

SCIENTIFIC REPORTS



OPEN

Band Gap Engineering of Hexagonal SnSe₂ Nanostructured Thin Films for Infra-Red Photodetection

Emma P. Mukhokosi, Saluru B. Krupanidhi & Karuna K. Nanda

We, for the first time, provide the experimental demonstration on the band gap engineering of layered hexagonal SnSe₂ nanostructured thin films by varying the thickness. For 50 nm thick film, the band gap is ~2.04 eV similar to that of monolayer, whereas the band gap is approximately ~1.2 eV similar to that of bulk for the 1200 nm thick film. The variation of the band gap is consistent with the theoretically predicted layer-dependent band gap of SnSe₂. Interestingly, the 400–1200 nm thick films were sensitive to 1064 nm laser irradiation and the sensitivity increases almost exponentially with thickness, while films with 50–140 nm thick are insensitive which is due to the fact that the band gap of thinner films is greater than the energy corresponding to 1064 nm. Over all, our results establish the possibility of engineering the band gap of SnSe₂ layered structures by simply controlling the thickness of the film to absorb a wide range of electromagnetic radiation from infra-red to visible range.

Single and few layered metal dichalcogenide materials such as WS₂, MoS₂, MoSe₂, MoTe₂, TaS₂ have been widely explored to develop various electronic devices such as transistors¹, photo-detectors², energy storage³, and humidity sensors^{4–6}, etc because of their unique layer dependent structural and electronic properties such as tunable band gaps^{4–9}. Among the layered materials, SnSe₂ is an earth abundant n-type binary semiconductor whose band gap can be tuned for a wide range of electromagnetic spectrum from 1–2 eV making SnSe₂ attractive material for various electronic device applications⁷. The semiconducting nature of SnSe₂ was discovered in 1955 during an investigation into the fundamental factors that are responsible for intrinsic semiconductivity in certain intermetallic compounds and compounds formed by the metalloids Se and Te⁸. The carrier concentration varies from 10¹⁷–10¹⁹ cm⁻³, electron mobilities μ_e between 0.6–85 cm²/V s^{9–15}, a direct and indirect band gap between 0.9–2.04 eV^{13,16–18}, high absorption coefficient of >10⁴ cm⁻¹. It crystallizes in the CdI₂ hexagonal lattice¹⁹ and is a prototype of transitional metal dichalcogenides, composed of two-dimensional Se-Sn-Se sheets stacked on the top of one another and characterized by strong covalent bonding between Se-Sn-Se atoms and weak interlayer Van der Waal's bonding.

SnSe₂ thin films have been prepared by spin coating^{20,21}, spray pyrolysis^{14,22,23}, chemical vapour deposition (CVD)^{15,18}, molecular beam epitaxy¹⁰, thermal evaporation of Sn and Se elements²⁴, and sputtering²⁵. Here, we deposited Sn films of various thicknesses on soda lime glass (SLG) substrate and annealed the films in selenium atmosphere for 1 h at 450 °C to obtain hexagonal SnSe₂. Interestingly, the band gap varied from 2.04 eV for 50 nm thick film to 1.20 eV for 1200 nm thick one. Furthermore, 400–1200 nm thick films were shown to be sensitive to 1064 nm laser irradiation, while films with 50–140 nm thick is insensitive which is due to the fact that the band gap of thinner films is greater than the energy corresponding to 1064 nm. Over all, our results establish the possibility of engineering the band gap of SnSe₂ layered structure by simply controlling the thickness of the film to absorb a wide range of electromagnetic radiation from infra-red to visible range.

Results and Discussion

The thickness of the deposited Sn films on SLG varied from 30–600 nm and after selenisation, it increased to 50–1200 nm as shown in Figure S1 (Supporting Information) and were measured using Dektak profilometer (deviation is ± 10 nm) and Atomic force microscopy (AFM) for 50 nm thick film only. The overall thickness of SnSe₂ films is almost double the thickness of the Sn films. X-ray diffraction (XRD) patterns of SnSe₂ thin films are

Materials Research Center, Indian Institute of Science, Bangalore, 560012, India. Correspondence and requests for materials should be addressed to K.K.N. (email: nanda@mrc.iisc.ernet.in)

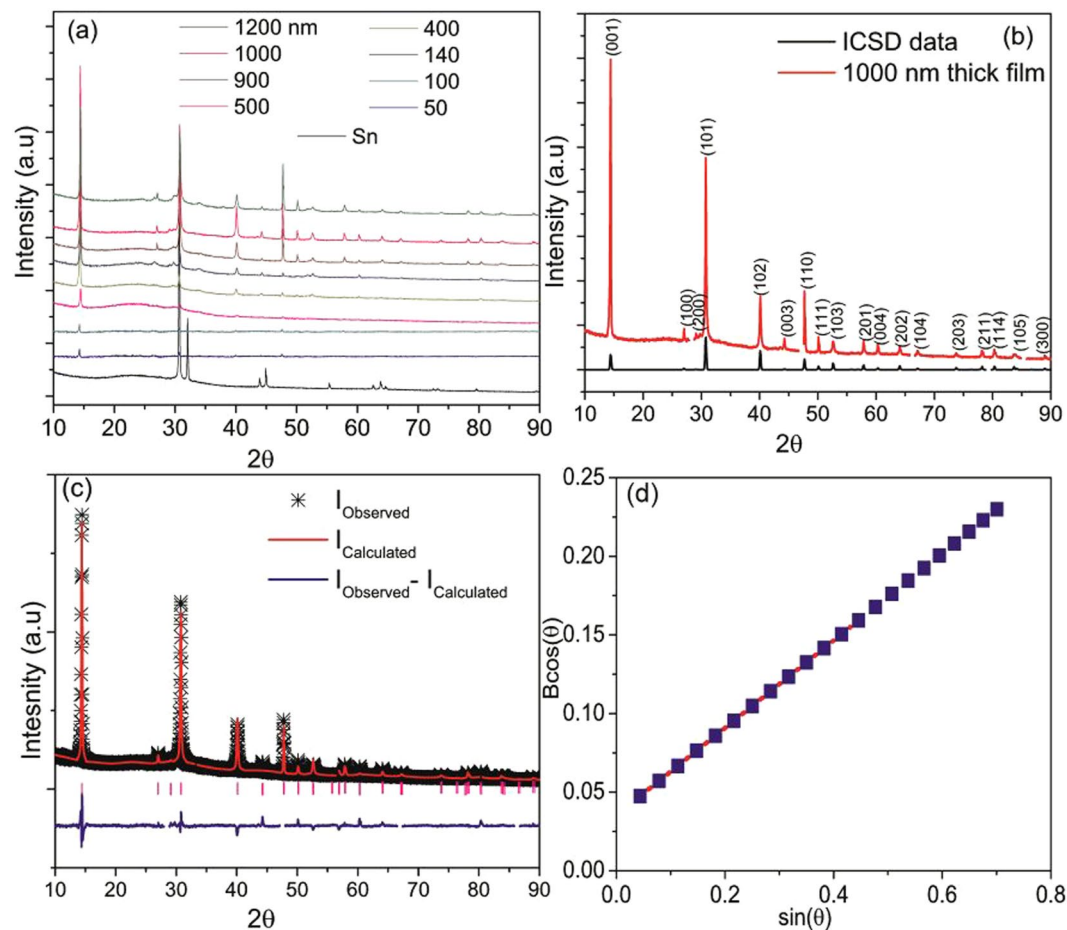


Figure 1. (a) XRD patterns of SnSe₂ films, (b) comparison of 1000 nm thick SnSe₂ film with ICSD data, (c) Rietveld refinement, and (d) Williamson-Hall plot of 1200 nm thick SnSe₂ film.

shown in Fig. 1a and matches well with the ICSD data as evident from Fig. 1b. It is observed that hexagonal pure phase of SnSe₂ with space group P-3 m1 was formed with no evidence of any secondary phases. Rietveld refinement using GSAS²⁶ software program was used to fit the crystal structure for the 1000 nm thick film as shown in Fig. 1c. The criteria for fitting is based on least squares refinement theory which gives indicators of the quality of refinement (SI). The details of the least squares refinement and associated parameters obtained in this study are shown in SI (Tables S1 and S2). The fitting parameters yield $a = b = 0.3818$ nm and $c = 0.6152$ nm with a strain of 0.12% from the Williamson-Hall plot (Fig. 1d).

Figure S2 shows the room temperature Raman spectra of SnSe₂ films. From Figure S2b, two Raman active modes at 115 and 183.5 cm⁻¹ are observed which belong to SnSe₂ hexagonal phase^{6,10,19,25,27,28}. The peak at 115 cm⁻¹ is for E_g mode and is due to in-plane stretching, while the peak located at 183.5 cm⁻¹ is for A_{1g} mode due to out of plane stretching of selenium atoms²⁹. The intensity of A_{1g} mode increases with the thickness (Figure S2c) as is the case of layer dependency⁷.

In order to investigate the chemical electronic states of the prepared SnSe₂ thin films, x-ray photoelectron spectroscopy (XPS) study was performed. In the survey spectrum of SnSe₂, the peaks corresponding to Sn_{3d}, C_{1s} and Se_{3d} were identified as shown in Fig. 2a. In addition, O peak observed may be from the SLG substrate. The spectra of Sn_{3d} and Se_{3d} were measured to determine the oxidation states of the constituent elements. The spectra of the constituent elements is shown in Fig. 2b and c. Sn_{3d} state splits into two states of Sn_{3d3/2} and Sn_{3d5/2} with binding energy of 493.01 and 484.55 eV respectively with a peak to peak separation of 8.46 eV, which is an indication of Sn⁴⁺^{30–32}. The peaks at 54.46 and 53.59 eV are related to Se_{3d3/2} and Se_{3d5/2} states, respectively^{30–34}. Figure S3 shows the energy dispersive x-ray (EDX) spectrum of 1200 nm thick SnSe₂ films. The presence of Cu and C in the spectrum emerges from carbon coated copper grids. We can conclude that the SnSe₂ thin films on SLG, confirms the oxidation states of constituent elements of Sn and Se with no presence of other oxidation states.

Transmission electron microscope (TEM), high resolution TEM (HRTEM) images and selected area electron diffraction (SAED) pattern shown in Fig. 3(a–f) confirm SnSe₂ hexagonal crystal structure. The strong reflections in SAED pattern shown in Fig. 3b and HRTEM fringes presented in Fig. 3c and d strongly confirm further the high crystallinity and layered hexagonal crystal structure of SnSe₂ with inter layer distances of about 0.643 nm¹⁵. The obtained fringes of ~0.6430 and 0.291 nm in HRTEM images shown in Fig. 3(c–e) and f correspond to (001) and (100) crystal plane of the hexagonal crystal system^{35,36}. In the SAED pattern (Fig. 3b), the crystal planes of (101), (003), (202) and (301) correspond to d-spacings of 0.2894, 0.2100, 0.1475 and 0.1034 nm, respectively

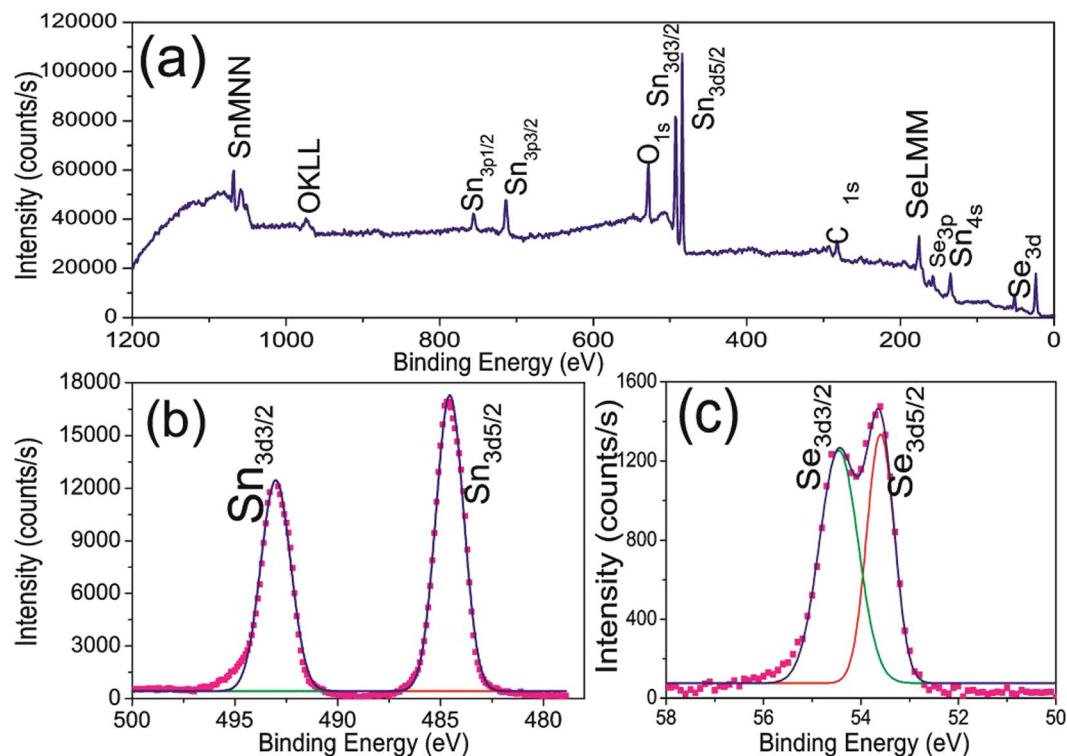


Figure 2. (a–c) Typical XPS survey, Se and Sn spectra. No presence of other oxidation states is realized.

of the hexagonal crystal system with space group P-3m1^{18,37,38}. These results are consistent with XRD planes for the hexagonal crystal system with space group P-3m1 shown in Fig. 1b. HRTEM images shown in Fig. 3(g–i) for 900, 400 and 140 nm thick films also show similar features as other thicknesses. The diffraction spots in SAED of Fig. 3b appear elongated and may suggest the formation of small crystalline domains inside the SnSe₂ layers which are disoriented with respect to each other^{39,40}.

The optical properties of SnSe₂ thin films on SLG substrate were evaluated by taking diffuse reflectance spectra (DRS) between 300–2000 nm of wavelength as shown in Fig. 4a. Using the Kubelka-Munk (KM) function⁴¹, the DRS was then converted to an equivalent absorption spectra as shown in Fig. 4b. The KM function at any wavelength is given by $F(R_{\infty}) = \frac{(1 - R_{\infty})^2}{2R_{\infty}} = \frac{\alpha}{S}$, where R_{∞} is the reflectance of the film relative to the reference material (i.e. $\frac{R_{\text{sample}}}{R_{\text{reference}}}$), α is the absorption coefficient and S is the scattering coefficient. The scattering coefficient is weakly dependent on energy and therefore, $F(R_{\infty})$ is assumed to be proportional to the absorption⁴². The optical band gap of SnSe₂ thin films were then estimated from Tauc plot⁴³ as shown in Fig. 4c and Figure S4 i.e. a plot of $(\alpha h\nu)^2$ against $h\nu$ for direct band gap material and the variation in band gap with thickness is shown in Fig. 4d. From Fig. 4d, it can be noted that the band gap depends on the thickness of the film which in turn depends on the number of layers⁷. It is interesting to note that the variation of energy corresponding to maximum absorption is similar to that of band gap (Figure S5) and the band gap varies linearly with the inverse of the thickness (inset of Fig. 4d). For 50 nm thick film, the band approximates 2.04 eV similar to that of monolayer thick film, whereas the band gap is approximately 1.2 eV for the 1200 nm thick film similar to that of bulk⁷. The variation of band gap with thickness is in excellent agreement with the theoretical prediction of layer dependency and is in accordance with other reports^{4,7,42–45}. The large band gap observed for 50 nm thick film may be due to quantum confinement of layered d-electron dichalcogenides (Se²⁻) and has been observed in other transitional metal dichalcogenides^{46–50}.

Room temperature Hall measurements for all the films revealed that SnSe₂ films are n-type materials with a mobility in the range of $\mu_e = 2.0\text{--}8.0 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, resistivity of $\rho = 5\text{--}30 \text{ } \Omega \text{ cm}$ and carrier concentration of $n_e = (0.95\text{--}6) \times 10^{17} \text{ cm}^{-3}$. The variation of resistivity, mobility and carrier concentration with thin film thickness are shown in Figure S6. It is interesting to note that the resistivity is optimum for a thickness of 700–900 nm, while the mobility peaks around 900 nm. Similarly, the carrier concentration is optimum around 700–800 nm. The better connectivity (Figure S7) of grains in the film is believed to be the reason for the better electrical properties. Over all, the mobility and carrier concentration are optimum for 700–900 nm thick films.

It can be noted from Fig. 4d that the band gap of thicker films (>400 nm) is below 1.3 eV, while that of thinner films (<140 nm) is above 1.6 eV. This indicates that thinner films can not be used for IR (1064 nm = 1.165 eV) photodetection. In order to substantiate this speculation, the photoresponse of SnSe₂ thin films under 1064 nm laser illumination was evaluated using the device configuration shown in Fig. 5a. The optical photograph of the typical device structure is shown in the background. Figure 5b shows the current (I)-voltage (V) linear behaviour of SnSe₂ under dark and 1064 nm laser illumination with different power density (0 to 250 mW/cm²) and

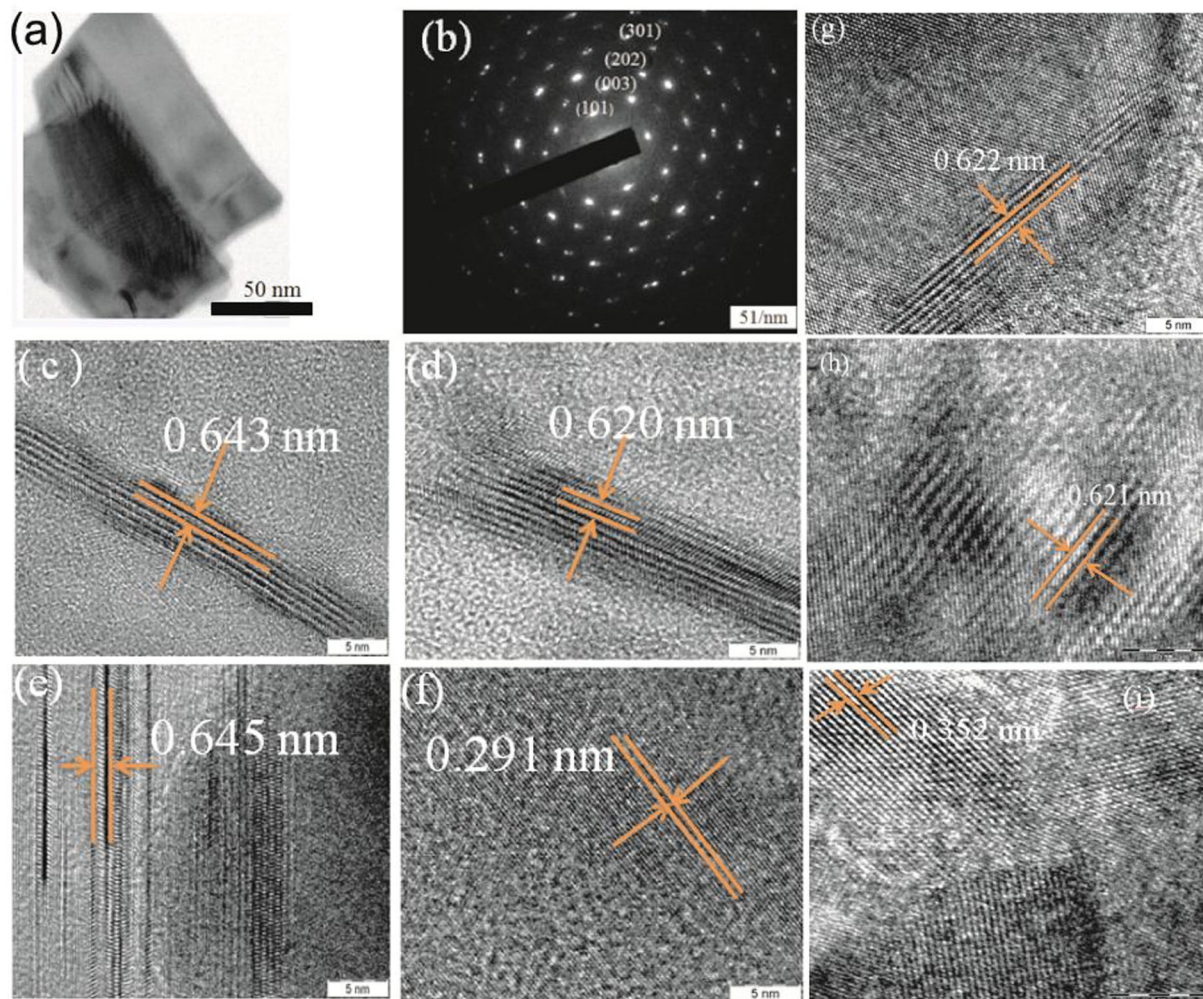


Figure 3. (a) TEM image and (b) SAED pattern and (c) HRTEM image of a 1000 nm thick SnSe₂ film. (d,e) HRTEM images of a 1200 nm thick film. HRTEM images of (g) 900 nm, (h) 400 nm and (i) 140 nm thick SnSe₂ films.

we observe that there is an increase in current upon the laser illumination on 1200 nm thick SnSe₂ film. The photocurrent is measured under alternating dark and 1064 nm laser light. Figure 5c and S8 show the temporal photoresponse of 1200 nm thick SnSe₂ film under 1064 nm laser illumination with different power densities upto 250 mW/cm² at a bias voltage of 10 and 5 V, respectively.

It can be noted that the photocurrent increases with increasing bias and power density. The dependence of photocurrent on power density shown in Fig. 5(d) is fitted using the power law⁵¹: $I_{ph} \propto P^m$ and the exponent m which determines the response characteristic of a photodetector with incident power density, is found to increase from 0.7 to 0.9 when the bias is increased from 5 to 10 V. The value of m indicates that trap states and interactions between the photogenerated carriers (electron-hole pairs) are involved in the recombination kinetics of photo-carriers⁵¹ at lower bias and the value of m is close to ideal unity value at higher bias which indicates that the photo-generated current can be attributed to efficient separation of electron-hole pairs with less trap states and interaction between photo-generated carriers.

Photoresponse of different films with a power density of 250 mW/cm² and a bias of 10 V are shown in Figure S9(a–f) and the photocurrent as a function of thickness is shown in Fig. 6(a). The sensitivity (defined as $S = I_{\lambda}/I_{\text{dark}}$, $I_{\lambda} = I_{\text{light}} - I_{\text{dark}}$, I is the current) as a function of thickness is shown in the inset of Fig. 6(a). It is interesting to note that thinner films do not respond to the 1064 nm irradiation, while the sensitivity increases with thickness for thicker films. As indicated before, the band gap of thinner films is higher than the energy corresponding to 1064 nm and hence, not responsive (Figure S10). In order to testify the band gap dependency, we irradiated 140 nm thick film with white light. Interestingly, the film is found to be sensitive to white light (Figure S11) ensuring the thinner film is insensitive to IR because of the higher band gap. The sensitivity is found to be 90.4% for visible light with 100 mW/cm². Figure 6b shows the variation of responsivity (defined as $R_{\lambda} = I_{\lambda}/(P_{\lambda}A)$, A is the effective surface area and P_{λ} is the power density). Figure 6c and d show the variation of external quantum efficiency (defined as $\text{EQE} = hcR_{\lambda}/q\lambda$) and detectivity (defined as $D^* = R_{\lambda}/(2qI_0)^{1/2}$)⁵² with power density. Similar dependency has also been observed for 5 V (Figure S12). It may be noted that R_{λ} , EQE and D^*

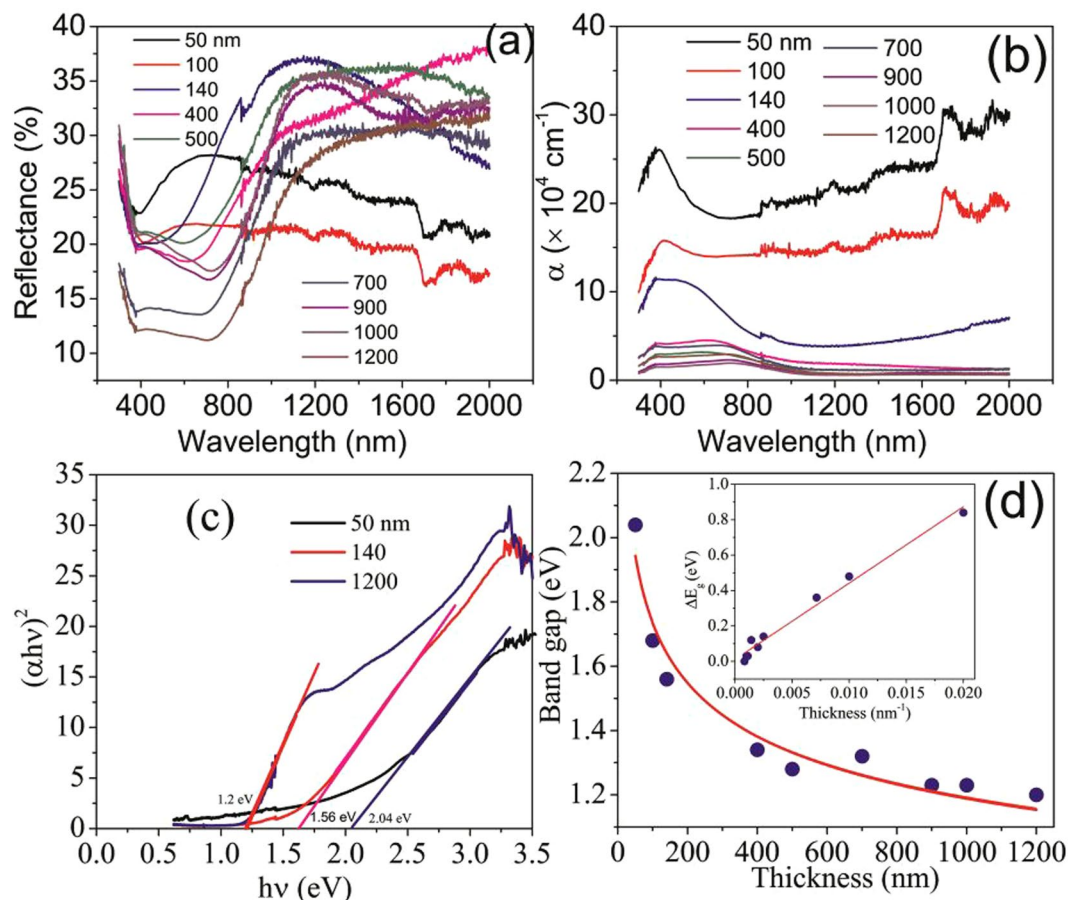


Figure 4. (a) DRS of SnSe₂ thin films on SLG substrate of various thicknesses, (b) absorbance versus wavelength of SnSe₂ thin films, (c) Tauc plot for 50, 140 and 1200 nm thin films, and (d) variation of band gap with thickness. Inset shows the change in band gap as a function of the inverse of the thickness. The solid lines are guide to the eye.

increase with thickness as well as with power density (Figure S13). At this point, we would like to note that the photocurrent and sensitivity (Fig. 6(a)) as well as the dark current (Figure S14) increases with the thickness of the films. As the thickness increases the band gap decreases that promotes the enhancement of the darkcurrent. Similarly, the decrease in band gap also promotes more charge carriers to be generated as depicted in Fig. 7 leading to higher photocurrent and hence, the sensitivity.

Figure S15(a–d) shows the ON and OFF IR response characteristic of the device which is well retained even after six cycle repetitions with their respective fitted growth and decay rate constants at 10 and 5 V and are determined by fitting one cycle of the photo-response curve using the second order exponential equation given by $I(t)_{\text{growth}} = I_{\text{dark}} + \alpha \exp[t/\tau_1] + \beta \exp[t/\tau_2]$ and $I(t)_{\text{decay}} = I_{\text{dark}} + \chi \exp[-t/\tau_1] + \gamma \exp[-t/\tau_2]$ respectively⁵³, where α , β , χ and γ are scaling constants, τ_1 and τ_2 are time constants, t is the time for ON or OFF cycles and I_{dark} is the dark current. From the fits, we estimate the time constants for growth and decay. When the bias voltage was set to 10 V, the photocurrent rises very fast within 0.38 s upon illumination followed by a slower component of 16 s before saturation. The average response time constant for this process is about 2.5 s. The average time constant was calculated from $\tau_{\text{average}} = (\alpha\tau_1 + \beta\tau_2)/(\alpha + \beta)$. After switching off the exciting laser, the photo-current decay follows a second order exponential relaxation process with an estimated time constant of 0.36 and 8.32 s with an average time constant of 3.68 s before reaching the initial dark current. When the bias voltage was 5 V, the response and decay of photoresponse follow a second order exponential relaxation process as is the case of 10 V. The time constants are 0.28 and 15.33 s with an average time constant of ≈ 7.76 s for the sensor response, while the time constants for decay are 0.59 and 14.47 s with an average time constant of ≈ 7.52 s. It can be noted that the fastest response/decay was obtained at a bias voltage of 10 V and is consistent with the power law and is attributed to efficient separation of electron-hole pairs. The general response is quiet slow as compared to other previously reported SnSe₂ layered device¹⁵. This slow response may be due to defects or charge impurity states inside the band gap which act as recombination centres for the photo-generated charge carriers^{2,54}. Previous reports with other layered 2D transitional metal dichalogenides have mainly focused on detecting visible light, a narrow range of the electromagnetic spectrum. The cut-off wavelength for our device is 1064 nm which is superior to those obtained by others^{15,55–57} and suggests that SnSe₂ of about 1200 nm thick grown on SLG can also be used as an efficient IR photodetector. The IR-response and sensitivity of other prepared films in this study are

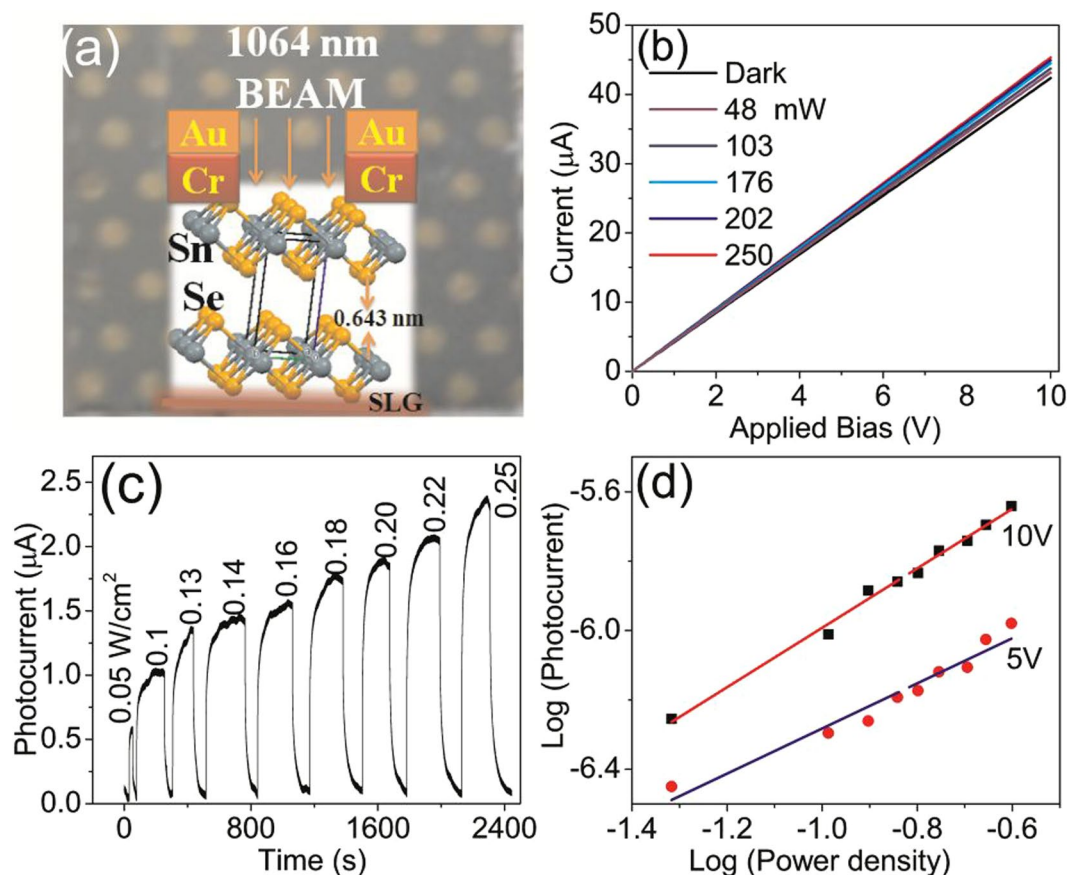


Figure 5. (a) Schematic of device architecture with the film in the background, (b) IV with varying power densities, (c) IR (1064 nm) photoresponse of 1200 nm thick SnSe₂ film with varying power densities under bias voltage of 10 V, and (d) Photocurrent versus power density ($I_{\text{photon}} \propto P^m$).

compared in Table S3. Overall, our results suggest that the SnSe₂ thin film can be explored as an excellent material for photodetection.

Conclusion

We have fabricated and characterized layered SnSe₂ thin films of different thickness on SLG substrate by DC sputtering of Sn metal target followed by selenisation. Hall measurements confirm that SnSe₂ is n-type material with carrier mobility between 2.0–8.0 cm² V⁻¹ s⁻¹, resistivity between 5–30 Ω cm and carrier concentration n_c between $(0.95\text{--}6) \times 10^{17}$ cm⁻³. We have observed that the band gap depends on thin film thickness. For 50 nm thick film, the band gap is ~2.04 eV similar to that of monolayer, whereas it is approximately 1.2 eV for the 1200 nm thick film similar to that of bulk. The IR photodetection response of SnSe₂ demonstrates a sensitivity of ~3% for a 1000 nm thick film and the response time constant is 0.38 s at a bias of 10 V. On the other hand, the sensitivity is 90.4% for 140 nm thick film at 100 mW/cm² which is insensitive to IR. Overall, our results suggest that the SnSe₂ thin film can be explored as an excellent material for photodetection and the sensitivity with response time can be improved further by increasing the applied bias.

Methods

Thin film Deposition. Prior to deposition, soda lime glass substrates (SLG) were cleaned by placing them in dilute HCl for 10 minutes, followed by sonication in de-ionised (DI) water and then boiled in iso-propanol alcohol (IPA) at 82.5 °C for about 15 minutes. The substrates were then purged with nitrogen gas to remove any contaminants such as water vapour. The vacuum chamber was first evacuated to a base pressure of about 5.0×10^{-6} mbar. Sn was then sputtered onto SLG between 15–47 W for 5–20 minutes using argon as plasma source flowing at 1.5 sccm. The chamber working pressure was maintained at 4.0×10^{-3} mbar throughout the deposition to obtain films of various thicknesses ranging from 30 to 600 nm. The substrate was maintained at 20 rotations per minute to obtain films of uniform thickness. The Sn films on SLG was annealed in a calibrated tube furnace as shown in Fig. 8. 1 g of Se powder in Alumina boat was placed in a zone with temperature of about 300 °C (above the melting point of Se), whereas Sn thin films on SLG was placed in a temperature zone of about 450 °C. Prior to selenization process, argon gas was passed through the tube furnace for 5 minutes to drive out the oxygen gas present. The tube furnace ramp rate was maintained at 3 °C per minute to a temperature of 450 °C and held constant for 1 h. The furnace was cooled down to room temperature naturally. Argon gas flow was maintained at 20 sccm throughout the annealing and cooling processes.

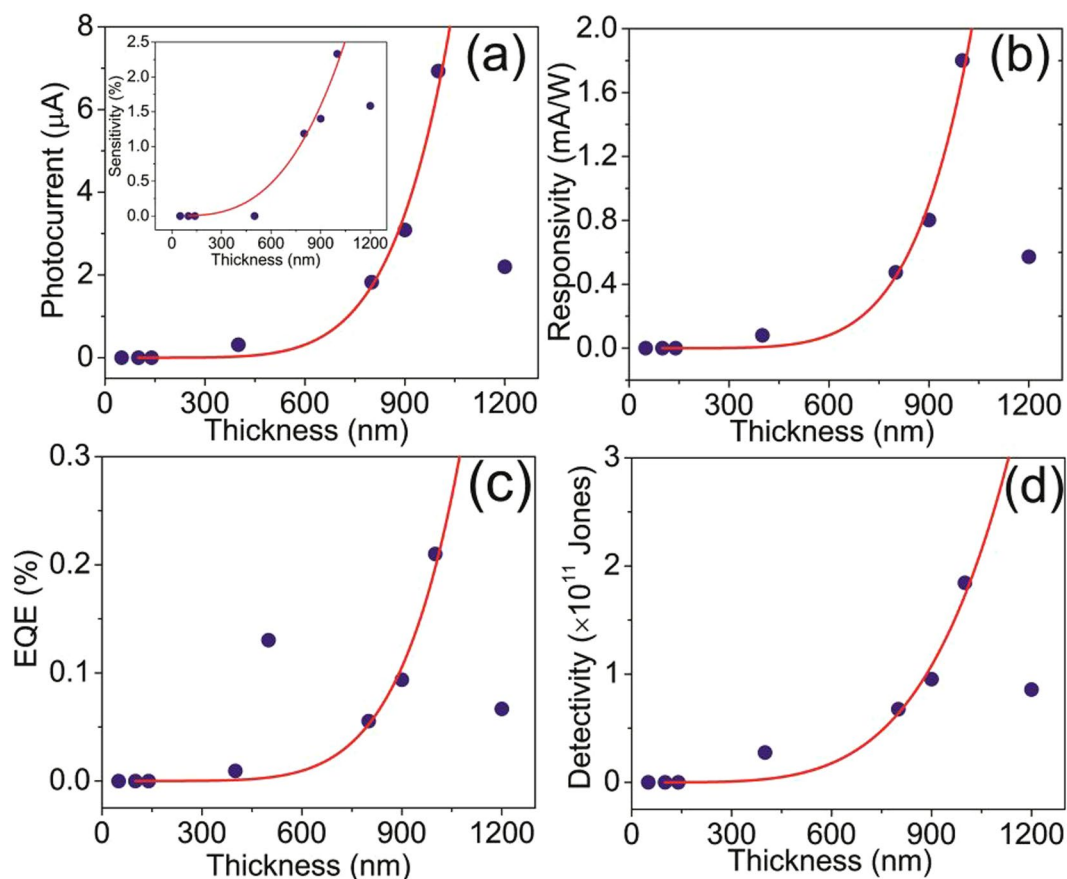


Figure 6. (a) Photocurrent with Sensitivity (inset), (b) Responsivity, (c) EQE, and (d) Detectivity versus film thickness at 10 V and 250 mW/cm^2 . The lines are guide to eye.

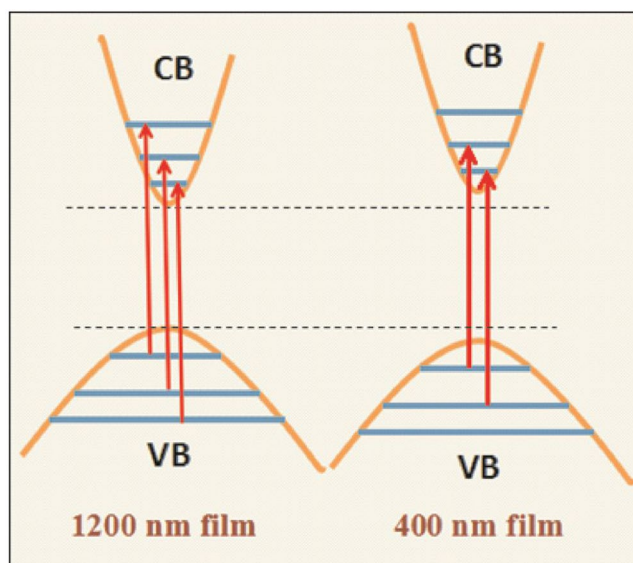


Figure 7. Energy band diagrams for 1200 and 400 nm thick film and the influence of light on the generation of charge carriers.

Thin film Characterization. The crystal structure of the films have been accessed by X-ray diffraction (XRD) using $\text{CuK}\alpha$ (1.5418 \AA) (X^{pert}-PRO PANALYtical instruments). The crystalline properties of the film were further determined using 200 kV FETEM (JEM-2100F) and were prepared by scratching the film surface. The powder collected was dispersed in IPA, sonicated for about 10 minutes and dropped onto carbon coated copper

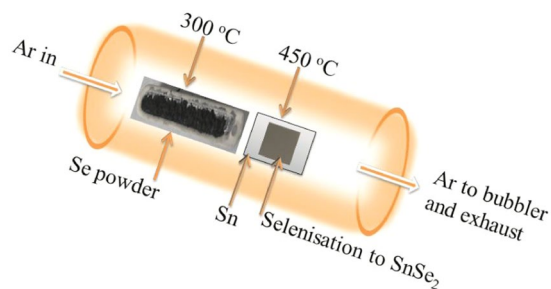


Figure 8. Schematic of a temperature profile inside the tubular furnace with an inner diameter of ~ 5.2 cm and outer diameter of ~ 6.2 cm.

grids. The surface morphology of the films, roughness and thickness were determined using non-contact mode AFM (A.P.E. Research A100-AFM). Veeco Dektak 6 M surface profilometer was used to measure the thickness of the film. Diffuse Reflection Spectrum (DRS) of the thin film was obtained using UV-Vis-NIR spectrophotometer (Perkin Elmer-lambda 750 Instruments). Raman study was carried out at room temperature in the range of $50\text{--}600\text{ cm}^{-1}$ using Visible LabRAM HR instruments with a 532 nm laser. X-ray photoelectron spectroscopy (XPS) measurements were performed using AXIS Ultra DLD X-ray photoelectron spectrometer with $\text{MgK}\alpha$ X-ray source. The system was maintained in ultra-high vacuum at a base pressure of 6.8×10^{-9} Torr. The C1s peak at 284.60 eV was taken as reference to correct the binding energy values of our samples.

Device fabrication and current-voltage measurements. The electrical contacts of Cr/Au (6 nm/80 nm) was thermally evaporated and deposited on top of SnSe_2 thin films. The current-voltage characteristics were measured using a Keithley SMU2400 source meter and 1064 nm laser with varying intensity. The Hall measurements were conducted at room temperature in the presence of 0.55 T magnetic field using Ecopia HMS 5000 Hall effect measurement system and measurements were taken in the Van der Pauw geometry.

Data Availability. The datasets generated during and/or analysed during the current study are available from the corresponding author on reasonable request.

References

- Choi, W. *et al.* High-detectivity multilayer MoS_2 phototransistors with spectral response from ultraviolet to infrared. *Adv. Mater.* **24**, 5832–5836 (2012).
- Chang, Y. *et al.* Monolayer MoSe_2 Grown by Chemical Vapor Deposition for Fast Photodetection. *ACS Nano* **8**, 8582–8590 (2014).
- Chhowalla, M. *et al.* The chemistry of two-dimensional layered transition metal dichalcogenide nanosheets. *Nat. Chem.* **5**, 263–275 (2013).
- Wang, Q. H., Kalantar-Zadeh, K., Kis, A., Coleman, J. N. & Strano, M. S. Electronics and optoelectronics of two-dimensional transition metal dichalcogenides. *Nat. Nanotechnol.* **7**, 699–712 (2012).
- Bharatula, L. D., Erande, M. B. Mulla, I. S. Rout, C. S. & Late, D. J. SnS_2 nanoflakes for efficient humidity and alcohol sensing at room temperature. *RSC Adv.* **6**, (2016).
- Pawbake, A. S., Date, A., Jadhkar, S. R. & Late, D. J. Temperature dependent raman spectroscopy and sensing behavior of few layer SnSe_2 nanosheets. *ChemistrySelect* **1**, 5380–5387 (2016).
- Gonzalez, J. M. & Oleynik, I. I. Layer-dependent properties of SnS_2 and SnSe_2 two-dimensional materials. *Phys. Rev. B - Condens. Matter Mater. Phys.* **94**, 1–10 (2016).
- Mooser, E. & Pearson, W. B. New Semiconducting Compounds. *Phys. Rev.* **101**, 492–493 (1956).
- Evans, B. L. & Hazelwood, R. A. Optical and electrical properties of SnSe_2 . *J. Phys. D. Appl. Phys.* **2**, 1507–1516 (2002).
- Lin, Z. *et al.* Molecular beam epitaxy of large-area SnSe_2 with monolayer thickness fluctuation. *2D Materials*, 2016, 4, 014006.
- Barrios-Salgado, E., Nair, M. T. S. & Nair, P. K. Thin films of n-type SnSe_2 produced from chemically deposited p-type SnSe . *Thin Solid Films* **598**, 149–155 (2016).
- Julien, C., Eddrief, M., Samaras, I. & Balkanski, M. Optical and electrical characterizations of SnSe , SnS_2 and SnSe_2 single crystals. *Mater. Sci. Eng.* **B15**, 70–72 (1992).
- Schlaf, R., Pettenkofer, C. & Jaegermann, W. Band lineup of a $\text{SnS}_2/\text{SnSe}_2/\text{SnS}_2$ semiconductor quantum well structure prepared by van der Waals epitaxy. *J. Appl. Phys.* **85**, 6550 (1999).
- Amalraj, L., Jayachandran, M. & Sanjeeviraja, C. Preparation and characterization of tin diselenide thin film by spray pyrolysis technique. *Mater. Res. Bull.* **39**, 2193–2201 (2004).
- Zhou, X. *et al.* Ultrathin SnSe_2 Flakes Grown by Chemical Vapor Deposition for High-Performance Photodetectors. *Adv. Mater.* **27**, 8035–8041 (2015).
- Gonzalez, J. M. & Oleynik, I. I. Layer-dependent properties of SnS_2 and SnSe_2 novel two-dimensional materials. <https://doi.org/10.1103/PhysRevB.94.125443>
- Guo, P., Luo, Y. W. & Jia, Y. Tuning band gap and optical properties of SnX_2 nanosheets: Hybrid functional studies. *Mod. Phys. Lett.* **B30**, 1650120 (2016).
- Huang, Y. *et al.* Designing the shape evolution of SnSe_2 nanosheets and their optoelectronic properties. *Nanoscale* **7**, 17375–17380 (2015).
- Anwar, S., Gowthamaraju, S., Mishra, B. K., Singh, S. K. & Anwar, S. Spray pyrolysis deposited tin selenide thin films for thermoelectric applications. *Mater. Chem. Phys.* **153**, 236–242 (2015).
- Mitzi, D. B., Milliron, D. J., Copel, M., Murray, C. & Kosbar, L. High mobility solution-deposited chalcogenide films for flexible applications. *2005 IEEE VLSI-TSA - Int. Symp. VLSI Technol. - VLSI-TSA-TECH. Proc. Tech. Pap* **4**, 41–44 (2005).
- Wang, R. Y. *et al.* Electronic and optical switching of solution-phase deposited SnSe_2 phase change memory material. *J. Appl. Phys.* **109**, (2011).

22. Narro-Rios, J. S., Ramachandran, M., Martínez-Escobar, D. & Sánchez-Juárez, A. Ultrasonic spray pyrolysis deposition of SnSe and SnSe₂ using a single spray solution. *J. Semicond.* **34**, 13001 (2013).
23. Anwar, S., Mishra, B. K. & Anwar, S. Optimized substrate temperature range for improved physical properties in spray pyrolysis deposited Tin Selenide thin films. *Mater. Chem. Phys.* **175**, 118–124 (2016).
24. Hady, D. A., Soliman, H., El-Shazly, A. & Mahmoud, M. S. Electrical properties of SnSe₂ thin films. *Vacuum* **52**, 375–381 (1999).
25. De Groot, C. H. *et al.* Highly selective chemical vapor deposition of tin diselenide thin films onto patterned substrates via single source diselenoether precursors. *Chem. Mater.* **24**, 4442–4449 (2012).
26. Larson, A. C. & Von Dreele, R. B. General Structure Analysis System (GSAS). *Structure* **748**, 86–748 (2004).
27. Smith, A. J., Meek, P. E. & Liang, W. Y. Raman scattering studies of SnS₂ and SnSe₂. *J. Phys. Chem* **C1321**, 1321 (1977).
28. Guo, C., Tian, Z., Xiao, Y., Mi, Q. & Xue, J. Field-effect transistors of high-mobility few-layer SnSe₂. 203104, (2016).
29. Cheng, C., Li, Z., Dong, N., Wang, J. & Chen, F. Tin diselenide as a new saturable absorber for generation of laser pulses at 1 μm. *Optics Express* **25**(6), 6132–6140 (2017).
30. Boscher, N. D., Carmalt, C. J., Palgrave, R. G. & Parkin, I. P. Atmospheric pressure chemical vapour deposition of SnSe and SnSe₂ thin films on glass. *Thin Solid Films* **516**, 4750–4757 (2008).
31. Kim, S. I. *et al.* Metallic conduction induced by direct anion site doping in layered SnSe₂. *Sci. Rep.* **6**, 19733 (2016).
32. Du, C.-Fb, Li, J.-R. & Huang, X.-Y. Microwave-assisted ionothermal synthesis of SnSex nanodots: A facile precursor approach towards SnSe, nanodots/graphene nanocomposites. *RSC Adv.* **6**, 9835–9842 (2016).
33. Patil, S. J., Lokhande, V. C., Lee, D. & Lokhande, C. D. Electrochemical impedance analysis of spray deposited CZTS thin film: Effect of Se introduction. *Opt. Mater. (Amst)*. **58**, 418–425 (2016).
34. Ma, R. *et al.* Fabrication of Cu₂ZnSn(S,Se)₄ (CZTSSe) absorber films based on solid-phase synthesis and blade coating processes. *Appl. Surf. Sci.* **368**, 8–15 (2016).
35. Serra, A., Rossi, M., Buccolieri, A. & Manno, D. Solid-to-solid phase transformations of nanostructured selenium-tin thin films induced by thermal annealing in oxygen atmosphere. *AIP Conf. Proc* **1603**, 31–39 (2014).
36. Saha, S., Banik, A. & Biswas, K. Few Layer Nanosheets of n-type SnSe₂. *Chem. - A Eur. J.* **44**, 1–12 (2016).
37. Fang, Z. *et al.* The enhanced photoelectrochemical response of SnSe₂ nanosheets. *Cryst Eng Comm* **16**, 2404 (2014).
38. Huang, L., Yu, Y., Li, C. & Cao, L. Substrate Mediation in Vapor Deposition Growth of Layered Chalcogenide Nanoplates: A Case Study of SnSe Substrate Mediation in Vapor Deposition Growth of Layered Chalcogenide Nanoplates: A Case Study of SnSe₂. *The Journal of Physical Chemistry C* **117**(no. 12), 6469–6475 (2013).
39. Song, Y. *et al.* Pseudocapacitance-tuned high-rate and long-term cyclability of NiCo₂S₄ hexagonal nanosheets prepared by vapor transformation for lithium storage. *J. Mater. Chem.* **A5**, 9022–9031 (2017).
40. Song, Y. *et al.* Bottom-up Approach Design, BandStructure, and Lithium Storage Properties of Atomically Thin gamma-FeOOH Nanosheets. *ACS Appl. Mater. Interfaces* **8**, 21334–21342 (2016).
41. Fiechter, S., Martinez, M., Schmidt, G., Henrion, W. & Tomm, Y. Phase relations and optical properties of semiconducting ternary sulfides in the system Cu-Sn-S. *J. Phys. Chem. Solids* **64**, 1859–1862 (2003).
42. Džimbeg-malčić, V., Barbarić-mikočević, Ž. & Itrić, K. Kubelka-Munk Theory in Describing Optical Properties of Paper (I). *Tehnički vjesnik* **1**, 117–124 (2011).
43. Chand, P., Gaur, A. & Kumar, A. Structural, optical and ferroelectric behavior of hydrothermally grown ZnO nanostructures. *Superlattices Microstruct.* **64**, 331–342 (2013).
44. Im, H. S. *et al.* Facile phase and composition tuned synthesis of tin chalcogenide nanocrystals. *RSC Adv.* **3**, 10349 (2013).
45. Bindu, K. & Nair, P. K. Semiconducting tin selenide thin films prepared by heating Se–Sn layers. *Semicond. Sci. Technol.* **19**, 1348–1353 (2004).
46. Mak, K. F., Lee, C., Hone, J., Shan, J. & Heinz, T. F. Atomically thin MoS₂: A new direct-gap semiconductor. *Phys. Rev. Lett.* **105**, 2–5 (2010).
47. Kuc, A., Zibouche, N. & Heine, T. Influence of Quantum Confinement on the Electronic Structure of the Transition Metal Sulfide TS₂. *Phys. Rev.* **B83**, 245213 (2011).
48. Yun, W. S., Han, S. W., Hong, S. C., Kim, I. G. & Lee, J. D. Thickness and strain effects on electronic structures of transition metal dichalcogenides: 2H-M_x2semiconductors (M = Mo, W; X = S, Se, Te). *Phys. Rev* **B85**, 33305 (2012).
49. Ding, Y. *et al.* First principles study of structural, vibrational and electronic properties of graphene-like MX₂ (M = Mo, Nb, W, Ta; X = S, Se, Te) monolayers. *Phys. B Condens. Matter* **406**, 2254–2260 (2011).
50. Liu, L., Kumar, S. B., Ouyang, Y. & Guo, J. Performance limits of monolayer transition metal dichalcogenide transistors. *IEEE Trans. Electron Devices* **58**, 3042–3047 (2011).
51. Jia, Z. *et al.* Enhanced Photoresponse of SnSe-Nanocrystals-Decorated WS₂ Monolayer Phototransistor. *ACS Appl. Mater. Interfaces* **8**, 4781–4788 (2016).
52. Murali, B. & Krupanidhi, S. B. Facile synthesis of Cu₂CoSnS₄ nanoparticles exhibiting red-edge-effect: Application in hybrid photonic devices. *J. Appl. Phys.* **114**, (2013).
53. Zang, C. *et al.* Photoresponse properties of ultrathin Bi₂Se₃ nanosheets synthesized by hydrothermal intercalation and exfoliation route. *Appl. Surf. Sci.* **316**, 341–347 (2014).
54. Tongay, S. *et al.* Defects activated photoluminescence in two-dimensional semiconductors: interplay between bound, charged, and free excitons. *Sci. Rep.* **3**, 2657 (2013).
55. Chang, R.-P. & Perng, D.-C. Near-infrared photodetector with Cu(In,Al)Se₂ thin film. *Appl. Phys. Lett.* **99**, 81103 (2011).
56. Chang, R. P. & Perng, D. C. Nano-structured Cu(In,Al)Se₂ near-infrared photodetectors. *Thin Solid Films* **529**, 238–241 (2013).
57. Akgul, F. A., Akgul, G., Gullu, H. H., Unalan, H. E. & Turan, R. Enhanced diode performance in cadmium telluride-silicon nanowire heterostructures. *J. Alloys Compd.* **644**, 131–139 (2015).

Acknowledgements

Emma P. Mukhokosi is grateful to Indian Institute of Science through the Office of International Relations for the Ph.D scholarship, grateful to Dr. Mustaque Ali, Dr. Sandra Dias and Mr. Nagabhushan Patel for their support in carrying out the experiments. The authors acknowledge Indian Institute of Science for providing the TEM facility.

Author Contributions

Idea was conceived through discussion of all authors. E.P.M. performed the experiments, S.B.K. and K.K.N. co-supervised the work. All authors contributed to analysis of data, discussion of the results and writing of the manuscript.

Additional Information

Supplementary information accompanies this paper at <https://doi.org/10.1038/s41598-017-15519-x>.

Competing Interests: The authors declare that they have no competing interests.

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2017