ORIGINAL ARTICLE

Discovery of highly selective and orally available benzimidazole-based phosphodiesterase 10 inhibitors with improved solubility and pharmacokinetic properties for treatment of pulmonary arterial hypertension

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Abstract   Optimization efforts were devoted to discover novel PDE10A inhibitors in order to improve solubility and pharmacokinetics properties for a long-term therapy against pulmonary arterial hypertension (PAH) starting from the previously synthesized inhibitor **A**. As a result, a potent and highly selective PDE10A inhibitor, **14**·3HCl (half maximal inhibitory concentration, $IC_{50} = 2.8$ nmol/L and >3500-fold selectivity) exhibiting desirable solubility and metabolic stability with a remarkable bioavailability of 50% was identified with the aid of efficient methods of binding free energy predictions. Animal PAH studies showed that the improvement offered by **14**·3HCl [2.5 mg/kg, oral administration (*p.o.*)] was comparable to tadalafil (5.0 mg/kg, *p.o.*), verifying the feasibility of PDE10A inhibitors for the anti-PAH treatment. The crystal structure of the PDE10A−**14** complex illustrates their binding pattern, which provided a guideline for rational design of highly selective PDE10A inhibitors.

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1. Introduction

Pulmonary arterial hypertension (PAH) is a rare but frequently fatal disease marked by particularly severe and progressive form of chronic pulmonary hypertension that frequently culminates in right ventricular heart-failure and premature death[1]. Although significant advances in understanding of the pathobiology of PAH have given rise to the development of several clinical therapeutics against this disease[2], an unmet medical need still exists because of significant morbidity and mortality as well as inadequate improvement for life quality of PAH patients[3,4]. Nowadays, clinical drugs for PAH therapy include endothelin receptor antagonists (bosentan), prostacyclin analogs (treprostinil), soluble guanylate cyclase (sGC) stimulator (riociguat) and PDE5 inhibitors (sildenafil and tadalafil) as well as their combination. However, the majority of U.S. Food and Drug Administration (FDA) proved anti-PAH drugs still fall "short of the mark" in reversing or halting the progression of this disease in the long run on account of the pathobiologic complexity relative to the various signaling pathway[5]. In addition, PDE5 inhibitors have been limited in clinical use because of the side effects from low PDE subtype selectivity and rapid metabolization such as sildenafil[6,7]. Therefore, a novel PDE5 inhibitor (TPN171) with fewer side effects and an excellent pharmacokinetic profile has being investigated in a phase II clinical trial for the treatment of PAH[8].

In the treatment of PAH, there is a growing body of evidence showing that nitric oxide (NO)-driven $3',5'$-cyclic guanosine monophosphate (cGMP) and prostacyclin (PGI2)-driven $3',5'$-cyclic adenosine monophosphate (cAMP) signaling pathways involve the up-regulation of the secondary messenger level and play a vital role in regulating vasomotor tone and vascular smooth muscle proliferation, thereby preventing pulmonary vascular from remodeling[2]. With this in mind, it is reasonable to hypothesize that the pharmacological dual regulation of both cAMP and cGMP levels may provide a synergistic effect for PAH treatment. Recently, PDE10 as a promising target for the therapy of PAH has been validated on the pathobiology[9]. PDE10 as a dual-substrate enzyme is capable of accommodating and degrading cAMP with a K_m value of 56 nmol/L and cGMP with a K_m value of 4.4 µmol/L, respectively[10]. PDE10 expression is detected not only in very high levels in human central nervous system (CNS) tissues, but also in the peripheral organs such as thyroid, testis, kidney, and lung[11], highlighting that PDE10 is of interest as an appealing drug target for a broad range of diseases including CNS disorders[12], lower urinary tract symptoms (LUTS)[13], lung/colorectal cancer[14,15] and PAH[9,16].

However, there is still limited study based on PDE10 in terms of pharmacological intervention as an effective anti-PAH therapeutic approach. On the positive side, our recent exciting result has demonstrated that a highly selective PDE10A inhibitor **A** (or original **2b**[16], IC$_{50}$ = 0.44 nmol/L, Scheme 1) served as a specific targeting chemical probe to exert significant pharmacodynamic effect on PAH treatment. Nevertheless, inhibitor **A** could not be dosed orally in mice due to its poor aqueous solubility and fast metabolism (rat *in vivo* blood terminal half-life $t_{1/2}$ = 0.58 h). Due to the fact that PAH is a chronic and progressive disease, it is of paramount importance to achieve oral administration during daily therapy. Therefore, a novel PDE10 inhibitor should be identified with more potential drug-like characteristics for PAH therapy including superior aqueous solubility and metabolic stability as well as pharmacokinetic properties. Herein, we reported our effort on further structural optimization of compound **A** aiming to improve metabolic stability and the pharmacokinetics properties in order to further validate PDE10 inhibition as a dependable drug target for PAH treatment by oral administration. As a result, compound **14** was designed and identified to show potent inhibition with an IC$_{50}$ of 2.8 nmol/L as well as selectivity of 3500-fold against other PDE subtypes. Furthermore, the pharmacokinetic properties of **14**·3HCl have been remarkably improved with an oral bioavailability up to ∼50%, thus enabling *in vivo* studies by oral intake. Importantly, **14**·3HCl significantly lowered the arterial pressure in PAH rats at a dosage of 2.5 mg/kg (*p.o.*). These findings further support the potential of PDE10 inhibition as an effective target for the anti-PAH therapy.
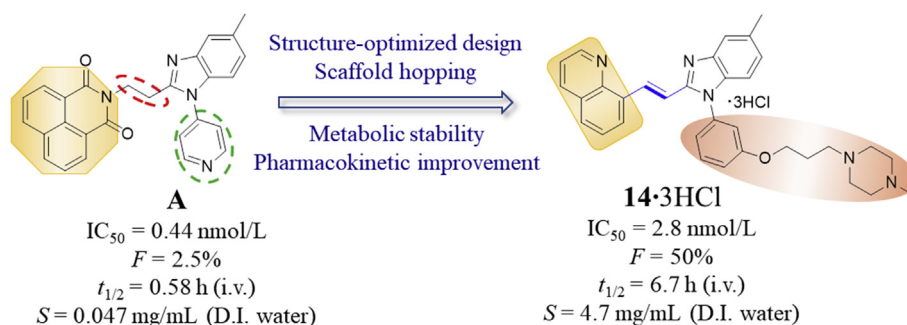
Scheme 1 Structure optimized design of novel PDE10 inhibitors aiming to improve solubility and pharmacokinetics properties.
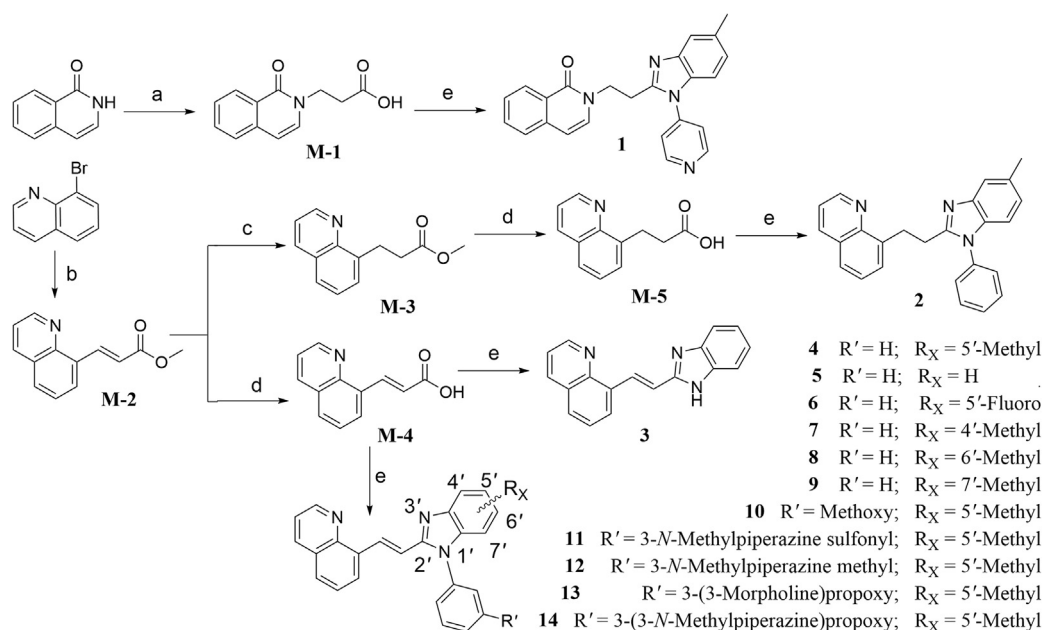
2. Results and discussion

2.1. Optimization of novel PDE10A inhibitors to improve metabolic stability and aqueous solubility

According to earlier structure−activity relationship studies and the X-ray crystal structure of the PDE10A−A complex[16], the substituted benzimidazole ring of A formed an H-bond interaction with Y693 in the selectivity pocket Q2[17,18], contributing to its high binding affinity to PDE10A as well as higher selectivity over other PDEs. Therefore, this scaffold is preferably retained in our structure-based design for novel congeneric inhibitors. Although the 1,8-naphthalimide moiety of A acted as an H-bond donor with the conserved residue Q726, this large rigid aromatic scaffold tended to aggregate via π−π stacking interactions, leading to a detrimental effect on aqueous solubility. It proved challenging to modify this structural element to improve aqueous solubility without reducing its inhibitory potency. Introduction of a hydrophilic group on to a non-critical binding site of the compound and extending out into a solvent-exposed region could be a favorable optimization approach.
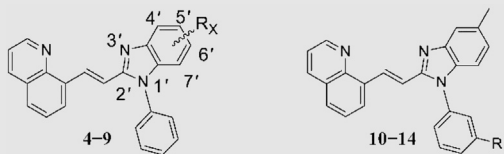
To test the above conjecture, molecular docking and molecular dynamics (MD) simulations based on the crystal complex of the PDE10A−A complex were first performed to preliminarily assess binding mode before the preparation of individual compounds (Supporting Information Fig. S1 and Table S1). The result of MD evaluations of novel compounds based on a benzimidazole moiety as an effective selective pocket binding group (SPBG) was available to estimate the binding affinity toward PDE10[16]. As a result, most of our newly designed compounds (Scheme 2) exhibited high potency inhibition towards PDE10A with IC_{50} values less than 100 nmol/L in the bioassay in vitro, while validating a similar binding affinity trend predicted among these compounds except individual differences. Our approach of structure-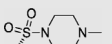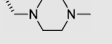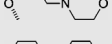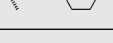based design made the utmost of binding free energy predictions to greatly reduce synthetic loads and costs, thereby accelerating the time-consuming hit-to-lead process. As our initial efforts focused on shrinking the rigid aromatic scaffold of 1,8-naphthalimide, isoquinolin-1(2H)-one and quinoline were utilized as relatively small scaffold to afford compounds 1 and 2 with IC_{50} values of ∼100 and 9.1 nmol/L, respectively. The higher inhibitory potency of compound 2 was divinable due to remarkable binding free energies ∼6 kcal/mol lower than that of 1 (Supporting Information Table S1). The high potent inhibition of compound 2 inspired us to work on further optimization on structural design (Scheme 2). To our surprise, compound 4 with a vinyl linker gave better inhibitory potency (IC_{50} = 2.7 nmol/L) compared to compound 2 with an ethyl linker. It was probably due to the fact that conformationally restricted internal C=C bridge reduced entropy compensation upon the ligand binding toward PDE10A[19]. More importantly, compound 4 exhibited a marked improvement in metabolic stability (rat liver microsomes, RLM $t_{1/2}$ = 17.4 min) as compared to compound 2 (RLM $t_{1/2}$ = 0.87 min, Supporting Information Table S2), which was attributed to limited conformations resulting in being more resistant to binding to drug metabolizing enzymes[20]. Higher stability and excellent binding affinity of compound 4 prompted us to carry out an in-depth investigation towards the effect of substitution on benzimidazole moiety. Unfortunately, none of the investigated compounds, 3−9 gained an advantage over compound 4 on evaluation of inhibitory potency against PDE10A (Table 1).

To investigate for in vivo efficacy, water solubility of the inhibitors had to be improved for oral administration. According to a similar molecular in silico docking method[16] of PDE10A with compound 4 (Supporting Information Fig. S2), the binding mode indicated that the aryl ring at the N-1′ position of benzimidazole moiety could further be structurally optimized, especially with substituents on the meta-/para-position. This N-1′ aryl moiety seems to be far from the key protein surroundings and extend out

4	R′ = H;	R_X = 5′-Methyl
5	R′ = H;	R_X = H
6	R′ = H;	R_X = 5′-Fluoro
7	R′ = H;	R_X = 4′-Methyl
8	R′ = H;	R_X = 6′-Methyl
9	R′ = H;	R_X = 7′-Methyl
10	R′ = Methoxy;	R_X = 5′-Methyl
11	R′ = 3-N-Methylpiperazine sulfonyl;	R_X = 5′-Methyl
12	R′ = 3-N-Methylpiperazine methyl;	R_X = 5′-Methyl
13	R′ = 3-(3-Morpholine)propoxy;	R_X = 5′-Methyl
14	R′ = 3-(3-N-Methylpiperazine)propoxy;	R_X = 5′-Methyl

Scheme 2   The general synthetic route summarized for the preparation of the compounds 1−14. Reagents and conditions: (a) methyl acrylate, $NaOCH_3$, methanol, reflux, 4 h; (b) methyl acrylate, TEA, $(PPh_3)_2PdCl_2$, N,N-dimethylformamide, Ar, 120 °C, overnight; (c) Pd/C, H_2, ethanol, room temperature, overnight; (d) 6 mol/L NaOH, ethanol, room temperature, overnight; (e) i) 2-(7-azabenzotriazol-1-yl)-N,N,N′,N′-tetramethyluronium hexafluorophosphate (HATU), N,N-diisopropylethylamine (DIPEA), CH_2Cl_2, overnight; ii) acetic acid, 90 °C, overnight.

Table 1 Inhibitory activities of compounds **1**−**14** against PDE10A.

Compd.	R_X	IC$_{50}$ (nmol/L)[b]	Compd.	R′	IC$_{50}$ (nmol/L)[b]
1[a]	−	∼100	**10**	−OMe	0.94 ± 0.02
2[a]	−	9.1 ± 0.7	**11**	O=S(=O)−N(piperazine)N−	0.46 ± 0.05
3[a]	−	>100			
4	5′-Me	2.7 ± 0.4	**12**	−N(piperazine)N−	2.4 ± 0.2
5	−H	25.0 ± 2.6			
6	5′-F	23.8 ± 2.1	**13**	O(morpholine)N−	0.94 ± 0.12
7	4′-Me	42.3 ± 2.0			
8	6′-Me	83.7 ± 2.0	**14**	O(morpholine)N(piperazine)N−	2.8 ± 0.1
9	7′-Me	35.7 ± 2.4			

[a]Structures as shown in Scheme 2.
[b]Values of PDE10A inhibitory rate are presented as the means ± SD ($n = 3$) with papaverine as a positive control (IC$_{50}$ = 0.1 μmol/L)[16].

into a solvent-exposed surface region. Meanwhile, the binding free energies of compounds **10**−**14** with values lower than −38 kcal/mol were found to speculate that their inhibitory activities against PDE10A could be better or comparable to that of compound **4**. As a result, all designed compounds, **10**−**14** with different hydrophilic groups showed nanomolar or subnanomolar inhibitory activities against PDE10A with IC$_{50}$ values lower than 2.8 nmol/L. Interestingly, the solubility of compounds substituted with the water-soluble morpholine and *N*-methyl piperazine were notably enhanced in the hydrochloride form, resulting in up to 11.9 and 4.7 mg/mL for **13**·HCl and **14**·3HCl solubility in deionized (D.I.) water (pH = 7.0), respectively. The C═C bridge between two aromatic rings within **14**·3HCl was insensitive to react with cysteine in aqueous phase (Supporting Information Fig. S3), indicating low probability formation of a covalent adduct *in vivo*[21].

2.2. Synthesis of compounds 1−14

The synthetic route for the newly developed compounds is outlined in Scheme 2. The key carboxyl intermediates **M-1**, **M-4**, and **M-5** were treated with different arylamines in the presence of HATU[22], followed by acid-catalyzed ring-forming condensation reaction to afford final products **1**−**12**, **13**·3HCl and **14**·3HCl in reasonable yields. All the new compounds were fully characterized by ^{1}H NMR, ^{13}C NMR spectroscopy and HR-MS, and found to fully agree with their proposed structures. To remove possible false hits, a pan-assay interfering compound substructures (PAINS) screening was performed by using the online program PAINS-Remover (http://cbligand.org/PAINS)[23]. All the synthesized compounds passed this test to minimize the risk of poor drug development choices, because starting hit compounds with problematic structures were alerted as PAINS on the account of being probably protein-reactive *in vivo*.

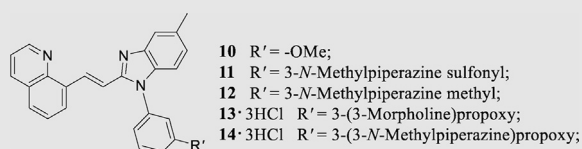### 2.3. Structure−activity relationships

The inhibitory effects of compounds on PDEs activities were investigated by the radioisotope procedure with ^{3}H-cGMP/^{3}H-cAMP while papaverine served as a positive control[16]. Most of newly synthesized compounds exhibited high binding affinity toward PDE10A with IC$_{50}$ values < 100 nmol/L. The introduction of the C═C bond instead of the C−C single bond afforded a 3-fold enhancement of PDE10A inhibition in compound **4** relative to compound **2**. Comparison of **3** with **4** showed that removal of the phenyl ring at the N-1′ position of benzimidazole moiety caused significantly decrease in inhibitory activity. As shown in Table 1, among compounds **4**−**9**, methyl-substitution only at 5′-position of the benzimidazole moiety gave rise to the pronounced inhibition; on the other hand, there was no improvement of inhibitory potency observed for 5′-fluoro analogue **6**, signifying the important hydrophobic effect of the methyl group. It was important to find that newly designed compounds **11**−**14** with different hydrophilic groups on the *meta*-position of the phenyl ring at the N-1′ position of benzimidazole moiety still afforded nanomolar or subnanomolar inhibitory activities against PDE10A with IC$_{50}$ values lower than 2.8 nmol/L, highlighting that our rational molecular docking approach played a crucial role in guiding the structure-based design for PDE10A inhibitors.

2.4. Remarkable pharmacokinetic profile of 14·3HCl

On the basis of their remarkable *in vitro* parameters, compounds **10**−**12**, **13**·3HCl and **14**·3HCl were progressed for evaluation of preliminary metabolic stability in a liver microsome assay. As shown in Table 2, compound **14**·3HCl showed favorable metabolic stability with RLM $t_{1/2}$ of 25 min, significantly more stable than other PDE10A inhibitors in Table 1, being regarded as a preferable lead to enable subsequent pharmacokinetic (PK) assessments.

The PK profile of compound **14**·3HCl in rats (approved by the Institutional Animal Care and Use Committee, Sun Yat-sen University, Guangzhou, China) is shown in detail in Table 3. To our delight, the PK (intravenous administration/i.v.) profile of compound **14**·3HCl showed significant improvement with relatively low clearance (CL = 1398 ± 217 mL/h/kg) and higher plasma exposure (the mean area under the plasma concentration−time

Table 2 Metabolic stability of **10−12**, **13·HCl** and **14·3HCl** in rat liver microsomes.

10 R′ = -OMe;
11 R′ = 3-*N*-Methylpiperazine sulfonyl;
12 R′ = 3-*N*-Methylpiperazine methyl;
13·3HCl R′ = 3-(3-Morpholine)propoxy;
14·3HCl R′ = 3-(3-*N*-Methylpiperazine)propoxy;

Compd.	$t_{1/2}$ (min)[a]	CL_{int} [mL/(min·g·protein)][b]
10	2.1	999.4
11	1.5	1390
12	11	198.1
13·3HCl	16	127.7
14·3HCl	25	85.1

[a]$t_{1/2}$: elimination half-life.
[b]CL_{int}: intrinsic body clearance.

Figure 1 Crystal structure of the PDE10-**14** (PDB code: 7BPI). (A) Surface presentation. The benzimidazole moiety and *N*-methylpiperazine group of **14** were involved in Q2 pocket and solvent region, respectively. (B) Ribbon model of **14** binding in the active site of PDE10A. The red and magentas dotted lines represent H-bonds and multiple $\pi-\pi$ stacking interactions, respectively.

curve, $AUC_{(0-\infty)} = 1819 \pm 297$ h·ng/mL) at a dose of 2.5 mg/kg, compared to **A**[16] (CL = 2337 ± 470 mL/h/kg and $AUC_{(0-\infty)} = 1099 \pm 215$ h·ng/mL). It is noteworthy that the **14·3HCl** had good bioavailability of 50%, almost 20-fold higher than **A** (Supporting Information Table S3), highlighting a successful structural optimization strategy by means of conformational restriction and introduction of the *N*-methyl piperazine side chain[24].

2.5. Crystal structure of PDE10A with **14**

In view of the promising potency against PDE10A, the cocrystal profile of the PDE10−**14** complex was pursued to reveal the binding motif. As illustrated in Fig. 1A, the quinoline group of **14** inserted in the hydrophobic clamp to form an H-bond with the invariant Q726 and had multiple $\pi-\pi$ stacking interactions with the hydrophobic clamp consisting of F729 and I692/F696, which mainly contributed to its high affinity with PDE10 A[10,16]. As expected, the benzimidazole moiety of **14** was sandwiched by the unique selectivity pocket Q2[17,18] of PDE10A and formed another strong H-bond with Y693, accounting for not only its high inhibitory potency but remarkable selectivity over other PDEs. Furthermore, the water-solubilizing *N*-methylpiperazine group extended to the solvent region in a chair conformation and displayed no detrimental effect on the binding motif in agreement with our predicted mode. The 5′-subtituted methyl group on the benzimidazole ring also exhibited the optimal interaction
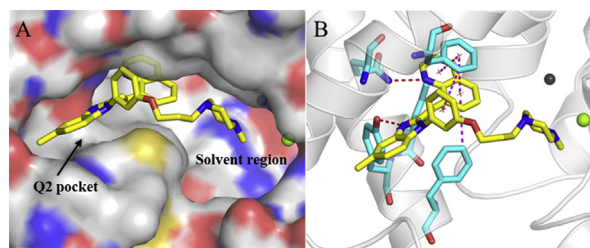
orientation fitting well within the selectivity pocket Q2 of PDE10A. Additionally, the site-directed mutagenetic PDE10A (449−770) were designed to obtain the Y693A/Q726A/F729A mutant, respectively. As a result, the inhibitory potency of the compound **14** against these mutations were significantly decreased by 13-, 30- and 58-fold, respectively, in the comparison with that of the wild type (Supporting Information Fig. S5), which added weight to the highlight of the key residues for inhibitor binding like Q726, F729 and Y693.

2.6. Excellent selectivity of compound **14·3HCl** over other PDE isoforms

As seen in Supporting Information Table S5, the structure-based sequence alignment of the substrate-binding pocket shows dramatic variation of amino acids across PDE families[10], which determines not only the shape and size of the binding pocket but also the binding affinity of different inhibitors. A selective binding pocket composed of adjacent amino acid residues of PDE10A has been demonstrated, which exclusively involves I692, Y693 and G725, respectively. The benzimidazole group of **14·3HCl** well occupied this narrow selectivity pocket[17,18], and showed a strong H-bond interaction with the specific Y693, accounting for the excellent selectivity over other PDEs. The selectivity profile of compound **14·3HCl** against other PDE isoforms is summarized in Table 4. Compound **14·3HCl** with the IC_{50} value of 2.8 nmol/L against PDE10A2 exhibited weak inhibition ($IC_{50} > 10$ μmol/L) towards the majority of the representative PDE subtypes including PDE1B, PDE2A, PDE3A, PDE4D2, PDE5A1, PDE7A1, PDE8A1

Table 3 Pharmacokinetic profile of compound **14·3HCl** in Sprague−Dawley rats.

Route	$t_{1/2}$ (h)	T_{max} (h)	C_{max} (ng/mL)	$AUC_{(0-t)}$ (h·ng/mL)	$AUC_{(0-\infty)}$ (h·ng/mL)	$MRT_{(0-t)}$ (h)	CL (mL/h/kg)	F (%)
i.v.[a]	6.7 ± 1.5	0.4 ± 0.4	230 ± 50	1666 ± 184	1819 ± 297	6.3 ± 0.8	1398 ± 217	−
p.o.[b]	5.2 ± 0.9	6.0 ± 2.0	272 ± 25	3310 ± 812	3551 ± 1008	8.4 ± 0.7	−	50 ± 12

AUC: area under the curve. MRT: mean residence time. CL: clearance. −Not applicable.
[a]i.v. = intravenous administration, dose = 2.5 mg/kg.
[b]p.o. = oral administration, dose = 10 mg/kg.

Table 4   Selectivity of **14**·3HCl across PDE families.

PDE	IC$_{50}$ (μmon/L)	Selectivity[a]
PDE10A2 (449−770)	2.8×10^{-3}	−
PDE1B (10−487)	>10	>3500
PDE2A (580−919)	>10	>3500
PDE3A (679−1087)	>10	>3500
PDE4D2 (86−413)	>10	>3500
PDE5A1 (535−860)	>10	>3500
PDE7A1 (130−482)	>10	>3500
PDE8A1 (480−820)	>10	>3500
PDE9A2 (181−506)	>10	>3500

[a]Selectivity fold $=$ IC$_{50(\text{PDEs})}$/IC$_{50(\text{PDE10A2})}$.

and PDE9A2, thereby showing more than 3500-fold selectivity across other PDE families. Apparently, the remarkable selectivity of compound **14**·3HCl over other PDE subtypes implied potential for PAH treatment with minimal interference from other PDEs as well as fewer side effects anticipated, supporting further pharmacodynamics assessment studies for **14**·3HCl.

2.7.   Pharmacological toxicity

After an oral administration, there was no obvious acute toxicity found for **14**·3HCl in the course of evaluation for pharmacological safety using the same protocol previously reported[16,25,26], in which this inhibitor was well tolerated up to a dose of 1.0 g/kg. Additionally, **14**·3HCl showed a moderate affinity towards human ether-à-go-go-related gene (hERG) potassium channel with an IC$_{50}$ value of 4.5 μmol/L (Supporting Information Fig. S4), corresponding to 1600-fold selectivity for PDE10A IC$_{50}$

(2.8 nmol/L) in vitro, indicating a good safety margin of 36-fold between hERG IC$_{50}$ and plasma peak concentration (C_{max}, calculated (Calcd.) as $\sim$124.6 nmol/L at an oral dosage of 2.5 mg/g at 6 h in Supporting Information Table S4). As a result, compound **14**·3HCl showed practical usability and potential to be used as a drug candidate for in vivo pharmacodynamic studies by means of oral delivery.

2.8.   Notable therapeutic effects against PAH in rats

Subsequently, pharmacodynamic effects of **14**·3HCl against PAH were evaluated in a monocrotaline (MCT)-induced PAH rat model[16,25,26] (rat source: Beijing Vital River Laboratory Animal Technology Co., Ltd.; approved by the Institutional Animal Care and Use Committee, Sun Yat-sen University, Guangzhou, China). The typical symptoms of PAH were observed after 21 consecutive days of MCT injection (60 mg/kg, once), as shown by an enhancement of the mean pulmonary artery pressure (mPAP), right ventricle hypertrophy index (RVHI%), and wall thickness percentage (WT%) of the model group in comparison with the control group (Fig. 2A−C). Treatment with **14**·3HCl or tadalafil decreased all the above symptoms of the rats in contrast to that of the model (no treatment) group, especially for mPAP. For RVHI% and WT%, similar trends were also observed. Oral administration of **14**·3HCl at a dose of 2.5 mg/kg approximately provided a plasma concentration of Calcd. $\sim$84 nmol/L at 10 h (equivalent to 336 nmol/L at an oral dose of 10 mg/kg in Supporting Information Table S4), which indicated 30-fold higher than the IC$_{50}$ (2.8 nmol/L) in vitro. It resulted in significantly ameliorating these PAH medical indicators in rats comparable to that of tadalafil at a dose of 5.0 mg/kg (p.o.), further indicating the feasibility of PDE10A inhibitors for the anti-PAH treatment.
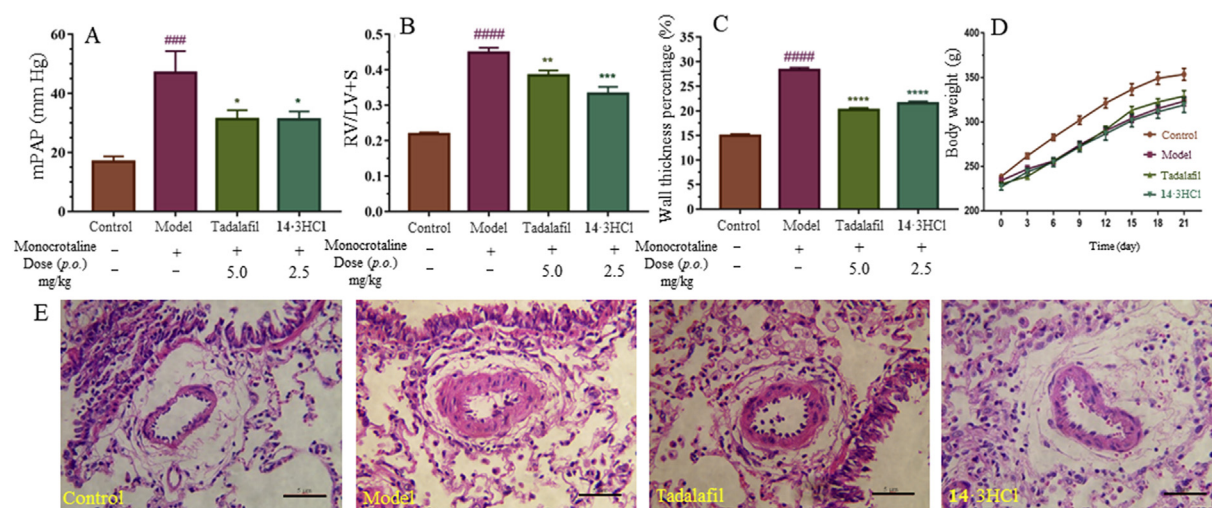

Figure 2   Effects of compound **14**·3HCl and tadalafil on pulmonary arterial rat model in vivo. (A) Mean pulmonary artery pressure (mPAP) of different groups. (B) Right ventricle hypertrophy index (RVHI%) of different groups. (C) Wall thickness percentage (WT%) of different groups. (D) Body weight of different groups. (E) Representative images for each group with hematoxylin and eosin straining. Data shown represent the mean $\pm$ SD ($n = 6-10$ animals per group). $^{\#\#\#\#}P<0.0001$ and $^{\#\#\#}P<0.001$ vs. the control group; $^{*}P<0.05$, $^{**}P<0.01$ and $^{***}P<0.001$ vs. the model group. Scale bar, 5 μm.

3. Conclusions

Given the previously-reported research findings of PDE10 inhibition as a potential anti-PAH target, there is a continuing interest in the discovery of novel PDE10A inhibitors with favorable druglike characteristics for a long-term therapy. Through optimization of the pharmacokinetic profile of the lead compounds *via* the structure-based design/optimization strategy using scaffold hopping and conformational restriction with the aid of efficient methods of binding free energy predictions, a novel, potent and highly selective PDE10A inhibitor **14** with $IC_{50} = 2.8$ nmol/L for PDE10A and more than 3500-fold selectivity over other PDEs was identified.

The crystal structure of the PDE10A−**14** complex illustrates the binding motif, which provided a guideline for rational design of highly selective PDE10A inhibitors. More importantly, **14·3HCl** was found to have favorable drug-like properties including high solubility ($S = 4.7$ mg/mL in D.I. water), remarkable *in vivo* metabolic stability ($t_{1/2} = 5.2$ h), low risk for hERG K^+ channel block, and remarkable oral bioavailability ($F = 50\%$). Furthermore, animal studies ascertained the therapeutic potential of PDE10 inhibition for the treatment of PAH by oral administration, in which **14·3HCl** significantly improved typical disease medical indicators in the rat model. In short, our findings clearly demonstrate that PDE10 inhibition is a promising potential therapeutic target, which opens a vista for discovery of potent PDE10A inhibitors for treatment of PAH.

4. Experimental

All the reagents were purchased from commercial suppliers (Aldrich®, Shanghai, China; BIDE®, Shanghai, China; or Innochem® Beijing, China), which are used directly without further purification. Chemical HG/T2354-92 silica gel (200−300 mesh, Haiyang®, Qingdao, China) was used for chromatography. ^{1}H NMR and ^{13}C NMR spectra were recorded at room temperature on a Bruker AVANCE III 400 and 500 instrument (Bruker®, Zürich, Switzerland) with tetramethylsilane (TMS) as an internal standard. The following abbreviations are used: s (singlet), d (doublet), dd (two doublets), t (triplet), q (quartet), and m (multiplet). Coupling constants were reported in Hz. High-resolution mass spectra were recorded on a Shimadzu LC−MS−IT-TOF mass spectrometer (Shimadzu®, Kyoto, Japan). The accurate number of HCl in the molecular structure of **14·3HCl** was deduced by an ion chromatography (IC) method with the aid of Dionex ICS-900 (Thermo Fisher Scientific®, Sunnyvale, USA). The purity of compounds was determined by reverse-phase high-performance liquid chromatography (HPLC) analysis confirming to be over 95%. HPLC instrument: Shimadzu LC-20AT (Shimadzu®, column: Hypersil BDS C18, 5.0 μm, 150 mm × 4.6 mm (elite); detector: SPD-20A UV−Vis detector, UV detection at 254 nm; elution, methanol/H_2O (100%−90%, v/v); column temperature: 25 °C; and flow rate = 0.8−1.0 mL/min.

4.1. General procedures for synthesis of compounds 4−14

To the solution of **M-4** (1.0 mmol) in dichloromethane (DCM, 10 mL) was added HATU (760 mg, 3 mmol), DIPEA (516 mg, 4.0 mmol), and appropriate amine (1.0 mmol) with stirring at room temperature for 12 h. After completion of reaction, the mixture was extracted with ethyl acetate (EA, 3 × 50 mL). The organic layer was washed with brine, dried over anhydrous sodium sulfate and concentrated *in vacuo* to afford crude intermediate directly used in the next step without purification. To the residue above was added acetic acid (10 mL) which was stirred at 90 °C overnight. Then the reaction mixture was alkalized with NaOH to pH 9−10, and extracted with EA (3 × 50 mL). The organic layer was washed with brine, dried over anhydrous sodium sulfate, filtered and concentrated *in vacuo*. The residue was purified by silica gel column chromatography (DCM/methanol, 10:1) to afford products **4−14** in two steps.

4.1.1. (E)-8-(2-(1-(3-Methoxyphenyl)-5-methyl-1H-benzo[d]imidazol-2-yl)vinyl)quinoline (10)

130 mg, yield 69.0%. ^{1}H NMR (400 MHz, dimethyl sulfoxide-d_6/DMSO-d_6) δ 9.05 (d, $J = 16.3$ Hz, 1H), 8.98 (dd, $J = 4.1, 1.8$ Hz, 1H), 8.41 (dd, $J = 8.3, 1.7$ Hz, 1H), 8.04 (d, $J = 7.1$ Hz, 1H), 7.98 (d, $J = 8.1$ Hz, 1H), 7.62 (dt, $J = 11.2, 5.0$ Hz, 3H), 7.56 (s, 1H), 7.36 (d, $J = 16.2$ Hz, 1H), 7.18 (m, 4H), 7.08 (dd, $J = 8.2, 1.1$ Hz, 1H), 3.85 (s, 3H), 2.46 (s, 3H). ^{13}C NMR (126 MHz, DMSO-d_6) δ 160.82, 150.99, 150.69, 145.67, 143.71, 137.10, 136.76, 134.82, 133.77, 132.62, 132.53, 131.35, 129.56, 128.70, 127.24, 127.08, 125.01, 122.35, 119.90, 119.14, 116.58, 115.29, 113.32, 110.39, 56.06, 21.72. HR-MS (ESI-TOF) m/z [M+H]$^+$ Calcd. for $C_{26}H_{21}N_3O$ 392.1757, Found 392.1750.

4.1.2. (E)-8-(2-(5-Methyl-1-(3-((4-methylpiperazin-1-yl)sulfonyl)phenyl)-1H-benzo[d]imidazol-2-yl)vinyl) quinoline (11)

52 mg, yield 48.4%. ^{1}H NMR (400 MHz, DMSO-d_6) δ 8.98 (m, 2H), 8.41 (dd, $J = 8.3, 1.7$ Hz, 1H), 8.08 (d, $J = 7.2$ Hz, 1H), 8.01 (m, 4H), 7.88 (s, 1H), 7.60 (m, 3H), 7.37 (d, $J = 16.2$ Hz, 1H), 7.13 (m, 2H), 2.98 (s, 4H), 2.47 (s, 3H), 2.27 (s, 4H), 2.11 (s, 3H). ^{13}C NMR (126 MHz, DMSO-d_6) δ 151.14, 150.80, 145.65, 137.19, 137.10, 136.62, 134.68, 133.61, 133.13, 132.90, 132.66, 132.03, 129.69, 128.70, 128.29, 127.34, 126.94, 126.74, 125.30, 122.35, 119.35, 116.49, 110.02, 53.88, 46.26, 45.70, 21.70. HR-MS (ESI-TOF) m/z [M+H]$^+$ Calcd. for $C_{30}H_{29}N_5O_2S$ 524.2115, Found 524.2108.

4.1.3. (E)-8-(2-(5-Methyl-1-(3-((4-methylpiperazin-1-yl)methyl)phenyl)-1H-benzo[d]imidazol-2-yl)vinyl) quinoline (12)

30 mg, yield 40.2%. ^{1}H NMR (400 MHz, CDCl$_3$) δ 8.99 (d, $J = 16.2$ Hz, 1H), 8.91 (dd, $J = 4.1, 1.7$ Hz, 1H), 8.15 (dd, $J = 8.3, 1.7$ Hz, 1H), 7.86 (d, $J = 7.2$ Hz, 1H), 7.78 (d, $J = 7.3$ Hz, 1H), 7.68 (s, 1H), 7.57 (t, $J = 7.7$ Hz, 1H), 7.51 (t, $J = 7.7$ Hz, 2H), 7.45 (dd, $J = 11.1, 6.8$ Hz, 3H), 7.41 (d, $J = 8.0$ Hz, 1H), 7.09 (m, 2H), 3.66 (s, 2H), 2.61 (s, 4H), 2.53 (s, 3H), 2.50 (s, 4H), 2.29 (s, 3H). ^{13}C NMR (126 MHz, CDCl$_3$) δ 151.20, 149.87, 146.21, 143.60, 140.26, 136.20, 136.14, 134.82, 134.61, 133.73, 132.84, 129.71, 129.07, 128.61, 128.50, 127.80, 127.34, 126.32, 126.10, 124.60, 121.45, 119.38, 117.38, 109.62, 61.88, 54.52, 51.93, 45.11, 21.67. HR-MS (ESI-TOF) m/z [M+H]$^+$ Calcd. for $C_{31}H_{31}N_5$ 474.2652, Found 474.2634.

4.1.4. (E)-4-(3-(3-(5-Methyl-2-(2-(quinolin-8-yl)vinyl)-1H-benzo[d]imidazol-1-yl)phenoxy)propyl)morpholine hydrochloride (13·3HCl)

30 mg, yield 57.5%. ^{1}H NMR (400 MHz, MeOD-d_4) δ 9.14 (m, 2H), 8.89 (d, $J = 8.0$ Hz, 1H), 8.34 (s, 1H), 8.30 (d, $J = 7.6$ Hz, 1H), 7.95 (dd, $J = 7.8, 4.5$ Hz, 1H), 7.88 (s, 1H), 7.77 (s, 1H), 7.73 (t, $J = 7.4$ Hz, 1H), 7.44 (m, 6H), 4.28 (s, 2H), 4.06 (d, $J = 12.6$ Hz, 2H), 3.86 (t, $J = 12.0$ Hz, 2H), 3.57 (d,

$J = 11.3$ Hz, 2H), 3.43 (s, 2H), 3.20 (s, 2H), 2.62 (s, 3H), 2.35 (s, 2H). ^{13}C NMR (126 MHz, MeOD-d_4) δ 160.13, 147.51, 147.40, 144.30, 138.83, 138.41, 133.05, 132.53, 131.87, 131.69, 131.43, 130.84, 129.46, 128.71, 128.65, 122.34, 119.69, 117.57, 113.63, 113.24, 112.99, 112.34, 65.46, 63.67, 54.66, 52.00, 23.38, 20.35. HR-MS (ESI-TOF) m/z [M+H]$^+$ Calcd. for $C_{32}H_{32}N_4O_2$ 505.2598, Found 505.2591.

4.1.5. (E)-8-(2-(5-Methyl-1-(3-(3-(4-methylpiperazin-1-yl) propoxy)phenyl)-1H-benzo[d]imidazole-2-yl)vinyl)quinoline hydrochloride (14·3HCl)

14, 2.0 g, yield 42.6%. To the solution of **14** (1.0 g 1.9 mmol) in DCM (10 mL) was added hydrochloric acid in 1,4-dioxane (4 mol/L, 3.8 mmol) with stirring at room temperature until multiple solid separated out. Then the yellow solid was filtered to obtain the product of **14**·3HCl (951 mg, yield 85.8%). ^{1}H NMR (400 MHz, MeOD-d_4) δ 9.12 (d, $J = 16.3$ Hz, 1H), 8.99 (d, $J = 3.1$ Hz, 1H), 8.47 (d, $J = 8.1$ Hz, 1H), 8.17 (d, $J = 6.2$ Hz, 1H), 8.11 (d, $J = 7.9$ Hz, 1H), 7.68 (m, 5H), 7.41 (m, 5H), 4.29 (s, 2H), 3.76 (s, 8H), 3.54 (s, 2H), 3.03 (s, 3H), 2.61 (s, 3H), 2.39 (s, 2H). ^{13}C NMR (126 MHz, MeOD-d_4) δ 160.12, 149.96, 148.24, 144.68, 140.72, 138.50, 138.11, 133.23, 131.81, 131.61, 131.41, 131.02, 130.93, 130.19, 128.87, 128.31, 126.78, 122.02, 119.72, 117.49, 113.62, 113.08, 112.20, 110.47, 65.27, 23.67, 20.35. HR-MS (ESI-TOF) m/z [M+H]$^+$ Calcd. for $C_{33}H_{35}N_5O$ 518.2914, Found 518.2895. Cl% Calcd. 16.9%, Found 16.2% by IC.

4.2. In vitro enzymatic activity assay

The catalytic domains of PDE1B (10−487), PDE2A (580−919), PDE3A (679−1087), PDE4D2 (86−413), PDE5A1 (535−860), PDE7A1 (130−482), PDE8A1 (480−820), PDE9A2 (181−506), and PDE10A (449−770) were purified using a similar protocol in our previously published protocol[16,27,28]. ^{3}H-cGMP was the substrate for the biological test against PDE1B, PDE2A and PDE5A while ^{3}H-cAMP for those of PDE3A, PDE4D, PDE7A, PDE8A, and PDE10A. Firstly, ^{3}H-cGMP/^{3}H-cAMP was diluted with the assay buffer (20−50 mmol/L Tris-HCl (pH 8.0), 10 mmol/L MgCl$_2$, and 1 mmol/L DTT) to 20,000−30,000 cpm per assay. The mixture was then performed at 25 °C for 15 min and terminated by the addition of 0.2 mol/L ZnSO$_4$. 0.2 mol/L Ba(OH)$_2$ was added with a precipitate formed. The unreacted ^{3}H-cGMP/^{3}H-cAMP was left in the supernatant. The radioactivity in the supernatant was measured in 2.5 mL of Ultima Gold liquid scintillation cocktail using a PerkinElmer 2910 liquid scintillation counter (PerkinElmer® Tri-carb2910TR, Waltham, USA). For the measurement of IC$_{50}$ of compounds, eight different concentrations were used, and each measurement was at least repeated three times.

5. Associated content

The atomic coordinates and structure factors have been deposited into the RCSB Protein Data Bank with accession number 7BPI.

Acknowledgments

Author contributions

Yuncong Yang conducted methodology and validation. Sirui Zhang investigated in validation. Qian Zhou investigated in methodology. Chen Zhang investigated in methodology and software. Yuqi Gao conducted validation and formal analysis. Hao Wang investigated the study. Zhe Li investigated in software and validation. Deyan Wu and Yinuo Wu validated the data. Yi-You Huang conducted methodology and validation, and writing-reviewed the manuscript. Lei Guo investigated the study, writing-reviewed and edited the manuscript. Hai-Bin Luo supervised the whole study, writing-reviewed and edited the manuscript. All the authors approved the final version of the manuscript.

Conflicts of interest

The authors declare no conflicts of interest.

Appendix A. Supporting information

Supporting data to this article can be found online at https://doi.org/10.1016/j.apsb.2020.04.003.

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