

# Argentatin C Analogues with Potential Antinociceptive Activity and Other Triterpenoid Constituents from the Aerial Parts of *Parthenium incanum*

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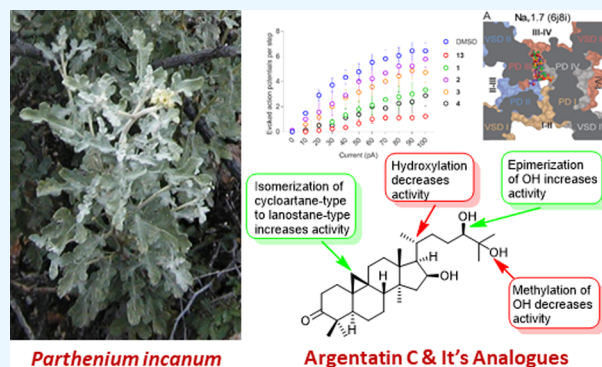
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**ABSTRACT:** Four new triterpenes, 25-dehydroxy-25-methoxyargentatin C (1), 20S-hydroxyargentatin C (2), 20S-hydroxyisoargentatin C (3), and 24-*epi*-argentatin C (4), together with 10 known triterpenes (5–14) were isolated from the aerial parts of *Parthenium incanum*. The structures of 1–4 were elucidated by detailed analysis of their spectroscopic data, and the known compounds 5–14 were identified by comparison of their spectroscopic data with those reported. Since argentatin C (11) was found to exhibit antinociceptive activity by decreasing the excitability of rat and macaque dorsal root ganglia (DRG) neurons, 11 and its new analogues 1–4 were evaluated for their ability to decrease the excitability of rat DRG neurons. Of the argentatin C analogues tested, 25-dehydroxy-25-methoxyargentatin C (1) and 24-*epi*-argentatin C (4) decreased neuronal excitability in a manner comparable to 11. Preliminary structure–activity relationships for the action potential-reducing effects of argentatin C (11) and its analogues 1–4, and their predicted binding sites in pain-relevant voltage-gated sodium and calcium channels (VGSCs and VGCCs) in DRG neurons are presented.



## INTRODUCTION

*Parthenium incanum* Kunth (Asteraceae; common names: mariola, New Mexico rubber plant) is a deciduous shrub native to the Southwestern United States and the Chihuahuan desert of Northern Mexico.<sup>1</sup> Plants of the genus *Parthenium* find applications in Native American medicine to treat pain conditions.<sup>2,3</sup> For example, boiled fresh leaves of *P. incanum* are used by Jicarilla Apaches as a medicinal tea and to treat pregnancy-related discomfort by rubbing preparations over the abdomen.<sup>2</sup> Triterpenes including argentatin A, argentatin B, isoargentatin B, and incanilin have been reported from *P. incanum*.<sup>4,5</sup> Previous investigations have shown that argentatins A and B exhibit antimicrobial<sup>6</sup> and anti-inflammatory<sup>7</sup> properties, and cytotoxic activity against several cancer cell lines.<sup>7–9</sup> We have recently reported the occurrence of argentatins A–C and isoargentatin A in *P. incanum* and antinociceptive activity of argentatin C.<sup>10</sup>

Chronic pain is a pathological state characterized by hyperexcitable sensory neurons, leading to nociception in the absence of painful stimuli. Chronic pain affects about 20–30% of the world population.<sup>11</sup> Historically, humans have used natural remedies derived from plants for medicinal purposes to treat musculoskeletal, neuropathic, and inflammatory pain conditions. Notable examples of these plant-based antinociceptive agents include morphine (from opium poppy),<sup>12</sup> salicin

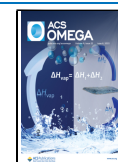
(from willow bark),<sup>13</sup> tetrahydrocannabinol (THC from cannabis),<sup>14</sup> capsaicin (from chili peppers),<sup>15</sup> resiniferatoxin (RTX) from the cactus-like plant *Euphorbia resinifera* (Euphorbiaceae),<sup>16</sup> mitragynine from *Mitragyna speciosa* (kratom; Rubiaceae),<sup>17</sup> salvinorin A from *Salvia divinorum* (Mexican sage bush; Lamiaceae),<sup>18</sup> polygodial from *Drimys winteri* (Winteraceae),<sup>19</sup> and tetrahydropalmitine from *Corydalis yanhusuo* (Papaveraceae).<sup>20</sup> Of these, opioids are the cornerstone of management for many types of pain.<sup>21,22</sup> However, opioids are only partially effective in some pain types and patient populations including neuropathic pain and are often associated with serious side effects including tolerance, addiction, and death that limit their clinical utility.<sup>23</sup> Thus, there is an urgent need for new pain therapies (antinociceptive agents) to circumvent the liabilities of opioid drugs.

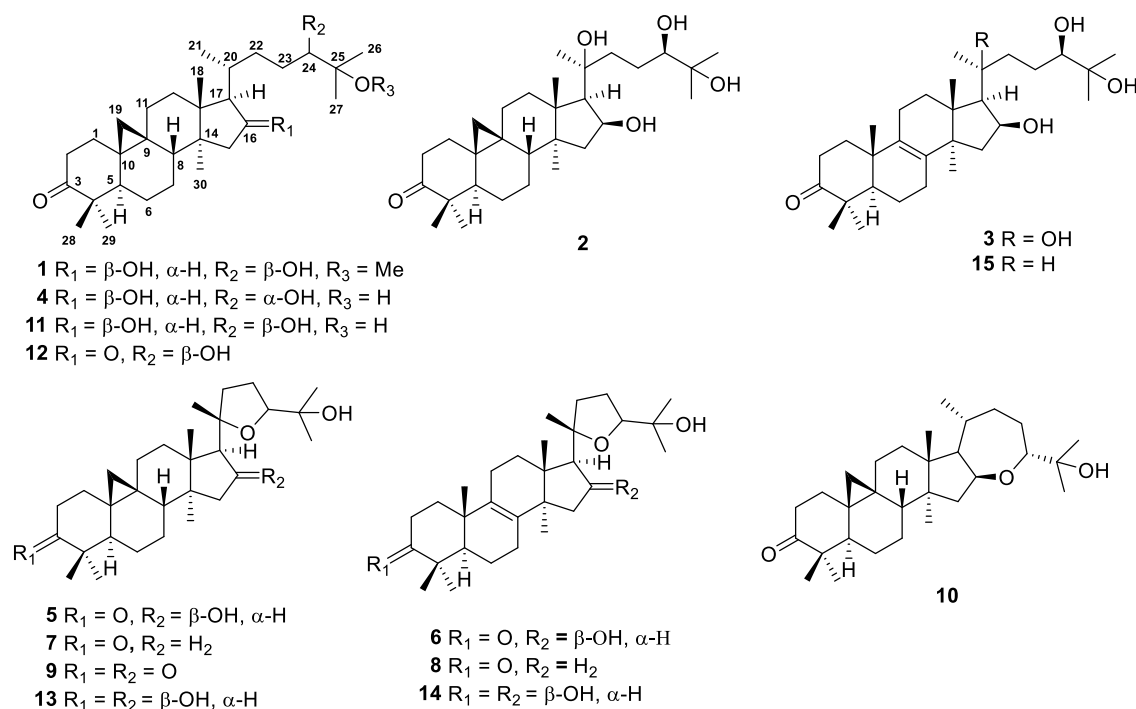
The International Association for the Study of Pain (IASP) defines pain as an unpleasant sensory and emotional

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**Figure 1.** Structures of triterpenoids **1–14** from *P. incanum* and isoargentatin C (**15**).

experience associated with, or resembling that associated with, actual or potential tissue damage.<sup>24</sup> Nociceptors are a subgroup of sensory neurons specialized in the detection of painful stimuli.<sup>25,26</sup> The cell bodies of pain receptors are located in either the dorsal root ganglia (DRG) or the trigeminal ganglia (TG), connecting the peripheral nervous system with the central nervous system (CNS). The key physiological processes for the peripheral transmission of sensory information are transduction, action potential transmission, and transmitter release. In the periphery, the nerve endings have transmembrane proteins called ion channels and receptors, which transduce a noxious stimulus into an electrical signal. Among them, specific voltage-gated sodium channels (VGSCs) and voltage-gated calcium channels (VGCCs) contribute to reach the threshold for action potential generation and determine cellular excitability.<sup>25,27</sup> In addition, these channels support propagation of action potentials to the central terminals within the spinal cord.<sup>28,29</sup> Sustained noxious stimuli or chronic diseases may affect stimulus detection, action potential generation, and transmission, which facilitate nociceptive processing, resulting in transition from acute pain to chronic pathological pain conditions.<sup>30</sup> The hyper-activity of VGSCs and VGCCs in the peripheral tissue injury or in damaged nerve promotes it to become hyper-excitable.<sup>28</sup> Therefore, the discovery of molecules that modulate the excitability in DRG neurons is key to developing antinociceptive agents that target the activity of these ion channels in nociceptive neurons.<sup>31</sup>

Our search for natural products-based antinociceptive agents has thus far resulted in the identification of betulinic acid from *Hyptis emoryi*,<sup>32</sup> physalin F from *Physalis acutifolia*,<sup>33</sup> (–)-hardwickiic acid from *Croton californicus*,<sup>34</sup> hautriwaic acid from *Eremocarpus setigerus*,<sup>34</sup> and argentatin C (**11**) from *P. incanum*.<sup>10</sup> Encouraged by the ability of **11** to attenuate postoperative pain via inhibition of VGSCs and T-type VGCCs,<sup>10</sup> and the reported use of *Parthenium* species to

treat pain conditions,<sup>2,3</sup> prompted us to undertake a detailed investigation of *P. incanum* for its antinociceptive constituents. Herein, we report the isolation and characterization of four new analogues of argentatin C (**1–4**) together with 10 known triterpenes (**5–14**) from the aerial parts of *P. incanum*, and evaluation of **1–4** for their potential antinociceptive activity.

## RESULTS AND DISCUSSION

The crude extract of the aerial parts of *P. incanum* on fractionation by solvent-solvent partitioning, size-exclusion chromatography, and column chromatography followed by HPLC purification afforded four new (**1–4**) and 10 known (**5–14**) triterpenoids (Figure 1). The known triterpenoids were identified as argentatin A (**5**),<sup>35,36</sup> isoargentatin A (**6**),<sup>35,37</sup> 16-deoxyargentatin A (**7**),<sup>35</sup> 16-deoxyisoargentatin A (**8**),<sup>35</sup> 16-dehydroargentatin A (**9**) (previously obtained by chromium trioxide oxidation of argentatin A),<sup>36a,38</sup> argentatin B (**10**),<sup>35,36</sup> argentatin C (**11**),<sup>35,36</sup> 16-dehydroargentatin C (**12**),<sup>35</sup> quisquagenin (**13**),<sup>35</sup> and isoquisquagenin (**14**),<sup>35</sup> by comparison of their spectroscopic data with those reported.

Metabolite **1**, obtained as a colorless amorphous solid, was determined to have the molecular formula  $\text{C}_{31}\text{H}_{52}\text{O}_4$  by a combination of HRESIMS,  $^{13}\text{C}$  NMR, and HSQC data, indicating 6 degrees of unsaturation. The  $^1\text{H}$  NMR spectrum of **1** (Table 1) exhibited six singlet methyls [ $\delta_{\text{H}}$  1.18 ( $\text{H}_3\text{-18}$ ), 1.11 ( $\text{H}_3\text{-26}$ ), 1.09 ( $\text{H}_3\text{-29}$ ), 1.08 ( $\text{H}_3\text{-27}$ ), 1.03 ( $\text{H}_3\text{-28}$ ) and 0.87 ( $\text{H}_3\text{-30}$ )], a methyl doublet [ $\delta_{\text{H}}$  0.92 ( $J = 6.4$  Hz,  $\text{H}_3\text{-21}$ )], two oxymethine protons [ $\delta_{\text{H}}$  4.46 (dt,  $J = 10.0, 4.8$  Hz,  $\text{H-16}$ ) and 3.65 (dd,  $J = 11.6, 3.2$  Hz,  $\text{H-24}$ )], one  $\text{OCH}_3$  [ $\delta_{\text{H}}$  3.21 (s)], and a pair of up-field doublets [ $\delta_{\text{H}}$  0.56 and 0.80 (AB, q,  $J = 4.4$  Hz,  $\text{H}_2\text{-19}$ )] characteristic of nonequivalent protons of a cyclopropyl methylene group in cycloartane-type triterpenoids.<sup>39</sup> The  $^{13}\text{C}$  NMR spectrum of **1** (Table 2), when analyzed with the help of HSQC data, revealed 31 signals due to a ketone carbonyl ( $\delta_{\text{C}}$  216.6), six quaternary carbons of which one is oxygenated ( $\delta_{\text{C}}$  77.2), six methines, including two

Table 1. <sup>1</sup>H NMR Spectroscopic Data of 1–4 and Argentatin C (11)

position	1	2	3	4	11
1	1.80 (m) 1.55 (m)	1.82 (m) 1.51 (m)	1.95 (m) 1.59 (m)	1.81 (dt,4.2, 13.2) 1.50 (m)	1.80 (dt,4.3, 13.2) 1.49 (m)
2	2.69 (td, 14.0, 6.4) 2.29 (dt, 14.0, 2.4)	2.69 (dt, 6.4, 13.8) 2.28 (ddd, 2.6, 4.4, 14.0)	2.57 (ddd, 7.0, 11.4, 15.8) 2.40 (m)	2.68 (dt, 6.4, 13.7) 2.26 (ddd, 13.7, 4.5, 2.7)	2.66 (dt, 6.6, 13.9) 2.26 (ddd, 13.9, 4.7,2.3)
5	1.68 (m)	1.67 (m)	1.57 (m)	1.66 (dt, 12.7, 5.8)	1.66 (m)
6	1.55 (m) 0.95 (m)	1.55 (m) 0.93 (dt, 2.0, 12.6)	1.61 (m) 1.02 (m)	1.53 (m) 0.93 (m)	1.54 (m) 0.94 (m)
7	1.32 (m) 1.12 (m)	1.38 (m) 1.09 (m)	2.05 (m) 1.04 (m)	1.35 (m) 1.11 (m)	1.35 (m) 1.12 (m)
8	1.70 (m)	1.65 m		1.65 m	1.68 (m)
11	2.06 (m) 1.15 (m)	2.05 (m) 1.14 (m)	2.04 (m) 1.03 (m)	2.02 (m) 1.13 (m)	2.03 (m) 1.12 (m)
12	1.74 (m) 1.65 (m)	1.84 (m) 1.64 (m)	1.87 (m) 1.76 (m)	1.66 (dt, 12.7, 5.8) 1.59 (dd, 12.7, 4.2)	2.00 (m) 1.14 (m)
15	2.02 (m) 1.35 (m)	2.03 (dd, 8.0, 13.1) 1.47 (5.8, 13.1)	1.95 (m) 1.78 (m)	2.00 (m) 1.37 (brd, 12.2)	2.01 (m) 1.31 (m)
16	4.46 (ddd, 8.3, 7.2, 5.0)	4.58 (dt, 13.8, 7.5)	4.61 (dt, 13.8, 7.6)	4.46 (ddd, 12.2, 7.5, 4.7)	4.45 (dt, 10.0, 4.8)
17	1.60 (m)	1.81 (d, 8.4)	1.71 (d, 8.4)	1.61 (dd, 11.5, 4.7)	1.59 (m)
18	1.18 (s)	1.42 (s)	1.06 (s)	1.15 (s)	1.14 (s)
19	0.80 (d, 4.4) 0.56 (d, 4.4)	0.80 (d, 4.4) 0.57 (d, 4.4)	1.11 (s)	0.78 (d, 4.4) 0.56 (d, 4.4)	0.78 (d, 4.4) 0.56 (d, 4.4)
20	1.92 (m)			1.81 (m)	1.87 (m)
21	0.92 (d, 6.4)	1.37 (s)	1.37 (s)	0.92 (d, 6.2)	0.88 (d, 6.4)
22	2.04 (m) 1.13 (m)	2.39 (dt, 4.6, 14.2) 1.23 (m)	2.37 (m) 1.24 (m)	1.72 (m) 1.04 (m)	1.60 (m) 1.07 (m)
23	1.82 (m) 1.08 (m)	2.17 (dt, 4.6, 14.2) 1.77 (m)	2.17 (dt, 4.6, 14.2) 1.77 (m)	1.66 (m) 1.21 (m)	1.53 (m) 1.36 (m)
24	3.65 (dd, 11.6, 3.2)	3.38 (brs)	3.37 (brs)	3.34 (d, 10.1)	3.51 (dd, 11.6, 2.8)
26	1.11 (s)	1.29 (s)	1.27 (s)	1.18 (s)	1.18 (s)
27	1.08 (s)	1.20 (s)	1.18 (s)	1.13 (s)	1.11 (s)
28	1.09 (s)	1.08 (s)	1.07 (s)	1.06 (s)	1.07 (s)
29	1.03 (s)	1.03 (s)	1.05 (s)	1.00 (s)	1.02 (s)
30	0.87 (s)	0.84 (s)	0.82 (s)	0.86 (s)	0.87 (s)
OMe	3.21 (s)				

oxygenated carbons ( $\delta_C$  72.4 and 73.8), 10 methylenes, seven methyls, and an OCH<sub>3</sub> carbon ( $\delta_C$  57.21). These data indicated that **1** is structurally related to argentatin C (**11**).<sup>35,36b</sup> In their <sup>1</sup>H and <sup>13</sup>C NMR spectra, the only difference between **1** and **11** was an additional singlet due to an OCH<sub>3</sub> group [ $\delta_H$  3.21 (s);  $\delta_C$  57.1 (CH<sub>3</sub>)] in **1**. The molecular formula of **1** (C<sub>31</sub>H<sub>52</sub>O<sub>4</sub>) indicated that compared to **11** (C<sub>30</sub>H<sub>50</sub>O<sub>4</sub>), it has an additional CH<sub>2</sub> suggesting that one of the OH groups in **11** may have undergone methylation to an OCH<sub>3</sub> group in **1**. The observed heteronuclear multiple bond correlation (HMBC) correlation of the OCH<sub>3</sub> protons ( $\delta_H$  3.21) with the oxygenated quaternary carbon ( $\delta_C$  77.2, C-25) (Figure 2) confirmed that the OH group at C-25 in **11** has been replaced by an OCH<sub>3</sub> group in **1**. The coupling patterns of H-16 and H-24 suggested  $\beta$ -orientation for OH-16 and *R* configuration at C-24, respectively.<sup>35</sup> Thus, the structure of **1** was established as 2*S*-dehydroxy-2*S*-methoxyargentatin C [(1*6* $\beta$ ,24*R*)-16,24-dihydroxy-25-methoxycycloartan-3-one].

The HRESIMS data of metabolite **2** exhibited a quasi-molecular ion peak at *m/z* 473.3601 (calcd for C<sub>30</sub>H<sub>49</sub>O<sub>4</sub> [M-H<sub>2</sub>O+H]<sup>+</sup> 473.3631), suggesting the molecular formula C<sub>30</sub>H<sub>50</sub>O<sub>5</sub> and indicating 6 degrees of unsaturation. The <sup>1</sup>H NMR spectrum of **2** (Table 1) exhibited signals due to seven methyls and a pair of up-field doublets at  $\delta_H$  0.80 and 0.57 (*J* = 4.4 Hz) characteristic of geminal methylene protons of a tetra-

substituted cyclopropane ring suggesting **2** to be a cycloartane-type triterpenoid.<sup>39</sup> The <sup>13</sup>C NMR spectrum of **2** (Table 2), when analyzed with the help of the HSQC data, indicated the presence of 30 carbons consisting of a ketone carbonyl ( $\delta_C$  216.5), seven quaternary carbons of which two are oxygenated ( $\delta_C$  78.6 and 76.0), five methines of which two are oxygenated ( $\delta_C$  73.8 and 69.7), 10 methylenes, and seven methyls. Since the carbonyl group accounted for only 1 degree of unsaturation, **2** should be pentacyclic. The <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data of **2** closely resembled those of argentatin C (**11**).<sup>35,36b</sup> (Tables 1 and 2), suggesting that **2** is an analogue of **11**. Comparison of the molecular formula of **2** (C<sub>30</sub>H<sub>50</sub>O<sub>5</sub>) with that of **11** (C<sub>30</sub>H<sub>50</sub>O<sub>4</sub>) indicated **2** to be an oxygenated analogue of **11**. The <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts of the C-21 methyl group of **2** ( $\delta_H$  1.37;  $\delta_C$  27.1) compared to those of **11** ( $\delta_H$  0.88;  $\delta_C$  17.7) indicated that this methyl is attached to an oxygenated carbon.<sup>40</sup> This was further confirmed by the HMBC correlation of H<sub>3</sub>-21 to C-20 (Figure 2). Based on their <sup>13</sup>C NMR chemical shifts,  $\beta$ -orientation for OH-16 and *S* configuration for C-20 were assigned.<sup>40a</sup> The <sup>13</sup>C NMR chemical shift of C-21 ( $\delta_C$  27.1) further supported the *S* configuration of C-20.<sup>41</sup> The absolute configuration of the chiral center at C-24 of **2** was determined to be *R* by the application of the advanced Mosher's ester method (Figure 3).<sup>42</sup> Metabolite **2** was thus identified as 2*S*-hydroxyargenta-

**Table 2.**  $^{13}\text{C}$  NMR Spectroscopic Data of 1–4 and Argentatin C (11)

position	1	2	3	4	11
1	33.5 CH <sub>2</sub>	33.3 CH <sub>2</sub>	35.8 CH <sub>2</sub>	33.4 CH <sub>2</sub>	33.3 CH <sub>2</sub>
2	37.5 CH <sub>2</sub>	37.5 CH <sub>2</sub>	34.6 CH <sub>2</sub>	37.5 CH <sub>2</sub>	37.4 CH <sub>2</sub>
3	216.6 C	216.5 C	217.7 C	216.6 C	216.6 C
4	50.2	50.2 C	47.4 C	50.2 C	50.2 C
5	48.5 CH	48.5 CH	51.2 CH	48.5 CH	48.4 CH
6	21.4 CH <sub>2</sub>	21.4 CH <sub>2</sub>	19.3 CH <sub>2</sub>	21.4 CH <sub>2</sub>	21.4 CH <sub>2</sub>
7	26.1 CH <sub>2</sub>	25.9 CH <sub>2</sub>	26.2 CH <sub>2</sub>	25.9 CH <sub>2</sub>	25.9 CH <sub>2</sub>
8	47.9 CH	47.5 CH	134.3 C	47.6 CH	47.5 CH
9	20.8 C	20.6 C	133.3 C	20.9 C	20.8 C
10	25.9 C	26.2 C	36.9 C	26.1 C	25.9 C
11	26.4 CH <sub>2</sub>	26.5 CH <sub>2</sub>	20.8 CH <sub>2</sub>	26.4 CH <sub>2</sub>	26.0 CH <sub>2</sub>
12	32.7 CH <sub>2</sub>	33.5 CH <sub>2</sub>	31.9 CH <sub>2</sub> *	32.6 CH <sub>2</sub>	32.5 CH <sub>2</sub>
13	45.3 C	46.2 C	45.3 C*	45.3 C	45.3 C
14	46.5 C	46.3 C	47.2 C*	46.8 C	46.6 C
15	47.9 CH <sub>2</sub>	48.4 CH <sub>2</sub>	43.9 CH <sub>2</sub>	47.7 CH <sub>2</sub>	47.8 CH <sub>2</sub>
16	72.4 CH	73.8 CH	74.1 CH	72.8 CH	72.7 CH
17	57.1 CH	58.6 CH	57.1 CH	57.0 CH	56.7 CH
18	18.9 CH <sub>3</sub>	21.5 CH <sub>3</sub>	19.2 CH <sub>3</sub>	20.0 CH <sub>3</sub>	18.9 CH <sub>3</sub>
19	29.8 CH <sub>2</sub>	30.2 CH <sub>2</sub>	18.4 CH <sub>3</sub>	29.8 CH <sub>2</sub>	29.8 CH <sub>2</sub>
20	26.5 CH	78.6 C	78.4 C	31.5 CH	26.3 CH
21	17.7 CH <sub>3</sub>	27.1 CH <sub>3</sub>	27.6 CH <sub>3</sub>	18.5 CH <sub>3</sub>	17.7 CH <sub>3</sub>
22	26.4 CH <sub>2</sub>	27.2 CH <sub>2</sub>	27.1 CH <sub>2</sub>	33.8 CH <sub>2</sub>	26.4 CH <sub>2</sub>
23	31.4 CH <sub>2</sub>	22.5 CH <sub>2</sub>	22.5 CH <sub>2</sub>	28.6 CH <sub>2</sub>	31.1 CH <sub>2</sub>
24	73.8 CH	69.7 CH	69.6 CH	80.3 CH	75.2 CH
25	77.2 C	76.0 C	76.0 C	73.2 C	73.0 C
26	25.6 CH <sub>3</sub>	27.4 CH <sub>3</sub>	27.4 CH <sub>3</sub>	23.1 CH <sub>3</sub>	22.8 CH <sub>3</sub>
27	25.9 CH <sub>3</sub>	28.1 CH <sub>3</sub>	28.1 CH <sub>3</sub>	26.6 CH <sub>3</sub>	26.8 CH <sub>3</sub>
28	20.8 CH <sub>3</sub>	20.8 CH <sub>3</sub>	26.1 CH <sub>3</sub>	20.8 CH <sub>3</sub>	20.7 CH <sub>3</sub>
29	22.2 CH <sub>3</sub>	22.1 CH <sub>3</sub>	21.3 CH <sub>3</sub>	22.2 CH <sub>3</sub>	22.1 CH <sub>3</sub>
30	20.0 CH <sub>3</sub>	19.9 CH <sub>3</sub>	24.9 CH <sub>3</sub>	19.0 CH <sub>3</sub>	19.9 CH <sub>3</sub>
OMe	57.1 CH <sub>3</sub>				

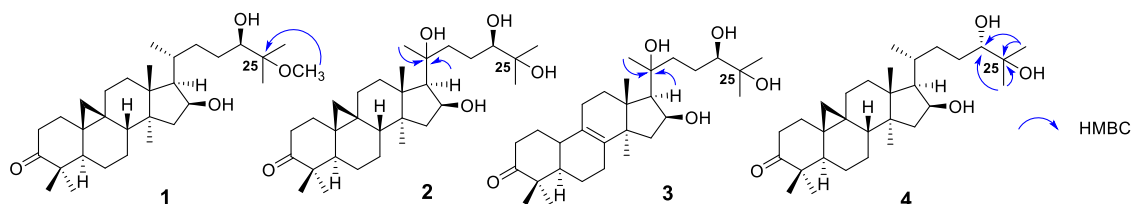
tin C [(16 $\beta$ ,20S,24R)-16,20,24,25-tetrahydroxycycloarten-3-one].

The HRESIMS,  $^1\text{H}$ , and  $^{13}\text{C}$  NMR spectroscopic data of 3 were consistent with the molecular formula  $\text{C}_{30}\text{H}_{50}\text{O}_5$ , indicating 6 degrees of unsaturation. In its  $^1\text{H}$  NMR spectrum (Table 1), 3 exhibited signals due to eight methyl groups, suggesting it to be a lanostane-type triterpenoid.<sup>35</sup> The  $^{13}\text{C}$  NMR spectrum of 3, when analyzed with the help of HSQC data, revealed 30 carbon signals including a carbonyl carbon ( $\delta_{\text{C}}$  217.7), seven quaternary carbons of which two are olefinic ( $\delta_{\text{C}}$  134.3 and 133.3) and two are oxygenated ( $\delta_{\text{C}}$  78.4 and 76.0), four methines of which two are oxygenated ( $\delta_{\text{C}}$  74.1 and 69.6), nine methylenes, and eight methyls. Since the carbonyl group and the double bond accounted for 2 degrees of unsaturation, 3 should be tetracyclic. These data revealed that the structure of 3 is closely related to that of the known lanostane triterpenoid, isoargentatin C (15).<sup>35</sup> The molecular

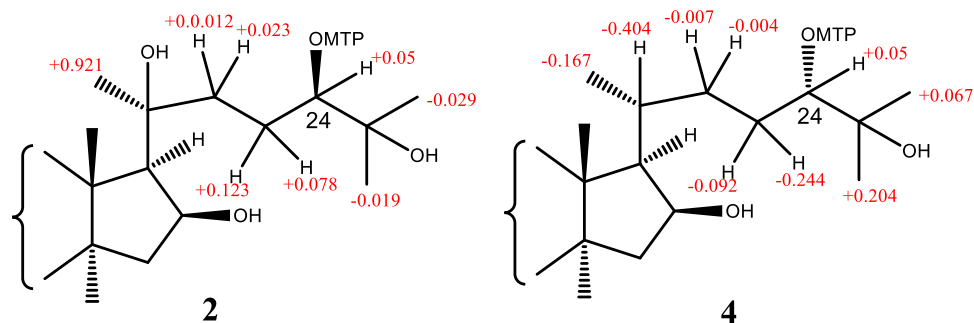
weight difference of 16 Da between 3 and 15 suggested that 3 is an oxygenated analogue of 15. The major difference in their NMR data was found to be the absence of the methyl doublet ( $\delta_{\text{H}}$  0.93) in 3, which was assigned to  $\text{H}_3$ -21 in 15. Instead, 3 showed a singlet ( $\delta_{\text{H}}$  1.37) due to a methyl group attached to an oxygenated carbon ( $\delta_{\text{C}}$  78.4), suggesting the presence of an OH group at C-20. This was confirmed by the presence of an HMBC correlation between  $\text{H}_3$ -21 and C-20 (Figure 2). Based on their  $^{13}\text{C}$  NMR chemical shifts, OH-16 was determined to have  $\beta$ -orientation and C-20 to have *S* configuration.<sup>40a</sup> The coupling pattern of H-24 and the chemical shift of C-24 were found to be similar to those of 20S-hydroxyargentatin C (2) (see above), suggesting the *R* absolute configuration for C-24. Thus, metabolite 3 was identified as 20S-hydroxyisoargentatin C [(16 $\beta$ ,20S,24R)-16,20,24,25-tetrahydroxylanost-8-en-3-one].

Metabolite 4 exhibited a  $[\text{M} + \text{Na}]^+$  ion at  $m/z$  497.3604 in its HRESIMS, consistent with the molecular formula  $\text{C}_{30}\text{H}_{50}\text{O}_4$  which is the same as that of argentatin C (11). The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data of 4 were similar to those of 11 (Tables 1 and 2), except for some signals of the side-chain moiety.<sup>35,36b</sup> Argentatin C (11) contains a 24 $\beta$ ,25-dihydroxylated side-chain moiety. The HMBC spectrum of 4 (Figure 2) suggested that it also contains a 24,25-dihydroxylated side chain as in 11. However, in their  $^{13}\text{C}$  NMR spectra, the signals due to the OH-24 bearing carbon (C-24) of 4 and 11 appeared at  $\delta_{\text{C}}$  80.3 and  $\delta_{\text{C}}$  75.2 ppm, respectively, suggesting that they have different configurations at C-24. Since the configuration of C-24 in 11 is known to be *R*,<sup>35,36</sup> the configuration of this carbon in 4 was suspected to be *S*, and this was confirmed by the application of the advanced Mosher's ester method (Figure 3).<sup>42</sup> Hence, metabolite 4 was identified as 24-*epi*-argentatin C [(16 $\beta$ ,24S)-16,24,25-trihydroxycycloarten-3-one].

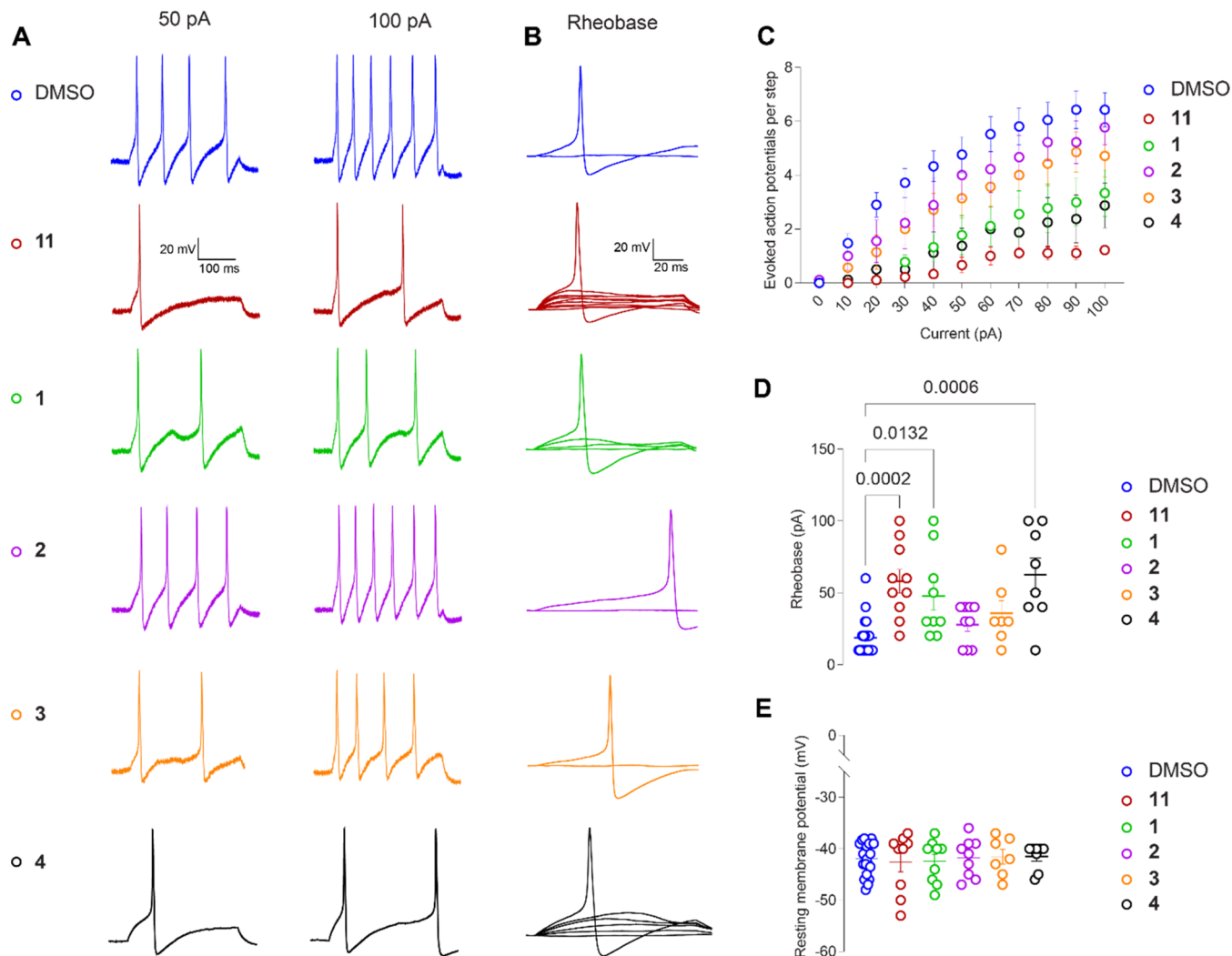
We have previously shown that the crude MeOH/ $\text{CH}_2\text{Cl}_2$  (1:1) extract of *P. incanum* and one of its major metabolites, argentatin C (11), exhibit antinociceptive activity by inhibiting depolarization-evoked calcium influx.<sup>10</sup> In addition, 11 decreased excitability as well as inhibited the current through VGSC and low-voltage-activated calcium channels in rat DRG neurons, whereas the other major metabolites of this extract, argentatin A (5), argentatin B (10), and isoargentatin A (6), had no effect on these neurons.<sup>10</sup> In the current study, the excitability of DRG neurons was tested in the presence of argentatin C (11) and its analogues (1–4). Compounds 1–4 and 11 were acutely incubated with DRG neurons in order to evaluate if these compounds could modulate the excitable properties of these cells which is attributed to VGSCs and VGCCs of rat DRG neurons. The isolated neurons were incubated in the presence of 20  $\mu\text{M}$  of each analogue for 15 min prior to recording and in the external solution during recording. Argentatin C (11) served as an internal positive control. As reported previously, 11 significantly decreased the number of action potentials (APs) observed in response to

**Figure 2.** Key HMBC correlations of 1–4.





**Figure 3.**  $\Delta\delta$  values [ $\Delta\delta$  values (in ppm) =  $\delta_S - \delta_R$ ] obtained for (*S*)- and (*R*)-MTPA esters of **2** and **4**.

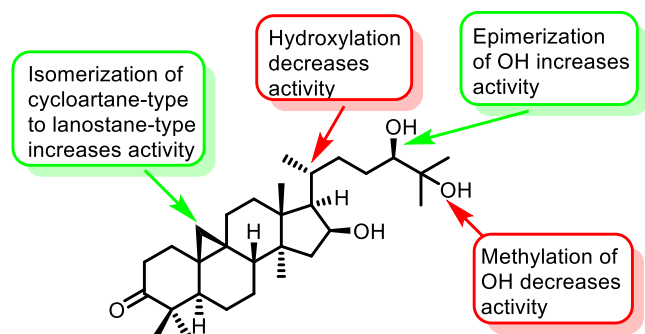


**Figure 4.** Inhibition of excitability in rat dorsal root ganglion (DRG) neurons by argentatin C (**11**) and its analogues **1–4**. (A) Representative traces of action potentials displayed by DRG neurons in culture after incubation with dimethyl sulfoxide (DMSO) or 20  $\mu$ M **11** and **1–4** in response to current injections at 50 and 100 pA for 300 ms. Representative traces (B) and quantification of the rheobase (D) [i.e., current required for eliciting the first action potential (AP)] showing significant increase in the presence of **11** and **1** when compared with DMSO. (C) Quantification of the number of action potentials evoked in response to current injection (see Table S1; Supporting Information). (E) Resting membrane potential (see Table S2; Supporting Information). [Error bars indicate mean  $\pm$  SEM, *p* values as indicated (*n* = 9–10 cells). Kruskal–Wallis test].

each step of injected current (Figure 4A,C and Table S1).<sup>10</sup> 2S-Dehydroxy-2S-methoxyargentatin C (**1**) and 24-*epi*-argentatin C (**4**) also inhibited the number of APs, whereas the excitability of rat DRG neurons in the presence of 20S-hydroxyargentatin C (**2**) and 20S-hydroxyisoargentatin C (**3**) remained unaffected (Figure 4C and Table S1). Argentatin C

(**11**) and its analogues **1** and **4** also increased the rheobase, defined as the minimum current necessary to evoke an AP as indicated by the following data compared to the negative control (0.1% DMSO  $18.57 \pm 2.79$ ): argentatin C (**11**)  $58.0 \pm 8.14$ ; 2S-dehydroxy-2S-methoxyargentatin C (**1**)  $47.8 \pm 9.97$ ; 20S-hydroxyargentatin C (**2**)  $27.78 \pm 4.65$ ; 20S-hydroxyisoar-

gentatin C (3)  $35.71 \pm 8.69$ ; and 24-*epi*-argentatin C (4)  $62.5 \pm 11.61$  (Figure 4B,D). Additionally, the resting membrane potential—an important mechanism regulating excitability—remained unchanged between conditions (Figure 4E and Table S2). Our results show that 25-dehydroxy-25-methoxyargentatin C (1), 24-*epi*-argentatin C (4), and argentatin C (11) diminished the excitability of rat DRG neurons. A summary of preliminary structure–activity relationship (SAR) data for argentatin C and its analogues is depicted in Figure 5. As the



**Figure 5.** Summary of structure–activity relationships of argentatin C analogues for inhibition of excitability in rat dorsal root ganglion (DRG) neurons.

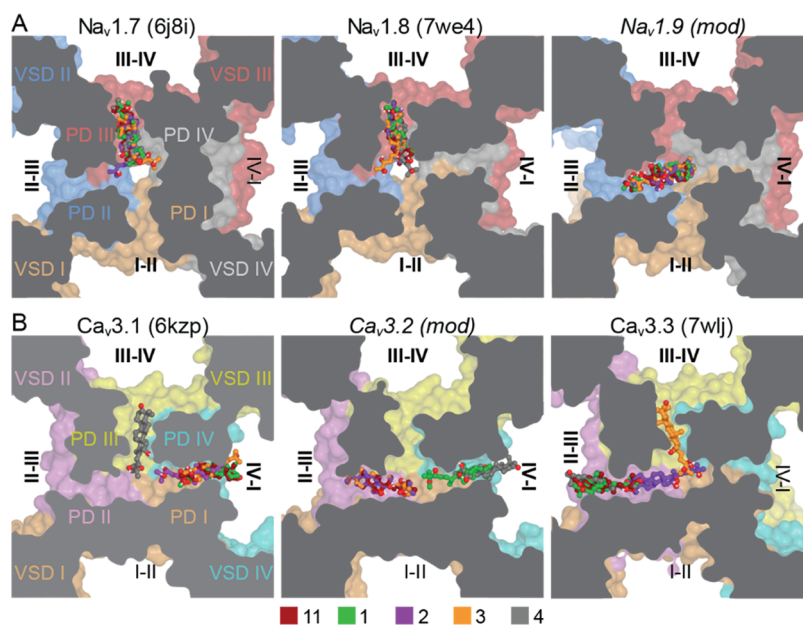
excitability depends on part of the activity of VGSCs and VGCCs, these results strongly suggest that like argentatin C (11), its analogues 1 and 4 inhibit these channels.<sup>10</sup> However, behavioral experiments in animal models with acute or chronic pain will be necessary to determine if these compounds could alleviate pain.

As we did previously for argentatin C (11),<sup>10</sup> we used *in silico* docking to predict the binding sites for its analogues 1–4 to pain-relevant VGSCs and VGCCs. Our previous analysis

identified the hydrophobic fenestrations and central pore as significantly more likely to be the sites of binding for 11,<sup>10</sup> so the analogues were docked at these locations as well. In the VGSCs, the five compounds are predicted to dock to the pore and a single fenestration: fenestration III–IV in NaV1.7 and NaV1.8 and fenestration II–III in NaV1.9 (Figure 6A). For the VGCC targets, the predictions were split between the pore and two fenestrations in each case (Figure 6B). These results confirm that the argentatin C analogues are able to bind to the VGSC and VGCC fenestrations. These hydrophobic access channels were predicted over 40 years ago by Hille as important antinociceptive drug access pathways.<sup>43</sup> Further, an increasing number of compounds have been found to utilize one or more fenestrations to anchor compounds within the pore of both VGSC (e.g., flecainide,<sup>44</sup> bulleyaconitine A,<sup>45</sup> XEN907 and TC-N1752,<sup>46</sup> and A-803467<sup>47</sup>) and VGCC (e.g., Z944,<sup>48</sup> cinnazarine,<sup>49</sup> PD173212,<sup>50</sup> mibefradil, otilonium bromide, and pimozone<sup>51</sup>) channels. Notably, the natural product bulleyaconitine A,<sup>52</sup> as well as TC-N1752,<sup>53</sup> A-803467,<sup>54</sup> Z994,<sup>55</sup> and CaV2.2 blocker 1<sup>56</sup> have been shown to have analgesic properties, supporting the fenestrations as sites that can be targeted for the development of more effective pain therapeutics. Unfortunately, the moderate resolution of the currently available cryoEM structures and inherent uncertainties of homology modeling lead to ambiguities in amino acid side-chain placement which likely has led to the highly similar docking scores (Table S3) and predicted contacts, precluding further interpretation of our docking results for argentatin C (11) and its analogues 1–4.

## EXPERIMENTAL SECTION

**General Experimental Procedures.** Optical rotations were measured in CH<sub>3</sub>OH with a JASCO Dip-370 digital polarimeter. One-dimensional (1D) and two-dimensional (2D) NMR spectra were recorded in CDCl<sub>3</sub> and pyridine-*d*<sub>5</sub>



**Figure 6.** Predicted binding sites of argentatin C (11) and analogues 1–4 in pain-relevant VGSCs and VGCCs. Docking poses for (A) VGSCs and (B) VGCCs. Top docked poses are shown as sticks with carbon atoms colored as labeled in legend, oxygen in red, and polar hydrogen in white. Channels are shown in surface representation, sliced at the level of the fenestrations (viewed from the extracellular side) and colored by domain as indicated (PD, pore domain; VSD, voltage-sensing domain). Fenestrations are labeled with the Roman numerals of the adjacent pore domains with bold text denoting open fenestrations.

with a Bruker Avance III 400 spectrometer at 400 MHz for  $^1\text{H}$  NMR and 100 MHz for  $^{13}\text{C}$  NMR using residual solvent resonances ( $\text{CDCl}_3$ :  $^1\text{H}$  NMR, 7.24;  $^{13}\text{C}$  NMR, 77.0; pyridine- $d_5$ :  $^1\text{H}$  NMR, 8.74, 7.58, 7.22;  $^{13}\text{C}$  NMR, 150.4, 135.9, 123.9) as internal references. The  $^1\text{H}$  NMR spectra of Mosher's esters were recorded in pyridine- $d_5$  with a Bruker DRX-500 spectrometer at 500 MHz. The chemical shift ( $\delta$ ) values are given in parts per million (ppm), and the coupling constants ( $J$  values) are in hertz. High-resolution mass spectra (HRMS) were recorded on a JEOL HX110A spectrometer. Circular dichroism (CD) spectra were obtained using a Jasco J-810 spectropolarimeter. Gel permeation chromatography was carried out using a Sephadex LH-20 (GE-Healthcare Bio-Sciences AB, Uppsala, Sweden). Normal-phase (NP) column chromatography was performed using Baker silica gel 40  $\mu\text{m}$  flash chromatography packing (J. T. Baker), and reversed-phase (RP) chromatography was carried out using BAKER-BOND  $\text{C}_{18}$  40  $\mu\text{m}$  preparative LC packing (J. T. Baker). Analytical thin-layer chromatography (TLC) was performed on aluminum-backed precoated 0.25 mm silica gel 60  $\text{F}_{254}$  plates (Merck, Darmstadt, Germany), and spots were visualized under UV light and spraying with a solution of anisaldehyde in  $\text{H}_2\text{SO}_4/\text{HOAc}$  followed by heating. HPLC purifications were carried out on a 10  $\times$  250 mm Phenomenex Luna 5  $\mu\text{m}$   $\text{C}_{18}$  column (3 mL/min flow rate) with a Waters Delta Prep system consisting of a PDA 996 detector.

**Plant Material.** Aerial parts of *P. incanum* were collected by Dr. Steven P. McLaughlin of the University of Arizona (collection no. 7043) on Sept. 14, 1996, in Pima County (Old Highway, 83 SE of junction with Interstate I-10; 31 59.67' N, 110 40.35' W, elevation ca. 4,000 feet). A voucher sample has been deposited at the University of Arizona Herbarium (Catalog No. ARIZ 336215).

**Extraction of *P. incanum*.** Freshly collected aerial parts of *P. incanum* were air-dried and powdered, and a portion (638.0 g) of this material was extracted with  $\text{CHCl}_3$ - $\text{CH}_3\text{OH}$  (1:1) at room temperature for 24 h. After filtration, the resulting filtrate was concentrated below 40  $^\circ\text{C}$  and under reduced pressure to afford the crude extract (137.0 g), which was stored in a sealed bottle at  $-5$   $^\circ\text{C}$  until use.

**Isolation and Identification of Metabolites.** A portion (10.0 g) of the above extract was partitioned between hexanes and 80% aq  $\text{CH}_3\text{OH}$ . The 80% aq  $\text{CH}_3\text{OH}$  layer was then diluted to 50% aq  $\text{CH}_3\text{OH}$  with water and extracted with  $\text{CHCl}_3$ . The resulting solutions were evaporated separately under reduced pressure to afford hexanes (4.43 g),  $\text{CHCl}_3$  (2.64 g), and 50% aq  $\text{CH}_3\text{OH}$  (2.29 g) fractions. A portion (4.3 g) of the hexanes fraction was subjected to gel permeation chromatography on a column of Sephadex LH-20 (200.0 g) made up in hexanes/ $\text{CH}_2\text{Cl}_2$  (1:4) and eluted with hexanes/ $\text{CH}_2\text{Cl}_2$  (1:4) followed by  $\text{CH}_3\text{OH}$ . Twenty-four fractions were collected, and fractions having similar TLC profiles were combined to give six major fractions [A (0.44 g), B (1.69 g), C (1.47 g), D (0.16 g), E (0.19 g), and F (0.65 g)]. Fraction B (1.60 g) was chromatographed over a column of silica gel (30.0 g) made up in hexanes and eluted with 200.0 mL each of hexanes, hexanes/ $\text{CHCl}_3$  (2:1),  $\text{CHCl}_3$  followed by  $\text{CHCl}_3/\text{CH}_3\text{OH}$  (95:5). Sixteen subfractions were collected [B1 (15.7 mg), B2 (15.1 mg), B3 (1.6 mg), B4 (2.6 mg), B5 (4.5 mg), B6 (4.0 mg), B7 (3.6 mg), B8 (11.2 mg), B9 (86.0 mg), B10 (101.1 mg), B11 (67.7 mg), B12 (57.0 mg), B13 (178.2 mg), B14 (58.1 mg), B15 (930.8 mg), and B16 (35.6 mg)]. The major fraction B15 (930.0 mg) containing triterpenoids was re-

chromatographed over a column of silica gel (30.0 g) made up in  $\text{CHCl}_3$  and eluted with  $\text{CHCl}_3$ ,  $\text{CHCl}_3/\text{CH}_3\text{OH}$  (99.5:0.5),  $\text{CHCl}_3/\text{CH}_3\text{OH}$  (99:1), followed by  $\text{CHCl}_3/\text{CH}_3\text{OH}$  (95:5) to give nine subfractions (B15<sub>A</sub>–B15<sub>I</sub>). Subfraction B15<sub>B</sub> (467.7 mg) was further separated by RP  $\text{C}_{18}$  (18.0 g) column chromatography using  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (90:10) followed by  $\text{CH}_3\text{OH}$  as eluant to give six subfractions (B15<sub>B1</sub>–B15<sub>B6</sub>). Subfraction B15<sub>B2</sub> (53.6 mg) on purification by reversed-phase high-performance liquid chromatography (RP-HPLC) afforded argentatin B [**10**, 31.1 mg,  $t_{\text{R}}$  33.0 min,  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (98:2)]. RP-HPLC purification of the subfraction B15<sub>B3</sub> (64.9 mg) using  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (98:2) as the mobile phase yielded 16-dehydroargentatin A (**9**, 2.2 mg,  $t_{\text{R}}$  32.0 min), 16-deoxyisoargentatin A (**8**, 3.1 mg,  $t_{\text{R}}$  37.5 min), 16-deoxyargentatin A (**7**, 8.2 mg,  $t_{\text{R}}$  44.0 min), and an additional amount of argentatin B (**10**, 16.8 mg,  $t_{\text{R}}$  35.0 min). RP-HPLC purification of the combined subfractions B15<sub>D</sub> (29.4 mg) and B15<sub>E</sub> (31.3 mg) afforded an additional amount of 16-dehydroargentatin A (**7**, 12.6 mg). Purification of the subfraction B15<sub>F</sub> (53.3 mg) by RP-HPLC followed by silica gel column chromatography gave 16-dehydroargentatin C (**12**, 2.2 mg), 25-dehydroxy-25-methoxyargentatin C (**1**, 1.7 mg), 20S-hydroxyargentatin C (**2**, 1.4 mg), and 20S-hydroxyisoargentatin C (**3**, 1.5 mg). Further purification of the subfraction B15<sub>G</sub> (24.1 mg) by RP-HPLC yielded argentatin A (**5**, 1.4 mg) and isoargentatin A (**6**, 0.8 mg).

The  $\text{CHCl}_3$  fraction (2.5 g) obtained above was subjected to size-exclusion chromatography over Sephadex LH-20 (300.0 g) made up in hexanes/ $\text{CH}_2\text{Cl}_2$  (1:4) and eluted with 1.0 L each of hexanes/ $\text{CH}_2\text{Cl}_2$  (1:4),  $\text{CH}_2\text{Cl}_2$ /acetone (4:1),  $\text{CH}_2\text{Cl}_2$ /acetone (2:3) followed by  $\text{CH}_3\text{OH}$ . Ninety-three fractions were collected, and fractions having similar TLC profiles were combined to give nine major fractions [C1 (455.8 mg), C2 (213.2 mg), C3 (149.7 mg), C4 (168.4 mg), C5 (378.0 mg), C6 (278.9 mg), C7 (243.6 mg), C8 (202.5 mg), and C9 (439.6 mg)]. Fraction C2 (213 mg) on further purification by RP  $\text{C}_{18}$  (20.0 g) column chromatography using combinations of  $\text{CH}_3\text{OH}$  and  $\text{H}_2\text{O}$  gave argentatin C (**11**, 19.8 mg). Fraction C3 (145 mg) was re-chromatographed over a column of RP  $\text{C}_{18}$  (18.0 g) made up in  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (80:20) and eluted with  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (80:20),  $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (80:20), and  $\text{CH}_3\text{OH}$ . Twenty-four fractions were collected and combined according to their TLC profile to give 10 subfractions [(C3A (11.0 mg), C3B (8.8 mg), C3C (6.1 mg), C3D (12.5 mg), C3E (5.8 mg), C3F (12.1 mg), C3G (25.5 mg), C3H (28.7 mg), C3I (9.4 mg), and C3J (18.4 mg)]. Purification of the subfraction C3H (28.7 mg) by RP- $\text{C}_{18}$  HPLC [ $\text{CH}_3\text{OH}/\text{H}_2\text{O}$  (92.5:7.5, v/v, 3 mL/min)] gave 24-*epi*-argentatin C (**4**, 15.2 mg,  $t_{\text{R}}$  = 23.5 min) as a white amorphous solid. The subfraction C3J was found to consist of an inseparable mixture of quisquagenin (**13**) and isoquisquagenin (**14**) in a ratio of 1:2 by careful analyses of its  $^1\text{H}$  and  $^{13}\text{C}$  NMR data (Figure S19).<sup>35</sup>

**25-Dehydroxy-25-methoxyargentatin C [(16 $\beta$ ,24R)-16,24-dihydroxy-25-methoxycycloartan-3-one] (**1**).** Amorphous white solid:  $[\alpha]_{\text{D}}^{25} +39$  ( $c$  0.16,  $\text{CH}_3\text{OH}$ ); ECD ( $\text{CH}_3\text{OH}$ )  $[\theta] -23,282$  (300 nm);  $^1\text{H}$  NMR data; see Table 1;  $^{13}\text{C}$  NMR data; see Table 2; positive HRESIMS  $m/z$  511.3761 [ $\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{31}\text{H}_{52}\text{O}_4\text{Na}$  511.3763).

**20S-Hydroxyargentatin C [(16 $\beta$ ,20S,24R)-16,20,24,25-tetrahydroxycycloartan-3-one] (**2**).** Amorphous white solid:  $[\alpha]_{\text{D}}^{25} +51$  ( $c$  0.14,  $\text{CH}_3\text{OH}$ ); ECD ( $\text{CH}_3\text{OH}$ )  $[\theta] 16,378$  (205 nm);  $^1\text{H}$  NMR data; see Table 1;  $^{13}\text{C}$  NMR data; see



Table 2; positive HRESIMS  $m/z$  473.3601  $[\text{MH}-\text{H}_2\text{O}]^+$  (calcd for  $\text{C}_{30}\text{H}_{49}\text{O}_4$  473.3631).

**20S-Hydroxyisoargentatin C [(16 $\beta$ ,20S,24R)-16,20,24,25-tetrahydroxylanost-8-en-3-one] (3).** Amorphous white solid:  $[\alpha]_{\text{D}}^{25} +17$  ( $c$  0.14,  $\text{CH}_3\text{OH}$ ); ECD ( $\text{CH}_3\text{OH}$ )  $[\theta]$   $-2217$  (296 nm);  $^1\text{H}$  NMR data; see Table 1;  $^{13}\text{C}$  NMR data; see Table 2; positive HRESIMS  $m/z$  513.3565  $[\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{30}\text{H}_{50}\text{O}_5\text{Na}$  513.3556).

**24-Epi-argentatin C [(16 $\beta$ ,24S)-16,24,25-trihydroxycycloartan-3-one] (4).** Amorphous white solid:  $[\alpha]_{\text{D}}^{25} +18$  ( $c$  0.12,  $\text{CH}_3\text{OH}$ ); ECD ( $\text{CH}_3\text{OH}$ )  $[\theta]$  1449 (223 nm),  $-527$  (300 nm);  $^1\text{H}$  NMR data; see Table 1;  $^{13}\text{C}$  NMR data; see Table 2; positive HRESIMS  $m/z$  497.3604  $[\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{30}\text{H}_{50}\text{O}_4\text{Na}$  497.3607).

### Preparation of (S)- and (R)-MTP Ester Derivatives of 2, and 4 by Advanced Mosher's Ester Procedure.

Each compound (1.5 mg) was transferred into a clean NMR tube and was dried completely *in vacuo*. Pyridine- $d_5$  (1.0 mL) and (R)-(-)- $\alpha$ -methoxy- $\alpha$ -trifluoromethylphenylacetyl chloride (5.0  $\mu\text{L}$ ) were added into the NMR tube immediately under a stream of  $\text{N}_2$ , and then the NMR tube was shaken carefully to mix the sample and the MTPA chloride evenly. The reaction in the NMR tube was monitored by  $^1\text{H}$  NMR and left at 25  $^\circ\text{C}$  for 2 h to give the (S)-MTPA derivatives (2a and 4a). Additional portions of 2 and 4 (1.5 mg each) in pyridine- $d_5$  (1.0 mL) were reacted in a second NMR tube with (S)-(+)- $\alpha$ -methoxy- $\alpha$ -trifluoromethylphenylacetyl chloride (5.0  $\mu\text{L}$ ) at 25  $^\circ\text{C}$  for 2 h to afford the (R)-MTPA derivatives (2b and 4b).  $^1\text{H}$  NMR data of 2a (500 MHz, pyridine- $d_5$ ):  $\delta$  5.138 (1H, brt,  $J = 3.1$  Hz, H-24), 2.315 (1H, dd,  $J = 14.8, 2.8$  Hz, Ha-23), 1.915 (1H, dd,  $J = 14.8, 3.2$  Hz, Hb-23), 1.616 (3H, s, H<sub>3</sub>-21), 1.430 (3H, s, H<sub>3</sub>-26), 1.308 (3H, s, H<sub>3</sub>-27), 1.298 (1H, m, Ha-22), 1.021 (1H, m, Hb-22).  $^1\text{H}$  NMR data of 2b (500 MHz, pyridine- $d_5$ ):  $\delta$  5.088 (1H, brt,  $J = 3.1$ , H-24), 2.192 (1H, dd,  $J = 14.6, 3.0$  Hz, Ha-23), 1.837 (1H, dd,  $J = 14.6, 2.9$  Hz, Hb-23), 1.595 (3H, s, H<sub>3</sub>-21), 1.449 (3H, s, H<sub>3</sub>-26), 1.337 (1H, s, H<sub>3</sub>-27), 1.286 (1H, m, Ha-22), 0.998 (1H, m, Hb-22).  $^1\text{H}$  NMR data of 4a (500 MHz, pyridine- $d_5$ ):  $\delta$  5.635 (1H, dd,  $J = 9.9, 2.2$  Hz, H-24), 2.183 (1H, m, Ha-23), 1.963 (1H, m, H-17), 1.949 (1H, m, H-20), 1.905 (Hb-23), 1.727 (1H, m, Ha-22), 1.666 (3H, s, H<sub>3</sub>-26), 1.483 (3H, s, H<sub>3</sub>-18), 1.445 (3H, s, H<sub>3</sub>-27), 1.437 (1H, m, Hb-22), 0.959 (3H, d,  $J = 6.2$  Hz, H<sub>3</sub>-21), 0.951 (3H, s, H<sub>3</sub>-30).  $^1\text{H}$  NMR data of 4b (500 MHz, pyridine- $d_5$ ):  $\delta$  5.597 (1H, d,  $J = 9.2$  Hz, H-24), 2.427 (1H, m, Ha-23), 2.353 (1H, m, Ha-22), 1.997 (1H, m, Hb-23), 1.734 (1H, m, Ha-22), 1.462 (3H, s, H<sub>3</sub>-26), 1.441 (1H, m, Hb-22), 1.378 (3H, s, H<sub>3</sub>-27), 1.126 (3H, d,  $J = 6.2$  Hz, H<sub>3</sub>-21).

**Animals.** Pathogen-free adult female Sprague–Dawley rats (100–200 g, Charles River Laboratories, Wilmington, MA) were kept in light (12 h light, 12 h dark cycle) and temperature (23  $\pm$  3  $^\circ\text{C}$ ) controlled rooms. Water and standard rodent diet were available *ad libitum*. All animal use was conducted in accordance with the National Institutes of Health guidelines, and the study was conducted in strict accordance with recommendations in the Guide for the Care and Use of Laboratory Animals of the New York University (Protocol No.: PROTO202100104). All animals were housed and bred at the Kriser Dental Center Animal Facility of the University of New York. Efforts were made to minimize animal suffering.

**Electrophysiology.** As previously described, dorsal root ganglia (DRG) were collected and dissociated from female Sprague–Dawley rats, aged 4–6 weeks old.<sup>10</sup> Compounds were applied for 15 min before and during the recording at 20

$\mu\text{M}$  using 0.1% DMSO as a vehicle control. The excitability was tested in response to current injection between 0 and 100 pA with increments of 10 pA in the presence of 0.1% DMSO (control), or 20  $\mu\text{M}$  of 11, 1, 2, 3, or 4. The external solution was composed of NaCl (154 mM), KCl (5.6 mM),  $\text{CaCl}_2$  (2 mM),  $\text{MgCl}_2$  (1 mM) D-glucose (10 mM), HEPES (8 mM), pH 7.4 and 319 milliOsmol/L. The internal solution consisted of KCl (137 mM), NaCl (10 mM),  $\text{MgCl}_2$  (1 mM), EGTA (1 mM), HEPES (10 mM), pH = 7.3 and 305 milliOsmol/L.

**Homology Modeling and Ligand Docking.** As previously reported,<sup>10</sup> the Baker Lab's structure prediction server<sup>57</sup> was used to generate homology models for the  $\alpha$  subunit of human NaV1.9 (UniProt ID Q9UI33-1) based on the structure of NaV1.7 with TTX bound (PDB 6j8i<sup>58</sup>) and for human CaV3.2 (UniProt ID O95180-1) based on the structure of CaV3.1 with Z944 bound (PDB 6kzp<sup>48</sup>). Briefly, 10 models were sampled for each target. Models with the lowest error estimates and the same number of open fenestrations as the template were selected for docking and trimmed to remove unreliably modeled intracellular loops prior to docking. Protein and ligand preparation and docking were conducted using Schrödinger Docking Suite tools.<sup>59</sup> Argentatin C (11) and its analogues 1–4, and Z944 were prepared from isomeric SMILES strings using LigPrep<sup>59</sup> with possible ionization states at pH 7.0. The cryoEM structure of human NaV1.7 (PDB 6j8i<sup>58</sup>), NaV1.8 (PDB 7we4<sup>47</sup>), CaV3.1 (PDB 6kzp<sup>48</sup>), CaV3.3 (PDB 7wLj<sup>51</sup>), and the trimmed homology models of NaV1.9 and CaV3.2 were prepared using the Protein Preparation Wizard,<sup>60</sup> and the charges on K1462 (canonical K1865) in PDB 6kzp, K1503 in the CaV3.2 model, and K1379 in PDB 7wLj were manually adjusted from +1 to 0 prior to docking. Our previous work identified the central pore and fenestrations as the most likely binding site for 11,<sup>10</sup> so we selected the same grid location for docking its analogues 1–4 with an enclosing box at the centroid of residues N388, F956, V1505, and V1820, and expanded the ligand-midpoint box to 32x32x32  $\text{Å}^3$  to completely enclose the fenestrations. Docking was performed with Glide Standard Precision (SP) mode<sup>61</sup> with enhanced sampling to obtain up to 10 poses at each site. We previously showed that redocking of Z944 to the prepared PDB 6kzp reproduced the bound pose (not shown) with a docking score of  $-6.6$  kcal/mol.<sup>10</sup> Docking scores for 11 and its analogues 1–4 were generally not significantly different at each target, precluding interpretation (average of top scores in kcal/mol:  $-7.3 \pm 0.1$  for NaV1.7,  $-6.9 \pm 0.2$  for NaV1.8,  $-6.6 \pm 0.2$  for NaV1.9,  $-6.5 \pm 0.6$  for CaV3.1,  $-7.1 \pm 0.8$  for CaV3.2,  $-5.2 \pm 0.3$  for CaV3.3). The top pose at each site was selected for visualization. Figure 6A,B was generated with PyMOL.<sup>62</sup>

## ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.3c02302>.

$^1\text{H}$ ,  $^{13}\text{C}$ , and 2D NMR, ECD, and HRESIMS spectra of 1–4; statistical summary of excitability of rat DRG neurons toward 11 and 1–4; and resting membrane potential and docking scores of 11 and 1–4 (PDF)



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### Notes

The authors declare the following competing financial interest(s): RK is the co-founder of Regulonix LLC, a company developing non-opioid drugs for chronic pain and ElutheriaTx Inc., a company developing gene therapy approaches for chronic pain. All other authors declare no competing financial interests.

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