

SCIENTIFIC REPORTS



OPEN

Origin of reduced magnetization and domain formation in small magnetite nanoparticles

Received: 03 January 2017

Accepted: 07 March 2017

Published: 10 April 2017

Zlatko Nedelkoski¹, Demie Kepaptsoglou², Leonardo Lari¹, Tianlong Wen^{3,4}, Ryan A. Booth³, Samuel D. Oberdick³, Pedro L. Galindo⁵, Quentin M. Ramasse², Richard F. L. Evans¹, Sara Majetich³ & Vlado K. Lazarov¹

The structural, chemical, and magnetic properties of magnetite nanoparticles are compared. Aberration corrected scanning transmission electron microscopy reveals the prevalence of antiphase boundaries in nanoparticles that have significantly reduced magnetization, relative to the bulk. Atomistic magnetic modelling of nanoparticles with and without these defects reveals the origin of the reduced moment. Strong antiferromagnetic interactions across antiphase boundaries support multiple magnetic domains even in particles as small as 12–14 nm.

Magnetic nanoparticles (NPs) are expected to be single magnetic domain below a critical size, typically on the order of the domain wall width for the corresponding bulk material. Single domain particles should have the maximum magnetic moment per volume, which is desirable for their use in ferrofluids^{1,2}, magnetic separation³, contrast agents^{4,5} for magnetic resonance imaging and magnetic hyperthermia^{6,7}, all of which use iron oxide NPs. However, many researchers observe reduced magnetization, relative to that of the bulk. The reduction has been attributed to surface spin disorder^{8,9} or to variations in crystallinity, as in the case of low temperature aqueous preparation methods¹⁰. However, increasing crystallinity and reducing surface roughness does not necessarily solve this problem^{11,12}. Indeed, in this work we demonstrate that even high quality magnetite NPs can have dramatic differences in their magnetic properties. High resolution electron microscopy is used to identify subtle differences in three kinds of NPs, and atomistic spin calculations are used to reveal the origin of the reduced magnetization and its anomalous temperature dependence.

By a combination of measurements and modelling we demonstrate that antiphase domain boundaries in NPs can greatly reduce their magnetic moment. APB defects are seen in substrate supported thin film growth of magnetite^{13–15}, and also in core-shell NPs¹⁶. Here we demonstrate that the strong antiferromagnetic super-exchange interactions across the APBs significantly decrease the magnetisation of the NPs due to formation of multiple magnetic domains even in NPs below 15 nanometres.

While the detection of APBs requires sophisticated atomic resolution electron microscopy, we show that besides the reduced M_s , zero field cooled magnetization measurements can be used to easily identify types of NPs with a high density of these defects, and therefore lower magnetization compared with that of single crystal NPs.

Results and Discussion

The experiments compared 12–14 nm diameter iron oxide nanoparticles synthesized by three well-established methods that were pioneered by the groups of Sun¹⁷, Colvin¹⁸ and Hyeon¹⁹. Hereafter, they will be referred to as Sun, Colvin and Hyeon NPs. All three synthesis methods involve high temperature inert atmosphere decomposition in organic solvents, leading to monodisperse, highly crystalline, spherical particles coated with surfactants. Large field of view electron microscopy images for all three sets of NPs are presented in Fig. 1a–c and Supplementary Fig. S1. The size distribution of the three samples is similar: Sun 12.3 ± 2.9 nm, Colvin 13.7 ± 1.6 nm and Hyeon 14.2 ± 2.0 nm.

¹Department of Physics, University of York, Heslington, York, YO10 5DD, UK. ²SuperSTEM, Sci-Tech Daresbury Campus, Daresbury, WA4 4AD, UK. ³Physics Department, Carnegie Mellon University, Pittsburgh, Pennsylvania 15213, USA. ⁴State Key Laboratory of Electronic Thin Films and Integrated Devices, University of Electronic Science and Technology of China, Chengdu, China. ⁵Department of Computer Science and Engineering, Universidad de Cádiz, 11510 Puerto Real, Spain. Correspondence and requests for materials should be addressed to S.M. (email: sara@cmu.edu) or V.K.L. (email: vlado.lazarov@york.ac.uk)

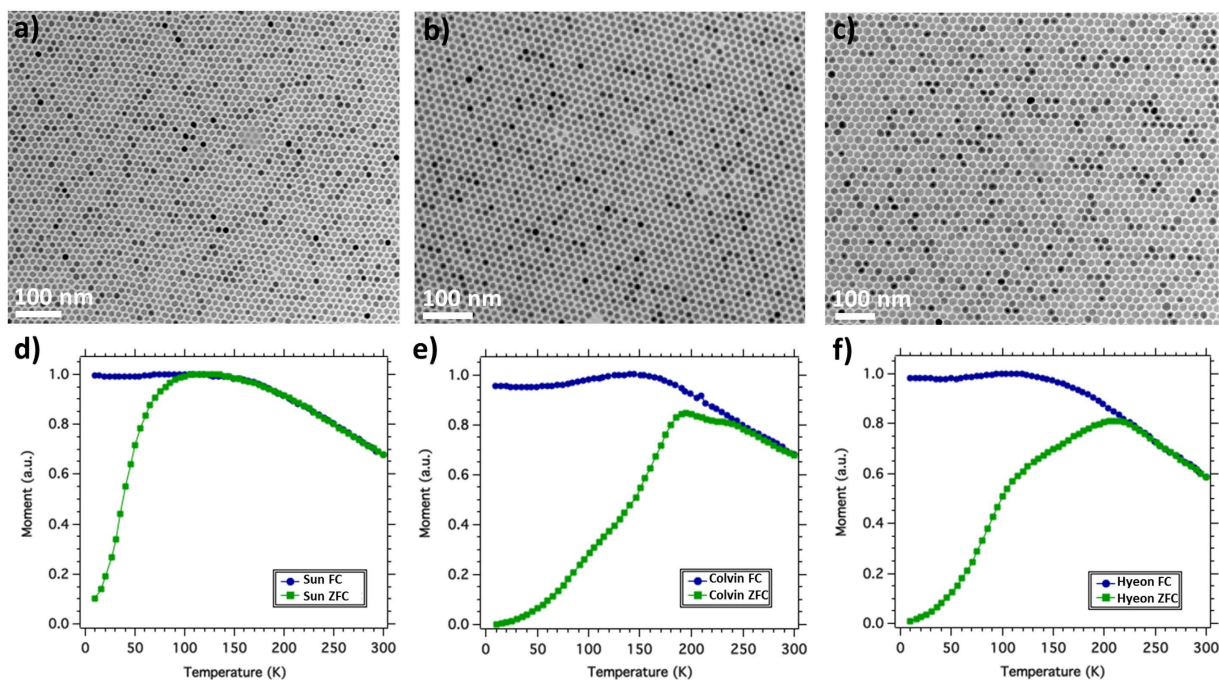


Figure 1. Conventional TEM images of (a) Sun, (b) Colvin, (c) Hyeon nanoparticles. Zero Field Cooled (ZFC) and Field Cooled (FC) magnetization curves for (d) Sun, (e) Colvin, (f) Hyeon nanoparticles.

The structural and chemical composition of the particles was characterized by a combination of selected area electron diffraction (SAED) and electron energy loss spectroscopy (EELS). In some syntheses of magnetite (Fe_3O_4) other phases are formed initially, then transformed through oxidation¹⁶. In contrast, as shown in Supplementary Fig. S2, all sets of the studied NPs in this work are single phase magnetite NPs. For more detailed chemical information, EELS measurements were performed in an aberration corrected scanning transmission electron microscope (STEM). The small size of the aberration corrected electron probe ($<1 \text{ \AA}$ in the conditions used for this experiment) reveals the local chemical composition within the nanoparticles. Furthermore, comparison of O K and Fe $L_{2,3}$ edge fine structures averaged over selected NPs with those acquired from a known 'bulk' Fe_3O_4 sample, acquired under identical optical conditions (Supplementary Fig. S3) confirms that the chemistry of all three sets nanoparticles is Fe_3O_4 like (Supplementary EELS discussion). In addition, atomically-resolved 2D EELS measurements for all three sets NPs (Supplementary Figs S4, S5 and S6) consistently showed no change in the Fe $L_{2,3}$ white-line intensity ratios across the observed NPs. This confirms the chemical uniformity of the particles and the absence of a core-shell structure.

In contrast to the apparent structural and chemical uniformity of the three sets of nanoparticles, as determined by TEM, SAED and EELS, the magnetic properties are strikingly different. The Sun particles have a specific saturation magnetization of $81 \pm 12 \text{ emu/g}$ at 10 K ($71 \pm 1 \text{ emu/g}$ at 300 K), values that are roughly 10% below that of the bulk. In comparison, the values for the Colvin particles were $37 \pm 1 \text{ emu/g}$ at 10 K ($36 \pm 1 \text{ emu/g}$ at 300 K), and for the Hyeon particles $39 \pm 4 \text{ emu/g}$ at 10 K ($40 \pm 4 \text{ emu/g}$ at 300 K). These Colvin and Hyeon particles have approximately half the magnetization of the Sun particles of comparable size. When hysteresis loops were measured after cooling in a large magnetic field, no exchange bias was observed.

Further details about the magnetic behaviour of the NPs are revealed by the zero field cooled (ZFC) and field cooled (FC) magnetization curves (Fig. 1d–f). The ZFC curve of the Sun particles shows a peak at 110–120 K, and a gradual decay at higher temperature. The Colvin particles also show a shoulder at ~ 110 –120 K, plus a peak at 180 K and another feature near 200 K. The Hyeon particles have a shoulder near 110 K and a peak slightly above 200 K. The peak in the ZFC magnetization curve is associated with the blocking temperature T_B of the nanoparticles, which is proportional to the anisotropy: $KV \sim 25 k_B T_B$, where K is the anisotropy energy density, V is the particle volume and k_B is the Boltzmann constant. With this interpretation, the Colvin and Hyeon particles would have higher K than the Sun particles, yet they have lower magnetization. ZFC curves with anomalous features like those in Fig. 1e,f have been reported for NPs made by many other methods including aqueous co-precipitation²⁰. ZFC curves like that of Fig. 1d have been reported for larger aqueous method particles²¹.

The puzzling magnetic behaviour in the ZFC curves cannot be explained by the conventional TEM and SAED observations alone, but requires deeper structural analysis. STEM imaging was used to determine the atomic structure and chemistry of selected nanoparticles, and to identify structural characteristics typical to each preparation method. The contrast dependence of high angle annular dark field (HAADF) imaging on the atomic number Z as $\sim Z^{1.7}$ enables direct identification of both iron tetrahedral (Fe_A) and octahedral (Fe_B) atomic columns in the NPs. Figure 2 shows atomically-resolved HAADF images of representative Sun, Colvin and Hyeon NPs. In all three cases the bulk-like magnetite structural ordering extends to the particle surfaces, and no core-shell structure is observed.

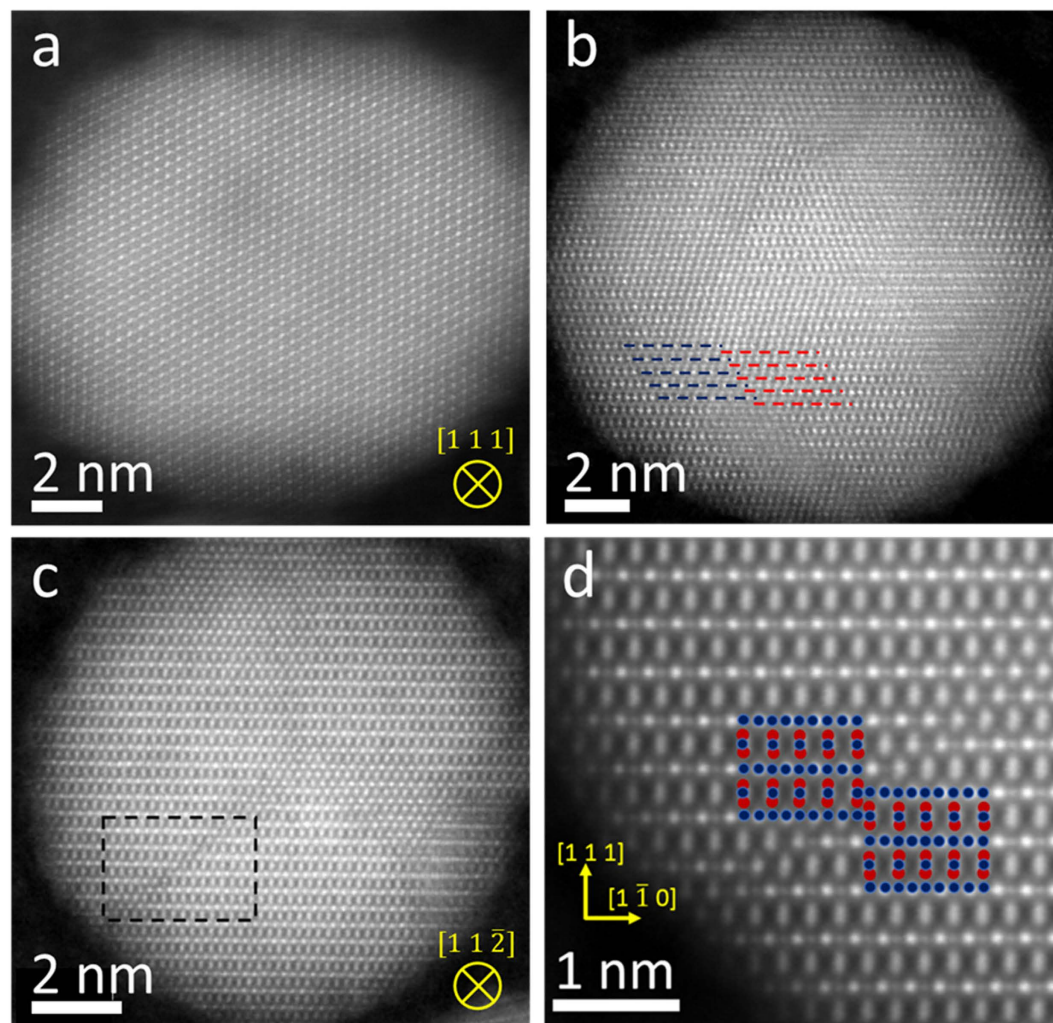


Figure 2. Atomically resolved HAADF STEM images of representative (a) Sun NP viewed along the [111] zone axis, (b) Colvin NP viewed along the [114] zone axis. (c) Hyeon NP along the [11-2] zone axis, obtained by rigid registration of a stack of images of the same area recorded in quick succession (resulting in high signal-to-noise and precision in the image). Dashed lines in (b) and (c) indicate the presence of the structural defects. (d) Magnified view of the dashed area shown in (c) with overlaid structural model emphasizing the defect region.

However, upon closer inspection, the HAADF images reveal certain structural characteristics that may be related to the observed differences in magnetic behavior. The Sun nanoparticles, such as that shown in Fig. 2a are predominantly spherical with no strain (Supplementary Fig. S7) or disorder across the entire particle, as in metallic NPs²². In contrast to the Sun NPs, structural defects are observed in both Colvin and Hyeon NPs. The prevalence of structural defects in Colvin and Hyeon NPs and absence of defects in Sun NPs is demonstrated by dark-field TEM images covering hundreds of NPs (Supplementary Figs S8, S9 and S10). In addition, the presence/absence of structural defects is illustrated in the atomic resolution HAADF STEM images for the three sets, presented in Supplementary Figs S11, S12 and S13. Figure 2b,c shows that the translational symmetry is broken and structural domains are formed. Most of the observed structural defects are antiphase domain boundaries (APBs), which can be described by fractional unit cells shift of $\frac{1}{4}a_0[110]$. In Fig. 2c the selected region shows the atomic arrangement of an APB with $\frac{1}{4}a_0 \langle 110 \rangle$ viewed along the [11-2] zone axis. The mismatch across the boundary (e.g. between octahedral planes across the boundary) is outlined. Figure 2d shows the break in the structural symmetry of the cationic (111) plane by a crystallographic shift of the octahedral atomic planes by $a_0\sqrt{\frac{1}{12}}$, imaged along the [11-2] viewing direction.

APBs, extensively studied in magnetite thin films^{13,23–27}, are correlated with anomalous properties such as very high magnetic saturation fields²³ and negative magnetoresistance^{15,28}. However, no direct evidence of APBs in single crystal magnetite NPs has been demonstrated. Recent work on the formation energy of highly stable APB defects²⁹ could explain why APBs are stable even in NPs. A lattice vector shift of $\frac{1}{4}a_0 \langle 110 \rangle$ creates an APB (Fig. 3a,b) with a very low formation energy (0.1 J/m²). Figure 3b illustrates how the 90° (Fig. 3a) Fe_B-O-Fe_B bulk-like ferromagnetic super-exchange interaction is changed to 180° antiferromagnetic interaction due to the lattice shift caused by the APB. APBs with different lattice shift vectors result in similarly increased AFM

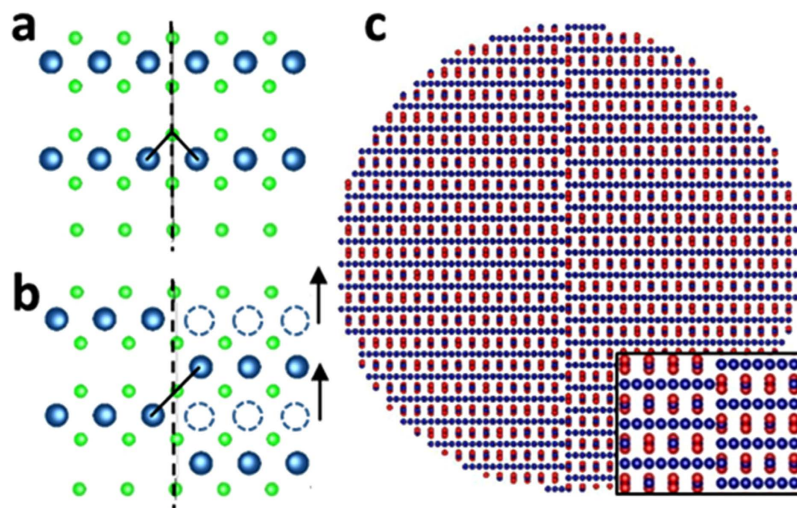


Figure 3. (a) [001] view of a (001) $\text{Fe}_B\text{-O}$ lattice plane in bulk magnetite. (b) The presence of a $\frac{1}{4}a_0 \langle 110 \rangle$ APB (dashed line) leads to shifted right hand side for $\frac{1}{4}a_0 \langle 110 \rangle$ (in-plane, arrows represent the shift vector), changing the $\text{Fe}_B\text{-O-Fe}_B$ angle from 90° as shown in (a) to 180° . The oxygen sub-lattice is invariant under the shift. (c) 10 nm nanoparticle model with single $\frac{1}{4}a_0 \langle 110 \rangle$ APB shown along the [11-2] zone axis. Tetrahedral Fe_A atoms are coloured in red, octahedral Fe_B in blue and oxygen atoms (not shown in (c)) in green. Drawing of the structural models was generated in VESTA³⁹ software.

interactions. Here we consider only the APBs with $\frac{1}{4} a_0 \langle 110 \rangle$ shift for two reasons: a) these APBs are experimentally observed in the studied NPs, and b) the formation energies of other type APBs are an order of magnitude larger, and are therefore much less likely to form in NPs.

These experimental findings provide a basis for an atomic level understanding of the different magnetic behavior, based on the strong correlation between the macroscopically measured magnetic properties of NPs and the presence of the APB defects observed by atomic level imaging. The magnetic properties of magnetite are determined by the short-range Fe-O-Fe super-exchange interactions. The dominant super-exchange interaction in Fe_3O_4 between tetrahedral Fe_A and octahedral Fe_B sublattices results in their antiferromagnetic alignment^{30,31}. Since there are twice as many Fe atoms on B sites as on A sites, this gives an overall net magnetisation of $4\mu_B$ per formula unit of Fe_3O_4 . The super-exchange interactions depend strongly on the angle and the length of the Fe-O-Fe bonds³². When the angle is 90° (the case of $\text{Fe}_B\text{-O-Fe}_B$ bonds) the super-exchange interaction is ferromagnetic. As the bond angle increases the super-exchange becomes antiferromagnetic (AFM), as in the case of $125^\circ \text{Fe}_A\text{-O-Fe}_B$ bond that is the dominant super-exchange interaction in magnetite. Any further increase in angle, increases the strength of AFM interaction, which reaches a maximum value for 180° bonds³³. Across the APB the Fe-O-Fe bond angles are distorted which significantly impacts the magnetic properties.

Atomistic spin calculations on model particles (with and without APBs) provide insight into the effect of APBs on NP magnetic moment. Using VAMPIRE software³⁴ we calculated the spin configurations at magnetic saturation for various model NPs and compared the results to same size NP without APBs (Fig. 4a). Figure 3c shows the atomic geometry of the simplest model, a spherical nanoparticle 10 nm in diameter, with a $\frac{1}{4}a_0 \langle 110 \rangle$ APB at the center of the NP. The calculations for this model NP show that there is a reduction in the saturation magnetization by 26% compared to the same size NP without the structural defect (i.e. magnetization value of $\sim 66 \text{ emu/g}$). Furthermore, looking at the spin configuration snapshots (Fig. 4a,b) one can see the formation of two magnetic domains due to the presence of the strong 180° antiferromagnetic bonds across the boundary (Fig. 4c). Within a domain, the Fe_A and Fe_B spins are antiparallel, as expected in bulk magnetite. However, across the APB, the Fe_B spins are canted because of the change in $\text{Fe}_B\text{-O-Fe}_B$ superexchange coupling. In Fig. 4b, where the NP has an APB across its center, the magnetization of the left (M_L) and right (M_R) sides of the particle are not parallel, even at 5T. The APB reduces the magnetic moment, relative to a defect-free NP.

This simple model demonstrates the effect of the APBs on the NP magnetic moment, but more complex NPs with multiple APBs, faceting, and surface anisotropy were also studied. When the number of coplanar APBs and their relative positions were varied, the magnetization reduction in all modeled NPs is determined by the number of $180^\circ \text{Fe}_B\text{-O-Fe}_B$ bonds per unit volume. In Fig. 4d we consider two non-coplanar APBs with boundaries on $\langle 11-2 \rangle$ type planes, a model inspired by the APBs geometry observed in the Hyeon NP shown in Fig. 2c,d. The predicted decrease in the saturation magnetization for this model NP is $\sim 34\%$ (i.e. the model nanoparticle has magnetization of $\sim 59 \text{ emu/g}$), a result quantitatively closer to the experimental data. Given the small nanoparticle size, surface effects can affect the magnetic properties. Nominally spherical particles can develop faceted surfaces either directly during synthesis or over time via Ostwald ripening. Supplementary Fig. S14 shows that this has little impact on the spin configuration and therefore the particle moment. Surface anisotropy can have important effects in NPs^{30,35}. However, the NPs here are coated with organic surfactants with weak spin-orbit coupling, hence the surface anisotropy contributions are negligible compared to the strong exchange interaction across the

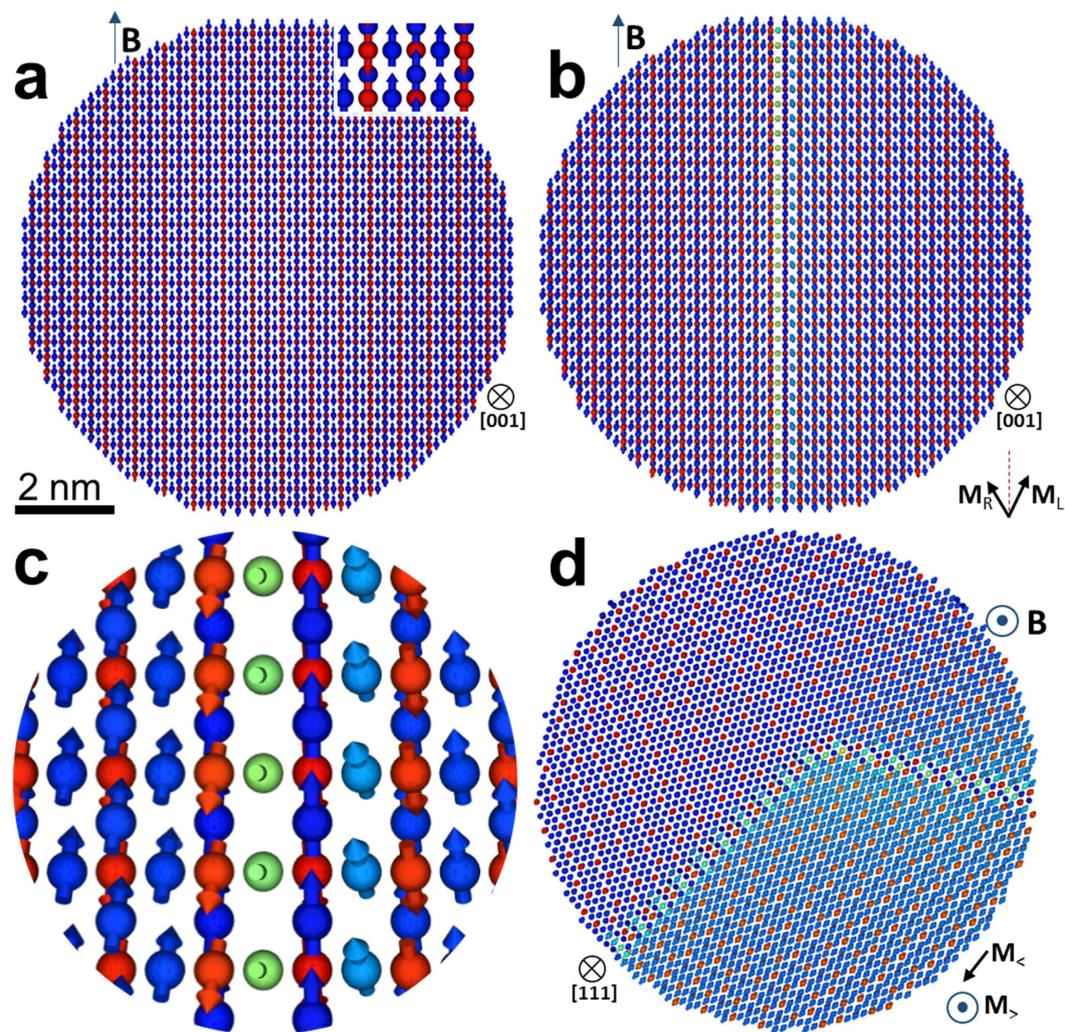


Figure 4. Spin configurations at 5T. (a) [001] zone axis of a 10 nm nanoparticle without APBs, which has a single magnetic domain. (b) [001] zone axis of a 10 nm nanoparticle with a $\frac{1}{4}a_0 \langle 110 \rangle$ APB, which leads to a coincident magnetic domain wall. The net magnetisations of the left and right part of the NP are labelled as M_R and M_L , respectively. (c) Magnified view of (b) near the APB plane showing the canting of the spins at the boundary. (d) 10 nm model with two non-coplanar APBs on $\langle 1\ 1\ \bar{2} \rangle$ type planes, with net magnetization $M_{>}$ (along the field) and $M_{<}$ (in-plane) for the upper and lower part respectively. The colour coding is in RGB fashion where the function which determines the colour is dependent only on the projection of the spin direction along the applied field. Two limiting cases are pure blue ($R=0; G=0; B=1$) for spins aligned along the field and pure red colour for spins in opposite direction. Net magnetization direction (determined by the Fe_B spins) in each of the domains helps identify the Fe_B sites.

APB defects. Even when large surface anisotropy was assumed, as shown in Supplementary Fig. S15 there was minimal distortion of the spin configuration and the magnetic moment was reduced by only 1%.

Conclusions

Nanoparticles prepared by the Sun, Colvin and Hyeon methods have similar structure and chemical composition, as confirmed by conventional TEM imaging, SAED, and EELS. However, their magnetic properties are very different. Sun NPs had close to bulk like specific magnetization while Hyeon and Colvin NPs showed less than half of the specific magnetization of bulk Fe_3O_4 . This anomalous magnetic behavior was correlated with the absence or presence of antiphase domain boundary structural defects. Atomic resolution STEM-HAADF imaging clearly shows the presence of abundant APB defects in the Colvin and Hyeon but not the Sun NPs. The dominant APB occurring in Colvin and Hyeon NPs is due to a $\frac{1}{4} a_0 \langle 110 \rangle$ lattice vector shift. Due to the very low energy formation of this type boundary they can form under the growth conditions of the Colvin and Hyeon methods. The density of APBs is directly correlated to the number of antiferromagnetic bonds that ultimately lead to the reduced magnetization. Atomistic spin calculations show that the magnetization reduction is greater when more antiferromagnetic bonds are created across the boundary planes. The strong character of the super-exchange interactions is responsible for the multi-domain magnetic behaviour of the Fe_3O_4 NPs, a unique property not expected in metallic magnetic-NPs.

In summary, the antiphase boundary structural defects create high angle antiferromagnetic bonds that reduce the magnetization in crystalline magnetite nanoparticles. Magnetic nanoparticles predicted to be mono-domain based on their size alone may in fact contain multiple magnetic domains. Because of their stability, APBs reduce the particle magnetic moment unless extremely large fields are applied. For applications where the magnitude of the particle magnetization is critical and the applied fields are modest, NPs with a low density of APBs are desirable. ZFC magnetization measurements are a convenient way to screen nanoparticles and optimize preparation methods in order to minimize these magnetization-reducing defects.

Methods

All samples were prepared using Schlenk line techniques, with inert atmosphere decomposition in high boiling point organic solvents. There were slightly different precursors, surfactants, and solvents. In all cases a single growth stage was used to obtain the final particle size. The Sun method¹⁷ uses Fe acetyl acetonate as a precursor and a combination of oleic acid and oleyl amine surfactants, in benzyl ether. The Colvin method¹⁸ uses an Fe oxyhydroxide precursor and oleic acid, in 1-octadecene. The Hyeon method¹⁹ uses an Fe oleate precursor and oleic acid, also in 1-octadecene. Transmission electron microscopy (TEM) was used to assess the size distribution of the particles and selected area electron diffraction (SAED) was used for preliminary phase identification. Dense assemblies of the particles were characterized by superconducting quantum interference device (SQUID) magnetometry to determine the saturation magnetization at 10 K and 300 K, at fields up to 5 T. Zero field cooled (ZFC) magnetization curves were measured after cooling to 10 K, turning on a small field (0.01 T), and measuring the magnetization as the temperature was increased up to 300 K. For field cooled (FC) magnetization curves, the magnetization was recorded as the samples were cooled from 300 K to 10 K in 0.01 T field.

To quantify the specific saturation magnetization, elemental analysis was used to determine the amount of iron. Here the magnetic moment at 5 T of a fixed volume of nanoparticle dispersion in toluene was measured. Atomic absorption measurements were made on the same samples over a series of dilutions, and comparison with a Fe calibration standard was used to determine the molar concentration of iron in the undiluted sample. The numerical values assumed that the iron was in the form of Fe₃O₄ and that the particles had the same density as that of bulk magnetite.

Structural characterisation has been performed by transmission electron microscopy (TEM) and Selected Area Diffraction (SAED) using a JEOL 2000 EX and a double aberration corrected JEOL JEM-2200FS, both operated at 200 kV.

Scanning transmission electron microscopy imaging and electron energy loss spectroscopy were performed in a Nion UltraSTEM100™ equipped with a Gatan Enfina spectrometer. The microscope was operated at 100 kV, with a convergence angle of 30 mrad; at these optical conditions the electron probe size is determined to be 0.9 Å; the inner detector angle for HAADF STEM imaging was 76 mrad. The native energy spread of the electron beam for EELS measurements was 0.3 eV; with the spectrometer dispersion set at 0.2 eV/channel, this yielded an effective energy resolution of 0.6 eV. The EELS collection angle was 31 mrad. For reading clarity the atomically resolved spectra presented in Supplementary Figs S4, S5 and S6 were de-noised by principal component analysis using the CiMe⁻ plugin³⁶ for Gatan's Digital Micrograph 2.3 software suite.

The spin dynamics of the magnetite nanoparticles containing APBs were simulated using VAMPIRE software package³⁴. The energetics of the system is described by a Heisenberg spin Hamiltonian of the form:

$$\mathcal{H} = -\sum_{i \neq j} J_{ij} \mathbf{S}_i \cdot \mathbf{S}_j - \frac{k_c}{2} \sum_i (S_x^4 + S_y^4 + S_z^4) - \sum_i \mu_i \mathbf{S}_i \cdot \mathbf{H}_{\text{app}},$$

where J_{ij} is the Heisenberg exchange interaction between neighbouring spins i and j , \mathbf{S}_i and \mathbf{S}_j are the local spin unit vectors on sites i and j respectively, k_c is the local cubic anisotropy constant, μ_i is the local atomic spin moment on each site i and \mathbf{H}_{app} is the external applied field vector. The demagnetizing field is included in the simulations using the macrocell method³⁴ where the atomic spin configuration is spatially averaged over the macrocell. A macrocell size of 2 nm was used in the calculations. A cubic magnetocrystalline anisotropy constant of $3.2 \cdot 10^{-25}$ J/atom is included in the simulations. We note that although local cubic anisotropy is included in the simulation, this is relatively weak and has no appreciable effect on the total magnetization of the particle, which is dominated by the exchange interactions. Between Fe_A and Fe_B sites of magnetite there are corresponding exchange interactions: $J_{AA} = -0.11$ meV, $J_{BB} = +0.63$ meV and $J_{AB} = -2.92$ meV determined from ab-initio calculations³⁷. At the APB, the 180° super exchange bonds lead to an enhanced antiferromagnetic exchange interactions with $J_{180} = -8.9$ meV, which was extrapolated from the known bulk exchange interactions using $\sim \cos^2(\theta)$ relationship³⁸. All the calculations were performed at temperature set to 0 K in order to show the ground state spin configuration. At higher temperature, spin fluctuations are present within domains and at the surface, but the domain structures remain the same. The simulations focused on spin configurations for NPs at saturation fields (5 T). Details of lower field hysteresis require a full treatment of the local anisotropy introduced by the APB, and are the subject of future work.

Data Availability. All data created during this research are available by request from the University of York Data Catalogue <https://dx.doi.org/10.15124/249cbf0c-8e88-426b-b3ba-53d490e027ed>.

References

- Berger, P. *et al.* Preparation and Properties of an Aqueous Ferrofluid. *Journal of Chemical Education* **76**, 943 (1999).
- Raj, K., Moskowitz, B. & Casciari, R. Advances in ferrofluid technology. *Journal of Magnetism and Magnetic Materials* **149**, 174 (1995).

3. Gupta, A. K. & Gupta, M. Synthesis and surface engineering of iron oxide nanoparticles for biomedical applications. *Biomaterials* **26**, 3995 (2005).
4. Laurent, S. *et al.* Magnetic iron oxide nanoparticles: synthesis, stabilization, vectorization, physicochemical characterizations, and biological applications. *Chem Rev* **108**, 2064 (2008).
5. Pereira, C., Pereira, A. M., Rocha, M., Freire, C. & Geraldes, C. F. G. C. Architected design of superparamagnetic Fe₃O₄ nanoparticles for application as MRI contrast agents: mastering size and magnetism for enhanced relaxivity. *Journal of Materials Chemistry B* (2015).
6. Thiesen, B. & Jordan, A. Clinical applications of magnetic nanoparticles for hyperthermia. *International journal of hyperthermia: the official journal of European Society for Hyperthermic Oncology, North American Hyperthermia Group* **24**, 467 (2008).
7. Pankhurst, Q. A., Thanh, N. T. K., Jones, S. K. & Dobson, J. Progress in applications of magnetic nanoparticles in biomedicine. *Journal of Physics D: Applied Physics* **42**, 224001 (2009).
8. Pratt, A. *et al.* Enhanced oxidation of nanoparticles through strain-mediated ionic transport. *Nat Mater* **13**, 26 (2014).
9. Kodama, R. H., Berkowitz, A. E., McNiff, J. E. J. & Foner, S. Surface Spin Disorder in NiFe₂O₄ Nanoparticles. *Phys Rev Lett* **77**, 394 (1996).
10. Massart, R. Preparation of aqueous magnetic liquids in alkaline and acidic media. *Magnetics, IEEE Transactions on* **17**, 1247 (1981).
11. Krycka, K. L. *et al.* Origin of Surface Canting within Fe₃O₄ Nanoparticles. *Phys Rev Lett* **113**, 147203 (2014).
12. Salafra, J. *et al.* Surfactant Organic Molecules Restore Magnetism in Metal-Oxide Nanoparticle Surfaces. *Nano Letters* **12**, 2499 (2012).
13. Gilks, D. *et al.* Origin of anomalous magnetite properties in crystallographic matched heterostructures: Fe₃O₄(111)/MgAl₂O₄(111). *Journal of Physics: Condensed Matter* **25**, 485004 (2013).
14. Eerenstein, W., Palstra, T. T. M., Hibma, T. & Celotto, S. Origin of the increased resistivity in epitaxial Fe₃O₄ films. *Physical Review B* **66**, 201101 (2002).
15. Margulies, D. T. *et al.* Origin of the Anomalous Magnetic Behavior in Single Crystal Fe₃O₄ Films. *Phys Rev Lett* **79**, 5162 (1997).
16. Wetterskog, E., Tai, C.-W., Grins, J., Bergström, L. & Salazar-Alvarez, G. Anomalous Magnetic Properties of Nanoparticles Arising from Defect Structures: Topotaxial Oxidation of Fe_{1-x}O|Fe_{3-x}O₄ Core|Shell Nanocubes to Single-Phase Particles. *ACS Nano* **7**, 7132 (2013).
17. Sun, S. *et al.* Monodisperse MFe₂O₄ (M = Fe, Co, Mn) Nanoparticles. *J Am Chem Soc* **126**, 273 (2004).
18. Yu, W. W., Falkner, J. C., Yavuz, C. T. & Colvin, V. L. Synthesis of monodisperse iron oxide nanocrystals by thermal decomposition of iron carboxylate salts. *Chemical Communications* **2306** (2004).
19. Park, J. *et al.* Ultra-large-scale syntheses of monodisperse nanocrystals. *Nat Mater* **3**, 891 (2004).
20. Daou, T. J. *et al.* Hydrothermal Synthesis of Monodisperse Magnetite Nanoparticles. *Chemistry of Materials* **18**, 4399 (2006).
21. Batlle, X. *et al.* Magnetic nanoparticles with bulklike properties (invited). *J Appl Phys* **109**, 07B524 (2011).
22. Kovács, A. *et al.* Direct Observation of a Surface Induced Disorder Process in Magnetic Nanoparticles. *Phys Rev Lett* **103**, 115703 (2009).
23. Margulies, D. T. *et al.* Anomalous moment and anisotropy behavior in Fe₃O₄ films. *Phys. Rev. B* **53**, 9175 (1996).
24. Hibma, T. *et al.* Anti-phase domains and magnetism in epitaxial magnetite layers. *J Appl Phys* **85**, 5291 (1999).
25. Luysberg, M., Sofin, R. G. S., Arora, S. K. & Shvets, I. V. Strain relaxation in Fe₃O₄/MgAl₂O₄ heterostructures: Mechanism for formation of antiphase boundaries in an epitaxial system with identical symmetries of film and substrate. *Physical Review B* **80**, 024111 (2009).
26. Eerenstein, W., Palstra, T. T. M., Saxena, S. S. & Hibma, T. Spin-Polarized Transport across Sharp Antiferromagnetic Boundaries. *Phys Rev Lett* **88**, 247204 (2002).
27. Eerenstein, W., Palstra, T. T. M. & Hibma, T. Spin-valve behaviour of anti-ferromagnetic boundaries in ultrathin magnetite films. *Thin Solid Films* **400**, 90 (2001).
28. Eerenstein, W., Kalev, L., Niesen, L., Palstra, T. T. M. & Hibma, T. Magneto-resistance and superparamagnetism in magnetite films on MgO and MgAl₂O₄. *Journal of Magnetism and Magnetic Materials* **258–259**, 73 (2003).
29. McKenna, K. P. *et al.* Atomic-scale structure and properties of highly stable antiphase boundary defects in Fe₃O₄. *Nat Commun* **5**, (2014).
30. Mazo-Zuluaga, J., Restrepo, J. & Mejía-López, J. Surface anisotropy of a Fe₃O₄ nanoparticle: A simulation approach. *Physica B: Condensed Matter* **398**, 187 (2007).
31. De Grave, E., Persoons, R. M., Vandenberghe, R. E. & de Bakker, P. M. Mossbauer study of the high-temperature phase of Co-substituted magnetites, Co_xFe_{3-x}O₄, 0 < x <= 0.04. *Physical review B, Condensed matter* **47**, 5881 (1993).
32. Goodenough, J. B. & Loeb, A. L. Theory of Ionic Ordering, Crystal Distortion, and Magnetic Exchange Due to Covalent Forces in Spinels. *Phys Rev* **98**, 391 (1955).
33. Robinson, D. W. Magnetism and the Chemical Bond. John B. Goodenough. Interscience (Wiley), New York, 1963. xvi + 394 pp. Illus. \$12.50. *Science* **143**, 33 (1964).
34. Evans, R. F. L. *et al.* Atomistic spin model simulations of magnetic nanomaterials. *Journal of Physics: Condensed Matter* **26**, 103202 (2014).
35. Yanes, R. *et al.* Effective anisotropies and energy barriers of magnetic nanoparticles with Néel surface anisotropy. *Physical Review B* **76**, 064416 (2007).
36. Lucas, G., Burdet, P., Cantoni, M. & Hébert, C. Multivariate statistical analysis as a tool for the segmentation of 3D spectral data. *Micron* **52–53**, 49 (2013).
37. Uhl, M. & Siberchicot, B. A first-principles study of exchange integrals in magnetite. *Journal of Physics: Condensed Matter* **7**, 4227 (1995).
38. Sawatzky, G. A., Geertsma, W. & Haas, C. Magnetic interactions and covalency effects in mainly ionic compounds. *Journal of Magnetism and Magnetic Materials* **3**, 37 (1976).
39. Momma, K. & Izumi, F. VESTA 3 for three-dimensional visualization of crystal, volumetric and morphology data. *J Appl Crystallogr* **44**, 1272 (2011).

Acknowledgements

S. A. M. acknowledges support from US DOE grant #DE-FG02-08ER46481. SuperSTEM is the U.K. National Facility for Aberration Corrected STEM, supported by the Engineering and Physical Sciences Research Council (EPSRC). V.K.L. acknowledges support from EPSRC (through grants EP/K03278X and EP/K013114/1). This work made use of the facilities of N8 HPC provided and funded by the N8 consortium and EPSRC (Grant No. EP/K000225/1). The Centre is co-ordinated by the Universities of Leeds and Manchester.

Author Contributions

Atomistic spin calculations were done by Z.N. and R.F.L.E. Iron NPs synthesis and magnetic characterization was done by T.W., R.A.B., S.D.O. and S.M. Electron microscopy and EEL spectroscopy were done by D.K., Q.M.R.,

Z.N., L.L. and V.K.L. Strain maps were produced by Z.N., V.K.L. and P.L.G. All authors contributed to the analysis and interpretation of the data. V.K.L and S.M. wrote the paper.

Additional Information

Supplementary information accompanies this paper at <http://www.nature.com/srep>

Competing Interests: The authors declare no competing financial interests.

How to cite this article: Nedelkoski, Z. *et al.* Origin of reduced magnetization and domain formation in small magnetite nanoparticles. *Sci. Rep.* 7, 45997; doi: 10.1038/srep45997 (2017).

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>

© The Author(s) 2017