



Isolated Heme A Synthase from Aquifex aeolicus Is a Trimer

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ABSTRACT The integral membrane protein heme A synthase (HAS) catalyzes the biosynthesis of heme A, which is a prerequisite for cellular respiration in a wide range of aerobic organisms. Previous studies have revealed that HAS can form homo-oligomeric complexes, and this oligomerization appears to be evolutionarily conserved among prokaryotes and eukaryotes and is shown to be essential for the biological function of eukaryotic HAS. Despite its importance, little is known about the detailed structural properties of HAS oligomers. Here, we aimed to address this critical issue by analyzing the oligomeric state of HAS from *Aquifex aeolicus* (AaHAS) using a combination of techniques, including size exclusion chromatography coupled with multiangle light scattering (SEC-MALS), cross-linking, laser-induced liquid bead ion desorption mass spectrometry (LILBID-MS), and single-particle electron cryomicroscopy (cryo-EM). Our results show that HAS forms a thermostable trimeric complex. A cryo-EM density map provides information on the oligomerization interface of the AaHAS trimer. These results provide structural insights into HAS multimerization and expand our knowledge of this important enzyme.

IMPORTANCE Heme A is a vital redox cofactor unique for the terminal cytochrome *c* oxidase in mitochondria and many microorganisms. It plays a key role in oxygen reduction by serving as an electron carrier and as the oxygen-binding site. Heme A is synthesized from heme O by an integral membrane protein, heme A synthase (HAS). Defects in HAS impair cellular respiration and have been linked to various human diseases, e.g., fatal infantile hypertrophic cardiomyopathy and Leigh syndrome. HAS exists as a stable oligomeric complex, and studies have shown that oligomerization of eukaryotic HAS is necessary for its proper function. However, the molecular architecture of the HAS oligomeric complex has remained uncharacterized. The present study shows that HAS forms trimers and reveals how the oligomeric arrangement contributes to the complex stability and flexibility, enabling HAS to perform its catalytic function effectively. This work provides the basic understanding for future studies on heme A biosynthesis.

KEYWORDS *Aquifex aeolicus*, cofactor biosynthesis, heme A synthase, hyperthermophilic bacterium, metalloproteins, protein oligomerization, respiratory chain, structural biology

eme A plays an essential role in cellular respiration as an obligatory cofactor in all mitochondrial and many prokaryotic cytochrome *c* oxidases (*CcO*) (1, 2). Heme A synthase (HAS) catalyzes the biosynthesis of heme A from heme O in three sequential steps: (i) a monohydroxylated heme O (heme I) is produced by oxygenation of the C-8 methyl group on pyrrole ring D of heme O; (ii) heme I is further oxidized to a dihydroxylated intermediate; and (iii) spontaneous dehydration of the geminal diol produces the C-8 aldehyde substituent of heme A (2–4). It was shown that the oxygen atom required for this reaction is derived from water (4), even Citation Zeng H, Zhu G, Zhang S, Li X, Martin J, Morgner N, Sun F, Peng G, Xie H, Michel H. 2020. Isolated heme A synthase from *Aquifex aeolicus* is a trimer. mBio 11:e02615-19. https:// doi.org/10.1128/mBio.02615-19.

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Received 5 November 2019 Accepted 29 May 2020 Published 30 June 2020 though the HAS activity is strongly dependent on molecular oxygen (3, 5). HAS contains a low-spin heme B; however, its role as a prosthetic group of HAS has not been fully elucidated.

HAS is located in the inner membrane of mitochondria and in the cytoplasmic membrane of prokaryotes. Based on the predicted topology and complement of cysteine residues present in extracellular loops, members of HAS can be classified into four classes (6). Class A is composed of archaeal HAS with only four transmembrane helices (TMHs), while members of classes B, C, and D contain eight TMHs. Classes B and C are found in many bacteria and archaea, which contain one or two pairs of cysteine residues in the extracellular loop region. Class D HAS lacks cysteine residues and represents eukaryotic homologs (Cox15).

HAS from Bacillus subtilis (BsHAS), which belongs to class B, has been relatively well characterized in previous studies (7-11). The functional relevance of several conserved cysteine and histidine residues has been investigated by site-directed mutagenesis (6, 7, 10). However, the complete catalytic mechanism cannot be elucidated due to the lack of an in vitro enzyme assay system (2). It has been shown that the purified HAS from Aeropyrum pernix (class A) and B. subtilis may form homo-oligomeric complexes (12). Oligomerization of the eukaryotic HAS has been reported for Cox15 from Saccharomyces cerevisiae (13, 14). Further studies showed that the Cox15 oligomeric complex interacts not only with several CcO assembly factors but also with CcO subunit I (Cox1), suggesting a critical role in CcO assembly and maturation (15, 16). Despite its importance, many questions regarding HAS oligomerization remain unanswered, including the exact oligomeric state and the complex architecture of HAS. Recently, the crystal structure of BsHAS was determined (17). This structure revealed that BsHAS is composed of eight TMHs arranged in two pseudosymmetrically related bundles. However, no information about the oligomeric state could be gained, because BsHAS was crystallized as a monomer.

Therefore, we undertook the present study to examine the oligomeric properties of HAS. The HAS from the hyperthermophilic bacterium *Aquifex aeolicus* (AaHAS), which belongs to class C, was isolated and studied by comprehensive biochemical, biophysical, and electron-microscopic analyses.

Purified AaHAS is a bo-type cytochrome. AaHAS was heterologously expressed in Escherichia coli and purified to high homogeneity using affinity chromatography. From size exclusion chromatography, the purified AaHAS eluted as a monodisperse peak which was monitored at both 280 and 415 nm, indicating the incorporation of heme cofactors (Fig. 1A). To characterize the heme cofactors, absorption spectra of asisolated, fully oxidized, and fully reduced AaHAS were recorded (see Fig. S1 in the supplemental material). Optical absorption spectra of as-isolated and fully reduced AaHAS were almost identical. In the reduced state, AaHAS has an absorption peak at 427 nm in the Soret region, a β -peak at 528 nm, and a broad α -absorption at 560 nm, indicating the perturbation of the low-spin heme B and probably of heme O. After oxidation with potassium ferricyanide, the Soret peak shifted to 415 nm and decreased in intensity. In addition, spectral properties of purified AaHAS are similar to those observed for BsHAS (9), with the exception that no absorption peak was observed at about 585 nm in the reduced-minus-oxidized difference spectrum, suggesting the absence of heme A in the purified AaHAS preparation. To determine whether heme O is incorporated, AaHAS was analyzed by high-performance liquid chromatography with UV detection coupled to mass spectrometry (Fig. S2). Both heme O and heme B, but not heme A, were identified in the purified AaHAS, which is consistent with our spectroscopic data. In addition, pyridine hemochrome assays were performed (Fig. S3), and a stoichiometry of 1.1 hemes (heme B+O) to 1 AaHAS monomer was found.

AaHAS forms a thermostable trimeric complex. The protein sample was subjected to N-terminal sequencing, and the results showed that the initiator methionine has been cleaved from the mature AaHAS. Therefore, the expected molecular mass of the monomeric apo-AaHAS is 34.96 kDa, considering the size of the affinity tag Strep-Tag



FIG 1 Biochemical and biophysical characterization of AaHAS. (A) Size exclusion profile of the purified AaHAS separated on a Superdex 200 10/300 GL column monitored by recording the absorbance at 280 nm (black) and 415 nm (red). (B) BN-PAGE gel of purified AaHAS. The molecular marker (M) used was the NativeMark unstained protein standard. (C) Thermal denaturation profile of AaHAS analyzed by nano-DSF. The corresponding transition temperatures are indicated. (D) SDS-PAGE analyses of AaHAS. Purified AaHAS was incubated with 10 mM dithio-threitol (DTT) at room temperature (RT), 70°C, and 90°C for 10 min prior to SDS-PAGE. The molecular marker (M) used was the PageRule prestained protein ladder. (E) SEC-MALS analysis of purified AaHAS. The readings of UV, refractive index (RI), and light scattering (RALS) detectors are shown in black, blue, and orange, respectively. The purple line indicates the contribution of the detergent DM, and the green line shows the calculated molecular marks

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II. To determine the oligomeric state of AaHAS, blue native polyacrylamide gel electrophoresis (BN-PAGE) analysis was conducted, and a single dominant band migrating at approximately 200 kDa was detected (Fig. 1B). This result is similar to that found in previous studies (13, 18), where COX15 from *S. cerevisiae* exists as an oligomeric complex as observed on BN-PAGE analysis. Since the migration behavior of membrane proteins in a BN-PAGE gel is strongly influenced by the bound detergents and Coomassie dye molecules, a conversion factor of 1.8 was used (19), yielding a mass of 110 kDa after correction, which is very close to what is expected for a trimer (105 kDa).

We analyzed the purified AaHAS by sodium dodecyl sulfate (SDS)-PAGE. Two distinct bands with apparent molecular masses of about 25 and 70 kDa were visible (Fig. 1D). To correlate the observed masses with different oligomeric states, a matrix-assisted laser desorption ionization time-of-flight mass-spectrometric analysis was conducted (data not shown). The actual sizes of the polypeptides present in the lower and upper bands were determined to be 34.95 kDa and 104.56 kDa, corresponding to the masses of the monomeric and trimeric forms of AaHAS, respectively. The purified AaHAS was also subjected to heat treatment prior to SDS-PAGE. The trimeric band remained largely unaffected after exposure to temperatures up to 70°C but disappeared completely when the synthase was heated to 90°C (Fig. 1D). It should be noted that the optimal growth temperature for A. aeolicus is 85°C. This unusually high thermal stability of the AaHAS trimer was further confirmed by nano-differential scanning fluorimetry, which showed two melting temperatures, 62.4°C and 75.3°C, probably due to the presence of two distinct thermally induced transitions (Fig. 1C). Interestingly, it was shown that HAS from the hyperthermophilic archaeon A. pernix is also very heat stable (12). In addition, chemical cross-linking experiments were performed using amine-reactive cross-linkers. Despite the heat treatment, when AaHAS was subjected to bissulfosuccinimidyl suberate (BS³) and disuccinimidyl suberate (DSS) cross-linking and subsequently analyzed by SDS-PAGE, the accumulation of dimeric and trimeric forms became clearly detectable (Fig. 1F).

To investigate the oligomeric state of AaHAS in detergent solution, we performed size exclusion chromatography-multiangle light scattering (SEC-MALS) experiments. The averaged masses of the AaHAS complex with and without the bound detergents were determined to be 209.02 kDa and 124.36 kDa (Fig. 1E). The latter value is slightly higher than expected but is still in agreement with the predicted molecular mass of the trimeric AaHAS with bound cofactors and lipids. In addition to SEC-MALS, laser-induced liquid bead ion desorption mass spectrometry (LILBID-MS) was also used. Under the harsh-laser condition, the LILBID-MS spectrum showed a dominating trimer of AaHAS, which is observed as a broad peak ranging from m/z 100,000 to 150,000 and as a doubly negatively charged species centered at 65,000 (Fig. 1G, top). The broadened feature is due to the mass of additional lipids, which are not removed in the LILBID process, suggesting that these lipids are tightly bound to the AaHAS, probably at the oligomerization interface. In addition, LILBID-MS measurements were also performed on additionally desalted protein samples. As shown in Fig. 1G (bottom), mainly monoand dimeric AaHAS species were observed, as indicated by two prominent m/z signals of 34,500 and 69,500. This result can be attributed to the decreased stability of AaHAS trimer when desalted. Nevertheless, trimeric AaHAS is still detectable as a singly negatively charged peak at m/z 105,500. Taken together, our LILBID-MS analyses showed that lipids may play an important role in the stabilization and assembly of the trimeric AaHAS complex.

FIG 1 Legend (Continued)

of the AaHAS trimeric complex. (F) Cross-linking analyses of AaHAS. Purified AaHAS was subjected to 0.5/3.5 mM BS³ or 0.5 mM DSS cross-linking at room temperature for 30 min. After quenching with 50 mM Tris-HCl (pH 7.5), samples were incubated at 90°C for 10 min and subjected to SDS-PAGE. (G) Identification of the trimeric AaHAS complex by LILBID-MS. LILBID spectra of purified AaHAS in the presence of 150 mM NaCl (top) and without NaCl (bottom) are shown. The red shadings in the upper spectrum represent the trimeric AaHAS with tightly bound lipids. The positions of the AaHAS monomer (M), dimer (D), and trimer (T) are indicated.



FIG 2 Structure of the AaHAS trimer determined by 3D reconstruction of single-particle cryo-EM images. (A) Typical cryo-EM micrograph of AaHAS imaged with a FEI Titan Krios and a Gatan K2 Summit camera. Bar, 200 Å. (B) Selection of 20 reference-free 2D class averages of AaHAS showing secondary structures in projection. (C) Surface-rendered views of the final 3D map. (D) Fitting of the homology model of AaHAS into the cryo-EM density map by rigid-body fitting in UCSF Chimera. The AaHAS model, which was built using the X-ray crystallographic coordinates of the HAS from *B. subtilis* (PDB entry 6A2J) as a template, is shown as cartoon representation and colored using a rainbow gradient from the N terminus (blue) to the C terminus (red). (E) Cross-section of the map, illustrating the arrangements of the transmembrane helices in the AaHAS trimer. The transmembrane helices of each AaHAS monomer are numbered 1 to 8.

The AaHAS trimer is mainly formed by hydrophobic interactions. To gain more insight into the structure of the AaHAS homo-oligomer complex, we used single-particle electron cryomicroscopy (Fig. 2A). A total of ~52,000 particles were selected from 528 micrographs, and the reference-free two-dimensional (2D) class averages clearly showed triangle-shaped top views and rod-shaped side views of AaHAS complex (Fig. 2B). Three-dimensional (3D) reconstruction and further refinement were performed using RELION (20), resulting in an AaHAS reconstruction at 4.6 Å resolution without applied symmetry (data not shown) and the final map at 4.2 Å resolution with applied C3 symmetry (Fig. 2C). At a low-density threshold, the map shows that a disordered belt of detergents and lipids surrounds the hydrophobic domain of

AaHAS complex and defines the position of the membrane (Fig. S5). Twenty-four transmembrane helices that can be grouped into three independent domains are clearly observed at intermediate density threshold, which leaves no doubt that AaHAS assembles as a trimer. Due to the lack of density in the loops between transmembrane helices, an unambiguous structural model cannot be constructed. Therefore, a homology model of AaHAS was built based on the crystal structure of BsHAS using the Swiss-Model server (21). Fitting of the resulting model into the cryo-EM map showed that, except for the loop region between TM1 and TM2, the majority of the AaHAS model matched relatively well with the experimental map (Fig. 2D). Three AaHAS protomers are arranged in a propeller-shaped manner surrounding the 3-fold symmetry axis. A cross-section view of the AaHAS map revealed that the C bundles (transmembrane region 5 [TM5] to TM8) of three AaHAS monomers are tightly associated with each other, forming the core of the trimeric complex (Fig. 2E). The central trimer interface is mainly constituted by TM5 and TM6 through hydrophobic interactions. Furthermore, extra density was found only in the C bundle, which can be attributed to the presence of heme B or heme O. The recently published crystal structure of BsHAS has revealed that heme B is present in the C bundle and the N bundle is the substrate-heme (heme O) binding domain (17). Therefore, the core of the trimeric complex is expected to be fairly rigid, since heme B is embedded in the C bundle of AaHAS via coordination of its heme iron by two histidine residues. On the other hand, the N bundles (TM1 to TM4) of AaHAS constitute the peripheral domain of the trimeric complex. The peripheral region is likely to be flexible, because the N bundle of AaHAS may undergo conformational changes to allow the uptake of the substrate heme O and the release of the reaction product heme A.

Conclusion. In this study, we investigated the oligomeric state of AaHAS using complementary methods. Our results explicitly indicate that AaHAS exists mainly as a trimer. We provide a structural basis for understanding how HAS forms stable oligomeric complexes. Our results might serve as a useful guide for further studies on the structure and function of HAS.

Methods. (i) Construction of expression vector. Genomic DNA from *Aquifex aeolicus* strain VF5 was isolated using the G-spin genomic DNA extraction kit (iNtRON Biotechnology). The coding sequence of the heme A synthase gene (*ctaA*; gene ID, 1192686) was amplified using the primer pair *ctaA*-F (5'-CACGGATCCATGAACACCAAC CTC) and *ctaA*-R (5'-GTGGAATTCTGCCAGCTCTCCTC), and the resulting 906-bp DNA fragment was cloned into the pJET1.2 vector (Thermo Fisher Scientific). The resulting plasmid, pJET1.2-*ctaA*, was digested with Ncol and EcoRI and subsequently ligated into the corresponding sites of pBADC3 (22), which contains a C-terminal Strep-Tag II, yielding the expression vector pBADC3-*ctaA*.

(ii) Protein production and purification. For heterologous overexpression, Escherichia coli Top10 cells transformed with pBADC3-ctaA vector were grown at 32°C in lysogeny broth (LB) medium supplemented with $100 \,\mu g/ml$ ampicillin to an optical density at 600 nm (OD₆₀₀) of 0.5. Production of heme A synthase was induced by addition of 0.02% (wt/vol) L-arabinose, and incubation was continued for 4 h. Cells were harvested by centrifugation and subsequently resuspended in lysis buffer (20 mM Tris-HCI [pH 7.4], 150 mM NaCl, 1 mM phenylmethylsulfonylfluoride [PMSF], 20 μg/ml DNase I]. Cells were disrupted by passing them through a French pressure cell at 12,000 lb/in² for 3 cycles. The cell lysate was centrifuged at 4°C and 10,000 \times g for 1 h. The supernatant containing the membranes was centrifuged at 4°C and 200,000 \times g for 3 h. The pelleted membranes were resuspended in membrane resuspension buffer (20 mM Tris-HCI [pH 7.4], 150 mM NaCl), and membrane proteins were solubilized by slow stirring with 1% (wt/vol) *n*-decyl- β -D-maltoside (DM) at 4°C for 1 h. The insoluble membrane fraction was removed by ultracentrifugation at 200,000 \times g for 1 h, and the supernatant containing the solubilized AaHAS was supplemented with avidin (0.05 mg avidin per mg of total solubilized protein) and subsequently incubated with StrepTactin resin (IBA) pre-equilibrated with purification buffer (20 mM Tris-HCI [pH 7.4], 150 mM NaCl, 0.15% [wt/vol] DM) at 4°C for 1 h. After incubation, the resin was washed with 15 column volumes (CV) of purification buffer. Strep-Tag II-tagged AaHAS was eluted with 5 CV of the purification buffer supplemented with 5 mM desthiobiotin. The protein was purified further on a Superdex 200 10/300 GL column (GE Healthcare) equilibrated with purification buffer. Peak fractions containing AaHAS were collected, concentrated, and stored at -80° C. The total protein concentration was determined using the bicinchoninic acid (BCA) assay (Pierce). Quantitative amino acid analysis was used to accurately determine the concentration of purified AaHAS at the Functional Genomics Center Zurich. An extinction coefficient of 192 mM⁻¹ cm⁻¹ was determined for the reduced AaHAS at 427 nm.

(iii) SDS-PAGE, BN-PAGE, and chemical cross-linking. Sodium dodecyl sulfatepolyacrylamide gel electrophoresis (SDS-PAGE) was performed using 4 to 12% bis-Tris NuPAGE gels (Invitrogen) followed by Coomassie blue staining. Blue native-PAGE was performed using Novex 4 to 16% bis-Tris gels (Invitrogen) according to the manufacturer's instructions. Chemical cross-linking experiments using bissulfosuccinimidyl suberate (BS³) or disuccinimidyl suberate (DSS) were performed according to the manufacturer's instructions (Thermo Fisher Scientific). The reaction was performed at room temperature for 30 min and quenched by dilution with 50 mM Tris-HCI (pH 7.5).

(iv) Nano-differential scanning fluorimetry. Nano-differential scanning fluorimetry (nano-DSF) was performed by using a Prometheus NT.48 nano-DSF instrument (NanoTemper Technologies). Measurements were performed at an AaHAS concentration of 1 mg/ml using the following settings: 15 to 98°C, 0.5°C/min.

(v) Size exclusion chromatography coupled with multiangle light scattering. SEC-MALS experiments were performed using a Viscotek SEC-MALS 20 system (Malvern) at 4°C. Samples (150 µl AaHAS at ~2 mg/ml) were loaded onto a Superdex 200 10/300 GL column pre-equilibrated with purification buffer. Data were analyzed using OmniSEC software (Malvern). The specific refractive index increment (*dn/dc*) of the protein was 0.187 ml/g. The *dn/dc* value of DM was taken as 0.12 ml/g (23). The molar extinction coefficient of AaHAS at 280 nm was calculated to be 50,435 M⁻¹ cm⁻¹ using the ProtParam server.

(vi) Laser-induced liquid bead ion desorption mass spectrometry. Prior to LILBID-MS, AaHAS (2 mg/ml) samples were buffer exchanged into 20 mM Tris-HCl (pH 7.4), 0.15% DM. Release of solvated AaHAS ions from solution was achieved by irradiating microdroplets (diameter, ~50 μ m; volume, ~65 pl) with pulsed infrared laser light ($\lambda = 3 \mu$ m) (24). The laser was a standard Nd:YAG laser and worked with a maximum energy of 23 mJ (harsh mode). Ion optics based on a Wiley-McLaren type accelerator are used to transfer the ions into a homebuilt time-of-flight (TOF) setup for mass analysis. More detailed information regarding LILBID was published previously (25). Ion detection was performed in negative-ion mode.

(vii) Single-particle cryo-electron microscopy. To prepare the cryo-electron microscopy (cryo-EM) grids, 3 μ l of the purified AaHAS (2.5 mg/ml) was applied to glow-discharged holey carbon film grids (GIG Au R 1/1, 300 mesh), which were plasma cleaned for 1 min using the Gatan Solarus system. Grids were blotted with Whatman 595 filter paper for 3 s at 16°C and 100% humidity and then flash frozen in liquid ethane using a Vitrobot (FEI). Grids were loaded onto a FEI Titan Krios transmission electron microscope operated at 300 kV. Images were collected using SerialEM (26) software on a K2 detector in the superresolution counting mode at a calibrated magnification of ×165,000 (yielding a pixel size of 0.41 Å) and a dose rate of 8.45 e⁻/Å²/s. A Quantum energy filter (Gatan), whose energy slot was set to 20 eV, was applied to remove any inelastic scattering. Defocus value varied from -1.5 to -2.5μ m. During movie data collection, frame alignment, exposure weighting, and contrast transfer function parameter estimation were performed using the cryo-EM platform with automated data collection scripts integrated with Motioncor2 (27) and CTFFIND4 (28) at the Institute of Biophysics of the Chinese Academy of Sciences. Particle picking was performed using

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Gautomatch (https://www2.mrc-lmb.cam.ac.uk/research/locally-developed-software/ zhang-software/) without a template, and particle analysis and 3D reconstruction were performed using RELION (20). Detailed information about data collection and processing is provided in Fig. S4 and Table S1.

Data availability. The EM map was deposited in EMDB with the accession number EMD-10987.

SUPPLEMENTAL MATERIAL

Supplemental material is available online only. FIG S1, TIF file, 0.1 MB. FIG S2, TIF file, 0.7 MB. FIG S3, TIF file, 0.1 MB. FIG S4, TIF file, 2 MB. FIG S5, TIF file, 1.9 MB. TABLE S1, DOCX file, 0.01 MB.

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