Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

6-Fluoroindan-1-one

Benjamin R. Slaw and Joseph M. Tanski*

Department of Chemistry, Vassar College, Poughkeepsie, NY 12604, USA
Correspondence e-mail: jotanski@vassar.edu

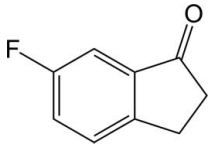
Received 6 June 2014; accepted 26 June 2014

Key indicators: single-crystal X-ray study; T = 125 K; mean σ(C–C) = 0.002 Å; R factor = 0.041; wR factor = 0.121; data-to-parameter ratio = 21.6.

The title compound, C_9H_7FO, crystallizes with two independent molecules in the asymmetric unit, in which corresponding bond lengths are the same within experimental error. The five-membered ring in each molecule is almost planar, with r.m.s. deviations of 0.016 and 0.029 Å. In the crystal, molecules form sheets parallel to (1 0 0) *via* C—H···O and C—H···F interactions with F···F contacts [3.1788 (16) and 3.2490 (16) Å] between the sheets.

Related literature

For the synthesis of 6-fluoroindan-1-one, see: Cui *et al.* (2004) and for its use in synthesis, see: Musso *et al.* (2003); Ślusarczyk *et al.* (2007); Yin *et al.* (2013). For the structure of the parent comound, 1-indanone, see: Morin *et al.* (1974) and Ruiz *et al.* (2004), the later containing a detailed analysis of the hydrogen bonding. For a related isomeric structure, 5-fluoroindan-1-one, see: Garcia *et al.* (1995). For more information on C—H···X interactions, see Desiraju & Steiner (1999) and on fluorine–fluorine interactions in the solid state, see: Baker *et al.* (2012). For van der Waals radii, see: Bondi (1964).



Experimental

Crystal data

C_9H_7FO
M_r = 150.15
Monoclinic, $P2_1/n$
a = 7.1900 (4) Å
b = 12.4811 (6) Å
c = 15.8685 (8) Å
β = 99.453 (1)°
V = 1404.69 (13) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.11 mm^{-1}
T = 125 K
0.37 × 0.26 × 0.04 mm

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker 2007)
T_{min} = 0.91, T_{max} = 1.00
22840 measured reflections
4298 independent reflections
3345 reflections with $I > 2\sigma(I)$
R_{int} = 0.029

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.041
$wR(F^2)$ = 0.121
S = 1.03
4298 reflections
199 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.40 e Å^{-3}
$\Delta\rho_{min}$ = −0.21 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C5—H5···O1^{i}	0.95	2.47	3.3873 (14)	161
C14—H14···O2^{ii}	0.95	2.65	3.5107 (14)	150
C2—H2B···F2^{iii}	0.99	2.46	3.2062 (13)	132
C6—H6···O2^{iv}	0.95	2.65	3.5338 (14)	154
C11—H11B···O1^{v}	0.99	2.52	3.3348 (13)	140
C15—H15···F1^{vi}	0.95	2.52	3.3664 (13)	148

Symmetry codes: (i) $-x+\frac{3}{2}, y+\frac{1}{2}, -z+\frac{1}{2}$; (ii) $-x+\frac{1}{2}, y-\frac{1}{2}, -z+\frac{1}{2}$; (iii) $x+\frac{1}{2}, -y+\frac{3}{2}, z+\frac{1}{2}$; (iv) $x+\frac{1}{2}, -y+\frac{3}{2}, z-\frac{1}{2}$; (v) $-x+1, -y+1, -z+1$; (vi) $-x+1, -y+1, -z$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*, *OLEX2* (Dolomanov *et al.*, 2009) and *Mercury* (Macrae *et al.*, 2006).

This work was supported by Vassar College. X-ray facilities were provided by the US National Science Foundation (grant No. 0521237 to JMT).

Supporting information for this paper is available from the IUCr electronic archives (Reference: KJ2241).

References

Baker, R. J., Colavita, P. E., Murphy, D. M., Platts, J. A. & Wallis, J. D. (2012). *J. Phys. Chem. A*, **116**, 1435–1444.
Bondi, A. (1964). *J. Phys. Chem.* **68**, 441–451.
Bruker (2007). *SAINT*, *SADABS* and *APEX2*. Bruker AXS Inc., Madison, Wisconsin, USA.
Cui, D., Zhang, C., Kawamura, M. & Shimada, S. (2004). *Tetrahedron Lett.* **45**, 1741–1745.
Desiraju, G. R. & Steiner, T. (1999). In *The Weak Hydrogen Bond*. Oxford University Press.
Dolomanov, O. V., Bourhis, L. J., Gildea, R. J., Howard, J. A. K. & Puschmann, H. (2009). *J. Appl. Cryst.* **42**, 339–341.
Garcia, J. G., Enas, J. D., VanBrocklin, H. F. & Fronczek, F. R. (1995). *Acta Cryst.* C**51**, 301–304.
Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J. (2006). *J. Appl. Cryst.* **39**, 453–457.
Morin, Y., Brassy, C. & Mellier, A. (1974). *J. Mol. Struct.* **20**, 461–469.
Musso, D. L., Cochran, F. R., Kelley, J. L., McLean, E. W., Selph, J. L., Rigdon, G. C., Orr, G. F., Davis, R. G., Cooper, B. R., Styles, V. L., Thompson, J. B. & Hall, W. R. (2003). *J. Med. Chem.* **46**, 399–408.
Ruiz, T. P., Fernández-Gómez, M., González, J. J. L., Koziol, A. E. & Roldán, J. M. G. (2004). *J. Mol. Struct.* **707**, 33–46.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Ślusarczyk, M., DeBorggraeve, W. M., Toppet, S. & Hoornaert, G. J. (2007). *Eur. J. Org. Chem.* **18**, 2987–2994.
Yin, H., Liu, M. & Shao, L. (2013). *Org. Lett.* **15**, 6042–6045.

supporting information

Acta Cryst. (2014). E**70**, o841 [doi:10.1107/S1600536814015049]

6-Fluoroindan-1-one

Benjamin R. Slaw and Joseph M. Tanski

1. Comment

The titular compound 6-fluoroindan-1-one may be synthesized by the $Tb(OTf)_3$-catalyzed cyclization of 3-(4-fluorophenyl)propanoic acid (Cui *et al.*, 2004). The substance has found laboratory applications in the synthesis of α-arylated compounds (Yin *et al.*, 2013), the synthesis of ethyl 2-(6-fluoro-1-hydroxy-1-indanyl)acetate, a potent muscle relaxant derivative (Musso *et al.*, 2003), and in the creation of methylene-bridged biologically active pteridine derivatives for potential hepatitis C treatments (Ślusarczyk *et al.*, 2007). The crystal structure of the parent compound, 1-indanone, has been reported previously (Morin *et al.*, 1974; Ruiz *et al.*, 2004), as has the structure of an isomer of the title compound, 5-fluoroindan-1-one (Garcia *et al.*, 1995).

The titular compound crystallizes with two molecules of 6-fluoroindan-1-one in the asymmetric unit (Figure 1). The carbonyl C—O bond lengths of 1.2172 (13) and 1.2179 (13) Å, as for the other bond lengths, are the same within the experimental error between the two independent molecules. These carbonyl C—O bond lengths are similar to those found in the structure of the parent comound, 1-indanone, 1.217 (2) Å (Ruiz *et al.*, 2004), and in the structure of the isomeric compound 5-fluoroindan-1-one, 1.218 (2) Å (Garcia *et al.*,1995). The C—F bond lengths in 6-fluoroindan-1-one, 1.3592 (12) and 1.3596 (11) Å, are also very similar to that found in the structure of the isomeric compound 5-fluoroindan-1-one, 1.354 (2) Å.

The molecules pack together in the solid state to form a two-dimensional sheet parallel to the 1 0 0 plane *via* several intermolecular C—H···O and C—F···H interactions (Figure 2, Table 2) measuring slightly less than the sum of the van der Waals radii (Bondi, 1964). The oxygen atom in each independent molecule forms two C—H···O interactions, while each independent molecule also forms one C—F···H interaction. For a discussion of C—H···*X* interactions, see Desiraju & Steiner (1999). There are also two long F···F interactions linking the two-dimensional sheets, (Figure 3, Table 1), which are somewhat longer than the sum of the van der Waals radii, 2.94 Å (Bondi, 1964). For a discussion of fluorine-fluorine interactions, which can vary widely in their metrical parameters and strength, see Baker *et al.* (2012).

2. Experimental

Crystalline 6-fluoroindan-1-one (I) was purchased from Aldrich Chemical Company, USA.

3. Refinement

All non-hydrogen atoms were refined anisotropically. Hydrogen atoms on carbon were included in calculated positions and refined using a riding model at C–H = 0.95 and 0.99 Å and $U_{iso}(H) = 1.2 \times U_{eq}(C)$ of the aryl and methylene C-atoms, respectively. The extinction parameter (EXTI) refined to zero and was removed from the refinement.

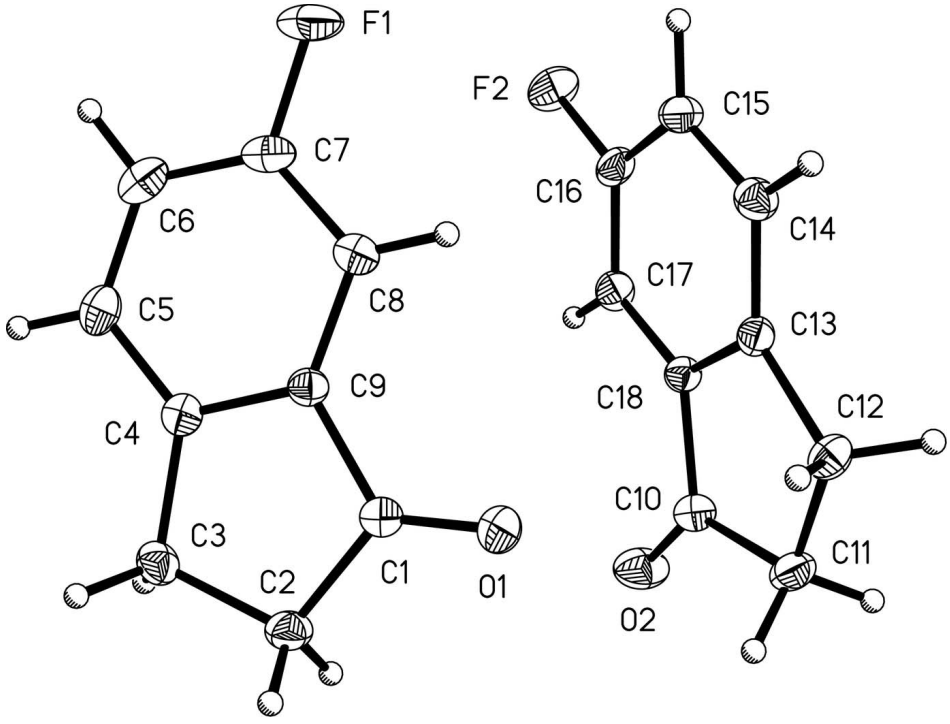


Figure 1

A view of the two independent molecules of the title compound, with atom numbering scheme. Displacement ellipsoids are shown at the 50% probability level.

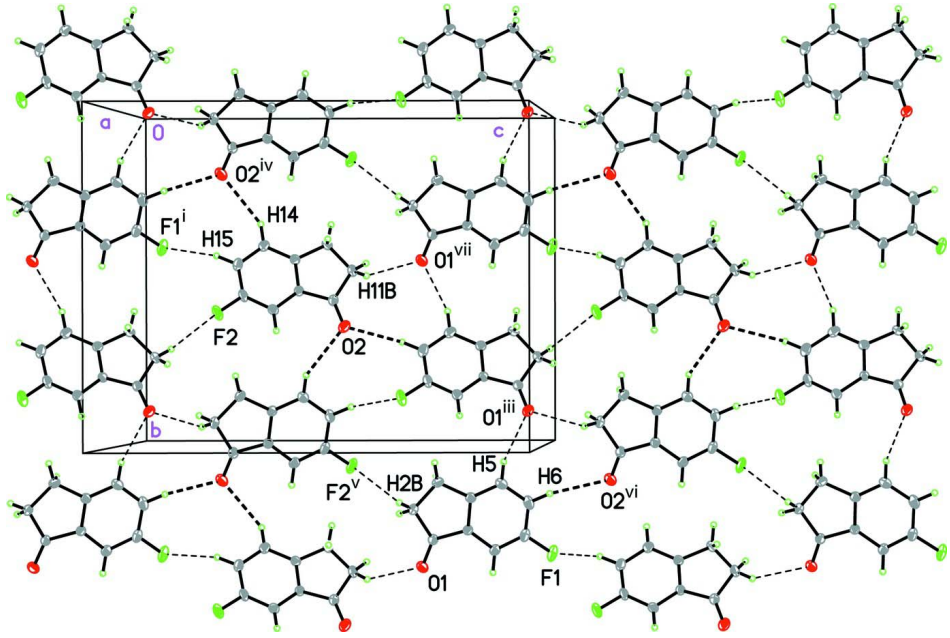


Figure 2

A view of the C—H···O and C—H···F interactions in the packing of 6-fluoroindan-1-one forming a sheet parallel to the 1 0 0 plane. Displacement ellipsoids are shown at the 50% probability level. Symmetry codes: (i) -*x* + 1, -*y* + 1, -*z*; (iii) -*x* + 3/2, *y* + 1/2, -*z* + 1/2; (iv) -*x* + 1/2, *y* - 1/2, -*z* + 1/2; (v) *x* + 1/2, -*y* + 3/2, *z* + 1/2; (vi) *x* + 1/2, -*y* + 3/2, *z* - 1/2; (vii) -*x* + 1, -*y* + 1, -*z* + 1.

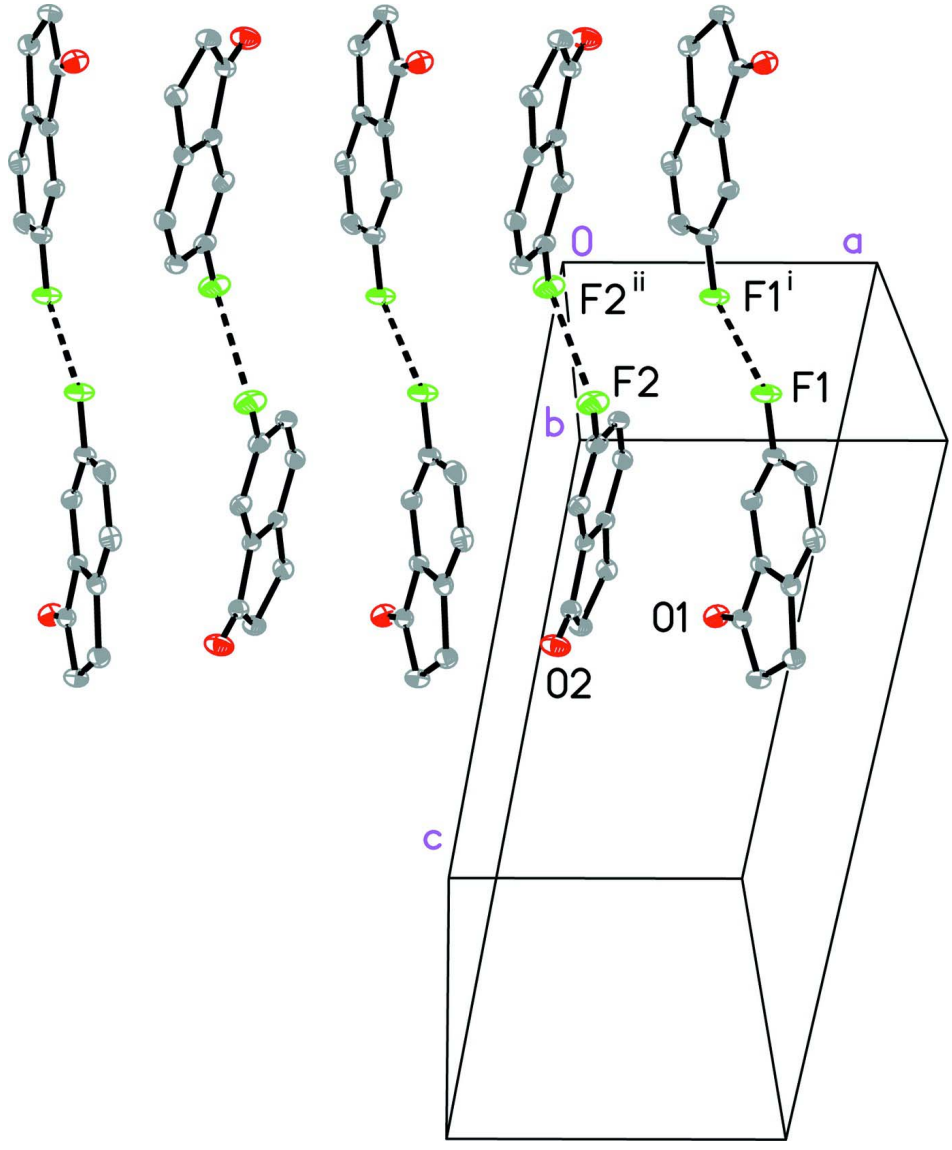


Figure 3

A view of the intermolecular F···F interactions in the packing of 6-fluoroindan-1-one. Distances F1···F1^{i} 3.1788 (16) Å, F2···F2^{ii} 3.2490 (16) Å. Displacement ellipsoids are shown at the 50% probability level; hydrogen atoms removed for clarity. Symmetry codes: (i) -*x* + 1, -*y* + 1, -*z*; (ii) -*x*, -*y* + 1, -*z*.

6-Fluoroindan-1-one

Crystal data

C_9H_7FO
M_r = 150.15
Monoclinic, $P2_1/n$
a = 7.1900 (4) Å
b = 12.4811 (6) Å
c = 15.8685 (8) Å
β = 99.453 (1)°
V = 1404.69 (13) Å^3
Z = 8

F(000) = 624
D_x = 1.420 Mg m^{-3}
Mo *Kα* radiation, λ = 0.71073 Å
Cell parameters from 9796 reflections
θ = 2.6–30.5°
μ = 0.11 mm^{-1}
T = 125 K
Plate, colourless
0.37 × 0.26 × 0.04 mm

Data collection

Bruker APEXII CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 8.3333 pixels mm^{-1}
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Bruker 2007)
T_{min} = 0.91, T_{max} = 1.00
22840 measured reflections
4298 independent reflections
3345 reflections with $I > 2\sigma(I)$
R_{int} = 0.029
θ_{max} = 30.5°, θ_{min} = 2.1°
h = −10→10
k = −17→17
l = −22→22

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.041
$wR(F^2)$ = 0.121
S = 1.03
4298 reflections
199 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0654P)^2 + 0.2949P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ = 0.001
$\Delta\rho_{max}$ = 0.40 e Å^{-3}
$\Delta\rho_{min}$ = −0.21 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
F1	0.57983 (11)	0.61035 (7)	0.04096 (4)	0.03306 (19)
F2	0.08173 (11)	0.60328 (6)	0.05655 (4)	0.03227 (19)
O1	0.58188 (12)	0.56668 (7)	0.37571 (5)	0.02425 (18)
O2	0.10060 (13)	0.64991 (7)	0.39289 (5)	0.0290 (2)
C1	0.63562 (14)	0.65257 (8)	0.35245 (6)	0.01661 (19)
C2	0.69797 (16)	0.74848 (9)	0.40858 (7)	0.0209 (2)
H2A	0.8032	0.7285	0.4543	0.025*
H2B	0.5922	0.7756	0.4354	0.025*
C3	0.76215 (15)	0.83443 (8)	0.35003 (7)	0.0198 (2)
H3A	0.6904	0.9017	0.3526	0.024*
H3B	0.8983	0.8498	0.3666	0.024*
C4	0.72154 (14)	0.78578 (8)	0.26152 (7)	0.0170 (2)
C5	0.74667 (15)	0.83033 (9)	0.18320 (7)	0.0217 (2)
H5	0.7957	0.9007	0.1806	0.026*
C6	0.69871 (16)	0.76990 (10)	0.10921 (7)	0.0238 (2)
H6	0.7148	0.7986	0.0555	0.029*

C7	0.62701 (15)	0.66714 (9)	0.11444 (7)	0.0218 (2)
C8	0.60039 (14)	0.62037 (9)	0.19016 (7)	0.0188 (2)
H8	0.5513	0.5499	0.1923	0.023*
C9	0.64971 (14)	0.68253 (8)	0.26346 (6)	0.01567 (19)
C10	0.14218 (14)	0.56199 (9)	0.36835 (6)	0.0184 (2)
C11	0.19726 (16)	0.46437 (9)	0.42394 (7)	0.0214 (2)
H11A	0.0886	0.4385	0.4493	0.026*
H11B	0.3016	0.4821	0.4707	0.026*
C12	0.26009 (15)	0.37845 (9)	0.36496 (7)	0.0195 (2)
H12A	0.3964	0.363	0.381	0.023*
H12B	0.1884	0.3112	0.3678	0.023*
C13	0.21781 (14)	0.42705 (8)	0.27662 (6)	0.01607 (19)
C14	0.24187 (15)	0.38210 (9)	0.19853 (7)	0.0195 (2)
H14	0.2889	0.3112	0.196	0.023*
C15	0.19576 (15)	0.44297 (9)	0.12451 (7)	0.0212 (2)
H15	0.2117	0.4142	0.0707	0.025*
C16	0.12627 (15)	0.54614 (9)	0.13004 (6)	0.0204 (2)
C17	0.10060 (15)	0.59323 (8)	0.20572 (7)	0.0189 (2)
H17	0.0525	0.6639	0.2079	0.023*
C18	0.14943 (14)	0.53097 (8)	0.27900 (6)	0.01578 (19)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0391 (4)	0.0439 (5)	0.0154 (3)	0.0011 (3)	0.0024 (3)	−0.0091 (3)
F2	0.0434 (4)	0.0365 (4)	0.0151 (3)	−0.0035 (3)	−0.0006 (3)	0.0100 (3)
O1	0.0307 (4)	0.0212 (4)	0.0204 (4)	−0.0055 (3)	0.0027 (3)	0.0036 (3)
O2	0.0401 (5)	0.0244 (4)	0.0227 (4)	0.0082 (4)	0.0053 (3)	−0.0045 (3)
C1	0.0165 (4)	0.0173 (5)	0.0158 (4)	0.0016 (4)	0.0022 (3)	−0.0003 (3)
C2	0.0274 (5)	0.0192 (5)	0.0165 (5)	−0.0002 (4)	0.0044 (4)	−0.0029 (4)
C3	0.0217 (5)	0.0158 (5)	0.0220 (5)	−0.0010 (4)	0.0041 (4)	−0.0034 (4)
C4	0.0163 (4)	0.0160 (5)	0.0192 (5)	0.0021 (3)	0.0039 (4)	0.0005 (4)
C5	0.0210 (5)	0.0195 (5)	0.0254 (5)	0.0013 (4)	0.0067 (4)	0.0051 (4)
C6	0.0224 (5)	0.0307 (6)	0.0193 (5)	0.0050 (4)	0.0070 (4)	0.0061 (4)
C7	0.0214 (5)	0.0287 (6)	0.0149 (5)	0.0047 (4)	0.0018 (4)	−0.0040 (4)
C8	0.0184 (5)	0.0195 (5)	0.0179 (5)	0.0009 (4)	0.0013 (4)	−0.0024 (4)
C9	0.0166 (4)	0.0154 (4)	0.0151 (4)	0.0010 (3)	0.0027 (3)	−0.0008 (3)
C10	0.0187 (5)	0.0205 (5)	0.0159 (4)	0.0010 (4)	0.0026 (3)	−0.0004 (4)
C11	0.0262 (5)	0.0234 (5)	0.0147 (4)	0.0006 (4)	0.0037 (4)	0.0026 (4)
C12	0.0214 (5)	0.0191 (5)	0.0182 (5)	0.0026 (4)	0.0040 (4)	0.0048 (4)
C13	0.0156 (4)	0.0166 (5)	0.0160 (4)	−0.0011 (3)	0.0027 (3)	0.0012 (3)
C14	0.0189 (5)	0.0195 (5)	0.0206 (5)	−0.0004 (4)	0.0044 (4)	−0.0027 (4)
C15	0.0212 (5)	0.0270 (5)	0.0159 (5)	−0.0047 (4)	0.0045 (4)	−0.0039 (4)
C16	0.0209 (5)	0.0258 (5)	0.0135 (4)	−0.0047 (4)	0.0003 (4)	0.0049 (4)
C17	0.0200 (5)	0.0178 (5)	0.0178 (5)	0.0000 (4)	−0.0002 (4)	0.0029 (4)
C18	0.0157 (4)	0.0169 (5)	0.0145 (4)	−0.0007 (3)	0.0019 (3)	0.0002 (3)

Geometric parameters (Å, º)

F1—C7	1.3592 (12)	C7—C8	1.3772 (15)
F1—F1^{i}	3.1788 (16)	C8—C9	1.3947 (14)
F2—C16	1.3596 (11)	C8—H8	0.95
F2—F2^{ii}	3.2490 (16)	C10—C18	1.4790 (14)
O1—C1	1.2172 (13)	C10—C11	1.5181 (15)
O2—C10	1.2179 (13)	C11—C12	1.5392 (15)
C1—C9	1.4802 (14)	C11—H11A	0.99
C1—C2	1.5152 (14)	C11—H11B	0.99
C2—C3	1.5387 (15)	C12—C13	1.5118 (14)
C2—H2A	0.99	C12—H12A	0.99
C2—H2B	0.99	C12—H12B	0.99
C3—C4	1.5140 (14)	C13—C18	1.3898 (14)
C3—H3A	0.99	C13—C14	1.3967 (14)
C3—H3B	0.99	C14—C15	1.3924 (15)
C4—C9	1.3905 (14)	C14—H14	0.95
C4—C5	1.4000 (14)	C15—C16	1.3892 (16)
C5—C6	1.3904 (16)	C15—H15	0.95
C5—H5	0.95	C16—C17	1.3764 (15)
C6—C7	1.3898 (17)	C17—C18	1.3946 (14)
C6—H6	0.95	C17—H17	0.95
F1···F1^{i}	3.1788 (16)	F2···F2^{ii}	3.2490 (16)
C7—F1—F1^{i}	145.61 (8)	C8—C9—C1	127.39 (9)
C16—F2—F2^{ii}	94.04 (6)	O2—C10—C18	126.26 (10)
O1—C1—C9	125.92 (9)	O2—C10—C11	126.27 (10)
O1—C1—C2	126.55 (9)	C18—C10—C11	107.46 (9)
C9—C1—C2	107.53 (8)	C10—C11—C12	106.29 (8)
C1—C2—C3	106.56 (8)	C10—C11—H11A	110.5
C1—C2—H2A	110.4	C12—C11—H11A	110.5
C3—C2—H2A	110.4	C10—C11—H11B	110.5
C1—C2—H2B	110.4	C12—C11—H11B	110.5
C3—C2—H2B	110.4	H11A—C11—H11B	108.7
H2A—C2—H2B	108.6	C13—C12—C11	104.47 (8)
C4—C3—C2	104.43 (8)	C13—C12—H12A	110.9
C4—C3—H3A	110.9	C11—C12—H12A	110.9
C2—C3—H3A	110.9	C13—C12—H12B	110.9
C4—C3—H3B	110.9	C11—C12—H12B	110.9
C2—C3—H3B	110.9	H12A—C12—H12B	108.9
H3A—C3—H3B	108.9	C18—C13—C14	119.66 (9)
C9—C4—C5	119.36 (10)	C18—C13—C12	111.56 (9)
C9—C4—C3	111.52 (9)	C14—C13—C12	128.76 (10)
C5—C4—C3	129.12 (10)	C15—C14—C13	118.83 (10)
C6—C5—C4	118.94 (10)	C15—C14—H14	120.6
C6—C5—H5	120.5	C13—C14—H14	120.6
C4—C5—H5	120.5	C16—C15—C14	119.39 (9)

C7—C6—C5	119.55 (10)	C16—C15—H15	120.3
C7—C6—H6	120.2	C14—C15—H15	120.3
C5—C6—H6	120.2	F2—C16—C17	118.56 (10)
F1—C7—C8	118.46 (10)	F2—C16—C15	117.94 (9)
F1—C7—C6	118.21 (10)	C17—C16—C15	123.51 (9)
C8—C7—C6	123.32 (10)	C16—C17—C18	115.98 (10)
C7—C8—C9	116.05 (10)	C16—C17—H17	122.0
C7—C8—H8	122.0	C18—C17—H17	122.0
C9—C8—H8	122.0	C13—C18—C17	122.62 (9)
C4—C9—C8	122.78 (9)	C13—C18—C10	109.79 (9)
C4—C9—C1	109.83 (9)	C17—C18—C10	127.58 (10)
O2—C10—C18—C17	−3.94 (18)	C3—C4—C5—C6	−179.93 (10)
O2—C10—C18—C13	174.94 (11)	C2—C3—C4—C9	−2.21 (11)
O2—C10—C11—C12	−173.07 (11)	C2—C3—C4—C5	177.79 (10)
O1—C1—C9—C8	1.81 (17)	C2—C1—C9—C8	−177.56 (10)
O1—C1—C9—C4	−178.32 (10)	C2—C1—C9—C4	2.31 (11)
O1—C1—C2—C3	177.04 (10)	C1—C2—C3—C4	3.48 (11)
F2^{ii}—F2—C16—C17	−142.91 (9)	C18—C13—C14—C15	−0.21 (15)
F2^{ii}—F2—C16—C15	37.18 (10)	C18—C10—C11—C12	6.57 (11)
F2—C16—C17—C18	−179.74 (9)	C16—C17—C18—C13	−0.78 (15)
F1^{i}—F1—C7—C8	−1.14 (19)	C16—C17—C18—C10	177.97 (10)
F1^{i}—F1—C7—C6	179.25 (9)	C15—C16—C17—C18	0.17 (16)
F1—C7—C8—C9	−179.49 (9)	C14—C15—C16—F2	−179.69 (9)
C9—C4—C5—C6	0.07 (15)	C14—C15—C16—C17	0.40 (16)
C9—C1—C2—C3	−3.60 (11)	C14—C13—C18—C17	0.82 (15)
C7—C8—C9—C4	0.06 (15)	C14—C13—C18—C10	−178.13 (9)
C7—C8—C9—C1	179.91 (10)	C13—C14—C15—C16	−0.38 (15)
C6—C7—C8—C9	0.10 (16)	C12—C13—C18—C17	179.76 (9)
C5—C6—C7—F1	179.42 (9)	C12—C13—C18—C10	0.81 (12)
C5—C6—C7—C8	−0.16 (17)	C12—C13—C14—C15	−178.95 (10)
C5—C4—C9—C8	−0.14 (15)	C11—C12—C13—C18	3.30 (11)
C5—C4—C9—C1	179.98 (9)	C11—C12—C13—C14	−177.87 (10)
C4—C5—C6—C7	0.07 (16)	C11—C10—C18—C17	176.42 (10)
C3—C4—C9—C8	179.86 (9)	C11—C10—C18—C13	−4.70 (12)
C3—C4—C9—C1	−0.02 (12)	C10—C11—C12—C13	−5.93 (11)

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*; (ii) −*x*, −*y*+1, −*z*.

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C5—H5···O1^{iii}	0.95	2.47	3.3873 (14)	161
C14—H14···O2^{iv}	0.95	2.65	3.5107 (14)	150
C2—H2*B*···F2^{v}	0.99	2.46	3.2062 (13)	132
C6—H6···O2^{vi}	0.95	2.65	3.5338 (14)	154

C11—H11*B*···O1^{vii}	0.99	2.52	3.3348 (13)	140
C15—H15···F1^{i}	0.95	2.52	3.3664 (13)	148

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*; (iii) −*x*+3/2, *y*+1/2, −*z*+1/2; (iv) −*x*+1/2, *y*−1/2, −*z*+1/2; (v) *x*+1/2, −*y*+3/2, *z*+1/2; (vi) *x*+1/2, −*y*+3/2, *z*−1/2; (vii) −*x*+1, −*y*+1, −*z*+1.