



Synergistic chiral iminium and palladium catalysis: Highly regio- and enantioselective [3 + 2] annulation reaction of 2-vinylcyclopropanes with enals

Haipan Zhu^{1,2}, Peile Du², Jianjun Li¹, Ziyang Liao¹, Guohua Liu², Hao Li^{*1} and Wei Wang^{*1,3}

Full Research Paper

[Open Access](#)

Address:

¹State Key Laboratory of Bioengineering Reactor, Shanghai Key Laboratory of New Drug Design and School of Pharmacy, East China University of Science and Technology, 130 Mei-long Road, Shanghai 200237, China, ²Department of Chemistry, Shanghai Normal University, Shanghai 200234, China and ³Department of Chemistry and Chemical Biology, University of New Mexico, Albuquerque, NM 87131-0001, USA

Email:

Hao Li^{*} - hli77@ecust.edu.cn; Wei Wang^{*} - wwang@unm.edu

* Corresponding author

Keywords:

[3 + 2] annulation; enals; synergistic catalysis; vinylcyclopropanes

Beilstein J. Org. Chem. **2016**, *12*, 1340–1347.

doi:10.3762/bjoc.12.127

Received: 29 March 2016

Accepted: 02 June 2016

Published: 29 June 2016

This article is part of the Thematic Series "Strategies in asymmetric catalysis".

Guest Editor: T. P. Yoon

© 2016 Zhu et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

A cooperative catalytic strategy of chiral iminium catalysis by regioselective activation of the C=C bond in enals and a transition metal promoting to open the 2-vinylcyclopropanes for highly regio- and enantioselective [3 + 2] cycloaddition reaction of 2-vinylcyclopropanes with α,β -unsaturated aldehydes has been developed.

Introduction

The power of "donor–acceptor" (D–A) cyclopropanes as versatile 1,3-dipolar components is fuelled by its capacity of serving a complementary approach to a wide array of 5-membered ring structures, which are difficult or impossible to access by classic [3 + 2] cycloaddition reactions [1–34]. In recent years, significant efforts have been devoted to developing a catalytic enantioselective version of the processes. In this context, the D–A cyclopropanes have been applied for the reaction with highly active dipolarophiles, such as electrophilic C=O [35], e.g., alde-

hydes [36–38], ketones [38,39], and imines [40], and nucleophilic enol ethers [38,41], enamides [42], and indoles [43]. Nonetheless, the reactions with the α,β -unsaturated aldehydes and ketones face important challenges. To the best of our knowledge, so far merely two catalyst manifolds have been realized to effect the transformations with C=C double bonds instead of C=O in the α,β -unsaturated systems. Tsuji described the first organometallic promoted non-asymmetric reaction between D–A cyclopropanes and methyl vinyl ketone and α,β -unsatu-

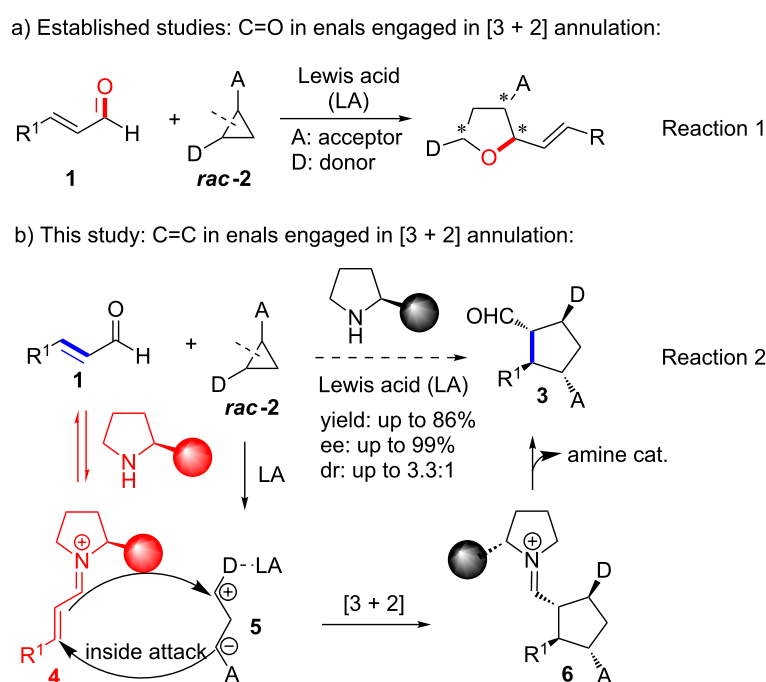
rated esters [44]. Trost and co-workers orchestrated the only example of the enantioselective reaction of D–A cyclopropanes with C=C double bonds with Meldrum's acid and alkylidenes or azlactone alkylidenes, catalyzed by the chiral Trost Pd(0)-complexes [45]. However, it is difficult to apply the catalytic system for the regio-controlled reaction with C=C bonds in α,β -unsaturated carbonyl compounds, particularly enals. The highly active aldehyde functionality reacts more favorably with the D–A cyclopropane resulting 1,3-dipoles, as elegantly demonstrated by Johnson and Waser for the formation of chiral tetrahydrofurans (Scheme 1, reaction 1) [36,38]. Achieving a regioselective control at the C=C bond rather than at C=O in enals represents a challenge and has not been reported.

Synergistic catalysis is a very important and useful strategy in organic synthesis by offering power for improving reaction efficiency and/or realizing impossible processes [46–55]. Recently, we developed an enantioselective addition of aldehydes to vinylpyridines and vinylarenes catalyzed by synergistic catalysis of iminium catalyst and Brønsted acid [56]. Herein we wish to disclose the first synergistic catalytic enantioselective [3 + 2] annulation reaction between 2-vinylcyclopropanes and enals via 1,4-addition (Scheme 1, reaction 2). The process proceeds highly regio- and enantioselectively with C=C bonds in enals. Notably, a synergistic catalytic system is implemented and makes this previously inaccessible [3 + 2] annulation transformation possible.

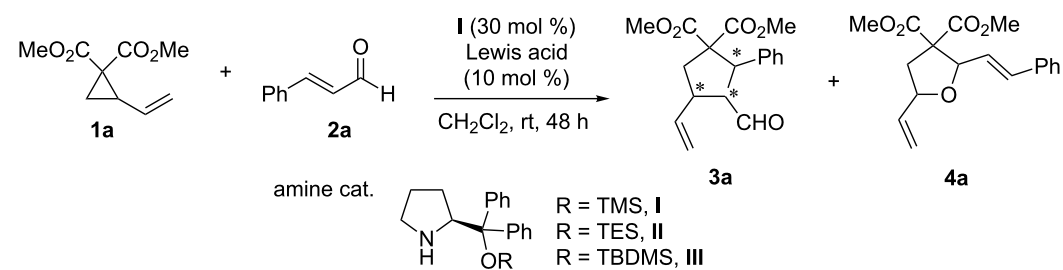
Results and Discussion

To render the [3 + 2] annulation reaction to selectively act on the C=C double bond rather than on the aldehyde in enals **1**, we proposed a new cooperative iminium and Lewis acid (LA) catalysis strategy (Scheme 1, reaction 2) [49,50,57–76]. The iminium catalysis plays an important dual role in the process. The formed iminium ion **4** derived from aldehyde **1** and an amine catalyst activates the C=C bond and sterically blocks the attack of the C=N iminium ion functionality posed by the bulky amine catalyst. In parallel, a LA promotes to open the D–A cyclopropanes **2**. The cooperative activation of two independent substrates by respective iminium and Lewis acid catalysis may enable an unprecedented catalytic regio- and enantioselective [3 + 2] annulation process, which offers a new approach to synthetically important heavily functionalized chiral cyclopentane structures **3**, bearing at least 3 stereogenic centers in this one-pot operation [77,78].

To test the feasibility of the designed [3 + 2] annulation process [79–94], we started our investigation by carrying out the reaction between the commonly used D–A system dimethyl 2-vinylcyclopropane-1,1-dicarboxylate (**1a**) and *trans*-cinnamaldehyde (**2a**) catalyzed in the presence of a LA and chiral amine **I** in CH₂Cl₂ at rt for 48 h (Table 1). A series of Lewis acids were initially screened. FeCl₃ and Cu(OTf)₂ gave the 1,2-cycloaddition product tetrahydrofuran **4a** (Table 1, entries 1 and 2). It is also disappointing that others Lewis acids, such as CuCl₂,



Scheme 1: Catalytic regio- and enantioselective [3 + 2] annulation reactions of 2-vinylcyclopropanes with enals.

Table 1: Screening of lewis acids.^a


Entry	LA	Yield (%) ^b , 3a	Yield (%) ^b , 4a
1	FeCl ₃	0	53
2	Cu(OTf) ₂	0	47
3	CuCl ₂	0	0
4	MgI ₂	0	0
5	ZnBr ₂	0	0
6	ZnCl ₂	0	0
7	FeCl ₂	0	0
8 ^c	Pd ₂ (dba) ₃	48	0

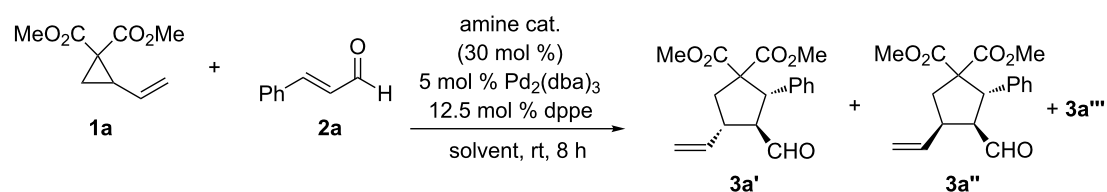
^aThe reaction was carried out with **1a** (36.8 mg, 0.2 mmol) and **2a** (26.4 mg, 0.2 mmol) in the presence of 10 mol % LA and 30 mol % amine **I** in 0.8 mL of CH₂Cl₂ at rt for 48 h. ^bIsolated yields; ^c5 mol % Pd₂(dba)₃ and 12.5 mol % dppe was used.

MgI₂, ZnBr₂, ZnCl₂ and FeCl₂ failed to promote these processes (Table 1, entries 3–7). Inspired by Trost's work of Pd(0)-catalyzed annulations of D–A cyclopropanes with C=C double bonds with Meldrum's acid and alkylidenes or azlactone alkylidenes [45], we probed the Pd₂(dba)₃-dppe complex for the 1,4-addition cycloaddition reaction (Table 1, entry 8). It was found that the reaction took place to afford the desired cyclopentane **3a**.

Encouraged by this result, we carried out further investigations of the co-catalysts promoted process (Table 2). First, we determined the diastereo- and enantioselectivity of the reaction. The ¹H NMR of the reaction crude mixture showed three diastereoisomers. The two major diastereoisomers were determined to be (2*S*,3*S*,4*S*)-**3a'** and (2*S*,3*S*,4*R*)-**3a''** in 2:1 ratio (Table 2, entry 1) based on single X-ray crystallographic analysis (see Scheme 2). Unfortunately, the third diastereoisomer **3a'''** was too hard to be separated to determine its stereochemistry. The enantioselectivities of two major diastereoisomers are even more encouraging (80 and 76% ee). Further investigations of solvents revealed the medium-dependent effect (Table 2, entries 1–8). No reaction happened in toluene (Table 2, entry 2). Disappointing outcomes were also received in DCE, ether, CH₃CN and EtOAc (Table 2, entries 3–6). Gratifyingly, in CHCl₃ this reaction proceeded smoothly to furnish the desired cyclopentanes in 63% yield with 99% ee for major **3a'** and 83% ee for minor **3a''** with a dr ratio of 1.7:1 (Table 2, entry 7). The reaction performed in THF was interesting: No reaction occurred at

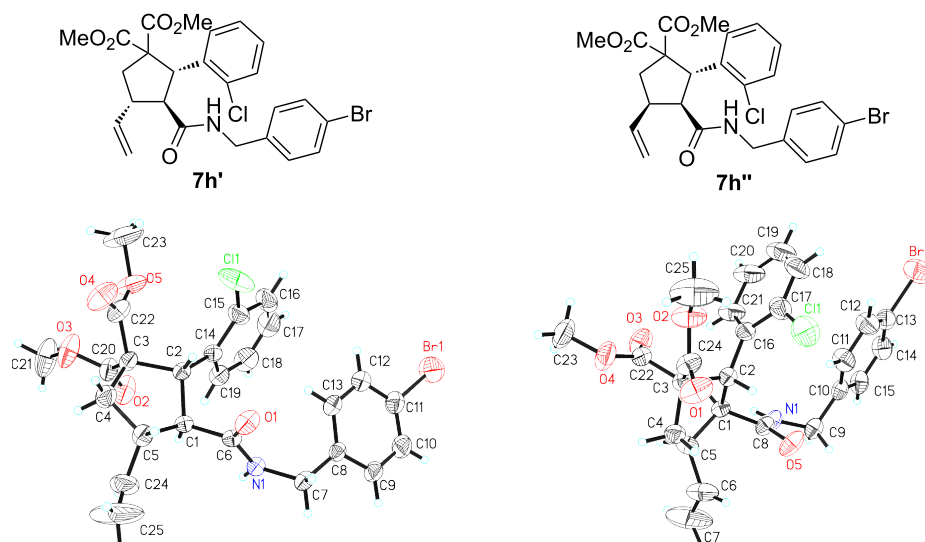
rt (Table 2, entry 8), but at 50 °C, 54% yield with high enantioselectivity for both isomers while **3a''** as the major product (dr: **3a''**:**3a'** = 5:1, Table 2, entry 9) was obtained. We decided to further optimize the reaction in CHCl₃ accordingly (Table 2, entries 10–13). A longer reaction time helped to increase the reaction yield (60 h, 76% yield entry 10). More steric hindered amine catalysts with bigger TES and TBDMS groups, **II** and **III**, were then probed and gave rise to the slight drop of enantioselectivity (Table 2, entries 11 and 12). A further optimization of reaction conditions found that the addition of additional 0.5 equiv **1a** into the reaction mixture in 4 portions significantly improved the reaction yield (83%, Table 2, entry 13). In order to improve the diastereoselectivity of this reaction, other cyclopentanes used in Trost's system were also tested in this reaction [45]. Unfortunately, the reactions proceeded slowly to afford the cycloaddition products in less than 10% yield.

We then selected the use of co-catalysts of Pd₂(dba)₃ and organocatalyst **I** in CHCl₃ at room temperature to evaluate the generality of this [3 + 2] annulation process by the variation of vinylcyclopropanes and enals (Table 3). The results exhibit that the synergistic catalyzed enantioselective [3 + 2] annulation process serves as a general approach to structurally chiral cyclopentanes bearing 3-consecutive stereogenic centers with high regio- and enantioselectivities. It was found that a wide range of aromatic α,β-unsaturated aldehydes can effectively participate in the process (Table 3, entries 1–12). The aromatic α,β-unsaturated aldehydes tethering electron-neutral, -with-

Table 2: The optimization of reaction conditions.^a

Entry	Amine cat.	Solvent	Yield (%) ^b	ee (3a' , 3a'') ^c	dr (3a' : 3a'') ^d
1	I	CH ₂ Cl ₂	48	80, 76	2:1
2	I	toluene	–	–	–
3	I	DCE	< 20	–	–
4	I	ether	< 20	–	–
5	I	CH ₃ CN	< 20	–	–
6	I	EtOAc	< 20	–	–
7	I	CHCl ₃	63	99, 83	1.7:1
8	I	THF	–	–	–
9 ^e	I	THF	54	90, 90	1:5
10 ^f	I	CHCl ₃	76	99, 83	1.7:1
11 ^f	II	CHCl ₃	76	96, 80	1.7:1
12 ^f	III	CHCl ₃	61	97, 82	1.7:1
13 ^{f,g}	I	CHCl ₃	83	99, 83	1.7:1

^aThe reaction was carried out with **1a** (36.8 mg, 0.2 mmol) and **2a** (26.4 mg, 0.2 mmol) in the presence of 5 mol % Pd₂(dba)₃, 12.5 mol % dppe and 30 mol % organic catalyst in 0.8 mL of solvent at rt for 48 h. ^bIsolated yields. ^cDetermined by HPLC analysis. ^dDetermined by ¹H NMR spectroscopy of the crude mixture. ^eThe reaction was run at 50 °C. ^fThe reaction was stirred for 60 h. ^gAdditional 0.5 equiv **1a** in 0.4 mL of CHCl₃ was added into the reaction mixture in 4 portions every 12 h.

**Scheme 2:** Single X-ray crystal structures of **7h'** and **7h''**.

drawing, and -donating substituents at the *para*-position of the phenyl ring gave good to high yields and excellent enantioselectivities for major isomer **3'** and minor **3'''** products, while the electronic effect on enantioselectivity is more pronounced for

minor **3''** (Table 3, entries 1–5). A similar trend is observed with the aromatic α,β -unsaturated aldehydes with electron-withdrawing at *meta*-position (Table 3, entries 6 and 7). Those with electron-withdrawing, and -donating groups at *ortho*-position

Table 3: Scope of the [3 + 2] annulation reaction of D–A cyclopropanes with enals.^a

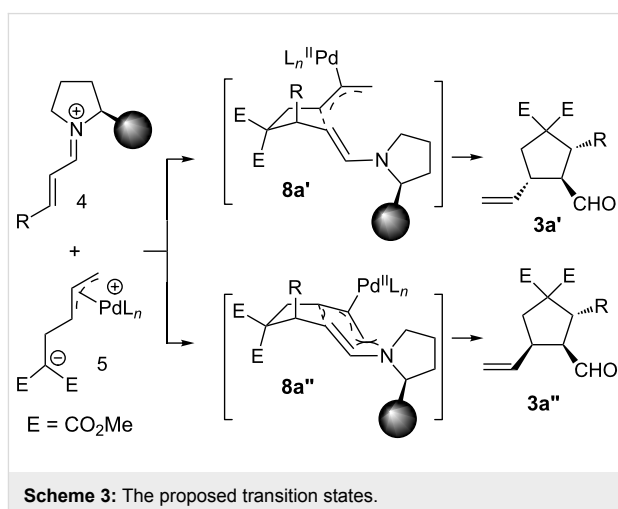
Entry	R ¹ , R ² , 3	Yield (%) ^b	ee (3a' , 3a'' , 3a''') ^c	dr (3a' : 3a'' : 3a''') ^d
1	H, Ph, 3a	83	99, 83, 99	1.7:1:0.7
2	H, 4-ClC ₆ H ₄ , 3b	86	97, 77, 99	2:1:0.4
3	H, 4-BrC ₆ H ₄ , 3c	84	94, 70, 99	2.5:1:0.5
4	H, 4-MeC ₆ H ₄ , 3d	70	99, 85, 99	2:1:0.7
5	H, 4-MeOC ₆ H ₄ , 3e	71	99, 99, 86	2:1:0.5
6	H, 3-FC ₆ H ₄ , 3f	85	97, 99, 76	2.5:1:0.6
7	H, 3-CF ₃ C ₆ H ₄ , 3g	61	90, 66, 99	3.3:1:0.5
8	H, 2-ClC ₆ H ₄ , 3h	65	98, 88, –	1.7:1:0.6
9	H, 2-BrC ₆ H ₄ , 3i	77	99, 90, –	2.5:1:0.8
10	H, 2-MeC ₆ H ₄ , 3j	72	99, 91, 99	3.3:1:0.4
11	H, 2-MeOC ₆ H ₄ , 3k	81	99, 99, 92	2:1:0.8
12	H, 2-furanyl, 3l	70	99, 97, 72	1.3:1:0.7
13 ^e	Ph, 4-ClC ₆ H ₄ , 3m	46	92, 86, 99	1.7:1:0.2

^aUnless specified, see experimental section for details. ^bIsolated yields. ^cDetermined by HPLC analysis. ^dDetermined by ¹H NMR spectroscopy of crude product. ^eThe reaction was run at 50 °C for 120 h.

furnished excellent enantioselectivities for both **3'** and **3''** products in cases studied (Table 3, entries 8–11). Moreover, the heteroaromatic furanyl α,β -unsaturated aldehyde **2l** can also be tolerated with good yield and 97% ee for the major product (Table 3, entry 12). More significantly, the more steric demanding D–A cyclopropane bearing a phenyl ring instead of H can effectively participate in the process to deliver the desired product with achieving an excellent level of enantioselectivity albeit a relatively low yield (Table 3, 46%, entry 13). It is noteworthy that although aliphatic enals also can engage in this [3 + 2] annulation reaction. Unfortunately, we could not separate them on chiral HPLC column for the determination of the enantioselectivity by all means we have attempted (data not shown).

The absolute configuration of cyclopentanes **3'** and **3''** were determined based on the derivatives **7h'** and **7h''** of **3h** (Scheme 2) [95].

We proposed two possible transition states (TS) **8a'** and **8a''** to rationalize the observed configurations (Scheme 3). The *trans*-C=C double bond in iminium ion **4** dictates the R group at pseudo axial position in the cyclic 5-membered ring TS **8a'** and **8a''**. This orientation avoids the A[1,3] strain induced by the catalyst-derived enamine. The Pd(II)- π 3 complex moiety at



pseudo axial and equatorial positions leads to respective TS **8a'** and **8a''**, while **8a'** is more stable due to the minimization of the A[1,3] interaction. Therefore, it is observed **3a'** produced from corresponding **8a'** as the major diastereomer whereas **3a''** as minor one.

Conclusion

We have developed a cooperative catalytic strategy for highly regio- and enantioselective [3 + 2] cycloaddition reactions of

vinylcyclopropanes with α,β -unsaturated aldehydes for the first time. The combination of a chiral iminium catalyst, which activates the C=C bond and blocks the C=O bond in enals, and a Lewis acid promoting to open the vinylcyclopropanes enables the annulation process to proceed with the challenging C=C bond. A high level of enantioselectivity could be achieved here. This previously unattainable [3 + 2] annulation transformation serves as a general approach to the preparation of new densely functionalized chiral cyclopentanes. This synergistic catalysis strategy holds great potentials for further exploration of new cycloaddition reactions involving enals and other D–A systems. The endeavor is being pursued in our laboratories.

Experimental

General procedure for the [3 + 2] annulation

A mixture of **1a** (0.2 mmol, 36.8 mg), **2a** (0.2 mmol, 26.4 mg), Pd₂(dba)₃ (0.01 mmol, 9.2 mg), dppe (0.025 mmol, 10 mg) and **I** (0.06 mmol, 18.5 mg) in 0.8 mL CHCl₃ was stirred for 60 h at rt. During this period, **1a** (0.1 mmol, 18.4 mg) in 0.4 mL CHCl₃ was added into the solution for total 4 times every 12 h, the mixture was purified by column chromatography on silica gel, eluted by petroleum ether/EtOAc = 20:1 to 10:1 to give the desired product **3a** in 83% yield as a colorless oil.

Supporting Information

Supporting Information File 1

Experimental and analytical data.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-12-127-S1.pdf>]

Acknowledgements

Financial support of this research from the program for Professor of Special Appointment (Eastern Scholar) at Shanghai Institutions of Higher Learning (No. 201226, H. L.), the National Science Foundation of China (No. 21372073, W. W.), the Fundamental Research Funds for the Central Universities and East China University of Science and Technology (start-up funds, H. L. and W. W.), and the China 111 Project (Grant B07023, H. L. and W. W.) is gratefully acknowledged.

References

- Danishefsky, S. *Acc. Chem. Res.* **1979**, *12*, 66–72. doi:10.1021/ar50134a004
- Verh , R.; de Kimpe, N. Synthesis and Reactivity of Electrophilic Cyclopropanes. In *The Carbonyl Group*; Rappoport, Z., Ed.; PATAI'S Chemistry of Functional Groups, Vol. 1 and 2; John Wiley & Sons: Chichester, United Kingdom, 1987; pp 445–564. doi:10.1002/0470023449.ch9
- Reissig, H.-U. Organic Synthesis via Cyclopropanes: Principles and Applications. In *Cyclopropyl Group*; Rappoport, Z., Ed.; PATAI'S Chemistry of Functional Groups, Vol. 1 and 2; John Wiley & Sons: Chichester, United Kingdom, 1987; pp 375–443. doi:10.1002/0470023449.ch8
- Salaun, J. Cyclopropane Derivatives and their Diverse Biological Activities. In *Small Ring Compounds in Organic Synthesis VI*; de Meijere, A., Ed.; Topics in Current Chemistry, Vol. 207; Springer: Berlin, Germany, 2000; pp 1–67. doi:10.1007/3-540-48255-5_1
- Reissig, H.-U.; Zimmer, R. *Chem. Rev.* **2003**, *103*, 1151–1196. doi:10.1021/cr010016n
- Yu, M.; Pagenkopf, B. L. *Tetrahedron* **2005**, *61*, 321–347. doi:10.1016/j.tet.2004.10.077
- Carson, C. A.; Kerr, M. A. *Chem. Soc. Rev.* **2009**, *38*, 3051–3060. doi:10.1039/b901245c
- Lebold, T. P.; Kerr, M. A. *Pure Appl. Chem.* **2010**, *82*, 1797–1812. doi:10.1351/PAC-CON-09-09-28
- Mel'nikov, M. Ya.; Budynina, E. M.; Ivanova, O. A.; Trushkov, I. V. *Mendeleev Commun.* **2011**, *21*, 293–301. doi:10.1016/j.mencom.2011.11.001
- Wang, Z. *Synlett* **2012**, *23*, 2311–2327. doi:10.1055/s-0032-1317082
- Tang, P.; Qin, Y. *Synthesis* **2012**, *44*, 2969–2984. doi:10.1055/s-0032-1317011
- Cavitt, M. A.; Phun, L. H.; France, S. *Chem. Soc. Rev.* **2014**, *43*, 804–818. doi:10.1039/C3CS60238A
- Schneider, T. F.; Kaschel, J.; Werz, D. B. *Angew. Chem., Int. Ed.* **2014**, *53*, 5504–5523. doi:10.1002/anie.201309886
- Stork, G.; Gregson, M. *J. Am. Chem. Soc.* **1969**, *91*, 2373–2374. doi:10.1021/ja01037a032
- Stork, G.; Grieco, P. A. *J. Am. Chem. Soc.* **1969**, *91*, 2407–2408. doi:10.1021/ja01037a059
- Stork, G.; Marx, M. *J. Am. Chem. Soc.* **1969**, *91*, 2371–2373. doi:10.1021/ja01037a031
- Stork, G.; Grieco, P. A. *Tetrahedron Lett.* **1971**, 1807–1810. doi:10.1016/S0040-4039(01)87467-0
- Corey, E. J.; Balanson, R. D. *Tetrahedron Lett.* **1973**, *14*, 3153–3156. doi:10.1016/S0040-4039(00)79797-8
- Danishefsky, S.; Dynak, J.; Hatch, E.; Yamamoto, M. *J. Am. Chem. Soc.* **1974**, *96*, 1256–1259. doi:10.1021/ja00811a068
- Danishefsky, S.; Tsai, M. Y.; Dynak, J. *J. Chem. Soc., Chem. Commun.* **1975**, 7–8. doi:10.1039/c3975000007
- Danishefsky, S.; McKee, R.; Singh, R. K. *J. Am. Chem. Soc.* **1977**, *99*, 7711–7713. doi:10.1021/ja00465a054
- de Meijere, A. *Angew. Chem., Int. Ed. Engl.* **1979**, *18*, 809–826. doi:10.1002/anie.197908093
- Wenkert, E.; Alonso, M. E.; Buckwalter, B. L.; Chou, K. J. *J. Am. Chem. Soc.* **1977**, *99*, 4778–4782. doi:10.1021/ja00456a040
- Piers, E.; Reissig, H.-U. *Angew. Chem., Int. Ed. Engl.* **1979**, *18*, 791–792. doi:10.1002/anie.197907911
- Wenkert, E. *Acc. Chem. Res.* **1980**, *13*, 27–31. doi:10.1021/ar50145a005
- Reissig, H.-U.; Hirsch, E. *Angew. Chem., Int. Ed. Engl.* **1980**, *19*, 813–814. doi:10.1002/anie.198008131
- Reissig, H.-U. *Tetrahedron Lett.* **1981**, *22*, 2981–2984. doi:10.1016/S0040-4039(01)81805-0
- Br ckner, C.; Reissig, H.-U. *J. Chem. Soc., Chem. Commun.* **1985**, 1512–1513. doi:10.1039/C39850001512
- Br ckner, C.; Reissig, H.-U. *Angew. Chem., Int. Ed. Engl.* **1985**, *24*, 588–589. doi:10.1002/anie.198505881

30. Grimm, E. L.; Zschiesche, R.; Reissig, H. U. *J. Org. Chem.* **1985**, *50*, 5543–5545. doi:10.1021/jo00350a022
31. Reißig, H.-U.; Reichelt, I.; Lorey, H. *Liebigs Ann. Chem.* **1986**, *1986*, 1924–1939. doi:10.1002/jlac.198619861113
32. Brueckner, C.; Reissig, H. U. *J. Org. Chem.* **1988**, *53*, 2440–2450. doi:10.1021/jo00246a010
33. Reissig, H.-U.; Holzinger, H.; Glomsda, G. *Tetrahedron* **1989**, *45*, 3139–3150. doi:10.1016/S0040-4020(01)80140-X
34. Reissig, H.-U. Donor-acceptor-substituted cyclopropanes: Versatile building blocks in organic synthesis. *Small ring compounds in organic synthesis III*; Top. Curr. Chem., Vol. 144; Springer: Berlin, Germany, 1988; pp 73–135. doi:10.1007/BFb0111229
35. Campbell, M. J.; Johnson, J. S.; Parsons, A. T.; Pohlhaus, P. D.; Sanders, S. D. *J. Org. Chem.* **2010**, *75*, 6317–6325. doi:10.1021/jo1010735
36. Parsons, A. T.; Johnson, J. S. *J. Am. Chem. Soc.* **2009**, *131*, 3122–3123. doi:10.1021/ja809873u
37. Benfatti, F.; de Nanteuil, F.; Waser, J. *Org. Lett.* **2012**, *14*, 386–389. doi:10.1021/ol203144v
38. de Nanteuil, F.; Serrano, E.; Perrotta, D.; Waser, J. *J. Am. Chem. Soc.* **2014**, *136*, 6239–6242. doi:10.1021/ja5024578
39. Benfatti, F.; de Nanteuil, F.; Waser, J. *Chem. – Eur. J.* **2012**, *18*, 4844–4849. doi:10.1002/chem.201103971
40. Parsons, A. T.; Smith, A. G.; Neel, A. N.; Johnson, J. S. *J. Am. Chem. Soc.* **2010**, *132*, 9688–9692. doi:10.1021/ja1032277
41. Xu, H.; Qu, J.-P.; Liao, S.; Xiong, H.; Tang, Y. *Angew. Chem., Int. Ed.* **2013**, *52*, 4004–4007. doi:10.1002/anie.201300032
42. de Nanteuil, F.; Waser, J. *Angew. Chem., Int. Ed.* **2011**, *50*, 12075–12079. doi:10.1002/anie.201106255
43. Xiong, H.; Xu, H.; Liao, S.; Xie, Z.; Tang, Y. *J. Am. Chem. Soc.* **2013**, *135*, 7851–7854. doi:10.1021/ja4042127
44. Shimizu, I.; Ohashi, Y.; Tsuji, J. *Tetrahedron Lett.* **1985**, *23*, 3825–3828. doi:10.1016/S0040-4039(00)89261-8
45. Trost, B. M.; Morris, P. J.; Sprague, S. J. *J. Am. Chem. Soc.* **2012**, *134*, 17823–17831. doi:10.1021/ja309003x
46. Sträter, N.; Lipscomb, W. N.; Klambunde, T.; Krebs, B. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 2024–2055. doi:10.1002/anie.199620241
47. Sawamura, M.; Sudoh, M.; Ito, Y. *J. Am. Chem. Soc.* **1996**, *118*, 3309–3310. doi:10.1021/ja954223e
48. Xu, H.; Zuend, S. J.; Woll, M. G.; Tao, Y.; Jacobsen, E. N. *Science* **2010**, *327*, 986–990. doi:10.1126/science.1182826
49. Krautwald, S.; Sarlah, D.; Schafroth, M. A.; Carreira, E. M. *Science* **2013**, *340*, 1065–1068. doi:10.1126/science.1237068
50. Allen, A. E.; MacMillan, D. W. C. *Chem. Sci.* **2012**, *3*, 633–658. doi:10.1039/c2sc00907b
51. Ahire, M. M.; Mhaske, S. B. *Angew. Chem., Int. Ed.* **2014**, *53*, 7038–7042. doi:10.1002/anie.201400623
52. Meazza, M.; Ceban, V.; Pitak, M. B.; Coles, S. J.; Rios, R. *Chem. – Eur. J.* **2014**, *20*, 16853–16857. doi:10.1002/chem.201404565
53. Logan, K. M.; Smith, K. B.; Brown, M. K. *Angew. Chem., Int. Ed.* **2015**, *54*, 5228–5231. doi:10.1002/anie.201500396
54. Zhu, S.; Zhang, J.; Chen, K.; Jiang, H. *Angew. Chem., Int. Ed.* **2015**, *54*, 9414–9418. doi:10.1002/anie.201504964
55. Peng, H.; Akhmedov, N. G.; Liang, Y.-F.; Jiao, N.; Shi, X. *J. Am. Chem. Soc.* **2015**, *137*, 8912–8915. doi:10.1021/jacs.5b05415
56. Wang, S.; Li, X.; Liu, H.; Xu, L.; Zhang, J.; Li, J.; Li, H.; Wang, W. *J. Am. Chem. Soc.* **2015**, *137*, 2303–2310. doi:10.1021/ja511143b
57. Shao, Z.; Zhang, H. *Chem. Soc. Rev.* **2009**, *38*, 2745–2755. doi:10.1039/b901258n
58. de Armas, P.; Tejedor, D.; García-Tellado, F. *Angew. Chem., Int. Ed.* **2010**, *49*, 1013–1016. doi:10.1002/anie.200906018
59. Zhong, C.; Shi, X. *Eur. J. Org. Chem.* **2010**, 2999–3025. doi:10.1002/ejoc.201000004
60. Rueping, M.; Koenigs, R. M.; Atodiresei, I. *Chem. – Eur. J.* **2010**, *16*, 9350–9365. doi:10.1002/chem.201001140
61. Zhou, J. *Chem. – Asian J.* **2010**, *5*, 422–434. doi:10.1002/asia.200900458
62. Loh, C. C. J.; Enders, D. *Chem. – Eur. J.* **2012**, *18*, 10212–10225. doi:10.1002/chem.201200287
63. Patil, N. T.; Shinde, V. S.; Gajula, B. *Org. Biomol. Chem.* **2012**, *10*, 211–224. doi:10.1039/C1OB06432K
64. Pellissier, H. *Tetrahedron* **2013**, *69*, 7171–7210. doi:10.1016/j.tet.2013.06.020
65. Du, Z.; Shao, Z. *Chem. Soc. Rev.* **2013**, *42*, 1337–1378. doi:10.1039/C2CS35258C
66. Ibrahim, I.; Córdova, A. *Angew. Chem., Int. Ed.* **2006**, *45*, 1952–1956. doi:10.1002/anie.200504021
67. Bihelovic, F.; Matovic, R.; Vulovic, B.; Saicic, R. N. *Org. Lett.* **2007**, *9*, 5063–5066. doi:10.1021/ol7023554
68. Usui, I.; Schmidt, S.; Breit, B. *Org. Lett.* **2009**, *11*, 1453–1456. doi:10.1021/ol9001812
69. Capdevila, M. G.; Benfatti, F.; Zoli, L.; Stenta, M.; Cozzi, P. G. *Chem. – Eur. J.* **2010**, *16*, 11237–11241. doi:10.1002/chem.201001693
70. Ikeda, M.; Miyake, Y.; Nishibayashi, Y. *Angew. Chem., Int. Ed.* **2010**, *49*, 7289–7293. doi:10.1002/anie.201002591
71. Yoshida, A.; Ikeda, M.; Hattori, G.; Miyake, Y.; Nishibayashi, Y. *Org. Lett.* **2011**, *13*, 592–595. doi:10.1021/ol1027865
72. Motoyama, K.; Ikeda, M.; Miyake, Y.; Nishibayashi, Y. *Eur. J. Org. Chem.* **2011**, 2239–2246. doi:10.1002/ejoc.201100044
73. Sinisi, R.; Vita, M. V.; Gualandi, A.; Emer, E.; Cozzi, P. G. *Chem. – Eur. J.* **2011**, *17*, 7404–7408. doi:10.1002/chem.201100729
74. Afewerki, S.; Ibrahim, I.; Rydfjord, J.; Breistein, P.; Córdova, A. *Chem. – Eur. J.* **2012**, *18*, 2972–2977. doi:10.1002/chem.201103366
75. Ma, G.; Afewerki, S.; Deiana, L.; Palo-Nieto, C.; Liu, L.; Sun, J.; Ibrahim, I.; Córdova, A. *Angew. Chem., Int. Ed.* **2013**, *52*, 6050–6054. doi:10.1002/anie.201300559
76. Krautwald, S.; Schafroth, M. A.; Sarlah, D.; Carreira, E. M. *J. Am. Chem. Soc.* **2014**, *136*, 3020–3023. doi:10.1021/ja5003247
77. Wang, H.; Yang, W.; Liu, H.; Wang, W.; Li, H. *Org. Biomol. Chem.* **2012**, *10*, 5032–5035. doi:10.1039/c2ob25682g
78. Sathishkannan, G.; Srinivasan, K. *Org. Lett.* **2011**, *13*, 6002–6005. doi:10.1021/ol2024423
79. Laugeois, M.; Ponra, S.; Ratovelomanana-Vidal, V.; Michelet, V.; Vitale, M. R. *Chem. Commun.* **2016**, *52*, 5332–5335. doi:10.1039/C6CC01775D
80. Goldberg, A. F. G.; O'Connor, N. R.; Craig, R. A.; Stoltz, B. M. *Org. Lett.* **2012**, *14*, 5314–5317. doi:10.1021/ol302494n
81. Zu, L.; Li, H.; Xie, H.; Wang, J.; Jiang, W.; Tang, T.; Wang, W. *Angew. Chem., Int. Ed.* **2007**, *46*, 3732–3734. doi:10.1002/anie.200700485
82. Wang, J.; Li, H.; Xie, H.; Zu, L.; Shen, X.; Wang, W. *Angew. Chem., Int. Ed.* **2007**, *46*, 9050–9053. doi:10.1002/anie.200703163
83. Vignola, N.; List, B. *J. Am. Chem. Soc.* **2004**, *126*, 450–451. doi:10.1021/ja0392566
84. Enders, D.; Wang, C.; Bats, J. W. *Angew. Chem., Int. Ed.* **2008**, *47*, 7539–7542. doi:10.1002/anie.200802532
85. Tan, B.; Shi, Z.; Chua, P. J.; Zhong, G. *Org. Lett.* **2008**, *10*, 3425–3428. doi:10.1021/ol801246m

86. Tan, B.; Chua, P. J.; Zeng, X.; Lu, M.; Zhong, G. *Org. Lett.* **2008**, *10*, 3489–3492. doi:10.1021/ol801273x
87. Hong, B.-C.; Nimje, R. Y.; Lin, C.-W.; Liao, J.-H. *Org. Lett.* **2011**, *13*, 1278–1281. doi:10.1021/ol1030487
88. Hong, B.-C.; Dange, N. S.; Hsu, C.-S.; Liao, J.-H.; Lee, G.-H. *Org. Lett.* **2011**, *13*, 1338–1341. doi:10.1021/ol200006e
89. Tan, B.; Candeias, N. R.; Barbas, C. F., III. *Nat. Chem.* **2011**, *3*, 473–477. doi:10.1038/nchem.1039
90. Albertshofer, K.; Tan, B.; Barbas, C. F., III. *Org. Lett.* **2012**, *14*, 1834–1837. doi:10.1021/ol300441z
91. Zhao, G.-L.; Ibrahem, I.; Dziedzic, P.; Sun, J.; Bonneau, C.; Córdova, A. *Chem. – Eur. J.* **2008**, *14*, 10007–10011. doi:10.1002/chem.200801082
92. Sun, W.; Zhu, G.; Wu, C.; Hong, L.; Wang, R. *Chem. – Eur. J.* **2012**, *18*, 6737–6741. doi:10.1002/chem.201200478
93. Remeš, M.; Veselý, J. *Eur. J. Org. Chem.* **2012**, 3747–3752. doi:10.1002/ejoc.201200334
94. Liu, G.; Shirley, M. E.; Van, K. N.; McFarlin, R. L.; Romo, D. *Nat. Chem.* **2013**, *5*, 1049–1057. doi:10.1038/nchem.1788
95. CCDC-1032267 and -1032268 for compounds **7h'** and **7h''**, respectively contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via <http://www.ccdc.cam.ac.uk>.

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/2.0>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (<http://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at:
doi:10.3762/bjoc.12.127