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C_2-symmetric bisamidines: Chiral Brønsted bases catalysing the Diels-Alder reaction of anthrones

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Full Research Paper

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Abstract

C_2-symmetric bisamidines **8** have been tested as chiral Brønsted bases in the Diels-Alder reaction of anthrones and *N*-substituted maleimides. High yields of cycloadducts and significant asymmetric inductions up to 76% *ee* are accessible. The proposed mechanism involves proton transfer between anthrone and bisamidine, association of the resulting ions and finally a cycloaddition step stereoselectively controlled by the chiral ion pair.

Introduction

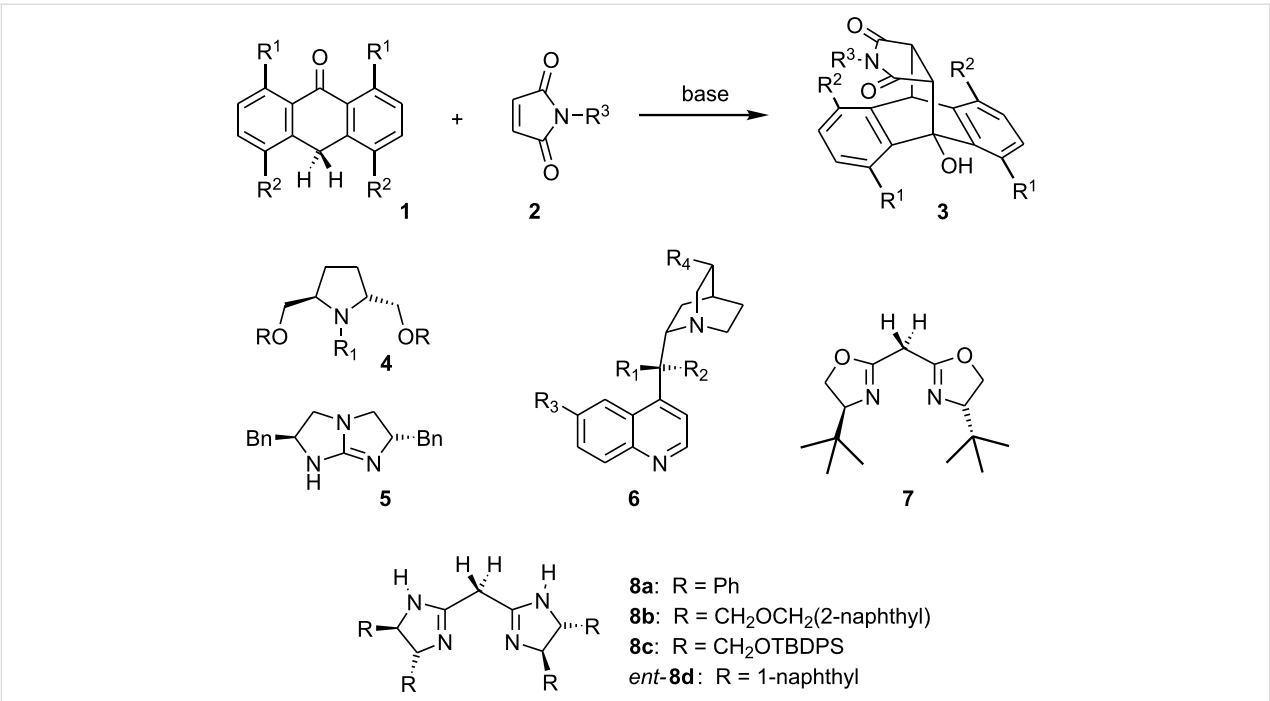
The cycloadditions of anthrones **1** and *N*-substituted maleimides **2** are prominent examples of asymmetric catalysis exerted by chiral Brønsted bases. Moderate to excellent stereoselectivities of products **3** have been reported using pyrrolidines **4** [1,2], cyclic guanidine **5** [3], or cinchona alkaloids **6** [4] as catalysts. Recently, we could promote this type of cycloaddition by metal-free bisoxazolines **7** in up to 70% *ee*, in spite of their limited Brønsted-basicity [5] (Scheme 1).

Our study was motivated by the structural similarity of bisoxazolines **7** and bisamidines **8**. Bisamidines **8**, readily accessible from malonodinitrile in two steps, prefer the conjugated tautomeric form (enamine-imine) in the monoprotonated state, which is characterised by an almost planar structure [6] (Scheme 2).

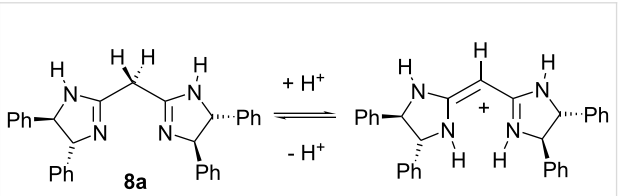
The aqueous pK_a of **8**·H$^+$ is approximately 11, sufficient to allow deprotonation of anthrones **1** (pK_a around 10, [7,8]) by bisamidines to a significant extent. Here we report on the use of neutral bisamidines **8** as asymmetric Brønsted base catalysts in the cycloaddition of anthrones **1** and maleimides **2**.

Results and Discussion

Analogous to the synthesis of compound **8a** [6], the other bisamidines were prepared as hydrochlorides in 60–79% yield from the corresponding chiral diamines **9** and bisimidate **10** in refluxing ethanol. Simple extraction in the presence of Na_2CO_3 afforded the neutral bases **8b–c** and *ent*-**8d** in almost quantitative yield. The *S*,*S* configurated diamines **9b** and **9c** were prepared from L-(+)-tartaric acid (*R*,*R*) via the vicinal diazide using Saalfrank's procedure [9]. **9d** was purchased as the



Scheme 1: Diels-Alder reaction of anthrones **1** and maleimides **2** catalyzed by chiral Brønsted bases **4**–**8**.



Scheme 2: Protonation states and tautomerism of C_2-symmetric bisamidine **8a** [6].

dihydrochloride salt and then deprotonated by aqueous sodium hydroxide. As an "artefact" of the sequence rule, the *S*,*S* configurated diamine **9d** leads to bisamidine *ent*-**8d** (Scheme 3).

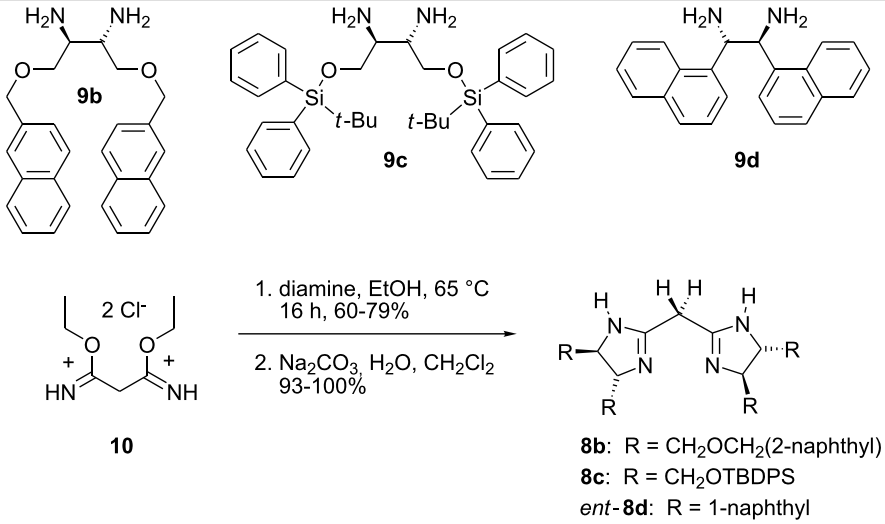
The anthrones **1b** (R^1: H; R^2: Cl) and **1c** (R^1: Cl; R^2: H) resulted from regioselective reductions of 1,8-dichloroanthraquinone [10,11]. Aliphatic side chains of compounds **2** could be introduced by a Mitsunobu alkylation of maleimide [12]. Alternatively, substituted maleimides were prepared by reaction of maleic anhydride with the corresponding amines followed by ring closure [13,14].

Cycloaddition kinetics of **1a** and **2a** was examined first by ^{1}H NMR in CD_2Cl_2 at room temperature. In the absence of catalyst, no product could be observed after 4 days. 5 mol% of the bisamidinium salt **8a**·H^+ with tetrakis(3,5-bis-(trifluoromethyl)phenyl)borate ($TFPB^-$) as weakly coordinating anion resulted in 7% yield of **3a** after 4 h. In contrast, only 1 mol% of the free Brønsted base **8a** led to a high rate increase in the first 30 min. After 90 min no further conversion was observed indicating product inhibition (Figure 1). Accordingly, the reaction runs best in the base-catalyzed mode. Compared to the bisoxazolines **7**, bisamidines **8** as stronger Brønsted bases induced much higher rates in all subsequent experiments.

In the next series of experiments, bisamidines **8a**–**c** and *ent*-**8d** were compared as catalysts of the cycloaddition forming **3a** from *N*-phenylmaleimide (**2a**) and anthrone (**1a**). Using 0.25 equiv of catalyst at room temperature, isolated yields between 71% and 86% were obtained after 30 min. The best enantioselectivity, albeit low, was induced by amidine **8c** (24% *ee*). As expected, in the presence of catalyst *ent*-**8d** product *ent*-**3a** was formed preferentially (Table 1).

In a solvent screening using 10 mol% of TBDPS-protected bisamidine **8c**, best results were obtained in dichloromethane (84% yield; 30% *ee*). Even higher yields were accessible in aromatic solvents, however, at the price of reduced stereoselectivity (Table 2).

Lowering the reaction temperature from 23 to −20 °C (**8c**, dichloromethane) retarded the cycloaddition but did not change enantioselectivities. After extended reaction times, excellent yields were still observed. Up to 39% *ee* was finally obtained at −70 °C. However, such conditions resulted in lower yields, even with increased catalyst loads and further extended reac-



Scheme 3: Synthesis of C_2-symmetric bisamidines **8b–c** and *ent*-**8d**.

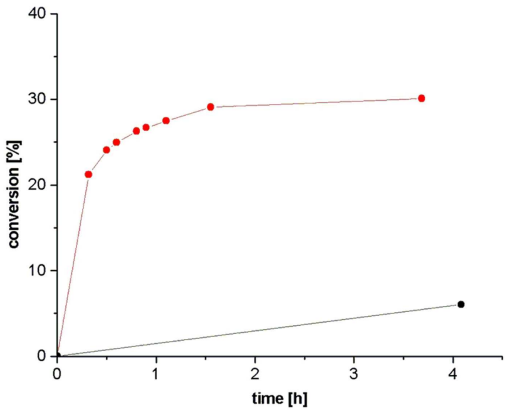


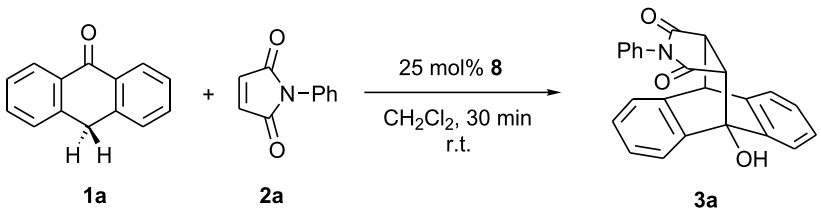
Figure 1: Kinetic measurements of **1a** with **2a** catalyzed by 5 mol% of **8a**·H^+·$TFPB^-$ (black line) and 1 mol% **8a** (free base; red line).

tion times. Best results, 96% yield and 36% *ee* with only 10 mol% of catalyst, were found at −40 °C (Table 3).

Having identified suitable experimental conditions, we explored the scope of the bisamidine-catalyzed Diels-Alder reaction. The results are summarized in Table 4. Both electron-donating and electron-withdrawing substituents were tolerated and furnished products in good to excellent yields and with moderate values of *ee*. A remarkable increase in enantioselectivity was observed using maleimide **2i**. The steric hindrance imposed by the large 2,6-diisopropylphenyl moiety of **2i** resulted in 76% *ee* at −70 °C but also lowered reaction rates.

Only 13% yield could be obtained under such conditions. Yields rose to 65% at room temperature (51% *ee*; entries 11 and 12). With other sterically hindered dienophiles such as *N-tert-*

Table 1: First evaluation step of chiral bisamidine catalysts.



entry[a]	catalyst	yield [%][b]	*ee* [%][c]
1	**8a**	86	11
2	**8b**	78	17
3	**8c**	85	24
4	*ent*-**8d**	71	−17[d]

[a]All reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.25 equiv of catalyst in 1 mL abs. dichloromethane at room temperature for 30 minutes. [b]Isolated yield after column chromatography. [c]The enantiomeric excess was determined by HPLC using a Chiralpak IA column. [d]A negative *ee* stands for an excess of *ent*-**3a**.

Table 2: Influence of the solvent on the bisamidine catalyzed Diels-Alder reaction.

entry[a]	solvent	yield [%][b]	*ee* [%][c]
1	dichloromethane	84	30
2	chloroform	86	18
3	benzene	98	21
4	toluene	99	16
5	α,α,α-trifluorotoluene	99	13
6	dibutyl ether	89	11

[a]All reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.1 equiv of **8c** in 1 mL abs. solvent at room temperature for 60 minutes. [b]Isolated yield after column chromatography. [c]The enantiomeric excess was determined by HPLC using a Chiralpak IA column.

Table 3: Influence of temperature on the Diels-Alder reaction.

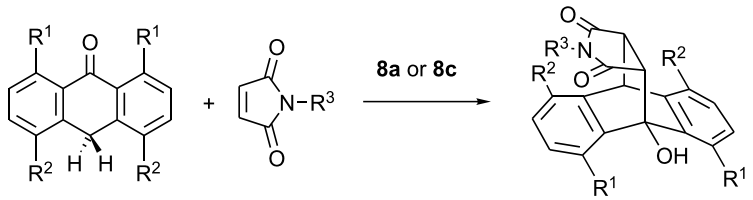
entry[a]	reaction temperature [°C]	reaction time [h]	yield [%][b]	*ee* [%][c]
1	23	1	84	30
2	0	24	96	29
3	−20	24	98	31
4	−40	48	96	36
5	−70	96	71	39

[a]All reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.1 (entry 1–4) or 0.25 equiv (entry 5) of **8c** in 1 mL abs. dichloromethane. [b]Isolated yield after column chromatography. [c]Enantiomeric excess was determined by HPLC using Chiralpak IA column.

butylmaleimide (**2c**), the level of *ee* remained low (entry 4). The halogen-substituted anthrones **1b**–**c** did not react with **2i** at −70 °C. At room temperature, however, **1b** and **2i** were efficiently transformed into **3m** by catalyst **8a** with 76% yield and 54% *ee*. A single recrystallisation step afforded an almost enantiopure product (96% *ee*). The *R*,*R* configuration of compound **3m** was determined by anomalous X-ray diffraction using a single crystal of **3m** with 96% *ee* (Figure 2).

A mechanistic rationalisation is proposed in Scheme 4. The catalyst deprotonates the anthrone in the initial step. This assumption is supported by the pK_a values of compounds **2a** (10, [7,8]) and **8**·H$^+$ (~11, [6]). Furthermore, the appearance of the yellow color of enolates (**1**·H$^+$) shows significant proton transfer when bisamidine **8a** is added to anthrones **1a**, **1b**, or **1c**. A chiral contact ion pair **A** is formed and controls the stereochemical course of the Diels-Alder reaction with maleimides. In the last step, the catalyst-product-complex **B** dissociates and regenerates the unprotonated bisamidine.

Table 4: Scope of the Diels-Alder-reaction.



1a-c **2a-i** **3a-m**

entry[a]	**1** [R^1, R^2]	R^3	condition[b]	**3**	yield [%][c]	*ee* [%][d]
1	**1a** [H, H,]	Ph (**2a**)	A	**3a**	96	36
2	**1b** [H, Cl]	**2a**	A	**3b**	95	41
3	**1a**	iPr (**2b**)	B	**3c**	74	26
4	**1a**	*t*-Bu (**2c**)	B	**3d**	45	30
5	**1a**	Cy (**2d**)	B	**3e**	83	42
6	**1c** [Cl, H]	**2d**	B	**3f**	90	19
7	**1a**	Bn (**2e**)	A	**3g**	95	20
8	**1a**	$CHPh_2$ (**2f**)	A	**3h**	85	26
9	**1a**	4-Br-(C_6H_4)- (**2g**)	B	**3i**	70	13
10	**1a**	4-MeO-(C_6H_4)- (**2h**)	A	**3j**	82	32
11	**1a**	2,6-iPr_2-(C_6H_3)- (**2i**)	B	**3k**	13	**76**
12	**1a**	**2i**	C	**3k**	65	51
13	**1c**	**2i**	C	**3l**	77	34
14	**1b**	**2i**	C	**3m**	76	54 (**96**)[e]

[a]All reactions were carried out using 0.1 mmol maleimide, 1.1 equiv anthrone in 1 mL abs. CH_2Cl_2. [b]A = 10 mol% **8c**, −40 °C, 48 h; B = 25 mol% **8a**, −70 °C, 96 h; C = 25 mol% **8a**, r.t., 3 h. [c]Isolated yield after column chromatography. [d]The enantiomeric excess was determined by HPLC using a Chiralpak IA column. [e]Recrystallized from 2-propanol/*n*-hexane.

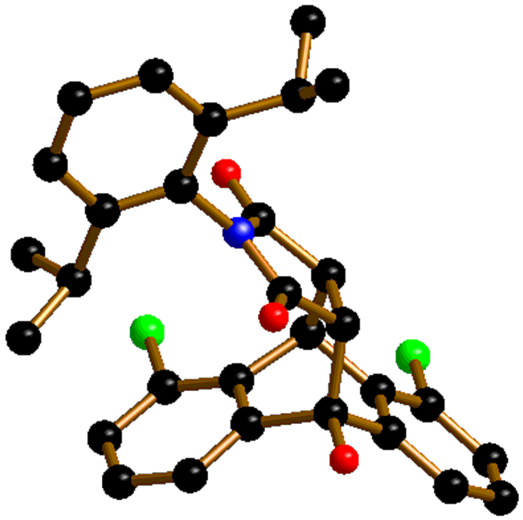


Figure 2: Molecular structure of **3m** (C: black; N: blue; O: red; Cl: green; hydrogen atoms are omitted for the sake of clarity).

Conclusion

C_2-symmetric bisamidines were shown to be potent chiral Brønsted base catalysts for the Diels-Alder reaction of *N*-substituted maleimides and anthrones. Compared to bisoxazolines **7**, much shorter reaction times under comparable conditions were sufficient with the more basic bisamidine catalysts **8** (~50-fold [5]). The higher intrinsic reactivity of the bisamidines allowed to run the reactions at lower temperatures. In both groups of catalysts, the phenyl substituted species induced the lowest enantioselectivities. Bisamidine **8a** performed better than the corresponding bisoxazoline. Increasing the size of substituents in catalysts **8b**–**d** also improved stereoselectivities, but not to high levels. This may be due to the flexible nature of the substituents present in bisamidines **8b** and **8c**. It is instructive, therefore, to compare with the bisoxazolines **7**. By far the best enantioselectivities were observed in this series with the *t*-Bu derivative (47% *ee* versus 3% for the phenyl analogue in the reaction of **1a** and **2a**). Keeping in mind that even the less selective bisamidine **8a** could induce up to 76% *ee* in favorable cases, replacing the phenyl moieties of **8a** by *t*-Bu is an attractive option for future studies on bisamidine-mediated organocatalytic transformations.

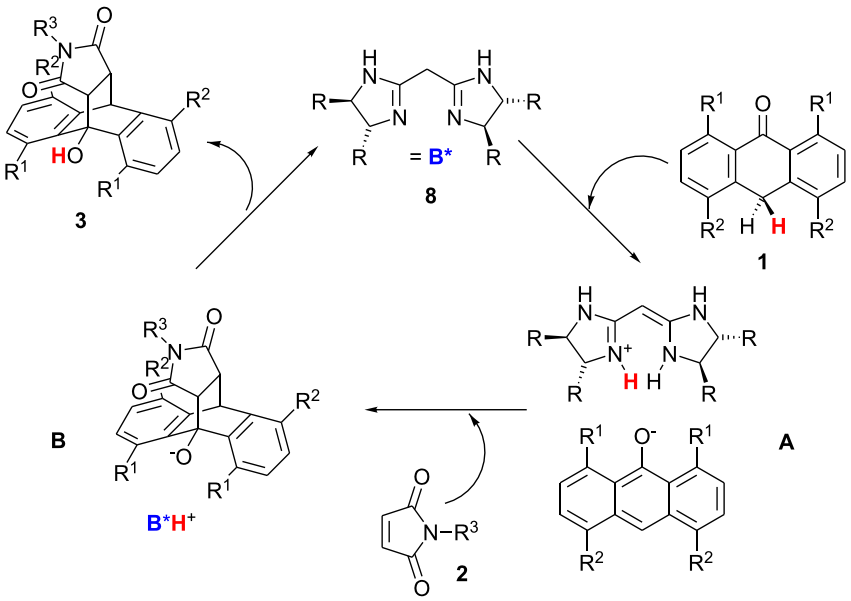
Supporting Information

Supporting Information File 1

Supporting information features characterisation data and copies of ^{1}H- and ^{13}C-NMR spectra of anthrones **1**, maleimides **2**, Diels-Alder adducts **3**, bisamidine hydro-chlorides **8b**–**d**·H^+·Cl^-, neutral bisamidines **8b**–**d** and diamines **9b**–**c**, plus copies of chromatograms obtained with chiral columns.
[http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S1.doc]

Supporting Information File 2

X-Ray data of compound **3k**
[http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S2.cif]



Scheme 4: Proposed mechanism of the Diels-Alder reaction.

Supporting Information File 3
X-Ray data of compound **3m**
[http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S3.cif]

Acknowledgments

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References

1. Uemae, K.; Masuda, S.; Yamamoto, Y. *J. Chem. Soc., Perkin Trans. 1* **2001,** 1002–1006. doi:10.1039/b100961n
2. Tokioka, K.; Masuda, S.; Fujii, T.; Hata, Y.; Yamamoto, Y. *Tetrahedron: Asymmetry* **1997,** *8,* 101–107. doi:10.1016/S0957-4166(96)00473-9
3. Shen, J.; Nguyen, T. T.; Goh, Y.-P.; Ye, W.; Fu, X.; Xu, J.; Tan, C.-H. *J. Am. Chem. Soc.* **2006,** *128,* 13692–13693. doi:10.1021/ja064636n
4. Fache, F.; Piva, O. *Tetrahedron Lett.* **2001,** *42,* 5655–5657. doi:10.1016/S0040-4039(01)01036-X
5. Akalay, D.; Dürner, G.; Göbel, M. W. *Eur. J. Org. Chem.* **2008,** 2365–2368. doi:10.1002/ejoc.200800179
6. Akalay, D.; Dürner, G.; Bats, J. W.; Bolte, M.; Göbel, M. W. *J. Org. Chem.* **2007,** *72,* 5618–5624. doi:10.1021/jo070534j
7. Freiermuth, B.; Hellrung, B.; Peterli, S.; Schultz, M.-F.; Wintgens, D.; Wirz, J. *Helv. Chim. Acta* **2001,** *84,* 3796–3809. doi:10.1002/1522-2675(20011219)84:12<3796::AID-HLCA3796>3.0.CO;2-T
8. McCann, G. M.; McDonnell, C. M.; Magris, L.; O'Ferrall, R. A. M. *J. Chem. Soc., Perkin Trans. 2* **2002,** 784–795. doi:10.1039/b109242c
9. Scheurer, A.; Mosset, P.; Saalfrank, R. W. *Tetrahedron: Asymmetry* **1999,** *10,* 3559–3570. doi:10.1016/S0957-4166(99)00353-5
10. House, H. O.; Hrabie, J. A.; VanDerveer, D. *J. Org. Chem.* **1986,** *51,* 921–929. doi:10.1021/jo00356a031
11. Prinz, H.; Wiegrebe, W.; Müller, K. *J. Org. Chem.* **1996,** *61,* 2853–2856. doi:10.1021/jo9520351
12. Walker, M. A. *J. Org. Chem.* **1995,** *60,* 5352–5355. doi:10.1021/jo00121a070
13. Cava, M. P.; Deana, A. A.; Muth, K.; Mitchell, M. J. *Org. Synth.* **1961,** *41,* 93–95.
14. Fielding, M. R.; Grigg, R.; Sridharan, V.; Thornton-Pett, M.; Urch, C. J. *Tetrahedron* **2001,** *57,* 7737–7748. doi:10.1016/S0040-4020(01)00740-2

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