Article

Chlorodifluoromethane as a C1 Synthon in the Assembly of *N*-Containing Compounds



Ma et al., iScience 19, 1–13 September 27, 2019 © 2019 The Author(s). https://doi.org/10.1016/ j.isci.2019.07.005

Check for

Article

Chlorodifluoromethane as a C1 Synthon in the Assembly of *N*-Containing Compounds

Xingxing Ma,¹ Jianke Su,¹ Xingang Zhang,^{2,*} and Qiuling Song^{1,3,4,5,*}

SUMMARY

The development of C1 synthons to afford the products that add one extra carbon has become an important research theme in the past decade, and significant progress has been achieved with CO₂, CO, HCOOH, and others as C1 units. Despite the great advance, the search for new C1 synthons that display unique reactivity, complement to the current C1 sources, and add more value to C1 chemistry is still desirable. Herein, we report a quadruple cleavage of chlorodifluoromethane to yield a C1 source, which was successfully employed in the construction of various *N*-containing compounds especially with pharmaceutical molecules under mild conditions. This strategy provides a useful method for late-stage modification of pharmaceutical compounds. Four bonds in CICF₂H were orderly cleaved under basic conditions in the absence of transition metals. Preliminary mechanistic studies revealed that (*E*)-*N*-phenylformimidoyl fluoride intermediate is involved in this process by *in situ* ¹H NMR studies and control experiments.

INTRODUCTION

The C1 chemistry has emerged as an elegant strategy for efficient preparation of homologous compounds, which added one extra carbon in modern chemical transformations (Aresta et al., 2014; Sakakura et al., 2007; Huang et al., 2011; Yan et al., 2012; Natte et al., 2017; Wakade et al., 2017; Senadi et al., 2019). There are ample significance and a plethora of characteristics for C1 chemistry, for instance, carbon chain increasing (Aresta et al., 2014), construction of importance functional groups (carboxylic or carbonyl groups) (Huang et al., 2011; Cokoja et al., 2011), incorporation of two or more organic small molecules to yield important products (Oh and Hu, 2013), and modification of the pharmaceutical or natural products for value-added bulk (Liu et al., 2015; Ma et al., 2018a). Among all known C1 synthons, CO₂, CO, and formic acid are the most famous ones, which have been widely used in various reaction processes, and many beautiful transformations have been developed with them, which further attracted more and more chemists devoted to this field (Aresta et al., 2014; Sakakura et al., 2007; Aresta et al., 2014; Huang et al., 2011; Cokoja et al., 2011; Oh and Hu, 2013; Sordakis et al., 2018; Gibson, 1969; Enthaler et al., 2010). Despite the significance and great advance of C1 chemistry, the search for C1 synthons that display unique reactivity, complement to the current C1 sources, and add more value to C1 chemistry is still highly desirable. Thus, direct introduction of one extra carbon from cheap and available materials under mild conditions to provide a cost-efficient, pragmatic, and valuable alternative avenue would be popular in the field of synthetic and pharmaceutical communities, which might have deep impact on industry as well.

Chlorodifluoromethane (CICF₂H) is well known as an inexpensive and abundant industrial raw material (Hudlicky and Pavlath, 1995) for the construction of various fluorinated compounds (Wang et al., 2014; Fier and Hartwig, 2012; Gu et al., 2014; Yu et al., 2017; Wu et al., 2019; Miao et al., 2018; Zhang et al., 2019), featuring thermodynamic stability and kinetic inertness as well as atomic economy as fluorine source. Therefore, efficient transformations of this easily accessible raw material to create valuable chemicals have deservedly gained great attention. The most common transformation of CICF₂H involves the formation of difluorocarbene (:CF₂) by the cleavage of both C-Cl and C-H bonds (Figure 1Aa) (Feng et al., 2017), usually under basic conditions with heteroatom nucleophiles, rendering the corresponding difluoromethylated heteroatom compounds (Hine and Porter, 1957; Nawrot and Jonczyk, 2007). Pyrolysis of CICF₂H at high temperature or pressure leads to the important raw industrial material tetrafluoroethylene (Hudlicky and Pavlath, 1995; Sung et al., 2004). Very recently a reaction by palladium-catalyzed cross-coupling between arylboronic acids and CICF₂H via a metal-difluorocarbene intermediate has been reported (Feng et al., 2017; Yu et al., 2019), representing the catalytic transformation of CICF₂H. Other conversion processes that do not involve difluorocarbene species were still difluoromethylation-related ones in which only one

¹The Institute of Next Generation Matter Transformation, College of Material Sciences Engineering at Huaqiao University, 668 Jimei Boulevard, Xiamen, Fujian 361021, China

²Key Laboratory of Organofluorine Chemistry, Center for Excellence in Molecular Synthesis, Shanghai Institute of Organic Chemistry, University of Chinese Academy of Sciences, CAS, Shanghai 200032. China

³State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, CAS, Shanghai 200032, China

⁴Key Laboratory of Molecule Synthesis and Function Discovery, Fujian Province University, College of Chemistry at Fuzhou University, Fuzhou, Fujian 350108, China

⁵Lead Contact

*Correspondence: xgzhang@mail.sioc.ac.cn (X.Z.),

qsong@hqu.edu.cn (Q.S.) https://doi.org/10.1016/j.isci. 2019.07.005

1



Figure 1. Various Transformations of CICF₂H

(A) Known transformations of $CICF_2H$.

(a) Double cleavage of $CICF_2H$ to lead to difluorocarbene species.

- (b) Single cleavage of $CICF_2H$ to lead to difluoromethyl radical.
- (c) Double cleavage of $CICF_2H$ with external F^- to lead to trifluoromethyl anion.
- (B) Our work.
- (d) Quadruple cleavage of $\mathsf{CICF}_2\mathsf{H}$ as a C1 source.

C-Cl bond was broken through a difluoromethyl radical pathway (Figure 1Ab) (Xu et al., 2018). Trifluoromethyl anion (CF₃⁻) is readily derived from the difluorocarbene species and external fluorine source via double cleavage of ClCF₂H (Figure 1Ac) (Zheng et al., 2015). Intriguingly and surprisingly, quadruple cleavage of ClCF₂H to provide versatile C1 synthons, by breaking one C-Cl bond, two stable C-F bonds, and one C-H bond orderly in a single-vessel reaction (Figure 1B), has never been reported to date, probably mainly because of the high BDE of C(sp^3)-F bonds (the bond dissociation energy of a single C-F bond: 485 KJ/mol) (O'Hagan, 2008).

Herein, we report a quadruple cleavage of chlorodifluoromethane as a type of C1 source to access valuable formimidamide derivatives that are widely employed as ligands or forming metal complexes as quasi-*N*-heterocyclic carbenes (NHCs) (Figure 1 (i) and (ii)) (Schröder et al., 2009, 2010; Bitterlich et al., 2007; Boogaerts and Nolan, 2010; Ohishi et al., 2008; Hopkinson et al., 2014). Despite the importance of these compounds, their elegant syntheses are very rare. Therefore, expanding the toolbox of methods for their synthesis will enrich diversity of this kind of compounds. Amines are very common raw materials as well as crucial building blocks with rich chirality (France et al., 2014); we envision that formimidamides could be readily generated in a single-vessel synthesis from two amines with CICF₂H as C1 source due to the special reactivity of CICF₂H. These processes represent a significant reaction modality for CICF₂H, which might promote and enrich C1 chemistry, organic fluorine chemistry (Gouverneur and Seppelt,

2015), as well as green chemistry (Horváth, 2007). Meanwhile, $CICF_2H$ provides a unique and alternative approach for the current known C1 sources: the control and comparison experiments with CO, CO₂, and formic acid as C1 synthons were performed under our standard conditions as well as the reported procedures; notably, no desired formimidamides were ever formed or obtained under those reaction conditions. These results further underscore the uniqueness and peculiarity of $CICF_2H$ as a C1 source (for further details, see also Schemes S1 and S2).

RESULTS AND DISCUSSION

Optimization of Reaction Conditions

Our design is based on our recent discovery in which ethyl bromodifluoroacetate (BrCF2COOEt) (Ma et al., 2018b; Deng et al., 2019) could act as a C1 source and formylating reagent with amines via quadruple cleavage under basic conditions (Ma et al., 2018a, 2018c, 2018d); we postulated that the guadruple cleavage of CICF₂H could also be occurred under the similar sets since it is known that difluorocarbene could be readily accessible from CICF₂H under basic conditions. Moreover, compared with BrCF₂COOEt, CICF₂H is obviously much cheaper and more atomic economical. We commenced our hypothesis by using low-cost and widely available aniline (1a) and N-methyl aniline (2a) as model substrates. To our delight, the yield of 76% of the anticipated product **3a** (Zhao et al., 2005) was obtained from the reaction of **1a** with **2a** under the CICF₂H atmosphere without water (entry 1). More delightfully, the yield was significantly increased to 83%-92% with the increase of dosage of water (entries 2-3); notably, excess water caused deteriorated effect on the reaction, since some unknown by-products were observed when the amount of water was increased to 20-30 equivalents, and the yield of the desired product was dropped to 83% (entry 2 versus entry 3, and for further details, see also Table S2). Replacing KOH with either Cs₂CO₃ or K₂CO₃ as the base resulted in lower yields (entries 4-5). To our surprise, this transformation was completely suppressed with Na₂CO₃ and NaHCO₃ as the base (entries 6-7). The solvent was so crucial that no reaction occurred in other solvents, such as in THF and dioxane (entries 8-9) (for further details, see also Table S3). In addition, the yield of the desired product was slightly higher at the ambient temperature than at 50°C. In terms of reaction time, the longer time (36 h) led to the best result (entry 11).

Substrate Scope for Intermolecular Transformation

With the optimal reaction conditions in hand (entry 2 in Table 1), we explored the generality and limitation of this transformation (Figure 2). First, a variety of para-substituted anilines with electron-donating groups (alkoxy, phenoxyl, alkyl, and N, N-dimethyl) (1b-1j), as well as electron-withdrawing groups, such as halogen (1k-1m) and nitro group (1n), delivered the desired formimidamide derivatives (3a-3n) in good to excellent yields under the standard conditions. We next examined [1,1'-biphenyl]-4-amine (1o) under this reaction condition to provide the desired product **30** in 77% yield. Besides, a large-scale (10 mmol) reaction of the N-methyl aniline (2a) has been carried out to afford 3a in 62% yield (for further details, see also Scheme S3). Similar result could be obtained for meta-substituted (m-Br) aniline (1p). Using the disubstituted 3,4-dimethylaniline (1q) and trisubstituted 2,4,6-trimethylaniline (1r), the corresponding products (3q and 3r) could be obtained in good yields (83%–85%). 5,6,7,8-Tetrahydronaphthalen-1-amine (1s) and 9H-fluoren-2-amine (1t) were carried out under the standard conditions to provide the target molecules (3s and 3t) in 80% and 76% yields, respectively. The absolute molecular structure of product 3t was unambiguously confirmed by X-ray crystallography analysis (Figure 2, and for further details, see also Table S1 and Data S1) (CCDC: 1874971). The fused polycyclic amines 9,9-diphenyl-9H-fluoren-2-amine (1u) and naphthalen-1-amine (1v) were subjected under the optimized reaction conditions, rendering the expected products (3u-3v) in moderate yields (68%-77%). Heterocyclic compounds, such as benzo[d]thiazol-2-amine (1w), were also amenable to this transformation, and the corresponding product was obtained in 62% yield. We then further investigated the scope of the N-substituted aniline derivatives with aniline (1a) under the viable reaction conditions. Delightedly, the corresponding products (3x-3z, 3aa-3ac) were obtained in good to excellent yields with good functional group tolerance. In addition, given the prevailing existence of amines in pharmaceutical molecules and natural products (Ma et al., 2018a; Brunet and Neibecker, 2001), we selected Benzocaine (1ad, local anesthetic), Amoxapine (2ae, antidepressant), 2-(piperazin-1-yl)-4-(trifluoromethyl)pyrimidine (2af, medical/material intermediates), and multi-functional Vildagliptin (2ag, inhibit glucagon/chiral reagent/medicinalintermediate) and exposed them under the standard conditions; the corresponding products were obtained in 59%, 76%, 61%, and 71% yields, respectively. Gratifyingly, the chiral molecule (S)-N-benzyl-1-phenylethan-1-amine (2ah) experienced the optimal reaction conditions to deliver (S, E)-N-benzyl-N'-phenyl-N-(1-phenylethyl) formimidamide (**3ah**) in 60% yield, which might be a potential chiral ligand to realize enantioselective-control reactions.



Entry	Base (3 Equiv)	H ₂ O (X Equiv)	Solvent (2 mL)	T (°C)	Yield (%) ^a
1	КОН	0	CH ₃ CN	r.t.	76
2	КОН	5	CH ₃ CN	r.t.	92 (88) ^b
3	КОН	30	CH ₃ CN	r.t.	83
4	Cs ₂ CO ₃	5	CH ₃ CN	r.t.	76
5	K ₂ CO ₃	5	CH ₃ CN	r.t.	79
6	Na ₂ CO ₃	5	CH ₃ CN	r.t.	N.D.
7	NaHCO ₃	5	CH ₃ CN	r.t.	N.D.
8	КОН	5	THF	r.t.	N.D.
9	КОН	5	Dioxane	r.t.	N.D.
10	КОН	5	CH ₃ CN	50	70
11 ^{c,d}	КОН	5	CH ₃ CN	r.t.	62° (79) ^d
12 ^e	КОН	5	CH ₃ CN	r.t.	N.D.

Table 1. Representative Results for Optimization of the Formation of (E)-N-methyl-N,N'-diphenylformimidamide (3a)

Reaction condition: aniline (1a, 1.2 equiv. 0.12 mmol), N-methylaniline (2a, 0.1 mmol), the atmosphere of chlorodifluoromethane (CICF₂H) (cat. 0.3 mmol), base (3 equiv.), solvent (2 mL), for 36 h.

r.t., room temperature; N.D., not detected. ^aGC yields.

^bIsolated yields.

^cFor 12 h.

^dFor 24 h.

^eNo CICF₂H.

In addition, we found that aliphatic secondary amine is compatible under the standard conditions as well; in terms of the substrate dicyclohexylamine (4), the corresponding product 5 was acquired in 76% yield (Equation 1 in Figure 3A). It is worth mentioning that our strategy is highly regio-selective, for example, when N^1 -isopropyl- N^3 -phenylbenzene-1,3-diamine (6) was investigated under viable reaction condition, only compound 7 was afforded with diphenylamine part intact (Equation 2). In addition, when the target 3a was hydrolyzing under 1 M HCl, the two original substrates (1a and 2a) as well as two formylated compounds 1a-1 and 2a-1 were obtained, respectively (Ma et al., 2018a) (for further details, see also Scheme S4), which infers that our strategy might be a potential method for drug delayed or sustainable release when two different pharmaceutical molecules are combined by one extra carbon with our strategy (Equation 3). We carried out, therefore, correlative experiment using Benzocaine (1ad) and Vildagliptin

CellPress



Reaction Condition 1: the primary amine (1, 0.12 mmol), the secondary amine (2, 0.1 mmol), KOH (3 equiv), H_2O (5 equiv), CH_3CN (2 mL), rt for 36 h under CICF₂H atmosphere; isolated yield. ^b **2a** 10 mmol.

Figure 2. Synthesis of Formimidamide Derivatives

(A) Scope of the primary amines.

(B) Scope of the secondary amines.

(C) Scope of R groups.

(D) Scope of the pharmaceutical molecules.



Reaction condition 1: the primary amine (1, 0.12 mmol), the secondary amine (2, 0.1 mmol), KOH (3 equiv), H_2O (5 equiv), CH_3CN (2 mL), rt for 36 h under $CICF_2H$ atmosphere; isolated yield. **Reaction condition 2**: aniline derivative (1, 0.2 mmol), K_2CO_3 (3 equiv), phenol (10 mol%), H_2O (5 equiv), CH_3CN (2 mL), 80 °C for 12 h under $CICF_2H$ atmosphere, isolated yield. **Reaction condition 3**: aniline derivative (1, 0.2 mmol), Cs_2CO_3 (3 equiv), S_8 (5 mol%), CH_3CN (5 mL), 80 °C for 26 h under $CICF_2H$ atmosphere, isolated yield.

Figure 3. High Regio- and Chemoselectivity of Our Strategy and Intriguing Products

(A) High regio-selectivity.

(B) The dispose of target product and the combination of two pharmaceutical molecules.

(C) Selective synthesis of formimidamides with two primary amines.

(D) Selective synthesis of N-difluoromethyl-formimidamides from two primary amines with ClCF₂H as both C1 source and difluoromethylating reagent.

(2ag) as the substrates under standard reaction *condition* 1; to our delight, the corresponding product 3ai was obtained in 70% yield (Equation 4). More interestingly, a highly chemoselective process was disclosed with two primary amines, in which N,N'-diphenylformimidamide and N-(difluoromethyl)-N,N'-diphenylformimidamide were obtained, respectively, by careful control of reaction conditions. Bases and additives played key roles on these two successful transformations: with K₂CO₃ as base, phenol and water as

additives (see *condition 2* in Figure 3 and for further details, see also Table S4), N,N'-diphenylformimidamides (8 and 9) were obtained in moderate yields; with Cs₂CO₃ as the base and S₈ as additive (see *condition 3* in Figure 3, and for further details, see also Table S5) (Zheng et al., 2017), N-(difluoromethyl)-N,N'diphenylformimidamides (10–15) were acquired in good yields. In the latter transformation, CICF₂H played a dual role as both C1 source and difluorocarbene source (Figure 3C).

Substrate Scope for Intramolecular Transformation

The success of the above-mentioned intermolecular transformation prompted us to exploit the intramolecular transformations, since the latter one always leads to cyclic compounds that are the essential skeletons in pharmaceutical and natural products (Sasaki et al., 2006; Kubo et al., 1993). Gratifyingly, when N¹-methylbenzene-1,2-diamine (16) was subjected to the standard conditions for intermolecular transformation, benzimidazole 17 was obtained in 90% yield, which could be readily converted into 2-bromo-benzimidazole 18 in the presence of NBS. Then, after a series of transformation, Telmisartan, a potent angiotensin Il receptor antagonist used in the treatment of essential hypertension, will be afforded (Figure 4A) (Martin et al., 2015). In addition, the transformation could be easily scaled up to 70 times from 16 to 17 without loss of the efficiency (for details, see also Scheme S6). Encouraged by this promising result, we next focused on the exploration of the formation of benzo[d]oxazoles and 1H-benzo[d]imidazoles compounds via intramolecular pattern, since it is well known that benzo[d]oxazoles and 1H-benzo[d]imidazoles are prevalent molecular scaffolds in various bioactive natural products, agrochemicals, and pharmaceuticals. After many attempts, an optimized condition was obtained (for further details, see also Tables S6 and S7). These transformations demonstrated a good functional group tolerance (Figure 4B). Different substituent groups on the benzene ring, including alkyl (19a, 19b), halo groups (19c, 19d), were all compatible, rendering the corresponding products (20a-20d) in moderate to good yield (68%-81%). Surprisingly, no desired products were detected when 2-amino-4-nitrophenol (19e) and 2-amino-5-nitrophenol (19f) underwent the same conditions, instead, the selective difluoromethylation of hydroxyl group occurred (20e' and 20f'). Good yields were achieved on various benzene-1,2-diamine compounds under the standard Reaction Condition 5 with K₂CO₃ as base in CH₃CN (2 mL) and H₂O (0.5 mL) at 100°C for 16 h (Figure 4C). Remarkably, the products of difluoromethylation of benzimidazoles (21-29) were acquired in moderate yields via the slight adjustment of reaction condition (see *reaction condition 6* for details); once again, $CICF_2H$ played a dual role as both C1 source and difluorocarbene source in this transformation (Figure 4D).

Mechanism Investigation

To gain more insights into the mechanism of the aforementioned transformations, some control experiments were performed. Initially, isotope labeling experiments were conducted, 84% (3a') and 78% (20g') of D atoms were incorporated into the final products correspondingly for intermolecular and intramolecular versions, and N-H of benzimidazole was replaced by N-D completely (Figures 5A and 5B). These results suggested that the hydrogen atom attached on the extra introduced carbon (from CICF₂H) was originated from H₂O in this process possibly (for further details, see also Scheme S8). The trace amount of the desired product **3a** was observed when benzimidazole was added into the reaction system as a difluorocarbene scavenger; instead, 1-(difluoromethyl)-1H-benzo[d]imidazole was detected by GC-MS (Figure 5C). When N-methyl-N-phenylformamide (30) or isocyanobenzene (31) was subjected to the earlier standard reaction conditions with amines, the corresponding target products 3a and 8 were not obtained (Figure 5D), indicating that the compounds 30 and 31 are not intermediates for this transformation, which is in sharp contrast to our previous reports in which isocyanides are the key intermediates for those transformations with BrCF₂COOEt (Ma et al., 2018c, 2018d). To thoroughly understand the reaction sequence, two more control experiments were carried out, in which the primary amine and the secondary amine were added to the reaction mixture stepwise instead of one-pot to check which one is the first amine species interacting with CICF₂H. It turned out that primary amine might react with CICF₂H before the secondary one since 47% of the desired product was obtained in the primary-secondary amine sequence, whereas no desired product was detected with the secondary-primary amine sequence (Figures 5E and 5F). Finally, we carried out comparison experiments for CO₂, CO and HCOOH as C1 source with amines 1a and 2a under various reaction conditions; no desired product 3a was detected (Figure 5G), which further highlighted the uniqueness of CICF₂H as the C1 source in these transformations.

Proposed Mechanism

To thoroughly figure out the possible reactive intermediate, we carried out *in situ* ¹H NMR studies between 4-ethoxyaniline (1c) and $ClCF_2H$ (Figure 6A). Since isocyanides have been ruled out to be the possible

CellPress



^a The scale of the original material **16** is 0.1 mmol; ^b The scale of the original material **16** is 7 mmol. (a) 1-Methylbenzimidazole (**17**) (5 mmol) and *N*-bromosuccinimide (3 equiv) in 25 mL of THF were heated under reflux for 1 h. *Condition 4*: The amine (0.2 mmol), K₂CO₃ (3 equiv), H₂O (5 equiv.), CH₃CN (2 mL), 50 °C for 12 h under the atmosphere of CICF₂H, isolated yield; *Condition 5*: The amine (0.2 mmol), K₂CO₃ (3 equiv), H₂O (0.5 mL), CH₃CN (2 mL), 100 °C for 16 h under the atmosphere of CICF₂H, isolated yield; *Condition 5*: The amine (0.2 mmol), K₂CO₃ (3 equiv), H₂O (0.5 mL), CH₃CN (2 mL), 100 °C for 16 h under the atmosphere of CICF₂H, isolated yield. *Condition 6*: The amine (0.1 mmol), K₂CO₃ (5 equiv), H₂O (30 µL), CH₃CN (2 mL), 110 °C for 48 h under the atmosphere of CICF₂H.

Figure 4. The Synthetic Route of the Telmisartan and the Intramolecular Reaction Scope

- (A) Telmisartan synthesis with our strategy.
- (B) Scope of benzoxazoles.
- (C) Scope of benzimidazoles.

(D) Scope of N-difluoromethyl benzimidazoles.

CellPress



Figure 5. Mechanistic Studies

intermediates for this transformation, we envision that a type of intermediate 3c' might be formed in this transformation. To our delight, *in situ* ¹H NMR studies indeed indicated the formation of (*E*)-*N*-(4-ethoxyphenyl)formimidoyl fluoride (3c'), which was increased continually at the first 6 h, whereafter it started to decline probably owing to its volatile property and the existence of various nucleophiles in the reaction system, such as H₂O and amines. The intermediate 3c' was totally consumed after 18 h or during the process of striping the solvent (for further details, see also Schemes S9–S11). To validate its presence, various nucleophiles (phenols, alcohols, amines, and carboxylic acids) were added into the system after 2 h, and the corresponding desired products were detected in GC-MS (Equation 1 in Figure 6B, and for further details, Scheme S12). In addition, one more control experiment was carried out with a CICF₂H balloon for 5 h;



Figure 6. In situ ¹H NMR, the Capture of Reaction Intermediates and the Proposed Mechanism

(A) In situ ¹H NMR studies.

(B) The capture of reaction intermediates.

(C) The plausible reaction mechanism.

the product 32, isocyanide 33, and N-(4-ethoxyphenyl)formamide (34) were obtained in 15%, 27%, and 38% yields, respectively (Equation 2 in Figure 6B), suggesting that the intermediate is a chemically active compound, which will further decompose into isocyanide by one more C-F bond cleavage easily. We have carried out the control experiments in the presence of radical scavengers; the reactions proceeded smoothly at room temperature to afford desired products in moderate yields. Those results suggest that the single electron transfer (SET) pathway could not be involved in this transformation (for further details, see also Scheme S13). On the basis of the above-mentioned results, a proposed mechanism for the reaction of CICF₂H as a C1 source is depicted in Figure 6C. The base coordinates with CICF₂H to generate difluorocarbene first. Then the primary amine traps the in situ generated difluorocarbene affording intermediate I, which is very sensitive under basic conditions to lead to monofluoroimine species via the cleavage of one C-F bond; subsequent inter- or intramolecular nucleophilic attack on the imine species (II and III) eventually delivers products 3, 5, 7, and 20 by S_NAr substitution (path a) or nucleophilic addition (path b) (Ma et al., 2018a). As either R² or R³ is H, the product could embark on capturing one more in situ generated difluorocarbene unceasingly to render the products (10-15). The products 21-29 were obtained in one-pot synthesis when compound 20 (X = NH) meets with excess $CICF_2H$ in basic conditions as a difluorocarbene scavenger.

Conclusion

In summary, we have disclosed a C1 source generated from chlorodifluoromethane (CICF₂H). This method allows the synthesis of a broad range of the formimidamides and benzo[*d*]oxazoles, benzo[*d*]imidazole derivatives via intermolecular and intramolecular reactions with good efficiency as well as high regio- and chemoselectivity under mild reaction condition. To our knowledge, this is the first example that CICF₂H proceeds quadruple cleavage to act as a C1 synthon and the valuable products were fabricated from readily available starting materials under transition-metal-free conditions. This process might enrich C1 chemistry, green chemistry, and fluorine chemistry as well as might partially solve the problem of the disposition of ODS. Preliminary mechanistic studies revealed that (*E*)-*N*-phenylformimidoyl fluoride intermediate is involved in this process, which is a distinct intermediate from BrCF₂COOEt case. Further studies toward the detailed mechanism and transformations and applications as well as exploration on more intriguing methodologies with this unusual C1 source are under way in our laboratory.

Limitation of the Study

Primary aliphatic amines showed poor or no reactivity toward this reaction system. In addition, reactive intermediate (e.g., **3c'**) was not isolated owing to its high reactivity.

METHODS

All methods can be found in the accompanying Transparent Methods supplemental file.

DATA AND CODE AVAILABILITY

The structures of **3t** reported in this article have been deposited in the Cambridge Crystallographic Data Centre under accession numbers CCDC: 1874971.

SUPPLEMENTAL INFORMATION

Supplemental Information can be found online at https://doi.org/10.1016/j.isci.2019.07.005.

ACKNOWLEDGMENTS

Financial support from the National Natural Science Foundation of China (21772046) and the Natural Science Foundation of Fujian Province (2016J01064) are gratefully acknowledged. We also thank Instrumental Analysis Center of Huaqiao University for analysis support. X.M. thanks the Subsidized Project for Cultivating Postgraduates' Innovative Ability in Scientific Research of Huaqiao University. Finally, we also thank Zhuoni Xie for reproducing the results of **3a**, **3c**, **4a**, **4b**, and **11**.

AUTHOR CONTRIBUTIONS

X.M. and J.S. performed the experiments and developed the reactions. X.Z. checked the manuscript and came up with suggestions for this transformation. Q.S. designed and directed the project and wrote the manuscript with the feedback of X.M.

DECLARATION OF INTERESTS

The authors declare no competing interests.

Received: May 9, 2019 Revised: June 9, 2019 Accepted: July 2, 2019 Published: September 27, 2019

REFERENCES

Aresta, M., Dibenedetto, A., and Angelini, A. (2014). Catalysis for the valorization of exhaust carbon: from CO_2 to chemicals, materials, and fuels. Technological use of CO_2 . Chem. Rev. 114, 1709–1742.

Bitterlich, B., Anilkumar, G., Gelalcha, G.G., Spilker, B., Grotevendt, A., Jackstell, A., Tse, M.K., and Beller, M. (2007). Development of a general and efficient iron-catalyzed epoxidation with hydrogen peroxide as oxidant. Chem. Asian J. 2, 521–529.

Boogaerts, I.I.F., and Nolan, S.P. (2010). Carboxylation of C-H bonds using N-heterocyclic carbene gold(I) complexes. J. Am. Chem. Soc. 132, 8858–8859.

Brunet, J.J., and Neibecker, D. (2001). Catalytic hydroamination of unsaturated carbon-carbon bonds. In Catalytic Heterofunctionalization from Hydroamination to Hydrozirconation, A. Togni and H. Grützmacher, eds. (Wiley-VCH Verlag GmbH), pp. 91–141.

Cokoja, M., Bruckmeier, C., Rieger, B., Herrmann, W.A., and Kihn, F.E. (2011). Transformation of carbon dioxide with homogeneous transitionmetal catalysts: a molecular solution to a global challenge? Angew. Chem. Int. Ed. 50, 8510–8537.

Deng, S., Chen, H., Ma, X., Zhou, Y., Lan, Y., and Song, Q. (2019). S₈-Catalyzed triple cleavage of bromodifluoro compounds for the assembly of N-containing heterocycles. Chem. Sci. https://doi.org/10.1039/c9sc01333d.

Enthaler, S., von Langermann, J., and Schmidt, T. (2010). Carbon dioxide and formic acid the couple for environmental-friendly hydrogen storage? Energy Environ. Sci. 3, 1207–1217.

Feng, Z., Min, O.Q., Fu, X.P., An, L., and Zhang, X. (2017). Chlorodifluoromethane triggered formation of difluoromethylated arenes catalysed by palladium. Nat. Chem. *9*, 918–923.

Fier, P.S., and Hartwig, J.F. (2012). Coppermediated difluoromethylation of aryl and vinyl iodides. J. Am. Chem. Soc. 134, 5524–5527.

France, S., Guerin, D.J., Miller, S.J., and Lectka, T. (2014). Nucleophilic chiral amines as catalysts in asymmetric synthesis. Chem. Rev. *103*, 2985– 3012.

Gibson, H.W. (1969). Chemistry of formic acid and its simple derivatives. Chem. Rev. *69*, 673–692.

Gouverneur, V., and Seppelt, K. (2015). Introduction: fluorine chemistry. Chem. Rev. *115*, 563–565.

Gu, Y., Leng, X., and Shen, Q. (2014). Cooperative dual palladium/silver catalyst for direct

difluoromethylation of aryl bromides and iodides. Nat. Commun. *5*, 5405–5412.

Hine, J., and Porter, J.J. (1957). Methylene derivatives as intermediates in polar reactions. VIII. Difluoromethylene in the reaction of chlorodifluoromethane with sodium methoxide. J. Am. Chem. Soc. 79, 5493–5496.

Hopkinson, M.N., Richter, C., Schedler, M., and Glorius, F. (2014). An overview of N-heterocyclic carbenes. Nature *510*, 485–496.

Horváth, I.T. (2007). Innovations and green chemistry. Chem. Rev. 107, 2169–2173.

Huang, K., Sun, C.-L., and Shi, Z.-J. (2011). Transition-metal-catalyzed C-C bond formation through the fixation of carbon dioxide. Chem. Soc. Rev. 40, 2435–2452.

M. Hudlicky, and A.E. Pavlath, eds. (1995). Chemistry of Organic Fluorine Compounds II: A Critical Review (DC American Chemical Society), pp. 1–1200.

Kubo, K., Imamiya, E., Sugiura, Y., Inada, Y., Furukawa, Y., Nishikawa, K., and Naka, T. (1993). Nonpeptide angiotensin II receptor antagonists. Synthesis and biological activity of benzimidazolecarboxylic acids. J. Med. Chem. 36, 2182–2195.

Liu, Q., Wu, L., Jackstell, R., and Beller, M. (2015). Using carbon dioxide as a building block in organic synthesis. Nat. Commun. 6, 5933.

Ma, X., Deng, S., and Song, Q. (2018a). Halodifluoroacetates as formylation reagents for various amines via unprecedented quadruple cleavage. Org. Chem. Front. *5*, 3505–3509.

Ma, X., Xuan, Q., and Song, Q. (2018b). N-H and O-H difluoromethylation of N-Heterocycles. Acta Chim. Sinica *76*, 972–976.

Ma, X., Zhou, Y., and Song, Q. (2018c). Synthesis of β -Aminoenones via cross-coupling of in-situ generated isocyanides with 1,3-Dicarbonyl compounds. Org. Lett. 20, 4777–4781.

Ma, X., Mai, S., Zhou, Y., Cheng, G.-J., and Song, Q. (2018d). Dual role of ethyl bromodifluoroacetate in the formation of fluorine-containing heteroaromatic compounds. Chem. Commun. (Camb.) 54, 8960–8963.

Martin, A.D., Siamaki, A.R., Belecki, K., and Gupton, B.F. (2015). A convergent approach to the total synthesis of telmisartan via a suzuki cross-coupling reaction between two functionalized benzimidazoles. J. Org. Chem. *80*, 1915–1919. Miao, W., Zhao, Y., Ni, C., Gao, B., Zhang, W., and Hu, J. (2018). Iron-catalyzed difluoromethylation of arylzincs with difluoromethyl 2-pyridyl sulfone. J. Am. Chem. Soc. 140, 880–883.

Natte, K., Neumann, H., Beller, M., and Jagadeesh, R.V. (2017). Transition-metalcatalyzed utilization of methanol as a C1 source in organic synthesis. Angew. Chem. Int. Ed. 56, 6384–6394.

Nawrot, E., and Jonczyk, A. (2007). Difluoromethyltrialkylammonium salts-their expeditious synthesis from chlorodifluoromethane and tertiary amines in the presence of concentrated aqueous sodium hydroxide. The catalytic process. J. Org. Chem. 72, 10258–10260.

Oh, Y., and Hu, X. (2013). Organic molecules as mediators and catalysts for photocatalytic and electrocatalytic CO_2 reduction. Chem. Soc. Rev. 42, 2253–2261.

O'Hagan, D. (2008). Understanding organofluorine chemistry. An introduction to the C-F bond. Chem. Soc. Rev. 37, 308–319.

Ohishi, T., Nishiura, M., and Hou, Z. (2008). Carboxylation of organoboronic esters catalyzed by N-heterocyclic carbene copper(I) complexes. Angew. Chem. Int. Ed. 47, 5792–5795.

Sakakura, T., Choi, J.-C., and Yasuda, H. (2007). Transformation of carbon dioxide. Chem. Rev. 107, 2365–2387.

Sasaki, H., Haraguchi, Y., Itotani, M., Kuroda, H., Hashizume, H., Tomishige, T., Kawasaki, M., Matsumoto, M., Komatsu, M., and Tsubouchi, H. (2006). Synthesis and antituberculosis activity of a novel series of optically active 6-Nitro-2,3dihydroimidazo[2,1-b]oxazoles. J. Med. Chem. *49*, 7854–7860.

Schröder, K., Enthaler, S., Join, B., Junge, K., and Bellera, M. (2010). Iron-catalyzed epoxidation of aromatic olefins and 1,3-dienes. Adv. Synth. Catal. 352, 1771–1778.

Schröder, K., Enthaler, K., Bitterlich, B., Schulz, T., Spannenberg, A., Tse, M.K., Junge, K., and Beller, M. (2009). Design of and mechanistic studies on a biomimetic iron-imidazole catalyst system for epoxidation of olefins with hydrogen peroxide. Chem. Eur. J. 15, 5471–5481.

Senadi, G.C., Kudale, V.S., and Wang, J.-J. (2019). Sustainable methine sources for the synthesis of heterocycles under metal- and peroxide-free conditions. Green Chem. 21, 979–985.

Sordakis, K., Tang, C., Vogt, L.K., Junge, H., Dyson, P.J., Beller, M., and Laurenczy, G. (2018). Homogeneous catalysis for sustainable hydrogen

CellPress

storage in formic acid and alcohols. Chem. Rev. 118, 372–433.

Sung, D., Moon, D., Lee, Y., and Hong, S.-I. (2004). Catalytic pyrolysis of difluorochloromethane to produce tetrafluoroethylene. Int. J. Chem. React. Eng. 2, 1542–6580.

Wakade, S.B., Tiwari, D.K., Prabhakar Ganesh, P.S.K., Phanindrudu, M., Likhar, P.R., and Tiwari, D.K. (2017). Transition-metal-free quinoline synthesis from acetophenones and anthranils via sequential one-carbon homologation/conjugate addition/annulation cascade. Org. Lett. *19*, 4948– 4951.

Wang, J., Sánchez-Roselló, K., Aceña, J.L., del Pozo, J., Sorochinsky, A.E., Fustero, S., Soloshonok, V.A., and Liu, H. (2014). Fluorine in pharmaceutical industry: fluorine-containing drugs introduced to the market in the last decade (2001-2011). Chem. Rev. 114, 2432–2506.

Wu, L., Wang, F., Chen, P., and Liu, G. (2019). Enantioselective construction of quaternary allcarbon centers via copper-catalyzed arylation of tertiary carbon-centered radicals. J. Am. Chem. Soc. 141, 1887–1892.

Xu, C., Guo, W.-H., He, X., Gao, Y.-L., Zhang, X.-Y., and Zhang, X. (2018). Difluoromethylation of (hetero)aryl chlorides with chlorodifluoromethane catalyzed by nickel. Nat. Commun. 9, 1170–1179.

Yan, Y., Zhang, Y., Feng, C., Zha, Z., and Wang, Z. (2012). Selective iodine-catalyzed intermolecular oxidative amination of C(sp³)-H bonds with *ortho*carbonyl-substituted anilines to give quinazolines. Angew. Chem. Int. Ed. *51*, 8077– 8081.

Yu, J., Lin, J.-H., and Xiao, J.-C. (2017). Reaction of thiocarbonyl fluoride generated from difluorocarbene with amines. Angew. Chem. Int. Ed. *129*, 16896–16900.

Yu, C., Su, J., Ma, X., Zhou, Y., and Song, Q. (2019). Difluoromethylation of tosylhydrazone compounds with chlorodifluoromethane under mild conditions. Asian J. Org. Chem. 8, 694–697. Zhang, K.-F., Bian, K.-J., Li, C., Sheng, J., Li, Y., and Wang, X.-S. (2019). Nickel-catalyzed carbofluoroalkylation of 1,3-enynes to access structurally diverse fluoroalkylated allenes. Angew. Chem. Int. Ed. *58*, 5069–5074.

Zhao, P., Krug, C., and Hartwig, J.F. (2005). Transfer of amido groups from isolated rhodium(i) amides to alkenes and vinylarenes. J. Am. Chem. Soc. 127, 12066–12073.

Zheng, J., Lin, J.-H., Deng, X.-Y., and Xiao, J.-C. (2015). 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU)-Promoted decomposition of difluorocarbene and the subsequent trifluoromethylation. Org. Lett. 17, 532–535.

Zheng, J., Cheng, R., Lin, J.-H., Yu, D.-H., Ma, L., Jia, L., Zhang, L., Wang, L., Xiao, J.-C., and Liang, S.H. (2017). An unconventional mechanistic insight into SCF₃ formation from difluorocarbene: preparation of ¹⁸F-Labeled α -SCF₃ carbonyl compounds. Angew. Chem. Int. Ed. 56, 3196– 3200. ISCI, Volume 19

Supplemental Information

Chlorodifluoromethane as a C1 Synthon

in the Assembly of *N*-Containing Compounds

Xingxing Ma, Jianke Su, Xingang Zhang, and Qiuling Song

Supplementary Figures



Figure S1. ¹H NMR spectrum of 3a, related to Figure 2



Figure S2. ¹³C NMR spectrum of 3a, related to Figure 2



Figure S3. ¹H NMR spectrum of 3b, related to Figure 2



Figure S4. ¹³C NMR spectrum of 3b, related to Figure 2



Figure S5. ¹H NMR spectrum of 3c, related to Figure 2



Figure S6. ¹³C NMR spectrum of 3c, related to Figure 2



Figure S7. ¹H NMR spectrum of 3d, related to Figure 2



Figure S8. ¹³C NMR spectrum of 3d, related to Figure 2



Figure S9. ¹H NMR spectrum of 3e, related to Figure 2



Figure S10. ¹³C NMR spectrum of 3e, related to Figure 2



Figure S11. ¹H NMR spectrum of 3f, related to Figure 2



Figure S12. ¹³C NMR spectrum of 3f, related to Figure 2



Figure S13. ¹H NMR spectrum of 3g, related to Figure 2



Figure S14. ¹³C NMR spectrum of 3g, related to Figure 2



Figure S15. ¹H NMR spectrum of 3h, related to Figure 2



Figure S16. ¹³C NMR spectrum of 3h, related to Figure 2



Figure S17. ¹H NMR spectrum of 3i, related to Figure 2



Figure S18. ¹³C NMR spectrum of 3i, related to Figure 2



Figure S19. ¹H NMR spectrum of 3j, related to Figure 2



Figure S20. ¹³C NMR spectrum of 3j, related to Figure 2



Figure S21. ¹H NMR spectrum of 3k, related to Figure 2



Figure S22. ¹³C NMR spectrum of 3k, related to Figure 2



Figure S23. ¹H NMR spectrum of 3l, related to Figure 2



Figure S24. ¹³C NMR spectrum of 3l, related to Figure 2



Figure S25. ¹H NMR spectrum of 3m, related to Figure 2



Figure S26. ¹³C NMR spectrum of 3m, related to Figure 2



Figure S27. ¹H NMR spectrum of 3n, related to Figure 2



Figure S28. ¹³C NMR spectrum of 3n, related to Figure 2



Figure S29. ¹H NMR spectrum of 30, related to Figure 2



Figure S30. ¹³C NMR spectrum of 30, related to Figure 2



Figure S31. ¹H NMR spectrum of 3p, related to Figure 2



Figure S32. ¹³C NMR spectrum of 3p, related to Figure 2



Figure S33. ¹H NMR spectrum of 3q, related to Figure 2



Figure S34. ¹³C NMR spectrum of 3q, related to Figure 2



Figure S35. ¹H NMR spectrum of 3r, related to Figure 2



Figure S36. ¹³C NMR spectrum of 3r, related to Figure 2



Figure S37. ¹H NMR spectrum of 3s, related to Figure 2



Figure S38. ¹³C NMR spectrum of 3s, related to Figure 2



Figure S39. ¹H NMR spectrum of 3t, related to Figure 2



Figure S40. ¹³C NMR spectrum of 3t, related to Figure 2



Figure S41. ¹H NMR spectrum of 3u, related to Figure 2



Figure S42. ¹³C NMR spectrum of 3u, related to Figure 2


Figure S43. ¹H NMR spectrum of 3v, related to Figure 2



Figure S44. ¹³C NMR spectrum of 3v, related to Figure 2



Figure S45. ¹H NMR spectrum of 3w, related to Figure 2



Figure S46. ¹³C NMR spectrum of 3w, related to Figure 2



Figure S47. ¹H NMR spectrum of 3x, related to Figure 2



Figure S48. ¹³C NMR spectrum of 3x, related to Figure 2



Figure S49. ¹H NMR spectrum of 3y, related to Figure 2



Figure S50. ¹³C NMR spectrum of 3y, related to Figure 2



Figure S51. ¹H NMR spectrum of 3z, related to Figure 2



Figure S52. ¹³C NMR spectrum of 3z, related to Figure 2



Figure S53. ¹H NMR spectrum of 3aa, related to Figure 2



Figure S54. ¹³C NMR spectrum of 3aa, related to Figure 2



Figure S55. ¹H NMR spectrum of 3ab, related to Figure 2



Figure S56. ¹³C NMR spectrum of 3ab, related to Figure 2



Figure S57. ¹H NMR spectrum of 3ac, related to Figure 2



Figure S58. ¹³C NMR spectrum of 3ac, related to Figure 2



Figure S59. ¹H NMR spectrum of 3ad, related to Figure 2



Figure S60. ¹³C NMR spectrum of 3ad, related to Figure 2



Figure S61. ¹H NMR spectrum of 3ae, related to Figure 2



Figure S62. ¹³C NMR spectrum of 3ae, related to Figure 2



Figure S63. ¹H NMR spectrum of 3af, related to Figure 2



Figure S64. ¹³C NMR spectrum of 3af, related to Figure 2



Figure S65. ¹H NMR spectrum of 3ag, related to Figure 2



Figure S66. ¹³C NMR spectrum of 3ag, related to Figure 2



Figure S67. ¹H NMR spectrum of 3ah, related to Figure 2



Figure S68. ¹³C NMR spectrum of 3ah, related to Figure 2



Figure S69. ¹H NMR spectrum of 5, related to Figure 3



Figure S70. ¹³C NMR spectrum of 5, related to Figure 3



Figure S71. ¹H NMR spectrum of 7, related to Figure 3



Figure S72. ¹³C NMR spectrum of 7, related to Figure 3



Figure S73. ¹H NMR spectrum of 3ai, related to Figure 3



Figure S74. ¹³C NMR spectrum of 3ai, related to Figure 3



Figure S75. ¹H NMR spectrum of 8, related to Figure 3



Figure S76. ¹³C NMR spectrum of 8, related to Figure 3



Figure S77. ¹H NMR spectrum of 9, related to Figure 3



Figure S78. ¹³C NMR spectrum of 9, related to Figure 3



Figure S79. ¹³C NMR spectrum of 10, related to Figure 3



Figure S80. ¹H NMR spectrum of 10, related to Figure 3



Figure S81. ¹⁹F NMR spectrum of 10, related to Figure 3



Figure S82. ¹³C NMR spectrum of 11, related to Figure 3



Figure S83. ¹H NMR spectrum of 11, related to Figure 3



Figure S84. ¹⁹F NMR spectrum of 11, related to Figure 3



Figure S85. ¹H NMR spectrum of 12, related to Figure 3



Figure S86. ¹³C NMR spectrum of 12, related to Figure 3



Figure S87. ¹⁹F NMR spectrum of 12, related to Figure 3



Figure S88. ¹H NMR spectrum of 13, related to Figure 3



Figure S89. ¹³C NMR spectrum of 13, related to Figure 3



Figure S90. ¹H NMR spectrum of 13, related to Figure 3



Figure S91. ¹H NMR spectrum of 14, related to Figure 3



Figure S92. ¹³C NMR spectrum of 14, related to Figure 3



Figure S93. ¹⁹F NMR spectrum of 14, related to Figure 3



Figure S94. ¹H NMR spectrum of 15, related to Figure 2



Figure S95. ¹³C NMR spectrum of 15, related to Figure 3



Figure S96. ¹⁹F NMR spectrum of 15, related to Figure 3



Figure S97. ¹H NMR spectrum of 17, related to Figure 4



Figure S98. ¹³C NMR spectrum of 17, related to Figure 4



Figure S99. ¹H NMR spectrum of 20a, related to Figure 4



Figure S100. ¹³C NMR spectrum of 20a, related to Figure 4



Figure S101. ¹H NMR spectrum of 20b, related to Figure 4



Figure S102. ¹³C NMR spectrum of 20b, related to Figure 4



Figure S103. ¹H NMR spectrum of 20c, related to Figure 4



Figure S104. ¹³C NMR spectrum of 20c, related to Figure 4



Figure S105. ¹H NMR spectrum of 20d, related to Figure 4



Figure S106. ¹³C NMR spectrum of 20d, related to Figure 4



Figure S107. ¹H NMR spectrum of 20e', related to Figure 4



Figure S108. ¹³C NMR spectrum of 20e', related to Figure 4



Figure S109. ¹⁹F NMR spectrum of 20e', related to Figure 4



Figure S110. ¹H NMR spectrum of 20f', related to Figure 4



Figure S111. ¹³C NMR spectrum of 20f', related to Figure 4



Figure S112. ¹⁹F NMR spectrum of 20f', related to Figure 4



Figure S113. ¹H NMR spectrum of 20g, related to Figure 4



Figure S114. ¹³C NMR spectrum of 20g, related to Figure 4


Figure S115. ¹H NMR spectrum of 20h, related to Figure 4



Figure S116. ¹³C NMR spectrum of 20h, related to Figure 4



Figure S117. ¹H NMR spectrum of 20i, related to Figure 4



Figure S118. ¹³C NMR spectrum of 20i, related to Figure 4



Figure S119. ¹H NMR spectrum of 20j, related to Figure 4



Figure S120. ¹³C NMR spectrum of 20j, related to Figure 4



Figure S121. ¹H NMR spectrum of 20k, related to Figure 4



Figure S122. ¹³C NMR spectrum of 20k, related to Figure 4



Figure S123. ¹H NMR spectrum of 20m, related to Figure 4



Figure S124. ¹³C NMR spectrum of 20m, related to Figure 4



Figure S125. ¹H NMR spectrum of 20n, related to Figure 4



Figure S126. ¹³C NMR spectrum of 20n, related to Figure 4



Figure S127. ¹H NMR spectrum of 200, related to Figure 4



Figure S128. ¹³C NMR spectrum of 200, related to Figure 4



Figure S129. ¹H NMR spectrum of 21, related to Figure 4



Figure S130. ¹³C NMR spectrum of 21, related to Figure 4



Figure S131. ¹⁹F NMR spectrum of 21, related to Figure 4



Figure S132. ¹H NMR spectrum of 22, related to Figure 4



Figure S133. ¹³C NMR spectrum of 22, related to Figure 4



Figure S134. ¹⁹F NMR spectrum of 22, related to Figure 4



Figure S135. ¹H NMR spectrum of 25, related to Figure 4



Figure S136. ¹³C NMR spectrum of 25, related to Figure 4



Figure S137. ¹⁹F NMR spectrum of 25, related to Figure 4



Figure S138. ¹H NMR spectrum of 26, related to Figure 4



Figure S139. ¹³C NMR spectrum of 26, related to Figure 4



Figure S140. ¹⁹F NMR spectrum of 26, related to Figure 4



Figure S141. ¹H NMR spectrum of 27, related to Figure 4



Figure S142. ¹³C NMR spectrum of 27, related to Figure 4



Figure S143. ¹⁹F NMR spectrum of 27, related to Figure 4



Figure S144. ¹H NMR spectrum of 28, related to Figure 4



Figure S145. ¹³C NMR spectrum of 28, related to Figure 4



Figure S146. ¹⁹F NMR spectrum of 28, related to Figure 4



Figure S147. ¹H NMR spectrum of 29, related to Figure 4



Figure S148. ¹³C NMR spectrum of 29, related to Figure 4



Figure S149. ¹⁹F NMR spectrum of 29, related to Figure 4



Figure S150. ¹H NMR spectrum of 32, related to Figure 6



Figure S151. ¹³C NMR spectrum of 32, related to Figure 6



Figure S152. ¹H NMR spectrum of 33, related to Figure 6



Figure S153. ¹³C NMR spectrum of 33, related to Figure 6



Figure S154. ¹H NMR spectrum of 34, related to Figure 6



Figure S155. ¹³C NMR spectrum of 34, related to Figure 6

Supplemental Item Legands

Table S1: Crystal data and structure refinement, related to Figure 2

Table S2. The effect of H₂O for this process, related to Table 1

Table S3. The effect of base and solvent for this reaction, related toTable 1

Table S4. The optimization experiment conditions of 8, related toFigure 3C.

Table S5. The optimization experiment conditions of 10, related toFigure 3D.

Table S6. The optimization experiment conditions of 20a, related toFigure 4B.

Table S7. The optimization experiment conditions of 20g, related toFigure 4C.

Transparent Methods

General Methods for Experiments

All chemicals were purchased from Adamas Reagent, Energy chemical company, Bide Pharmatech Ltd and Shang Fluoro company (ClCF₂H). Unless otherwise stated, all experiments were conducted in a sealed tube under ClCF₂H atmosphere. Reactions were monitored by TLC or GC-MS analysis. Flash column chromatography was performed over silica gel (200-300 mesh).

¹H-NMR and ¹³C-NMR spectra were recorded in CDCl₃ and DMSO-d₆ on a Bruker Avance III 500 MHz NMR spectrometer (500 MHz ¹H, 125 MHz ¹³C (CPD), 470 MHz ⁹F (CPD)) at room temperature. Chemical shifts were reported in ppm on the scale relative to CDCl₃ (δ = 7.26 for ¹H-NMR, δ = 77.00 for ¹³C-NMR) as an internal reference. Coupling constants (*J*) were reported in Hertz (Hz).

General Procedure for the Transformations of CO₂, CO and HCOOH as C1 sources.



Reaction Condition 1: the amine (1a, 0.12 mmol), *N*-metnyl amine (2a, 0.1 mmol), KOH (3 equiv), H₂O (5 equiv), CH₃CN (2 mL), (HCOOH 3 equiv) rt for 36 h under CO₂/CO atmosphere.

Reaction condition 2: the amine (**1a**, 0.12 mmol), *N*-metnyl amine (**2a**, 0.1 mmol), K₂CO₃ (3 equiv), phenol (10 mol%), H₂O (5 equiv), CH₃CN (2 mL), (HCOOH 3 equiv) 80 °C for 12 h under CO₂/CO atmosphere.

Reaction condition 3: the amine (**1a**, 0.12 mmol), *N*-metnyl amine (**2a**, 0.1 mmol), Cs₂CO₃ (3 equiv), S₈ (5 mol%), CH₃CN (5 mL), (HCOOH 3 equiv) 80 °C for 26 h under CO₂/CO atmosphere.

Reaction Condition 4: the amine (1a, 0.12 mmol), *N*-metnyl amine (2a, 0.1 mmol), K₂CO₃ (3 equiv), H₂O (5 equiv.), CH₃CN (2 mL), (HCOOH 3 equiv) 50 °C for 12 h under CO₂/CO atmosphere.

Reaction Condition 5: the amine (1a, 0.12 mmol), N-metnyl amine (2a, 0.1 mmol), K₂CO₃ (3 equiv), H₂O (0.5

mL), CH₃CN (2 mL), (HCOOH 3 equiv) 100 °C for 16 h under the atmosphere of ClCF₂H, isolated yield. *Reaction Condition 6*: the amine (**1a**, 0.12 mmol), *N*-metnyl amine (**2a**, 0.1 mmol), K₂CO₃ (5 equiv), H₂O (30 mL), CH₃CN (2 mL), (HCOOH 3 equiv) 110 °C for 48 h under CO₂/CO atmosphere.

Scheme S1. The transformations of CO₂, CO and HCOOH as C1 sources, related to Figure 1

No desired product **3a** was obtained when we carried out many experiments by using CO_2 , CO and HCOOH as C1 sythons under *reaction conditions* 1/2/3/4/5/6.



Scheme S2. the transformation for CO₂ and CO as C1 sources under transition metals, related to Figure 1

In addition, various transformations for using CO_2 and CO as C1 sources in presence of transition metals (TM = Pd, Rh, Ni, Zn, Al) were also carried out according to the reported literature procedures (Tlili et al., 2015; Huang et al., 2011), unfortunately, no desired product **3a** was detected.

General Procedure for Large-scale Reaction of the N-methyl Aniline (2a).



Scheme S3. Large-scale reaction of the N-methyl aniline (2a), related to Figure 2

Large-scale reaction of the N-methyl aniline

In a dried Schlenk round flask (1500 mL) were placed amine **1a** (12 mmol, 1.2 equiv, 1.1 g), *N*-methyl amine **2a** (10 mmol, 1 equiv, 1.07 g) and KOH (30 mmol, 3 equiv, 1.7 g). Then the flask was fulled with ClCF₂H. Whereafter the solvent was added into Schlenk tube by injector. The resulting mixture was stirred at room temperature for 4 days. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatograph (silica gel, Petroleum ether : Ethyl acetate = 50:1, v/v) to give the desired product **3a** (62%, 1.31 g).



Scheme S4. The experiment for the decomposition of target product, related to Figure 3B

The Decomposition of Target Product 3a

To a mixture of (*E*)-*N*-methyl-*N*,*N'*-diphenyl- formimidamide 3a (0.2 mmol) in MeOH (2 mL), 1 M HCl was added to the seal tube. The resulting mixture was stirred at 80 °C for 3 h. Upon completion of the reaction, the compounds 1a, 2a, 1a-1 and 2a-1 were detected via TLC and GC-MS.

General Procedure for the synthesis of 3, 8-9 and 10-15.



Scheme S5. General process for the synthesis of 3, 8-9 and 10-15, related to Figure 2, Figure 3C and Figure 3D.

Preparations of target product 3

In a dried Schlenk tube were placed the primary amines 1 (0.12 mmol, 1.2 equiv), the secondary aniline 2 (0.1 mmol), KOH (0.3 mmol, 3 equiv) and H₂O (0.5 mmol, 5 equiv). Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (2 mL) were added into the mixtures via an injector. The resulting mixture was stirred at room temperature for 36 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 80:1, v/v) to give the desired product (**3**).

Preparations of target product 8 and 9

In a dried Schlenk tube were placed the anilines **1** (0.2 mmol, 1.2 equiv), K_2CO_3 (0.6 mmol, 3 equiv), phenol (0.02 mmol, 10 mol%) and H_2O (1 mmol, 5 equiv), Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (2 mL) were added into the mixtures via an injector. The resulting mixtures

was stirred at 80 °C for 12 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 50:1, v/v) to give the desired product (8-9).

Preparations of target product 10-15

In a dried Schlenk tube were placed the aniline **1** (0.2 mmol), Cs_2CO_3 (0.6 mmol, 3 equiv), S_8 (0.02 mmol, 10 mol%) and H₂O (1 mmol, 5 equiv), Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (5 mL) is added into the mixtures via an injector. The resulting mixture was stirred at 80 °C for 26 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 100:1, v/v) to give the desired product (**10-15**).

General Procedure for the Large-scale Synthesis of 18



Scheme S6. General process for the large-scale synthesis of 18, related to Figure 4A.

Preparations of target product 17

In a dried Schlenk tube were placed N^1 -methylbenzene-1,2-diamine **16** (7 mmol), KOH (21 mmol, 3 equiv) and H₂O (35 mmol, 5 equiv), Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and solvent is added into the mixtures via an injector. The resulting mixture was stirred at room temperature for 48 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 10:1, v/v) to give the desired product **17** in 78% yield.

Preparations of target product 18

Methylbenzimidazole (17) (5 mmol) and *N*-bromosuccinimide (15 mmol) in 30 mL of THF were heated under reflux for 1 h. The solvent was removed by a rotary evaporator, and the residue was recrystallized from EtOAc to afford **18** in 90% yield as a white solid.

General process for the synthesis of 20 and 21-29.



Scheme S7. General process for the synthesis of 20 and 21-29, related to Figure 4B-4D.

Preparations of target products 20a-20d, 20e' and 20f'

In a dried Schlenk tube were placed 2-aminophenol compounds (19a-19f) (0.2 mmol), K_2CO_3 (0.6 mmol, 3 equiv) and H_2O (1 mmol, 5equiv). Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (2 mL) is added into the mixtures via an injector. The resulting mixture was stirred at 50 °C for 12 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatographyy (silica gel, Petroleum ether : Ethyl acetate = 100:1, v/v) to give the desired product.

Preparations of target products 20g-20o

In a dried Schlenk tube were placed benzene-1,2-diamine compounds (0.2 mmol), K_2CO_3 (0.6 mmol, 3 equiv) and H_2O (0.5 mL). Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (2 mL) is added into the mixtures via an injector. The resulting mixture was stirred at 100 °C for 16 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 10:1, v/v) to give the desired product.

Preparations of target products 21-29

In a dried Schlenk tube were placed benzene-1,2-diamine compounds (0.1 mmol), K_2CO_3 (0.5 mmol, 5 equiv) and H_2O (0.5 mL). Then the tube was vacuumized for removing the air. Subsequently, chlorodifluoromethane and CH₃CN (2 mL) is added into the mixtures via an injector. The resulting mixture was stirred at 100 °C for 48 h. Upon completion of the reaction, the solvent was evaporated under reduced pressure and the residue was purified by flash column chromatography (silica gel, Petroleum ether : Ethyl acetate = 50:1, v/v) to give the desired product **21-29**.

The experiment for H-scrambling of the target product 3a



Scheme S8. The experiment for H-scrambling of the target product 3a, related to Figure 5a-5b

We carried out a H-scrambling experiment by exposing the product 3a to the standard condition in the presence of D₂O. No corresponding deuterium-labeling product 3a-D was detected (Scheme S8).

In-situ ¹H NMR of 3c'



Scheme S9. In-situ ¹H NMR of 3c', related to Figure 6A.

General Procedure for In-situ ¹H NMR of 3c'

There still is plentiful ClCF₂H dissolved in solvent (CD₃CN) in the first couple of hours, which caused a problem to detect compound **3c'** by NMR analysis (Scheme S9-S11). In order to see ¹H peak of compound **3c'** more clearly, we carried out an experiment for about 1 hour and stripped the excess ClCF₂H at low temperature, the

resulting mixture was analyzed by in situ NMR. The below figure has shown the change of possible reactive intermediate at different time.



Scheme S10. In-situ ¹H NMR of 3c', related to Figure 6A.



Scheme S11. the change of possible reactive intermediate at different time, related to Figure 6A.



The experiments for capturing of reaction intermediate 3c'

^a HRMS (ESI, m/z) found; ^b HRMS (ESI, m/z) calcd

Scheme S12. Various nucleophiles for capturing 3c', related to Figure 6B.

General Procedure for various nucleophiles for capturing 3c'

In order to validate the presence of the compound 3c', various nucleophiles, such as phenols, 1,1,2,3,3,3-hexafluoropropan-1-ol, benzoic acid, hexanoic acid and diphenylmethanolwere added into the system after 2 h (entries 1-5) , the corresponding desired products were detected by GC-MS (MW : 241, 329, 269, 263 and 331). In addition, we conducted the tests of HRMS (ESI, m/z). Delightedly, we detected corresponding m/z of various anticipated products (Scheme S12).





Scheme S13. The experiments for capturing of radical, related to Figure 6C.

We have carried out the control experiments in presence of radical scavengers (TEMPO, BHT, ethene-1,1-diyldibenzene and (1-cyclopropylvinyl)benzene), the reactions proceeded smoothly at room temperature to afford desired products in moderate yields. Those results suggest the SET pathway could not be involved in this transformation.

Characterization data for products

(E)-N-methyl-N,N'-diphenylformimidamide (3a) (CAS number:

32189-59-6)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 88%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.41 – 7.35 (m, 2H), 7.34 – 7.29 (m,

2H), 7.20 – 7.12 (m, 3H), 7.11 – 7.00 (m, 3H), 3.52 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 151.5, 151.2, 145.1, 129.5, 129.1, 124.1, 123.4, 121.3, 119.9, 34.1.

(E)-N'-(4-methoxyphenyl)-N-methyl-N-phenylformimidamide (3b)



The reaction was performed following the general

procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20 mg, 82%).¹H NMR (500 MHz, CDCl₃) δ 8.09 (s, 1H), 7.37 (dd, *J* = 8.6, 7.4 Hz, 2H), 7.18 – 7.11 (m, 3H), 7.01 – 6.96 (m, 2H), 6.86 (d, *J* = 8.9 Hz, 2H), 3.80 (s, 3H), 3.50 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 156.1, 150.7, 145.2, 144.8, 129.5, 123.9, 121.9, 119.7, 114.4, 55.5, 34.0.

HRMS (ESI, m/z) calcd for C₁₅H₁₆N₂O[M+H]⁺: 241.1335; found: 241.1337

(E)-N'-(4-ethoxyphenyl)-N-methyl-N-phenylformimidamide (3c)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (22 mg, 85%).¹H NMR (500 MHz, CDCl₃) δ 8.09 (s, 1H), 7.37 (dd, *J* = 8.4, 7.5 Hz, 2H), 7.20 –

7.10 (m, 3H), 7.02 – 6.94 (m, 2H), 6.91 – 6.81 (m, 2H), 4.01 (q, J = 7.0 Hz, 2H), 3.50 (s, 3H), 1.41 (t, J = 7.0 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 155.5, 150.6, 145.2, 144.6, 129.5, 123.99, 121.9, 119.7, 115.1, 63.7, 34.1, 15.0. HRMS (ESI, m/z) calcd for C₁₆H₁₉N₂O[M+H]⁺: 255.1492; found: 255.1491.

(E)-N-methyl-N'-(4-phenoxyphenyl)-N-phenylformimidamide (3d)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (24 mg, 81%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.41 – 7.35 (m, 2H), 7.34 – 7.28 (m, 2H), 7.19

- 7.13 (m, 3H), 7.09 - 6.96 (m, 7H), 3.52 (s, 3H). ¹³C NMR (125MHz, CDCl₃) δ 158.22 (s), 1529, 151.0, 147.4, 145.1, 129.6, 124.2, 122.6, 122.2, 120.3, 119.9, 118.0, 34.1.

HRMS (ESI, m/z) calcd for C₂₀H₁₉N₂O[M+H]⁺: 303.1492; found: 303.1491

(E)-N-methyl-N'-(4-(methylthio)phenyl)-N-phenylformimidamide

(**3e**)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (21 mg, 83%). ¹H NMR (500 MHz, CDCl₃) δ 8.08 (s, 1H), 7.38 (dd, *J* = 8.6, 7.4 Hz, 2H), 7.26 – 7.22 (m,

2H), 7.19 – 7.12 (m, 3H), 7.03 – 6.95 (m, 2H), 3.51 (s, 3H), 2.47 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 151.0, 149.3, 145.0, 132.1, 129.5, 128.8, 124.3, 121.8, 120.0, 34.2, 17.2.

HRMS (ESI, m/z) calcd for $C_{15}H_{17}N_2S[M+H]^+$: 257.1107; found: 257.1109.

(E)-N-methyl-N-phenyl-N'-(p-tolyl)formimidamide (3f)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a vellow oil (19 mg, 86%). ¹H NMR (500 MHz, CDCl₃) δ 8.10 (s, 1H), 7.42 -7.33 (m, 2H), 7.20 - 7.06 (m, 5H), 7.00 - 6.90 (m, 2H),

3.51 (s, 3H), 2.33 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 150.9, 149.0 145.2, 132.8, 129.7, 129.5, 124.0, 121.0, 119.8, 34.1, 20.9.

HRMS (ESI, m/z) calcd for C₁₅H₁₇N₂[M+H]⁺: 225.1386; found: 225.1388.

(E)-N'-(4-ethylphenyl)-N-methyl-N-phenylformimidamide (3g)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20 mg, 83%). ¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.41 – 7.30 (m, 2H), 7.20 – 7.08 (m, 5H), 7.03 – 6.92

(m, 2H), 3.51 (s, 3H), 2.63 (q, J = 7.6 Hz, 2H), 1.24 (t, J = 7.6 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 150.9, 149.1, 145.2, 139.3, 129.5, 128.5, 124.0, 121.1, 119.8, 34.1, 28.3, 15.8.

HRMS (ESI, m/z) calcd for C₁₆H₁₉N₂[M+H]⁺: 239.1543; found: 239.1543.

(E)-N'-(4-(tert-butyl)phenyl)-N-methyl-N-phenylformimidamide (3h)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (21 mg, 79%). ¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 1H), 7.41 - 7.29 (m, 4H), 7.20 - 7.09 (m, 3H), 7.03 - 6.91

(m, 2H), 3.51 (s, 3H), 1.33 (s, 9H). ¹³C NMR (125MHz, CDCl₃) δ 150.9, 148.8, 146.2, 145.2, 129.5, 126.0, 123.9, 120.7, 119.7, 34.3 34.0, 31.5.

HRMS (ESI, m/z) calcd for C₁₈H₂₃N₂[M+H]⁺: 267.1856; found: 267.1853.

(E)-N'-(4-isopropylphenyl)-N-methyl-N-phenylformimidamide (3i)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (22 mg, 86%). ¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H),

7.41 – 7.32 (m, 2H), 7.19 – 7.10 (m, 5H), 7.01 – 6.93 (m, 2H), 3.51 (s, 3H), 2.89 (dt, J

= 13.8, 6.9 Hz, 1H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 150.9, 149.2, 145.2, 143.9, 129.4, 127.0, 123.9, 121.0, 119.7, 34.0, 33.5, 24.2. HRMS (ESI, m/z) calcd for C₁₇H₂₁N₂[M+H]⁺: 253.1699; found: 253.1701.

(E)-N'-(4-(dimethylamino)phenyl)-N-methyl-N-phenylformimidamie

(3j)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 77%). ¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 1H), 7.39 – 7.31 (m, 2H), 7.18 – 7.07 (m, 3H), 7.02 – 6.94

(m, 2H), 6.79 – 6.64 (m, 2H), 3.50 (s, 3H), 2.92 (s, 6H). 13 C NMR (125 MHz, CDCl₃) δ 150.9, 148.8, 146.2, 145.2, 129.5, 126.0, 123.9, 120.7, 119.7, 34.3, 34.0, 31.5. HRMS (ESI, m/z) calcd for C₁₆H₂₀N₃[M+H]⁺: 254.1652; found: 254.1655.

(E)-N'-(4-fluorophenyl)-N-methyl-N-phenylformimidamide (3k)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (17 mg, 75%). ¹H NMR (500 MHz, CDCl₃) δ 8.05 (s, 1H),

7.46 – 7.31 (m, 2H), 7.15 (dd, *J* = 10.9, 4.2 Hz, 3H), 7.06 – 6.87 (m, 4H), 3.50 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 160.6, 158.6, 151.2, 147.6, 145.1, 129.5, 124.2, 122.2, 120.0, 115.7, 115.5, 34.2.

HRMS (ESI, m/z) calcd for $C_{14}H_{13}F_5N_2[M+H]^+$: 229.1136; found: 229.1137.

(E)-N'-(4-chlorophenyl)-N-methyl-N-phenylformimidamide (31)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 77 %). ¹H NMR (500 MHz, CDCl₃) δ 8.05 (s, 1H),

7.44 – 7.34 (m, 2H), 7.29 – 7.22 (m, 2H), 7.20 – 7.10 (m, 3H), 7.04 – 6.91 (m, 2H), 3.50 (s, 3H). $^{\rm 13}{\rm C}$ NMR (125 MHz, CDCl₃) δ 151.3, 150.1, 145.0, 129.5, 129.1, 128.5, 124.4, 122.5, 120.1, 34.3.

HRMS (ESI, m/z) calcd for $C_{14}H_{14}ClN_2[M+H]^+$: 245.0840; found:245.0838.

(*E*)-*N*'-(4-bromophenyl)-*N*-methyl-*N*-phenylformimidamide (3m)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (21 mg,

71%). ¹H NMR (500 MHz, CDCl₃) δ 8.06 (s, 1H), 7.44 – 7.35 (m, 4H), 7.21 – 7.14 (m, 3H), 6.98 – 6.88 (m, 2H), 3.51 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 151.3, 150.6, 144.9, 132.0, 129.5, 124.5, 123.0, 120.1 116.2, 34.3. HRMS (ESI, m/z) calcd for C₁₄H₁₄BrN₂[M+H]⁺: 289.0335; found: 289.0336.

(E)-N-methyl-N'-(4-nitrophenyl)-N-phenylformimidamide (3n)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (15 mg,58%).¹H NMR (500 MHz, CDCl₃) δ 8.22 – 8.13 (m, 2H), 8.10 (s, 1H), 7.46 – 7.37 (m, 2H), 7.22 (dd,

J = 17.7, 7.7 Hz, 3H), 7.13 – 7.05 (m, 2H), 3.55 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 152.1, 144.4, 143.5, 129.7, 125.3, 121.4, 120.7, 34.7. HRMS (ESI, m/z) calcd for C₁₄H₁₄N₃O₂[M+H]⁺: 256.1081; found: 256.1079.

(*E*)-*N'*-([1,1'-biphenyl]-4-yl)-*N*-methyl-*N*-phenylformimidamide (30)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (22 mg, 77%).¹H NMR (500 MHz, CDCl₃) δ 8.17 (s, 1H),

7.61 (dd, J = 7.3, 1.0 Hz, 2H), 7.57 (dt, J = 9.0, 1.8 Hz, 2H), 7.41 (dt, J = 19.8, 7.7 Hz, 4H), 7.32 (td, J = 7.5, 1.1 Hz, 1H), 7.22 – 7.10 (m, 5H), 3.55 (d, J = 0.7 Hz, 3H).¹³C NMR (125 MHz, CDCl₃) δ 151.1, 150.8, 145.1, 141.0, 136.2, 129.5, 128.7, 127.8, 126.7, 124.2, 121.6, 120.0, 34.2.

HRMS (ESI, m/z) calcd for C₂₀H₁₉N₂[M+H]⁺: 289.1543; found: 289.1544.

(E)-N'-(3-bromophenyl)-N-methyl-N-phenylformimidamide (3p)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (21 mg, 71%).¹H NMR (500 MHz, CDCl₃) δ 8.05 (s, 1H), 7.43 – 7.34 (m, 2H), 7.23 – 7.11 (m, 6H), 7.03 – 6.92

(m, 1H), 3.50 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 153.0, 151.5, 144.8, 130.3, 129.5, 126.1, 124.6, 124.1, 1227, 120.3, 34.3.

HRMS (ESI, m/z) calcd for C₁₄H₁₃BrN₂[M+H]⁺: 289.0335; found: 289.0333.

(E)-N'-(3,4-dimethylphenyl)-N-methyl-N-phenylformimidamide (3q)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20

mg, 83%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.37 (dd, J = 8.6, 7.4 Hz, 2H), 7.19 – 7.10 (m, 3H), 7.07 (d, J = 7.9 Hz, 1H), 6.86 (d, J = 2.0 Hz, 1H), 6.80 (dd, J = 7.9, 2.3 Hz, 1H), 3.51 (s, 3H), 2.25 (d, J = 8.3 Hz, 6H).¹³C NMR (125 MHz, CDCl₃) δ 150.8, 149.3, 145.2, 137.2, 131.5, 130.3, 129.4, 123.9, 122.6, 119.8, 118.3, 34.0, 19.9, 19.1.

HRMS (ESI, m/z) calcd for C₁₆H₁₈N₂[M+H]⁺: 239.1543; found: 239.1546.

(E)-N'-mesityl-N-methyl-N-phenylformimidamide (3r)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (22 mg, 85%). ¹H NMR (500 MHz, CDCl₃) δ 7.84 (s, 1H),

7.35 (dd, J = 8.6, 7.5 Hz, 2H), 7.16 – 7.04 (m, 3H), 6.92 – 6.81 (m, 2H), 3.54 (s, 3H), 2.26 (s, 3H), 2.16 (s, 6H).¹³C NMR (125 MHz, CDCl₃) δ 151.2, 146.9, 145.1, 131.7, 129.4, 129.0, 128.6, 123.5, 119.2, 33.6, 20.7, 18.7.

HRMS (ESI, m/z) calcd for C₁₇H₂₁N₂[M+H]⁺: 253.1699; found: 253.1704.

(E)-N-methyl-N-phenyl-N'-(5,6,7,8-tetrahydronaphthalen-1-yl)formi

midamide (3s)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (21 mg, 80%).¹H NMR (500 MHz, CDCl₃) δ 8.03 (s, 1H),

7.42 – 7.32 (m, 2H), 7.20 – 7.09 (m, 3H), 7.05 (t, J = 7.6 Hz, 1H), 6.84 (d, J = 7.6 Hz, 1H), 6.64 (d, J = 7.6 Hz, 1H), 3.52 (s, 3H), 2.79 (dt, J = 12.7, 6.2 Hz, 4H), 1.86 – 1.74 (m, 4H).¹³C NMR (125 MHz, CDCl₃) δ 150.0, 149.5, 145.2, 137.9, 130.7, 129.4, 125.7, 124.4, 123.6, 119.3, 116.0, 33.7, 29.9, 25.4, 23.4, 23.2. HRMS (ESI, m/z) calcd for C₁₈H₂₁N₂[M+H]⁺:265.1699; found:265.1702.

N-methyl-*N*-phenylformimidamide (3t)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (23 mg, 76%).¹H NMR (500 MHz,

CDCl₃) δ 8.20 (s, 1H), 7.73 (t, *J* = 7.1 Hz, 2H), 7.52 (d, *J* = 7.4 Hz, 1H), 7.45 – 7.33 (m, 3H), 7.26 (dd, *J* = 11.1, 3.7 Hz, 2H), 7.22 – 7.13 (m, 3H), 7.09 (dd, *J* = 8.0, 1.9 Hz, 1H), 3.89 (s, 2H), 3.56 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 151, 150.6, 1452, 144.6, 143.0, 141.9, 137.3, 129.5, 126.7, 125.8 124.9, 124.1, 120.3, 120.0, 119.3, 117.8, 37.0, 34.2

HRMS (ESI, m/z) calcd for C₂₁H₁₉N₂[M+H]⁺: 299.1543; found: 299.1547.
(E)-N'-(9H-fluoren-2-yl)-(E)-N'-(9,9-diphenyl-9H-fluoren-2-yl)-N-met

hyl-N-phenylformimidamide (3u)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (31 mg, 68%). ¹H NMR (500 MHz, CDCl3) δ 8.09 (s, 1H), 7.73 (dd, *J* = 7.8, 4.5 Hz, 2H), 7.38 (ddd, *J* = 14.9, 7.4, 6.1 Hz, 4H), 7.30 –

7.11 (m, 16H), 7.06 (dd, J = 8.0, 1.9 Hz, 1H), 3.52 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 152.5, 151.3, 151.1, 150.9, 146.1, 145.1, 140.3, 135.7, 129.8, 128.2 127.4, 126.8, 126.5, 126.1, 124.2, 120.7, 120.4, 120.0, 119.5, 65.5, 34.5. HRMS (ESI, m/z) calcd for C₂₁H₁₉N₂[M+H]⁺: 299.1543; found: 299.1541.

(E)-N-methyl-N'-(E)-N-methyl-N'-(naphthalen-1-yl)-N-phenylformim

idamide (3v)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20 mg, 77%).¹H NMR (500 MHz, CDCl₃) δ 8.24 (s, 1H), 7.78

 $(dd, J = 18.2, 8.3 Hz, 3H), 7.46 - 7.29 (m, 6H), 7.24 - 7.10 (m, 3H), 3.58 (s, 3H).^{13}C$ NMR (125MHz, CDCl₃) δ 151.4, 149.3, 145.1, 134.6, 130.7, 129.5, 128.78, 127.6 127.11, 126.1, 124.2, 123.0, 120.0, 116.4, 34.3.

HRMS (ESI, m/z) calcd for $C_{18}H_{16}N_2[M+H]^+$: 261.1386; found: 261.1383.

(E)-N'-(benzo[d]thiazol-2-yl)-N-methyl-N-phenylformimidamide (3w)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (17 mg, 62%).¹H NMR (500 MHz, CDCl₃) δ 8.80 (s, 1H), 7.73 (dd, *J* = 22.2, 8.0 Hz, 2H), 7.43 (t, *J* = 7.9 Hz, 2H), 7.39

- 7.35 (m, 1H), 7.28 (dd, J = 8.6, 0.8 Hz, 3H), 7.22 (t, J = 7.6 Hz, 1H), 3.59 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 172.9, 155.2, 151.9, 143.9, 133.7, 129.7, 126.2, 125.9, 123.3, 121.5 121.4, 121.2, 35.3.

HRMS (ESI, m/z) calcd for C₁₅H₁₃N₃S[M+H]⁺: 268.0903; found: 268.0902.

(E)-N-methyl-N'-phenyl-N-(p-tolyl)formimidamide (3x)



The reaction was performed following the general procedure. The residue was purified by flash

column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 85%).¹H NMR (500 MHz, CDCl₃) δ 8.05 (s, 1H), 7.31 (t, *J* = 7.8 Hz, 2H), 7.18 (d, *J* = 8.2 Hz, 2H), 7.13 – 6.95 (m, 5H), 3.49 (s, 3H), 2.35 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 151.6, 151.3, 142.7, 134.0, 130.0, 129.1, 123.2, 121.3, 120.2, 34.3, 20.7.

HRMS (ESI, m/z) calcd for C₁₅H₁₇N₂[M+H]⁺: 225.1386; found: 225.1390.

(E)-N-(4-chlorophenyl)-N-methyl-N'-phenylformimidamide (3y)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20 mg, 83%).¹H NMR (500 MHz, CDCl₃) δ 8.04 (s, 1H), 7.35

 $\begin{array}{l} -7.28 \ (m, \, 4H), \ 7.11 - 7.06 \ (m, \, 3H), \ 7.06 - 7.00 \ (m, \, 2H), \ 3.49 \ (s, \, 3H).^{13} C \ NMR \ (125 \ MHz, \, CDCl_3) \ \delta \ 151.2, \ 150.6, \ 143.7, \ 129.4, \ 129.1, \ 123.6, \ 121.2, \ 120.9, \ 34.2. \\ HRMS \ (ESI, \ m/z) \ calcd \ for \ C_{14}H_{13} ClN_2 [M+H]^+: \ 245.0840; \ found: \ 245.0836. \end{array}$

(E)-N-methyl-N'-phenyl-N-(o-tolyl)formimidamide (3z)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (15 mg, 68%)¹H NMR (500 MHz, CDCl₃) δ 7.73 (s, 1H), 7.32 –

7.24 (m, 5H), 7.17 (dd, J = 6.8, 2.1 Hz, 1H), 7.06 (t, J = 8.0 Hz, 3H), 3.39 (s, 3H), 2.35 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 152.8, 151.6, 134.8, 131.5, 129.0, 127.3, 127.0, 123.0, 121.3, 31.5, 18.2.

HRMS (ESI, m/z) calcd for C₁₅H₁₇N₂[M+H]⁺: 225.1386; found: 25.1389.

(E)-N-ethyl-N,N'-diphenylformimidamide (3aa)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (18 mg, 82%).¹H NMR (500 MHz, CDCl₃) δ 7.98 (s, 1H), 7.41 – 7.34 (m,

2H), 7.33 - 7.27 (m, 2H), 7.20 - 7.13 (m, 3H), 7.10 - 7.00 (m, 3H), 4.09 (q, J = 7.1 Hz, 2H), 1.31 (d, J = 7.1 Hz, 3H).¹³C NMR (125MHz, CDCl₃) δ 151.7, 150.5, 144.0, 129.5, 129.1, 124.4, 123.2, 121.3, 121.1, 41.9, 12.8.

HRMS (ESI, m/z) calcd for $C_{15}H_{17}N_2[M+H]^+$: 225.1386; found:225.1390.

(*E*)-*N*-ethyl-*N*'-phenyl-*N*-(*p*-tolyl)formimidamide (3ab)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg,

80%). ¹H NMR (500 MHz, CDCl₃) δ 7.92 (s, 1H), 7.29 (dd, J = 8.1, 7.5 Hz, 2H), 7.17 (d, J = 8.1 Hz, 2H), 7.09 – 7.00 (m, 5H), 4.05 (q, J = 7.1 Hz, 2H), 2.35 (s, 3H), 1.30 – 1.28 (m, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 151.9, 150.7, 141.6, 134.3, 130.0, 129.0, 123.0, 121.6, 121.3, 42.0, 20.8, 12.8.

HRMS (ESI, m/z) calcd for C₁₆H₁₈N₂[M+H]⁺: 239.1543; found: 239.1542.

(*E*)-*N*-butyl-*N*,*N*'-diphenylformimidamide (3ac)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 76%).¹H NMR (500 MHz, CDCl3) δ 7.99 (s, 1H), 7.40 – 7.34 (m, 2H), 7.29 (t, *J* = 7.8 Hz, 2H), 7.16 (dd, *J* = 12.0,

7.6 Hz, 3H), 7.10 – 6.99 (m, 3H), 4.13 – 3.93 (m, 2H), 1.70 (tt, J = 7.7, 6.7 Hz, 2H), 1.44 – 1.35 (m, 2H), 0.94 (t, J = 7.4 Hz, 3H).¹³C NMR (125 MHz, CDCl₃) δ 155.3, 151.8, 150.9, 144.3, 129.4, 129.0, 124.4, 123.1, 121.3, 46.5, 29.5, 20.2, 13.9. HRMS (ESI, m/z) calcd for C₁₆H₁₈N₂[M+H]⁺: 239.1543; found: 239.1544.

Ethyl (E)-4-(((methyl(phenyl)amino)methylene)amino)benzoate (3ad)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (17 mg, 59%).¹H NMR (500 MHz,

 $\begin{array}{l} CDCl_3) \ \delta \ 8.13 \ (s, \ 1H), \ 8.07 - 7.98 \ (m, \ 2H), \ 7.47 - 7.36 \ (m, \ 2H), \ 7.21 \ (dd, \ J = 7.7, \ 4.8 \\ Hz, \ 3H), \ 7.12 - 7.00 \ (m, \ 2H), \ 4.38 \ (q, \ J = 7.1 \ Hz, \ 2H), \ 3.55 \ (s, \ 3H), \ 1.41 \ (t, \ J = 7.1 \\ Hz, \ 3H).^{13}C \ NMR \ (125 \ MHz, \ CDCl_3) \ \delta \ 166.7, \ 155.7, \ 151.6, \ 144.8, \ 130.9, \ 129.6 \ , \ 125.2, \ 124.7, \ 121.1, \ 120.4, \ 60.7, \ 34.4, \ 14.4. \end{array}$

HRMS (ESI, m/z) calcd for $C_{17}H_{18}N_2O_2[M+H]^+$: 283.1441; found: 283.1442.

(E)-1-(4-(2-chlorodibenzo[b,f][1,4]oxazepin-11-yl)piperazin-1-yl)-N-p

henylmethanimine (3ae)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil 32 mg, 76%). ¹H NMR (500 MHz, CDCl₃) δ 7.59 (s, 1H), 7.42 (dd, *J* = 8.7, 2.6 Hz, 1H), 7.36 (d, *J* = 2.6 Hz, 1H), 7.32 – 7.27 (m, 2H), 7.23 – 7.16 (m, 2H), 7.12 (ddd, *J* = 13.6, 5.3, 3.7 Hz, 2H), 7.08 – 6.97 (m, 4H), 3.59 (s, 8H).¹³C NMR (125 MHz, CDCl₃) δ

159.4, 158.8, 152.3, 151.8, 151.4, 139.8, 132.8, 130.5, 129.1, 128.9, 127.2, 125.9, 124.9, 122.9, 121.2, 120.2, 47.4. HRMS (ESI, m/z) calcd for C₂₄H₂₁ClN₄O[M+H]⁺: 417.1477; found: 417.1479.

(*E*)-*N*-phenyl-1-(4-(4-(trifluoromethyl)pyrimidin-2-yl)piperazin-1-yl)

methanimine (3af)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (20.5 mg, 61%).¹H NMR (500 MHz, CDCl₃) δ 8.53 (d, *J* = 4.8 Hz,

1H), 7.61 (s, 1H), 7.28 (t, J = 7.8 Hz, 2H), 7.04 (t, J = 7.3 Hz, 1H), 6.99 (d, J = 7.4 Hz, 2H), 6.82 (d, J = 4.8 Hz, 1H), 3.99 – 3.94 (m, 4H), 3.60 (s, 4H).¹³C NMR (125 MHz, CDCl₃) δ 161.4, 160.3, 156.8, 156.5, 156.2, 155.9, 152.3, 151.4, 129.1, 123.0, 121.1, 119.9, 119.4, 118.9, 105.2, 43.6.

HRMS (ESI, m/z) calcd for $C_{16}H_{16}F_3N_5[M+H]^+$: 336.1431; found: 336.1430.

(E)-N-(2-((S)-2-cyanopyrrolidin-1-yl)-2-oxoethyl)-N-((1r,3R,5R,7S)-3-

hydroxyadamantan-1-yl)-N'-phenylformimidamide compound with

11-methane and methane (1:1:2) (3ag)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a yellow oil (29 mg, 71%).¹H NMR (500 MHz, CDCl₃) δ 7.92 (s, 1H), 7.25 (dd, *J* = 14.9, 7.3 Hz, 2H), 7.01 (dt, *J* = 14.7, 7.4 Hz, 1H), 6.87 (dd, *J* = 16.8, 7.5 Hz, 2H), 5.69 – 4.72 (m,

1H), 4.34 (dd, J = 28.7, 15.6 Hz, 1H), 4.07 (t, J = 14.5 Hz, 1H), 3.82 – 3.47 (m, 2H), 2.41 – 1.81 (m, 15H), 1.71 (s, 4H). ¹³C NMR (125 MHz, CDCl₃) δ 169.5, 168.9, 152.0, 151.6, 149.7, 149.6, 129.1, 129.0, 122.8, 122.6, 121.3, 121.2, 119.4, 118.6, 69.5, 69.4, 59.2, 58.8, 49.9, 49.7, 47.5, 46.8, 46.6, 46.2, 45.0, 44.4, 43.8, 41.1, 41.1, 34.6, 32.3, 30.7, 29.9, 29.7, 25.4, 23.2.

(S,E)-N-benzyl-N'-phenyl-N-(1-phenylethyl)formimidamide (3ah)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (19 mg, 60%).¹H NMR (500 MHz, CDCl₃) δ 7.97 (s, 1H), 7.57 – 7.15 (m, 12H), 7.05 (ddd, *J* = 9.5, 5.2, 1.0 Hz,

3H), 4.90 (s, 1H), 4.51 (d, J = 65.1 Hz, 2H), 1.59 (d, J = 7.1 Hz, 3H).¹³C NMR (125 MHz, CDCl₃) δ 152.3, 151.9, 141.6, 138.4, 129.1, 128.7, 128.6, 128.3, 127.7, 127.1, 126.90 – 125.52 (m), 122.7, 121.4, 58.3, 48.5, 20.6. HRMS (ESI, m/z) calcd for C₂₂H₂₂N₂[M+H]⁺: 315.1856; found: 315.1858.

(E)-N,N-dicyclohexyl-N'-phenylformimidamide (5)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (22 mg, 76%).¹H NMR (500 MHz, CDCl3) δ 7.70 (s, 1H), 7.28

(dd, J = 10.8, 4.8 Hz, 2H), 7.02 – 6.93 (m, 3H), 1.77 (dd, J = 69.6, 11.6 Hz, 11H), 1.58 – 1.31 (m, 9H), 1.19 – 1.10 (m, 2H), 0.89 (dt, J = 12.6, 6.5 Hz, 1H).¹³C NMR (125 MHz, CDCl₃) δ 153.3, 150.6, 128.9, 122.0, 121.3, 29.7, 25.6. HRMS (ESI, m/z) calcd for C₁₉H₁₉N₂[M+H]⁺:285.2325; found: 285.2321.

(Z)-N-isopropyl-N'-phenyl-N-(3-(phenylamino)phenyl)formimidamid

e (7)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 80:1, v/v) to give the product as a yellow oil (18 mg, 54%). ¹H NMR (500 MHz, CDCl₃) δ 7.73 (s, 1H), 7.35 – 7.28 (m, 4H), 7.13 –

6.99 (m, 10H), 5.84 (s, 1H), 1.28 (d, *J* = 6.8 Hz, 7H). ¹³C NMR (125 MHz, CDCl₃) δ 153.0, 152.1, 142.6, 142.4, 133.9, 129.8, 129.5, 129.0, 123.5, 122.6, 121.6, 121.3, 118.4, 117.4, 114.8, 23.1, 21.1.

4-(((E)-((2-((S)-2-cyanopyrrolidin-1-yl)-2-oxoethyl)((1r,3R,5R,7S)-3-h

ydroxyadamantan-1-yl)amino)methylene)amino)benzoate (3ai)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 20:1, v/v) to give the product as a yellow oil (33.5 mg, 70%). ¹H NMR (500 MHz, CDCl₃) δ 7.92 (s, 1H), 7.88 (dd, *J* = 8.4, 4.6 Hz, 2H), 6.84 (dd, *J* = 16.5, 8.5 Hz, 2H), 5.48 – 4.63 (m, 1H), 4.38 – 4.18 (m, 3H), 4.15 – 3.93 (m, 2H), 3.78 – 3.66 (m,

2H), 3.62 - 3.34 (m, 1H), 2.59 (d, J = 138.4 Hz, 1H), 2.37 - 2.07 (m, 7H), 2.03 - 1.98 (m, 1H), 1.98 - 1.80 (m, 6H), 1.67 (s, 4H), 1.56 - 1.47 (m, 2H), 1.33 (t, J = 7.1 Hz, 3H), 1.24 - 1.12 (m, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 171.2, 171.1, 168.9, 168.4,

166.8, 166.7, 156.3, 155.9, 150.1, 150.0, 130.9, 130.8, 124.3, 121.0, 120.9, 119.1, 118.5, 74.6, 69.3, 69.2, 67.1, 60.6, 60.5, 60.4, 59.6, 59.2, 56.7, 49.7, 49.5, 47.4, 46.8, 46.6, 46.2, 45.0, 44.5, 43.6, 41.0, 41.0, 34.6, 34.5, 32.3, 30.6, 29.9, 25.4, 23.2, 21.1, 21.0, 16.2, 14.4, 14.2.

(E)-N,N'-diphenylformimidamide (8)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 50:1, v/v) to give the product as a yellow oil (11 mg, 53%). ¹H NMR (500 MHz, CDCl₃) δ 8.20 (s, 1H), 7.31

(t, J = 7.8 Hz, 4H), 7.24 – 6.64 (m, 6H).¹³C NMR (125 MHz, CDCl₃) δ 129.4, 123.4, 118.9.

HRMS (ESI, m/z) calcd for $C_{13}H_{13}N_2[M+H]^+$:197.1073; found: 197.1076.

(E)-N,N'-di-p-tolylformimidamide (9)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 50:1, v/v) to give the product as a yellow oil (11 mg, 48%).¹H NMR (500 MHz, CDCl₃) δ 8.17 (s, 1H), 7.11

(d, *J* = 8.1 Hz, 4H), 6.94 (d, *J* = 8.0 Hz, 4H), 2.33 (s, 6H).¹³C NMR (125 MHz, CDCl₃) δ 149.3, 132.8, 129.9, 119.5, 118.9, 29.7, 20.7.

HRMS (ESI, m/z) calcd for $C_{15}H_{17}N_2[M+H]^+$:225.1386; found: 225.1389.

(E)-N-(difluoromethyl)-N,N'-diphenylformimidamide (10)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a yellow oil (17 mg, 67%).¹H NMR (500 MHz, CDCl₃) δ 7.84 (t, *J* =

2.8 Hz, 1H), 7.57 (t, J = 52.5 Hz, 1H), 7.45 (dd, J = 6.9, 1.6 Hz, 2H), 7.42 – 7.37 (m, 3H), 7.32 (dd, J = 10.7, 5.0 Hz, 2H), 7.14 (t, J = 7.4 Hz, 1H), 7.06 – 7.00 (m, 2H).¹³C NMR (125 MHz, CDCl₃) δ 149.4, 149.1, 136.4, 129.5, 129.2, 128.2, 127.4, 124.6, 121.1 110.2 (s, J=243.75Hz).¹⁹F NMR (470 MHz, CDCl₃) δ -97.8, -102.8. HRMS (ESI, m/z) calcd for C₁₄H₁₃F₂N₂[M+H]⁺:247.1041; found: 247.1042.

(*E*)-*N*-(difluoromethyl)-*N*,*N*'-di-p-tolylformimidamide (11)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a yellow oil (17 mg, 61%).¹H NMR (500 MHz, CDCl₃) δ 7.82 (t, *J* =

2.8 Hz, 1H), 7.55 (dd, J = 67.5, 54.9 Hz, 1H), 7.17 – 7.11 (m, 2H), 6.94 (d, J = 8.2 Hz, 2H), 2.42 (s, 3H), 2.35 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 149.0, 147.0, 138.2, 134.1, 133.7, 130.1, 129.7, 127.5, 120.9, 110.2(s, J=237.5Hz), 21.1, 20.8.¹⁹F NMR (470 MHz, CDCl₃) δ -97.8, -103.0.

HRMS (ESI, m/z) calcd for C₁₆H₁₇F₂N₂[M+H]⁺:275.1354; found: 275.1356.

(E)-N-(difluoromethyl)-N,N'-bis(4-ethylphenyl)formimidamide (12)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a yellow oil (18 mg, 59%).¹H NMR (500 MHz, CDCl₃) δ 7.81 (t, *J* = 2.8 Hz, 1H), 7.53 (t, *J* = 61.2

Hz, 1H), 7.27 (s, 4H), 7.14 (d, J = 8.4 Hz, 2H), 6.98 – 6.91 (m, 2H), 2.69 (q, J = 7.6 Hz, 2H), 2.63 (q, J = 7.6 Hz, 2H), 1.26 (s, 3H), 1.23 (t, J = 7.6 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 149.0, 147.1, 144.4, 140.6, 133.9, 128.9, 128.5, 127.5, 121.0, ,110.2 (s, J=242.5Hz), 28.5, 28.3, 15.7, 15.4.

¹⁹F NMR (470 MHz, CDCl₃) δ -98.5, -103.0.

HRMS (ESI, m/z) calcd for $C_{18}H_{21}F_2N_2[M+H]^+$:303.1667; found: 303.1668.

(E)-N-(difluoromethyl)-N,N'-bis(4-isopropylphenyl)formimidamide

(13)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a yellow oil (21 mg, 62%).¹H NMR (500 MHz,

CDCl₃) δ 7.81 (t, J = 2.8 Hz, 1H), 7.55 (t, J = 61.2 Hz, 1H), 7.28 (s, 4H), 7.20 – 7.13 (m, 2H), 7.00 – 6.90 (m, 2H), 2.92 (ddt, J = 30.5, 13.8, 6.9 Hz, 2H), 1.28 (s, 3H), 1.26 (s, 3H), 1.24 (s, 3H), 1.23 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 149.0, 147.2, 145.2, 134.0, 127.4, 127.1, 120.9, 110.2(s,J=247.5Hz), 33.8, 33.6, 24.1, 23.9. ¹⁹F NMR (470 MHz, CDCl₃) δ -98.0, -103.0.

HRMS (ESI, m/z) calcd for C₂₀H₂₅F₂N₂[M+H]⁺:331.1980; found: 331.1986.

(E)-N,N'-bis(4-(tert-butyl)phenyl)-N-(difluoromethyl)formimidamide

(14)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a yellow oil (23 mg, 65%). ¹H NMR (500 MHz, CDCl₃) δ 7.85 (t, *J* = 2.8 Hz, 1H), 7.59 (t, *J* = 55.2

Hz, 1H), 7.48 – 7.44 (m, 2H), 7.44 – 7.28 (m, 5H), 7.03 – 6.94 (m, 2H), 1.37 (s, 9H), 1.34 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 151.2, 149.1, 147.5, 146.8, 133.7, 126.9, 126.4, 126.0, 120.7, 110.2 (t, *J* = 242.5 Hz), 34.7, 34.4, 31.38, 31.1. ¹⁹F NMR (470 MHz, CDCl₃) δ -98.1, -102.9.

(E)-N,N'-bis(4-bromophenyl)-N-(difluoromethyl)formimidamide (15)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a yellow oil (23 mg, 57%). ¹H NMR (500 MHz, CDCl₃) δ 7.78 (t, *J* = 2.4 Hz, 1H), 7.66 (t, *J* = 38.0

Hz, 1H), 7.59 - 7.56 (m, 2H), 7.43 - 7.40 (m, 2H), 7.26 - 7.23 (m, 2H), 6.91 - 6.84 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 148.6, 135.0, 132.8, 132.2, 129.1, 122.8, 117.9 (t, *J* = 571.3 Hz), 100.0. ¹⁹F NMR (470 MHz, CDCl₃) δ -99.5, -102.9.

1-methyl-1*H*-benzo[*d*]imidazole (17) (CAS number:1632-83-3)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a yellow oil (12mg, 90%).¹H NMR (500 MHz, CDCl₃) δ 7.86 (s, 1H), 7.81

(dd, J = 7.1, 1.2 Hz, 1H), 7.39 (dd, J = 7.0, 1.1 Hz, 1H), 7.35 – 7.26 (m, 2H), 3.84 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 143.7, 143.5, 134.5, 122.94 (s), 122.1, 120.3, 109.3, 31.0.

6-methylbenzo[d]oxazole (20a) (CAS number:10531-80-3)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (21 mg, 81%).¹H NMR ($500 \text{ MHz}, \text{CDCl}_3$) $\delta 8.03$ (s,

1H), 7.65 (d, J = 8.1 Hz, 1H), 7.39 (s, 1H), 7.18 (dd, J = 8.1, 0.7 Hz, 1H), 2.50 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 152.1, 150.3, 137.8, 136.1, 125.9, 119.9, 111.1, 21.8.

5-methylbenzo[d]oxazole (20b) (CAS number:10531-78-9)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (21 mg,77%).¹H NMR (500 MHz,

CDCl₃) δ 8.07 (s, 1H), 8.07 (s, 1H), 7.40 (d, J = 8.2 Hz, 1H), 7.40 (d, J = 8.2 Hz, 1H), 7.27 (t, J = 7.8 Hz, 1H), 7.27 (t, J = 7.8 Hz, 1H), 7.16 (d, J = 7.5 Hz, 1H), 2.64 (s, 3H), 2.64 (s, 3H).¹³C NMR (125 MHz, CDCl₃) δ 151.8 , 149.7, 139.2, 131.0, 125.3, 125.1, 108.2, 16.5.

5-chlorobenzo[d]oxazole (20c) (CAS number:17200-29-2)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (22 mg,70%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.77 (d, *J* = 2.0 Hz, 1H), 7.51 (d, *J* = 8.7

Hz, 1H), 7.36 (dd, J = 8.7, 2.0 Hz, 1H).¹³C NMR (125 MHz, CDCl₃) δ 153.7, 148.5, 141.1, 130.2, 126.1, 120.6, 111.8.

5-bromobenzo[d]oxazole (20d) (CAS number:132244-31-6)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (25 mg,68%).¹H NMR (500 MHz, CDCl₃) δ 8.09 (s, 1H), 7.93 (d, *J* = 1.7 Hz, 1H), 7.49 (dt, *J* =

20.0, 5.2 Hz, 2H).¹³C NMR (125 MHz, CDCl₃) δ 153.5, 149.0, 141.6, 128.8, 123.7, 117.44, 112.3.

2-(difluoromethoxy)-5-nitroaniline (20e')



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (36 mg,89%).¹H NMR (500 MHz, CDCl₃) δ 7.63 (d, *J* = 2.7

Hz, 1H), 7.59 (dd, J = 8.8, 2.7 Hz, 1H), 7.12 (d, J = 8.8 Hz, 1H), 6.58 (t, J = 72.8 Hz, 1H), 4.20 (s, 2H).¹³C NMR (125 MHz, CDCl₃) δ 145.7, 142.3, 139.0, 117.8 (s, J=261.25Hz), 113.5, 110.6.¹⁹F NMR (470 MHz, CDCl₃) δ -81.0.

2-(difluoromethoxy)-4-nitroaniline (20f')



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 100:1, v/v) to give the product as a colourless oil (34 mg,84%).¹H NMR (500 MHz, CDCl₃) δ 8.03 – 7.90 (m,

2H), 6.76 (d, J = 8.8 Hz, 1H), 6.57 (t, J = 72.9 Hz, 1H), 4.66 (s, 2H).¹³C NMR (125 MHz, CDCl₃) δ 145.1, 138.2, 136.0, 118.1, 116.1,116.0(s,J=262.5Hz) 114.0.¹⁹F NMR (470 MHz, CDCl₃) δ -80.8.

benzo[d]oxazole (20g) (CAS number:51-17-2)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (20 mg, 86%).¹H NMR (500 MHz,

DMSO) δ 12.46 (s, 1H), 8.22 (s, 1H), 7.59 (s, 2H), 7.29 – 7.06 (m, 2H). $^{13}\mathrm{C}$ NMR (126 MHz, DMSO) δ 142.40 (s), 122.16 (s).

6-methylbenzo[d]oxazole (20h) (CAS number:4887-83-6)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (24 mg, 92%).¹H NMR (500 MHz, DMSO) δ 12.47 (s, 1H), 8.18 (s, 1H), 7.39 (d, *J* = 7.6 Hz, 1H), 7.07 (t, *J* = 7.6 Hz, 1H), 6.98

(d, J = 7.2 Hz, 1H), 2.52 (s, 3H).¹³C NMR (125 MHz, DMSO) δ 141.77 (s), 122.16 (s), 17.24 (s).

5-fluoro-1*H*-benzo[*d*]imidazole (20i) (CAS number:1977-72-6)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (21 mg, 78%).¹H NMR (500 MHz, DMSO) δ 12.56 (s, 1H), 8.25 (s, 1H), 7.58 (dd, *J* = 8.4, 4.9 Hz,

1H), 7.42 – 7.36 (m, 1H), 7.07 – 7.01 (m, 1H).¹³C NMR (125 MHz, DMSO) δ 159.9, 158.0, 143.8, 110.5.

7-chloro-1*H*-benzo[*d*]imidazole (20j) (CAS number:16931-35-4)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (26 mg, 86%).¹H NMR (500 MHz, DMSO) δ 12.80 (s, 1H), 8.32 (s, 1H), 7.53 (s, 1H), 7.30 – 7.17

(m, 2H).¹³C NMR (125 MHz, DMSO) δ 143.3, 123.5, 121.6, 111.4.

5-bromo-1*H*-benzo[*d*]imidazole (20k) (CAS number:4887-88-1)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (29 mg, 75%).¹H NMR (500 MHz, DMSO) δ 12.63 (s, 1H), 8.26 (s, 1H), 7.79 (s, 1H), 7.55 (s, 1H),

7.32 (d, J = 8.2 Hz, 1H).¹³C NMR (125 MHz, DMSO) δ 143.8, 100.0.

6-nitro-1*H*-benzo[*d*]imidazole (20l) (CAS number:94-52-0)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a yellow solid (14 mg, 42%). ¹H NMR (500 MHz,

DMSO) δ 8.63 – 8.41 (m, 2H), 8.17 – 7.97 (m, 1H), 7.75 (ddd, J = 8.4, 7.5, 1.8 Hz,

1H). ¹³C NMR (126 MHz, DMSO) δ 147.17 (d, J = 9.3 Hz), 143.09 (s), 118.00 (d, J = 9.4 Hz), 115.29 (s), 113.18 (s).

5-5,6-dimethyl-1*H*-benzo[*d*]imidazole (20m) (CAS number:582-60-5)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (23mg, 79%).¹H NMR (500 MHz,

DMSO) δ 8.05 (s, 1H), 7.34 (s, 2H), 2.29 (s, 6H).¹³C NMR (125 MHz, DMSO) δ 141.4, 130.5, 115.8, 20.4.

5,6-difluoro-1*H*-benzo[*d*]imidazole (20n) (CAS number: 78581-99-4)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (24mg, 77%).¹H NMR (500 MHz, DMSO) δ 12.65 (s, 1H), 8.28 (s, 1H), 7.64 (d, *J* = 22.7 Hz,

2H).¹³C NMR (125 MHz, DMSO) δ 144.5.

5,6-dichloro-1*H*-benzo[*d*]imidazole (200) (CAS number:6478-73-5)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (28mg, 75%).¹H NMR (500 MHz, 1H) 8 35 (s 1H) 7 88 (s 2H) ¹³C NMR (125 MHz DMSO) δ

DMSO) δ 12.75 (s, 1H), 8.35 (s, 1H), 7.88 (s, 2H).13C NMR (125 MHz, DMSO) δ 145.2

1-(difluoromethyl)-1*H*-benzo[*d*]imidazole (21)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (11.4 mg, 61%). ¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 1H),

7.93 – 7.76 (m, 1H), 7.61 (dd, J = 5.4, 3.6 Hz, 1H), 7.37 (ddd, J = 85.8, 55.8, 43.2 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 143.9, 139.1, 130.6, 124.8, 124.2, 121.0, 111.1, 109.0 (t, J = 248.8 Hz) ¹⁹F NMR (470 MHz, CPD, CDCl₃) δ -93.7.

1-(difluoromethyl)-5(6)-methyl-1*H*-benzo[*d*]imidazole (22)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (11 mg, 58%).¹H NMR (500 MHz,

CDCl₃) δ 8.05 (d, J = 11.4 Hz, 1H), 7.71 (d, J = 8.3 Hz, 1H), 7.62 (s, 1H), 7.48 (d, J = 8.3 Hz, 1H), 7.41 (d, J = 4.1 Hz, 1H), 7.29 (s, 1H), 7.20 (d, J = 11.8, 4.5 Hz, 1H), 7.17 (s, 1H), 2.50 (d, J = 8.8 Hz, 3H).¹³C NMR (125 MHz, CDCl₃) δ 144.2, 142.0,

139.1, 138.6, 135.6, 134.1, 130.7, 128.5, 126.2, 125.7, 120.7, 120.4, 110.98 (d, *J* = 5.5 Hz), 110.56 (s), 108.97 (s), 106.99 (s), 21.79 (s), 21.51 (s).¹⁹F NMR (470 MHz, CDCl₃) δ -93.6, -93.7.

1-(difluoromethyl)-4,7-dimethyl-1*H*-benzo[*d*]imidazole (23)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (11 mg, 63%).¹H NMR (500 MHz, CDCl₃) δ 8.13 (d, J = 27.9 Hz, 1H), 7.69 (d, J = 8.1 Hz, 1H), 7.61 (s, 1H), 7.49 (s, 1H), 7.45 – 7.38 (m, 1H), 7.37 (s, 1H), 7.33 – 7.26 (m, 1H), 7.18 $(t, J = 6.0 \text{ Hz}, 1\text{H}), 2.66 \text{ (d}, J = 18.6 \text{ Hz}, 3\text{H}).^{13}\text{C NMR} (125 \text{ MHz}, \text{CDCl}_3) \delta 143.2),$ 139.7, 138.1, 131.1, 130.3, 127.2, 124.7, 124.5, 124.1, 121.6, 118.7, 110.9, 109.0,

108.4, 19.2, 16.6.¹⁹F NMR (471 MHz, CDCl₃) δ -87.3, -93.8.

1-(difluoromethyl)-5(6)-fluoro-1*H*-benzo[*d*]imidazole (24)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (10 mg, 53%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (d, J = 20.5 Hz, 1H), 7.78 (dd, J = 8.9, 4.7 Hz,

1H), 7.56 (dd, J = 8.9, 4.5 Hz, 1H), 7.52 (dd, J = 8.9, 2.4 Hz, 1H), 7.41 (d, J = 9.8 Hz, 1H), 7.35 - 7.27 (m, 1H), 7.15 (m, J = 13.5, 9.1, 2.2 Hz, 1H). ¹³C NMR (125 MHz, $CDCl_3$) 161.5 (d, J = 46.3 Hz), 159.5 (d, J = 37.5 Hz), 140.5, 139.5, 122.8 (d, J = 10.0Hz), 112.9 (d, J = 47.5 Hz), 112.6 (t, J = 113.8 Hz), 98.4 (d, J = 28.8 Hz).¹⁹F NMR (470 MHz, CDCl₃) δ -93.7, -94.0, -115.6, -118.0.

5(6)-chloro-1-(difluoromethyl)-1*H*-benzo[*d*]imidazole (25)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (12 mg, 58%).¹H NMR (500 MHz,

 $CDCl_3$) δ 8.14 (s, 1H), 7.85 (d, J = 1.8 Hz, 1H), 7.56 (d, J = 8.7 Hz, 1H), 7.45 (s, 1H), 7.40 (dd, J = 8.6, 1.9 Hz, 1H), 7.33 (s, 1H), 7.21 (s, 1H).¹³C NMR (125 MHz, CDCl₃) δ 144.8, 140.2, 129.9, 129.1, 125.4, 120.9, 112.0, 108.8 (J = 248.75 Hz,).¹⁹F NMR (470 MHz, CDCl₃) δ -93.8.

5(6)-bromo-1-(difluoromethyl)-1*H*-benzo[*d*]imidazole (26)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (15 mg, 60%).¹H NMR (500 MHz, CDCl₃) δ 8.09 (d, J = 7.1 Hz, 1H), 7.97 (d, J = 0.6 Hz, 1H),

7.77 (s, 1H), 7.68 (d, J = 8.6 Hz, 1H), 7.52 – 7.45 (m, 2H), 7.43 (d, J = 6.2 Hz, 1H),

7.31 (d, J = 6.1 Hz, 1H), 7.19 (d, J = 6.1 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 145.2, 142.9, 140.1, 139.6, 131.4, 129.4, 128.0, 127.7, 123.9, 122.2, 118.2, 117.3, 114.4, 112.4, 108.9 (t, J = 250.0 Hz), 108.8 (t, J = 250.0 Hz). ¹⁹F NMR (470 MHz, CPD, CDCl₃) δ -93.8.

1-(difluoromethyl)-5,6-dimethyl-1*H*-benzo[*d*]imidazole (27)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (13 mg, 69%).¹H NMR (500 MHz, CDCl₃) δ 7.99 (s, 4H), 7.99 (s, 4H), 7.59 (s, 4H), 7.38 (s, 1H), 7.38 (d,

J = 5.7 Hz, 5H), 7.37 (s, 4H), 7.26 (s, 2H), 7.26 (d, J = 2.0 Hz, 3H), 7.14 (s, 1H), 7.14 (s, 1H), 2.38 (d, J = 8.9 Hz, 25H), 2.38 (d, J = 8.9 Hz, 25H).¹³C NMR (125 MHz, CDCl₃) δ 142.5, 138.3, 134.2, 133.2, 129.0, 120.9, 111.24, 109.00 (t,J = 247.5Hz), 20.5, 20.3.¹⁹F NMR (470MHz, CDCl₃) δ -93.6.

1-(difluoromethyl)-5,6-difluoro-1*H*-benzo[*d*]imidazole (28)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (10 mg, 51%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.62 (dd, *J* = 10.0, 7.2 Hz, 1H), 7.44 (dd,

J = 9.3, 6.8 Hz, 1H), 7.41 (s, 1H), 7.29 (s, 1H), 7.17 (s, 1H).¹³C NMR (126 MHz, CDCl₃) δ 150.15 (d, J = 15.5 Hz), 149.80 (d, J = 14.8 Hz), 148.19 (d, J = 15.5 Hz), 147.86 (d, J = 15.0 Hz), 140.44 (s), 139.35 (d, J = 9.0 Hz), 125.7, 108.8, 108.7 (t,J = 228.8Hz), 108.5, 106.8, 100.0, 99.8.¹⁹F NMR (470MHz, CDCl₃) δ -94.0, -137.9, -138.0, -140.3, -140.3.

5,6-dichloro-1-(difluoromethyl)-1*H*-benzo[*d*]imidazole (29)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (13 mg, 55%).¹H NMR (500 MHz, CDCl₃) δ 8.11 (s, 1H), 7.95 (s, 1H), 7.75 (s, 1H), 7.40 (s, 1H),

7.28 (s, 1H), 7.16 (s, 1H).¹³C NMR (125 MHz, CDCl₃) δ 143.2, 140.7, 129.3, 128.8, 122.3, 112.9, 110.7(t,*J* = 228.8Hz).¹⁹F NMR (470 MHz, CDCl₃) δ -93.9.

(E)-N,N'-bis(4-ethoxyphenyl)formimidamide (32)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a white solid (8.5 mg, 15%). ¹H NMR

 $(500 \text{ MHz}, \text{CDCl}_3) \delta 8.02 \text{ (s, 1H)}, 6.97 \text{ (d, } J = 8.6 \text{ Hz}, 4\text{H}), 6.89 - 6.81 \text{ (m, 4H)}, 4.01 \text{ (m$

(q, J = 7.0 Hz, 4H), 1.40 (t, J = 7.0 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 155.4, 120.2, 115.3, 63.8, 14.9.

1-ethoxy-4-isocyanobenzene (33)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a colourless oil (8 mg, 27%). ¹H NMR (500 MHz, CDCl₃) δ 7.29 (d, *J* = 8.9 Hz, 2H), 6.87 – 6.82 (m, 2H), 4.03 (q,

J = 7.0 Hz, 2H), 1.42 (t, J = 7.0 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 162.3, 159.3, 127.8, 115.0, 63.9, 14.7.

N-(4-ethoxyphenyl)formamide (34)



The reaction was performed following the general procedure. The residue was purified by flash column chromatography (silica gel, Petroleum ether: AcOEt = 10:1, v/v) to give the product as a colourless oil (8 mg, 27%). ¹H NMR (500 MHz, CDCl₃) δ 8.50 (d, *J* = 11.5 Hz, 0.49 H), 8.31 (d, *J* = 1.8 Hz, 0.49

H), 8.09 (s, 0.46 H), 7.47 – 7.35 (m, 1.42 H), 7.06 – 6.96 (m, 1H), 6.90 – 6.81 (m, 2H), 4.00 (qd, J = 7.0, 4.3 Hz, 2H), 1.40 (q, J = 6.9 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 163.1, 158.9, 157.0, 156.1, 129.8, 129.4, 121.8, 121.7, 115.5, 114.8, 63.8, 63.7, 14.8, 148.

References

- 1. Tlili, A., Blondiaux, E., Frogneux, X., and Cantat, T. (2015). Reductive functionalization of CO₂ with amines: an entry to formamide, formamidine and methylamine derivatives. Green Chem. *17*, 157–168.
- 2. Huang, K., Sun, C-L., and Shi, Z-J. (2011). Transition-metal-catalyzed C–C bond formation through the fixation of carbon dioxide. Chem. Soc. Rev. *40*, 2435–2452.