

Synthesis and Molecular Docking Study of New Thiazole Derivatives as Potential Tubulin Polymerization Inhibitors

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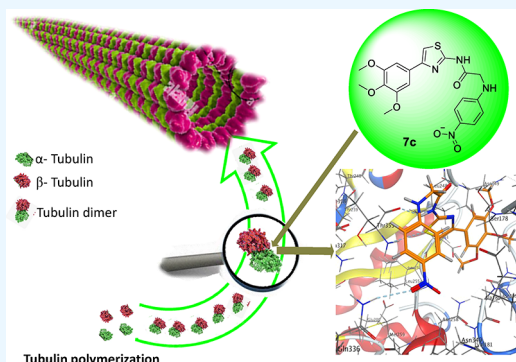
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ABSTRACT: A new series of 2,4-disubstituted thiazole derivatives containing 4-(3,4,5-trimethoxyphenyl) moiety was synthesized and evaluated for their potential anticancer activity as tubulin polymerization inhibitors. All designed compounds were screened for cytotoxic activity against four human cancer cell lines, namely, HepG2, MCF-7, HCT116, and HeLa, using 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide assay, with combretastatin A-4 as a reference drug. Compounds **5c**, **6d**, **7c**, **8**, and **9a,b** showed superior activity against the tested cell lines, with IC_{50} values ranging from 3.35 ± 0.2 to $18.69 \pm 0.9 \mu\text{M}$. Further investigation for the most active cytotoxic agents as tubulin polymerization inhibitors was also performed in order to explore the mechanism of their antiproliferative activity. The obtained results suggested that compounds **5c**, **7c**, and **9a** remarkably inhibit tubulin polymerization, with IC_{50} values of 2.95 ± 0.18 , 2.00 ± 0.12 , and $2.38 \pm 0.14 \mu\text{M}$, respectively, which exceeded that of the reference drug combretastatin A-4 (IC_{50} $2.96 \pm 0.18 \mu\text{M}$). Molecular docking studies were also conducted to investigate the possible binding interactions between the targeted compounds and the tubulin active site. The interpretation of the results showed clearly that compounds **7c** and **9a** were identified as the most potent tubulin polymerization inhibitors with promising cytotoxic activity and excellent binding mode in the docking study.



1. INTRODUCTION

Cancer is a group of diseases characterized by uncoordinated and uncontrolled cell growth with the potential to spread into or invade nearby tissues in a process called metastasis, which is the major cause of death from cancer.^{1,2} Factors that are associated with the high risk of cancer are tobacco use, lack of physical activity, alcohol use, low vegetable and fruit intake, and obesity. These factors are thought to account for approximately one third of cancer mortalities.³ Cancer pathogenesis involves several gene mutations which eventually lead to abnormal cell proliferation.⁴ The increased cancer incidence worldwide led to an impressive progress in the discovery of new, safer, and curative chemotherapeutic agents.⁵ In the scope of designing novel antitumor agents, we are particularly interested in the present work with thiazoles as one of the bioactive heterocyclic compounds that exhibited countless biological activities such as antimicrobial,^{6,7} antiprotozoal,^{8,9} antitumor,^{10,11} antiviral,^{12,13} antioxidant,¹⁴ antidiabetic,^{15,16} anti-inflammatory,^{17,18} anticonvulsant,¹⁹ antipsychotic,²⁰ and antihypertensive activities.²¹ Thiazole derivatives containing one or more thiazole rings represent a class of heterocyclic compounds of considerable importance as antitumor agents due to their remarkable affinity toward different biological targets involved in cancer pathogenesis such as tiazofurin²² and bleomycin.²³ Additionally, considerable attention and extensive efforts have been made to

enhance the antitumor activity of the 2-aminothiazole core in anticancer therapeutic areas as exemplified by dasatinib (**A**),²⁴ thia-netropsin (**B**),²⁵ and more recently alpelisib (**C**)²⁶ that was approved for medical use in 2019. Moreover, some 2-amino-4-phenylthiazol derivatives have been reported to inhibit tubulin polymerization, disrupt microtubule formation, and suppress cell division.^{27,28} However, combretastatin A-4 (CA-4) (**D**) is a natural trimethoxyphenyl (TMP) containing stilbenoid²⁹ that targets tubulin at the colchicine-binding site. CA-4 and other TMP-containing analogues such as combretoxazalone analogue (**E**) were reported as tubulin polymerization inhibitors.^{30,31} Inspired by the abovementioned facts, our objective was directed to gather two bioactive entities (3,4,5-trimethoxyphenyl and 2-aminothiazole) into one compact structure to be the main target skeleton for the development of new 2,4-disubstituted thiazole derivatives to suppress tubulin polymerization as an extremely attractive target for clinically effective anticancer drugs. Moreover, structure diversity was accom-

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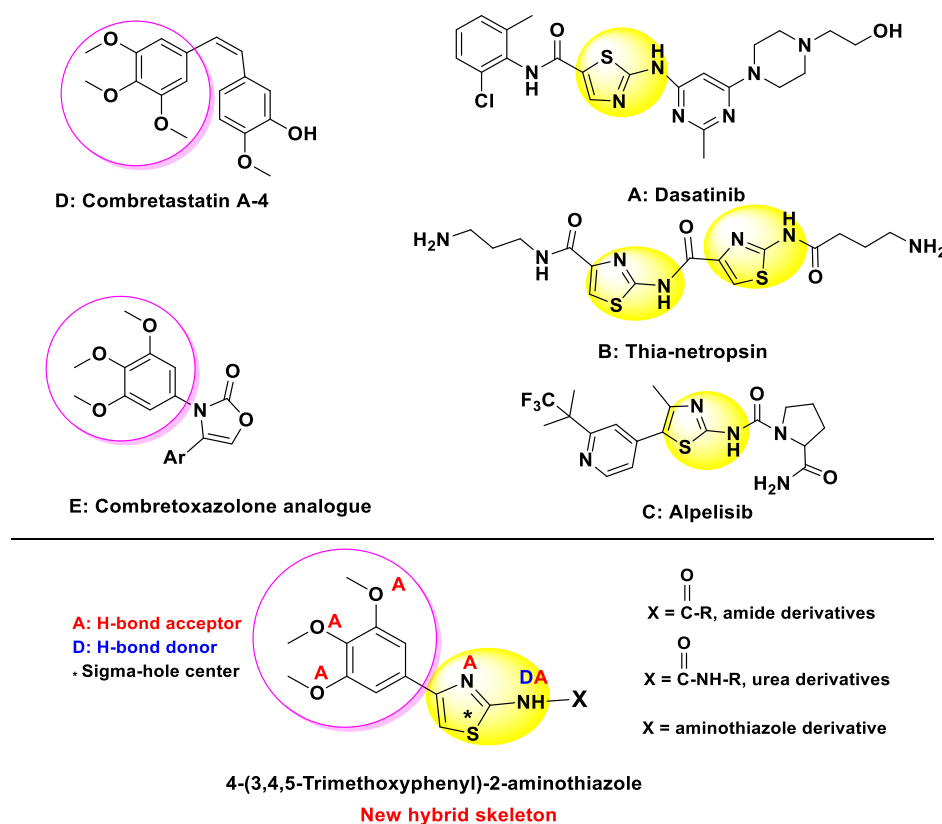


Figure 1. Structures of the reported lead compounds (A–E) and the newly designed targeted compounds.

plished by introducing variable bioactive side arms at the 2-position of the selected backbone, particularly amide derivatives, as in the synthesized compounds **4**, **5a–c**, **6a–e**, **7a–e**, **8**, and **9a,b**, and urea derivatives as our target compounds **11**, **12**, and **13a,b**. In addition, thiazolyl-2,4-diamine was also introduced at position 2 as in the new bis-thiazole compound **10**. The cytotoxic activity of the tested compounds and their potential inhibitory effect on tubulin polymerization were evaluated. In addition, molecular docking methodology was conducted to identify the possible binding interactions between the target compounds and the tubulin active site. The structures of the reported lead compounds (A–E) and the newly designed targeted compounds are shown in **Figure 1**.

2. RESULTS AND DISCUSSION

2.1. Chemistry. **2.1.1. Synthesis of Compounds 1–5 (Scheme 1).** Chloroacetylation of 2-aminothiazole **3** with chloroacetyl chloride was conducted in 1,4-dioxane in the presence of triethylamine to afford the chloroacetamide compound **4** as a key intermediate. It is noteworthy that the structure of compound **4** was verified by infrared (IR) spectroscopy that showed the appearance of a characteristic absorption band for the C=O group at 1690 cm^{-1} , which was also confirmed by ^{13}C NMR by a signal at δ 165.6 ppm.

The key intermediate **4** is considered as an α -halo amide that possesses key structural features, including an easily substituted chlorine atom, an active methylene group, and an amide moiety, which make it a versatile synthon in organic chemistry. Replacement of the α -halogen atom by different nitrogen nucleophiles is among the available methods for the synthesis of α -amino amides and creation of a new C–N bond. Hence, substitution reaction of compound **4** with primary aliphatic

amines such as 2-aminoethanol, *n*-butylamine, or isopropylamine was proceeded successfully in dry dimethyl formamide (DMF) in the presence of anhydrous K_2CO_3 to afford compounds **5a–c**.

2.1.2. Synthesis of Compounds 6–10 (Scheme 2). The target compounds **6a–e** were obtained by the reaction of the chloroacetamide compound **4** with a variety of secondary heterocyclic amines in dry DMF, and ^1H NMR spectra proved the inclusion of the secondary amines into their structures. For example, compounds **6a–d** showed either two sets of multiplets, one multiplet, two sets of triplets, or two sets of multiplets integrated for eight protons of the piperazine ring, respectively.

Moreover, the key intermediate **4** was reacted with different aniline derivatives in ethanol in the presence of HCl to afford the corresponding anilinoacetamide derivatives **7a–e**. The successful incorporation of the (un)substituted aniline moiety into the structure of compounds **7a–e** was verified based on the spectral data. For example, the ^1H NMR spectra of the later compounds showed the appearance of additional aromatic protons in the expected regions of the spectra. Moreover, their structures were further confirmed by mass spectroscopy that showed molecular ion peaks in accordance with the molecular formulae. In addition, compound **4** was treated with hydrazine hydrate in ethanol to furnish the corresponding hydrazinoacetamide **8**, which was subsequently condensed with isatin derivatives in ethanol and in the presence of catalytic drops of acetic acid. The carbonyl group at position-3 of isatin is more reactive than the carbonyl group at position-2; hence, it is readily involved in the condensation reaction with hydrazine hydrate, affording isatinmonohydrazone derivatives **9a,b**. The IR spectrum of compound **8** showed absorption bands at 3109 and 3188 cm^{-1} , representing NH_2 and NH of the hydrazine moiety, respectively.

However, the protons of the hydrazine moiety resonated as two singlets at δ 3.40 and δ 4.20 ppm, representing NH_2 and NH , respectively, which are D_2O exchangeable. The successful condensation of isatin derivatives with compound **8** was proved on the basis of the spectral data of isatin hydrazone products **9a,b**. Cyclization of the chloroacetamide compound **4** with thiourea was achieved by refluxing acetone in the presence of anhydrous K_2CO_3 , yielding the corresponding N^4 -phenylthiazole-2,4-diamine derivative **10**. The ^1H NMR spectrum of the later compound confirmed the presence of thiazole diamine functionality, and the structure was further verified by the presence of a molecular ion peak at m/z 364 in mass spectroscopy.

2.1.3. Synthesis of Compounds 11–13 (Scheme 3). A successful synthesis of 1-(2-chloroethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea **11** was achieved through the addition of the aminothiazole derivative **3** to 2-chloroethylisocyanate in dry DMF in the presence of anhydrous K_2CO_3 , and its structure was verified on the basis of the spectral data. The 2-chloroethyl urea derivative **11** was reacted with aniline, *N*-ethylpiperazine, or morpholine to afford the target compounds **12** and **13a,b**, respectively.

2.2. Biological Screening. **2.2.1. In Vitro Antitumor Evaluation.** The in vitro cytotoxicity study of the newly synthesized compounds was performed on four different cell lines, namely, HepG2, HCT-116, MCF-7, and HeLa cell lines, by employing MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide) assay^{32,33} using CA-4 as a reference drug. The tested compounds exhibited different degrees of cytotoxic activity against the tested cell lines, and the results are shown in Table 1.

2.2.1.1. Cytotoxic Activity of the Amide-Based Targeted Compounds (4–9). According to the results of biological screening, the key intermediate 2-chloroacetamide derivative **4** showed a weak cytotoxic effect (Table 1, Scheme 1). However, compounds **5a–c** generally showed strong-to-moderate activity against the four tested cell lines. Among these compounds, compound **5a**, with a hydroxyethyl amino acetamido side chain, showed moderate cytotoxic activity against the tested cell lines, while compound **5b** with a butylamino substituent showed a strong lethal effect on the HepG2 cell line with IC_{50} value of $10.92 \pm 0.5 \mu\text{M}$ and moderate effect on the other tested cell lines. Substitution with an isopropylamino acetamido side chain as in compound **5c** exploited a strong cytotoxic effect against HepG2, HCT-116, MCF-7, and HeLa cell lines with IC_{50} values of 11.87 ± 0.6 , 12.88 ± 0.6 , 15.49 ± 0.8 , and $16.02 \pm 0.8 \mu\text{M}$, respectively (Table 1, Scheme 1). Compound **6a** with an unsubstituted piperazine moiety showed selective cytotoxicity against the HepG2 cell line with IC_{50} value of $8.17 \pm 0.4 \mu\text{M}$ and moderate activity against the rest of the tested cell lines. It was noticed that substituting the piperazine moiety with a methyl or ethyl group as in compounds **6b** and **6c**, respectively, decreased their cytotoxic activity. However, connecting 4-fluorophenylpiperazine moiety to the acetamido side chain as in compound **6d** greatly enhanced its lethal effect over HepG2 with IC_{50} values of $4.03 \pm 0.2 \mu\text{M}$, which is better than that of the reference drug CA-4 (IC_{50} $4.50 \pm 0.2 \mu\text{M}$). Moreover, compound **6d** showed high cytotoxic activity over HCT-116, MCF-7, and HeLa cell lines with IC_{50} values of 15.74 ± 0.8 and 9.64 ± 0.5 , and $18.69 \pm 0.9 \mu\text{M}$, respectively. Replacing the piperazine moiety by morpholine afforded compound **6e** with strong cytotoxic activity against HepG2 and HCT-116. In compound **6e**, the piperazine ring was replaced by morpholine (Table 1, Scheme

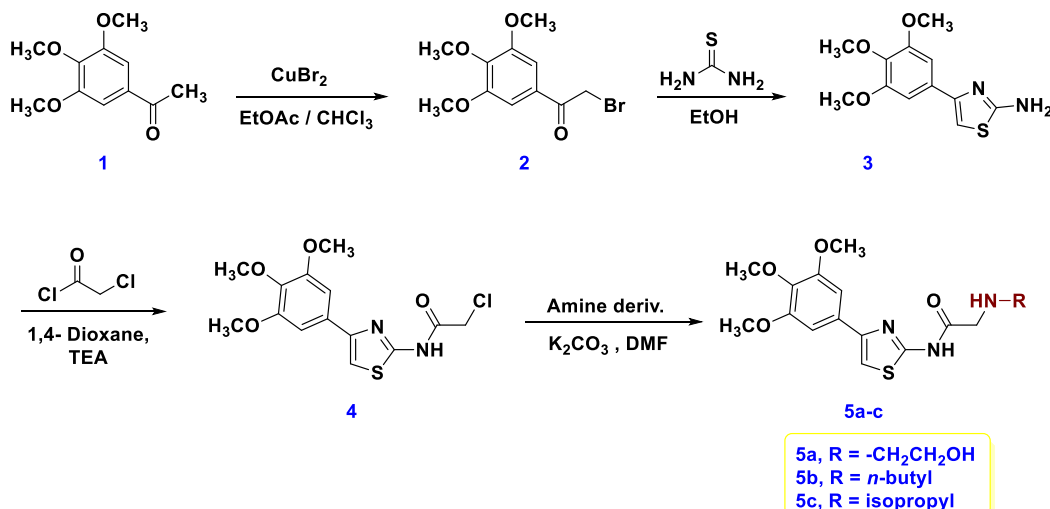
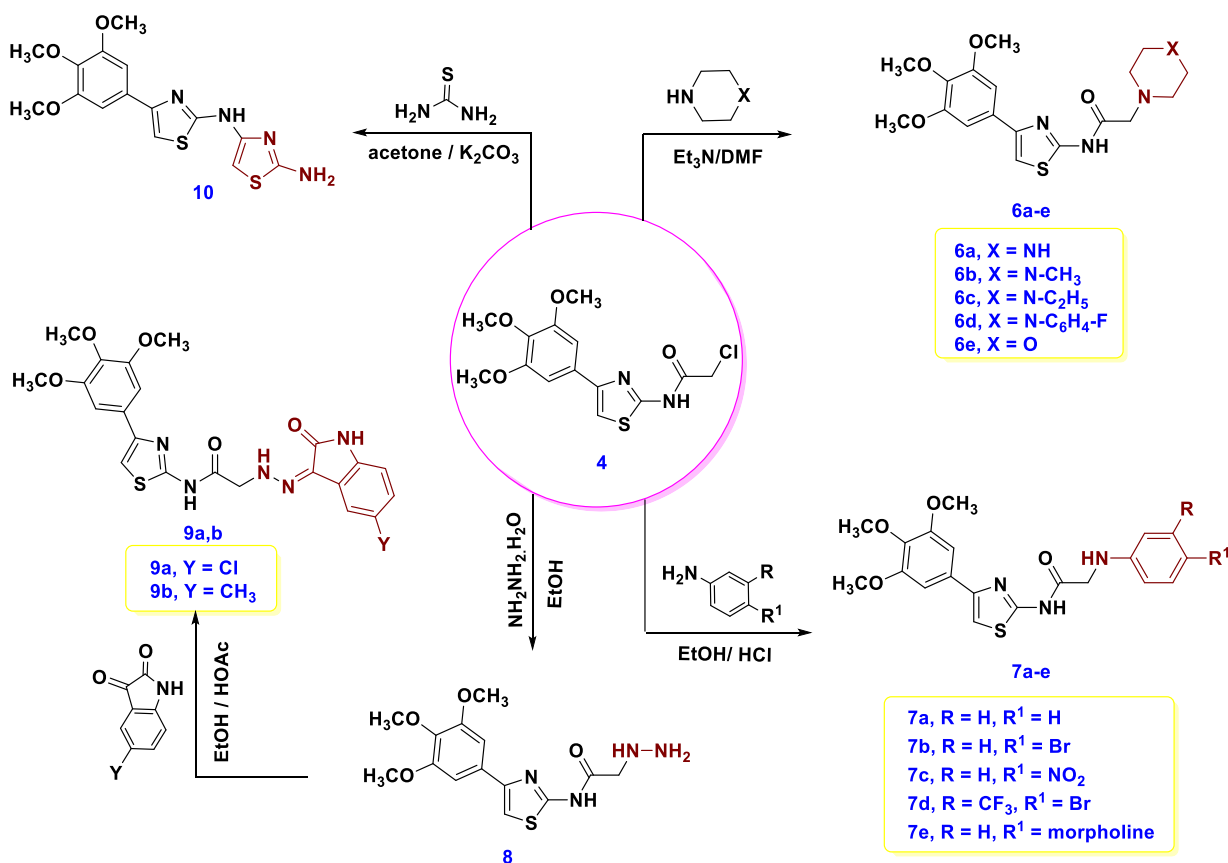
Table 1. In Vitro Cytotoxic Activity of the Designed Compounds

comp. no	IC_{50} (μM) ^a			
	HepG2 ^b	HCT-116 ^c	MCF-7 ^d	HeLa ^e
CA-4 ^f	4.50 ± 0.2	5.23 ± 0.3	4.17 ± 0.2	5.57 ± 0.3
4	62.57 ± 3.1	69.03 ± 3.5	62.91 ± 3.1	76.92 ± 3.8
5a	22.03 ± 1.1	29.36 ± 1.5	30.00 ± 1.5	50.86 ± 2.5
5b	10.92 ± 0.5	24.38 ± 1.2	22.80 ± 1.1	28.30 ± 1.4
5c	11.87 ± 0.6	12.88 ± 0.6	15.49 ± 0.8	16.02 ± 0.8
6a	8.17 ± 0.4	28.94 ± 1.4	40.87 ± 2.0	47.74 ± 2.4
6b	74.61 ± 3.7	83.68 ± 4.2	65.27 ± 3.3	87.32 ± 4.4
6c	33.86 ± 1.7	37.18 ± 1.9	41.36 ± 2.1	48.19 ± 2.4
6d	4.03 ± 0.2	15.74 ± 0.8	9.64 ± 0.5	18.69 ± 0.9
6e	19.76 ± 1.0	18.03 ± 0.9	39.08 ± 2.0	58.84 ± 2.9
7a	94.13 ± 4.7	52.48 ± 2.6	>100	54.09 ± 2.7
7b	34.17 ± 1.7	9.35 ± 0.5	44.84 ± 2.2	13.12 ± 0.7
7c	6.26 ± 0.3	7.30 ± 0.4	10.01 ± 0.5	9.85 ± 0.5
7d	28.10 ± 1.4	33.10 ± 1.7	41.42 ± 2.1	29.78 ± 1.5
7e	25.65 ± 1.3	31.15 ± 1.6	34.98 ± 1.7	41.45 ± 2.1
8	3.35 ± 0.2	5.32 ± 0.3	8.66 ± 0.4	6.15 ± 0.3
9a	7.32 ± 0.4	10.13 ± 0.5	9.80 ± 0.5	10.59 ± 0.5
9b	8.04 ± 0.4	9.93 ± 0.5	11.55 ± 0.6	13.42 ± 0.7
10	49.62 ± 2.5	55.30 ± 2.8	60.62 ± 3.0	48.03 ± 2.4
11	92.97 ± 4.6	86.69 ± 4.3	95.77 ± 4.8	89.86 ± 4.5
12	>100	88.46 ± 4.4	>100	91.82 ± 4.6
13a	14.17 ± 0.7	20.08 ± 1.0	16.95 ± 0.8	32.39 ± 1.6
13b	25.67 ± 1.3	27.42 ± 1.4	32.03 ± 1.6	43.93 ± 2.2

^a IC_{50} (μM): expressed as mean \pm SD 1–10 (very strong); 11–20 (strong); 21–50 (moderate); 51–100 (weak); and above 100 (nontoxic). ^bHuman hepatocellular carcinoma cell line (HepG2). ^cHuman breast adenocarcinoma cell line (MCF-7). ^dHuman colorectal carcinoma cell line (HCT-116). ^eEpitheloid cervix carcinoma cell line (HeLa). ^fCA-4: combretastatin A-4.

2). Reaction of 2-chloro-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide **4** with different (un)substituted aniline derivatives afforded the corresponding anilinoacetamides **7a–e** representing the aromatic substitution. The results from biological screening revealed that the unsubstituted aniline derivative **7a** showed weak cytotoxic activities against the HepG2, HCT-116, and MCF-7 cell lines and that it was nontoxic on the HeLa cell line. However, 4-substitution of the aniline moiety with bromo, nitro, or morpholino groups exerted a positive impact on increasing the activity as in compounds **7b**, **7c**, and **7e**, respectively. For example, compound **7b**, with a bromo substituent, showed a selective cytotoxic activity against HCT-116 and HeLa cell lines with IC_{50} values of 9.35 ± 0.5 and $13.12 \pm 0.7 \mu\text{M}$, respectively. Moreover, compound **7c**, with 4-nitroanilino acetamide side arm, displayed a notable cytotoxic and broad-spectrum activity among this series, with IC_{50} values ranging from 6.26 ± 0.3 to $10.01 \pm 0.5 \mu\text{M}$ against the four tested cell lines. However, disubstitution of the aniline moiety as in compound **7d** decreases the cytotoxic activity level comparable to the monosubstituted derivatives, while it still has more lethal effect than the unsubstituted compound **7a** (Table 1, Scheme 2). Remarkable cytotoxic activity was noticed for compound **8** obtained by hydrazinolysis of the chloroacetamide derivative **4** and was identified as the most potent and broad-spectrum cytotoxic agent among all the tested compounds. Compound **8** showed the most lethal effect over the HepG2 cell line with IC_{50} value of $3.35 \pm 0.2 \mu\text{M}$ which exceeded that of CA-4 (IC_{50} value of $4.50 \pm 0.2 \mu\text{M}$). Moreover,

Scheme 1. Synthesis of the Designed 2-Chloroacetamide Derivative 4 and 2-Substituted Amino Acetamides 5a–c

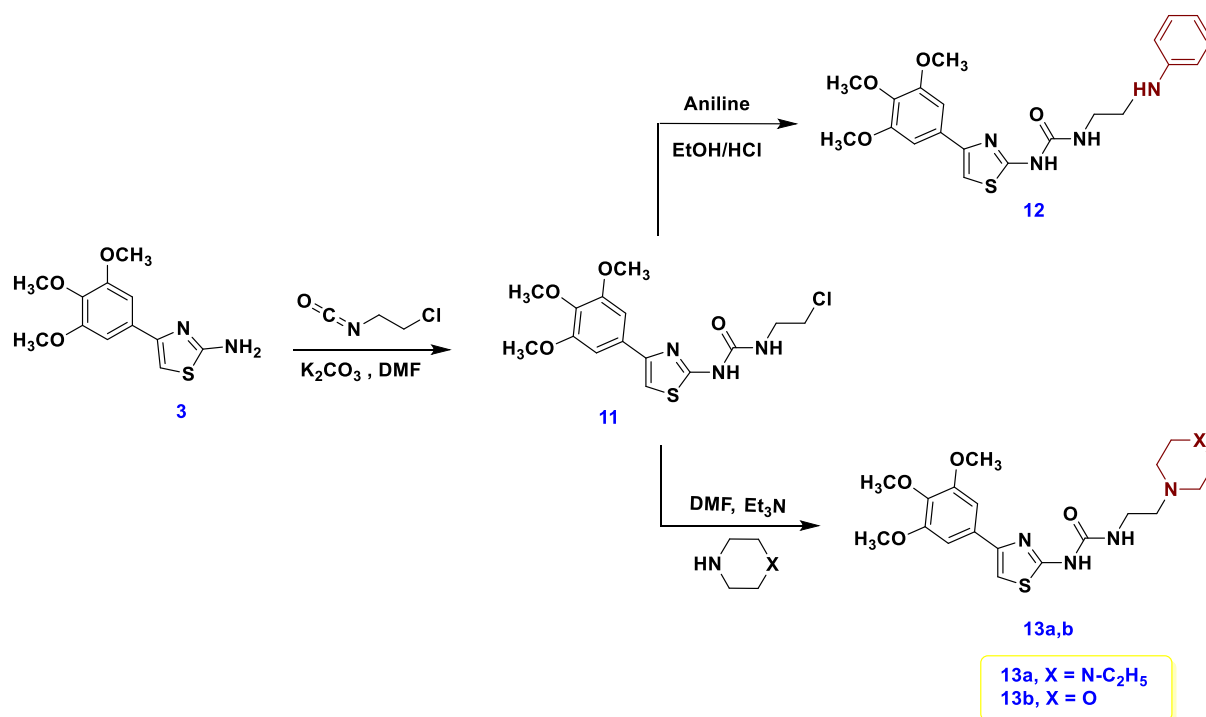
Scheme 2. Synthesis of the Designed 2-Substituted Acetamides 6–9 and *N*⁴-Phenylthiazole-2,4-Diamine Derivative 10

this compound showed an equipotent cytotoxic activity against HCT-116 with IC_{50} value of $5.32 \pm 0.3 \mu\text{M}$, which is comparable to that of the reference drug (IC_{50} value of $5.23 \pm 0.3 \mu\text{M}$). In addition, marked cytotoxic activity level was observed for compound 8 over both MCF-7 and HeLa cell lines with IC_{50} values of 8.66 ± 0.4 and $6.15 \pm 0.3 \mu\text{M}$, respectively (Table 1, Scheme 2). Regarding the isatinhydrazone derivatives, both methyl and chloro compounds 9a and 9b showed potent cytotoxicity against the four tested cancer cell lines with IC_{50} values ranging from 7.32 ± 0.4 to $13.42 \pm 0.7 \mu\text{M}$ (Table 1, Scheme 2).

2.2.1.2. Cytotoxic Activity of the Bis-Thiazole Compound 10. Compound 10 with two thiazole moieties in its core structure showed moderate-to-weak cytotoxicity against the tested cell lines, as revealed from the IC_{50} values (Table 1, Scheme 2).

2.2.1.3. Cytotoxic Activity of the Urea-Based Target Compounds (11–13). According to the results of biological screening, 1-(chloromethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea 11 showed a weak cytotoxic effect. Compound 12, in which an aniline moiety was connected to the methylurea side arm, showed the least activity against HCT-116 and HeLa

Scheme 3. Synthesis of the Designed Urea Derivatives 11–13



cell lines among the tested compounds and complete loss of cytotoxicity over HepG2 and MCF-7 cell lines. However, the corresponding piperazinylmethyl urea derivative **13a** showed strong cytotoxicity over HepG2, HCT-116, and MCF-7 cell lines with IC_{50} values of 14.17 ± 0.7 , 20.08 ± 1.0 , and $16.95 \pm 0.8 \mu\text{M}$, respectively. This compound also exerted a moderate effect against the HeLa cell line with IC_{50} values of $32.39 \pm 1.6 \mu\text{M}$, and the activity slightly decreased when the piperazine ring was replaced by morpholine as in compound **13b** (Table 1, Scheme 3).

It is conspicuously observed that most of our newly synthesized compounds (16 out of 22) showed their best antiproliferative activity against the HePG2 cell line and the rest of the compounds; **6e**, **7a**, **7b**, **11**, and **12** displayed their best antiproliferative activity against the HCT-116 cell line, whereas compound **6b** was seen to be the only compound that exhibited its best antiproliferative activity against the MCF-7 cell line. However, among these aforementioned 18 compounds selective for the HePG2 cell line, there are five compounds **6d**, **7c**, and **8–9b** that showed very strong activity (below $10 \mu\text{M}$) in the HePG2 cell line, with consistently good IC_{50} values across the investigated series of cell lines. Notably, all of these comprise two hydrogen-bond donors, for example, **7c** and **8**, or three hydrogen-bond donors, for example, **9a–b**, with the exception of **6d** that has only one.

2.2.2. In Vitro Tubulin Polymerization Inhibition Assay. To investigate whether the antitumor activity of the proposed compounds was related to the inhibition of the tubulin system, an in vitro tubulin polymerization inhibition assay was performed. Ten selected cytotoxic compounds including **5c**, **6a,d,e**, **7c,d**, **8** and **9a,b** and **13a** were evaluated for their tubulin polymerization inhibition activity, with CA-4 as the reference drug. IC_{50} values of the tested compounds were calculated and listed in Table 2 and graphically represented in Figure 2. Compounds **5c**, **7c**, and **9a** exhibited the highest inhibition of tubulin assembly among the tested compounds with IC_{50} values

Table 2. Results of the in Vitro Tubulin Polymerization Inhibition Assay and Docking Interaction Energy

comp. no	IC_{50} (μM) ^a	tubulin polymerization inhibition	docking interaction energy (kcal/mol)
CA-4 ^b		2.96 ± 0.18	-13.42
5c		2.95 ± 0.18	-13.88
6a		4.50 ± 0.27	-13.41
6d		5.00 ± 0.30	-13.09
6e		3.72 ± 0.23	-13.99
7c		2.00 ± 0.12	-14.15
7d		3.42 ± 0.21	-13.91
8		3.91 ± 0.24	-13.20
9a		2.38 ± 0.14	-14.50
9b		4.21 ± 0.26	-14.20
13a		5.11 ± 0.31	-13.25

^a IC_{50} (μM): expressed as mean \pm SD. ^bCA-4: combretastatin A-4.

2.95 ± 0.18 , 2.00 ± 0.12 , and $2.38 \pm 0.14 \mu\text{M}$, respectively, which are better than that of the standard drug ($\text{IC}_{50} = 2.96 \pm 0.18 \mu\text{M}$). The rest of the tested compounds showed a certain level of inhibition, with IC_{50} values ranging from 3.42 ± 0.21 to $5.11 \pm 0.31 \mu\text{M}$.

2.3. Docking Studies. Molecular docking was carried out for all the newly synthesized compounds into the colchicine binding site of our molecular target tubulin to confirm their capability for binding and explain their binding modes in comparison with CA-4 (Supporting Information data and Table 2). The values of free binding energies of the docked compounds **5c**, **6e**, **7b–d**, **9a,b**, and **12** ranged from -13.88 to -14.50 kcal/mol, which were higher than that of CA-4 (-13.42 kcal/mol). Compounds **5b**, **6a,b,d**, **7a**, **8**, and **13a** scored free binding energy in the range of -13.02 to -13.41 kcal/mol, while the rest of docked compounds showed less free binding energies (-12.99 to -12.09 kcal/mol) than that of CA-4. The high level of tubulin polymerization inhibition by the promising

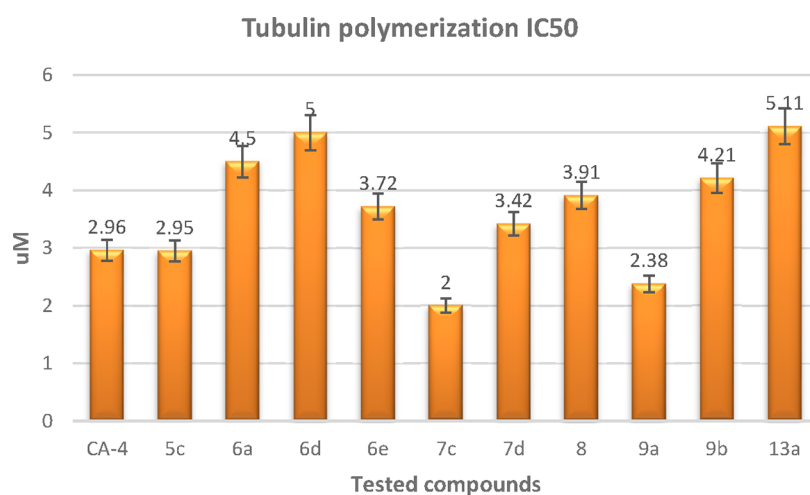


Figure 2. Tubulin polymerization inhibition chart of the tested compounds vs CA-4 (reference drug) expressed as IC₅₀ (μM).

compounds **5c**, **7c**, and **9a** prompted us to use these compounds along with the least active **13a** as representative examples to obtain a consistent and more precise picture of the biological active molecules at the atomic level and provide new insights that can be used to design novel therapeutic agents (Table 2 and Figures 3 and 4).

In the binding profile of the reference drug CA-4 (Figure 3, upper panel), the terminal 2-methoxylated phenol moiety constructed a conspicuous H-bond *via* its hydroxyl group that acted as a H-donor with the H-acceptor-conserved amino acid SerA178 along with strong hydrophobic/hydrophilic interactions in between, easily detected from the deep blue and cyan shadows around OH and SerA178, respectively. Having an olefinic bridge, CA-4 is trapped in *cis* conformer that gave rise to the confrontation of 3,4,5-trimethoxy groups to the conserved amino acids LeuB248, LysB254, LeuB255, AlaB316, LysB352, and ThrB353, establishing intense hydrophobic/hydrophilic interactions, paving the way for the stability of the ligand/complex to score free binding energy of -13.42 kcal/mol. The molecular docking results of compound **5c** (Figure 3, middle panel) demonstrated a thiazole ring strongly involved in noncovalent bindings: sulfur bond with AsnB249 and arene-H bond with AsnA101. The amidic nitrogen formed a H-bond with the H-acceptor-conserved amino acid SerA178 in the pocket core. Additionally, the noticeable hydrophobic/hydrophilic interactions appeared in both terminals of the ligand (trimethoxy groups and isopropyl fragment) as a blue shadow, improved the overall ligand recognition, and augmented the ligand/receptor complex stability with free binding energy of -13.88 kcal/mol. However, compound **7c** (Figure 3, lower panel) showed a higher stability of ligand/receptor complex with free binding energy of -14.15 kcal/mol. ThrB353 built a bifurcated bond with sulfur atom and amidic NH, GlnB247 acted as a H-bond donor to amidic carbonyl, and GlnB336 acted as a H-bond donor to the double-bond oxygen atom of the terminal nitro group.

Furthermore, 5-chloro-3-oxoindolin-2-ylidene in compound **9a** (Figure 4, upper panel) formed an arene-H bond through its pyrrole ring that acted as a nonclassical Lewis base with the H-donor-conserved amino acid AsnB249. Moreover, the hydrogen atom of the amino group of the hydrazino moiety established a H-bond with ThrA179. Besides, the excellent binding mode of thiazole moiety formed two different types of interactions; its sulfur established a noncovalent bond with AsnA101, and its

aromaticity with its electron cloud as a nonclassical Lewis base formed an arene-H bond with LeuB248, improving significantly the stability of the ligand/receptor complex to score the best free binding energy of -14.50 kcal/mol among the docked compounds. Docking of urea derivative **13a** (Figure 4, lower panel), which showed the least tubulin inhibition activity among the tested compounds, demonstrated that ThrB353 acted as the wedge to fix two sites of the ligand; it bound with thiazole sulfur and the hydrogen atom of the interior amino group of urea moiety by the sulfur-bond and H-bond, respectively. In spite of the cyan shadow around LeuB28, LysB352, LeuB255, and AsnA101 that reflected the hydrophobic/hydrophilic interactions between the ligand and the tubulin hot spot, the ligand scored free binding energy of -13.25 kcal/mol, comparatively lower than those of **5c**, **7c**, and **9a**. As per the molecular docking results, it can be stated that the thiazole ring with its variable bioactive side arms has played a great impact in destabilising tubulin through a variety of binding interactions. Sulfur has σ -hole features that give rise to the formation of σ -hole bonding as a recently recognized noncovalent bonding type similar to halogen bond and thus has attracted the attention in medicinal chemistry³⁴ and became a fundamental tool in rational drug design.^{35–39} Furthermore, TMP with its steric effect has steered efficiently the ligands into the appropriate binding mode by its miscellaneous interactions, so it is a successful pharmacophore to inhibit the tubulin assembly, in agreement with Tron et al.⁴⁰ CA-4 as an olefinic structure with *cis*-restricted rotation has two obvious hindrances; first, it can convert rapidly to a *trans*-isomer which is 100-fold less active, by the effect of light, heat, or even protic solvent,⁴¹ that is, having poor metabolic stability ascribed to its olefinic bridge, and second, it is weakly soluble in water.⁴² Therefore, there is an eminent need for alternatives; so, we recommend, herein, that compounds **7c** and **9a** are excellent candidates for further clinical studies as tubulin assembly inhibitors.

3. CONCLUSIONS

In this study, newly synthesized 2-substituted-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide, 2-substituted-4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea, and (4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)thiazole-2,4-diamine were subjected to antitumor evaluation as potential tubulin polymerization inhibitors. The obtained data revealed that compounds **5c**, **6d**, **7c**, **8**, and **9a,b** showed strong activity against the tested cell lines

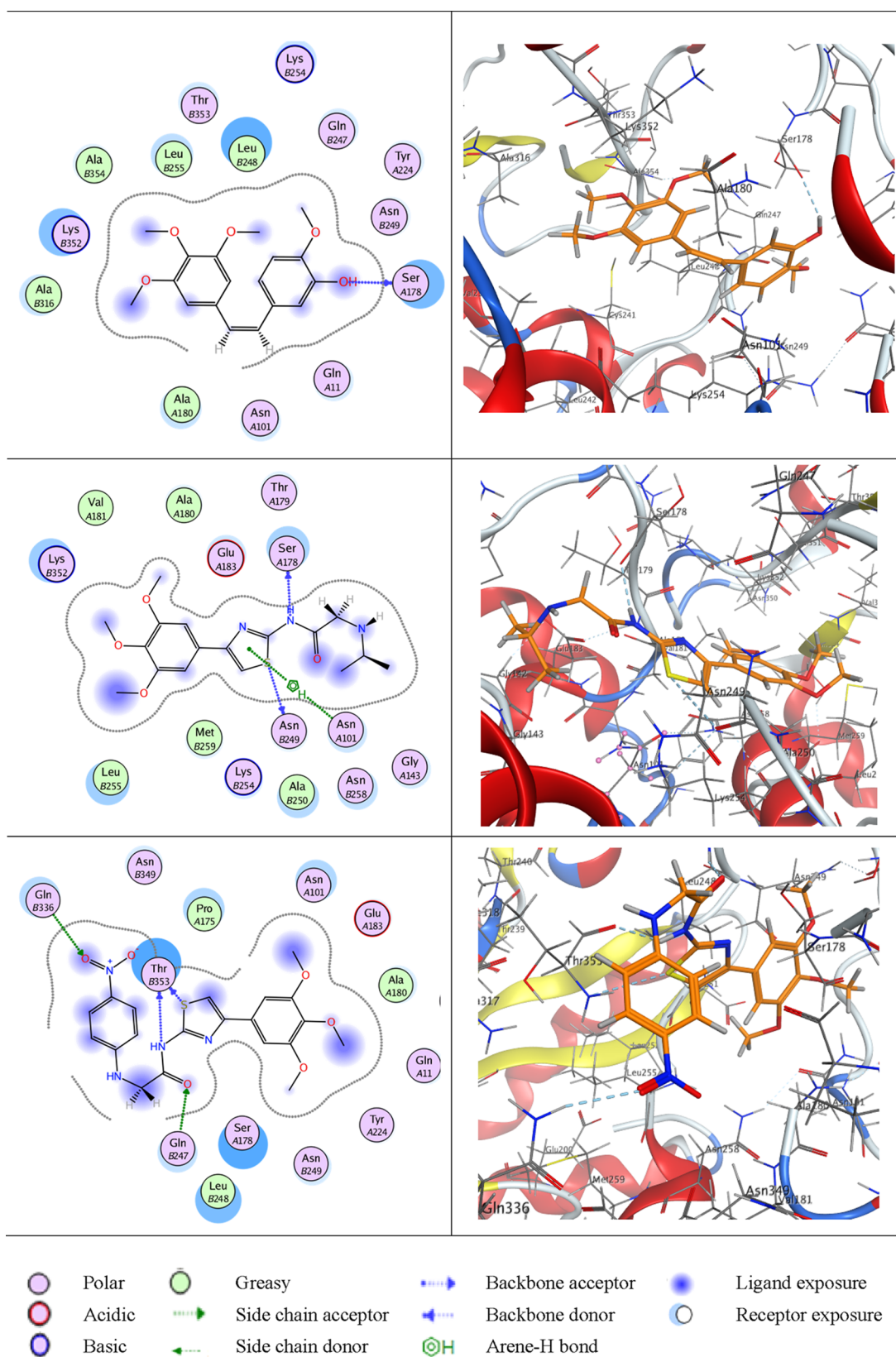


Figure 3. 2D and 3D interactions of CA-4 (upper panel), compounds 5c (middle panel), and 7c (lower panel) with the receptor pocket of tubulin (PDB ID: 4O2B).

(HepG2, MCF-7, HCT116, and HeLa in comparison with CA-4. In addition, the high level of inhibition of tubulin polymerization by compounds 5c, 7c, and 9a with IC_{50} better

than that of the CA-4 reference drug indicated that their antitumor activity may be mediated through this mechanism, and this was further supported by the docking study. There was

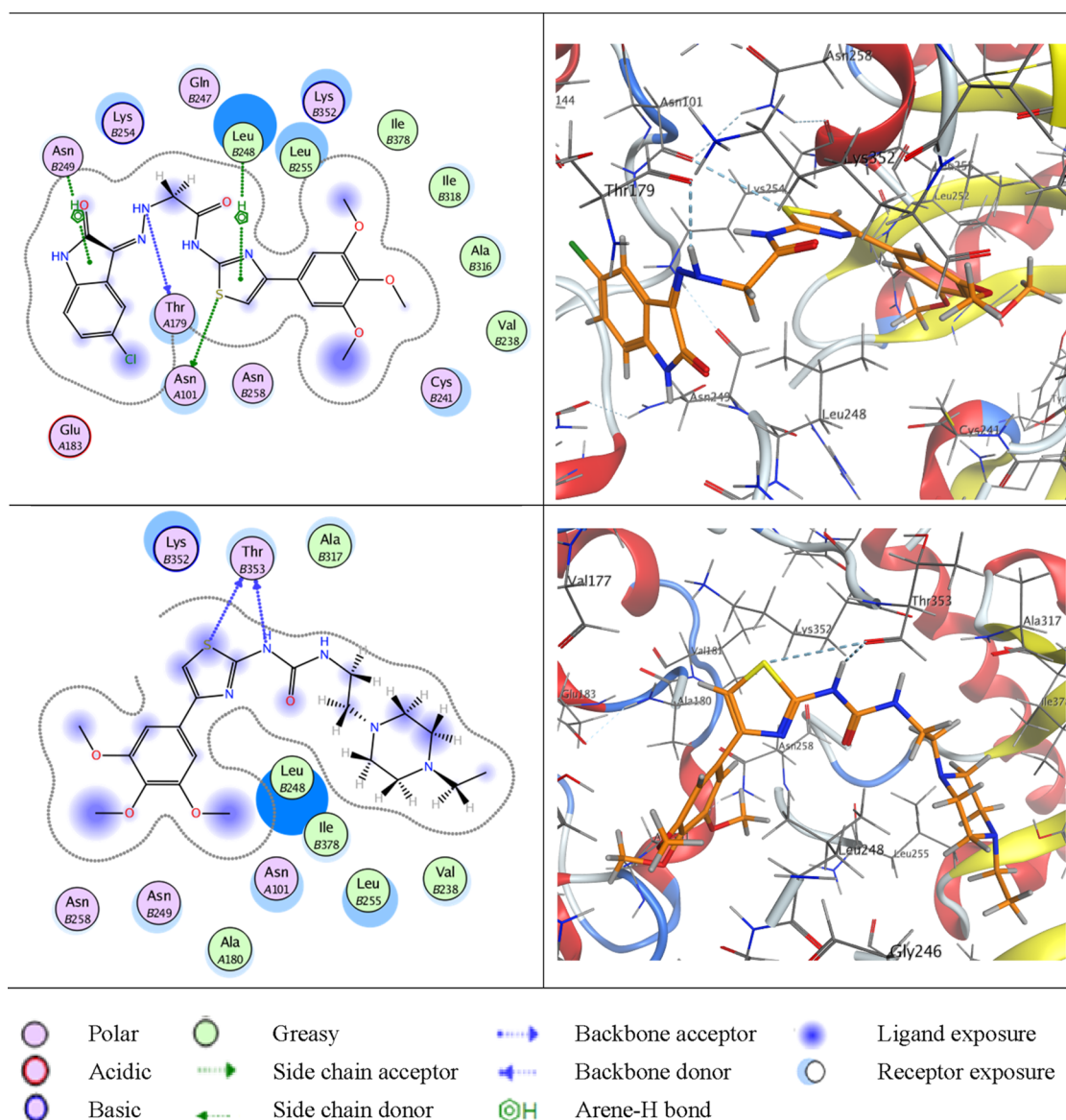


Figure 4. 2D and 3D interactions of compounds **9a** (upper panel) and **13a** (lower panel) with the receptor pocket of tubulin (PDB ID: 4O2B).

good agreement between the results of the docking study and the tubulin inhibition assay. The promising cytotoxic compounds **7c** and **9a** inhibit tubulin polymerization with IC_{50} 2.00 ± 0.12 and $2.38 \pm 0.14 \mu\text{M}$, respectively, and they also score free binding energy of -14.15 and -14.50 kcal/mol, respectively. Hence, these new compounds might be used as the leading units to develop novel tubulin polymerization inhibitors as potential anticancer agents.

4. EXPERIMENTAL SECTION

4.1. Chemistry. Melting points ($^{\circ}\text{C}$) were recorded using a Stuart melting point apparatus and are uncorrected. IR spectra were recorded on a Mattson 5000 FT-IR spectrophotometer (ν in cm^{-1}) using a potassium bromide disk at the Faculty of Pharmacy, Mansoura University. ^1H NMR and ^{13}C NMR spectra were determined by using a NMR spectrometer, Bruker 400, UK, at the Faculty of Pharmacy, Mansoura University. A proper amount of each compound was dissolved in either $\text{DMSO}-d_6$ or CDCl_3 before measurement. Tetramethylsilane was used as the internal standard. MS was performed on a Hewlett Packard 5988

spectrometer, at Al-Azhar University, Cairo, Egypt. Elemental microanalyses of the synthesized compounds were performed by the method proposed by Sullivan et al., at Al-Azhar University, Cairo, Egypt. The completion of reactions was monitored using thin-layer chromatography (TLC) plates, precoated with silica gel 60 F254 (E. Merck), and the spots were visualized under UV light (254, 365 nm). Pet. ether/EtOAc (1:1) or (3:1) were adopted as elution solvents. All chemicals and reagents were purchased from Sigma-Aldrich, and the solvents were obtained from El-Gomhouria Company for Pharmaceuticals and Chemicals. 2-Bromo-1-(3,4,5-trimethoxyphenyl)ethan-1-one (**2**)⁴³ and 4-(3,4,5-trimethoxyphenyl)thiazol-2-amine (**3**)⁴⁴ were prepared according to the previous report.

4.1.1. Procedure for the Synthesis of 2-Chloro-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (4). To a mixture of compound **3** (2.6 g, 0.01 mol) in 1,4-dioxane (25 mL) and triethylamine (1 mL), chloroacetyl chloride (0.8 mL, 0.01 mol) was added dropwise at (5–10) $^{\circ}\text{C}$. The reaction mixture was stirred at 70 $^{\circ}\text{C}$ for 6 h. After the completion of the reaction, the solvent was reduced and then the reaction mixture was poured

into crushed ice, and the obtained solid was recrystallized from ethanol.

Brown crystals, mp 173–176 °C, yield (70%), IR (KBr, ν , cm^{-1}): 3450 (N–H amide), 1690 (C=O amide), 1658 (C=N), 1592 (C=C), 1236 (C–O), 761 (C–Cl), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 3.67 (s, 3H, $p\text{-OCH}_3$), 3.82 (s, 6H, $m\text{-OCH}_3$), 4.39 (s, 2H, COCH_2Cl), 7.18 (s, 2H, H-2 and H-6 Ar), 7.71 (s, 1H, thiazole-H), 12.50 (s, 1H, NH, D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 42.7 (CCH_2Cl), 56.4 ($m\text{-OCH}_3$), 60.6 ($p\text{-OCH}_3$), 103.6 (2,6-Ar-C), 108.7 (thiazole-C-5), 130.3 (1-Ar-C), 137.9 (4-Ar-C), 149.4 (thiazole-C-4), 153.6 (3,5-Ar-C), 157.7 (thiazole-C-2), 165.6 (C=O), MS (m/z , Rel. Abundance): 342.07 (M^+ , 55.31), 343.16 ($\text{M} + 1$, 15.91), 344.25 ($\text{M} + 2$, 15.75), 109.08 (100), Anal. Calcd for $\text{C}_{14}\text{H}_{15}\text{ClN}_2\text{O}_4\text{S}$ calculated %: C, 49.05; H, 4.41; N, 8.17; S, 9.35. Found %: C, 49.08; H, 4.45; N, 8.14; S, 9.33.

4.1.2. General Procedure for the Synthesis of 2-Substituted *N*-(4-(3,4,5-Trimethoxyphenyl)thiazol-2-yl)acetamide Derivatives (5a–c). A mixture of compound 4 (0.34 g, 0.001 mol), aliphatic amines (0.001 mol), and anhydrous K_2CO_3 (0.14 g, 0.001 mol) in dry DMF (10 mL) was heated under reflux for 8–9 h. After the reaction completion, the reaction mixture was poured onto crushed ice, and the precipitated product was filtered, dried, and recrystallized from acetone.

4.1.2.1. 2-((2-Hydroxyethyl)amino)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (5a). Brown crystals, mp 85–88 °C, yield (71%), IR (KBr, ν , cm^{-1}): 3410 (OH), 3347 (N–H amide), 3114 (2° amine NH), 1662 (C=O amide), 1589 (C=C), 1234 (C–O), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 3.09 (s, 2H, COCH_2), 3.40 (t, $J = 12.80$ Hz, 2H, NHCH_2), 3.57 (t, $J = 17.60$ Hz, 2H, CH_2OH), 3.67 (s, 3H, $p\text{-OCH}_3$), 3.84 (s, 6H, $m\text{-OCH}_3$), 4.80 (s, 1H, OH, D_2O exchangeable), 5.20 (s, 1H, CH_2NH , D_2O exchangeable), 7.10 (s, 2H, H-2 and H-6 Ar), 7.20 (s, 1H, thiazole-H), 12.50 (s, 1H, NHCO , D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 52.0 (NHCH_2), 56.3 ($m\text{-OCH}_3$), 60.5 ($p\text{-OCH}_3$), 60.5 (CH_2OH), 61.6 ($\text{COCH}_2\text{-NH}$), 101.6 (thiazole-C-5), 103.3 (2,6-Ar-C), 131.2 (1-Ar-C), 137.3 (4-Ar-C), 150.2 (thiazole-C-4), 153.3 (3,5-Ar-C), 168.5 (thiazole-C-2), 171.0 (C=O), MS (m/z , Rel. Abundance): 367.1 (M^+ , 33.67), 265.9 (100), Anal. Calcd for $\text{C}_{16}\text{H}_{21}\text{N}_3\text{O}_5\text{S}$ calculated %: C, 52.30; H, 5.76; N, 11.44; S, 8.73. Found %: C, 52.34; H, 5.72; N, 11.41; S, 8.72.

4.1.2.2. 2-(Butylamino)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (5b). Brown crystals, mp 87–92 °C, yield (64%), IR (KBr, ν , cm^{-1}): 3490 (N–H amide), 3331 (2° amine NH), 1655 (C=O amide), 1590 (C=C), 1235 (C–O), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 0.88 (t, $J = 16.00$ Hz, 3H, CH_2CH_3), 1.32 (m, 4H, $\text{CH}_2\text{CH}_2\text{CH}_3$), 2.70 (t, $J = 16.00$ Hz, 2H, NHCH_2), 3.40 (s, 2H, COCH_2), 3.52 (s, 1H, NH, D_2O exchangeable), 3.70 (s, 3H, $p\text{-OCH}_3$), 3.84 (s, 6H, $m\text{-OCH}_3$), 7.10 (s, 2H, H-2 and H-6 Ar), 7.64 (s, 1H, thiazole-H), 12.50 (s, 1H, NHCO , D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 13.5 (butyl amine-C-4), 20.3 (butyl amine-C-3), 32.4 (butyl amine-C-2), 49.1 (butyl amine-C-1), 52.8 (COCH_2), 56.9 ($m\text{-OCH}_3$), 60.7 ($p\text{-OCH}_3$), 101.5 (2,6-Ar-C), 107.2 (thiazole-C-5), 128.1 (1-Ar-C), 141.1 (4-Ar-C), 151.1 (thiazole-C-4), 154.1 (3,5-Ar-C), 165.2 (thiazole-C-2), 169.6 (C=O), MS (m/z , Rel. Abundance): 381.62 (M^+ , 19.52), 379.29 (100), Anal. Calcd for $\text{C}_{18}\text{H}_{25}\text{N}_3\text{O}_4\text{S}$ calculated %: C, 56.97; H, 6.64; N, 11.07; S, 8.45. Found %: C, 57.01; H, 6.67; N, 11.05; S, 8.46.

4.1.2.3. 2-(Isopropylamino)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (5c). Black crystals,

mp 178–181 °C, yield (68%), IR (KBr, ν , cm^{-1}): 3424 (N–H amide), 3331 (2° amine NH), 1655 (C=O amide), 1588 (C=C), 1235 (C–O), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 1.01 (d, $J = 6.40$ Hz, 6H, CH_3), 1.05 (m, 1H, CH_3 , CHCH_3), 3.20 (s, 2H, COCH_2), 3.40 (s, 1H, CH_2NH , D_2O exchangeable), 3.70 (s, 3H, $p\text{-OCH}_3$), 3.83 (s, 6H, $m\text{-OCH}_3$), 7.15 (s, 2H, H-2 and H-6 Ar), 7.26 (s, 1H, thiazole-H), 12.50 (s, 1H, NHCO , D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 23.2 (isopropyl- CH_3), 50.5 ($\text{COCH}_2\text{-NH}$), 51.3 (isopropyl- CH), 57.34 ($m\text{-OCH}_3$), 61.9 ($p\text{-OCH}_3$), 100.5 (2,6-Ar-C), 105.2 (thiazole-C-4), 127.5 (1-Ar-C), 139.1 (4-Ar-C), 150.3 (thiazole-C-5), 153.3 (3,5-Ar-C), 164.5 (thiazole-C-2), 168.7 (C=O), MS (m/z , Rel. Abundance): 364.45 (M^+ , 37.24), 109.38 (100), Anal. Calcd for $\text{C}_{17}\text{H}_{23}\text{N}_3\text{O}_4\text{S}$ calculated %: C, 55.87; H, 6.34; N, 11.50; S, 8.77. Found %: C, 55.84; H, 6.35; N, 11.48; S, 8.81.

4.1.3. General Procedure for the Synthesis of Substituted 4-(3,4,5-Trimethoxyphenyl)thiazol-2-yl)acetamide Derivatives (6a–e). A mixture of compound 4 (1 g, 0.003 mol) and the appropriate secondary amines (0.003 mol) in dry DMF (10 mL) containing few drops of triethylamine (3–4 drops) was heated under reflux for 8 h. The reaction mixture was left to cool, then poured onto ice-cold water, and the formed precipitate was filtered, dried, and recrystallized from ethanol.

4.1.3.1. 2-(Piperazin-1-yl)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (6a). Brown crystals, mp 258–261 °C, yield (59%), IR (KBr, ν , cm^{-1}): 3449 (N–H amide), 3114 (2° amine NH), 1656 (C=O amide), 1555 (C=N), 1267 (C–O), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 1.78 (s, 1H, piperazine-NH, D_2O exchangeable), 2.47 (m, 4H, piperazine- CH_2), 2.75 (m, 4H, piperazine- CH_2), 3.28 (s, 2H, COCH_2), 3.69 (s, 3H, $p\text{-OCH}_3$), 3.84 (s, 6H, $m\text{-OCH}_3$), 7.21 (s, 2H, H-2 and H-6 Ar), 7.64 (s, 1H, thiazole-H), 12.50 (s, 1H, NH, D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 44.1 (3,5-piperazine- CH_2), 55.7 ($m\text{-OCH}_3$), 56.1 (2,6-piperazine- CH_2), 60.9 ($p\text{-OCH}_3$), 63.0 (COCH_2), 101.3 (2,6-Ar-C), 106.1 (thiazole-C-5), 128.5 (1-Ar-C), 139.7 (4-Ar-C), 149.9 (thiazole-C-4), 153.6 (3,5-Ar-C), 163.7 (thiazole-C-2), 169.1 (C=O), MS (m/z , Rel. Abundance): 392.97 (M^+ , 42.88), 392 (100), Anal. Calcd for $\text{C}_{18}\text{H}_{24}\text{N}_4\text{O}_4\text{S}$ calculated %: C, 55.09; H, 6.16; N, 14.28; S, 8.17. Found %: C, 55.07; H, 6.12; N, 14.31; S, 8.20.

4.1.3.2. 2-(4-Methylpiperazin-1-yl)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (6b). Brown crystals, mp 135–138 °C, yield (66%), IR (KBr, ν , cm^{-1}): 3451 (N–H amide), 1682 (C=O amide), 1658 (C=N), 1588 (C=C), 1455 (N– CH_3), 1236 (C–O), ^1H NMR (400 MHz, $\text{DMSO-}d_6$): δ 2.15 (s, 3H, NCH_3), 2.33 (m, 8H, piperazine- CH_2), 3.29 (s, 2H, COCH_2), 3.69 (s, 3H, $p\text{-OCH}_3$), 3.84 (s, 6H, $m\text{-OCH}_3$), 7.21 (s, 2H, H-2 and H-6 Ar), 7.64 (s, 1H, thiazole-H), 12.50 (s, 1H, NH, D_2O exchangeable), ^{13}C NMR (100 MHz, $\text{DMSO-}d_6$): δ 46.2 (N– CH_3), 52.9 (2,6-piperazine- CH_2), 55.1 ($m\text{-OCH}_3$), 56.4 (3,5-piperazine- CH_2), 60.5 ($p\text{-OCH}_3$), 60.7 ($\text{CH}_2\text{-N}$), 103.5 (2,6-Ar-C), 108.1 (thiazole-C-5), 130.5 (1-Ar-C), 137.7 (4-Ar-C), 149.2 (thiazole-C-4), 153.3 (3,5-Ar-C), 158.1 (thiazole-C-2), 169.3 (C=O), MS (m/z , Rel. Abundance): 406.02 (M^+ , 43), 407.25 ($\text{M}^+ + 1$, 28.64), 336.1 (100), Anal. Calcd for $\text{C}_{19}\text{H}_{26}\text{N}_4\text{O}_4\text{S}$ calculated %: C, 56.14; H, 6.45; N, 13.78; S, 7.89. Found %: C, 56.11; H, 6.48; N, 13.75; S, 7.85.

4.1.3.3. 2-(4-Ethylpiperazin-1-yl)-*N*-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (6c). Black crystals, mp 133–136 °C, yield (62%), IR (KBr, ν , cm^{-1}): 3446 (N–H amide), 1653 (C=O amide), 1591 (C=C), 1239 (C–O), ^1H NMR (400 MHz, CDCl_3): δ 1.25 (t, $J = 14.50$ Hz, 3H,

CH₂CH₃), 2.43 (q, 2H, CH₂CH₃), 2.54 (t, *J* = 9.50 Hz, 4H, piperazine-CH₂), 2.69 (t, *J* = 8.00 Hz, 4H, piperazine-CH₂), 3.32 (s, 2H, COCH₂), 3.80 (s, 3H, *p*-OCH₃), 3.95 (s, 6H, *m*-OCH₃), 7.04 (s, 2H, H-2 and H-6 Ar), 8.00 (s, 1H, thiazole-H), 12.50 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 15.4 (N-CH₂-CH₃), 49.1 (N-CH₂CH₃), 54.3 (2,6-piperazine-CH₂), 55.7 (*m*-OCH₃), 56.9 (3,5-piperazine-CH₂), 60.3 (*p*-OCH₃), 62.5 (COCH₂), 101.3 (2,6-Ar-C), 106.1 (thiazole-C-5), 128.5 (1-Ar-C), 138.7 (4-Ar-C), 149.6 (thiazole-C-4), 152.4 (3,5-Ar-C), 160.5 (thiazole-C-2), 168.1 (C=O), MS (*m/z*, Rel. Abundance): 420.07 (M⁺, 100), Anal. Calcd for C₂₀H₂₈N₄O₄S calculated %: C, 57.12; H, 6.71; N, 13.32; S, 7.62. Found %: C, 57.15; H, 6.75; N, 13.31; S, 7.64.

4.1.3.4. 2-(4-(4-Fluorophenyl)piperazin-1-yl)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**6d**). Dark yellow crystals, mp 97–100 °C, yield: (60%), IR (KBr, *ν*, cm⁻¹): 3386 (N-H amide), 1679 (C=O amide), 1590 (C=C), 1230 (C-O), 1126 (C-F), ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.37 (m, 4H, piperazine-CH₂), 2.54 (m, 4H, piperazine-CH₂), 3.30 (s, 2H, COCH₂), 3.70 (s, 3H, *p*-OCH₃), 3.85 (s, 6H, *m*-OCH₃), 7.09 (d, 2H, *J* = 16.00 Hz, fluorophenyl H-2, H-6), 7.22 (s, 2H, H-2 and H-6 Ar), 7.28 (d, 2H, *J* = 12.00 Hz, fluorophenyl H-3, H-5), 7.65 (s, 1H, thiazole-H), 12.06 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 52.9 (3,5-piperazine-CH₂), 53.0 (*m*-OCH₃), 56.4 (2,6-piperazine-CH₂), 60.5 (*p*-OCH₃), 61.6 (COCH₂-N), 103.6 (2,6-Ar-C), 108.1 (thiazole-C-5), 115.2 (2,6-fluoro-phenyl c), 115.4 (3,5-fluoro-phenyl c), 130.5 (1-Ar-C), 134.8 (thiazole-C-4), 137.8 (4-Ar-C), 149.3 (1-fluoro-phenyl c), 153.6 (3,5-Ar-C), 157.7 (4-fluoro-phenyl c), 160.5 (thiazole-C-2), 162.9 (C=O), MS (*m/z*, Rel. Abundance): 486.42 (M⁺, 34.40), 487.06 (M + 1, 6.53), 109.16 (100), Anal. Calcd for C₂₄H₂₇FN₄O₄S calculated %: C, 59.25; H, 5.59; N, 11.52; S, 6.59. Found %: C, 59.28; H, 5.56; N, 11.50; S, 6.61.

4.1.3.5. 2-Morpholino-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**6e**). Black crystals, mp 102–104 °C, yield: (65%), IR (KBr, *ν*, cm⁻¹): 3450 (N-H amide), 1637 (C=O amide), 1530 (C=C), 1235 (C-O), ¹H NMR (400 MHz, CDCl₃): δ 3.20 (s, 2H, COCH₂), 3.40 (t, 4H, *J* = 10.00 Hz, morpholine-CH₂), 3.65 (t, 4H, *J* = 10.50 Hz, morpholine-CH₂), 3.72 (s, 3H, *p*-OCH₃), 3.86 (s, 6H, *m*-OCH₃), 7.05 (s, 2H, H-2 and H-6 Ar), 7.08 (s, 1H, thiazole-H), 12.40 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 55.6 (3,5-morpholine-CH₂), 56.4 (*m*-OCH₃), 61.3 (*p*-OCH₃), 62.6 (COCH₂), 64.2 (2,6-morpholine-CH₂), 103.5 (2,6-Ar-C), 107.1 (thiazole-C-5), 130.5 (1-Ar-C), 137.7 (4-Ar-C), 149.2 (thiazole-C-4), 153.6 (3,5-Ar-C), 158.1 (thiazole-C-2), 169.3 (C=O), MS (*m/z*, Rel. Abundance): 394.17 (M⁺, 41.68), 393.33 (100), Anal. Calcd for C₁₈H₂₃N₃O₅S calculated %: C, 54.95; H, 5.89; N, 10.68; S, 8.15. Found %: C, 54.97; H, 5.87; N, 10.64; S, 8.11.

4.1.4. General Procedure for the Synthesis of 2-Substituted N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide Derivatives (**7a–e**). A mixture of compound **4** (1 g, 0.003 mol) and substituted anilines (0.003 mol) in absolute ethanol (10 mL) and few drops of HCl (5–6 drops) was refluxed overnight. The reaction progress was monitored by TLC after the completion of the reaction. The reaction mixture was poured into ice-cold water, and the resultant precipitate was filtered, dried, and recrystallized from methanol.

4.1.4.1. 2-(Phenylamino)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**7a**). Black crystals, mp 165–168 °C, yield (65%), IR (KBr, *ν*, cm⁻¹): 3340 (N-H amide), 3091 (2°

amine NH), 1629 (C=O amide), 1583 (C=C), 1251 (C-O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.70 (s, 3H, *p*-OCH₃), 3.77 (s, 2H, COCH₂), 3.86 (s, 6H, *m*-OCH₃), 6.70 (s, 1H, aniline-NH, D₂O exchangeable), 7.18 (s, 2H, H-2 and H-6 Ar), 7.35 (s, 1H, thiazole-H), 7.41–7.50 (m, 5H, aniline-H), 10.70 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 54.1 (CH₂-N), 56.6 (*m*-OCH₃), 60.2 (*p*-OCH₃), 100.3 (2,6-Ar-C), 105.8 (thiazole-C-5), 114.6 (aniline-C-2,6), 119.5 (aniline-C-4), 127.8 (1-Ar-C), 130.0 (aniline-C-3,5), 140.2 (4-Ar-C), 147.1 (aniline-C-1), 150.5 (thiazole-C-4), 153.6 (3,5-Ar-C), 164.8 (thiazole-C-2), 168.7 (C=O), MS (*m/z*, Rel. Abundance): 399.14 (M⁺, 5.83), 109.02 (100), Anal. Calcd for C₂₀H₂₁N₃O₄S calculated %: C, 60.14; H, 5.30; N, 10.52; S, 8.03. Found %: C, 60.18; H, 5.32; N, 10.56; S, 8.01.

4.1.4.2. 2-((4-Bromophenyl)amino)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**7b**). Brown crystals, mp 147–150 °C, yield (64%), IR (KBr, *ν*, cm⁻¹): 3340 (N-H amide), 3192 (2° amine NH), 1630 (C=O amide), 1588 (C=N), 1560 (C=C), 1251 (C-O), 657 (C-Br), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.70 (s, 3H, *p*-OCH₃), 3.78 (s, 2H, COCH₂), 3.86 (s, 6H, *m*-OCH₃), 6.70 (s, 1H, aniline-NH, D₂O exchangeable), 7.17 (s, 2H, H-2, H-6 Ar), 7.32 (s, 1H, thiazole-H), 7.34 (d, *J* = 4.00 Hz, 2H, aniline-H), 7.65 (d, *J* = 8.00 Hz, 2H, aniline-H), 9.07 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 55.6 (CH₂-N), 56.4 (*m*-OCH₃), 60.0 (*p*-OCH₃), 100.5 (2,6-Ar-C), 105.0 (thiazole-C-5), 114.1 (aniline-C-2,6), 115.9 (aniline-C-4), 127.5 (1-Ar-C), 133.1 (aniline-C-3,5), 139.7 (4-Ar-C), 146.9 (aniline-C-1), 150.4 (thiazole-C-4), 153.8 (3,5-Ar-C), 164.5 (thiazole-C-2), 168.4 (C=O), MS (*m/z*, Rel. Abundance): 478.86 (M⁺, 32.51), 479.55 (M + 1, 18.52), 480.75 (M + 2, 30.58), 267.55 (100), Anal. Calcd for C₂₀H₂₀BrN₃O₄S calculated %: C, 50.22; H, 4.21; N, 8.78; S, 6.70. Found %: C, 50.25; H, 4.24; N, 8.74; S, 6.72.

4.1.4.3. 2-((4-Nitrophenyl)amino)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**7c**). Black crystals, mp 100–103 °C, yield (60%), IR (KBr, *ν*, cm⁻¹): 3479 (N-H amide), 3362 (2° amine NH), 1631 (C=O amide), 1595 (C=C), 1501 (N-O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.71 (s, 3H, *p*-OCH₃), 3.78 (s, 2H, COCH₂), 3.85 (s, 6H, *m*-OCH₃), 6.59 (t, *J* = 13.60 Hz, 2H, aniline-H), 6.90 (s, 1H, aniline-NH, D₂O exchangeable), 7.10 (s, 2H, H-2, H-6 Ar), 7.40 (s, 1H, thiazole-H), 7.93 (t, *J* = 14.00 Hz, 2H, aniline-H), 12.50 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 55.4 (CH₂-N), 56.3 (*m*-OCH₃), 60.7 (*p*-OCH₃), 100.5 (2,6-Ar-C), 105.3 (thiazole-C-4), 115.6 (aniline-C-2,6), 127.3 (1-Ar-C), 128.2 (aniline-C-3,5), 136.4 (aniline-C-4), 140.3 (4-Ar-C), 151.2 (thiazole-C-5), 153.0 (3,5-Ar-C), 153.8 (aniline-C-1), 164.8 (thiazole-C-2), 168.7 (C=O), MS (*m/z*, Rel. Abundance): 444.02 (M⁺, 7.47), 52.08 (100), Anal. Calcd for C₂₀H₂₀N₄O₆S calculated %: C, 54.05; H, 4.54; N, 12.61; S, 7.21. Found %: C, 54.02; H, 4.58; N, 12.63; S, 7.23.

4.1.4.4. 2-((4-Bromo-3-(trifluoromethyl)phenyl)amino)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (**7d**). Black crystals, mp 103–105 °C, yield (57%), IR (KBr, *ν*, cm⁻¹): 3336 (N-H amide), 3084 (2° amine NH), 1628 (C=O amide), 1560 (C=C), 1257 (C-O), 1126 (C-F), 662 (C-Br), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.70 (s, 2H, COCH₂), 3.78 (s, 3H, *p*-OCH₃), 3.86 (s, 6H, *m*-OCH₃), 6.74 (s, 1H, aniline-NH, D₂O exchangeable), 6.92 (d, *J* = 6.00 Hz, 2H, aniline-H), 7.21 (s, 2H, H-2, H-6 Ar), 7.50 (s, 1H, thiazole-H), 7.54 (d, *J* = 7.60 Hz, 1H, aniline-H), 12.50 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 56.3 (CH₂-N), 57.3 (*m*-OCH₃), 61.2 (*p*-OCH₃), 100.8 (2,6-Ar-C), 105.3

(thiazole-C-4), 109.4 (aniline-C-4), 112.1 (aniline-C-2), 118.1 (aniline-C-6), 123.5 (CF₃), 127.8 (1-Ar-C), 132.0 (aniline-C-3), 133.1 (aniline-C-5), 139.6 (4-Ar-C), 147.1 (aniline-C-1), 150.7 (thiazole-C-5), 154.0 (3,5-Ar-C), 164.3 (thiazole-C-2), 168.7 (C=O), MS (*m/z*, Rel. Abundance): 546.33 (M⁺, 41.60), 548.01 (M + 2, 37.77), 91.88 (100), Anal. Calcd for C₂₁H₁₉BrF₃N₃O₄S calculated %: C, 46.17; H, 3.51; N, 7.69; S, 5.87. Found %: C, 46.20; H, 3.53; N, 7.68; S, 5.90.

4.1.4.5. 2-((4-Morpholinophenyl)amino)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (7e). Gray crystals, mp 239–242 °C, yield (60%), IR (KBr, ν , cm⁻¹): 3453 (N–H), 3362 (2° amine NH), 1672 (C=O amide), 1655 (C=N), 1585 (C=C), 1234 (C–O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.2 (m, 4H, morpholine-H), 3.70 (s, 3H, *p*-OCH₃), 3.74 (m, 4H, morpholine-H), 3.75 (s, 2H, COCH₂), 3.86 (s, 6H, *m*-OCH₃), 5.70 (s, 1H, aniline NH, D₂O exchangeable), 6.50 (m, 2H, aniline-H), 6.80 (m, 2H, aniline-H), 7.15 (s, 2H, H-2, H-6 Ar), 7.28 (s, 1H, thiazole-H), 12.50 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 53.7 (morpholine-C-3,5), 55.5 (CH₂–N), 56.7 (*m*-OCH₃), 60.4 (*p*-OCH₃), 67.0 (morpholine-C-2,6), 100.5 (2,6-Ar-C), 105.1 (thiazole-C-5), 117.5 (aniline-C-2,6), 127.5 (1-Ar-C), 129.1 (aniline-C-3,5), 137.6 (aniline-C-1), 138.8 (aniline-C-4), 139.5 (4-Ar-C), 150.3 (thiazole-C-4), 153.5 (3,5-Ar-C), 164.6 (thiazole-C-2), 168.4 (C=O), MS (*m/z*, Rel. Abundance): 484.41 (M⁺, 53.45), 90.97 (100).

4.1.5. Procedure for the Synthesis of 2-Hydrazinyl-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (8). Hydrazine hydrate 64–65% (1.5 mL, 0.03 mol) was added to a stirred solution of compound 4 (1 g, 0.003 mol) in absolute ethanol (10 mL). The reaction mixture was refluxed for 6 h. The reaction mixture was left to cool, and the precipitate was collected by filtration and recrystallized from methanol.

Brown crystals, mp 135–138 °C, yield (75%), IR (KBr, ν , cm⁻¹): 3403 (N–H amide), 3188 (2° amine NH), 3109 (1° amine NH), 1635 (C=O amide), 1590 (C=C), 1234 (C–O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.40 (s, 2H, NH₂, D₂O exchangeable), 3.67 (s, 2H, COCH₂), 3.69 (s, 3H, *p*-OCH₃), 3.84 (s, 6H, *m*-OCH₃), 4.20 (s, 1H, NH, D₂O exchangeable), 7.19 (s, 2H, H-2 and H-6 Ar), 7.62 (s, 1H, thiazole-H), 12.30 (s, 1H, NHCO D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 56.3 (*m*-OCH₃), 60.5 (*p*-OCH₃), 60.6 (COCH₂–NH), 101.7 (thiazole-C-5), 103.5 (2,6-Ar-C), 130.5 (1-Ar-C), 137.4 (4-Ar-C), 149.1 (thiazole-C-4), 153.4 (3,5-Ar-C), 168.5 (thiazole-C-2), 169.2 (C=O), MS (*m/z*, Rel. Abundance): 338.35 (M⁺, 50.13), 308.27, (100), Anal. Calcd for C₁₄H₁₈N₄O₄ calculated %: C, 49.69; H, 5.36; N, 16.56; S, 9.47. Found %: C, 49.66; H, 5.39; N, 16.59; S, 9.44.

4.1.6. Procedure for the Synthesis of 2-(2-Substituted hydrazinyl)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide Derivatives (9a–b). A mixture of hydrazide 8 (0.44 g, 0.002 mol) and isatin derivative (0.002 mol) was refluxed in absolute ethanol (10 mL) in the presence of few drops of glacial acetic acid (5–6 drops) for 24 h. Progress of the reaction was monitored through TLC. After the completion of the reaction, the solvent was evaporated to afford the target product. The product was filtered and recrystallized from methanol.

4.1.6.1. (Z)-2-(2-(5-Chloro-2-oxoindolin-3-ylidene)-hydrazinyl)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (9a). Orange crystals, mp 170–173 °C, yield (60%), IR (KBr, ν , cm⁻¹): 3400 (N–H amide), 3280 (N–H isatin), 3239 (N–H hydrazine), 1686 (C=O amide), 1587 (C=C), 1237 (C–O), 812 (C–Cl), ¹H NMR (400 MHz, DMSO-*d*₆): δ

3.50 (s, 2H, COCH₂), 3.70 (s, 3H, *p*-OCH₃), 3.84 (s, 6H, *m*-OCH₃), 7.19 (s, 2H, H-2 and H-6 Ar), 7.30 (m, 1H, NH–N, D₂O exchangeable), 7.42 (s, 1H, isatin-H), 7.68 (m, 1H, isatin-H) 7.80 (s, 1H, thiazole-H), 7.90 (m, 1H, isatin-H), 10.80 (s, 1H, isatin NH, D₂O exchangeable), 12.50 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 53.5 (COCH₂–NH), 56.4 (*m*-OCH₃), 60.6 (*p*-OCH₃), 103.5 (2,6-Ar-C), 111.9 (thiazole-C-5), 117.4 (3a-isatin C) 123.8 (7-isatin C), 126.8 (4-isatin C), 127.2 (1-Ar-C), 127.5 (3-isatin C), 129.2 (5-isatin C), 132.1 (6-isatin C), 137.6 (4-Ar-C), 137.9 (2-isatin C), 138.2 (thiazole-C-4), 153.3 (3,5-Ar-C), 153.6 (7a-isatin C), 163.1 (thiazole-C-2), 168.6 (C=O), MS (*m/z*, Rel. Abundance): 501.53 (M⁺, 14.29), 503.59 (M + 1, 6.05) 72.08 (100), Anal. Calcd for C₂₂H₂₀ClN₅O₅S calculated %: C, 52.64; H, 4.02; N, 13.95; S, 6.39. Found %: C, 52.68; H, 4.06; N, 13.93; S, 6.35.

4.1.6.2. (Z)-2-(2-(5-Methyl-2-oxoindolin-3-ylidene)-hydrazinyl)-N-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)acetamide (9b). Orange crystals, mp 139–142 °C, yield (57%), IR (KBr, ν , cm⁻¹): 3392 (N–H amide), 3337 (N–H hydrazine), 3235 (N–H isatin), 1687 (C=O amide), 1626 (C=N), 1591 (C=C), 1450 (CH₃), 1241 (C–O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.27 (s, 3H, CH₃), 3.50 (s, 2H, COCH₂), 3.69 (s, 3H, *p*-OCH₃), 3.85 (s, 6H, *m*-OCH₃), 6.50 (d, 1H, isatin H-7), 7.18 (s, 2H, H-2 and H-6 Ar), 7.20 (s, 1H, NH=N, D₂O exchangeable), 7.30 (m, 1H, isatin H-6), 7.40 (s, 1H, isatin H-4), 7.80 (s, 1H, thiazole-H), 10.60 (s, 1H, isatin NH, D₂O exchangeable), 12.50 (s, 1H, NHCO, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 22.0 (CH₃ isatin), 53.6 (COCH₂–NH), 56.3 (*m*-OCH₃), 60.4 (*p*-OCH₃), 102.9 (2,6-Ar-C), 108.9 (thiazole-C-5), 117.5 (3a-isatin C), 122.6 (7-isatin C), 127.1 (1-Ar-C), 129.2 (4-isatin C), 130.4 (6-isatin C), 133.1 (3-isatin C), 135.5 (5-isatin C), 136.8 (7a-isatin C), 137.9 (2-isatin C), 138.5 (4-Ar-C), 145.2 (thiazole-C-4), 153.3 (3,5-Ar-C), 163.7 (thiazole-C-2), 168.6 (C=O), MS (*m/z*, Rel. Abundance): 481.74 (M⁺, 11.24), 485.6 (M + 1, 6.28), 118.23 (100), Anal. Calcd for C₂₃H₂₃N₅O₅S calculated %: C, 57.37; H, 4.81; N, 14.54; S, 6.66. Found %: C, 57.33; H, 4.84; N, 14.57; S, 6.64.

4.1.7. Procedure for the Synthesis of N⁴-(4-(3,4,5-Trimethoxyphenyl)thiazol-2-yl)thiazole-2,4-diamine (10). A mixture of compound 4 (0.34 g, 0.001 mol) and thiourea (0.08 g, 0.001 mol) and anhydrous K₂CO₃ (0.14 g, 0.001 mol) in acetone (10 mL) was refluxed overnight. After the completion of the reaction, the solvent was evaporated to afford the target product. The product was filtered and recrystallized from methanol.

Brown crystals, mp 180–183 °C, yield (66%), IR (KBr, ν , cm⁻¹): 3421 (2° amine N–H), 3188 (1° amine NH₂), 2160 (N=C=S), 1589 (C=C), 1125 (C–O), ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.69 (s, 3H, *p*-OCH₃), 3.84 (s, 6H, *m*-OCH₃), 5.30 (s, 1H, thiazole'-H), 7.05 (s, 2H, H-2 and H-6 Ar), 7.20 (s, 2H, NH₂, D₂O exchangeable), 7.60 (s, 1H, thiazole-H), 9.00 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (100 MHz, DMSO-*d*₆): δ 56.3 (*m*-OCH₃), 60.5 (*p*-OCH₃), 103.5 (2,6-Ar-C), 108.4 (thiazole-C-4), 110.5 (thiazole'-C-4), 130.4 (1-Ar-C), 137.8 (4-Ar-C), 142.1 (thiazole'-C-5), 149.3 (thiazole-C-5), 153.6 (3,5-Ar-C), 158.0 (thiazole-C-2), 168.4 (thiazole'-C-2), MS (*m/z*, Rel. Abundance): 364 (M⁺, 33.12), 149 (100), Anal. Calcd for C₁₅H₁₆N₄O₃S₂ calculated %: C, 49.44; H, 4.43; N, 15.37; S, 17.59. Found %: C, 49.47; H, 4.40; N, 15.39; S, 17.62.

4.1.8. Procedure for the Synthesis of 1-(2-Chloroethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea (11). A mixture of compound 3 (0.5 g, 0.002 mol), 2-chloroethyl isocyanate (0.3

mL, 0.002 mol), and anhydrous K_2CO_3 (0.3 g, 0.002 mol) was heated under reflux in dry DMF (10 mL) for 12 h. The reaction mixture was left to cool, concentrated under reduced pressure, and poured onto crushed ice. The separated precipitate was filtered and dried.

Brown crystals, mp 205–208 °C, yield (52%), IR (KBr, ν , cm^{-1}): 3487, 3328 (NH amide), 1657 (C=O amide), 1589 (C=C), 1238 (C–O), 738 (C–Cl), 1H NMR (400 MHz, $(CD_3)_2CO$): δ 3.65 (t, $J = 10.00$ Hz, 2H, $NHCH_2$), 3.70 (t, $J = 10.00$ Hz, 2H, CH_2Cl), 3.73 (s, 3H, $p-OCH_3$), 3.87 (s, 6H, $m-OCH_3$), 7.28 (s, 2H, H-2 and H-6 Ar), 7.52 (s, 1H, thiazole-H), 8.20 (t, $J = 15.00$ Hz, 1H, $NHCH_2$, D_2O exchangeable), 9.60 (s, 1H, $NHCO$, D_2O exchangeable), ^{13}C NMR (100 MHz, DMSO- d_6): δ 42.3 ($NHCH_2CH_2$), 42.9 (CH_2Cl), 56.5 ($m-OCH_3$), 60.9 ($p-OCH_3$), 100.5 (2,6-Ar-C), 105.3 (thiazole-C-4), 127.9 (1-Ar-C), 139.1 (4-Ar-C), 149.9 (thiazole-C-5), 153.6 (3,5-Ar-C), 154.5 (C=O), 164.4 (thiazole-C-2), MS (m/z , Rel. Abundance): 371.96 (M^+ , 18.84), 373.73 ($M + 1$, 11.96), 169.22 (100), Anal. Calcd for $C_{15}H_{18}ClN_3O_4S$ calculated %: C, 48.45; H, 4.88; N, 11.30; S, 8.62. Found %: C, 48.47; H, 4.86; N, 11.34; S, 8.59.

4.1.9. Procedure for the Synthesis of 1-(2-(Phenylamino)ethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea (12). A mixture of compound 11 (0.5 g, 0.001 mol) and aniline (0.1 mL, 0.001 mol) in absolute ethanol (10 mL) and few drops of HCl (5–6 drops) was heated under reflux overnight. The reaction progress was monitored by TLC; after the completion of the reaction, the reaction mixture was poured into ice-cold water, and the precipitate was filtered, dried, and recrystallized from methanol.

Brown crystals, mp 278–281 °C, yield: (70%), IR (KBr, ν , cm^{-1}): 3451, 3380 (N–H amide), 3250 (2° amine N–H), 1682 (C=O amide), 1658 (C=N), 1588 (C=C), 1236 (C–O), 1H NMR (400 MHz, DMSO- d_6): δ 3.50 (t, 2H, $J = 16.80$ Hz, CH_2N), 3.69 (s, 3H, $p-OCH_3$), 3.70 (t, 2H, $J = 17.60$ Hz, $CONHCH_2$), 3.84 (s, 6H, $m-OCH_3$), 4.07 (t, 1H, $J = 15.60$ Hz, $CONHCH_2$, D_2O exchangeable), 4.14 (t, 1H, $J = 16.00$ Hz, CH_2NH , D_2O exchangeable), 7.15 (m, 3H, aniline H-2, H-4, H-6), 7.21 (s, 2H, H-2 and H-6 Ar), 7.55 (s, 2H, aniline H-3, H-5), 7.60 (s, 1H, thiazole-H), 9.80 (s, 1H, $NHCO$, D_2O exchangeable), ^{13}C NMR (100 MHz, DMSO- d_6): δ 41.8 (ethyl-C-1), 43.5 (ethyl-C-2), 56.7 ($m-OCH_3$), 61.2 ($p-OCH_3$), 101.2 (2,6-Ar-C), 105.2 (thiazole-C-4), 114.3 (aniline-C-2,6), 121.3 (aniline-C-4), 127.8 (1-Ar-C), 130.0 (aniline-C-3,5), 139.7 (4-Ar-C), 148.0 (aniline-C-1), 150.7 (thiazole-C-5), 153.8 (3,5-Ar-C), 154.7 (C=O), 164.8 (thiazole-C-2), MS (m/z , Rel. Abundance): 428.39 (M^+ , 13.94), 343.66 (100), Anal. Calcd for $C_{21}H_{24}N_4O_4S$ calculated %: C, 58.86; H, 5.65; N, 13.08; S, 7.48. Found %: C, 58.83; H, 5.67; N, 13.06; S, 7.45.

4.1.10. General Procedure for the Synthesis of 1-Substituted-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea (13a–b). A mixture of compound 11 (0.5 g, 0.001 mol) and appropriate secondary amines (0.001 mol), *N*-ethyl piperazine (0.001 mol), and morpholine (0.001 mol) in DMF (10 mL) containing few drops of triethylamine was refluxed for 8 h. After the completion of the reaction, the reaction mixture was cooled and poured onto ice-cold water, and the formed precipitate was filtered, dried, and recrystallized from ethanol.

4.1.10.1. 1-(2-(4-Ethylpiperazin-1-yl)ethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea (13a). Black crystals, mp 217–220 °C, yield (58%), IR (KBr, ν , cm^{-1}): 3421, 3255 (N–H amide), 1657 (C=O amide), 1586 (C=C), 1408 (CH_3), 1235 (C–O), 1H NMR (400 MHz, DMSO- d_6): δ 1.01 (t, 3H, $J =$

14.00 Hz, CH_2CH_3), 2.28 (m, 8H, piperazine-Hs), 3.20 (q, 2H, CH_2CH_3), 3.50 (t, 2H, $J = 16.00$ Hz, CH_2N), 3.68 (s, 3H, $p-OCH_3$), 3.84 (s, 6H, $m-OCH_3$), 4.13 (t, 2H, $J = 16.00$ Hz, $NHCH_2$), 7.21 (s, 2H, H-2 and H-6 Ar), 7.56 (s, 1H, thiazole-H), 8.30 (s, 1H, $NHCH_2$, D_2O exchangeable), 9.70 (s, 1H, $NHCO$, D_2O exchangeable), ^{13}C NMR (100 MHz, DMSO- d_6): δ 12.4 (CH_2CH_3), 37.8 ($NHCH_2CH_2$), 45.0 ($m-OCH_3$), 52.1 ($NHCH_2CH_2$), 53.4 (CH_2CH_3), 56.4 (piperazine-C), 60.6 ($p-OCH_3$), 103.7 (2,6-Ar-C), 107.3 (thiazole-C-5), 130.6 (1-Ar-C), 137.7 (4-Ar-C), 149.0 (thiazole-C-4), 153.5 (3,5-Ar-C), 157.6 (C=O), 158.7 (thiazole-C-2), MS (m/z , Rel. Abundance): 449.86 (M^+ , 13.79), 333.93 (100), Anal. Calcd for $C_{21}H_{31}N_5O_4S$ calculated %: C, 56.10; H, 6.95; N, 15.58; S, 7.13. Found %: C, 56.13; H, 6.92; N, 15.61; S, 7.09.

4.1.10.2. 1-(2-Morpholinoethyl)-3-(4-(3,4,5-trimethoxyphenyl)thiazol-2-yl)urea (13b). Brown crystals, mp 236–239 °C, yield (70%), IR (KBr, ν , cm^{-1}): 3420, 3258 (N–H amide), 1655 (C=O amide), 1589 (C=C), 1235 (C–O), 1000 (C–O morpholine), 1H NMR (400 MHz, DMSO- d_6): δ 3.20 (m, 4H, morpholine-H), 3.30 (m, 2H, CH_2N), 3.50 (m, 4H, morpholine-H), 3.60 (t, 2H, $J = 16.00$ Hz, $NHCH_2$), 3.69 (s, 3H, $p-OCH_3$), 3.84 (s, 6H, $m-OCH_3$), 7.21 (s, 2H, H-2 and H-6 Ar), 7.56 (s, 1H, thiazole-H), 8.50 (s, 1H, $NHCH_2$, D_2O exchangeable), 9.70 (s, 1H, $NHCO$, D_2O exchangeable), ^{13}C NMR (100 MHz, DMSO- d_6): δ 53.4 ($NHCH_2CH_2$), 56.4 ($NHCH_2CH_2$), 58.6 ($m-OCH_3$), 60.5 ($p-OCH_3$), 60.9 (3,5-morpholine-C), 66.0 (2,6-morpholine-C), 103.7 (2,6-Ar-C), 107.3 (thiazole-C-5), 130.6 (1-Ar-C), 137.7 (4-Ar-C), 149 (thiazole-C-4), 153.5 (3,5-Ar-C), 158.3 (C=O), 158.7 (thiazole-C-2), MS (m/z , Rel. Abundance): 422.53 (M^+ , 31.40), 125.73 (100), Anal. Calcd for $C_{19}H_{26}N_4O_5S$ calculated %: C, 54.01; H, 6.20; N, 13.26; O, 18.93; S, 7.59. Found %: C, 54.05; H, 6.23; N, 13.22; S, 7.61.

4.2. Biological Screening. **4.2.1. Materials.** The cell lines and the reference drug CA-4 were obtained from ATCC via the Holding Company for Biological Products and Vaccines (VACSERA), Cairo, Egypt. Fetal bovine serum was obtained from GIBCO, UK, and the other reagents such as RPMI-1640 medium, MTT, and dimethyl sulfoxide (DMSO) were obtained from Sigma Co., St. Louis, USA.

4.2.2. In Vitro Antitumor Evaluation (MTT Assay). All the synthesized compounds (4–13) were evaluated for their inhibition of the growth of a panel of four different human cancer cell lines, HEPG-2, HCT-116, MCF-7, and HeLa in comparison with CA-4 as positive controls using the MTT assay.^{32,33} This colorimetric assay is based on the conversion of the yellow tetrazolium bromide (MTT) to a purple formazan derivative by mitochondrial succinate dehydrogenase in viable cells. Cell lines were cultured in RPMI-1640 medium with 10% fetal bovine serum. Antibiotics added were 100 units/mL penicillin and 100 μ g/mL streptomycin at 37 °C in a 5% CO_2 incubator. The cell lines were seeded in a 96-well plate at a density of 1.0×10^4 cells/well at 37 °C for 48 h under 5% CO_2 . After incubation, the cells were treated with different concentrations of compounds and incubated for 24 h. After 24 h of drug treatment, 20 μ L of MTT solution at 5 mg/mL was added and incubated for 4 h. DMSO in a volume of 100 μ L is added into each well to dissolve the purple formazan formed. Colorimetric assay is performed and recorded at an absorbance of 570 nm using a plate reader (EXL 800, USA). The cytotoxic activity was expressed as the concentration of the compound that caused 50% growth inhibition (IC_{50} , mean \pm SD).

4.2.3. *In Vitro* Tubulin Polymerization Inhibition Assay. To further characterize the interaction of the proposed compounds with the microtubule system, the selected 10 compounds including **5c**, **6a,d,e**, **7b,c**, **8**, **9a,b**, and **13a** were evaluated for their *in vitro* inhibition of tubulin polymerization, with CA-4 as the reference drug.

Standard polymerization reactions (minus tubulin ligands) were carried out as described in the polymerization protocol. Briefly, the standard polymerization reaction contains 100 μL volume of 4 mg/mL tubulin in 80 mM PIPES, pH 6.9, 0.5 mM EGTA, 2 mM MgCl_2 , and 1 mM GTP. Polymerization was started by incubation at 37 $^\circ\text{C}$, followed by absorption readings at 340 nm. Under these conditions, polymerization will reach a maximal OD_{340} between 0.15 and 0.25 within 30 min. The three phases of polymerization are shown: I (nucleation), II (growth), and III (steady state). In this experimental setup (100 μL volume in a spectrophotometer with a 0.5 cm pathlength), OD_{340} of 0.1 is approximately equal to 1 mg/mL of polymer mass. Thus, under the conditions described, approximately 40% of tubulin is polymerized, leaving the flexibility for detecting the enhancers and inhibitors of polymerization. The tubulin polymerization inhibition activity for each compound was expressed as IC_{50} values, and the data were represented as mean \pm SD from three independent experiments.

4.3. Docking Methodology. The coordinates of the target co-crystallized with colchicine were retrieved from the Protein Data Bank (PDB ID: 4O2B).⁴⁵ All hydrogen atoms were added the crystal 3D structure of the protein with their standard geometry, followed by their energy minimization. The tested compounds and CA-4 were drawn into Marvin Sketch of Marvin suite (<http://www.chemaxon.com>) to generate the lowest energy conformer. Dock module of MOE (Molecular Operating Environment) version MOE 2019.0102,2⁴⁶ on a computer having Pentium 1.6 GHz workstation, 512 MB memory using windows operating system, was utilized in docking studies. Our tested compounds were docked into the rigid binding pocket, the active site of colchicine, of the protein using flexible ligand mode. From the ligand conformations, the placement phase generates poses. The free energy of binding of the ligand from a given pose is estimated using GBVI/WSA ΔG as a force field-based scoring function.⁴⁷

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.2c05077>.

Putative binding modes (2D and 3D) of all newly synthesized compounds with tubulin (PDB ID: 4O2B) and free binding energies between the ligands of all newly synthesized compounds and tubulin (PDB ID: 4O2B) (PDF)

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Notes

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