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# Chemical Investigation of *Tetradium ruticarpum* Fruits and Their Antibacterial Activity against *Helicobacter pylori*

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**ABSTRACT:** The fruit of *Tetradium ruticarpum*, known as *Evodiae* Fructus, is a traditional herbal medicine used to treat gastric and duodenal ulcers, vomiting, and diarrhea. The traditional usage can be potentially associated with the antibacterial activity of *T. ruticarpum* fruits against *Helicobacter pylori*. However, so far, the antibacterial activity of *T. ruticarpum* fruits and antibacterial components against *H. pylori* has not been investigated despite the traditional folk use. The current study was conducted to investigate the bioactive chemical components of *T. ruticarpum* fruits and evaluate their antibacterial activity against *H. pylori*. Phytochemical investigation of the EtOH extract of *T. ruticarpum* fruits led to the isolation and identification of nine compounds (1–9), including phellolactone (1), the absolute configuration of which has not yet been determined. The chemical structures of the isolated compounds were elucidated by analyzing the spectroscopic data from one-dimensional (1D) and two-dimensional (2D) NMR and high-resolution electrospray ionization mass spectrometry (HR-ESIMS) experiments. Specifically, the absolute configuration of



compound 1 was established by the application of computational methods, including electronic circular dichroism (ECD) calculation and the NOE/ROE-based interproton distance measurement technique via peak amplitude normalization for the improved cross-relaxation (PANIC) method. In the anti-*H. pylori* activity test, compound 3 showed the most potent antibacterial activity against *H. pylori* strain 51, with 94.4% inhibition (MIC<sub>50</sub> and MIC<sub>90</sub> values of 22 and 50  $\mu$ M, respectively), comparable to that of metronidazole (97.0% inhibition, and MIC<sub>50</sub> and MIC<sub>90</sub> values of 17 and 46  $\mu$ M, respectively). Moreover, compound **5** exhibited moderate antibacterial activity against *H. pylori* strain 51, with 58.6% inhibition (MIC<sub>50</sub> value of 99  $\mu$ M), which was higher than that of quercetin (34.4% inhibition) as a positive control. Based on the bioactivity results, we also analyzed the structure–activity relationship of the anti-*H. pylori* activity. Conclusion: These findings demonstrated that *T. ruticarpum* fruits had antibacterial activity against *H. pylori* activity. Conclusion: These findings demonstrated that *T. ruticarpum* fruits had antibacterial activity against *H. pylori* and could be used in the treatment of gastric and duodenal ulcers. Meanwhile, the active compound, 1-methyl-2-(8E)-8-tridecenyl-4(1H)-quinolinone (**3**), identified herein also indicated the potential application in the development of novel antibiotics against *H. pylori*.

## **1. INTRODUCTION**

Tetradium ruticarpum (A. Juss.) Hartley is a small shrub native to temperate and tropical regions of Asia, including China and Korea, and its dried nearly ripe fruit, also known as Evodiae Fructus or Wu zhu yu in Chinese, has been used as a traditional herbal medicine for more than 2000 years to treat gastric and duodenal ulcers, headache, abdominal pain, vomiting, diarrhea, dysmenorrhea, and postpartum hemorrhage.<sup>1,2</sup> The fruit of T. ruticarpum was first recorded in "Shennong's Classic of Materia Medica", and the fruit of T. ruticarpum is named variously in Pharmacopoeias (e.g., Evodia fruit in Korean Pharmacopoeia, Euodia fruit in Japanese Pharmacopoeia, Euodiae fructus in Chinese Pharmacopoeia).<sup>1</sup> However, T. ruticarpum has also been reported to exhibit potential toxicity by long-term use or excessive doses, and owing to its toxicity, T. ruticarpum is usually processed and/or combined with other herbs in the formulation of traditional Chinese medicine.<sup>2</sup>

Recent pharmacological studies of *T. ruticarpum* fruit have demonstrated that the extracts of *T. ruticarpum* fruit and its active constituents show diverse therapeutic efficacies, including antiproliferative,<sup>3</sup> anti-inflammatory,<sup>4</sup> antibacterial,<sup>5</sup> cytotoxic,<sup>5</sup> antiadipogenic,<sup>6</sup> antioxidant,<sup>7</sup> central nervous system homeostasis,<sup>8</sup> and cardiovascular protective activities.<sup>9</sup> To date, more than 160 chemical components, including alkaloids, terpenoids, steroids, limonoids, phenylpropanoids, flavonoids, phenolic acids, anthraquinones, and essential oils, have been isolated from the *T. ruticarpum* fruits.<sup>1</sup> Especially, among the main components of *T. ruticarpum* fruit, evodi-

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Figure 1. Chemical structures of compounds 1-9.

amine and rutecarpine were found to act as partial agonists and antagonists of transient receptor potential vanilloid 1 (TRPV1).<sup>10</sup> In addition, rutecarpine has been reported to inhibit COX-2,<sup>11</sup> and evodiamine has been shown to possess anticancer activities both *in vitro* and *in vivo* by inhibiting proliferation, invasion, and metastasis by inducing apoptosis in various tumor cell lines.<sup>12</sup>

Of note, one of the major traditional uses of T. ruticarpum fruits is to treat gastric and duodenal ulcers, vomiting, and diarrhea,<sup>1</sup> and its traditional usage can be potentially associated with the antibacterial activity of *T. ruticarpum* fruits against *H.* pylori. To date, the antibacterial activity of T. ruticarpum fruit has been widely investigated. Quinolone alkaloids from T. ruticarpum fruits have been reported to inhibit the growth of Mycobacterium fortuitum, Mycobacterium smegmatis, and Mycobacterium phlei, indicating their therapeutic potential for tuberculosis.<sup>13</sup> Other quinolone alkaloids, euocarpines A-E, exhibit antibacterial activity against Staphylococcus aureus ATCC25923, Staphylococcus epidermidis ATCC12228, and Bacillus subtilis ATCC6633.<sup>5</sup> In addition, rhetsinine, an indole alkaloid, exhibits potent antibacterial activity against Xanthomonas oryzae pv. oryzae, Xanthomonas oryzae pv. oryzicola, and Xanthomonas oryzae pv. oryzae,14 and essential oils derived from T. ruticarpum fruits exhibited antibacterial activity against Bacillus subtilis ATCC 6633 and Staphylococcus aureus ATCC 6538.<sup>15</sup> However, to the best of our knowledge, the antibacterial activity of T. ruticarpum fruits and their responsible components against H. pylori has not been investigated, although T. ruticarpum fruits have been used traditionally to treat gastric and duodenal ulcers. Thus, the anti-H. pylori compounds from T. ruticarpum fruits should be elucidated to support the therapeutic application of T. ruticarpum fruits in the treatment of gastric and duodenal ulcers.

As part of our continuing natural product discovery research for the identification of bioactive components from diverse natural resources,<sup>16–20</sup> we explored antibacterial components from an ethanolic (EtOH) extract of *T. ruticarpum* fruit. We performed solvent partition of EtOH extracts of *T. ruticarpum* fruit and applied the solvent-partitioned fractions to antibacterial activity tests in our bioactivity screening, where we observed that the CH<sub>2</sub>Cl<sub>2</sub> fraction showed strong antibacterial activity against Helicobacter pylori strain 51. Intensive phytochemical investigation of the active CH<sub>2</sub>Cl<sub>2</sub> fraction, aided by liquid chromatography-mass spectrometry (LC/ MS)-based analysis, led to the isolation of nine compounds (1-9), including phellolactone (1), whose absolute configuration has not yet been determined. In this study, the chemical structure of phellolactone (1) was elucidated by conventional spectroscopic data analysis, including one-dimensional (1D) and two-dimensional (2D) nuclear magnetic resonance (NMR) and high-resolution electrospray ionization mass spectrometry (HR-ESIMS), as well as computational methods including electronic circular dichroism (ECD) calculations and the NOE/ROE-based interproton distance measurement technique via peak amplitude normalization for improved cross-relaxation (PANIC). Herein, we describe the separation and structural elucidation of compounds 1-9 and evaluation of their anti-H. pylori activity.

## 2. RESULTS AND DISCUSSION

2.1. Separation of Compounds from T. ruticarpum Fruit. Dried T. ruticarpum fruits were extracted with 50% EtOH under reflux, yielding a crude ethanolic extract by evaporation in vacuo. The resultant EtOH extract was sequentially subjected to solvent partitioning using four organic solvents: n-hexane, CH<sub>2</sub>Cl<sub>2</sub>, EtOAc, and BuOH. Four main solvent-partitioning fractions with increasing polarity were obtained: n-hexane-, CH2Cl2-, EtOAc-, and BuOH-soluble fractions. The LC/MS analysis of each fraction using reference to the in-house UV library database in our LC/ MS system revealed that the CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction mainly contains alkaloids and/or phenolic acids. In addition, the major traditional usage of T. ruticarpum fruits to treat gastric and duodenal ulcers can be potentially associated with the antibacterial activity of T. ruticarpum fruits against H. pylori, and thus the anti-H. pylori activities of T. ruticarpum fruit extract and its solvent-partitioning fractions were evaluated using a clinical strain of H. pylori 51 in our bioactivity screening test. Although the crude extract showed no inhibitory activity against H. pylori strain 51, the CH<sub>2</sub>Cl<sub>2</sub>soluble fraction showed potent antibacterial activity against *H*.

*pylori* strain 51 with 94.3% inhibition at a final concentration of 100  $\mu$ g/mL, which was comparable to that of the positive control, metronidazole (97.0% inhibition at a final concentration of 100  $\mu$ M). Based on these results, the CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction was subjected to phytochemical investigation by repeated column chromatography and semipreparative high-performance liquid chromatography (HPLC) with the guidance of LC/MS analysis, which led to the isolation of nine compounds (1–9) (Figure 1).

2.2. Structural Elucidation of the Isolated Compounds 1–9. Compound 1 was isolated as a white, amorphous powder. The molecular formula of 1 was established as  $C_{13}H_{14}O_8$ , determined by negative-ion HR-ESIMS, which revealed an  $[M - H]^-$  ion peak at m/z 297.0618 (calcd. for  $C_{13}H_{13}O_8$ , 297.0610). The <sup>1</sup>H NMR data (Table 1) of compound 1, assigned by the aid of heteronuclear

Table 1. <sup>1</sup>H (850 MHz) and <sup>13</sup>C NMR (212.5 MHz) Data for Compound 1 in CD<sub>3</sub>OD ( $\delta$  ppm)<sup>*a*</sup>

position	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	$\delta_{\rm C}{}^{b}$				
2		176.2 C				
3	4.55 s	70.1 CH				
4		75.5 C				
$5\alpha$	4.41 d (10.0)	72.3 CH <sub>2</sub>				
$5\beta$	4.26 d (10.0)					
6a	4.39 d (11.5)	64.6 CH <sub>2</sub>				
6b	4.43 d (11.5)					
1'		120.4 C				
2'	7.59 d (2.0)	112.3 CH				
3'		147.4 C				
4′		151.8 C				
5'	6.86 d (8.0)	114.6 CH				
6'	7.61 dd (8.0, 2.0)	123.9 CH				
7′		166.2 C				
3'-OCH <sub>3</sub>	3.91 s	54.9 CH <sub>3</sub>				
<sup><i>a</i></sup> Coupling constants	(Hz) are given in parenthe	eses. <sup><i>b</i>13</sup> C NMR data				
were assigned based on 115QC and HMDC experiments.						

single quantum correlation (HSQC) experiment, showed the presence of signals for one set of ABX signals at  $\delta_{\rm H}$  7.61 (1H, dd, J = 8.0, 2.0 Hz, H-6'), 7.59 (1H, d, J = 2.0 Hz, H-2'), 6.86 (1H, d, J = 8.0 Hz, H-5'), indicating a 1,3,4-trisubstituted aromatic ring, two oxymethylene groups [ $\delta_{\rm H}$  4.41 (1H, d,  $J = 10.0, \text{H-}5\alpha)/4.26$  (1H, d,  $J = 10.0, \text{H-}5\beta$ ) and 4.39 (1H, d, J = 11.5, H-6a)/4.43 (1H, d, J = 11.5, H-6b)], an oxymethine group [ $\delta_{\rm H}$  4.55 (1H, s, H-3)], and one methoxy group [ $\delta_{\rm H}$  3.91 (3H, s, 3'-OCH<sub>3</sub>)]. The <sup>13</sup>C NMR data (Table 1) of 1, combined with heteronuclear multiple bond correlation

(HMBC) experiment, revealed 13 carbon resonances, including six carbons for the aromatic ring; two carbonyl carbons at  $\delta_{\rm C}$  166.2 and 176.2; four oxygenated carbons at  $\delta_{\rm C}$  64.6, 70.1, 72.3, and 75.5; and one methoxy carbon at  $\delta_{\rm C}$  54.9.

The planar structure of 1 was elucidated by interpretation of 2D NMR data ( $^{1}H-^{1}H$  COSY and HMBC). The  $^{1}H-^{1}H$ COSY correlations of H-5'/H-6' and HMBC correlations of H-2'/C-4' and C-6', H-5'/C-3' and C-6', and H-6'/C-1' and C-3', as well as the HMBC correlation of OCH<sub>3</sub>/C-3', confirmed the presence of a 1,3,4-trisubstituted aromatic ring. Importantly, the  $\gamma$ -lactone moiety was deduced from the key HMBC correlations of H-3/C-2 ( $\delta_{\rm C}$  176.2), C-4 ( $\delta_{\rm C}$  75.5), C-5 ( $\delta_{\rm C}$  72.3), and C-6 ( $\delta_{\rm C}$  64.6), and H-5/C-2, C-3 ( $\delta_{\rm C}$  70.1), and C-6 (Figure 2). Furthermore, the  $\gamma$ -lactone moiety was connected to the aromatic ring via the linkage between C-6 and the ester moiety (C-7'), which was confirmed by the key HMBC correlations of H-6/C-7', H-2'/C-7', and H-6'/C-7' (Figure 2). Based on the 2D NMR data, the complete planar structure of 1 was determined to be  $2\beta$ ,  $3\beta$ -dihydroxy- $3\alpha$ vanilloylmethyl- $\gamma$ -lactone, which was previously reported as phellolactone from Phellodendron chinense bark.<sup>21</sup> However, to the best of our knowledge, the absolute configuration of phellolactone has not yet been determined. Thus, it would be worthwhile to determine its absolute configuration.

First, the relative configuration of the  $\gamma$ -lactone moiety was established based on the NOESY experiment, where the crosspeak of H-5 $\beta$ /H-3 confirmed the hydroxyl group of C-3 to be in the  $\alpha$  configuration (Figure 2). However, despite intensive NOESY data analysis of 1, the configuration of the hydroxyl group at C-4 was not confirmed by the experimental spectroscopic data because it was difficult to distinguish the NOESY cross-peak between H-5 and H-6, and it was predicted that the NOESY correlation of H-3/H-6 could be detected in both the  $4\alpha$ -OH and  $4\beta$ -OH cases as the interproton distance between H-3 and H-6 in 1 was calculated within 5.00 Å. The inversion of the 4-OH configuration significantly affected the conformational change of the  $\gamma$ -lactone moiety. Therefore, to accomplish a configurational analysis of the  $\gamma$ -lactone moiety, we applied the NOE/ROE-based interproton distance measurement technique via the PANIC method.<sup>22-24</sup> The PANIC-based calibrated interproton distance between H-3 and H-6b is 2.21 Å (Figure S10). To predict the interproton distance between H-3 and H-6b in the situation of  $4\alpha$ -OH or  $4\beta$ -OH in 1, molecular mechanics and quantum mechanicsoptimized three-dimensional (3D) structures of 1A ( $3\alpha$ -OH and  $4\alpha$ -OH) and **1B** ( $3\alpha$ -OH and  $4\beta$ -OH) were generated and calculated. The predicted interproton distance between H-3 and H-6b of 1A (2.34 Å) was matched well with the experimental value (2.21 Å), rather than 1B (3.85 Å), which



Figure 2. (A) Key  ${}^{1}H - {}^{1}H$  COSY (blue hyphen) and HMBC (red forward-curved arrow) correlations for 1. (B) Key nuclear Overhauser effect spectroscopy (NOESY) correlations of 1.

suggested *cis*-configuration between 3-OH and 4-OH in 1 (Figure 3).



Figure 3. Optimized 3D structures of 1A and 1B with predicted interproton distance between H-3 and H-6b.

Finally, with reasonable structural elucidation of **1A**, the absolute configuration determination of **1** was confirmed by quantum chemical calculations for ECD simulations. Two possibilities with absolute configurations of 3R,4R and 3S,4S were calculated, and the experimental ECD spectrum of **1** was compared (Figure 4). The experimental ECD data were in good agreement with the calculated ECD data for (3R,4R)-1. Thus, the absolute configuration of the phellolactone isolated in this study was determined to be 3R,4R and compound **1** was named (3R,4R)-phellolactone.

The other isolated compounds were identified as 1-methyl-2-undecyl-4(1*H*)-quinolone (2),<sup>25</sup> 1-methyl-2-(8*E*)-8-tridecenyl-4(1*H*)-quinolinone (3),<sup>26</sup> 1-methyl-2-(6*Z*,9*Z*)-6,9-pentadecadien-1-yl-4(1*H*)-quinolinone (4),<sup>26</sup> dihydroevocarpine (5),<sup>27</sup> 4-hydroxy-8-methoxy-3-(3-methyl-2-buten-1-yl)-2(1*H*)quinolinone (6),<sup>28</sup> dehydroevodiamine (7),<sup>29</sup> crocusatin C (8),<sup>30</sup> and (+)-*trans*-abscisic acid (9)<sup>31</sup> (Figure 1) by comparison of their NMR spectral and physical data with those reported earlier and the data from LC/MS analysis.

2.3. Evaluation of Antibacterial Activity of the Isolated Compounds against *H. pylori*. *H. pylori* is a

major public health issue worldwide, affecting approximately 50% of the global population.<sup>32</sup> Eradication of *H. pylori* helps treat both gastric and duodenal ulcers, and even gastric cancer, because the presence of *H. pylori* is known to be associated with gastric or duodenal pathologies.<sup>33</sup> The antibacterial activity of *T. ruticarpum* fruits against *H. pylori* can be involved with the major traditional usage of *T. ruticarpum* fruits to treat gastric and duodenal ulcers. In our bioactivity screening test of the *T. ruticarpum* fruit extract and its derived fractions, we found that the CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction showed potent antibacterial activity against *H. pylori* strain 51 with 94.3% inhibition. Next, compounds 1–9 isolated for antibacterial activity against *H. pylori* strain 51 (Table 2). Among the

Table 2. Anti-H. pylori Activity of Compounds 1–9 against H. pylori Strain 51 Treated with 100  $\mu$ M of Each Compound

compound	inhibition (%)	$MIC_{50}$ ( $\mu M$ )	$MIC_{90}$ ( $\mu M$ )
1	8.1		
2	0.0		
3	94.4	22	50
4	14.8		
5	58.6	99	>100
6	15.7		
7	20.0		
8	7.1		
9	16.1		
quercetin <sup>a</sup>	34.4		
metronidazole <sup><i>a</i></sup>	97.0	17	46
<sup><i>a</i></sup> Positive controls.			

isolates, compound **3** exhibited the most potent antibacterial activity against *H. pylori* strain 51, with 94.4% inhibition at a final concentration of 100  $\mu$ M, comparable to that of metronidazole (97.0% inhibition) as a positive control, and it showed MIC<sub>50</sub> and MIC<sub>90</sub> values of 22 and 50  $\mu$ M, respectively. Compound **5** displayed moderate antibacterial activity against *H. pylori* strain 51 with 58.6% inhibition, which was higher than that of quercetin (34.4% inhibition) as a positive control, with an MIC<sub>50</sub> value of 99  $\mu$ M. The other



Figure 4. Experimental and calculated ECD spectra of compound 1.

compounds showed weak activity or failed to exhibit anti-H. pylori infection, respectively (Table 2). Based on these findings, quinolone alkaloids (3 and 5) with 13 carbons on the side chain showed anti-H. pylori activity, and the presence of double bonds and the length of the side carbon chain in quinolone alkaloids may be important for anti-H. pylori activity. Interestingly, the structure-activity relationship analysis confirmed that this study corresponds exactly to previous studies showing that quinolone alkaloids with 13 carbons on the side chain are the most potent antibacterial compounds.<sup>5</sup> The mechanisms of anti-H. pylori activity include antiadhesion, oxidative stress to H. pylori, the inhibition of bacterial enzymes such as urease and N-acetyltansferase, and hydrophilic/hydrophobic interaction.<sup>34,35</sup> Further studies will be required to elucidate the exact mechanism by which active compounds 3 and 5 inhibit H. pylori growth. Effects of the active compounds on the human body with safety issues and their cytotoxicity to other cells are also required in the following study.

## 3. CONCLUSIONS

In this study, phytochemical investigation of the EtOH extract of T. ruticarpum fruits, a traditional herbal medicine used to treat gastric and duodenal ulcers, led to the isolation and identification of nine compounds (1-9), including phellolactone (1), whose absolute configuration has not yet been determined. The complete structure of phellolactone (1), including its absolute configuration, was elucidated using conventional spectroscopic data analysis, including 1D and 2D NMR and HR-ESIMS, and computational methods, including ECD calculation and NOE/ROE-based interproton distance measurement technique via the PANIC method. All of the compounds (1-9) were isolated from the anti-*H. pylori*-active CH2Cl2-soluble fraction, and from anti-H. pylori activity test of the isolates, compound 3 showed the most potent antibacterial activity against H. pylori strain 51, with 94.4% inhibition (MIC<sub>50</sub> and MIC<sub>90</sub> values of 22 and 50  $\mu$ M, respectively), comparable to that of metronidazole (97.0% inhibition, and MIC<sub>50</sub> and MIC<sub>90</sub> values of 17 and 46  $\mu$ M, respectively). Moreover, compound 5 exhibited moderate antibacterial activity against H. pylori strain 51, with 58.6% inhibition (MIC<sub>50</sub> value of 99  $\mu$ M), which was higher than that of quercetin (34.4% inhibition) as a positive control. Accordingly, the present study at least partially elucidated the antibacterial constituents from the T. ruticarpum fruits and revealed the anti-H. pylori principles, which provide the scientific basis of this plant as the herb in the folk application to treat gastric and duodenal ulcers.

## 4. EXPERIMENTAL SECTION

**4.1. General Experimental Procedures.** Optical rotations were measured using a JASCO P-2000 polarimeter (JASCO, Easton, MD). Ultraviolet (UV) spectra were acquired using an Agilent 8453 UV-visible spectrophotometer (Agilent Technologies, Santa Clara, CA). ECD spectra were obtained using a JASCO J-1500 spectropolarimeter (JASCO). Infrared (IR) spectra were recorded on a Bruker IFS-66/S FT-IR spectrometer (Bruker, Karlsruhe, Germany). NMR spectra were recorded with a Bruker AVANCE III HD 850 NMR spectrometer with a 5 mm TCI CryoProbe operating at 850 MHz (<sup>1</sup>H) and 212.5 MHz (<sup>13</sup>C), with chemical shifts given in ppm ( $\delta$ ) for <sup>1</sup>H and <sup>13</sup>C NMR analyses. All HR-ESIMS data

were obtained using an Agilent G6545B quadrupole time-offlight mass spectrometer (Agilent Technologies) coupled to an Agilent 1290 Infinity II high-performance liquid chromatography (HPLC) instrument using an Agilent Eclipse Plus C18 column (2.1  $\times$  50 mm, 1.8  $\mu$ m; flow rate: 0.3 mL/min). Preparative HPLC was performed using a Waters 1525 Binary HPLC pump with a Waters 996 Photodiode Array Detector (Waters Corporation, Milford, MA) and an Agilent Eclipse C18 column (250  $\times$  21.2 mm, 5  $\mu$ m; flow rate: 5 mL/min; Agilent Technologies). Semipreparative HPLC was performed using a Shimadzu Prominence HPLC System with SPD-20A/ 20AV Series Prominence HPLC UV-vis detectors (Shimadzu, Tokyo, Japan) and a Phenomenex Luna C18 column (250  $\times$ 10 mm, 5  $\mu$ m; flow rate, 2 mL/min; Phenomenex, Torrance, CA). LC/MS analysis was performed on an Agilent 1200 Series HPLC system equipped with a diode array detector and 6130 Series ESI mass spectrometer using an analytical Kinetex C18 100 Å column (100  $\times$  2.1 mm, 5  $\mu$ m; flow rate: 0.3 mL/ min; Phenomenex). Silica gel 60 (230-400 mesh; Merck, Darmstadt, Germany) and RP-C18 silica gel (Merck, 230-400 mesh) were used for column chromatography. Sephadex-LH20 (Pharmacia, Uppsala, Sweden) was used as the packing material for the molecular sieve column chromatography. Thin-layer chromatography (TLC) was performed using precoated silica gel F254 plates and RP-C18 F254s plates (Merck), and spots were detected under UV light or by heating after spraying with anisaldehyde-sulfuric acid.

**4.2. Plant Material.** Dried *T. ruticarpum* fruit was purchased from the Kyungdong herbal market in Seoul, Korea in July 2020, and the plant material was identified by one of the authors (K. H. Kim). A voucher specimen (OSU-2020) was deposited in the herbarium of the School of Pharmacy, Sungkyunkwan University, Suwon, Korea.

4.3. Extraction and Isolation. Completely dried fruits of T. ruticarpum (600 g) were extracted three times with 50% EtOH under reflux (2.0 L  $\times$  3) and then filtered. The filtrate was subsequently evaporated in vacuo to obtain the crude EtOH extract (172.7 g). The extract was dissolved in distilled water (1.0 L) and partitioned with hexane, dichloromethane  $(CH_2Cl_2)$ , ethyl acetate (EtOAc), and *n*-butanol (*n*-BuOH). Four layers with increasing polarity, including hexane-, CH<sub>2</sub>Cl<sub>2</sub>-, EtOAc-, and *n*-BuOH-soluble fractions, were obtained as 1.2, 0.9, 2.1, and 18.7 g, respectively. Through TLC and LC/MS analysis of each fraction, we found that the CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction (0.9 g) contains mainly alkaloids and/ or phenolic acids, and in our bioactivity screening, the CH<sub>2</sub>Cl<sub>2</sub>soluble fraction showed antibacterial activity against H. pylori strain 51. Thus, the CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction was loaded onto a silica gel chromatography column and subjected to a gradient solvent system of  $CH_2Cl_2$ -MeOH (60:1-1:1, v/v) to give seven eluted fractions (M1–M7). Fraction M1 (487.7 mg) was subjected to medium-pressure liquid chromatography to obtain four fractions (M11-M14). The M11 fraction (100.5 mg) was subjected to Sephadex-LH20 column chromatography with 100% MeOH to obtain three fractions (M111-M113). Compounds 2 (0.9 mg,  $t_{\rm R}$  = 28.4 min), 3 (0.5 mg,  $t_{\rm R}$  = 31.9 min), 4 (0.7 mg,  $t_R$  = 34.1 min), and 5 (1.0 mg,  $t_R$  = 42.3 min) were isolated from subfraction M111 (11.6 mg) via semipreparative reversed-phase HPLC with 75% MeCN/H<sub>2</sub>O (isocratic system, flow rate: 2 mL/min). The M12 fraction (168.6 mg) was subjected to C18 silica gel reversed-phase column chromatography to obtain five fractions (M121-M125). Compound 1 (0.4 mg,  $t_{\rm R}$  = 36.5 min) was isolated

from subfraction M121 (12.9 mg) via semipreparative reversed-phase HPLC with 30% MeOH/H<sub>2</sub>O (isocratic system, flow rate: 2 mL/min). Compound 8 (1.2 mg,  $t_{\rm R}$  = 50.8 min) was isolated from subfraction M122 (57.1 mg) using semipreparative reversed-phase HPLC with 30% MeOH/H<sub>2</sub>O (isocratic system, flow rate: 2 mL/min). Compounds 6 (0.8 mg,  $t_{\rm R}$  = 99.8 min) and 9 (1.4 mg,  $t_{\rm R}$  = 106.1 min) were isolated from subfraction M125 (44.4 mg) using semipreparative reversed-phase HPLC with 55% MeOH/H2O (isocratic system, flow rate: 2 mL/min). Fraction M4 (229.7 mg) was subjected to C18 silica gel reversed-phase column chromatography to obtain six fractions (M41-46). The M45 fraction (134.7 mg) was subjected to Sephadex-LH20 column chromatography to obtain four fractions (M451-M454). Compound 7 (6.0 mg,  $t_{\rm R}$  = 12.3 min) was isolated from subfraction M451 (62.1 mg) using semipreparative reversedphase HPLC with 50% MeOH/H<sub>2</sub>O (isocratic system, flow rate: 2 mL/min).

4.3.1. (3*R*,4*R*)-Phellolactone (1). White amorphous powder;  $[\alpha]_{25}^{25}$  -85 (*c* 0.02 MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 203 (1.82), 222 (3.51), 264 (1.56), 269 (0.78); CD (MeOH)  $\lambda_{max}$ ( $\Delta \varepsilon$ ) 212 (-0.07), 231 (-0.12), 251 (+0.04) nm; IR (KBr)  $\nu_{max}$ : 3410, 1706, 1599, 1516, 1462 cm<sup>-1</sup>; <sup>1</sup>H (850 MHz) and <sup>13</sup>C (212.5 MHz) NMR data (Table 1); HR-ESIMS (negativeion mode) *m*/*z* 297.0618 [M - H]<sup>-</sup> (calcd for C<sub>13</sub>H<sub>13</sub>O<sub>8</sub>, 297.0610).

**4.4. PANIC Analysis.** PANIC analysis was carried out according to a previously reported method, with slight modifications.<sup>22</sup> Slices of the 2D NOESY spectrum of 1 were obtained using the MestReNova software (version 14.1.2-25024). As a reference,<sup>22</sup> the interproton distance between two ortho-relationship aromatic protons, H-5' and H-6' ( $r_{reference}$ ), was set as 2.5 Å. The corresponding NOE intensity (NOE<sub>reference</sub>) was obtained by integrating the signal of H-5' from the slice of 2D NOESY spectrum irradiated at H-6' normalized to -1000. The NOE intensity of H-6b was acquired by irradiation at H-3 normalized to -1000 and integrated (NOE<sub>unknown</sub>). The calibrated interproton distance ( $r_{unknown}$ ) was calculated with the following equation

$$\frac{\text{NOE}_{\text{unknown}}}{\text{NOE}_{\text{reference}}} = \left(\frac{r_{\text{reference}}}{r_{\text{unknown}}}\right)^{6}$$

The MacroModel (version 2021-4, Schrödinger LLC) program was used to generate molecular mechanics (MM)- and quantum mechanics (QM)-optimized 3D structures and measure the specific interproton distance.<sup>36</sup>

**4.5. ECD Calculation.** Initial conformational searches were performed at the MMFF94 force field using the MacroModel (version 2021-4, Schrödinger LLC) program with a mixed torsional/low-mode sampling method, in which a gas phase with a 50 kJ/mol energy window and 10,000 maximum iterations were employed. The Polak–Ribiere conjugate gradient protocol was established with 10,000 maximum iterations and a 0.001 kJ (mol Å)<sup>-1</sup> convergence threshold on the root-mean-square gradient to minimize conformers. The conformers proposed in this study (found within 20 kJ/mol in the MMFF force field) were selected for geometry optimization using TmoleX 4.3.2 with the density functional theory settings of B3-LYP/6-31+G(d,p).<sup>37</sup>

ECD calculations for the (3R,4R)-1 and (3S,4S)-1 conformers (20 conformers each) were performed at an identical theory level and basis sets. The calculated ECD spectra were

simulated by superimposing each transition, where  $\sigma$  is the bandwidth at a height of 1/e.  $\Delta E_i$  and  $R_i$  are the excitation energy and rotatory strength for transition *i*, respectively.<sup>37</sup> In this study, the value of  $\sigma$  was 0.2 eV. The excitation energies and rotational strengths of the ECD spectra were calculated based on the Boltzmann populations of conformers, and ECD visualization was performed using SigmaPlot 14.0.

$$\Delta \epsilon(E) = \frac{1}{2.297 \times 10^{-39}} \frac{1}{\sqrt{2\pi\sigma}} \sum_{A}^{i} \Delta E_{i} R_{i} \mathrm{e}^{[-(E - \Delta E_{i})^{2}/(2\sigma)^{2}]}$$

**4.6.** Helicobacter pylori Culture. The clinical strain of *H. pylori* 51 was provided by the *H. pylori* Korean Type Culture Collection, School of Medicine, Gyeongsang National University, Korea. The strain was grown and maintained on Brucella agar (BD Co., Sparks, MD) supplemented with 10% horse serum (Gibco, New York). The culture conditions were 37 °C, 100% humidity, and 10% CO<sub>2</sub> for 2–3 days.

4.7. Determination of Minimal Inhibitory Concentration (MIC) Values. MICs were determined using the broth dilution method as previously reported.<sup>38,39</sup> Bacterial colony suspension  $(20 \ \mu\text{L})$  equivalent to  $2-3 \times 10^8$  CFU/ mL and 20  $\mu$ L of twofold diluted test samples and controls were added to each well of a six-well plate containing Brucella broth medium supplemented with 10% horse serum. The final volume was 2 mL. After 24 h of incubation, the bacterial growth was evaluated by measuring the optical density at 600 nm using a spectrophotometer. The MIC<sub>50</sub> and MIC<sub>90</sub> values were defined as the lowest concentrations of samples at which bacterial growth was inhibited by 50 and 90%, respectively, and were calculated using Excel (Microsoft, Redmond, WA). All values were obtained from two independent experiments.

#### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.2c02380.

HR-ESIMS, 1D and 2D NMR spectra of compounds 1, and slice of 2D NOESY spectrum for PANIC (PDF)

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#### **Author Contributions**

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#### Notes

The authors declare no competing financial interest.

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