Base-displaced intercalation of the 2-amino-3-methylimidazo[4,5-*f*]quinolone *N*²-dG adduct in the *Nar*I DNA recognition sequence

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ABSTRACT

2-Amino-3-methylimidazo[4,5-f]quinolone (IQ), а heterocyclic amine found in cooked meats, undergoes bioactivation to a nitrenium ion, which alkylates guanines at both the C8-dG and N^2 -dG positions. The conformation of a site-specific N^2 -dG-IQ adduct in an oligodeoxynucleotide duplex containing the iterated CG repeat restriction site of the Narl endonuclease has been determined. The IQ moiety intercalates, with the IQ H4a and CH₃ protons facing the minor groove, and the IQ H7a, H8a and H9a protons facing the major groove. The adducted dG maintains the anti-conformation about the glycosyl bond. The complementary dC is extruded into the major groove. The duplex maintains its thermal stability, which is attributed to stacking between the IQ moiety and the 5'- and 3'-neighboring base pairs. This conformation is compared to that of the C8-dG-IQ adduct in the same sequence, which also formed a 'basedisplaced intercalated' conformation. However, the C8-dG-IQ adopted the syn conformation placing the Watson-Crick edge of the modified dG into the major groove. In addition, the C8-dG-IQ adduct was oriented with the IQ CH₃ group and H4a and H5a facing the major groove. These differences may lead to differential processing during DNA repair and replication.

INTRODUCTION

Although browning meats during cooking imparts flavor, it also leads to the formation of heterocyclic amines (HCAs) such as 2-amino-3-methylimidazo[4,5-f]quinoline (IQ) (1–5). IQ has been produced in cooked meats at ppb levels (6,7) and has also been detected in tobacco

smoke (8). HCAs and their metabolites have been isolated from human urine (9). Human exposures to HCAs, estimated to be $\sim 60 \text{ ng/day}$ (10), are modest, but are likely to be involved in cancer etiology (11,12).

IQ is a potent mutagen (13). It is less prevalent than 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) (14), but it is 200-fold more mutagenic in Salmonella reversion (Ames) assays (3) and it is an order of magnitude more mutagenic than aflatoxin B_1 . In these assays (15–18), HCAs such as IQ are active in point and frameshift tester strains (19). In bacteria, mutations occur primarily at G:C base pairs (20,21). IQ is a potent inducer of two-base frameshifts in CG repeats. Similar levels of mutations are seen in mammalian hprt (22) and ef-2 (23) gene assays. Base-pair substitutions are the predominant mutations observed in mammalian cells (24-26). Sister chromatid exchange has been observed in rodent cells (27-29). IQ induces tumors in the organs of rodents and in the livers of monkeys (30–33). Liver, forestomach and lung tumors have been observed in IO treated mice (34), whereas liver, intestine, zymbal gland, clitoral gland, skin (35), mammary glands, liver and ear ducts tumors have been observed in exposed rats (36). The TD_{50} values for IQ are 0.7 and 14.7 mg/kg/day in rats and mice, respectively (37). Human exposures to HCAs have been associated with pancreatic (38), colon (39), prostate (40) and breast cancers (41,42).

The genotoxicity of IQ derives primarily from its oxidation by CYP P450 1A2 to an *N*-hydroxylamine (43–47) although extra-hepatic CYP P450s oxidize HCAs with lower efficiencies (Scheme 1) (48). The *N*-hydroxylamine is acetylated by *N*-acetyl transferases, particularly NAT2 (49–51). In humans, the *NAT2* fast acetylator polymorphism correlates with increased genotoxicity and cancer (52–54). The nitrenium ion resulting from solvolysis of *N*-acetoxy-IQ is the ultimate electrophile (26,48). It reacts predominately at the C8 atom of guanine, while a minor alkylation product is formed at the N^2 atom of guanine (55–57). In addition, IQ can be converted to a

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Scheme 1. The bioactivation of IQ. Cytochrome P450-mediated *N*-hydroxylation followed by *N*-acetylation of the hydroxylamine and deacetylation forms an electrophilic aryl nitrenium ion. The aryl nitrenium ion alkylates guanine in DNA either via the IQ amine nitrogen to form the C8-dG-IQ adduct, or alternatively, alkylates DNA via the C5 position of the IQ ring to form the N^2 -dG-IQ adduct.

reactive and genotoxic *N*-nitrosamine, which shows similar regioselectivity for DNA alkylation (58,59).

Levels of C8 and N^2 -dG-IQ adducts have been measured in rat and primate tissues using mass spectrometry (60,61) and ³²P post-labeling methodology. Turesky *et al.* (62) have monitored C8-dG-IQ adduct formation in human hepatocytes using tandem liquid chromatographyelectrospray ionization mass spectrometry. Levels range from 7 to 26 adducts per 10⁷ bases. While less abundant, the N^2 -dG-IQ adduct is more persistent than is the C8-dG-IQ adduct, suggesting that it is repaired less efficiently (63). The N^2 -dG-IQ adduct may therefore play a significant role in the genotoxicity of IQ.

We have synthesized phosphoramidite reagents of the C8- and N^2 -dG-IQ adducts in which the Buchwald-Hartwig palladium-catalyzed N-arylation was the key C-N bonding-forming step (64-67). These adducts have been site-specifically incorporated into the NarI restriction sequence, 5'-d(CG¹G²CX³CC)-3', using automated solidphase synthesis. The G^3 position is a hot spot for two-base frameshift deletions in bacterial mutagenesis assays, while the G^1 position is not (68–71). In addition, human DNA polymerase (hpol) n produces two-base deletions when replicating past the N^2 -dG-IQ adduct at position G^3 , in vitro (72). Thus, this sequence provides a platform for investigating sequence-specific conformational perturbation of DNA structure by IQ adducts, in relationship to their biological processing (68-71,73,74). Previously, we determined the conformation of the C8-dG-IQ adduct at the G^3 position of the *NarI* sequence, which exhibited a base-displaced intercalated conformation (75).

Presently, we have determined the conformation of the N^2 -dG-IQ adduct in 5'-d(C¹T²C³G⁴G⁵C⁶X⁷C⁸C⁹ A¹⁰T¹¹C¹²)-3':5'-d(G¹³A¹⁴T¹⁵G¹⁶G¹⁷C¹⁸G¹⁹C²⁰C²¹G²²A²³ G^{24})-3'; $X = N^2$ -dG-IQ adduct (Chart 1). This duplex contains the recognition sequence of the NarI restriction endonuclease, in which the G^3 nucleotide (X³ in the NarI sequence and X^7 in this study) represents a hot spot for two-base deletions. The IQ moiety intercalates, with the IO H4a and CH₃ protons facing the minor groove, and the IQ H7a, H8a and H9a protons facing the major groove. The adducted nucleotide maintains the anti-conformation about the glycosyl bond. The complementary dC is extruded into the major groove. Nevertheless, the duplex maintains its thermal stability. This is attributed, in part, to stacking between the IO moiety and the 5'- and 3'-neighboring base pairs. The base-displaced intercalated conformation of the N^2 -dG-IQ adduct differs from that of the C8-dG-IQ adduct, and may also be compared with that of the N^2 -dG AAF adduct (76), providing insight as to the persistence of the N^2 -dG-IQ adduct (63) and its processing during replication and repair events.

MATERIALS AND METHODS

Sample preparation

The N^2 -dG-IO-adducted oligodeoxynucleotide 5'-d(CTC GGCXCCATC)-3' was synthesized as described (67). The complement strand 5'-d(GATGGCGCCGAG)-3' was synthesized by Midland Certified Reagents Co. (Midland, TX, USA) and purified by anion exchange chromatography. HPLC chromatographic utilized a Supelcosil LS-18-DB analytical base-deactivated C-18 column (Sigma-Aldrich Inc., St. Louis, MO, USA), using a gradient from 5% to 12% acetonitrile in ammonium formate buffer (pH 7), over 35 min, or a gradient from 5% to 30% acetonitrile in NaH₂PO₄ buffer (pH 5), over 25 min. The oligodeoxynucleotides were characterized by negative mode MALDI-TOF spectrometry in a hydroxypicolinic acid matrix. The oligodeoxynucleotides were annealed at 1:1 ratio at room temperature in 180 µl buffer containing 10 mM NaH₂PO₄, 100 mM NaCl and 5 µM Na₂EDTA (pH 7).

Thermal melting experiments

UV melting temperatures were collected on Cary 100 Bio UV spectrometer using 0.5 OD of duplex in 1 ml of solution containing 0.1 M NaCl, 10 mM NaH₂PO₄ and 0.05 mM Na₂EDTA (pH 7.0). The temperature was increased from 25 to 75°C at a rate of 1°C per min.

NMR spectroscopy

The N^2 -dG-IQ modified and the unmodified duplexes were prepared at concentrations of 570 and 810 μ M, respectively, and placed into 3 mm diameter micro NMR sample tubes (77). The samples were prepared in 0.1 M NaCl, 50 μ M Na₂EDTA and 10 mM NaH₂PO₄ (pH 7.0). To observe non-exchangeable protons, the samples were Α

С



2-Amino-3-methylimidazo[4,5-f]quinoline (IQ)





N²(2-amino-3-methyl-3H-imidazo[4,5-f]quinol-5-yl)-dG



C8-[(3-methyl-3H-imidazo[4,5-f]quinolin-2-yl)amino]-dG

Ε

3-(deoxyguanosin-N²-yl)-2-acetylaminofluorene Adduct

в

D

Narl Oligodeoxynucleotide Sequence

Chart 1. (A) Structure of IQ. (B) Structure of the N^2 -dG-IQ adduct, showing the numbering of guanine base and IQ protons. (C) Structure of the C8-dG-IQ adduct. (D) Structure of the N^2 -dG-AAF adduct. (E) The duplex containing the NarI sequence, showing the numbering of the nucleotides. The N^2 -dG-IQ adduct is positioned at X⁷, which corresponds to the G³ frameshift-prone position of the *NarI* sequence.

exchanged with D₂O. ¹H NMR spectra were recorded at 600 or 800 MHz. The spectra were collected at 15° C; NOESY experiments were conducted at mixing times of 150, 200 and 250 ms with a relaxation delay of 1.8 s. Additional experiments were conducted with a longer relaxation delay to evaluate NOE distances arising from the adenine H2 protons, which typically exhibit longer T_1 relaxation values. The data were collected with 512 points in the t1 dimension and 2048 points in the t2 dimension. Chemical shifts were referenced to water. For the observation of exchangeable protons, the samples were dissolved in 9:1 H₂O:D₂O. ¹H NMR spectra were recorded at 600 or 800 MHz at 15°C. The data were collected with 512 points in the t1 dimension and 2048 points in the t2 dimension. A mixing time of 250 ms was used. Water suppression was performed using the WATERGATE pulse sequence (78). The spectra were processed using the TOPSPIN software (Bruker Biospin Inc., Billerica, MA, USA).

NMR experimental restraints

The spectral data were evaluated using the program SPARKY (79). The intensities of NOE cross peaks were measured by volume integrations. The bounds for overlapped peaks were optimized manually. Noise was assigned half the intensity of the weakest peak, and motion was assumed to be isotropic. Experimental intensities were combined with intensities obtained from complete relaxation matrix analysis (CORMA) of starting

model to generate a hybrid intensity matrix (80,81). The intensities were converted to distances with the program MARDIGRAS (82), which refined the hybrid intensity matrix. Calculations were performed using 150, 200 and 250 ms mixing time data and 2, 3 and 4 ns isotropic correlation times. Evaluation of the resulting distance data allowed creation of upper and lower bound distance restraints that were used in restrained molecular dynamics (rMD) calculations.

Restrained molecular dynamics calculations

An unmodified B-DNA model (83) was used as a starting structure. The guanine at position G^7 was replaced by the N^2 -dG-IQ adduct using the program INSIGHT II (Accelrys Inc., San Diego, CA, USA). Partial charges for the N^2 -dG-IQ adduct were calculated with the B3LYP/6-31G* basis set in GAUSSIAN (84). The starting structure was energy minimized for 1000 cycles. Simulated annealing protocols (85) used for the rMD calculations were conducted with the parm99 force field (86), using the program AMBER (87). Force constants of 32 kcal/mol/Å^2 were applied for distance and torsion angle restraints. The generalized Born model (88) was used for solvation. The salt concentration was 0.1 M. The molecule was coupled to the bath to control the temperature during simulated annealing (89). First, calculations were performed for 20 ps (20 000 steps) and recording data every ps by the following protocol: during steps 0-1000, the system was heated from 0 to 600 K with a coupling of 0.5 ps. During steps 1001-2000, the system was kept at 600 K. The system was then cooled from 600 to 100 K during steps 2001-18000 with a coupling of 4 ps. Further cooling from 100 to 0 K occurred during steps 18001-20000 with a coupling of 1 ps. After initial cycles of refinement a longer 100 ps (100 000 steps) calculation was performed by the following protocol: During steps 0-5000, the system was heated from 0 to 600 K with a coupling of 0.5 ps. During steps 5001-10000, the system was kept at 600 K. The system was cooled from 600 to 100 K during steps 10001-90000 with a coupling of 4 ps. Additional cooling from 100 to 0 K occurred during steps 90 001-100 000 with a coupling of 1 ps. Structure coordinates were saved after each cycle and were subjected to potential energy minimization. CORMA (80.81) was used to compare intensities calculated from these emergent structures with the distance restraints. Helicoidal analysis was performed using the CURVES+ web server (90,91).

RESULTS

Oligodeoxynucleotide containing the N^2 -dG-IQ adduct

The N^2 -dG-IQ adduct was incorporated into 5'-d(CTCGG CXCCATC)-3' using automated solid-phase synthesis (67). The position of the N^2 -dG-IQ adduct was located at position X^7 , corresponding to position G^3 in the NarI sequence. The modified oligodeoxynucleotide was purified by C18 reverse phase HPLC and characterized by MALDI-TOF mass spectrometry in negative ion mode [m/z 3777.7,calcd for (M - H), 3776.6]. Thermal melting (T_m) profiles of $0.5 A_{260}$ units of the IQ-modified duplex were monitored at 100 mM NaCl (1 ml volume) as a function of temperature by absorbance at 260 nm. An unmodified duplex was evaluated under the same conditions to provide a basis of comparison. The $T_{\rm m}$ of the modified duplex was 63°C, within experimental error of the unmodified duplex (Supplementary Figure S1). Thus, the N^2 -dG-IQ adduct did not reduce the stability of this oligodeoxynucleotide. This result differed from our previous report, which had indicated that this adduct destabilized this duplex (67). Subsequent analysis of the previous sample by mass spectrometry revealed that the complement strand was not correct, accounting for the discrepancy. Table 1 lists the correct $T_{\rm m}$ values of the N²-dG-IQ adduct at the three positions of the NarI sequence.

NMR

The modified duplex yielded well-resolved NMR spectra with narrow line shapes for the non-exchangeable protons at 15° C. The best spectral quality for the exchangeable protons was obtained at 5° C.

Non-exchangeable DNA protons

The base aromatic and deoxyribose anomeric protons were assigned using established procedures (Figure 1) (92,93). The intensity of the X^7 H8 to X^7 H1' NOE was not changed in the presence of the adduct, indicating minimal change in the conformation of the glycosyl torsion angle. In the complementary strand, the intensity of the NOE

Table 1. Thermal melting temperatures (T_m measurements) of NarI duplexes containing the N^2 -dG-IQ adducts

NarI N ² -dG-IQ modified duplex	$T_{\rm m}~(^{\circ}{\rm C})$	$\Delta T_{\rm m}^*$ (°C)
5'-CTCXGCGCCATC-3' 3'-GAGCCGCGGTAG-5'	62	-1
5'-CTCG <u>X</u> CGCCATC-3' 3'-GAGCCGCGGTAG-5'	64	+1
5'-CTCGGCXCCATC-3' 3'-GAGCCGCGGTAG-5'	63	0

between C^{18} H1' and G^{19} H8 was weakened. The N^2 -dG-IQ adduct did not induce breaks in the sequential pattern of NOEs between the aromatic base protons and the anomeric protons. With the exception of the adduct site, the internucleotide NOEs were characteristic of a B-type duplex. The adenine H2 protons were assigned based upon NOEs to the thymine imino protons of the respective A:T base pairs. With the deoxyribose H1' assignments in hand, the remainder of the deoxyribose protons was assigned from a combination of NOESY and COSY data. The assignments of the non-exchangeable DNA protons are summarized in Supplementary Table S1.

Exchangeable DNA protons

The imino and amino proton regions of the NOESY spectrum are shown in Figure 2. The assignments were made using established methods (94). The N^2 -dG-IQ adduct perturbed Watson–Crick hydrogen bonding. At the $X^7{:}C^{18}$ base pair, the X^7 imino proton resonance was broadened, probably due to an enhanced rate of exchange with water. The amino protons for C^{18} were not detected. No NOE was observed between the X^7 and G^{19} imino protons, perhaps due to the broadening of the X^7 imino proton. The chemical shifts of the X^7 and G^{17} imino protons were almost isochronous. It was not possible to determine if a NOE between these two protons existed. All other base pairs were assigned, with the exception of the two terminal base pairs $C^{1}:G^{24}$ and $C^{12}:G^{13}$. The imino protons from the terminal base pairs were exchange broadened. Overall, the data suggested that the duplex maintained Watson-Crick hydrogen bonding, with the exception of the modified base pair (Figure 2B). The assignments of the exchangeable protons are summarized in Supplementary Table S2.

IQ protons

The IQ protons, consisting of the CH₃ group, the H4a proton, and the H7a, H8a and H9a spin system, were assigned from a combination of COSY and NOESY data (Figure 3). The CH₃ resonance was observed at 3.57 ppm. It displayed an intense NOE to the H4a proton, whose resonance was observed at 8.55 ppm. A ³J coupling between the H8a proton (δ 6.55 ppm) and the H9a proton (δ 7.65 ppm) was observed in the COSY spectrum. The H8a proton also exhibited an NOE to the H7a proton (δ 7.6 ppm). The ³J coupling between H8 and H7a exhibited weak intensity in the COSY spectrum. This was attributed to presence of the nitrogen atom in the ring, which broadened the H7a resonance. This effect



Figure 1. Expanded plot of the 250 ms NOESY spectrum showing NOEs between the base aromatic and deoxyribose anomeric protons of the N^2 -dG-IQ modified duplex. (A) Bases C¹ to C¹² of the modified strand. (B) Bases G¹³ to G²⁴ of the complementary strand. The spectrum was acquired at 800 MHz at 15°C.



Figure 2. Expanded plots of the NOESY spectrum, showing the NOEs between the exchangeable imino and amino protons of the N^2 -dG-IQ-modified duplex. (A) Sequential NOE connectivity for the imino protons of base pairs $T^2:A^{23}$ to $T^{11:}A^{14}$. (B) NOE connectivity within Watson–Crick base pairs and between the imino protons and the amino protons. The lettered cross-peaks are assigned as follows: a', $X^7 N^2 H \rightarrow C^6 N^4 Hb$; a, $X^7 N^2 H \rightarrow C^6 N^4 Ha$; b', $G^{17} N1H \rightarrow C^8 N^4 Ha$; b, $G^{17} N1H \rightarrow C^8 N^4 Hb$; c', $G^{19} N1H \rightarrow C^6 N^4 Hb$; c, $G^{19} N1H \rightarrow C^6 N^4 Ha$; d', $G^{16} N1H \rightarrow C^9 N^4 Hb$; d, $G^{16} N1H \rightarrow C^9 N^4 Hb$; e, $G^{22} N1H \rightarrow C^3 N^4 Hb$; e, $G^{23} N^2 H^2$; j, $T^2 N^3 H \rightarrow A^{23} H^2$; 1, $X^7 N^2 H \rightarrow IQ H4a$; 2, $IQ H7a \rightarrow X^7 N^2 H$; 3, $X^{19} N1H \rightarrow IQ H4a$. The spectrum was collected at 800 MHz at 5°C.

was also observed for the C8-dG-IQ adduct, for which the COSY cross-peak between H7a and H8a was only observed between 25 and 35° C (95). The IQ amine proton was not assigned.

Chemical shift perturbations

The N^2 -dG-IQ adduct resulted in localized chemical shift perturbations, involving the modified base pair X^7 :C¹⁸ and the neighboring C⁶:G¹⁹ and C⁸:G¹⁷ base pairs

(Figure 4). At the modified $X^{7}:C^{18}$ base pair, the X^{7} H8 resonance shifted 0.4 ppm downfield relative to the G^{7} H8 resonance in the unmodified duplex. In contrast, the C^{18} H6 and C^{18} H1' resonances shifted 1 and 0.8 ppm downfield, respectively. At the 5'-neighbor $C^{6}:G^{19}$ base pair, the C^{6} H6 resonance shifted upfield by 0.2 ppm, whereas the C^{6} H1' resonance shifted downfield by 0.4 ppm. The G^{19} H8 and H1' resonances each shifted upfield by 0.4 ppm. At the 3'-neighbor $C^{8}:G^{17}$ base pair,



Figure 3. Expanded tile plot of the 250 ms NOESY spectrum showing the assignments of the IQ ring protons. (A) The IQ H4a proton was observed at 8.55 ppm. (B) The IQ H7a proton was observed at 7.65 ppm. (C) The IQ H9a proton was observed at 7.65 ppm. (D) The IQ H8a proton was observed at 6.55 ppm. (E) The IQ CH₃ protons were observed at 3.57 ppm. The spectrum was collected at 800 MHz at 15° C.

the C⁸ H6 resonance shifted downfield by 0.3 ppm, whereas the C⁸ H1' resonance shifted upfield by 0.2 ppm. The G¹⁷ H8 resonance shifted downfield by 0.1 ppm. The G¹⁷ H1' resonance showed negligible chemical shift perturbation. The resonances for the remaining base pairs in the duplex also showed negligible chemical shift perturbations. In the imino proton region of the spectrum, the X⁷ and G¹⁷ N1H imino resonances, at 11.57 and 11.59 ppm, respectively, exhibited upfield chemical shifts of >1 ppm from those of the unmodified duplex, at 13.24 and 13.16 ppm, respectively. The X⁷ N²H amine resonance was observed at 9.5 ppm.

NOEs between IQ and DNA

The CH₃, H4a, H7a and H8a protons of IQ exhibited NOEs to the C⁸, G¹⁷, C¹⁸ and G¹⁹ bases (Table 2). The pattern of NOEs involving H9a was difficult to establish due to resonance overlap with G¹⁷. The CH₃ group showed medium strength NOEs to X⁷ H1' and C⁸ H6 and weak NOEs to C⁸ H1' and C⁸ H5. It also showed an NOE to the G¹⁹ N1H imino proton. The H4a proton showed a strong NOE to X⁷ H1', and medium NOEs to C⁸ H5, the X⁷ N2H amine proton and the G¹⁹ N1H imino proton exhibited weak NOEs to X^7 H2' and X⁷ H2'' The H7a proton exhibited weak NOEs from X⁷ to C¹⁸ were observed; these were of medium strength NOE between H9a and C¹⁸ H1', and H8a and medium strength between C¹⁸ H2' and H2''. The H8a proton showed a medium strength NOE to G¹⁹ H3' and weak NOEs to G¹⁷ H1' and G¹⁹ H8. Some 30% of the NOEs from the IQ ring were to protons in the complementary strand, whereas another 46% were to other IQ protons and protons of the modified base. The remaining 24% of the NOEs were to neighbor bases in the modified strand.



Figure 4. Chemical shift perturbations of the deoxyribose H1' protons (gray) and the pyrimidine H6 or purine H8 aromatic protons (black), for the N^2 -dG-IQ modified duplex. (A) Nucleotides $C^{1}-C^{12}$ in the modified strand. (B) Nucleotides $G^{13}-G^{24}$ in the complementary strand. The $\Delta\delta$ (ppm) values were calculated as $\delta_{\text{modified duplex}}-\delta_{\text{unmodified duplex}}$. Positive $\Delta\delta$ values represent upfield chemical shift perturbations. Negative $\Delta\delta$ values represent downfield chemical shift perturbations.

Conformational Refinement

After the unmodified duplex was constructed using B-DNA coordinates (83), the guanine at position G^7 was replaced by the N^2 -dG-IQ adduct. The partial charges for the N^2 -dG-IO adduct are provided in Supplementary Figure S2. Potential energy minimization provided an energy minimized starting duplex. A total of 329 distance restraints consisting of 127 inter- and 202 intranucleotide distances (Table 3) were obtained using the program MARDIGRAS (81,82), from 15°C NOESY data. Similar distance restraints were obtained if the data were collected at 150, 200 or 250 ms mixing times. These restraints included 16 DNA-IQ distances. A total of 49 Watson-Crick hydrogen-bonding restraints were applied for all of the base pairs except for the modified \hat{X}^{7} :C¹⁸ base pair. An additional 100 phosphodiester backbone and 20 deoxyribose pseudorotation restraints for base pairs not proximal to the site of modification were obtained from canonical values derived from B-DNA (83), consistent with the spectroscopic data indicating that the duplex maintained a B-DNA like structure. A series of rMD calculations were performed using a simulated annealing protocol in which the generalized Born solvation model (88) was used, with a salt concentration of 0.1 M. The emergent structures were subjected to potential energy minimization before further analysis, which involved a 100 ps rMD calculation using the protocol described above, again followed by potential energy minimization.

The pairwise rmsd analysis of structures emergent from the rMD calculations was used to measure the precision of the structural refinement. Ten structures were chosen

IQ proton	NOEs to oligodeoxynucleotide protons
CH ₃	X^7 H1': medium; C^8 H6: medium; C^8 H1': weak; C^8 H5: weak
H4a	X^7 H1': strong; C^8 H5: medium; X^7 H2': weak; X^7 H2'': weak; X^7 N2H: medium; G^{19} N1H: medium
H7a	G^{17} H1': weak; X^7 H2: weak
H8a	G^{19} H3': medium; C^{18} H2': medium; C^{18} H2'': medium; G^{17} H1': weak; G^{19} H8: weak; G^{17} H2'': weak
H9a	C^{18} H1': medium

Table 2. Summary of NOEs observed between N^2 -dG-IQ(X7) adduct protons and oligodeoxynucleotide protons and their intensities

Table 3. NMR restraints used for the N^2 -dG-IQ structure calculations and refinement statistics

NOE restraints	
Internucleotide	127
Intranucleotide	202
Total	329
Backbone torsion angle restraints	100
H-bonding restraints	49
Deoxyribose restraints	20
Total number of restraints	498
Refinement statistics	
Number of distance restraint violations	56
Number of torsion restraint violations	50
Total distance penalty/maximum penalty (kcal/mol)	2.3/0.18
Total torsion penalty/maximum penalty (kcal/mol)	2.8/0.17
r.m.s. distances (Å)	0.012
r.m.s. angles (°)	2.5
Distance restraint force field (kcal/mol/Å ²)	32
Torsion restraint force field (kcal/mol/deg ²)	32

based on the lowest deviations from the experimental distance and dihedral restraints (Figure 5). These exhibited an rmsd of 0.012 Å in distances and 2.5° in torsion angles (Table 3). There were 56 distance violations with a maximum penalty of 0.187 kcal/mol and a total distance penalty of 2.3 kcal/mol. There were 50 torsion angle violations with a maximum penalty of 0.177 kcal/mol and a total torsion angle penalty of 2.8 kcal/mol. The maximum pairwise rmsd distances were 1.12 Å. These structures were averaged and subjected to potential energy minimization.

The accuracy of the refined structures was assessed by complete relaxation matrix analyses (80,81), which compared intensities calculated from the refined structures with the distance restraints (Figure 6). The sixth root residual R_x^1 value of the average structure was 8.4%, and the individual values for intra-nucleotide restraints (8.5%) and inter-nucleotide restraints (8.3%) were of similar magnitudes. This indicated agreement with the NOE data. Nucleotide G¹⁹ exhibited a greater R_x^1 value of 17.1%, suggesting that it was not as well-refined. This was attributed to several NOEs involving G¹⁹ being overlapped with other resonances. The structural statistics are summarized in Table 4.

Conformation of the N^2 -dG-IQ Adduct

The modified nucleotide (X^7) remained in the anti-conformation about the glycosyl bond. It was displaced toward the major groove. The IQ ring was intercalated



Figure 5. Superposition of ten potential energy minimized structures emergent from the rMD calculations of the N^2 -dG-IQ modified duplex, using distance restraints from the 250 ms NOESY data. The positions of the modified X^7 nucleotide and the C¹⁸ nucleotide in the complementary strand are as indicated. The maximum pairwise rmsd between these 10 structures was 1.12 Å.

and oriented such that the H4a proton and the CH₃ group faced into the minor groove, whereas the H7a, H8a and H9a protons faced into the major groove (Figure 7). The IQ ring was angled by $\sim 15^{\circ}$ with respect to the modified guanine, but otherwise remained largely in plane with the damaged base. The helix was unwound between C⁶ and X⁷, with a reduced helicoidal twist of 30°. This was partially compensated by an increased twist of 9° between X^7 and C^8 . At base pair $X^7:C^{18}$, the roll of the X^7 purine decreased by 24° . This was compensated at base pair $C^8:G^{17}$, where the roll decreased by 12° . Consequently, the N²-dG-IQ adduct induced a bend of 10° to the duplex. The IO ring exhibited stacking with the flanking base pairs (Figure 8). IQ was stacked between G^{17} and G^{19} of the complementary strand of the $C^6:G^{19}$ and $C^8:G^{17}$ base pairs. The complementary nucleotide, C^{18} , extruded into the major groove and did not exhibit stacking with the neighboring base pairs. The base opening between X^7 and C^{18} increased



Figure 6. Sixth root residuals (R^1_x) calculated using complete relaxation matrix calculations from the average of 10 structures emergent from the rMD calculations of the N^2 -dG-IQ modified duplex. The black bars represent intra-nucleotide sixth root residuals, and the gray bars represent internucleotide sixth root residuals. (A) Nucleotides C^1-C^{12} in the modified strand. (B) Nucleotides $G^{13}-G^{24}$ in the complementary strand.

Table 4.	Structural	statistics	for	the	N ² -dG-IQ	modified	duplex
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Average structure (obtained	from 10 structures)		
RMS pairwise difference be RMS difference from avera	etween structures ge structure	1.12 0.75	
	COF	RMA analysis for average structure ^a	
	Intranucleotide	Internucleotide	Total
$R^{1}_{x}^{b}$ Average error ^c	0.085	0.083	0.084 0.0037

^aThe mixing time was 250 ms.

^bR¹_x is the 6th root R factor: $\sum \left[\left(\left(I_o \right)_i^{1/6} \right) - \left(\left(I_c \right)_i^{1/6} \right) / \sum \left(\left(I_o \right)_i^{1/6} \right) \right].$

^cAverage error: $\sum (I_c - I_o)/n$, where the I_c are NOE intensities calculated from the refined structure, the I_o are experimental NOE intensities.

by 76°. This disrupted Watson–Crick hydrogen bonding. The other base pairs maintained Watson–Crick hydrogen bonding.

DISCUSSION

The N^2 -dG-IQ DNA adduct has been of interest following reports that it is more persistent than the C8-dG-IQ adduct in rodents and primates that were fed IQ in their diet (63). The synthesis of this adduct into oligodeoxynucleotides (67) has allowed the conformation of the N^2 -dG-IQ adduct at the G³ position of this sequence to be determined. This is a hot spot for two-base frameshift deletions in bacterial mutagenesis assays (68–71,73,74). In addition, human DNA polymerase (hpol) η produces two-base deletions when replicating past the N^2 -dG-IQ adduct at the reiterated G³ position of the NarI sequence, *in vitro* (72).

Conformation of the N^2 -dG-IQ adduct

The IQ ring intercalates when the N^2 -dG-IQ adduct is positioned at the frameshift-prone G³ position of the NarI sequence (Figure 7). The strong NOE intensities of the IQ H4a and CH₃ protons to the X⁷ and C⁸ H1'



Figure 7. Expanded view of the average structure calculated from 10 structures emergent from the rMD calculations of the N^2 -dG-IQ(X⁷) modified duplex, showing base pairs C⁶:G¹⁹, X⁷:C¹⁸ and C⁸:G¹⁷. The view is from the major groove. The modified base pair X⁷:C¹⁸ is shown in magenta, with the IQ moiety shown in blue.

protons (Table 1) indicate that these protons face into the minor groove and establish the conformation about the bond between N^2 -dG and C5 of the IQ moiety. In contrast, NOEs involving the H8a proton of the IQ ring are primarily to bases G^{17} , C^{18} and G^{19} of the complementary strand (Table 1). The chemical shifts of the IQ H7a, H8a and H9a protons are observed between 6.5 and 8.0 ppm, which is 1.3–2.0 ppm upfield as compared to the N^2 -IQ-dG nucleoside. This is consistent with the intercalated conformation and stacking of the IQ ring below the 5'-neighboring G^{19} of the complementary strand and above the 3'-neighboring $C^8:G^{17}$ base pair (Figure 8). Chemical shift perturbations corroborate the NOE data (Figure 4). The IQ H4a proton resonance, observed at 9.6 ppm, is 0.4 ppm upfield from the resonance observed for the modified N^2 -dG-IQ nucleoside (67), consistent with its location below G¹⁹ and above C⁸ (Figure 8). The IQ moiety displaces the complementary C^{18} base from the duplex, and flips it into the major groove. This is supported by smaller perturbations in chemical shifts for the H4a and CH₃ protons as compared to the H7a, H8a and H9a aromatic protons of IO. The displacement of the modified nucleotide X^7 toward the major groove (Figure 8) is supported by the downfield chemical shift change of 0.4 ppm for the X⁷ H8 and H1' protons of the modified base. The C⁸ H6 proton resonance also experiences a downfield shift of 0.3 ppm. The stacking interactions of the IQ ring with the flanking bases C⁸, G¹⁷ and G¹⁹ are reflected in the thermodynamic analysis of the adduct, in which the thermal melting temperature of 63°C is unchanged from that of the unmodified duplex.

Comparison to the N^2 -acetylaminofluorene-dG adduct

The other N^2 -dG arylamine adduct that has been subjected to conformational analysis, although not in the *Nar*I sequence of interest herein, is that arising from *N*-acetylaminofluorene (AAF; Chart 1) (76). The N^2 -dG-AAF adduct conformation has also been examined using computational approaches (96). Zaliznyak *et al.* (76) have shown that the AAF moiety resides in the minor grove with its long axis directed toward the 5'-end of the modified strand. This shields the hydrophobic AAF ring



Figure 8. Expanded views of the average structure calculated from 10 structures emergent from the rMD calculations of the N^2 -dG-IQ(X⁷) modified duplex. Base stacking of the modified X⁷:C¹⁸ base pair with the 5'-neighbor and 3'-neighbor base pairs. (A) Stacking of C⁶:G¹⁹ above X⁷:C¹⁸. (B) Stacking of X⁷:C¹⁸ above C⁸:G¹⁷. Expanded views of the average structure of the corresponding C8-dG-IQ adduct at X⁷. (75) (C) Stacking of C⁶:G¹⁹ above X⁷:C¹⁸. (D) Stacking of X⁷:C¹⁸ above C⁸:G¹⁷. In each instance, the modified base pair X⁷:C¹⁸ is shown in magenta, with the IQ moiety shown in blue.

from water. Similar to the N^2 -dG-IO adduct, the modified nucleotide maintains the anti-conformation about the glycosyl bond. Notably, the N^2 -dG-AAF adduct increases the stability of the DNA, which has been attributed to a favorable entropic effect (76). The present data reveal that the base-displaced intercalated conformation of the N^2 -dG-IQ adduct at position G³ of the NarI sequence differs from that of the N^2 -dG-AAF adduct, suggesting that the conformations of N^2 -dG arylamine adducts vary rather than following a common motif. At the molecular level, the factors governing whether planar aromatic molecules such as AAF or IQ favor DNA groove binding versus intercalation are not well established, but may be influenced both by their electronic structures and their respective geometries (97). Replication bypass studies have revealed that the N^2 -dG-AAF adduct largely blocked DNA synthesis, but with some bypass and misincorporation of dATP opposite the lesion (98).

Comparison to the C8-dG-IQ adduct

When the C8-dG-IQ adduct was placed into the NarI sequence at the frameshift-prone G^3 position, the IQ

ring also intercalated into the duplex and the complementary C^{18} base was extruded into the major groove. The conformation of the C8-dG-IQ adduct also was characterized as base-displaced intercalated (75). Thus, at the G^3 position within the NarI sequence, both the C8-dG-IQ and N^2 -dG-IQ adducts share a motif in which the IQ ring intercalates and C^{18} is extruded into the major groove. However, the two conformations are distinctive. Apart from the difference in the regiochemistry of alkylation (C8 versus N^2 ; Scheme 1), a major difference between the C8-dG-IO and N^2 -dG-IO adducts is that the C8-dG-IO-modified guanine adopts a svn conformation about the glycosyl bond, whereas the N^2 -dG-IQmodified guanine maintains the anti-conformation about the glycosyl bond (Figures 8 and 9). In addition, for the C8-dG-IQ adduct, rotation of the glycosyl bond into the syn conformation places the Watson-Crick hydrogen bonding edge of the modified dG into the major groove. The X^7 imino and amino protons are exposed to solvent. For the C8-dG-IQ adduct the orientation of the IQ ring with respect to the base is opposite to that of the N^2 -IQ adduct, such that the IQ CH₃ group and H4a and H5a protons face the major groove rather than the minor groove (75). The orientation of the C8-dG-IQ adduct in the duplex rotates the bulk of the IQ aromatic ring away from the flanking bases, resulting in a loss of base-stacking interactions, as shown in Figure 8. In comparison, the N^2 -dG-IQ adduct appears to have more favorable stacking interactions with G¹⁹. These differences may lead to differential processing during both DNA repair and DNA replication.

Structure-activity relationships

The N^2 -dG-IQ adduct is less efficiently removed from genomic DNA by nucleotide excision repair (63,99). The NER machinery is thought to recognize bulky DNA damage that is destabilizing and distortive to the duplex (100-103). It has been proposed that the thermal stabilization of the N^2 -dG-AAF adduct hinders NER (76). We observe that the $T_{\rm m}$ of the N²-dG-IQ adduct at position G^3 within the NarI sequence does not destabilize the duplex (Table 1), correcting our original report (67). The $T_{\rm m}$ of the N²-dG-IQ modified duplex is 63°C, and does not differ significantly from the unmodified duplex. This is remarkable given that the intercalated IQ moiety disrupts Watson-Crick hydrogen bonding and that the complementary C^{18} base is displaced into the major groove. The stability of the N^2 -dG-IQ modified duplex likely arises from favorable stacking between the IQ moiety and the neighboring base pairs (Figure 8). It is also interesting to note that unlike the N^2 -dG-IQ adduct, the C8-dG-IO adduct, which does not stack with the neighboring bases as well at this position (Figure 8) thermally destabilizes the duplex, reducing the $T_{\rm m}$ by 4°C. Yeo et al. (104) examined AAF and AF C8-dG adducts within the NarI sequence and observed a correlation between the degree of destabilization induced by the lesions, binding affinities to the damage recognition protein XPC-RAD23B and overall NER efficiencies. Likewise, Zaliznyak et al. (76) attributed the increased

stability of the N^2 -dG-AAF adduct to its orientation within the minor groove and the entropy-favored release of waters from the duplex. Similar conclusions were reached by Cai et al. (105) who correlated thermodynamic stabilities and van der Waals interaction energies with repair efficiencies for stereoisomeric intercalated N^6 -dA PÂH adducts. Their studies showed that intercalated adducts with fewer DNA structural distortions and increased van der Waals interactions with neighboring bases correlated with reduced repair efficiencies. The HCA PhIP adduct has been compared with the cis- $B[a]P-N^2-dG$ adduct in duplex DNA and in a nucleotide deletion duplex, and it was concluded that local stabilization induced by these adducts governs the ability of the β-hairpins of NER proteins to recognize the damage (106). In summary, it seems plausible that the thermal stability of the N^2 -dG-IQ adduct may, in part, explain the persistence of the N^2 -dG-IQ adduct in rats and primates.

If not repaired, the N^2 -dG-IQ adduct is anticipated to be genotoxic. Indeed, IQ is an order of magnitude more mutagenic than is aflatoxin B_1 in Ames assays. The mutations occur primarily at G:C base pairs (20,21). The replication of the N^2 -dG-IQ-adduct within the NarI sequence is influenced by the identity of the DNA polymerase. Because the damaged guanine remains in the anti-conformation about the glycosyl bond (Figures 7 and 8), one might anticipate that the N^2 -dG-IQ lesion should block Watson-Crick base pairing with incoming dNTPs during lesion bypass. Stover et al. (107) incorporated the N^2 -dG-IQ-adduct into the G¹ and G³ positions of the NarI sequence and examined replication of the oligodeoxynucleotides with Escherichia coli polymerases (pol) I (exonuclease deficient Klenow fragment), exonuclease deficient pol II and the Solfolobus solfataricus P2 DNA polymerase IV (Dpo4), in vitro. At the G^3 position, the N^2 dG-IQ adduct blocked the E. coli polymerases. Pol II exo⁻ favored correct incorporation of dCTP over dGTP but was unable to extend either of these initial insertion products. In contrast, the Dpo4 polymerase bypassed the N^2 -dG-IQ adduct and produced an error-free product. The present studies do not necessarily predict the structure of the N^2 -dG-IQ adduct during trans-lesion bypass. Consequently, it will be of interest to prepare complexes of bypass polymerases with N^2 -dG-IQ modified template:primers in an effort to determine how the N^2 -dG-IQ adduct is accommodated during lesion bypass and how polymerases, e.g. the Dpo4 polymerase (108) allow bypass of this lesion.

Bypass of the N^2 -dG-IQ adduct has been reported to be dependent upon its position in the NarI sequence. Choi et al. (72) have demonstrated that the human DNA polymerase (hpol) η can extend primers beyond template N^2 -dG-IQ adducts. Pol η correctly inserts dCTP and incorrectly inserts dATP. Analyses of hpol η extension products reveal that a -2 bp deletion occurs with the G³ N^2 -dG-IQ adduct. In contrast, at the G¹ position replication past the N^2 -dG-IQ adduct results in error-free incorporation of dCTP, but further extension is inhibited and the polymerase stalls. In contrast, hpol η does not yield -2 bp deletions with the C8-dG-IQ adduct located at position G^3 . While further studies will be necessary to probe the basis for these observations, it is of interest to note that the stability of the N^2 -dG-IQ adduct placed opposite a 2-bp deletion increases as compared to the fully complementary duplex, suggesting that the adduct may stabilize a 2-bp strand slippage intermediate (67). At the G¹ position, the N^2 -dG-IQ adduct is bypassed and extended by the *E. coli* polymerases and the Dpo4 polymerase, and error-free product is observed. Thus, it will also be of interest to examine the structure of the N^2 -dG-IQ adduct when positioned at position G¹ of the *Nar*I sequence.

Summary

Analysis of the N^2 -dG-IQ adduct placed at position G³ of the NarI sequence (68-71,73,74), where it has been observed to cause -2 bp deletions when bypassed by hpol η (72), reveals that it adopts a base-displaced intercalated conformation in which the H4a and CH₃ protons of the IQ ring face the minor groove and the H7a, H8a and H9a protons face the major groove. The IQ ring is shielded from water and stacks with the 5'- and 3'-neighbor base pairs. Remarkably, despite this conformational perturbation, the N^2 -dG-IQ adduct does not destabilize the duplex, which may correlate with the observation that it is refractory to repair by NER (63,99). In addition, the IQ moiety disrupts the potential for Watson-Crick hydrogen bonding with incoming dNTPs, which perhaps explains why this lesion blocks DNA synthesis by many polymerases.

ACCESSION NUMBERS

The structural coordinates were deposited in the Protein Data Bank (www.rcsb.org): The PDB ID code for the N^2 -dG-IQ duplex is 2MAV.

SUPPLEMENTARY DATA

Supplementary Data are available at NAR Online.

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REFERENCES

- Wakabayashi,K., Nagao,M., Esumi,H. and Sugimura,T. (1992) Food-derived mutagens and carcinogens. *Cancer Res.*, 52, 2092s–2098s.
- Layton, D.W., Bogen, K.T., Knize, M.G., Hatch, F.T., Johnson, V.M. and Felton, J.S. (1995) Cancer risk of heterocyclic amines in cooked foods: an analysis and implications for research. *Carcinogenesis*, 16, 39–52.
- Sugimura, T. (1997) Overview of carcinogenic heterocyclic amines. Mutat. Res., 376, 211–219.
- Sugimura, T., Wakabayashi, K., Nakagama, H. and Nagao, M. (2004) Heterocyclic amines: mutagens/carcinogens produced during cooking of meat and fish. *Cancer Sci.*, 95, 290–299.
- Turesky,R.J. and Le Marchand,L. (2011) Metabolism and biomarkers of heterocyclic aromatic amines in molecular epidemiology studies: lessons learned from aromatic amines. *Chem. Res. Toxicol.*, 24, 1169–1214.
- Kataoka,H., Nishioka,S., Kobayashi,M., Hanaoka,T. and Tsugane,S. (2002) Analysis of mutagenic heterocyclic amines in cooked food samples by gas chromatography with nitrogenphosphorus detector. *Bull. Environ. Contam. Toxicol.*, 69, 682–689.
- Felton,J.S., Knize,M.G., Salmon,C.P., Malfatti,M.A. and Kulp,K.S. (2002) Human exposure to heterocyclic amine food mutagens/carcinogens: relevance to breast cancer. *Environ. Mol. Mutagen.*, 39, 112–118.
- 8. Yamashita, M., Wakabayashi, K., Nagao, M., Sato, S., Yamaizumi, Z., Takahashi, M., Kinae, N., Tomita, I. and Sugimura, T. (1986) Detection of 2-amino-3-methylimidazo[4,5f]quinoline in cigarette smoke condensate. Jpn J. Cancer Res.: Gann, 77, 419–422.
- 9. Ushiyama,H., Wakabayashi,K., Hirose,M., Itoh,H., Sugimura,T. and Nagao,M. (1991) Presence of carcinogenic heterocyclic amines in urine of healthy volunteers eating normal diet, but not of inpatients receiving parenteral alimentation. *Carcinogenesis*, **12**, 1417–1422.
- Kobayashi, M., Hanaoka, T., Nishioka, S., Kataoka, H. and Tsugane, S. (2002) Estimation of dietary HCA intakes in a largescale population-based prospective study in Japan. *Mutat. Res.*, 506–507, 233–241.
- International Agency for Research on Cancer (1993). IARC Monographs on the Evaluation of Carcinogenic Risks to Humans, Vol. 56. WHO Press, Lyon, France.
- National Toxicology Program. (2005) National Toxicology Program. Report on Carcinogenesis, 11th edn. U.S. Department of Health and Human Services, Public Health Service. Research Triangle Park, NC.
- 13. Sugimura, T. and Sato, S. (1983) Mutagens-carcinogens in foods. *Cancer Res.*, 43, 2415s–2421s.
- 14. Hecht, S.S. (2002) Tobacco smoke carcinogens and breast cancer. *Environ. Mol. Mutagen.*, **39**, 119–126.
- Oda,Y., Yamazaki,H., Watanabe,M., Nohmi,T. and Shimada,T. (1995) Development of high sensitive umu test system: rapid detection of genotoxicity of promutagenic aromatic amines by *Salmonella typhimurium* strain NM2009 possessing high O-acetyltransferase activity. *Mutat. Res.*, 334, 145–156.
- Josephy, P.D., Gruz, P. and Nohmi, T. (1997) Recent advances in the construction of bacterial genotoxicity assays. *Mutat. Res.*, 386, 1–23.
- Josephy, P.D., Evans, D.H., Parikh, A. and Guengerich, F.P. (1998) Metabolic activation of aromatic amine mutagens by simultaneous expression of human cytochrome P450 1A2, NADPH-cytochrome P450 reductase, and N-acetyltransferase in *Escherichia coli. Chem. Res. Toxicol.*, **11**, 70–74.

- Oda,Y., Aryal,P., Terashita,T., Gillam,E.M., Guengerich,F.P. and Shimada,T. (2001) Metabolic activation of heterocyclic amines and other procarcinogens in *Salmonella typhimurium umu* tester strains expressing human cytochrome P4501A1, 1A2, 1B1, 2C9, 2D6, 2E1, and 3A4 and human NADPH-P450 reductase and bacterial O-acetyltransferase. *Mutat. Res.*, 492, 81–90.
- 19. Nagao, M. (2000) In: Nagao, M. and Sugimura, T. (eds), Foodborne Carcinogens: Heterocyclic Amines. Wiley, New York.
- Kosakarn,P., Halliday,J.A., Glickman,B.W. and Josephy,P.D. (1993) Mutational specificity of 2-nitro-3,4-dimethylimidazo[4,5*f*]quinoline in the lacI gene of *Escherichia coli*. *Carcinogenesis*, 14, 511–517.
- Watanabe, M. and Ohta, T. (1993) Analysis of mutational specificity induced by heterocyclic amines in the *lacZ* gene of *Escherichia coli*. *Carcinogenesis*, 14, 1149–1153.
- Terada, M., Nagao, M., Nakayasu, M., Sakamoto, H., Nakasato, F. and Sugimura, T. (1986) Mutagenic activities of heterocyclic amines in Chinese hamster lung cells in culture. *Environ. Health Perspect.*, 67, 117–119.
- Thompson, L.H., Tucker, J.D., Stewart, S.A., Christensen, M.L., Salazar, E.P., Carrano, A.V. and Felton, J.S. (1987) Genotoxicity of compounds from cooked beef in repair-deficient CHO cells versus *Salmonella* mutagenicity. *Mutagenesis*, 2, 483–487.
- Felton, J.S., Fultz, E., Dolbeare, F.A. and Knize, M.G. (1994) Effect of microwave pretreatment on heterocyclic aromatic amine mutagens/carcinogens in fried beef patties. *Food Chem. Toxicol.*, 32, 897–903.
- Schut,H.A. and Snyderwine,E.G. (1999) DNA adducts of heterocyclic amine food mutagens: implications for mutagenesis and carcinogenesis. *Carcinogenesis*, **20**, 353–368.
- Turesky, R.J. (2002) Heterocyclic aromatic amine metabolism, DNA adduct formation, mutagenesis, and carcinogenesis. *Drug Metab. Rev.*, 34, 625–650.
- Thompson,L.H., Carrano,A.V., Salazar,E., Felton,J.S. and Hatch,F.T. (1983) Comparative genotoxic effects of the cookedfood-related mutagens Trp-P-2 and IQ in bacteria and cultured mammalian cells. *Mutat. Res.*, **117**, 243–257.
- Aeschbacher, H.U. and Ruch, E. (1989) Effect of heterocyclic amines and beef extract on chromosome aberrations and sister chromatid exchanges in cultured human lymphocytes. *Carcinogenesis*, 10, 429–433.
- Tohda, H., Oikawa, A., Kawachi, T. and Sugimura, T. (1980) Induction of sister-chromatid exchanges by mutagens from amino acid and protein pyrolysates. *Mutat. Res.*, 77, 65–69.
- Ohgaki, H., Hasegawa, H., Kato, T., Suenaga, M., Ubukata, M., Sato, S., Takayama, S. and Sugimura, T. (1986) Carcinogenicity in mice and rats of heterocyclic amines in cooked foods. *Environ. Health Perspect.*, 67, 129–134.
- Adamson, R.H., Thorgeirsson, U.P., Snyderwine, E.G., Thorgeirsson, S.S., Reeves, J., Dalgard, D.W., Takayama, S. and Sugimura, T. (1990) Carcinogenicity of 2-amino-3methylimidazo[4,5-*f*]quinoline in nonhuman primates: induction of tumors in three macaques. *Jpn J. Cancer Res.*, 81, 10–14.
- Heflich, R.H. and Neft, R.E. (1994) Genetic toxicity of 2acetylaminofluorene, 2-aminofluorene and some of their metabolites and model metabolites. *Mutat. Res.*, 318, 73–114.
- Thorgeirsson, U.P., Snyderwine, E.G., Gomez, D.E. and Adamson, R.H. (1996) Dietary heterocyclic amines as potential human carcinogens: experimental data from nonhuman primates. *In Vivo*, **10**, 145–152.
- 34. Ohgaki, H., Kusama, K., Matsukura, N., Morino, K., Hasegawa, H., Sato, S., Takayama, S. and Sugimura, T. (1984) Carcinogenicity in mice of a mutagenic compound, 2-amino-3-methylimidazo[4,5f]quinoline, from broiled sardine, cooked beef and beef extract. *Carcinogenesis*, 5, 921–924.
- 35. Takayama,S., Nakatsuru,Y., Masuda,M., Ohgaki,H., Sato,S. and Sugimura,T. (1984) Demonstration of carcinogenicity in F344 rats of 2-amino-3-methyl-imidazo[4,5-f]quinoline from broiled sardine, fried beef and beef extract. *Gann*, **75**, 467–470.
- 36. Tanaka, T., Barnes, W.S., Williams, G.M. and Weisburger, J.H. (1985) Multipotential carcinogenicity of the fried food mutagen 2-amino-3-methylimidazo[4,5-f]quinoline in rats. Jpn J. Cancer Res., 76, 570–576.

- 37. Sugimura, T., Nagao, M. and Wakabayashi, K. (2000) Complex Factors Pertinent to Human Hazard and Risk. Wiley, New York.
- Anderson, K.E., Hammons, G.J., Kadlubar, F.F., Potter, J.D., Kaderlik, K.R., Ilett, K.F., Minchin, R.F., Teitel, C.H., Chou, H.C., Martin, M.V. *et al.* (1997), Metabolic activation of aromatic amines by human pancreas. *Carcinogenesis*, 18, 1085–1092.
- 39. Lang, N.P., Butler, M.A., Massengill, J., Lawson, M., Stotts, R.C., Hauer-Jensen, M. and Kadlubar, F.F. (1994) Rapid metabolic phenotypes for acetyltransferase and cytochrome P4501A2 and putative exposure to food-borne heterocyclic amines increase the risk for colorectal cancer or polyps. *Cancer Epidemiol. Biomarkers Prev.*, **3**, 675–682.
- 40. Shirai, T., Sano, M., Tamano, S., Takahashi, S., Hirose, M., Futakuchi, M., Hasegawa, R., Imaida, K., Matsumoto, K., Wakabayashi, K. *et al.* (1997), The prostate: A target for the carcinogenicity of 2-amino-1-methyl-6-phenylimidazo[4,5-b] pyridine derived from cooked foods. *Cancer Res.*, 57, 195–198.
- Snyderwine, E.G. (1994) Some perspectives on the nutritional aspects of breast cancer research. Food-derived heterocyclic amines as etiologic agents in human mammary cancer. *Cancer*, 74, 1070–1077.
- Ronco, A., De Stefani, E., Mendilaharsu, M. and Deneo-Pellegrini, H. (1996) Meat, fat and risk of breast cancer: a casecontrol study from Uruguay. *Int. J. Cancer*, 65, 328–331.
- 43. Yamazoe, Y., Shimada, M., Kamataki, T. and Kato, R. (1983) Microsomal activation of 2-amino-3-methylimidazo[4,5-*f*]quinoline, a pyrolysate of sardine and beef extracts, to a mutagenic intermediate. *Cancer Res.*, 43, 5768–5774.
- 44. Boobis,A.R., Lynch,A.M., Murray,S., de la Torre,R., Solans,A., Farre,M., Segura,J., Gooderham,N.J. and Davies,D.S. (1994) CYP1A2-catalyzed conversion of dietary heterocyclic amines to their proximate carcinogens is their major route of metabolism in humans. *Cancer Res.*, 54, 89–94.
- 45. Shimada, T., Hayes, C.L., Yamazaki, H., Amin, S., Hecht, S.S., Guengerich, F.P. and Sutter, T.R. (1996) Activation of chemically diverse procarcinogens by human cytochrome P-450 1B1. *Cancer Res.*, 56, 2979–2984.
- 46. Hammons,G.J., Milton,D., Stepps,K., Guengerich,F.P., Tukey,R.H. and Kadlubar,F.F. (1997) Metabolism of carcinogenic heterocyclic and aromatic amines by recombinant human cytochrome P450 enzymes. *Carcinogenesis*, 18, 851–854.
- 47. Turesky,R.J., Constable,A., Richoz,J., Varga,N., Markovic,J., Martin,M.V. and Guengerich,F.P. (1998) Activation of heterocyclic aromatic amines by rat and human liver microsomes and by purified rat and human cytochrome P450 1A2. *Chem. Res. Toxicol.*, **11**, 925–936.
- Guengerich, F.P. (2002) N-hydroxyarylamines. Drug Metab. Rev., 34, 607–623.
- Minchin, R.F., Reeves, P.T., Teitel, C.H., McManus, M.E., Mojarrabi, B., Ilett, K.F. and Kadlubar, F.F. (1992) N-and O-acetylation of aromatic and heterocyclic amine carcinogens by human monomorphic and polymorphic acetyltransferases expressed in COS-1 cells. *Biochem. Biophys. Res. Commun.*, 185, 839–844.
- Hein,D.W., Doll,M.A., Rustan,T.D., Gray,K., Feng,Y., Ferguson,R.J. and Grant,D.M. (1993) Metabolic activation and deactivation of arylamine carcinogens by recombinant human NAT1 and polymorphic NAT2 acetyltransferases. *Carcinogenesis*, 14, 1633–1638.
- Hickman, D., Pope, J., Patil, S.D., Fakis, G., Smelt, V., Stanley, L.A., Payton, M., Unadkat, J.D. and Sim, E. (1998) Expression of arylamine N-acetyltransferase in human intestine. *Gut*, 42, 402–409.
- 52. Le Marchand, L., Hankin, J.H., Pierce, L.M., Sinha, R., Nerurkar, P.V., Franke, A.A., Wilkens, L.R., Kolonel, L.N., Donlon, T., Seifried, A. *et al.* (2002) Well-done red meat, metabolic phenotypes and colorectal cancer in Hawaii Mutat. *Res.*, 506–507, 205–214.
- Ishibe, N., Sinha, R., Hein, D.W., Kulldorff, M., Strickland, P., Fretland, A.J., Chow, W.H., Kadlubar, F.F., Lang, N.P. and Rothman, N. (2002) Genetic polymorphisms in heterocyclic amine metabolism and risk of colorectal adenomas. *Pharmacogenetics*, 12, 145–150.

- 54. Metry,K.J., Neale,J.R., Doll,M.A., Howarth,A.L., States,J.C., McGregor,W.G., Pierce,W.M. Jr and Hein,D.W. (2010) Effect of rapid human N-acetyltransferase 2 haplotype on DNA damage and mutagenesis induced by 2-amino-3-methylimidazo-[4,5f]quinoline (IQ) and 2-amino-3,8-dimethylimidazo-[4,5f]quinoxaline (MeIQx). Mutat. Res., 684, 66–73.
- 55. Snyderwine, E.G., Roller, P.P., Adamson, R.H., Sato, S. and Thorgeirsson, S.S. (1988) Reaction of N-hydroxylamine and N-acetoxy derivatives of 2-amino-3-methylimidazolo[4,5*f*]quinoline with DNA. Synthesis and identification of N-(deoxyguanosin-8-yl)-IQ. *Carcinogenesis*, 9, 1061–1065.
- 56. Nagaoka,H., Wakabayashi,K., Kim,S.B., Kim,I.S., Tanaka,Y., Ochiai,M., Tada,A., Nukaya,H., Sugimura,T. and Nagao,M. (1992) Adduct formation at C-8 of guanine on *in vitro* reaction of the ultimate form of 2-amino-1-methyl-6-phenylimidazo[4,5*b*]pyridine with 2'-deoxyguanosine and its phosphate esters. *Jpn. J. Cancer Res.*, 83, 1025–1029.
- 57. Turesky,R.J., Rossi,S.C., Welti,D.H., Lay,J.O. Jr and Kadlubar,F.F. (1992) Characterization of DNA adducts formed in vitro by reaction of N-hydroxy-2-amino-3-methylimidazo[4,5*f*]quinoline and N-hydroxy-2-amino-3,8-dimethylimidazo[4,5*f*]quinoxaline at the C-8 and N² atoms of guanine. *Chem. Res. Toxicol.*, **5**, 479–490.
- Zenser, T.V., Lakshmi, V.M., Schut, H.A., Zhou, H.J. and Josephy, P. (2009) Activation of aminoimidazole carcinogens by nitrosation: mutagenicity and nucleotide adducts. *Mutat. Res.*, 673, 109–115.
- Murata, M. and Kawanishi, S. (2011) Mechanisms of oxidative DNA damage induced by carcinogenic arylamines. *Front. Biosci.* J. Virt. Lib., 16, 1132–1143.
- 60. Gangl, E.T., Turesky, R.J. and Vouros, P. (2001) Detection of *in vivo* formed DNA adducts at the part-per-billion level by capillary liquid chromatography/microelectrospray mass spectrometry. *Anal. Chem.*, 73, 2397–2404.
- 61. Soglia, J. R., Turesky, R.J., Paehler, A. and Vouros, P. (2001) Quantification of the heterocyclic aromatic amine DNA adduct N-(deoxyguanosin-8-yl)-2-amino-3-methylimidazo[4,5-f]quinoline in livers of rats using capillary liquid chromatography/ microelectrospray mass spectrometry: a dose-response study. *Anal. Chem.*, **73**, 2819–2827.
- Nauwelaers, G., Bessette, E.E., Gu, D., Tang, Y., Rageul, J., Fessard, V., Yuan, J.M., Yu, M.C., Langouet, S. and Turesky, R.J. (2011) DNA adduct formation of 4-aminobiphenyl and heterocyclic aromatic amines in human hepatocytes. *Chem. Res. Toxicol.*, 24, 913–925.
- 63. Turesky,R.J., Markovic,J. and Aeschlimann,J.M. (1996) Formation and differential removal of C-8 and N2-guanine adducts of the food carcinogen 2-amino-3-methylimidazo[4,5f]quinoline in the liver, kidney, and colorectum of the rat. *Chem. Res. Toxicol.*, **9**, 397–402.
- 64. Wang,Z. and Rizzo,C.J. (2001) Synthesis of the C8-deoxyguanosine adduct of the food mutagen IQ. *Org. Lett.*, **3**, 565–568.
- Elmquist,C.E., Stover,J.S., Wang,Z. and Rizzo,C.J. (2004) Sitespecific synthesis and properties of oligonucleotides containing C8-deoxyguanosine adducts of the dietary mutagen IQ. J. Am. Chem. Soc., 126, 11189–11201.
- 66. Stover, J.S. and Rizzo, C.J. (2004) Synthesis of the N²-deoxyguanosine adduct of the potent dietary mutagen IQ. Org. Lett., 6, 4985–4988.
- Stover, J.S. and Rizzo, C.J. (2007) Synthesis of oligonucleotides containing the N²-deoxyguanosine adduct of the dietary carcinogen 2-amino-3-methylimidazo[4,5-f]quinoline. *Chem. Res. Toxicol.*, **20**, 1972–1979.
- Koffel-Schwartz, N., Verdier, J.M., Bichara, M., Freund, A.M., Daune, M.P. and Fuchs, R.P.P. (1984) Carcinogen-induced mutation spectrum in wild-type, *uvrA* and *umuC* strains of *Escherichia coli. J. Mol. Biol.*, **177**, 33–51.
- 69. Koehl, P., Burnouf, D. and Fuchs, R.P.P. (1989) Construction of plasmids containing a unique acetylaminofluorene adduct located within a mutation hot spot. A new probe for frameshift mutagenesis. J. Mol. Biol., 207, 355–364.
- 70. Koehl,P., Valladier,P., Lefevre,J.F. and Fuchs,R.P. (1989) Strong structural effect of the position of a single acetylaminofluorene

adduct within a mutation hot spot. Nucleic Acids Res., 17, 9531-9541.

- Broschard, T.H., Koffel-Schwartz, N. and Fuchs, R.P. (1999) Sequence-dependent modulation of frameshift mutagenesis at NarI-derived mutation hot spots. J. Mol. Biol., 288, 191–199.
- 72. Choi, J.Y., Stover, J.S., Angel, K.C., Chowdhury, G., Rizzo, C.J. and Guengerich, F.P. (2006) Biochemical basis of genotoxicity of heterocyclic arylamine food mutagens: human DNA polymerase eta selectively produces a two-base deletion in copying the N²-guanyl adduct of 2-amino-3-methylimidazo[4,5-f]quinoline but not the C8 adduct at the NarI G³ site. J. Biol. Chem., 281, 25297–25306.
- Fuchs, R.P., Schwartz, N. and Daune, M.P. (1981) Hot spots of frameshift mutations induced by the ultimate carcinogen N-acetoxy-N-2-acetylaminofluorene. *Nature*, 294, 657–659.
- Hoffmann,G.R. and Fuchs,R.P. (1997) Mechanisms of frameshift mutations: Insight from aromatic amines. *Chem. Res. Toxicol.*, 10, 347–359.
- 75. Wang, F., Demuro, N.E., Elmquist, C.E., Stover, J.S., Rizzo, C.J. and Stone, M.P. (2006) Base-displaced intercalated structure of the food mutagen 2-amino-3-methylimidazo[4,5-f]quinoline in the recognition sequence of the *NarI* restriction enzyme, a hottspot for -2 bp deletions. J. Am. Chem. Soc., **128**, 10085–10095.
- 76. Zaliznyak, T., Bonala, R., Johnson, F. and de los Santos, C. (2006) Structure and stability of duplex DNA containing the 3-(deoxyguanosin- N^2 -yl)-2-acetylaminofluorene (dG(N^2)-AAF) lesion: a bulky adduct that persists in cellular DNA. *Chem. Res. Toxicol.*, **19**, 745–752.
- 77. Voehler, M.W., Collier, G., Young, J.K., Stone, M.P. and Germann, M.W. (2006) Performance of cryogenic probes as a function of ionic strength and sample tube geometry. *J. Magn. Reson.*, **183**, 102–109.
- Piotto, M., Saudek, V. and Sklenar, V. (1992) Gradient-tailored excitation for single-quantum NMR spectroscopy of aqueous solutions. J. Biomol. NMR, 2, 661–665.
- 79. Goddard, T.D. and Kneller, D.G. (2006) SPARKY v. 3.113. University of California, San Francisco.
- James, T.L. (1991) Relaxation matrix analysis of two-dimensional nuclear Overhauser effect spectra. *Curr. Opin. Struct. Biol.*, 1, 1042–1053.
- Keepers, J.W. and James, T.L. (1984) A theoretical study of distance determination from NMR. Two-dimensional nuclear Overhauser effect spectra. J. Magn. Reson., 57, 404–426.
- Borgias, B.A. and James, T.L. (1990) MARDIGRAS-a procedure for matrix analysis of relaxation for discerning geometry of an aqueous structure. J. Magn. Reson., 87, 475–487.
- Arnott,S. and Hukins,D.W.L. (1972) Optimised parameters for A-DNA and B-DNA. *Biochem. Biophys. Res. Comm.*, 47, 1504–1509.
- 84. Frisch, M.J., Trucks, G.W., Schlegel, H.B., Scuseria, G.E., Robb, M.A., Cheeseman, J.R., Montgomery, J.A., Vreven, T., Kudin, K.N., Burant, J.C. *et al.* (2004), Gaussian 03, Gaussian, Inc., Wallingford, CT.
- Kirkpatrick, S., Gelatt, C.D. Jr and Vecchi, M.P. (1983) Optimization by simulated annealing. *Science*, 220, 671–680.
- Wang,J., Wolf,R.M., Caldwell,J.W., Kollman,P.A. and Case,D.A. (2004) Development and testing of a general amber force field. *J. Comput. Chem.*, 25, 1157–1174.
- Case, D.A., Cheatham, T.E. 3rd, Darden, T., Gohlke, H., Luo, R., Merz, K.M. Jr, Onufriev, A., Simmerling, C., Wang, B. and Woods, R.J. (2005) The AMBER biomolecular simulation programs. J. Comput. Chem., 26, 1668–1688.
- Bashford, D. and Case, D.A. (2000) Generalized Born models of macromolecular solvation effects. *Annu. Rev. Phys. Chem.*, 51, 129–152.
- Berendsen,H.J.C., Postma,J.P.M., van Gunsteren,W.F., DiNola,A. and Haak,J.R. (1984) Molecular dynamics with coupling to an external bath. J. Phys. Chem., 81, 3684–3690.
- Lavery, R., Moakher, M., Maddocks, J.H., Petkeviciute, D. and Zakrzewska, K. (2009) Conformational analysis of nucleic acids revisited: Curves+. *Nucleic Acids Res.*, 37, 5917–5929.
- 91. Blanchet, C., Pasi, M., Zakrzewska, K. and Lavery, R. (2011) CURVES+ web server for analyzing and visualizing the helical,

backbone and groove parameters of nucleic acid structures. *Nucleic Acids Res.*, **39**, W68–W73.

- Patel,D.J., Shapiro,L. and Hare,D. (1987) DNA and RNA: NMR studies of conformations and dynamics in solution. *Q. Rev. Biophys.*, 20, 35–112.
- 93. Reid, B.R. (1987) Sequence-specific assignments and their use in NMR studies of DNA structure. *Q. Rev. Biophys.*, **20**, 2–28.
- 94. Boelens, R., Scheek, R.M., Dijkstra, K. and Kaptein, R. (1985) Sequential assignment of imino- and amino-proton resonances in ¹H NMR spectra of oligonucleotides by two-dimensional NMR spectroscopy. Application to a lac operator fragment. *J. Magn. Reson.*, 62, 378–386.
- 95. Elmquist, C.E., Wang, F., Stover, J.S., Stone, M.P. and Rizzo, C.J. (2007) Conformational differences of the C8-deoxyguanosine adduct of 2-amino-3-methylimidazo[4,5-f]quinoline (IQ) within the NarI recognition sequence. *Chem. Res. Toxicol.*, 20, 445–454.
- 96. Grad, R., Shapiro, R., Hingerty, B.E. and Broyde, S. (1997) A molecular mechanics and dynamics study of the minor adduct between DNA and the carcinogen 2-(acetylamino)fluorene (dG-N²-AAF). *Chem. Res. Toxicol.*, **10**, 1123–1132.
- 97. Brown,K., Hingerty,B.E., Guenther,E.A., Krishnan,V.V., Broyde,S., Turteltaub,K.W. and Cosman,M. (2001) Solution structure of the 2-amino-1- methyl-6-phenylimidazo[4,5b]pyridine C8-deoxyguanosine adduct in duplex DNA. *Proc. Natl Acad. Sci. U.S.A.*, **98**, 8507–8512.
- Shibutani,S. and Grollman,A.P. (1993) Nucleotide misincorporation on DNA templates containing N-(deoxyguanosin-N²-yl)-2-(acetylamino)fluorene. *Chem. Res. Toxicol.*, 6, 819–824.
- 99. Turesky, R.J., Box, R.M., Markovic, J., Gremaud, E. and Snyderwine, E.G. (1997) Formation and persistence of DNA adducts of 2-amino-3-methylimidazo[4,5-f]quinoline in the rat and nonhuman primates. *Mutat. Res.*, **376**, 235–241.
- 100. Gunz,D., Hess,M.T. and Naegeli,H. (1996) Recognition of DNA adducts by human nucleotide excision repair. Evidence for a thermodynamic probing mechanism. J. Biol. Chem., 271, 25089–25098.

- 101. Batty,D.P. and Wood,R.D. (2000) Damage recognition in nucleotide excision repair of DNA. *Gene*, **241**, 193–204.
- 102. Sugasawa,K., Okamoto,T., Shimizu,Y., Masutani,C., Iwai,S. and Hanaoka,F. (2001) A multistep damage recognition mechanism for global genomic nucleotide excision repair. *Genes Dev.*, 15, 507–521.
- 103. Geacintov, N.E., Broyde, S., Buterin, T., Naegeli, H., Wu, M., Yan, S. and Patel, D.J. (2002) Thermodynamic and structural factors in the removal of bulky DNA adducts by the nucleotide excision repair machinery. *Biopolymers*, 65, 202–210.
- 104. Yeo, J.E., Khoo, A., Fagbemi, A.F. and Scharer, O.D. (2012) The efficiencies of damage recognition and excision correlate with duplex destabilization induced by acetylaminofluorene adducts in human nucleotide excision repair. *Chem. Res. Toxicol.*, 25, 2462–2468.
- 105. Cai, Y., Geacintov, N.E. and Broyde, S. (2012) Nucleotide excision repair efficiencies of bulky carcinogen-DNA adducts are governed by a balance between stabilizing and destabilizing interactions. *Biochemistry*, **51**, 1486–1499.
- 106. Reeves, D.A., Mu, H., Kropachev, K., Cai, Y., Ding, S., Kolbanovskiy, A., Kolbanovskiy, M., Chen, Y., Krzeminski, J., Amin, S. *et al.* (2011) Resistance of bulky DNA lesions to nucleotide excision repair can result from extensive aromatic lesion-base stacking interactions. *Nucleic Acids Res.*, **39**, 8752–8764.
- 107. Stover, J.S., Chowdhury, G., Zang, H., Guengerich, F.P. and Rizzo, C.J. (2006) Translesion synthesis past the C8- and N²-deoxyguanosine adducts of the dietary mutagen 2-Amino-3methylimidazo[4,5-f]quinoline in the NarI recognition sequence by prokaryotic DNA polymerases. *Chem. Res. Toxicol.*, **19**, 1506–1517.
- 108. Ling,H., Boudsocq,F., Woodgate,R. and Yang,W. (2001) Crystal structure of a Y-family DNA polymerase in action: a mechanism for error-prone and lesion-bypass replication. *Cell*, **107**, 91–102.