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2-(5-Bromopyridin-3-yl)-5-[3-(4,5,6,7-tetrahydrothieno[3,2-*c*]pyridine-5-ylsulfonyl)thiophen-2-yl]-1,3,4-oxadiazole

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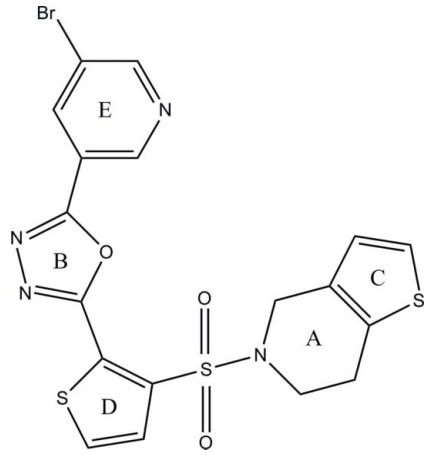
Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.004 Å; R factor = 0.046; wR factor = 0.123; data-to-parameter ratio = 27.6.

In the title compound, $C_{18}H_{13}BrN_4O_3S_3$, the tetrahydropyridine ring adopts a half-chair conformation with the central methylene-C atom of the NCH_2CH_2 unit at the flap. The dihedral angles between the tetrahydropyridine ring and the pyridine and two thiophene rings are 69.34 (13) 5.66 (13) and 68.63 (13)°, respectively, while the dihedral angle between the 1,3,4-oxadiazole and tetrahydropyridine rings is 54.76 (13)°. The molecule is stabilized by an intramolecular C—H···N interaction. In the crystal, adjacent molecules are connected *via* bifurcated C—H···(N,O) hydrogen bonds, forming a chain along the *b* axis.

Related literature

For applications of 4,5,6,7-tetrahydrothieno[3,2-*c*]pyridine derivatives, see: Lopez-Rodriguez *et al.* (2001); Roth *et al.* (1994); Ying & Rusak (1997). For ring conformational analysis, see: Cremer & Pople (1975).



Experimental

Crystal data

$C_{18}H_{13}BrN_4O_3S_3$
M_r = 509.41
Monoclinic, $P2_1/c$
a = 7.0327 (14) Å
b = 7.6488 (15) Å
c = 36.939 (7) Å
β = 91.315 (5)°
V = 1986.5 (7) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 2.41 mm^{-1}
T = 296 K
0.35 × 0.13 × 0.05 mm

Data collection

Bruker APEXII DUO CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)
T_{min} = 0.482, T_{max} = 0.885
22958 measured reflections
7230 independent reflections
4160 reflections with $I > 2\sigma(I)$
R_{int} = 0.051

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.046
$wR(F^2)$ = 0.123
S = 1.02
7230 reflections
262 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.76 e Å^{-3}
$\Delta\rho_{min}$ = −0.73 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7A···N2	0.97	2.52	3.283 (4)	136
C10—H10A···O3^{i}	0.93	2.49	3.330 (3)	150
C10—H10A···N2^{i}	0.93	2.42	3.183 (3)	139

Symmetry code: (i) $x, y+1, z$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Nuclear Sciences, Government of India, for the Young Scientist award.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2791).

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supplementary materials

Acta Cryst. (2011). E**67**, o2743-o2744 [**doi:10.1107/S1600536811038529**]

2-(5-Bromopyridin-3-yl)-5-[3-(4,5,6,7-tetrahydrothieno[3,2-*c*]pyridine-5-ylsulfonyl)thiophen-2-yl]-1,3,4-oxadiazole

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Comment

4,5,6,7-Tetrahydrothieno[3,2-c]pyridine derivatives are extensively studied in medicinal chemistry due to their various biological activities (Lopez-Rodriguez *et al.*, 2001). 4,5,6,7-Tetrahydrothieno[3,2-c] pyridine oxadiazole derivatives are mainly used in CNS functions and disorders such as schizophrenia (Roth *et al.*, 1994), depression, epilepsy, migraine, and control of circadian rhythm (Ying & Rusak, 1997).

The molecular structure of the title compound, Fig. 1, contains five rings, namely, A (N3/C1,C2,C5–C7), B (N1/N2/O1/C12,C13), C (S3/C2–C5), D (S1/C8–C11) and E (N4/C14–C18). The tetrahydropyridine (N3/C1,C2,C5–C7) ring adopts a half-chair conformation with puckering parameters Q = 0.497 (3) Å, θ = 131.5 (3)° and φ = 141.6 (4)° with the flap atom at C7 [maximum deviation of -0.338 (3) Å]. The dihedral angle between the least-squares planes of the rings are A/B = 54.76 (13)°, A/C = 5.66 (13)°, A/D = 68.63 (13)°, A/E = 69.34 (13)°, B/C = 56.97 (14)°, B/D = 13.90 (14)°, B/E = 15.62 (13)°, C/D = 70.85 (13)°, and C/E = 70.88 (13)°.

In the crystal structure, (Fig. 2), adjacent molecules are connected *via* bifurcated C—H···N and C—H···O (Table 1) hydrogen bonds forming one-dimensional chains along the *b*-axis.

Experimental

To a mixture of 3-(6,7-dihydrothieno[3,2-c]pyridine-5(4*H*)-ylsulfonyl) thiophene-2-carbohydrazide (0.5 g, 0.0014 mol) and 5-bromopyridine-3-carboxylic acid (0.29 g, 0.0014 mol), neutral alumina (0.5 g) and $POCl_3$, (1.1 g, 0.007 mol) were added. The resulting mixture was irradiated in a microwave oven for 5 min. Mass analysis of the crude reaction mixture confirmed completion of the reaction. The reaction mixture was concentrated and the residue was purified by column chromatography to get title compound which was recrystallised using acetone. Yield: 68%, m.p. 429–431 K.

Refinement

All hydrogen atoms were positioned geometrically [C–H = 0.93 or 0.97 Å] and were refined using a riding model, with U_{iso}(H) = 1.2 U_{eq}(C).

Figures

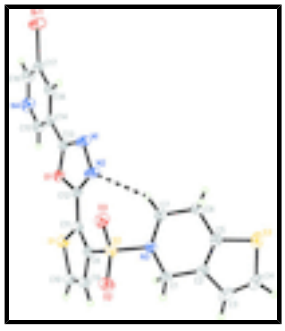

Fig. 1. The molecule of the title compound, showing 30% probability displacement ellipsoids. The dashed line represents a C—H···N interaction.

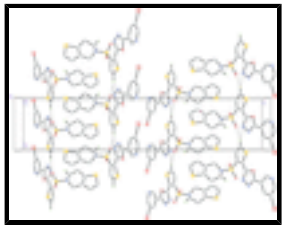

Fig. 2. A view of the crystal packing for the title compound (I). The dashed lines represent C—H···O and C—H···N hydrogen bonds.

2-(5-Bromopyridin-3-yl)-5-[3-(4,5,6,7-tetrahydrothieno[3,2-*c*]pyridine- 5-ylsulfonyl)thiophen-2-yl]-1,3,4-oxadiazole

Crystal data

$C_{18}H_{13}BrN_4O_3S_3$	$F(000) = 1024$
$M_r = 509.41$	$D_x = 1.703$ Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 3605 reflections
$a = 7.0327$ (14) Å	$\theta = 2.7$–$25.9°$
$b = 7.6488$ (15) Å	$\mu = 2.41$ mm^{-1}
$c = 36.939$ (7) Å	$T = 296$ K
$\beta = 91.315$ (5)°	Plate, colourless
$V = 1986.5$ (7) Å^3	0.35 × 0.13 × 0.05 mm
$Z = 4$	

Data collection

Bruker APEXII DUO CCD area-detector diffractometer	7230 independent reflections
Radiation source: fine-focus sealed tube	4160 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.051$
φ and ω scans	$\theta_{max} = 32.7°$, $\theta_{min} = 2.2°$
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)	$h = -10 \rightarrow 8$
$T_{min} = 0.482$, $T_{max} = 0.885$	$k = -11 \rightarrow 11$
22958 measured reflections	$l = -55 \rightarrow 52$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.046$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.123$	H-atom parameters constrained
$S = 1.02$	$w = 1/[\sigma^2(F_o^2) + (0.0445P)^2 + 0.8759P]$ where $P = (F_o^2 + 2F_c^2)/3$
7230 reflections	$(\Delta/\sigma)_{max} < 0.001$
262 parameters	$\Delta\rho_{max} = 0.76$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.73$ e Å^{-3}

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
S1	0.93996 (10)	0.96690 (8)	0.10025 (2)	0.04559 (16)
S2	1.36853 (8)	0.62674 (8)	0.146401 (17)	0.03714 (14)
S3	1.17503 (12)	0.51687 (12)	0.30392 (2)	0.0593 (2)
Br1	0.24445 (4)	−0.00049 (4)	0.017218 (10)	0.06497 (12)
O1	0.7735 (2)	0.6343 (2)	0.07860 (4)	0.0371 (4)
O2	1.5532 (2)	0.7035 (3)	0.14769 (5)	0.0505 (5)
O3	1.3410 (3)	0.4632 (2)	0.12866 (5)	0.0464 (4)
N1	0.7985 (3)	0.3595 (3)	0.09508 (7)	0.0503 (6)
N2	0.9456 (3)	0.4530 (3)	0.11199 (7)	0.0513 (6)
N3	1.3043 (3)	0.6029 (3)	0.18809 (5)	0.0358 (4)
N4	0.2436 (3)	0.5346 (4)	0.02560 (8)	0.0582 (6)
C1	1.3498 (4)	0.7466 (3)	0.21341 (7)	0.0443 (6)
H1A	1.2736	0.8485	0.2073	0.053*
H1B	1.4829	0.7781	0.2117	0.053*
C2	1.3091 (3)	0.6885 (3)	0.25127 (7)	0.0397 (5)
C3	1.3612 (4)	0.7808 (4)	0.28326 (8)	0.0555 (7)
H3A	1.4313	0.8839	0.2833	0.067*
C4	1.2989 (4)	0.7039 (5)	0.31349 (9)	0.0619 (8)

H4A	1.3206	0.7472	0.3368	0.074*
C5	1.2078 (3)	0.5430 (4)	0.25829 (7)	0.0426 (5)
C6	1.1309 (4)	0.4206 (4)	0.22997 (8)	0.0515 (7)
H6A	1.0054	0.3804	0.2365	0.062*
H6B	1.2134	0.3195	0.2281	0.062*
C7	1.1199 (4)	0.5162 (4)	0.19417 (8)	0.0453 (6)
H7A	1.0920	0.4343	0.1747	0.054*
H7B	1.0190	0.6026	0.1945	0.054*
C8	1.2147 (3)	0.7874 (3)	0.12720 (7)	0.0367 (5)
C9	1.2684 (4)	0.9644 (3)	0.13041 (8)	0.0443 (6)
H9A	1.3834	1.0013	0.1407	0.053*
C10	1.1341 (4)	1.0751 (3)	0.11683 (8)	0.0486 (6)
H10A	1.1467	1.1961	0.1166	0.058*
C11	1.0374 (3)	0.7667 (3)	0.11097 (6)	0.0366 (5)
C12	0.9260 (3)	0.6133 (3)	0.10159 (6)	0.0359 (5)
C13	0.7017 (3)	0.4705 (3)	0.07619 (7)	0.0367 (5)
C14	0.5288 (3)	0.4345 (3)	0.05505 (6)	0.0374 (5)
C15	0.4063 (4)	0.5660 (4)	0.04346 (8)	0.0492 (6)
H15A	0.4393	0.6814	0.0484	0.059*
C16	0.1984 (4)	0.3691 (4)	0.01865 (8)	0.0541 (7)
H16A	0.0842	0.3453	0.0064	0.065*
C17	0.3143 (3)	0.2309 (4)	0.02888 (7)	0.0442 (6)
C18	0.4814 (3)	0.2626 (3)	0.04777 (7)	0.0415 (5)
H18A	0.5600	0.1713	0.0554	0.050*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0598 (4)	0.0272 (3)	0.0493 (4)	0.0054 (2)	−0.0094 (3)	0.0027 (3)
S2	0.0358 (3)	0.0367 (3)	0.0390 (3)	0.0038 (2)	0.0026 (2)	−0.0008 (2)
S3	0.0600 (4)	0.0788 (5)	0.0392 (4)	0.0036 (4)	0.0025 (3)	0.0054 (4)
Br1	0.04669 (17)	0.0620 (2)	0.0854 (3)	−0.00962 (13)	−0.01522 (15)	−0.01519 (17)
O1	0.0451 (9)	0.0303 (7)	0.0356 (9)	0.0030 (6)	−0.0073 (7)	0.0014 (7)
O2	0.0351 (9)	0.0595 (12)	0.0571 (12)	−0.0017 (8)	0.0089 (8)	0.0006 (9)
O3	0.0534 (11)	0.0408 (9)	0.0449 (11)	0.0115 (8)	0.0011 (8)	−0.0084 (8)
N1	0.0554 (13)	0.0304 (10)	0.0638 (15)	−0.0004 (9)	−0.0274 (11)	0.0006 (10)
N2	0.0576 (13)	0.0280 (9)	0.0671 (16)	−0.0017 (9)	−0.0277 (12)	−0.0001 (10)
N3	0.0317 (9)	0.0376 (10)	0.0379 (11)	−0.0042 (7)	−0.0015 (8)	−0.0015 (8)
N4	0.0522 (14)	0.0607 (15)	0.0610 (16)	0.0138 (11)	−0.0163 (12)	0.0063 (12)
C1	0.0475 (14)	0.0396 (12)	0.0459 (15)	−0.0084 (10)	0.0007 (11)	−0.0064 (11)
C2	0.0303 (11)	0.0482 (13)	0.0404 (14)	0.0004 (9)	−0.0031 (10)	−0.0065 (11)
C3	0.0443 (14)	0.0685 (19)	0.0534 (18)	−0.0038 (13)	−0.0043 (13)	−0.0197 (15)
C4	0.0549 (16)	0.087 (2)	0.0439 (17)	0.0084 (15)	−0.0062 (13)	−0.0193 (16)
C5	0.0370 (12)	0.0512 (14)	0.0395 (14)	0.0016 (10)	−0.0015 (10)	0.0004 (11)
C6	0.0537 (15)	0.0537 (15)	0.0471 (16)	−0.0187 (12)	0.0028 (12)	0.0018 (13)
C7	0.0367 (12)	0.0570 (16)	0.0420 (14)	−0.0119 (11)	−0.0036 (10)	−0.0012 (12)
C8	0.0431 (12)	0.0300 (10)	0.0369 (13)	0.0003 (9)	0.0010 (10)	0.0013 (9)
C9	0.0537 (15)	0.0346 (11)	0.0445 (15)	−0.0070 (10)	−0.0015 (12)	−0.0003 (11)

C10	0.0658 (17)	0.0285 (11)	0.0512 (16)	−0.0034 (11)	−0.0048 (13)	0.0019 (11)
C11	0.0484 (13)	0.0253 (9)	0.0358 (13)	0.0025 (9)	−0.0029 (10)	0.0002 (9)
C12	0.0454 (12)	0.0281 (10)	0.0338 (12)	0.0029 (9)	−0.0075 (10)	−0.0024 (9)
C13	0.0427 (12)	0.0313 (10)	0.0357 (12)	0.0030 (9)	−0.0055 (10)	−0.0023 (9)
C14	0.0410 (12)	0.0401 (12)	0.0308 (12)	0.0033 (10)	−0.0042 (9)	0.0000 (10)
C15	0.0543 (15)	0.0456 (14)	0.0475 (16)	0.0079 (12)	−0.0071 (12)	0.0021 (12)
C16	0.0401 (13)	0.0674 (19)	0.0542 (17)	0.0042 (13)	−0.0107 (12)	0.0033 (15)
C17	0.0385 (12)	0.0524 (14)	0.0415 (14)	0.0002 (11)	−0.0056 (11)	−0.0016 (12)
C18	0.0373 (12)	0.0436 (12)	0.0432 (14)	0.0049 (10)	−0.0076 (10)	−0.0003 (11)

Geometric parameters (Å, °)

S1—C10	1.698 (3)	C3—C4	1.345 (4)
S1—C11	1.720 (2)	C3—H3A	0.9300
S2—O3	1.4233 (19)	C4—H4A	0.9300
S2—O2	1.4248 (18)	C5—C6	1.496 (4)
S2—N3	1.625 (2)	C6—C7	1.512 (4)
S2—C8	1.774 (2)	C6—H6A	0.9700
S3—C4	1.707 (4)	C6—H6B	0.9700
S3—C5	1.718 (3)	C7—H7A	0.9700
Br1—C17	1.884 (3)	C7—H7B	0.9700
O1—C13	1.353 (3)	C8—C11	1.381 (3)
O1—C12	1.362 (3)	C8—C9	1.410 (3)
N1—C13	1.284 (3)	C9—C10	1.356 (4)
N1—N2	1.394 (3)	C9—H9A	0.9300
N2—C12	1.291 (3)	C10—H10A	0.9300
N3—C1	1.473 (3)	C11—C12	1.448 (3)
N3—C7	1.479 (3)	C13—C14	1.456 (3)
N4—C15	1.329 (4)	C14—C18	1.381 (3)
N4—C16	1.329 (4)	C14—C15	1.386 (4)
C1—C2	1.501 (4)	C15—H15A	0.9300
C1—H1A	0.9700	C16—C17	1.382 (4)
C1—H1B	0.9700	C16—H16A	0.9300
C2—C5	1.349 (4)	C17—C18	1.375 (3)
C2—C3	1.417 (4)	C18—H18A	0.9300
C10—S1—C11	92.19 (12)	N3—C7—C6	108.8 (2)
O3—S2—O2	119.45 (11)	N3—C7—H7A	109.9
O3—S2—N3	107.45 (11)	C6—C7—H7A	109.9
O2—S2—N3	106.69 (11)	N3—C7—H7B	109.9
O3—S2—C8	110.45 (11)	C6—C7—H7B	109.9
O2—S2—C8	105.95 (11)	H7A—C7—H7B	108.3
N3—S2—C8	106.06 (11)	C11—C8—C9	112.6 (2)
C4—S3—C5	91.51 (15)	C11—C8—S2	129.09 (17)
C13—O1—C12	102.63 (17)	C9—C8—S2	118.17 (19)
C13—N1—N2	106.4 (2)	C10—C9—C8	112.7 (2)
C12—N2—N1	106.31 (19)	C10—C9—H9A	123.7
C1—N3—C7	114.6 (2)	C8—C9—H9A	123.7
C1—N3—S2	117.19 (16)	C9—C10—S1	112.13 (19)
C7—N3—S2	117.26 (16)	C9—C10—H10A	123.9

C15—N4—C16	117.9 (2)	S1—C10—H10A	123.9
N3—C1—C2	109.1 (2)	C8—C11—C12	132.5 (2)
N3—C1—H1A	109.9	C8—C11—S1	110.41 (17)
C2—C1—H1A	109.9	C12—C11—S1	117.10 (17)
N3—C1—H1B	109.9	N2—C12—O1	112.0 (2)
C2—C1—H1B	109.9	N2—C12—C11	130.1 (2)
H1A—C1—H1B	108.3	O1—C12—C11	117.87 (19)
C5—C2—C3	112.2 (3)	N1—C13—O1	112.6 (2)
C5—C2—C1	122.4 (2)	N1—C13—C14	126.3 (2)
C3—C2—C1	125.3 (2)	O1—C13—C14	121.0 (2)
C4—C3—C2	113.0 (3)	C18—C14—C15	119.0 (2)
C4—C3—H3A	123.5	C18—C14—C13	118.6 (2)
C2—C3—H3A	123.5	C15—C14—C13	122.3 (2)
C3—C4—S3	111.7 (2)	N4—C15—C14	123.0 (3)
C3—C4—H4A	124.2	N4—C15—H15A	118.5
S3—C4—H4A	124.2	C14—C15—H15A	118.5
C2—C5—C6	124.5 (2)	N4—C16—C17	122.6 (3)
C2—C5—S3	111.6 (2)	N4—C16—H16A	118.7
C6—C5—S3	124.0 (2)	C17—C16—H16A	118.7
C5—C6—C7	108.6 (2)	C18—C17—C16	119.7 (3)
C5—C6—H6A	110.0	C18—C17—Br1	119.8 (2)
C7—C6—H6A	110.0	C16—C17—Br1	120.4 (2)
C5—C6—H6B	110.0	C17—C18—C14	117.8 (2)
C7—C6—H6B	110.0	C17—C18—H18A	121.1
H6A—C6—H6B	108.3	C14—C18—H18A	121.1
C13—N1—N2—C12	−0.5 (3)	C8—C9—C10—S1	0.5 (3)
O3—S2—N3—C1	170.10 (17)	C11—S1—C10—C9	−0.6 (2)
O2—S2—N3—C1	40.9 (2)	C9—C8—C11—C12	179.0 (3)
C8—S2—N3—C1	−71.75 (19)	S2—C8—C11—C12	−5.7 (4)
O3—S2—N3—C7	−47.6 (2)	C9—C8—C11—S1	−0.3 (3)
O2—S2—N3—C7	−176.79 (18)	S2—C8—C11—S1	174.96 (15)
C8—S2—N3—C7	70.6 (2)	C10—S1—C11—C8	0.5 (2)
C7—N3—C1—C2	45.9 (3)	C10—S1—C11—C12	−178.9 (2)
S2—N3—C1—C2	−170.77 (16)	N1—N2—C12—O1	0.1 (3)
N3—C1—C2—C5	−13.3 (3)	N1—N2—C12—C11	179.3 (3)
N3—C1—C2—C3	170.1 (2)	C13—O1—C12—N2	0.3 (3)
C5—C2—C3—C4	0.0 (4)	C13—O1—C12—C11	−179.0 (2)
C1—C2—C3—C4	176.9 (3)	C8—C11—C12—N2	15.1 (5)
C2—C3—C4—S3	0.0 (3)	S1—C11—C12—N2	−165.6 (2)
C5—S3—C4—C3	0.0 (2)	C8—C11—C12—O1	−165.8 (2)
C3—C2—C5—C6	179.2 (3)	S1—C11—C12—O1	13.5 (3)
C1—C2—C5—C6	2.2 (4)	N2—N1—C13—O1	0.7 (3)
C3—C2—C5—S3	0.0 (3)	N2—N1—C13—C14	−177.0 (2)
C1—C2—C5—S3	−177.00 (19)	C12—O1—C13—N1	−0.6 (3)
C4—S3—C5—C2	0.0 (2)	C12—O1—C13—C14	177.3 (2)
C4—S3—C5—C6	−179.2 (2)	N1—C13—C14—C18	−14.6 (4)
C2—C5—C6—C7	−20.7 (4)	O1—C13—C14—C18	167.9 (2)
S3—C5—C6—C7	158.4 (2)	N1—C13—C14—C15	162.4 (3)
C1—N3—C7—C6	−67.1 (3)	O1—C13—C14—C15	−15.1 (4)

S2—N3—C7—C6	149.6 (2)	C16—N4—C15—C14	−0.2 (4)
C5—C6—C7—N3	49.6 (3)	C18—C14—C15—N4	0.2 (4)
O3—S2—C8—C11	29.5 (3)	C13—C14—C15—N4	−176.8 (3)
O2—S2—C8—C11	160.2 (2)	C15—N4—C16—C17	−0.7 (5)
N3—S2—C8—C11	−86.6 (2)	N4—C16—C17—C18	1.6 (4)
O3—S2—C8—C9	−155.4 (2)	N4—C16—C17—Br1	−178.6 (2)
O2—S2—C8—C9	−24.7 (2)	C16—C17—C18—C14	−1.5 (4)
N3—S2—C8—C9	88.4 (2)	Br1—C17—C18—C14	178.66 (19)
C11—C8—C9—C10	−0.1 (3)	C15—C14—C18—C17	0.7 (4)
S2—C8—C9—C10	−175.9 (2)	C13—C14—C18—C17	177.8 (2)

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C7—H7A···N2	0.97	2.52	3.283 (4)	136
C10—H10A···O3^{i}	0.93	2.49	3.330 (3)	150
C10—H10A···N2^{i}	0.93	2.42	3.183 (3)	139

Symmetry codes: (i) *x*, *y*+1, *z*.

Fig. 1

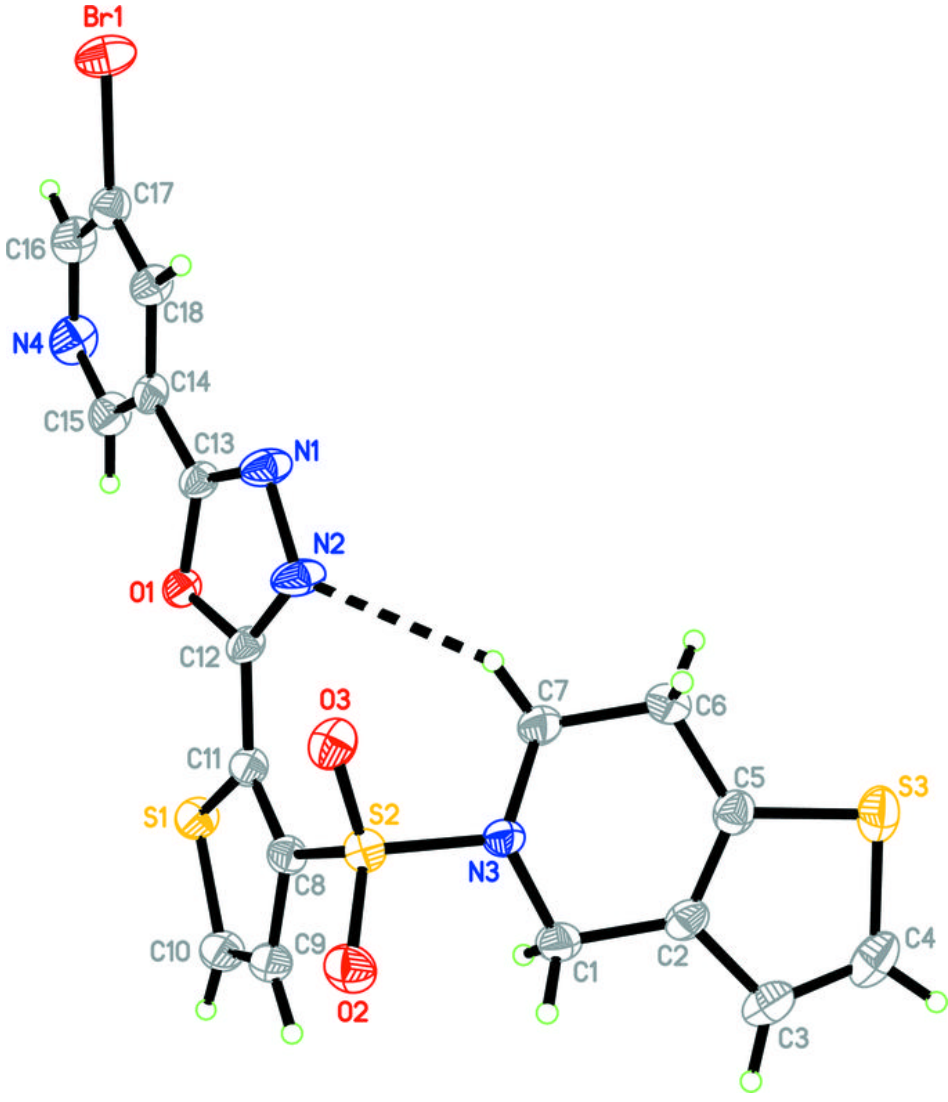
Br1
C17
C16
C18
N4
C14
C15
C13
N1
N2
O1
C12
O3
C7
C6
C11
C5
S3
S1
S2
N3
C8
C10
C9
O2
C1
C2
C4
C3

Fig. 2