The use of modified and non-natural nucleotides provide unique insights into pro-mutagenic replication catalyzed by polymerase eta

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ABSTRACT

This report evaluates the pro-mutagenic behavior of 8-oxo-guanine (8-oxo-G) by quantifying the ability of high-fidelity and specialized DNA polymerases to incorporate natural and modified nucleotides opposite this lesion. Although high-fidelity DNA polymerases such as pol δ and the bacteriophage T4 DNA polymerase replicating 8-oxo-G in an error-prone manner, they display remarkably low efficiencies for TLS compared to normal DNA synthesis. In contrast, pol n shows a combination of high efficiency and low fidelity when replicating 8-oxo-G. These combined properties are consistent with a pro-mutagenic role for pol η when replicating this DNA lesion. Studies using modified nucleotide analogs show that pol nrelies heavily on hydrogen-bonding interactions during translesion DNA synthesis. However, nucleobase modifications such as alkylation to the N2 position of guanine significantly increase error-prone synthesis catalyzed by pol η when replicating 8-oxo-G. Molecular modeling studies demonstrate the existence of a hydrophobic pocket in pol n that participates in the increased utilization of certain hydrophobic nucleotides. A model is proposed for enhanced promutagenic replication catalyzed by pol n that couples efficient incorporation of damaged nucleotides opposite oxidized DNA lesions created by reactive oxygen species. The biological implications of this model toward increasing mutagenic events in lung cancer are discussed.

INTRODUCTION

DNA replication is the biological process in which an organism's genome is copied by the action of at least one DNA polymerase. The vast majority of DNA polymerases use the coding information present on the templating strand of DNA (or RNA) to guide each nucleotide incorporation event. The efficiency and fidelity of this process are often attributed to the formation of complementary hydrogenbonding interactions that define a correct base pair (1,2). The mutual recognition of adenine (A) by thymine (T) and of guanine (G) by cytosine (C) involves complementarity in shape, size and hydrogen-bonding interactions made between each base pair. Using these interactions, most replicative DNA polymerases synthesize DNA with an incredible degree of fidelity, making only 1 mistake every million opportunities (3) at remarkably high speeds of >500 bp/s (4).

While the molecular mechanism of normal DNA synthesis is well understood, our knowledge of how DNA polymerases replicate damaged DNA is far less defined. Much of this deficiency arises from two major complications. These first involves the diversity of DNA lesions formed inside the cell while the second reflects the number of DNA polymerases that can replicate each lesion. For example, there are over a hundred different DNA lesions that can form within a cell (5), and many of these lesions can drastically influence DNA polymerase activity (6). For instance, lesions which significantly alter the proper conformation of the DNA template such as abasic sites and thymine dimers are strong replication blocks for most highfidelity DNA polymerases (7-9). If left unrepaired, these types of lesions would constantly interrupt DNA synthesis catalyzed by high-fidelity DNA polymerases and produce devastating effects on cell proliferation and viability. To avoid catastrophic failures caused by damaged DNA, prokaryotic and eukaryotic cells possess several 'specialized' DNA polymerases that can efficiently replicate vari-

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ous DNA lesions. The ability of DNA polymerases to bypass damaged DNA is a process termed translesion DNA synthesis (TLS) and can result in error-free or error-prone replication. Although error-prone activity during TLS can reduce genomic fidelity, it is essential as most cells would die without the ability of specialized DNA polymerases to efficiently replicate unrepaired DNA lesions.

However, not all forms of DNA damage produce such dire effects on the efficiency of DNA replication. In fact, many DNA lesions have simple alterations to the hydrogen-bonding potential of the nucleobase that can enhance pro-mutagenic replication (10-12). At the cellular level, this increase in mutagenic DNA synthesis is a hallmark of several hyperproliferative diseases such as cancer. Indeed, lung cancer represents an important example as the oxygen-rich environment of the lung can produce reactive oxygen species (ROS) that can generate promutagenic DNA lesions such as 2,6-diamino-4-hydroxy-5formamidopyrimidine (faPyG) and 8-oxo-guanine (8-oxo-G) (Figure 1A). Since both lesions are miscoding, they can be replicated in an error-free or error-prone manner (13,14). For example, 8-oxo-G can be replicated incorrectly via the incorporation of adenosine-2'-deoxyriboside monophosphate (dAMP), leading to an overall increase in G:C to T:A transversion mutations. At the molecular level, the misreplication of 8-oxo-G depends upon the svn- versus anti-conformation of the oxidized nucleobase (Figure 1B). Error-free replication occurs when cytosine-2'-deoxyriboside monophosphate (dCMP) is inserted opposite the anti-conformation of 8-oxo-G. In contrast, promutagenic synthesis occurs when dAMP insertion opposite the syn-conformation of 8-oxo-G which resembles thymine with respect to hydrogen-bonding interactions. A number of laboratories have performed kinetic and structural studies with different DNA polymerases to define the replication of 8-oxo-G (15–22). In general, the results from these studies demonstrate that the efficiency and selectivity for inserting dCMP versus dAMP opposite 8-oxoG varies significantly depending on which DNA polymerase is used as well as the context of DNA sequence. For example, studies with high-fidelity DNA polymerases from different sources demonstrate an unusual preference for misinserting dAMP opposite the oxidized DNA lesion compared to the correct partner, dCMP (15–17). In contrast, studies with various specialized DNA polymerases have yielded conflicting results. For example, Dpo4, the error prone DNA polymerase from Sulfolobus solfataricus is reported to catalyze errorfree replication of 8-oxo-G (18). In contrast, human pol κ is more proficient at catalyzing error-prone synthesis when replication 8-oxo-G (19). In addition, several studies have compared the activity of pol η isolated from yeast or human cells. Carlson and Washington demonstrated that pol η can efficiently incorporate dAMP opposite 8-oxo-G (20). Likewise, McCulloch *et al.* showed that yeast pol η replicates 8-oxo-G more accurately than its high-fidelity counterpart, pol δ (21). Finally, Patra *et al.* reported that human pol n replicates 8-oxo-G in a predominantly error-free manner by preferentially inserting dCMP (22).

To better understand the molecular mechanisms of TLS and induced-mutagenesis, we compared the abilities of high-fidelity and specialized DNA polymerases to replicate 8-oxo-G. Our quantitative analyses demonstrate that the incredible efficiency of high-fidelity DNA polymerases for replicating normal DNA is reduced ~1000-fold when replicating the oxidized lesion, 8-oxo-G. This reduction in efficiency suggests that these polymerases are unlikely to replicate this miscoding lesion under physiological conditions. Consistent with this model, the specialized DNA polymerase, pol n, replicates 8-oxo-G approximately 100-fold more efficiently than either high-fidelity DNA polymerase. In addition, pol m misinserts dAMP opposite 8-oxo-G with a catalytic efficiency comparable to dCMP insertion. The combination of high efficiency and promiscuity for misinserting dAMP likely contributes to the mutagenic potential of the oxidized DNA lesion. Furthermore, we identified a specific nucleotide, N^2 -MedGTP, which is unique in its utilization and selectivity by pol η when replicating 8-oxo-G. This result suggests that inappropriate modifications to DNA and nucleotide pools can cause a synergistic increase in pro-mutagenic DNA synthesis. The biological ramifications of these findings are discussed in the context of disease models such as lung cancer in which increased levels of ROS can enhance disease development and progression.

MATERIALS AND METHODS

Materials

 $[\gamma - {}^{32}P]$ -ATP was purchased from MP Biomedical (Irvine, CA, USA). Unlabeled deoxynucleoside triphosphate (dNTPs) (ultrapure) were obtained from Pharmacia. Unlabeled dNTPs (ultrapure) were obtained from Pharmacia. Magnesium acetate, magnesium chloride and Trizma base were from Sigma. Urea, acrylamide and bis-acrylamide were from National Diagnostics (Rochester, NY, USA). Oligonucleotides, including those containing 8-oxo-guanine, were synthesized by Operon Technologies (Alameda, CA, USA). All modified nucleotides including N⁶-MedATP, 6-Cl-PTP, 6-Cl-2APTP, O⁶-MedGTP, N²-MedGTP, 2-6-dATP, dITP and 8-oxo-dGTP were obtained from Trilink Biotechnologies (San Diego, CA, USA). All non-natural nucleotides including IndTP. 5-MeITP. 5-Et-ITP, 5-EvITP, 5-NITP, 4-NITP and 6-NITP were synthesized and purified as previously reported (23-26). All other materials were obtained from commercial sources and were of the highest available quality. Exonucleasedeficient bacteriophage T4 DNA polymerase (gp43exo⁻ (Asp-219 to Ala mutation)) was purified and quantified as previously described (27). Recombinant human polymerase delta (pol δ) and human polymerase eta (pol η) were purified as previously described (28,29). All DNA polymerases used in this study were >97% pure as assessed by sodium dodecylsulphate-polyacrylamide denaturing gel electrophoresis.

Kinetic parameters for nucleotide incorporation

Kinetic studies using the bacteriophage T4 DNA were performed using an assay buffer consisting of TrisOAc (25 mM pH 7.5), KOAc (150 mM), DTT (10 mM) and 10 mM MgAcetate at pH 7.5. Kinetic studies using polymerase δ and η were performed using an assay buffer consisting of TrisOAc (50 mM pH 7.5), bovine serum albumin (1 mg/ml),



Figure 1. (A) Overview describing the ability of reactive oxygen species (ROS) to modify natural nucleobases such as guanine to generate miscoding DNA lesions such as 8-oxo-guanine (8-oxo-G). (B) The replication of 8-oxo-guanine can be error-free via dCMP insertion or error-prone via dAMP insertion.

DTT (10 mM) and 5 mM MgCl₂ at pH 7.5. All assays were performed at 25°C unless otherwise noted. The kinetic parameters $(k_{\text{cat}}, K_{\text{m}} \text{ and } k_{\text{cat}}/K_{\text{m}})$ for nucleotides were measured as previously described (30). Briefly, a typical assay was performed by pre-incubating DNA substrate (500 nM) with a limiting amount of polymerase (10 or 25 nM) in assay buffer and Mg²⁺. Reactions were initiated by adding variable concentrations of nucleotide substrate (1–500 μ M). At variable time intervals, 5 μ l aliquots of the reaction were quenched by adding an equal volume of 200 mM ethylenediaminetetraacetic acid (EDTA). Polymerization reactions were monitored by analyzing products on 20% sequencing gels as previously described (29). Gel images were obtained with a Packard PhosphorImager by using the OptiQuant software supplied by the manufacturer. Product formation was quantified by measuring the ratio of ³²P-labeled extended and un-extended primer. The ratios of product formation are corrected for substrate in the absence of polymerase (zero point). Corrected ratios are then multiplied by the concentration of primer/template used in each assay to yield total product. Steady-state rates were obtained from the linear portion of the time course and were fit to Equation(1)

$$y = \mathbf{m}\mathbf{x} + \mathbf{b} \tag{1}$$

where m is the slope of the line corresponding to the rate of the polymerization reaction (nM s⁻¹), b is the *y*-intercept and *t* is time. Data for the dependency of rate as a function of nucleotide concentration were fit to the Michaelis–Menten equation (Equation 2):

$$\nu = V_{\text{max}} * [\text{dNTP}] / (K_{\text{m}} + [\text{dNTP}])$$
(2)

where ν is the rate of product formation (nM s⁻¹), V_{max} is the maximal rate of polymerization, K_{m} is the Michaelis constant for dXTP and [dXTP] is the concentration of nucleotide substrate. The turnover number, k_{cat} , is V_{max} divided by the final concentration of polymerase used in the experiment.

Extension beyond 8-oxo-guanine

Single turnover conditions were used to measure extension beyond various mispairs containing 8-oxo-G in the template. Pol η (500 nM) was incubated with 250 nM DNA (13/200G-mer) in assay buffer and mixed with a fixed concentration of nucleotide substrate and 5 mM MgCl₂. In all cases, the concentration of nucleotide was maintained at its respective $K_{\rm m}$ value. An aliquot of the reaction was quenched with 500 mM EDTA after 60 or 120 s (time to achieve 4 half-lives as calculated from k_{cat} values). This time point represents insertion opposite the DNA lesion. An aliquot containing 200 µM of dTTP and dGTP was then added to initiate elongation beyond the formed mispair. After an additional 60 s, an aliquot of the reaction was quenched. Products of the polymerization reactions were processed and analyzed as described above to define the efficiency of elongation.

Molecular modeling studies

Crystal structures containing putative structural models of pol η in complex with DNA template containing 8-oxo-G were obtained from the Research Collaboratory for Structural Bioinformatics Protein Data Bank (www.pdb.org).

The structural models of pol η containing DNA_{8-og}-dNTP complex were made from the co-crystal structure of pol η -DNA_{8-og}-dATP/dCTP (PDB ID and 403R/403S) (22). Pol η is structurally and functionally homologous in both the crystal structures.

Using the molecular modeling program, *Molecular Operating Environment*, modified dNTP structures for N²-MedGMP and 6-NIMP were created by modifying dAMP in the crystal structure PDB ID: 403R. Following the modifications, the active site of pol η was minimized using MOE modeling software (MOE 2014). The molecular modeling program *Chimera* was used to determine interglycosyl distances and distance between the templating base and active amino residues using the modified structural model of Pol η -DNA_{8-og}-N²MeGMP/6-NIMP. Hydrophobicity surface models were generated for the polymerase containing DNA and dNTP using the Kyle Doolittle hydrophobicity scale. The scale of hydrophobicity is represented in decreasing order as Red>White>Blue.

RESULTS

Comparison of normal and translession DNA synthesis activity catalyzed by high-fidelity and specialized DNA polymerases

This study begins by comparing the ability of two high fidelity DNA polymerases (bacteriophage T4 DNA polymerase (gp43 exo⁻) and human pol δ) and a specialized DNA polymerase (human pol η) to incorporate natural dNTPs opposite normal and damaged nucleobases. Defined DNA substrates containing G, C or 8-oxo-G at position 14 of the template (Figure 2A) were used to accurately quantify the activity of each DNA polymerase. Denaturing gel electrophoresis images provided in Figure 2B illustrate the ability of these DNA polymerases to perform normal synthesis (dCMP insertion opposite G), incorrect synthesis (dAMP insertion opposite G) and TLS (dCMP or dAMP insertion opposite 8-oxo-G). During normal replication, both high-fidelity polymerases are remarkably faithful as they insert dCMP opposite G much more efficiently than dAMP. However, gp43 exo⁻ and pol δ exhibit pro-mutagenic tendencies by efficiently incorporating dAMP opposite 8-oxo-G. In contrast, the specialized polymerase, pol n, displays less fidelity when replicating undamaged DNA, a results that is consistent with previous reports (31,32). Furthermore, pol η displays more promiscuous behavior when replicating 8-oxo-G as dATP and dCTP are utilized with nearly equal efficiencies. This result is consistent with work performed by Carlson and Washington (20) and McCulloch *et al.* (21) showing that pol η utilizes dATP with high efficiency when replicating 8-oxo-G.

To more accurately assess the kinetic behavior of these polymerases, k_{cat} , K_m and k_{cat}/K_m values were measured for the insertion of dCTP and dATP opposite G and 8-oxo-G. Figure 2C provides Michaelis–Menten plots for dCMP incorporation opposite G while Figure 2D provides Michaelis–Menten plots for dCMP incorporation opposite 8-oxo-G catalyzed by gp43 exo⁻. The kinetic parameters for gp43 exo⁻ during normal and TLS are summarized in Table 1. As expected, gp43 exo⁻ displays a high catalytic efficiency of $1.7 * 10^5 \text{ M}^{-1}\text{s}^{-1}$ during normal DNA synthesis, and this

is achieved by a relatively low $K_{\rm m}$ value of 4.7 μ M for dCTP and a fast $k_{\rm cat}$ value of 0.94 s⁻¹. In contrast, the catalytic efficiency for replicating 8-oxo-G decreases by ~20-fold, and this reflects a 10-fold increase in the $K_{\rm m}$ for dCTP coupled with a 2-fold reduction in $k_{\rm cat}$. Surprisingly, the efficiency for dAMP insertion opposite 8-oxo-G is only 4-fold lower than dCMP (compare $k_{\rm cat}/K_{\rm m}$ values of 3000 M⁻¹s⁻¹ for dATP versus 11 000 M⁻¹s⁻¹ for dCTP). This effect is caused exclusively by a large increase in the $K_{\rm m}$ value for dATP (compare $K_{\rm m}$ values of 54 versus 250 μ M for dCTP and dATP, respectively).

Identical experiments performed using pol δ , the highfidelity human DNA polymerase, yield the kinetic parameters for normal and TLS summarized in Table 1. These data indicate that pol δ is also highly proficient during normal DNA synthesis, displaying a low $K_{\rm m}$ of 0.64 μ M and a fast k_{cat} of 1 s⁻¹ for incorporating dCMP opposite G. Similar to gp43 exo⁻, the efficiency for replicating 8-oxo-G is reduced significantly as the $K_{\rm m}$ for dCTP and $k_{\rm cat}$ are both adversely affected. In particular, the $K_{\rm m}$ for dCTP is 50-fold higher than that measured during normal DNA synthesis while the k_{cat} value is reduced ~30-fold when replicating the oxidized lesion. This later result indicates that the ratelimiting step for polymerase turnover is sensitive to the nature of the DNA lesion, and this represents an interesting difference between pol δ and gp43exo⁻ when replicating 8oxo-G that is currently being investigated.

The kinetic parameters for pol η are considerably different than both high-fidelity DNA polymerases. As summarized in Table 1, the $K_{\rm m}$ value for dCTP is only 2-fold higher when replicating the oxidized DNA leison (compare 1.3 versus 3.0 µM for G and 8-oxo-G, respectively). Likewise, the k_{cat} value during TLS is only 2-fold lower than for replicating normal DNA (compare 0.40 s⁻¹ versus 0.88 s⁻¹ for 8oxo-G and G, respectively). Collectively, these data demonstrate that pol n misinserts dAMP opposite 8-oxo-G with a high catalytic efficiency of 24 000 $M^{-1}s^{-1}$. In fact, the value measured for pol η is 60-fold higher than that measured for the efficiency of pol δ misinserting dAMP opposite 8-oxo-G. Since catalytic efficiencies are often used as predictors for enzyme activity *in vivo*, our results suggest that pol η is more proficient at catalyzing error-prone replication of 8oxo-G compared to pol δ under cellular conditions.

Quantifying the incorporation of modified nucleotide analogs by Pol $\boldsymbol{\eta}$

Several models have been proposed to explain why pol η displays promiscuous and low replicative fidelity, especially when replicating normal or undamaged DNA (33–37). One generalized model invokes the contributions of hydrogen-bonding interactions made between the incoming nucleotide with the templating nucleobase and/or with amino acids residing within the polymerase's active site (33–37). An alternative model suggests that pol η possesses a 'loose' active site that is larger than most high-fidelity DNA polymerases and thus capable of better accommodating DNA lesions (38). The concept that the relative 'tightness' of the active site of a DNA polymerase was predicated on work initially described by Kool *et al.* demonstrating that base analogs such as 2,4-difluorotoluene which lack classi-



Figure 2. (A) DNA substrate used in these studies. X in the template at position 14 denotes G, C or 8-oxo-G. (**B**) Denaturing gel electrophoresis images comparing dCMP insertion opposite G (13/20G) and the insertion of dCMP and dAMP insertion opposite 8-oxo-G (13/20OG) by the bacteriophage T4 DNA polymerase, pol δ and pol η . (**C**) Michaelis–Menten plots for dCMP incorporation opposite G catalyzed by the bacteriophage T4 polymerase. (**D**) Michaelis–Menten plots for dCMP incorporation opposite 8-oxo-G catalyzed by the bacteriophage T4 polymerase.

Table 1. Summary of kinetic constants for the incorporation of natural nucleotides opposite guanine and 8-oxo-guanine by the bacteriophage T4 DNA polymerase (gp43 exo⁻), pol δ and pol η

DNA	dNTP	$K_{\rm m}$ (μM)	$k_{\rm cat} ({\rm s}^{-1})$	$k_{\text{cat}/}K_{\text{m}} (\mathrm{M}^{-1}\mathrm{s}^{-1})$
T4 DNA polymerase				
13/20G	dCTP	4.7 ± 0.1	0.94 ± 0.02	$(1.7 \pm 0.1)^* 10^5$
13/20G	dATP	ND	ND	ND
13/20OG	dCTP	54 ± 16	0.59 ± 0.07	$(1.1 \pm 0.4)^* 10^4$
13/20OG	dATP	250 ± 40	0.76 ± 0.13	$(3.0 \pm 0.3)^* 10^3$
Pol δ				
13/20G	dCTP	0.64 ± 0.07	1.03 ± 0.02	$(1.6 \pm 0.5)^* 10^6$
13/20G	dATP	ND	ND	ND
13/20OG	dCTP	34 ± 5	0.033 ± 0.002	$(9.8 \pm 0.4)^* 10^3$
13/20OG	dATP	258 ± 14	0.097 ± 0.003	$(0.4 \pm 0.1)^* 10^3$
Pol η				
13/20G	dCTP	1.3 ± 0.3	0.88 ± 0.10	$(6.7 \pm 0.3)^* 10^5$
13/20G	dATP	49 ± 13	1.11 ± 0.08	$(2.2 \pm 0.4)^{*10^4}$
13/20OG	dCTP	3.0 ± 0.6	0.40 ± 0.02	$(1.3 \pm 0.1)^* 10^5$
13/20OG	dATP	47 ± 7	1.13 ± 0.06	$(2.4 \pm 0.3)^* 10^4$

cal hydrogen-bonding groups are efficiently and selectively incorporated opposite natural nucleobases such as adenine (39). This suggested that the shape of the formed base pair rather than H-bonding interactions serve as the primary source for nucleotide selection and incorporation by DNA polymerases.

To further investigate these mechanisms, we quantified the incorporation of several distinct purine nucleoside triphosphates that contain alterations to hydrogenbonding functional groups (Figure 3A). We classified these modified nucleotides into three distinct groups. The first includes alkylated nucleotides such as N⁶-MedGTP, O⁶-MedGTP and N²-MedGTP that were used to evaluate the contributions of hydrogen-bonding interactions, shape complementarity/steric fit and nucleobase hydrophobicity. The second group includes analogs in which a hydrogen-bonding group is replaced with a halogen (6Cl-dATP and 6-Cl-2APTP). These substitutions provide a way to more accurately evaluate the contributions of shape complementarity/steric fit and nucleobase hydrophobicity. The final group are purines in which hydrogen-bonding groups are removed (dITP) or added (2,6-dATP) to more directly interrogate the contributions of hydrogen-bonding interactions.

Figure 3B provides representative Michaelis–Menten plots comparing the utilization of dATP with N⁶-MedATP and 6-Cl-PTP when replicating 8-oxo-G. Surprisingly, alkylation at the N6-position of dATP actually causes a slight increase in the catalytic efficiency for insertion opposite 8oxo-G compared to the natural substrate, dATP. In this case, the $K_{\rm m}$ of 46 μ M for N⁶-MedATP is identical to dATP ($K_{\rm m} = 47 \ \mu$ M) whereas the $k_{\rm cat}$ value for N⁶-MedATP is increased slightly (compare 1.35 s⁻¹ versus 1.13 s⁻¹ for N⁶-MedATP and dATP, respectively). At face value, this result



Figure 3. (A) Structures of modified purine nucleotides used in this study. (B) Michaelis–Menten plots comparing the utilization of dATP (\bigcirc), N⁶-MedATP (\bullet) and 6-Cl-PTP (\blacksquare) during the replication of 8-oxo- G catalyzed by polymerase η . (C) Gel electrophoresis data comparing the incorporation of modified natural nucleotides opposite G and 8-oxo-G catalyzed by pol eta. G represents guanine while OG represents 8-oxoguanine. (D) Michaelis–Menten plots for the insertion of N²-MedGTP opposite G (\bigcirc) and 8-oxo-G (\bullet) catalyzed by polymerase eta.

suggests that altering hydrogen-bonding interactions has a minimal effect of the activity of pol n. However, data obtained with the halogenated analog, 6-Cl-PTP, suggests otherwise. In fact, replacing a hydrogen-bonding group (-NH₂) with a halogen (non-hydrogen bonding group) lowers the overall catalytic efficiency by producing negative effects on both $K_{\rm m}$ and $k_{\rm cat}$. As summarized in Table 2, the reduced efficiency for utilizing 6-Cl-PTP is caused by a 3-fold increase in $K_{\rm m}$ coupled with a 3-fold decrease in $k_{\rm cat}$. Similar results are obtained for the majority of nucleotide analogs tested here as modification to one or more hydrogen-bonding groups present on the incoming nucleotide causes a reduction in incorporation efficiencies (Table 2). For example, the catalytic efficiencies for 6-Cl-2APTP and O⁶-MedGTP are lower than dATP for insertion opposite 8-oxo-G, suggesting that hydrogen bonding groups are indeed needed to optimize pol n activity when replicating 8-oxo-G. However, the position of the hydrogen bonding group also appears critical as analogs such as 2,6-dATP and ITP have lower incorporation efficiencies compared to dATP. Indeed, 8-oxodGTP displays the lowest activity amongst the substrates tested here as its catalytic efficiency is 50-fold lower than dATP and 16-fold lower than dGTP. This reduction is likely caused by alterations in the anti-versus syn-conformation of the nucleotide (40).

Surprisingly, there is one remarkable instance in which modifications to a hydrogen-bonding group actually causes a substantial increase in nucleotide utilization during the replication of 8-oxo-G. Specifically, N²-MedGTP is utilized \sim 10- and 20-fold more effectively than dATP and dGTP, respectively. This increase in utilization occurs via a large

enhancement in the $K_{\rm m}$ value for N²-MedGTP rather than an effect on $k_{\rm cat}$. In this case, the $K_{\rm m}$ of 0.64 μ M for N²-MedGTP is 70-fold lower that dATP ($K_{\rm m} = 47 \,\mu$ M) whereas the $k_{\rm cat}$ value for N²-MedGTP is 10-fold lower (compare 0.11 s⁻¹ versus 1.13 s⁻¹ for N²-MedGTP and dATP. This represents an unusual case as the $K_{\rm m}$ values for the majority of modified nucleotides tested here are either identical or are slightly higher than dATP. In fact, the $K_{\rm m}$ value for most analogs tested here remains fairly constant at ~50 μ M while the $k_{\rm cat}$ value is the kinetic parameter that varies more significantly amongst these modified analogs.

To verify that these effects are not limited to TLS, we also measured if these modifications affect the efficiency of replicating undamaged DNA. As illustrated in Figure 3C, as pol n inserts most modified nucleotides opposite 8-oxo-G much more efficiently than opposite unmodified G. To verify that the preferential utilization occurs only during TLS, we measured the kinetic parameters for their incorporation opposite G. As summarized in Table 2, the catalytic efficiencies for most modified analogs are \sim 10- to 100-fold lower during normal synthesis compared to the replication of 8-oxo-G. In most cases, these reductions are caused by reductions in k_{cat} rather than through perturbations in the $K_{\rm m}$. Collectively, these kinetic data indicate that modifications to a natural dNTP can increase the efficiency in the error-prone replication of 8-oxo-G catalyzed by pol n. However, these modifications do not significantly affect the fidelity of pol η when replicating undamaged DNA. Again, N2-MedGTP represents the best example of this phenomenon as it displays a \sim 40-fold higher selectivity for insertion opposite 8-oxo-G than G (Figure 3D and Table 2). The greater selectivity of

			koot / Km		
Nucleotide	$K_{\rm m}$ ($\mu { m M}$)	k_{cat} (s ⁻¹)	$(M^{-1}s^{-1})$	Efficiency ^a	Selectivity ^b
dATP	47 ± 7	1.13 ± 0.06	$(2.4 \pm 0.3)^*10^4$	100%	1.14
N ⁶ -MedATP	46 ± 15	1.35 ± 0.15	$(2.9 \pm 0.2)^* 10^4$	120%	2.23
6-Cl-PTP	137 ± 7	0.39 ± 0.08	$(2.8 \pm 0.4)^* 10^3$	12%	8.48
6-Cl-2APTP	14.3 ± 3.2	0.09 ± 0.01	$(6.3 \pm 0.6)^* 10^3$	26%	1.26
dGTP	32 ± 10	0.25 ± 0.03	$(7.8 \pm 0.8)^* 10^3$	33%	1.04
O ⁶ -MedGTP	50 ± 15	0.55 ± 0.13	$(1.1 \pm 1.0)^* 10^4$	46%	17.2
N ² -MedGTP	0.64 ± 0.05	0.11 ± 0.02	$(1.7 \pm 0.1)^* 10^5$	720%	41
2,6-dATP	72 ± 19	0.31 ± 0.01	$(4.3 \pm 0.5)^* 10^3$	18%	11.9
dITP	64 ± 25	0.17 ± 0.03	$(2.7 \pm 0.9)*10^3$	11%	5.6
8-oxo-dGTP	143 ± 36	0.068 ± 0.005	$(4.8 \pm 0.8)^* 10^2$	2%	2.5
13/20G					
dATP	49 ± 13	1.11 ± 0.08	$(2.2 \pm 0.4)^* 10^4$	100%	
N ⁶ -MedATP	70 ± 30	0.88 ± 0.01	$(1.3 \pm 0.5)*10^3$	6%	
6-Cl-PTP	77 ± 16	0.025 ± 0.001	$(3.3 \pm 0.9)^*10^2$	1.5%	
6-Cl-2APTP	15.9 ± 3.9	0.08 ± 0.01	$(5.0 \pm 0.8)^* 10^3$	23%	
dGTP	2.4 ± 0.9	0.018 ± 0.001	$(7.5 \pm 0.9)^* 10^3$	34%	
O ⁶ -Me-dGTP	150 ± 70	0.096 ± 0.017	$(6.4 \pm 0.8)^* 10^2$	2.9%	
N ² -Me-dGTP	5.5 ± 2.1	0.023 ± 0.004	$(4.2 \pm 0.8)^* 10^3$	19%	
2,6-dATP	76 ± 17	0.027 ± 0.002	$(3.6 \pm 0.8)^* 10^2$	1.6%	
dITP	25 ± 4	0.012 ± 0.001	$(4.8 \pm 06)^*10^2$	2.2%	
8-oxo-dGTP	183 ± 41	0.035 ± 0.003	$(1.9 \pm 0.5)^* 10^2$	0.9%	

Table 2. Summary of kinetic constants measured for the incorporation of natural and modified nucleotides opposite guanine (G) and 8-oxoguanine (8-oxo-G) by pol η

N²-MedGTP for damaged DNA, coupled with its 10-fold higher catalytic efficiency, highlights a pro-mutagenic role for pol η during the replication of damaged DNA.

Incorporation of 5-substituted indolyl nucleotide analogs by Pol $\boldsymbol{\eta}$

To further evaluate the role of hydrogen-bonding, shape complementarity and hydrophobicity, we quantified the incorporation of a series of 5-substituted indolyl-2'deoxynucleotide analogs depicted in Figure 4A. Data provided in Figure 4B shows the ability of pol η to incorporate these nucleoside analogs opposite 8-oxo-G. The kinetic parameters for these analogs are summarized in Table 3 and reveal that these 5-substituted indolyl-2'-deoxynucleotide analogs are poor substrates for pol η . Specifically, the catalytic efficiencies for these indolyl-based nucleotides are between 5- to 1000-fold lower than dATP. This decrease efficiency is caused primarily by reductions in k_{cat} rather than through significant perturbations in K_m. The effect on k_{cat} values again suggest that the rate-limiting step for pol n turnover is highly sensitive to proper hydrogen bonding interactions made between the templating nucleobase with the incoming dNTP. While these data suggest that hydrogen-bonding interactions are important for the polymerization cycle, they also point to a role for steric fit during TLS. This is evident as the overall size of the incoming nucleotide influences the catalytic efficiency of nucleotide utilization. Figure 4C provides molecular models highlighting how the overall shape and size of the incoming nucleobase can influence its kinetic behavior. For example, 5-MeITP, a good pairing partner for the syn-conformation of 8-oxo-G, possesses a $k_{\text{cat}}/K_{\text{m}}$ that is 3-fold higher than the smaller analog, IndTP. Furthermore, the $k_{\rm cat}/K_{\rm m}$ decreases as the size of the incoming nucleotide increases be-

yond an optimal interglycosyl distance of 10.6 Å for a normal Watson–Crick base pair. In particular, the k_{cat}/K_m for 5-EtITP is 10-fold lower than that for 5-MeITP while that for 5-EyITP is 100-fold lower than 5-MeITP. This later result is interesting as 5-EtITP and 5-EyITP are similar in size. Thus, the reduction in catalytic efficiency likely reflects the influence of entropy on nucleotide utilization by pol n. In this case, the conjugated double bond present on 5-EyITP reduces rotational freedom of the functional group, and this entropic penalty could diminishes its utilization as a substrate. Consistent with this mechanism, 5-NITP displays the highest catalytic efficiency amongst all 5-substituted indolyl-nucleotides tested here. We argue that rotational freedom of the nitro group improves its efficiency for incorporation. However, other features such as increased pi-electron density associated with the nitro moiety may also play a role as this has been previously demonstrated for high-fidelity DNA polymerases when replicating non-instructional DNA lesions (41,42).

The denaturing gel provided in Figure 4B also compares the incorporation of these non-natural nucleotides opposite G versus 8-oxo-G. Visual inspection reveals that most analogs are preferentially incorporated opposite 8-oxo-G as they are poorly inserted opposite G. The kinetic parameters summarized in Table 3 confirm that the efficiency for their incorporation opposite G is lower, thus indicating that they are also poor substrates for pol η during normal DNA synthesis.

The position of the nitro moiety influences the efficiency and selectivity for incorporation

Previous work in our lab demonstrated that the position of the nitro moiety on the indole-nucleotide influences the selectivity for utilization by high fidelity DNA polymerase

Nucleotide	$K_{\rm m}$ (μ M)	$k_{\rm cat} (^{-1})$	$k_{\rm cat}/K_{\rm m}~({\rm M}^{-1}{\rm s}^{-1})$	Efficiency	Selectivity
13/200G					
dÁTP	47 ± 7	1.13 ± 0.06	$24\ 000\pm 3000$	100%	1.09
IndTP	61 ± 18	0.050 ± 0.005	820 ± 120	3.4%	3.0
5-MeITP	18 ± 5	0.047 ± 0.001	2610 ± 200	10.9%	11.6
5-EtITP	36 ± 16	0.007 ± 0.001	200 ± 50	0.8%	0.24
5-EyITP	56 ± 36	0.0015 ± 0.0003	27 ± 15	0.1%	0.075
5-NITP	63 ± 12	0.344 ± 0.020	5460 ± 550	23%	0.30
4-NITP	23 ± 8	0.132 ± 0.016	5740 ± 350	24%	8.57
6-NITP	3.9 ± 1.6	0.283 ± 0.032	$72\ 600\pm1200$	303%	1.80
13/20G					
dATP	49 ± 13	1.11 ± 0.08	$22\ 000 \pm 4000$	100%	
IndTP	81 ± 18	0.022 ± 0.001	270 ± 80	1.2%	
5-MeITP	40 ± 15	0.009 ± 0.12	225 ± 80	1.0%	
5-EtITP	10 ± 3	0.008 ± 0.001	830 ± 100	3.8%	
5-EvITP	26 ± 5	0.009 ± 0.001	360 ± 50	1.6%	
5-NITP	46 ± 15	0.83 ± 0.12	$18\ 000\pm 1200$	82%	
4-NITP	57 ± 18	0.038 ± 0.009	670 ± 150	3.0%	
6-NITP	2.7 ± 0.6	0.108 ± 0.008	$40~000\pm1500$	180%	

Table 3. Summary of kinetic constants measured for the incorporation of non-natural nucleotides opposite guanine and 8-oxoguanine by polymerase η

during the replication of normal and damaged DNA (26). Here, we tested if pol η shows similar behavior by preferentially utilizing one of these nitro-containing nucleotides during normal or TLS.

The kinetic parameters for the utilization of 4-NITP, 5-NITP and 6-NITP by pol η are summarized in Table 3. Close inspection reveals that pol n utilizes 4-NITP with an identical catalytic efficiency compared to 5-NITP when replicating 8-oxo-G. However, the $K_{\rm m}$ for 4-NITP is ~2.5fold lower than 5-NITP while the k_{cat} for 4-NITP is 2.5fold lower than 5NITP. Surprisingly, 6-NITP is a very efficient substrate for pol η as the $k_{\text{cat}}/K_{\text{m}}$ for 6-NITP is 13-fold higher than either 4-NITP or 5-NITP and 3-fold greater than dATP. In all cases, the higher catalytic efficiency for 6-NITP is caused exclusively by a 15-fold decrease in its $K_{\rm m}$ value while the k_{cat} value is similar to the other substituted indolyl-nucleotides. While 6-NITP is highly efficient for insertion opposite 8-oxo-G, it displays only a 2-fold preference for the oxidized lesion compared to G. A more interesting effect is observed with 4-NITP as it displays a \sim 10fold higher selectivity for incorporation opposite 8-oxo-G compared to G. Much of this selectively arises as 4-NITP is poorly utilized by pol η when replicating undamaged DNA. Collectively, these data demonstrate that the efficiency and selectivity of substrate utilization by pol η can be influenced by permutations in the position of the nitro moiety.

Extension beyond 8-oxo-guanine

The mutagenic potential of a DNA lesion depends not only upon which nucleotide is misinserted opposite it but also by the ability of the polymerase to elongate beyond the formed mispair. As such, we explored the ability of pol η to extend beyond two natural nucleotides (dCMP and dAMP), three modified purines (N⁶-MedAMP, O⁶-MedGMP and N²-MedGMP), and one substituted indolyl nucleotide (6-NIMP) after they are incorporated opposite 8-oxo-G. These specific nucleotides were chosen as they are efficiently inserted opposite the oxidized lesion. Protocols outlined in Figure 5A were used to measure the ability of pol η to insert and extend various nucleotides opposite 8-oxo-G. Assays were performed using single turnover reaction conditions in order to maximize signal-to-noise ratios in product formation. Under these conditions, pol η (500 nM) and 13/20OG-mer (250 nM) were pre-incubated in assay buffer and then mixed with a fixed concentration of nucleotide (dNTP) and 5 mM Mg²⁺ to initiate insertion opposite the lesion. After four half-lives, an aliquot of dTTP and dGTP $(200 \,\mu M \text{ final concentration})$ was added to initiate the elongation reaction. Aliquots of the reactions were quenched with EDTA at variable times and analyzed by denaturing gel electrophoresis to assess elongation. Representative gel electrophoresis data provided in Figure 5B shows that pol η effectively performs pro-mutagenic DNA synthesis when replicating 8-oxo-G as it easily extends beyond either dCMP and dAMP when paired opposite 8-oxo-G. An interesting feature is that pol n can elongate one base beyond dAMP even in the absence of the next correct nucleotide. Furthermore, pol n easily extends beyond the dAMP mispair to yield full length product when supplied with the correct nucleotides needed for elongation.

Pol η also efficiently extends beyond most of the modified nucleotides substrates tested here when paired opposite 8-oxo-G. Similar to what was observed for the elongation of the dAMP:8-oxo-G mispair, pol n extends beyond these modified purines, even in the absence of the next correct nucleotide. However, slight differences are observed in the efficiency of elongation when supplied with nucleotides needed for complete elongation of the primer. For example, the halogenated analog, 6-Cl-PTP is elongated with a higher efficiency compared to either of the alkylated nucleotides, N⁶-MedATP and O⁶-MedGTP. Surprisingly, pol η incorporates 6-NITP opposite 8-oxo-G and subsequently extends it one base pair in the absence of complementary dNTPs. However, pol η poorly extends beyond the nonnatural nucleotide when supplied with dTTP and dGTP. This later result is consistent with previous reports demonstrating that high-fidelity DNA polymerases have severe difficulties in elongating substituted 5- and 6-nitroindolylnucleotides (26).



Figure 4. (A) Structures of non-natural nucleotides used in this study. (B) Gel electrophoresis data comparing the incorporation of non-natural nucleotides opposite G and 8-oxo-G catalyzed by pol η . (C) The top panel provides molecular models comparing the structures of basepairs between G : C, 8-oxo-G : C and 8-oxo-G : A. The bottom panel provides molecular models comparing the structures of basepairs between non-natural nucleotides paired opposite 8-oxo-G. The left panel provides models for 5-MeInd paired opposite 8-oxo-G (top and side view). The middle panel provides models for 5-EtInd paired opposite 8-oxo-G (top and side view). The right panel provides models for 5-EyInd paired opposite 8-oxo-G (top and side view).

Perhaps the most unusual behavior is again exhibited by N²-MedGTP as pol η efficiently inserts N²-MedGTP opposite 8-oxo-G and continues to elongate the mispair in the absence of complementary dNTP. This is highly unusual as pol η effectively misinserts N²-MedGMP opposite a templating A (position 15) before inserting N²-MedGTP opposite two templating C bases (positions 16 and 17). The ability of pol η to efficiently extend beyond 8-oxo-G again highlights the pro-mutagenic behavior of this alkylated nucleotide. It should be noted that this type of pro-mutagenic synthesis is highly selective for replication of the oxidized DNA lesion as pol η poorly incorporates and elongates N²-MedGMP when replicating undamaged G.

DISCUSSION

Mutagenesis is defined as the inappropriate change in an organism's genomic information. This process is influenced by a number of cellular factors including the number and diversity of DNA lesions that can form after exposure to DNA damaging agents. For example, hyperoxic conditions present in organs such as the lung can generate ROS which in turn can produce a number of distinctive DNA lesions. If left unrepaired, these lesions can be misreplicated by various DNA polymerases, and the resulting mutations that arise can significantly increase the risk of initiating cancer.

To better understand how misreplicating damaged DNA influences cancer initiation and progression, we evaluated the ability of high-fidelity and specialized DNA polymerases to replicate 8-oxo-G, a DNA lesion that is commonly formed by ROS. Our results validate that 8-oxo-G is a pro-mutagenic DNA lesion as dAMP is efficiently inserted opposite the lesion by several different DNA polymerases. In the case of high-fidelity DNA polymerases, the efficiency for inserting dAMP is only 5-fold lower than that for inserting dCMP, the correct pairing partner for 8-oxo-G. This reduced fidelity contrasts what is observed during the replication of undamaged DNA in which the efficiency for dCMP incorporation opposite G is 10⁶-fold higher than the misincorporation of dAMP. At face value, the reduced fidelity for replicating 8-oxo-G suggests that high fidelity polymerases are primarily responsible for misreplicating this oxidized lesion under cellular conditions. However, the kinetic data presented here suggests an alternative model in which the activity pol η plays a more important role in the pro-mutagenic behavior of 8-oxo-G. This model is predicated on the fact that high-fidelity DNA polymerases display remarkable low catalytic efficiencies for inserting either dAMP or dCMP opposite 8-oxo-G ($\sim 10^3$ M⁻¹s⁻¹) compared to normal DNA synthesis ($\sim 10^6 \text{ M}^{-1}\text{s}^{-1}$). As outlined in Figure 6, the reduced efficiency for replicating 8oxo-G would likely stall pol δ at the oxidized base, and this would facilitate polymerase switching to allow the specialized polymerase, pol n, to replace its more faithful counterpart. This is consistent with work published by Markkanen et al. (43) demonstrating that pol δ stalls at 8-oxo-G and is replaced with pol λ , another specialized DNA polymerase. While this group showed that pol λ promotes error free bypass of 8-oxo-G, the kinetic data presented here indicates that pol n can also by-pass 8-oxo-G as it is also more efficient at replicating the oxidized lesion than pol δ (compare $10^4 \text{ M}^{-1}\text{s}^{-1}$ versus $10^3 \text{ M}^{-1}\text{s}^{-1}$). Furthermore, the catalytic efficiencies measured for pol n inserting either dCMP or dAMP opposite 8-oxo-G are nearly identical, and this equality predicts that pol η will misreplicate 8-oxo-G with a relatively high frequency of $\sim 50\%$. This is a significantly higher frequency compared to pol δ which is predicted to correctly insert dCMP about 80% of the time. These differences in replication efficiencies and fidelity are predicted to favor the ability of pol η to replicate 8-oxo-G in an errorprone manner under cellular conditions causing an increase in mutagenic events associated with diseases such as cancer. Several recent reports support this potential mechanism. For example, higher pol n expression correlates with poor outcomes in patients with non-small-cell lung cancer (44).



Figure 5. (A) Protocols used to measure insertion and extension beyond various nucleotides opposite 8-oxo-G. Assays were performed using single turnover reaction conditions in which pol η (500 nM) and 13/200G-mer (250 nM) were pre-incubated and then mixed with a fixed concentration of nucleotide (dNTP) and 5 mM Mg²⁺. After four half-lives, an aliquot of dTTP and dGTP (200 μ M final concentration) was added to initiate the elongation reaction. Aliquots of the reactions were quenched with EDTA at variable times and analyzed by denaturing gel electrophoresis to assess elongation. (B) Representative gel electrophoresis data comparing the ability of pol η to extend beyond various mispairs containing 8-oxo-G. See text for further details.



Figure 6. Hypothetical model for the coordination of pol δ and pol η activity during the replication of damaged DNA. In this model, pol δ stalls upon encountering the oxidized DNA lesion. Stalling would facilitate polymerase switching and allow pol η to replace pol δ . pol η is more efficient at misinserting dAMP opposite 8-oxo-G, and this predicts an increase in promutagenic replication. In addition, certain modified nucleotides including N²-MedGTP are utilized more efficiently by pol η and are thus predicted to increase mutagenesis.

In particular, high expression of pol η is associated with shorter survival times in non-small cell lung cancer patients treated with platinum drugs (44). Finally, high levels of pol η correlate with resistance to chemotherapeutic agents such as cisplatin (45). In this case, drug resistance likely reflects the ability of pol η to incorporate nucleotides opposite and beyond the crosslinked lesion. In fact, *in vitro* studies have demonstrated that pol η can efficiently replicate cisplatinated DNA (46,47). Collectively, the results from our study demonstrate that pol η can efficiently replicate 8-oxo-G in an error-prone manner, and this offers new insights into how the activity of specialized DNA polymerases can contribute to cancer initiation and progression.

In addition, the kinetic data here show that certain modifications to the incoming nucleotide can significantly increase the frequency of mutagenic events catalyzed by pol η during TLS. The best example of this phenomenon is exhibited by N²-MedGTP which is inserted opposite 8-oxo-G with a remarkably high catalytic efficiency compared to natural dNTPs. In particular, N²-MedGTP is utilized ~10-fold more efficiently than either dATP or dCTP. Furthermore, N²-MedGTP is surprisingly selective for insertion opposite 8-oxo-G compared to undamaged G. This is best illustrated by examining the ability of pol η to extend beyond mispairs containing N²-MedGMP placed opposite 8-oxoG versus undamaged G (Figure 5B). As illustrated, pol η inserts and extends beyond N²-MedGTP at least 1000-fold better when replicating the oxidized lesion compared to unmodified G. Collectively, the high efficiency for insertion opposite and extension beyond damaged DNA predicts that N²-MedGTP is a highly pro-mutagenic nucleotide, especially when utilized by pol η during TLS. However, cell-based studies are clearly needed to verify that the kinetic phenomena described here can be recapitulated under cellular conditions.

Close inspection of the data summarized in Table 2 also points to an interesting trend in which nucleobase modifications have a more pronounced effect on the overall selectivity for incorporation opposite 8-oxo-G compared to G. For example, replacing a hydrogen bonding moiety such as the exocyclic amino group of dATP with a halogen (6-Cl-PTP) increases the efficiency of insertion opposite 8-oxo-G by \sim 8.5-fold compared to its insertion opposite G. Similar effects are observed with other analogs including N⁶-MedATP (2.2-fold), O⁶-MedGTP (17.2-fold) and N^2 -MedGTP (41-fold) as all of these modified nucleotides show higher selectivity for incorporation opposite damaged DNA compared to normal DNA. Again, this change in selectivity results from specific nucleobase modifications as the natural substrate, dATP displays no selectivity for replicating damaged DNA as the catalytic efficiencies for insertion opposite normal G and 8-oxo-G are identical. Similar effects are observed for the incorporation of dGTP opposite the natural or oxidized nucleobase. The provocative implication of these results is that pol η uses different molecular mechanisms to replicate normal and damaged DNA. This may occur by the ability of pol η to use different biophysical features such as hydrogen-bonding interactions, shape complementarity and nucleobase hydrophobicity as a consequence of the physical nature of the templating base, i.e. normal versus damaged DNA.

To interrogate this potential mechanism, we performed molecular modeling studies that compare the predicted interactions of the natural substrate, dATP, in the active site of pol n with two modified nucleotides, N²-MedGTP and 6-NITP. These particular nucleotides were chosen as they are incorporated opposite 8-oxo-G with higher catalytic efficiencies compared to dATP, and in the case of N^2 -MedGTP, with increased selectivity for insertion opposite 8-oxo-G. The model provided in Figure 7A shows that dAMP interacts with 8-oxo-G in the syn-conformation and is mediated almost exclusively via hydrogen bonding interactions. This mode of interaction is consistent with structural data obtained from both high fidelity and specialized DNA polymerases that show nearly identical interactions between dAMP and the oxidized nucleobase (17,19,22). The model provided in Figure 7B shows the structure of the active site of pol y containing N²-MedGMP paired opposite 8-oxo-G in the syn conformation. The major difference is that there are few, if any, direct hydrogen bonding interactions made between 8-oxo-G with the incoming alkylated nucleobase. In addition, the predicted interglycosyl distance between N²-MedGMP and 8-oxo-G is slightly longer compared to that predicted for dAMP paired opposite 8-oxo-G (compare 9.2 Å versus 8.9 Å, respectively). While both features predict that N²-MedGTP should be poorly incorporated opposite 8-oxo-G, our kinetic data demonstrate that the efficiency for utilizing N²-MedGTP is almost an order of magnitude higher than dATP utilization. We propose that this increase in efficiency reflects favorable entropic contributions made between the hydrophobic methyl group of N²-MedGTP with hydrophobic amino acids within the active site of pol η . Indeed, closer inspection of this model reveals the presence of a hydrophobic pocket that lines the back wall of the active site of pol n. Of particular importance are two isoleucine residues, Ile48 and Ile114, that exist within \sim 5 Å of the methyl group of N²-MedGTP and could interact with the hydrophobic methyl group of N²-MedGTP via van der Waals interactions. These entropic interactions may explain why the $K_{\rm m}$ of 0.64 μ M for N²-MedGTP is ~70-fold lower than the $K_{\rm m}$ of 47 μ M measured for dATP. The model provided in Figure 7C showing the molecular interactions of 6-NITP with 8-oxo-G is consistent with this mechanism. As expected for this non-natural nucleoside, there are no hydrogen bonding contacts made between 6-NIMP with 8oxo-G. However, the nitro group of 6-NIMP exists in a similar location as the methyl group of N²-MedGTP. In fact, the model in Figure 7C shows that this hydrophobic nitro group is close to Ile48 and Ile114, and the close proximity could allow for favorable interaction with these hydrophobic amino acids through van der Waals contacts. Again, these favorable entropic interactions may explain why the measured $K_{\rm m}$ of 3.9 μ M measured for 6-NITP is ~12-fold lower than that measured for dATP. It is interesting to note that the predicted distances between the nitro group of 6-NIMP and Ile48/Ile114 are longer than those measured between N^2 -MedGTP and these hydrophobic amino acids, and this increase in distance may explain the differences in measure $K_{\rm m}$ values between the two modified nucleotides. In addition, the predicted interglycosyl distance between 6-NIMP and 8-oxo-G is 10 Å and slightly longer than those for an dAMP:8-oxo-G or N²-Me-dGMP:8-oxo-G base pair. Thus,



Figure 7. Molecular models of 8-oxo-G paired opposite dAMP (panel A), N²-MedGMP (panel B), and 6-NIMP (panel C) in the active site of pol eta. These models were generated using molecular modeling programs, *MOE* and *Chimera*. The amino acid residues are color coded based on the Kyle Doolittle hydrophobicity scale, where blue denotes hydrophilic residues and red denotes hydrophobic residues. Ile114 and Ile48 are hydrophobic residues color coded in yellow for convenience. The close proximity of Ile114 and Ile48 (5–6 A⁰) to the hydrophobic methyl group of N²-MedGMP is proposed to play a key role in achieving an high catalytic efficiency $(1.7 \times 10^5 \text{ M}^{-1} \text{s}^{-1})$ for insertion opposite 8-oxo-G. Similarly, 6-NIMP with a nitro moiety at its sixth position is also inserted with a high efficiency $(7.26 \times 10^4 \text{ M}^{-1} \text{s}^{-1})$ opposite 8-oxo-G. This decrease could be attributed to a slight increase in distance between the nitro moiety and the isoleucine residues.

a lack of steric fit and shape complementarity might also explain the higher $K_{\rm m}$ value measure for 6-NITP compared to N²-MedGTP.

This new information provides additional insight into to the mechanism by which pol η selects nucleotides when replicating damaged DNA. Current structural data of specialized DNA polymerases have produced two mutually exclusive mechanisms to account for their activities in replicating various DNA lesions. Compared to high-fidelity DNA polymerases which possess more constrained active sites, the active sites of specialized DNA polymerases such as pol η are enlarged such that they can accommodate structurally diverse DNA lesions, many of which are significantly larger and bulkier compared to undamaged nucleobases (34). Based on this difference, models invoking steric fit and shape complementarity are often used to explain the high efficiency displayed by pol η when replicating damaged DNA as well as its error-prone behavior when replicating undamaged DNA. However, structural studies also highlight an important role for hydrogen-bonding interactions. For example, the structure of pol n bound to a cisplatinated DNA lesion shows that dCTP properly pairs with the first templating nucleobase via conventional hydrogen bonding interactions (48). This result suggest that enthalpic contributions associated with forming Watson-Crick hydrogen bonds between the incoming nucleotide and templating base are the primary driving force for nucleotide utilization. Our kinetic data support a model for enthlapic contributions as modifications to hydrogen bonding groups present on the incoming dNTP tend to generate negative effects on their utilization by pol n during TLS.

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