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Synthesis, physiochemical property and antimicrobial activity of novel quaternary ammonium salts

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ABSTRACT

Twenty-four novel 5-phenyl-1,3,4-oxadiazole-2-thiol (POT) analogues, benzo[*d*]oxazole-2-thiol, benzo[*d*]thiazole-2-thiol and 5-methyl-1,3,4-thiadiazole-2-thiol-substituted *N*,*N*-bis(2-hydroxyethyl) quaternary ammonium salts (QAS) (**5a-d, 6a-d, 7a-d, 10a-d, 13a-d, 16a-d**) were prepared and characterised by FTIR, NMR and elemental analysis. Part of target compounds (**5d, 6d, 7d, 10d, 13d, 16d**) displayed potent antimicrobial effect against ten common pathogens (*S. aureus, α-H-tococcus, β-H-tococcus, E. coli, P. aeruginosa, Proteus vulgaris, Canidia Albicans, Cytospora mandshurica, Physalospora piricola, Aspergillus niger*) and had relatively low cytotoxity against two human cell lines (HaCat and LO2). TEM and SEM images of *E. coli* and *S. aureus* morphologies treated with **7d** showed that the antibacterial mechanism might be the QAS fixing on cell wall surfaces and puncturing to result in the release of bacterial cytoplasm. This study provides new information of QAS, which could be used to design novel antimicrobial agents applied in clinic or agriculture.



ARTICLE HISTORY

Received 3 August 2017 Revised 27 September 2017 Accepted 5 October 2017

KEYWORDS

5-Phenyl-1,3,4-oxadiazole-2thiol; quaternary ammonium salts; antibacterial activity; antifungal activity; cytotoxicity

Introduction

As one of the leading causes of death worldwide, outbreaks of infectious diseases triggered by bacteria, viruses and fungi lead to over one-fourth of global deaths annually^{1,2}. Thus, there is an urgent demand for exploring more efficient, broad-spectrum and long-lasting antimicrobial agents. Because of excellent antibacterial activities, guaternary ammonium salts (QAS) are widely used by the pharmaceutical industry for their recognised activity against fungi, bacteria, viruses and parasites³. For example, QASs are used as antiseptics and disinfectants of medical apparatus and instruments, and a variety of clinical application, such as preoperative disinfection of unbroken skin and trauma⁴⁻⁹. With more and more resistant organisms continue to emerge on clinic, the ideal antimicrobial agents should have to possess strong antibacterial activity, be safe in relation to humans and should not persist in the environment for a long time¹⁰. Some of the quaternary ammonium compounds fulfil these conditions have been invented and received increasing attentions as antibacterial agents^{11–13}. Especially, long-chain alkyl-QAS compounds show unsurpassed antibacterial effects. It is well-known that the antibacterial activity and drug toxicity are influenced by the alkyl chain length of these agents^{14,15}. As the growth of the alkyl chains, the antibacterial activities gradually increase, until reaching a reasonable limit. Unfortunately, drug toxicity of QAS bearing longer-chain alkyl groups also strengthen as well as benzyl-QAS. Tremendous research efforts have been devoted to development of alternative antibacterial therapeutics that are not easily forming resistance, such as quaternary ammonium salts (QAS), which destruct cell membrane of bacteria and induce the leakage of intracellular components from bacterial cells¹⁶. Thus, it becomes urgent to design and prepare one kind of novel antibacterial with higher antibacterial activities and lower drug toxicity. Many discrete QAS¹⁷ and polymerised quaternised materials (such as quaternised chitosan^{18,19}) were

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reported with good antibacterial activities because of guick disinfection and sterilisation.

In the previous work of our group, N-methyl-N-R-N,N-bis(2hydroxyethyl) ammonium bromides (BNQAS, C12QAS, C14QAS, C16QAS, C18QAS) were prepared from N-methyldiethanolamine and halohydrocarbon with long-chain alkyl bromides²⁰. These QASs showed good antibacterial abilities against E. coli, S. aureus, B. subtilis and antifungal activities against Cytospora mandshurica, Botryosphaeria ribis, Physalospora piricola and Glomerella cinqulata. The strategy provided a facile way to design and develop new types of antimicrobial agents for application in preventing the fruit rot, especially apple. Meanwhile, these QASs also were used to prepare nontoxic O-quaternised antimicrobial chitosan materials²¹. Simultaneously, C12QAS (1, Figure 1) with the best antibacterial abilities was also used to prepare novel block antibacterial polyurethane (BAPU) and terminated antibacterial polyurethane (TAPU) through prepolymerisation reaction¹⁹. However, QAS monomers display higher cytotoxicity towards normal cell lines²⁰, which may limit their clinical utility as antibacterial material. So it is a big challenge that how to improve their antimicrobial activities on the basis of lower cytotoxicity.

Motivated by our interest in the antimicrobial agents, chemical modifications of QAS monomers were carried out immediately. Due to the quaternary ammonium group was a key active group, N,N-bis(2-hydroxyethyl) ammonium bromide of C12QAS (1, Figure 1) was retained. Studied through literature researches, we



C12QAS (1)

Figure 1. Representative target compounds in previous work.

found many 1,3,4-oxadiazole-2-thiol derivatives were reported as potent inhibitors of Gram-negative and Gram-positive bacteria²²⁻²⁵. Very recently, we tried to change long-chain alkyl group of C12QAS (1, Figure 1) into 5-phenyl-1,3,4-oxadiazole-2thiol $(POT)^{26}$ substituted alkyl groups and synthesised dozens of POT-QASs²⁷. Among these, bearing *N*,*N*-bis(2-hydroxyethyl) and POT groups, POT-C12QAS (2, Figure 1) displayed unsurpassed antimicrobial activities and low cytotoxicity. The structure-activity relationships indicated that POT and flexible dihydroxyethyl group in QAS were necessary for antibacterial activities.

Furthermore, in order to improve antimicrobial activities and decrease cytotoxicity, we tried to modify or replace the POT fragment in POT-QAS with 5-(4-fluorophenyl)-1,3,4-oxadiazole-2-thiol (POTF), 5-(4-trifluoromethyl phenyl)-1,3,4-oxadiazole-2-thiol (POTT), 5-(4-tert-butylphenyl)-1,3,4-oxadiazole-2-thiol (POTB), benzo[d]oxazole-2-thiol, benzo[d]thiazole-2-thiol and 5-methyl-1,3,4-thiadiazole-2-thiol fragments. Based on the consideration, six series of novel QASs (5a-d, 6a-d, 7a-d, 10a-d, 13a-d, 16a-d, Scheme 1) were prepared. These novel QASs were characterised with proton nuclear magnetic resonance (¹H NMR and ¹³C NMR) pectroscopy, stransmission Fourier transform infrared (FTIR) spectra, elemental analyses and melting point. Their antibacterial, antifungal properties and cytotoxicity were tested and the antibacterial mechanism of quaternary ammonium salts to E. coli and S. aureus was explored by SEM and TEM.

Results and discussion

Structural analysis

In this study, 24 5-phenyl-1,3,4-oxadiazole-2-thiol (POT) analogues, benzo[d]oxazole-2-thiol, benzo[d]thiazole-2-thiol and 5-methyl-1,3,4-thiadiazole-2-thiol substituted N,N-bis(2-hydroxyethyl) quaternary ammonium salts (5a-d, 6a-d, 7a-d, 10a-d, 13a-d, 16a-d) were prepared from Aromatic thiols, α, ω -dibromoalkanes and N-methyldiethanolamine in good yields (Scheme 1). Their structures were



Scheme 1. The synthesis and structures of six series of noval QASs.

characterised by FTIR, NMR and elemental analysis. From the FTIR spectra, strong characteristic absorption band of hydroxyl and stretching vibration band of C-O bonds can be found. For example, the two bands of **5a** lie in 3307 cm⁻¹ and 1054 cm⁻¹, respectively, which demonstrates the existence of –OH of QAS. In addition, the distinct band in 788 cm⁻¹ belongs to the stretching vibration band of linear paraffin in FTIR spectra of **5a**. In other QAS, the stretching vibration bands of linear paraffin are 731 cm⁻¹ (**6a**), 739 cm⁻¹ (**7a**), 756 cm⁻¹ (**10a**), 741 cm⁻¹ (**13a**) and 729 cm⁻¹ (**16a**), respectively. In ¹H NMR spectra, the single peak of hydroxy of **5a** can be found at δ 5.31 ppm, the multiple peaks of benzene ring disappeared in ¹H NMR spectra at δ 8.16–7.22 ppm, and the single peaks of methyl of

5a at δ 3.06 ppm. The similar structure characteristic also can prove their structure for other compounds and they are shown in experimental part. Additionally, ¹³C NMR, element analysis further confirms the correctness of their structures.

Antimicrobial properties

In this paper, antimicrobial activities of 24 QASs were detected by quantitative tests. The data of minimal inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) for 24 QASs against Gram-positive bacteria (*S. aureus*, α -*H*-tococcus, β -*H*-tococcus), Gram-negative bacteria (*E. coli*, *P. aeruginosa*, *Proteus vulgaris*)

Compd	S. aureus	α-H-tococcus	β -H-tococcus	E. coli	P. aeruginosa	Proteus vulgaris	Canidia albicans	Cytospora mandshurica	Physalospora piricola	Aspergillus niger
BZK	3.125	3.125	1.56	25	25	25	12.5	6.25	3.125	3.125
CA	3.125	6.25	1.56	6.25	25	12.5	12.5	6.25	1.56	3.125
2	6.25	6.25	6.25	25	12.5	25	6.25	3.125	3.125	3.125
5a	100	50	25	100	100	100	50	50	12.5	50
5b	25	12.5	12.5	25	50	50	25	25	6.25	50
5c	25	12.5	6.25	25	50	50	25	12.5	3.125	25
5d	12.5	12.5	6.25	25	50	25	12.5	12.5	3.125	25
ба	50	50	25	50	100	100	50	12.5	6.25	25
6b	25	12.5	12.5	25	50	50	25	12.5	3.125	12.5
6c	25	12.5	6.25	25	50	25	25	12.5	3.125	12.5
6d	6.25	6.25	3.125	25	50	25	12.5	6.25	1.56	6.25
7a	50	25	12.5	50	50	50	25	12.5	12.5	25
7b	25	12.5	6.25	25	50	25	12.5	6.25	3.125	12.5
7c	12.5	12.5	3.125	25	50	25	6.25	6.25	1.56	6.25
7d	6.25	6.25	1.56	12.5	25	12.5	6.25	3.125	1.56	6.25
10a	100	50	25	100	100	100	50	50	25	50
10b	50	25	12.5	50	50	50	50	25	12.5	25
10c	25	25	6.25	50	50	50	25	6.25	6.25	12.5
10d	12.5	12.5	6.25	25	25	25	12.5	6.25	3.125	6.25
13a	50	50	12.5	50	100	100	25	25	25	25
13b	25	25	12.5	50	50	50	25	6.25	6.25	12.5
13c	12.5	25	6.25	25	25	25	12.5	3.125	3.125	6.25
13d	6.25	12.5	1.56	25	25	25	6.25	1.56	3.125	3.125
16a	100	50	50	100	200	100	50	50	25	50
16b	50	50	25	50	100	50	25	25	12.5	25
16c	25	25	6.25	50	50	50	25	12.5	12.5	12.5
16d	12.5	12.5	6.25	25	50	25	12.5	12.5	6.25	12.5

Table 1. Minimal inhibitory concentration (MIC) of QASs (μ g/mL).

Table 2. Minimal bactericidal concentration (MBC) of QASs (µg/mL).

Compd	S. aureus	α-H-tococcus	β -H-tococcus	E. coli	P. aeruginosa	Proteus vulgaris	Canidia albicans	Cytospora mandshurica	Physalospora piricola	Aspergillus niger
BZK	3.125	3.125	1.56	25	25	25	25	6.25	3.125	6.25
CA	6.25	6.25	3.125	12.5	25	12.5	12.5	6.25	3.125	6.25
2	12.5	6.25	6.25	50	25	50	12.5	6.25	3.125	3.125
5a	100	50	25	100	100	100	50	50	12.5	50
5b	50	25	12.5	50	50	50	25	25	6.25	50
5c	25	12.5	12.5	50	50	50	25	12.5	6.25	25
5d	12.5	12.5	6.25	25	50	25	12.5	12.5	3.125	25
6a	50	50	25	50	100	100	50	25	12.5	25
6b	25	25	12.5	50	50	50	25	12.5	6.25	25
6c	25	12.5	6.25	25	50	25	25	12.5	3.125	12.5
6d	12.5	6.25	3.125	25	50	25	12.5	6.25	1.56	6.25
7a	50	25	12.5	50	50	50	25	12.5	12.5	25
7b	25	12.5	6.25	25	50	25	12.5	12.5	3.125	12.5
7c	12.5	12.5	6.25	25	50	25	12.5	6.25	3.125	12.5
7d	6.25	6.25	1.56	12.5	25	12.5	6.25	3.125	1.56	6.25
10a	100	50	25	100	200	100	100	50	25	50
10b	50	50	25	100	100	50	50	25	25	25
10c	25	25	12.5	50	50	50	25	12.5	6.25	12.5
10d	12.5	12.5	6.25	25	25	25	12.5	6.25	6.25	12.5
13a	50	50	25	100	100	100	50	25	25	25
13b	50	25	12.5	50	100	50	25	12.5	12.5	12.5
13c	12.5	25	6.25	25	50	25	25	6.25	3.125	6.25
13d	6.25	12.5	3.125	25	25	25	6.25	6.25	3.125	6.25
16a	200	100	50	200	200	200	50	50	25	50
16b	100	50	25	100	100	50	50	25	25	25
16c	25	25	12.5	50	50	50	25	25	12.5	25
16d	25	12.5	6.25	25	50	25	12.5	12.5	6.25	12.5

For the six series of QASs (**5a-d**, **6a-d**, **7a-d**, **10a-d**, **13a-d**, **16a-d**), all of their structures contained nitrogen-containing heterocyclics and *N*,*N*-bis(2-hydroxyethyl) quaternary ammonium salt pharmacophores linked by medium- or long-chain alkyl groups (hexyl, octyl, decyl and dodecyl). Compared with our previous research, similar result was found that their MIC and MBC values decreased in turn following the increase of chain length. In other words, compounds **5d**, **6d**, **7d**, **10d**, **13d** and **16d** could regard as the most potent compounds in each series.

Although compounds 5a-d, 6a-d and 7a-d belonged to POT-QAS derivatives, they exhibited diverse antibacterial activities. For Gram-positive bacteria (S. aureus, α -H-tococcus, β -H-tococcus) and Gram-negative bacteria (E. coli, P. aeruginosa, Proteus vulgaris), compound 7d bearing with 5-(4-tert-butylphenyl)-1,3,4-oxadiazole-2-thiol (POTB) substituent group exhibited the best antibacterial activities among all six series (corresponding MIC values, Table 2). It is just similar to unmodified POT-C12QAS: ca. 6.25 µg/mL against S. aureus, α -H-tococcus ca. 1.56 μ g/mL against β -H-tococcus; ca. 12.5 µg/mL against E. coli, Proteus vulgaris, ca. 25 µg/mL against P. aeruginosa, Table 1; MBC values were equal to corresponding MIC values, Table 2. It should be pointed out that the antibacterial activities of 7d were comparable to positive controls (BZK and CA), and more potent than unmodified POT-C12QAS (2). Furthermore, Introduced halogen substitution, especially fluoro substitution into the structure was hopeful to enhance the antibacterial effect^{28,29}. But to our disappointment, fluoro (5d) or trifluoromethyl (6d) substituted derivatives did not exhibit desired antibacterial activities, and their MIC and MBC values only reached to medium level (e.g. MIC values for 5d: ca. 12.5 µg/mL against S. aureus, α -H-tococcus, ca. 6.25 μ g/mL against β -H-tococcus; ca. 25 µg/mL against E. coli, Proteus vulgaris, ca. 50 µg/mL against P. aeruginosa, Table 1). These foregoing results suggest that electron-donating substituents (tertiary butyl) on POT might enhance antibacterial activities of POT-QAS; on the contrary, electron-withdrawing substituents (fluoro or trifluoromethyl) might give little contribution to the activities.

In order to investigate the influence of other nitrogen heterocyclics substituted in QAS, POT fragment was changed into benzo[*d*]oxazole-2-thiol (**10a-d**), benzo[*d*]thiazole-2-thiol (**13a-d**) or 5-methyl-1,3,4-thiadiazole-2-thiol (**16a-d**), respectively. Among these three non-POT series, compound **13d** exhibited relatively potent antibacterial activities (MIC: ca. $6.25 \,\mu$ g/mL against *S. aureus*, ca. 12.5 μ g/mL against α -*H*-tococcus, ca. 1.56 μ g/mL

Table 3. Cytotoxicity of QASs for HaCat and LO2 cells.

against β -H-tococcus; ca. 25 µg/mL against *E. coli, Proteus vulgaris, P. aeruginosa*, Table 1; MBC: ca. 3.125 µg/mL against β -H-tococcus, other MBC values were equal to corresponding MIC values, Table 2), which was close to positive controls (**BZK** and **CA**). However, besides compound **10d** possessed fairly activities against Gramnegative bacteria (MIC: ca. 25 µg/mL against *E. coli, Proteus vulgaris, P. aeruginosa*, Table 1; MBC values were equal to corresponding MIC values, Table 2), other non-POT derivatives did not exhibit noticeable activities. Based on the above results, POT-QAS derivatives were partially superior to non-POT-QAS derivatives, and this might be because POT-QAS derivatives with flexible alkyl chain length and POT fragment could more easily pass through cell membrane into the cell to damage bacteria^{20,27,30–32}.

In the antifungal activity assay of the target compounds, four strains of common fungi in clinical practice and agriculture (Canidia albicans, Cytospora mandshurica, Physalospora piricola, Aspergillus niger) were tanken as the target fungi. As shown in Tables 1 and 2, to our delight, most compounds bearing with the long-chain alkyl group (dodecyl, e.g. 5d, 6d, 7d, 10d, 13d and 16d) displayed fairly good antifungal activities. The most noticeable compound was 13d (MIC: ca. 6.25 µg/mL against Canidia albicans, ca. 1.56 µg/mL against Cytospora mandshurica, ca. 3.125 µg/ mL against Physalospora piricola, Aspergillus niger, Table 1; MBC: ca. 6.25 µg/mL against Cytospora mandshurica, other MBC values were equal to corresponding MIC values, Table 2), with a 2-mercaptobenzothiazole substituent in C12QAS, exhibited more potent activities than positive controls (BZK and CA). In addition, POT-C12QAS (2) derivative 7d (MIC: ca. 6.25 µg/mL against Canidia albicans, ca. 3.125 µg/mL against Cytospora mandshurica, ca. 1.56 µg/ mL against Physalospora piricola, ca. 6.25 µg/mL against Aspergillus niger, Table 1; MBC values were equal to corresponding MIC values, Table 2) also displayed almost the same antifungal activities as positive controls (BZK and CA). These encouraging results indicated that either POTT (7d) or 2-mercaptobenzothiazole (13d) substituent in C12QAS could enhance the antifungal activities. More than that, both of which might develop into promising antifungal agents in clinic or agriculture.

Cytotoxicity

These six series of antimicrobial compounds may be applied to daily or clinical disinfectants for unbroken skin and trauma, or agricultural fungicides alternatively. Thus, their cytotoxicity for skin or body tissues are crucial. In this study, the cytotoxicity of target compounds was evaluated for human immortalised epidermal (HaCat) and human normal liver (LO2) cell line using MTT assay. The results were showed in Table 3 summarising the corresponding IC₅₀ values of QASs. The IC₅₀ values of **5a**, **6a**, **10a**, **16a**, **16b** were about more than 50 μ g/mL for LO2 and HaCat. In addition, compounds **5b**, **6b**, **10b**, **13a**, **16c**, **16d** also showed lower IC₅₀

Compd	IC ₅₀ (μ	ıg/mL)	Compd	IC ₅₀ (μg/mL)			IC ₅₀ (μg/mL)	
	LO2	HaCat		LO2	HaCat	Compd	LO2	HaCat
BZK	2.12 ± 0.43	3.19 ± 0.26	CA	2.06 ± 0.06	2.54 ± 0.12	2	14.23 ± 2.15	18.27 ± 2.21
5a	83.51 ± 5.67	76.52 ± 4.58	7a	44.26 ± 3.31	23.84 ± 2.13	13a	64.28 ± 4.12	45.06 ± 3.23
5b	46.75 ± 3.32	50.83 ± 3.65	7b	16.78 ± 1.35	9.05 ± 0.92	13b	29.58 ± 2.86	17.84 ± 1.56
5c	23.86 ± 1.68	19.39 ± 1.23	7c	12.90 ± 1.37	6.94 ± 0.53	13c	16.84 ± 1.61	10.51 ± 1.02
5d	6.95 ± 0.62	6.14 ± 0.53	7d	5.08 ± 0.51	4.16 ± 0.43	13d	8.96 ± 0.46	6.02 ± 0.35
ба	74.35 ± 4.12	62.57 ± 3.82	10a	88.27 ± 5.89	54.62 ± 3.16	16a	>100	>100
6b	38.72 ± 2.92	32.57 ± 2.69	10b	42.92 ± 2.37	30.38 ± 2.89	16b	86.52 ± 5.33	65.38 ± 3.12
6с	19.56 ± 1.62	12.06 ± 1.57	10c	25.32 ± 1.78	15.86 ± 1.23	16c	43.18 ± 2.67	36.87 ± 2.81
6d	5.86 ± 0.61	5.25 ± 0.37	10d	9.96 ± 0.76	7.14 ± 0.51	16d	30.12 ± 2.32	27.03 ± 2.53

values (more than 25 µg/mL). These data demonstrated that compounds 5a, 6a, 16a, 16b were non-toxic for LO2 and HaCat. Especially, lower-toxic compounds 5 b, 6 b, 16c, 16d with better activities might become superior antimicrobial agents. However, the IC₅₀ values of each series for LO2 and HaCat decreased gradually and finally reached to 4.16 µg/mL. For example, the IC_{50} values of **7a-d** were orderly $44.26 \pm 3.31 \,\mu\text{g/mL}$, $16.78 \pm 1.35 \,\mu$ g/mL, $12.90 \pm 1.37 \,\mu$ g/mL, $5.08 \pm 0.51 \,\mu$ g/mL for LO2 as well as $23.84 \pm 2.13 \,\mu\text{g/mL}$, $9.05 \pm 0.92 \,\mu\text{g/mL}$, $6.94 \pm 0.53 \,\mu\text{g/mL}$ mL,4.16 \pm 0.43 μ g/mL for HaCat, respectively. These results indicated that their cytotoxicity and antimicrobial activities synchronously increased following the increased of chain length from hexyl group to dodecyl group²⁷, and all of them exhibited weaker cytotoxicity compared with positive controls (BZK and CA). In addition, it should be pointed out that compounds 7d and POT-C12QAS (2) with POT and dodecyl group displayed weaker cytotoxicity than N-methyl-N-dodecyl-N,N-bis(2-hydroxyethyl) ammonium bromides (C12QAS), which IC₅₀ value was ca. 4.00 μ g/mL²⁰. This result proved that introduction of POT fragment in C12QAS could effectively reduce cytotoxicity and improve antimicrobial activities.

Haemolysis

The surfactant may be absorbed and penetrate to the cell membrane, where it makes osmotic phenomena by altering the permeability of membrane, which in turn causes the cellular lysis³³. Haemolysis assay gives quantitative estimation about haemoglobin release when red blood cells are treated with QASs. Haemolytic assay of QASs at different concentration was carried out in Table 4. It was observed that the haemolysis rate of these samples increased successively following the increase of concentration, which showed concentration-dependent haemolysis. More importantly, all QASs did not cause haemolysis under the concentration of 50 µg/mL except **10d** and **13d**. For **5a-b**, **6a-b**, **10a-c**, **13a-b**, **and 16a-d**, their safe concentration can more than 100 µg/mL.

Table 4. Haemolysis rate (%) of QASs with different concentration (µg/mL).

Compared with BZK and CA, all QASs were safe to be used when they circulated into the blood in low concentration.

SEM images and TEM images of E. coli and S. aureus morphologies

Treatment of bacteria with 7d caused physical and morphological alterations as well as cell wall surface deterioration, and very few intact cells were detected in samples of E. coli and S. aureus. Scanning electron microscope (SEM) was carried out to follow the morphological alterations of E. coli and S. aureus and the results are shown in Figures 2 and 3. The E. coli were uniformly rod-shaped in their morphology, they had typical dimensions and a smooth cell surface and were present in large numbers (Figure 2(a)). After contacting the addition of compound 7d (Figures 2(c) and 3(c)) for 4 h, bacteria reveal detrimental effect on the morphology of the cell membrane, showing a large surface collapse and wrinkled abnormalities on the morphology of E. coli and S. aureus. More obviously, bacterial cytoplasm release from cell membranes and bacteria have been on the verge of death. This may be caused by the antibacterial agents fixing on membrane surfaces³⁴. As shown in Figure 2(e) and 3(e), all bacteria died after contacting compound 7d for 12h. The alteration of E. coli and S. aureus morphology may be attributed to the puncturing of cell membranes to result in the release of bacterial cytoplasm³⁵. Very few intact cells were detected in samples of E. coli and S. aureus treated with 7d.

TEM analysis showed normal *E. coli* and *S. aureus* cell wall and cytoplasmic membrane were intact, and their cytoplasm was homogenously electron dense (Figure 2(b) and 3(b)). A lot of incomplete cells were detected in samples of *E. coli* and *S. aureus* treated with **7d** (Figures 2(d,f) and 3(d,f)). Leakage of intracellular substance was observed in those cells that could be seen. Cells examined by TEM showed regional to complete loss of electron density of both their

Compd	Haemolysis rate (%)									
	6.25	12.50	25.00	50.00	100.00	200.00	400.00			
BZK	1.51 ± 0.31	1.39 ± 0.38	12.69 ± 1.87	63.85 ± 4.56	92.36 ± 3.78	96.28 ± 2.97	98.53 ± 1.23			
CA	0.86 ± 0.28	2.51 ± 0.87	5.29 ± 1.22	37.05 ± 3.78	62.18 ± 3.53	86.02 ± 3.82	95.67 ± 3.54			
2	0.18 ± 0.05	0.42 ± 0.13	0.93 ± 0.41	4.11 ± 0.89	17.07 ± 2.34	82.75 ± 3.18	90.61 ± 4.23			
5a	0.21 ± 0.03	0.53 ± 0.15	1.37 ± 0.87	1.54 ± 0.96	3.25 ± 1.23	6.21 ± 3.21	35.64 ± 4.87			
5b	0.29 ± 0.08	0.68 ± 0.23	1.87 ± 1.35	2.19 ± 1.29	3.92 ± 1.02	21.08 ± 4.55	65.49 ± 4.13			
5c	0.43 ± 0.16	0.75 ± 0.19	2.31 ± 0.67	4.89 ± 0.53	33.13 ± 4.87	52.28 ± 6.34	67.15 ± 5.89			
5d	0.49 ± 0.12	0.87 ± 0.27	2.43 ± 0.82	4.81 ± 0.74	32.41 ± 6.78	66.71 ± 5.21	66.92 ± 4.12			
ба	0.15 ± 0.08	0.35 ± 0.17	0.51 ± 0.19	0.99 ± 0.12	3.38 ± 1.23	7.09 ± 2.89	40.96 ± 5.32			
6b	0.18 ± 0.06	0.51 ± 0.08	0.43 ± 0.10	0.70 ± 0.18	4.61 ± 0.31	26.16 ± 3.68	80.37 ± 4.98			
6с	0.57 ± 0.15	0.42 ± 0.13	0.67 ± 0.13	3.56 ± 1.32	35.76 ± 5.10	82.52 ± 6.12	89.39 ± 4.35			
6d	0.64 ± 0.19	1.05 ± 0.28	2.02 ± 3.45	4.85 ± 4.65	36.43 ± 6.35	85.62 ± 4.86	92.46 ± 3.87			
7a	1.03 ± 0.56	2.50 ± 1.08	1.01 ± 0.58	1.98 ± 0.68	5.66 ± 1.23	20.76 ± 3.28	34.73 ± 4.78			
7b	1.41 ± 0.43	1.48 ± 0.68	2.20 ± 0.68	2.59 ± 0.89	13.84 ± 2.73	30.63 ± 3.76	85.75 ± 5.67			
7c	0.18 ± 0.06	0.75 ± 0.19	1.19 ± 0.76	1.92 ± 0.87	14.08 ± 0.69	38.82 ± 3.12	79.03 ± 4.08			
7d	1.51 ± 0.91	1.71 ± 0.12	2.32 ± 1.86	4.77 ± 1.57	12.54 ± 4.79	39.62 ± 3.93	87.75 ± 3.26			
10a	0.13 ± 0.06	0.40 ± 0.08	0.43 ± 0.10	0.83 ± 0.43	0.31 ± 0.11	1.27 ± 0.85	1.26 ± 0.82			
10b	0.17 ± 0.09	0.36 ± 0.09	0.12 ± 0.05	0.09 ± 0.02	0.85 ± 0.32	1.89 ± 0.72	1.52 ± 0.87			
10c	0.36 ± 0.08	1.37 ± 0.18	1.75 ± 0.53	3.87 ± 1.41	1.89 ± 0.67	34.67 ± 4.91	53.50 ± 6.04			
10d	1.28 ± 0.87	2.36 ± 1.23	3.31 ± 1.13	6.02 ± 1.11	39.95 ± 3.25	60.77 ± 4.37	74.32 ± 4.98			
13a	0.24 ± 0.05	0.27 ± 0.12	0.50 ± 0.12	0.81 ± 0.23	1.12 ± 0.87	1.23 ± 0.92	6.27 ± 2.32			
13b	0.37 ± 0.13	0.68 ± 0.37	1.03 ± 0.45	1.17 ± 0.74	4.05 ± 0.68	8.66 ± 1.85	44.95 ± 3.91			
13c	0.26 ± 0.05	0.70 ± 0.12	0.30 ± 0.09	1.40 ± 0.73	27.70 ± 3.96	51.09 ± 3.27	73.46 ± 4.76			
13d	0.56 ± 0.13	1.50 ± 0.72	3.98 ± 1.03	11.18 ± 2.86	48.65 ± 3.54	65.17 ± 3.91	83.68 ± 4.12			
16a	0.17 ± 0.04	0.21 ± 0.10	0.35 ± 0.12	0.47 ± 0.11	0.78 ± 0.17	1.23 ± 0.92	3.55 ± 1.03			
16b	0.19 ± 0.03	0.32 ± 0.26	0.45 ± 0.27	0.51 ± 0.76	0.80 ± 0.65	1.54 ± 0.92	4.37 ± 2.19			
16c	0.23 ± 0.09	0.44 ± 0.17	0.49 ± 0.21	0.58 ± 0.21	0.82 ± 0.27	1.63 ± 0.98	5.45 ± 1.72			
16d	0.28 ± 0.06	0.57 ± 0.22	0.73 ± 0.27	0.86 ± 0.51	1.53 ± 0.87	5.94 ± 2.13	21.39 ± 3.12			



Figure 2. Images of *E. coli* obtained by SEM (a, c & e; scale bar = $1.0 \,\mu$ m) and TEM (b, d & f; scale bar = $0.1 \,\mu$ m) following treatment with culture medium MHB (control; a & b), compound 7d under the concentration $100 \,\mu$ g/mL for 4 h (c & d), 12 h (e & f).



Figure 3. Images of *S. aureus* obtained by SEM (a, c & e; scale bar = $1.0 \,\mu$ m) and TEM (b, d & f; scale bar = $0.1 \,\mu$ m) following treatment with culture medium MHB (control; a & b), compound 7d under the concentration 100 μ g/mL for 4 h (c & d), 12 h (e & f).

cell envelope and cytoplasm. In addition, large quantities of electron-dense extracellular material were observed³⁶.

Conclusions

In this study, 24 N,N-bis(2-hydroxyethyl)-substituted guaternary ammonium salts bromides (5a-d, 6a-d, 7a-d, 10a-d, 13a-d, 16ad) were prepared based on 5-phenyl-1,3,4-oxadiazole-2-thiol (POT) analogues, benzo[d]oxazole-2-thiol, benzo[d]thiazole-2-thiol, 5methyl-1,3,4-thiadiazole-2-thiol and N,N-bis(2-hydroxyethyl) as well as several $\alpha_{t}\omega$ -dibromoalkanes. The results displayed that the antibacterial activities and antifungal activities of QASs were basically related to the introduction of POT, chain length and N,N-bis(2hydroxyethyl) quaternary ammonium salts. The long-chain and two flexible hydroxyethyl groups in QAS could help QAS effectively pass through cell membrane into the cell to passivate enzyme and damage bacteria. Especially, compounds 5d, 6d, 7d, 10d, 13d, 16d showed unsurpassed antimicrobial activities against all the experimental bacteria and fungi. Meanwhile, introduction of POT fragment in QAS could effectively reduce cytotoxicity and improve antimicrobial activity. Haemolysis assay showed all QASs did not cause haemolysis under the concentration of 50 µg/mL, in which concentration all QASs were safe to be used as antibacterial and antifungal agents. SEM images and TEM images of E. coli and S. aureus morphologies of 7d showed antibacterial mechanism might be the antibacterial agents fixing on membrane surfaces and puncturing of cell membranes to result in the release of bacterial cytoplasm. This study provided a facile strategy to design and develop new types of antibacterial and antifungal agents for application as daily or clinical disinfectants, as well as agricultural fungicides.

Disclosure statement

No potential conflict of interest was reported by the authors.

Funding

We are grateful for financial support from the National Natural Science Foundation of China (No. 21702018), the Foundation of Shandong province (Nos. ZR2013BM022, 2014CGZH1316, 2015GGX102013, BS2014YY049), the Foundation of Yantai city (Nos. 2016ZH078, 2013ZH095, 2014ZH092), and a project of Shandong province Higher Educational Science and Technology program (No. J17KA225).

References

- 1. Fan Z, Senapati D, Khan SA, et al. Popcorn-shaped magnetic core-plasmonic shell multifunctional nanoparticles for the targeted magnetic separation and enrichment, label-free SERS imaging, and photothermal destruction of multidrug-resistant bacteria. Chem Eur J 2013;19:2839–47.
- Xue Y, Xiao H, Zhang Y. Antimicrobial polymeric materials with quaternary ammonium and phosphonium salts. Int J Mol Sci 2015;16:3626–55.
- Duquebenítez SM, Ríosvásquez LA, Ocampocardona R, et al. Synthesis of novel quaternary ammonium salts and their in vitro antileishmanial activity and U-937 cell cytotoxicity. Molecules 2016;21:381.
- Shelton RS, Campen MG, Tilford CH, et al. Quaternary ammonium salts as germicides; non-acylated quaternary

ammonium salts derived from aliphatic amines. J Am Chem Soc 1946;68:753–5.

- 5. McDonnell G, Russell AD. Antiseptics and disinfectants: activity, action, and resistance. Clin Microbiol Rev 1999;12:147–79.
- Badawy HT, Pasetto P, Mouget JL, et al. Bacterial adhesion and growth reduction by novel rubber-derived oligomers. Biochem Biophy Res Commun 2013;438:691–6.
- Lu G, Wu D, Fu R. Studies on the synthesis and antibacterial activities of polymeric quaternary ammonium salts from dimethylaminoethyl methacrylate. React Funct Polym 2007;67:355–66.
- 8. Gong SQ, Niu LN, Kemp LK, et al. Quaternary ammonium silane-functionalized, methacrylate resin composition with antimicrobial activities and self-repair potential. Acta Biomaterialia 2012;8:3270–82.
- Obłąk E, Piecuch A, Guz-Regner K, et al. Antibacterial activity of gemini quaternary ammonium salts. FEMS Microbiol Lett 2014;350:190–8.
- Lukáč M, Lacko I, Bukovský M, et al. Synthesis and antimicrobial activity of a series of optically active quaternary ammonium salts derived from phenylalanine. CEJC 2010;8:194–201.
- 11. Tischer M, Pradel G, Ohlsen K, et al. Quaternary ammonium salts and their antibacterial potential: Targets or nonspecific interactions. ChemMedChem 2012;7:22–31.
- 12. Li J, Song XJ, Yang RJ. Synthesis and characterization of *N*,*N*-dimethyl dodecyl-(4-vinylbenzyl) ammonium chloride. Key Eng Mater 2014;575:71–5.
- Babbs M, Collier HOJ, Austin WC, et al. Salts of decamethylene-bis-4-aminoquinaldinium (dequadin); a new antimicrobial agent. J Pharm Pharmacol 1956;8:110–19.
- Franklin TJ, Snow GA, Biochemistry of antibacterial action.
 3rd ed. London: Chapman and Hall 1981;10.
- 15. Docherty KM, Kulpa CF. Jr, Toxicity and antimicrobial activity of imidazolium and pyridinium ionic liquids. Green Chem 2005;7:185–9.
- 16. Zhang Y, Li YY, Li JH, et al. Synthesis and antibacterial characterization of waterborne polyurethanes with gemini quaternary ammonium salt. Sci Bull 2015;60:1114–21.
- 17. Ma S, Izutani N, Imazato S, et al. Assessment of bactericidal effects of quaternary ammonium-based antibacterial monomers in combination with colloidal platinum nanoparticles. Dent Mater J 2012;31:150–6.
- Sajomsang W. Synthetic methods and applications of chitosan containing pyridylmethyl moiety and its quaternized derivatives: a review. Carbohyd Polym 2010;80:631–47.
- Wang CH, Hou GG, Du ZZ, et al. Synthesis, characterization and antibacterial properties of polyurethane material functionalized with quaternary ammonium salt. Polym J 2016;48:259–65.
- 20. Liu WS, Wang CH, Sun JF, et al. Synthesis, characterization and antibacterial properties of dihydroxy quaternary ammonium salts with long chain alkyl bromides. Chem Biol Drug Des 2015;85:91–7.
- 21. Wang CH, Liu WS, Sun JF, et al. Non-toxic O-quaternized chitosan materials with better water solubility and antimicrobial function. Int J Biolog Macromol 2016;84:418–27.
- 22. Siddiqa A, Rehman AU, Abbasi MA, et al. Synthesis and antibacterial screening of S-substituted derivatives of 5-(3,4methylenedioxyphenyl)-1,3,4-oxadiazol-2-thiol. Pak J Pharm Sci 2016;29:27–34.
- 23. Salahuddin Mazumder A, Shaharyar M. Synthesis, antibacterial and anticancer evaluation of 5-substituted (1,3,4-oxadiazol-2-yl)quinoline. Med Chem Res 2015;24:2514–28.

- 24. Khalid H, Aziz-ur-Rehman Abbasi MA, et al. Synthesis, spectral analysis and anti-bacterial study of *N*-substituted derivatives of 2-(5-(1-(phenylsulfonyl)piperidin-4-yl)-1,3,4-oxadiazol-2-ylthio)acetamide. J Saudi Chem Soc 2016;20:S615–23.
- 25. Aggarwal N, Kumar R, Dureja P, et al. Synthesis of novel nalidixic acid-based 1,3,4-thiadiazole and 1,3,4-oxadiazole derivatives as potent antibacterial agents. Chem Biol Drug Des 2012;79:384–97.
- 26. Tripathi P, Pal A, Jancik V, et al. Metal-assisted transformation of *N*-benzoyldithiocarbazate to 5-phenyl-1,3,4-oxadiazole-2-thiol in the presence of ethylenediamine, and its first row transition metal complexes. Polyhedron 2007;26: 2597–602.
- 27. Wang CH, Xie XR, Liu WS, et al. Quaternary ammonium salts substituted by 5-phenyl-1,3,4-oxadiazole-2-thiol as novel antibacterial agents with low cytotoxicity. Chem Biol Drug Des 2017;90:943–52.
- 28. Kobal E. Synthesis and antibacterial activity of fluorobenzoylcephalosporins. Int J Pharm 1991;75:131–9.
- 29. Bi FC, Guo LW, Wang YH, et al. Design, synthesis and biological activity evaluation of novel 2,6-difluorobenzamide derivatives through FtsZ inhibition. Bioorg Med Chem Lett 2017;27:958–62.
- 30. Aries V, Hill MJ. Degradation of steroids by intestinal bacteria. II. Enzymes catalysing the oxidoreduction of the 3α -, 7α -and 12α -hydroxyl groups in cholic acid, and the

dehydroxylation of the 7-hydroxyl group. BBA-Lipids Lipid Metab 1970;202:535–43.

- 31. Ultee A, Bennik MHJ, Moezelaar R. The phenolic hydroxyl group of carvacrol is essential for action against the foodborne pathogen Bacillus cereus. Appl Environ Microbiol 2002;68:1561–8.
- 32. Makino M, Ohta S, Zenda H. Study on new anti-rust disinfectants. III. Effect of alkyl chain length of *N*-alkyl-N-(2-hydroxy-3-phenoxy)propyl-*N*,*N*-dimethylammonium butyl phosphate on the antibacterial activity. Yakugaku Zasshi 1994;114:73–9.
- Li J, Guo YY. Basic evaluation of typical nanoporous silica nanoparticles in being drug carrier: Structure, wettability and hemolysis. Mater Sci Eng C 2017;73:670–3.
- 34. Sharma A, Bajpai VK, Baek K. Determination of antibacterial mode of action of Allium sativum essential oil against foodborne pathogens using membrane permeability and surface characteristic parameters. J Food Safety 2013;33:197–208.
- 35. Wang J, Sun H, Gao X, et al. Enhancing antibiofouling performance of polysulfone (PSF) membrane by photo-grafting of capsaicin derivative and acrylic acid. Appl Surf Sci 2014;317:210–19.
- 36. Dholvitayakhun A, Trachoo N, Narkkong NA, et al. Using scanning and transmission electron microscopy to investigate the antibacterial mechanism of action of the medicinal plant Annona squamosa, Linn. J Herb Med 2017;7:31–6.