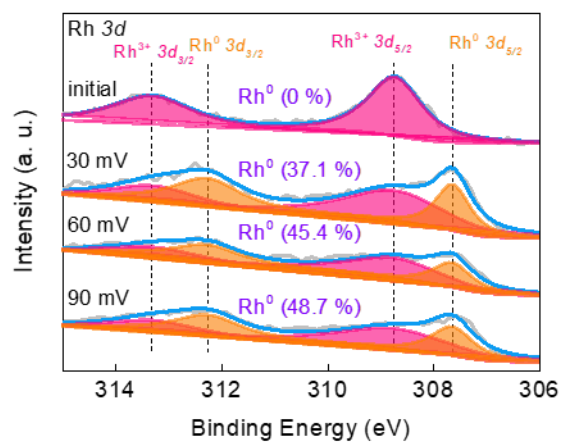
Supplementary Fig. 14 **a** HAADF-STEM of Cs₃Rh₂I₉/NC and **b** decomposed Cs₃Rh₂I₉/NC. The Cs₃Rh₂I₉ nano-clusters are unstable under the high-energy measurement.



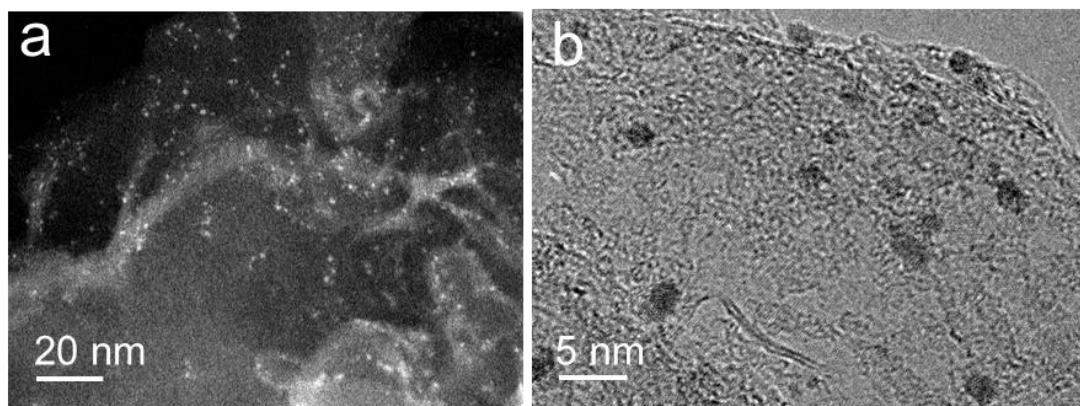
Supplementary Fig. 15 The in-situ Raman spectra of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ under the CV measurement.



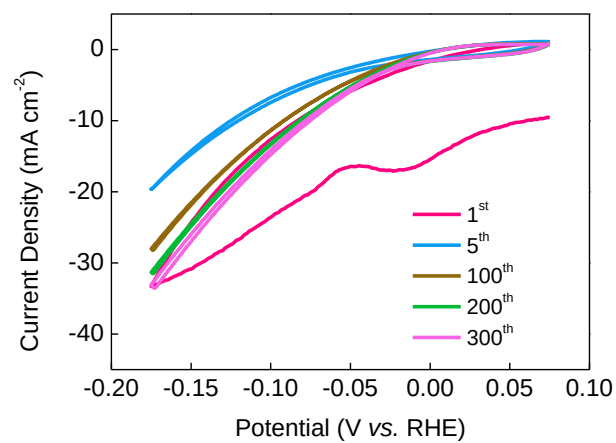
Supplementary Fig. 16 High resolution XPS spectra of **a)** Cs 3*d* and **b)** I 3*d* for $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ before and after reduction.



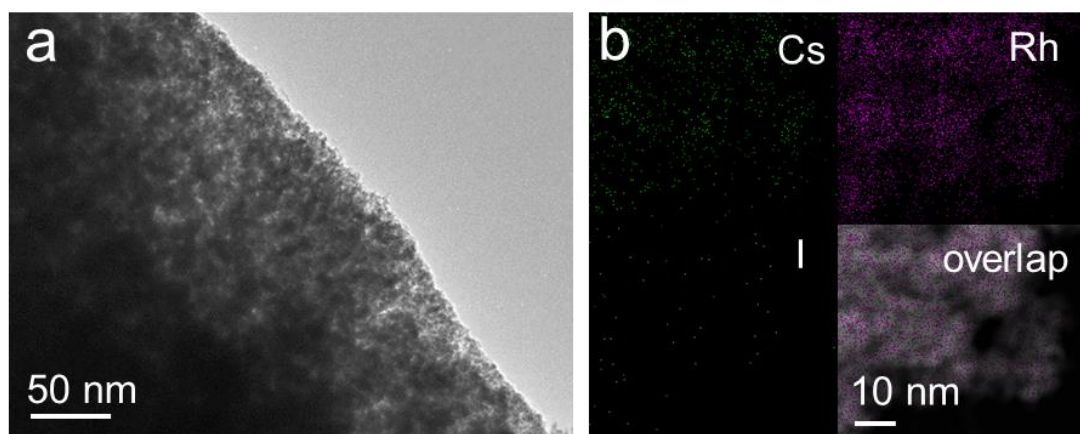
Supplementary Fig. 17 The ex-situ XPS spectra for Cs₃Rh₂I₉/NC of different overpotential under the CV measurement.



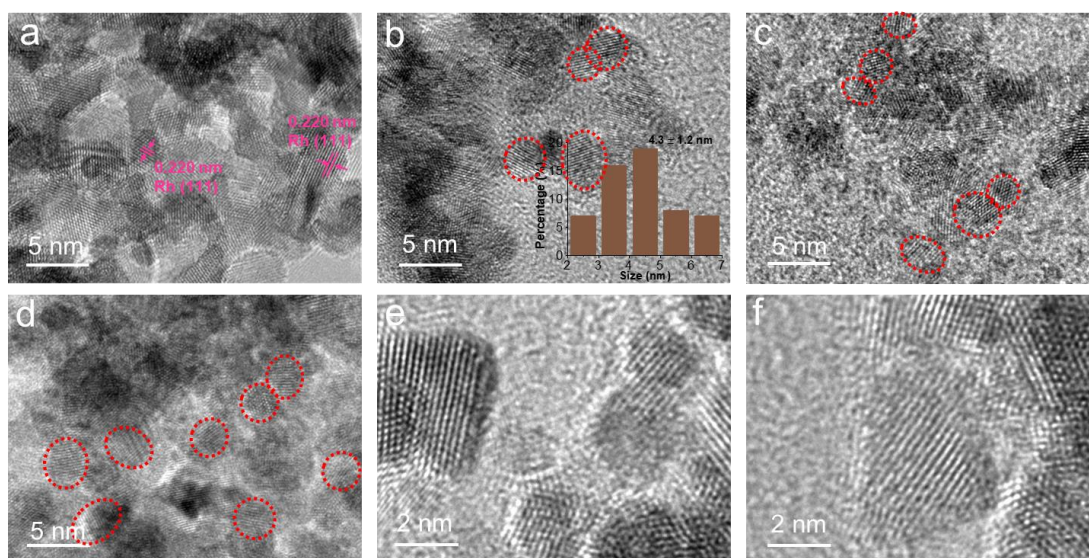
Supplementary Fig. 18 **a** HAADF image and **b** HRTEM of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ after reduction.



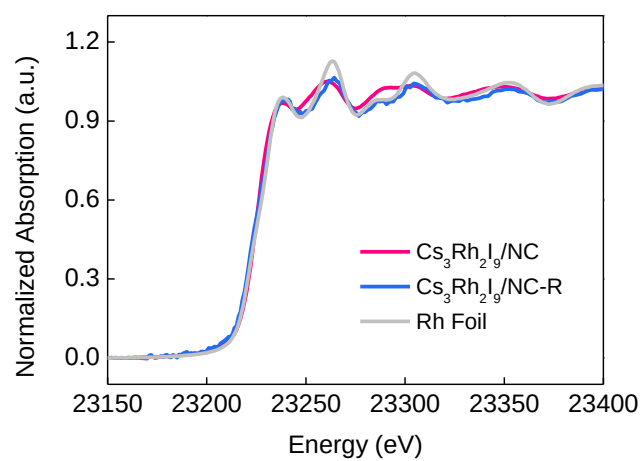
Supplementary Fig. 19 The CV curves from 1st to 300th cycle at 100 mV s⁻¹ for Cs₃Rh₂I₉.



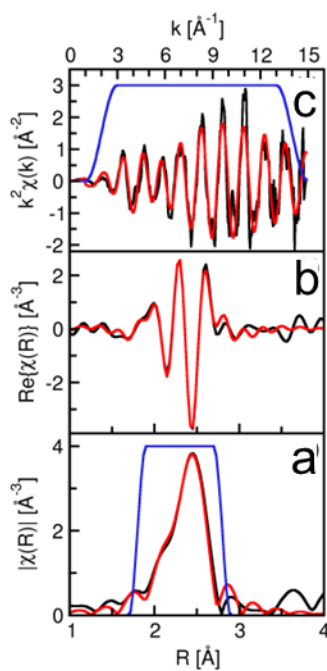
Supplementary Fig. 20 **a** TEM, **b** EDX images of $\text{Cs}_3\text{Rh}_2\text{I}_9$ after reduction ($\text{Cs}_3\text{Rh}_2\text{I}_9\text{-R}$).



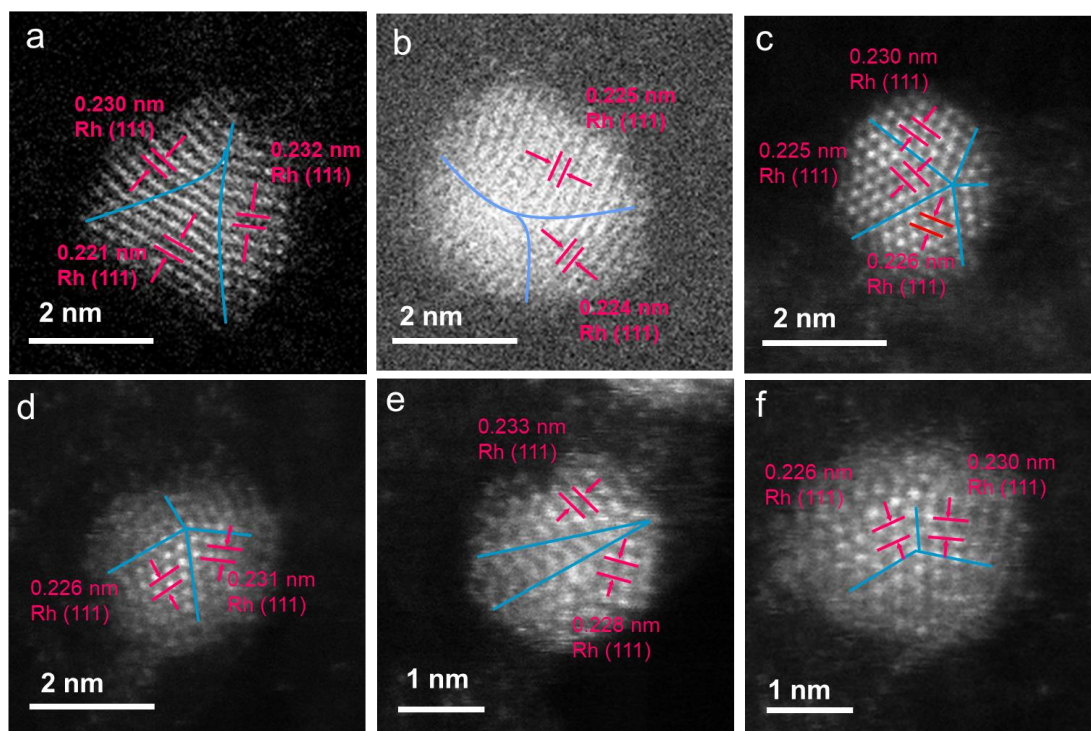
Supplementary Fig. 21 HRTEM of $\text{Cs}_3\text{Rh}_2\text{I}_9\text{-R}$. The inset in **b** shows particle size distribution.



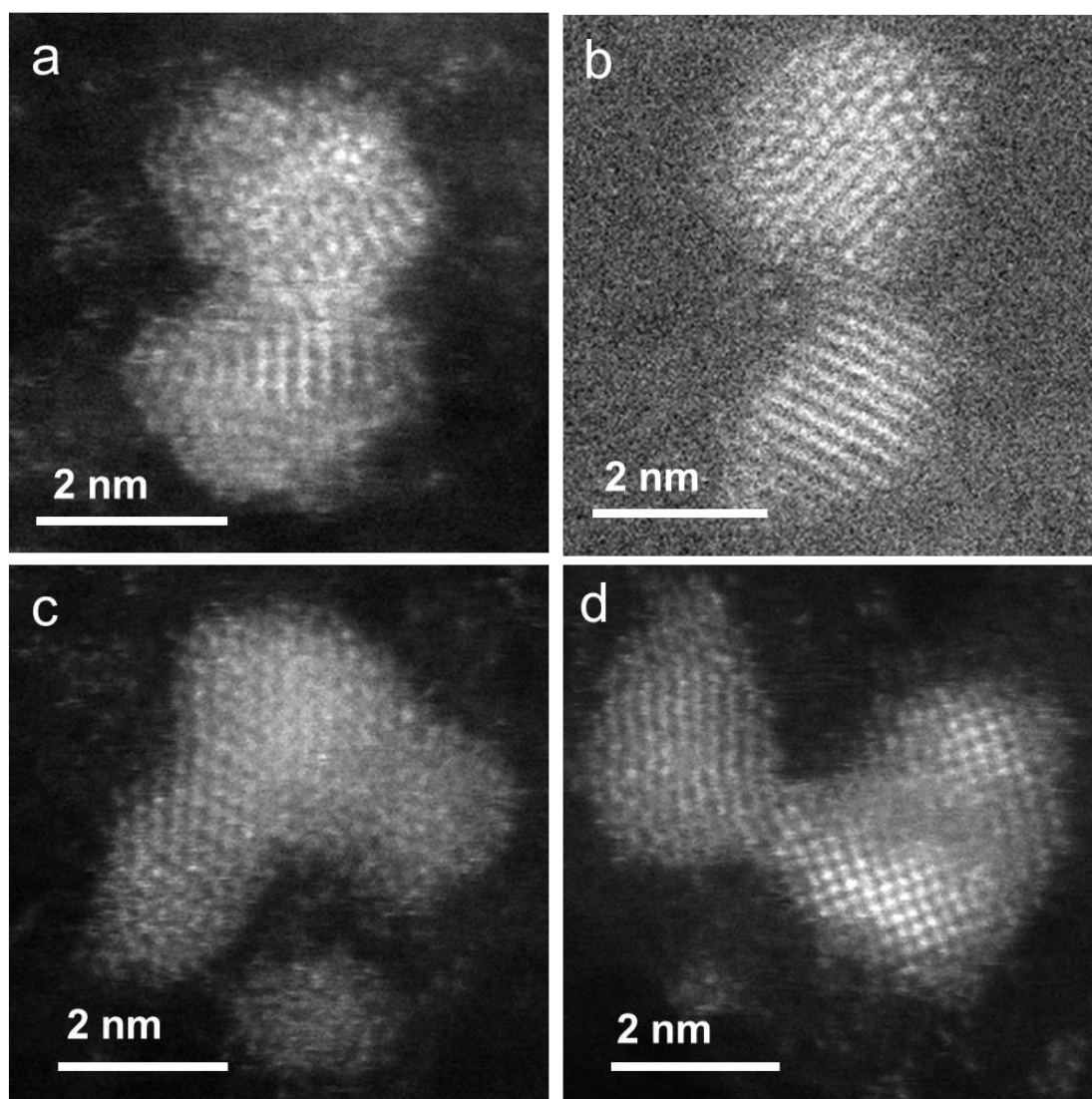
Supplementary Fig. 22 The X-ray absorption near edge structure spectra of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$, $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$, and Rh foil at Rh K-edge.



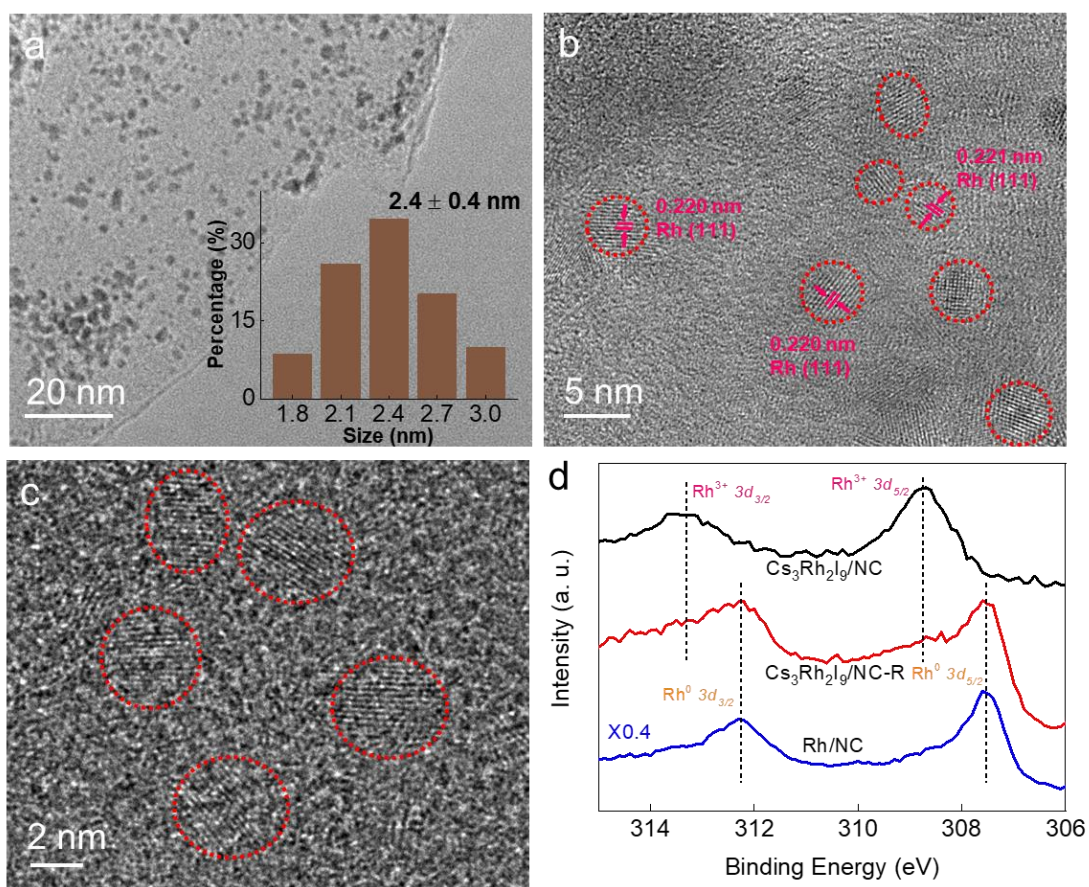
Supplementary Fig. 23 EXAFS data and fitting results. **a-c** Rh edge of reduced $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ sample showing magnitude of FT, real part of FT and weighted $\chi(k)$. In all figures, the data is shown in black, the fit is shown in red, and the FT and fitting windows are shown in blue.



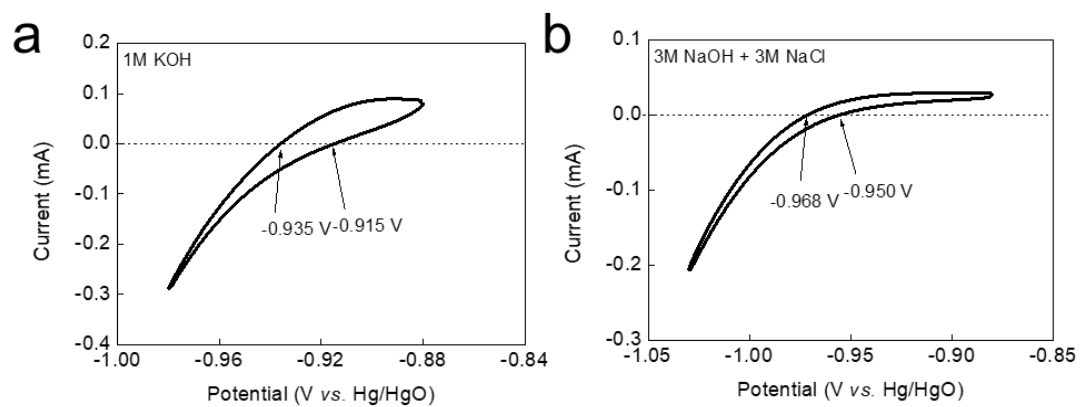
Supplementary Fig. 24 HAADF-STEM images of reduced $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ ($\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$). And these images show that the $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$ is composed of twinned Rh nanoparticles.



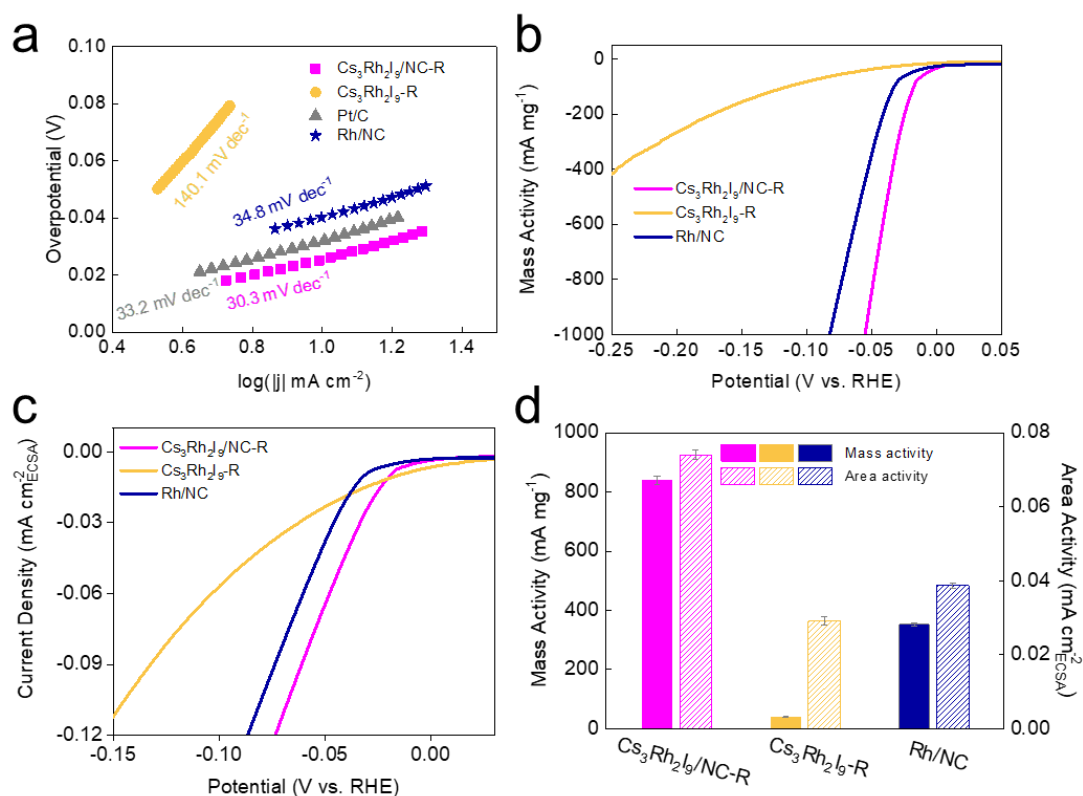
Supplementary Fig. 25 HAADF-STEM images of adjacent Rh particles in reduced $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ ($\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$).



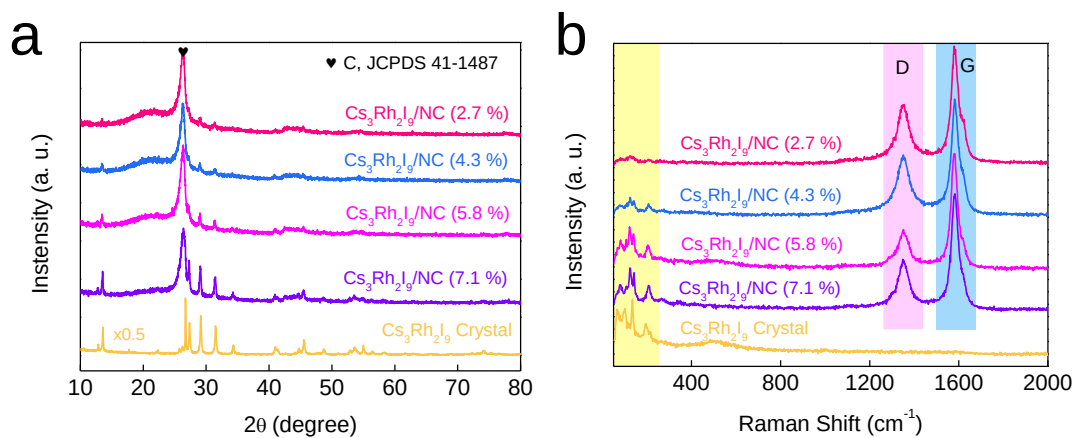
Supplementary Fig. 26 **a** TEM images of Rh/NC. The inset shows the Rh particle size distribution. **b** and **c** HRTEM of Rh/NC. The red circles show the Rh particle. **d** XPS spectra of Rh 3d for $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$, $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$ and Rh/NC.



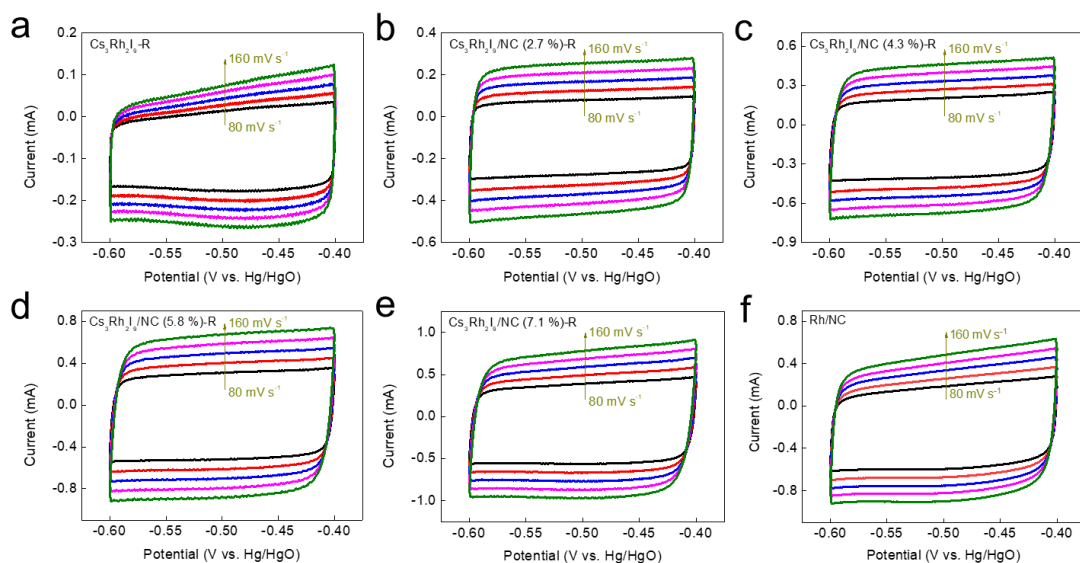
Supplementary Fig. 27 CV curves of RHE calibration in 1.0 M KOH **(a)** and 3.0 M NaOH+3.0 M NaCl **(b)**.



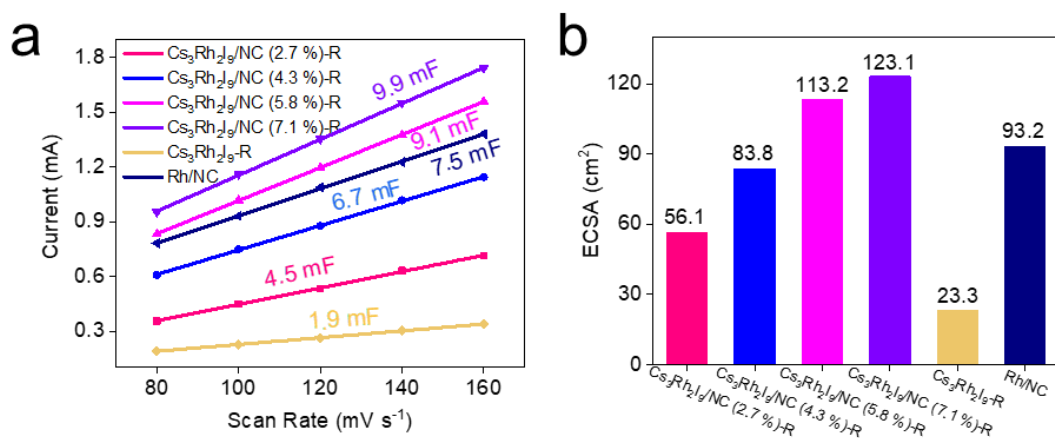
Supplementary Fig. 28 The HER activity of different electrocatalysts on rotating GCE (1600 rpm) in 1.0 M KOH. **a** Tafel plots corresponding to the curves in Figure 5a. **b** Mass activity normalized to the mass of Rh. **c** Area activity normalized to the ECSA. **d** Comparison of mass activity and area activity at the overpotential of 50 mV. The catalyst loading amount on GCE is 0.764 mg cm⁻². All data shows the mean and standard deviation through three repeated measurements. For Rh-based samples, the calculated Rh loading amounts on GCE are 0.045, 0.090 and 0.053 mg cm⁻² for Cs₃Rh₂I₉/NC-R, Cs₃Rh₂I₉-R, and Rh/NC, respectively.



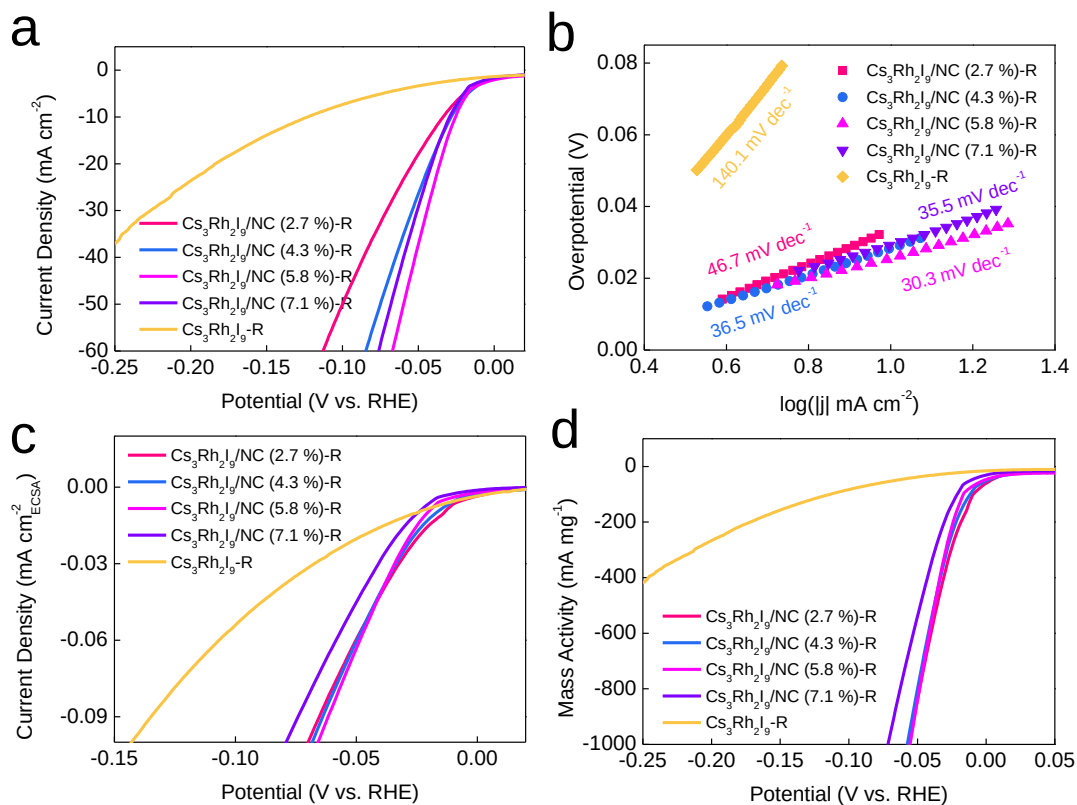
Supplementary Fig. 29 a XRD patterns and **b** Raman spectra of $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ with different Rh contents.



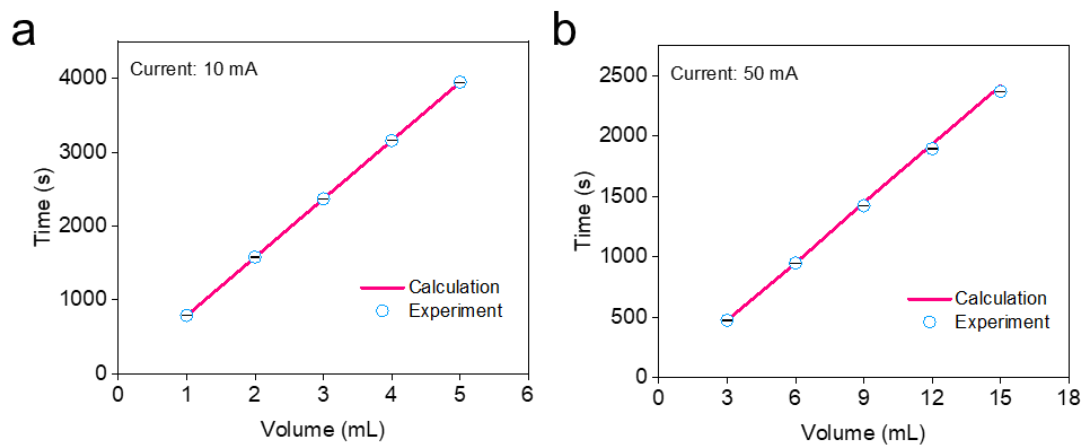
Supplementary Fig. 30 CV curves of $\text{Cs}_3\text{Rh}_2\text{I}_9\text{-R}$ (a), $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ (2.7 wt. %)-R (b), $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ (4.3 wt. %)-R (c), $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ (5.8 wt. %)-R (d), $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC}$ (7.1 wt. %)-R (e), and Rh/NC (f) at different scan rates in 1.0 M KOH.



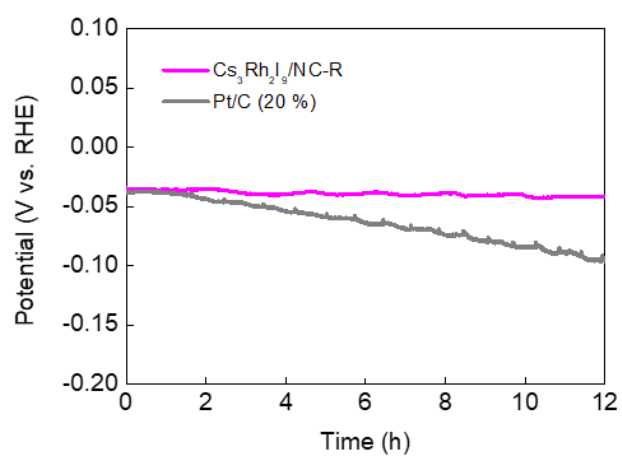
Supplementary Fig. 31 **a** The currents (-0.50 V vs. Hg/HgO) as a function of scan rate, and **b** the electrochemical surface area (ECSA) for various Rh-based samples.



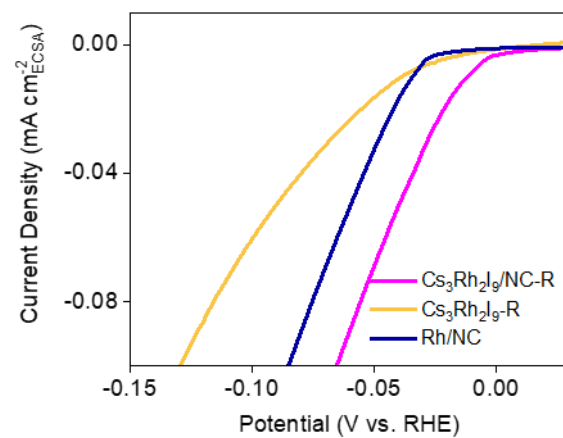
Supplementary Fig. 32 The HER activity of different Rh-based samples coated on rotating GCE (1600 rpm) in 1.0 M KOH. **a** LSV curves. **b** Tafel plots. **c** The area activity normalized to the ECSA. **d** Mass activity normalized to the mass of Rh. The catalyst loading amount on GCE is 0.764 mg cm^{-2} .



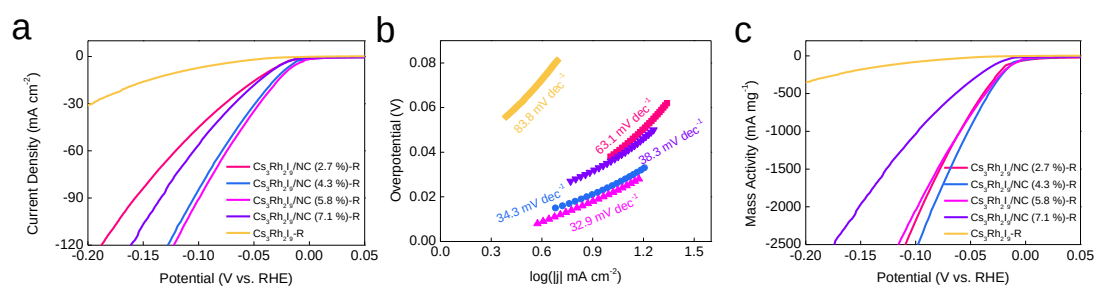
Supplementary Fig. 33 Total amount of H_2 produced over time at a current of 10 mA (a) and 50 mA (b) in 1 M KOH for $Cs_3Rh_2I_9/NC-R$, showing the nearly 100% faradaic efficiency.



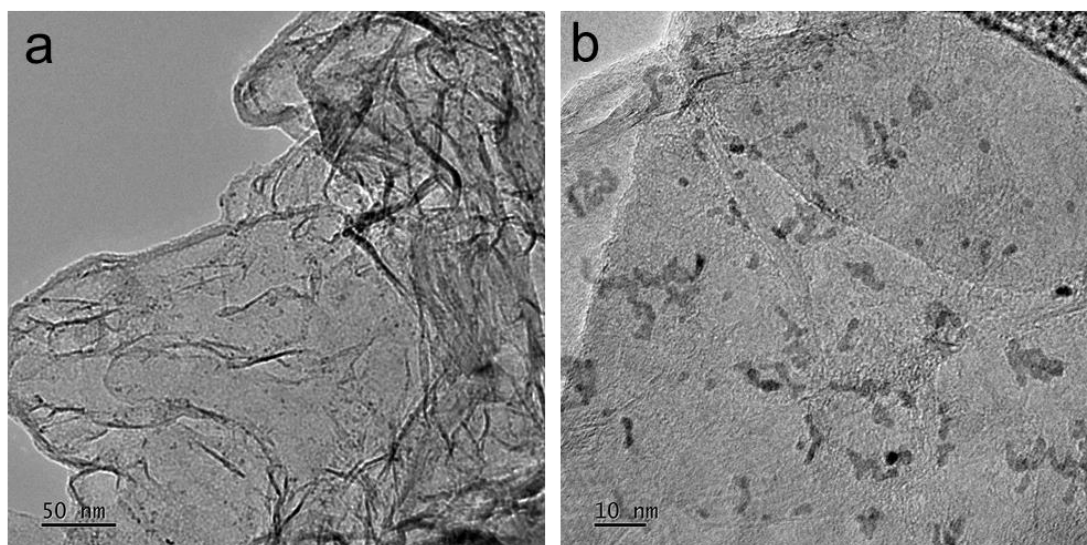
Supplementary Fig. 34 Long-term HER stability at the current density of 10 mA cm^{-2} in 1.0 M KOH.



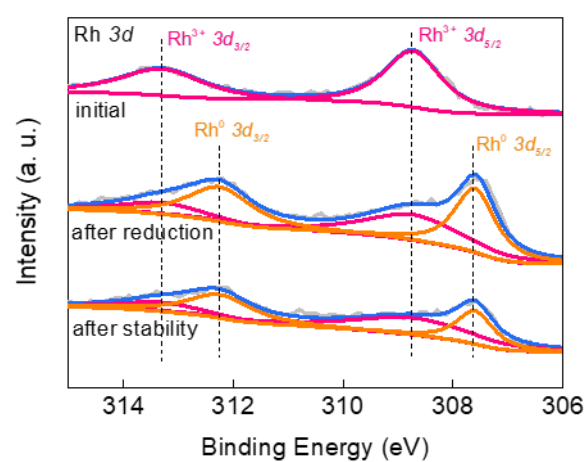
Supplementary Fig. 35 Area activity normalized to ECSA in chlorine-alkali electrolyte for $\text{Cs}_3\text{Rh}_2\text{I}_9/\text{NC-R}$, $\text{Cs}_3\text{Rh}_2\text{I}_9\text{-R}$, and Rh/NC .



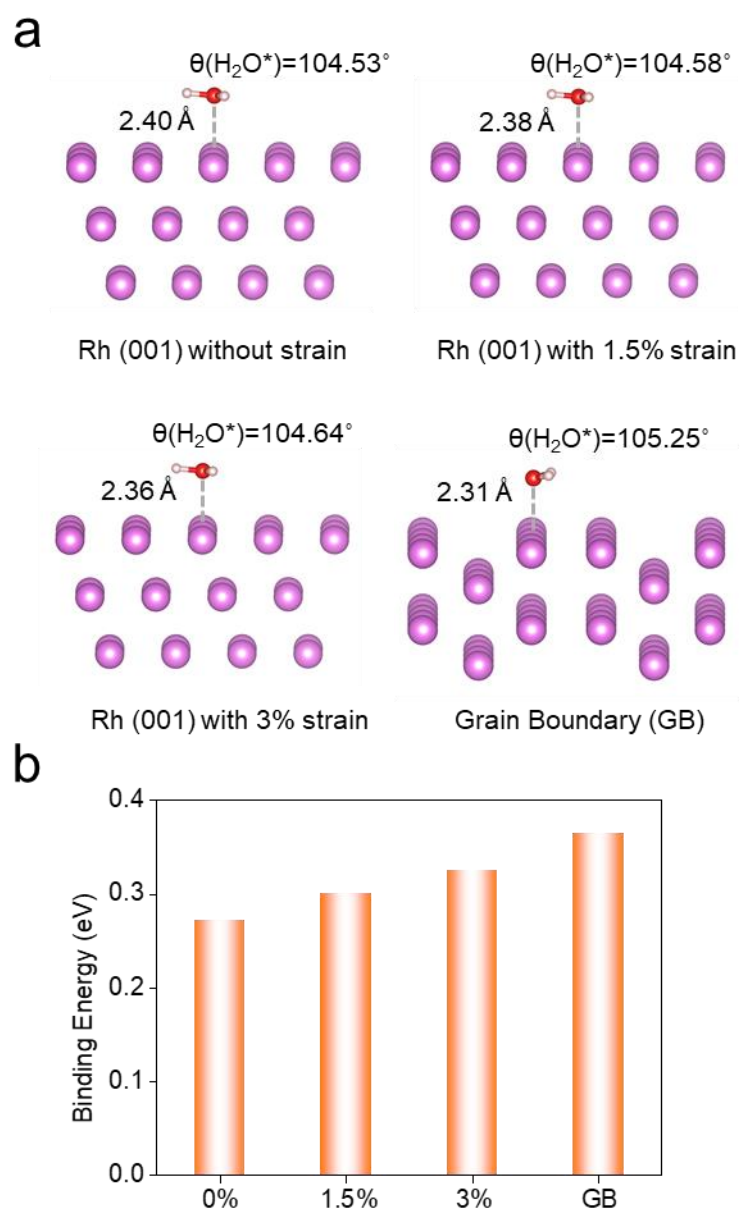
Supplementary Fig. 36 The HER activity of different Rh-based samples coated on rotating GCE (1600 rpm) in chlorine-alkali electrolyte. **a** LSV curves. **b** Tafel plots. **c** Mass activity normalized to the mass of Rh. The catalyst loading amount on GCE is 0.764 mg cm^{-2} .



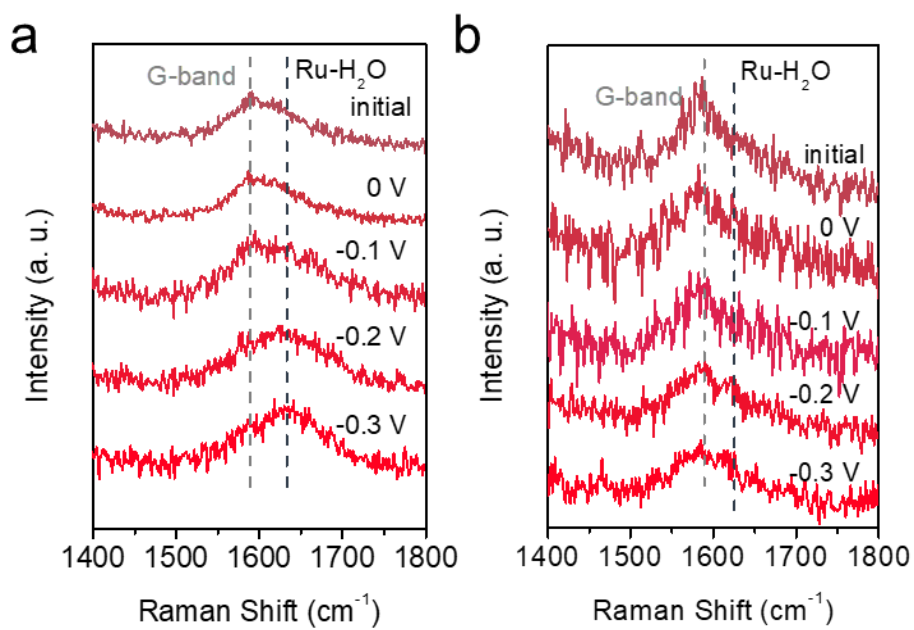
Supplementary Fig. 37 **a** TEM and **b** HRTEM images of Cs₃Rh₂I₉/NC-R after the durable measurement.



Supplementary Fig. 38 XPS spectra of Cs₃Rh₂I₉/NC-R after the durable measurement.



Supplementary Fig. 39 **a** Structural models of adsorbed water molecule on Rh (001), Rh (001) with tensions, and Rh with (110) GB. **b** Binding energy of water molecule.



Supplementary Fig. 40 In situ Raman spectra of interfacial water on Cs₃Rh₂I₉/NC-R (a) and Rh/NC (b).

Supplementary Table 1. Crystallographic data and structural refinement for Cs₃Rh₂I₉.

Chemical formula	Cs ₃ Rh ₂ I ₉
M_r	1746.67
Crystal system, space group	Hexagonal, $P6_3/mmc$
Temperature (K)	298
a, c (Å)	7.9644 (7), 20.0225 (18)
V (Å ³)	1099.9 (2)
Z	2
Radiation type	Mo $K\alpha$
μ (mm ⁻¹)	19.00
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	5410, 403, 322
R_{int}	0.035
$(\sin \theta/\lambda)_{max}$ (Å ⁻¹)	0.595
$R [F^2 > 2\sigma(F^2)]$, $wR (F^2)$, S	0.054, 0.140, 1.15
No. of reflections	403
No. of parameters	19
$\Delta\rho_{max}, \Delta\rho_{min}$ (e Å ⁻³)	1.52, -1.33

Supplementary Table 2. Fractional atomic coordinates, equivalent isotropic displacement parameters ($\text{\AA}^2$) of $\text{Cs}_3\text{Rh}_2\text{I}_9$.

Atom	Wyck	x	y	z	U_{iso}^*/U_{eq}
Cs 1	2 <i>b</i>	0	0	1/4	0.0698 (12)
Cs 2	4 <i>f</i>	2/3	1/3	0.57460 (15)	0.0739 (10)
Rh	4 <i>f</i>	2/3	1/3	0.33522 (13)	0.0547 (10)
I 1	6 <i>h</i>	0.51255 (16)	0.48745 (16)	1/4	0.0592 (9)
I 2	12 <i>k</i>	0.82483 (13)	0.6497 (3)	0.41004 (7)	0.0660 (8)

Supplementary Table 3. The molar ratio of Rh, I and Cs elements from Figure S3.

Element	Line type	K factor	Absorption correction	wt%	wt% sigma	Atomic percentage
Rh	L	1.848	1.00	11.76	0.48	14.25
I	L	1.986	1.00	66.52	0.66	65.36
Cs	L	2.020	1.00	21.73	0.61	20.39
total				100.00		100.00

Supplementary Table 4. Structure parameters extracted from the Rh K-edge EXAFS fittings of Rh foil, Cs₃Rh₂I₉, and Cs₃Rh₂I₉/NC.

Samples	Path	N		σ^2 (Å ²)	ΔE (eV)	R (Å)
Rh foil	Rh–Rh	12		0.0035±0.0005	3.0±0.7	2.682±0.004
Cs ₃ Rh ₂ I ₉	Rh–I	5.0±0.3		0.0041±0.0004	0.3±0.4	2.651±0.003
	Path	Fraction		σ^2 (Å ²)	ΔE (eV)	R (Å)
Cs ₃ Rh ₂ I ₉ /NC*	Rh–I	0.43	±0.13	0.004±0.001	0.5±1.4	2.651
	Rh–Rh	0.57		0.004±0.001		2.682

*The fitting model used for Cs₃Rh₂I₉/NC takes the Rh–Rh path from Rh foil and the Rh–I path from Cs₃Rh₂I₉ fits and holds their distance constant, allowing path and the relative fraction of the paths and the σ^2 for each path to vary.

Supplementary Table 5. ICP-MS results of Rh/Cs/I content in Cs₃Rh₂I₉/NC-R.

Elements	Mass ratio	Atomic ratio
Rh:Cs:I	1:0.0084:0.0024	1:0.0065:0.0019

Supplementary Table 6. Structure parameters extracted from the Rh K-edge EXAFS fittings of Cs₃Rh₂I₉/NC-R.

Samples	Path	N	σ^2 (Å ²)	ΔE (eV)	R (Å)
Cs ₃ Rh ₂ I ₉ /NC-R	Rh–Rh	8.0±1.0	0.0031±0.0006	3.8±0.8	2.683±0.005

Supplementary Table 7. HER activity of recent advanced electrocatalysts in 1 M KOH.

Catalysts	Overpotential at 10 mA cm ⁻²	Tafel slope	Reference
PS-MoNi@NF	30 mV	37 mV dec ⁻¹	[1]
Ni-Mo-N/CFC	43 mV	70 mV dec ⁻¹	[2]
Ni@C	27 mV	38 mV dec ⁻¹	[3]
Ru ₁ /D-NiFe LDH	18 mV	29 mV dec ⁻¹	[4]
Mo-NiO/Ni	50 mV	86 mV dec ⁻¹	[5]
F-Ni ₃ S ₄ /NF	29 mV	46 mV dec ⁻¹	[6]
Ru@Ni-MOF	22 mV	40 mV dec ⁻¹	[7]
Pt-Ni NTA	23 mV	38 mV dec ⁻¹	[8]
MoO ₂ -FeP@C	103 mV	48 mV dec ⁻¹	[9]
Cu-Ni nanocages	140 mV	79 mV dec ⁻¹	[10]
Ni-CeF ₃ -VN	33 mV	37 mV dec ⁻¹	[11]
Pt _{SA} -NiO/Ni	26 mV	27 mV dec ⁻¹	[12]
Cr-Ni NHs	75 mV	72 mV dec ⁻¹	[13]
Co ₁ /PCN	89 mV	52 mV dec ⁻¹	[14]
Ni ₅ P ₄ -Ru	54 mV	52 mV dec ⁻¹	[15]
FD-MoS ₂	164 mV	36 mV dec ⁻¹	[16]
Pt ₁ /N-C	46 mV	37 mV dec ⁻¹	[17]
NiRu _{0.13} -BDC	34 mV	32 mV dec ⁻¹	[18]
Cs ₃ Rh ₂ I ₉ /NC-R	25 mV	30 mV dec ⁻¹	This work

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