

NMR Spectroscopy

International Edition: DOI: 10.1002/anie.201713158 German Edition: DOI: 10.1002/ange.201713158

Site-Specific Studies of Nucleosome Interactions by Solid-State NMR **Spectroscopy**

ShengOi Xiang⁺, Ulric B. le Paige⁺, Velten Horn, Klaartje Houben, Marc Baldus, and Hugo van Ingen*

Abstract: Chromatin function depends on a dense network of interactions between nucleosomes and a wide range of proteins. A detailed description of these protein-nucleosome interactions is required to reach a full molecular understanding of chromatin function in both genetics and epigenetics. Herein, we show that the structure, dynamics, and interactions of nucleosomes can be interrogated in a residue-specific manner by using state-of-the-art solid-state NMR spectroscopy. Using sedimented nucleosomes, high-resolution spectra were obtained for both flexible histone tails and the non-mobile histone core. Through co-sedimentation of a nucleosomebinding peptide, we demonstrate that protein-binding sites on the nucleosome surface can be determined. We believe that this approach holds great promise as it is generally applicable, extendable to include the structure and dynamics of the bound proteins, and scalable to interactions of proteins with higherorder chromatin structures, including isolated and cellular chromatin.

Lukaryotic genomes are present in the cell as a complex of DNA and histone proteins, called chromatin. The fundamental building block of chromatin, the nucleosome, is an

_	
[*]	Dr. S. Xiang, ^[+] Dr. K. Houben, Prof. Dr. M. Baldus
	NMR Spectroscopy Research Group
	Bijvoet Center for Biomolecular Research
	Padualaan & 3584 CH Utrecht (The Netherlands)
	E-mail: h.vaningen@uu.nl
	U. B. le Paige, ^[+] V. Horn, Dr. H. van Ingen
	Macromolecular Biochemistry
	Leiden Institute of Chemistry, Leiden University Einsteinweg 55, 2333 CC Leiden (The Netherlands)
	U. B. le Paige, ^[+] V. Horn, Dr. H. van Ingen Current address: NMR Spectroscopy Research Group Bijvoet Center for Biomolecular Research, Utrecht University Padualaan 8, 3584 CH Utrecht (The Netherlands)
	Dr. K. Houben Current address: DSM Food Specialties
	DSM Biotechnology Center
	Alexander Flemminglaan 1, 2613 AX Delft (The Netherlands)
[+]	These authors contributed equally to this work.
Ø	Supporting information and the ORCID identification number(s) the author(s) of this article can be found under:
0	https://doi.org/10.1002/anie.201713158.
0	© 2018 The Authors. Published by Wiley-VCH Verlag GmbH & KGaA. This is an open access article under the terms of the Crea Commons Attribution Non-Commercial NoDerivs License. whic

Co. ative :h permits use and distribution in any medium, provided the original work is properly cited, the use is non-commercial, and no modifications or adaptations are made.

assembly of a histone octamer protein core around which approximately 147 base pairs (bp) of DNA are wrapped^[1] (Figure 1 A). Nucleosomes compact and protect DNA and, importantly, serve as the binding platform for a multitude of proteins that regulate gene transcription, DNA repair, and replication.^[2] Recently, the structures of various nucleosomeprotein complexes have been solved, offering new and exciting insight into chromatin function.^[3] The wide range of interaction modes in terms of affinities, binding sites, and contribution of dynamics calls for a wide range of techniques to describe the molecular basis of these interactions. The combination of state-of-the-art solid-state NMR (ssNMR) spectroscopy and sedimentation has developed into an attractive method for the detailed characterization of soluble biomolecular systems.^[4] Sedimentation has been widely used to study the folding and compaction of nucleosomes and nucleosomal arrays.^[5] Jaroniec and co-workers were the first to realize the great potential of ssNMR spectroscopy in the context of nucleosomes through their study on histone tails in nucleosomal arrays.^[6] Herein, we introduce the use of nucleosome sedimentation, ultra-fast magic angle spinning (MAS), and ¹H-detected ssNMR spectroscopy to characterize the structure and dynamics of nucleosomes and their protein complexes. High-quality ssNMR spectra were obtained for both flexible and rigid parts of the nucleosome, allowing nearcomplete and residue-specific assignment. As proof-of-principle, we exploited the sensitivity of the amide backbone chemical shifts to map the binding surface of the N-terminal segment of the LANA protein on the nucleosome surface.^[7]

We reconstituted milligram amounts of mononucleosomes from recombinantly expressed core histones (H2A, H2B, H3, and H4) and a high-affinity DNA template^[8] according to established methods (see the Supporting Information for a detailed description and Figure S1).^[9] The modular nature of the nucleosome (molecular weight ca. 210 kDa) afforded selective isotope labeling of one histone protein per sample (here H2A or H3, each ca. 14 kDa), thereby alleviating signal overlap. Isotope-labeled histones were labeled uniformly with ¹⁵N/¹³C isotopes in combination with fractional deuteration.^[10] Typically, ca. 2 mg of the nucleosomes were sedimented directly into a 1.3 mm ssNMR rotor by ultracentrifugation. Proton 1D spectra established that the sediment is highly hydrated (Figure S1).

Sedimented nucleosomes with either labeled H2A or H3 give excellent 2D 1H-15N correlation spectra in terms of sensitivity, resolution, and signal dispersion (Figure 1). The spectral quality suggests that the nucleosomes in the sediment are well-folded and homogenously arranged on a microscopic scale. Spectra were recorded using either dipolar- or scalar-

© 2018 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

for





Figure 1. Nucleosome sedimentation yields high-quality protein "fingerprint" NMR spectra covering both the globular core and flexible tails of the histones. A) The nucleosome is formed by an octamer of four core histones, H2A, H2B, H3, and H4, and binds ca. 147 bp of DNA around its surface (crystal structure from PDB ID 1KX5^[15]). B) ¹H-detected 2D NH spectra of sedimented nucleosomes with either ²H/¹⁵N/¹³C-labeled H2A or H3, using either dipolar-coupling-based (top) or scalar-coupling-based (bottom) magnetization transfer. Slices along the ¹H/¹⁵N dimension in the dipolar spectra are shown to highlight typical linewidths.

coupling-based magnetization transfer to identify and characterize rigid and dynamic parts in the histones, respectively.^[11] As expected for folded nucleosomes, the dipolarcoupling-based spectra exhibit high chemical shift dispersion in line with the folded core of the histones (Figure 1B, top). The scalar-coupling-based spectra show fewer resonances, which are clustered around $\delta({}^{1}\text{H}) = 8.2 \text{ ppm}$, characteristic for dynamic and unstructured histone tails (Figure 1B, bottom). For both H2A and H3, the observed peak pattern is highly similar to that in the solution-state spectra of nucleosomes,^[12] allowing the assignment of nearly all observed resonances (Figure S2). For H3, all resonances correspond to the N-terminal tail (residues 3-35), indicating that the H3 tail of sedimented nucleosomes is unstructured and highly flexible, as in solution^[13] and in Mg²⁺-precipitated nucleosomal arrays.^[6] For H2A, part of the N-terminal (residues 3-8), and C-terminal tails (residues 117-123) are highly dynamic, as judged from their observation in the J-based ssNMR spectrum. Chemical shift changes to isolated H2A/H2B dimer or H3 peptide, where DNA is absent, suggest that the N-terminal tails of both H2A and H3 are (transiently) bound to nucleosomal DNA (Figure S2), in line with previous NMR studies.[13,14]

The high quality of the spectra, in particular for the core region of the nucleosome, enabled the per-residue characterization of histone structure, dynamics, and interactions. We performed a de novo sequential resonance assignment of H2A within the sedimented nucleosomes, aided by the assignments of H2A in the H2A–H2B dimer in solution (unpublished data).

Using 3D ¹H-detected CANH/CA(CO)NH and CONH/ CO(CA)NH experiments,^[16] we assigned 93% of the amide backbone resonances between residues 18–112, which corresponds to nearly all resonances in the dipolar-based ¹H-¹⁵N spectrum of H2A (Figure 2A,B and Figure S3). Notably, the significant resolution and sensitivity gains afforded by ¹H compared to ¹³C detection were instrumental herein (Figure S4). Based on the assigned H, N, C α , and C' chemical shifts, the H2A secondary structure propensities were analyzed using TALOS.^[17] The characteristic three-helix histone fold is clearly present, as well as the two β -strands that mediate interactions with H2B in the H2A-H2B dimer (Figure 2 C). Importantly, the C-terminal β-strand, which forms a "docking" sheet with H4 in the nucleosome crystal structure,^[18] is also observed. This region of H2A, the docking sequence (residues 97-116), binds to H3-H4 and thus stabilizes the histone octamer. In our dipolar ssNMR spectra, resonances for several stretches in this region, including parts of the 3^{10} helix observed in the crystal, are missing, suggesting increased mobility. In addition, this region has, compared to the histone-fold region, slightly decreased order parameters, as predicted from the chemical shifts (Figure 2 C).^[19]

Interestingly, residues 15-17, which are

part of the N-terminal α -helix in the crystal structure, are absent in both the dipolar and J-based spectra, and residues 18–20 exhibit low helical propensities and low predicted order parameters, suggesting that the α N helix is not a stably folded structure (Figure 2C). Comparison of the chemical shifts of H2A in sedimented nucleosomes to those of the H2A–H2B dimer in solution shows significant changes for regions that interact with either other histones or DNA (Figure S5). As an example of the sensitivity to the chemical environment afforded by ¹H detection, we note the peak doubling observed for V42 (Figure 2D). As the 601 DNA is not palindromic, this residue is hydrogen-bonded to either a TpA or GpA base pair step in the two copies of H2A in the nucleosome (Figure 2E).^[20]

We next used the assigned amide backbone resonances as sensitive reporters on nucleosome-protein interactions. We chose to study the interaction with the N-terminal domain of the viral LANA protein, for which a crystal structure of the complex with the nucleosome is available.^[7] LANA binds to the acidic patch, a negatively charged region on the nucleosome surface formed by acidic residues from H2A and H2B.^[1a,2c] A 20 residue peptide derived from the LANA N-terminus binds tightly ($K_{\rm D} = 0.16 \,\mu\text{M}$) to nucleosomes, in agreement with previous studies using longer fragments^[21] (Figure S6A). The LANA peptide (2 kDa) was added in fivefold molar excess to a dilute solution of H2A-labeled nucleosomes from the same batch as the sample used for assignment, followed by sedimentation into the rotor. Comparison of the remaining supernatant and pure LANA peptide solution by NMR spectroscopy suggested that the peptide had been successfully co-sedimented with the nucleosome (Figure S6 B). The dipolar ¹H-¹⁵N correlation spectrum of the nucleosome-LANA complex, measured under identical conditions to the spectrum of free nucleosomes, shows perturbations specifically for a subset of resonances (Figure 3 A and Figure S7). Importantly, there is no peak doubling upon binding, indicating that LANA is bound to both H2A copies in the nucleosome whereas only one copy is bound in the crystal structure.^[7a] Transfer of H2A assignments to the LANA-bound state was aided by making use of ¹³Ca chemical GDCh

Communications



Figure 2. Resonance assignment and secondary structure of H2A in sedimented nucleosomes. A) 2D NH spectrum with (tentative) assignments indicated in (gray) black. Side chain resonances in light colors. B) Representative strips illustrating the sequential backbone assignment based on 3D CANH and CA(CO)NH (red and green) or CONH and CO(CA)NH (magenta and blue) spectra. C) Secondary structure propensities (colored bars) and predicted S² values (black line) based on assigned backbone chemical shifts. Secondary structure in the nucleosome crystal structure (PDB ID 2PYO^[18]) and isolated H2A–H2B dimer (unpublished results) shown at the top. Asterisks indicate tentative assignments. D, E) Peak doubling of V42 observed in the 2D NH (left) and 3D CANH (right) spectra (D), correlated to the asymmetric environment in 601 nucleosomes (PDB UD 3LZ0^[20]). Dashed lines indicate hydrogen bonds (E).



Figure 3. Mapping of the LANA binding site on the nucleosome surface by ssNMR spectroscopy. A) Overlay of the 2D NH spectra of H2A in sedimented nucleosomes (yellow) and nucleosome–LANA complexes (blue). Residues with significant chemical shift changes labeled in bold, peak displacement indicated with arrows. B) Weighted chemical shift perturbations (CSPs) per residue. The 10% trimmed mean values (dashed line) plus one (orange line) or two (red line) standard deviations (σ) are indicated. Residues with significant peak broadening in the bound state are labeled with asterisks. C) CSPs color-coded on the structure of H2A. Residues with high CSPs co-localize with the binding site of LANA (PDB ID 1ZLA). D) NMR data driven model of the LANA–nucleosome complex, showing the best-scoring water-refined solution (green). This model corresponds to the crystal structure (magenta) within 1.5 Å backbone RMSD.

Angew. Chem. Int. Ed. 2018, 57, 4571–4575 © 2018 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim www.angewandte.org 4573

shifts from a 3D CANH spectrum. The lack of significant changes in the $C\alpha$ shifts indicated that there is little to no structural change in H2A upon binding and underscores the benefit of ¹H detection. Thus an unambiguous comparison of the ¹⁵N/¹H chemical shifts in the free and LANA-bound states was possible for the majority of the H2A residues (Figure 3B). Notably, resonances from the acidic patch residues E55, E63, E90, and E91 show clear chemical shift changes. In addition, a number of residues, with L92 in Figure 3 A given as an example, display significant peak broadening in the LANA-bound state, either reflecting increased local ¹H density or local structural or dynamic disorder as a result of binding. Displaying the observed chemical shift perturbations (CSPs) on the crystal structure reveals a striking co-localization of the changes with the interaction surface of LANA seen in the crystal (Figure 3C).^[7a] The highest CSPs are observed for the residues in the center of the acidic patch, which are in contact with the LANA arginine anchor.

We next used our ssNMR data and published mutagenesis data on the LANA–nucleosome interaction^[7a] to generate a molecular model of the complex using HADDOCK.^[22] The best scoring solution in terms of energetics and agreement with experimental data closely resembles the binding mode observed in the crystal, illustrating the potential of this approach to reveal the architecture of nucleosome–protein complexes (Figure 3D and Figure S8). We note that for this approach to work well, the structures of the two interacting molecules must be known, and the conformational changes upon binding must be limited.^[23]

In conclusion, we have reported the first high-resolution solid-state NMR spectroscopy study on a non-mobile region of the nucleosome and demonstrated the accurate mapping of binding sites on the nucleosome surface, which was the main goal of this study. High-quality spectra of histone proteins in sedimented nucleosomes allowed us to obtain near-complete assignments of histone H2A. Our data indicate that the H2A N- and C-terminal regions, including the αN element and docking domain, have increased flexibility compared to the core, and that its N- and C-terminal tails are highly dynamic, yet DNA-bound. It is noteworthy that these regions are rich in post-translational modifications, suggesting that the structural flexibility may be exploited by the enzymes that install, read, or remove these epigenetic marks. In addition, this is the first high-resolution study of histone structure and nucleosome binding in the crowded, solvated environment of the sediment, thereby mimicking the cellular conditions. Next to this biochemical insight, our approach provides a starting point for the more detailed analysis of the structure and dynamics of nucleosome complexes using the entire toolbox of biomolecular ssNMR spectroscopy. Whereas the solution NMR methyl-TROSY approach^[24] is superior in terms of spectral sensitivity and resolution,^[25] the approach described here enables the observation of all non-proline residues on the interface and poses fewer labeling requirements. Even though fractional deuteration and considerable measurement times are required, a major advantage is the lack of an intrinsic size limit for ssNMR, making the approach extendable to studies of larger complexes, including larger chromatin substrates such as nucleosomal arrays. We anticipate that in particular for dynamic complexes, for interactions with smaller proteins that cannot easily be captured by cryo-EM or crystallography, or when plasticity of histones is thought to play a role,^[26] the use of sedimented nucleosome–protein complexes will be an effective and complementary method in the structural biology of chromatin function.

Acknowledgements

This work received financial support from the Dutch Science Foundation NWO (VIDI grant 723.013.010 to H.v.I., VENI grant 722.016.002 to S.X., NWO-Groot grant 175.010.2009.002 to M.B., and grant 184.032.207 to the uNMR-NL National Roadmap Large-Scale Facility of the Netherlands). We thank Jelmer Eerland for solution-state assignments of H2A and the Leiden MacBio group members for their support.

Conflict of interest

The authors declare no conflict of interest.

Keywords: magic-angle spinning · NMR spectroscopy · nucleosomes · protein-protein interactions · proton detection

How to cite: Angew. Chem. Int. Ed. 2018, 57, 4571–4575 Angew. Chem. 2018, 130, 4661–4665

- a) K. Luger, A. W. Mader, R. K. Richmond, D. F. Sargent, T. J. Richmond, *Nature* **1997**, *389*, 251–260; b) R. D. Kornberg, Y. Lorch, *Cell* **1999**, *98*, 285–294.
- [2] a) K. Luger, M. L. Dechassa, D. J. Tremethick, *Nat. Rev. Mol. Cell Biol.* 2012, *13*, 436–447; b) R. K. McGinty, S. Tan, *Chem. Rev.* 2015, *115*, 2255–2273; c) R. K. McGinty, S. Tan, *Curr. Opin. Struct. Biol.* 2016, *37*, 54–61.
- [3] a) V. Speranzini, S. Pilotto, T. K. Sixma, A. Mattevi, *EMBO J.* 2016, *35*, 376–388; b) X. Liu, M. Li, X. Xia, X. Li, Z. Chen, *Nature* 2017, *544*, 440–445; c) L. Farnung, S. M. Vos, C. Wigge, P. Cramer, *Nature* 2017, *550*, 539–542; d) J. Kitevski-LeBlanc, A. Fradet-Turcotte, P. Kukic, M. D. Wilson, G. Portella, T. Yuwen, S. Panier, S. Duan, M. D. Canny, H. van Ingen, C. H. Arrowsmith, J. L. Rubinstein, M. Vendruscolo, D. Durocher, L. E. Kay, *eLife* 2017, *6*, e23872.
- [4] a) S. Luca, J. F. White, A. K. Sohal, D. V. Filippov, J. H. van Boom, R. Grisshammer, M. Baldus, Proc. Natl. Acad. Sci. USA 2003, 100, 10706-10711; b) I. Bertini, C. Luchinat, G. Parigi, E. Ravera, B. Reif, P. Turano, Proc. Natl. Acad. Sci. USA 2011, 108, 10396-10399; c) C. Gardiennet, A. K. Schutz, A. Hunkeler, B. Kunert, L. Terradot, A. Bockmann, B. H. Meier, Angew. Chem. Int. Ed. 2012, 51, 7855-7858; Angew. Chem. 2012, 124, 7977-7980; d) J. M. Lamley, D. Iuga, C. Öster, H.-J. Sass, M. Rogowski, A. Oss, J. Past, A. Reinhold, S. Grzesiek, A. Samoson, J. R. Lewandowski, J. Am. Chem. Soc. 2014, 136, 16800-16806; e) E. Barbet-Massin, C.-T. Huang, V. Daebel, S.-T. D. Hsu, B. Reif, Angew. Chem. Int. Ed. 2015, 54, 4367-4369; Angew. Chem. 2015, 127, 4441-4444; f) H. R. W. Dannatt, M. Felletti, S. Jehle, Y. Wang, L. Emsley, N. E. Dixon, A. Lesage, G. Pintacuda, Angew. Chem. Int. Ed. 2016, 55, 6638-6641; Angew. Chem. 2016, 128, 6750-6753.
- [5] a) J. C. Hansen, J. Ausio, V. H. Stanik, K. E. Van Holde, *Biochemistry* **1989**, 28, 9129–9136; b) M. Shogren-Knaak, H.

Angew. Chem. Int. Ed. 2018, 57, 4571-4575

Ishii, J. M. Sun, M. J. Pazin, J. R. Davie, C. L. Peterson, *Science* 2006, *311*, 844–847; c) A. Bertin, S. Mangenot, M. Renouard, D. Durand, F. Livolant, *Biophys. J.* 2007, *93*, 3652–3663; d) N. Korolev, A. Allahverdi, Y. Yang, Y. Fan, A. P. Lyubartsev, L. Nordenskiold, *Biophys. J.* 2010, *99*, 1896–1905; e) K. Maeshima, R. Rogge, S. Tamura, Y. Joti, T. Hikima, H. Szerlong, C. Krause, J. Herman, E. Seidel, J. DeLuca, T. Ishikawa, J. C. Hansen, *EMBO. J.* 2016, *35*, 1115–1132.

- [6] M. Gao, P. S. Nadaud, M. W. Bernier, J. A. North, P. C. Hammel, M. G. Poirier, C. P. Jaroniec, *J. Am. Chem. Soc.* 2013, 135, 15278– 15281.
- [7] a) A. J. Barbera, J. V. Chodaparambil, B. Kelley-Clarke, V. Joukov, J. C. Walter, K. Luger, K. M. Kaye, *Science* 2006, 311, 856–861; b) J. V. Chodaparambil, A. J. Barbera, X. Lu, K. M. Kaye, J. C. Hansen, K. Luger, *Nat. Struct. Mol. Biol.* 2007, 14, 1105–1107.
- [8] P. T. Lowary, J. Widom, J. Mol. Biol. 1998, 276, 19-42.
- [9] P. N. Dyer, R. S. Edayathumangalam, C. L. White, Y. Bao, S. Chakravarthy, U. M. Muthurajan, K. Luger, *Methods Enzymol.* 2004, 375, 23–44.
- [10] D. Mance, T. Sinnige, M. Kaplan, S. Narasimhan, M. Daniëls, K. Houben, M. Baldus, M. Weingarth, *Angew. Chem. Int. Ed.* 2015, 54, 15799–15803; *Angew. Chem.* 2015, 127, 16025–16029.
- [11] a) O. C. Andronesi, S. Becker, K. Seidel, H. Heise, H. S. Young, M. Baldus, *J. Am. Chem. Soc.* **2005**, *127*, 12965–12974; b) M. E. Ward, E. Ritz, M. A. M. Ahmed, V. V. Bamm, G. Harauz, L. S. Brown, V. Ladizhansky, *J. Biomol. NMR* **2015**, *63*, 375–388.
- [12] B. R. Zhou, H. Feng, R. Ghirlando, H. Kato, J. Gruschus, Y. Bai, J. Mol. Biol. 2012, 421, 30–37.
- [13] A. Stützer, S. Liokatis, A. Kiesel, D. Schwarzer, R. Sprangers, J. Soding, P. Selenko, W. Fischle, *Mol. Cell* **2016**, *61*, 247–259.
- [14] H. Kato, J. Gruschus, R. Ghirlando, N. Tjandra, Y. Bai, J. Am. Chem. Soc. 2009, 131, 15104–15105.
- [15] C. A. Davey, D. F. Sargent, K. Luger, A. W. Maeder, T. J. Richmond, J. Mol. Biol. 2002, 319, 1097-1113.

[16] D. H. Zhou, A. J. Nieuwkoop, D. A. Berthold, G. Comellas, L. J. Sperling, M. Tang, G. J. Shah, E. J. Brea, L. R. Lemkau, C. M. Rienstra, J. Biomol. NMR 2012, 54, 291–305.

Angewandte

Chemie

- [17] Y. Shen, A. Bax, J. Biomol. NMR 2013, 56, 227-241.
- [18] C. R. Clapier, S. Chakravarthy, C. Petosa, C. Fernandez-Tornero, K. Luger, C. W. Muller, *Proteins Struct. Funct. Bioinf.* 2008, 71, 1–7.
- [19] M. V. Berjanskii, D. S. Wishart, J. Am. Chem. Soc. 2005, 127, 14970–14971.
- [20] D. Vasudevan, E. Y. Chua, C. A. Davey, *J. Mol. Biol.* **2010**, *403*, 1–10.
- [21] C. Beauchemin, N. J. Moerke, P. Faloon, K. M. Kaye, J. Biomol. Screening 2014, 19, 947–958.
- [22] G. C. P. van Zundert, J. P. G. L. M. Rodrigues, M. Trellet, C. Schmitz, P. L. Kastritis, E. Karaca, A. S. J. Melquiond, M. van Dijk, S. J. de Vries, A. M. J. J. Bonvin, *J. Mol. Biol.* 2016, 428, 720–725.
- [23] a) A. M. J. J. Bonvin, *Curr. Opin. Struct. Biol.* 2006, *16*, 194–200;
 b) M. F. Lensink, S. Velankar, S. J. Wodak, *Proteins Struct. Funct. Bioinf.* 2017, 85, 359–377.
- [24] V. Tugarinov, P. M. Hwang, J. E. Ollerenshaw, L. E. Kay, J. Am. Chem. Soc. 2003, 125, 10420-10428.
- [25] a) H. Kato, H. van Ingen, B. R. Zhou, H. Feng, M. Bustin, L. E. Kay, Y. Bai, *Proc. Natl. Acad. Sci. USA* 2011, *108*, 12283–12288;
 b) R. van Nuland, F. M. van Schaik, M. Simonis, S. van Heesch, E. Cuppen, R. Boelens, H. M. Timmers, H. van Ingen, *Epigenet. Chromatin* 2013, *6*, 12.
- [26] K. K. Sinha, J. D. Gross, G. J. Narlikar, Science 2017, 355, eaaa3761.

Manuscript received: December 21, 2017

- Revised manuscript received: February 8, 2018
- Accepted manuscript online: February 21, 2018
- Version of record online: March 13, 2018