

Green Synthesis of Cerium Oxide Nanoparticles, Characterization, and Their Neuroprotective Effect on Hydrogen Peroxide-Induced Oxidative Injury in Human Neuroblastoma (SH-SY5Y) Cell Line

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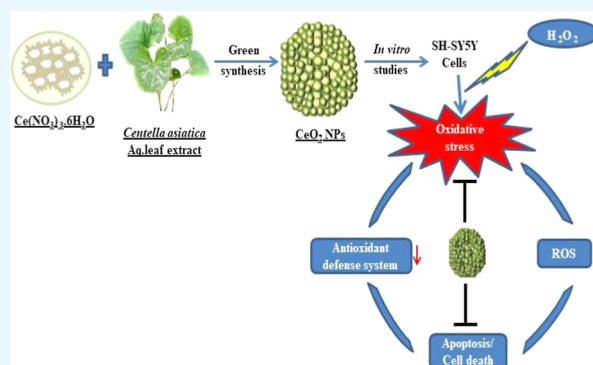


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ABSTRACT: Cerium oxide nanoparticles (CeO₂NPs) have a broad scale of applications in the biomedical field due to their excellent physicochemical and catalytic properties. The present study aims to synthesize the CeO₂NPs from *Centella asiatica* (*C. asiatica*) leaf extract, which has been used in Indian traditional medicine for its neuroprotective properties. The CeO₂NPs were characterized by ultraviolet–visible, X-ray diffraction, Fourier transform infrared, Raman spectroscopy, scanning electron microscopy–energy dispersive X-ray spectroscopy, and high-resolution transmission electron microscopy. The antioxidant property was evaluated by 2,2-di (4-*tert*-octyl phenyl)-1-picrylhydrazyl and OH radical assays. The neuroprotective potential was assessed against the oxidative stress (OS) induced by H₂O₂ in the human neuroblastoma (SH-SY5Y) cell line. CeO₂NPs exhibited significant DPPH and OH radical scavenging activity. Our results revealed that CeO₂NPs significantly increased H₂O₂-induced cell viability, decreased lactate dehydrogenase, protein carbonyls, reactive oxygen species generation, apoptosis, and upregulated antioxidant enzyme activity. Our findings suggest that the CeO₂NPs protect the SH-SY5Y cells from OS and apoptosis, which could potentially counter OS-related neurodegenerative disorders.



1. INTRODUCTION

Oxidative stress (OS) triggered by reactive oxygen species (ROS) has been implicated in several neurological diseases.¹ ROS, such as hydrogen peroxide (H₂O₂), hydroxyl radicals, and superoxide radicals, are generated in physiological conditions and can cause damage to proteins, lipids, and nucleic acids.² H₂O₂ is linked with the production of reactive hydroxyl radicals generated through Fenton's reaction and can promote apoptosis in neuronal cells.³ Therefore, identifying therapeutic strategies that inhibit ROS generation and treatment of these OS-linked neurodegenerative diseases. Nanotechnology represents a novel and promising approach due to the unique nanoparticle properties that allow them to penetrate the blood–brain barrier to manage various neurodegenerative disorders.⁴ Among them, cerium oxide nanoparticles (CeO₂NPs) play a vital role in a wide range of applications.⁵ CeO₂NPs recently gained interest in biological applications due to their potent free radical quenching activity due to the Ce³⁺/Ce⁴⁺ redox stoichiometry of the surface.⁶ They mimic superoxide dismutase (SOD), catalase (CAT), peroxidase, hydroxyl, and nitric oxide (NO) scavenging properties.⁷ Thus,

researchers are drawn toward synthesizing nanoparticles (NPs) using biological sources deployed as stabilizing and reducing agents, such as plants, bacteria, and fungus. The synthesis of NPs using plant extract is cost-effective, eco-friendly, fast, and simple.⁸

Medicinal plants traditionally used in ayurvedic medicine have gained recognition worldwide for their antioxidant activity, attributed to the quenching of free radicals.⁹ *Centella asiatica* L (*C. asiatica*) is a small creeping perennial herb belonging to the Apiaceae or Umbelliferae family that flourishes in moist and damp areas of India, Indonesia, and other parts of Asia.¹⁰ *C. asiatica* is a natural antioxidant used as a nerve tonic. It is known to exert protective action against age-associated modifications in the brain's antioxidant defense system.¹¹ Based on this, the present study aimed to synthesize

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nanoceria using different volumes of *C. asiatica* extract and investigate the neuroprotective potential of H₂O₂-induced OS in SH-SY5Y cells.

2. MATERIALS AND METHODS

2.1. Chemicals and Reagents. Cerium nitrate hexahydrate [Ce(NO₃)₃·6H₂O], deoxy-2-ribose, 2,2-di (4-*tert*-octyl phenyl)-1-picrylhydrazyl (DPPH), 4',6-diamidino 2 phenylindole (DAPI), 2',7' dichlorofluorescein diacetate (DCFH-DA), 2,4-dinitrophenylhydrazine (DNPH), epinephrine, glutathione reductase, reduced glutathione, guanidine hydrochloride, propidium iodide, RNase A, *t*-peroxyhydroxide, thiobarbituric acid, annexin V, and trizol were procured from Sigma (St. Louis, MO, USA). Dulbecco's modified Eagle's medium (DMEM), Ham's F12K mixture, fetal bovine serum (FBS), antibiotic-antimycotic solution, and 3-(4,5-dimethylthiazol-2)-2,5-diphenyl tetrazolium bromide (MTT) were purchased from Himedia chemicals (Mumbai). All solvents were of spectral grade, and other chemicals were of analytical grade.

2.2. Preparation of Aqueous Extract. Fresh leaves of *C. asiatica* were bought from local markets in Bengaluru, India. After washing thoroughly with water, the leaves were shade dried and finely pulverized with a blender. The aqueous extract was made by suspending the powder in warm water. The mixture was kept at room temperature for 24 h and then filtered using Whatman no.1 paper. Further, the filtrate was freeze-dried and stored at -20 °C for further studies.

2.3. Synthesis of CeO₂NPs. The CeO₂NPs were synthesized by solution combustion method according to Lakshmeesha et al.¹² Briefly, various volumes of *C. asiatica* extract (5, 10, 20, and 30 mL) were mixed with the precursor (cerium nitrate hexahydrate) and stirred for 10 min on a magnetic stirrer. The resultant solution was calcined in a preheated muffle furnace at 570 ± 10 °C for 10–20 min.

2.4. Characterization of CeO₂NPs. Fourier transform infrared (FTIR) spectral analysis of the samples was performed using a PerkinElmer FTIR spectrophotometer (Spectrum-1000). The UV–vis spectroscopy was carried out using an SL 159 ELICO UV–vis spectrophotometer. The crystalline nature and phase purity of the NPs were studied using a powder X-ray diffractometer (Shimadzu 7000s). Raman analysis was carried out using a Raman spectrometer (ID Raman reader). The surface morphology was assessed by scanning electron microscopy (SEM) (JEOL JSM 840A) equipped with an energy-dispersive X-ray (EDX) system (Kevex Sigma TM Quasar, USA). The size and selected area diffraction (SAED) patterns were obtained using high-resolution transmission electron microscopy (HR-TEM) (TecnaiG2 20).

2.5. Antioxidant Assays. **2.5.1. DPPH Scavenging Assay.** DPPH assay was performed according to the procedure of Ravikiran et al.¹³ 0.5 mL of green synthesized nanoceria were added with 0.1 mM DPPH solution (1 mL) prepared in 95% ethanol and incubated for 20 min at room temperature (RT). The absorbance was recorded at 517 nm against a reagent blank.

Scavenging effect (%)

$$= \frac{\text{Abs}_{\text{control}(517)} - \text{Abs}_{\text{sample}(517)}}{\text{Abs}_{\text{control}(517)}} \times 100$$

where Abs_{control} = absorbance of the control and Abs_{sample} = absorbance of the test sample.

2.5.2. Hydroxyl Radical Assay. Hydroxyl radical scavenging assay was performed by the procedure of Halliwell and Gutteridge.¹⁴ 20 μL of various concentrations of the sample were mixed with a reaction mixture composed of 120 μL of deoxyribose (20 mM), phosphate buffer (400 μL, 0.1 M), H₂O₂ (40 μL, 20 mM), FeSO₄ (40 μL, 500 mM), and the total volume was made up to 800 μL with distilled water. The reaction mixture was allowed to stand at 37 °C for 30 min, and the reaction was blocked by adding 2.8% trichloroacetic acid (TCA) (0.5 mL) and 0.6% *tert*-butyl alcohol (TBA) (0.4 mL). Finally, the reaction mixture was placed in a boiling water bath for 20 min. The absorbance was measured at 532 nm.

Scavenging effect (%)

$$= \frac{\text{Abs}_{\text{control}(532)} - \text{Abs}_{\text{sample}(532)}}{\text{Abs}_{\text{control}(532)}} \times 100$$

where Abs_{control} = absorbance of the control and Abs_{sample} = absorbance of the test sample.

2.6. In Vitro Studies. **2.6.1. Cell Culture and Treatment.** The SH-SY5Y cells were obtained from the National Centre for Cell Science, Pune. The cells were cultured in 25 cm³ tissue culture flask (1 × 10⁶ cells/well) in DMEM and Ham's F12 medium (1/1), fortified with 10% FBS, penicillin (100 U/mL), and streptomycin (100 μg/mL). The cells were maintained at 37 °C in a CO₂ incubator, and the medium was replenished every 2–3 days upon attaining 70–80% confluency. The cell viability was assessed after exposure to varying concentrations of H₂O₂ (3.125 to 400 μM) for 24 h to determine the IC₅₀ value. The protective effect of the biosynthesized CeO₂NPs on SH-SY5Y cells was estimated by pretreatment with H₂O₂ for a period of 24 h, followed by treatment with different concentrations of nanoceria (10–320 μg/mL) for 24 h. The control cells were also maintained without CeO₂ and H₂O₂ treatment. The cell viability was assessed by MTT assay following the protocol of Mosmann.¹⁵

2.6.2. MTT Assay. The SH-SY5Y cells were pretreated with H₂O₂ (135 μM) for 24 h, followed by the treatment with CeO₂NPs (10, 20, 40, 80, 160, and 320 μg/mL) for the next 24 h. The cells were then incubated with MTT (5 mg/mL in PBS) for 4 h at 37 °C. The blue formazan crystals that appeared in live cells were solubilized with DMSO. The absorption was recorded at 570 nm in an enzyme-linked immunosorbent assay (ELISA) plate reader (Mode Tecan 1650). The results were expressed as a percentage of the MTT reduction relative to the absorbance of the control cells.

2.6.3. Lactate Dehydrogenase Release Assay. The cytotoxicity was estimated by measuring the lactate dehydrogenase (LDH) release into the culture medium (LDH kit, DIATEK Diagnostics). The cells (SH-SY5Y) were plated in a six-well culture plate (1 × 10⁶ cells/well) and treated with H₂O₂ (135 μM) alone or with CeO₂NPs. After the treatment, the LDH leakage was estimated according to the maker's protocol. The sample absorbance was measured spectrophotometrically at 340 nm, and the LDH leakage was expressed as % LDH release.

2.6.4. Cellular Uptake Studies. The cellular uptake of CeO₂NPs was investigated through flow cytometry analysis following the method of Chauhan et al.¹⁶ The nanoparticle uptake analysis was based on the intensity of the side scatter (SSC) recorded for each cell. The intracellular density of

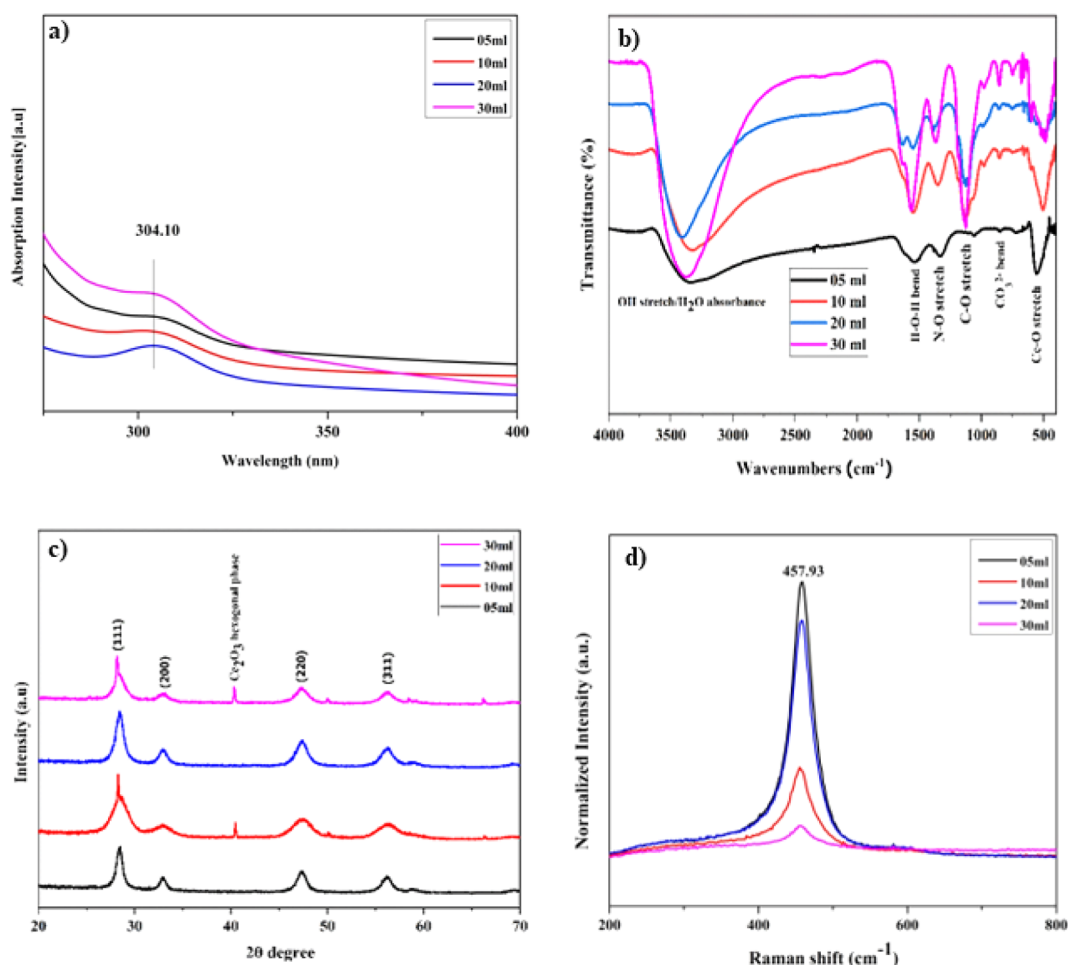


Figure 1. Characterization of CeO₂NPs synthesized using different volumes of *C. asiatica* extract. (a) UV-vis spectroscopy, (b) FTIR, (c) XRD pattern, and (d) Raman spectroscopy.

internalized NPs was directly proportional to the intensity of SSC.

2.6.5. Preparation of Cell Extract. The SH-SY5Y cells were plated in 6-well plates at a density of 1×10^6 cells/well. The cells were pretreated with $135 \mu\text{M}$ H₂O₂ followed by cell treatment with biosynthesized CeO₂NPs for 24 h. The collected cells were suspended in ice-cold PBS (pH 7.4) and centrifuged at 10,000g (Plastocrafts, Superspin RV/FM) at 4 °C for 20 min. The supernatant was utilized to estimate the enzymatic antioxidants and markers of OS. The protein content was estimated by the method of Lowry et al.¹⁷ using bovine serum albumin (BSA).

2.6.6. Measurement of Antioxidant Enzyme Activities. The SOD activity was estimated by the method of Misra and Fridovich.¹⁸ The tissue supernatant (100 μL) was mixed with 880 μL of carbonate buffer (0.05 M, pH 10.2) and ethylenediaminetetraacetic acid (EDTA) (0.1 mM), followed by the addition of 20 μL of epinephrine (30 mM, 0.05% acetic acid). The enzyme activity was estimated by recording the absorbance at 480 nm for 5 min. One unit of enzyme activity is the amount of enzyme that resulted in the inhibition (50%) of epinephrine deterioration.

The determination of CAT activity was performed following the protocol of Aebi.¹⁹ The tissue supernatant (100 μL) was added to an equivalent volume of absolute alcohol and incubated at RT (30 min), followed by triton X-100 addition.

The desired amount of this reaction mixture was mixed with an equal volume of H₂O₂ (0.066 M) in phosphate buffer, and the decline in absorbance was recorded for 1 min at 240 nm. The enzyme activity was measured using an extinction coefficient of 43.6 M cm^{-1} . One unit of enzyme activity was equivalent to the moles of H₂O₂ degraded $\text{min}^{-1} \text{ mg}^{-1}$ of protein.

The protocol of Flohe and Gunzler²⁰ was used to measure glutathione peroxidase (GPx) activity. The reaction mixture contains phosphate buffer (500 μL), 100 μL of reduced glutathione (GSH, 0.01 M), 100 μL of reduced nicotinamide adenine dinucleotide phosphate (NADPH) (1.5 mM), and glutathione reductase (100 μL). The tissue extract (100 μL) was added with a known volume of the reaction mixture and incubated (37 °C for 10 min). After incubation, 50 μL of 12 mM *t*-butyl hydroperoxide was mixed with a tissue reaction mixture (450 μL). The absorbance was noticed at 340 nm for 3 min. The calculation of enzyme activity was done by taking the molar absorption coefficient of $6.229 \times 10^3 \text{ M cm}^{-1}$. A unit of activity equals mM NADPH oxidized/min/mg protein.

2.6.7. Markers of OS. **2.6.7.1. Measurement of Protein Oxidation.** Protein carbonyls (PCs) were quantified by the procedure of Levine et al.²¹ The supernatant (100 μL) from the tissue extract was incubated with 10 mM (2,4-dinitrophenyl)hydrazine (DNPH) (0.5 mL) in HCl (2 M) in the dark for 60 min. 20% TCA (0.5 mL) was then added to precipitate the protein and centrifuged at 4 °C for 3 min at

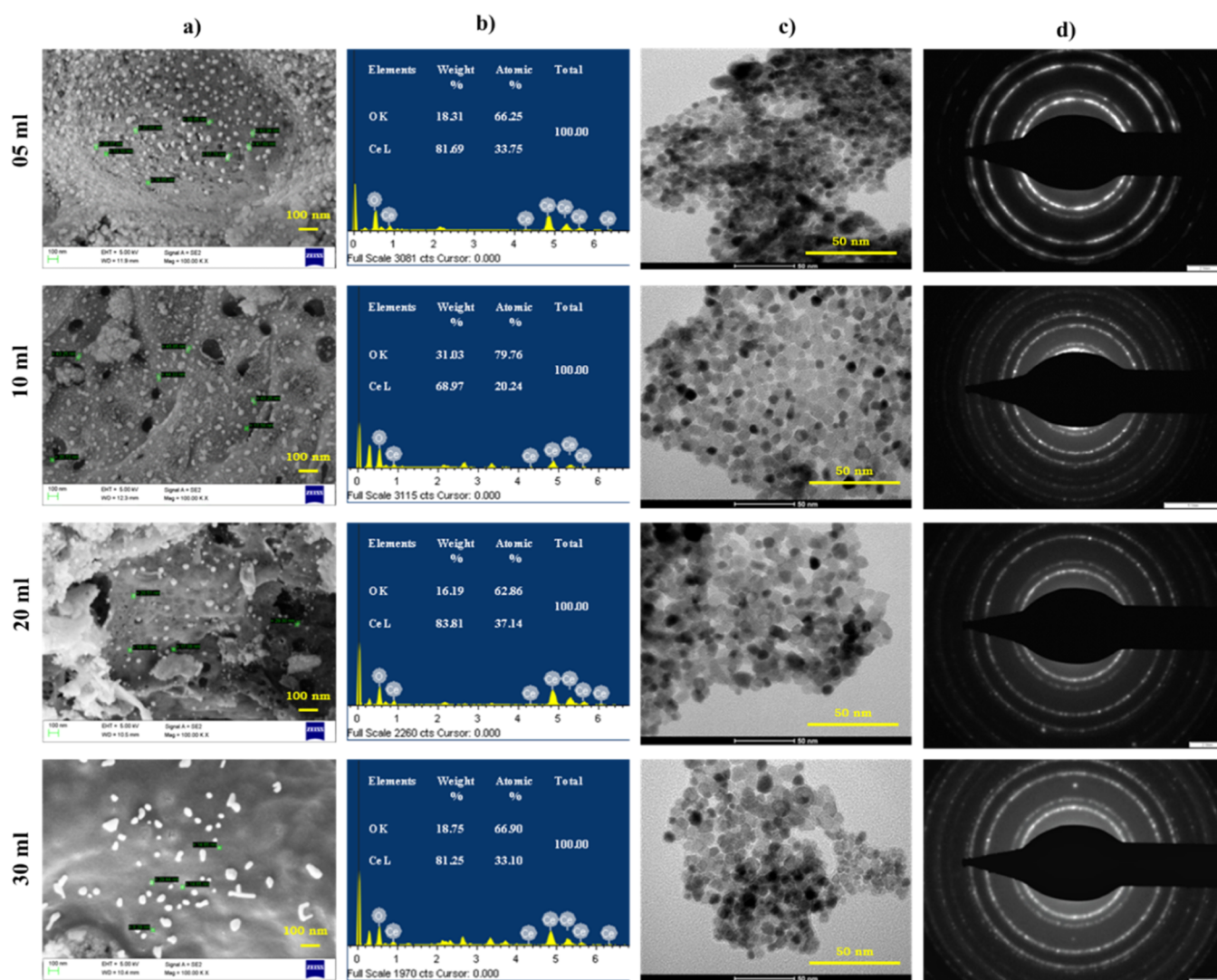


Figure 2. Morphology of CeO₂NPs synthesized using different volumes of *C. asiatica* extract. (a) SEM images, (b) EDX analysis, (c) HR-TEM images, and (d) SAED patterns.

10,000g. The unsolubilized DNPH was removed by washing the pellet with 1/1 ethyl acetate/ethanol twice by centrifuging for 5 min (3400g). Then, the pellet was suspended in 6 M guanidine hydrochloride (1.5 mL) in phosphate buffer of pH 6.5, and the absorbance was measured at 370 nm.

2.6.7.2. Evaluation of ROS Inhibition by Flow Cytometry. ROS inhibition was quantified by the method of Liu et al.²² using a flow cytometer. The cell monolayer was washed and supplemented with fresh media and incubated for 2 h before the treatment. DCFH-DA solution (25 μ M) was prepared in DMEM media, and 1 mL of this solution was poured into each well. After the solution was incubated for 45 min, the DCFH-DA solution was removed carefully. The cells were rinsed with 1 \times PBS once without disturbing the cell monolayer. The cells were treated with H₂O₂ (135 μ M) for 24 h and then subjected to treatment with CeO₂NPs (20 and 40 μ g/mL) for 24 h. 5 mM N-acetylcysteine (NAC) served as a positive control (ROS inhibitor). Post-incubation, the cells were processed for flow cytometry analysis by washing in 1 \times PBS and suspending the cells in sheath fluid. The fluorescent signal was then detected using the FL1H detector.

2.6.8. Apoptosis Analysis. **2.6.8.1. DAPI Staining.** DAPI staining was done following the modified protocol of Lingappa et al.²³ The cells were treated as mentioned previously and fixed using 4% paraformaldehyde in PBS (10 min), followed by washing with PBS and staining with DAPI (1 μ g/mL in PBS) for 20 min. After staining, the cells were observed under a confocal microscope (CKX41, Olympus Corporation, Japan). The apoptotic cells were identified using their condensed, fragmented, and degraded nuclei.

2.6.8.2. Apoptosis Detection Using PI and Annexin V-FITC by Flow Cytometry. Before apoptosis was introduced, the cells were cultured for 24 h. Pretreatment was done with 135 μ M H₂O₂ for 24 h, and subsequent treatments with different concentrations (20 and 40 μ g/mL) of biosynthesized CeO₂NPs for another 24 h were performed. Apoptosis was detected following the method of Zheng et al.²⁴ After treatment, the cells were processed and washed with cold PBS (twice), and then 1 mL of 1 \times binding buffer. 500 μ L of the cell suspension was added with 10 and 5 μ L of PI and annexin V, respectively, followed by incubation for 15 min in the dark at RT. Post incubation, the cells were subjected to flow cytometric analysis.

2.7. Statistical Analysis. The data is represented as mean \pm SE. One-way analysis of variance (ANOVA) followed by Tukey's test was performed to analyze the antioxidant enzyme activities, markers of OS, and apoptosis. Probability values of less than 0.05 were statistically significant. The statistical analysis was carried out using a GraphPad prism.

3. RESULTS AND DISCUSSION

3.1. Characterization of CeO₂NPs. The prominent absorption peak observed at 304 nm confirms the formation of CeO₂NPs (Figure 1a). This peak may be attributed to the transfer of charge from the 2p valence band of O²⁻ to the Ce⁴⁺ 4f band.²⁵ The absorption band of the biosynthesized CeO₂ samples slightly shifts toward the lesser wavelength side, suggesting the higher bandgap in energy based on the vibrational frequencies of ions. CeO₂NPs exhibit strong absorption bands at 304 nm in the UV range, and no band of Ce (IV) in the visible region indicates the occurrence of a +3 ionic state of Ce ions. FTIR spectrum of CeO₂NPs showed several significant IR peaks in the range of 4000–400 cm⁻¹, as shown in Figure 1b. The spectra of the samples reveal a firm Ce metal–oxygen stretching band at about 512 cm⁻¹. The peaks at 3375.80 and 1622.71 cm⁻¹ represent the hydroxyl stretch and water, respectively.

The phase formation and crystalline structure of the synthesized CeO₂NPs were studied using powder X-ray diffraction (PXRD) showing the patterns scanned at 2 θ in the range of 10°–80° (Figure 1c). These profiles confirmed the polycrystalline nature and indicated the pure face-centered cubic phase structure. The peaks correspond to the crystal planes (111), (200), (220), (311), (222), and (400), which were well-matched with JCPDS no: 81-0792 of the CeO₂ cubic crystal group with space group *Fm* $\bar{3}$ *m* (225). In contrast, a slightly Ce₂O₃ hexagonal phase (JCPDS-74-1145) was observed when the *C. asiatica* extract was used for a few samples.

Scherer's formula was used to calculate the average crystallite size.²⁶ The average crystallite size and structural parameters are presented in Table S1. Observed broadening of the diffraction peaks in *C. asiatica*-extract-mediated samples indicates the reduction of crystallite sizes to nanoregime. The same was confirmed by HR-TEM analysis in the subsequent section. In order to validate the obtained PXRD data, a Rietveld refinement pattern was performed for CeO₂NPs prepared from a 20 mL *C. asiatica* extract sample using the FULLPROF program.²⁷ The refined parameters such as atomic functional positions and occupancy for the as-formed CeO₂NPs sample are tabulated in Table S2. Few peaks left out with error corresponded to the Ce₂O₃ of the hexagonal phase (JCPDS-74-1145) as an impurity observed in the prepared sample. The cubic structure of the synthesized CeO₂NPs was also confirmed by Raman spectroscopy (Figure 1d). The intense peak at 457 cm⁻¹ confirms the Raman active F_{2g} mode of CeO₂, which may be due to the Ce–O8 vibrational symmetrical stretching mode.²⁸ More intense Raman peaks were observed in *C. asiatica*-mediated CeO₂ samples, and it became maximum for 5 mL and very close to the maximum for 20 mL of *C. asiatica* samples, indicating increased oxygen vacancies in these samples. The uniformity of the strain within each particle and the average strain formed by all the particles influence the position of the Raman line.²⁹ In the current study, the particle sizes were found to be uniform, and hence, collective strain emerging from all the particles is similar. Also,

the peak asymmetries and shifting were not observed much due to a significantly less quantity of nonuniform strain contributed by most of the particles.

SEM microscopy at different magnification illustrates that the particles were uniform and spherical, and the mean size was less than 35 nm (Figure 2a). The EDX spectra of the synthesized CeO₂NPs indicate Ce and O elements and the high purity of the synthesized nanoceria (Figure 2b). The characterization result evidences the formation of CeO₂NPs, which had particle size in the nano regime and were pure in their chemical composition. SEM images showed (Figure 2a) an increase in porosity and slightly decreased particle size of the samples with an increase in the volume of the *C. asiatica* extract. The CeO₂NPs were agglomerated to form spherical-like clusters, with sizes ranging from micrometers to a few tenths of microns. The presence of agglomerated spherical clusters and numerous voids and pores seen in the samples may be due to the production of gases during solution combustion synthesis.³⁰

The internal morphology of the sample (CeO₂NPs obtained using 20 mL *C. asiatica* extract) was characterized by HR-TEM and selected area electron diffraction (SAED). The HR-TEM image displayed spherical-shaped particles of \sim 10 nm size (Figure 2c). The SAED pattern indicates several weak Scherrer rings conforming to the reflections of the cubic CeO₂ sample and also representing the high crystallinity of the NPs (Figure 2d). These reflection planes were in good accordance with PXRD results. HR-TEM images revealed the presence of 3.124 Å interplanar spacing, which correlated to the distance of the crystallographic cubic planes (111) of CeO₂ lattice fringes, suggesting the formation of high-quality cubic CeO₂NPs. HR-TEM analysis revealed a crystallite size of about 10 nm, which reasonably complies with the size estimated using the PXRD pattern by Scherrer's equation.

3.2. Antioxidant Assays. The free radical scavenging ability was evaluated by measuring the DPPH and hydroxyl radical activity. DPPH is a stable free radical and widely accepted method to determine an antioxidant compounds' free radical quenching ability.³¹ The DPPH reduction was assessed by observing the decline in absorbance due to the formation of stable DPPH molecules. 10 mL of the prepared CeO₂NPs revealed significant dose-dependent activity, as apparent by its lesser IC₅₀ value of 6.95 \pm 1.16 μ g/mL (Table 1). Our results are comparable with the reports of Nezhad et al.³² wherein green synthesized CeO₂NPs significantly inhibited the free radicals.

The hydroxyl [OH] radical generated by the Fe²⁺ ions with a highly reactive oxygen center was estimated by the extent of

Table 1. DPPH and Hydroxyl Radical Scavenging Activities of CeO₂NPs Synthesized Using Different Volumes of *C. asiatica* Extract^a

CeO ₂ NPs synthesized using different volumes of extract	free radical scavenging activity (IC ₅₀ μ g/mL)	OHDPPH
05 mL	09.21 \pm 1.20 ^b	21.75 \pm 1.17
10 mL	06.95 \pm 1.16	12.21 \pm 1.19 ^b
20 mL	11.26 \pm 1.18 ^b	11.23 \pm 1.20
30 mL	11.82 \pm 1.36 ^b	17.93 \pm 1.15
Vit C	03.99 \pm 1.76	20.37 \pm 1.16

^aThe data is analyzed by one-way ANOVA and expressed as mean \pm standard error (SE) from three independent experiments. ^b*p* < 0.05, samples compared with vitamin C.

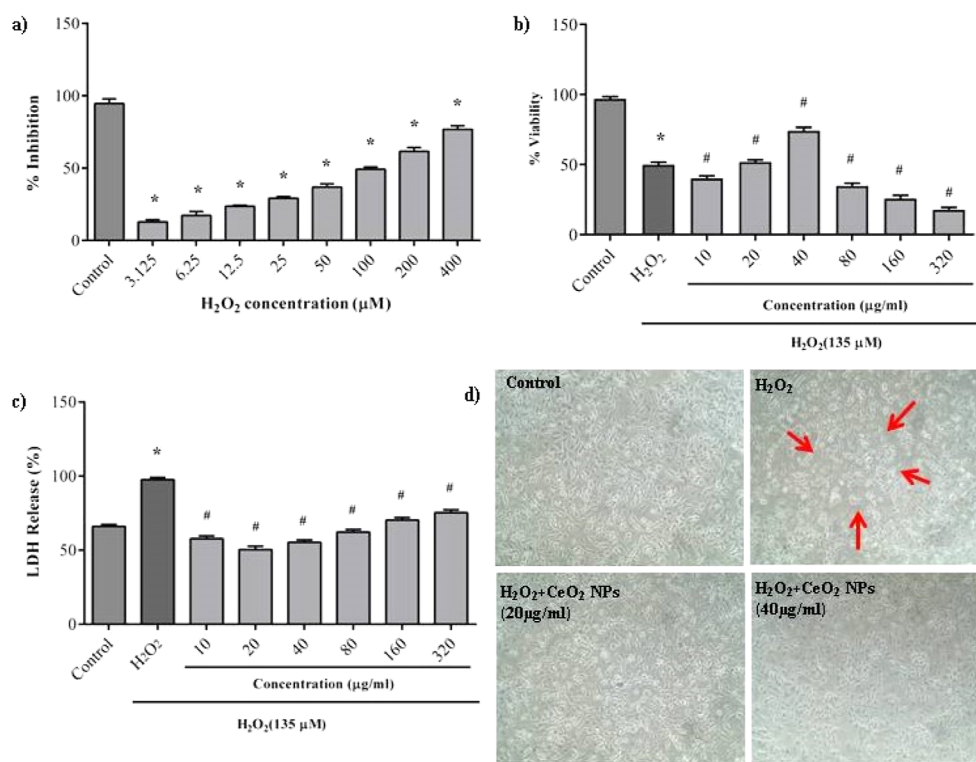


Figure 3. Effects of CeO₂NPs on H₂O₂-induced cytotoxicity in SH-SY5Y cells. (a) Cells treated with varying concentration of H₂O₂ (3.125–400 μg/mL) for 24 h. (b) Cells pretreated with H₂O₂ followed by incubation with CeO₂NPs for 24 h, and cell viability was determined by MTT assay. (c) Cells pretreated with H₂O₂ followed by incubation with CeO₂NPs for 24 h, and cell cytotoxicity was measured by LDH assay. (d) Morphology of SH-SY5Y cells observed under phase contrast microscope. The data is expressed as mean ± SE from three independent experiments and analyzed by one-way ANOVA followed by Tukey's test. **p* < 0.05 significance compared with control; #*p* < 0.05 significance compared with H₂O₂-treated cells.

deoxyribose degradation.³³ As depicted in Table 1, a remarkable feature in OH radical activity is that the CeO₂NPs prepared with 10 and 20 mL of plant extract exhibited higher inhibitory activity (12.21 ± 1.19 and 11.23 ± 1.20 μg/mL) compared to that of vitamin C. The antioxidant properties of CeO₂NPs are attributable to the greater oxygen vacancy that enhances its reactivity leading to higher radical scavenging ability.³⁴

3.3. In Vitro Studies. **3.3.1. Dose Fixation.** CeO₂NPs synthesized using different volumes of plant extract (5, 10, 20, and 30 mL) were screened to determine the neuroprotective activity against OS induced by H₂O₂ in SH-SY5Y cells. The effective volume of plant-extract-synthesized CeO₂NPs was assessed based on the MTT, LDH, and the levels of antioxidant enzyme activities and protein oxidation. 20 mL of prepared CeO₂NPs was most effective in neuroprotective activity and enhanced antioxidant status. Hence, 20 mL of prepared CeO₂NPs was selected for further studies.

3.3.2. Effect of H₂O₂ on Cell Viability through MTT Assay. H₂O₂ is one of the most potent inducers of OS by producing highly reactive OH radicals through the Fenton reaction.³⁵ H₂O₂ is a physiologically and pathologically significant cytotoxicant that can quickly diffuse through the plasma membrane.³⁶ MTT assay, which is a sensitive and reliable indicator of the cells' metabolic activity, was used to determine the viability of the SH-SY5Y cells upon H₂O₂-induced OS. This assay is based on the capacity of functional mitochondria to convert MTT dye into formazan crystals.^{37,38} As evident from Figure 3, the SH-SY5Y cells exhibited dose-dependent cytotoxicity upon treatment with various concentrations of

H₂O₂ (3.125–400 μM) with an IC₅₀ value of 135 μM. Hence, this concentration was fixed to assess the protective activity of CeO₂NPs against H₂O₂ exposure in further experiments.

3.3.3. Effect of Synthesized CeO₂NPs against H₂O₂-Induced Cytotoxicity. To evaluate the cytoprotective ability of CeO₂NPs on H₂O₂-mediated cytotoxicity, the SH-SY5Y cells were treated with varying concentrations of CeO₂NPs for 24 h after exposure to H₂O₂. As shown in Figure 3a, the H₂O₂-treated cells significantly decreased cell viability. However, treatment of CeO₂NPs at 20 and 40 μg/mL remarkably attenuated H₂O₂-induced cytotoxicity by 74.44% and 51.77% over the control cells. The higher doses of CeO₂NPs reduced the cell viability of the H₂O₂-exposed cells (Figure 3b). We further examined the levels of LDH leakage, which is an important marker of the membrane integrity and cell death. Our results (Figure 3c) revealed higher LDH leakage in H₂O₂-treated cells. However, treatment with 20 and 40 μg/mL of CeO₂NPs significantly reduced the LDH leakage (50.45% and 55.45%) over cells treated with H₂O₂ alone.

The morphological observations also supported the protective effects of CeO₂NPs. H₂O₂-exposed cells showed irregular shape, membrane blebbing, detachment, shrinkage, and higher cell mortality. Meanwhile, treatment with CeO₂NPs prevented these morphological changes, depicting the neuroprotective activity of CeO₂NPs against H₂O₂-mediated cytotoxicity (Figure 3d).

3.3.4. Cellular Uptake Studies. Cellular uptake of CeO₂NPs in the SH-SY5Y cells was studied using flow cytometry analysis based on SSC signals. The CeO₂NPs were internalized within 24 h of exposure in a concentration-dependent manner, as

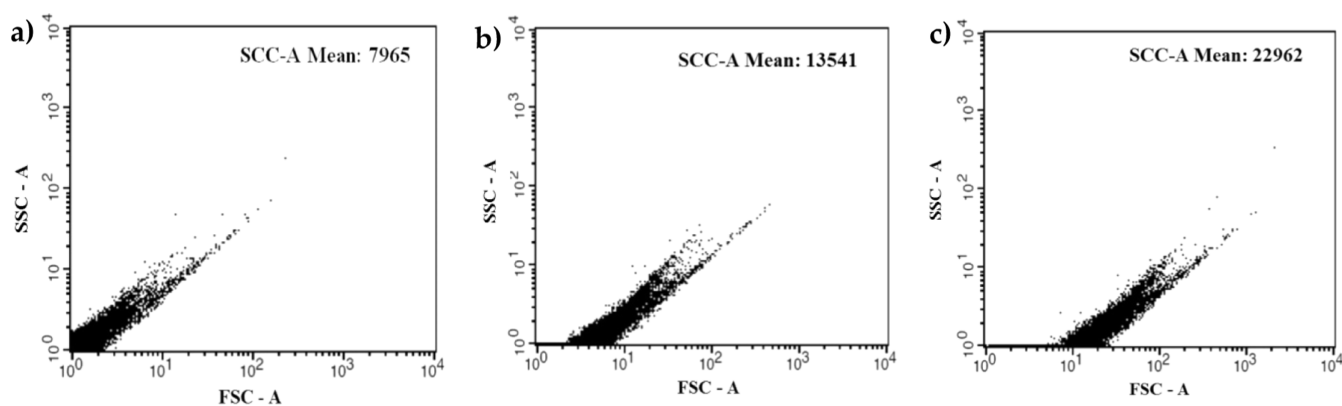


Figure 4. Cellular uptake of CeO₂NPs in SH-SY5Y cells. (a) Control, (b) CeO₂NPs (20 µg/mL), and (c) CeO₂NPs (40 µg/mL).

indicated by increasing SSC intensity compared to that of control cells. As shown in Figure 4, the maximum uptake of CeO₂NPs was found at 40 µg/mL (22,962). The increase in SSC intensity indicates the increased cell density, which can be correlated to the successful uptake of CeO₂NPs in the SH-SY5Y cells. Our findings were in accordance with the studies of Patel et al.³⁹

3.3.5. Antioxidant Enzyme Activities. SOD, CAT, and GPx are the first lines of an antioxidant defense system. These enzymes protect the cells by scavenging free radicals produced by various biochemical reactions in the body, thereby making less toxic metabolites and inhibiting OS.^{40,41} To evaluate the antioxidant ability of nanoceria, we assessed the SOD, CAT, and GPx activities. Our results revealed diminished antioxidant enzyme activity in H₂O₂-treated cells with respect to control, which might be due to the increased generation of ROS. However, a prominent increase in the enzymatic activity was seen in the CeO₂NP-treated cells compared to that in H₂O₂-treated cells. The maximum activity was noticed in 20 µg/mL synthesized CeO₂NPs [SOD (63.92%), CAT (72.36%), and GPx (34.68%)] as depicted in Figure 5. The upregulation of the enzyme activities is attributable to the transition capacity of phytosynthesized NPs to switch between Ce³⁺ and Ce⁴⁺ ions, which mimic SOD, CAT, and peroxidase activity.⁴²

3.3.6. Markers of OS. **3.3.6.1. Determination of PC Content.** PCs, an important biomarker of protein oxidation, significantly increased by 11.92% in H₂O₂-treated cells. The treatment of CeO₂NPs markedly attenuated the PC content, with a maximum decrease seen in 20 µg/mL CeO₂NP-treated cells (4.82%) compared to that in H₂O₂-treated cells (Figure 6). The reduction in protein carbonyl levels can be accredited to the excellent antioxidant properties of CeO₂NPs, which are in line with the studies of Soren et al.⁴³ and Arya et al.⁴⁴

3.3.6.2. Intracellular ROS Measurement. To determine the protective activity of CeO₂NPs on H₂O₂-induced ROS generation, we utilized a cell-permeable DCFH-DA fluorescent probe. As shown in Figure 7, ROS generation significantly increased in H₂O₂-treated cells (58.27%) compared to that in control. The elevated ROS levels induced by H₂O₂ were attenuated considerably in CeO₂NP-treated SH-SY5Y cells, suggesting the efficient scavenging ability of CeO₂NPs in preventing OS.

3.3.7. Apoptotic Analysis. Apoptosis is a means of regulating cell death, which is crucial for cellular homeostasis. It is marked by nuclear changes leading to the formation of apoptotic bodies and cell death.⁴⁵ The protective effect of CeO₂NPs on H₂O₂-induced apoptosis was determined by

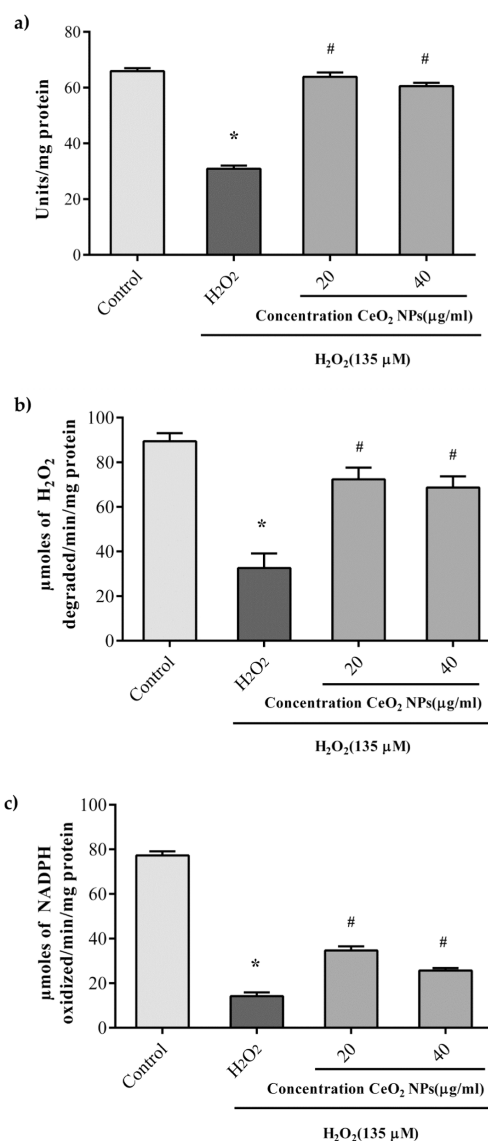


Figure 5. Effect of CeO₂NPs on antioxidant enzyme activities in H₂O₂-treated cells: (a) SOD, (b) CAT, and (c) GPx. The data is analyzed from three independent experiments by one-way ANOVA followed by Tukey's test. **p* < 0.05 significance compared with control; #*p* < 0.05 significance compared with H₂O₂-treated cells.

DAPI staining. As shown in Figure 8, the H₂O₂-treated cells showed distinct apoptotic characteristics such as chromatin

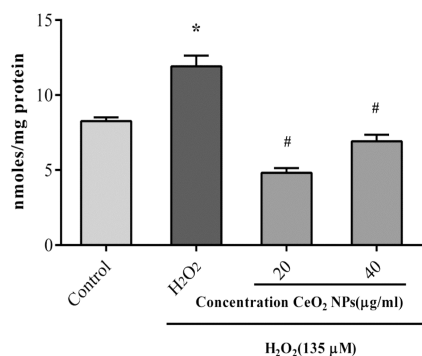


Figure 6. Effect of CeO₂NPs on PC levels in H₂O₂-treated SH-SY5Y cells. The data is analyzed from three independent experiments by one-way ANOVA followed by Tukey's test. **p* < 0.05 significance compared with control; #*p* < 0.05 significance compared with H₂O₂-treated cells.

condensation, nuclear fragmentation, and cell shrinkage. The treatment of cells with CeO₂NPs drastically decreased the number of apoptotic cells. Annexin-V-FITC/PI costaining was

further carried out to estimate the apoptosis through flow cytometric analysis quantitatively. The apoptotic rate in the H₂O₂-treated cells increased from 5.74% to 64% compared with that of the control. Interestingly, treatment of 20 and 40 µg/mL of CeO₂NPs remarkably attenuated SH-SY5Y cell apoptosis by 31% and 44%, respectively (Figure 8).

4. CONCLUSIONS

The green synthesis of CeO₂NPs using *C. asiatica* leaf extract is a simple, rapid, and eco-friendly method. The UV-vis, XRD, FTIR, SEM-EDAX, and HR-TEM analyses confirm the nanoregime purity, crystallite structures, size, morphology, and optical properties. The synthesized NPs exhibited excellent free radical scavenging activity against DPPH and OH radicals. Our results revealed that CeO₂NPs might protect the SH-SY5Y cells against OS induced by H₂O₂ by modulating antioxidant status and apoptosis due to effective redox switching in nanoscale. These findings suggest that CeO₂NPs may be a potential therapeutic agent for treating OS-related neurodegenerative disorders. However, further studies must be performed to ascertain its role in nanomedicine.

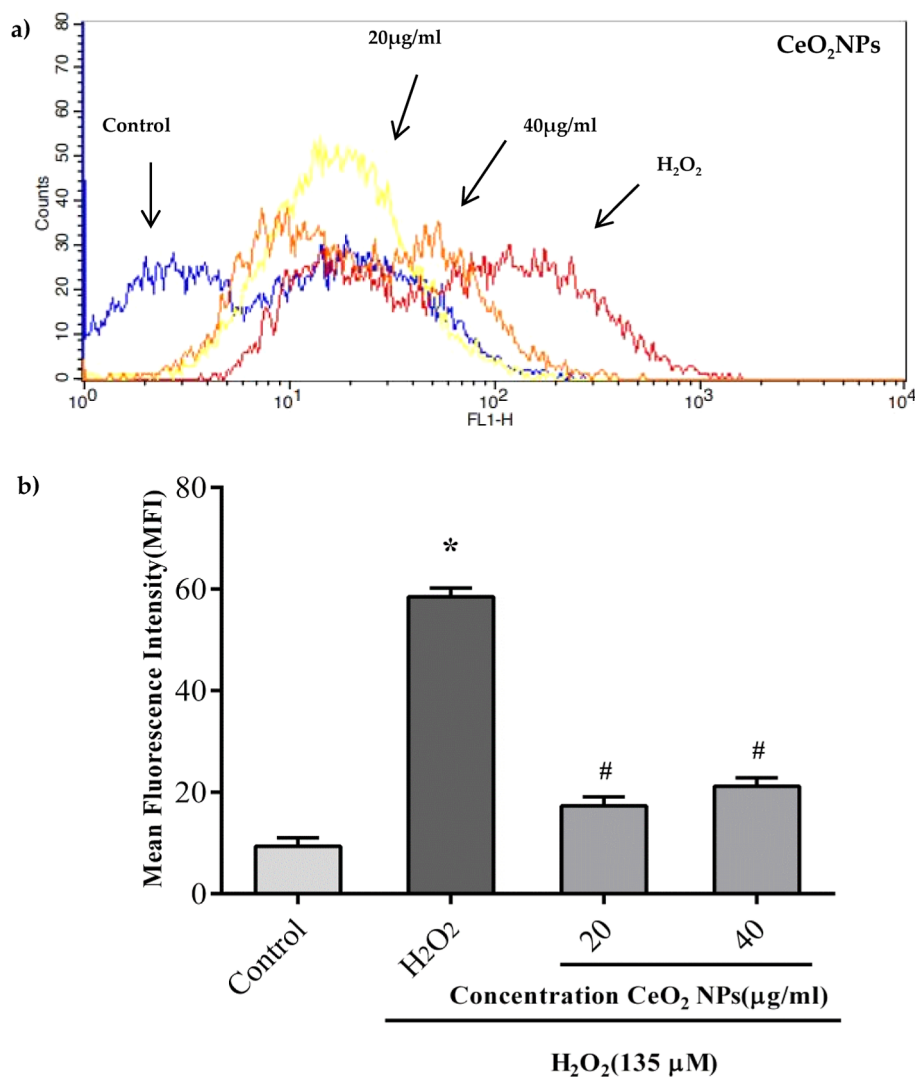


Figure 7. Effect of CeO₂NPs on ROS generation in H₂O₂-treated SH-SY5Y cells. (a) ROS determination by flow cytometry. (b) Quantitative analysis of the ROS fluorescence intensity. The data is analyzed from three independent experiments by one-way ANOVA followed by Tukey's test. **p* < 0.05 significance compared with control; #*p* < 0.05 significance compared with H₂O₂-treated cells.

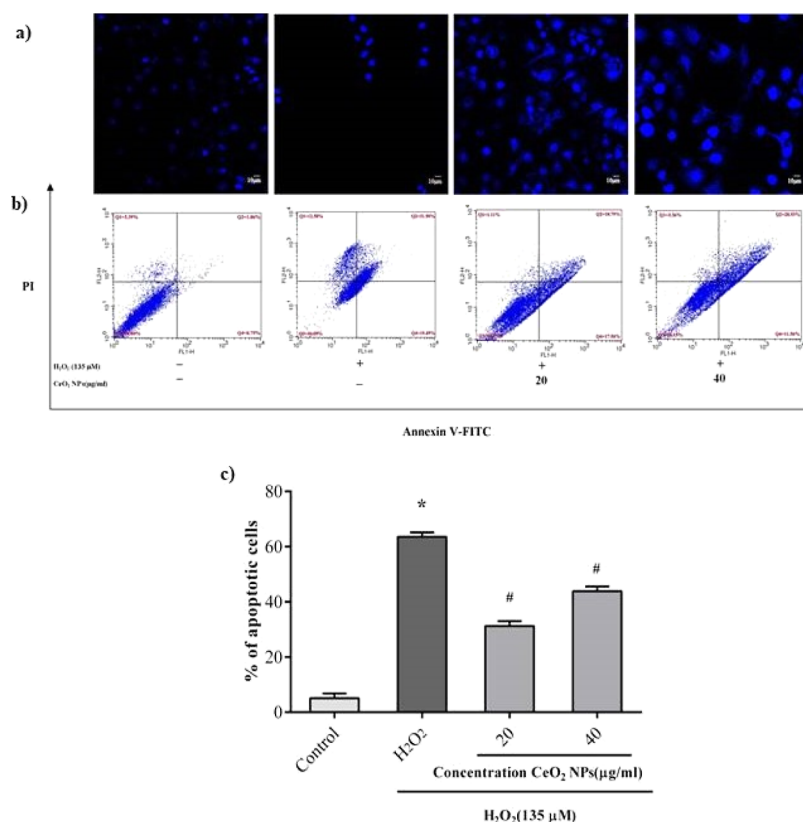


Figure 8. Effect of CeO₂NPs on H₂O₂-induced apoptosis in SH-SY5Y cells. (a) Morphology of cell nucleus using DAPI staining. (b) Flow cytometry. (c) Percentage of apoptotic cells. Data were expressed as mean ± SE (*n* = 3). Statistical analysis was performed by *t*-test. **p* < 0.05 significance compared with control; #*p* < 0.05 compared with H₂O₂-treated cells.

■ ASSOCIATED CONTENT

Data Availability Statement

Data is included within the article and in the [Supporting Information](#).

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.3c07505>.

Various structural parameters of CeO₂ NPs and Rietveld refined data for the CeO₂NPs obtained using *C. asiatica* extract (PDF)

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Notes

The authors declare no competing financial interest.

Ethical Approval: This study does not contain any work related to human participants or animals performed by any of the authors.

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