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Ethyl 27-oxo-15-oxa-2,20-diazahexacyclo[18.6.1.0^{1,8}.0^{2,6}.0^{9,14}.0^{21,26}]heptacos-9,11,13,21,23,25-hexaene-7-carboxylate

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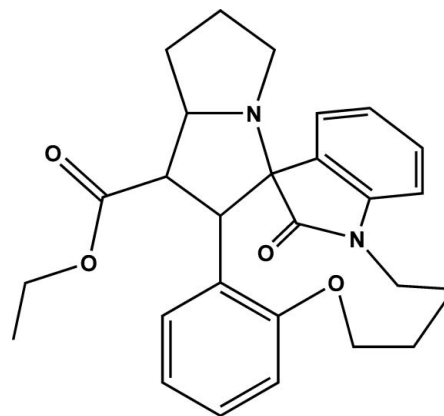
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; disorder in main residue; R factor = 0.045; wR factor = 0.125; data-to-parameter ratio = 17.6.

In the title compound, $\text{C}_{27}\text{H}_{30}\text{N}_2\text{O}_4$, the pyrrolidine ring adopts a twisted conformation. The indoline ring system is almost perpendicular to the mean plane of the pyrrolidine ring, making a dihedral angle of $81.7(8)^\circ$. In the crystal, molecules are linked into centrosymmetric dimers with graph-set motif $R_2^2(16)$ via pairs of $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds. The terminal ethyl group of the ester group is disordered over two sets of sites, with a site-occupancy ratio of 0.587 (11): 0.413 (11).

Related literature

For the biological activity of spiro-pyrrolidine derivatives, see: Obniska *et al.* (2003); Peddi *et al.* (2004); Christoph *et al.* (2011); Stylianakis *et al.* (2003); Waldmann (1995); Suzuki *et al.* (1994); Huryn *et al.* (1991). For a related structure, see: Ganesh *et al.* (2012). For puckering parameters, see: Cremer & Pople (1975) and for asymmetry parameters, see: Nardelli *et al.* (1983).



Experimental

Crystal data

$\text{C}_{27}\text{H}_{30}\text{N}_2\text{O}_4$
 $M_r = 446.53$
 Triclinic, $P\bar{1}$
 $a = 8.9327(5)$ Å
 $b = 10.0068(5)$ Å
 $c = 14.6379(11)$ Å
 $\alpha = 103.988(4)^\circ$
 $\beta = 95.023(4)^\circ$
 $\gamma = 113.775(3)^\circ$
 $V = 1136.41(13)$ Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 293$ K
 $0.30 \times 0.30 \times 0.25$ mm

Data collection

Bruker APEXII CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2008)
 $T_{\min} = 0.974$, $T_{\max} = 0.978$
 20472 measured reflections
 5603 independent reflections
 4378 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.125$
 $S = 1.08$
 5603 reflections
 319 parameters
 40 restraints
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.23$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.22$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C11}-\text{H11}\cdots\text{O1}^i$	0.93	2.49	3.3957 (19)	164
$\text{C12}-\text{H12}\cdots\text{O2}^{ii}$	0.93	2.59	3.446 (2)	153
$\text{C13}-\text{H13}\cdots\text{O4}^{ii}$	0.93	2.47	3.3986 (17)	175

 Symmetry codes: (i) $-x + 1, -y + 2, -z + 1$; (ii) $x + 1, y, z$.

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 2012); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT6853).

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supplementary materials

Acta Cryst. (2013). E69, o23–o24 [doi:10.1107/S1600536812049082]

**Ethyl 27-oxo-15-oxa-2,20-diazahexacyclo-
[18.6.1.0^{1,8}.0^{2,6}.0^{9,14}.0^{21,26}]heptacos-9,11,13,21,23,25-hexaene-7-carboxylate**

**Sibi Narayanan, Thothadri Srinivasan, Santhanagopalan Purushothaman, Raghavachary
Raghunathan and Devadasan Velmurugan**

Comment

Spiro-pyrrolidine derivatives are unique tetracyclic 5-HT(2 A) receptor antagonists (Obniska *et al.*, 2003; Peddi *et al.*, 2004). These derivatives possess anticancer (Christoph *et al.*, 2011) and anti-influenza virus (Stylianakis *et al.*, 2003) activities. Highly functionalized pyrrolidines have gained much interest in the past few years as they constitute the main structural element of many natural and synthetic pharmacologically active compounds (Waldmann, 1995). Optically active pyrrolidines have been used as intermediates, chiral ligands or auxiliaries in controlled asymmetric synthesis (Suzuki *et al.*, 1994; Huryn *et al.*, 1991). In view of these importance and in continuation of our work on the crystal structure analysis of spiro-pyrrolidine derivatives, the crystal structure of the title compound has been carried out and the results are presented here.

X-Ray analysis confirms the molecular structure and atom connectivity as illustrated in Fig. 1. The bond lengths and angles are within normal ranges and comparable to those found in related structures (Ganesh *et al.*, 2012). Terminal atoms C1 & C2 is substituted at propanate group, which is disordered over two positions [C1A/C1B & C2A/C2B] with a site-occupancy ratio of 0.604 (5):0.396 (5). The sum of the angles at N1 & N2 [339 (1)° & 348.7 (1)°] of the pyrrolidine and indole rings are in accordance with sp^3 hybridization.

The indoline ring system is essentially planar, with maximum deviation of 0.038 (2) Å for atom C16. The pyrrolidine ring system makes dihedral angle of 81.7 (8)° with the indoline ring system, it clearly shows that both the rings are perpendicular to each other. The propanate group assumes an extended conformation which can be seen from the torsion angle [C4/C3/O2/C2= 179.3 (2)°].

The dihedral angle of the pyrrolidine ring and the benzene ring (C21—C26) is 36.5 (1)°. The atom O4 deviates by 0.102 (1) Å from the leastsquares plane of the indole ring. The pyrrolidine ring adopts *twisted* conformation [it is twisted about C1—C2], with the puckering parameters q_2 and φ (Cremer & Pople, 1975) and the smallest displacement asymmetric parameter, Δ_s , (Nardelli *et al.*, 1983) as follows: $q_2=0.4071$ (2) Å, $\varphi=82.0$ (8)°, $\Delta_s(C4)=7.61$ (2). The crystal packing is stabilized by C—H...O interactions.

Experimental

A solution of (*E*)-ethyl 3-(2-(4-(2,3-dioxindolin-1-yl)butoxy)phenyl)acrylate (200 mg, 0.5 mmol) and *L*-proline (70 mg, 0.61 mmol) was refluxed in dry toluene under N₂ atmosphere for 8hrs under Dean-Stark apparatus. After the completion of reaction as indicated by TLC, toluene was evaporated under reduced pressure. The crude product was washed with water and extracted with dichloromethane (4x20mL). The combined organic layers were dried (MgSO₄) and filtered, concentrated in vacuum. The crude product was purified by column chromatography using hexane: EtOAc (8:2) as

eluent.

Refinement

Atoms C1 and C2 are disordered over two positions (C1A/C1B & C2A/C2B) with refined occupancies of 0.587 (11):0.413 (11). The O-C and C-C distances of the disordered atoms were restrained to be equal. The displacement parameters of the disordered atoms were restrained to be equal for bonded atoms. All H atoms were fixed geometrically and allowed to ride on their parent C atoms, with C—H distances fixed in the range 0.93–0.97 Å with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H, $1.2U_{\text{eq}}(\text{C})$ for other H atoms.

Computing details

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINTE* (Bruker, 2008); data reduction: *SAINTE* (Bruker, 2008); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008) and *PLATON* (Spek, 2009).

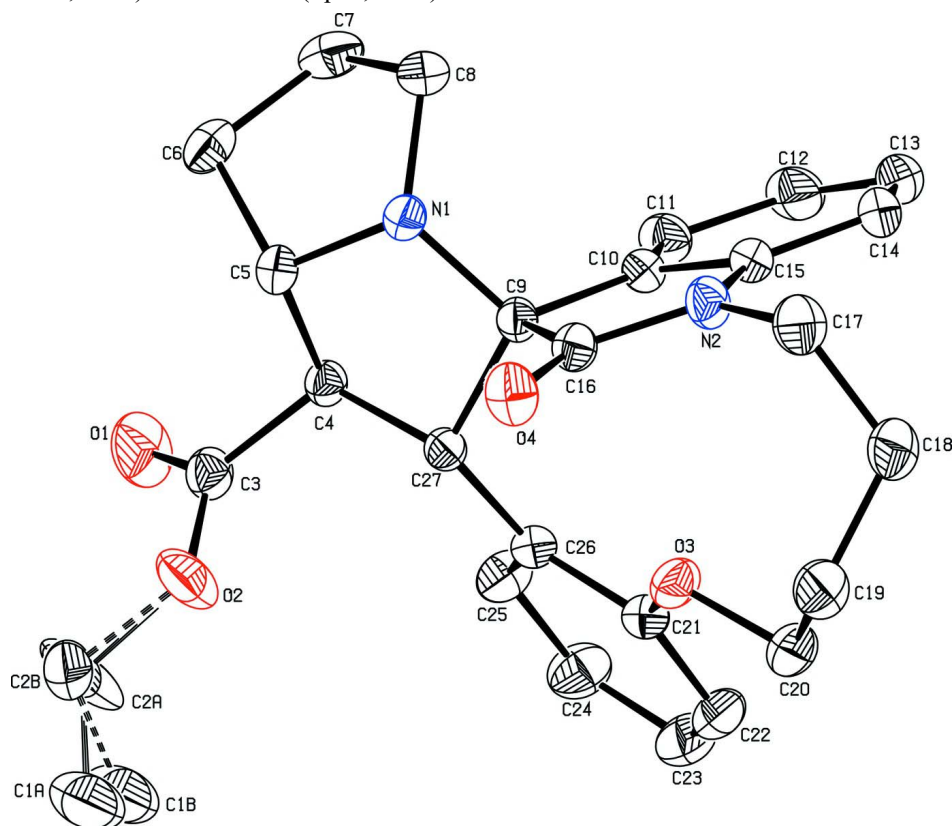


Figure 1

The molecular structure of the title compound, showing displacement ellipsoids drawn at the 30% probability level. H atoms are omitted for figure clarity.

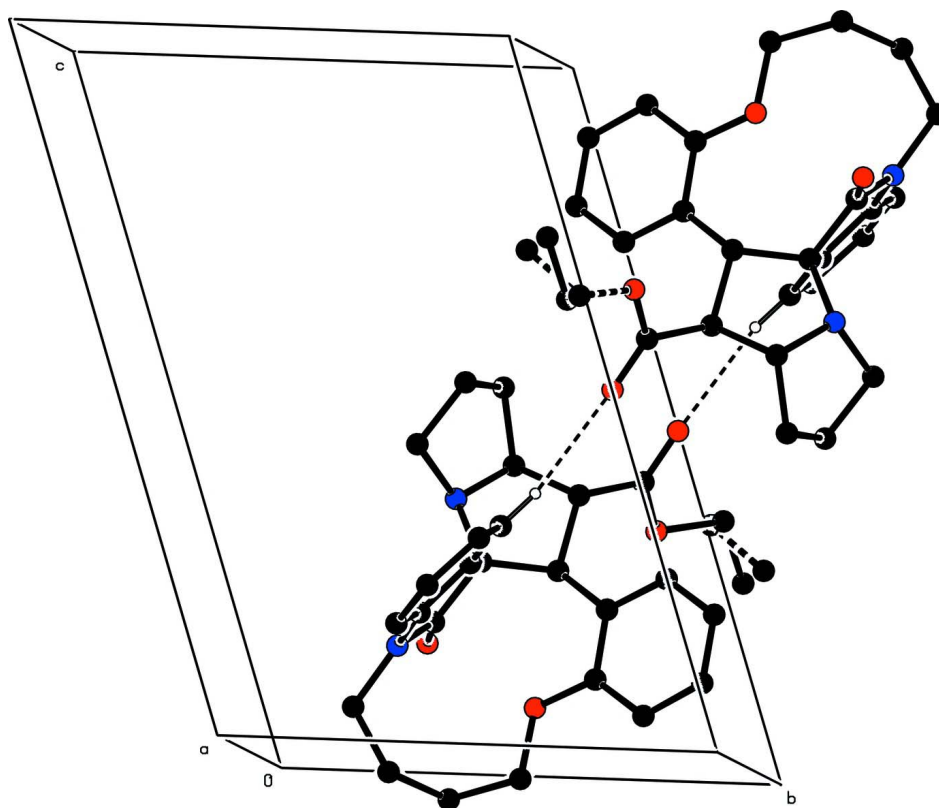


Figure 2

The crystal structure showing the formation of the centrosymmetric $R_2^2(16)$ dimer. H atoms not involved in hydrogen bonds have been omitted.

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Crystal data

$C_{27}H_{30}N_2O_4$
 $M_r = 446.53$
 Triclinic, $P\bar{1}$
 Hall symbol: -P 1
 $a = 8.9327$ (5) Å
 $b = 10.0068$ (5) Å
 $c = 14.6379$ (11) Å
 $\alpha = 103.988$ (4)°
 $\beta = 95.023$ (4)°
 $\gamma = 113.775$ (3)°
 $V = 1136.41$ (13) Å³

$Z = 2$
 $F(000) = 476$
 $D_x = 1.305$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 5603 reflections
 $\theta = 1.5$ – 28.3 °
 $\mu = 0.09$ mm⁻¹
 $T = 293$ K
 Block, white crystalline
 $0.30 \times 0.30 \times 0.25$ mm

Data collection

Bruker APEXII CCD area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω and ϕ scans

Absorption correction: multi-scan
 (*SADABS*; Bruker, 2008)
 $T_{\min} = 0.974$, $T_{\max} = 0.978$
 20472 measured reflections
 5603 independent reflections
 4378 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.025$
 $\theta_{\text{max}} = 28.3^\circ$, $\theta_{\text{min}} = 1.5^\circ$
 $h = -11 \rightarrow 11$

$k = -13 \rightarrow 13$
 $l = -18 \rightarrow 19$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.125$
 $S = 1.08$
 5603 reflections
 319 parameters
 40 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0559P)^2 + 0.215P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.23 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.22 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1A	-0.0430 (10)	1.0338 (8)	0.2839 (4)	0.0877 (18)	0.587 (11)
H1A1	0.0125	1.0358	0.2306	0.132*	0.587 (11)
H1A2	-0.0730	1.1174	0.2975	0.132*	0.587 (11)
H1A3	-0.1420	0.9386	0.2680	0.132*	0.587 (11)
C2A	0.0677 (12)	1.0483 (8)	0.3675 (6)	0.089 (3)	0.587 (11)
H2A1	0.0188	1.0599	0.4238	0.106*	0.587 (11)
H2A2	0.1740	1.1376	0.3795	0.106*	0.587 (11)
C1B	0.0656 (17)	1.0966 (9)	0.2980 (6)	0.088 (3)	0.413 (11)
H1B1	0.1845	1.1562	0.3183	0.131*	0.413 (11)
H1B2	0.0130	1.1639	0.3003	0.131*	0.413 (11)
H1B3	0.0393	1.0304	0.2334	0.131*	0.413 (11)
C2B	0.0054 (13)	1.0057 (11)	0.3611 (9)	0.072 (3)	0.413 (11)
H2B1	-0.1149	0.9451	0.3426	0.086*	0.413 (11)
H2B2	0.0346	1.0696	0.4274	0.086*	0.413 (11)
C3	0.21727 (18)	0.92280 (15)	0.41102 (10)	0.0423 (3)	
C4	0.26073 (15)	0.78973 (13)	0.38728 (9)	0.0328 (3)	
H4	0.3792	0.8281	0.4159	0.039*	
C5	0.15967 (15)	0.66458 (14)	0.43037 (9)	0.0346 (3)	
H5	0.0410	0.6419	0.4175	0.042*	
C6	0.22279 (18)	0.69639 (17)	0.53715 (10)	0.0466 (3)	
H6A	0.2708	0.8051	0.5713	0.056*	
H6B	0.1334	0.6416	0.5664	0.056*	
C7	0.3552 (2)	0.63749 (19)	0.53718 (11)	0.0550 (4)	

H7A	0.4601	0.7115	0.5293	0.066*
H7B	0.3743	0.6130	0.5960	0.066*
C8	0.2771 (2)	0.49510 (18)	0.45101 (11)	0.0503 (4)
H8A	0.3628	0.4711	0.4251	0.060*
H8B	0.2022	0.4082	0.4687	0.060*
C9	0.24212 (14)	0.55059 (13)	0.28972 (9)	0.0314 (2)
C10	0.40719 (14)	0.54634 (14)	0.28034 (9)	0.0335 (3)
C11	0.56843 (16)	0.64768 (16)	0.32716 (11)	0.0433 (3)
H11	0.5907	0.7389	0.3736	0.052*
C12	0.69804 (17)	0.6109 (2)	0.30370 (12)	0.0531 (4)
H12	0.8079	0.6793	0.3342	0.064*
C13	0.66598 (19)	0.4754 (2)	0.23631 (12)	0.0552 (4)
H13	0.7541	0.4521	0.2231	0.066*
C14	0.50416 (19)	0.37281 (19)	0.18772 (11)	0.0491 (4)
H14	0.4821	0.2810	0.1419	0.059*
C15	0.37664 (16)	0.41162 (15)	0.20968 (9)	0.0373 (3)
C16	0.11869 (15)	0.40657 (14)	0.20770 (9)	0.0362 (3)
C17	0.1341 (2)	0.19559 (16)	0.08382 (11)	0.0528 (4)
H17A	0.0135	0.1469	0.0769	0.063*
H17B	0.1753	0.1226	0.0927	0.063*
C18	0.1747 (2)	0.23314 (18)	-0.00835 (11)	0.0552 (4)
H18A	0.1372	0.1376	-0.0597	0.066*
H18B	0.2953	0.2869	0.0004	0.066*
C19	0.0979 (2)	0.33016 (19)	-0.04075 (11)	0.0568 (4)
H19A	0.0621	0.2915	-0.1103	0.068*
H19B	-0.0013	0.3154	-0.0142	0.068*
C20	0.2087 (2)	0.49895 (18)	-0.01325 (10)	0.0531 (4)
H20A	0.3192	0.5157	-0.0248	0.064*
H20B	0.1635	0.5445	-0.0528	0.064*
C21	0.33250 (17)	0.72236 (15)	0.12529 (10)	0.0396 (3)
C22	0.4290 (2)	0.80905 (19)	0.07220 (11)	0.0536 (4)
H22	0.4166	0.7651	0.0066	0.064*
C23	0.5431 (2)	0.95975 (19)	0.11637 (12)	0.0592 (4)
H23	0.6082	1.0165	0.0806	0.071*
C24	0.5606 (2)	1.02605 (18)	0.21268 (12)	0.0565 (4)
H24	0.6379	1.1274	0.2427	0.068*
C25	0.46165 (18)	0.94020 (16)	0.26501 (11)	0.0473 (3)
H25	0.4736	0.9863	0.3302	0.057*
C26	0.34589 (15)	0.78868 (14)	0.22383 (9)	0.0350 (3)
C27	0.23201 (14)	0.69994 (13)	0.28114 (8)	0.0314 (2)
H27	0.1165	0.6710	0.2510	0.038*
N1	0.18398 (13)	0.53096 (12)	0.37981 (7)	0.0361 (2)
N2	0.20527 (14)	0.32851 (12)	0.16915 (8)	0.0398 (3)
O1	0.28603 (18)	1.03033 (14)	0.48165 (9)	0.0743 (4)
O2	0.09278 (16)	0.90670 (14)	0.34755 (9)	0.0684 (3)
O3	0.22133 (12)	0.57163 (11)	0.08703 (6)	0.0447 (2)
O4	-0.03046 (11)	0.36456 (11)	0.18380 (7)	0.0494 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1A	0.106 (4)	0.079 (3)	0.104 (3)	0.063 (3)	0.008 (3)	0.038 (3)
C2A	0.119 (6)	0.077 (4)	0.087 (4)	0.078 (4)	-0.009 (4)	0.002 (3)
C1B	0.131 (7)	0.069 (4)	0.074 (4)	0.046 (4)	0.027 (4)	0.032 (3)
C2B	0.092 (5)	0.072 (4)	0.087 (6)	0.058 (4)	0.041 (5)	0.038 (4)
C3	0.0498 (7)	0.0370 (7)	0.0411 (8)	0.0205 (6)	0.0135 (6)	0.0094 (6)
C4	0.0343 (6)	0.0329 (6)	0.0297 (6)	0.0152 (5)	0.0065 (5)	0.0064 (5)
C5	0.0346 (6)	0.0354 (6)	0.0328 (6)	0.0155 (5)	0.0097 (5)	0.0072 (5)
C6	0.0565 (8)	0.0451 (8)	0.0320 (7)	0.0163 (6)	0.0131 (6)	0.0103 (6)
C7	0.0570 (9)	0.0661 (10)	0.0420 (8)	0.0227 (8)	0.0038 (7)	0.0265 (7)
C8	0.0655 (9)	0.0601 (9)	0.0476 (9)	0.0391 (8)	0.0218 (7)	0.0304 (7)
C9	0.0315 (5)	0.0331 (6)	0.0300 (6)	0.0157 (5)	0.0067 (5)	0.0075 (5)
C10	0.0341 (6)	0.0377 (6)	0.0340 (6)	0.0193 (5)	0.0086 (5)	0.0128 (5)
C11	0.0360 (6)	0.0470 (8)	0.0474 (8)	0.0183 (6)	0.0062 (6)	0.0162 (6)
C12	0.0349 (7)	0.0702 (10)	0.0638 (10)	0.0254 (7)	0.0118 (7)	0.0319 (9)
C13	0.0528 (8)	0.0895 (12)	0.0568 (10)	0.0507 (9)	0.0274 (7)	0.0396 (9)
C14	0.0629 (9)	0.0646 (9)	0.0431 (8)	0.0460 (8)	0.0218 (7)	0.0208 (7)
C15	0.0433 (7)	0.0437 (7)	0.0347 (7)	0.0255 (6)	0.0126 (5)	0.0156 (5)
C16	0.0379 (6)	0.0353 (6)	0.0342 (7)	0.0161 (5)	0.0082 (5)	0.0081 (5)
C17	0.0673 (9)	0.0375 (7)	0.0460 (9)	0.0239 (7)	0.0076 (7)	-0.0002 (6)
C18	0.0703 (10)	0.0500 (9)	0.0403 (8)	0.0310 (8)	0.0085 (7)	-0.0018 (7)
C19	0.0667 (10)	0.0619 (10)	0.0342 (8)	0.0330 (8)	-0.0024 (7)	-0.0020 (7)
C20	0.0743 (10)	0.0599 (9)	0.0305 (7)	0.0375 (8)	0.0112 (7)	0.0090 (6)
C21	0.0473 (7)	0.0439 (7)	0.0353 (7)	0.0260 (6)	0.0094 (6)	0.0148 (6)
C22	0.0738 (10)	0.0600 (9)	0.0394 (8)	0.0345 (8)	0.0225 (7)	0.0230 (7)
C23	0.0705 (10)	0.0612 (10)	0.0576 (10)	0.0276 (8)	0.0278 (8)	0.0351 (8)
C24	0.0612 (9)	0.0471 (8)	0.0554 (10)	0.0138 (7)	0.0143 (8)	0.0226 (7)
C25	0.0530 (8)	0.0448 (8)	0.0388 (8)	0.0155 (6)	0.0103 (6)	0.0140 (6)
C26	0.0392 (6)	0.0386 (6)	0.0322 (6)	0.0201 (5)	0.0081 (5)	0.0134 (5)
C27	0.0331 (6)	0.0327 (6)	0.0287 (6)	0.0164 (5)	0.0046 (5)	0.0072 (5)
N1	0.0436 (6)	0.0373 (6)	0.0323 (6)	0.0210 (5)	0.0125 (4)	0.0115 (4)
N2	0.0451 (6)	0.0357 (6)	0.0365 (6)	0.0207 (5)	0.0071 (5)	0.0024 (4)
O1	0.0918 (9)	0.0525 (7)	0.0627 (8)	0.0368 (7)	0.0019 (7)	-0.0139 (6)
O2	0.0874 (8)	0.0622 (7)	0.0668 (8)	0.0558 (7)	-0.0023 (6)	0.0045 (6)
O3	0.0589 (6)	0.0458 (5)	0.0286 (5)	0.0243 (5)	0.0090 (4)	0.0078 (4)
O4	0.0336 (5)	0.0502 (6)	0.0515 (6)	0.0142 (4)	0.0025 (4)	0.0029 (5)

Geometric parameters (\AA , $^\circ$)

C1A—C2A	1.443 (6)	C11—C12	1.3947 (19)
C1A—H1A1	0.9600	C11—H11	0.9300
C1A—H1A2	0.9600	C12—C13	1.373 (2)
C1A—H1A3	0.9600	C12—H12	0.9300
C2A—O2	1.487 (5)	C13—C14	1.385 (2)
C2A—H2A1	0.9700	C13—H13	0.9300
C2A—H2A2	0.9700	C14—C15	1.3827 (18)
C1B—C2B	1.438 (7)	C14—H14	0.9300
C1B—H1B1	0.9600	C15—N2	1.4011 (17)

C1B—H1B2	0.9600	C16—O4	1.2123 (15)
C1B—H1B3	0.9600	C16—N2	1.3701 (16)
C2B—O2	1.478 (6)	C17—N2	1.4524 (18)
C2B—H2B1	0.9700	C17—C18	1.521 (2)
C2B—H2B2	0.9700	C17—H17A	0.9700
C3—O1	1.1925 (18)	C17—H17B	0.9700
C3—O2	1.3145 (18)	C18—C19	1.530 (2)
C3—C4	1.5038 (17)	C18—H18A	0.9700
C4—C5	1.5333 (16)	C18—H18B	0.9700
C4—C27	1.5351 (17)	C19—C20	1.500 (2)
C4—H4	0.9800	C19—H19A	0.9700
C5—N1	1.4720 (16)	C19—H19B	0.9700
C5—C6	1.5241 (19)	C20—O3	1.4440 (16)
C5—H5	0.9800	C20—H20A	0.9700
C6—C7	1.520 (2)	C20—H20B	0.9700
C6—H6A	0.9700	C21—O3	1.3643 (16)
C6—H6B	0.9700	C21—C22	1.3930 (19)
C7—C8	1.511 (2)	C21—C26	1.4064 (18)
C7—H7A	0.9700	C22—C23	1.380 (2)
C7—H7B	0.9700	C22—H22	0.9300
C8—N1	1.4756 (17)	C23—C24	1.370 (2)
C8—H8A	0.9700	C23—H23	0.9300
C8—H8B	0.9700	C24—C25	1.3896 (19)
C9—N1	1.4867 (15)	C24—H24	0.9300
C9—C10	1.5093 (15)	C25—C26	1.3843 (19)
C9—C16	1.5442 (17)	C25—H25	0.9300
C9—C27	1.5669 (16)	C26—C27	1.5194 (16)
C10—C11	1.3734 (18)	C27—H27	0.9800
C10—C15	1.3934 (18)		
C1A—C2A—O2	107.4 (5)	C14—C13—H13	119.5
C1A—C2A—H2A1	110.2	C15—C14—C13	117.71 (14)
O2—C2A—H2A1	110.2	C15—C14—H14	121.1
C1A—C2A—H2A2	110.2	C13—C14—H14	121.1
O2—C2A—H2A2	110.2	C14—C15—C10	121.80 (13)
H2A1—C2A—H2A2	108.5	C14—C15—N2	127.97 (13)
C2B—C1B—H1B1	109.5	C10—C15—N2	110.23 (10)
C2B—C1B—H1B2	109.5	O4—C16—N2	125.24 (12)
H1B1—C1B—H1B2	109.5	O4—C16—C9	126.60 (11)
C2B—C1B—H1B3	109.5	N2—C16—C9	108.07 (10)
H1B1—C1B—H1B3	109.5	N2—C17—C18	113.75 (12)
H1B2—C1B—H1B3	109.5	N2—C17—H17A	108.8
C1B—C2B—O2	103.0 (6)	C18—C17—H17A	108.8
C1B—C2B—H2B1	111.2	N2—C17—H17B	108.8
O2—C2B—H2B1	111.2	C18—C17—H17B	108.8
C1B—C2B—H2B2	111.2	H17A—C17—H17B	107.7
O2—C2B—H2B2	111.2	C17—C18—C19	115.73 (13)
H2B1—C2B—H2B2	109.1	C17—C18—H18A	108.3
O1—C3—O2	123.36 (13)	C19—C18—H18A	108.3

O1—C3—C4	123.34 (14)	C17—C18—H18B	108.3
O2—C3—C4	113.27 (12)	C19—C18—H18B	108.3
C3—C4—C5	111.85 (10)	H18A—C18—H18B	107.4
C3—C4—C27	118.71 (10)	C20—C19—C18	116.20 (14)
C5—C4—C27	102.81 (9)	C20—C19—H19A	108.2
C3—C4—H4	107.6	C18—C19—H19A	108.2
C5—C4—H4	107.6	C20—C19—H19B	108.2
C27—C4—H4	107.6	C18—C19—H19B	108.2
N1—C5—C6	105.91 (10)	H19A—C19—H19B	107.4
N1—C5—C4	103.94 (9)	O3—C20—C19	110.62 (13)
C6—C5—C4	115.22 (11)	O3—C20—H20A	109.5
N1—C5—H5	110.5	C19—C20—H20A	109.5
C6—C5—H5	110.5	O3—C20—H20B	109.5
C4—C5—H5	110.5	C19—C20—H20B	109.5
C7—C6—C5	103.05 (11)	H20A—C20—H20B	108.1
C7—C6—H6A	111.2	O3—C21—C22	123.50 (13)
C5—C6—H6A	111.2	O3—C21—C26	116.04 (11)
C7—C6—H6B	111.2	C22—C21—C26	120.46 (13)
C5—C6—H6B	111.2	C23—C22—C21	120.37 (14)
H6A—C6—H6B	109.1	C23—C22—H22	119.8
C8—C7—C6	102.04 (12)	C21—C22—H22	119.8
C8—C7—H7A	111.4	C24—C23—C22	120.23 (14)
C6—C7—H7A	111.4	C24—C23—H23	119.9
C8—C7—H7B	111.4	C22—C23—H23	119.9
C6—C7—H7B	111.4	C23—C24—C25	119.26 (15)
H7A—C7—H7B	109.2	C23—C24—H24	120.4
N1—C8—C7	105.66 (11)	C25—C24—H24	120.4
N1—C8—H8A	110.6	C26—C25—C24	122.54 (14)
C7—C8—H8A	110.6	C26—C25—H25	118.7
N1—C8—H8B	110.6	C24—C25—H25	118.7
C7—C8—H8B	110.6	C25—C26—C21	117.12 (12)
H8A—C8—H8B	108.7	C25—C26—C27	121.56 (11)
N1—C9—C10	116.46 (9)	C21—C26—C27	121.25 (11)
N1—C9—C16	106.22 (9)	C26—C27—C4	114.73 (10)
C10—C9—C16	101.89 (9)	C26—C27—C9	117.37 (9)
N1—C9—C27	104.52 (9)	C4—C27—C9	101.60 (9)
C10—C9—C27	115.18 (9)	C26—C27—H27	107.5
C16—C9—C27	112.44 (9)	C4—C27—H27	107.5
C11—C10—C15	119.80 (11)	C9—C27—H27	107.5
C11—C10—C9	131.67 (12)	C5—N1—C8	108.13 (10)
C15—C10—C9	108.54 (10)	C5—N1—C9	110.23 (9)
C10—C11—C12	118.65 (14)	C8—N1—C9	120.66 (10)
C10—C11—H11	120.7	C16—N2—C15	110.85 (10)
C12—C11—H11	120.7	C16—N2—C17	123.63 (11)
C13—C12—C11	121.05 (14)	C15—N2—C17	124.23 (11)
C13—C12—H12	119.5	C3—O2—C2B	124.6 (5)
C11—C12—H12	119.5	C3—O2—C2A	111.3 (3)
C12—C13—C14	120.93 (13)	C21—O3—C20	117.33 (11)
C12—C13—H13	119.5		

O1—C3—C4—C5	89.09 (17)	C22—C21—C26—C27	175.06 (12)
O2—C3—C4—C5	-89.09 (14)	C25—C26—C27—C4	-2.05 (16)
O1—C3—C4—C27	-151.44 (14)	C21—C26—C27—C4	-178.96 (10)
O2—C3—C4—C27	30.38 (16)	C25—C26—C27—C9	-121.25 (13)
C3—C4—C5—N1	165.51 (10)	C21—C26—C27—C9	61.84 (15)
C27—C4—C5—N1	37.05 (11)	C3—C4—C27—C26	67.92 (14)
C3—C4—C5—C6	-79.07 (14)	C5—C4—C27—C26	-168.04 (9)
C27—C4—C5—C6	152.47 (10)	C3—C4—C27—C9	-164.39 (10)
N1—C5—C6—C7	28.37 (13)	C5—C4—C27—C9	-40.35 (11)
C4—C5—C6—C7	-85.91 (13)	N1—C9—C27—C26	155.06 (10)
C5—C6—C7—C8	-38.37 (13)	C10—C9—C27—C26	25.97 (15)
C6—C7—C8—N1	34.89 (14)	C16—C9—C27—C26	-90.16 (12)
N1—C9—C10—C11	-70.23 (17)	N1—C9—C27—C4	29.09 (11)
C16—C9—C10—C11	174.70 (13)	C10—C9—C27—C4	-99.99 (11)
C27—C9—C10—C11	52.69 (18)	C16—C9—C27—C4	143.88 (10)
N1—C9—C10—C15	110.05 (12)	C6—C5—N1—C8	-6.82 (13)
C16—C9—C10—C15	-5.02 (12)	C4—C5—N1—C8	115.00 (11)
C27—C9—C10—C15	-127.02 (11)	C6—C5—N1—C9	-140.65 (10)
C15—C10—C11—C12	-1.28 (19)	C4—C5—N1—C9	-18.83 (12)
C9—C10—C11—C12	179.03 (12)	C7—C8—N1—C5	-17.71 (14)
C10—C11—C12—C13	-1.0 (2)	C7—C8—N1—C9	110.39 (13)
C11—C12—C13—C14	1.7 (2)	C10—C9—N1—C5	121.72 (11)
C12—C13—C14—C15	-0.2 (2)	C16—C9—N1—C5	-125.67 (10)
C13—C14—C15—C10	-2.1 (2)	C27—C9—N1—C5	-6.59 (12)
C13—C14—C15—N2	178.71 (13)	C10—C9—N1—C8	-5.43 (16)
C11—C10—C15—C14	2.89 (19)	C16—C9—N1—C8	107.18 (12)
C9—C10—C15—C14	-177.35 (11)	C27—C9—N1—C8	-133.74 (11)
C11—C10—C15—N2	-177.82 (11)	O4—C16—N2—C15	177.33 (12)
C9—C10—C15—N2	1.94 (14)	C9—C16—N2—C15	-5.83 (14)
N1—C9—C16—O4	60.93 (16)	O4—C16—N2—C17	9.9 (2)
C10—C9—C16—O4	-176.69 (13)	C9—C16—N2—C17	-173.31 (12)
C27—C9—C16—O4	-52.82 (17)	C14—C15—N2—C16	-178.20 (13)
N1—C9—C16—N2	-115.86 (10)	C10—C15—N2—C16	2.56 (15)
C10—C9—C16—N2	6.52 (12)	C14—C15—N2—C17	-10.8 (2)
C27—C9—C16—N2	130.40 (10)	C10—C15—N2—C17	169.95 (12)
N2—C17—C18—C19	-66.50 (18)	C18—C17—N2—C16	96.75 (16)
C17—C18—C19—C20	97.32 (18)	C18—C17—N2—C15	-69.06 (18)
C18—C19—C20—O3	-76.86 (17)	O1—C3—O2—C2B	-9.2 (4)
O3—C21—C22—C23	-177.87 (14)	C4—C3—O2—C2B	169.0 (4)
C26—C21—C22—C23	2.2 (2)	O1—C3—O2—C2A	8.6 (5)
C21—C22—C23—C24	-0.9 (3)	C4—C3—O2—C2A	-173.2 (5)
C22—C23—C24—C25	-0.4 (3)	C1B—C2B—O2—C3	106.8 (10)
C23—C24—C25—C26	0.6 (2)	C1B—C2B—O2—C2A	52.4 (13)
C24—C25—C26—C21	0.6 (2)	C1A—C2A—O2—C3	167.2 (7)
C24—C25—C26—C27	-176.39 (13)	C1A—C2A—O2—C2B	-58.7 (15)
O3—C21—C26—C25	178.04 (12)	C22—C21—O3—C20	1.99 (19)
C22—C21—C26—C25	-1.99 (19)	C26—C21—O3—C20	-178.03 (11)
O3—C21—C26—C27	-4.92 (17)	C19—C20—O3—C21	173.01 (12)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C11—H11···O1 ⁱ	0.93	2.49	3.3957 (19)	164
C12—H12···O2 ⁱⁱ	0.93	2.59	3.446 (2)	153
C13—H13···O4 ⁱⁱ	0.93	2.47	3.3986 (17)	175

Symmetry codes: (i) $-x+1, -y+2, -z+1$; (ii) $x+1, y, z$.