

N-Cyclohexyl-2-(5-methoxy-1H-indol-3-yl)-2-oxoacetamide

Jing Liu,^a Yong-Feng Liu,^b Shi Zhang,^b Ying Gao^b and Hong Chen^{c*}

^aSchool of Pharmacy, Tianjin Medical University, Tianjin 300070, People's Republic of China, ^bRoom of Pharmacognosy, Medical College of Chinese People's Armed Police Forces, Tianjin 300162, People's Republic of China, and ^cTianjin Key Laboratory for Biomarkers of Occupational and Environmental Hazards, Tianjin 300162, People's Republic of China
Correspondence e-mail: tjch2010@yahoo.cn

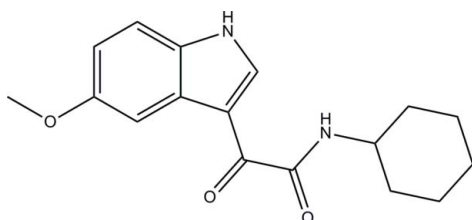
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Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.035; wR factor = 0.076; data-to-parameter ratio = 8.5.

In the title compound, $\text{C}_{17}\text{H}_{20}\text{N}_2\text{O}_3$, the cyclohexane ring adopts a chair conformation. In the crystal, intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into layers parallel to the ac plane.

Related literature

For the biological activity of indole derivatives, see: Souli *et al.* (2008); Chai *et al.* (2006); Radwan *et al.* (2007); Karthikeyan *et al.*, (2009). For details of the synthesis, see: Bacher *et al.* (2001). For similar structures, see: Feng *et al.* (2008); Sonar *et al.* (2006).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{20}\text{N}_2\text{O}_3$
 $M_r = 300.35$
Monoclinic, $P2_1$
 $a = 5.083$ (3) Å
 $b = 27.336$ (13) Å

$c = 5.220$ (3) Å
 $\beta = 91.977$ (12)°
 $V = 724.9$ (6) Å³
 $Z = 2$
Mo $K\alpha$ radiation

$\mu = 0.10$ mm⁻¹
 $T = 113$ K

$0.20 \times 0.18 \times 0.10$ mm

Data collection

Rigaku Saturn CCD area-detector diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)
 $T_{\min} = 0.981$, $T_{\max} = 0.991$

7342 measured reflections
1764 independent reflections
1569 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.076$
 $S = 1.02$
1764 reflections
208 parameters
1 restraint

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.23$ e Å⁻³
 $\Delta\rho_{\min} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O2}^{\text{i}}$	0.96 (3)	1.90 (3)	2.840 (2)	164 (3)
$\text{N2}-\text{H2}\cdots\text{O3}^{\text{ii}}$	0.88 (3)	2.09 (3)	2.926 (3)	160 (3)

Symmetry codes: (i) $x + 1, y, z - 1$; (ii) $x - 1, y, z$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *XP* in *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5025).

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supplementary materials

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***N*-Cyclohexyl-2-(5-methoxy-1*H*-indol-3-yl)-2-oxoacetamide**

J. Liu, Y.-F. Liu, S. Zhang, Y. Gao and H. Chen

Comment

Indole and their derivatives are well known as substances exhibiting various biological activities, such as anti-cancer (Souli *et al.*, 2008), anti-viral (Chai *et al.*, 2006), anti-tubercular (Karthikeyan *et al.*, 2009) and anti-inflammatory (Radwan *et al.*, 2007). In our search for new indole derivatives with improved activities, we have synthesized the title compound, (I). Here we report its crystal structure.

In (I) (Fig. 1), all bond lengths and angles are normal and in a good agreement with those reported previously (Feng *et al.*, 2008; Sonar *et al.*, 2006). The cyclohexane ring (C12—C17) adopts a chair conformation. In the crystal structure, intermolecular N—H···O hydrogen bonds (Table 1) link the molecules into layers parallel to the *ac* plane.

Experimental

The target compound was synthesized by two steps. Oxalyl chloride was added dropwise to a solution of 5-methoxy-indole in dry ether, the crude product 5-methoxyindol-3-yl-glyoxyl chloride, cyclohexylamine, two drops of triethylamine in dry dichloromethane. The reaction mixture was washed with water and dried over MgSO₄ and concentrated *in vacuo* (Bacher *et al.*, 2001). The residue was resolved in a methanol solution. Slow evaporation over two weeks at room temperature gave light-yellow crystals suitable for X-ray analysis.

Refinement

C-bound H atoms were geometrically positioned (C—H = 0.95–1.00 Å), and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5 U_{\text{eq}}(\text{C})$. Amino H atoms were located on a difference map and refined isotropically.

Figures

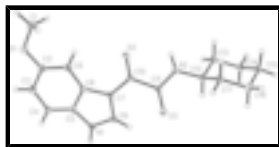


Fig. 1. View of the title compound, with displacement ellipsoids drawn at the 40% probability level.

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Crystal data

C₁₇H₂₀N₂O₃

$M_r = 300.35$

Monoclinic, $P2_1$

$F(000) = 320$

$D_x = 1.376 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

supplementary materials

Hall symbol: P 2₁y
 $a = 5.083$ (3) Å
 $b = 27.336$ (13) Å
 $c = 5.220$ (3) Å
 $\beta = 91.977$ (12)°
 $V = 724.9$ (6) Å³
 $Z = 2$

Cell parameters from 2636 reflections
 $\theta = 1.5$ – 27.9 °
 $\mu = 0.10$ mm⁻¹
 $T = 113$ K
Prism, colourless
 $0.20 \times 0.18 \times 0.10$ mm

Data collection

Rigaku Saturn CCD area-detector diffractometer

Radiation source: rotating anode multilayer

Detector resolution: 14.63 pixels mm⁻¹
 ω and φ scans

Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)

$T_{\min} = 0.981$, $T_{\max} = 0.991$

7342 measured reflections

1764 independent reflections

1569 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.042$

$\theta_{\max} = 27.9$ °, $\theta_{\min} = 1.5$ °

$h = -6 \rightarrow 6$

$k = -36 \rightarrow 31$

$l = -6 \rightarrow 6$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.035$

$wR(F^2) = 0.076$

$S = 1.02$

1764 reflections

208 parameters

1 restraint

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0439P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.003$

$\Delta\rho_{\max} = 0.23$ e Å⁻³

$\Delta\rho_{\min} = -0.18$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger. Along with the meaningless absolute structure parameter value (and s.u. value) obtained from any refinement with Friedel pairs, as justification of the merging of Friedel-pair data.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	−0.1844 (3)	0.03318 (6)	0.3071 (3)	0.0242 (4)
O2	−0.1506 (3)	0.20016 (5)	0.8241 (3)	0.0181 (3)
O3	0.3519 (3)	0.28545 (6)	0.8311 (3)	0.0209 (4)
N1	0.4663 (4)	0.19176 (7)	0.2050 (3)	0.0175 (4)
N2	−0.0791 (4)	0.29907 (6)	0.8997 (3)	0.0167 (4)
C1	−0.3621 (4)	0.03287 (9)	0.5127 (5)	0.0241 (5)
H1A	−0.2627	0.0360	0.6761	0.036*
H1B	−0.4607	0.0021	0.5106	0.036*
H1C	−0.4851	0.0603	0.4927	0.036*
C2	−0.0281 (4)	0.07411 (8)	0.2854 (4)	0.0191 (5)
C3	0.1505 (4)	0.07156 (8)	0.0851 (4)	0.0207 (5)
H3	0.1508	0.0435	−0.0223	0.025*
C4	0.3242 (4)	0.10905 (8)	0.0430 (4)	0.0185 (4)
H4	0.4446	0.1075	−0.0920	0.022*
C5	0.3169 (4)	0.14924 (7)	0.2052 (4)	0.0158 (4)
C6	0.3895 (4)	0.22142 (8)	0.3936 (4)	0.0182 (4)
H6	0.4632	0.2526	0.4319	0.022*
C7	0.1850 (4)	0.19958 (8)	0.5251 (4)	0.0147 (4)
C8	0.1372 (4)	0.15268 (8)	0.4031 (4)	0.0153 (4)
C9	−0.0386 (4)	0.11418 (8)	0.4453 (4)	0.0170 (4)
H9	−0.1603	0.1156	0.5792	0.020*
C10	0.0403 (4)	0.22048 (7)	0.7283 (4)	0.0150 (4)
C11	0.1203 (4)	0.27147 (7)	0.8258 (4)	0.0154 (4)
C12	−0.0360 (4)	0.35000 (7)	0.9780 (4)	0.0168 (4)
H12	0.1085	0.3636	0.8739	0.020*
C13	−0.2834 (4)	0.38039 (8)	0.9209 (4)	0.0187 (4)
H13A	−0.4302	0.3675	1.0208	0.022*
H13B	−0.3340	0.3777	0.7366	0.022*
C14	−0.2356 (5)	0.43413 (8)	0.9899 (4)	0.0194 (5)
H14A	−0.4000	0.4529	0.9587	0.023*
H14B	−0.0995	0.4478	0.8793	0.023*
C15	−0.1453 (4)	0.43918 (8)	1.2721 (4)	0.0184 (4)
H15A	−0.1048	0.4739	1.3106	0.022*
H15B	−0.2888	0.4287	1.3832	0.022*
C16	0.0984 (4)	0.40803 (8)	1.3293 (4)	0.0205 (5)
H16A	0.1486	0.4106	1.5137	0.025*
H16B	0.2466	0.4206	1.2301	0.025*
C17	0.0493 (4)	0.35425 (8)	1.2602 (4)	0.0185 (5)
H17A	0.2123	0.3351	1.2934	0.022*
H17B	−0.0897	0.3408	1.3682	0.022*
H1	0.603 (6)	0.2006 (11)	0.090 (6)	0.042 (8)*
H2	−0.240 (5)	0.2875 (10)	0.891 (5)	0.027 (7)*

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0251 (8)	0.0167 (8)	0.0310 (9)	-0.0030 (7)	0.0032 (7)	-0.0044 (7)
O2	0.0158 (7)	0.0174 (8)	0.0214 (8)	-0.0014 (6)	0.0052 (6)	-0.0016 (6)
O3	0.0157 (7)	0.0193 (8)	0.0277 (9)	-0.0012 (6)	0.0022 (6)	-0.0038 (6)
N1	0.0182 (9)	0.0182 (9)	0.0167 (9)	0.0019 (7)	0.0070 (7)	-0.0002 (7)
N2	0.0132 (9)	0.0159 (9)	0.0211 (10)	-0.0008 (7)	0.0020 (7)	-0.0023 (7)
C1	0.0225 (12)	0.0194 (11)	0.0303 (13)	-0.0028 (10)	0.0008 (10)	0.0029 (9)
C2	0.0180 (11)	0.0165 (11)	0.0226 (11)	0.0014 (8)	-0.0016 (9)	-0.0004 (8)
C3	0.0249 (12)	0.0198 (11)	0.0171 (11)	0.0046 (9)	-0.0033 (9)	-0.0045 (9)
C4	0.0190 (10)	0.0210 (12)	0.0155 (10)	0.0043 (9)	0.0015 (8)	-0.0021 (8)
C5	0.0169 (10)	0.0169 (10)	0.0137 (10)	0.0016 (8)	0.0016 (8)	0.0023 (8)
C6	0.0183 (10)	0.0175 (11)	0.0189 (10)	0.0013 (9)	0.0011 (8)	0.0005 (8)
C7	0.0133 (9)	0.0139 (10)	0.0168 (10)	0.0014 (8)	0.0012 (8)	0.0007 (8)
C8	0.0128 (10)	0.0176 (10)	0.0155 (10)	0.0027 (8)	0.0008 (8)	0.0006 (8)
C9	0.0160 (10)	0.0179 (11)	0.0171 (10)	0.0005 (9)	0.0017 (8)	0.0031 (8)
C10	0.0144 (10)	0.0138 (10)	0.0169 (10)	0.0016 (8)	0.0003 (8)	0.0007 (8)
C11	0.0163 (10)	0.0159 (11)	0.0141 (10)	-0.0001 (8)	0.0025 (8)	0.0006 (8)
C12	0.0173 (10)	0.0137 (10)	0.0196 (11)	0.0004 (8)	0.0028 (9)	-0.0011 (8)
C13	0.0197 (10)	0.0164 (10)	0.0200 (11)	0.0005 (9)	0.0019 (9)	-0.0015 (8)
C14	0.0204 (11)	0.0154 (10)	0.0225 (11)	0.0022 (8)	0.0035 (9)	0.0023 (8)
C15	0.0199 (11)	0.0148 (10)	0.0209 (11)	0.0004 (8)	0.0053 (9)	-0.0024 (8)
C16	0.0227 (11)	0.0195 (11)	0.0194 (11)	0.0010 (9)	0.0011 (9)	-0.0013 (8)
C17	0.0199 (11)	0.0174 (11)	0.0182 (11)	0.0038 (9)	0.0001 (9)	-0.0015 (8)

Geometric parameters (\AA , $^\circ$)

O1—C2	1.379 (3)	C7—C10	1.431 (3)
O1—C1	1.427 (3)	C7—C8	1.448 (3)
O2—C10	1.238 (2)	C8—C9	1.403 (3)
O3—C11	1.237 (2)	C9—H9	0.9500
N1—C6	1.344 (3)	C10—C11	1.534 (3)
N1—C5	1.389 (3)	C12—C17	1.526 (3)
N1—H1	0.96 (3)	C12—C13	1.528 (3)
N2—C11	1.331 (3)	C12—H12	1.0000
N2—C12	1.465 (3)	C13—C14	1.530 (3)
N2—H2	0.88 (3)	C13—H13A	0.9900
C1—H1A	0.9800	C13—H13B	0.9900
C1—H1B	0.9800	C14—C15	1.534 (3)
C1—H1C	0.9800	C14—H14A	0.9900
C2—C9	1.379 (3)	C14—H14B	0.9900
C2—C3	1.409 (3)	C15—C16	1.524 (3)
C3—C4	1.375 (3)	C15—H15A	0.9900
C3—H3	0.9500	C15—H15B	0.9900
C4—C5	1.388 (3)	C16—C17	1.532 (3)
C4—H4	0.9500	C16—H16A	0.9900
C5—C8	1.405 (3)	C16—H16B	0.9900

C6—C7	1.399 (3)	C17—H17A	0.9900
C6—H6	0.9500	C17—H17B	0.9900
C2—O1—C1	116.63 (17)	O3—C11—N2	123.5 (2)
C6—N1—C5	109.37 (18)	O3—C11—C10	121.94 (18)
C6—N1—H1	122.6 (18)	N2—C11—C10	114.56 (18)
C5—N1—H1	128.1 (18)	N2—C12—C17	112.11 (17)
C11—N2—C12	120.74 (17)	N2—C12—C13	110.36 (18)
C11—N2—H2	119.9 (18)	C17—C12—C13	110.58 (17)
C12—N2—H2	119.2 (18)	N2—C12—H12	107.9
O1—C1—H1A	109.5	C17—C12—H12	107.9
O1—C1—H1B	109.5	C13—C12—H12	107.9
H1A—C1—H1B	109.5	C12—C13—C14	110.68 (18)
O1—C1—H1C	109.5	C12—C13—H13A	109.5
H1A—C1—H1C	109.5	C14—C13—H13A	109.5
H1B—C1—H1C	109.5	C12—C13—H13B	109.5
C9—C2—O1	124.05 (18)	C14—C13—H13B	109.5
C9—C2—C3	121.8 (2)	H13A—C13—H13B	108.1
O1—C2—C3	114.11 (19)	C13—C14—C15	110.64 (18)
C4—C3—C2	121.0 (2)	C13—C14—H14A	109.5
C4—C3—H3	119.5	C15—C14—H14A	109.5
C2—C3—H3	119.5	C13—C14—H14B	109.5
C3—C4—C5	117.46 (19)	C15—C14—H14B	109.5
C3—C4—H4	121.3	H14A—C14—H14B	108.1
C5—C4—H4	121.3	C16—C15—C14	110.60 (18)
C4—C5—N1	129.47 (18)	C16—C15—H15A	109.5
C4—C5—C8	122.22 (19)	C14—C15—H15A	109.5
N1—C5—C8	108.29 (18)	C16—C15—H15B	109.5
N1—C6—C7	109.98 (19)	C14—C15—H15B	109.5
N1—C6—H6	125.0	H15A—C15—H15B	108.1
C7—C6—H6	125.0	C15—C16—C17	111.42 (19)
C6—C7—C10	127.11 (19)	C15—C16—H16A	109.3
C6—C7—C8	106.15 (17)	C17—C16—H16A	109.3
C10—C7—C8	126.63 (17)	C15—C16—H16B	109.3
C9—C8—C5	119.91 (19)	C17—C16—H16B	109.3
C9—C8—C7	133.89 (18)	H16A—C16—H16B	108.0
C5—C8—C7	106.20 (18)	C12—C17—C16	109.81 (17)
C2—C9—C8	117.55 (18)	C12—C17—H17A	109.7
C2—C9—H9	121.2	C16—C17—H17A	109.7
C8—C9—H9	121.2	C12—C17—H17B	109.7
O2—C10—C7	123.40 (19)	C16—C17—H17B	109.7
O2—C10—C11	118.42 (18)	H17A—C17—H17B	108.2
C7—C10—C11	118.15 (17)		
C1—O1—C2—C9	-1.3 (3)	C5—C8—C9—C2	-0.7 (3)
C1—O1—C2—C3	177.67 (19)	C7—C8—C9—C2	179.7 (2)
C9—C2—C3—C4	0.6 (3)	C6—C7—C10—O2	174.9 (2)
O1—C2—C3—C4	-178.4 (2)	C8—C7—C10—O2	-0.7 (3)
C2—C3—C4—C5	0.1 (3)	C6—C7—C10—C11	-2.9 (3)
C3—C4—C5—N1	-179.4 (2)	C8—C7—C10—C11	-178.55 (19)

supplementary materials

C3—C4—C5—C8	-1.2 (3)	C12—N2—C11—O3	4.1 (3)
C6—N1—C5—C4	178.7 (2)	C12—N2—C11—C10	-175.09 (18)
C6—N1—C5—C8	0.3 (2)	O2—C10—C11—O3	148.5 (2)
C5—N1—C6—C7	-0.2 (2)	C7—C10—C11—O3	-33.6 (3)
N1—C6—C7—C10	-176.2 (2)	O2—C10—C11—N2	-32.4 (3)
N1—C6—C7—C8	0.1 (2)	C7—C10—C11—N2	145.54 (19)
C4—C5—C8—C9	1.5 (3)	C11—N2—C12—C17	-85.5 (2)
N1—C5—C8—C9	-179.95 (19)	C11—N2—C12—C13	150.80 (19)
C4—C5—C8—C7	-178.77 (19)	N2—C12—C13—C14	-177.23 (17)
N1—C5—C8—C7	-0.2 (2)	C17—C12—C13—C14	58.2 (2)
C6—C7—C8—C9	179.8 (2)	C12—C13—C14—C15	-56.8 (2)
C10—C7—C8—C9	-3.9 (4)	C13—C14—C15—C16	55.7 (2)
C6—C7—C8—C5	0.1 (2)	C14—C15—C16—C17	-56.4 (2)
C10—C7—C8—C5	176.4 (2)	N2—C12—C17—C16	178.52 (17)
O1—C2—C9—C8	178.57 (19)	C13—C12—C17—C16	-57.9 (2)
C3—C2—C9—C8	-0.4 (3)	C15—C16—C17—C12	57.4 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O2 ⁱ	0.96 (3)	1.90 (3)	2.840 (2)	164 (3)
N2—H2 \cdots O3 ⁱⁱ	0.88 (3)	2.09 (3)	2.926 (3)	160 (3)

Symmetry codes: (i) $x+1, y, z-1$; (ii) $x-1, y, z$.

Fig. 1

