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catena-Poly[[*(5,5'*-dimethyl-2,2'-bipyridine- κ^2 N,N')cadmium]-di- μ -bromido]

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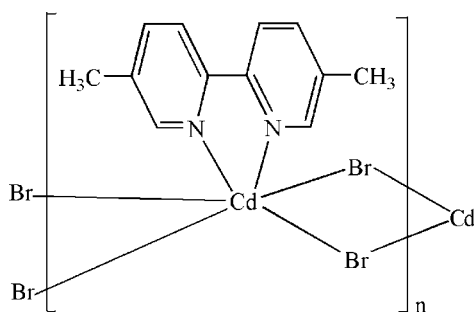
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.009$ Å;
R factor = 0.041; wR factor = 0.091; data-to-parameter ratio = 17.3.

In the crystal of the title polymeric compound, $[\text{CdBr}_2(\text{C}_{12}\text{H}_{12}\text{N}_2)]_n$, the Cd^{II} cation is located on a twofold rotation axis. The Cd^{II} cation is six-coordinated in a distorted octahedral geometry formed by two N atoms from the 5,5'-dimethyl-2,2'-bipyridine ligand and four bridging Br^- anions. The bridging function of the Br^- anions leads to a polymeric chain running along the c axis.

Related literature

For related structures, see: Ahmadi *et al.* (2008, 2010); Albada *et al.* (2004); Amani *et al.* (2007, 2009); Han *et al.* (2006); Kalateh *et al.* (2010); Karaca *et al.* (2009); Khalighi *et al.* (2008); Maheshwari *et al.* (2007); Tadayon Pour *et al.* (2008); Zhang (2007).



Experimental

Crystal data

$[\text{CdBr}_2(\text{C}_{12}\text{H}_{12}\text{N}_2)]$
 $M_r = 456.45$
Monoclinic, $C2/c$
 $a = 19.637$ (5) Å

$b = 9.6563$ (15) Å
 $c = 7.485$ (2) Å
 $\beta = 104.76$ (2) $^\circ$
 $V = 1372.4$ (6) Å 3

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 7.39$ mm $^{-1}$

$T = 298$ K
 $0.12 \times 0.11 \times 0.09$ mm

Data collection

Bruker APEXII CCD area-detector diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2001)
 $T_{\text{min}} = 0.435$, $T_{\text{max}} = 0.548$

5378 measured reflections
1346 independent reflections
1015 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.110$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.091$
 $S = 1.03$
1346 reflections

78 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.85$ e Å $^{-3}$
 $\Delta\rho_{\text{min}} = -0.70$ e Å $^{-3}$

Table 1

Selected bond lengths (Å).

Cd1—N1	2.352 (4)	Cd1—Br1 ⁱ	2.9351 (10)
Cd1—Br1	2.6676 (8)		

Symmetry code: (i) $x, -y, z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

We are grateful to the Islamic Azad University, Omidieh Branch, for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5547).

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supplementary materials

Acta Cryst. (2012). E68, m846 [doi:10.1107/S1600536812023860]

catena-Poly[[*(5,5'*-dimethyl-2,2'-bipyridine- κ^2 N,N')cadmium]-di- μ -bromido]

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Comment

5,5'-Dimethyl-2,2'-bipyridine (5,5'-dmbipy), is a good bidentate ligand, and numerous complexes with 5,5'-dmbipy have been prepared, such as that of zinc (Khalighi *et al.*, 2008), indium (Kalateh *et al.*, 2010), iron (Amani *et al.*, 2007), platinum (Amani *et al.*, 2009; Maheshwari *et al.*, 2007), copper (Albada *et al.*, 2004), gold (Karaca *et al.*, 2009), cadmium (Ahmadi *et al.*, 2008,2010) and mercury (Tadayon Pour *et al.*, 2008). Here, we report the synthesis and structure of the title compound.

The asymmetric unit of the title compound, (Fig. 1), contains one half-molecule; a twofold rotation axis passes through the Cd atom. The Cd^{II} atom is six-coordinated in a distorted octahedral configuration by two N atoms from 5,5'-dimethyl-2,2'-bipyridine and four bridging Br atoms. The bridging function of the bromide atoms leads to a one-dimensional chain structure. The Cd—Br and Cd—N bond lengths and angles (Table 1) are within normal range [Cd(phen)(μ -Br)₂]_n, (Zhang, 2007) and [Cd(bipy)(μ -Br)₂]_n, (Han *et al.*, 2006) [where phen is 1,10-phenanthroline and bipy is 2,2'-bipyridine].

Experimental

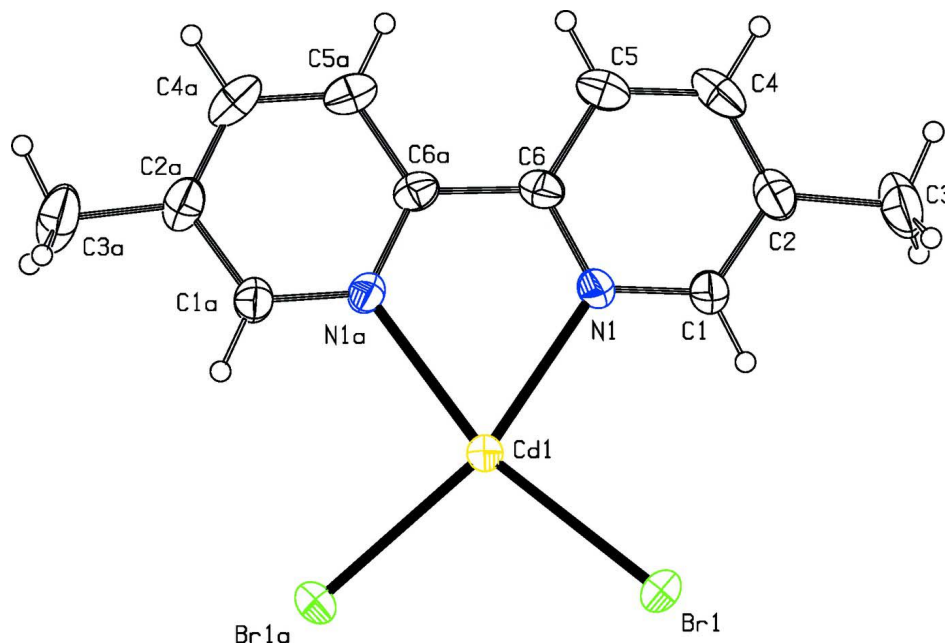
For the preparation of the title compound, a solution of 5,5'-dimethyl-2,2'-bipyridine (0.25 g, 1.33 mmol) in methanol (10 ml) was added to a solution of CdBr₂·4H₂O (0.46 g, 1.33 mmol) in methanol (10 ml) at room temperature. The suitable crystals for X-ray diffraction experiment were obtained by methanol diffusion to a colorless solution in DMSO. Suitable crystals were isolated after one week (yield; 0.45 g, 74.1%).

Refinement

H atoms were positioned geometrically with C—H = 0.93 Å and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Computing details

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT* (Bruker, 2007); program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

**Figure 1**

The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level [Symmetry codes: (a) 1 - x , y , 1/2 - z].

catena-Poly[[5,5'-dimethyl-2,2'-bipyridine- κ^2N,N' cadmium]-di- μ -bromido]

Crystal data

[CdBr₂(C₁₂H₁₂N₂)]

$M_r = 456.45$

Monoclinic, $C2/c$

$a = 19.637$ (5) Å

$b = 9.6563$ (15) Å

$c = 7.485$ (2) Å

$\beta = 104.76$ (2)°

$V = 1372.4$ (6) Å³

$Z = 4$

$F(000) = 864$

$D_x = 2.209$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5378 reflections

$\theta = 2.2$ – 26.0 °

$\mu = 7.39$ mm⁻¹

$T = 298$ K

Prism, colorless

$0.12 \times 0.11 \times 0.09$ mm

Data collection

Bruker APEXII CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2001)

$T_{\min} = 0.435$, $T_{\max} = 0.548$

5378 measured reflections

1346 independent reflections

1015 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.110$

$\theta_{\max} = 26.0$ °, $\theta_{\min} = 2.2$ °

$h = -22 \rightarrow 24$

$k = -10 \rightarrow 11$

$l = -9 \rightarrow 9$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.091$

$S = 1.03$

1346 reflections

78 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier map
 Hydrogen site location: inferred from neighbouring sites
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0403P)^2]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.004$
 $\Delta\rho_{\max} = 0.85 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.70 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.3751 (3)	0.2708 (6)	-0.0036 (8)	0.0498 (15)
H1	0.3565	0.1843	-0.0435	0.060*
C2	0.3367 (3)	0.3894 (7)	-0.0747 (9)	0.0549 (16)
C3	0.2663 (4)	0.3740 (9)	-0.2097 (11)	0.085 (3)
H3A	0.2355	0.3221	-0.1537	0.102*
H3B	0.2718	0.3261	-0.3175	0.102*
H3C	0.2465	0.4639	-0.2444	0.102*
C4	0.3660 (3)	0.5147 (7)	-0.0143 (9)	0.0600 (18)
H4	0.3423	0.5960	-0.0591	0.072*
C5	0.4299 (3)	0.5211 (6)	0.1118 (8)	0.0506 (15)
H5	0.4494	0.6067	0.1527	0.061*
C6	0.4659 (3)	0.4006 (5)	0.1792 (8)	0.0421 (13)
N1	0.4378 (2)	0.2774 (5)	0.1198 (6)	0.0412 (11)
Cd1	0.5000	0.07821 (6)	0.2500	0.0476 (2)
Br1	0.41572 (3)	-0.09792 (6)	0.02140 (9)	0.0507 (2)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.041 (3)	0.047 (3)	0.054 (4)	-0.004 (3)	-0.001 (3)	0.008 (3)
C2	0.041 (3)	0.066 (4)	0.055 (4)	0.003 (3)	0.008 (3)	0.020 (3)
C3	0.048 (4)	0.109 (6)	0.086 (6)	0.006 (4)	-0.005 (4)	0.034 (5)
C4	0.055 (4)	0.061 (4)	0.069 (4)	0.026 (3)	0.026 (3)	0.029 (4)
C5	0.062 (4)	0.040 (3)	0.056 (4)	0.012 (3)	0.027 (3)	0.007 (3)
C6	0.044 (3)	0.036 (3)	0.049 (3)	0.003 (2)	0.017 (3)	0.007 (2)
N1	0.033 (2)	0.038 (2)	0.049 (3)	0.0021 (18)	0.003 (2)	0.006 (2)
Cd1	0.0456 (4)	0.0318 (3)	0.0530 (4)	0.000	-0.0102 (3)	0.000
Br1	0.0474 (4)	0.0423 (3)	0.0550 (4)	-0.0094 (2)	-0.0006 (3)	-0.0064 (2)

Geometric parameters (Å, °)

C1—N1	1.339 (6)	C5—C6	1.388 (8)
C1—C2	1.400 (8)	C5—H5	0.9300
C1—H1	0.9300	C6—N1	1.339 (7)
C2—C4	1.366 (10)	C6—C6 ⁱ	1.481 (11)
C2—C3	1.497 (9)	Cd1—N1	2.352 (4)
C3—H3A	0.9600	Cd1—N1 ⁱ	2.352 (4)
C3—H3B	0.9600	Cd1—Br1	2.6676 (8)
C3—H3C	0.9600	Cd1—Br1 ⁱ	2.6676 (8)
C4—C5	1.366 (9)	Cd1—Br1 ⁱⁱ	2.9351 (10)
C4—H4	0.9300	Cd1—Br1 ⁱⁱⁱ	2.9352 (10)
N1—C1—C2	122.3 (6)	C5—C6—C6 ⁱ	123.0 (4)
N1—C1—H1	118.8	C6—N1—C1	120.0 (5)
C2—C1—H1	118.8	C6—N1—Cd1	117.6 (3)
C4—C2—C1	117.3 (5)	C1—N1—Cd1	122.4 (4)
C4—C2—C3	123.3 (6)	N1—Cd1—N1 ⁱ	70.3 (2)
C1—C2—C3	119.4 (6)	N1—Cd1—Br1	94.81 (10)
C2—C3—H3A	109.5	N1 ⁱ —Cd1—Br1	163.48 (11)
C2—C3—H3B	109.5	N1—Cd1—Br1 ⁱ	163.48 (11)
H3A—C3—H3B	109.5	N1 ⁱ —Cd1—Br1 ⁱ	94.81 (10)
C2—C3—H3C	109.5	Br1—Cd1—Br1 ⁱ	100.78 (4)
H3A—C3—H3C	109.5	N1—Cd1—Br1 ⁱⁱ	84.78 (12)
H3B—C3—H3C	109.5	N1 ⁱ —Cd1—Br1 ⁱⁱ	89.13 (12)
C5—C4—C2	120.2 (6)	Br1—Cd1—Br1 ⁱⁱ	96.79 (3)
C5—C4—H4	119.9	Br1 ⁱ —Cd1—Br1 ⁱⁱ	87.96 (3)
C2—C4—H4	119.9	N1—Cd1—Br1 ⁱⁱⁱ	89.13 (12)
C4—C5—C6	120.4 (6)	N1 ⁱ —Cd1—Br1 ⁱⁱⁱ	84.78 (12)
C4—C5—H5	119.8	Br1—Cd1—Br1 ⁱⁱⁱ	87.96 (2)
C6—C5—H5	119.8	Br1 ⁱ —Cd1—Br1 ⁱⁱⁱ	96.79 (3)
N1—C6—C5	119.7 (5)	Br1 ⁱⁱ —Cd1—Br1 ⁱⁱⁱ	172.56 (3)
N1—C6—C6 ⁱ	117.3 (3)	Cd1—Br1—Cd1 ⁱⁱⁱ	92.04 (3)
N1—C1—C2—C4	0.5 (10)	C1—N1—Cd1—N1 ⁱ	-177.4 (6)
N1—C1—C2—C3	-178.9 (6)	C6—N1—Cd1—Br1	-172.9 (4)
C1—C2—C4—C5	-0.6 (10)	C1—N1—Cd1—Br1	10.0 (5)
C3—C2—C4—C5	178.8 (6)	C6—N1—Cd1—Br1 ⁱ	26.4 (8)
C2—C4—C5—C6	0.2 (10)	C1—N1—Cd1—Br1 ⁱ	-150.8 (4)
C4—C5—C6—N1	0.2 (10)	C6—N1—Cd1—Br1 ⁱⁱ	90.7 (4)
C4—C5—C6—C6 ⁱ	-177.8 (7)	C1—N1—Cd1—Br1 ⁱⁱ	-86.4 (4)
C5—C6—N1—C1	-0.3 (9)	C6—N1—Cd1—Br1 ⁱⁱⁱ	-85.0 (4)
C6 ⁱ —C6—N1—C1	177.8 (6)	C1—N1—Cd1—Br1 ⁱⁱⁱ	97.8 (4)
C5—C6—N1—Cd1	-177.5 (4)	N1—Cd1—Br1—Cd1 ⁱⁱⁱ	88.96 (12)
C6 ⁱ —C6—N1—Cd1	0.5 (9)	N1 ⁱ —Cd1—Br1—Cd1 ⁱⁱⁱ	63.9 (4)
C2—C1—N1—C6	-0.1 (9)	Br1 ⁱ —Cd1—Br1—Cd1 ⁱⁱⁱ	-96.53 (2)
C2—C1—N1—Cd1	177.0 (5)	Br1 ⁱⁱ —Cd1—Br1—Cd1 ⁱⁱⁱ	174.26 (2)
C6—N1—Cd1—N1 ⁱ	-0.2 (3)	Br1 ⁱⁱⁱ —Cd1—Br1—Cd1 ⁱⁱⁱ	0.0

Symmetry codes: (i) $-x+1, y, -z+1/2$; (ii) $x, -y, z+1/2$; (iii) $-x+1, -y, -z$.