



Article High-Resolution Magic Angle Spinning NMR of KcsA in Liposomes: The Highly Mobile C-Terminus

Gary S. Howarth D and Ann E. McDermott *

Department of Chemistry, Columbia University, New York, NY 10031, USA

* Correspondence: aem5@columbia.edu

Abstract: The structure of the transmembrane domain of the pH-activated bacterial potassium channel KcsA has been extensively characterized, yet little information is available on the structure of its cytosolic, functionally critical N- and C-termini. This study presents high-resolution magic angle spinning (HR-MAS) and fractional deuteration as tools to study these poorly resolved regions for proteoliposome-embedded KcsA. Using ¹H-detected HR-MAS NMR, we show that the C-terminus transitions from a rigid structure to a more dynamic structure as the solution is rendered acidic. We make previously unreported assignments of residues in the C-terminus of lipid-embedded channels. These data agree with functional models of the C-terminus-stabilizing KcsA tetramers at a neutral pH with decreased stabilization effects at acidic pH. We present evidence that a C-terminal truncation mutation has a destabilizing effect on the KcsA selectivity filter. Finally, we show evidence of hydrolysis of lipids in proteoliposome samples during typical experimental timeframes.

Keywords: solid-state nuclear magnetic resonance (SS NMR); high-resolution nuclear magnetic resonance (HR-MAS NMR); KcsA; liposomes; lipid chemical shifts; lipid hydrolysis; membrane protein structure



Citation: Howarth, G.S.; McDermott, A.E. High-Resolution Magic Angle Spinning NMR of KcsA in Liposomes: The Highly Mobile C-Terminus. *Biomolecules* 2022, *12*, 1122. https:// doi.org/10.3390/biom12081122

Academic Editor: Wylie Benjamin

Received: 27 June 2022 Accepted: 4 August 2022 Published: 15 August 2022

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/).

1. Introduction

KcsA, a pH-activated, inward-rectifying potassium channel from the Gram-positive soil bacterium *Streptomyces lividans*, has played a unique role in the structural biology of ion channels, being the first to be characterized in detail. The continuing progress on structural, biochemical, electrophysiological, and biophysical characterization of KcsA makes it a uniquely rich model system for transmembrane allosteric coupling [2,3], ion channel activation and inactivation mechanism [4,5], lipid–protein interactions [6,7], and virtually any other question concerning ion channels and membrane proteins. However, we do not yet have a full-length, atomic-resolution structure of KcsA in a lipid membrane.

In situ, the KcsA channel is closed at neutral pH, briefly activates to let K⁺ ions flow into the cell when the cytosol is rendered acidic, and then enters an inactivation period. Functional channels are composed of four identical subunits of 160 amino acids, each with two transmembrane-spanning domains. While the transmembrane segments are relatively well-characterized, the most mobile portions of KcsA, its extracellular termini, present a unique set of challenges to resolve while in the membrane, despite their important roles.

Several efforts [5,8–11], most notably by Chill et al. [12], have provided nearly complete assignments for KcsA in detergent micelles. Solution studies have provided remarkable information about the dynamics [13], secondary structure [12], and the major [14] and minor pH sensors [15]. Solution studies of KcsA often involve truncation mutants and high temperatures (e.g., >45 °C).

Logically, channels can only be considered functional in a lipid bilayer. KcsA activation is lipid-dependent. Therefore, details of the lipid environment are likely to be important to structural elements relevant to function. KcsA has been shown to have an open-probability dependence on the lipid headgroup with a requirement of anionic lipids in the inner leaflet [6]. The stability of the KcsA tetramer is also lipid-dependent with anionic headgroups (e.g., phosphatidic acid (PA), phosphoglycerol (PG), and phosphoserine (PS)) providing greater stability, especially at a low pH compared to lipids with net-neutral or net-positive headgroups [16]. Crystal structures [17–20], biochemical experiments [21], and NMR data [22] have shown that KcsA routinely co-purifies with a diacyl lipid with a PG headgroup. Protein–lipid affinity experiments have shown that KcsA has a high affinity for the net-negatively charged lipids (PA, PE, and PG), with PG affinity being the highest of all [23].

High-resolution cross-polarization magic angle spinning solid-state NMR (CP-MAS) has emerged as one of the few techniques that can investigate membrane channels embedded in a lipid bilayer. There have been many successful efforts to use CP-MAS to make residue assignments to resonances of KcsA in proteoliposomes [24–28]. The protein residues that are detected and assigned are notably incomplete, emphasizing primarily the transmembrane helices (Figure 1). Yet, unobserved regions of KcsA are functionally critical.

The KcsA C-Terminus

The KcsA C-terminus is an example of a critical region that remains unresolved when channels are in a lipid membrane. The C-terminus of KcsA has been proposed to have a high degree of α -helix character and to project perpendicular to the membrane surface [1,12,29,30]. Truncations of and mutations within the C-terminus lead to reduced tetramer stability [31,32] and affect gating and pH dependence [15,29,32].



Figure 1. Solid-state NMR backbone assignments of KcsA in liposomes as described by the McDermott Group (*) [3,26,33,34] or Baldus Group (†) [28,35,36]. Structural regions are indicated to the right of residue numbers.

Where biochemical experiments have indicated the functional importance of KcsA's C-terminal domain [5,8–11], structural studies provide insight into the potential mechanism by which the C-terminus regulates KcsA function. Recombinant KcsA expressed as only as its C-terminal domain is soluble and has been shown to exist principally as a monomer at a neutral pH and to oligomerize as the solution decreases to pH 5 [9,15]. Magnetic resonance studies of full-length channels in MSP1D1 lipid nanodiscs indicate that the C-termini of individual channels associate at a neutral pH and dissociate at pH 5 [37]. However, the lipids used in that investigation were relatively short (14 carbon acyl chain), saturated DMPC lipids, and single-component phosphocholine (PC) bilayers have vastly reduced the open probability of KcsA compared to when an anionic lipid headgroup is at least a component of the bilayer [6].

X-ray and small-angle neutron scattering data support a model where the C-termini within a tetrameric channel are close in space at a neutral pH and become much more widely separated at pH 5 [30]. Those same data suggest that the changes in conformation of the bundle of C-termini accompany a change in conformation in the transmembrane domain, but that changes to the transmembrane domain do not occur in channels with truncated C-termini.

These previous studies in aggregate suggest a putative model where the C-termini within a tetramer electrostatically coordinate to stabilize channels at a neutral pH, and at an acidic pH the bundle of C-termini become more loosely associated, leading to allostery affecting transmembrane gate conformation. This model suggests that the bundle of C-termini would be rigid at neutral pH and more dynamic at acidic pH. These putative changes in dynamics provide motivation to study the C-terminus of KcsA by hybrid solid-state solution-state NMR methods. Arguably, proteoliposomes provide a gold standard for authentic membrane environments. With that perspective in mind, we show that high-resolution magic angle spinning NMR (HR-MAS) can be utilized to probe the C-terminus of KcsA to support this model while channels are embedded in a native-like membrane environment.

2. Methods

2.1. Protein Expression, Purification, and Reconstitution

Wild-type KcsA was expressed as an N-terminal His₆ fusion protein cloned onto a PASK90 plasmid, which confers ampicillin resistance, and was over-expressed in *Escherichia coli* cells using established protocols developed by our group [38].

Expression of fractional deuteration with uniform ¹³C and ¹⁵N KcsA was accomplished with minor modifications from [36]. Briefly, KcsA culture using Bl21(DE3) *E. coli* cells (New England Biolabs, Ipswich, MA, USA). Cells were grown at 35 °C in LB selection media to $OD_{600} = 0.8$ and collected by centrifugation. The cell pellet was rinsed in approximately 50 mL of D₂O (Cambridge Isotopes, Cambridge, MA, USA). The cells were then suspended in 500 mL of M9 minimal media solution (See Supplemental Information) using 98% D₂O. The M9 media was supplemented with 1.5 g per liter U-¹³C-glucose and 0.5 g ¹⁵N-ammonia. This culture was allowed to grow for 1 h at 37 °C, then induced with anhydrotetracyclin (aTC) (Sigma-Aldrich, Saint Louis, MO, USA), and allowed to express at 25 °C for 10 h.

KcsA was purified in the same manner regardless of labelling strategy. Pellets were lysed by French Pressure cell (Thermo, Waltham, MA, USA), detergent solubilized in buffer containing decyl-β-maltopyranoside (Anatrace, Maumee, OH, USA), and purified by His-Select nickel affinity (Sigma-Aldrich, Saint Louis, MO, USA) chromatography detergent (DM).

We prepared a KcsA C-terminal truncation construct, containing only residues 1–125 (KcsA- Δ 125), cleaving residues 126–160, by incubating purified KcsA in detergent with bovine α -chymotrypsin (Sigma-Aldrich, Saint Louis, MO, USA) for 3 h at 35 °C. KcsA was isolated using Ni-affinity gel as described above.

Liposomes were formed in a 9:1 mass ratio of 1,2-dioleoyl-sn-glycero-3-phosphoethan olamine (DOPE) to 1,2-dioleoyl-sn-glycero-3-phospho-L-serine (DOPS) (Avanti, Alabaster,

AL, USA) by solubilizing a thin film of the lipids in detergent buffer and mixing with detergent solubilized KcsA in a 1:1 lipid-to-protein mass ratio. The lipid-protein solution was dialyzed into the final buffer with three buffer changes. The presence of KcsA as a tetramer in the liposomes was verified by SDS-PAGE before and after NMR experiments.

2.2. NMR

J-coupled-based experiments were performed on a Bruker magnet with a proton field of 750 MHz using a 4 mm high-resolution magic angle spinning probe (HR-MAS) equipped with ¹H/¹³C/¹⁵N/²H channels and a 40 G/cm gradient coil oriented along the magic angle. HSQCs were collected using the Bruker sequence hsqcetgpsi for ¹³C-resolved spectra, and hsqcetf3gpsi2 for ¹⁵N resolved spectra. hCCH-TOCSY spectra were collected with full-rotor period synchronized TOCSY 10 kHz spinlocks using the Bruker sequence hcchdigp3d2. HcCH-TOCSY indicates ¹³C as F2 and ¹H as F1 and hCCH-TOCSY indicates 13 C in both indirect dimensions. Site-specific T_2 measurements were collected by adding a single rotor-synchronized ¹³C spin-echo between the two inept transfers and increasing the delay of the echo over at least five steps until magnetization had decayed by at least 90%. Quantitative ¹H spectra were collected with calibrated ¹H 90° (typically ~9 μ s), with recycle delays of at least 5 s (the longest T_1 in samples was typically ~0.9 s) with presaturation on the H₂O resonance during the recycle delay and a presaturation field strength of approximately 25 Hz. Typically, 4 to 16 scans were collected to reach a signal-to-noise ratio of more than 20 for the CH₂ region. Cross-polarization MAS was performed with a Bruker 750 Avance I spectrometer using a 3.2 mm $^{1}H/^{13}C/^{15}N$ e-free probe at 16.6 kHz at 275 K rotor set temperature.

2.3. Sucrose Gradients

Sucrose Gradients were prepared by layering equal volumes of two sucrose concentrations (depending on upper and lower bounds of gradient desired) in buffer in 12 mL ultracentrifuge tubes (Beckman, Brea, CA, USA), then applying a preprogramed algorithm to mix the layers using a Gradient Master (BioComp Instruments, Fredericton, NB, CA) and cooling to 4 °C overnight. Proteoliposomes were prepared as described above except, a rhodamine-conjugated lipid (1,2-dioleoyl-sn-glycero-3-phosphoethanolamine-N-(lissamine rhodamine B sulfonyl) (Rhod-PE) (Avanti, Alabaster, AL, USA) was added to the lipid mixture to more easily visualize and quantify the mixtures. The lipid to protein ratio was 1:1 by mass. The lipid mixture was 900:10:1 PE-PS-Rhod PE by mass. A total of 20 mg of KcsA proteoliposomes was suspended in buffer and added to the top of the gradient. The tube was subjected to ultracentrifugation at 25,000 RPM, corresponding to a relative centrifugal field ranging from $47,200 \times g$ to $107,000 \times g$, for 24 h at 4 °C with the slowest acceleration and no brake during deceleration. The bottom of the tube was pierced with a 20 Ga needle that was then removed and the tube was allowed to flow under gravity at $4 \,^{\circ}$ C. Fractions were measured by UV-VIS at 280 nm for the presence of protein and 560 nm for the presence of lipids as a qualitative measure. To quantify protein we solubilized KcsA proteoliposomes in Triton-X100 and measured against a standard curve using bromophenol blue. We performed SDS-PAGE and a Pierce silver stain kit (ThermoFisher, Waltham, MA, USA) was used to visualize protein in gradient fractions.

See the Supplementary Information for more detailed materials and methods.

3. Results

Since the C-terminus of KcsA is not resolved by CP-MAS nor by crystallography, we reasoned that the order and rigidity required by these methods is likely lacking. All regions of KcsA when channels are detergent-solubilized are accessible to solution-state NMR. However, the long rotational correlation time of channels in bilayers leads to no well-resolved resonances in traditional solution-state NMR due to anisotropic effects.

Therefore, we pursued a strategy to reduce anisotropy and increase radio field homogeneity using magic angle spinning (MAS). We made *J*-based NMR measurements, which work well on isotropic systems, to increase the likelihood of capturing signals from dynamic regions of the protein. An HR-MAS probe is ideally suited for these purposes. HR-MAS detects the ¹H signal, and as ¹H has much a much higher gyromagnetic ratio and typically much shorter T_1 relaxation periods than ¹³C and ¹⁵N, detecting on ¹H leads to practical gains of 100 times or greater signal-to-noise ratio than detecting on the heavier nuclei. Additionally, unlike CP-MAS probes, HR-MAS probes have a gradient coil, enabling modern, gradient-selected pulse sequences.

3.1. HR-MAS Selectively Detects Signals from the KcsA C-Terminus

We prepared proteoliposomes with a C-terminal truncation construct, containing only residues 1–125 (KcsA- Δ 125) by cleaving residues 126–160 by enzyme reaction (Supplementary Materials Figure S1). We collected both ¹⁵N (Figure 2) and ¹³C (Supplementary Materials Figure S2) HSCQ spectra of the KcsA- Δ 125 truncation mutant in liposomes and compared these to spectra of proteoliposomes containing full-length KcsA. Comparing full-length KcsA to KcsA- Δ 125, we conclude that much of the protein signal in spectra of the full-length construct arise from the C-terminus. The relatively few signals observed in the ¹H-¹³C HSQC of the KcsA- Δ 125 sample arise mainly from the exogenous phospholipids into which the protein is reconstituted. Similarly, ¹H-¹⁵N HSQC spectra (Figure 2) also show that most (though not all) resonances are eliminated when the C-terminus. Throughout this study, SDS-PAGE was used to verify that all samples were of tetrameric KcsA before and after NMR experiments. Protein-free liposomes were examined to assign resonances arising from lipids. We do not present data of KcsA- Δ 125 at low pH, as each time such a sample was prepared, we detected only monomeric KcsA.



Figure 2. ¹H-¹⁵N HSQC by HR-MAS of full-length wild-type KcsA (red), and KcsA- Δ 125 (navy); both samples pH 7.25, 50 mM K⁺, 308 K, 5 kHz MAS.

3.2. Cleavage of the KcsA C-Terminus Causes Conformational Heterogenity near the Selectivity Filter

To inspect for changes to the KcsA transmembrane domain caused by cleavage of its C-terminus, we collected CP-MAS $^{13}C^{-13}C$ proton-driven spin diffusion (DARR) spectra (Supplementary Materials Figure S3). Chemical shift assignments from previous studies in our group [26] were used to assign the resonances that are present in the CP-MAS data of this study. These data indicate the KcsA- Δ 125 transmembrane domain is well-folded. However, several transmembrane resonances that are typically observed in full-length KcsA spectra are missing from the KcsA- Δ 125 sample, corresponding with the residues



Ile-38, Leu-59, Thr-74, Asp-80, Tyr-82, and Thr-85. We map these absences onto a cartoon of KcsA in Figure 3.

Figure 3. Cartoon of KcsA adapted from PDB: 1F6G [29], with two subunits hidden for clarity. Cytosolic N-terminus shown in orange (residues 1–22), transmembrane domain in blue (23–124), and cytosolic C-terminus in magenta (125–160). Resonances missing from CP-MAS ¹³C-¹³C spectrum of KcsA– Δ 125 are indicated with blue labels. Newly assigned Lys-131 indicated. Approximate locations of candidates for type-assigned resonances labeled in black with single letter amino acid code. Visualized with Pymol [39].

The most likely explanation of these transmembrane residues missing from KcsA- Δ 125 data is that the C-terminal truncation leads to much greater conformational heterogeneity of the regions at the interface between KcsA's loop regions and selectivity filter. Alternatively, or in addition to conformational heterogeneity, the deletion of the C-terminus may lead to increased dynamics within the selectivity filter, thereby making it less amenable to cross-polarization experiments, which rely on rigidity for efficient magnetization transfer. Previous studies have found a decreased open probability of C-terminal truncation mutants [2]. Regardless of the specific mechanism, these data provide evidence that the C-terminus is involved in stabilizing the KcsA selectivity filter and suggest a mechanism for the decreased open probability caused by truncating the C-terminus [6,31,32].

3.3. Low pH Leads to More Dynamics of the C-Terminus of KcsA Embedded into Liposomes

Both ¹H-¹³C and ¹H-¹⁵N HSQC HR-MAS spectra of KcsA in liposomes at neutral pH contain few resonances, most of which originate from the C-terminus. When the pH of the sample is lowered to pH 4, both the number and the distribution of resonances in spectra increases. Figure 4 presents an example. The dramatic increase in the number of resolved resonances and an overall increase in signal strength suggest that KcsA is undergoing much greater conformational dynamics at a low pH. The ¹³C and ¹⁵N chemical shifts of resonances that appear in our HR-MAS spectra do not correspond with shifts appearing in our solid-state spectra. Additionally, we have demonstrated in previous studies that

there is no comparable reduction in peaks in CP-MAS spectra when moving from pH 7 to pH 4 [26]. Thus, we conclude these previously unresolved resonances arise from KcsA termini. We were able to collect a variety of 3D correlations (HNCO, HNCA, HcCH-TOCSY, and hCCH-TOCSY). We are unable to unambiguously assign resonances in ¹⁵N correlation spectra by residue or by type, but we are able to make assignments in our ¹³C data, and those details are given in Section 3.7.



Figure 4. ¹H-¹⁵N HSQC of protonated KcsA at pH 7.25 (red), and fractionally deuterated KcsA at pH 4.0 (turquoise) by HR-MAS. 5 kHz MAS, 308 K, 50 mM K⁺.

3.4. Leucine as a C-Terminal Conformational Indicator

The conformational and dynamical changes that the protein undergoes from neutral to low pH can be observed in ${}^{1}\text{H}{-}{}^{13}\text{C}$ HSQC data. This is convenient, as a high-quality ¹H-¹³C HSQC can be collected in approximately three hours, where a ¹H¹⁵N HSQC with similar signal to noise requires approximately 16 h. We find the leucine ${}^{1}H^{-13}C\delta$ (methyl) correlations in the HSQC to be a reliable indicator of state. We made the residue type assignment from characteristic shifts of connected spin systems from 3D TOCSY data (see Section 3.7 for more shift assignments). The number and chemical shifts of peaks in HSQC data are replicable at both neutral and acidic pH, reflecting systematic changes to the entire spectrum, and to the pair of leucine in particular (Figure 5). We cannot confirm whether we are detecting the same leucine residues in the sequence at each pH condition. These leucine resonances are not present in the C-terminal truncation construct, confirming that these resonances arise from the C-terminus, which has a total of three leucine residues (Leu-144, Leu-151, Leu-155). The difference in the shift dispersion (line broadening) at a low pH is a strong indication that the conformation of the C-terminus is more heterogeneous than the conformation at neutral pH. These observations support a model where the KcsA C-termini at neutral pH are relatively well-defined, and the termini bundle becomes more heterogeneous at low pH.



Figure 5. ¹H-¹³C HSQC spectra of leucine H δ of KcsA in liposomes. pH 7.25 (red contours) and pH 4.0 (turquoise contours) are shown, with fully protonated KcsA samples (**A**) and fractionally deuterated KcsA samples (**B**).

3.5. ¹³C T₂ Relaxation

We sought to estimate the relative anisotropy of resonances using site-specific ¹³C transverse relaxation of the protein signal of uniformly ¹H, ¹³C, and ¹⁵N-labelled KcsA in liposomes. We collected HSQC data with a variable-length dephasing period before T_2 evolution (Supplementary Materials Table S1: ¹³C T_2 s). The average protein ¹³C T_2 relaxation rate was 2.9 ms, and the average for lipid headgroup ¹³C was 5.2 ms at 308 K, 9 kHz MAS, and 10 kHz heteronuclear decoupling. The optimal transfer time for a single C α -N polarization transfer for the refocused-INEPT is expected to be 45 to 70 ms [40]. Our relaxation data predict a maximum theoretical signal yield of only 4% when compared to yield possible in the absence of relaxation, as predicted from the density function of the refocused INEPT experiment, $\Gamma(t) = sin(\pi J_{N-C\alpha}t)exp(\frac{t}{T_2})$ [41]. Under these conditions, even the highest-sensitivity ¹H-detected backbone experiments, such as the HNCaH and HNCO (out-and-back), which require multiple INEPT transfers, are not feasible with fully protonated samples.

3.6. Fractional Deuteration

Anisotropic interactions, particularly dipolar couplings, lead to shorter T_2 relaxation values for proteins. Residual dipolar couplings with ¹H nuclei vastly attenuate *J*-based coherence transfer in backbone experiments. Replacing 1 H with 2 H in proteins is a well-established method of reducing these dipolar couplings, thus increasing T_2 times and improving resolution and coherence transfer efficiency [42,43]. Although reports of high [5,11,37] and perdeuterated [12] KcsA exist in the literature, the expression of perdeuterated KcsA led to poor yields in our research. We followed previous work that found that expressing proteins in D_2O minimal media supplemented with U-¹H, ¹³C-glucose and ¹⁵N-ammonium chloride leads to proteins that have very high levels of deuteration at the H-C α position and fractional deuteration in the sidechains that vary by amino acid [35,44]. Here, we expressed KcsA in this manner and incorporated it into liposomes. ¹³C T_2 relaxation profiles were determined using a 1D back-INEPT experiment with a dephasing period. We present profiles for the methyl region, the aliphatic region, Glu- ${}^{13}C\beta$, and Glu- ${}^{13}C\gamma$ as these are readily identified on the 1D spectrum of our samples (Figure 6). The results show that the relaxation profile is vastly improved in fractionally deuterated samples over uniformly protonated samples. The Glu-C γ profiles are similar for both samples, which agrees well with previous observations that both glutamate γ protons tend to remain protonated in this labeling scheme, and a portion of the Glu-H β remains protonated in this labeling scheme [35]. With the fully protonated sample, signal was insufficient to collect 2D HNCA or HNCO correlation experiments, while with fractional ²H-KcsA, full 3D datasets were obtained, showing 13 well-resolved resonances (Supplementary Materials Figures S8 and S9). Yet, these improvements still leave backbone walk experiments out of reach.



Figure 6. The T_2 relaxation time is lengthened due to fractional deuteration of KcsA. (A) ¹H-¹³C double-INEPT (1D refocused HSQC) spectrum of U-¹H, ¹³C, ¹⁵N-KcsA in proteoliposomes (9:1 DOPE-DOPS) with peak annotations. Assignments are based on 2D and 3D data. (B) ¹³C relaxation of selected peaks from 1D double-INEPT for fully protonated KcsA (black), and fractionally deuterated KcsA (red). Both samples U-¹³C, ¹⁵N at pH 4, 50 mM K⁺, 9 kHz MAS, 308 K.

3.7. Chemical Shift Data

Table 1 provides chemical shifts of spin systems identified in 3D hCCH-TOCSY data of U-¹H,¹³C,¹⁵N-KcsA in proteoliposomes at pH 7.2. Spin systems were type assigned based on characteristic chemical shifts. Lys-131 is assigned based on it being the only lysine residue in the C-terminus and that spin system is absent in KcsA- Δ 125 spectra. The deviation from random coil of type-assigned ¹³C α resonances from Table 1 are plotted in Figure 7.

Residue	Cα	Cβ	Cγ	Cδ1	Cδ2	CG	Hα	Ηβ1	Ηβ2	Hγ	Нδ
Lys 131	57.5	32.6				42.1	3.83				
Arg a	57.1	30.4	25.6	41.8			3.86	2.09		1.63	3.15
Arg b	57.0	30.2	25.6	41.8			3.83	2.06		1.63	3.14
Arg c	57.4	30.6	27.8	42.1			3.77	1.96		1.62	3.14
Leu a	56.3	42.7		23.7	24.9		3.81	1.82	1.76		
Leu b	56.3	42.6		23.8	24.8		3.81	1.82	1.76	1.82	
Phe/Tyr a	58.9	38.3					4.01	3.28			
Phe/Tyr b	58.9	38.4					4.02	3.13			
Glu a	57.5	29.7	36.4				3.83	2.12		2.44	
Glu b	57.1	29.6	36.4				4.08	2.20		2.44	

Table 1. Amino acid types identified in hCCH-TOCSY experiment of KcsA in 9:1 DOPE-DOPS at pH 7.2.

Chemical Shift Deviation from Random Coil



Figure 7. Chemical shift deviations from random coil of residues in the KcsA C-terminus at neutral pH condition KcsA using hCCH-TOCSY data, U-¹H, ¹³C, ¹⁵N-KcsA in proteoliposomes (9:1 DOPE-DOPS) at pH 7.2, 50 mM K⁺.

3.8. Lipid Stability in NMR Samples

Solid-state NMR samples routinely spend a week or more in the probe. We made near-complete ¹H and ¹³C assignments of lipids (except highly degenerate resonances in the aliphatic chain) in KcsA proteoliposome samples (Table 2) using hCCH-TOCSY experiments (Supplementary Materials Figure S7). At least six resonances frequently appeared in KcsA proteoliposome data that do not correspond to known lipid species or that are consistent with protein chemical shifts. Chemical shifts of correlated spin systems correspond to the literature values of free glycerol and free ethanolamine [45,46]. Neither glycerol nor ethanolamine are used in the preparation of our samples, and the signal strength from these compounds is too high to suggest trace contaminants. Chemical logic suggests that these components result from degradation of the lipids. The chemical shifts of degradation products are documented in Table 2, and their presence (or absence) in various samples is further with additional detail is in Supplementary Materials Tables S2 and S3. We and others have previously reported that exogenous lipids co-purify with KcsA expressed in E. coli containing a phosphoglycerol headgroup [22]. The copurified lipids can be identified because they are expected (like the purified protein) to have uniform enrichment of ^{13}C and ¹⁵N, which is in contrast to exogenously lipids that are added during reconstitution, which would have natural abundance of ¹³C and ¹⁵N content. The most plausible source of ethanolamine is from hydrolyzed exogenous PE lipid headgroups. We have not identified

the catalyst for this hydrolysis. In our samples, HSQC signal for free glycerol is invariably much greater than that of free ethanolamine (when ethanolamine is detectable at all). These data suggest to us that at least some of the glycerol arises from hydrolysis of co-purifying ¹³C-enriched PG. Phosphoglycerol or free glycerol signal is present in nearly all KcsA samples, with most containing signals from both.

Table 2. Chemical shift assignments for liposome lipids and lipid hydrolysis products measured in this paper at pH 4–7, 50 mM KCl, 308 K, 5 kHz MAS. See Figure 8 for nomenclature.

Site	¹ H (ppm)	STD	¹³ C (ppm)	STD
Fatty acid 12 (ω)	0.88	0.018	16.6	0.12
Fatty acid 11 ($\omega - 1$)	1.3	0.017	25.3	0.06
Fatty acid 8 (CH2-HC=C)	2.02	0.038	29.9	0.19
Fatty acid 6 (CH2)	1.29	0.022	32.1	0.1
Fatty acid 10 (ω – 2)	1.27	0.02	34.6	0.08
Fatty acid 4 (α -CH2)	2.33	0.033	36.6	0.24
ΡΕ β	3.24	0.037	43.3	0.37
Ethanolamine β	3.19	0.088	44.2	0.1
Ethanolamine α	3.84	0.095	60.3	0.2
PE α	4.11	0.027	64.6	0.48
PG γ + glycerol a	3.67	0.025	65.4	0.1
PG γ + glycerol b	3.61	0.044	65.4	0.09
G3 a	4.23	0.028	67	2.2
G3 b	4.46	0.027	67	2.3
G1	4.3	0.23	67.5	3.4
PG α a	3.94	0.022	70	1.8
PG a b	3.88	0.025	70	1.8
PG β	3.93	0.028	72	2.2
G2	5.26	0.012	73.3	0.15
glycerol (CH1)	3.83	0.035	74	2.4
Fatty Acid 9 (HC=C)	5.31	0.027	132.2	0.12



Figure 8. Structure of lipids of interest to this study and their nomenclature.

In Figure 9, we demonstrate that a direct polarization ¹H experiment with presaturation conducted at 308 K in an MAS probe is sufficient to diagnose the presence of lipid hydrolysis products. Although the spectra we display were collected in an HR-MAS probe, we also find that this experiment can be accomplished effectively in an ¹H/¹³C/¹⁵N e-free CP-MAS probe configured for ¹H detection. The sharp peaks arising from mobile solutes of lipid hydrolysis were identified by chemical shift and confirmed through ¹H-¹³C HSQC and hCCH-TOCSY data (see Supplementary Materials Figure S4 for example assignment data). The principal small molecule (sharp-peaked) signal is from buffer components and water-soluble hydrolysis products of lipids such as glycerol and ethanolamine.

We took a survey of preserved samples from previous studies that used KcsA liposomes from our group and for each, we investigated the presence of lipid hydrolysis product by ¹H-¹³C HR-MAS HSQC. The spectra were examined for the presence of PG, ethanolamine, and glycerol, and the chemical shifts of these compounds were then used to assign peaks in quantitative ¹H spectra of the same samples. The results are summarized in Supplementary Materials Table S2. The shifts from HSQC data were used to assign the peaks in the ¹H data, and thereby the presence or absence of the glycerol and ethanolamine was verified.



Figure 9. Quantitative ¹H direct-excitation MAS spectra (with water presaturation) of KcsA proteoliposome samples. Samples that contained free ethanolamine signal are displayed in red (A and B). The samples that contain only PE (intact lipid) and no ethanolamine are displayed in black (C and D). All spectra were collected at 308 K, 5 kHz MAS. Spectra are normalized to the bulk CH₂ signal (~1.2 ppm). All samples are U-¹H-¹³C-¹⁵N-KcsA in 9:1 DOPE liposomes, LPR = 1, by mass with additional details shown in the figure. Free glycerol, ethanolamine (E.A.), and citrate peaks are labeled.

3.9. Effect of MAS Centrifugation on Samples

Magic angle spinning places samples under intense centrifugal forces. So, we conducted experiments to attempt to mitigate these forces. The centrifugal forces on the sample during MAS are described by:

$$RCF = 4.025 \times 10^{-3} rQ^2 \tag{1}$$

where *RCF* is given in *g*-force (g), r is the radius from the center of the rotor in millimeters and Q the spinning frequency in Hz, wherein sample fractions at the rotor wall experience the greatest forces.

We hypothesized that varying MAS might lead to changes in protein structure that could be identified in spectra. To test this, we compared ¹H-¹³C HSQC spectra of KcsA in

liposomes at 5 kHz versus 15 kHz MAS, increasing the RFC by a factor of 9 at the rotor wall, but did not identify any systematic differences. Further, we have found MAS speeds more than 10 kHz for these samples lead to rotor crashes, whereas we had no such difficulty at 5 kHz.

As an alternative, we sought to balance centrifugal forces with buoyant forces by density matching the sample to the buffer. We pelleted KcsA following dialysis, and then subjected the proteoliposomes to 25 rounds of freeze–thaw using liquid nitrogen and a 30 °C water bath. This led to the formation of proteoliposomes of unilamellar as well as multilamellar or oligomellar structures (Supplementary Materials Figure S6) similar to previous reports [47,48]. To characterize the proteoliposomes resulting from this freeze–thaw procedure, we loaded the samples onto a buffer and uniform sucrose gradient (5–60%) and performed isopycnic ultracentrifugation. In order to visualize the proteoliposomes on the column and quantify the lipid concentration, we included a rhoadamine-congated lipid (1,2-dioleoyl-sn-glycero-3-phosphoethanolamine-N-(lissamine rhodamine B sulfonyl) (Rhod PE) in the lipid mixture. The lipid mixture had a mass ratio of 900:10:1 PE-PS-Rhod PE, and the total lipid to protein mass ratio was 1:1.

Following centrifugation, aliquots were taken, and portions of each aliquot were used to determine protein concentration by solubilizing proteoliposomes with a solution of Triton-X100 and bromophenyl blue and comparing UV-VIS absorbance at 610 nm with a standard curve [49]. Total lipids were extracted from aliquots using chloroform and methanol [50] and quantified using a standard curve based on UV-VIS absorbance of Rhod-PE at 560 nm. Density of aliquots was determined by analytical balance and calibrated micropipette using low retention tips.

The isopycnic gradient experiment reveals distinct sets of populations distinguished by their densities (Table 3). All the data described in this paper (and many other studies) would presumably contain both of these populations as we were unable to efficiently separate these populations.

Table 3. Fractions of KcsA and DOPE–DOPS–Rhod-PE proteoliposomes from isopycnic sucrosegradient ultracentrifugation.

Aliquot Density (g/cm ³)	Lipid: Protein	Proportion of Loaded Protein (%)
1.02 ± 0.01	7 ± 1	20 ± 10
1.07 ± 0.01	0.8 ± 0.1	70 ± 10

We hypothesized that placing liposomes in a density-matched matrix would diminish the deleterious mechanical effects of magic angle spinning. However, despite many repeated attempts at isopycnic preparations, we were unable to achieve total protein concentrations of greater than 5% in aliquots. As NMR is a relatively insensitive technique, and therefore sample concentration is a critical feature of obtaining strong signal in a reasonable amount of time, we sought alternative methods to density-match the sample.

MAS can be expected to exert pressure on the proteoliposomes, potentially changing their morphology or hydration, and therefore affect both function and NMR detection. To reduce net force on the proteoliposomes, we sought to prepare an isopycnic solution of the proteoliposomes. To roughly match the density of the KcsA pellet to the buffer, we 'floated' the unilamellar proteoliposomes pellet by adding aliquots of 60% w/w sucroseaugmented buffer to a tube containing the pellet, then vortexed and centrifuged in a bench-top centrifuge and observed whether the pellet sank or floated. We found that this process can be reversed (i.e., cause a floating pellet to sink) by adding an aliquot of buffer without sucrose to the tube. The process can be repeated in either direction indefinitely. Thus, this experiment was dubbed a reversible 'elevator' experiment and provided a means to titrate the density to match KcsA proteoliposomes. Because we used transparent rotor inserts, we were able to verify that the floating pellet does indeed migrate to the center of the rotor during MAS (Figure 10). From the observation that the sample floats in a high-sucrose background, we assert that under these conditions, the sample experiences greater buoyant force than centrifugal force. Further, the floating sample migrates to the center of the rotor, where *g*-forces diminish to zero. We estimate the maximum radius of a pelleted sample at 0.96 mm (the inner diameter of the rotor insert) and the maximum radius of a floating sample at 0.63 mm (Supplementary Materials Figure S11), reducing the maximum centrifugal force by a factor of 0.66.

We did not identify dramatic differences in ¹H-¹³C or ¹H-¹⁵N HSQC by HR-MAS spectra of KcsA unilamellar proteoliposomes that were 'floating' versus pelleted (data not shown). Cryo-electron microscopy (Supplementary Materials Figure S6) shows that before MAS, many small unilamellar vesicles are present. Following MAS, no unilamellar vesicles could be found. Instead, post-MAS vesicles in pelleted and floating samples are less well-ordered and show multilamellar, oligolamellar, and aggregated morphology. This suggests that the floating samples are not protected from changes to lipid morphology under MAS conditions.



Figure 10. KcsA proteoliposomes in MAS rotor inserts post-experiment. Left: 'floating' sample in sucrose-augmented buffer, and Right: 'sinking' sample with no sucrose in buffer.

4. Discussion

We present evidence that KcsA in model liposomes assumes distinct conformations at pH 7.25, where the channel is inactive, and at pH 4.0, under which conditions the channels can activate. Here, we use the same lipid system used previously to model KcsA activation and inactivation states [25] and allosteric coupling between sites within KcsA [3,26]. Thus, sample conditions in this study are appropriate for studies that connect channel structure to channel function.

The conformational switch presented here is consistent with previously proposed structural models. A sequential spin-labeled ESR structure of KcsA in bilayers indicated significant mobility of the C-terminus at pH 4 [29]. On the other hand, full-length KcsA with its C-terminus stabilized by Fab antibodies at neutral pH was rigid enough to capture using X-ray crystallography [1].

This study and others [37,51] report two distinct dynamical regimes of the KcsA C-terminus as a function of pH. Dramatic changes can be observed in the ¹H-¹⁵N HSQC data (Figure 2), with many more peaks with typically much narrower line widths at low pH compared to neutral pH. This is particularly clear when comparing fractionally deuterated samples but is also observed with fully protonated samples, showing reproducibility and independence of the labeling scheme. These data indicate that as the pH is lowered, the KcsA the C-terminus moves more freely.

Our analysis of hCCH-TOCSY spectra resolved ten groups of peaks that can be assigned by amino acid type (summarized in Table 1). These data alone established an additional new site-specific spectral assignment (Lys-131). We also identify three arginine, two leucine, two aromatic (Phe or Tyr), and glutamic acid spin systems. We are unable to definitively distinguish between Phe/Tyr, though Tyr is more likely than the respective alternative based on canonical chemical shift data. All these spin systems are absent in spectra of KcsA- Δ 125. So, the resonances are almost certainly from the C-terminus. Previous X-ray structures of full-length KcsA in the closed confirmation [1] and whose data agree with EPR structural data [29] suggest that, in the C-terminus, Phe-125, Arg-127, Lys-131, Glu-135, Arg-139, Arg-142, Glu-146, Arg-159, Arg-160 are solvent-exposed and are therefore most likely to have resolved side-chain resonances. The most compact sequence of amino acids that conforms to the evidence would be from Phe-125 to Leu-151 (FVRHSEKAAEEAYTRTTRALHERFDRL) (see Figure 3 for a visualization of these potential sites). Even within this short sequence, several arginine and glutamic acid residues would be unresolved amongst the amino acids we have type-assigned. According to the previous structural models, all the candidate sites for hydrophilic residues (Lys, Arg X3, Glu X2) are on the solvent-exposed surface, whereas the leucine side chains are likely to be facing other protein subunits and the phenylalanine and tyrosine.

The ¹³C α resonances of type-assigned residues have systematically higher shift values than mean values for random coil (Figure 7), suggesting that the residues we detect are likely to be in helical arrangement [52]. Several other studies have examined chemical shift deviations of KcsA, typically finding that the protein has particularly elevated shift values (often > 6 ppm) for tetramers in detergent micelles [12] and similar behavior in the C-terminal domain alone solubilized in water when it spontaneously tetramerizes at a high concentration [9]. Both those studies, however, also find small regions within the C-terminus at neutral pH that have less dramatic shift elevations (+1–2 ppm), which are putatively characterized as more dynamic, less rigid helices. Crystal structures [1,29] and EPR [53] measurements of KcsA in lipid mimics at neutral pH further support the hypothesis of the C-terminus existing as series of linked helices connected by more dynamic regions. Our study, conducted in the lipid bilayer, comports with this structural model. Specifically, it is likely that the resonances we detect are from the more mobile portions and that the most rigid portions of the C-terminus in the tightest helices remain unresolved under our experimental conditions.

This work identifies leucine 1 H- 13 C δ HSQC peaks that provide markers for the low versus high pH states (Figure 5). Specifically, we reproducibly observe a set of sharp leucine peaks at neutral pH and at low pH more peaks that are generally broader. This same behavior has been previously observed of the C-terminal domain alone solubilized in water [9]. EPR measurements in particular [53] suggest thatLeu-151 undergoes significant conformational rearrangement when the channel goes from a neutral-pH to a low-pH environment. We resolve additional leucine resonances at low pH, underscoring our conclusions that the KcsA C-terminus becomes more mobile at low pH. Yet, it is not clear why in our data the leucines in the C-terminus show the opposite trend of other resonances in becoming broader. These data cannot distinguish if the apparent line broadening is due to relaxation effects or due to conformational heterogeneity. Regardless, these findings support the use of leucine as markers for high-pH and low-pH conditions.

The data presented here show that HR-MAS is a viable method to investigate the structure of KcsA's C-terminal domain. There are several interesting functional questions ready to be addressed with this system. For example, from previous studies we know that depleting the system of K⁺ in the low pH state leads to structural changes and subsequently to the inactivation of KcsA with a strong allosteric connection between binding of potassium at the selectivity filter and protonation of pH sensors at Glu-118 and Glu-120 [3]. HR-MAS could detect whether the C-terminus also has an allosterically induced conformational change, for example reverting to the more rigid structure upon depletion of K⁺ from the selectivity filter.

4.1. Considerations for Lipid Model Systems

In these pages, we demonstrate that ¹H-detected NMR can be conducted on KcsA embedded in proteoliposomes, a highly cell-like model system, whereas previous ¹H-detected studies of KcsA have been performed in model systems somewhat more removed from a native-like environment. Lipid–protein nanodiscs have shown promise for the study of KcsA. Work on KcsA in PC nanodiscs showed that the C-terminus has a pH-dependent conformation, which this work supports in 9:1 DOPE-DOPS liposomes [37]. This suggests that the lipid milieu does not determine the C-terminal helix bundle dissociation at low pH. One suggested function of the C-terminus helix bundling is that the contacts made across monomers may aid the stabilization of the KcsA tetramer at neutral pH [51]. Those data were collected on a solution-NMR instrument and found that resonances were too broad and the T₂ values too short to conduct 3D backbone experiments. Here, we have demonstrated the ability to conduct 3D experiments in liposomes, and we might see significant increases in resolution and improvements in T_2 relaxation by combining both nanodiscs and HR-MAS. Here, proteoliposomes are not expected to be rapidly tumbling. Instead, the protein diffusion through the lipid bilayer is the largest source of translational motion for the entire protein, and then individual domains, such as the C-terminus, are likely to undergo further localized movement that can decrease anisotropic interactions. Lipid composition varies the rate at which lipids and proteins diffuse through the membrane surface. The lipid composition and the protein-to-lipid ratio could be tuned to produce more rapid diffusion, which might lead to improved relaxation and resolution characteristics. Lateral diffusion of lipid probes is inversely proportional to bilayer thickness [54], so shorter chains might prove helpful. Lipid hydration is very important in determining diffusion, with maximum diffusion rates occurring above 40 percent water by mass [55]. Lipid acyl chain composition changes the diffusion rates, with DPPC (saturated lipid with 16 carbon chain), diffusing at more than two times the rate of DOPC (single cis-double bond per chain and 18 carbon chains) [56]. Headgroups play a dramatic role as well, with DOPG diffusing at twice the rate of DOPC [54], and single-component PG membranes have been shown to produce functioning channels. The rate of protein diffusion is not only lipid-dependent but also protein-dependent [54], and there is no good framework, of which we are aware, to predict the rate of diffusion of a particular protein in a mixture of lipids. Therefore, a lipid screen might be able to optimize the rate of diffusion of KcsA in the membrane. In this study, we used a lipid mixture selected here to previously connect structure and function. Future lipid surveys would be wise to begin with lipids shown to support active channels.

4.2. Effect of Magic Angle Spinning on Samples

Many protocols commonly used during MAS studies, including strong radio frequency (RF) irradiation for spectral decoupling and excessive *g*-forces from high spin rates, may be detrimental to proteoliposomes. Biophysical studies of membrane proteins operate on the assumption that the lipid mixture of the model systems is well-defined and does not change over the course of an experiment. The availability and distribution of conformational states may conceivably be affected by these forces. The bulk properties of the lipids, such as viscosity, lateral pressure, and other factors that affect the ability of protein to diffuse through the bilayer, are also likely to be influenced by these networks of protons, where neighboring proton pairs typically have couplings more than 30 kHz. Therefore, experiments often apply MAS rates and RF fields to achieve 100 kHz or more of decoupling to obtain narrow lines, applying extreme centrifugal forces on samples.

For example, 3.2 mm rotors typically have a maximum safe spinning frequency of 24 kHz and an inner diameter of 2.17 mm, meaning at top frequency, objects at the rotor wall (from Equation (1) above) experience a relative centrifugal force of $5.0 \times 10^6 \times g$. The most advanced equipment can result in substantially stronger forces. Even relatively slow spinning frequencies are known to damage biological tissue. Lipid-laden mouse adipocytes experience nearly 20% lysis after 2 h of MAS at $23 \times g$ [57]. Spinning at $20 \times g$ for 1 h substantially alters human prostate tissue morphology [48]. Under our experimental conditions, proteoliposomes migrate to the rotor wall during MAS, effectively pelleting samples.

Isopycnic sample preparation by sucrose gradient ultracentrifugation led to dilute proteoliposome samples in a high concentration of sucrose, which contributes significant background signal to spectra. These are obviously suboptimal conditions for NMR. While not useful for spectroscopy, these experiments did uncover a bimodal distribution of proteoliposome densities. This finding underscores a major theme throughout these studies: the typical methods used to prepare proteoliposomes for NMR leads to samples that are of heterogenous nature, and solid-state NMR samples deserve more scrutiny by complimentary biophysical techniques.

The alternative method of adding highly concentrated sucrose to samples to make the buffer slightly denser than the proteoliposomes to form a "floating pellet" led to a scenario in which the liposomes migrate to the region where the least amount of centrifugal force is present, namely the center of the rotor, where the *g*-force approaches zero. This allowed for much higher sample concentrations, but with still significantly less sample than a fully packed rotor and with significant small molecule background signal. The lack of dramatic changes to spectra does not justify the use of this technique routinely. However, the slight perturbations we observe suggest that this topic may warrant further investigation.

These studies reiterate our group's previously reported findings that the structure of proteoliposome samples is heterogeneous and poorly defined. Future investigations into the effect of magic angle spinning on sample structure would likely benefit from an alternative lipid selection. We have shown that proteoliposomes with DOPC (phosphocholine) headgroups form more predictably spherical structures that would be a better candidate for future investigations [22].

4.3. Lipid Degradation of KcsA-Proteoliposomes

This work supports previous findings [58] that fatty acid esters can be readily hydrolyzed under certain conditions. We have observed that on the benchtop in buffer at neutral pH, DOPE and DOPS mixtures appear to be unreacted after two months of incubation. It was surprising to find evidence of hydrolysis of fatty acid chains and lipid headgroups in proteoliposome samples with this lipid composition. In one sample, we analyzed that for samples prepared for past experiments, the lipid head groups appear to be entirely hydrolyzed. The catalytic agent leading to hydrolysis in proteoliposome samples has not yet been identified.

Free glycerol ¹H NMR peaks for the hydrolyzed samples are of the same magnitude as the peaks for free ethanolamine, suggesting that the glycerol peaks do not arise from the co-purifying lipids, but rather from the lipids added for reconstitution of liposomes. The most likely cause of the intense glycerol peaks in the ¹H spectra is that the lipid glycerol backbone is completely hydrolyzed from the phosphate and the fatty acid moieties. Many samples do not show any evidence for free glycerol but have signals consistent with phosphoglycerol lipid. From this, we can conclude that ¹³C-enriched PG co-purifies intact with KcsA in most cases we examined.

This study suggests that proteopliposome samples demand routine quality validation measurements, particularly for studies of the effect of lipid composition. HR-MAS requires specialized equipment and the data acquisition time for an ¹H-¹³C HSQC is at least 3 h, making this possible but somewhat onerous. However, here we show that a 1D proton spectrum is adequate to diagnosis the presence of hydrolyzed head groups. Such a procedure requires no (or very little) change to a CP-MAS experimental setup and can be collected in a matter of minutes.

5. Conclusions and Perspective

For many membrane proteins, structure and dynamics information on functionally crucial dynamic loops and termini is lacking. This study demonstrated the use of hybrid solution- and solid-state NMR methods to identify signals for mobile segments of an intrinsic membrane protein in proteoliposomes under functionally relevant conditions. We documented changes in dynamics of the C-terminus upon pH-triggered activation. We show that cleavage of the C-terminus has a destabilizing effect on distant residues in loops near the selectivity filter. We also show evidence that under typical experimental conditions, lipids in proteoliposomes degrade into small molecules with sharp resonances

18 of 21

and demonstrate fast and practical NMR experiments that can detect the presence of degradation products.

Supplementary Materials: The following are available online at https://www.mdpi.com/article/ 10.3390/biom12081122/s1, Figure S1: SDS PAGE of KcsA C-terminus cleavage, Figure S2: HSQC of KcsA proteoliposomes, Figure S3: CP-MAS NMR of KcsA proteoliposomes, Figure S4: Example of KcsA type assignment from hCCH-TOCSY data, Figure S5: Isopycnic gradient ultracentrifugation of KcsA proteoliposomes, Figure S6: Electron micrographs of KcsA liposomes, Figure S7: Lipid degradation in KcsA proteoliposome samples from TOCSY Data, Figure S8: Planes of intra-residue 3D HNCA of KcsA proteoliposomes, Figure S9: Planes of inter-residue (i + 1) 3D HNCO of KcsA proteoliposomes, Figure S10: Residue types identified in hCCH-TOCSY data of KcsA, Figure S11: HR-MAS sample radius, Table S1: KcsA ¹³C T_2 s, Table S2: ¹H-¹³C HSQC per sample chemical shifts and relative intensities of various species, Table S3: KcsA proteoliposome sample details, and Detailed Materials and Methods [59–65].

Author Contributions: Conceptualization, G.S.H. and A.E.M.; methodology, G.S.H. and A.E.M.; software, G.S.H.; validation, G.S.H. and A.E.M.; formal analysis G.S.H. and A.E.M.; investigation, G.S.H.; resources, G.S.H. and A.E.M.; data curation, G.S.H.; writing—original draft preparation, G.S.H.; writing—review and editing, G.S.H. and A.E.M.; visualization, G.S.H.; supervision, A.E.M.; project administration, G.S.H. and A.E.M.; funding acquisition, A.E.M. All authors have read and agreed to the published version of the manuscript.

Funding: This work was supported by NIH Grant R01 GM088724. Ann McDermott is a member of the New York Structural Biology Center (NYSBC). The NMR data were collected at the NYSBC with support from the Center on Macro-molecular Dynamics by NMR Spectroscopy, a Biomedical Technology Research Resource supported by the National Institutes of Health (NIH) through Grant P41 GM-118302. The NYSBC is also enabled by a grant from the Empire State Division of Science Technology and Innovation and by Office of Research Infrastructure Programs/NIH Facility Improvement Grant CO6RR015495. Data collected using the 750 MHz Avance I spectrometer is supported by the NIH (S10OD016432). The 900 MHz NMR spectrometers were purchased with funds from the NIH (P41GM066354), the Keck Foundation, New York State Assembly, and U.S. Dept. of Defense. Electron micrographs in this work were collected at the Simons Electron Microscopy Center and National Resource for Automated Molecular Microscopy located at the New York Structural Biology Center, supported by grants from the Simons Foundation (SF349247), NYSTAR, and the NIH National Institute of General Medical Sciences (GM103310), and the NIH National Center for Research Resources (C06 RR017528-01—CEM).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: NMR data and pulse sequences are available at https://comdnmr. nysbc.org/comd-nmr-dissem, (accessed on 27 June 2022).

Conflicts of Interest: The authors declare no conflict of interest.

References

- Uysal, S.; Vasquez, V.; Tereshko, V.; Esaki, K.; Fellouse, F.A.; Sidhu, S.S.; Koide, S.; Perozo, E.; Kossiakoff, A. Crystal Structure of Full-Length KcsA in Its Closed Conformation. *Proc. Natl. Acad. Sci. USA* 2009, 106, 6644–6649. [CrossRef] [PubMed]
- Cuello, L.G.; Jogini, V.; Cortes, D.M.; Pan, A.C.; Gagnon, D.G.; Dalmas, O.; Cordero-Morales, J.F.; Chakrapani, S.; Roux, B.; Perozo, E. Structural Basis for the Coupling between Activation and Inactivation Gates in K(+) Channels. *Nature* 2010, 466, 272–275. [CrossRef] [PubMed]
- Xu, Y.; Bhate, M.P.; McDermott, A.E. Transmembrane Allosteric Energetics Characterization for Strong Coupling between Proton and Potassium Ion Binding in the KcsA Channel. *Proc. Natl. Acad. Sci. USA* 2017, *114*, 8788–8793. [CrossRef]
- Gao, L.; Mi, X.; Paajanen, V.; Wang, K.; Fan, Z. Activation-Coupled Inactivation in the Bacterial Potassium Channel KcsA. *Proc. Natl. Acad. Sci. USA* 2005, 102, 17630–17635. [CrossRef] [PubMed]
- Imai, S.; Osawa, M.; Takeuchi, K.; Shimada, I. Structural Basis Underlying the Dual Gate Properties of KcsA. Proc. Natl. Acad. Sci. USA 2010, 107, 6216–6221. [CrossRef] [PubMed]
- Iwamoto, M.; Oiki, S. Amphipathic Antenna of an Inward Rectifier K+ Channel Responds to Changes in the Inner Membrane Leaflet. Proc. Natl. Acad. Sci. USA 2013, 110, 749–754. [CrossRef]

- Molina, M.L.; Giudici, A.M.; Poveda, J.A.; Fernández-Ballester, G.; Montoya, E.; Renart, M.L.; Fernández, A.M.; Encinar, J.A.; Riquelme, G.; Morales, A.; et al. Competing Lipid-Protein and Protein-Protein Interactions Determine Clustering and Gating Patterns in the Potassium Channel from Streptomyces Lividans (KcsA). J. Biol. Chem. 2015, 290, 25745–25755. [CrossRef] [PubMed]
- 8. Baker, K.A.; Tzitzilonis, C.; Kwiatkowski, W.; Choe, S.; Riek, R. Conformational Dynamics of the KcsA Potassium Channel Governs Gating Properties. *Nat. Struct. Mol. Biol.* **2007**, *14*, 1089–1095. [CrossRef] [PubMed]
- 9. Kamnesky, G.; Shaked, H.; Chill, J.H. The Distal C-Terminal Region of the KcsA Potassium Channel Is a PH-Dependent Tetramerization Domain. *J. Mol. Biol.* **2012**, *418*, 237–247. [CrossRef]
- 10. Kim, D.M.; Dikiy, I.; Upadhyay, V.; Posson, D.J.; Eliezer, D.; Nimigean, C.M. Conformational Heterogeneity in Closed and Open States of the KcsA Potassium Channel in Lipid Bicelles. *J. Gen. Physiol.* **2016**, *148*, 119–132. [CrossRef]
- 11. Takeuchi, K.; Takahashi, H.; Kawano, S.; Shimada, I. Identification and Characterization of the Slowly Exchanging PH-Dependent Conformational Rearrangement in KcsA. J. Biol. Chem. 2007, 282, 15179–15186. [CrossRef]
- 12. Chill, J.H.; Louis, J.M.; Miller, C.; Bax, A. NMR Study of the Tetrameric KcsA Potassium Channel in Detergent Micelles. *Protein Sci.* 2006, *15*, 684–698. [CrossRef]
- 13. Chill, J.; Louis, J.; Baber, J.; Bax, A. Measurement of 15N Relaxation in the Detergent-Solubilized Tetrameric KcsA Potassium Channel. J. Biomol. NMR 2006, 36, 123–136. [CrossRef]
- 14. Thompson, A.N.; Posson, D.J.; Parsa, P.V.; Nimigean, C.M. Molecular Mechanism of PH Sensing in KcsA Potassium Channels. *Proc. Natl. Acad. Sci. USA* **2008**, *105*, 6900–6905. [CrossRef]
- 15. Pau, V.P.T.; Zhu, Y.; Yuchi, Z.; Hoang, Q.Q.; Yang, D.S.C. Characterization of the C-Terminal Domain of a Potassium Channel from Streptomyces Lividans (KcsA). *J. Biol. Chem.* **2007**, *282*, 29163–29169. [CrossRef]
- 16. van den Brink-van der Laan, E.; Chupin, V.; Killian, J.A.; de Kruijff, B. Stability of KcsA Tetramer Depends on Membrane Lateral Pressure. *Biochemistry* **2004**, *43*, 4240–4250. [CrossRef]
- 17. Zhou, Y.; Morais-Cabral, J.H.; Kaufman, A.; MacKinnon, R. Chemistry of Ion Coordination and Hydration Revealed by a K+ Channel-Fab Complex at 2.0 A Resolution. *Nature* **2001**, *414*, 43–48. [CrossRef]
- Zhou, Y.; MacKinnon, R. The Occupancy of Ions in the K+ Selectivity Filter: Charge Balance and Coupling of Ion Binding to a Protein Conformational Change Underlie High Conduction Rates. J. Mol. Biol. 2003, 333, 965–975. [CrossRef]
- Zhou, M.; MacKinnon, R. A Mutant KcsA K+ Channel with Altered Conduction Properties and Selectivity Filter Ion Distribution. J. Mol. Biol. 2004, 338, 839–846. [CrossRef]
- Lenaeus, M.J.; Vamvouka, M.; Focia, P.J.; Gross, A. Structural Basis of TEA Blockade in a Model Potassium Channel. *Nat. Struct. Mol. Biol.* 2005, 12, 454–459. [CrossRef]
- Valiyaveetil, F.I.; MacKinnon, R.; Muir, T.W. Semisynthesis and Folding of the Potassium Channel KcsA. J. Am. Chem. Soc. 2002, 124, 9113–9120. [CrossRef] [PubMed]
- Zhang, D.; Howarth, G.S.; Parkin, L.A.; McDermott, A.E. NMR Studies of Lipid Regulation of the K+ Channel KcsA. *Biochim. Biophys. Acta (BBA) Biomembr.* 2021, 1863, 183491. [CrossRef] [PubMed]
- 23. Demmers, J.A.A.; van Dalen, A.; de Kruijff, B.; Heck, A.J.R.; Killian, J.A. Interaction of the K+ Channel KcsA with Membrane Phospholipids as Studied by ESI Mass Spectrometry. *FEBS Lett.* **2003**, *541*, 28–32. [CrossRef]
- Weingarth, M.; Prokofyev, A.; van der Cruijsen, E.A.W.; Nand, D.; Bonvin, A.M.J.J.; Pongs, O.; Baldus, M. Structural Determinants of Specific Lipid Binding to Potassium Channels. J. Am. Chem. Soc. 2013, 135, 3983–3988. [CrossRef] [PubMed]
- 25. Bhate, M.P.; McDermott, A.E. Protonation State of E71 in KcsA and Its Role for Channel Collapse and Inactivation. *Proc. Natl. Acad. Sci. USA* **2012**, *109*, 15265–15270. [CrossRef] [PubMed]
- Wylie, B.J.; Bhate, M.P.; McDermott, A.E. Transmembrane Allosteric Coupling of the Gates in a Potassium Channel. *Proc. Natl. Acad. Sci. USA* 2014, 111, 185–190. [CrossRef] [PubMed]
- 27. Schneider, R.; Ader, C.; Lange, A.; Giller, K.; Hornig, S.; Pongs, O.; Becker, S.; Baldus, M. Solid-State NMR Spectroscopy Applied to a Chimeric Potassium Channel in Lipid Bilayers. *J. Am. Chem. Soc.* **2008**, *130*, 7427–7435. [CrossRef] [PubMed]
- Lange, A.; Giller, K.; Hornig, S.; Martin-Eauclaire, M.-F.; Pongs, O.; Becker, S.; Baldus, M. Toxin-Induced Conformational Changes in a Potassium Channel Revealed by Solid-State NMR. *Nature* 2006, 440, 959–962. [CrossRef] [PubMed]
- 29. Cortes, D.M.; Cuello, L.G.; Perozo, E. Molecular Architecture of Full-Length KcsA. J. Gen. Physiol. 2001, 117, 165–180. [CrossRef]
- Zimmer, J.; Doyle, D.A.; Günter Grossmann, J. Structural Characterization and PH-Induced Conformational Transition of Full-Length KcsA. *Biophys. J.* 2006, 90, 1752–1766. [CrossRef]
- 31. Raja, M. The Role of Extramembranous Cytoplasmic Termini in Assembly and Stability of the Tetrameric K(+)-Channel KcsA. *J. Membr. Biol.* **2010**, 235, 51–61. [CrossRef]
- Molina, M.L.; Encinar, J.A.; Barrera, F.N.; Fernández-Ballester, G.; Riquelme, G.; González-Ros, J.M. Influence of C-Terminal Protein Domains and Protein-Lipid Interactions on Tetramerization and Stability of the Potassium Channel KcsA. *Biochemistry* 2004, 43, 14924–14931. [CrossRef]
- Bhate, M.P.; Wylie, B.J.; Tian, L.; McDermott, A.E. Conformational Dynamics in the Selectivity Filter of KcsA in Response to Potassium Ion Concentration. J. Mol. Biol. 2010, 401, 155–166. [CrossRef] [PubMed]
- Pérez-Conesa, S.; Keeler, E.G.; Zhang, D.; Delemotte, L.; McDermott, A.E. Informing NMR Experiments with Molecular Dynamics Simulations to Characterize the Dominant Activated State of the KcsA Ion Channel. J. Chem. Phys. 2021, 154, 1–15. [CrossRef]

- Mance, D.; Sinnige, T.; Kaplan, M.; Narasimhan, S.; Daniëls, M.; Houben, K.; Baldus, M.; Weingarth, M. An Efficient Labelling Approach to Harness Backbone and Side-Chain Protons in 1H-Detected Solid-State NMR Spectroscopy. *Angew. Chem.* 2015, 127, 16025–16029. [CrossRef]
- Weingarth, M.; van der Cruijsen, E.A.W.; Ostmeyer, J.; Lievestro, S.; Roux, B.; Baldus, M. Quantitative Analysis of the Water Occupancy around the Selectivity Filter of a K + Channel in Different Gating Modes. *J. Am. Chem. Soc.* 2014, 136, 2000–2007. [CrossRef] [PubMed]
- Qasim, A.; Sher, I.; Hirschhorn, O.; Shaked, H.; Qasem, Z.; Ruthstein, S.; Chill, J.H. Investigation of a KcsA Cytoplasmic PH Gate in Lipoprotein Nanodiscs. *ChemBioChem* 2019, 52900, 813–821. [CrossRef] [PubMed]
- Bhate, M.P.; Wylie, B.J.; Thompson, A.; Tian, L.; Nimigean, C.; McDermott, A.E. Preparation of Uniformly Isotope Labeled KcsA for Solid State NMR: Expression, Purification, Reconstitution into Liposomes and Functional Assay. *Protein Expr. Purif.* 2013, 91, 119–124. [CrossRef] [PubMed]
- 39. Schrödinger, LCC. The PyMOL Molecular Graphics System; Version 2.5; Schrödinger, LCC: New York, NY, USA, 2022.
- 40. Delaglio, F.; Torchia, D.A.; Bax, A. Measurement Of15N-13C J Couplings in Staphylococcal Nuclease. J. Biomol. NMR 1991, 1, 439–446. [CrossRef]
- 41. Lian, L.-Y.; Roberts, G. (Eds.) Protein NMR Spectroscopy; Elsevier: Chichester, UK, 2007; ISBN 9780121644918.
- 42. Kay, L.E.; Gardner, K.H.; Rosen, M.K.; Pawson, T.; Parris, W.E.; Willis, R.C. Selective Methyl Group Protonation of Perdeuterated Proteins. *J. Mol. Biol.* 2002, 263, 627–636. [CrossRef]
- LeMaster, D.M. Uniform and Selective Deuteration in Two-Dimensional NMR of Proteins. *Annu. Rev. Biophys. Biophys. Chem.* 1990, 19, 243–266. [CrossRef] [PubMed]
- 44. Shekhtman, A.; Ghose, R.; Goger, M.; Cowburn, D. NMR Structure Determination and Investigation Using a Reduced Proton (REDPRO) Labeling Strategy for Proteins. *FEBS Lett.* **2002**, 524, 177–182. [CrossRef]
- 45. Souvorov, A.; Benson, D.A.; Lipman, D.J.; Kenton, D.L.; Wheeler, D.L.; Church, D.M.; Maglott, D.R.; Sequeira, E.; Yaschenko, E.; Schuler, G.D.; et al. Glycerol. *Biol. Magn. Reson. Data Bank* **2011**. [CrossRef]
- 46. Souvorov, A.; Benson, D.A.; Lipman, D.J.; Kenton, D.L.; Wheeler, D.L.; Church, D.M.; Maglott, D.R.; Sequeira, E.; Yaschenko, E.; Schuler, G.D.; et al. Ethanolamine. *Biol. Magn. Reson. Data Bank* **2006**. [CrossRef]
- Mac Donald, R.C.; Mac Donald, R.I.; Menco, B.P.; Takeshita, K.; Subbarao, N.K.; Hu, L.R. Small-Volume Extrusion Apparatus for Preparation of Large, Unilamellar Vesicles. *Biochim. Biophys. Acta (BBA) Biomembr.* 1991, 1061, 297–303. [CrossRef]
- Stuart, M.C.A.; Boekema, E.J. Two Distinct Mechanisms of Vesicle-to-Micelle and Micelle-to-Vesicle Transition Are Mediated by the Packing Parameter of Phospho-Lipid-Detergent Systems. *Biochim. Biophys. Acta (BBA) Biomembr.* 2007, 1768, 2681–2689. [CrossRef]
- 49. Greenberg, C.S.; Craddock, P.R. Rapid Single-Step Membrane Protein Assay. Clin. Chem. 1982, 28, 1725. [CrossRef] [PubMed]
- 50. Bligh, E.G.; Dyer, W.J. A Rapid Method of Total Lipid Extraction and Purification. *Can. J. Biochem. Physiol.* **1959**, 37, 911–917. [CrossRef] [PubMed]
- Kamnesky, G.; Hirschhorn, O.; Shaked, H.; Chen, J.; Yao, L.; Chill, J.H. Molecular Determinants of Tetramerization in the KcsA Cytoplasmic Domain. *Protein Sci.* 2014, 23, 1403–1416. [CrossRef]
- 52. Wishart, D.S.; Sykes, B.D. The 13C Chemical-Shift Index: A Simple Method for the Identification of Protein Secondary Structure Using 13C Chemical-Shift Data. *J. Biomol. NMR* **1994**, *4*, 171–180. [CrossRef]
- Iwamoto, M.; Shimizu, H.; Inoue, F.; Konno, T.; Sasaki, Y.C.; Oiki, S. Surface Structure and Its Dynamic Rearrangements of the KcsA Potassium Channel upon Gating and Tetrabutylammonium Blocking. *J. Biol. Chem.* 2006, 281, 28379–28386. [CrossRef] [PubMed]
- 54. Ramadurai, S.; Duurkens, R.; Krasnikov, V.V.; Poolman, B. Lateral Diffusion of Membrane Proteins: Consequences of Hydrophobic Mismatch and Lipid Composition. *Biophys. J.* 2010, *99*, 1482–1489. [CrossRef] [PubMed]
- 55. Filippov, A.; Orädd, G.; Lindblom, G. Effect of NaCl and CaCl2 on the Lateral Diffusion of Zwitterionic and Anionic Lipids in Bilayers. *Chem. Phys. Lipids* **2009**, *159*, 81–87. [CrossRef]
- Orädd, G.; Westerman, P.W.; Lindblom, G. Lateral Diffusion Coefficients of Separate Lipid Species in a Ternary Raft-Forming Bilayer: A Pfg-NMR Multinuclear Study. *Biophys. J.* 2005, *89*, 315–320. [CrossRef] [PubMed]
- Weybright, P.; Millis, K.; Campbell, N.; Cory, D.G.; Singer, S. Gradient, High-Resolution, Magic Angle Spinning1H Nuclear Magnetic Resonance Spectroscopy of Intact Cells. *Magn. Reson. Med.* **1998**, *39*, 337–345. [CrossRef] [PubMed]
- Grit, M.; de Smidt, J.H.; Struijke, A.; Crommelin, D.J.A. Hydrolysis of Phosphatidylcholine in Aqueous Liposome Dispersions. *Int. J. Pharm.* 1989, 50, 1–6. [CrossRef]
- 59. Gill, S.C.; von Hippel, P.H. Calculation of Protein Extinction Coefficients from Amino Acid Sequence Data. *Anal. Biochem.* **1989**, 182, 319–326. [CrossRef]
- Heginbotham, L.; Kolmakova-Partensky, L.; Miller, C. Functional Reconstitution of a Prokaryotic K + Channel. J. Gen. Physiol. 1998, 111, 741–749. [CrossRef] [PubMed]
- 61. Shaka, A.J.; Keeler, J.; Freeman, R. Evaluation of a New Broadband Decoupling Sequence: WALTZ-16. J. Magn. Reson. 1983, 53, 313–340. [CrossRef]
- 62. Schleucher, J.; Schwendinger, M.; Sattler, M.; Schmidt, P.; Schedletzky, O.; Glaser, S.J.; Sørensen, O.W.; Griesinger, C. A General Enhancement Scheme in Heteronuclear Multidimensional NMR Employing Pulsed Field Gradients. *J. Biomol. NMR* **1994**, *4*, 301–306. [CrossRef] [PubMed]

- 63. Palmer, A.G.; Cavanagh, J.; Wright, P.E.; Rance, M. Sensitivity Improvement in Proton-Detected Two-Dimensional Heteronuclear Correlation NMR Spectroscopy. *J. Magn. Reson.* **1991**, *93*, 151–170. [CrossRef]
- 64. Kay, L.E.; Keifer, P.; Saarinen, T. Pure Absorption Gradient Enhanced Heteronuclear Single Quantum Correlation Spectroscopy with Improved Sensitivity. *J. Am. Chem. Soc.* **1992**, *114*, 10663–10665. [CrossRef]
- 65. Doyle, D.A.; Morais Cabral, J.; Pfuetzner, R.A.; Kuo, A.; Gulbis, J.M.; Cohen, S.L.; Chait, B.T.; MacKinnon, R. The structure of the potassium channel: molecular basis of K+ conduction and selectivity. *Science* **1998**, *280*, 69–77. [CrossRef] [PubMed]