

Preventing the return of smallpox: molecular modeling studies on thymidylate kinase from *Variola virus*

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Smallpox was one of the most devastating diseases in the human history and still represents a serious menace today due to its potential use by bioterrorists. Considering this threat and the non-existence of effective chemotherapy, we propose the enzyme thymidylate kinase from *Variola virus (Var*TMPK) as a potential target to the drug design against smallpox. We first built a homology model for *Var*TMPK and performed molecular docking studies on it in order to investigate the interactions with inhibitors of *Vaccinia virus* TMPK (*Vac*TMPK). Subsequently, molecular dynamics (MD) simulations of these compounds inside *Var*TMPK and human TMPK (*Hss*TMPK) were carried out in order to select the most promising and selective compounds as leads for the design of potential *Var*TMPK inhibitors. Results of the docking and MD simulations corroborated to each other, suggesting selectivity towards *Var*TMPK and, also, a good correlation with the experimental data.

Keywords: smallpox; Variola virus; thymidylate kinase; docking; molecular dynamics

1. Introduction

Smallpox is an acute disease, highly contagious, and exclusive of human beings. It is caused by two variants of viruses from the genera Orthopoxvirus known as Variola major and Variola minor and presents lethality around 30% (Behbehani, 1983; Gileva et al., 2006; Hendrickson, Wang, Hatcher, & Lefkowitz, 2010; Liszewski et al., 2006; Ryan & Ray, 2004; Sakhatskyy et al., 2008). As one of the most devastating diseases affecting human kind, smallpox had been present in all continents for centuries, decimating populations in several pandemics, and changing the course of history (Chapman, Nichols, Martinez, & Raymond, 2010; Hammarlund et al., 2010; Kennedy, Ovsyannikova, Jacobson, & Poland, 2009; Kennedy, Ovsyannikova, & Poland, 2009). Despite declared eradicated by the World Health Organization (WHO) in 1980, there are no guaranties that this disease would not come back in future because orthopoxvirus are capable of surviving for decades crystallized in nature at low temperature and humidity (Harper, 1961; Hug, 1976). Besides, due to the interruption of the vaccination programs, the world population is now more susceptible to infections with *orthopoxvirus*. This is a serious menace to the public health either by the risk of natural outbreaks, like the *monkeypox* and *cowpox* episodes occurred in Germany and France in 2002 and 2009 respectively, or by the use of smallpox as a biological weapon by the terrorist groups (Alzhanova & Früh, 2010; Campe et al., 2009; Handley, Buller, Frey, Bellone, & Parker, 2009; Johnston et al., 2012; Kennedy, Ovsyannikova, Jacobson, et al., 2009; Mätz-Rensing et al., 2006; Ninove et al., 2009; Schmiedeknecht et al., 2010).

Soon after the eradication of smallpox, all countries members of WHO agreed in destroying eventual strains or sending it to the two unique laboratories in the world accredited to keep strains of these viruses: the Center of Disease Control in Atlanta, USA, and the Institute of Viruses Preparations in Moscow, Russia. The remaining strains in these labs should have been destroyed by January of 2002 but the fear of smallpox being used as a biological agent has postponed this decision (Baker, Bray, & Huggins, 2003; Mätz-Rensing et al., 2010). Besides, WHO also

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admits the existence of these viruses in clandestine labs around the world that could be accidentally or intentionally spread out (Hammarlund et al., 2003; Lindler, Lebeda, & Korch, 2005).

Considering the high infectiveness and lethality of smallpox, allied to the potential risk represented by its eventual use by bioterrorists, it is imperative to search for new drugs to combat this disease. This situation is aggravated by the fact that there is no chemotherapy against smallpox approved by the Food and Drugs Administration (FDA). The proposition of molecular targets for the drug design against smallpox is, therefore, a worldwide emergency. Thus, we proposed in the present work the enzyme thymidylate kinase from Variola virus (VarTMPK) as a potential target. A homology model was built and further submitted to docking and molecular dynamics (MD) studies in order to investigate the interactions of known inhibitors of Vaccinia virus TMPK (VacTMPK) inside VarTMPK active site. Also, the selectivity of each compound was investigated by performing the same studies inside human TMPK (HssTMPK). These studies led to the selection of the most promising and selective compounds as lead for the design of potential VarTMPK inhibitors. Results of the docking and MD studies corroborated to each other and, also, suggested a good correlation with the experimental data. We also observed the main interactions responsible for molecular recognition and suggested insights to the design of efficient and selective VarTMPK inhibitors.

2. Materials and methods

2.1. Homology modeling

The primary sequence of VarTMPK was downloaded from Expasy (http://au.expasy.org/) using the UniProtKB database (http://www.uniprot.org/). The search for templates with 3D structures available in the Protein Data Bank (PDB) (Berman et al., 2000) was performed with the basic local alignment search tool program (BLAST) (Altschul, Gish, Miller, Myers, & Lipman, 1990; Altschul et al., 1997). The dimeric crystallographic structure of VacTMPK at 2.40 Å resolution with factor R = .216 (PDB code: 2V54) (Caillat et al., 2008) was the one presenting the biggest sequential identity with VarTMPK being, for this reason, chosen as template to build the 3D model for VarTMPK. The alignment of the monomeric sequences was done with the software Swiss-Pdb Viewer (Guex & Peitsch, 1997) and, after some manual adjustments, submitted to the SWISS-MODEL server (Guex & Peitsch, 1997) in order to generate the final monomer model.

In order to check for the quality of our model, it was submitted to the validation server of the PDB (Berman et al., 2000) and to the Verify3D structure evaluation server (Bowie, Luthy, & Eisenberg, 1991; Luthy, Bowie, & Eisenberg, 1992), to check for the chemical environment of each residue.

After validation, the model was superposed to the 3D structure of *Vac*TMPK with the program Swiss-Pdb Viewer (Guex & Peitsch, 1997) in order to copy the coordinates of thymidine-5'-diphosphate (TDP), the cofactor Mg^{2+} , and the co-crystallized water molecules from the *Vac*TMPK active site to build the complex used in our docking studies.

To verify the similarity between the residues of the active sites of *Var*TMPK and *Hss*TMPK and to determine the degree of identity between them, we aligned the *Var*TMPK and *Hss*TMPK (PDB code: 1E2G) structures using Swiss-Pdb Viewer (Guex & Peitsch, 1997).

2.2. Docking energy calculations

The 3D structure of HssTMPK with resolution of 1.7 Å, complexed with TDP and Mg^{2+} (Ostermann et al., 2000) was obtained from the PDB (Berman et al., 2000) under the code 1E2G, and the 3D structures of the compounds used in this study (Figure 1) were built using the program PC Spartan® (Hehre, Deppmeier, & Klunzinger, 1999) and had their partial atomic charges calculated by the RM1 semiempirical molecular orbital method (Rocha, Freire, & Simas, 2006). These compounds are potential inhibitors of VacTMPK and had its activities investigated by Caillat et al. (2008). Calculations of their docking energies inside the actives sites of VarTMPK and HssTMPK were performed using the software Molegro Virtual Docker® (Thomsen & Christensen, 2006) (MVD). The binding sites of VarTMPK and HssTMPK were restricted into spheres with radius of 6Å around TDP and all the residues inside these spheres were set to be flexible. The influence of the water molecules was also considered during the docking studies because the solvent has an important role for protein-ligand interactions as it can modify the characteristics of the active site, mediate the formation of H-bonds, and also increase the entropy of the system due to the hydrophobic effect. To validate the docking protocol used, we first performed the re-docking of TDP over the active sites of VarTMPK and HssTMPK. Due to the stochastic nature of the docking algorithm, about 20 runs were performed for each compound with 30 poses (conformation and orientation of the ligand) returned to the analysis of the overlap with TDP inside VarTMPK and HssTMPK and, also, the ligand-protein interactions. The best pose of each compound was selected for the further steps of MD simulations.

2.3. MD simulations

In order to check the docking results, we chose the best conformations obtained from the docking studies to run further steps of MD simulations. Before performing the simulations it was necessary to parameterize the ligands,



Figure 1. Structures of the compounds studied.

so they could be recognized by the forcefield GROMOS 96 53a6 (Oostenbrink, Villa, Mark, & Van Gunsteren, 2004) from the program GROMACS 4.5.4 (Hess, Kutzner, Van Der Spoel, & Lindahl, 2008). The parameterization was carried out in the Dundee PRODRG Server (Van Aalten et al., 1996) and the charges distributions were calculated by the CHELPG (Brenaman & Wiberg, 1990) method using the Gaussian® (Frisch et al., 2001) program at DFT method (Hohenberg & Kohn, 1964) with functional B3LYP and basis function 6-31G (d,p). The dynamic behaviors of the complexes VarTMPK-ligand and HssTMPK-ligand were simulated using the GRO-MACS 4.5.4 package (Hess et al., 2008), in cubic boxes of approximately 640 and 940 nm³, containing around 7800 and 8000 water molecules of the SPC type, respectively, using periodic boundary conditions (Berendsen, Postma, Van Gunsteren, & Hermans, 1981). In this study, for both pressure and temperature, we used the Berendsen's weakcoupling algorithm scheme which maintained temperature and pressure at 300 K and 1 bar. These systems were submitted to four steps of energy minimization as follows: steepest descent with position restrained of the ligands and convergence criterion of $50.00 \text{ kcal mol}^{-1} \text{ nm}^{-1}$, followed by steepest descent without restriction, conjugate gradients and, finally, quasi-Newton-Raphson until an energy of $10.00 \,\text{kcal mol}^{-1} \,\text{nm}^{-1}$.

The minimized complexes were then submitted to MD simulations in two steps. Initially, we performed 500 ps of MD at 300 K with PR for the entire system, except the water molecules, in order to ensure a balance of the solvent molecules around protein and ligand. Subsequently, there were performed 20 ns of MD without any restriction, using 2 fs of integration time and a cutoff of 10 Å for long-distance interactions. The electrostatic interactions were calculated using the Particle Mesh Ewald method and a total of 1000 conformations were obtained during each simulation. In this step, the lists of pairs (pairlists) were updated every 500 steps, all Arg and Lys residues were assigned with positive charges and the residues Glu and Asp were assigned with negative charges.

To analyze the structures generated after the optimization and MD steps, we used the visual molecular dynamics (VMD) (Humphrey, Dalke, & Schulten, 1996) and Swiss-Pdb Viewer (Guex & Peitsch, 1997) programs. Plots of variation of total energy, distance, random mean-square deviation (RMSD), and H-bonds formed during the MD simulation were generated with the Origin® program (Edwards, 2002). Qualitative spatial RMSD pictures were generated in MolMol program (Koradi, Billeter, & Wüthrich, 1996) and the figures of the frames of MD simulations were generated in the PyMOL program (DeLano & Bromberg, 2004).

3. Results and discussion

3.1. Homology modeling

The alignment between the sequences of *Vac*TMPK and *Var*TMPK (Figure 2) shows that these enzymes share 98.5% of sequential identity with each other as reported before (Caillat et al., 2008). This is an excellent result considering literature reports that high quality models can be achieved when the similarity to the template is superior to 50%. Such models are comparable to structures determined by NMR and X-ray techniques and, therefore, suitable to the drug design. It was also observed that only four residues (Thr30, Thr103, Ala148, and Glu165) are different and none of them belong to the active site. This result corroborates the premise that potential inhibitors of *Vac*TMPK could, eventually, become inhibitors of *Var*TMPK.

After building the 3D structure of *Var*TMPK, we superimposed model and template using Swiss-Pdb Viewer (Guex & Peitsch, 1997) and copied the structures of TDP, Mg^{2+,} and the water molecules from the template to the active site of our model in order to build the system for the docking and MD studies. The RMSD of .18 Å obtained on the superposition evidences the high sequential identity between model and template.

The Ramachandram plot (Ramachandran & Sasisekharan, 1968) of our model (data not shown) presented 99.5% of the residues in the most favored regions; a result considered very good for a homology model (Laskowski, Macarthur, Moss, & Thornton, 1993). Regarding the main chain properties, there were not found any bad contacts neither distortions of C_{α} nor energy problems with H-bonds. Also, there were not

| VarTMPK | 1 | MSRGALIVFE | GLDKSGKTTQ | CMNIMESIPT | NTIKYLNFPQ | RSTVTGKMID |
|---------|-----|------------------------|------------|------------|------------|------------|
| VacTMPK | 1 | MSRGALIVFE | GLDKSGKTTQ | CMNIMESIPA | NTIKYLNFPQ | RSTVTGKMID |
| | | ****** | ****** | ********** | ****** | ****** |
| | | | | | | |
| VarTMPK | 51 | DYLTRKKTYN | DHIVNLLFCA | NRWEFASFIQ | EQLEQGITLI | VDRYAFSGVA |
| VacTMPK | 51 | DYLTRKKTYN | DHIVNLLFCA | NRWEFASFIQ | EQLEQGITLI | VDRYAFSGVA |
| | | ******** | ******* | ******* | ******* | ******** |
| | | | | | | |
| VarTMPK | 101 | YATAKGASMT | LSKSYESGLP | KPDLVIFLES | GSKEINRNVG | EEIYEDVAFQ |
| VacTMPK | 101 | YAAAKGASMT | LSKSYESGLP | KPDLVIFLES | GSKEINRNVG | EEIYEDVTFQ |
| | | ** [*] ****** | **** | **** | **** | ****** |
| | | | | | | |
| VarTMPK | 151 | QKVLQEYKKM | IEEGEDIHWQ | IISSEFEEDV | KKELIKNIVI | EAIHTVTGPV |
| VacTMPK | 151 | QKVLQEYKKM | IEEG-DIHWQ | IISSEFEEDV | KKELIKNIVI | EAIHTVTGPV |
| | | ****** | **** | **** | **** | **** |
| | | | | | | |
| VarTMPK | 201 | GQLWM | | | | |
| VacTMPK | 200 | GQLWM | | | | |
| | | **** | | | | |

Figure 2. Alignment of VarTMPK and VacTMPK sequences. Non-matching residues are shown in red.



Figure 3. Re-docking of TDP in the active sites of *Var*TMPK (right) and *Hss*TMPK (left), TDP from crystal is colored by element, while the TDP conformation obtained by docking is colored in yellow.



Figure 4. Left: Interactions of BrdU with aminoacids and water molecules inside VarTMPK. Right: superposition of the best conformation of BrdU (in CPK) with TDP (in yellow) after docking.

found distortions on the torsion angles of the side chains. The 3D–1D averaged score for each residue was calculated by the Verify3D program (Bowie et al., 1991) and presented a variation between .03 and .58. The Z-scores calculated using PROVE (Pontius, Richelle, & Wodak, 1996) showed that our model is comparable to a typical native structure. Besides, G factor (one measure of the normality degree of the protein properties) obtained of -.24 was below the maximum value allowed for homology models (Laskowski et al., 1993).

The 3D structure of *Hss*TMPK was superimposed to our model, using SPDBViewer, in order to verify the degree of similarity among the active site residues and the possibility of designing selective *Var*TMPK inhibitors. The superposition presented a RMSD of 1.19 Å and the sequential identity observed between these two enzymes

Table 1. Docking results for the compounds docked in VarTMPK and their K_i values, obtained by Caillat et al. (2008), with VacTMPK.

| Compound | Residue | Dist. (Å) | Energy (kcal mol^{-1}) | Intermolecular energy (kcal mol^{-1}) | H-bond energy (kcal mol^{-1}) | ${}^{a}K_{i}$ |
|----------|---------|-----------|---------------------------|--|----------------------------------|---------------|
| BrdU | Asp13 | 3.15 | -2.25 | -127.16 | -10.49 | 7 μΜ |
| | Arg41 | 3.10 | -1.71 | | | |
| | Arg72 | 2.87 | -2.50 | | | |
| | | 3.09 | -2.09 | | | |
| | Tyr144 | 3.21 | -1.94 | | | |
| | H_2O | 3.09 | -2.50 | | | |
| | | 3.26 | -1.70 | | | |
| CldU | Asp13 | 3.04 | -2.50 | -129.63 | -8.11 | 15 µM |
| | Arg41 | 2.74 | -1.55 | | | |
| | Arg72 | 2.60 | -2.34 | | | |
| | | 2.95 | -1.69 | | | |
| | Gly98 | 3.35 | 02 | | | |
| IdU | Asp13 | 3.16 | -2.18 | -125.97 | -10.43 | 23 µM |
| | Arg41 | 3.10 | -1.75 | | | |
| | Arg72 | 2.79 | -2.50 | | | |
| | | 2.99 | -2.05 | | | |
| | Tyr144 | 3.21 | -1.94 | | | |
| | H_2O | 3.10 | -2.50 | | | |
| | | 3.26 | -1.68 | | | |
| dT | Asp13 | 3.32 | -1.41 | -123.49 | -9.18 | 25 μΜ |
| | Arg41 | 3.10 | -1.59 | | | |
| | Arg72 | 2.82 | -1.97 | | | |
| | | 2.63 | -2.44 | | | |
| | Tyr101 | 3.47 | 62 | | | |
| | Tyr144 | 3.37 | -1.14 | | | |
| | H_2O | 3.07 | -2.50 | | | |
| | | 3.39 | -1.04 | | | |
| | | 3.19 | -2.04 | | | |
| BvdU | Arg72 | 2.60 | -1.99 | -119.06 | -4.11 | .5 mM |
| | | 2.60 | -2.01 | | | |
| | Tyr94 | 3.57 | 11 | | | |
| | H_2O | 2.61 | -2.50 | | | |
| | | 3.10 | -2.50 | | | |
| FdU | Asp13 | 3.11 | -2.45 | -116.32 | -9.49 | .7 mM |
| | Arg41 | 2.74 | -1.57 | | | |
| | Arg72 | 3.10 | -2.09 | | | |
| | | 2.94 | -2.47 | | | |
| | Tyr101 | 3.42 | 90 | | | |
| | H_2O | 3.54 | 31 | | | |
| dU | Asp13 | 3.10 | -2.49 | -105.17 | -11.20 | 1.0 mM |
| | Arg41 | 3.10 | -1.15 | | | |
| | Arg72 | 3.09 | -2.32 | | | |
| | | 3.00 | -2.50 | | | |
| | Tyr101 | 3.45 | 77 | | | |
| | Tyr144 | 3.20 | -1.98 | | | |
| | H_2O | 3.31 | -1.43 | | | |
| | | 3.41 | 95 | | | |
| | | 2.86 | -2.50 | | | |

^aExperimental data.²⁷

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| Table 2 | Docking results for | the compounds d | locked in HeeTMPK | and their K. values | obtained by Caillat et al. (2008). | |
|----------|---------------------|-----------------|------------------------|-----------------------|------------------------------------|--|
| 14010 2. | Docking results for | the compounds d | JUCKEU III 1155 I WILK | and then K_i values | obtained by Camat et al. (2008). | |

| Compound | Residue | Dist. (Å) | Energy (kcal mol^{-1}) | Intermolecular energy (kcal mol^{-1}) | H-bond energy (kcal mol^{-1}) | ${}^{a}K_{i}$ |
|----------|------------------|--------------|---------------------------------|--|--|---------------|
| BrdU | Asp15 | 3.09 | -2.50 | -101.04 | -7.66 | .2 mM |
| | _ | 2.73 | -2.50 | | | |
| | Arg76 | 3.27 | 99 | | | |
| | $\Lambda ra07$ | 2.48 | -1.50 | | | |
| | H ₂ O | 3.42 | 17 - 90 | | | |
| | 1120 | 2.96 | -2.50 | | | |
| | | 2.80 | -2.50 | | | |
| | | 2.61 | -2.50 | | | |
| | | 2.77 | -2.50 | | | |
| 177 | 4 17 | 3.10 | -2.50 | 101.52 | (00 | 25. 14 |
| dT | Asp15 | 3.29 | -1.56 | -101.53 | -6.00 | .25 mM |
| | Arg/6 | 3.34 | 20 | | | |
| | $\Delta ra 97$ | 2.83 | -2.50 | | | |
| | ngy | 2.62 | -1.02 | | | |
| | H ₂ O | 3.10 | -2.50 | | | |
| | 2 | 2.93 | -2.50 | | | |
| | | 3.10 | -2.50 | | | |
| | | 2.68 | -2.50 | | | |
| IdU | Asp15 | 3.17 | -2.11 | -95.79 | -8.47 | .35 mM |
| | | 3.10 | -2.49 | | | |
| | Arg/6 | 2.90 | -2.50 | | | |
| | Alg9/ | 5.20 2.74 | 49 - 89 | | | |
| | H ₂ O | 2.74 | -250 | | | |
| | 1120 | 2.10 | -2.50 | | | |
| | | 3.24 | -1.77 | | | |
| | | 2.30 | 03 | | | |
| CldU | Asp15 | 3.17 | -2.50 | -103.52 | -7.77 | .37 mM |
| | | 2.87 | -2.50 | | | |
| | Arg76 | 3.54 | 16 | | | |
| | A === 0.7 | 2.57 | -2.29 | | | |
| | Alg97 | 3.54 | 33 -2.50 | | | |
| | 1120 | 2.80 | -2.50 | | | |
| | | 2.62 | -2.50 | | | |
| | | 2.60 | -2.50 | | | |
| | | 2.82 | -2.50 | | | |
| | | 3.10 | -2.50 | | | |
| BvdU | Pro43 | 3.56 | 10 | -90.61 | -7.43 | 1.1 mM |
| | Arg45 | 2.29 | .07 | | | |
| | Arg/6 | 2.88 | -1.76 | | | |
| | Tyr08 | 2.60 | -2.31 | | | |
| | Ser101 | 3 45 | 4 5 - 74 | | | |
| | Serror | 2.54 | -2.04 | | | |
| | Phe105 | 3.54 | 011 | | | |
| | H_2O | 3.05 | -2.50 | | | |
| dU | Asp15 | 3.01 | -2.50 | -86.80 | -8.71 | 2.5 mM |
| | Arg76 | 3.08 | -2.13 | | | |
| | | 2.92 | -2.50 | | | |
| | Arg97 | 2.64 | -1.0/ | | | |
| | ЧО | 2.84 | 51 | | | |
| | п ₂ 0 | 3.00 | -2.30 -2.50 | | | |
| | | 2.60 | -2.50 | | | |
| FdU | Asp15 | 3.10 | -2.50 | -74.83 | -8.88 | >5 mM |
| | Arg76 | 2.75 | -2.04 | | | |

(Continued)

| Compound | Residue | Dist. (Å) | Energy (kcal mol^{-1}) | Intermolecular energy (kcal mol^{-1}) | H-bond energy (kcal mol^{-1}) | ^a K _i |
|----------|------------------|-----------|---------------------------------|--|--|-----------------------------|
| | | 2.85 | -1.88 | | | |
| | Arg97 | 2.67 | -1.27 | | | |
| | H ₂ O | 3.10 | -2.49 | | | |
| | 2 | 3.26 | -1.67 | | | |
| | | 3.40 | -1.01 | | | |
| | | 3.10 | -2.50 | | | |
| | | 3.07 | -2.50 | | | |
| | | 3.22 | -1.89 | | | |

Table 2. (Continued)

^aExperimental data.²⁷



Figure 5. Correlation between the intermolecular energies and the K_i, obtained by Caillat et al. (2008), and values of each compound for *Var*TMPK (left) and *Hss*TMPK (right).



Figure 6. Temporal RMSD values for VarTMPK (left) and HssTMPK (right) and the compounds studied. Black lines refer to the enzymes without ligands.

was of 43.1%. The active site residues of *Var*TMPK were determined by analogy with the *Vac*TMPK active site residues former identified by Caillat et al. (2008). Analyzing the alignment with *Hss*TMPK (data not shown), we verified that among 14 residues of the active site 7 are different. This means that there is a similarity of 50% between the active sites and suggests that there is room for the drug design of selective inhibitors.

3.2. Docking studies

Figure 3 shows the re-docking results of TDP inside the active sites of *Var*TMPK and *Hss*TMPK. The RMSD

values observed were of 1.58 and .86 Å, respectively. These values validate the docking methodology considering that, according to literature, a RMSD lower than 2.00 Å is considered acceptable (Kontoyanni, McClellan, & Sokol, 2004; Leach, Shoichet, & Peishoff, 2006; Warren et al., 2006).

The cavities of the active sites of *Var*TMPK and *Hss*TMPK were predicted by MVD software (Thomsen & Christensen, 2006) as having 78,336 and 90,112 Å³, respectively. The inhibitors (Figure 1) were docked into these cavities, according to the procedure described in the methodology, in order to evaluate their interactions



Figure 7. Spatial RMSD of the systems VarTMPK/BrdU (right) and HssTMPK/BrdU (left). BrdU was omitted in the figure for better clarity.



Figure 8. Frames of BrdU in the binding site of *Var*TMPK during the 20.0 ns of MD simulation. The H-bonds formed along the simulation are shown in yellow.

to provide insights to the design of selective inhibitors. The best poses for each compound were selected on the basis of the interaction energies with the enzymes (intermolecular/electrostatic and H-bond) and best superposition to TDP, as illustrated in Figure 4 for the system BrdU/*Var*TMPK).

Tables 1 and 2 present the active site residues of both enzymes participating in H-bonds with each compound studied, the energy values obtained, and the respective experimental values of K_i for these compounds inside *Vac*TMPK, reported by Caillat et al. (2008). Our theoretical results present a good correlation to the experimental values with R^2 of .978 and .976 for *Var*TMPK and *Hss*TMPK, respectively (see Tables 1 and 2 and Figure 5).

The lower values of energies observed in Table 1 related to the interactions of the ligands with *Var*TMPK suggest that these compounds could have more affinity for this enzyme. It was also observed that, except for fluxuridine (FdU), the insertion of halogen groups in the position 5' of the pyrimidine ring reduces the energy values. Brivudin (BvdU) presented lower energy inside *Var*TMPK than in *Hss*TMPK (-119.06 kcal mol⁻¹ *versus* – 90.61 kcal mol⁻¹). This could be related to the narrowest cavity of *Hss*TMPK that disable this enzyme to accommodate bulkier molecules inside its active site, allowing a major specificity to substrates, like pirimidines with larger groups in the position 5' or purine nucleotides like dGMP, as reported by Caillat et al. (2008).

3.3. MD studies

The best poses obtained in the docking studies were submitted to 20 ns of MD simulations inside *Var*TMPK and *Hss*TMPK in order to corroborate the docking results and assess the dynamical behavior of the compounds.

The energy plots for the 14 systems simulated showed that the total energy tends to stability since the first ps of simulation.

In the plots of temporal RMSD presented in Figure 6, it is also possible to observe the equilibration of the systems since the first ps. The RMSD values never overpassed .10 nm during the simulations except for IdU inside *Var*TMPK that stabilized at .15 nm, and for BvdU and FdU inside *Hss*TMPK that stabilized around .20 nm.

The spatial RMSD for the systems *Var*TMPK/BrdU and *Hss*TMPK/BrdU are shown in Figure 7. As can be seen, the regions with major fluctuations during the simulation (larger tubes in the picture) correspond to the extremities and loops. On the other hand, the residues in the active site region and in the α -helix and β -sheets presented lower fluctuations (lower thickness in the tubes). This behavior was common to all the 14 systems studied.

The docking studies showed that BrdU was able to establish H-bonds with residues Asp13, Ar41, Arg72, Tyr144, and two water molecules (Figure 3). The MD simulations showed that this compound was able to establish until 6 H-bond during the simulation and keep between 2 and 4. The frames of the MD simulation for this compound inside *Var*TMPK (Figure 8) show the interactions with residues Asp13, Phe68, Arg72, Tyr101, Glu142, and water molecules as suggested in the docking studies, except for the H-bond with Tyr144.

Regarding CldU, MD showed the formation of 4 Hbonds with *Var*TMPK during the simulation, including H-bonds with residues Glu142 and Tyr144 not observed in the dockings. The frames extracted for this compound also showed the possibility of hydrophobic interactions with Phe68 (Figure 9). This interaction was observed to all compounds inside *Var*TMPK but not inside *Hss*TMPK and, therefore, can contribute for an additional stabilization and permanence of them inside *Var*TMPK.

In the same way as for CldU, IdU formed about 4 H-bonds with *Var*TMPK during the simulation and an additional interaction with Tyr101 not plotted in the docking study.



Figure 9. Hydrophobic interactions observed along the MD simulation between CldU (in tubes) and the side chain of Phe68.



Figure 10. H-bonds formed among BrdU and residues Asp13, Arg72, Tyr101, and Glu142 during 20,000 ps of MD simulation.

The analysis described above were performed for all compounds studied inside both enzymes and, in general, good concordances with the docking results were observed. The results obtained point the interactions with residues Asp13, Phe68, Arg72, Tyr101, Glu142, and Tyr144 as essential for the drug design. As BrdU was the ligand with the most promissing results, we show in plots of Figures 10 and 11 the Hbonds formed among this compound and residues Asp13, Arg72, Tyr101, and Glu142 during the MD simulations and, also, the variation of distance between the aromatic rings of BrdU and Phe68. As can be seen, BrdU was able to form and keep until 3 Hbonds with Asp13, while its aromatic ring was kept at about 6.0 Å from the aromatic ring of Phe68, suggesting the formation of the hydrophobic interaction.



Figure 11. Variation of distance between the aromatic rings of BrdU and Phe68 during 20,000 ps of MD simulation.

Our studies suggest that, despite of its larger size, the cavity of *Hss*TMPK is narrower and have no empty space to accommodate the modifications in the pyrimidine rings of the compounds. This could be observed analyzing the results for BvdU. So, the derivatives of the compounds studied should present modifications in the position 5' of their pyrimidine rings in order to be more promising as selective inhibitors.

4. Conclusion

The docking and MD results obtained in the present work corroborated to each other and suggested a good selectivity of the compounds face to *Hss*TMPK. Also, the good correlations with experimental data suggest our docking protocol as a suitable tool for an initial prediction of affinities of potential new inhibitors.

Our results pointed out CldU and BrdU as the most promising inhibitors of *Var*TMPK. These compounds should, therefore, be submitted to experimental studies. Additionally, we observed that the interactions with residues Asp13, Phe68, Arg72, Tyr101, Glu142, and Tyr144 and the differences in the shapes of the active sites of *Var*TMPK and *Hss*TMPK should be taken in account in the drug design of selective inhibitors for *Var*TMPK, through modifications on the positions 3' and 5' of the pyrimidine rings of CldU and BrdU.

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