

DNA strand break induction of aqueous plasmid DNA exposed to 30 MeV protons at ultra-high dose rate

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ABSTRACT

Radiation cancer therapy with ultra-high dose rate exposure, so called FLASH radiotherapy, appears to reduce normal tissue damage without compromising tumor response. The aim of this study was to clarify whether FLASH exposure of proton beam would be effective in reducing the DNA strand break induction. We applied a simple model system, pBR322 plasmid DNA in aqueous $1 \times \text{TE}$ solution, where DNA single strand breaks (SSBs) and double strand breaks (DSBs) can be precisely quantified by gel electrophoresis. Plasmid DNA were exposed to 27.5 MeV protons in the conventional dose rate of 0.05 Gy/s (CONV) and ultra-high dose rate of 40 Gy/s (FLASH). With both dose rate, the kinetics of the SSB and DSB induction were proportional to absorbed dose. The SSB induction of FLASH was significantly less than CONV, which were 8.79 ± 0.14 (10^{-3} SSB per Gy per molecule) and 10.8 ± 0.68 (10^{-3} SSB per Gy per molecule), respectively. The DSB induction of FLASH was also slightly less than CONV, but difference was not significant. Altogether, 27.5 MeV proton beam at 40 Gy/s reduced SSB and not DSB, thus its effect may not be significant in reducing lethal DNA damage that become apparent in acute radiation effect.

Keywords: FLASH; proton; high dose rate; DNA strand breaks; plasmid DNA

INTRODUCTION

FLASH radiotherapy targets tumors while minimizing the damage to the surrounding normal tissues by ultra-high dose rate (> 40 Gy/s) [1,2], which exceeds currently employed clinical dose rates by about a factor of 100–1000. The advantageous feature of FLASH, the so called sparing effect, enables radiation therapy to maintain the effectiveness of tumor killing, or increase dose delivery at the tumor region without increasing the toxicity in normal health tissues [3–6]. FLASH effect studies have begun in electrons [3] and photons [4, 7, 8]. The first clinical FLASH radiotherapy trial with electrons proved its feasibility

and safety with favorable outcomes both on normal skin protection and tumor control. A recent overview reported a protective effect in normal tissues for FLASH ranging from about 1.4 to 1.8 [6].

The FLASH effect depends on the balance between dose, oxygen concentration, radical production and reactions, which contribute to the reduction of biological toxicity (9–12). There are two major hypotheses for the sparing effect of FLASH. First is the radiochemical depletion of oxygen at FLASH dose rates that suppresses the fixation of indirect radiation-induced DNA damage, which results in a sparing effect conferred to the irradiated tissue [8]. Second is the improved

immune response, due to the fast exposure time leading to less irradiation of circulating immune cells by FLASH radiotherapy compared to CONV radiotherapy, which results in a reduction of radiation-induced chronic inflammation and other deteriorative effects [9, 13].

For protons, recent papers have reported the sparing effects in mammalian cells [14, 15], and animal models [5]. It is well known that protons and heavier charged particle therapy have advantages compared to modern photon and electron therapy [16, 17]. Thus, there are emerging needs to characterize the biological effects of proton irradiation due to the advantageous features that encourage the adoption of proton radiotherapy [18, 19]. However, as for charged particles, the subject remains controversial [20, 21]. Recently, Kusumoto *et al.* [22] reported the radiation chemical yields (G values) of 7-hydroxy-coumarin-3-carboxylic acid (7OH-C3CA), which is produced by water radiolysis using coumarin-3-carboxylic acid (C3CA) solution as a radical scavenger of hydroxyl radicals. They have clearly demonstrated that increasing the dose rate from 0.05 to 160 Gy/s significantly reduced the value of G (7OH-C3CA) due to the oxygen depletion of 27.5 MeV protons. The proton-FLASH effect has been investigated in many radiobiological studies with cells and mice. Furthermore, simulation studies and a kinetic model are vigorously developed to understand the mechanism of the FLASH effect [10, 23]. However, studies are limited by the availability of irradiators that can provide such dose rates. The mechanism behind the sparing effect of the proton-FLASH in correlation with DNA damage induction needs further investigation to clarify the mechanistic aspect of the proton-FLASH effect.

Here, we report the first investigation on physical and chemical damage processes in DNA of proton-Flash versus conventional proton irradiation, which were assessed by investigating the DNA single strand breaks (SSBs) and double strand breaks (DSBs) induction rate in a simple model system, plasmid DNA in aqueous conditions, where SSBs and DSBs were quantified by agarose gel electrophoresis. Plasmid pBR322 DNA were exposed to 27.5 MeV protons in the conventional dose rate of 0.05 Gy/s (CONV) and ultra-high dose rate of 40 Gy/s (FLASH) to the and the SSB and DSB induction rates were validated.

METHODS AND MATERIALS

Sample preparation

The pBR322 plasmid DNA solution (4361 bp, 0.5 $\mu\text{g}/\mu\text{l}$ in $1 \times \text{TE}$) were purchased from Takara Bio Inc, Shiga, Japan and was over 90% in supercoiled form (form 1) and without any linear form (form 3). Plasmid DNA solution were diluted in $1 \times \text{TE}$ buffer (10 mM Tris-HCl, 1 mM EDTA, pH 7.4) to be 50 ng/ μL and each DNA samples were prepared in 0.2 mL PCR tubes (PCR-02F2, BMBio Equipment Co., LTD) containing 50 μL of DNA solution for proton beam irradiation.

Irradiation

Irradiation experiments were performed at the AVF-930 cyclotron facility [24] in the Institute for Quantum Medical Science (iQMS), National Institutes for Quantum Science and Technology (QST), Chiba, Japan. Irradiation set ups are described in detail elsewhere [22]. Briefly, protons were accelerated up to 27.5 MeV, where energy were 27.5 MeV at the sample irradiated position after penetrating beam monitors, beam exit window, air gaps and other necessary equipment

installed in the beam line. Beam fields were confirmed using EBT3 GAF chromic film (Ashland Advance Materials, NJ) to assure the sample tubes were set in uniform beam field within $\pm 5\%$ difference. The beam intensity was monitored with a parallel plate ionization chamber installed in front of the sample and it was characterized with the Markus ionization chamber for absorbed dose. The thickness of the PCR tube was 0.5 mm and that of solution was 4 mm coaxial with beam trajectory. An average linear energy transfer (LET) in the solution was calculated to be 2.3 keV/ μm with SRIM code [25].

The beam currents were controlled to be 0.2 nA and 300 nA, which the absorbed dose rates were 0.05 Gy/s and 40 Gy/s, respectively. The two dose rates were chosen to compare the DNA strand break yields of plasmid DNA in solution condition between conventional dose rate (CONV) and high dose rate (FLASH). Absorbed doses were controlled by the time width of pulse signals to the beam deflector installed in the beam line.

Agarose gel electrophoresis and quantification of DNA strand breaks

DNA solution of 10 μl (pBR322 DNA: 500 ng) were mixed with 1 μL of $10 \times$ gel-loading blue buffer (40% sucrose and 0.25% bromophenol blue) and then electrophoresed in 1.4% agarose gel (LO3, Takara-Bio Inc., Shiga, Japan) in $1 \times \text{TAE}$ buffer (40 mM Tris-acetate, 5 mM sodium acetate and 1 mM EDTA; pH 7.8 adjusted with acetic acid), at 4.2 V/cm for 4 h at 4°C. Under these electrophoresis conditions, two SSBs (one on the opposite strand) with less than 6 base pair apart were detected as DSB [26, 27]. Gels were then stained in 1 $\mu\text{g}/\text{ml}$ ethidium bromide (EtBr) solution for 1 hr, and subsequently washed twice in water for 30 min each. Fluorescence gel images were recorded using Molecular Imager Phoros FX system (Bio-Rad Laboratories, Inc.), which excite the EtBr with 532 nm laser and records the fluorescence after 605 nm band path emission filter. The gel image was exported as raw TIFF format and analyzed using Multi Gauge Version 2.3 (Fujifilm Holdings Corporation, Tokyo).

Using the image analysis software, the total fluorescent intensities in the three bands corresponding to the three forms of plasmid DNA, super-helical closed circular (form 1, no strand break), open circular (form 2, with SSB) and linear (form 3, with DSB) were determined. To account for the reduced uptake of the ethidium bromide by form 1 DNA, a correction factor of 1.42 described by Lloyd *et al.* [28], was used. The numbers of SSB (N_{SSB}), and DSB (N_{DSB}) per DNA molecule were calculated using equations (1–3) described by Povirk *et al.* [29], where F_{1-3} represent the fractions of forms 1–3, respectively:

$$F_1 + F_2 + F_3 = 1 \quad (1)$$

$$N_{\text{SSB}} = \ln(F_1) - \ln(1 + N_{\text{DSB}}) \quad (2)$$

$$N_{\text{DSB}} = \frac{F_3}{1 - F_3} \quad (3)$$

RESULTS AND DISCUSSION

Figure 1 shows the representative result of agarose gel electrophoresis. The pBR322 plasmid DNA were separated into non-strand break

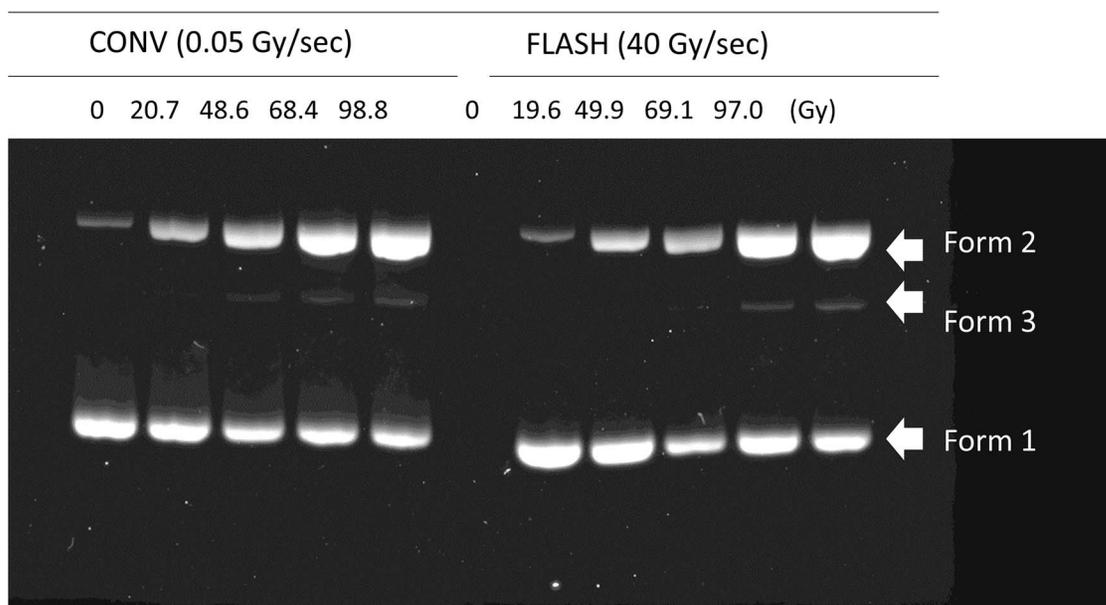


Fig. 1. Representative image of agarose gel electrophoresis. CONV and FLASH irradiated pBR322 plasmid DNA in solution were electrophoresed and were isolate according to their molecular forms. Induction rate (N_{SSB} and N_{DSB} per Gy) analyzed from the shown gel image were, 9.7×10^{-3} ($N_{SSB}/\text{molecule}/\text{Gy}$) and 9.4×10^{-5} (N_{DSB}/Gy) for CONV and 8.8×10^{-3} ($N_{SSB}/\text{molecule}/\text{Gy}$) and 9.5×10^{-5} ($N_{DSB}/\text{molecule}/\text{Gy}$) for FLASH, respectively. Fluorescence values of each band detected in the lanes are indicated in supplemented [Table S1](#).

supercoiled DNA, open circular DNA with an SSB and linear DNA resulted from a DSB, which were form 1, form 2 and form 3 mentioned in the previous section. It is apparent that the fluorescence of the non-damaged DNA (Form 1) decreased as form 2 increased, depending on the increased absorbed dose. Form 3 was detectable from 20 Gy and quantified from the fluorescence gel images. [Figure 2](#) show the induced number of SSB and DSB per plasmid molecule (N_{SSB} and N_{DSB}) as a function of absorbed dose at CONV (blue, circle) and FLASH dose rate (red, circle). Plots in the figures are from three independent irradiation experiments. Both N_{SSB} and N_{DSB} were increased proportional to the absorbed dose, and they were fitted to a linear function with the least-squares method. The slopes are the induction rate of N_{SSB} and N_{DSB} per absorbed dose, R_{SSB} and R_{DSB} , respectively. The averaged values of R_{SSB} and R_{DSB} for CONV and FLASH and the standard errors of three independent experiments are summarized in [Table 1](#).

Significant suppression was observed in R_{SSB} of FLASH compared to that of CONV. R_{SSB} for CONV and FLASH were 10.8 ± 0.68 [$\times 10^{-3}$ ($N_{SSB}/\text{molecule}/\text{Gy}$)] and 8.79 ± 0.14 [$\times 10^{-3}$ ($N_{SSB}/\text{molecule}/\text{Gy}$)], respectively. R_{SSB} compared to CONV are shown as R_{eff} , a ratio FLASH:CONV, which was 1:0.81, in another words, FLASH suppressed the R_{SSB} nearly 19% compared to CONV. In the chemical stage of radiation damage, most of the hydrated electrons and hydrogen radicals produced by water radiolysis react with the dissolved molecular oxygen resulting cytotoxic superoxide anion and the perhydroxyl radicals. However, under the FLASH dose-rate it is estimated that the oxygen is consumed by hydrated electrons and hydrogen radicals, and its rediffusion into the irradiated volume can be excluded, a transient acute radiation-induced hypoxia increases the

radio-resistance [23]. Kusumoto *et al.* [22] measured the G values of 7OH-C3CA of 27.5 MeV proton beam in wide range of dose rate from 0.05 Gy/s to 160 Gy/s and found that $G(7\text{OH-C3CA})$ reduces with increasing dose rate. They concluded that under FLASH dose rate, oxygen molecules were rapidly consumed by the hydrogen radicals and hydrated electrons produced by the water radiolysis, leading the lower $G(7\text{OH-C3CA})$ values. From the work others [22, 23], it can be estimated that FLASH irradiation would reduce the production of cytotoxic superoxide anion and the perhydroxyl radicals by oxygen depletion, resulting suppression of DNA damage induction. If the case, the FLASH effect would be strongly dependent on the radiation quality that results radiological process of indirect action and high oxygen enhancement ratio. The proton beam in this study was a 27.5 MeV proton which has $G(\text{OH})100$ eV of ~ 3 and OER that are equivalent to that of X-rays and gamma-rays [22, 30, 38, 39].

However, the degree of suppression of SSB can be considered small compared to the suppression rate of $G(7\text{OH-C3CA})$ value of FLASH compared to CONV, which was nearly 50%. As mentioned earlier, the plasmid DNA were exposed to proton beam in $1 \times \text{TE}$ buffer which contains 10 mM Tris(hydroxymethyl)aminomethane (Tris) that react as radical scavenger against hydroxyl radicals with the scavenging capacity of 1.5×10^7 (s^{-1}) [31–33]. Tris modification by hydroxyl radicals are estimated to consume oxygen molecules which oxidizes primary radicals to peroxy species [33, 34]. Oxygen depletion effects of proton-FLASH may have effect following chemical reaction, but negligible on initial chemical process against the hydroxyl radicals. Notably, it was previously reported that $G(\text{OH})$ value do not change and are consistent in the dose rate in the range of 0.05 Gy/s to 160 Gy/s

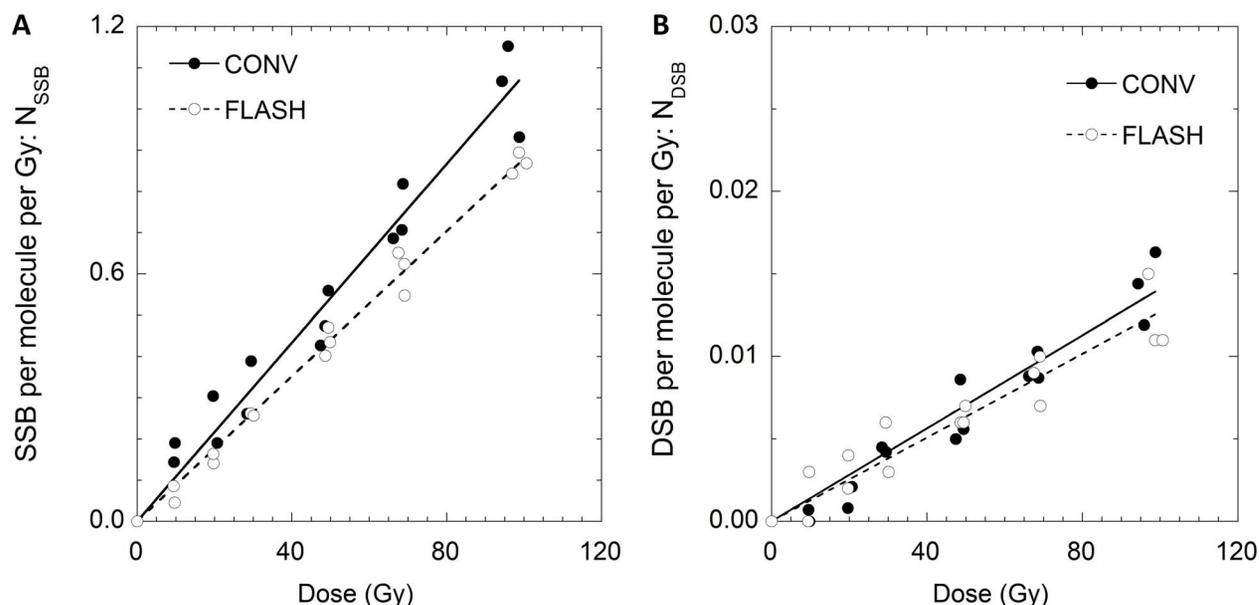


Fig. 2. Number of SSB (N_{SSB} , Panel A) and DSB (N_{DSB} , Panel B) and per plasmid irradiated at CONV and FLASH dose rate as a function of absorbed dose in Gy. The solid lines are the fit obtained by linear regression of the experimental data.

Table 1. SSB and DSB induction rate in pBR322 plasmid DNA by FLASH and CONV irradiation

Exposure	Dose Rate [Gy/s]	Induction rate, R		Ratio R_{SSB}/R_{DSB}
		$\times 10^{-3} [N_{SSB}, N_{DSB}/Gy]$		
		R_{SSB}	R_{DSB}	
CONV	0.05	10.8 ± 0.68	0.118 ± 0.021	91.9 ± 11.4
FLASH	40	8.79 ± 0.14	0.108 ± 0.016	81.1 ± 6.9
$^2R_{eff}$		0.81 ± 0.05	0.92 ± 0.18	0.88 ± 0.019
P-value		0.044	0.57	0.29

¹Induced number of SSB and DSB per molecule per Gy. ²Ratio of FLASH/CONV ³ standard errors from 3 independent experiments are shown in \pm . P-values are from student's t-test.

[22]. Therefore, it can be estimated that the Tris scavenged the OH radicals equally in CONV and FLASH, but other radicals decreased in FLASH due to oxygen depletion effect, which may have contributed to the significant but small reduction of R_{SSB} .

DSB induction rate, R_{DSB} for CONV and FLASH were $0.118 \pm 0.021 [\times 10^{-3} (N_{SSB}/molecule/Gy)]$ and $0.108 \pm 0.016 [\times 10^{-3} (N_{DSB}/molecule/Gy)]$, respectively. R_{eff} for R_{DSB} was 0.92 ± 0.12 , however, the difference was not significant. In our set up, three different molecular forms of plasmid DNA were isolated and quantified with agarose electrophoresis that detected DSBs as a fraction of linear type form 3. Plasmid DNA will need two SSBs induced in the distance less than 6 bp to form 3, in other words, when the DSBs were induced by two indirect actions, such as two OH radical attacks, the OH radicals must be produced in area of 4 nm^2 ($6 \text{ bp} \times 0.34 \text{ nm/bp}$). DSB would not be induced by OH radicals alone, since a sufficient number of OH radicals would not be produced per track of 27.5 MeV proton. However, in terms of 'spurs' will be produced along the proton tracks

which is defined to contain energy up 100 eV and have average of three ion pairs in the size of 4 nm in diameter area. The dimension of 'spurs' relatively identical to the DNA helix, if overlapped with DNA, it will result in multiple and various complex damage. On the other hand, the contribution of oxygen depletion effect of proton-FLASH, which suppressed hydrogen radicals and hydrated electrons are the suppression of indirect action of the biological effect, that are mostly detected as reduced values in SSB induction and not DSB. In addition, there are calculations that show a single hydrated electron can induced base damages but is not effective enough to induce a DNA strand break [36], which explains why there was no significant reduction in DSB at the FLASH condition. In fact, others reported that acute responses in mammalian cells exposed to proton-FLASH did not result in suppression on DSB induction and cell survival [15]. Indeed, the FLASH effect would strongly correlate on radiation types. For example, charged particles near the Bragg peak are well known to have high LET with highly localized dose distribution along its

tracks [37] that result in a small oxygen enhancement ratio [38, 39]. It is explained as a result of small $G(\text{OH})$ value [24, 32] and low consumption of the molecular oxygen per 100 eV at charged particles with high LET that drastically decrease near the Bragg peak [12]. Thus, studies that reported little impact of proton-FLASH on acute effects in mammalian cells [15] may be due to the radiation quality of low energetic charged particles that were used [5]. Still, hydrogen radicals and hydrated electrons can contribute to DNA damage at the vicinity of induced DNA strand breaks, such as clustered DNA damage. Most importantly, induction of clustered DNA damage at vicinity DSBs are critical and become a challenge to the repair system of living cells [40–43]. Therefore, the possibility of the sparing effect of FLASH due to the oxygen depletion effect are undeniable, and further investigation will be necessary to confirm the physical and chemical damage process to DNA.

In the present study, we have evaluated the SSB and DSB induction rate in plasmid DNA of aqueous conditions by proton beams under the CONV and FLASH. As a result, proton-FLASH reduced SSB induction, but not DSBs. In conclusion, the FLASH effect with 27.5 MeV protons may not be sufficient in reducing complex/clustered DNA damage, which become the main cause of lethality. On the other hand, SSBs are relatively easily repaired compared to DSBs in the living cells, thus the FLASH effect would be effective in reducing the non-lethal damage that may lead to late effects, such as cell senescence, genomic instability and cell transformation. Moreover, further investigation on the changes in the types DNA damages with parameters such as higher dose rate, proton energy and oxygen pressure will be necessary to clarify the underlying physical and chemical process resulting the biological effect by proton-FLASH.

SUPPLEMENTARY DATA

Supplementary data is available at *RADRES Journal* online.

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CONFLICT OF INTEREST

The authors declare they have no conflicts of interest.

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