

Functionalized Pyridines: Synthesis and Toxicity Evaluation of Potential Insecticidal Agents against *Aphis craccivora*

Hany M. Abd El-Lateef,* Mai M. Khalaf, Mohamed Gouda, Mahmoud Kandeel, Amer A. Amer, Antar A. Abdelhamid, Ali M. Drar, and Mohamed A. Gad*



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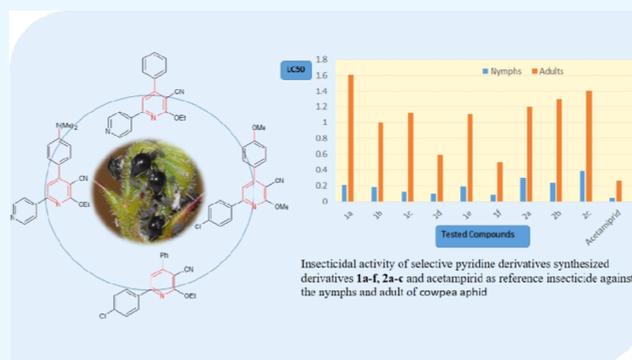


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ABSTRACT: Using ultrasound technology instead of traditional methods, some pyridine derivatives were prepared by a simple procedure via a four-component reaction of different aromatic aldehydes, acetyl aryl, sodium alkoxide, and malononitrile, and additionally, all prepared compounds were monitored for insecticidal activities toward nymphs and adults of cowpea aphid. Though a lot of insecticides are discovered as a novelty on the other hand, neonicotinoid compounds are reflected as the most affected insecticides against aphids and many other pests. Thus, some of the pyridine derivatives were chemically prepared as analogues to a large group of insecticides called neonicotinoids. Under laboratory conditions, the toxicity of these components was measured toward adults and nymphs of *Aphis craccivora*. With respect to the LC₅₀ values, components **1f**, **1d**, and **1c** have the utmost insecticidal bioactivity, with values of 0.080, 0.098, and 0.127 mg/L. This work covers the way to discover novel compounds for the prospective use as insecticidal representatives.



1. INTRODUCTION

More than 40% of the yearly food production is missing because of pest invasion, so an efficient pest management through employing a varied group of pesticides is a prerequisite for confronting pests and increasing crop production.¹ Due to the ability of pests for developing opposition to conventional insecticides, there is a continuing requirement for discovering and developing novel insecticides.² Pyridines exist widely in nature as vitamins, alkaloids, and coenzymes and own altered biological effects such as anti-microbial, antitumor, and antioxidant properties.^{3–8} On the other hand, the pyridine ring seems to be the main framework of many neonicotinoid insecticides (Figure 1).^{9,10} The great insecticidal effectiveness related to neonicotinoids has been demonstrated against numerous pests and homopteran insects, like aphids, which are one of the most dangerous pests that attack crops in Egypt, such as cotton, wheat, bean, maize, and others, causing huge damage to crops and a great economic loss; sucking plant sap and weakening it not only a reason to control aphids but the reason is aphids transmit viruses to plants; neonicotinoids were successful in controlling different types of aphids on different crop types.

However, many neonicotinoids have been banned in different countries around the world due to their toxicity to pollinators, especially honeybee. So, the search for new neonicotinoid analogues has become necessary, with the hope of finding a new one that has high toxicity against

pests and is safer for honeybees. Also, their relative safety concerning aquatic life and mammals is well-known.¹¹ Using ultrasound technology in organic designing has acquired great attention in recent years, such as a green and valuable technique in particular organic production paralleled with a classical technique.^{12–14} In view of the previously mentioned facts, the present work aims to prepare some pyridine derivatives by a simple procedure via a four-component reaction using ultrasound technology instead of traditional methods. Moreover, all prepared compounds will partition for insecticide effectiveness toward nymphs and adults of cowpea aphids.

2. MATERIALS AND TECHNIQUES

2.1. Chemistry. MP of the prepared products was analyzed using the Fisher–John mechanical apparatus. The infrared spectrum was measured by a KBr disc technique. The NMR spectrum was measured using a Bruker 400 MHz spectrometer by Si (CH₃)₄ (TMS) as a source of discernment and invention. Mass spectra were consummately measured using Jeol JMS-

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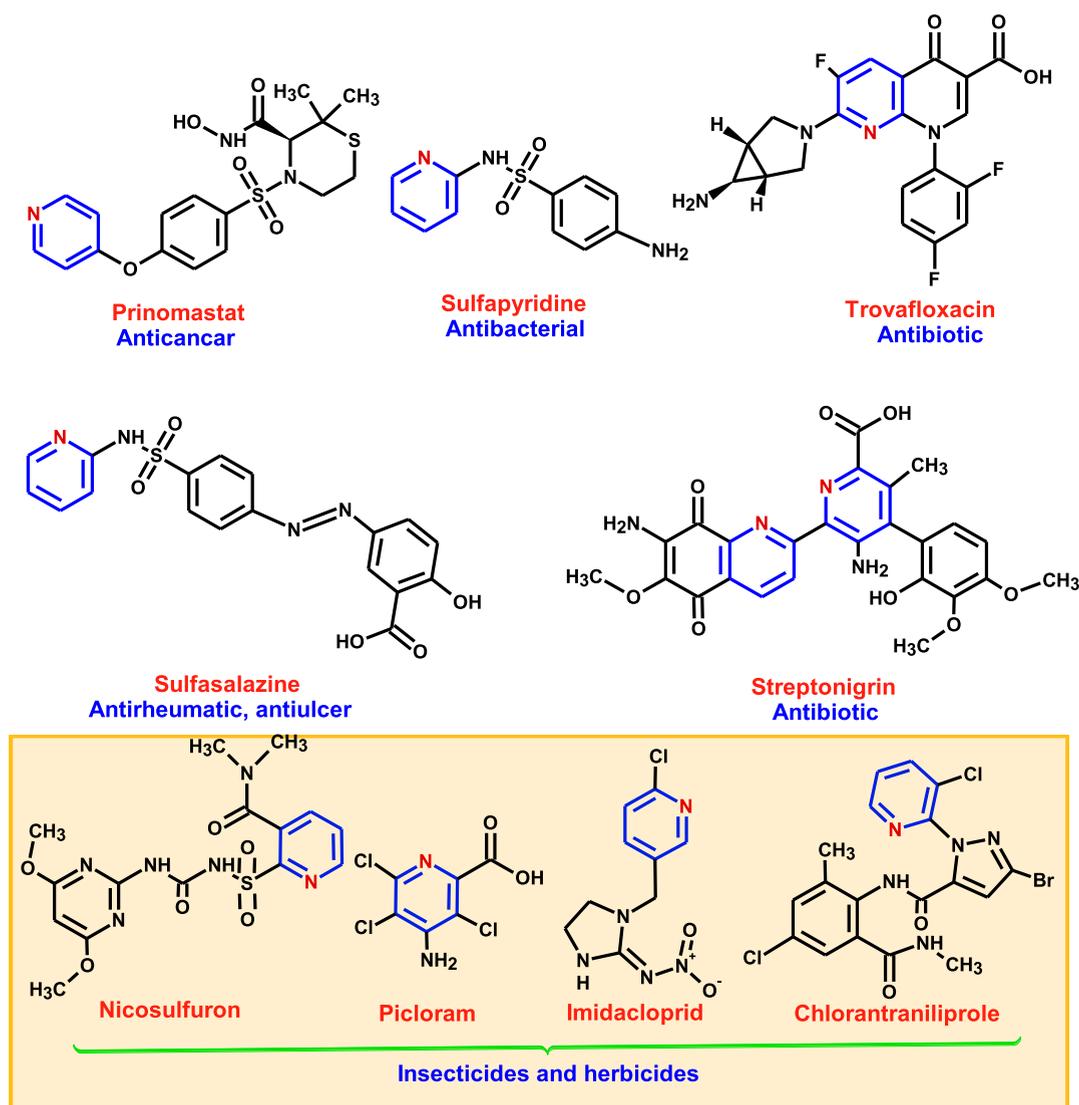


Figure 1. Some examples of drugs and insecticides containing pyridine scaffolds.

400. Ultrasonication was carried out in a Power sonic410 apparatus (DAIHAN LABTECH Co., LTD, KOERA). The electric supply was 20 V, AC 60 Hz; ultrasonic watts 500 W/2 A. serial no. 2016120709. All compounds 1a–j and 2a–d were synthesized by different methods.¹⁵

2.2. Insect Collection and Rearing. We found pyridine derivatives that were highly poisonous to both adults and nymphs of cowpea aphids. Cowpea aphid was collected from the insect laboratory, (PPRI), Agricultural Research Center and screened for insecticidal effectiveness of the objective products toward laboratory strain of adults and nymphs of cowpea aphid.

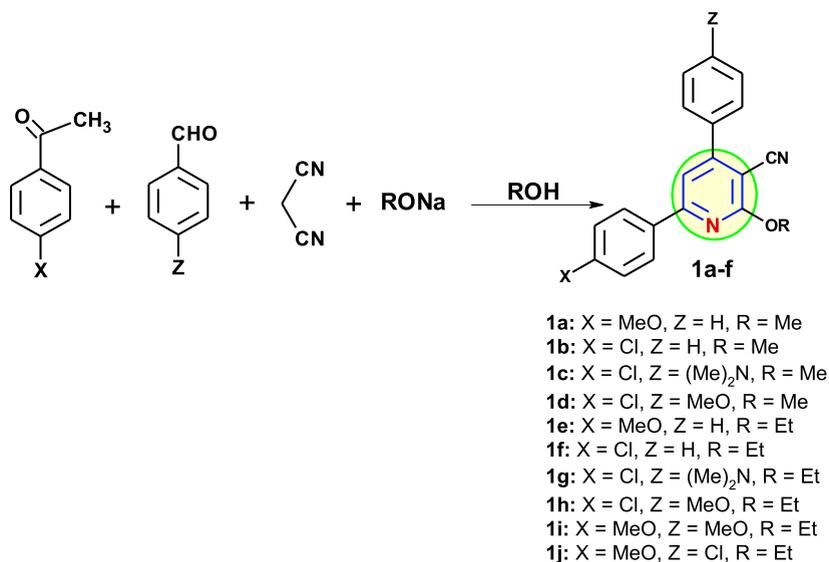
2.3. Laboratory Bioassay. The toxicological activities of target pyridine analogues were calculated using the leaf dipping technique under the same reported laboratory conditions.^{16–35} Thus, we prepared five concentrations of all compounds dissolved in drops of dimethylformamide with the use of 0.1% Tween-80 as a surfactant. An identical size of 30 adults and also 30 nymphs of cowpea aphid insects were dipped for 10 s in each concentration of each compound “repeated three times”. Target pests were kept at 25 °C to dry for approximately 1/2 h. In a control experiment, the insects

were immersed in distilled water with “0.1% Tween-80”. These procedures were carried out at 25 °C with “5%” relative humidity. The tested insects were transferred to glass jars containing water after they were dried. In addition, aphids were incapable to move and counted dead. Experimental results of all synthesized compounds were investigated by employing Abbott’s formula.^{36–38}

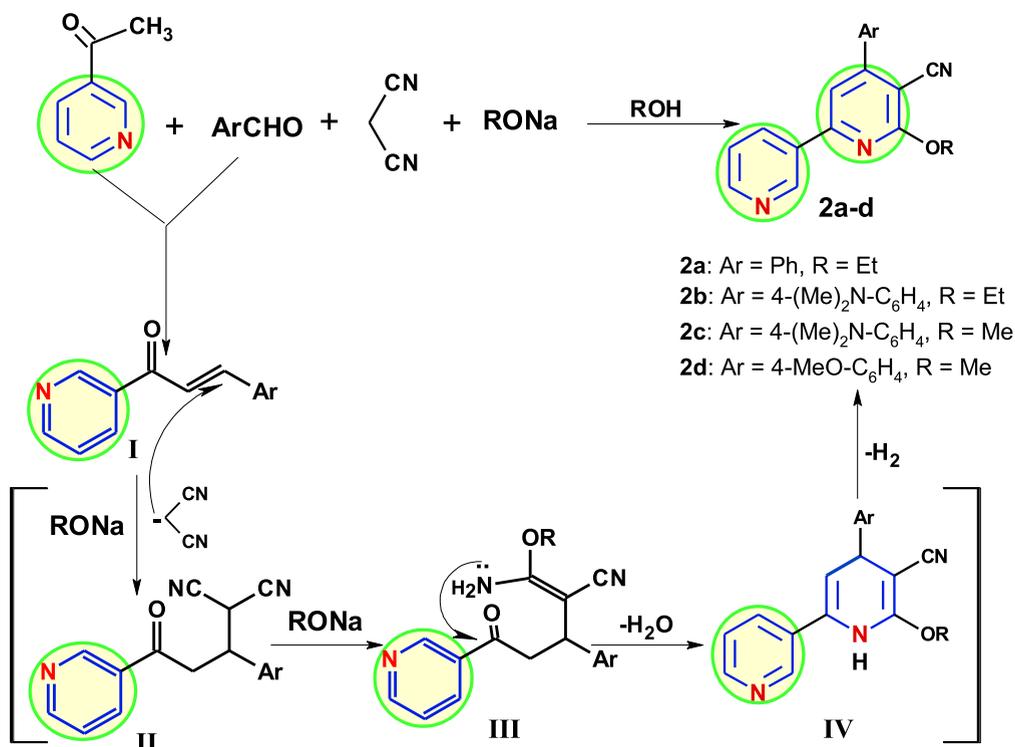
3. RESULTS AND DISCUSSION

3.1. Synthesis. The prepared products: 2-methoxy-6-(*p*-methoxyphenyl)-4-phenyl-3-cyanopyridine (1a), 6-(*p*-chlorophenyl)-2-methoxy-4-phenyl-3-cyanopyridine (1b), 6-(*p*-chlorophenyl)-4-[*p*-(dimethylamino)phenyl]-2-methoxy-3-cyanopyridine (1c), 6-(*p*-chlorophenyl)-2-methoxy-4-(*p*-methoxyphenyl)-3-cyanopyridine (1d), 2-ethoxy-6-(*p*-methoxyphenyl)-4-phenyl-3-cyanopyridine (1e), 6-(*p*-chlorophenyl)-2-ethoxy-4-phenyl-3-cyanopyridine (1f), 6-(*p*-chlorophenyl)-4-[*p*-(dimethylamino)phenyl]-2-ethoxy-3-cyanopyridine (1g), 6-(*p*-chlorophenyl)-2-ethoxy-4-(*p*-methoxyphenyl)-3-cyanopyridine (1h), 2-ethoxy-4,6-bis(*p*-methoxyphenyl)-3-cyanopyridine (1i), and 4-(*p*-chlorophenyl)-2-ethoxy-6-(*p*-methoxyphenyl)-3-cyanopyridine (1j) were synthesized by

Scheme 1. Synthesis of Compounds 1a–j



Scheme 2. Synthesis of Compounds 2a–d



simple and efficient procedures via a four-component reaction between acetyl aryl derivatives, aryl carbaldehydes, propane-dinitrile, and sodium methoxide in methanol and/or sodium ethoxide in ethanol at 50 °C under ultrasound irradiation (Scheme 1).

In light of these data, it was worthy to employ the same methodologies for the synthesis of heterocyclic compounds, which contain two pyridine rings in the same compound, hoping to obtain these compounds with enhanced biological activity applications. Hence, the equivalent procedures were extended for the preparation of 4-phenyl-6-ethoxy-2,3'-bipyridine-5-cyanide (**2a**), 4-[*p*-(dimethylamino)phenyl]-2,3'-bipyridine-6-ethoxy-5-cyanide (**2b**), 4-[*p*-(dimethylamino)-

phenyl]-6-methoxy-2,3'-bipyridine-5-cyanide (**2c**), and 4-(*p*-methoxyphenyl)-2,3'-bipyridine-6-methoxy-5-cyanide (**2d**) by the reaction of aryl carbaldehydes, 3-acetylpyridine, cyano acetonitrile, and RONA under the same reaction conditions (Scheme 2).

The reaction mechanism for the production of compounds **2a–d** was presumed through the formation of benzylideneacetophenone derivatives **I**, followed by the addition of the cyano acetonitrile anion at α , β -unsaturated carbonyl to generate intermediate **II** which reacted with RO[−] for forming adduct **III**, the adduct **III** cyclized via intramolecular cyclizing to produce **IV**, and subsequent dehydrogenation of **IV** produce the alkoxybipyridines-3-carbonitrile **2a–d** (Scheme 2). Chemical

structures of compounds **1a–j** and **2a–d** were determined using infrared spectroscopy, NMR spectral analysis, and melting points. For example, the IR spectrum of 6-ethoxy-4-phenyl-2,3'-bipyridine-5-cyanide (**2a**) does not have any absorption peak characteristic of =CO , while the band characteristic of CN was observed at 2223 cm^{-1} . Its ^1H NMR spectra show four peaks at 9.37, 8.63, 8.52, and 7.09 ppm characteristic of a pyridine nucleus. Also, it contains a multiplied signal at 7.93 attributed to 2H aromatic and 1H of the pyridine ring, and the aromatic protons show other multiplied signals at 7.57–7.09 ppm. ^{13}C NMR spectra of **2a** illustrate the peaks at δ 61.96 (exchangeable in dept-135) and 14.97 distinguishing ethoxide carbons besides aryl carbons at δ 164.52, 155.79, 152.80, 150.70, 150.33, 148.50, 136.52, 134.75, 134.45, 134.27, 130.99, 129.41, 124.05, 111.93, and 107.62.

3.1.1. General Method of Designing Functionalized Pyridines 1a–j & 2a–d. An equimolar of 0.01 mol acetophenone derivatives and aromatic aldehydes was mixed with RONA (0.011 moles Na in 30 mL of CH_3OH or $\text{CH}_3\text{CH}_2\text{OH}$, respectively) and stirred for 10 min at $25\text{ }^\circ\text{C}$. Then, cyanoacetonitrile (0.01 mol) was added, and the reaction mixture was allowed to irradiate through an ultrasonic generator at $60\text{ }^\circ\text{C}$ for 1–3 h. Irradiation was continued until the completion of the reaction (thin-layer chromatography monitoring), and the reaction mixture was refrigerated to $25\text{ }^\circ\text{C}$ and poured on 90 mL of H_2O . Then, the precipitated compounds were filtrated, washed many times with H_2O , and recrystallized from the appropriate solvents.

3.1.1.1. 2-Methoxy-6-(4-methoxyphenyl)-4-phenyl-3-cyanopyridine (1a). Yellow solid, Yield 79%; mp $181\text{--}183\text{ }^\circ\text{C}$ (reported $181\text{ }^\circ\text{C}$); IR: 2948, 2851 (CH-aliphatic), 2218 (NC); ^1H NMR: δ 8.245 (s, 2H, aromatic), 7.730 (s, 3H, 2H, arom + 1H, pyridine), 7.586 (s, 3H, aromatic), 7.077 (s, 2H, arom), 4.134 (s, 3H, OMe), 3.844 (s, 3H, OMe); ^{13}C NMR: δ 164.794, 162.017, 157.684, 156.662, 136.534, 130.372, 129.623, 129.243, 128.980, 115.861, 114.824, 113.244, 91.731, 55.898, 54.824; Anal. Calcd for $\text{C}_{20}\text{H}_{16}\text{N}_2\text{O}_2$: C, 75.93; H, 5.10; N, 8.86; Found: C, 76.00; H, 5.16; N, 8.80.

3.1.1.2. 6-(p-Chlorophenyl)-2-methoxy-4-phenyl-3-cyanopyridine (1b). Brown crystal, Yield 82%; mp $181\text{--}182\text{ }^\circ\text{C}$; IR: 2988, 2947 (CH-aliphatic), 2226 (NC); ^1H NMR: δ 8.312 (d, $J = 6.9\text{ Hz}$, 2H, aromatic), 7.864 (s, 1H, 1H, pyridine), 7.756 (s, 2H, aromatic), 7.601 (s, 5H, aromatic), 4.154 (s, 3H, OMe); ^{13}C NMR: δ 164.829, 157.060, 156.544, 136.253, 136.066, 130.549, 129.729, 129.390, 129.284, 129.056, 115.599, 114.352, 93.245, 55.046; Anal. Calcd for $\text{C}_{19}\text{H}_{13}\text{ClN}_2\text{O}$: C, 71.14; H, 4.08; N, 8.73; Found: C, 71.17; H, 4.18; N, 8.70.

3.1.1.3. 6-(p-Chlorophenyl)-4-[p-(dimethylamino)phenyl]-2-methoxy-3-cyanopyridine (1c). Brown crystal, Yield 78%; mp $286\text{--}287\text{ }^\circ\text{C}$; IR: 2214 (C \equiv N); ^1H NMR: Yield 78%; mp $286\text{--}287\text{ }^\circ\text{C}$; IR: 2995, 2944 (CH-aliphatic), 2214 (NC); ^1H NMR: δ 8.285 (s, 2H, aromatic), 7.767 (s, 1H, 1H, pyridine), 7.683 (s, 2H, aromatic), 7.597 (s, 2H, aromatic), 6.860 (s, 2H, aromatic), 4.118 (s, 3H, OMe), 3.017 (s, 6H, N(Me) $_2$); Analysis Calculated/found $\text{C}_{21}\text{H}_{18}\text{ClN}_3\text{O}$: C, 69.32/69.39; H, 4.99/4.88; N, 11.55/11.47.

3.1.1.4. 6-(p-Chlorophenyl)-2-methoxy-4-(p-methoxyphenyl)-3-cyanopyridine (1d). Yellowish powder, Yield 82%; mp $238\text{--}239\text{ }^\circ\text{C}$; IR: 2982, 2946 (CH-aliphatic), 2221 (NC); ^1H NMR: δ 8.306 (d, $J = 6\text{ Hz}$, 2H, aromatic), 7.818 (s, 1H, 1H, pyridine), 7.748 (d, $J = 6.2\text{ Hz}$, 2H, aromatic), 7.605 (d, $J = 5.9\text{ Hz}$, 2H, aromatic), 7.149 (d, $J = 6.3\text{ Hz}$, 2H, aromatic),

4.140 (s, 3H, OCH $_3$), 3.865 (s, 3H, OCH $_3$); ^{13}C NMR: δ 161.439, 156.621, 156.318, 136.158, 135.967, 130.689, 129.691, 129.366, 128.340, 114.845, 114.093, 92.732, 55.949, 54.977; Anal. Calcd for $\text{C}_{21}\text{H}_{18}\text{ClN}_3\text{O}$: C, 69.32; H, 4.99; N, 11.55; Found: C, 69.39; H, 4.88; N, 11.47.

3.1.1.5. 2-Ethoxy-6-(4-methoxyphenyl)-4-phenyl-3-cyanopyridine (1e). White powder, Yield 77%; mp $180\text{--}182\text{ }^\circ\text{C}$; IR: 2983, 2948, 2899 (CH-aliphatic), 2217 (NC); ^1H NMR: δ 8.214 (s, 2H, aromatic), 7.716 (s, 3H, 2H, aromatic + 1H, pyridine), 7.582 (s, 3H, aromatic), 7.070 (s, 2H, aromatic), 4.618 (q, $J = 6.9\text{ Hz}$, 2H, CH $_2$), 3.840 (s, 3H, OMe), 1.445 (t, $J = 7\text{ Hz}$, 3H, Me); ^{13}C NMR: δ 164.454, 161.984, 157.679, 156.679, 136.608, 130.333, 129.694, 129.557, 129.227, 128.968, 115.864, 114.827, 113.079, 91.795, 63.423, 55.890, 14.806; Analysis Calculated/found $\text{C}_{20}\text{H}_{15}\text{N}_2\text{O}_2$: C, 80.26/80.38; H, 5.01/5.24; N, 9.36/9.49.

3.1.1.6. 6-(4-Chlorophenyl)-2-ethoxy-4-phenyl-3-cyanopyridine (1f). White powder, Yield 75%; mp $150\text{--}152\text{ }^\circ\text{C}$; IR: 2985, 2974 (CH-aliphatic), 2221 (NC); ^1H NMR: δ 8.269 (s, 2H, aromatic), 7.791 (s, 1H, 1H, pyridine), 7.741 (s, 2H, aromatic), 7.591 (m, 5H, aromatic), 4.456 (q, $J = 3.9\text{ Hz}$, 2H, CH $_2$), 1.456 (t, $J = 3.9\text{ Hz}$, 3H, Me); ^{13}C NMR: δ 164.471, 157.043, 156.521, 136.315, 136.037, 130.498, 129.652, 129.372, 129.256, 129.037, 115.589, 114.150, 93.282, 63.669, 14.768; Analysis Calculated/found $\text{C}_{20}\text{H}_{15}\text{ClN}_2\text{O}$: C, 71.75/71.66; H, 4.52/4.39; N, 8.37/8.29.

3.1.1.7. 6-(4-Chlorophenyl)-4-[p-(dimethylamino)phenyl]-2-ethoxy-3-cyanopyridine (1g). White powder, Yield 73%; mp $211\text{--}212\text{ }^\circ\text{C}$; IR: 2972, 2919, 2849 (CH-aliphatic), 2217 (NC); ^1H NMR: δ 8.240 (s, 2H, aromatic), 7.725 (m, 5H, 4H, arom + 1H, pyridine), 6.847 (s, 2H, aromatic), 4.604 (q, $J = 7.7\text{ Hz}$, 2H, OCH $_2$), 3.010 (s, 6H, N(Me) $_2$), 1.436 (t, $J = 7.8\text{ Hz}$, 3H, Me); Analysis Calculated/found $\text{C}_{22}\text{H}_{20}\text{ClN}_3\text{O}$: C, 69.93/69.90; H, 5.33/5.23; N, 11.12/11.11.

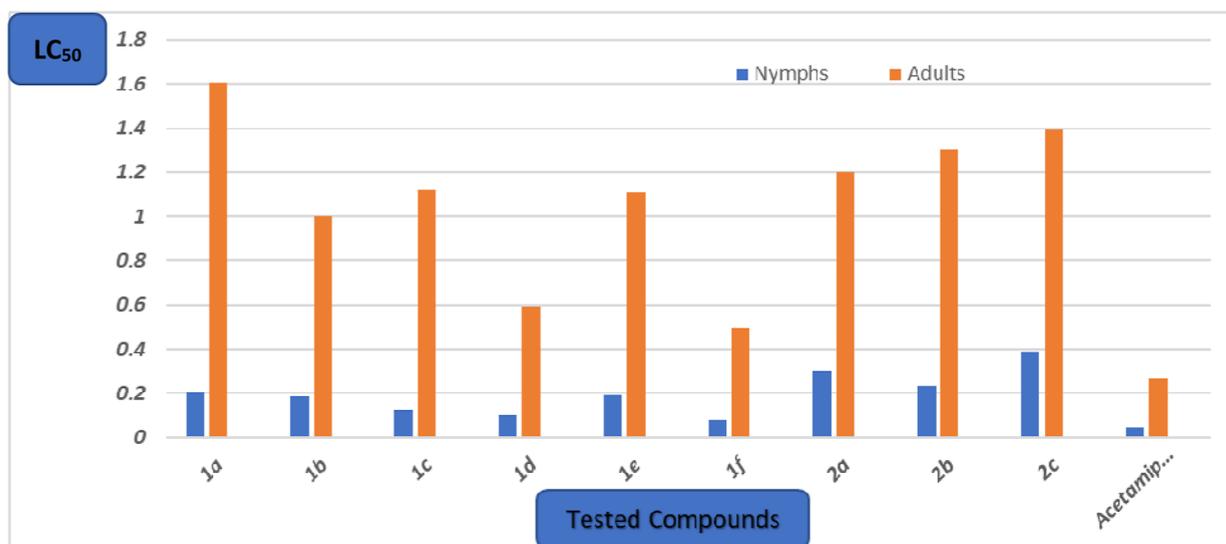
3.1.1.8. 6-(p-Chlorophenyl)-2-ethoxy-4-(p-methoxyphenyl)-3-cyanopyridine (1h). Yellowish powder, Yielding compound 68%; mp $162\text{--}163\text{ }^\circ\text{C}$; IR: 2956, 2847 (CH-aliphatic), 2218 (NC); ^1H NMR: δ 8.302 (d, $J = 8.6\text{ Hz}$, 2H, aromatic), 7.820 (s, 1H, 1H, pyridine), 7.757 (d, 2H, $J = 8.7\text{ Hz}$, aromatic), 7.615 (d, 2H, $J = 8.5\text{ Hz}$, aromatic), 7.156 (d, $J = 8.5\text{ Hz}$, 2H, aromatic), 4.634 (q, $J = 6.9\text{ Hz}$, 2H, CH $_2$), 3.865 (s, 3H, OMe), 1.448 (t, $J = 7\text{ Hz}$, 3H, Me); ^{13}C NMR: δ 164.611, 161.406, 156.626, 156.312, 136.203, 135.923, 130.660, 129.625, 129.359, 128.413, 115.896, 114.823, 113.900, 92.791, 63.575, 55.937, 14.785; Analysis Calculated/found $\text{C}_{21}\text{H}_{17}\text{ClN}_2\text{O}_2$: C, 69.14/69.05; H, 4.70/4.62; N, 7.68/7.51.

3.1.1.9. 2-Ethoxy-4,6-bis(p-methoxyphenyl)-3-cyanopyridine (1i). Yellow powder, Yield 65%; mp $186\text{--}167\text{ }^\circ\text{C}$; IR: 2951, 2836 (CH-aliphatic), 2217 (NC); ^1H NMR: δ 8.198 (s, 2H, aromatic), 7.699 (m, 3H, 2H, aromatic + 1H, pyridine), 7.098 (m, 2H, aromatic), 4.596 (q, $J = 6.3\text{ Hz}$, 2H, CH $_2$), 3.845 (s, 6H, 2OMe), 1.422 (t, $J = 6.6\text{ Hz}$, 3H, Me); ^{13}C NMR: δ 164.574, 161.905, 161.263, 157.455, 156.237, 130.547, 129.802, 129.491, 128.705, 116.157, 114.784, 112.807, 91.363, 63.324, 55.890, 14.814; Analysis Calculated/found $\text{C}_{22}\text{H}_{20}\text{N}_2\text{O}_3$ (360): C, 73.32/73.21; H, 5.59/5.41; N, 7.77/7.71.

3.1.1.10. 4-(p-Chlorophenyl)-2-ethoxy-6-(p-methoxyphenyl)-3-cyanopyridine (1j). White powder, Yield 69%; mp $230\text{--}231\text{ }^\circ\text{C}$; IR: 2940, 2841 (CH-aliphatic), 2218 (NC); ^1H NMR: δ 8.165 (s, 2H, aromatic), 7.715 (m, 2H, aromatic), 7.640 (m, 3H, 2H, aromatic + 1H, pyridine), 7.041 (s, 2H,

Table 1. Insecticidal Effectiveness of Components 1a–f, 2a–c, and Acetamiprid as an Insecticide Reference toward Cowpea Aphid Insects (Nymphs and Adults) after 1 Day of Management

comp.	nymphs			adults		
	LC ₅₀ (mg/L)	slope	toxic ratio	LC ₅₀ (mg/L)	sSlope	toxic ratio
1a	0.207	0.394 ± 0.275	0.217	1.606	0.571 ± 0.267	0.166
1b	0.185	0.288 ± 0.265	0.243	1.002	0.527 ± 0.271	0.266
1c	0.127	0.396 ± 0.271	0.354	1.121	0.522 ± 0.267	0.238
1d	0.098	0.322 ± 0.267	0.459	0.593	0.492 ± 0.271	0.450
1e	0.189	0.319 ± 0.264	0.238	1.107	0.574 ± 0.273	0.241
1f	0.080	0.397 ± 0.265	0.562	0.498	0.461 ± 0.275	0.536
2a	0.304	0.419 ± 0.272	0.148	1.198	0.251 ± 0.269	0.222
2b	0.233	0.429 ± 0.274	0.193	1.302	0.557 ± 0.261	0.205
2c	0.388	0.416 ± 0.271	0.115	1.399	0.573 ± 0.270	0.190
acetamiprid	0.045	0.381 ± 0.283	1	0.267	0.387 ± 0.283	1

**Figure 2.** Insecticidal activity of selective pyridine synthesized derivatives 1a–f, 2a–c, and acetamiprid as reference insecticides on aphids.

aromatic), 4.567 (q, $J = 6.3$ Hz, 2H, CH₂), 3.825 (s, 3H, OMe), 1.414 (t, $J = 6.2$ Hz, 3H, Me); ¹³C NMR: δ 164.387, 162.012, 157.773, 155.240, 135.415, 135.315, 130.842, 129.539, 129.242, 115.707, 114.775, 112.911, 91.654, 63.457, 55.855, 4.760; Analysis Calculated/found C₂₁H₁₇ClN₂O₂: C, 69.14/69.05; H, 4.70/4.59; N, 7.68/7.59.

3.1.1.11. 5-Cyano-4-phenyl-6-Ethoxy-2,3'-bipyridine (2a). White powder, Yield, 80%; mp 232–233 °C; IR: 2919, 2849 (CH-aliphatic), 2222 (NC); ¹H NMR: δ 9.371 (s, 1H, 1H, pyridine), 8.637 (d, $J = 4.6$ Hz, 1H, pyridine), 8.521 (m, 1H, pyridine nucleus), 7.929 (m, 3H, 2H, arom + 1H, pyridine), 7.574–7.515 (m, 3H, aromatic), 7.089 (s, 1H, pyridine nucleus), 4.505 (q, $J = 7.1$ Hz, 2H, CH₂), 1.407 (t, $J = 6.9$ Hz, 3H, Me); ¹³C NMR: δ 164.522, 155.788, 152.797, 150.700, 150.331, 148.502, 136.517, 134.750, 134.454, 134.274, 130.993, 129.413, 124.048, 111.933, 107.622, 61.963, 14.970; Analysis Calculated/found C₁₉H₁₅N₃O: C, 75.73/75.79; H, 5.02/4.99; N, 13.94/13.87.

3.1.1.12. 5-Cyano-4-[p-(dimethylamino)phenyl]-2,3'-bipyridine-6-ethoxy-5-cyanide (2b). White powder, Yield, 83%; mp 262–263 °C; IR: 2986, 2901, 2813 (CH-aliphatic), 2218 (NC); ¹H NMR: δ 9.375 (s, 1H, pyridine), 8.681 (s, 1H, pyridine), 8.529 (d, $J = 7.9$ Hz, 1H, pyridine), 7.784 (s, 1H, pyridine), 7.685 (d, $J = 8.6$ Hz, 2H, arom), 7.537 (d, $J = 8.5$ Hz, 1H, pyridine), 6.854 (d, $J = 8.9$ Hz, 2H, arom), 4.621 (q, $J =$

5.2 Hz, 2H, CH₂), 3.018 (s, 6H, N(Me)₂), 1.444 (t, $J = 5.3$ Hz, 3H, Me); Analysis Calculated/found C₂₁H₂₀N₄O: C, 73.23/73.18; H, 5.85/5.89; N, 16.27/16.25.

3.1.1.13. 5-Cyano-4-[p-(dimethylamino)phenyl]-6-methoxy-2,3'-bipyridine (2c). White crystal, Yield 80%; mp 264–265 °C; IR: 2947, 2858 (CH-aliphatic), 2216 (NC); ¹H NMR: δ 9.402 (s, 1H, pyridine), 8.694 (s, 1H, pyridine), 8.564 (d, $J = 7.7$ Hz, 1H, pyridine), 7.808 (s, 1H, pyridine), 7.694 (d, $J = 8.3$ Hz, 2H, aromatic), 7.549 (s, 1H, pyridine), 6.861 (d, $J = 8.0$ Hz, 2H, aromatic), 4.145 (s, 3H, OMe), 3.023 (s, 6H, N(CH₃)₂); Analysis Calculated/found C₂₀H₁₈N₄O: C, 72.71/72.68; H, 5.49/5.41; N, 16.96/17.00.

3.1.1.14. 5-Cyano-6-methoxy-4-(p-methoxyphenyl)-2,3'-bipyridine-5-cyanide (2d). Yellowish powder, Yield 82%; mp 241–142 °C; IR: 2954, 2846 (CH-aliphatic), 2220 (NC); ¹H NMR: δ 9.429 (s, 1H, pyridine), 8.709 (s, 1H, pyridine), 8.604 (d, $J = 7.6$ Hz, 1H, pyridine), 7.884 (s, 1H, pyridine), 7.767 (d, $J = 8.1$ Hz, 2H, aromatic), 7.569 (s, 1H, pyridine), 7.168 (d, $J = 6$ Hz, 2H, arom), 4.177 (s, 3H, OMe), 3.877 (s, 3H, OMe); ¹³C NMR: δ 165.092, 161.494, 156.693, 155.447, 151.552, 149.102, 135.266, 132.914, 130.749, 128.246, 124.246, 115.823, 114.871, 114.475, 93.150, 55.962, 55.100; Analysis Calculated/found C₁₉H₁₅N₃O₂: C, 71.91/71.79; H, 4.76/4.68; N, 13.24/13.16.

3.2. Insecticidal Bioefficacy Screening. The concentration which kills 50% of the aphid population is a more valuable point statistically, for comparative purposes, as shown in Table 1. In order to estimate the LC₅₀ values and “95%” educible limits of upper and lower confidence, standard error, slope, chi-square, and correlation coefficient, the mortality data of aphids were analyzed using probity analysis through some statistics (LDP-line) software.

3.2.1. Toxicological Activity toward Nymphs of Cowpea Aphids after 1 Day of Treatment. From Table 1 and Figure 2, it can be seen that all tested 11 analogues exhibit a moderate or feeble toxicological effect toward cowpea aphids (nymphs) whose LC₅₀ values varied from 0.080 to 0.385 mg/L after 24 h of treatment, and some of them possess toxicological activity in close to a reference acetamiprid. In particular, the LC₅₀ values of compounds 1a–f, 2a–c, and acetamiprid are 0.207, 0.185, 0.127, 0.098, 0.189, 0.080, 0.304, 0.233, 0.385, and 0.045 mg/L, respectively; among all tested compounds, pyridine derivatives 1f and 1d possess the highest insecticidal activity with LC₅₀ values of 0.080 and 0.098 mg/L, respectively.

3.2.2. Toxicological Activity toward Adults of Cowpea Aphids after 1 Day of Treatment. The insecticidal properties of compounds 1a through f, 2a through c, and acetamiprid against adult insects of cowpea aphids after 1 day of treatment possess moderate to feeble toxicological effectiveness toward the cowpea aphid (adults) with LC₅₀ values varied from 0.498 to 1.606 mg/L after 1 day of treatment, and only one of them possesses toxicological activity in close to that of acetamiprid reference. In particular, the LC₅₀ values of compounds 1a–f, 2a–c, and acetamiprid are 1.606, 1.002, 1.121, 0.593, 1.107, 0.498, 1.198, 1.302, 1.399, and 0.267 mg/L, respectively. Among all tested compounds, pyridine derivatives 1f and 1d possess the highest toxicological effectiveness with LC₅₀ values of 0.498 and 0.593 mg/L, respectively. In comparison with other tested compounds, compound 1f showed the highest toxicity and is close in activity to a reference insecticide.

3.3. Structure–Activity Relationship. It is evident from the results in Table 1 and Figure 2 that all components exhibited higher insecticidal activity due to the presence of pyridine rings, and their substituents attached to pyridine rings lead to a variety in toxicity; all analogues showed higher activity toward nymphs of cowpea aphids than the adults of cowpea aphids after 1 day of treatment; adults possess more resistance toward the target synthesized compounds than the nymphs' ones. Due to the inclusion of the 4-chlorophenyl group, two cyano groups, phenyl moiety, and an ethoxy group in its structure, compound 1f has the highest activity in this series. Due to the inclusion of the 4-chlorophenyl group, cyano groups, and phenyl moiety in its structures, compound 1d exhibits strong toxicological activity toward adults and nymphs of cowpea aphids. Additionally, compounds 1c and 1b of the tetrahydroisoquinoline family are more hazardous than the analogue 1e, which may be because of the 4-chlorophenyl moiety in their structures. Compound 1a may have insecticidal properties. The compounds 1a–f demonstrate that the occurrence of the chlorophenyl group might reflect better effectiveness than the pyridine group. On considering the toxicity line and slope, we noticed that the slope increased in the following order 1f > 1d > 1c > 1b > 1e > 1a > 2b > 2a > 2c. This order revealed the toxicity response of the tested cowpea aphid which showed variation in response toward the target insecticidal compounds.

4. CONCLUSIONS

In this article, it was worth designing new pyridine derivatives that act as an insecticide. Conferring to the poisonous value in Table 1 and Figure 2 through the employment of a computerized program of regression analysis, LC₅₀ and slope values of the objective components were counted and described as ppm unit. The insecticidal activities of components 1a–f, 2a, b, 3, 4a, and b in contradiction to the adults and nymph insects of aphids were compared with that of acetamiprid as a reference insecticide against cowpea aphid. In this section, the structure–activity relationship was established, in which component 3 is more active in contradiction to adults and nymphs of cowpea aphids than all the other derivative pyridines. The great effectiveness of component 3 may be owing to the presence of chlorophenyl moieties, two carbonitrile groups, and phenyl moieties in its configuration. This study's results encourage further testing of these promising compounds and the exploitation of their toxicological activity to produce new chemical insecticides against sucking pests.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.3c03831>.

IR spectra, ¹H NMR spectra, ¹³C NMR spectra, and elemental analysis of pyridine derivatives (PDF)

■ AUTHOR INFORMATION

Corresponding Authors

Hany M. Abd El-Lateef – Department of Chemistry, College of Science, King Faisal University, Al-Ahsa 31982, Saudi Arabia; Department of Chemistry, Faculty of Science, Sohag University, Sohag 82524, Egypt; Email: hmahmed@kfu.edu.sa, Hany_shubra@science.sohag.edu.eg

Mohamed A. Gad – Research Institute of Plant Protection, Agriculture Research Center, Giza 12112, Egypt; orcid.org/0000-0001-7850-7520; Email: mohamedgad87@arc.sci.eg

Authors

Mai M. Khalaf – Department of Chemistry, College of Science, King Faisal University, Al-Ahsa 31982, Saudi Arabia; Department of Chemistry, Faculty of Science, Sohag University, Sohag 82524, Egypt

Mohamed Gouda – Department of Chemistry, College of Science, King Faisal University, Al-Ahsa 31982, Saudi Arabia

Mahmoud Kandeel – Department of Biomedical Sciences, College of Veterinary Medicine, King Faisal University, Al-Ahsa 31982, Saudi Arabia; Department of Pharmacology, Faculty of Veterinary Medicine, Kafrelsheikh University, Kafrelsheikh 33516, Egypt

Amer A. Amer – Department of Chemistry, Faculty of Science, Sohag University, Sohag 82524, Egypt; orcid.org/0000-0001-7112-9918

Antar A. Abdelhamid – Department of Chemistry, Faculty of Science, Sohag University, Sohag 82524, Egypt; Chemistry Department, Faculty of Science, Albaha University, Albaha 1988, Saudi Arabia

Ali M. Drar – Research Institute of Plant Protection, Agriculture Research Center, Giza 12112, Egypt

Complete contact information is available at:

<https://pubs.acs.org/10.1021/acsomega.3c03831>

Notes

The authors declare no competing financial interest. The raw/processed result produced in this work is existing upon request from the agreeing author.

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