

Synthesis and Biological Evaluation of Lysophosphatidic Acid Analogues Using Conformational Restriction and Bioisosteric Replacement Strategies

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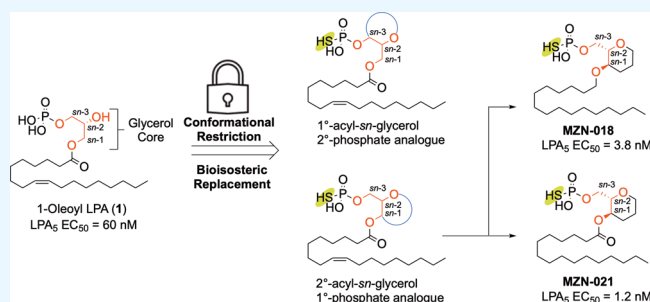
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ABSTRACT: Lysophosphatidic acid (LPA) is a key player in many physiological and pathophysiological processes. The biological activities of LPA are mediated through interactions with—at least—six subtypes of G-protein-coupled receptors (GPCRs) named LPA_{1–6}. Developing a pharmacological tool molecule that activates LPA subtype receptors selectively will allow a better understanding of their specific physiological roles. Here, we designed and synthesized conformationally restricted 25 1-oleoyl LPA analogues **MZN-001** to **MZN-025** by incorporating its glycerol linker into dihydropyran, tetrahydropyran, and pyrrolidine rings and varying the lipophilic chain. The agonistic activities of these compounds were evaluated using the TGF α shedding assay. Overall, the synthesized analogues exhibited significantly reduced agonistic activities toward LPA₁, LPA₂, and LPA₆, while demonstrating potent activities toward LPA₃, LPA₄, and LPA₅ compared to the parent LPA. Specifically, **MZN-010** showed more than 10 times greater potency ($EC_{50} = 4.9$ nM) than the standard 1-oleoyl LPA ($EC_{50} = 78$ nM) toward LPA₅ while exhibiting significantly lower activity on LPA₁, LPA₂, and LPA₆ and comparable potency toward LPA₃ and LPA₄. Based on the **MZN-010** scaffold, we synthesized additional analogues with improved selectivity and potency toward LPA₅. Compound **MZN-021**, which contains a saturated lipophilic chain, exhibited 50 times more potent activity ($EC_{50} = 1.2$ nM) than the natural LPA against LPA₅ with over a 45-fold higher selectivity when compared to those of other LPA receptors. Thus, **MZN-021** was found to be a potent and selective LPA₅ agonist. The findings of this study could contribute to broadening the current knowledge about the stereochemical and three-dimensional arrangement of LPA pharmacophore components inside LPA receptors and paving the way toward synthesizing other subtype-selective pharmacological probes.



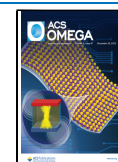
INTRODUCTION

Lysophosphatidic acid (LPA) is a kind of bioactive lipid that acts as an extracellular signaling molecule, eliciting wide arrays of physiological and pathophysiological functions. Most of the LPA effects are mediated via at least six G-protein-coupled receptors (GPCRs),¹ named LPA_{1–6}. The first subtype of this family (LPA₁) was discovered in the mid-1990s from studies using neural cells for its ability to induce cell rounding.² Following this, five LPA receptors were identified in the late 1990s and early 2000s. LPA_{1–3} are classified as part of the endothelial differentiation gene (Edg) subfamily of GPCRs, whereas LPA_{4–6} are categorized as non-Edg family members of LPA receptors and belong to the purinergic GPCR cluster.³ LPA is involved in many biological events including cellular proliferation, cancer cell invasion, calcium mobilization, platelet activation, brain development, and hair follicle formation, among others.^{4–10} Hence, developing potent LPA receptor modulators that can selectively activate or block specific receptors would enhance our understanding of the

distinct functions of each receptor and uncover its potential benefits for therapeutic applications.

1-Oleoyl LPA (**1**) (Figure 1) is the most widely and extensively used LPA molecular species reported in research with potent biological effects^{1,3} and agonism toward all six LPA receptors. Structurally, it has a glycerol backbone with a phosphate group at the *sn*-3 position and an oleoyl fatty chain at the *sn*-1 position.¹¹ Generally, LPA exists as two isomers, *sn*-1 and *sn*-2 acyl LPA, depending on the relative position of the acyl chain. Several studies have been conducted by different groups in an attempt to establish a structure–activity relationship (SAR) of LPA toward LPA receptors and to

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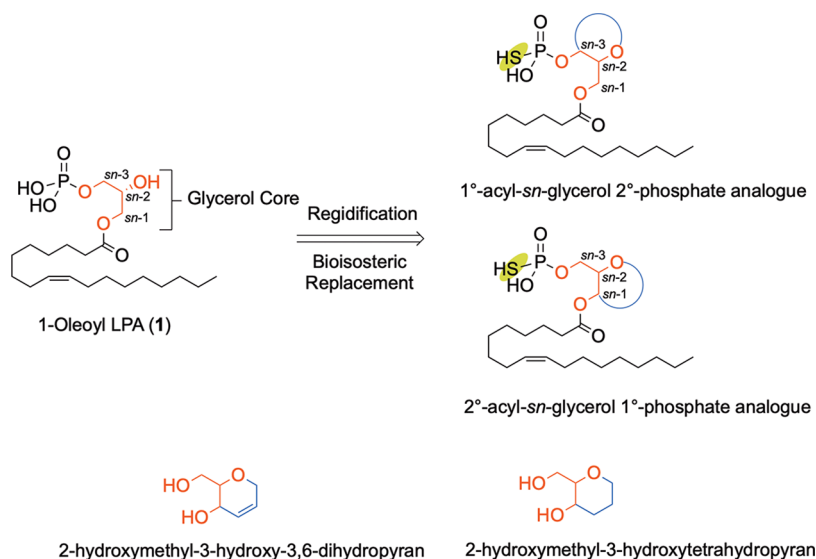


Figure 1. Molecular design of the LPA analogs.

obtain a better understanding of its pharmacophore. These studies have concluded that the phosphate group or phosphate isosteres that have the capability of retaining the negative charge under physiological pH are essential for agonistic activity.¹² Only two classes of phosphate isosteres proved to be active in this context, namely, phosphonate derivatives especially those containing α -hydroxy, keto, and fluorophosphonate^{13,14} and phosphorothioate analogues,^{12,15,16} indicating its direct involvement in receptor activation, which has been proven by X-ray crystallography and single-particle cryo-EM analysis of LPA₆ and LPA₁, respectively.^{17–19} Recently, a nonlipid LPA₂-selective agonist was developed. The sulfamoyl benzoic acid (SBA) group in this agonist was predicted to occupy the phosphate group binding site, based on the model used in the study.^{20–22} However, no report on the use of this SBA head in glycerol-based LPA analogues has been published.

Regarding the lipophilic chain, the change in this part appeared to be more tolerated.²³ Naturally occurring LPA species consist of LPA having varied acyl chains of saturated and unsaturated types with lengths ranging from 14 to 22 carbons. These LPA species have differential biological activities, implying that different degrees of selectivity toward LPA receptors can be achieved by changing the acyl chain moiety.^{24,25} Furthermore, the more stable LPA analogues based on amide and ether isosteres proved to be active and potent LPA receptor agonists.^{13,23}

The *sn*-2 hydroxy group does not seem to be essential for the activation of, at least, some LPA receptors. Analogues having no hydroxy group at this position or masked OH group in form of methyl ether, short-chain ester, or cyclic phosphatidic acid have shown potent and sometimes selective activity.^{17,26–31} The glycerol backbone itself as a linker between the lipophilic tail and the phosphate head is not required for optimal agonistic activity since it was replaced with many scaffolds without much loss of ability to activate receptors. For example, oleyl thiophosphate, a fatty alcohol thiophosphate, has been reported to have partial agonistic activity toward LPA_{1,3} and full agonistic activity toward LPA₂, indicating the importance but not essentiality of the glycerol linker.²⁷ Replacing the glycerol core with other linkers, such as unsubstituted and small-sized substituted ethanolamines, has been reported as a

successful strategy for developing LPA receptors agonists.^{13,32,33}

Being a highly flexible molecule, LPA has several conformations to adopt, which explains the ability of LPA to activate many receptors. Strategies to restrict this flexibility by modifying the glycerol core into more rigid substructures have been reported in the field of lipid mediators. For example, LPA analogues containing conformationally restricted aryl substructures such as resorcinol and aminophenols have been synthesized and evaluated for their agonistic activity. However, none of these compounds showed equipotent activity to LPA.³⁴ In another study, a linker that secures a nonflattened three-dimensional arrangement of the LPA pharmacophore compared to the flattened aromatic systems previously examined was used by employing a tetrahydrofuran scaffold as a linker to synthesize a set of 2-oleoyl LPA analogues. Among this series, compound **T13** showed a very potent and selective LPA₃ agonistic activity (Figure 2).³⁵ A recent

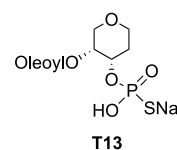
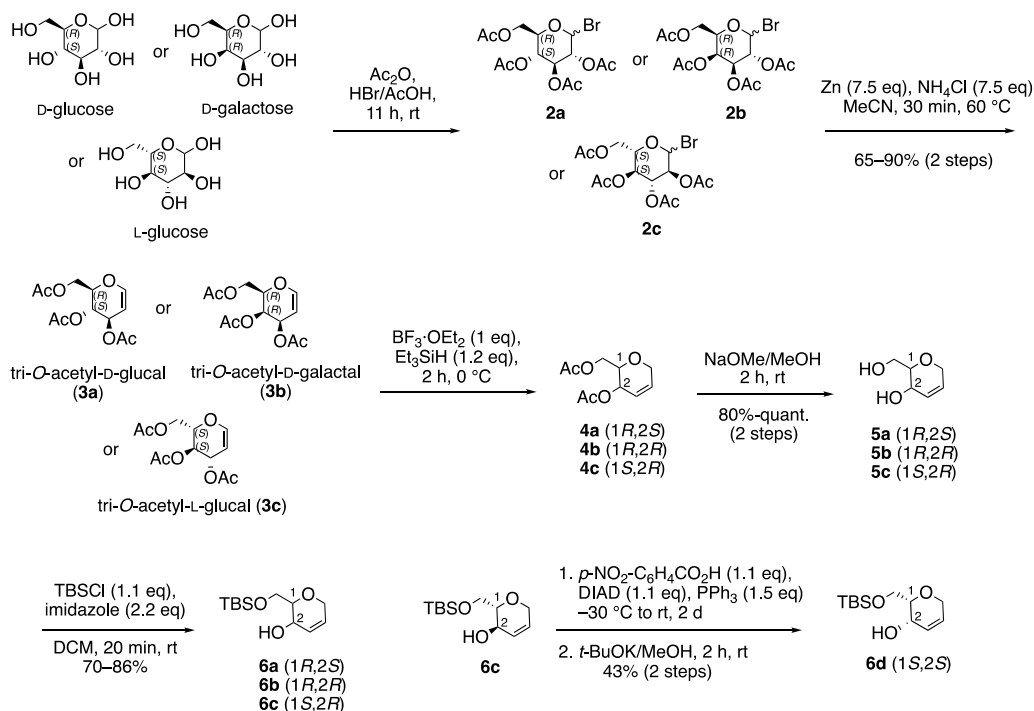


Figure 2. Structure of **T13**.

evaluation of **T13** activity against LPA_{1–6} using a more robust assay, TGF α shedding assay, showed that although **T13** has a subnanomolar activity on LPA₃; it also has more potent activity than 1-oleoyl LPA on LPA₅.³⁶ Aside from cyclic phosphatidic acids that contain a cyclic phosphate group,³¹ no previous report has been published using this strategy to design 1-oleoyl LPA (rather than 2-oleoyl LPA) analogues. Besides, many ligands have been designed and synthesized for LPA_{1–3}, but ligands for LPA_{4–6} are still limited.

Seeking to enhance our understanding of the three-dimensional arrangement of the LPA pharmacophore components and its relation to the LPA agonistic activity and trying to develop a selective and potent LPA₃ pharmacological probe, here, we describe the design and synthesis of a number of conformationally restricted 1-oleoyl

Scheme 1. Synthesis of the Common Intermediates 6a–d



LPA analogues by employing 2-hydroxymethyl-3-hydroxytetrahydropyran and 2-hydroxymethyl-3-hydroxy-3,6-dihydropyran as glycerol substitute linkers (Figure 1). These skeletons have two chiral centers and two regioisomers arising from the functionalization of two different hydroxy groups. A total of 16 compounds of all possible stereoisomers and regioisomers were initially synthesized to evaluate their selectivity, and then more analogues have been synthesized based on the molecular structure of the most potent one toward LPA₃.

RESULTS AND DISCUSSION

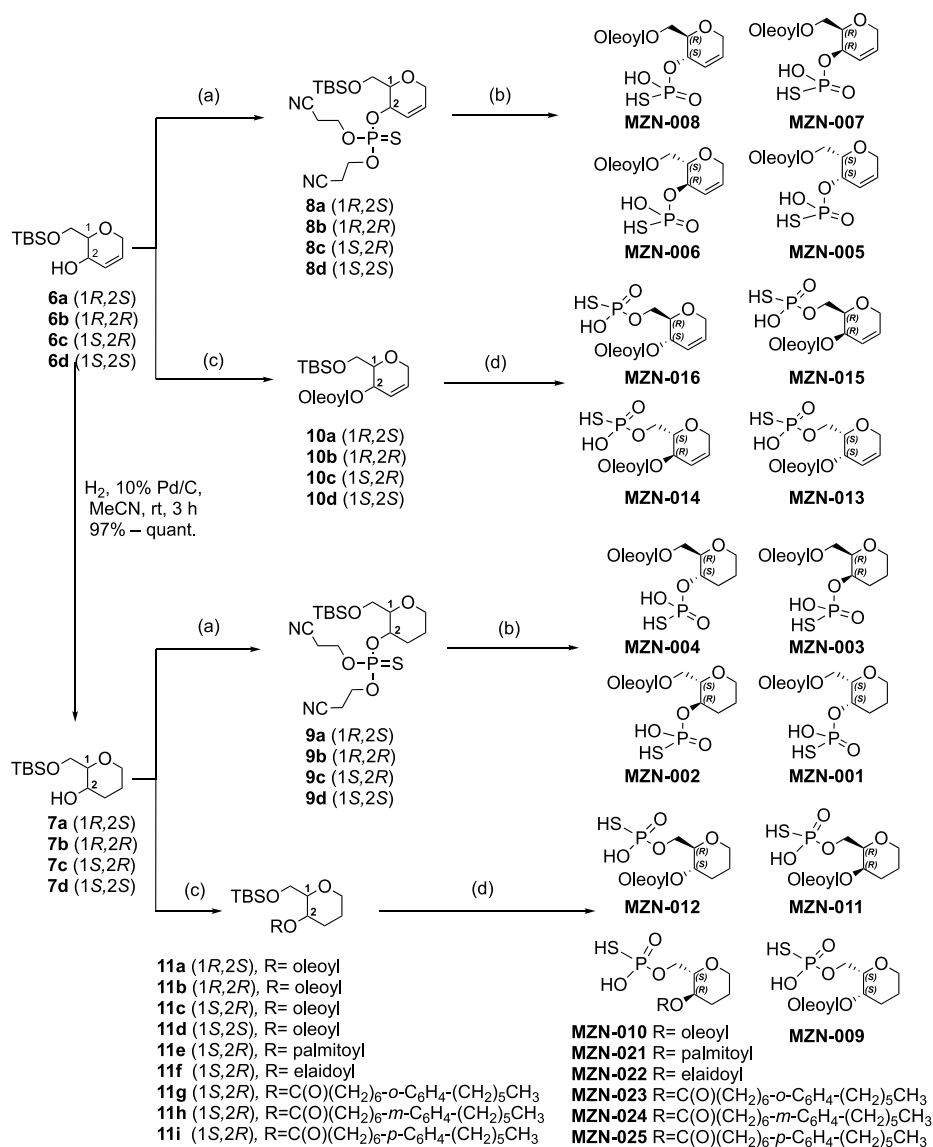
Molecular Design. As shown in Figure 1, 1-oleoyl LPA (1) is composed of two substructures, a polar phosphate moiety and a lipophilic chain, connected via a glycerol linker through ester bonds. Both substructures proved to be essential for LPA activity. Owing to its numerous rotatable bonds and, therefore, conformational flexibility, LPA can adopt many active conformations, which may explain its pan-agonistic effect.³⁵ We hypothesize that restricting this flexibility and removing the H-bond donating potential at the *sn*-2 hydroxy group by incorporating its glycerol moiety into more rigid skeletons than glycerol (namely, 2-hydroxymethyl-3-hydroxytetrahydropyran and 2-hydroxymethyl-3-hydroxy-3,6-dihydropyran) could generate molecules with selective agonistic activity. In addition, these conformational constraints could reduce the entropic cost arising from freezing the ligand inside the receptor into a specific conformation during the binding event, increasing its potency.³⁷

Incorporating glycerol into these two scaffolds could be achieved in two ways, generating two types of analogues, as depicted in Figure 1. Connecting carbon 3 of the glycerol core to the *sn*-2 hydroxy group produces 1°-acyl-*sn*-glycerol 2°-phosphate analogues while linking the hydroxy group with carbon 1 generates 2-acyl-*sn*-glycerol-1-phosphate analogues.³⁸ In addition, cyclization produces another stereogenic center in either case. We considered synthesizing all possible stereo-

isomers of both types of each scaffold (16 compounds) and evaluating their LPA_{1–6} agonistic activity. Bioisosteric replacement of phosphate into phosphorothioate has been previously reported to generate less readily hydrolyzable molecules.¹⁶ Using this strategy is expected to bring more metabolic stability to the designed molecules. Further analogues were synthesized based on the most potent analogues toward LPA₃ by varying the lipophilic chain or/and the linker.

Chemistry. *Synthesis of 2-Hydroxymethyl-3-hydroxytetrahydropyran and 2-Hydroxymethyl-3-hydroxy-3,6-dihydropyran-Based LPA Analogues.* We started the synthesis by employing enantiomerically pure monosaccharides, namely, D-glucose, D-galactose, and L-glucose, to synthesize the corresponding glycols 3a, 3b, and 3c (Scheme 1). Treating these monosaccharides separately with acetic anhydride followed by the addition of HBr/AcOH solution generated bromides 2a, 2b, and 2c, which in turn reduced to the corresponding tri-O-acetylated glycols using the Zn/NH₄Cl mixture.^{39,40}

The glycols 3a, 3b, and 3c were subjected to Ferrier rearrangement employing BF₃·OEt₂/Et₃SiH, followed by alcoholysis using NaOMe/MeOH to produce diols 5a, 5b, and 5c.³⁸ Selective protection of the primary alcohol of 5a, 5b, and 5c using a TBS-protecting group generated 6a, 6b, and 6c. To invert the stereochemistry of the carbon bearing secondary alcohol in 6c to produce its epimer 6d, Mitsunobu reaction was employed for this step, followed by base-promoted hydrolysis of ester product.⁴¹ The corresponding saturated analogues of 7a–7d were obtained by catalytic hydrogenation of 6a–6d over Pd/C (Scheme 2), respectively. The formation of thiophosphate diesters 8 and 9 was achieved by treating compounds 6 and 7, respectively, with phosphorodiamidite (CNCH₂CH₂OP[N(*i*Pr)₂]₂)/1H-tetrazole, followed by oxidation using elemental sulfur. Acid-catalyzed deprotection of silyl ethers 8 and 9 was carried out producing alcohols, which, in turn, were esterified by oleoyl chloride/DMAP, followed by β-

Scheme 2. Synthesis of MZN-001–MZN-016 and MZN-021–MZN-025^a

^aReaction conditions: (a) CNCH₂CH₂OP[N(*i*Pr)₂]₂ (2 equiv), ethylene cyanohydrin (1.6 equiv), 1*H*-tetrazole (2.5 equiv), then S (2 equiv), DCM, RT, 4 h, 66–88%. (b) 1. Amberlyst 15 (H⁺), MeOH, 14 h, RT; 2. Oleoyl chloride (1.5 equiv) (or carboxylic acid, DIC), DMAP (1.5 equiv), DCM, RT; 3. DBU, MSTFA, Pyridine, RT, 1 h, 21–58% (three steps). (c) Oleoyl chloride (1.5 equiv), DMAP (1.5 equiv), DCM, RT, or RCOOH, DIC, DMAP, DCM, RT, 55–97%. (d) 1. HF/Py, THF, RT, 16 h; 2. CNCH₂CH₂OP[N(*i*Pr)₂]₂ (2 equiv), ethylene cyanohydrin (1.6 equiv), 1*H*-tetrazole (2.5 equiv), and then S (2 equiv), DCM, RT, 4 h; 3. DBU, MSTFA, pyridine, RT, 1 h, 34–81% (three steps).

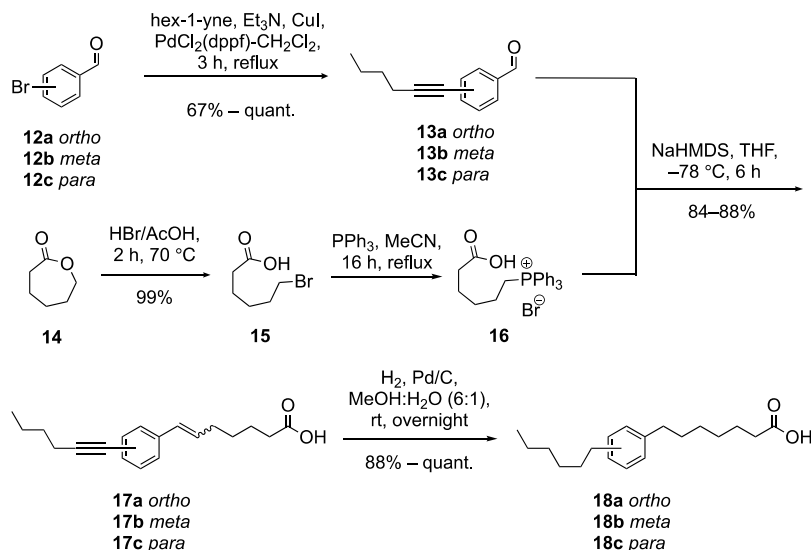
elimination of the cyanoethyl group to provide the target compounds 1°-acyl-*sn*-glycerol-2°-phosphate analogues **MZN-001** to **MZN-008**. To synthesize the 2°-acyl-*sn*-glycerol-1°-phosphate analogues (Scheme 2), we changed the reaction sequence, starting with the esterification of compounds **6** and **7** by the appropriate carboxylic acid or acid chloride followed by deprotection of the TBS group using HF/Py, and then thiophosphorylation of free alcohol and β-elimination of the cyanoethyl group, generating compounds **MZN-009** to **MZN-016**, **MZN-021**, and **MZN-022**.

Synthesis of Unnatural Fatty Acids. The synthesis of benzene-containing fatty acids **18a**, **18b**, and **18c** is summarized in Scheme 3.¹⁸ Aldehydes **13a**, **13b**, and **13c** were constructed by utilizing Sonogashira coupling of the corresponding aldehydes **12a–12c** and 1-hexyne. Stirring lactone **14** with 30% HBr/AcOH at room temperature for 1

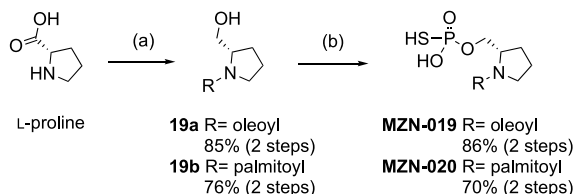
h produced compound **15**, which was treated with PPh₃ in refluxing MeCN to obtain triphenylphosphonium bromide **16** as the second component for the following Wittig reaction. A dropwise addition of **16** to a THF solution of NaHMDS at −78 °C, followed by the addition of aldehydes **13**, furnished the corresponding alkenes **17a**, **17b**, and **17c**. Catalytic hydrogenation of these enynes over Pd/C afforded the target acids **18a**, **18b**, and **18c**. These acids were utilized to synthesize **MZN-023** to **MZN-025** in a similar way that **MZN-009** to **MZN-016** were synthesized. However, rather than using the acid chloride for the esterification step, acids **18a–c** were directly esterified by stirring them with alcohol **7c** and DIC in DCM (Scheme 2).

Synthesis of Pyrrolidine-Based LPA Analogues. Because of the nucleophilicity difference between amine and alcohol functionalities, the synthesis of pyrrolidine-based LPA was

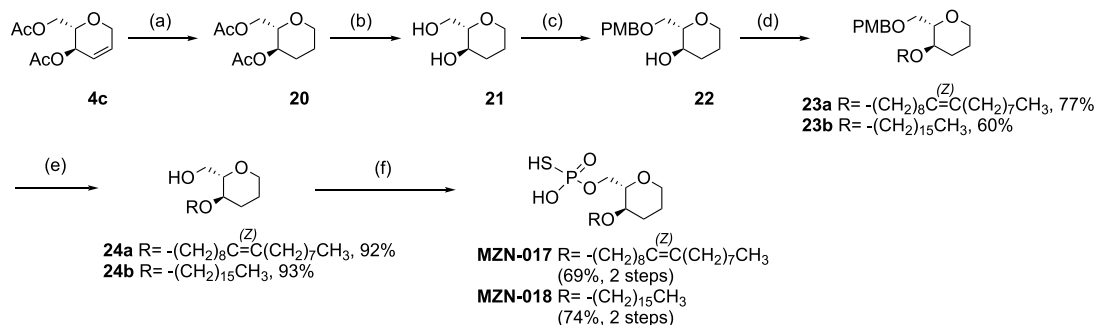
Scheme 3. Synthesis of Benzene-Containing Fatty Acids 18



much simpler, with no need for protecting groups. The synthesis commenced by reducing the amino acid *L*-proline using LiAlH₄ to generate *L*-prolinol, which was then directly coupled with oleoyl chloride and palmitoyl chloride to obtain compounds **19a** and **19b**, respectively (Scheme 4). Thiophosphorylation using the phosphorodiamidite/sulfur oxidation method, followed by the removal of the cyanoethyl group, yielded **MZN-019** and **MZN-020**.

Scheme 4. Synthesis of MZN-019 and MZN-020^a

^aReactions conditions: (a) 1. LiAlH₄, THF, reflux, 2 h; 2. Acid chloride, Et₃N (3 equiv), 0 °C to RT, 6 h. (b) 1. CNCH₂CH₂OP-[N(*i*Pr)₂]₂ (2 equiv), ethylene cyanohydrin (1.6 equiv), 1*H*-tetrazole (2.5 equiv), then S (2 equiv), DCM, RT, 4 h; 2. DBU, MSTFA, pyridine, RT, 30 min.

Scheme 5. Synthesis of MZN-017 and MZN-018^a

^aReactions conditions: (a) H₂, 10% Pd/C, MeOH, 99%. (b) NaOMe, MeOH, RT, 97%. (c) *n*-Bu₂SnO (1.3 equiv), MS3A, toluene, reflux, overnight, and then PMBCl (1.1 equiv), TBAI (1.3 equiv), reflux, 5 h, 73%. (d) KOH (4 equiv), RBr (2 equiv), toluene, reflux, overnight. (e) TfOH (0.5 equiv), resorcinol dimethyl ether (3 equiv), DCM, RT, 20 min. (f) 1. CNCH₂CH₂OP-[N(*i*Pr)₂]₂ (2 equiv), ethylene cyanohydrin (1.6 equiv), 1*H*-tetrazole (2.5 equiv), then S (2 equiv), DCM, RT, 4 h; 2. DBU, MSTFA, pyridine, RT, 30 min.

Table 1. Biological Evaluation of MZN-001–MZN-016 on LPA_{1–6} Using the TGF α Shedding Assay

	LPA ₁	LPA ₂	LPA ₃	LPA ₄	LPA ₅	LPA ₆
	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>
18:1-LPA (1)	7.6 nM (−8.12) [30%] <1>	13 nM (−7.90) [37%] <1>	73 nM (−7.14) [50%] <1>	21 nM (−7.70) [20%] <1>	78 nM (−7.11) [43%] <1>	35 nM (−7.45) [28%] <1>
MZN-001	1500 nM (−5.83) [27%] <0.0045>	1700 nM (−5.77) [29%] <0.0056>	41 nM (−7.38) [42%] <1.5>	170 nM (−6.76) [11%] <0.067>	390 nM (−6.41) [46%] <0.21>	NA (NA) [NA] <NA>
MZN-002	NA (NA) [NA] <NA>	2300 nM (−5.63) [11%] <0.0016>	13 nM (−7.89) [48%] <5.4>	160 nM (−6.80) [17%] <0.11>	110 nM (−6.96) [50%] <0.84>	NA (NA) [NA] <NA>
MZN-003	1600 nM (−5.79) [9.5%] <0.0015>	>10 μ M (>−4) [ND] <ND>	57 nM (−7.25) [30%] <0.77>	180 nM (−6.74) [17%] <0.095>	230 nM (−6.64) [46%] <0.37>	NA (NA) [NA] <NA>
MZN-004	530 nM (−6.27) [22%] <0.01>	1600 nM (−5.81) [16%] <0.0036>	37 nM (−7.43) [42%] <1.7>	130 nM (−6.89) [22%] <0.17>	73 nM (−7.14) [30%] <0.75>	980 nM (−6.01) [10%] <0.013>
MZN-005	250 nM (−6.60) [32%] <0.032>	740 nM (−6.13) [30%] <0.014>	11 nM (−7.97) [46%] <6.3>	85 nM (−7.07) [15%] <0.18>	64 nM (−7.19) [53%] <1.5>	490 nM (−6.31) [13%] <0.034>
MZN-006	440 nM (−6.36) [25%] <0.014>	1900 nM (−5.73) [28%] <0.0051>	4.7 nM (−8.33) [54%] <17>	48 nM (−7.32) [17%] <0.37>	36 nM (−7.45) [51%] <2.6>	590 nM (−6.23) [4.1%] <0.0087>
MZN-007	1300 nM (−5.9) [18%] <0.0035>	>10 μ M (>−4) [ND] <ND>	11 nM (−7.94) [53%] <6.8>	300 nM (−6.53) [19%] <0.064>	160 nM (−6.79) [54%] <0.61>	1500 nM (−5.81) [26%] <0.021>
MZN-008	71 nM (−7.15) [27%] <0.095>	500 nM (−6.30) [35%] <0.024>	2.6 nM (−8.59) [54%] <30>	26 nM (−7.59) [17%] <0.69>	59 nM (−7.23) [37%] <1.1>	680 nM (−6.17) [28%] <0.051>
MZN-009	470 nM (−6.33) [25%] <0.013>	8200 nM (−5.09) [74%] <0.0031>	73 nM (−7.13) [45%] <0.89>	300 nM (−6.52) [4.6%] <0.016>	34 nM (−7.46) [54%] <2.9>	NA (NA) [NA] <NA>
MZN-010	590 nM (−6.23) [21%] <0.009>	>10 μ M (>−4) [ND] <ND>	51 nM (−7.29) [42%] <1.2>	24 nM (−7.62) [18%] <0.74>	4.9 nM (−8.31) [52%] <19>	1100 nM (−5.96) [35%] <0.041>
MZN-011	210 nM (−6.68) [7.2%] <0.0086>	>10 μ M (>−4) [ND] <ND>	50 nM (−7.30) [37%] <1.1>	13 nM (−7.89) [19%] <1.5>	34 nM (−7.47) [53%] <2.9>	110 nM (−6.98) [29%] <0.35>
MZN-012	310 nM (−6.51) [31%] <0.025>	1400 nM (−5.85) [33%] <0.0079>	12 nM (−7.91) [48%] <5.7>	36 nM (−7.44) [18%] <0.5>	23 nM (−7.65) [51%] <4.1>	190 nM (−6.71) [25%] <0.16>
MZN-013	710 nM (−6.15) [33%] <0.012>	2000 nM (−5.70) [26%] <0.0045>	43 nM (−7.37) [51%] <1.8>	390 nM (−6.41) [15%] <0.04>	26 nM (−7.58) [55%] <3.8>	700 nM (−6.15) [6.9%] <0.012>
MZN-014	290 nM	1300 nM	18 nM	2.3 nM	9.1 nM	370 nM

Table 1. continued

	LPA ₁	LPA ₂	LPA ₃	LPA ₄	LPA ₅	LPA ₆
MZN-015	(−6.54)	(−5.88)	(−7.75)	(−8.64)	(−8.04)	(−6.44)
	[26%]	[41%]	[46%]	[17%]	[50%]	[13%]
	<0.023>	<0.011>	<3.8>	<7.6>	<9.9>	<0.045>
	230 nM	860 nM	10 nM	8.1 nM	27 nM	100 nM
MZN-016	(−6.63)	(−6.06)	(−7.98)	(−8.09)	(−7.57)	(−7.00)
	[27%]	[39%]	[49%]	[22%]	[53%]	[32%]
	<0.029>	<0.015>	<6.9>	<2.8>	<3.6>	<0.39>
	61 nM	270 nM	2.5 nM	24 nM	9.5 nM	49 nM
	(−7.22)	(−6.57)	(−8.60)	(−7.62)	(−8.02)	(−7.31)
	[32%]	[39%]	[55%]	[22%]	[53%]	[32%]
	<0.13>	<0.049>	<32>	<0.93>	<10>	<0.82>

donating potential of *sn*-2-OH of these compounds. In contrast, all these compounds exhibited overall potent agonistic activities toward LPA_{3,4,5} with those having *trans* configuration (even numbers) exhibiting more potent activity compared to the corresponding *cis* isomers (odd numbers). It is also noted that dihydropyran-containing compounds generally showed more potent agonistic activities toward all LPA receptors than the tetrahydropyran-containing analogues (MZN-001–004 vs MZN-005–008; MZN-009–012 vs MZN-013–016).

1°-Acyl-*sn*-glycerol-2°-phosphate analogues MZN-001–008 were more potent toward LPA₃ and less potent or equipotent toward other receptors than **1**. Specifically, MZN-007 had an EC₅₀ of 11 nM toward LPA₃ compared to that of 73 nM for **1**. It was significantly less potent toward LPA_{1,2,4,6} and twofold less potent toward LPA₅ than **1**. MZN-002 demonstrated a close activation potency toward the six LPA receptors to MZN-007 while MZN-008 showed very strong activity at LPA₃ (EC₅₀ = 2.6 nM) with comparable activity to parent LPA at LPA_{4,5}. Again, those analogues of this type with *trans* configuration were more potent toward LPA₃. On the other hand, 2°-acyl-*sn*-glycerol-1°-phosphate analogues showed more potent activities toward LPA_{3,4,5} and less potent activities toward LPA_{1,2,6} than 1-oleoyl LPA. One such analogue is MZN-014, which exhibited a strong potency on LPA_{3,4,5}, approximately 4- to 10-fold more potent than parent LPA. This difference in selectivity between the two types could be attributed to the free movement of the phosphate head in 2°-acyl-*sn*-glycerol-1°-phosphate analogues compared to 1°-acyl-*sn*-glycerol-2°-phosphate analogues, which could adjust itself precisely inside the binding pocket, activating potently more than one receptor. Among all these analogues, MZN-010 was the most potent LPA₅ agonist with EC₅₀ = 4.9 nM, approximately 16 times more potent than the standard **1** with comparable potency toward LPA_{3,4}.

Interestingly, MZN-005 and its enantiomer MZN-007 were equally effective at activating LPA₃, and similarly, MZN-014 and its enantiomer MZN-016 were equally effective against LPA₅. Other enantiomer pairs showed very close values (MZN-006 and MZN-008 on LPA₃, and MZN-013 and MZN-015 on LPA₃), which may indicate a lack of stereospecific recognition of this set of compounds by the mentioned receptors. A similar pattern of lacking stereospecific recognition has been noticed for the natural LPA where the natural one and its unnatural enantiomer demonstrated almost equal activity in some assays.⁴⁴ On the contrary, nonglycerol-based LPA analogues tend to exhibit a stereoselective preference for LPA receptors.¹²

Motivated by the fact that there are no selective pharmacological tools for LPA₅, we chose the skeleton of MZN-010, being the most potent at LPA₅, to be a starting point to synthesize more analogues in an attempt to enhance the potency and selectivity at LPA₅. We hypothesized that the dihedral angle between the phosphate group and the lipophilic chain might be an important determinant for selectivity. Therefore, the synthesis of other analogues bearing a more rigid structure with a relatively close dihedral angle to that shown by MZN-010 should enhance both selectivity and potency. To examine this hypothesis, we synthesized MZN-019 and MZN-020 utilizing a pyrrolidine ring, which could achieve a close dihedral angle, as depicted in Figure 3, and

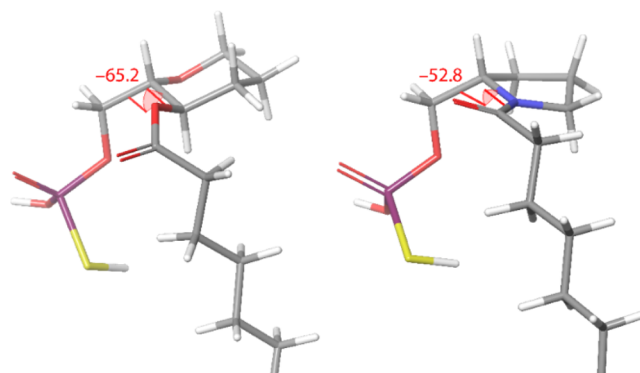


Figure 3. Dihedral angles of compounds MZN-010 (left) and MZN-020 (right) (measured after being energy-minimized using Maestro).

examined their potency against LPA_{1–6}. As expected, both ligands showed very potent activity toward LPA₅ compared to **1** and MZN-010 with a slight preference for the ligand bearing the saturated fatty acid chain (MZN-020 EC₅₀ at LPA₅ = 1.1 nM) over the unsaturated one MZN-019 (see Tables 2 and 3). However, the enhancement of activity at LPA₅ was associated with simultaneous enhancement at LPA_{3,4}. A previous report demonstrated that LPA₅ had a preference for alkyl LPA over acyl LPA;⁴⁵ hence, we speculated that the high electron density of the amide and ester functionalities could contribute to the activation of LPA_{3,4} but not LPA₅, which may explain the enhancement shown by MZN-019 and MZN-020 at LPA_{3,4}. Accordingly, we considered the synthesis of both saturated (16:0) and unsaturated (18:1) chain ether analogues based on the MZN-010 structure. We also considered changing the lipophilic chain of the ester analogues by changing the alkene configuration in the oleic acid, examining another unsaturated system, namely, a benzene ring, which could interact with the

Table 2. Biological Evaluation of the First Set of MZN-010 Analogues on LPA₁₋₆ Using the TGF α Shedding Assay

	LPA ₁	LPA ₂	LPA ₃	LPA ₄	LPA ₅	LPA ₆
	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>
18:1-LPA (1)	28 nM (−7.55) [30%] <1>	32 nM (−7.50) [38%] <1>	220 nM (−6.67) [45%] <1>	18 nM (−7.76) [15%] <1>	160 nM (−6.80) [40%] <1>	42 nM (−7.38) [25%] <1>
MZN-017	940 nM (−6.03) [5.4%] <0.0054>	>10 μ M (>−4) [ND] <ND>	85 nM (−7.07) [34%] <1.9>	94 nM (−7.03) [13%] <0.16>	13 nM (−7.89) [44%] <13>	260 nM (−6.58) [24%] <0.16>
MZN-018	NA (NA) [NA] <NA>	1000 nM (−5.98) [16%] <0.012>	170 nM (−6.77) [23%] <0.65>	39 nM (−7.4) [6.4%] <0.19>	3.8 nM (−8.42) [43%] <44>	330 nM (−6.48) [12%] <0.06>
MZN-019	270 nM (−6.58) [26%] <0.09>	1600 nM (−5.80) [42%] <0.022>	26 nM (−7.58) [52%] <9.4>	8.2 nM (−8.08) [16%] <2.3>	3 nM (−8.52) [43%] <55>	340 nM (−6.47) [34%] <0.17>

Table 3. Biological Evaluation of the Second Set of MZN-010 Analogues on LPA₁₋₆ Using the TGF α Shedding Assay

	LPA ₁	LPA ₂	LPA ₃	LPA ₄	LPA ₅	LPA ₆
	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>	EC ₅₀ (pEC ₅₀) [Emax] <RIA>
18:1-LPA (1)	9.5 nM (−8.02) [28%] <1>	17 nM (−7.78) [48%] <1>	70 nM (−7.15) [57%] <1>	15 nM (−7.83) [21%] <1>	60 nM (−7.23) [44%] <1>	13 nM (−7.88) [31%] <1>
MZN-020	450 nM (−6.35) [22%] <0.017>	630 nM (−6.20) [44%] <0.024>	30 nM (−7.53) [56%] <2.4>	20 nM (−7.70) [19%] <0.67>	1.1 nM (−8.96) [52%] <64>	290 nM (−6.54) [24%] <0.036>
MZN-021	320 nM (−6.49) [7.5%] <0.008>	>10 μ M (>−4) [ND] <ND>	140 nM (−6.84) [31%] <0.27>	55 nM (−7.26) [21%] <0.27>	1.2 nM (−8.91) [51%] <55>	380 nM (−6.42) [19%] <0.022>
MZN-022	480 nM (−6.32) [16%] <0.011>	>10 μ M (>−4) [ND] <ND>	78 nM (−7.11) [48%] <0.77>	34 nM (−7.47) [23%] <0.49>	6.3 nM (−8.20) [51%] <11>	450 nM (−6.35) [11%] <0.011>
MZN-023	1100 nM (−5.97) [15%] <0.0049>	>10 μ M (>−4) [ND] <ND>	310 nM (−6.50) [27%] <0.11>	380 nM (−6.42) [18%] <0.034>	44 nM (−7.35) [55%] <1.7>	>10 μ M (>−4) [ND] <ND>
MZN-024	630 nM (−6.20) [18%] <0.0097>	>10 μ M (>−4) [ND] <ND>	240 nM (−6.63) [40%] <0.21>	58 nM (−7.24) [22%] <0.28>	8.1 nM (−8.09) [53%] <9>	>10 μ M (>−4) [ND] <ND>
MZN-025	370 nM (−6.43) [13%] <0.012>	>10 μ M (>−4) [ND] <ND>	260 nM (−6.58) [10%] <0.05>	49 nM (−7.31) [23%] <0.34>	11 nM (−7.95) [51%] <6.1>	>10 μ M (>−4) [ND] <ND>

target to form π -stacking or CH– π interactions, as well as testing saturated analogues.

The results of the new set of compounds are summarized in Table 3. The elaidoyl derivative MZN-022 showed almost

equipotent activity to MZN-10 at LPA₁₋₆, indicating that the configuration of the double bond in the lipophilic chain had no significant effect on the activity. However, replacing this unsaturated system with a benzene moiety results in different

patterns of activity and potency. All benzene derivatives **MZN-023–025** lost their activities against LPA_{2,6}. Furthermore, the *ortho* analogue **MZN-023** exhibited much weaker activity at LPA_{1,3,4} and slightly stronger activity at LPA₅ with an EC₅₀ value of 44 nM for LPA₅ compared to 1-oleoyl LPA's EC₅₀ value of 60 nM. On the other hand, *meta* derivative **MZN-024** and *para* derivative **MZN-025** showed much stronger activity at LPA_{4,5} than **MZN-023**. Replacing the unsaturated chain with the saturated one had a positive effect on LPA₅ activation in both ether and ester analogues with approximately three- to fourfold increases in potency compared to the unsaturated ones. For example, the palmityl derivative **MZN-018** showed a potent LPA₅ agonistic activity (EC₅₀ = 3.8 nM compared to 13 nM for oleyl ether **MZN-017**) with a partial agonistic effect on both LPA_{3,4}. Its ester analogue **MZN-021** was the most potent analogue at LPA₅ with an EC₅₀ value of 1.2 nM, 50 times more potent than **1**. In addition, it was approximately 2 and 3.5 times less potent at LPA_{3,4} than **1** with a partial agonistic effect at LPA₃ and no activity at LPA₂. The results demonstrate that **MZN-021** displayed robust and selective agonistic activity targeting LPA₅, exhibiting over a 45-fold higher selectivity when compared with other LPA receptors. Ether derivatives **MZN-017** and **MZN-018** exhibited lower maximum effects (E_{max}) on LPA₄ than the corresponding ester analogues **MZN-10** and **MZN-021**, suggesting that groups with high electron densities, such as esters and amides, could contribute to a higher E_{max} by increasing the stability of the ligand/LPA₄ complex by acting as hydrogen bond acceptors.

CONCLUSIONS

In this study, a group of LPA analogues containing dihydropyran and tetrahydropyran skeletons were designed, synthesized, and biologically evaluated for their agonistic effect using the TGF α shedding assay. By employing conformational restriction, bioisosteric replacement, and lipophilic chain variation, we succeeded in developing compounds **MZN-018** and **MZN-021** with improved potency and selectivity toward LPA₅ compared to 1-oleoyl LPA (**1**) (EC₅₀ = 3.8, 1.2, and 60 nM, respectively). The results of this study support the idea that the high flexibility of LPA is responsible for its pan-agonistic activity. Restricting this flexibility by using ring systems as a core linker between the phosphate and the acyl or alkyl lipophilic chain alters the agonistic activity toward LPA_{1–6}. The findings also clearly indicate that different lipophilic chains have different levels of agonistic activity. This could potentially have important implications for further study and understanding of compounds' selectivity toward LPA receptors. Research is ongoing in our laboratory to enhance the potency further by varying the lipophilic chains.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.3c07668>.

Experimental procedures, dose–response curves, and copies of ¹H and ¹³C NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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