



Article

Deposition and Characterization of RP-ALD SiO₂ Thin Films with Different Oxygen Plasma Powers

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Abstract: In this study, silicon oxide (SiO₂) films were deposited by remote plasma atomic layer deposition with Bis(diethylamino)silane (BDEAS) and an oxygen/argon mixture as the precursors. Oxygen plasma powers play a key role in the quality of SiO₂ films. Post-annealing was performed in the air at different temperatures for 1 h. The effects of oxygen plasma powers from 1000 W to 3000 W on the properties of the SiO₂ thin films were investigated. The experimental results demonstrated that the SiO₂ thin film growth per cycle was greatly affected by the O₂ plasma power. Atomic force microscope (AFM) and conductive AFM tests show that the surface of the SiO₂ thin films, with different O₂ plasma powers, is relatively smooth and the films all present favorable insulation properties. The water contact angle (WCA) of the SiO₂ thin film deposited at the power of 1500 W is higher than that of other WCAs of SiO₂ films deposited at other plasma powers, indicating that it is less hydrophilic. This phenomenon is more likely to be associated with a smaller bonding energy, which is consistent with the result obtained by Fourier transformation infrared spectroscopy. In addition, the influence of post-annealing temperature on the quality of the SiO₂ thin films was also investigated. As the annealing temperature increases, the SiO₂ thin film becomes denser, leading to a higher refractive index and a lower etch rate.

Keywords: SiO₂ thin film; oxygen plasma power; atomic layer deposition



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1. Introduction

Silicon dioxide (SiO₂) is a versatile material that is used for many applications. SiO₂ and more generally ultra-thin oxide films have been extensively described as good components for modern nano technologies such as dielectric materials in silicon microelectronic devices, anticorrosion films, or non-exhaustive applications of nanoscale films in catalysis [1]. SiO₂ films can be prepared by many methods, such as chemical vapor deposition (CVD) [2–4], thermal oxidation [5–7], and atomic layer deposition (ALD) [8–15]. Among these preparation methods, the ALD method is highlighted because the film growth can be precisely controlled based on self-limiting and sequential surface reactions. ALD can be used to prepare various thin films, such as alumina, graphene-augmented alumina [16–19], Hafnium dioxide, SiO₂, and so on. Many studies on the deposition of SiO₂ films through ALD have been conducted. In 2018, M. Ziegler et al. [11] prepared 3D nanostructures using silica with tri(dimethylamino)silane and oxygen plasma. During the same year, Yoo-Jin Choi et al. [12] reported that SiO₂ deposited by ALD functions as gate dielectric thin films. In 2019, Wen Zhou et al. [13] investigated ALD SiO₂ for the channel isolation of colloidal

quantum dot phototransistors. D. Arl et al. [1] reported less contaminated and porous SiO₂ films grown by ALD at room temperature. Marc, J. M. Merckx et al. [15] studied the removal and reapplication of the inhibitor molecules during the area-selective ALD of SiO₂. There are various ALD process modes. Remote plasma atomic layer deposition (RP-ALD) is a widely used mode. The RP-ALD possesses the high reactivity of many plasma species; it can thus reduce the deposition temperatures without degrading the prepared film quality. After comparing many investigations, Bis(diethylamino)silane (BDEAS) is the usual precursor for depositing SiO₂ due to its chemical stability. Argon (Ar) and oxygen (O₂) are the most commonly used gases that can effectively generate plasma because of their stability and unique electronegativity [20,21], respectively. Ar and O₂ molecules are activated into excited atoms. The functional groups of the precursor are then oxidized by the excited atoms. However, systematic investigations on the oxygen plasma power for the property of SiO₂ films are rare.

SiO₂ films were grown by RP-ALD with different oxygen plasma powers. The variation of the oxygen plasma power on the effects of the electrical, optical, chemical, structural, and morphological properties of the films was investigated. Furthermore, the annealing temperature is another parameter that affected the quality of the thin films prepared by the RP-ALD. Therefore, the relationship between the property changes of the SiO₂ thin films and the variation in the annealing temperature is also discussed.

2. Experimental Section

Four-inch (100) oriented p-type Czochralski polished Si wafers were used as substrates. The Si wafer had a resistivity of about 50 Ω-cm and a thickness of about 450 μm. Prior to the deposition of the SiO₂ thin films, all wafers were cleaned using deionized water for 10 s, a diluted hydrofluoric (HF) acid solution (2%) for 1 min, and deionized water for 10 s. After cleansing, all Si wafers were dried using pure nitrogen (N₂) gas with a purity of 99.99%, then transferred onto the substrate holder of a commercial RP-ALD system (Picosun R-200, Espoo, Finland). The RP-ALD system had six source channels. One of the channels was connected to the BDEAS bubbler, and the rest of the channels were connected to N₂. The plasma was produced in a microwave cavity using inductive coupling of radio frequency (RF) power with the mixture of Ar and O₂ gases. These two gases both have a high purity of 99.999%. The SiO₂ thin films were grown using BDEAS (Aimou Yuan, Nanjing, China) and remote O₂ plasma as the precursors with the carrier gas N₂. The BDEAS precursor foreline was heated to 45 °C in order to avoid any precursor condensation. The base pressure of the system was 100 Pa. One BDEAS pulse and one O₂ plasma pulse were 1.6 s and 8 s respectively. The N₂ purge times for BDEAS and O₂ were both 5 s. The substrate holder was heated to 250 °C and kept at the same temperature in the deposition process. O₂ plasma at different powers (1000, 1500, 2000, 2500, and 3000 W) was used. Table 1 lists the detailed experiment parameters of the SiO₂ thin films. The SiO₂ thin films with a plasma of 2000 W were separately prepared and annealed at 300–850 °C in the air for 1 h to investigate their densification. Various SiO₂ thin films were obtained by chemical etching using a diluted HF acid (2% in deionized H₂O), followed by a deionized water rinse.

The O₂ plasma power emission spectra were assessed by an optical emission spectrometer (OES). The thickness and refractive index of the SiO₂ thin films deposited on the silicon wafers were measured by a spectroscopic ellipsometer (SENTECH SE 800 DUV, Berlin, Germany) using an air roughness model, which consisted of an “air, SiO₂, silicon” three-layer structure. The refractive index was extracted from the spectroscopic ellipsometer measurements assuming a Cauchy dispersion model. AFM (Bruker, Karlsruhe, Germany) measurements were performed to investigate the surface morphology of the SiO₂ thin films. The AFM images shown in this study are 5 μm × 5 μm scans with a resolution of 256 points × 256 lines. All the AFM evaluations to calculate roughness were conducted from the central areas of the samples. Conductive atomic force microscopy (CAFM) was also measured in the air to analyze the insulation properties of the SiO₂ thin films. The bonding configuration of the SiO₂ thin films was examined by Fourier transformation

infrared (FTIR) spectroscopy (Bruker, Karlsruhe, Germany). The water contact angle (WCA) and the surface free energy (SFE) of the SiO₂ thin films were also evaluated by a contact angle analyzer (Dongguan Shengding Precision Instrument Co., Ltd., Dongguan, China).

Table 1. Deposition parameters of RP-ALDSiO₂ thin films.

Parameters	Value
Bubbler temperature (°C)	45
Substrate temperature (°C)	250
BDEAS pulse time (s)	1.6
BDEAS purge time (s)	5
O ₂ pulse time (s)	8
O ₂ flow stabilization (s)	1.4
O ₂ RF power on (s)	6
O ₂ purge time (s)	5
Flow rate of Ar (sccm)	80
Flow rate of O ₂ (sccm)	150
O ₂ plasma power (W)	1000–3000
Flow rate of BDEAS carry gas (sccm)	120
Flow rate of BDEAS dilute gas (sccm)	400

3. Results and Discussion

The image of the visual observation of glow discharge at the plasma power of 2000 W is presented for reference in Figure 1. The O₂ plasma pulse was produced by a dielectric tube. The distance between the sensing point and the substrate was 430 mm. According to a research study carried out by the Zhen Zhu group, there is a recognizable effect of the oxygen plasma power on the SiO₂ thin films' impurity levels [22]. The OES method has also proven to be useful as a tool for the monitoring of film deposition by PECVD [23]. Figure 2a shows the OES spectra of SiO₂ thin films prepared with an O₂/Ar mixture plasma power of 1000, 1500, 2000, 2500, and 3000 W. The main features of the OES spectra correspond to the emission of Ar ions and O radicals. Ar ions and O radicals' emission spectra were assigned according to the signals from pure Ar and O₂ plasma [24]. The emission spectra are caused by the transition of the electrons. When the plasma concentration is lower, such as between 1000 and 1500 W, the intensity of the Ar ions and O radicals is low in the wavelength region of 680 to 800 nm, indicating that the ionization of the gases is low. Although the intensity of Ar ions and O radicals is enhanced at the plasma power of 1500 W, there is not enough energy to fully oxidize the BDEAS. When the plasma power is increased to 2000 W, the intensity of the O radicals increases significantly and surpasses the intensity of Ar ions. The spectra are similar when the powers are between 2000 and 3000 W. Figure 2b shows the summed intensity of Ar ions and O radicals at various plasma powers. When the plasma power is increased, both the summed intensities of Ar ions and O radicals are increased and then tend to saturate. When the plasma power is at 1000 and 1500 W, the summed intensity of Ar ions and O radicals is considerably low, suggesting that the ionization of the gases is quite low. When the plasma power is above 2000 W, the summed intensity of the O radicals increases sharply and begins to saturate. The summed intensity of O radicals exceeds the summed intensity of Ar ions, indicating that the ionization of the gases is sufficient. However, the summed intensity of Ar ions continues to increase, which may increase the plasma bombardment etching effect, resulting in the degradation of the SiO₂ film quality.

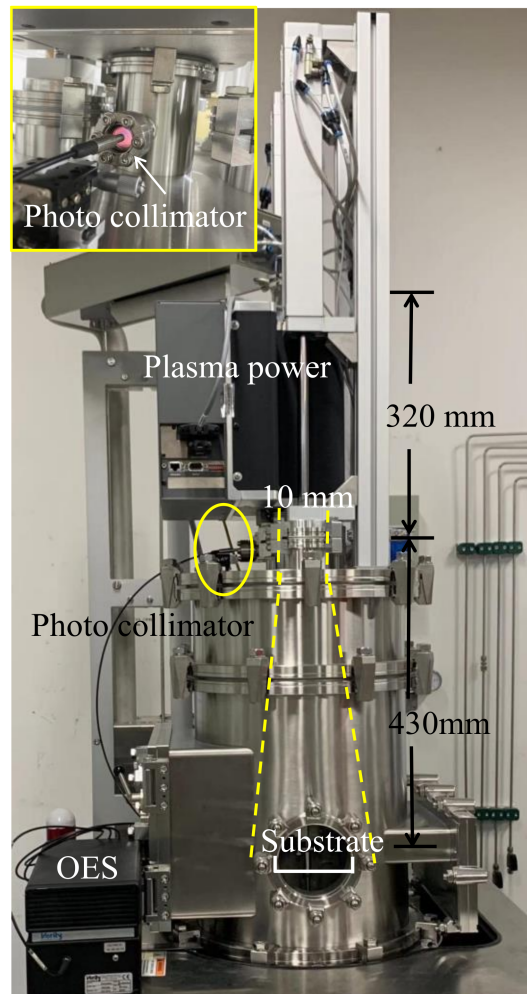


Figure 1. Outline image of the ALD system and visual observation of the glow discharge at 2000 W.

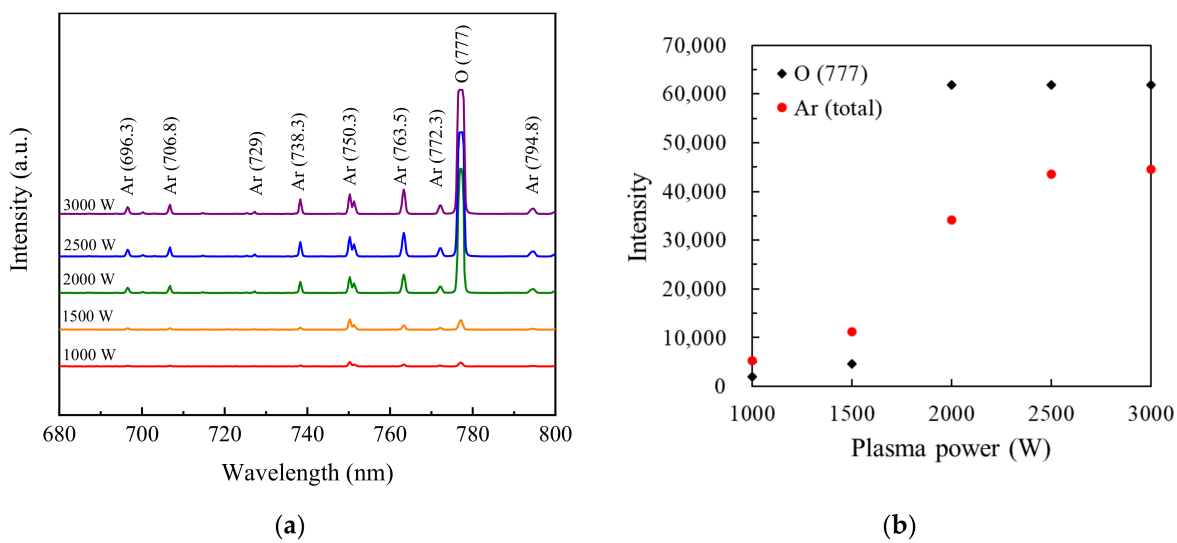


Figure 2. (a) OES spectra for SiO₂ thin films prepared with O₂/Ar mixture plasma at different plasma powers; (b) summed intensity for Ar ions and O radical spectra at various plasma powers.

Figure 3a reveals a linear increase in the thickness of the SiO₂ thin films as a result of different cycles for five different plasma powers. As shown in Figure 3a, the SiO₂ thin films'

thickness increases as the ALD cycle increases from 100 to 1000. The linear relationship between the SiO₂ thin films' thickness and the cycles has confirmed that the SiO₂ thin films were prepared in a self-limiting manner [25]. Figure 3b shows the relationship between O₂ plasma power and the deposition rate of the SiO₂ thin films. The deposition rate was determined using the SiO₂ thin film thickness divided by the total number of ALD cycles. The growth rate is about 0.1 nm/cycle at a plasma power of 1000 W. The growth rate sharply plunged to a minimum of about 0.04 nm/cycle when the plasma power increased to 1500 W. When the plasma further increased, the growth rate rose again and became saturated. The highest deposition rate at a plasma power of 1000 W may arise from the deficient energy of oxygen radicals, which grow loose SiO₂ molecules or SiO. According to the OES measurement, the intensity of O radicals is relatively low. Therefore, when the SiO₂ thin films are deposited, the surface reactions may deviate from the ideal ALD process, giving rise to an abnormally high deposition rate at 1000 W. Under the circumstances, the SiO₂ thin films were prepared with a considerably loose structure instead of a typical ALD film structure. When the plasma power goes up to 1500 W, the BDEAS reacts with the hydroxyl groups which are on the substrate surface, and the excess BDEAS molecules and byproducts are purged by N₂. Then, greater amounts of oxygen radicals oxidize the BDEAS absorbed on the substrate surface to form an OH terminated surface. However, the amount of oxygen radicals is still insufficient to form a dense SiO₂ film. When the plasma power is higher than 2000 W, adequate oxygen radicals take part in the surface oxidation reaction, leading to a higher deposition rate. When the plasma power is changed from 2000 W to 3000 W, the Ar ions boost the bombardment etching effect, as more Ar ions are excited. This is confirmed by the OES measurement. The deposited rate is gained by the trade-off between the plasma bombardment etching effect and the full oxidation. It is interesting to note that this variation is also found in the deposition of tin oxide prepared by the same ALD system in our lab [26].

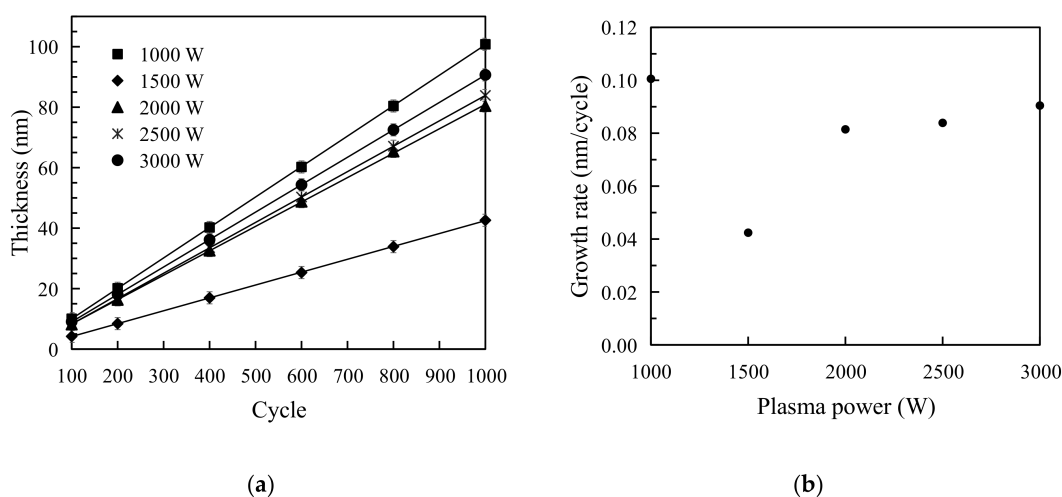


Figure 3. (a) Increase in the thickness of SiO₂ thin films with the number of RP-ALD cycles; (b) deposition rate of the RP-ALD SiO₂ thin films prepared at different plasma powers.

Figure 4a shows the measured wavelength-dependent refractive index spectra for the RP-ALD SiO₂ films with various plasma powers. As can be seen from the graph, the refractive index fluctuates between 1.46 and 1.47, which is comparable to the value of 1.465 for the thermally grown SiO₂ film prepared at 1000 °C in dry O₂ [6]. The values at plasma power of 1000 and 1500 W are higher. This could be related to the stoichiometric ratio of the SiO₂ thin films. When depositing the SiO₂ thin films at a lower power of 1000 and 1500 W, the O radicals were not enough to oxidize the BDEAS absorbed on the Si substrate, resulting in oxygen deficit. Thus, some SiO with a higher refractive index of about 1.8~1.9 [27] or SiO_x (1 < x < 2) was formed at the interface. As the O₂ plasma power

was increased to 2000, 2500, and 3000 W, more O radicals were produced with sufficient radicals to form SiO₂ films. Therefore, the refractive indexes at a higher O₂ plasma power show little change. Figure 3b displays the refractive index spectra of the RP-ALD SiO₂ thin films at various annealing temperatures. The films were all deposited at a plasma power of 2000 W. From this diagram, the refractive index of the annealing samples are all higher than the refractive index of the non-annealing samples. When the annealing temperature increase from 400 to 850 °C, the refractive index of the annealing SiO₂ thin films continuously increases. The noticeable increase in the refractive index may be a reflection of the change in film density, as they are closely related. When the SiO₂ thin films were annealed from 400 to 850 °C, they became compact, giving rise to a higher refractive index.

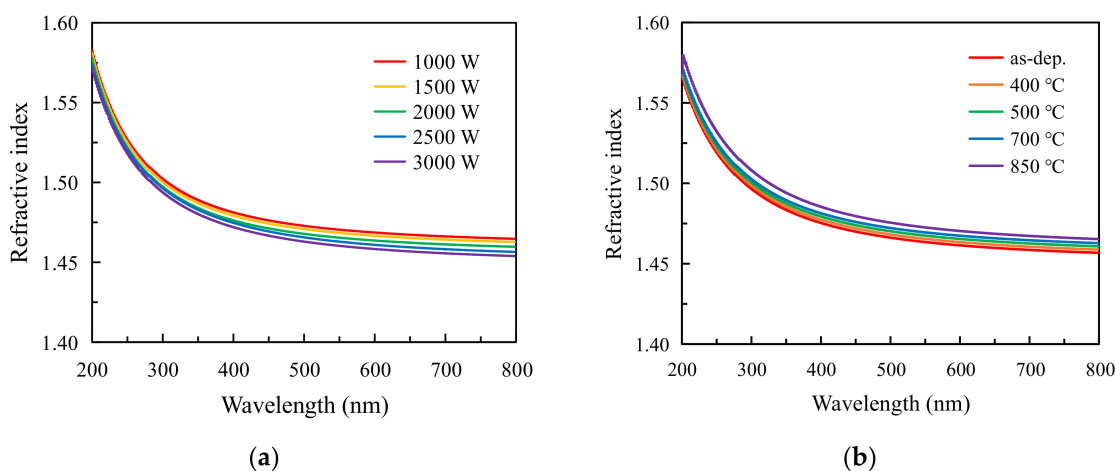


Figure 4. Refractive index spectra for the RP-ALD SiO₂ thin films (a) with various plasma powers; (b) with and without different annealing temperatures.

The wet etch rate is a key parameter to determine the densification of the annealing SiO₂ thin films. In this study, the wet etch rate was obtained by immersing the SiO₂ thin films in a diluted 2% HF acid solution. Figure 5a shows the etched thickness of the SiO₂ thin films deposited at 2000 W versus the etching time. The etched SiO₂ thickness gradient was measured using a spectroscopic ellipsometer. The etch rate of the samples in the HF acid solution is determined by the relationship between the SiO₂ thickness and the etching time gradient. The etching rate is denoted in nanometer per second. As shown in the figure, the thickness of as deposited SiO₂ thin film rapidly decreases when the etching time increases, indicating that the SiO₂ thin film has insufficient densification. In addition, the etch rate decreases as the annealing temperature increases. The densification of SiO₂ films is improved by the annealing process. Figure 5b depicts a variation in the etch rate of the SiO₂ thin films with 70 s etching when the temperature annealed from 300 to 800 °C. It is noted that the etch rate decreases sharply and then decreases slowly when the annealing temperature gradually increases. The etch rate at an annealing temperature of 200 °C is about 1.6 nm/s, while it decreases to about 0.3 nm/s when the annealing temperature is 800 °C. The SiO₂ thin films were easily decomposed in a diluted HF acid solution because they were not fully densified when their annealing temperatures were lower [28]. This result is also supported by the change in refractive index, because it commonly corresponds to the densification of the SiO₂ thin films. When the annealing temperature increases, the SiO₂ thin film becomes denser, leading to a lower etch rate. The SiO₂ films' etching mechanism in HF acid solution can refer D. Martin Knotter's study. According to a research study carried out by D. Martin Knotter [29], the dissolution of vitreous silicon dioxide in HF-based solutions depends on the number of silicon atoms bonded to four bridging oxygen atoms, the pH, and the concentration of HF₂⁻.

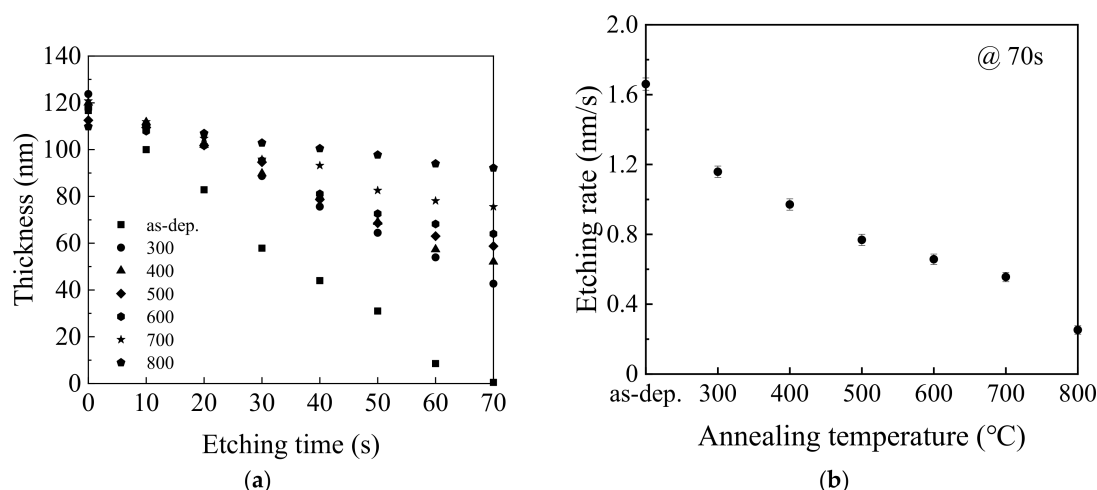


Figure 5. (a) Thickness of the RP-ALD SiO₂ thin films at various annealing temperatures after immersing in diluted HF acid; (b) etching rate of the RP-ALD SiO₂ thin films during the 70 s.

The morphological evolution of the SiO₂ thin films prepared at various O₂ plasma powers was confirmed by AFM. Figure 6 shows the AFM images for the SiO₂ thin films deposited at different O₂ plasma powers: (a) 1000 W, (b) 1500 W, (c) 2000 W, (d) 2500 W, and (e) 3000 W. Since plasma may change surface roughness, the surface morphology of the SiO₂ thin films deposited at various plasma powers was studied. The average surface roughness (Ra) values are also plotted in Figure 6f to indicate surface roughness. As can be seen in the figure, the Ra values, which vary from 0.4 nm to 0.5 nm with the plasma power from 1000 W to 3000 W, show little change. This result indicates that all of the surfaces of the RP-ALD SiO₂ films are relatively smooth. Although the SiO₂ thin films deposited by a higher plasma power of 2500 or 3000 W undergo Ar ions bombardment, they still have low Ra roughness values. It is worth noting that when the O₂ plasma is 3000 W, more needle tips emerge on the image, possibly due to the stronger Ar ions bombardment.

Scanning probe microscopes are a powerful tool for investigating the electrical quality of dielectrics [30,31]. CAFM has widely been used to measure the electrical conduction of dielectrics. In our study, CAFM measurements were performed in the air and the current maps were obtained by using a constant voltage to the tip with the substrate grounded. The effect of the plasma power on the electrical quality of the SiO₂ thin films was investigated on the current images of the thin films measured at 3.0 V on the same regions as shown in Figure 6. Figure 7 shows the current maps of the SiO₂ thin films with various plasma powers. From the images, it can be seen that all the currents of the SiO₂ thin films are relatively small, with values on the order of femtoamperes (fA), suggesting that all the samples have favorable insulation properties. When the plasma power is 1000 W, most of the currents of the SiO₂ thin films are approximately 80 fA due to the considerably loose structure of the film. When the plasma power increases to 1500 W, most of the currents of the SiO₂ thin films are between 75 and 40 fA. It is important to note that most of the currents of the thin films with a plasma power of 2000 W are about 20 fA, which is the lowest value, when compared with other samples. When the plasma power further increases to 2500 W and 3000 W, most of the current values of the thin films are around 90 fA, which may be induced by the ions' bombardment effect, leading to a reduction in the SiO₂ thin film quality.

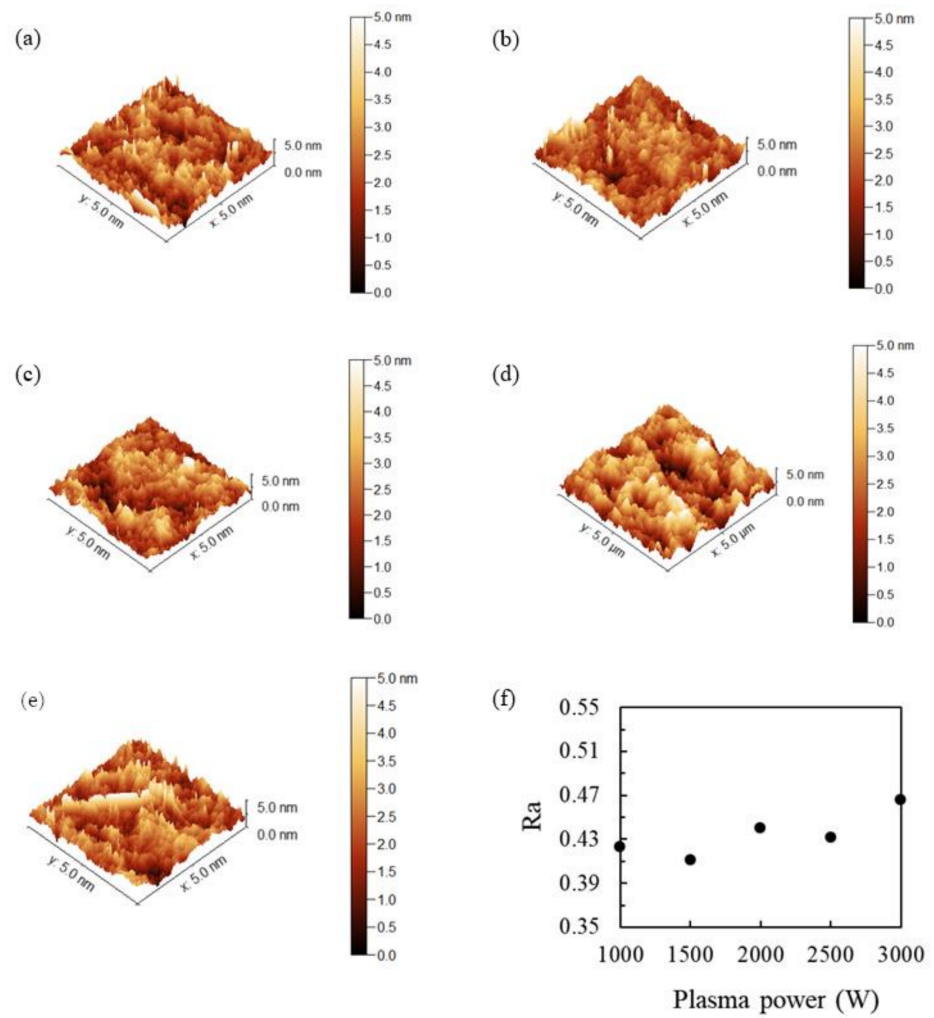


Figure 6. AFM images of the SiO₂ thin films with plasma power of (a) 1000 W, (b) 1500 W, (c) 2000 W, (d) 2500 W, and (e) 3000 W. (f) Ra values as a function of various plasma powers.

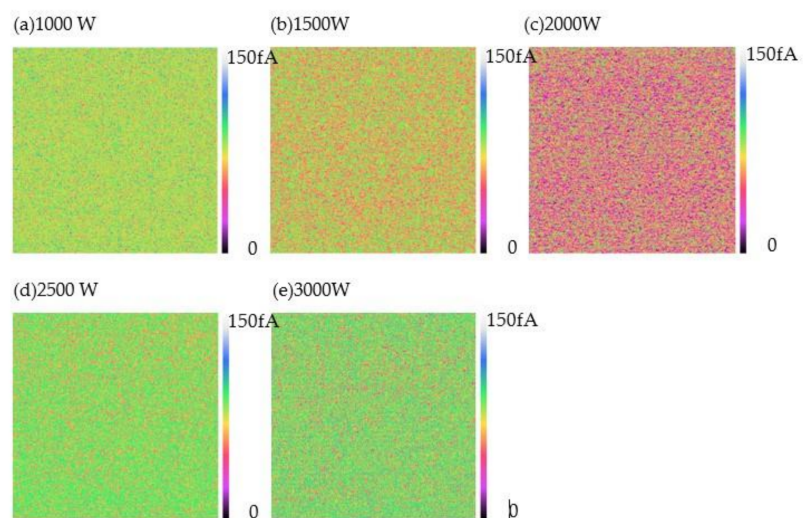


Figure 7. CAFM images of the RP-ALD SiO₂ thin films with different powers (a) 1000 W, (b) 1500 W, (c) 2000 W, (d) 2500 W and (e) 3000 W.

The chemical composition of the SiO₂ thin films was measured by FTIR. Figure 8 displays a typical FTIR spectrum of the SiO₂ thin films prepared at different plasma powers. Peaks in the range of 1100–1000 cm⁻¹ could be attributed to the absorption of Si-O vibrations. The strong absorption at about 1067 cm⁻¹ indicates the presence of Si-O-Si bonds in the SiO₂ films. These features are consistent with Si-O vibrational features that have been observed previously [9]. Furthermore, the absorbance strength of the thin film prepared at 1500 W is less than that of other films. The relatively weak absorbance from the Si-O-Si combination bands at a plasma power of 1500 W is primarily caused by the lower SFE.

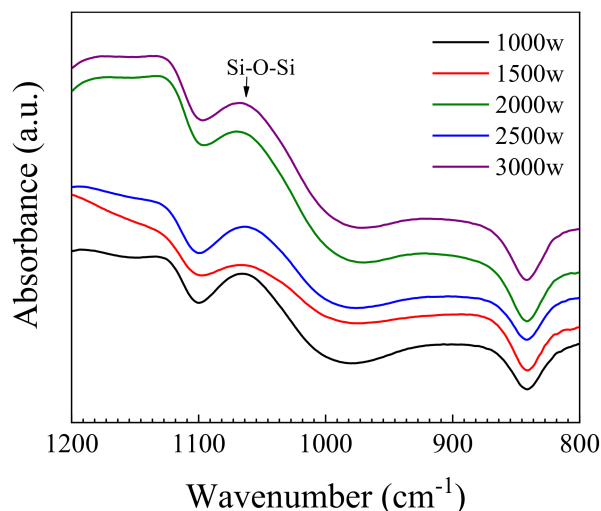


Figure 8. FTIR spectra of the RP-ALD SiO₂ films with various plasma powers.

The wetting characteristic of SiO₂ films has many applications. Figure 9a shows the WCA of the SiO₂ thin films with various plasma powers and different annealing temperatures. The WCA is known as the angle at which a water-vapor interface meets the solid surface. The WCA can be used as a measurement to display wetting phenomena. On a solid surface, a drop of water may remain a spherical drop or spread to cover the solid surface. When the WCA is less than 90°, wetting occurs. When the WCA is greater than 90°, the water does not wet the surface. In this study, the volume of liquid dropped to the SiO₂ film surface was 2 μL every time. As can be seen from the figure, the WCAs vary between 20 and 57°, which is indicative of a hydrophilic surface. The hydrophilic nature of the SiO₂ surface is due to the hydrogen bonding between silanol groups and water molecules [32]. On the surface of the as deposited SiO₂ thin films, the WCAs show minor changes, as the change of the O₂ plasma power. However, the WCA of the as deposited SiO₂ thin film at the power of 1500 W is higher than that of other WCAs of as deposited SiO₂ films prepared by other plasma powers, indicating that it is less hydrophilic. This phenomenon is more likely to be associated with the smaller bonding energy, which is in accordance with the result obtained by FTIR. According to findings by Murat Barisik and Beskok, the WCA varies with the variation of the interaction intensity of Si-O [33]. The change of the WCAs for the SiO₂ thin films annealing at different temperatures is shown in Figure 9b. The samples were all prepared with a plasma power of 2000 W. The shapes of the water droplets created on the SiO₂ thin films are also included in the figure. It is interesting to note that the higher the annealing temperature, the greater the WCA. When the SiO₂ thin films were annealed, the WCAs of the SiO₂ thin films sharply increased from 22.8° at 400 °C to 53.5° at 850 °C. The WCA of the thin film annealing at 850 °C is comparable to the value of 50° studied by Eun-Kyeong Kim [34], which was prepared by thermal annealing Si wafer in an oxidizing atmosphere. To further study the change in the surface state of the SiO₂ thin films with various plasma powers, the SFE was also

measured. Figure 10 shows the variation in SFE for five different plasma powers. The SFE was calculated by the Fowkes method without considering the polar force component [35]:

$$\gamma_L(1 + \cos \theta) = 2\sqrt{\gamma_S^d \gamma_L^d}, \quad (1)$$

where θ is the WCA between solid and liquid, γ_L is the surface tension of liquid, and γ_S^d and γ_L^d refer to the dispersion force component of SFE for the solid and liquid. Fowkes method is based on the Young equation. It is derived for a droplet on an ideal surface that is homogeneous, rigid, and smooth. Corresponding to the WCA, the SFE is lower as the WCA is higher. The result is in accordance with the FTIR and the WCA analyses in Figures 8 and 9, respectively.

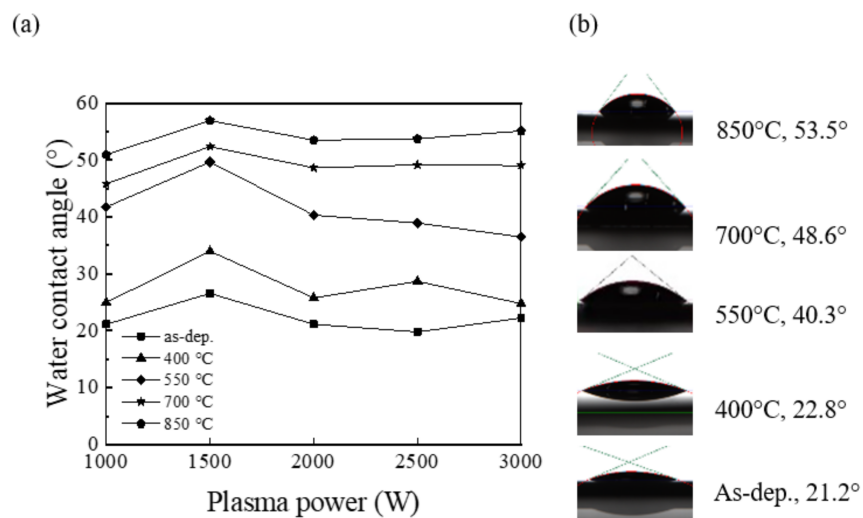


Figure 9. (a) WCA for the RP-ALD SiO₂ films with various plasma powers at different annealing temperatures; (b) WCA for the SiO₂ films when the plasma power is 2000 W at different annealing temperatures.

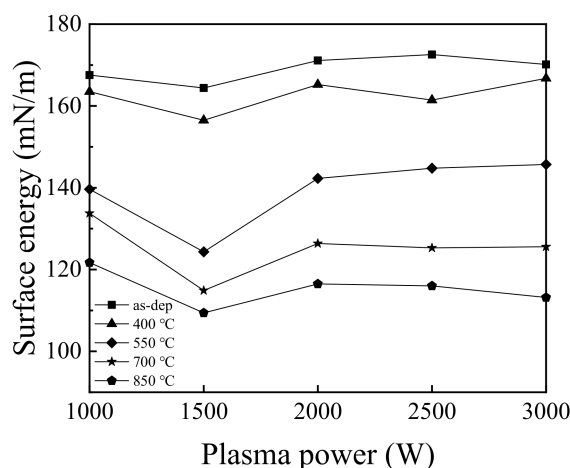


Figure 10. SFE for the RP-ALD SiO₂ films with various plasma powers at different annealing temperatures.

4. Conclusions

In this study, SiO₂ thin films were prepared by remote plasma atomic layer deposition. The effects of the oxygen plasma power and annealing temperature on the SiO₂ properties

were investigated. The oxygen plasma power plays a key role in fabricating the SiO₂ thin films, as they affect the components of the Ar ions and O radicals. The plasma power of 2000 W is the optimum power for preparing high-quality SiO₂ thin films, because at this power adequate oxygen radicals take part in the surface oxidation reaction with less ion bombardment. AFM and CAFM tests show that all the SiO₂ thin films are relatively smooth and have insulating properties. When the annealing temperature of a SiO₂ film is increased, the refractive index increases while the etch rate decreases due to the density of the films. The variation in the bonding energy gives rise to the change in the hydrophilic properties. Therefore, the SiO₂ thin films surface becomes hydrophobic and the surface free energy decreases as the post-annealing temperature increases. This research achievement is greatly beneficial to the application of RP-ALD SiO₂ thin films in future semiconductor-related industries.

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Conflicts of Interest: The authors declare that they have no competing interests.

References

1. Arl, D.; Roge, V.; Adjeroud, N.; Pistillo, B.R.; Sarr, M.; Bahlawane, N.; Lenoble, D. SiO₂ thin film growth through a pure atomic layer deposition technique at room temperature. *R. Soc. Chem.* **2020**, *10*, 18073. [[CrossRef](#)]
2. Davide, B.; Alberto, G.; Chiara, M.; Eugenio, T.; Giberto, R. A soft Plasma Enhanced-Chemical Vapor Deposition process for the tailored synthesis of SiO₂ films. *Thin Solid Films* **2008**, *516*, 7393–7399.
3. Kim, H.-S.; Eom, S.-K.; Seo, K.-S.; Kim, H.; Cha, H.-Y. Time-dependent dielectric breakdown of recessed AlGaIn/GaN-on-Si MOS-HFETs with PECVD SiO₂ gate oxide. *Vacuum* **2018**, *155*, 428–433. [[CrossRef](#)]
4. Viet, H.N.; Abderrahime, S.; Cesar, A.M.D.L.H.; Fadi, Z.; Chiara, C.; Juan, R.-Z.; Moustapha, J.; Marceline, B.; Christophe, V.; Olivier, A.; et al. Atmospheric plasma-enhanced spatial chemical vapor deposition of SiO₂ using trivinylmethoxysilane and oxygen plasma. *Chem. Mater.* **2020**, *32*, 5153–5161.
5. Heun, S.; Kremmer, S.; Erolani, D.; Wurmbauer, H.; Teichert, C. LEEM and XPEEM studies of C-AFM induced surface modifications of thermally grown SiO₂. *J. Electron Relat. Phenom.* **2005**, *144–117*, 1163–1166. [[CrossRef](#)]
6. Chongsawangvirod, S.; Irene, E.A.; Kalnitsky, A.; Tay, S.P.; Ellul, J.P. Refractive index profiles of thermally grown and chemically vapor deposited films on silicon. *J. Electrochem. Soc.* **1990**, *137*, 3536–3541. [[CrossRef](#)]
7. Hill, D.; Blasco, X.; Porti, M.; Aymerich, X. Characterising the surface roughness of AFM grown SiO₂ on Si. *Microelectron. Reliab.* **2001**, *41*, 1077–1079. [[CrossRef](#)]
8. Burton, B.B.; Kang, S.W.; Rhee, S.W.; George, S.M. SiO₂ atomic layer deposition using tris(dimethylamino)silane and hydrogen peroxide studied by in situ transmission FTIR spectroscopy. *J. Phys. Chem. C* **2009**, *113*, 8249–8257. [[CrossRef](#)]
9. Yi, L.; Akiko, K.; Hiroki, K.; Kenji, I.; Makoto, S.; Masaru, H. Chemical reactions during plasma-enhanced atomic layer deposition of SiO₂ films employing aminosilane and O₂/Ar plasma at 50 °C. *Jpn. J. Appl. Phys.* **2014**, *53*, 010305.
10. Matti, P.; Markus, B.; Oili, M.E.Y.; Riikka, L.P.; Lauri, K.; Helena, R.; Sakari, S.; Saima, A.; Harri, L.; Liu, X.; et al. Thermal and plasma enhanced atomic layer deposition of SiO₂ using commercial silicon precursors. *Thin Solid Films* **2014**, *558*, 93–98.
11. Ziegler, M.; Yuksel, S.; Goerke, S.; Cialla-May, D.; Hubner, U.; Wang, D.; Schmidt, H.; Schaaf, P. Controlled synthesis of self-assembled 3D nanostructures using metastable atomic layer deposition. *Mater. Today Chem.* **2018**, *10*, 112–119. [[CrossRef](#)]
12. Choi, Y.-J.; Bae, S.-M.; Kim, J.-H.; Kim, E.-H.; Hwang, H.-S.; Park, J.-W.; Yang, H.; Choi, E.; Hwang, J.-H. Robust SiO₂ gate dielectric thin films prepared through plasma-enhanced atomic layer deposition involving di-sopropylamino silane (DIPAS) and oxygen plasma: Application to amorphous oxide thin film transistors. *Ceram. Int.* **2018**, *44*, 1556–1565. [[CrossRef](#)]
13. Zhou, W.; Zheng, L.; Cheng, X.H.; Zhou, W.J.; Ye, P.Y.; Shen, L.Y.; Zhang, D.L.; Gu, Z.Y.; Yu, Y.H. Plasma-enhanced atomic layer deposition of SiO₂ for channel isolation of colloidal quantum dots phototransistors. *Superlattices Microstruct.* **2019**, *125*, 281–286. [[CrossRef](#)]
14. Li, S.Z.; Xu, J.H.; Wang, L.X.; Yang, N.; Ye, X.J.; Yuan, X.; Xiang, H.T.; Liu, C.; Li, H.B. Effect of post-deposition annealing on atomic layer deposited SiO₂ film for silicon surface passivation. *Mater. Sci. Semicond. Proc.* **2020**, *106*, 104777. [[CrossRef](#)]

15. Merckx, J.M.; Jongen, R.J.; Mameli, A.; Lemaire, P.C.; Sharma, K.; Hausmann, D.M.; Kessels, W.M.M.; Mackus, A.J.M. Insight into the removal and reapplication of small inhibitor molecules during area-selective atomic layer deposition of SiO₂. *J. Vacuum Sci. Technol. A* **2021**, *39*, 012402. [[CrossRef](#)]
16. Guo, S.H.; Zheng, R.; Jiang, J.T.; Yu, J.H.; Dai, K.; Yan, C. Enhanced thermal conductivity and retained electrical insulation of heat spreader by incorporating alumina-deposited graphene filler in nano-fibrillated cellulose. *Compos. Part B* **2019**, *178*, 107489. [[CrossRef](#)]
17. Amr, O.; Abdelmoty, E.; Saleh, K.; Abdalla, A. Thermal, electrical and mechanical properties of graphene/nano-alumina/epoxy composites. *Mater. Chem. Phys.* **2021**, *257*, 123809.
18. Saheed, B.; Duyen, C.; Alex, B.F.M.; Stephen, C. Effects of atomic-layer-deposition alumina on proton transmission through single-layer graphene in electrochemical hydrogen pump cells. *ACS Appl. Energy Mater.* **2020**, *3*, 1364–1372.
19. Abhay, S.; Lars, E.; Patrick, R.W.; David, M.A.; Birong, L.; Peter, B.; Timothy, J.B. Atomic layer deposition alumina-mediated graphene transfer for reduced process contamination. *Phys. Status Solidi RRL* **2019**, *13*, 1900424.
20. Hsu, C.-H.; Cho, Y.-S.; Liu, T.-X.; Chang, H.-W.; Lien, S.-Y. Optimization of residual stress of SiO₂/organic silicon stacked layer prepared using inductively coupled plasma deposition. *Surf. Coat. Technol.* **2017**, *320*, 293–297. [[CrossRef](#)]
21. Lien, S.-Y.; Cho, Y.-S.; Hsu, C.-H.; Shen, K.-Y.; Zhang, S.; Wu, W.-Y. Mechanism of dense silicon dioxide films deposited under 100 °C via inductively coupled plasma chemical vapor deposition. *Surf. Coat. Technol.* **2019**, *359*, 247–251. [[CrossRef](#)]
22. Zhu, Z.; Modanese, C.; Sippola, P.; Sabatino, M.D.; Savin, H. Nanometer-scale depth-resolved atomic layer deposited SiO₂ thin films analyzed by glow discharge optical emission spectroscopy. *Phys. Status Solidi A* **2017**, *215*, 1700864. [[CrossRef](#)]
23. Lien, S.-Y.; Chang, Y.-Y.; Cho, Y.-S.; Wang, J.-H.; Weng, K.-W.; Chao, C.-H.; Chen, C.-F. Characterization of HF-PECVD a-Si:H thin film solar cells by using OES studies. *J. Non-Cryst. Solids* **2011**, *357*, 161–164. [[CrossRef](#)]
24. Lee, C.; Lieberman, M.A. Global model of Ar, O₂, Cl₂, and Ar/O₂ high-density plasma discharges. *J. Vacuum Sci. Technol. A* **1995**, *13*, 368–380. [[CrossRef](#)]
25. Ma, H.-P.; Yang, J.-H.; Yang, J.-G.; Zhu, L.-Y.; Huang, W.; Yuan, G.-J.; Feng, J.-J.; Jen, T.-C.; Lu, H.-L. Systematic study of the SiO_x film with different stoichiometry by plasma-enhanced atomic layer deposition and its application in SiO_x/SiO₂ super-lattice. *Nanomaterials* **2019**, *9*, 55. [[CrossRef](#)] [[PubMed](#)]
26. Hsu, C.-H.; Zhang, Z.-X.; Huang, P.-H.; Wu, W.-Y.; Ou, S.-L.; Lien, S.-Y.; Huang, C.-J.; Lee, M.-K.; Zhu, W.-Z. Effect of plasma power on the structural properties of tin oxide prepared by plasma-enhanced atomic layer deposition. *Ceram. Int.* **2021**, *47*, 8634–8641. [[CrossRef](#)]
27. Lopez, F.; Bernabeu, E. Refractive index of vacuum-evaporated SiO thin films: Dependence on substrate temperature. *Thin Solid Films* **1990**, *191*, 13–19. [[CrossRef](#)]
28. Lee, S.; Kim, J.M.; Kim, C.; Kim, H.; Kang, H.J.; Ha, M.-W.; Kim, H.J. Densification of silicon dioxide formed by plasma-enhanced atomic layer deposition of 4H-silicon carbide using argon post-deposition annealing. *Ceram. Int.* **2018**, *44*, 13565–13571. [[CrossRef](#)]
29. Knotter, D.M. Etching Mechanism of Vitreous Silicon Dioxide in HF-Based Solutions. *J. Am. Chem. Soc.* **2000**, *122*, 4345–4351. [[CrossRef](#)]
30. Patrick, F.; Wouter, P.; Wilfried, V. Conductive atomic force microscopy studies of thin SiO₂ layer degradation. *Appl. Phys. Lett.* **2006**, *88*, 222104.
31. Iglesias, V.; Porti, M.; Nafria, M.; Aymerich, X.; Dudek, P.; Schroeder, T.; Bersuker, G. Correlation between the nanoscale electrical and morphological properties of crystallized hafnium oxide-based metal oxide semiconductor structures. *Appl. Phys. Lett.* **2010**, *97*, 262906. [[CrossRef](#)]
32. Nagashio, K.; Yamashita, T.; Nishimura, T.; Kita, K.; Toriumi, A. Electrical transport properties of graphene on SiO₂ with specific surface structures. *J. Appl. Phys.* **2011**, *110*, 024513. [[CrossRef](#)]
33. Murat, B.; Ali, B. Wetting characterization of silicon (1,0,0) surface. *Mol. Simul.* **2013**, *39*, 700–709.
34. Kim, E.-K.; Kim, J.-Y.; Kim, S.S. Significant change in water contact angle of electro-spray-synthesized SiO₂ films depending on their surface morphology. *Surf. Interface Anal.* **2012**, *45*, 656–660. [[CrossRef](#)]
35. Li, N.N.; Li, G.L.; Wang, H.D.; Kang, J.J. Research progress of surface free energy's computing methods and the influence on the properties of material surface. *Mater. Rep.* **2015**, *29*, 30–40.