



Iron Corrosion via Direct Metal-Microbe Electron Transfer

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ABSTRACT The concept that anaerobic microorganisms can directly accept electrons from Fe(0) has been controversial because direct metal-microbe electron transfer has previously only been indirectly inferred. Fe(0) oxidation was studied with *Geobacter sulfurreducens* strain ACL, an autotrophic strain that was previously shown to grow with electrons derived from a graphite cathode as the sole electron donor. Strain ACL grew with Fe(0) as the sole electron donor and fumarate as the electron acceptor. However, it appeared that at least a portion of the electron transfer was via H₂ produced nonenzymatically from the oxidation of Fe(0) to Fe(II). H₂, which accumulated in abiotic controls, was consumed during the growth of strain ACL, the cells were predominately planktonic, and genes for the uptake hydrogenase were highly expressed. Strain ACL_{HF} was constructed to prevent growth on H₂ or formate by deleting the genes for the uptake of hydrogenase and formate dehydrogenases from strain ACL. Strain ACL_{HF} also grew with Fe(0) as the sole electron donor, but H₂ accumulated in the culture, and cells heavily colonized Fe(0) surfaces with no visible planktonic growth. Transcriptomics suggested that the outer surface c-type cytochromes OmcS and OmcZ were important during growth of strain ACL_{HF} on Fe(0). Strain ACL_{HF} did not grow on Fe(0) if the gene for either of these cytochromes was deleted. The specific attachment of strain ACL_{HF} to Fe(0), coupled with requirements for known extracellular electrical contacts, suggest that direct metal-microbe electron transfer is the most likely option for Fe(0) serving as an electron donor.

IMPORTANCE The anaerobic corrosion of iron structures is expensive to repair and can be a safety and environmental concern. It has been known for over 100 years that the presence of anaerobic respiratory microorganisms can accelerate iron corrosion. Multiple studies have suggested that there are sulfate reducers, methanogens, and acetogens that can directly accept electrons from Fe(0) to support sulfate or carbon dioxide reduction. However, all of the strains studied can also use H₂ as an electron donor for growth, which is known to be abiotically produced from Fe(0). Furthermore, no proteins definitely shown to function as extracellular electrical contacts with Fe(0) were identified. The studies described here demonstrate that direct electron transfer from Fe(0) can support anaerobic respiration. They also map out a simple genetic approach to the study of iron corrosion mechanisms in other microorganisms. A better understanding of how microorganisms promote iron corrosion is expected to lead to the development of strategies that can help reduce adverse impacts from this process.

KEYWORDS *Geobacter*, autotroph, cytochrome, electromicrobiology, extracellular electron transfer, zero-valent iron

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The mechanisms by which microorganisms accelerate the corrosion of iron are of interest because of the costs and threats to the environment and human safety associated with the corrosion of steel (1–3). Iron corrodes when metallic iron [Fe(0)] is oxidized to Fe(II):



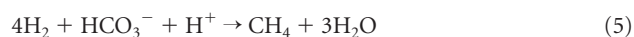
This oxidation reaction must be coupled with a corresponding reduction reaction, which under anaerobic conditions is typically the reduction of protons to produce H₂:



With the net reaction:



Microbial removal of H₂ theoretically favors additional H₂ formation and iron corrosion (3). Thus, H₂-consuming microorganisms, such as sulfate-reducing (equation 4), methane-producing (equation 5), and acetogenic bacteria (equation 6), may have an important influence on corrosion (4–6).



Removal of Fe(II), the other product of Fe(0) oxidation, also favors Fe(0) oxidation. Microbial production of sulfide during sulfate reduction provides an Fe(II) sink:



Thus, investigations into the mechanisms of corrosion in the presence of sulfate reducers must account for this influence on corrosion rates.

A number of studies have suggested that the most important microbial contribution to corrosion is direct metal-microbe electron transfer, in which the electrons derived from Fe(0), rather than a H₂ intermediate, serve as the electron donor for anaerobic respiration (1, 2, 4–9). It is important to recognize that there was no direct demonstration of direct electron transfer in any of these studies. All of the microbes proposed to directly accept electrons from Fe(0) could use H₂ as an electron donor, and the possibility of H₂ serving as an electron carrier during Fe(0) oxidation was not ruled out. Rather, the claim for direct electron transfer was based on the observation that the microbes that were proposed to be capable of direct electron transfer oxidized Fe(0) more quickly than did other closely related H₂-utilizing strains.

However, the enrichment of the microbes proposed to be capable of direct electron transfer on Fe(0) may have selected for other characteristics that promote Fe(0) oxidation with the production of H₂ (10). For example, many of the microbes proposed to be capable of direct electron transfer from Fe(0) appear to be more effective in colonizing surfaces, which may result in more effective H₂ removal at the point of production. Another consideration is that the slow release of H₂ from Fe(0) oxidation may enrich for strains with higher affinities for H₂, thereby enhancing H₂ uptake at the Fe(0) surface compared with closely related strains that were enriched and isolated with high concentrations of H₂.

Studies with microbial strains incapable of using H₂ as an electron donor for growth have more definitively demonstrated direct electron transfer from other types of insoluble electron donors. Direct electron transfer from graphite electrodes to support anaerobic respiration was demonstrated with *Geobacter metallireducens*, which is unable to respire H₂ (11, 12), and a strain of *G. sulfurreducens* in which the gene for the uptake hydrogenase (13) was deleted (14). In a similar manner, construction of a strain that eliminated the possibility that H₂ or formate could function as an electron donor for *G. sulfurreducens* clarified its ability to participate as the electron-accepting partner in direct interspecies electron transfer (15, 16).

Multiheme outer surface c-type cytochromes are key electrical contacts between

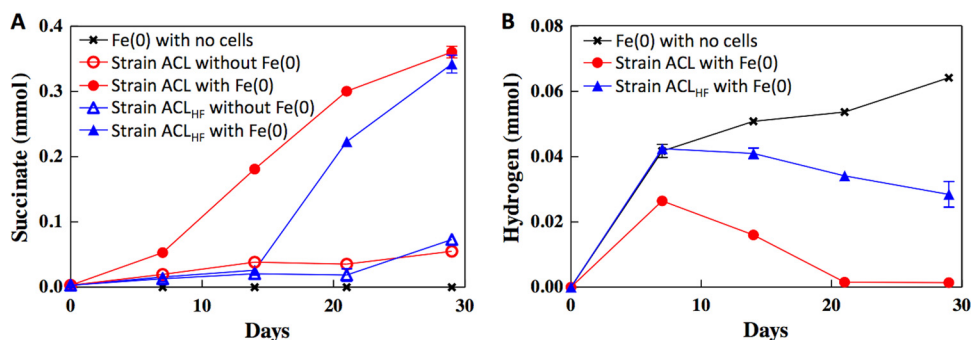


FIG 1 (A and B) Succinate (A) and hydrogen (B) concentrations over time when strain ACL and strain ACL_{HIF} were grown with and without Fe(0) as the potential electron donor and fumarate as the electron acceptor. Error bars represent one standard deviation of the mean of the results from triplicate cultures.

Geobacter species and other species, electrodes, and minerals (17–19). Some microbes proposed to be capable of directly accepting electrons from Fe(0) have multiheme outer surface *c*-type cytochromes (20, 21). However, the ability of cytochromes to function as the direct electrical contacts on the outer surface has not been demonstrated.

The recent construction of a strain of *G. sulfurreducens* capable of autotrophic growth (22) has provided the opportunity to further explore the possibility of direct electron transfer from Fe(0) in a genetically tractable microbe known to be highly effective in extracellular electron exchange. Here, we report evidence for direct electron transfer from Fe(0) under conditions in which the possibility of H₂ (or formate) serving as an intermediate electron carrier has been eliminated, and we identify likely multiheme *c*-type cytochrome electrical contacts with Fe(0).

RESULTS AND DISCUSSION

H₂ as an electron carrier during growth of strain ACL with Fe(0) as the electron donor. *G. sulfurreducens* strain ACL grew in medium with Fe(0) as the sole electron donor with the reduction of fumarate to succinate (Fig. 1). The culture was sequentially transferred (5% inoculum) with similar rates of succinate production. No succinate was produced in the controls of cells without Fe(0) or medium with Fe(0) but no cells (Fig. 1A).

H₂ accumulated in the presence of Fe(0) when cells were not added (Fig. 1B), consistent with the well-known abiotic corrosion of Fe(0) (23). There was no H₂ production in the absence of Fe(0). In the presence of cells of strain ACL and Fe(0), H₂ accumulated and then declined, indicating that H₂ was being consumed in the presence of strain ACL.

Cultures of strain ACL growing on Fe(0) were visibly turbid (Fig. 2), and few cells were associated with the Fe(0) particles (Fig. 3B and D). These results suggested that strain ACL was primarily growing with H₂ as the electron donor because cells do not need to attach to the Fe(0) particle surface to metabolize H₂. This is similar to previous studies in which *G. sulfurreducens* did not attach during syntrophic growth with a H₂-producing partner (16). In contrast, *G. sulfurreducens* grows in aggregates with an electron-donating partner when electrons are delivered via direct interspecies electron transfer (15, 16, 24).

Direct electron transfer from Fe(0) with strain ACL_{HIF}. Previous studies have demonstrated that deletion of the gene for the uptake hydrogenase (*hybL*) and the formate dehydrogenase (*fdnG*) yielded a strain of *G. sulfurreducens* that could not grow with H₂ or formate as the electron donor (16). Therefore, in order to better evaluate the possibility of direct electron transfer from Fe(0), the previously described (16) *G. sulfurreducens* Δ *fdnG* Δ *hybL* mutant strain unable to use H₂ or formate (16) was modified with the insertion of the *aclA* and *aclB* genes for citrate lyase in the same

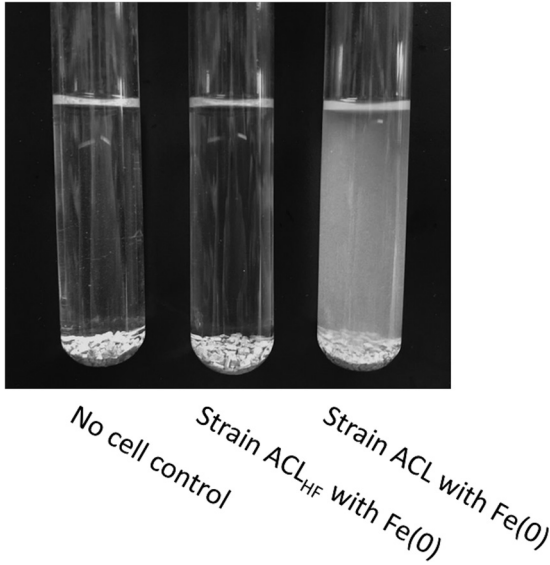


FIG 2 Appearance of cultures after growth of strain ACL or strain ACL_{HF} in medium with Fe(0) provided as the electron donor and fumarate provided as the electron acceptor.

manner previously described (22) to construct strain ACL (Fig. S1). This new strain was designated strain ACL_{HF}.

When strain ACL_{HF} was inoculated into medium with Fe(0) as the electron donor and fumarate as the electron acceptor, succinate production lagged initially but once

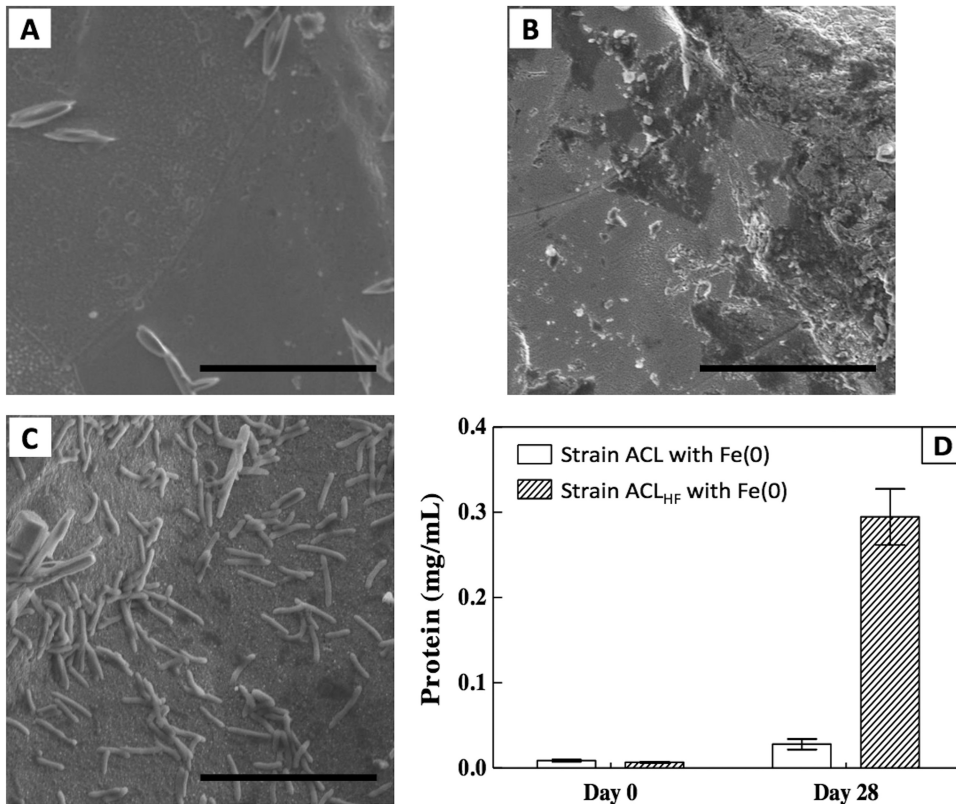


FIG 3 Scanning electron micrograph images of Fe(0) particles. (A) No-cell control. (B) Strain ACL after 28 days of incubation. (C) Strain ACL_{HF} after 28 days of incubation. Scale bar = 10 μ m. (D) Protein concentrations detected on Fe(0) particles. Error bars represent one standard deviation of the mean of the results from triplicate samples.

TABLE 1 Ten most highly upregulated genes in ACL_{HF} strain compared to the ACL strain^a

Name	Annotation	Abbreviation	Main role	Specific role	ACL _{HF} log ₂ RPKM	ACL log ₂ RPKM	Fold upregulated in ACL _{HF}
GSU3506	DUF2917 domain protein		Unknown function		3.99	0.62	17.22
GSU0012	Protoporphyrinogen oxidase	<i>hemG</i>	Biosynthesis of cofactors,	Heme, porphyrin,	4.95	2.41	7.67
			prosthetic groups, and carriers	and cobalamin			
GSU0769	Protein RarD	<i>rarD</i>	Transport and binding proteins	Unknown substrate	3.08	1.11	5.96
GSU0013	Transcriptional regulator, MarR family	<i>marR</i>	Regulatory functions	DNA interactions	4.20	2.40	5.84
GSU2503	c-type cytochrome	<i>omcT</i>	Energy metabolism	Electron transport	4.50	2.06	5.32
GSU2504	c-type cytochrome	<i>omcS</i>	Energy metabolism	Electron transport	4.97	2.80	4.45
GSU0018	Transcriptional regulator, GntR family	<i>gntR</i>	Regulatory functions	DNA interactions	3.77	2.26	4.22
GSU3410	Putative membrane protein		Unknown function		8.35	6.83	3.11
GSU3395	L-Proline dehydrogenase	<i>putA</i>	Energy metabolism	Amino acids and amines	2.93	1.99	2.86
GSU3409	Putative membrane protein		Unknown function		5.07	3.84	2.64

^aGenes that were significantly upregulated in the ACL_{HF} strain compared to the ACL strain when grown on Fe(0) with fumarate as the electron acceptor. The median log₂ RPKM value for the ACL_{HF} strain was 1.53, and the median log₂ RPKM value for the ACL strain was 3.31.

initiated was more rapid than succinate production by strain ACL (Fig. 1A). Unlike strain ACL, strain ACL_{HF} did not consume substantial H₂ during the reduction of fumarate to succinate (Fig. 1B).

In contrast to the turbid strain ACL cultures, cultures of strain ACL_{HF} lacked turbidity despite comparable succinate production (Fig. 2). Scanning electron microscopy revealed that over time, strain ACL_{HF} heavily colonized the Fe(0) particles (Fig. 3C), and protein concentrations on the Fe(0) surface also increased over time (Fig. 3D). Attachment is required for direct electron transfer (18). These results suggest that strain ACL_{HF} adapted to the inability to use H₂ as an electron donor by attaching to Fe(0) for Fe(0) oxidation.

Transcriptomic and genetic analyses of direct electron transfer mechanisms. In order to gain further insight into the potential mechanisms for electron transfer from Fe(0), the transcriptomes of strain ACL and strain ACL_{HF} grown on Fe(0) were analyzed (Table S1). In strain ACL, log₂ reads per kilobase per million (RPKM) values for genes associated with the uptake hydrogenase complex (*hybS*, *hybA*, *hybB*, *hybL*, *hybP*, and *hybT*) ranged from 4.4 ± 0.2 to 8.1 ± 0.1, which was substantially higher than the median log₂ RPKM value for genome-wide expression of strain ACL (2.5 ± 0.1). In contrast, log₂ RPKM values for genes coding for formate dehydrogenase (*fdnG*, *fdnH*, and *fdnI*) ranged from 1.03 ± 0.5 to 2.03 ± 0.5 and were well below the median log₂ RPKM. These results further suggest that H₂ was an important intermediary carrier for electron transfer from Fe(0) to strain ACL but that formate was not.

The median log₂ RPKM value for the ACL_{HF} cells (1.3 ± 0.6) was substantially lower than the median log₂ RPKM value for ACL cells (2.5 ± 0.1), yet the transcript abundances for the genes for the multiheme c-type cytochromes OmcS and OmcT were more than 4-fold higher in strain ACL_{HF} than in strain ACL (Table 1). *omcS* and *omcT* are adjacent on the *G. sulfurreducens* chromosome (25). *omcT* is cotranscribed with *omcS*, but *omcS* can also be transcribed separately (25). OmcS is one of the most abundant outer surface proteins during growth on Fe(III) and Mn(IV) oxides (25–27); *omcS* is highly expressed when *G. sulfurreducens* functions as the electron-accepting partner for direct interspecies electron transfer in coculture with *G. metallireducens* (15), and OmcS is important for electron transfer to anodes under some conditions (28). In contrast, OmcT is in low abundance under all growth conditions that have been evaluated (25, 27). None of the other genes with higher expression in the ACL_{HF} strain had annotations that suggested that they could have a direct function in electron transfer from Fe(0) (Table 1).

When *omcS* was deleted, the mutant strain failed to grow on Fe(0) (Fig. 4). The capacity for growth on Fe(0) was restored when *omcS* expression was complemented in *trans* (Fig. 4). These results suggested that OmcS is an important component in

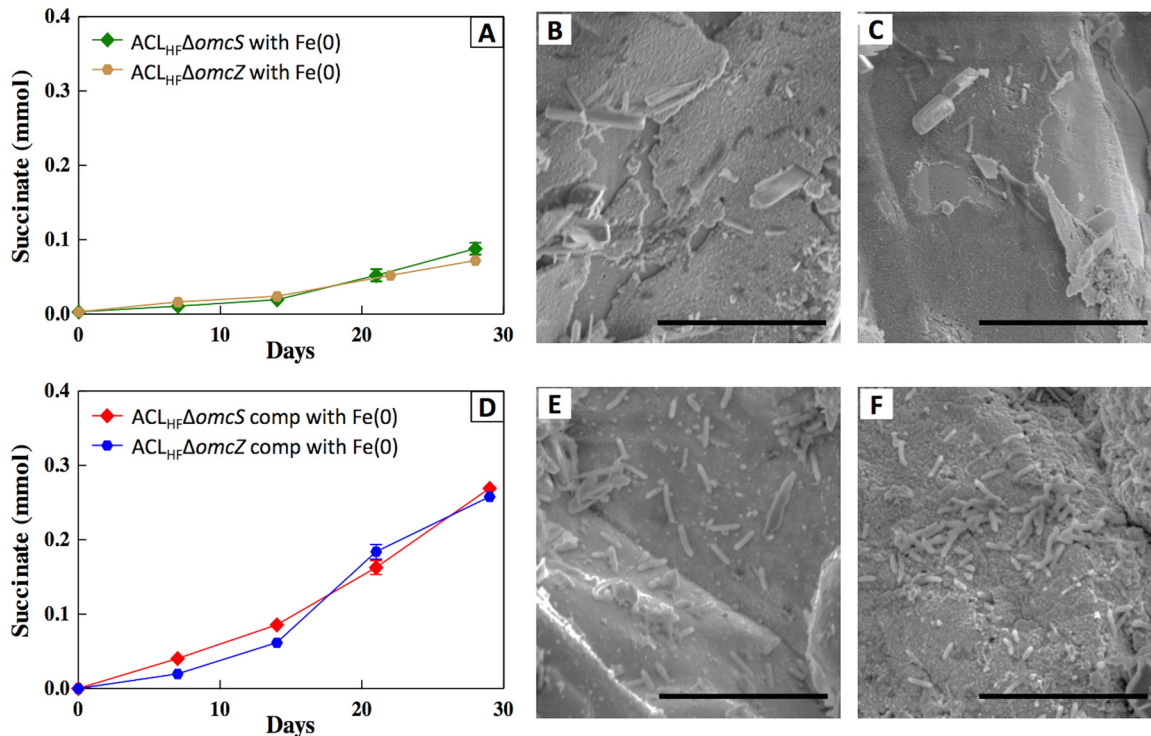


FIG 4 Impact of deleting the gene for the outer surface *c*-type cytochrome OmcS or OmcZ. (A) Succinate production from fumarate reduction in strains with either *omcS* or *omcZ* deleted. (B and C) Scanning electron micrographs demonstrating the lack of cell growth on Fe(0) in the *omcS* (B) or *omcZ* (C) mutants. (D) Succinate production from fumarate reduction when the *omcS* or *omcZ* deletion mutants were complemented *in trans*. (E and F) Growth of cells on Fe(0) when the $\Delta omcS$ mutants (E) or $\Delta omcZ$ mutants (F) were complemented *in trans*. Error bars represent one standard deviation from the mean of the results from triplicate cultures. Scale bar = 10 μ m.

electron transport from Fe(0). Under some conditions, the multiheme *c*-type cytochrome OmcZ, rather than OmcS, is an important electrical contact for electron transfer to electrodes (29, 30). The log₂ RPKM values for *omcZ* transcripts in strain ACL (5.1 ± 0.1) and strain ACL_{HF} (4.5 ± 0.6) were comparable. Considering the much lower median log₂ RPKM value for strain ACL_{HF}, this result indicated a higher relative expression of *omcZ* in strain ACL_{HF}. Deletion of the gene for *omcZ* prevented growth on Fe(0) (Fig. 4). Growth was restored with *in trans* complementation of *omcZ*. These results suggest that OmcZ is also involved in electron transfer from Fe(0) into the cell.

Immunogold labeling has demonstrated that both OmcS (26, 31) and OmcZ (32) can associate with the outer cell surface and thus are properly localized to function as electrical contacts between cells and Fe(0). Under some conditions, OmcS may also extend at distance from the cell, either attached to pili (26, 33, 34) or as filaments composed of OmcS (35). However, filament extensions are unlikely to be important for electron transfer from Fe(0) because the cells are in close contact with the Fe(0) surface.

The inability of strain ACL_{HF} to grow on Fe(0) in the absence of *omcS* or *omcZ* contrasts with the previous finding that the deletion of *omcS* or *omcZ* did not inhibit electron uptake of wild-type *G. sulfurreducens* from negatively poised graphite cathodes (36). However, there are substantial differences between the two studies. In the cathode study, electron uptake, not growth, in a pregrown biofilm was monitored. Different electron transport pathways into the cell may be required to generate sufficient ATP to support growth. The graphite cathodes and Fe(0) have very different surface properties, including a much lower potential for the Fe(0)/Fe(II) redox couple (-470 mV standard hydrogen electrode [SHE] [1]) than the electrode poise (-300 mV SHE [36]). Furthermore, the previous studies on electron uptake from cathodes (36)

were conducted with wild-type *G. sulfurreducens*, which required acetate in the medium as a carbon source. As previously discussed in detail (22), the added acetate may have also served as an electron donor, resulting in different routes for electron flux in the wild-type strain from those found in the autotrophic strain, which was grown in the absence of acetate.

Implications. To our knowledge, strain ACL_{HF} is the first microorganism for which it can reliably be concluded that Fe(0) functions as a direct electron donor. Unlike previous studies, the possibility of H₂ or formate serving as an electron carrier was eliminated, and the need for outer surface redox-active electrical contacts was confirmed. Although it was suggested that a nitrate-reducing strain of *Prolixibacter* that could not grow with H₂ as the sole electron donor might be capable of directly accepting electrons from Fe(0) (37), H₂ was consumed in Fe(0)-amended cultures, indicating that there was a mechanism for H₂ uptake. In contrast, strain ACL_{HF} did not consume the H₂ produced from abiotic Fe(0) oxidation.

Strain ACL_{HF} can serve as a model strain for developing a better understanding of direct electron transfer from Fe(0) and its potential role in corrosion. These studies can build on previous studies regarding extracellular electron exchange in *G. sulfurreducens* (38–41) to develop testable hypotheses regarding the mechanisms for electron transfer from Fe(0). Studies with strain ACL_{HF} also have the advantage that, unlike studies with sulfate reducers, the production of sulfide, which can nonenzymatically promote Fe(0) oxidation (1), is avoided. Growth of strain ACL_{HF} on Fe(0) also does not result in the formation of organic acids, which can promote corrosion (42, 43).

Furthermore, the finding that direct electron transfer from Fe(0) is possible provides further incentive for developing genetic approaches to better evaluate whether the direct electron transfer that has been proposed for other respiratory classes of microorganisms is feasible. Constructing strains that are unable to use H₂ or formate as an electron donor is essential for definitive studies. For example, Fe(0) is rapidly oxidized in the presence of *Desulfovibrio ferrophilus* strain IS5, which is proposed to directly accept electrons from Fe(0) (1). Adapting strategies that are well developed for the genetic manipulation of other *Desulfovibrio* species (44–46), coupled with biochemical localization of putative electrical contacts, would more rigorously test the hypothesis of direct electron transfer from Fe(0) for this microbe, as well as potential alternative strategies, such as a role for flavin shuttles (47). Similar approaches applied to diverse microorganisms are likely to identify unifying mechanisms for direct electron transfer from Fe(0) that will greatly aid in elucidating the capacity for direct metal-microbe electron transfer (DMMET) in the microbial world and may lead to strategies to mitigate corrosion.

MATERIALS AND METHODS

Bacterial strains and growth condition. All strains were routinely maintained under strict anaerobic conditions at 30°C in NB medium (per liter of medium: 10 ml 100X NB salts, 10 ml 100X NB mineral elixir, 15 ml DL vitamins, 0.04 g CaCl₂·2H₂O, 0.1 g MgSO₄·7H₂O, 1.8 g NaHCO₃, 0.5 g Na₂CO₃·H₂O, 1 μM Na₂SeO₄) in which fumarate (40 mM) was supplied as the electron acceptor and acetate (15 mM) was the electron donor (48). For growth on Fe(0), the acetate was omitted, and zero-valent iron granules (1 to 2 mm in diameter; Alfa Aesar, Inc.) were provided as the electron donor in 10 ml of medium dispensed into 28-ml anaerobic pressure tubes under an N₂:CO₂ [80:20 (vol/vol)] atmosphere.

Geobacter sulfurreducens strain ACL, which is capable of autotrophic growth due to the introduction of the citrate lyase genes *aclA* and *aclB* (22), was obtained from our laboratory culture collection. Strain ACL_{HF} was constructed by using the previously described (22) procedure to insert these same citrate lyase genes into the chromosome of a previously described (16) *G. sulfurreducens* strain in which the genes for the uptake hydrogenase (*hybL*) and formate dehydrogenase (*fdnG*) were deleted.

Additional strains were constructed in the strain ACL_{HF} background in which the gene for OmcS or the gene for OmcZ was deleted. The sequences of all primer pairs used for the construction of deletion mutant strains and their complements are listed in Table S2. All mutants were constructed by replacement of the gene of interest with a gentamicin resistance cassette, as previously described (49).

Primer pairs were designed to amplify regions approximately 500 bp upstream and downstream of the target genes and to add AvrII (CCTAGG) restriction sites to the PCR products. These products were ligated into the pCR2.1 TOPO cloning vector, resulting in the formation of PCR2.1up5'+3'dn. Plasmids carrying *omcS* and *omcZ* mutant alleles were linearized by digestion with KpnI (GGTACC) and SacI (GAGCTC), respectively. Linearized plasmids were transformed into ACL_{HF} competent cells by electropo-

ration, as previously described (48), and resulted in the formation of $ACL_{HF} \Delta omcS$ and $ACL_{HF} \Delta omcZ$ strains. Isolated colonies were grown on agar plates supplemented with fumarate (50 mM), acetate (10 mM), and gentamicin (20 μ g/ml).

Deletion mutants made in this study were complemented by transformation with recombinant plasmids carrying a constitutive *lac* promoter (pCM66) and the *omcS* or *omcZ* gene with their native ribosome binding sites (50). Primers (Table S2) were designed to introduce XbaI (TCTAGA) and BamHI (GGATCC) restriction sites into the *omcS* amplicon and EcoRI (GAATTC) and HindIII (AAGCTT) restriction sites into the *omcZ* amplicon. After the complementary plasmids were constructed, they were introduced into $ACL_{HF} \Delta omcS$ and $ACL_{HF} \Delta omcZ$ mutant cells by electroporation, as previously described (48).

Transcriptomics. Cells were harvested from triplicate 50-ml cultures of strain ACL_{HF} and strain ACL grown as described above with Fe(0) as the potential electron donor and fumarate as the acceptor. Cultures were mixed with RNAprotect (Qiagen) in a 1:1 ratio, and iron particles and cells were pelleted by centrifugation at $3,000 \times g$ for 15 min at 4°C, as previously described (22). Pellets were then immediately frozen in liquid nitrogen and stored at -80°C. Total RNA was extracted from all six Fe(0)-containing cell pellets according to the previously described protocol (51) and cleaned with the RNeasy minikit (Qiagen). All RNA samples were then treated with Turbo DNA-free DNase (Ambion, Austin, TX). In order to ensure that samples were not contaminated with genomic DNA, PCR with primers targeting the 16S rRNA gene was done with RNA that had not been reverse transcribed. Further enrichment of mRNA was done with the MICROBExpress kit (Ambion), according to the manufacturer's instructions.

Directional multiplex libraries were prepared with the ScriptSeq v2 RNA sequencing (RNA-seq) library preparation kit (Epicentre), and paired-end sequencing was performed with a HiSeq 2000 platform at the Deep Sequencing Core Facility at the University of Massachusetts Medical School in Worcester, MA.

All raw data generated by Illumina sequencing were quality checked by visualization of base quality scores and nucleotide distributions with FastQC (<http://www.bioinformatics.babraham.ac.uk/projects/fastqc/>). Initial raw nonfiltered forward and reverse sequencing libraries contained an average of $52,114,626 \pm 10,712,759$ reads that were ~100 bp long. Sequences from all of the libraries were trimmed and filtered with Trimmomatic (52) with the sliding window approach set to trim bases with quality scores lower than 3, strings of 3+ Ns, and reads with a mean quality score lower than 20. Bases were also cut from the start and end of reads that fell below a threshold quality of 3, and any reads smaller than 100 bp were eliminated from the library. These parameters yielded an average of $47,913,813 \pm 14,703,090$ trimmed quality reads per RNA-seq library.

All paired-end reads were then merged with FLASH (53), resulting in $19,462,441 \pm 9,267,108$ reads, with an average read length of 147 ± 45 bp. After merging the quality control (QC)-filtered reads, SortMeRNA (54) was used to separate all rRNA reads from nonribosomal reads.

Trimmed and filtered mRNA reads from the triplicate samples for the three different culture conditions were mapped against the *Geobacter sulfurreducens* strain PCA genome (NCBI RefSeq accession no. [NC_002939.5](https://www.ncbi.nlm.nih.gov/ncbi01/NC_002939.5)) downloaded from GenBank at the National Center for Biotechnology Information (NCBI) website (<https://www.ncbi.nlm.nih.gov/>). Mapped reads were normalized with the reads per kilobase per million (RPKM) mapped reads method (55, 56) using the ArrayStar software (DNASTar). Analysis of reads from all three biological replicates for each condition demonstrated that the results were highly reproducible. Therefore, all reported values were obtained after merging and averaging replicates. Expression levels were considered significant only when the \log_2 RPKM value was higher than that of the median \log_2 RPKM.

Analytical methods. The formation of succinate was monitored over time with Shimadzu high-performance liquid chromatography (HPLC) with an Aminex HPX-87H ion exclusion column (300 mm by 7.8 mm) and an eluent of 8.0 mM sulfuric acid, as previously described (57).

Hydrogen was measured from the headspace of cultures at regular intervals using strictly anaerobic sampling techniques. Headspace gas was monitored with a gas chromatograph (GC; Agilent Technologies G1530A, USA) equipped with a Carboxen-1010 Plot column (30 m by 0.53 mm) and a thermal conductivity detector. The oven temperature was 40°C, and the detector temperature was set at 225°C. The carrier gas was N₂.

For protein extraction from Fe(0) particles, 10-ml cultures were centrifuged at $3,000 \times g$ for 20 min at 4°C, and 2 ml of 5% SDS solution was added to the pelleted particles. Pellets were then steam treated for 15 min and centrifuged at $16,100 \times g$ for 20 min at 4°C. Supernatant was collected, and protein concentrations were determined using the Pierce bicinchoninic acid (BCA) protein assay kit (Thermo Fisher Scientific, USA), according to the manufacturer's instructions.

Scanning electron microscopy. Fe(0) particles were collected from cultures when succinate production plateaued and were fixed with 2.5% glutaraldehyde in 0.1 M phosphate buffer for 12 h at 4°C. They were then washed 3 times in 0.1 M phosphate buffer at 4°C for 10 min and then dehydrated in successive ethanol-water mixtures of 35%, 50%, 70%, 80%, 90%, 95%, and 100% for 10 min. The 100% ethanol step was repeated 3 times. Samples were further dehydrated in a 50% hexamethyldisilazane (Sigma-Aldrich, St. Louis, MO, USA) ethanol solution by gentle mixing for 3 min at room temperature, immersed in pure hexamethyldisilazane for 3 min at room temperature, and dried with a stream of high-purity nitrogen for 30 min. Scanning electron microscopy was conducted with an ultrahigh-resolution field emission scanning electron microscope (FEI Magellan 400; Nanolab Technologies, CA, USA).

Data availability. Illumina sequence reads have been submitted to the NCBI database under BioProject number [PRJNA510956](https://www.ncbi.nlm.nih.gov/bioproject/PRJNA510956) and submission number SUB4929798.

SUPPLEMENTAL MATERIAL

Supplemental material for this article may be found at <https://doi.org/10.1128/mBio.00303-19>.

FIG S1, TIF file, 1.5 MB.

TABLE S1, XLSX file, 0.2 MB.

TABLE S2, DOCX file, 0.1 MB.

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REFERENCES

- Enning D, Garrelfs J. 2014. Corrosion of iron by sulfate-reducing bacteria: new views of an old problem. *Appl Environ Microbiol* 80:1226–1236. <https://doi.org/10.1128/AEM.02848-13>.
- Kato S. 2016. Microbial extracellular electron transfer and its relevance to iron corrosion. *Microb Biotechnol* 9:141–148. <https://doi.org/10.1111/1751-7915.12340>.
- Li Y, Xu D, Chen C, Li X, Jia R, Zhang D, Sand W, Wang F, Gu T. 2018. Anaerobic microbiologically influenced corrosion mechanisms interpreted using bioenergetics and bioelectrochemistry: a review. *J Mat Sci Technol* 34:1713–1718. <https://doi.org/10.1016/j.jmst.2018.02.023>.
- Enning D, Venzlaff H, Garrelfs J, Dinh HT, Meyer V, Mayrhofer K, Hassel AW, Stratmann M, Widdel F. 2012. Marine sulfate-reducing bacteria cause serious corrosion of iron under electroconductive biogenic mineral crust. *Environ Microbiol* 14:1772–1787. <https://doi.org/10.1111/j.1462-2920.2012.02778.x>.
- Uchiyama T, Ito K, Mori K, Tsurumaru H, Harayama S. 2010. Iron-corroding methanogen isolated from a crude-oil storage tank. *Appl Environ Microbiol* 76:1783–1788. <https://doi.org/10.1128/AEM.00668-09>.
- Kato S, Yumoto I, Kamagata Y. 2015. Isolation of acetogenic bacteria that induce biocorrosion by utilizing metallic iron as the sole electron donor. *Appl Environ Microbiol* 81:67–73. <https://doi.org/10.1128/AEM.02767-14>.
- Dinh HT, Kuever J, Mussmann M, Hassel AW, Stratmann M, Widdel F. 2004. Iron corrosion by novel anaerobic microorganisms. *Nature* 427:829–832. <https://doi.org/10.1038/nature02321>.
- Venzlaff H, Enning D, Srinivasan J, Garrelfs J, Stratmann M, Widdel F, Stratmann M. 2013. Accelerated cathodic reaction in microbial corrosion of iron due to direct electron uptake by sulfate-reducing bacteria. *Corros Sci* 66:88–96. <https://doi.org/10.1016/j.corsci.2012.09.006>.
- Beese P, Venzlaff H, Srinivasan J, Garrelfs J, Stratmann M, Mayrhofer K. 2013. Monitoring of anaerobic microbially influenced corrosion via electrochemical frequency modulation. *Electrochim Acta* 105:239–247. <https://doi.org/10.1016/j.electacta.2013.04.144>.
- Phillips J, Van den Driessche N, De Paepe K, PrévotEAU A, Gralnick JA, Arends JBA, Rabaey K. 2018. A novel *Shewanella* isolate enhances corrosion by using metallic iron as the electron donor with fumarate as the electron acceptor. *Appl Environ Microbiol* 84:e01154-18. <https://doi.org/10.1128/AEM.01154-18>.
- Lovley DR, Phillips EJ. 1988. Novel mode of microbial energy metabolism: organic carbon oxidation coupled to dissimilatory reduction of iron or manganese. *Appl Environ Microbiol* 54:1472–1480.
- Lovley DR, Giovannoni SJ, White DC, Champagne JE, Phillips EJ, Gorby YA, Goodwin S. 1993. *Geobacter metallireducens* gen. nov. sp. nov., a microorganism capable of coupling the complete oxidation of organic compounds to the reduction of iron and other metals. *Arch Microbiol* 159:336–344. <https://doi.org/10.1007/BF00290916>.
- Coppi MV, O'Neil RA, Lovley DR. 2004. Identification of an uptake hydrogenase required for hydrogen-dependent reduction of Fe(III) and other electron acceptors by *Geobacter sulfurreducens*. *J Bacteriol* 186:3022–3028. <https://doi.org/10.1128/JB.186.10.3022-3028.2004>.
- Gregory KB, Bond DR, Lovley DR. 2004. Graphite electrodes as electron donors for anaerobic respiration. *Environ Microbiol* 6:596–604. <https://doi.org/10.1111/j.1462-2920.2004.00593.x>.
- Summers ZM, Fogarty HE, Leang C, Franks AE, Malvankar NS, Lovley DR. 2010. Direct exchange of electrons within aggregates of an evolved syntrophic coculture of anaerobic bacteria. *Science* 330:1413–1415. <https://doi.org/10.1126/science.1196526>.
- Rotaru AE, Shrestha PM, Liu FH, Ueki T, Nevin K, Summers ZM, Lovley DR. 2012. Interspecies electron transfer via hydrogen and formate rather than direct electrical connections in cocultures of *Pelobacter carbinolicus* and *Geobacter sulfurreducens*. *Appl Environ Microbiol* 78:7645–7651. <https://doi.org/10.1128/AEM.01946-12>.
- Lovley DR. 2011. Live wires: direct extracellular electron exchange for bioenergy and the bioremediation of energy-related contamination. *Energy Environ Sci* 4:4896–4906. <https://doi.org/10.1039/c1ee02229f>.
- Lovley DR. 2012. Electromicrobiology. *Annu Rev Microbiol* 66:391–409. <https://doi.org/10.1146/annurev-micro-092611-150104>.
- Lovley DR. 2017. Happy together: microbial communities that hook up to swap electrons. *ISME J* 11:327–336. <https://doi.org/10.1038/ismej.2016.136>.
- Beese-Vasbender PF, Nayak S, Erbe A, Stratmann M, Mayrhofer K. 2015. Electrochemical characterization of direct electron uptake in electrical microbially influenced corrosion of iron by the lithoautotrophic SRB *Desulfopila corrodens* strain IS4. *Electrochim Acta* 167:321–329. <https://doi.org/10.1016/j.electacta.2015.03.184>.
- Deng X, Dohmae N, Nealson KH, Hashimoto K, Okamoto A. 2018. Multi-heme cytochromes provide a pathway for survival in energy-limited environments. *Sci Adv* 4:eaa05682. <https://doi.org/10.1126/sciadv.aao5682>.
- Ueki T, Nevin KP, Woodard TL, Aklujkar MA, Holmes DE, Lovley DR. 2018. Construction of a *Geobacter* strain with exceptional growth on cathodes. *Front Microbiol* 9:1512. <https://doi.org/10.3389/fmicb.2018.01512>.
- Reardon EJ. 1995. Anaerobic corrosion of granular iron: measurement and interpretation of hydrogen evolution rates. *Environ Sci Technol* 29:2936–2945. <https://doi.org/10.1021/es00012a008>.
- Shrestha PM, Rotaru AE, Aklujkar M, Liu F, Shrestha M, Summers ZM, Malvankar N, Flores DC, Lovley DR. 2013. Syntrophic growth with direct interspecies electron transfer as the primary mechanism for energy exchange. *Environ Microbiol Rep* 5:904–910. <https://doi.org/10.1111/1758-2229.12093>.
- Mehta T, Coppi MV, Childers SE, Lovley DR. 2005. Outer membrane c-type cytochromes required for Fe(III) and Mn(IV) oxide reduction in *Geobacter sulfurreducens*. *Appl Environ Microbiol* 71:8634–8641. <https://doi.org/10.1128/AEM.71.12.8634-8641.2005>.
- Leang C, Qian X, Mester T, Lovley DR. 2010. Alignment of the c-type cytochrome OmcS along pili of *Geobacter sulfurreducens*. *Appl Environ Microbiol* 76:4080–4084. <https://doi.org/10.1128/AEM.00023-10>.
- Ding YH, Hixson KK, Aklujkar MA, Lipton MS, Smith RD, Lovley DR, Mester T. 2008. Proteome of *Geobacter sulfurreducens* grown with Fe(III) oxide or Fe(III) citrate as the electron acceptor. *Biochim Biophys Acta* 1784:1935–1941. <https://doi.org/10.1016/j.bbapap.2008.06.011>.
- Holmes DE, Chaudhuri SK, Nevin KP, Mehta T, Methe BA, Liu A, Ward JE, Woodard TL, Webster J, Lovley DR. 2006. Microarray and genetic analysis of electron transfer to electrodes in *Geobacter sulfurreducens*. *Environ Microbiol* 8:1805–1815. <https://doi.org/10.1111/j.1462-2920.2006.01065.x>.
- Nevin KP, Kim B-C, Glaven RH, Johnson JP, Woodard TL, Methé BA, DiDonato RJ, Jr, Covalla SF, Franks AE, Liu A, Lovley DR. 2009. Anode biofilm transcriptomics reveals outer surface components essential for

- high current power production in *Geobacter sulfurreducens* fuel cells. *PLoS One* 4:e5628. <https://doi.org/10.1371/journal.pone.0005628>.
30. Richter H, Nevin KP, Jia H, Lowy DA, Lovley DR, Tender LM. 2009. Cyclic voltammetry of biofilms of wild type and mutant *Geobacter sulfurreducens* on fuel cell anodes indicates possible roles of OmcB, OmcZ, type IV pili, and protons in extracellular electron transfer. *Energy Environ Sci* 2:506–516. <https://doi.org/10.1039/b816647a>.
 31. Flanagan KA, Leang C, Ward J, Lovley D. 2017. Improper localization of the OmcS cytochrome may explain the inability of the *xapD*-deficient mutant of *Geobacter sulfurreducens* to reduce Fe(III) oxide. *bioRxiv* <https://doi.org/10.1101/114900>.
 32. Inoue K, Leang C, Franks AE, Woodard TL, Nevin KP, Lovley DR. 2011. Specific localization of the *c*-type cytochrome OmcZ at the anode surface in current-producing biofilms of *Geobacter sulfurreducens*. *Environ Microbiol Rep* 3:211–217. <https://doi.org/10.1111/j.1758-2229.2010.00210.x>.
 33. Malvankar NS, Tuominen MT, Lovley DR. 2012. Comment on “On electrical conductivity of microbial nanowires and biofilms” by S. M. Strycharz-Glaven, R. M. Snider, A. Guiseppi-Elie and L. M. Tender, *Energy Environ. Sci.*, 2011, 4, 4366. *Energy Environ Sci* 5:6247. <https://doi.org/10.1039/c2ee02613a>.
 34. Yun J, Malvankar NS, Ueki T, Lovley DR. 2016. Functional environmental proteomics: elucidating the role of a *c*-type cytochrome abundant during uranium bioremediation. *ISME J* 10:310–320. <https://doi.org/10.1038/ismej.2015.113>.
 35. Filman D, Marino S, Ward J, Yang L, Mester Z, Bullitt E, Lovley D, Strauss M. 2018. Structure of a cytochrome-based bacterial nanowire. *bioRxiv* <https://doi.org/10.1101/492645>.
 36. Strycharz SM, Glaven RH, Coppi MV, Gannon SM, Perpetua LA, Liu A, Nevin KP, Lovley DR. 2011. Gene expression and deletion analysis of mechanisms for electron transfer from electrodes to *Geobacter sulfurreducens*. *Bioelectrochemistry* 80:142–150. <https://doi.org/10.1016/j.bioelechem.2010.07.005>.
 37. Iino T, Ito K, Wakai S, Tsurumaru H, Ohkuma M, Harayama S. 2015. Iron corrosion induced by nonhydrogenotrophic nitrate-reducing *Prolixibacter* sp. strain MIC1-1. *Appl Environ Microbiol* 81:1830–1837. <https://doi.org/10.1128/AEM.03741-14>.
 38. Smith JA, Nevin KP, Lovley DR. 2015. Syntrophic growth via quinone-mediated interspecies electron transfer. *Front Microbiol* 6:121. <https://doi.org/10.3389/fmicb.2015.00121>.
 39. Voordeckers JW, Kim BC, Izallalen M, Lovley DR. 2010. Role of *Geobacter sulfurreducens* outer surface *c*-type cytochromes in reduction of soil humic acid and anthraquinone-2,6-disulfonate. *Appl Environ Microbiol* 76:2371–2375. <https://doi.org/10.1128/AEM.02250-09>.
 40. Nevin KP, Lovley DR. 2000. Lack of production of electron-shuttling compounds or solubilization of Fe(III) during reduction of insoluble Fe(III) oxide by *Geobacter metallireducens*. *Appl Environ Microbiol* 66:2248–2251. <https://doi.org/10.1128/AEM.66.5.2248-2251.2000>.
 41. Aklujkar M, Coppi MV, Leang C, Kim BC, Chavan MA, Perpetua LA, Giloteaux L, Liu A, Holmes DE. 2013. Proteins involved in electron transfer to Fe(III) and Mn(IV) oxides by *Geobacter sulfurreducens* and *Geobacter uraniireducens*. *Microbiology* 159:515–535. <https://doi.org/10.1099/mic.0.064089-0>.
 42. Pak KR, Lee HJ, Lee HK, Kim YK, Oh YS, Choi SC. 2003. Involvement of organic acid during corrosion of iron coupon *Desulfovibrio desulfuricans*. *J Microbiol Biotech* 13:937–941.
 43. Iverson WP. 1987. Microbial corrosion of metals. *Adv Appl Microbiol* 32:1–36. [https://doi.org/10.1016/S0065-2164\(08\)70077-7](https://doi.org/10.1016/S0065-2164(08)70077-7).
 44. Keller KL, Bender KS, Wall JD. 2009. Development of a markerless genetic exchange system for *Desulfovibrio vulgaris* Hildenborough and its use in generating a strain with increased transformation efficiency. *Appl Environ Microbiol* 75:7682–7691. <https://doi.org/10.1128/AEM.01839-09>.
 45. Keller KL, Wall JD. 2011. Genetics and molecular biology of the electron flow for sulfate respiration in *Desulfovibrio*. *Front Microbiol* 2:135. <https://doi.org/10.3389/fmicb.2011.00135>.
 46. Wall JD, Hemme CL, Rapp-Giles B, Ringbauer JA, Casalot L, Glibin T. 2003. Genes and genetic manipulations of *Desulfovibrio*, p 85–98. *In* Ljungdahl LG, Adams MW, Barton LL, Ferry JG, Johnson MK (ed), *Biochemistry and physiology of anaerobic bacteria*. Springer, New York, NY.
 47. Zhang PY, Xu DK, Li YC, Yang K, Gu TY. 2015. Electron mediators accelerate the microbiologically influenced corrosion of 304 stainless steel by the *Desulfovibrio vulgaris* biofilm. *Bioelectrochemistry* 101:14–21. <https://doi.org/10.1016/j.bioelechem.2014.06.010>.
 48. Coppi MV, Leang C, Sandler SJ, Lovley DR. 2001. Development of a genetic system for *Geobacter sulfurreducens*. *Appl Environ Microbiol* 67:3180–3187. <https://doi.org/10.1128/AEM.67.7.3180-3187.2001>.
 49. Aklujkar M, Lovley DR. 2010. Interference with histidyl-tRNA synthetase by a CRISPR spacer sequence as a factor in the evolution of *Pelobacter carbinolicus*. *BMC Evol Biol* 10:230. <https://doi.org/10.1186/1471-2148-10-230>.
 50. Marx CJ, Lidstrom ME. 2001. Development of improved versatile broad-host-range vectors for use in methylotrophs and other Gram-negative bacteria. *Microbiol* 147:2065–2075. <https://doi.org/10.1099/00221287-147-8-2065>.
 51. Holmes DE, Risso C, Smith JA, Lovley DR. 2012. Genome-scale analysis of anaerobic benzoate and phenol metabolism in the hyperthermophilic archaeon *Ferroglobus placidus*. *ISME J* 6:146–157. <https://doi.org/10.1038/ismej.2011.88>.
 52. Bolger AM, Lohse M, Usadel B. 2014. Trimmomatic: a flexible trimmer for Illumina sequence data. *Bioinformatics* 30:2114–2120. <https://doi.org/10.1093/bioinformatics/btu170>.
 53. Magoč T, Salzberg SL. 2011. FLASH: fast length adjustment of short reads to improve genome assemblies. *Bioinformatics* 27:2957–2963. <https://doi.org/10.1093/bioinformatics/btr507>.
 54. Kopylova E, Noe L, Touzet H. 2012. SortMeRNA: fast and accurate filtering of ribosomal RNAs in metatranscriptomic data. *Bioinformatics* 28:3211–3217. <https://doi.org/10.1093/bioinformatics/bts611>.
 55. Klevebring D, Bjursell M, Emanuelsson O, Lundberg J. 2010. In-depth transcriptome analysis reveals novel TARs and prevalent antisense transcription in human cell lines. *PLoS One* 5:e9762. <https://doi.org/10.1371/journal.pone.0009762>.
 56. Mortazavi A, Williams BA, McCue K, Schaeffer L, Wold B. 2008. Mapping and quantifying mammalian transcriptomes by RNA-seq. *Nat Methods* 5:621. <https://doi.org/10.1038/nmeth.1226>.
 57. Nevin KP, Richter H, Covalla SF, Johnson JP, Woodard TL, Orloff AL, Jia H, Zhang M, Lovley DR. 2008. Power output and coulombic efficiencies from biofilms of *Geobacter sulfurreducens* comparable to mixed community microbial fuel cells. *Environ Microbiol* 10:2505–2514. <https://doi.org/10.1111/j.1462-2920.2008.01675.x>.