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Anticancer efficacy of magnetite nanoparticles synthesized using aqueous extract of brown seaweed *Rosenvingea intricata***, South Andaman, India**

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Cancer is a global issue and hence various eforts are being made. Iron oxide is considered a signifcant biochemical agent in the biomedical arena for cancer treatment. Marine macroalgae-mediated iron oxides especially, magnetite (Fe3O4) nanoparticles (NPs) are a prospective alternative to diagnose and treat cancer owing to their fuorescent and magnetic properties. We intend to appraise the usability of the aqueous extract of *Rosenvingea intricata* **(***R. intricata***) in Fe3O4 NPs synthesis and to study their cytotoxic efects against human hepatocarcinoma (Hep3B) and pancreatic (PANC1) cancer cells. In the present study,** *R. intricata* **were collected from the coastal region of South Andaman, India. Aqueous extracts of** *R. intricata* **were utilized to synthesize Fe3O4 NPs via the co-precipitation method. Phycosynthesized Fe3O4 NPs exhibited wide peak at 400–600 nm from ultraviolet–visible difused refectance spectroscopic analysis which validated the formation of NPs. Band edge emission peak at 660 nm in fluorescent spectra confirmed the quantum confinement in Fe₃O₄ NPs. Fourier transform infrared spectroscopy confrmed the role of** *R. intricata* **as a capping and reducing agent with functional groups such as O–H, C–H, C=O, N=O, C=C, C–O, C–N, and C–S arising from amino acids, polysaccharides, aliphatic hydrocarbons, esters, amides, lignins, alkanes, aliphatic amines, and sulfates. Physicochemical properties such as crystallite size (14.36 nm), hydrodynamic size (84.6 nm), irregular morphology, elemental composition, particle size (125 nm), crystallinity, and saturation magnetization (0.90007 emu/g) were obtained from x-ray difractometer, dynamic light scattering, scanning electron microscopy, energy dispersive x-ray spectrometer, high-resolution transmission electron microscopy, selected area electron difraction and vibrating sample magnetometer techniques, respectively. The cell viability showed dose-dependent cytotoxic efects and enhanced the apoptosis against Hep3B and PANC1 cancer cells.** *R. intricata* **extract capped Fe3O4 NPs could be the most appropriate and efective nanomaterial for cancer treatment and management.**

Keywords *Rosenvingea intricata*, Magnetite, Nanotechnology, Biomolecules, Cancer, Apoptosis

Cancer is one of the major reasons for mortality worldwide and is a great apprehension among researchers for therapeutic findings¹. The World Health Organization (WHO) has stated that by 2040, it is expected to reach [2](#page-15-1)7.5 million new cancer cases and 16.3 million mortalities². It is instigated by uninhibited uncharacteristic cell development^{[1](#page-15-0)}. At present, chemotherapy and radiation therapy are leading in the treatment of cancer which has substantial side efects like hair loss, nausea, vomiting, reduced red blood cells, defciency of leukocytes, and low platelet count which affects the patient's well-being and quality of life during treatment³. Hence, the necessity to find an efficient treatment to overcome the disadvantages of the above treatment must be found⁴. Herein, nanomedicine is a signifcant root that fnds a solution to cure cancer and kill cancerous cells using diferent

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nanoparticles (NPs) without many side efects[5](#page-16-0) . Nanotechnology deals with materials at the molecular level and has enormously reformed the arena of science⁶. NPs with size 1–100 nm have eminent properties compared to bulk materials^{[7](#page-16-2)}.

Several methods are available for the synthesis of NPs including physical and chemical methods with the use of toxic chemicals that are exclusive and injurious to human beings^{[8](#page-16-3)}. Hence, evolved the greener approach for the nontoxic synthesis of NPs using aqueous extracts of different organisms like plants^{9,[10](#page-16-5)}, algae^{11,12}, bacteria^{[13](#page-16-8),[14](#page-16-9)}, and fungi^{[15–](#page-16-10)[17](#page-16-11)}. Amongst these, algae-assisted production of NPs is comparatively less discovered. This method is not only cost-effective but causes insignificant menace to ecological health¹⁸. Although, algae are abundantly present in bulk extents at coastal areas^{[19](#page-16-13),[20](#page-16-14)}. However, less research has been carried out globally.

Seaweed or macroalgae is a signifcant renewable source in the coastal area. Red, brown, and green seaweeds play crucial roles in developing several pharmaceutical products²¹. In the field of medicine, seaweeds are widely used to cure stones in gall bladder, stomach disorders, eczema, cancer, kidney failure, scabies, asthma, atherosclerosis, cardiovascular disorders, lung ailments, ulcers, etc.²². The presence of biomolecules such as terpenoids, sterols, lectins, pigments, phlorotannins, ketones, alkanes, fatty acids, polysaccharides, carotenoid, glycoproteins, haloforms, hydroquinones, and phenolic compounds aids the use of seaweeds for drug discovery²³. In the past decades, derivatives of brown seaweed *Rosenvingea intricata* (*R. intricata*) have shown antifungal, anticancer, antiallergic, antidiabetic, antiaging, anti-inflammatory, and antioxidant activities^{[24](#page-16-18)}. Seaweed-mediated NPs have been demonstrated as efficient nanomaterials for the treatment of cancer^{[25](#page-16-19)}.

Metal oxide NPs are assumed to be stable and benign for human beings and among iron oxide NPs especially magnetite (Fe_3O_4) being a semiconductor material finds applications in wastewater decontamination, sensors, purging of organic dyes, carbon capture, energy devices, magnetic storage devices, magnetic hyperthermia treat-ment, targeted drug delivery, and crop development^{[26](#page-16-20),[27](#page-16-21)}. Realizing the medical importance of *R. intricata* and Fe₃O₄ NPs, the objectives of this work are as follows: (1) Synthesis of Fe₃O₄ NPs using aqueous extract of *R. intricata*, (2) characterization of Fe₃O₄ NPs using various spectroscopic techniques and (3) testing its efficacy against human hepatocarcinoma (Hep3B) and pancreatic (PANC1) cancer cells. Hence, Fe₃O₄ NPs were synthesized using an aqueous extract of *R. intricata* collected from the tropical coastal region of Collinpur, South Andaman, India. Synthesized Fe3O4 NPs were characterized using various spectroscopic methods. 3-(4,5-dimethylethiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay, morphological analysis and dual staining assay were used to determine the anticancer efficacy of *R. intricata* mediated Fe₃O₄ NPs.

Materials and methods

Study area description

Mature and well-grown thallus of *R. intricata* were collected from the coastal area of the Collinpur region, which was sandy coast along with various types of rocky substratum spread over the shoreline that helps in the sustenance of the luxuriant development of seaweed fora. Latitude (11° 41′ 33.9" N) and Longitude (92° 35′ 55.1"E) of *R. intricata* collection site was noted from handheld GPS (Garmin etrex Vista H), and the study area map (Fig. [1](#page-2-0)) was generated using ArcGIS sofware (ArcGIS, Version: 10.8.2).

Collection and identifcation of seaweed

The collected seaweed sample was preserved and species-level identification was done based on the external and internal morphology of specimens and sourcing the original protocol and standard for[a28.](#page-16-22) *Rosenvingea intricata* (=*Rosenvingea endiviifolia*) is one of the lesser-known wild species that dwell in the east (Tamil Nadu and Andhra Pradesh) and west coast (Gujarat) of India and Andaman Nicobar Islands^{[21](#page-16-15)[,29](#page-16-23),30}. For the present study, the biomass of this species was collected from the Collinpur region of South Andaman. The species level of identification is done by acquiring the morphological and anatomical character sourcing the original protologue of the species^{[31](#page-16-25)} and referring to the standard Indian seaweed flora^{29[,32](#page-16-26)[,33](#page-16-27)}. The collected specimens have been preserved in the form of herbarium (Specimen No: ACOSTI-NIOT-027) and deposited at the Atal Centre for Ocean Science and Technology, National Institute of Ocean Technology, Port Blair.

Processing of seaweed

The collected seaweed sample was well rinsed with seawater followed by freshwater and Milli Q water to eliminate all debris such as epiphytes, sand particles including other adhering detritus materials. The water content in the seaweed sample was removed using blotting sheets. Cleaned seaweed was shade-dried for 5–7 days and destined for pulverization to get a coarse powder $34,35$.

Preparation of aqueous extract of *R. intricata*

10 g dry powder of *R. intricata* was mixed with 100 mL milli Q water and boiled at 90 °C for 15 min. It was vacuum-fltered to yield the aqueous extract of *R. intricata.* Stored at -4 °C for further usage.

Phycosynthesis of Fe3O4 NPs using aqueous extract of *R. intricata*

To 50 mL aqueous extract of *R. intricata*, 0.6 molar (M) ferric chloride hexahydrate (FeCl₃.6H₂O) was added and stirred for 1 h at 80 °C at 600 rpm. Further, 0.5 M sodium hydroxide (NaOH) was added dropwise to this for the precipitation of NPs. pH is maintained to be 11 (Fig. [2](#page-3-0)). It was further stirred for 30 min, centrifuged, and dried in a hot air oven (Model: ASI-HO-121212) at 60 °C. Powdered using an agate mortar and pestle¹¹.

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Location of the seaweed collection site

Figure 1. Location of the seaweed collection site map (ArcGIS 10.8.2, inset: *R. intricata*).

Characterization of Fe3O4 NPs

Various methods for the description of synthesized Fe₃O₄ NPs in the present work were used for NPs characterization such as:

Ultra violet‑visible difused refectance spectrophotometer (UV–Vis DRS)

The optical behavior of Fe₃O₄ NPs obtained using an aqueous extract of *R. intricata* was analyzed using UV-Vis DRS (Model: Jasco V-750). The sample was prepared by compacting it into a wafer and fitting the wafer to the refectance sample holder. Optical band gap energy is estimated using Eq. ([1](#page-2-1)).

$$
\alpha h v = (h v - E_g)^n \tag{1}
$$

where α = absorption coefficient, hv = photon energy, E_g = optical band gap energy, and n = ½ for a direct transition³⁶.

Fluorescence analysis

Fluorescence property was studied using a fuorometer (Model: PerkinElmer LS 45[\)32](#page-16-26). *R. intricata-*mediated $Fe₃O₄$ NPs were diluted with Milli Q water and the fluorescence spectra were attained by exciting the sample using LED lamps with a resolution of 0.5 nm.

Fourier transform infra‑red spectrometer (FTIR)

Biomolecules adhered to the sample were premeditated using FTIR (Model: PerkinElmer Spectrum Two)³⁶. Briefly, 3 mg *R. intricata-mediated* Fe₃O₄ NPs was mixed and ground in an agate mortar with 250 mg of pure spectroscopic grade potassium bromide (KBr) earlier dried at 300 °C in a muffle furnace (Model: Gefran 400) for 4 days. The ultrapure transparent pellet of 13 mm diameter was attained by smearing a force weight of 10 tons for nearly 10 min. IR absorbance was scanned in the range 450–4000 cm−1.

X‑Ray difractometer (XRD)

Crystalline phases of Fe3O4 NPs were analyzed using XRD (Model: Bruker D8 Advance) with a Cu radiation source. Briefy, the sample was ground into a fne powder using an agate mortar and pestle. It was flled in the empty sample holder and gently pressed using a glass slide. The excess sample was removed from the sample

Synthesis of $R.$ intricata-mediated $Fe₃O₄ NPs$

Figure 2. Schematic illustration of *R. intricata*-mediated phycosynthesis of Fe₃O₄ NPs as a promising agent against pancreatic and liver cancer cells.

holder edges. The sample holder was carefully placed in the XRD slot. Scanned in the range of 20°–70°. The crystallite size was determined using Scherrer's Eq. [\(2\)](#page-2-1):

$$
D = K\lambda/\beta\cos\theta\tag{2}
$$

where D = crystallite size (nm), K = shape factor, λ = wavelength of X-ray radiation, β = full with at half maximum (FWHM) and θ = Bragg angle³⁶.

Dynamic light scattering (DLS)

Measurement of size is a main stage in the design of NPs production and its thriving usage. DLS utilizes scattered light to measure the extent of diffusion in nanomaterials^{29,30}. Particle size measurement was done using DLS (Model: Micromeritics Nano Plus). Fe₃O₄ NPs are distributed in Milli Q water and sonicated for 30 min. The measurement state is maintained with a refractive index of 1.3328 and viscosity of the dispersion as 0.8878 (cP) at temperature 25° $C^{36,37}$ $C^{36,37}$ $C^{36,37}$.

High‑resolution scanning electron microscope (HRSEM) coupled with energy dispersive X‑ray spectrometer (EDX) HRSEM analysis is apt for the visualization of the morphology and size of the nanomaterials with their exterior texture³⁵. The sample was mounted on a stub of metal with adhesive coated with gold and then visualized in the scanning electron microscope HRSEM (Model: F E I Quanta FEG 200). HRSEM coupled with EDX was used for the elemental profiling 36 .

High‑resolution transmission electron microscope (HRTEM) coupled with selected area electron difraction (SAED) Surface texture and particle size were observed using HRTEM (Model: Jeol/JEM 2100). The sample was prepared through drop-coating on carbon-coated TEM grids³⁶. HRTEM coupled with SAED was used for the identifica-tion of crystalline phases of Fe₃O₄ NPs^{[36](#page-16-30)}. The d spacing was calculated using the electron diffraction Eq. ([3\)](#page-3-1)

$$
L \lambda = d R \tag{3}
$$

where L is the distance between the sample and electron micrograph, λ is the X-ray wavelength, and R is the radius of diffraction rings $26,36$ $26,36$ $26,36$.

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Vibrating sample magnetometer (VSM)

Magnetic properties of *R. intricata*-mediated Fe₃O₄ NPs are determined using VSM (Model: Lakeshore, USA, 7407) at moderate conditions and the magnetization value is analytically noted with the applied magnetic feld to an extent of 15,000 Gauss. The squareness ratio is estimated using relation ([4](#page-4-0)).

$$
Squares ratio = \frac{M_r}{M_s}
$$
 (4)

where M_r = remanent magnetization and M_s = saturation magnetization^{[36,](#page-16-30)37}.

Anticancer studies

Cell culture conditions

The Human liver cancer cells Hep3B and pancreatic cancer cells PANC1 were purchased from the National Centre for Cell Science (NCCS, Pune, India). Cells were cultured in Dulbecco's modifed Eagle's medium with high glucose (DMEM-HG) media ((Make: Himedia; Cat No: AT151-5L)) containing 10% FBS (Make: Himedia; at No: RM9955-500 ml), Antibiotic Antimycotic solution 100X (Make: Himedia; Cat No: A002) and 37 °C temperature provided with 5% $CO₂$. During subculture, once every 2-3 days, the medium was replaced with a fresh one. Cell-based experiments were performed afer reaching 80% to 90% confuency.

In vitro cytotoxic studies using MTT assay

The human liver cancer cells (Hep3B) and human pancreatic cancer cells (PANC1) cells (Table [1\)](#page-4-1) were indepen-dently seeded (1 X 10^{[4](#page-15-3)} cells/well) in 96 well plates with DMEM medium augmented with 10% FBS, 1X Antibiotic Antimycotic solution in CO₂ incubator at 37 °C with 5% CO₂. Cells were washed with 200 μ L of 1 × phosphatebuffered saline (PBS), afterwards the cells were treated with different doses of *R. intricata*-mediated Fe₃O₄ NPs (25, 50, 100, 250, and 500 μ g/mL) and with half-maximal inhibitory concentration (IC₅₀) of Doxorubicin (3.87 μ g/ mL for Hep3B) and (8 μg/mL for PANC1)) as a positive control in serum-free DMEM medium and incubated for 24 h. The medium was aspirated from cells towards the end of the treatment period. Subsequently, the cells were washed with sterile PBS and imaged using an inverted phase contrast microscope (Make: Optika, Germany). MTT (0.5 mg/mL) prepared in $1 \times PBS$ was added to 100 µl which was added to each well and incubated for a period of 4 h at 37 °C in a CO₂ incubator. Once the incubation time was accomplished, the MTT was cast off from the cells and washed using 200 μL of PBS. The washing phase is for the exclusion of residual non-reduced MTT in the supernatant. Formed formazan crystals were dissolved in 100 μL of dimethyl sulfoxide (DMSO) and carefully mixed. Progress of color intensity was estimated at 570 nm. The dissolution of formazan crystals yielded a purple-blue color. Absorbance was measured at 570 nm using a microplate reader (Model: Lark LIPR-9608). Blank well contained only medium served as a negative control. Experiments were repeated thrice and the IC₅₀ values were attained from the mean of triplicate measurements^{[36](#page-16-30)[–38](#page-16-32)}. The amount of cytotoxicity was evaluated from the MTT protocol. From produced purple formazan crystals, the metabolic functioning of live cells is measured^{[37](#page-16-31)}.

The Percentage of cell viability is calculated using Eq. (5) (5) .

Percentage of cell viability =
$$
\frac{OD\text{ Control} - OD\text{ sample}}{OD\text{ Control}} \times 100\%
$$
 (5)

where OD control=absorbance value of untreated cells, OD sample=absorbance value of treated cells^{[37](#page-16-31)}.

Morphological analysis

Hep3B and PANC1 cancer cells were cultured on a coverslip in DMEM medium overnight and treated with *R. intricata-mediated Fe₃O₄ NPs for 24 h. Then, the medium was discarded and washed with* $1 \times PBS$ *, later the* coverslip was fxed with ethanol and acetic acid solution (3:1 v/v). Coverslips were slightly placed on glass slides and the change in shape of cells was detected using a phase contrast microscope³⁸.

Apoptosis analysis using acridine orange/ethidium iodide assay

Apoptosis is a biotic progression usually categorized by dissimilar structural phenomena by actuating the precise signaling pathways counting the caspase death receptor pathway^{[39](#page-16-33)}. Hep3B and PANC1 cancer cells were developed individually in a 6-well plate for 24 h before attachment. Cells were treated with IC_{50} concentrations of *R*. *intricata*-mediated Fe₃O₄ NPs and incubated for 24 h in a CO₂ incubator. The cells were trypsinized, centrifuged, washed twice with $1\times$ PBS, and resuspended in PBS solution. The cell suspension was withdrawn on the slide trailed by adding 1 mL of acridine orange (1 mg/mL) and 1 mL of ethidium bromide (1 mg/mL) and placed with a coverslip. Additional fuorescent dye was washed with PBS and the apoptotic features were captured under a fluorescence microscope³⁸.

Cell line	Organism	Morphology	Tissue	Disease
Hep3B	Human	Epithelial	Liver	Hepatocellular carcinoma
PANC ₁	Human	Epithelial	Pancreas	Pancreatic carcinoma

Table 1. Details of cell lines used.

Statistical analysis

All tests were conducted a minimum of thrice in triplicate. T-test was used for the approximation of intergroup variations using GraphPad Prism 5 software. The data was expressed as mean and standard error mean (SEM). When a *p* value was fewer than 0.05, it was assumed statistically substantial.

Results and discussion

Optical properties of *Rosenvingea* **intricata‑mediated Fe3O4 NPs**

UV‑DRS analysis

The absorbance peak at 260 nm (Fig. [3a](#page-5-0)) corresponds to the transition of ligand to metal charge transfer (LMCT) in the tetra-coordinated Fe³⁺ (t₁—t₂ and t₁—e) and the broadness in the peak around 400–550 nm was accredited to the existence of Fe³⁺ and Fe²⁺ ions at the octahedral sites. Also, the enlargement of the absorbance band denotes NPs synthesis. Peak amplification owes to the oxygen to metal charge transfer $(O^2 - Fe^{3+})^{40}$ $(O^2 - Fe^{3+})^{40}$ $(O^2 - Fe^{3+})^{40}$. The direct optical band gap was calculated to be 2.30 eV by inferring the tangent curve⁴¹ (Fig. [3b](#page-5-0)).

Band gap arises due to the electrically conductive nature of a material. Semiconductors are materials with a narrow band gap. The versatility of semiconductors is that this optical phenomenon makes these materials appropriate for photocatalytic and antiaging applications. The band gap is calculated to be 2.30 eV which is quite narrow proving the possibility of using *R. intricata-*mediated Fe3O4 NPs in sunscreen lotions and wastewater treatment plants as catalysts 37 .

Fluorescence analysis

The fluorescence spectrum (Fig. [4\)](#page-6-0) of *R. intricata*-mediated Fe₃O₄ NPs exhibited sturdy emission with an intense peak at 660 nm which is credited to the band edge emission and the external imperfections arising from iron vacancy at the tetrahedral site. Surfactants capped $Fe₃O₄$ NPs depict sharp emission which deters NPs from energy exchange with the neighboring atoms and disrupts the expansion of the wave function of an electron. Tis phenomenon is owing to the efect of quantum confnement and the surface state in nanometers and biomolecule interactions. Biogenic capping agents enable control over particle size thereby enhancing intense fuorescence and in biological applications, it aids biofunctionalization for better adherence of biomolecules and drugs. During the explosion process, the emission is an indication of the production of ions. Some of these ions get stuck to the exterior part of NPs and produce emissive centers. Supposed fuorescence is caused by the interface between emissive centers and the interface surrounding. Some allowed transitions are permitted through this emission superfcial interface energy bands. Symbolically, semiconductor NPs holding photoexcited charge carriers possess a crucial part in medical and environmental arenas. Especially, for localized cancer treatment and wastewater decontamination, emission-bound excitation in NPs makes them competing materials $37,42$ $37,42$.

Structural properties of *R. intricata* **mediated Fe3O4 NPs**

FTIR analysis

Table [2](#page-6-1) and Fig. [5](#page-7-0) depict the FTIR spectrum data of the seaweed aqueous extract and *R. intricata-*mediated Fe3O4 NPs. Strong absorption around 3430 cm−1 and 3420 cm−1 is due to the existence of O–H groups from the polysaccharides and amino acids, inherent to seaweeds. These electron-rich biomolecules with hydroxyl (OH⁻) groups arising from amino acids and polysaccharides have the efficacy of reducing iron ions (Fe³⁺ and

UV-DRS spectrum and Tauc plot

Figure 3. (**a**) Ultraviolet–visible spectrum and (**b**) Tauc plot for direct optical band gap of *R. intricata* aqueous extract-mediated $Fe₃O₄$ NPs.

Fluorescence spectrum

Figure 4. Fluorescence spectrum of *R. intricata* aqueous extract-mediated Fe₃O₄ NPs.

 Fe^{2+}) from their trivalent and divalent oxidation states into metallic Fe⁰. Further, the chemical activity aided by these functional groups converts zero-valent iron into magnetite NPs^{[5](#page-16-0)}. The short peak at 2927 cm⁻¹ validates the existence of a C–H stretch of aliphatic hydrocarbon[s43](#page-16-37). Peaks at 2027 cm−1 and 2026 cm−1 are due to the C–H stretching vibrations⁵. Strong bands at 1634 cm⁻¹ and 1630 cm⁻¹ denoted C=O, N=O stretch signifying a mark of ester and amide causative for distinguishing essence of seaweed and also confrming the presence of lignin with (C=C) group⁴⁴. The peak at 1458 cm⁻¹ is due to the C–N stretching vibration and C–H bending mode of alkanes^{[5](#page-16-0)}. Peaks around 1355 cm⁻¹, 1385 cm⁻¹, and 1345 cm⁻¹ signify the C–O stretching and O–H bending vibrations⁵. Peaks overhead 1271 cm^{−1} denote the phenol groups which are characteristic of seaweeds with excess antioxidants, that could be employed in the progress of efficient food and their extract could be a potential candidate in nutraceutical preparation. This peak denotes the C–O stretching of ring oxygen^{[5,](#page-16-0)45}. Peaks at 1119 cm⁻¹, 1016 cm⁻¹, 1119 cm⁻¹ and 1020 cm⁻¹ confirmed the C–N stretch of aliphatic amines⁴⁶. Feeble absorption at 620 cm−1 is indicative of S=O, C–S, and C=S modes of sulfates existing in the seaweed[47](#page-17-0). *R. intricata* possesses polysaccharides, amino acids, aliphatic hydrocarbons, alkanes, aliphatic amines, and sulfdes, which are the major biomolecules present in it. It is believed that aquatic macroalgae produce a varied diversity of organic constituents that are accountable for their distinctive essence and freshness^{[48](#page-17-1)}. Peaks at 613 cm⁻¹ and 438 cm⁻¹ of *R. intricata*-mediated $Fe₃O₄$ NPs denote the Fe–O stretch of ferrites^{[36](#page-16-30)}.

XRD analysis

XRD is a suitable method to analyze the crystalline property of a material[49.](#page-17-2) Figure [6](#page-7-1) denoted the XRD pattern of *R. intricata* aqueous extract-mediated Fe₃O₄ NPs. Broad peaks at 2θ = 30.42°, 35.44°, 43.11°, 57.02°, and 63.15° resemble (220), (311), (400), (511) and (440) planes of magnetite $Fe₃O₄$ with an inverse cubic spinel structure (JCPDS file no. $89-2355$)^{[36](#page-16-30)}.

The crystallite size was appraised to be 14.36 nm using Scherrer's equation.

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Figure 5. FTIR spectra of aqueous extract of *R. intricata* and its aqueous extract-mediated Fe3O4 NPs.

XRD spectrum

Figure 6. XRD spectra of *R. intricata* aqueous extract-mediated Fe₃O₄ NPs.

DLS analysis

Hydrodynamic particle size distribution was analyzed with a light scattering technique revealing that the average particle size of Fe₃O₄ NPs in colloidal dispersion is measured to be 84.6 ± 4.6 nm (Fig. [7](#page-8-0)) with a polydispersity index (PDI) of 0.479. It is observed that the size distribution of Fe₃O₄ NPs is much unvarying with a narrow distribution range³⁶.

Figure 7. DLS plot of *R. intricata* aqueous extract-mediated $Fe₃O₄$ NPs.

Morphological properties of *R. intricata* **mediated Fe3O4 NPs**

SEM with EDX analysis

HRSEM analysis (Fig. [8](#page-8-1)a and b) confrmed the irregular shape with minor agglomerates with a rough surface. HRSEM coupled with EDX confrmed the presence of iron at 0.8, 6.3, and 6.9 keV in higher levels with minimum levels of carbon, chlorine, and sodium from the seaweed extract^{[50](#page-17-4)}. Peak at 0.5 keV signifies the occurrence of oxygen with a weight percent of 19.74% (Fig. [9](#page-9-0)a and b). The presence of iron and oxygen to maximum denotes the successful synthesis of $Fe₃O₄$ NPs.

TEM with SAED analysis

HRTEM investigation proved the uniform size distribution, shape, and defects in the structure. HRTEM images (Fig. [10a](#page-9-1) and b), confrm the spherical-shaped NPs with a size of 125 nm and lattice fringe of 0.19 nm denot-ing the plane (400). The mean particle size distribution is 125 nm (Fig. [10](#page-9-1)c). SAED pattern (Fig. 10d) displays well-defned concentric bright circles aligned to the planes obtained from XRD data. Concentric bright circles consisting of small dots confirmed that the *R. intricata-mediated* Fe₃O₄ is highly crystalline. The existence of distinctive rings with tiny spots denoted the presence of planes conforming magnetite $36,51$ $36,51$.

Magnetic properties of *R. intricata***‑mediated Fe3O4 NPs**

VSM analysis

VSM hysteresis loop **(**Fig. [11](#page-10-0)**)** discloses that the synthesized material is superparamagnetic with saturation magnetization (M_s), remanent magnetization (M_r), and coercivity (B_c) values as 0.90007 emu/g, 0.17876 emu/g, and 335.72 G. Reduced magnetization (M_s) arises from spherical form, smaller size, exterior anisotropy, and disordered spins at NPs exterior layer³⁷. The squareness ratio is estimated to be 0.1986. A squareness ratio less than 0.5 is a symbolic sign of small single-domain, randomly arranged spherical particles. Superparamagnetic nature aids in targeted drug delivery in cancer treatment^{[36](#page-16-30),[37](#page-16-31)}.

Figure 8. (a) and (b) HRSEM images of *R. intricata* aqueous extract-mediated Fe₃O₄ NPs.

SEM images

EDX spectrum

Figure 9. (**a**) EDX spectrum and (**b**) Bar graph of elemental constituents of *R. intricata* aqueous extractmediated $Fe₃O₄$ NPs.

TEM and SAED images

Figure 10. (**a**) TEM image and (**b**) HRTEM image (**c**) Particle size histogram and d) SAED pattern of *R. intricata* aqueous extract-mediated $Fe₃O₄$ NPs.

Anticancer properties of *R. intricata* **mediated Fe3O4 NPs MTT assay**

Cytotoxicity of *R. intricata* mediated Fe₃O₄ NPs studied against Hep3B and PANC1 human liver and pancreatic cancer cells exhibited dose-dependent effect. The impact of *R. intricata-mediated Fe₃O₄ NPs on live cells is* expressed (Figs. [12](#page-10-1) and [13;](#page-11-0) Table [3](#page-11-1)). From linear regression, IC₅₀ values are calculated as 311.7 µg/mL against

VSM hysteresis loop

Figure 11. VSM hysteresis curve of *R. intricata* aqueous extract mediated Fe₃O₄ NPs.

% cell viability of Hep3B cancer cells

Figure 12. In vitro*,* cytotoxic activity of liver Hep3B cancer cells treated with various R. intricata aqueous extract concentrations mediated Fe₃O₄ NPs (Negative control=untreated cells; positive control=Doxorubicin 3.87 μg/mL). Results were stated as mean relative expression±standard error of three replicates. Results were considered signifcant for ****P*≤0.001 and ns—non-signifcant.

Hep3B cancer cells and 460.5 µg/mL for PANC1 cancer cells. As the concentration of *R. intricata*-mediated Fe₃O₄ NPs upsurges, cell viability declines. Further, cytotoxicity is predominant against Hep3B liver cancer cells while related to PANC1 pancreatic cancer cells^{[36](#page-16-30)} (Table [4](#page-11-2)).

% cell viability of PANC1 cancer cells

extract concentrations mediated $Fe₃O₄$ NPs (Negative control=untreated cells; positive control=Doxorubicin 8 μg/mL). Results were stated as mean relative expression±standard error of three replicates. Results were considered signifcant for ****P*≤0.001 and ns—non-signifcant.

Table 3. Statistical analysis of MTT cell proliferation assay results against Hep3B cancer cells.

Table 4. Statistical analysis of MTT cell proliferation assay results against PANC1 cancer cells.

% cell viability of PANC1 *cancer* **cells**

Cancer cell morphology analysis

Morphological changes of *R. intricata* mediated Fe₃O₄ NPs treated Hep3B and PANC1 cancer cells were visualized under a phase contrast microscope (Figs. [14](#page-12-0) and [15\)](#page-12-1). Control cells (untreated) without NPs displayed no alterations in structure with pure and monodispersed and fne matured cells (Figs. [14](#page-12-0)a and [15](#page-12-1)a). But the cancer cells treated with *Rosenvingea intricata* mediated Fe₃O₄ NPs (Figs. [14](#page-12-0)b-f and [15](#page-12-1)b-f) showed important morphological alterations such as asymmetrical morphology, engorged cells, and cell agglomeration thus confrming

Morphological analysis of Hep3B cancer cells

Figure 14. Cell morphological analysis of diferent concentrations of *R. intricata* aqueous extract mediated Fe3O4 NPs {(**a**) 0 μg/mL, (**b**) 25 μg/mL, (**c**) 50 μg/mL, d) 100 μg/mL, (**e**) 250 μg/mL and (**f**) 500 μg/mL} against Hep3B human liver cancer cell line.

Morphological analysis of PANC1 cancer cells

Figure 15. Cell morphological analysis of diferent concentrations of *R. intricata* aqueous extract mediated Fe3O4 NPs {(**a**) 0 μg/mL, (**b**) 25 μg/mL, (**c**) 50 μg/mL, (**d**) 100 μg/mL, (**e**) 250 μg/mL and (**f**) 500 μg/mL} against PANC1 human pancreatic cancer cell line.

the reduction in the number of viable cells, detached cells and changed the morphology of Hep3B and PANC1 cancer cells^{37[,38](#page-16-32)}.

Morphological analysis of PANC1 *cancer* **cells Apoptotic analysis**

Obtained results depicted exceptional apoptotic effect after 24 h treatment with *R. intricata* mediated Fe₃O₄ NPs. The yellow color cells designated the early apoptotic cells in treated Hep3B and PANC1 cancer cells and colorless cells visualized in control along with green color cells representing healthy cancer cells **(**Figs. [16](#page-13-0) and [17](#page-13-1)**)**. Apoptotic efficacy of *R. intricata* mediated Fe₃O₄ NPs in Hep3B liver cancer cells and PANC1 pancreatic cancer cells suggestively improved the cell mortality over apoptosis, not necrosis[52](#page-17-5). Normally, apoptosis is characterized by

Figure 16. Morphological changes analyzed by phase contrast microscope and apoptotic efect of *R. intricata* aqueous extract mediated Fe₃O₄ NPs (311.7 µg/mL) against Hep3B human liver cancer cell line.

Apoptotic effect on PANC1 cancer cells

Figure 17. Morphological changes analyzed by phase contrast microscope and apoptotic efect of *R. intricata* aqueous extract mediated Fe₃O₄ NPs (460.5 µg/mL) against PANC1 human pancreatic cancer cell line.

fuorescence intensity afer the variations in structural features such as reduced or split chromatin with orange (early apoptosis) and red color (late apoptosis) fluorescence^{[53](#page-17-6)}. The above results support the benefit of *R. intricata*mediated Fe₃O₄ NPs in nanobiology for improved cancer cure.

Apoptotic efect on PANC1 *cancer* **cells**

From Figs. [16](#page-13-0) and [17.](#page-13-1) it was clear that *R. intricata* mediated Fe3O4 NPs showed dose-dependent cytotoxicity against Hep3B and PANC1 cancer cells. Different properties like superparamagnetism $(M_s = 0.90007 \text{ emu/g})$, smaller size (125 nm), spherical morphology, and surface adherence of biomolecules from *R. intricata* have enabled the cytotoxicity through targeted activity, and easy release of metal ions. Superparamagnetism aids targeted anticancer efficacy due to the magnetic field. This advantage makes them adaptable in targeted drug delivery with fewer side effects. VSM studies found that the R. intricata-mediated Fe₃O₄ NPs are superparamagnetic, augmenting the prospective utilization for targeted drug delivery without affecting the healthy cells²⁶. Amidst diferent shapes of nanomaterials, spherical morphology has notable advantages like uniform surface functionalization, thus excess drugs can be efficiently loaded on the exterior portion of nanomaterials for better biological applications. It is also assumed that the total cell uptake in the drug release system is due to the smaller size of the nanomaterials. Smaller the size of nanomaterials better circulation in the human body with superior penetration via tiny capillaries. Functional groups adhered to the nanomaterials ofen enhance the biological properties with minimal toxicity to the human body. In the present research, the use of aqueous extract of *R. intricata* has endorsed dose-dependent cytotoxic effects owing to the biomolecules present in it^{[36](#page-16-30),[37](#page-16-31)}.

Mechanism of anticancer activity

Based on the review of literature, the hypothesis on the mechanism of anticancer activity by Fe_3O_4 NPs (Fig. [18](#page-14-0)), suggests that iron ions released generated reactive oxygen species (ROS) into the cancer cell which destructs the mitochondria through oxidative stress. Thereby, inhibiting the replication of DNA resulting in cell death. Nanomaterials with magnetic properties pierce into the nuclear membrane and cause damage to DNA. ROS at excess levels leads to hydrogen bond breakage in the structure of DNA^{[36](#page-16-30),[37](#page-16-31)}.

Mechanism of anticancer activity by *R. intricata‑***mediated Fe3O4 NPs**

Comparative studies. The anticancer efficacy of different NPs against various cancer cells is compared with previous reports as given in Table [5.](#page-15-4)

Conclusion

This is the first report on the usage of marine brown seaweed *R. intricata* for the synthesis of Fe₃O₄ NPs. The use of ferric chloride hexahydrate and aqueous extract of *R. intricata* at moderate conditions produced Fe₃O₄ NPs. Herein, the direct optical band gap was estimated to be 2.30 eV from UV–Vis DRS analysis. From fuorescence studies, band edge emission is observed with a sharp peak at 660 nm. The biomolecules present in the aqueous extract of *R. intricata* played multiple roles as reducing and capping agents for the synthesis of Fe₃O₄ NPs which was proven from FTIR. Crystalline phases were confrmed from XRD with inverse cubic spinel structure. Hydrodynamic size distribution from DLS proved the smaller size of $Fe₃O₄$ NPs. Irregular morphology with slight aggregates was visualized from HRSEM. Iron and oxygen were present to be maximum from EDX analysis. Sphere-shaped with an average particle size of 125 nm and highly crystalline was witnessed from HRTEM and SAED techniques. The superparamagnetic nature with lesser saturation magnetization (0.90007 emu/g) was confirmed from VSM analysis. From these findings, it is concluded that Fe_3O_4 NPs with magneto-fluorescence properties overlay platform for fuorescent biosensor and magnetic hyperthermia applications in the future.

Mechanism of anticancer activity by R. intricata-mediated Fe,O, NPs

Figure 18. Schematic depiction of the credible mechanism of anticancer activity of *R. intricata*-mediated Fe₃O₄ NPs.

Table 5. Comparative analysis of anticancer efficacy of different NPs.

Owed to the inimitable physicochemical properties of $Fe₃O₄$ NPs, this green synthesized nanomaterial has latent applications in the arena of environmental remediation and biomedicine for catalysis, pollutant detection, cell imaging, and hyperthermia treatment. A further advantage is that the superparamagnetic behavior of $Fe₃O₄$ NPs aids in the easy retrieval of catalyst from reaction solution with an exterior magnet. Likewise, in the feld of biomedicine, this property enables targeted drug delivery. *R. intricata-*mediated Fe3O4 NPs augmented the anticancer efficiency against Hep3B and PANC1 cancer cells with IC_{50} values 311.7 and 460.5 µg/mL, correspondingly. Cell morphological analysis and dual staining assay revealed the characteristic apoptotic changes denoting cell death. These findings endorse that the synthesized Fe₃O₄ NPs can be an appropriate alternative for the treatment of human liver and pancreatic cancer. Also, the fndings can be converted into technology with suitable testing and more trials in other human cell lines and testing in animal models.

Data availability

The data that support the results of this research are available on request from the corresponding author, Swathi Pon Sakthi Sri V. The data are not publicly available owing to their comprising material that could compromise the privacy of researchers.

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S.P.S.S.V. Conceptualization, Methodology, Data Analysis, and Writing-Original Draf. A.S.K.Y. Resources and Methodology. M.S. Methodology and Sofware usage. D.K.J. Project administration, Writing-Review and Editing. N.V.V.K. Writing-Review and Editing. G.D. Supervision, Writing-Review, and Editing.

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Additional information

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