



Article The Fabrication of Portland Composite Cement Based on Pozzolan Napa Soil

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Abstract: The objective of this study is to investigate Napa soil's potential as an alternative additive in producing Portland composite cement. The Napa soil of Tanah Datar district, West Sumatra, Indonesia is a natural material which contains SiO₂ and Al₂O₃ as its major components. The parameters used were the fineness of the cement particles, the amount left on a 45 μ m sieve, the setting time, normal consistency, loss on ignition, insoluble parts, compressive strength and chemical composition. The composition of Napa soils (% *w*/*w*) used as variables include 4, 8, 12 and 16%. Furthermore, 8% pozzolan was used as a control in this research. The results showed that the compressive strength of Napa soil cement which contained 4% Napa soil was much better compared to that of the control on the 7th and 20th day. Furthermore, all the analyzed Napa soil cements met the standard of cement as stipulated in Indonesian National Standard, SNI 7064, 2016.

Keywords: cement; pozzolans; Napa soil; composite materials; compressive strength

1. Introduction

Cement, as the binding material for stones and concretes, plays an important role in every construction activity; therefore, one cannot separate cement and construction [1–3]. In 1824, Joseph Aspidin from England carried out the calcination of the mixture of limestone and clay. Through the developments of science and technology, research on high-quality cement has been carried out to reduce the disadvantages of ordinary cement [4].

Portland composite cement (PCC) is a new variant type of commercial cement that is similar in characteristics with Portland cement in general, but has better quality, is ecofriendly and cheap [3]. The composition of the cement materials includes clinker, gypsum and additive materials such as fly ash and trace or pozzolan [3,4]. Therefore, the Indonesian cement industry uses pozzolan as an additive material.

Pozzolan is a material rich in aluminum and silicon [5], which in its finely divided form and in the presence of water, chemically reacts with calcium hydroxide. This reaction gives rise to hydrated calcium silicate and other byproducts [4,6]. Although pozzolan does not bind materials as effectively as cement, the water it contains causes it to form binding substances through certain reactions [7]. Therefore, pozzolan may be substituted with another alternative material with high silica oxide content to produce cement [8]. Unfortunately, the high abundance of Napa soil in West Sumatera has not been fully utilized (Figure 1a). In West Sumatera, an estimates 150,000,000 tons (154 Ha) of Napa soils can be found in various areas such as Pesisir Selatan Regency, Tanah Datar Regency, Solok Regency, and Lima Puluh Kota Regency (Figure 1b).



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3'N

0

3*S

6'S





(b)

Figure 1. (a) Napa soil distribution in Sumatera Island. Topography of West Sumatera, (b) and map of West Sumatera province which contains the largest amount of Napa soil [9].

Napa soil is produced naturally, originating from the weathering of feldspathic rocks and carried out by endogenous forces that do not move from the parent rock or origin. As this soil does not move, it has a purer nature, in the sense that it is not mixed with other types of soil and material. Napa soil contains alumina silicate in the ratio of SiO_2/Al_2O_3 , which is approximately 1.25:3.43. In addition, it contains SiO_2 , Al_2O_3 , Fe_2O_3 , TiO_2 , CaO, K_2O and Na₂O, with percentages of 70.43, 20.52, 3.67, 0.40, 2.70, 1.26% and trace, respectively [9]. Meanwhile, Napa soil has not been effectively used to its full potential. Therefore, in this research, Napa soil was used to produce a type of PCC that varies in additive materials such as clinker, gypsum and limestone [10], where it was used as a natural pozzolan. Based on its composition, Napa soil was categorized as natural pozzolan to produce PCC.

The main compounds in cement include: elite $(3CaO \cdot SiO_2 (C_3S))$; belite $2CaO \cdot SiO_2 (C_2S)$; aluminate, $3CaO \cdot Al_2O_3 (C_3A)$; and ferrite, $4CaO \cdot Al_2O_3 \cdot Fe_2O_3 (C_4AF)$. When cement and water react together, the product formed is called calcium–silicate–hydrated (C-S-H) gel, and it is the main agent that binds the cement and particles together within the concrete. The following chemical equations illustrate the reactions between C₂S and C₃S [11,12].

$$2(3CaO \cdot SiO_2) + 6H_2O \rightarrow 3CaO \cdot 2SiO_2 \cdot 3H_2O + 3Ca (OH)_2$$
(1)

$$2(2CaO \cdot SiO_2) + 4H_2O \rightarrow 3CaO \cdot 2SiO_2 \cdot 3H_2O + Ca (OH)_2$$
⁽²⁾

2. Materials and Methods

2.1. Tools and Materials

The tools used in this research were crusher ME-100 Jaw Crusher 5"X8", mini mill, mixer and its dish, stirrer Toni TechnikToniMI, mixing knife, Vicat tools, ebonite ring and glass plate, Zwick Roell Toni SET, alpine air jet 200LS-N, cube mold, pressing equipment, oven Carbolite CWF 1300, Blaine tool Toni Technik Toni Trol. Glass equipment, X-ray fluorescence (XRF) model Epsilon3 PANanalytical Netherlands, X-ray diffraction (XRD) model XPERT PRO PANanalytical Netherlands, and thermogravimetric analyzer (TGA) model DTG 60 Shimadzu, Kyoto, Japan.

The materials used were Napa soil from subdistrict of Lintau of Tanah Datar district, clinker, gypsum, limestone, water, Herzog pill, NaOH, HCl (Merck, Jakarta, Indonesia, pa grade), MM indicator and Ottawa sand.

2.2. Preparation of Napa Soil

Lumps of Napa soil were crushed using crusher. Then, its fine particles were dried under sunlight in order to evaporate its water content and then sieved to pass through a 90 µm sieve.

2.3. Preparation of Composite Cement

Cement samples were prepared by thoroughly mixing various ratios of clinker, gypsum, limestone and pozzolan or Napa soil bases. In order to determine the characteristics of the cement produced, the analysis of its particles' fineness, loss of ignition, insoluble residue, particles left on 45 μ m sieve, normal consistency, setting time and the compressive strength of the mortar produced was carried out [13]. Meanwhile, XRF (X-ray Fluorescence) was used to determine its chemical composition.

2.4. Characterization of Composite Cement

2.4.1. XRF Analysis

About 30 g of Napa soil cement and Napa soil were added using 2 pellets of Herzog pill, and ground. The fine sample produced was added into the XRF analyzer to carry out the characterization step.

2.4.2. Blaine Analysis

Determining the cement's fineness was carried out using Blaine analysis. About 2.8948 g sample was placed into the Blaine apparatus for the characterization process.

2.4.3. Lost on Ignition (LOI) Analysis

Analysis of LOI was carried out according to the loss of water and CO_2 when the cement was ignited. Firstly, 1.000 g of cement was ignited at the temperature of 1000 °C for 15 min. Then, it was placed in a desiccator for the cooling process.

2.4.4. Insoluble Residue Analysis

One gram of the sample was placed into 250 mL beaker glass, followed by the addition of 25 mL of HCl and aquadest dropwise until it reached 100 mL. The solution was boiled, filtered and rinsed using aquadest. Then, the precipitate formed was placed in a beaker which contained 1% of 100 mL NaOH. Additionally, two drops of methyl red indicator and a few drops of HCl were added until the color of the solution become red. It was then filtered and rinsed with hot water and NH₄NO₃ solution. Finally, the solid was placed in a furnace at 1000 °C for 30 min and weighed. The percentage of insoluble part was calculated using this formula:

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% insoluble part = (mass of precipitate/mass of sample) \times 100% (3)
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2.4.5. Sieving Analysis

A 20 g of sample was placed on 45 μ m sieve and analyzed using sieving machine for 3 min. After that, the remaining samples left in the sieve were weighed. The percentage of sieving analysis was calculated using this formula:

% sieving analysis = (mass after sieving/mass before sieving) \times 100% (4)

2.4.6. Normal Consistency Analysis

A 650 g of sample was placed in the mixing dish followed by the addition of 156 mL of water. Then, the cement was added to the water and mixed for 30 s and left to set for 15 s. The paste that set on the bowl was collected and further mixed for another 1 min. After that, the paste produced was collected again, molded into a ball shape and pushed through the Vicat ring.

2.4.7. Setting Time Analysis

Water was added to 650 g of cement to produce paste according to normal consistency analysis. Then, the penetrating test was carried out using Vicat pins of 1 diameter every 10 min. The first setting was marked when the pin penetrated more than 25 mm and stopped when the Vicat pin could not penetrate further, which implies that it has reached the final setting.

2.4.8. Compressive Strength Analysis

Compressive strength analysis was carried out based on the standard procedure of PT. Semen Padang, which refers to the Indonesian National Standard (SNI 2049:2015). The samples used for this analysis were 500 g of cement, 1375 g of standard sand and 276 mL of water. Firstly, the cement was placed in the dish and water was added. Using a mixer, the samples were mixed with a low velocity (140 \pm 5) gradient per minute for 30 s and the sand was slowly added to the mixture with low velocity. Meanwhile, the velocity was increased to a gradient of (285 \pm 10) per minute for 30 s. The mixer was stopped and left for 1.5 min, and within the first 15 s the dough attached to the wall of the bowl was lowered and the stirring bowl was closed with a lid to prevent evaporation of water in the paste. Then, the mixer was turned on for 60 s at medium velocity, then switched off, and subsequently turned on again for another 15 s. After pouring the mixture into each cube, they were each slowly pressed 32 times at ± 4 mm cube. All residual mixtures on the cube mold were scraped with a knife, while the surface was softened with a spoon. After molding, the sample was placed in a humid environment for one whole day, and its surface was covered to prevent contact with water. Subsequently, the sample was removed from the mold, soaked in water containing calcium oxide, and then placed in an antirust humid room until it was analyzed. Evaluation of its compressive strength was performed on day 3, 7, and 28. Compression of the sample was carried out using tools that were calibrated.

3. Results and Discussion

The use of Napa soil from the subdistrict of Lintau of Tanah Datar district as an alternative substitution material for pozzolan was carried out in different variations, while the pozzolan was used as the control. The addition of Napa soil is the source of SiO₂ that reacts with the byproduct of the reaction between $3\text{CaO}\cdot\text{SiO}_2$, $2\text{CaO}\cdot\text{SiO}_2$ and water (Equations (2) and (3)). $3\text{CaO}\cdot\text{SiO}_2\cdot3\text{H}_2\text{O}$ is the important factor that contributes to the strength of the cement; the more $3\text{CaO}\cdot\text{SiO}_2\cdot3\text{H}_2\text{O}$ formed, the greater the strength of the mortar to withstand a load [12]. Then, the Ca(OH)₂, which is the byproduct of the reaction between C₃S and C₂S, reacts with SiO₂ in the pozzolan as shown in the equation below:

$$3Ca(OH)_2 + SiO_2 + H_2O \rightarrow 3CaO \cdot SiO_2 \cdot 3H_2O,$$
 (5)

The reform of $3CaO \cdot SiO_2 \cdot 3H_2O$ increases the compressive strength of the mortar. $Ca(OH)_2$ should be minimized as a byproduct because it fills a bigger space in the structure that causes breakages in the binding of the hardening cement paste and cracks in the

cementation [14,15]. The variation in the amount of the Napa soil ((w/w)), as additive, in producing composite cement is shown in Table 1.

Sample	Clinker (%)	Gypsum (%)	Limestone (%)	Pozzolan (%)	Napa Soil (%)
PCC	76%	4%	12%	8%	-
CNS1	76%	4%	16%	-	4%
CNS2	76%	4%	12%	-	8%
CNS3	76%	4%	8%	-	12%
CNS4	76%	4%	4%	-	16%

Table 1. The variation of materials in composite cement.

3.1. XRF Analysis of Composite Cement

The chemical composition of composite cement using XRF is shown in Table 2. The table shows that the quantities of SiO_2 , Al_2O_3 , and Fe_2O_3 and MgO are proportional to the quantity of Napa soil in composite cement, which is due to SiO_2 being the main composite of the Napa soil. Meanwhile, the composition of CaO is inversely proportional to the quantity of Napa soil, which is good, as the source of CaO is limestone, which is important to balance the proportion of Napa soil. In addition, MgO is one of the compounds that affect the quality of the cement and it is important to make sure that the level of MgO in the cement is less than 5.0% [16]. This is because the presence of high amounts of Mg(OH)₂ and periclase can weaken the cement used in construction.

Table 2. Chemical composition ((www)) of cement with variation of Napa soil.

Sample	SiO ₂	Al_2O_3	Fe ₂ O ₃	CaO	MgO	SO ₃
PCC	22.00	6.10	3.29	59.36	0.91	1.79
CNS1	19.98	5.49	3.16	61.96	0.96	1.84
CNS2	21.27	6.04	3.37	60.04	1.00	1.75
CNS3	22.79	6.63	3.64	57.89	1.08	1.77
CNS4	24.72	7.34	3.92	55.49	1.11	1.67

After the determination of the percentages of SiO₂, Al₂O₃, Fe₂O₃, CaO, MgO, and SO₃, the compositions of C₃S (3CaO·SiO₂), C₂S (2CaO·SiO₂), C₃A (3CaO·Al₂O₃) and C₄AF (4CaO·Al₂O₃·Fe₂O₃) were calculated using the formula stipulated in Indonesian National Standard (SNI 2049:2015) [17]:

 $\begin{array}{l} C_3S = (4.071 \times \% CaO) - (7.600 \times \% SiO_2) - (6.718 \times \% Al_2O_3) - (1.430 \times \% Fe_2O_3) - \\ (2.852 \times \% SO_3) \\ C_2S = (2.867 \times \% SiO_2) - (0.7544 \times C_3S) \\ C_3A = (2.650 \times \% Al_2O_3) - (1.692 \times \% Fe_2O_3) \\ C_4AF = (3.043 \times \% Fe_2O_3) \end{array}$

The composition of C_3S , C_2S , C_3A , and C_4AF of cement with variation of Napa soil shown in Table 3. C_3S contributes significantly to the strength in the beginning. Meanwhile, C_2S also contributes to the cement strength in the longer term. C_3A affects the compressive strength after 28 days and becomes less effective after one or two years. C_4AF does not affect the compressive strength significantly [18].

Table 3. The composition of C₃S, C₂S, C₃A, and C₄AF of cement with variation of Napa Soil.

Sample	C ₃ S	C ₂ S	C ₃ A	C ₄ AF
PCC	23.67	45.22	10.60	10.01
CNS1	53.74	16.74	9.20	9.62
CNS2	32.39	36.55	10.30	10.25
CNS3	7.67	59.55	11.41	11.08
CNS4	-21.63	87.20	12.82	11.93

3.2. Analysis of the Fineness

The fineness analysis of the composite cement particle was carried out using a Blaine instrument as stipulated in Indonesian National Standard of SNI 2049:2015. The bar chart that shows the relationship between the fineness of the cement particle and the percentage of Napa soil composition is shown in Figure 2. The fineness of the cement particle and the surface area of the cement are proportional to the amount of Napa soil due to its softness and brittleness. The composite cement produced by using Napa soil as an additive material has smooth and fine granules, so the cement tends to form agglomerates when analyzed using SEM (Figure 3) [19].



Figure 2. The relationship between the fineness of the cement particle and the percentage of Napa soil composition.



Figure 3. SEM image (A) Napa soil (NS), (B) CompositecCement with Napa soil (CNS) and (C) Portland composite cement (PCC).

The quantity of Napa soil used in producing composite cement is proportional to its grinding time until it reaches $3800-4200 \text{ cm}^2/\text{g}$. In Figure 4, CNS4 gives a good result as it



requires only a short setting time and high compressive strength, probably due to its fine cement particles [13].

Figure 4. Analysis of Napa soil usage to the setting time (minutes) with its variation.

3.3. Analysis of the Loss on Ignition

The value of the loss of ignition (LOI) should be minimal because it is proportional to the level of mineral that can be parsed in the cement. The effect of using Napa soil on the loss of ignition (LOI) is shown in Figure 5. Composite cement with 16% Napa soil shows a low LOI value, which means the amount of the mineral parsed is low, which is due to low quantity of limestone in composite cement from all cement samples. The percentage of limestone in cement is inversely proportional to the quantity of Napa soil in cement. From the results, it is clear that mortar with high LOI, containing ground bagasse ash, has slower comprehensive strength development when compared to those with low LOI. However, both of them displayed similar compressive strength over time [20].



Figure 5. The effect of Napa soil usage on the loss of ignition (LoI).

3.4. Analysis of Insoluble Part of Cement

The effect of Napa soil usage on the percentage of the insoluble part is shown in Figure 6. It was discovered that the high percentage of Napa soil added enhanced the percentage of insoluble part of cement, which is due to high content of silica in Napa soil. The identification of the insoluble part was carried out by dissolving cement as shown in the reaction below:



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Figure 6. The effect of Napa soil usage on the insoluble part.

When cement reacts with hydrochloric acid, HCl, it forms a water-soluble salt compound, such as CaCl, AlCl₃, and FeCl₃, H_2SiO_2 and H_4SiO_4 , in the form of a gel that is difficult to dissolve. If cement reacts with NaOH, it will form a compound in the form of salt, as shown in the reaction:

 $\begin{array}{l} H_2SiO_2 + NaOH \rightarrow Na_2SiO_4 \mbox{ (Dissolved)} \\ H_4SiO_4 + NaOH \rightarrow Na_4SiO_4 \mbox{ (Undissolved)} \\ Na_4SiO_4 + HCl \rightarrow H_4SiO_4 + NaCl \end{array}$

Meanwhile, H₄SiO₄ is calcinated at 1000 °C for 1 h to form

$$H_4SiO_4 \rightarrow H_2O + SiO_2$$

SiO₂ cannot be dissolved in water and acid solvents. The insoluble part of the cement should be minimal as possible and SiO₂ needs to completely react. CNS1 shows the low percentage of insoluble parts. CNS2 that has the same composition, with a lower insoluble part percentage than the control. This shows that the cement containing Napa soil has lower percentage of insoluble parts than the cement containing pozzolan. Portland cement contains pozzolan, a non-cementing substance called insoluble residue, which influences properties of the cement such as compressive strength [21]. Therefore, in order to control the amount of non-cementing materials within the cement, ASTM standards prevent the quantity of insoluble residue from exceeding 0.75%. This is lower than the British standard of 1.5%. However, the normal consistency and setting times are unaffected by the addition of insoluble residue to the cement. Despite all these data, this hypothesis needs to be further researched in order to be considered completely reliable [22].

3.5. Analysis of the Rest on Sieve of 45 µm

Figure 7 shows the analysis of Napa soil usage on a 45 μ m sieve. It shows that the quantity of Napa soil used in producing the cement is inversely proportional to the percentage of the 45 μ m sieve. The relationship chart between the use of Napa soil used does not affect the amount of cement that escapes from the 45 μ m sieve, with an average of 17.215%. The particle size of cement is one of the important factors that influences the speed of its reaction with water, and also influences the compressive strength of the cement. If the cement has a fine particle size, when making mortar, it covers the cavities in the mortar mixture. Therefore, the product is denser and has high compressive strength. Of all cement samples, the control had the lowest number of particles above the sieves. It can be understood that many control cements pass through the 45 μ m sieve, which speeds up the binding time and increases the compressive strength of the cement. Arvaniti et al. [23] discovered that the fly ash which was tested met the STM C618 [24] fineness criterion, with not less than 34% of the substance remaining on the 45 μ m sieve. Furthermore, 20% of the slag from the blast furnace met the ASTM C989 [25] fineness criterion.



Figure 7. The effect of Napa soil usage on the percentage remaining on 45 µm sieve.

3.6. Analysis of Normal Consistency

The effect of the of Napa soil usage and its normal consistency is shown in Figure 8. It shows that composite cement that contains 16% Napa soil (CNS4) has higher values of normal consistency than cement CNS1, CNS2, and CNS3, which is due to the size of the cement particles as well as low surface area that reduces the water distribution and eventually requires a greater volume of water to be of normal consistency.



Figure 8. The effect of the utilizing of Napa soil to normal consistency.

3.7. Analysis the Setting Time

Generally, the cement hydration process coupled with the hydration degree makes a key contribution to properties of hardened cement [26]. The content of C_3A , particle size and gypsum in a cement will significantly affect its setting time. The effect of C_3A , which reacts with water, on the setting time is shown in the reaction below:

$$3CaO \cdot Al_2O_3 + 6H_2O \rightarrow 3CaO \cdot Al_2O_3 \cdot 6H_2O$$

 $3CaO \cdot Al_2O_3$ reacts with gypsum to form ettringite, which covers the surface of the 3CaO, while $Al_2O_3 \cdot 6H_2O$ and $3CaO \cdot Al_2O_3$ block the hydration reaction of $3CaO \cdot Al_2O_3$ and prevent the hardening process. Increasing amount of pozzolanic materials can increase the hydration process, hence increasing the required time period [27]. This influences the amount of time the material takes to set [15].

From Figure 9, it can be deduced that the setting time is proportional to the amount of Napa soil in the produced cement. When a large amount of Napa soil added to the cement, this produces a larger particle size, which leads to a longer setting time. Furthermore, the SNI standard starting setting time is 45 min. The maximal final setting time is 375 min. It means that all samples used meet the national standard for cement. The correlation between the percentage of Napa soil and the setting time of the cement is shown in Figure 9.





3.8. Analysis of Napa Soil Usage to the Compressive Strength

Napa soil's effect on compressive strength is displayed in Figure 10, which shows a decrease with increasing composition of the Napa soil as the additive contributing silicaalumina. The cements CNS2 and PCC have the same composition, but the control contains pozzolan, while CNS2 contains Napa soil. Meanwhile, at days 7 and 28, the compressive strength of the mortar of the cement containing 4% Napa soil (CNS1) was higher than that of the pozzolan.



Figure 10. The relationship between the percentage of Napa soil (n = 5) and compressive strength of the cement.

The amount remaining on the 45 μ m sieve is another factor that influenced the compressive strength; the less the material that rested on the sieve, the greater the amount of cement that has a particle size smaller than 45 μ m. Therefore, the mortar formed more solidly and the small particles of the cement filled its pores. The quantity of Cement CNS4 left on the sieve was 15.93%.

XRD analysis was performed to determine the mineral content of composite samples such as calcium silicate hydrate (C-S-H), calcium aluminosilicate hydrate (C-A-S-H), portlandite (Ca(OH)₂), and ettringite minerals in the sample (Figure 11). The intensity of calcium hydroxide (Ca(OH)₂) $2\theta = 20.71$ and ettringite $2\theta = 22.9$ increased with the increasing percentage in the sample. The peak of calcium silicate hydrate (at $2\theta = 26.55$) increased with the percentage in the sample. An increase in intensity also occurred at $2\theta = 29.41$, which indicates a very fast hydration process in the sample. The higher intensity of calcium hydroxide (Ca(OH)₂) was able to form C-S-H and increase the strength of the sample.



Figure 11. XRD analysis of portland composite cement based on pozzolan Napa soil on day 28.

The effect of increasing temperature on the mass loss (weight loss) of the sample and determining the temperature of the decomposition process of the cement sample using TGA analysis is shown in Figure 12.



Figure 12. TGA analysis of portland composite cement based on pozzolan Napa soil on day 28.

The TGA analysis shows a significant weight loss for all of the cement samples. The first weight loss, at 30 °C to 100 °C, occurred due to the evaporation of water from the pores on the surface of samples. Second, the weight loss shown at 100 °C to 450 °C occurred due to dehydration of calcium silicate hydrate and C-A-S-H. The drastic change at 475 °C to 750 °C related to the decomposition of Ca(OH)₂ during the hydration process, as shown in equation below:

$$Ca(OH)_2 \rightarrow CaO + H_2O$$
 (evaporation) (6)

Decarbonation process of CaCO₃ at 800 °C:

$$Ca(OH)_2 + CO_2 \rightarrow CaCO_3 + H_2O \text{ (carbonation)}$$
(7)

$$CaCO_3 \rightarrow CaO + CO_2$$
 (decarbonation) (8)

Then, the percentage of calcium hydroxide and water content was calculated as:

Mass of calcium hydroxide (%) =
$$[(W_{400} - W_{580})/W_{580}] \times (74/18) \times 100$$
 (9a)

where the molecular weights of calcium hydroxide and water are 74 and 1, respectively.

Water content (%) =
$$[(W_{80} - W_{580})/W_{580}] \times 100$$
 (9b)

Figure 13 shows the calculated percentage of calcium hydroxide in the samples. Sample CNS12% (CNS3) contains the highest percentage of calcium hydroxide, while the water content is 1.529%. The lowest percentage of calcium hydroxide was found in sample CNS8% (CNS2), which was caused by the formation of CSH gel in the sample.



Figure 13. (a) Mass of calcium hydroxide, (b) Mass of bound water.

4. Conclusions

All analyzed samples met Indonesia National Standard of SNI 2049:2015. The use of Napa soil as pozzolan in producing cement showed good result percentages compared to control. Meanwhile, the parameters used were the loss of ignition, the amount remaining on 45 μ m sieve, normal consistency, fineness of cement particle, insoluble part, setting time, compressive strength and chemical composition. From this study, the results led to the conclusions below:

- (a) Composite cement with 16% Napa soil shows a low LOI value, which means that the mineral parsed is low and this is due to low quantity of limestone in composite cement from all cement samples.
- (b) The increase in the amount of Napa soil used does not affect the relative amount of cement that escapes the 45 μm sieve, with an average of 17.215%.
- (c) The compressive strength of the mortar on days 7 and 28 of the cement containing 4% Napa soil was higher compared to that of pozzolan. Furthermore, the XRF analysis of the cement showed that the most common chemical compounds produced in the cement were SiO₂ and Al₂O₃.

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