

OPEN

# Dynamical Quantum Filtering via Enhanced Scattering of *para*-H<sub>2</sub> on the Orientationally Anisotropic Potential of SrTiO<sub>3</sub>(001)

Koji Shimizu<sup>1</sup>, Wilson Agerico Diño<sup>1,2\*</sup>, Hiroshi Nakanishi<sup>1,3</sup>, Hideaki Kasai<sup>1,3,4</sup>, Kotaro Takeyasu<sup>4</sup>, Katsuyuki Fukutani<sup>4</sup> & Ayako Yajima<sup>5</sup>

Quantum dynamics calculation, performed on top of density functional theory (DFT)-based total energy calculations, show dynamical quantum filtering via enhanced scattering of *para*-H<sub>2</sub> on SrTiO<sub>3</sub>(001). We attribute this to the strongly orientation-dependent (electrostatic) interaction potential between the H<sub>2</sub> (induced) quadrupole moment and the surface electric field gradient of ionic SrTiO<sub>3</sub>(001). These results suggest that ionic surfaces could function as a scattering/filtering media to realize rotationally state-resolved H<sub>2</sub>. This could find significant applications not only in H<sub>2</sub> storage and transport, but also in realizing materials with pre-determined characteristic properties.

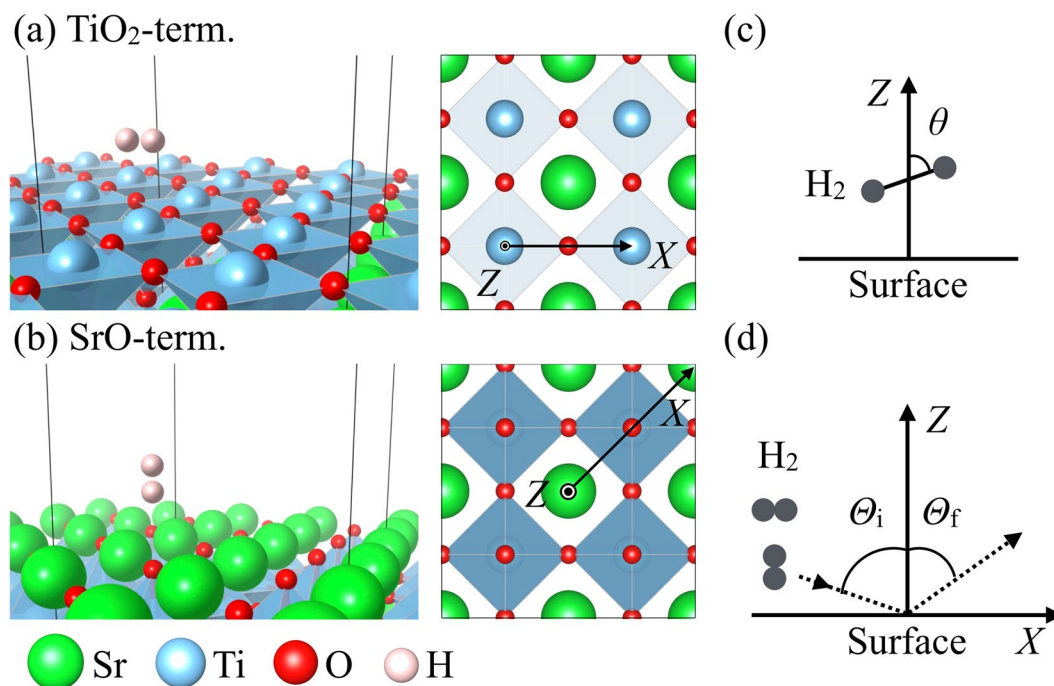
The behavior/dynamics of H<sub>2</sub> on surfaces strongly depend on the H<sub>2</sub> molecular orientation/rotational states (rotational quantum number  $J$ , magnetic quantum number  $m$ ) (cf., e.g., ref. <sup>1</sup> and references therein). On transition metal surfaces such as copper (Cu) and palladium (Pd), H<sub>2</sub> exhibits rotational alignment (cf., e.g., refs. <sup>1,2</sup> and references therein). One could then tune or design the structure of the metal (alloy) surfaces to dynamically filter the quantum rotational states of desorbing or scattered H<sub>2</sub> (dynamical quantum filtering), and control the H<sub>2</sub> dynamics (cf., e.g., refs. <sup>1-7</sup> and references therein). The resulting H<sub>2</sub> dynamics could, in turn, be used to probe local surface reactivity<sup>8,9</sup> (e.g., via the H<sub>2</sub>(D<sub>2</sub>) diffraction spectra<sup>10-12</sup>). On ionic crystal surfaces, the H<sub>2</sub> quadrupole moment interacts with the surface local electric field gradient to couple the translational and rotational degrees-of-freedom<sup>13</sup>. On SrTiO<sub>3</sub>(001) (STO(001)), an ionic crystal material<sup>14-21</sup> with tunable surface terminations (cf., e.g., refs. <sup>22,23</sup>), H<sub>2</sub> adsorbs with the H-H bond oriented parallel (polar coordinate  $\theta = \pi/2$ ) to the TiO<sub>2</sub>-terminated surface at the Ti-site, and perpendicular ( $\theta = 0$ ) to the SrO-terminated surface on top of the O-site (cf., Fig. 1)<sup>24</sup>. The strong orientationally anisotropic potential ( $\Delta E_{\text{anisotropy}}^{\theta} = |E_{\theta=0} - E_{\theta=\pi/2}|$ ) results in adsorbed H<sub>2</sub> with (hindered) rotational states ( $J, m$ ) different from that of gas phase H<sub>2</sub>. These strongly hindered adsorption states lead to ( $J, m$ )-dependent thermal desorption energies<sup>24-30</sup>, suggesting the possibility of separating *para*-H<sub>2</sub> [ $p$ -H<sub>2</sub> ( $J = 0, m = 0$ )] and *ortho*-H<sub>2</sub> [ $o$ -H<sub>2</sub> ( $J = 1, m = \pm 1$ )] through an adsorption-desorption process. This could find significant applications not only in H<sub>2</sub> storage and transport applications, but also in realizing materials with pre-determined characteristic properties.

As with metal (alloy) surfaces<sup>5,8,9</sup>, the H<sub>2</sub> dynamics would be susceptible to the positive and negative charges that corrugate ionic crystal surfaces. In the following, we will show that on STO(001), under the influence of the orientationally anisotropic potential, on top of the surface lateral corrugation,  $p$ -H<sub>2</sub> scatter strongly at specific angles from the TiO<sub>2</sub>-terminated and SrO-terminated STO(001). This dynamical filtering/scattering selectivity allows for more economical (less heat consumption) and more efficient means to rotationally separate  $o$ -H<sub>2</sub> and  $p$ -H<sub>2</sub>, than the usual adsorption-desorption process<sup>1-5,24-30</sup>.

## Results and Discussion

**H<sub>2</sub>/SrTiO<sub>3</sub>(001) System.** Figure 1 shows a H<sub>2</sub> interacting with STO(001).  $X$  gives the surface lateral coordinate of the H<sub>2</sub> center-of-mass (CM) along the most corrugated directions on the two STO(001) terminations, viz., along [100] for TiO<sub>2</sub>-termination (Ti-O-Ti row) and along [110] for SrO-termination (Sr-O-Sr row), respectively

<sup>1</sup>Department of Applied Physics, Osaka University, Osaka, 565-0871, Japan. <sup>2</sup>Center for Atomic and Molecular Technologies, Osaka University, Osaka, 565-0871, Japan. <sup>3</sup>National Institute of Technology, Akashi College, Hyogo, 674-8501, Japan. <sup>4</sup>Institute of Industrial Science, The University of Tokyo, Tokyo, 153-8505, Japan. <sup>5</sup>Kawasaki Heavy Industries, Ltd., Hyogo, 673-8666, Japan. \*email: [wilson@dyn.ap.eng.osaka-u.ac.jp](mailto:wilson@dyn.ap.eng.osaka-u.ac.jp)

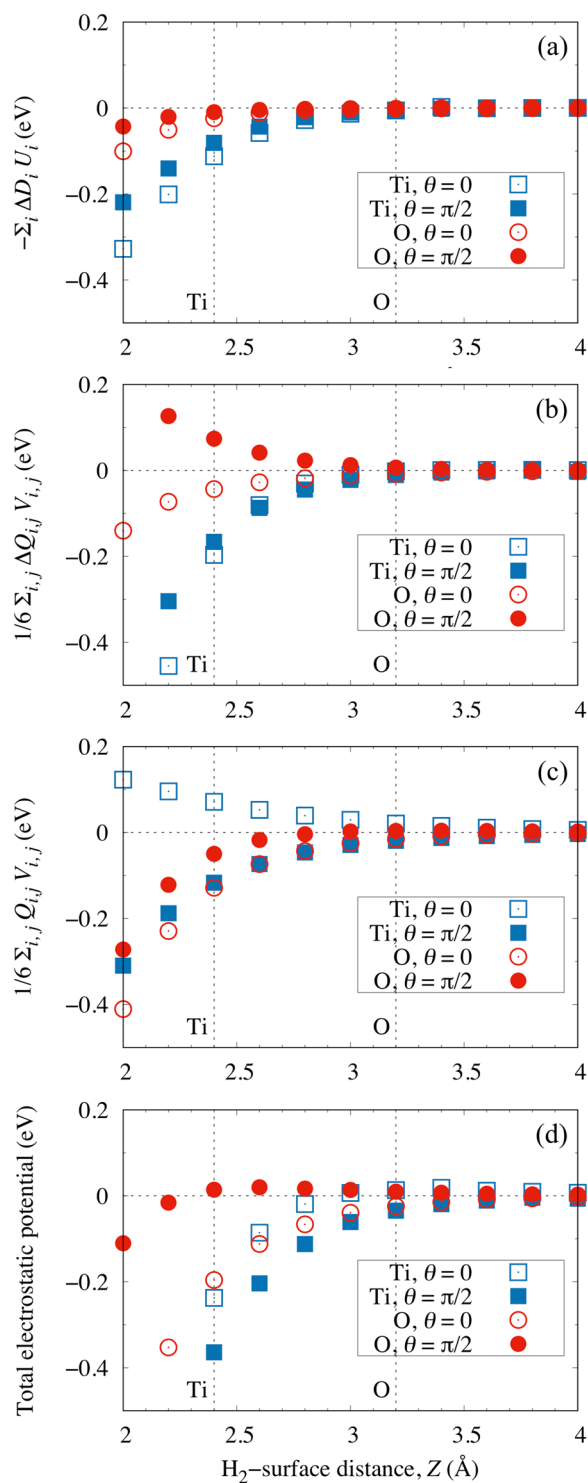


**Figure 1.** A depiction of  $\text{H}_2$  adsorption (a) atop the Ti-site and (b) atop the O-site on the  $\text{TiO}_2$ - and SrO-terminated  $\text{SrTiO}_3(001)$ , with corresponding preferential orientations  $\theta = \pi/2$  and  $0$ , respectively. The lower left panel shows which colored balls correspond to which element. (c) A depiction of  $\text{H}_2$  with the  $\text{H}_2$  center-of-mass (CM) at a distance  $Z$  from the surface, and the H-H bond at an orientation  $\theta$  with respect to the surface normal. (d) A depiction of  $\text{H}_2$  scattering on  $\text{SrTiO}_3(001)$  with angle of incidence  $\Theta_i$  and scattering angle  $\Theta_f$ .  $X$  corresponds to the surface lateral position of the  $\text{H}_2$  CM along the (a)  $[100]$  and (b)  $[110]$  direction on the  $\text{TiO}_2$ - and SrO-terminated  $\text{SrTiO}_3(001)$ , respectively.

(cf., Fig. 1(a,b)).  $Z$  gives the normal distance of the  $\text{H}_2$  CM from the surface.  $\theta$  gives the polar angular orientation of the H-H bond with respect to the surface normal.  $\phi$  (not shown) gives the azimuthal angular orientation of the H-H bond about the surface normal, with respect to the  $X$ -axis, at each site on  $\text{STO}(001)$ .  $\Theta_{i(f)}$  gives  $\text{H}_2$  incidence (scattering) angle measured with respect to the surface normal (cf., Fig. 1(c,d)). In the following, given  $\text{H}_2(J_i, m_i, E_i, \Theta_i)$ , we determine the probability of finding  $\text{H}_2(J_f, m_f, E_f, \Theta_f)$ .  $\text{H}_2(J_i, m_i, E_i, \Theta_i)$  indicates a  $\text{H}_2$  with initial rotational state  $(J_i, m_i)$ , impinging  $\text{STO}(001)$  with an initial incident translational energy  $E_i$  and at an incidence angle  $\Theta_i$  with respect to the surface normal.  $\text{H}_2(J_f, m_f, E_f, \Theta_f)$  indicates a  $\text{H}_2$  with final rotational state  $(J_f, m_f)$ , scattered from  $\text{STO}(001)$  with a final translational energy  $E_f$  and at a scattering angle  $\Theta_f$  with respect to the surface normal.

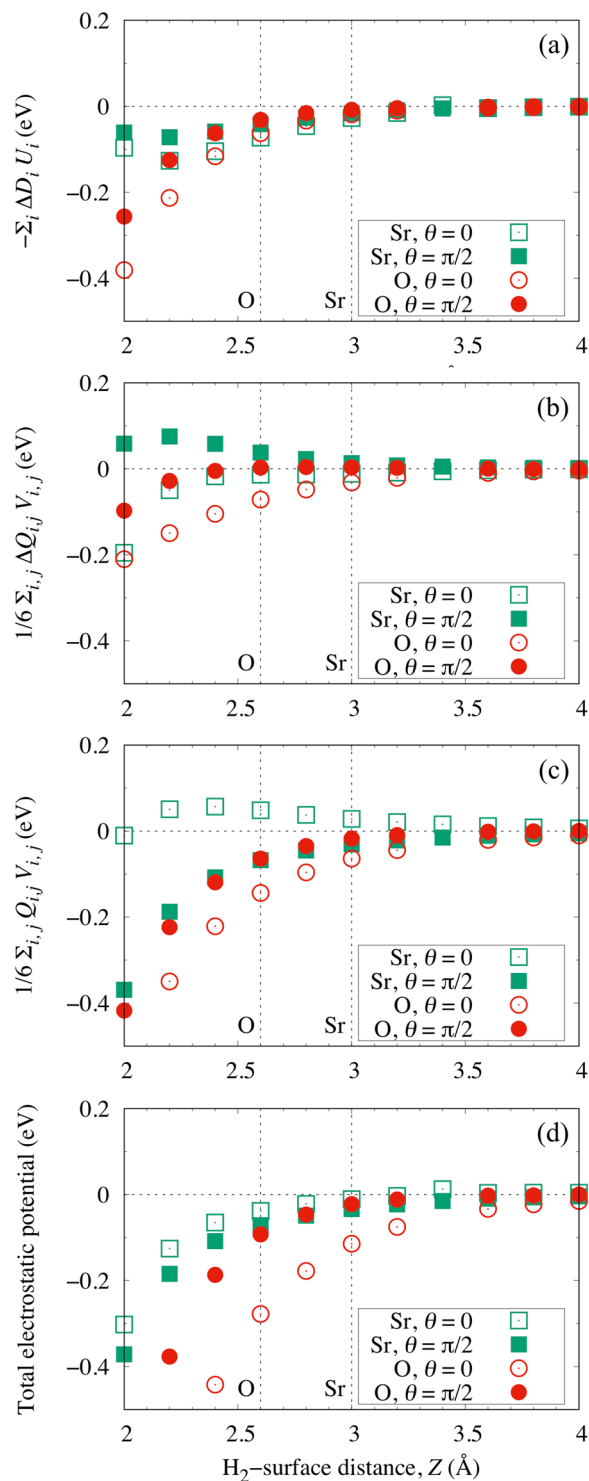
**Orientationally anisotropic electrostatic potential.** In Figs. 2 and 3, we plot the (electrostatic interaction energies) dot products of the (induced) dipole moment with the surface electric field  $U_i$  and the (induced) quadrupole moment with the gradient of the surface electric field  $V_{ij}$ , for the  $\text{TiO}_2$ -terminated and SrO-terminated surfaces, respectively. We can see that the orientational ( $\theta$ ) anisotropy  $\Delta E_{\text{anisotropy}}^\theta$  becomes important when the impinging  $\text{H}_2$  comes sufficiently near the surface, viz., at  $Z \leq 2.4 \text{ \AA}$  above the Ti-site and  $Z \leq 2.6 \text{ \AA}$  above the O-site on the  $\text{TiO}_2$ - and SrO-terminated surfaces, respectively. On the other hand, far from the surface, viz., at  $Z \geq 3.2 \text{ \AA}$  above the O-site and  $Z \geq 3.0 \text{ \AA}$  above the Sr-site of the  $\text{TiO}_2$ - and SrO-terminated surfaces, respectively, only a small  $\Delta E_{\text{anisotropy}}^\theta$  can be observed. This orientational anisotropy  $\Delta E_{\text{anisotropy}}^\theta$ , on top of the surface lateral corrugation, would prove to be useful in our attempt to control the  $\text{H}_2$  scattering dynamics, as we will discuss in detail in the next sections.

**$\text{H}_2$  Scattering along  $\text{STO}(001)[100]$  on  $\text{TiO}_2$ -terminated  $\text{STO}(001)$ .** In Fig. 4(a-c), we show the calculated (scattering) probabilities/spectra  $P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i = 80 \text{ meV}, E_f, \Theta_i = 15.9^\circ, \Theta_f)$  of finding  $\text{H}_2(J_i, m_i, E_i = 80 \text{ meV}, \Theta_i = 15.9^\circ)$  scattered as  $\text{H}_2(J_f, m_f, E_f, \Theta_f)$  along the  $[100]$  direction of a  $\text{TiO}_2$ -terminated  $\text{STO}(001)$ . (The corresponding initial surface perpendicular translational energy  $E_i \times \cos^2 \Theta_i = 80 \text{ meV} \times \cos^2(15.9^\circ) \sim 74.0 \text{ meV}$  and initial surface lateral translational energy  $E_i \times \sin^2 \Theta_i = 80 \text{ meV} \times \sin^2(15.9^\circ) \sim 6.00 \text{ meV}$ ). This corresponds to the maximum  $p\text{-H}_2$  to  $o\text{-H}_2$  ratio  $R_{J_i=0/J_f=1}(E_i = 80 \text{ meV}, \Theta_i = 15.9^\circ, \Theta_f)$  on a  $\text{TiO}_2$ -terminated  $\text{STO}(001)$  (cf., Table 1). The resulting trends can be explained by inspecting the corresponding orientational anisotropy and surface lateral corrugation of the potential encountered by the impinging  $\text{H}_2(J_i, m_i, E_i = 80 \text{ meV}, \Theta_i = 15.9^\circ)$  (i.e.,  $\langle Y_f^m | V(Z, \theta, X) | Y_i^m \rangle$ , cf., Fig. 4(d)). Note that along  $\text{STO}(001)[100]$ ,  $\text{H}_2(J_i = 0, m_i = 0)$  encounters a strongly corrugated and orientationally anisotropic PES (cf., solid curve in Fig. 4(d)), that favors molecular



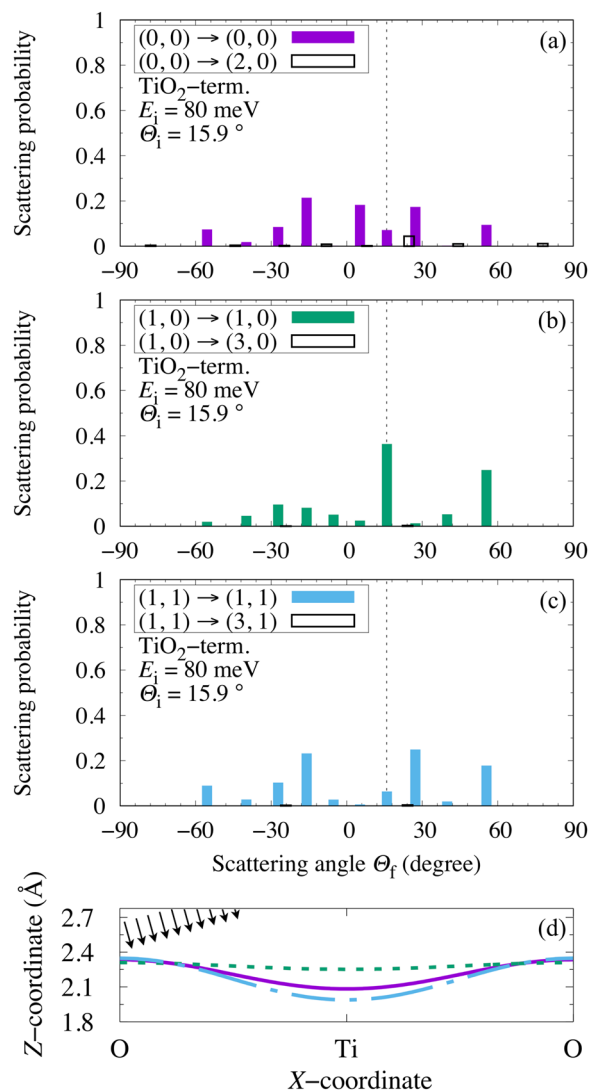
**Figure 2.** Calculated electrostatic potential contributions from (a) the induced dipole moment  $\Delta D_i$  with the surface electric field  $U_i$ , (b) the induced quadrupole moment  $\Delta Q_{ij}$  and (c) the H<sub>2</sub> quadrupole moment  $Q_{ij}$  with the gradient of the surface electric field  $V_{ij}$ , on the Ti-site (squares) and the O-site (circles) with  $\theta = 0$  (open symbols) and  $\theta = \pi/2$  (filled symbols) for the TiO<sub>2</sub>-terminated STO(001). (d) The sum of the contributions (a–c).  $i, j = (x, y, z)$ . The dotted vertical lines intercepting the abscissas indicate locations of the corresponding potential energy minima (H<sub>2</sub>-surface equilibrium distance) at each respective surface site.

adsorption atop the Ti-site, with H-H bond oriented parallel to the surface, and an adsorption energy  $E_{\text{ads}} = -191$  meV<sup>24</sup>. Atop the O-site, H<sub>2</sub> preferentially adsorbs with the H-H bond oriented perpendicular to the surface, with  $E_{\text{ads}} = -72.5$  meV<sup>24</sup>. As a result, in Fig. 4(a), we find a corresponding spectra dominated by off-specularly scattered H<sub>2</sub>, with dominant components coming from backscattering. Most of the H<sub>2</sub> ( $J_1 = 0, m_1 = 0$ ) would be



**Figure 3.** Calculated electrostatic potential contributions from (a) the induced dipole moment  $\Delta D_i$  with the surface electric field  $U_i$ , (b) the induced quadrupole moment  $\Delta Q_{i,j}$  and (c) the H<sub>2</sub> quadrupole moment  $Q_{i,j}$  with the gradient of the surface electric field  $V_{i,j}$  on the Sr-site (squares) and the O-site (circles) with  $\theta = 0$  (open symbols) and  $\theta = \pi/2$  (filled symbols) for the SrO-terminated STO(001). (d) The sum of the contributions (a–c).  $i, j = (x, y, z)$ . The dotted vertical lines intercepting the abscissas indicate the locations of the corresponding potential energy minima (H<sub>2</sub>-surface equilibrium distance) at each respective surface site.

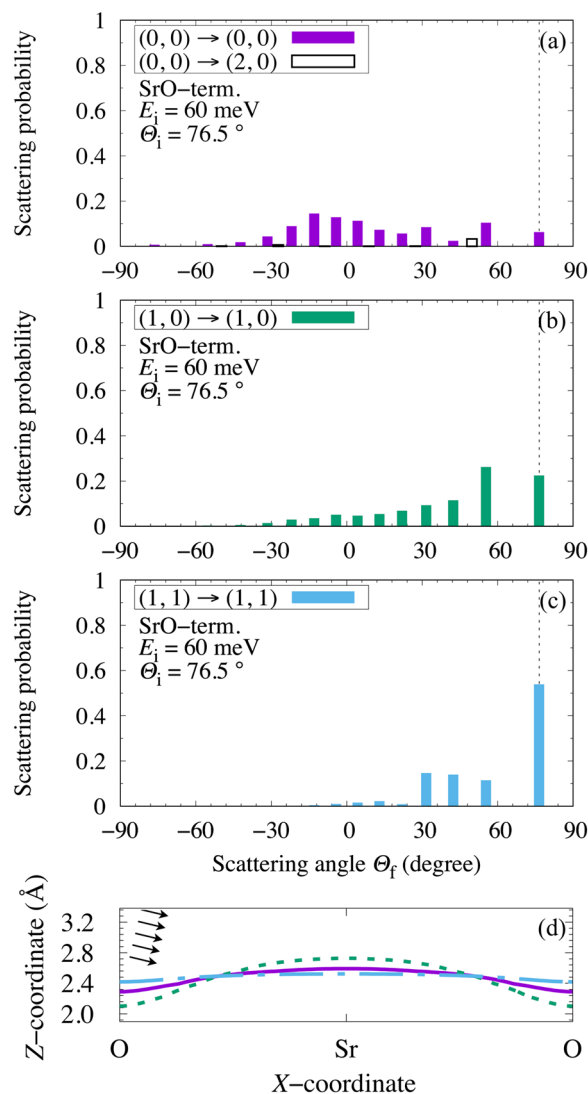
molecularly adsorbed (due to *reorientation/steering*<sup>31,32</sup>), and only a small fraction would be elastically scattered (due to *shadow effect*<sup>9</sup>). Considering the angle of incidence, the inelastically scattered H<sub>2</sub> would come from those hitting the repulsive part of the potential well near the O-site (*shadow effect*, cf., solid curve in Fig. 4(d)), resulting in a change in surface lateral momentum and off-specular scattering angles. Energy transfer from the



**Figure 4.** (a–c) Calculated  $\text{H}_2$  scattering probability  $(J_i, |m_i|) \rightarrow (J_f, |m_f|)$  from initial rotational state  $(J_i, |m_i|)$  to final rotational state  $(J_f, |m_f|)$  on the  $\text{TiO}_2$ -terminated  $\text{STO}(001)$  as a function of the scattering angle  $\Theta_f$  for incident energy  $E_i = 80$  meV and incident angle  $\Theta_i = 15.9^\circ$ . The dotted vertical lines intercepting the abscissas indicate specular scattering. (d) Solid, dotted, and dash-dot lines indicate constant energy surfaces of  $\langle Y_f^m | V(Z, \theta, X) | Y_i^m \rangle$  (with  $80 [\text{meV}] \times \cos^2 \Theta_i$ ; corresponding to  $\text{H}_2$  surface normal translational energy) encountered by a  $\text{H}_2$  impinging with initial rotational states  $(J_i, m_i)$ :  $[(0, 0), (1, 0), (1, 1)]$ , respectively. Same scales used for X- and Z-coordinates. The arrows indicate the direction of incidence of the impinging  $\text{H}_2$ .

$E_i$ [meV]	$\Theta_i$ [°]	$\Theta_f$ [°]	$R_{J=0/J=1}$	$\overline{R}_{J=0/J=1}$
10.0	15.0	15.0	1.45	0.470
20.0	47.1	0.00	1.91	0.155
30.0	48.4	48.4	1.14	0.397
40.0	40.4	-7.44	1.92	0.0967
50.0	20.3	6.65	2.65	0.449
60.0	18.5	31.9	1.70	0.0753
70.0	43.3	43.3	1.64	0.265
80.0	15.9	5.26	4.96	0.271

**Table 1.** Calculated  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  and  $\overline{R}_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  on  $\text{TiO}_2$ -terminated  $\text{STO}(001)$  at the incident energy range  $10 \text{ meV} \leq E_i \leq 80 \text{ meV}$ . For each incident energy, we list only the maximum  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  with the corresponding incident and scattering angles.



**Figure 5.** (a–c) Calculated H<sub>2</sub> scattering probability  $(J_i, |m_i|) \rightarrow (J_f, |m_f|)$  from initial rotational state  $(J_i, |m_i|)$  to final rotational state  $(J_f, |m_f|)$  on the SrO-terminated STO(001) as a function of the scattering angle  $\Theta_f$ , for incident energy  $E_i = 60$  meV and incident angle  $\Theta_i = 76.5^\circ$ . The dotted vertical lines intercepting the abscissas indicate specular scattering. (d) Solid, dotted, and dash-dot lines indicate constant energy surfaces of  $\langle Y_f^m | V(Z, \theta, X) | Y_i^m \rangle$  (with  $60 [\text{meV}] \times \cos^2 \Theta_i$ ; corresponding to H<sub>2</sub> surface normal translational energy) encountered by a H<sub>2</sub> impinging with initial rotational states  $(J_i, m_i)$ : [(0, 0), (1, 0), (1, 1)], respectively. Same scales used for X- and Z-coordinates. The arrows indicate the direction of incidence of the impinging H<sub>2</sub>.

translational degree-of-freedom (DOF) to the rotational DOF allows the molecule more time to sample the anisotropic surface through *reorientation/steering*. Those that succeed would molecularly adsorb. Those that fail, would be rotationally de-excited on the way back to the gas phase, because of the smaller anisotropic potential further (out in the vacuum) from the surface. Thus, the scattering spectra shows negligible rotationally excited H<sub>2</sub>, i.e., H<sub>2</sub>( $J_f = 2, m_f = 0$ ). H<sub>2</sub>( $J_f = 1, m_f = \pm 1$ ) also encounters a strongly anisotropic PES (cf., dash-dot curve in Fig. 4(d)). Thus, we see the same trend in Fig. 4(c), i.e., a corresponding spectra dominated by off-specularly scattered H<sub>2</sub>, with dominant components coming from backscattering. On the other hand, H<sub>2</sub>( $J_f = 1, m_f = 0$ ) encounters an almost flat PES (cf., dotted curve in Fig. 4(b)). As a result, we see strong (dominant) specular scattering of H<sub>2</sub>( $J_f = 1, m_f = 0$ ) (Fig. 4(b)). Finally, we could observe a *p*-H<sub>2</sub> to *o*-H<sub>2</sub> ratio as large as ca. 4.96 (cf., Table 1) compared to that of *normal*-H<sub>2</sub> (*n*-H<sub>2</sub>) (i.e.,  $> 1/3$  at 300 K) at scattering angle  $\Theta_f = 5.26^\circ$  (Fig. 4).

**H<sub>2</sub> Scattering along STO(001)[110] on SrO-terminated STO(001).** In Fig. 5(a–c), we show the calculated (scattering) probabilities/spectra corresponding to the maximum *para*-to-*ortho* ratio  $R_{J=0/J=1}$  ( $E_i = 60$  meV,  $\Theta_i = 76.5^\circ$ ,  $\Theta_f$ ), along the [110] direction of a SrO-terminated STO(001) (cf., Table 2). (The initial surface perpendicular translational energy  $E_i \times \cos^2 \Theta_i = 60 \text{ meV} \times \cos^2(76.5^\circ) \sim 3.27$  meV and the initial surface lateral translational energy  $E_i \times \sin^2 \Theta_i = 60 \text{ meV} \times \sin^2(76.5^\circ) \sim 56.7$  meV.) Again, the resulting trends can be explained by inspecting the corresponding orientational anisotropy and surface lateral corrugation of the

$E_i$ [meV]	$\Theta_i$ [°]	$\Theta_f$ [°]	$R_{J=0/J=1}$	$\overline{R_{J=0/J=1}}$
10.0	0.00	47.1	1.32	0.124
20.0	7.44	22.9	5.97	0.496
30.0	47.8	-47.8	11.4	0.529
40.0	15.9	-55.5	4.06	0.137
50.0	47.5	47.5	2.90	0.541
60.0	76.5	-76.5	16.1	0.0294
70.0	7.96	56.2	1.57	0.0194
80.0	57.3	-57.3	1.73	0.00328

**Table 2.** Calculated  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  and  $\overline{R_{J=0/J=1}}(E_i, \Theta_i, \Theta_f)$  on SrO-terminated STO(001) at the incident energy range  $10 \text{ meV} \leq E_i \leq 80 \text{ meV}$ . For each incident energy, we list only the maximum  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  with the corresponding incident and scattering angles.

potential encountered by the impinging  $\text{H}_2(J_i, m_i, E_i = 60 \text{ meV}, \Theta_i = 76.5^\circ)$  (i.e.,  $\langle Y_J^m | V(Z, \theta, X) | Y_J^m \rangle$ , cf., Fig. 5(d)). Note that along STO(001)[110] on the SrO-terminated STO(001),  $\text{H}_2(J_i = 0, m_i = 0)$  encounters a strongly corrugated and orientationally anisotropic PES (cf., solid curve in Fig. 5(d)), that now favors molecular adsorption atop the O-site, with H-H bond oriented perpendicular to the surface, and an adsorption energy  $E_{\text{ads}} = -151 \text{ meV}^{24}$ . Atop the Sr-site,  $\text{H}_2$  adsorbs with the H-H bond oriented parallel to the surface, and  $E_{\text{ads}} = -111 \text{ meV}^{24}$ . As a result, in Fig. 5(a), we find a corresponding spectra dominated by off-specularly scattered  $\text{H}_2$ , with dominant components coming from backscattering. Again, most of the  $\text{H}_2(J_i = 0, m_i = 0)$  would be molecularly adsorbed (due to *reorientation/steering*<sup>31,32</sup>), and only a small fraction would be elastically scattered (possibly due to *shadow effect*<sup>9</sup>). Considering the angle of incidence, the inelastically scattered  $\text{H}_2$  would come from those hitting the repulsive part of the potential well near the Sr-site (*shadowing effect*, cf., solid curve in Fig. 4(d)), resulting in a change in surface lateral momentum and off-specular scattering angles. Note that the impinging  $\text{H}_2$  has a larger surface lateral momentum as compared to the corresponding surface perpendicular component. This allows the impinging  $\text{H}_2$  more time to explore the anisotropic surface. Those that succeed would molecularly adsorb via *reorientation/steering*. Those that fail, would be rotationally de-excited on the way back to the gas phase, because of the smaller anisotropic potential further (out in the vacuum) from the surface. Thus, the scattering spectra shows negligible rotationally excited  $\text{H}_2$ , i.e.,  $\text{H}_2(J_f = 2, m_i = 0)$ .  $\text{H}_2(J_i = 1, m_i = \pm 1)$  also encounters a strongly anisotropic PES (cf., dash-dot curve in Fig. 5(d)). But now recall that the preferred adsorption site is at the O-site, with the H-H bond oriented perpendicular to the surface. Thus, we see strong specular scattering in Fig. 5(c).  $\text{H}_2(J_i = 1, m_i = 0)$  shows higher backscattering probabilities (cf., Fig. 5(b)) due to the larger surface anisotropy along X, making it more susceptible to reorientation/steering. Finally, we could observe a *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio as large as ca. 16.1 (cf., Table 2) compared to that of *n*- $\text{H}_2$  (i.e.,  $>1/3$  at 300 K) at scattering angle  $\Theta_f = -76.5^\circ$  (Fig. 5).

**(Normalized) *p*- $\text{H}_2$  to *o*- $\text{H}_2$  Ratios.** In Tables 1 and 2, we show the calculated *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$  and the normalized *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio  $\overline{R_{J=0/J=1}}(E_i, \Theta_i, \Theta_f)$  of  $\text{H}_2$  scattered from the  $\text{TiO}_2$ -terminated and SrO-terminated STO(001), respectively. The incident energy ranges from  $10 \text{ meV} \leq E_i \leq 80 \text{ meV}$ . Assuming Maxwell-Boltzmann distribution, at 300 K, 90% of the impinging  $\text{H}_2$  would have kinetic energies less than 80 meV. In general, the SrO-terminated STO(001) show a much higher *p*- $\text{H}_2$  yield compared to the  $\text{TiO}_2$ -terminated STO(001). On  $\text{TiO}_2$ -terminated STO(001), we observe a maximum *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio of  $R_{J=0/J=1}^{\text{max}} = 4.96$  for  $E_i = 80.0 \text{ meV}, \Theta_i = 15.9^\circ, \Theta_f = 5.26^\circ$ . Whereas on SrO-terminated STO(001), we observe a maximum *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio of  $R_{J=0/J=1}^{\text{max}} = 16.1$  for  $E_i = 60 \text{ meV}, \Theta_i = 76.5^\circ, \Theta_f = -76.5^\circ$ . The results suggest that, for all incident energies  $E_i$  considered, the *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratios at specific scattering angles  $\Theta_f$  exceed that of *n*- $\text{H}_2$ , i.e.,  $R_{J=0/J=1}(E_i, \Theta_i, \Theta_f) > 1/3$ , at room temperature ( $T = 300 \text{ K}$ ).

## Conclusion

Here, we reported increased *para*- $\text{H}_2$  (*p*- $\text{H}_2$ ) to *ortho*- $\text{H}_2$  (*o*- $\text{H}_2$ ) ratio of  $\text{H}_2$  scattered from  $\text{SrTiO}_3(001)$  (viz., ca. 4.96 and 16.1 along the  $\text{SrTiO}_3(001)[100]$  and  $\text{SrTiO}_3(001)[110]$  of the  $\text{TiO}_2$ - and SrO-terminated  $\text{SrTiO}_3(001)$ , respectively). For reference, *normal*- $\text{H}_2$  (*n*- $\text{H}_2$ ) have a *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio of  $1/3$ , at room temperature ( $T = 300 \text{ K}$ ). We attribute this to the strongly orientation-dependent (electrostatic) interaction potential between the  $\text{H}_2$  (induced) quadrupole moment and the surface electric field of ionic  $\text{SrTiO}_3(001)$ . These results suggest that ionic surfaces (with tunable surface terminations) could function as a scattering/filtering media to realize rotationally state-resolved  $\text{H}_2$ . This could find significant applications not only in  $\text{H}_2$  storage and transport, but also in realizing materials with pre-determined characteristic properties.

We can compare the present results to previous reports for the inelastic scattering of  $\text{H}_2$  on  $\text{LiF}(001)$ , at normal incidence ( $\Theta_i = 0^\circ$ ) and incident energy of  $E_i = 100 \text{ meV}^{33}$ . From the scattering probability data<sup>33</sup>, we estimate a maximum *p*- $\text{H}_2$  to *o*- $\text{H}_2$  ratio  $R_{J=0/J=1}$  of ca. 2 (as compared to ca. 4.96 and 16.1 along  $\text{SrTiO}_3(001)[100]$  and  $\text{SrTiO}_3(001)[110]$  of the  $\text{TiO}_2$ - and SrO-terminated  $\text{SrTiO}_3(001)$ , respectively). (Note that the results of ref. <sup>33</sup> have since been confirmed experimentally<sup>34</sup>).

Considering the Debye temperature of  $\text{SrTiO}_3$  (ca.  $413.3 \text{ K}^{35}$ ), we may expect some thermal modulation/attenuation. With Ti having a smaller mass than Sr, we would expect more pronounced effect (modulation/attenuation) on  $\text{TiO}_2$ -terminated  $\text{SrTiO}_3(001)$ , as compared to SrO-terminated  $\text{SrTiO}_3(001)$ . Consider for example a

surface temperature of 100 K and incidence angle  $\Theta_i$ : [0°, 66°]. For TiO<sub>2</sub>-terminated SrTiO<sub>3</sub>(001), we estimated Debye-Waller factor values (cf., e.g., ref.<sup>12</sup> for more details) ranging from ca. 0.3 to 0.6, increasing with increasing incidence angle  $\Theta_i$ . In comparison, for SrO-terminated SrTiO<sub>3</sub>(001), we estimated<sup>12</sup> Debye-Waller factor values ranging from ca. 0.6 to 0.8, increasing in value with increasing incidence angle  $\Theta_i$ . The modulation/attenuation would become more pronounced with increased temperature. However, regardless of the degree of modulation/attenuation, the maximum *p*-H<sub>2</sub> to *o*-H<sub>2</sub> ratio for each surface termination remained almost the same (viz., ca. 4.48 and 16.7 along the SrTiO<sub>3</sub>(001)[100] and SrTiO<sub>3</sub>(001)[110] of the TiO<sub>2</sub>- and SrO-terminated SrTiO<sub>3</sub>(001), respectively, at 100 K).

Finally, because of the wide spread of the scattering angle, we could expect small scattering probabilities at each particular scattering angle. Furthermore, the larger the (surface lateral) incident energy, the more number of (surface lateral) diffraction channels (would be) involved (excited). As a result, the normalized *p*-H<sub>2</sub> to *o*-H<sub>2</sub> ratio  $R_{j=0/j=1}$  shows small values (cf., Tables 1 and 2). Thus, so far, we can collect only a small amount of H<sub>2</sub> through any given scattering event. (But with high purity!) For engineering applications, scattering at optimum surface angles should resolve this problem.

## Methods

**Model Hamiltonian.** To study the dynamics of H<sub>2</sub> scattering on STO(001) (Fig. 1), we performed quantum dynamical calculations (cf., e.g., refs.<sup>1-5,8,9,24,31,32</sup>) by solving the corresponding time-independent Schrödinger equation for H<sub>2</sub>, in the vibrational ground state ( $\nu = 0$ ), under the influence of an orientationally anisotropic potential energy (hyper-) surface (PES), using the coupled-channel method<sup>36-42</sup>. The dynamical variables we considered include the H<sub>2</sub> center-of-mass (CM) distance  $Z$  from the surface, the H<sub>2</sub> bond-length  $r$ , the polar and azimuthal angular orientations of the H-H bond with respect to the surface,  $\theta$  and  $\phi$ , respectively, and the position of the H<sub>2</sub> CM  $X$ , along STO(001)[100] and STO(001)[110] of the TiO<sub>2</sub>- and SrO-terminated STO(001), respectively. Considering that the energy scale of the H<sub>2</sub> molecular vibration ( $\hbar\omega = 516$  meV) exceeds the energy range relevant to our current study (i.e.,  $E_i$ : [10, 80] meV), we can neglect the molecular vibrational excitations, and fix the H<sub>2</sub> interatomic distance at  $r = 0.74$  Å. The small variation of the potential energy with respect to  $\phi$  allows us to further neglect the  $\phi$ -dependence<sup>24</sup>. Thus, we can reduce the original 6-Dimensional (6-D) Hamiltonian (for the diatomic molecule-surface system) to the following simplified 3-D form:

$$H = -\frac{\hbar^2}{2m} \left( \frac{\partial^2}{\partial Z^2} + \frac{\partial^2}{\partial X^2} \right) - \frac{\hbar^2}{2I} \left[ \frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \left( \sin\theta \frac{\partial}{\partial\theta} \right) + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\phi^2} \right] + V(Z, \theta, X). \quad (1)$$

$m$  and  $I$  correspond to the H<sub>2</sub> total mass and moment of inertia, respectively.

**Potential Energy (Hyper-) Surface: PES.** The 3-D PES  $V(Z, \theta, X)$  in Eq. (1) comes from previously performed density functional theory (DFT)-based total energy calculations<sup>43-45</sup> for H<sub>2</sub> adsorption at the Ti and O surface sites on the TiO<sub>2</sub>-terminated STO(001), the Sr and O surface sites on the SrO-terminated STO(001), and H<sub>2</sub> polar orientations  $\theta = 0$  and  $\pi/2$ <sup>24</sup>. We fitted the potential energy curves for each configuration using Morse potentials:

$$V_{A(B),\theta} = D_{A(B),\theta} \times \{ \exp[ - 2\alpha_{A(B),\theta}(Z - Z_{A(B),\theta}^{\text{eq}})] - 2\exp[\alpha_{A(B),\theta}(Z - Z_{A(B),\theta}^{\text{eq}})] \}, \quad (2)$$

and connected the Morse potentials for  $\theta = 0$  and  $\pi/2$  at each surface site and for different surface sites to get

$$V(Z, \theta, X) = [V_{A,\theta=\pi/2}\sin^2\theta + V_{A,\theta=0}\cos^2\theta]\cos^2gX + [V_{B,\theta=\pi/2}\sin^2\theta + V_{B,\theta=0}\cos^2\theta]\sin^2gX. \quad (3)$$

$D_{A(B),\theta}$ ,  $\alpha_{A(B),\theta}$ ,  $Z_{A(B),\theta}^{\text{eq}}$  give the corresponding potential depth, potential width, and equilibrium (normal/perpendicular) distance of H<sub>2</sub> from the surface cation sites (A:[Ti, Sr]) and oxygen site (B), at H-H bond angle  $\theta$ , respectively. The reciprocal lattice constant  $g_X (= \pi/a_X)$ , with corresponding direct lattice constants  $a_X^{[100]} = 3.91$  Å and  $a_X^{[110]} = 5.52$  Å along [100] and [110] of the TiO<sub>2</sub>- and SrO-terminated STO(001), respectively<sup>46</sup>. Table 3 shows the fitted parameters for each configuration. We used spherical harmonics and plane waves as basis sets for the rotational motion and the translational motion (perpendicular to the surface and along the surface direction  $X$ ), respectively.

To extract the contribution of the electrostatic interaction between the H<sub>2</sub> quadrupole moment and the surface local electric field, as discussed above, we used the charge density distribution obtained from previous DFT-based total energy calculations<sup>24</sup>. We calculated the induced dipole  $\Delta D_i$  ( $i = x, y, z$ ) and quadrupole moments  $\Delta Q_{ij}$  ( $i, j = x, y, z$ ) from the charge density difference ( $\Delta\rho(Z) = \rho_{\text{H}_2/\text{STO}}(Z) - \rho_{\text{H}_2} - \rho_{\text{STO}}$ ) as a function of  $Z$  for the Ti and O sites (TiO<sub>2</sub>-termination) and Sr and O sites (SrO-termination) with  $\theta = 0$  and  $\pi/2$ . We also calculated the H<sub>2</sub> quadrupole moment  $Q_{ij}$  from the charge density distribution of the isolated system. We used the pristine STO(001) to calculate the surface electric field and its gradient.

**Scattering probability.** Consider a H<sub>2</sub> impinging with an initial rotational state ( $J_i, m_i$ ), incident energy  $E_i$ , and angle of incidence with respect to the surface normal  $\Theta_i$  (cf., e.g., Fig. 1(c,d), and Table 4). Using the coupled-channel method<sup>36-42</sup>, we calculated the probability  $P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f)$  of finding the H<sub>2</sub> scattered with a final rotational state ( $J_f, m_f$ ), final kinetic energy  $E_f$ , scattering at an angle of  $\Theta_f$  with respect to the surface normal. (We carefully checked the convergence for calculations with maximum quantum numbers  $J_{\text{max}} = 10$  and  $|G_{X_{\text{max}}}| = 30$ ). From the calculated scattering probability  $P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f)$  of H<sub>2</sub> on STO(001), we evaluated



Surface	adsorption	$\theta$	$D$	$\alpha$	$Z^{\text{eq}}$
Termination	site	[rad]	[eV]	[ $\text{\AA}^{-1}$ ]	[ $\text{\AA}$ ]
TiO <sub>2</sub>	Ti site-A	0	0.0390	1.40	3.18
		$\pi/2$	0.172	1.47	2.37
	O site-B	0	0.0665	1.30	2.97
		$\pi/2$	0.0486	1.24	3.14
SrO	Sr site-A	0	0.0409	1.21	3.57
		$\pi/2$	0.108	1.39	2.93
	O site-B	0	0.152	1.18	2.55
		$\pi/2$	0.0322	0.755	3.70

**Table 3.** Fitted values of the potential depth  $D$ , potential width  $\alpha$ , and equilibrium distance from the surface  $Z^{\text{eq}}$ , at each surface site, in Eq. (3).

	TiO <sub>2</sub> -termination	SrO-termination
	$a_X^{[100]} = 3.91 \text{ \AA}$	$a_X^{[110]} = 5.52 \text{ \AA}$
	$E_i = 80 \text{ meV}$	$E_i = 60 \text{ meV}$
	$E_i^{\text{total}}(J=0) = 80.0 \text{ meV}$	$E_{\text{total}}(J=0) = 60.0 \text{ meV}$
	$E_i^{\text{total}}(J=1) = 95.2 \text{ meV}$	$E_{\text{total}}(J=1) = 75.2 \text{ meV}$
$G_{iX}$	incident angle $\Theta_i [^\circ]$	incident angle $\Theta_i [^\circ]$
0	0.00	0.00
1	5.26	4.29
2	10.6	8.60
3	<u>15.9<sup>†</sup></u>	13.0
4	21.5	17.4
5	27.3	22.0
6	33.3	26.7
7	39.9	31.6
8	47.1	36.7
9	55.5	42.3
10	66.3	48.4
11	—	55.3
12	—	63.8
13	—	<u>76.5<sup>‡</sup></u>

**Table 4.** Some relevant physical values corresponding to  $\text{H}_2(E_i, J_i, m_i, \Theta_i)$  impinging a TiO<sub>2</sub>- and SrO-terminated SrTiO<sub>3</sub>(001). Incident angles presented in this work doubly underlined.  $^{\dagger}E_{\parallel} = E_X = E_i \sin^2 15.9^\circ \sim 6.004 \text{ meV}$ .  $^{\ddagger}E_{\perp} = E_Z = E_i \cos^2 15.9^\circ \sim 74.00 \text{ meV}$ .  $^{\S}E_{\parallel} = E_X = E_i \sin^2 76.5^\circ \sim 56.73 \text{ meV}$ .  $^{\P}E_{\perp} = E_Z = E_i \cos^2 76.5^\circ \sim 3.270 \text{ meV}$ . •  $E_i$ : incident energy. •  $(J_i, m_i)$ : initial rotational state. •  $\Theta_i$ : incident angle with respect to the surface normal. •  $E_i^{\text{total}} = E_i + E_{J_i}$ : total initial kinetic energy. •  $E_{J_i} = B_{\text{H}_2} J_i(J_i + 1)$ : initial rotational energy. •  $B_{\text{H}_2} = 7.6 \text{ meV}$ : H<sub>2</sub> rotational constant. •  $E_{\perp} = E_Z = E_i \cos^2 \Theta$ : surface normal translational energy. •  $E_{\parallel} = E_X = E_i \sin^2 \Theta = \frac{\hbar^2}{2M} [gG_{iX}]^2$ : surface lateral translational energy. •  $M$ : H<sub>2</sub> total mass. •  $g = \frac{\pi}{a_X}$ : reciprocal lattice constant. •  $a_X$ : direct lattice constant. •  $G_{iX}$ : surface lateral wave number (integer).

the corresponding  $p\text{-H}_2$  to  $o\text{-H}_2$  ratio  $R_{p\text{-H}_2(J=0)/o\text{-H}_2(J=1)}(E_i, \Theta_i, \Theta_f) = R_{J=0/J=1}(E_i, \Theta_i, \Theta_f)$ , given the incident energy  $E_i$ , incident angle  $\Theta_i$ , and scattering angle  $\Theta_f$ , i.e.,

$$R_{J=0/J=1}(E_i, \Theta_i, \Theta_f) = \frac{\int dE_f P_{J_i=0 \rightarrow J_f=0}^{m_i=0 \rightarrow m_f=0}(E_i, E_f, \Theta_i, \Theta_f)}{\int dE_f \sum_{m=-1}^1 P_{J_i=1 \rightarrow J_f=1}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f)} = \frac{P_0^0(E_i, \Theta_i, \Theta_f)}{\sum_{m=-1}^1 P_1^m(E_i, \Theta_i, \Theta_f)}. \quad (4)$$

Note that

$$\int dE_f \sum_{J_f, m_f, \Theta_f} P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f) = 1. \quad (5)$$

And since we are considering the case when  $J_i = J_f$  and  $m_i = m_f$  in Eq. (4) we have

$$P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f) = P_{J_i \rightarrow J_f}^{m_i \rightarrow m_f}(E_i, E_f, \Theta_i, \Theta_f) = P_J^m(E_i, \Theta_i, \Theta_f). \quad (6)$$

We also calculated the normalized  $p$ -H<sub>2</sub> to  $o$ -H<sub>2</sub> ratio  $\overline{R_{J=0/J=1}}(E_i, \Theta_i, \Theta_f)$ , i.e.,

$$\overline{R_{J=0/J=1}}(E_i, \Theta_i, \Theta_f) = R_{J=0/J=1}(E_i, \Theta_i, \Theta_f) \times \frac{P_0^0(E_i, \Theta_i, \Theta_f) + \sum_{m=-1}^1 P_1^m(E_i, \Theta_i, \Theta_f)}{\sum_{\Theta_f=-90^\circ}^{90^\circ} [P_0^0(E_i, \Theta_i, \Theta_f) + \sum_{m=-1}^1 P_1^m(E_i, \Theta_i, \Theta_f)]} \quad (7)$$

In Eq. (7), we multiplied the angle specific  $p$ -H<sub>2</sub> to  $o$ -H<sub>2</sub> ratio by a normalization factor, so as to evaluate the efficiency of rotational state separation with respect to the incident  $n$ -H<sub>2</sub>. Note that the  $p$ -H<sub>2</sub> to  $o$ -H<sub>2</sub> ratio of  $n$ -H<sub>2</sub> corresponds to 0.333 (1/3) at room temperature ( $T = 300$  K)

Received: 14 January 2020; Accepted: 17 March 2020;

Published online: 03 April 2020

## References

- Diño, W. A., Kasai, H. & Okiji, A. Orientational effects in dissociative adsorption/associative desorption dynamics of H<sub>2</sub> (D<sub>2</sub>) on Cu and Pd. *Prog. Surf. Sci.* **63**, 63–134 (2000).
- Diño, W. A., Kasai, H. & Okiji, A. Rotational Alignment in the Associative Desorption Dynamics of Hydrogen Molecules from Metal Surfaces. *J. Phys. Soc. Jpn.* **67**, 1517–1520 (1998).
- Diño, W. A., Kasai, H. & Okiji, A. Dynamical quantum filtering in hydrogen-surface reactions. *Surf. Sci.* **418**, L39–L44 (1998).
- Miura, Y., Kasai, H. & Diño, W. A. Dynamical quantum filtering in the scattering dynamics of H<sub>2</sub> on Cu(001). *J. Phys.: Condens. Matter* **14**, L479–L486 (2002).
- Diño, W. A., Kasai, H. & Okiji, A. Dissociative Adsorption Dynamics of H<sub>2</sub> (D<sub>2</sub>) on Metal Alloy Surfaces—A First Step Towards Surface Reaction Design/Control—. *J. Phys. Soc. Jpn.* **69**, 993–996 (2000).
- Ogura, S. & Fukutani, K. Dynamic Blocking by CO of Hydrogen Transport across Pd<sub>70</sub>Au<sub>30</sub>(110) surfaces. *J. Phys. Chem. C* **121**, 3373–3380 (2017).
- Namba, K. *et al.* Acceleration of hydrogen absorption by palladium through surface alloying with gold. *Proc. Nat. Acad. Sci.* **115**, 7896–7900 (2018).
- Diño, W. A. Can we probe local surface reactivity with hydrogen molecules? *J. Phys.: Condens. Matter* **14**, 4379–4384 (2002).
- Diño, W. A. *et al.* Probing Local Surface Reactivity with Hydrogen Molecules—Realizing an Atom/Molecule Scanning Probe—. *J. Vac. Soc. Jpn.* **46**, 391–396 (2003).
- Barredo, D., Laurent, G., Nieto, P., Fariás, D. & Miranda, R. High-resolution elastic and rotationally inelastic diffraction of D<sub>2</sub> from NiAl(110). *J. Chem. Phys.* **133**, 124702 (2010).
- Nieto, P., Barredo, D., Fariás, D. & Miranda, R. In-Plane and Out-of-Plane Diffraction of H<sub>2</sub> from Ru(001). *J. Phys. Chem. A* **115**, 7283–7290 (2011).
- Fariás, D. & Miranda, R. Diffraction of molecular hydrogen from metal surfaces. *Prog. Surf. Sci.* **86**, 222–254 (2011).
- Hill, N. R. Quadrupole interaction in the scattering of H<sub>2</sub> from the surface of LiF: Rotational transitions. *Phys. Rev. B* **19**, 4269–4276 (1979).
- Meyer, B., Padilla, J. & Vanderbilt, D. Theory of PbTiO<sub>3</sub>, BaTiO<sub>3</sub>, and SrTiO<sub>3</sub> surfaces. *Faraday Discuss.* **114**, 395–405 (1999).
- Noguera, C. Polar oxide surfaces. *J. Phys.: Condens. Matter* **12**, R367–R410 (2000).
- Piskunov, S., Heifets, E., Eglitis, R. & Borstel, G. Bulk properties and electronic structure of SrTiO<sub>3</sub>, BaTiO<sub>3</sub>, PbTiO<sub>3</sub> perovskites: an ab initio HF/DFT study. *Comp. Mater. Sci.* **29**, 165–178 (2004).
- Wang, D., K., T., Ye, J. & Kimura, T. Photophysical and Photocatalytic Properties of SrTiO<sub>3</sub> Doped with Cr Cations on Different Sites. *J. Phys. Chem. B* **110**, 15824–15830 (2006).
- Konta, R., Ishii, T., Kato, H. & Kudo, A. Photocatalytic Activities of Noble Metal Ion Doped SrTiO<sub>3</sub> under Visible Light Irradiation. *J. Phys. Chem. B* **108**, 8992–8995 (2004).
- Jia, Q., Wu, X., Foltyn, S. & Tiwari, P. Structural and electrical properties of Ba<sub>0.5</sub>Sr<sub>0.5</sub>TiO<sub>3</sub> thin films with conductive SrRuO<sub>3</sub> bottom electrodes. *Appl. Phys. Lett.* **66**, 2197–2199 (1995).
- Van der Berg, R., Blom, P., Cillessen, J. & Wolf, R. Field dependent permittivity in metal-semiconducting SrTiO<sub>3</sub> Schottky diodes. *Appl. Phys. Lett.* **66**, 697–699 (1995).
- Tasker, P. W. The stability of ionic crystal surfaces. *J. Phys. C: Solid State Phys.* **12**, 4977–4984 (1979).
- Kawasaki, M. *et al.* Atomic Control of the SrTiO<sub>3</sub> Crystal Surface. *Science* **266**, 1540–1542 (1994).
- Aballe, L. *et al.* Instability and Surface Potential Modulation of Self-Patterned (001) SrTiO<sub>3</sub> Surfaces. *Chem. Mater.* **27**, 6198–6204 (2015).
- Shimizu, K. *et al.* Molecular Adsorption, Hindered Rotation, and Species Separation of H<sub>2</sub>/SrTiO<sub>3</sub>(001). *J. Phys. Soc. Jpn.* **86**, 073601–1–4 (2017).
- Sandler, Y. The Adsorption and Ortho-Para Conversion of Hydrogen on Diamagnetic Solids. II. The Relative Adsorbabilities of Orthohydrogen and Parahydrogen. *J. Phys. Chem.* **58**, 58–61 (1954).
- Briquez, S. *et al.* Adsorption of ortho and para H<sub>2</sub> on NaCl(001). *J. Chem. Phys.* **109**, 6435–6449 (1998).
- Silvera, I. F. & Nielsen, M. Inelastic Neutron Scattering and Separation Coefficient of Absorbed Hydrogen: Molecular Alignment and Energy Levels. *Phys. Rev. Lett.* **37**, 1275–1278 (1976).
- Fukutani, K. & Sugimoto, T. Physisorption and ortho-para conversion of molecular hydrogen on solid surfaces. *Prog. Surf. Sci.* **88**, 279–348 (2013).
- Sugimoto, T. & Fukutani, K. Effects of Rotational-Symmetry Breaking on Physisorption of Ortho- and Para-H<sub>2</sub> on Ag(111). *Phys. Rev. Lett.* **112**, 146101–1–5 (2014).
- Ohno, S. *et al.* Rotational state modification and fast ortho-para conversion of H<sub>2</sub> trapped within the highly anisotropic potential of Pd(210). *Phys. Rev. B* **97**, 085436–1–5 (2018).
- Diño, W. A., Kasai, H. & Okiji, A. Effects of Rotational State Excitations on the Dissociative Adsorption Dynamics of D<sub>2</sub>/Cu(111). *J. Phys. Soc. Jpn.* **64**, 2478–2487 (1995).
- Diño, W. A., Kasai, H. & Okiji, A. Role of Rotational Motion in the Dissociative Adsorption and Associative Desorption Dynamics of D<sub>2</sub>/Cu(111). *Phys. Rev. Lett.* **78**, 286–289 (1997).
- Kroes, G.-J. & Mowrey, R. Scattering of H<sub>2</sub> by LiF(001) studied using a new model potential. I. Prediction of large differences in diffraction of cold beams of para-H<sub>2</sub> and normal-H<sub>2</sub>. *J. Chem. Phys.* **103**, 2186–2201 (1995).
- Bertino, M. F. *et al.* Observation of Large Differences in the Diffraction of Normal- and Para- H<sub>2</sub> from LiF(001). *Phys. Rev. Lett.* **81**, 5608–5611 (1998).
- Lawless, W. N. Debye temperatures of ferroelectrics. *Phys. Rev. B* **17**, 1458–1459 (1978).
- Kasai, H. & Okiji, A. Rotational Cooling and Vibrational Heating in Associative Desorption. *Prog. Theor. Phys. Suppl.* **106**, 341–348 (1991).
- Kasai, H. & Okiji, A. Dynamics of vibrational excitation in molecule-surface interactions. *Prog. Surf. Sci.* **44**, 101–183 (1993).

38. Kasai, H., Brenig, W. & Müller, H. Rotational state distribution of NO molecules scattered from surfaces. *Z. Phys. B* **60**, 489–495 (1985).
39. Brenig, W., Kasai, H. & Müller, H. Coupled channel problems with Morse potentials: A simple and fast method. *Surf. Sci.* **161**, 608–626 (1985).
40. Brenig, W., Brunner, T., Groß, A. & Russ, R. Numerically stable solution of coupled channel equations: The local reflection matrix. *Z. Phys. B* **93**, 91–101 (1993).
41. Brenig, W. & Russ, R. Numerically stable solution of coupled-channel equations: the local transmission matrix. *Surf. Sci.* **315**, 195–203 (1994).
42. Brenig, W., Groß, A. & Russ, R. Numerically stable solution of coupled channel equations: the wave function. *Z. Phys. B* **97**, 311–317 (1995).
43. Kresse, G. & Furthmüller, J. Efficiency of ab-initio total energy calculations for metals and semiconductors using a plane-wave basis set. *Comput. Mater. Sci.* **6**, 15–50 (1996).
44. Kresse, G. & Furthmüller, J. Efficient iterative schemes for ab initio total-energy calculations using a plane-wave basis set. *Phys. Rev. B* **54**, 11169–11186 (1996).
45. Grimme, S. Semiempirical GGA-type density functional constructed with a long-range dispersion correction. *J. Comput. Chem.* **27**, 1787–1799 (2006).
46. Okazaki, A. & Kawaminami, M. Lattice constant of strontium titanate at low temperatures. *Mat. Res. Bull.* **8**, 545–550 (1973).
47. Momma, K. & Izumi, F. VESTA 3 for three-dimensional visualization of crystal, volumetric and morphology data. *J. Appl. Cryst.* **44**, 1272–1276 (2011).

## Acknowledgements

This work is supported in part by MEXT Grants-in-Aid for Scientific Research (18H05518, 17K06818, 17H01057, 15H05736, 15KT0062, 15K14147, and 26248006); NEDO Project “R&D Towards Realizing an Innovative Energy Saving Hydrogen Society based on Quantum Dynamics Applications”, Grants-in-Aid for JSPS Fellows; and Kawasaki Heavy Industries, Ltd. Some of the numerical calculations presented here done using the computer facilities at the following institutes: CMC (Osaka University), ISSP, KEK, NIFS, and YITP. Structure figures were plotted using the VESTA package<sup>47</sup>.

## Author contributions

K.S., W.A.D., H.N., K.F., and H.K. conceived the model and performed the calculation. All authors contributed to the discussion, analyses, and writing the manuscript.

## Competing interests

A.Y. is an employee of Kawasaki Heavy Industries, Ltd. The other authors declare no competing interests.

## Additional information

**Correspondence** and requests for materials should be addressed to W.A.D.

**Reprints and permissions information** is available at [www.nature.com/reprints](http://www.nature.com/reprints).

**Publisher's note** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



**Open Access** This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2020